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Cyanobacterial Growth on Municipal Wastewater Requires Low Temperatures Travis C. Korosh^{a,b}, Andrew Dutcher^c, Brian F. Pfleger ^{a,d}, and Katherine D. McMahon ^{c,e} * Department of Chemical and Biological Engineering, University of Wisconsin-Madison, Madison, WI 53706, United States Environmental Chemistry and Technology Program, University of Wisconsin-Madison, Madison, WI 53706, United States Department of Civil and Environmental Engineering, University of Wisconsin-Madison, Madison, WI 53706, United States Microbiology Doctoral Training Program, University of Wisconsin-Madison, Madison, WI 53706, United States Department of Bacteriology, University of Wisconsin-Madison, Madison, WI 53706, United States * Corresponding author. 5552 Microbial Science Building, 1550 Linden Drive, Madison, WI 53706, United States. Phone: +1 608 890 2836. Fax: +1 608 262-9865. E-mail address: trina.mcmahon@wisc.edu. Keywords: Nutrient removal, cyanobacteria, Synechococcus strain PCC 7002

ABSTRACT

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Side-streams in wastewater treatment plants can serve as concentrated sources of nutrients (i.e. nitrogen and phosphorus) to support the growth of photosynthetic organisms that ultimately serve as feedstock for production of fuels and chemicals. However, other chemical characteristics of these streams may inhibit growth in unanticipated ways. Here, we evaluated the use of liquid recovered from municipal anaerobic digesters via gravity belt filtration as a nutrient source for growing the cyanobacterium Synechococcus sp. strain PCC 7002. The gravity belt filtrate (GBF) contained high levels of complex dissolved organic matter (DOM), which seemed to negatively influence cells. We investigated the impact of GBF on physiological parameters such as growth rate, membrane integrity, membrane composition, photosystem composition, and oxygen evolution from photosystem II. At 37°C, we observed an inverse correlation between GBF concentration and membrane integrity. Radical production was also detected upon exposure to GBF at 37°C. However, at 27°C the dose dependent relationship between GBF concentration and lack of membrane integrity was abolished. Immediate resuspension of strains in high doses of GBF showed markedly reduced oxygen evolution rates relative to the control. Together, this suggests that one mechanism responsible for GBF toxicity to Synechococcus is the interruption of photosynthetic electron flow and subsequent phenomena. We hypothesize this is likely due to the presence of a phenolic compounds within the DOM.

INTRODUCTION

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The need to develop non-petroleum based platforms for fuel and chemical production is driving researchers to explore alternatives that harness renewable energy sources while minimizing other environmental impacts such as freshwater depletion, eutrophication, and the use of arable land for non-food production. Cyanobacteria are particularly attractive such platforms due to their genetic tractability, rapid growth rates, halotolerance, and ability to be grown on non-productive land with simple nutrient requirements (1, 2). According to published life cycle assessments, a large portion of the associated costs of culturing photoautotrophs are tied to upstream costs, such as CO₂ delivery and fertilizer application (3). High phosphorous/nitrogen removal rates and energy efficiencies have been reported for photobioreactor and open pond cultivation systems using wastewater streams rich in nitrogen and phosphorus (4, 5). Therefore, it may be possible to offset the requirement for fertilizer by reclaiming nutrients from wastewater. This approach could yield the sought-after non-petroleum based alternative while also providing a more effective means of nutrient and metal removal than conventional wastewater treatment (6-8). Side-streams from common wastewater treatment facilities such as supernatants or filtrates from solids-separation processes are particularly promising nutrient sources, assuming that cyanobacterial strains can efficiently use them.

Of the many streams available in common wastewater facilities, the liquid fraction of anaerobic digestate is thought to be the most attractive nutrient source (8–11). Although digestate is rich in the inorganic constituents necessary for growth, it also contains dissolved organic matter (DOM) that has been shown to limit photosynthetic activity (12). DOM is a heterogenous mixture of aliphatic and aromatic compounds derived from the decomposition of living organisms (13). The chemical nature of wastewater-derived DOM is largely governed by the

type of treated waste and the treatment process, but it is largely comprised of hydrophilic, fulvic, and humic substances (14, 15). Humics can induce damaging permeability in model and bacterial membranes (16, 17). Various studies have also demonstrated that fulvic and humic acids can enhance the solubility of many organic compounds (18), which in turn would augment their bioavailability and potential toxicity. Many of these compounds are also photo-reactive, producing toxic hydrogen peroxide and hydroxyl radicals (19, 20), which is of significant concern for phototroph cultivation.

The mode of DOM toxicity is thought to involve interactions with the protein-pigment complex of photosystem II (PSII) in photosynthetic organisms, although the exact molecular mechanism remains unclear (21). When the rate of light induced damage to PSII exceeds its rate of repair, growth suppression and chlorosis result from the phenomena known as photoinhibition (22). When damage by photoinhibition is sufficient to hamper the natural ability to consume electrons generated by photosynthesis, reactive oxygen species (ROS) are concomitantly produced as an undesired by-product. Prolonged oxidative stress halts protein translation through oxidation of specific cysteine residues in the ribosomal elongation factors (23). Given these findings, it is increasingly evident that under conditions of sustained stress, regulation of electron flow is critical to maintain homeostasis in photosynthetic organisms (24, 25). Thus, it is important to understand the mechanisms by which DOM may be interrupting electron flow in order to capitalize on the potential of cyanobacteria to remediate wastewater and generate high-value chemicals.

In this study, we tested the practicality of using combined streams from a municipal wastewater plant as a nutrient source for cyanobacterial cultivation. We used the euryhaline cyanobacteria *Synechococcus* sp. strain PCC 7002 (PCC 7002) due to its exceptional tolerance to

high-light intensity, salt, and other environmental stresses (26, 27). Initial attempts to grow PCC 7002 in this nutrient source under standard environmental conditions of 1% (v/v) CO₂, a temperature of 37°C, and illumination of 200 µmol photons m⁻²s⁻¹ resulted in photobleached (white-yellow) cultures. In an effort to explain this observation while developing more feasible cultivation conditions, we assessed the effects of many environmental parameters during PCC 7002 cultivation in wastewater-based media by monitoring changes in growth rate, photopigment abundance, oxygen evolution rates, membrane integrity, and membrane composition. We observed a concentration and temperature dependent effect of GBF towards membrane and photosystem degradation. High GBF concentrations were associated with elevated DOM levels, and produced ROS in 37°C grown cultures. At 27°C, cultures adapted to growth on GBF had elevated levels of total fatty acids, high unsaturated fatty acid content, and elevated membrane integrity. Decreased photosynthetic oxygen production rates were noted upon exposure to high levels of GBF. This suggests bioavailability of the photoinhibitory compound in GBF is governed by changes in membrane content and composition that occur during growth at relatively high temperatures.

MATERIALS AND METHODS

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Media and Growth Conditions

Synechococcus sp. strain PCC 7002 was obtained from the Pasteur Culture Collection of Cyanobacteria. Experiments were performed with a strain of PCC 7002 harboring a gentamicin resistance marker in the A2842 locus to maintain axenic cultures. Strains were grown and maintained on Medium A⁺ (28) with 1.5% Bacto-Agar with gentamicin (30 µg/mL). Strains were

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cultured in 250 ml baffled flasks with 50 mL Medium A^+ with 5 μM NiSO₄ (29) with 1% CO₂-enriched air at 150 rpm in a Kuhner ISF1-X orbital shaker. Temperature was maintained at 37°C or 27°C and light intensity was fixed at approximately 200 μ mol photons m⁻²s⁻¹ or 100 μ mol photons m⁻²s⁻¹ via a custom LED panel. Strains were pre-acclimated to the culture conditions overnight before inoculating in fresh media. Optical density at 730 nm was measured in a Tecan M1000 plate reader.

Wastewater-derived media was obtained from the Nine Springs Wastewater Treatment Plant (Dane County, Wisconsin, USA). The plant is configured for biological nutrient removal via a modified University of Cape Town process with no internal nitrate recycling and stable performance yielding high secondary effluent quality (total phosphorus < 1 mg P/L, ammonia < 1 mg N/L, nitrate ~ 15 mg N/L) (30). Anaerobic digesters are used for solids stabilization and the resulting digested material is passed over a gravity belt filter for dewatering. The system includes an Ostara WASSTRIP process to recover phosphorus. This filtrate (GBF) served as the primary source of phosphorus and reduced nitrogen for our cultures, and effluent from the postmainstream secondary treatment clarifier (secondary effluent) served as a diluent. GBF was filtered through a paper filter to remove any exceptionally large flocs, then stored at -80°C until use. Secondary effluent was collected one to four days before each experiment and held under refrigeration at approximately 2°C. Experimental media was comprised primarily of secondary clarifier effluent and GBF, combined in different proportions. Unless otherwise noted, GBF was used at a concentration of 12.5% (v/v) in secondary effluent. To meet complete nutrient requirements, the diluted GBF media was supplemented with trace metals and vitamin B12 at the concentrations found in Medium A⁺, as well as KH₂PO₄ at a molar ratio of 1:32 soluble reactive phosphorus (SRP) to bioavailable nitrogen (the sum of NH₄⁺ and NO₃⁻). All media was buffered

with Tris-HCl and adjusted to pH 8.0 with potassium hydroxide or hydrochloric acid before sterilization by autoclaving.

Staining, Flow Cytometry, and Fluorescence Measurements

The membrane integrity of cyanobacteria cells was recorded by a flow cytometer (FACSCalibur, BD Biosciences, San Jose, CA, USA). After growth in the respective media, cells were centrifuged (2 minutes, 5000 RCF), decanted, and resuspended in 1 mL of Tris-Buffered Saline (TBS) solution (pH 8.0). To identify membrane-compromised cells, SYTOX Green (Life Technologies) was also added to each sample (1 µM). SYTO 59 (Life Technologies) was added to each sample (1 µM) as a nucleic acid counterstain. As a control for a permeabilized membrane, cells were resuspended in 190 proof ethanol. SYTOX green fluorescence was visualized using 488 nm laser excitation and emission area was read using a 530/30 nm bandpass filter. The 633 nm laser coupled with a 661/16 bandpass filter was used for SYTO 59 visualization. Analysis of the cytometric data was carried out with CellQuest Pro (BDBiosciences, San Jose, CA, USA) software. To assess reactive oxygen species production, cells $(OD_{730} = 1)$ were incubated overnight in Medium A, GBF, GBF + 1 mM Dithiothreitol (DTT), GBF + 1 mM N-acetylcysteine (NAC), or Medium A + 100 μM methyl viologen as a positive control at 37°C with 5% CO₂ at a light intensity of 200 µmol photons m⁻²s⁻¹. Cells were washed in TBS, and either Sytox Green (1 µM) or CellROX Orange reagent (Life Technologies) (5 µM) was added. After 30 min incubation in

darkness at 37°C, fluorescence was measured (Ex/Em 545/565 nm for Cell ROX Orange) and

(Ex/Em 504/523 nm for Sytox Green) in a Tecan M1000 plate reader.

GBF Characterization

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SRP, ammonia, nitrate, and nitrite concentrations were determined for all secondary clarifier effluent and GBF samples used in these experiments. In addition, the GBF was tested for total suspended solids (TSS), volatile suspended solids (VSS) total solids (TS), and chemical oxygen demand (COD). Ammonia, SRP, and COD were determined by colorimetric tests using reagents from Hach. Nitrate and nitrite were determined using high performance liquid chromatography (Shimadzu) with a C18 column and photodiode array detector (31). TSS, VSS, and TS were measured according to Standard Method 2540 D, 2540 E, and 2540 B, respectively, with 47mm diameter glass fiber filters (Whatman) used for TSS and VSS (32). Fluorescence EEM measurements were conducted using a M1000 Tecan plate reader using a UV-transparent plate (Costar 3635). Fluorescence intensity was normalized to quinine sulfate units (QSU), where 1 QSU is the maximum fluorescence intensity of 1 ppm of quinine sulfate in 0.1 N H₂SO₄ at Ex/Em = 350/450. Rayleigh scatter effects were removed from the data set.

Biochemical analyses

Batch cultures were further assayed for dry cell weight (DCW), fatty acid content, oxygen evolution rates, and chlorophyll a content. Cells were concentrated by centrifugation, washed in TBS, and lyophilized overnight to obtain dry cell weights (DCW). Fatty acids from approximately 10 mg of DCW with 10 mg ml⁻¹ pentadecanoic acid as an internal standard were converted to methyl-esters, extracted with n-hexane and analyzed by GC-FID on a Restek Stabilwax column (60m, 0.53 mm ID, 0.50 μ m) (33). Photosynthetic oxygen evolution from whole cells was measured with a Unisense MicroOptode oxygen electrode with 10 mM NaHCO₃ illuminated with a slide projector at PPFD from 76-2700 μ mol m⁻² s⁻¹ for 30 min at room temperature (34). Cells were collected by centrifugation and resuspended in the appropriate media to give an OD₇₃₀ = 1.0. Chlorophyll a measurements were done via a 100% chilled

methanol extraction procedure (35). Chlorophyll a was calculated via the following equation:

$$Chl_a = 16.29 * A^{665} - 8.54 * A^{652}$$
 (36).

RESULTS

GBF and secondary effluent characteristics

We measured nutrient concentrations from the batches of GBF collected over the 6 month experimental period (Table 1 and Table 2), to ascertain if it was a stable and reliable nutrient source for cultivating the cyanobacteria. Sampling points from the Nine Springs Wastewater Treatment Plant (Dane County, Wisconsin, USA) are shown in Figure 1. In order to calculate nutrient stoichiometries, we took the sum of NH₃-N and NO₃-N to be the bioavailable N. Nutrient levels in the GBF were markedly more variable than in the secondary effluent. The average molar ratio of bioavailable N to SRP was 35 ± 7 in GBF diluted with secondary effluent (12.5% v/v), as compared to 32 in Medium A⁺.

We also measured DOM quality in the secondary effluent and GBF using excitation–emission matrix (EEM) fluorescence spectroscopy (37) because we hypothesized that DOM was linked to toxicity during cultivation. Secondary effluent contained diffuse constituents, including humic [excitation wavelengths (>280 nm) and emission wavelengths (>380 nm)] and fulvic acid-like [excitation wavelengths (<250 nm) and emission wavelengths (>350 nm)] spectra relative to Medium A⁺ (Figure 2). The distinction between the two substances is historically based on solubility (38), but compositionally, fulvic acids contain more acidic functional groups than humic acids (39). The fulvic acid content in media preparations rose with increasing concentrations of GBF. Additionally, at high concentrations of GBF, a distinct region [excitation wavelengths (270 -290 nm) and emission wavelengths (340-400 nm)] was attributed

to the presence of soluble microbial products, which include aromatic amino acids, carbohydrates, or phenols (37, 40). The exact composition of these products varies with plant configuration, but they are refractory to microbial degradation (41).

Exposure to GBF at high temperatures generates radicals

Preliminary growth experiments using GBF as a nutrient source were unsuccessful under standard cultivation conditions, due to cell photobleaching. We hypothesized that DOM in the media was in some way compromising membrane integrity, possibly via intracellular production of reactive oxygen species (ROS). Thus, we measured the ROS content and loss of membrane integrity in response to overnight GBF exposure at 37°C (Figure 3). We also examined the capacity of reducing agents Dithiothreitol (DTT) or N-acetylcysteine (NAC) to quench the media toxicity, since they have anti-oxidant properties due to the direct reduction of disulfide bonds or as precursors for the anti-oxidant glutathione (42). To measure their effect on the ROS production and culture viability after GBF exposure, we assessed ROS content and membrane integrity with either concurrent addition or preincubation of these quenching compounds in the diluted (12.5% v/v) GBF media. Addition of 100 μM methyl viologen to Medium A⁺ served as a positive control. Exposure of cells to 12.5%-GBF media resulted in marked ROS production and membrane permeability, which no thiol treatment alleviated.

Dose-dependent toxicity of GBF is a function of growth temperature

To evaluate the effects of GBF dosage on PCC 7002 physiology, we measured growth rates in GBF concentrations ranging from 6.25%-12.5% (v/v) as a function of temperature (27°C vs 37°C) and light intensity (100 μmol m⁻² s⁻¹ vs 200 μmol m⁻² s⁻¹) (Table 3). Medium A⁺ served as a control. Higher temperatures depressed growth rates in GBF-based media. At 37°C,

growth rates with 6.25% GBF at both 100 μ mol m⁻² s⁻¹ and 200 μ mol m⁻² s⁻¹ were most comparable to Medium A⁺. Higher GBF concentrations had a more extreme effect on growth rates. However, this dose dependent effect of GBF on growth rate was abolished when the cultivation temperature was lowered to 27°C. At 200 μ mol m⁻² s⁻¹ and 27°C, GBF cultures grew twice as slowly as the control and there was no significant difference between the tested GBF concentrations. Under 100 μ mol m⁻² s⁻¹ and 27°C, growth rates were comparable across media conditions. Thus, successful cultivation using the more concentrated GBF media required adjusting both the light and temperature regimes.

Photobleaching is a common symptom of oxidative stress in photosynthetic organisms and is caused by the accumulation of ROS (43). To investigate the effects of GBF media on the photosynthetic pigmentation, we also performed whole cell absorbance scans at two different temperatures (27°C vs 37°C). Light intensity was held constant at 200 µmol m⁻² s⁻¹. High GBF concentrations yielded enhanced chlorophyll, phycobilisome, and carotenoid degradation at 37°C (Figure 4a). At 27°C, photosynthetic pigments maintained intact relative to the control, regardless of GBF concentration, implying less ROS production at this temperature (Figure 4b).

We wondered whether the observed toxicity was imposed on cells early or late during the 72-hour cultivation period. To track the dynamics of GBF induced toxicity, we employed forward scatter flow cytometry using SYTO 59 as a counterstain to identify cells, which were subsequently visualized for membrane integrity using Sytox Green. Two distinct phases were identified upon exposure to GBF, which we interpreted as initial and chronic toxicity (Figure 4). Initial toxicity was defined as the change in Sytox Green positive events for samples analyzed during exponential growth, while chronic toxicity accounted for the change in Sytox Green events during linear growth. As was the case for 72-hour based growth rate, a relationship

between both initial and chronic toxicity and increasing GBF concentrations was found at 37°C (Figure 5a). While we still detected considerable initial toxicity with GBF exposure at 27°C, there was less of an effect of dosage on chronic toxicity (Figure 5b). Altogether, this suggested that there was a temperature and cell-concentration dependent adaptation that ameliorated the susceptibility of cultures to GBF induced toxicity.

Exposure to GBF retards oxygen evolution

To better delineate the cause of initial toxicity associated with high GBF concentrations, we measured maximal oxygen evolution rates for strains briefly exposed to GBF media while increasing the light intensity. Measurements of photosynthetic oxygen evolution would allow for an indirect assessment of PSII activity and electron transfer. Cultures were grown to early linear phase in Medium A^+ or GBF media, washed, and resuspended in the appropriate media. Resuspension media was saturated with HCO_3^- (10 mM) in order to prevent inorganic carbon limitation. As expected, cells grown and assayed in Medium A^+ at 37°C showed a clear increase in O_2 evolution rate as the light intensity approached saturation at 2700 μ mol m⁻² s⁻¹, reaching a maximal rate of 240 \pm 8 μ mol O_2 (mg Chl a)⁻¹ h⁻¹ (Figure 6a). Cells grown in Medium A^+ at 37°C but assayed in 12.5% GBF had diminished O_2 evolution rates at all light intensities, plateauing with a rate of 79 \pm 6 μ mol O_2 (mg Chl a)⁻¹ h⁻¹ at an intensity of 270 μ mol m⁻² s⁻¹ (Figure 6a). Thus, exposure to GBF under these conditions immediately caused a decrease in O_2 production.

Next, we examined the effect of temperature in a similar experiment. Assays carried out

in Medium A^+ after growth in Medium A^+ at 27°C showed much higher maximal O_2 evolution rates at all tested light intensities than with cells grown at 37°C (Figure 6b), peaking at a rate 556 \pm 102 μ mol O_2 (mg Chl a)⁻¹ h⁻¹). This was expected because elevated O_2 evolution rates in low

temperature grown cells have been previously reported and were attributed to a substantial change in photosystem stoichiometry (44). The O_2 evolution rates of cells grown in Medium A^+ at 27°C and then resuspended in 12.5% GBF stayed relatively constant at all of the tested intensities and were roughly 5-fold lower than in controls, with a maximal rate of $109 \pm 22 \,\mu$ mol O_2 (mg Chl a)⁻¹ h⁻¹ at an intensity of 75 μ mol m⁻² s⁻¹. We compared the above rates to those from cultures grown in 12.5% GBF at 27°C to test if adaptation to GBF was met with changes in photosynthetic activity. At tested light intensities, O_2 evolution rates with 27°C GBF-adapted cultures were not statistically different than with unadapted cultures. Finally, we conducted the inverse experiment, using cultures grown in GBF media at 27°C but assayed in Medium A⁺ under saturating light. Interestingly, they displayed the highest evolution rate of any tested condition, at 944 \pm 96 μ mol O_2 (mg Chl a)⁻¹ h⁻¹ (Figure 6b). This suggested that there is a period of dynamic photosynthetic adaptation to overcome the stress of GBF, and that when the stress is alleviated the cells have an enhanced capacity for photosynthetic activity.

Acclimation to GBF changes lipid content and composition

Based on the results described above, we hypothesized that the temperature dependent photosynthetic adaptation is related to changes in thylakoid membrane content and composition. We extracted total fatty acids of cultures grown at 27°C for 72 hours in 12.5% GBF or Medium A^+ , and analyzed the content after derivatization. We could detect and resolve all major saturated and unsaturated fatty acid species (Table 4). Cultures grown in GBF had greater totals of assayed fatty acid species (27 \pm 7 mg FAME gDCW⁻¹) when compared to cells grown in Medium A^+ (15 \pm 1 mg FAME gDCW⁻¹) (Table 4). The most abundant fatty acid in all samples was 16:0 (42-45% of the total fatty acids). C18:2 Δ 9,12 fatty acids comprised a significant fraction of Medium

 A^+ grown cultures with 22% of the total fatty acid species. However, in GBF-grown cells the amount of C18:2 Δ 9,12 fatty acids was only 15%, while C18:3 Δ 9,12,15 fatty acids were twice as high in GBF grown cells (16%) than in Medium A^+ grown cells (9%) (Table 4). This suggests that cells were altering their membrane composition when grown in GBF, as compared to standard growth in Medium A^+ .

DISCUSSION

The amount and distribution of arable land and potable water are projected to change over the next several decades due to climate change, while the rise in global population and standard of living are expected to increase demands (45). Integration of microalgal cultivation with industrial and municipal wastewater treatment circumvents many of the resource concerns raised over biofuel production (2), while simultaneously removing additional nutrients and pollutants present in the wastewater (46). However, under standard environmental conditions we were unable to obtain robust growth of PCC 7002 using a diluted municipal side stream as a nutrient source. We hypothesized that this effect may be due to the presence of DOM, which has been demonstrated to cause a decrease in photosynthetic performance in various strains of cyanobacteria (47–49). We investigated the effects of light intensity and temperature on the physiology of digestate grown cultures, in an effort to find conditions conducive to biomass generation and better understand the mechanisms of DOM toxicity.

We found that cultivation temperature was an important factor that allowed for the growth of PCC 7002 under high GBF concentrations. Numerous physiological processes are altered at low temperatures (50). Notably, the fatty acyl chains in the both the cytoplasmic and thylakoid membrane undergo a transition from a fluid to nonfluid state (51). The immediate

cessation of oxygen evolution (Figure 6) and high rates of initial toxicity (Figure 5) upon GBF exposure with cells cultivated at tested temperatures suggests that the toxic compound rapidly crosses bacterial membranes. We propose that due to the decreased fluidity of the outer membrane at 27°C relative to 37°C, accessibility of humic and fulvic acids to the hydrophobic domains in the phospholipid bylayer is decreased, lessening the intracellular transport of the toxic compound(s) and initial and chronic toxicity.

Upon a shift to a lower temperature, cyanobacteria also alter the expression of desaturases to increase the unsaturated fatty acid content and maintain optimal membrane function (52). Optimal fluidity of thylakoid membranes is a critical factor in photosynthetic electron transport, due to the mobility of co-utilized redox components to both photosynthetic and respiratory complexes to ensure ideal electron flow (53, 54). It has been shown that temperature influences the kinetics governing the redox state of plastiquinone (PQ) through alterations in thylakoid membrane composition and fluidity (55). We believe that this temperature dependent alteration of the thylakoid membrane to circumvent GBF induced changes in the redox state is also an important component of the adaptive response, given the increase in unsaturated fatty acids observed during cultivation in GBF-based media (Table 4).

We propose that the herbicidal effect of GBF is partially due to the electron scavenging properties of phenolic moieties within in the DOM, given the symptoms of chlorophyll-bleaching as shown in Figure 3 and pronounced ROS production as evidenced by Figure 2 (56). These phenolic moieties are likely a component of the "soluble microbial products" found in our EEM scans (Figure 2). At low concentrations, phenolic photosynthetic electron transfer inhibitors such as 2,5-Dibromo-3-methyl-6-isopropyl-p-benzoquinone (DBMIB), alter the redox potential of the PQ pool of PSII by blocking forward electron transfer to the cytochrome b₆/f

complex (57). DBMIB treatment has also been shown to substantially decrease inner mitochondrial membrane fluidity (58). This strong reduction of the PQ pool by DBMIB induces the phycobilisomes to physically move from PSII to PSI in a transition known as "state 2", thereby decreasing the ratio of reducing power to proton-motive force generated by photosynthesis (59). Marine *Synechococcus*, including PCC 7002, have also been shown to transition to state 2 upon shifts to lower temperature (60). Chronic exposure to phenolic herbicides eventually leads to radical-catalyzed back reactions that trigger the formation of ROS that facilitate complex destruction and cell death (43, 61–63). Some phenolic herbicides may also act as arylating agents, causing covalent binding to macromolecules via Michael addition and a depletion of thiol pools (64). The inability of the reducing agents we tested to maintain membrane integrity suggests that GBF-induced cytotoxicity is likely caused by redox cycling, and not arylation. Future efforts to increase tolerance to these toxicants might include disruption of the inhibitor binding sites (65, 66) or optimizing membrane fluidity (67).

CONCLUSIONS

Under standard cultivation conditions of 37°C with the cyanobacteria *Synechococcus* sp. strain PCC 7002, there was a dose-dependent relationship between liquid anaerobic digestate concentration and toxicity. This was met with ROS production and photopigment degradation. Digestate contained constituents of dissolved organic matter that were likely affecting photosynthetic electron transport. Decreasing the cultivation temperature to 27°C enabled robust cultivation at high digestate concentrations, resulting in high biomass productivities. This temperature dependent tolerance may be due to changes in membrane fluidity. Our study highlights the contributions of dissolved organic matter to photosynthetic growth and physiology in wastewater based media, as well as a potential mechanism for tolerance in low temperatures.

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FIGURE LEGENDS

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Figure 1. Flow diagram and nutrient streams obtained from the Nine Springs Wastewater

Treatment Plant (Dane County, Wisconsin, USA). Stars indicate sampling points.

Figure 2. Excitation Emission Matrix of wastewater medias. Fluorescence values were

normalized to 1 ppm of quinine sulfate in 0.1 N H_2SO_4 at Ex/Em = 350/450 nm.

Figure 3. Reactive oxygen species and membrane integrity assay. Cells were grown to early

linear phase in Medium A⁺ or 12.5% GBF with the appropriate treatment under continuous

illumination (200 µE m⁻² s⁻¹) with 1 % CO₂ at 37°C. Fluorescence values were normalized to

 OD_{730} . The values represent the mean \pm SD of biological triplicates.

Figure 4. Absorption spectra of GBF exposed strains as a function of temperature. Cultures

were grown in Medium A⁺, 6.25 %, or 12.5 % (v/v) GBF media under continuous illumination

(200 µE m⁻² s⁻¹) with 1 % CO₂ at 37°C (A) or 27°C (B) for 72 hours. The spectra were recorded

in dilute cell suspensions and normalized to an OD₇₃₀. The peak at 637 nm is due to

phycobilisomes, the peak at 683 nm is due to chlorophyll a, and the peak at 438 nm is due to

carotenoids.

Figure 5. Initial and chronic toxicity of cultures grown in Medium A⁺, 6.25 %, 9.9 %, or 12.5 %

(v/v) GBF media under continuous illumination (200 µE m⁻² s⁻¹) with 1 % CO₂ at 37°C (A) or

27°C (B) for 72 hours. Initial toxicity was defined as the change in Sytox Green positive events for samples analyzed during exponential growth. Chronic toxicity accounted for the change in Sytox Green events during linear growth. A measure of < 0 chronic toxicity means that the samples recovered following a period of initial toxicity.

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Figure 6. Rates of oxygen evolution as a function of acclimation temperature and light intensity. Cells were grown to early linear phase in Medium A^+ or 12.5% GBF under continuous illumination (200 μ E m⁻² s⁻¹) with 1 % CO₂ at 37°C (A) or 27°C (B). Cells were pelleted, resuspended in Medium A^+ or 12.5% GBF, and the rate of maximal oxygen evolution was measured with 10 mM HCO₃⁻ as an electron acceptor at increasing light intensities. The values represent the mean \pm SE of biological duplicates.

TABLES

Table 1. Nutrient composition of batch of 100% GBF used for subsequent experiments

Analyte	Concentration ^a (mg L ⁻¹) (±SD)			
NH ₃ -N	1180 ± 135			
NO_3 – N	7.5 ± 0.04			
SRP	78 ± 10			
COD	735 ± 4			
TSS	467 ± 123			
VSS	18 ± 2			
TS	2350 ± 36			
TS (Glass Fiber Filtered)	2030 ± 42			
TS (0.45 um membrane filtered)	2000 ± 151			

^aValues shown represent the mean \pm standard deviation (SD) of at least three technical replicates.

Table 2. Characteristics of GBF and secondary effluent over the 6 month experimental period.

Source	Analyte	Concentration ^a (mg L ⁻¹) (±COV)		
	NH ₃ -N	920 ± 22%		
GBF	NO_3-N	$3.9 \pm 43\%$		
	SRP	$54 \pm 27\%$		
	NH ₃ -N	n.d.		
Secondary Effluent	NO_3-N	$19.3 \pm 6.6\%$		
27.1	SRP	n.d.		

^aValues shown represent the mean \pm coefficient of variation (COV) of at least three technical replicates.

Table 3. Growth Rates with varying Temperature and Light Intensity

Media	Light Intensity	Temperature	Growth Rate (±SD)		
	$(\mu E m^{-2} s^{-1})$	(°C)	(OD day ⁻¹)		
A^+	200	37	0.66 ± 0.04		
6.25% GBF	200	37	0.58 ± 0.05		
12.5% GBF	200	37	0.05 ± 0.01		
A^{+}	100	37	0.28 ± 0.01		
6.25% GBF	100	37	0.24 ± 0.01		
9.9% GBF	100	37	0.11 ± 0.02		
12.5% GBF	100	37	0.02 ± 0.00		
A^{+}	200	27	0.73 ± 0.08		
6.25% GBF	200	27	0.32 ± 0.03		
9.9% GBF	200	27	0.42 ± 0.06		
12.5% GBF	200	27	0.28 ± 0.04		
A^{+}	100	27	0.25 ± 0.01		
6.25% GBF	100	27	0.26 ± 0.02		
9.9% GBF	100	27	0.24 ± 0.01		
12.5% GBF	100	27	0.26 ± 0.02		

^a Linear growth rates of cultures grown in Medium A⁺ or GBF with the appropriate treatment under continuous illumination (200 μ E m⁻² s⁻¹ or 100 μ E m⁻² s⁻¹) with 1 % CO₂ at 37°C or 27°C. The values represent the mean \pm SD of biological triplicates.

Table 4. Fatty acid content and composition of cultures grown in Medium A^+ or 12.5% GBF media under continuous illumination (200 $\mu E m^{-2} s^{-1}$) at 27°C for 72 hours

		Fatty Acids Species					
Growth Conditions ¹	- !	C16:0	C16:1 Δ9	C18:0	C18:1 \(\Delta 9	C18:2 Δ9,12	C18:3 Δ9,12,1
Medium A ⁺	mg FA gDCW⁻¹	6.7 ± 0.7	2.0 ± 0.1	0.2 ± 0.1	1.5 ± 0.2	3.3 ± 0.2	1.4 ± 0.
	% FA	44 ± 1	13 ± 1	2 ± 0	10 ± 0	22 ± 1	9 ± 1
GBF	mg FA gDCW⁻¹	11.6 ± 3.1	3.5 ± 0.8	1.3 ± 0.3	2.2 ± 0.7	3.9 ± 0.8	4.4 ± 1.
	% FA	43 ± 0	13 ± 0	5 ± 0	8 ± 0	15 ± 1	16 ± 1

¹ Strains were grown in the Medium A⁺ or 12.5% GBF media under continuous illumination (200 $\mu E m^{-2} s^{-1}$) at 27°C for 72 hours. Fatty acids were extracted and derivatized as previously described. The values represent the mean \pm SD of two independent experiments.

FIGURES

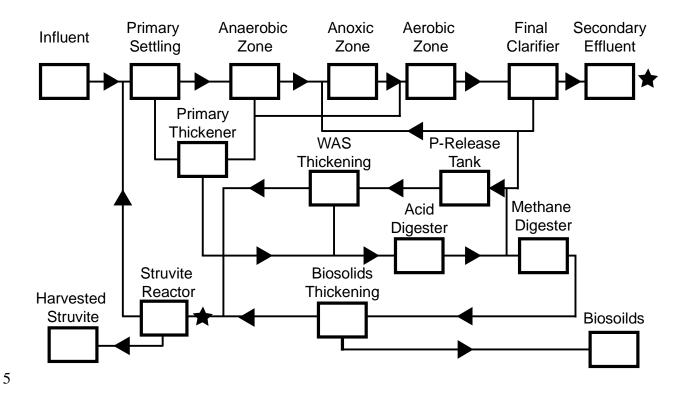


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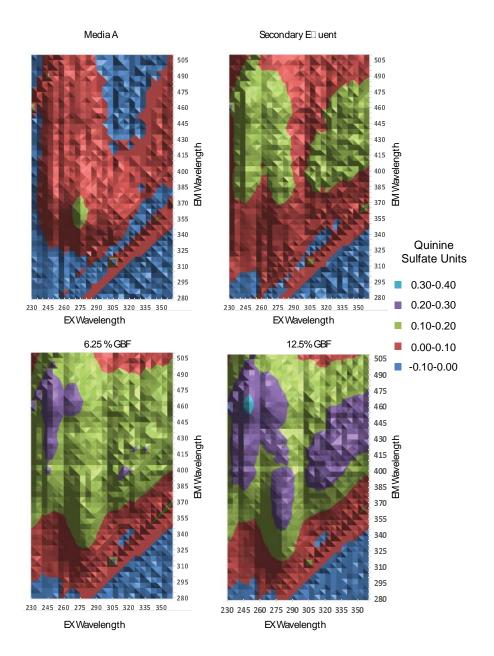


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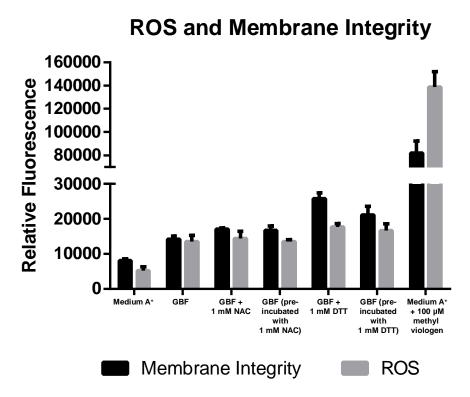


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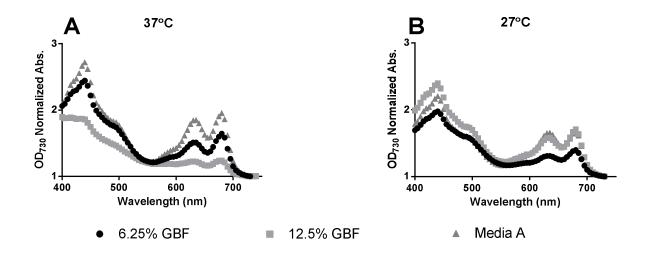


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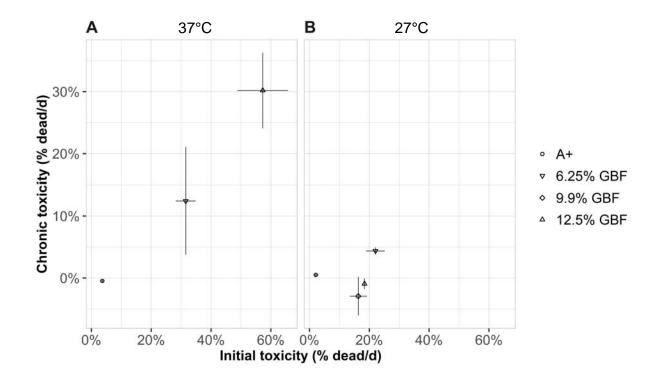


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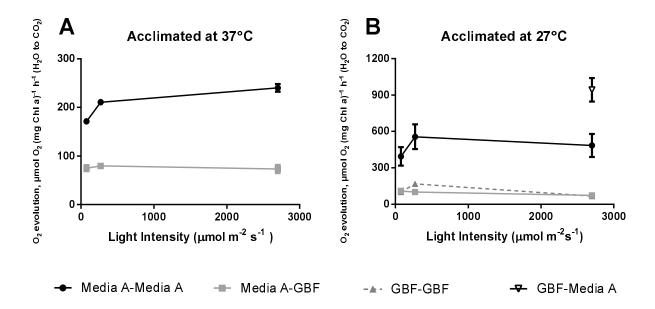


Figure 6. Rates of oxygen evolution as a function of acclimation temperature and light intensity. Cells were grown to early linear phase in Medium A^+ or 12.5% GBF under continuous illumination (200 μE m⁻² s⁻¹) with 1 % CO₂ at 37°C (A) or 27°C (B). Cells were pelleted, resuspended in Medium A^+ or 12.5% GBF, and the rate of maximal oxygen evolution was measured with 10 mM HCO₃⁻ as an electron acceptor at increasing light intensities. The values represent the mean \pm SE of biological duplicates.