

A High-Throughput Extraction and Analysis Method for Steroidal Glycoalkaloids in Tomato

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Abstract

Tomato steroidal glycoalkaloids (tSGAs) are a class of cholesterol-derived metabolites uniquely produced by the tomato clade. These compounds provide protection against biotic stress due to their fungicidal and insecticidal properties. Although commonly reported as being anti-nutritional, both *in vitro* as well as pre-clinical animal studies have indicated that some tSGAs may have a beneficial impact on human health. However, the paucity of quantitative extraction and analysis methods presents a major obstacle for determining the biological and nutritional functions of tSGAs. To address this problem, we developed and validated the first comprehensive extraction and UHPLC-MS/MS quantification method for tSGAs. Our extraction method allows for up to 16 samples to be extracted simultaneously in 20 minutes with $93.0 \pm 6.8\%$ and $100.8 \pm 13.1\%$ recovery rates for tomatidine and alpha-tomatine, respectively. Our ultra-high-performance liquid chromatography tandem mass spectrometry (UHPLC-MS/MS) method was able to chromatographically separate analytes derived from 16 tSGAs representing 9 different tSGA masses, as well as two internal standards, in 13 minutes. Tomato steroidal glycoalkaloids that did not have available standards were annotated using high resolution mass spectrometry as well as product ion scans that provided fragmentation data. Lastly, we utilized our method to survey a variety of commonly consumed tomato-based products. Total tSGA concentrations ranged from 0.7 to 3.4 mg/serving and represent some of the first reported tSGA concentrations in tomato-based products. Our validation studies indicate that our method is sensitive, robust, and able to be used for a variety of applications where concentrations of biologically relevant tSGAs need to be quantified.

41 **1 Introduction**

42 Solanaceous plants produce a spectrum of cholesterol derived compounds called steroidal
43 glycoalkaloids. While each solanaceous clade produces its own unique assortment of steroidal
44 glycoalkaloids, these metabolites share commonality in their role as phytoanticipins and anti-
45 herbivory agents (Etalo et al., 2015; Fontaine et al., 1948; Irving et al., 1945; Ökmen et al., 2013).
46 Tomato (*Solanum lycopersicum* and close relatives) is no exception, and over 100 tomato steroidal
47 glycoalkaloids (tSGAs, Fig. 1) have been suggested (Iijima et al., 2013, 2008). Although these
48 compounds are typically reported as anti-nutritional (Ballester et al., 2016; Cárdenas et al., 2016,
49 2015; Itkin et al., 2013), other studies suggest a health-promoting role. In fact, emerging evidence
50 suggests that some tSGAs may play a role in positive health outcomes associated with tomato
51 consumption (Cayen, 1971; Choi et al., 2012; Cooperstone et al., 2017; Lee et al., 2004). While
52 these compounds continue to be evaluated both *in planta* and *in vivo*, there is a lack of quantitative
53 and validated methods to extract and measure tSGAs from tomatoes; a critical need for additional
54 research in this area.

55 Tomato steroidal glycoalkaloids are typically extracted by grinding individual samples
56 using a mortar and pestle, or blender and then solubilizing analytes with polar solvent systems,
57 typically methanol. This approach is time consuming because each sample is handled individually.
58 Additionally, this technique has been used for relative profiling, and has not been evaluated for its
59 ability to extract tSGAs quantitatively. Tomato steroidal glycoalkaloids such as alpha-tomatine,
60 have previously been quantified using gas and liquid chromatography (Kozukue and Friedman,
61 2003; Lawson et al., 1992; Rick et al., 1994), as well as a number of bioassays including cellular
62 agglutination (Schlösser and Gottlieb, 1966) and radioligand assays using radioactive cholesterol
63 (Eltayeb and Roddick, 1984). These methods are unreliable, suffer from poor sensitivity, have
64 poor selectivity for different alkaloids, and are time consuming. Recent advances in analytical
65 chemistry have enabled researchers to discover other tSGA species in tomato fruits using high
66 resolution mass spectrometry (Iijima et al., 2013, 2008; Zhu et al., 2018), however these methods
67 are qualitative. A small number of quantitative methods using mass spectrometry have been
68 developed, but only for individual or few of tSGAs (Baldina et al., 2016; Caprioli et al., 2014).
69 Thus, there is a need to develop validated extraction and quantification methods in order to
70 continue to study the role these compounds have in both plant and human health.

71 To address the lack of suitable approaches to extract and quantify tSGAs, we developed
72 and validated a high-throughput extraction and ultra-high-performance liquid chromatography
73 tandem mass spectrometry (UHPLC-MS/MS) method suitable for tomato and tomato-based
74 products. Our extraction method is able to process 16 samples in parallel in 20 minutes (1.25
75 min/sample) and our UHPLC-MS/MS method can chromatographically separate, detect, and
76 quantify 16 tSGAs (using two external and two internal standards) representing 9 different tSGA
77 masses (Fig. 2) in 13 minutes per sample. This is the first comprehensive targeted method to
78 quantify a broad panel of tSGAs. Here, we present the experiments used to develop and validate
79 our method as well as an application providing baseline information of tSGA concentrations in
80 commonly consumed tomato products.

81

82 **2 Materials and Methods**

83 *Reagents and standards:* Acetonitrile (LC-MS grade), formic acid (LC-MS grade), isopropanol
84 (LC-MS grade), methanol (HPLC grade), and water (LC-MS grade) were purchased from Fisher

85 Scientific (Pittsburgh, PA). Alpha-tomatine ($\geq 90\%$ purity) and solanidine ($\geq 99\%$ purity) were
86 purchased from Extrasynthese (Genay, France). Alpha-solanine ($\geq 95\%$ purity) and tomatidine
87 ($\geq 95\%$ purity) were purchased from Sigma Aldrich (St. Louis, MO). Stock solutions were prepared
88 by weighing each analyte into glass vials and dissolving into methanol prior to storage at $-80\text{ }^{\circ}\text{C}$.
89 Standard curves were prepared by mixing 15 nmol of alpha-tomatine and 1 nmol of tomatidine in
90 methanol. The solution was evaporated to dryness under a stream of ultra-high purity (5.0 grade)
91 nitrogen gas. The dried residue was then resuspended in 900 μL of methanol, briefly sonicated (\sim
92 5 s), and then diluted with an additional 900 μL of water. An 8-point dilution series was then
93 prepared, and analyte concentrations ranged from 3.81 pmol/mL to 8.34 nmol/mL (11.14
94 femtomoles to 25 picomoles injected).

95
96 To utilize alpha-solanine and solanidine as internal standards (IS), 1.25 nmol and 22.68
97 pmol of alpha solanine and solanidine, respectively, were spiked into each vial of the alpha-
98 tomatine/tomatidine external standard dilution series described above. The spike intensity of
99 alpha-solanine and solanidine was determined by calculating the amount needed to achieve target
100 peak areas of tSGAs typically seen in tomato samples.

101
102 *Sample material:* For UHPLC-MS/MS and UHPLC-Quadrupole Time-of-Flight Mass
103 Spectrometry (UHPLC-QTOF-MS) method development experiments, 36 unique accessions of
104 tomato including *Solanum lycopersicum*, *Solanum lycopersicum* var. *cerasiforme*, and *Solanum*
105 *pimpinellifolium* were combined and pureed to create a tomato reference material expected to span
106 the diversity of tSGAs reported in nature. For spike-in recovery experiments, red-ripe processing-
107 type tomatoes (OH8245; courtesy of David M. Francis) were diced, mixed together by hand, and
108 stored at $-20\text{ }^{\circ}\text{C}$ until analysis. Items used for the tomato product survey were purchased from
109 supermarkets in Columbus, OH in July 2019. Three unique brands of tomato paste, tomato juice,
110 diced tomatoes, whole peeled tomatoes, ketchup, pasta sauce, and tomato soup were analyzed for
111 tSGAs. Additionally, four heirloom, two fresh-market, one processing, and one cherry variety of
112 unprocessed tomatoes were also analyzed.

113
114 *Extraction of tSGAs:* Five grams of diced OH8245 tomato ($\pm 0.05\text{ g}$) were weighed in 50 mL
115 falcon tubes. Two $3/8'' \times 7/8''$ angled ceramic cutting stones (W.W. Grainger: Lake Forest, IL;
116 Item no.: 5UJX2) were placed on top of the tomato sample and 100 μL of internal standard was
117 added, followed by 15 mL of methanol. Samples were then extracted for 5 minutes at 1400 RPM
118 using a Geno/Grinder 2010 (SPEX Sample Prep: Metuchen, NJ). Sample tubes were immediately
119 centrifuged for 5 minutes at $3000 \times g$ and $4\text{ }^{\circ}\text{C}$. Two mL aliquots of supernatant from each sample
120 were then transferred to glass vials and diluted with 1 mL of water. Samples were then filtered into
121 LC vials using a $0.22\text{ }\mu\text{m}$ nylon filter (CELLTREAT Scientific Products: Pepperell, MA).

122 Tomato products sourced from grocery stores were extracted as described above except
123 fresh fruits of each type were blended in a coffee grinder prior to extraction. To account for
124 differences in water content among the tomato products, 500 μL aliquots from each sample were
125 dried down under nitrogen gas, re-dissolved in 1.5 mL of 50% methanol, and filtered using a $0.22\text{ }\mu\text{m}$
126 filter prior to analysis.

127

128 *UHPLC-MS/MS Quantification of tSGAs:* Tomato steroidal glycoalkaloids were
129 chromatographically separated on a Waters (Milford, MA) Acquity UHPLC H-Class System using
130 a Waters C18 Acquity bridged ethylene hybrid (BEH) 2.1 x 100 mm, 1.7 μ m particle size column
131 maintained at 40 °C. The autosampler compartment was maintained at 20 °C. A gradient method
132 with Solvent A (water + 0.1% (v/v) formic acid) and Solvent B (Acetonitrile + 0.1% (v/v) formic
133 acid) at a flow rate of 0.4 mL/min was utilized as follows: 95% A for 0.25 minutes, 95% A to 80%
134 A for 1.0 minute, 80% A to 75% A for 2.5 minutes, 75% A held for 0.5 minutes, 75% A to 68%
135 A for 1.7 minutes, 38% A to 15% A for 1.7 minutes, 0% A held for 3.0 minutes, and back to 95%
136 A for 2.35 minutes to re-equilibrate the column. Each run lasted 13 minutes and the sample needle
137 was washed for 10 seconds with 1:1 methanol:isopropanol before and after each injection to
138 minimize carryover. Column eluent was directed into a Waters TQ Detector tandem mass
139 spectrometer and source parameters and transitions can be found in Table 1. Dwell times were
140 optimized for each analyte to allow for 12-15 points across each peak. Quantification was carried
141 out using 6-8 point external calibration curves, depending on the extent of linearity for a given
142 analyte. Relative quantification was used for tSGAs (quantified using alpha-tomatine) and their
143 aglycones (quantified using tomatidine) that did not have commercially available standards.
144 Additionally, signals were normalized to alpha-solanine and solanidine for glycosylated and
145 aglycone analytes to correct for instrument variability.

146

147 *UHPLC-QTOF/MS Confirmation of tSGA Identities:* We verified the identities of our tSGA
148 analytes using an Agilent 1290 Infinity II UHPLC coupled with an Agilent 6545 QTOF-MS.
149 Identical column and chromatographic separation conditions were used as described above for our
150 MS/MS method. The QTOF-MS used an electrospray ionization source operated in positive mode
151 and data were collected from 50-1700 m/z for both full-scan and MS/MS experiments. Gas
152 temperature was set to 350 °C, drying gas flow was 10 L/min, nebulizer gas flow was 10 L/min,
153 nebulizer was 35 psig, and sheath gas flow and temperature was 11 L/min and 375 °C, respectively.
154 For MS/MS experiments on the QTOF-MS, identical parameters were used except for the selection
155 of tSGA masses of interest and a two-minute retention time window around each analyte to
156 maximize duty cycle of the instrument. Collision energy for all tSGAs was set to 70 eV and all
157 aglycones were fragmented with 45 eV.

158

159 *Limit of Detection (LOD) and Limit of Quantification (LOQ):* Limit of detection and LOQ were
160 calculated using six replicates of the lowest concentration standard curve calibrant sample (3.81
161 and 0.254 femtomoles on column for alpha-tomatine and tomatidine, respectively) and
162 determining their signal to noise ratios. Moles on column at 3/1 and 10/1 signal to noise were then
163 determined for alpha-tomatine and tomatidine to calculate LOD and LOQ.

164

165 *Spike Recovery Experiments:* Ten, 5 g (\pm 0.01 g) replicates of diced OH8245 processing tomatoes
166 were weighed into 50 mL falcon tubes. Five samples were extracted as outlined previously with
167 the addition of a 100 μ L methanolic solution containing 1.67 nmol of alpha-tomatine, 1.25 nmol
168 of alpha solanine, 12.4 pmol of tomatidine, and 22.68 pmol of solanidine (spiked tomato) while
169 another five samples were extracted without IS solution (non-spiked tomato; 100 μ L of methanol
170 used in its place). The IS was allowed to integrate into the sample matrix for 30 min. Another set

171 of five samples were prepared by substituting tomato for 5 mL of water for tomato and extracted
172 with the addition of 100 μ L of the methanolic IS solution mentioned previously (spiked mock
173 sample). Percent recovery was estimated using the following equation:

$$174 \quad \text{Recovery (\%)} = \frac{\text{Spiked Tomato}}{\text{Non - spiked Tomato} + \text{Spiked Mock Sample}}$$

175
176 *Intra/Interday Variability Experiments:* Eight OH8245 tomato fruits were blended together and 5
177 g aliquots (\pm 0.05 g) were distributed among 18, 50 mL tubes, and frozen at -20 $^{\circ}$ C. Over three
178 days, six tubes were randomly selected from the freezer each day and tSGAs were extracted and
179 quantified as outlined above by a single individual. Intraday variability was determined by
180 computing the coefficient of variation for an analyte within a day. Interday variability was
181 calculated by taking the coefficient of variation of all samples run over the three-day period.

182
183 *Autosampler Stability Experiments:* A quality control sample containing multiple tomato species,
184 as described, above was extracted with the addition of 100 μ L of IS solution as outlined previously.
185 Over a period of 12 hours, the quality control sample was injected and analyzed by UHPLC-
186 MS/MS at hourly intervals. The vial cap was replaced after each injection to prevent sample
187 evaporation between injections and the autosampler compartment was maintained at 20 $^{\circ}$ C.

188
189 **3 Results and Discussion:**

190 3.1 Development of High-Throughput Extraction and UHPLC-MS/MS Quantification
191 Methods

192
193 *Development of high-throughput extraction method:* Generally, tomato samples are pulverized in
194 a mortar in the presence of liquid nitrogen or homogenized using a blender prior to extracting
195 tSGAs. Tomato steroidal glycoalkaloids are considered semi-polar metabolites and are typically
196 extracted via physical disruption in a methanolic solvent system (Ballester et al., 2016; Iijima et
197 al., 2013, 2008; Mintz-Oron et al., 2008; Moco et al., 2006). Current methods are time consuming
198 since each sample needs to be processed individually. Our protocol features a combined
199 homogenization/extraction step using a Geno/Grinder system that can process up to 16 samples at
200 once. Given a five-minute homogenization/extraction, five-minute centrifugation, and an
201 approximately ten-minute dilution/filtration step, our extraction method can process 16 samples
202 every 20 minutes (1.25 min/sample) making it ideal for screening large tomato populations or large
203 sample sets of tomato products. Moreover, the tomato sample is able to stay frozen until the
204 extraction begins which prevents potential enzymatic modification and degradation of analytes.

205
206 *Selection of Precursor Ions:* Over 100 tSGAs have been tentatively identified in tomato using high-
207 resolution mass spectrometry and some with MS/MS fragmentation (Iijima et al., 2013, 2008; Zhu
208 et al., 2018). However, we do not know the specific concentrations of tSGA accumulating in fruits.
209 To study tSGAs further, quantitative analysis methods are necessary. In order to maximize the

210 amount of tSGAs detected and separated in our method, we first compiled a target list of
211 biologically relevant tSGAs by surveying the literature (Alseekh et al., 2015; Cichon et al., 2017;
212 Cooperstone et al., 2017; Fujiwara et al., 2004; Hövelmann et al., 2019; Iijima et al., 2009; Zhu et
213 al., 2018). Tomato steroidal glycoalkaloid species were prioritized based on their perceived
214 abundance in the tomato clade, previous structural characterization, and having an established
215 record of being impacted by or a part of biological processes such as ripening or plant defense,
216 respectively. Using this process, 18 masses covering at least 25 different tSGA species were
217 selected for chromatographic separation and quantification.

218 A 50% aqueous methanolic extract from a reference material comprised of red-ripe
219 *Solanum lycopersicum*, *Solanum lycopersicum* var. *cerasiforme*, and *Solanum pimpinellifolium*
220 fruits was used for method development on a Waters Acquity UHPLC H-Class System connected
221 to a TQ Detector triple quadrupole mass spectrometer with electrospray ionization operated in
222 positive ion mode. A gradient progressing from 5% to 100% acetonitrile over 15 minutes run on a
223 Waters 2.1 x 100 mm (1.7 μ m particle size) column at 0.4 mL/min was used to separate as many
224 potential analytes as possible. Selected Ion Recordings (SIRs) of masses of interest were utilized
225 to identify potential tSGA species. Since only two alkaloids of interest are available commercially
226 (alpha-tomatine and tomatidine), elution order, accurate mass, and fragmentation patterns were
227 used to assign identity all other tSGAs. Source parameters of the MS were then adjusted to the
228 maximize signal of both identified and tentatively identified tSGAs Those tSGAs which were
229 readily detectable in our pooled tomato quality control samples were used in our final method.
230 While studied more extensively than many other tSGAs, we were not able to detect and quantify
231 beta-, gamma-, and delta-tomatine in our reference material.

232

233 *Use of Internal Standards:* We tested three, commercially available potato-derived alkaloids for
234 their suitability as internal standards to correct for inter and intraday variability created in the MS.
235 Alpha-solanine, alpha-chaconine, and solanidine (aglycone of alpha solanine) were selected based
236 on their similarity in structure, ionization efficiencies and retention times to tomato-derived
237 alkaloids. However, alpha-chaconine was excluded due to co-elution with alpha-tomatine. We
238 determined 1.25 nmol and 22.68 pmol of alpha solanine and solanidine, respectively, should be
239 added to each sample (41.7 femtomoles of alpha solanine and 0.756 femtomoles of solanidine on
240 column) to achieved comparable peak areas to those observed for tSGAs and their aglycones such
241 as tomatidine and tomatidenol (Fig. 2). Alpha-solanine and solanidine multiple reaction
242 monitoring (MRMs) experiments were then optimized in tandem with tSGAs of interest as follows.

243

244 *Optimization of MS parameters:* Desolvation temperature, desolvation gas flow rate, and cone
245 voltage were experimentally optimized. All other source parameters remained at their
246 recommended default settings and are reported in the footer of (Table 1). For all experiments, vial
247 caps were replaced after each injection to prevent any possible effects from evaporation through
248 the pierced septa. To optimize the desolvation temperature, a 50% aqueous methanolic solution of
249 alpha-tomatine and tomatidine was injected and desolvation temperatures ranging from 350 °C to
250 500 °C at 25 °C increments were tested. A 500 °C desolvation temperature resulted in the highest
251 signal. Desolvation gas flow was tested in a similar manner starting from 600 L/hr to 1000 L/hr in
252 100 L/hr increments. Likewise, the 1000 L/h flow rate resulted in the most signal for both analytes.
253 Alpha-tomatine and tomatidine were used in these experiments because of their commercial

254 availability, their structural similarity to other tSGAs of interest, and their intended use for relative
255 quantification of all other tSGAs and their aglycones. Finally, cone voltage was optimized by
256 injecting a 50% aqueous methanolic extract of our tomato reference material and measuring the
257 signal of each SIR. Cone voltages ranged from 20 to 90 V and successive injections were made in
258 5 V increments. Optimal cone voltages were specific to each mass and are notated in Table 1. With
259 source parameters set to optimize the signal of all precursor ions of interest, product ion scans were
260 then conducted to tentatively identify tSGAs and aid in the development of MRM experiments,
261 which were ultimately used for quantification.

262 Since each SIR yielded multiple peaks, information from product ion scans was leveraged
263 to determine if each peak was actually a tSGA. Product ion scan experiments were created for each
264 mass of interest and multiple collision energies (20, 45, and 65 eV) were tested. The resulting
265 spectra generated for each peak allowed us to eliminate peaks that were isobaric with tSGAs of
266 interest, but had product ions inconsistent with proposed structures. Masses such as 254.9 and
267 272.9 m/z were particularly useful in identifying alkaloids as they are likely derived from the
268 fragmentation of the steroidal backbone characteristic of all tSGAs (Supplementary Information)
269 and have been previously reported in the literature (Caprioli et al., 2014; Cichon et al., 2017; Iijima
270 et al., 2013; Sonawane et al., 2018; Zhu et al., 2018). Additionally, tSGAs with the prefix
271 “dehydro” exhibit a desaturation on the B ring of the steroidal backbone between carbons 5 and 6
272 (Iijima et al., 2013; Itkin et al., 2011; Ono et al., 1997; Sonawane et al., 2018). We observed that
273 common fragments derived from the steroidal backbone of these alkaloids, such as 252.9 and
274 270.8, were accordingly 2 m/z less than their saturated counterparts. The 272.9 fragment
275 corresponds to the A-D rings of the steroidal backbone and its corresponding water loss product
276 (Sonawane et al., 2018). Elution order of analytes was used to help tentatively identify tSGAs
277 detected in our reference sample based on previous reports (Alseekh et al., 2015; Zhu et al., 2018).
278 Multiple collision energies allowed us to select product ions that were abundant and consistently
279 produced under different conditions. These product ions then became candidate ions for MRM
280 development.

281 MRM experiments allowed us to confidently detect and quantify tSGAs of interest and
282 increase sensitivity by minimizing interference of co-eluting compounds. We created MRM
283 experiments for each mass using optimized source conditions and four product ions with the
284 highest signal/noise ratio. Initially, our 50% aqueous methanolic reference sample extract was
285 injected and each transition was tested at 5 eV. The experiments were rerun at increasing collision
286 energies at 15 eV increments up to 95 eV. Afterwards, a 20 eV window broken into 5 eV
287 increments was determined for each transition and the experiments were re-run. Optimized MRMs
288 are displayed in Table 1. To maximize duty cycle, two transitions with the best signal to noise ratio
289 were retained. The gradient was then optimized to chromatograph each analyte. All tSGAs were
290 quantified using a standard curve generated with alpha-tomatine while aglycone species used
291 tomatidine. Due to the structural similarity among tSGA species quantified in our method, we
292 hypothesize that ionization efficiencies will be similar amongst our analytes. Lastly, MRMs were
293 developed for the potato derived alkaloids alpha-solanine and solanidine used as IS. These IS
294 allowed us to correct for instrument derived variability that normally occurs with mass
295 spectrometers.

296

297 *Development of Chromatographic Gradient:* Method development related to the MS was initially
298 carried out using a simple 13-minute gradient outlined above. While this run time is shorter than
299 many of the previously published studies characterizing tSGAs using high-resolution MS (Iijima
300 et al., 2013, 2008; Zhu et al., 2018) we aimed to create a more efficient method that would be able
301 to accommodate large sample sets. Of the two columns tested (Waters C18 Acquity bridged
302 ethylene hybrid (BEH) 2.1 x 100 mm, 1.7 μm and Waters C18 Acquity high strength silica (HSS)
303 2.1 x 100 mm, 1.8 μm), the BEH column was able to better resolve analytes of interest with a
304 particular benefit observed in the nonpolar aglycone steroidal alkaloids. We adjusted our gradient
305 conditions in such a way that all separation of analytes occurred within a six-minute window with
306 an additional five minutes devoted to cleaning and reequilibrating the column to reduce carryover
307 (Fig. 2). Additionally, the needle wash was set to rinse the needle and injection port for ten seconds
308 before and after an injection with 1:1 methanol:isopropanol to further reduce carryover. We
309 observed multiple peaks for many of our masses indicating the presence of multiple isobaric tSGAs
310 (likely including structural isomers) (Fig. 2). In the case of esculeoside B, multiple diastereomers
311 have been previously reported in tomato products which explains our observation of multiple peaks
312 for this analyte (Hövelmann et al., 2019; Manabe et al., 2013; Nohara et al., 2015). Validation
313 experiments, including confirmation of peak identities using high-resolution mass spectrometry,
314 were next carried out using the finalized chromatographic gradient.

315

316 3.2 Validation of Extraction and UHPLC-MS/MS Methods

317

318 *Confirmation of Analytes using High-Resolution Mass Spectrometry:* Accurate mass spectrometry
319 was used to confirm the identities of analytes quantified by our UHPLC-MS/MS method. We
320 transferred our method to an Agilent 1290 Infinity II connected to an Agilent 6545 QTOF and
321 profiled tSGAs both in high resolution full scan mode (50-1700 m/z) and through targeted
322 fragmentation experiments. Both types of experiments were consistent with our identities of all
323 tSGAs and aglycones in our UHPLC-MS/MS method (Table 2). Retention times differed slightly
324 between the UHPLC-MS/MS method and the UHPLC-QTOF-MS experiments due to differences
325 in dead volume between the two instruments. However, relative elution order remained the same.

326 Targeted MS/MS experiments using the UHPLC-QTOF-MS allowed us to determine
327 common spectral characteristics for each tSGA (Fig. 1 and Supplementary Information). Using
328 commercially available alpha-tomatine and tomatidine and exploiting the presence of
329 dehydrotomatine and tomatidenol (dehydrotomatine) as impurities within these standards, we were
330 able to collect MS/MS fragmentation data on these four analytes. We found that all tSGAs and
331 aglycones fragmented in predictable ways that allow for identification. Common masses produced
332 by each tSGA in our method can be found in Table 2. These data allow us to tentatively identify
333 all analytes in our UHPLC-MS/MS with a high degree of confidence.

334

335 *LOD and LOQ:* Previous chromatography-based methods to quantify both potato and tSGAs relied
336 on photodiode array detectors and set 208 nm (Del Giudice et al., 2015; Kozukue et al., 2004;
337 Kozukue and Friedman, 2003; Tajner-Czopek et al., 2014). Given that the molar extinction
338 coefficient for alpha-tomatine is only 5000 $\text{M}^{-1}\text{cm}^{-1}$, (Keukens et al., 1994), photodiode array
339 detectors are not sensitive enough for detecting low quantities of these compounds, nor

340 distinguishing between different alkaloids. Moreover, photodiode array detectors are often set to
341 200 nm to quantify tSGAs which is a non-specific wavelength where many compounds (including
342 mobile phases) can absorb light (Friedman and Levin, 1998, 1992; Keukens et al., 1994). Mass
343 spectrometers offer substantial gains in sensitivity through the use of MRM experiments and the
344 ability to differentiate numerous analytes in a single run. Our UHPLC-MS/MS method for
345 quantifying tSGAs was able to detect and quantify alpha-tomatine and tomatidine in the low
346 femtomole-on-column range (Table 3). Given our extraction method, tSGAs could be present in
347 picomolar concentrations in tomato and still be quantified.

348

349 *Spike Recovery:* Spike addition experiments were conducted to assess the performance of our high-
350 throughput extraction method. Both tomato and potato derived external alkaloid standards were
351 used to determine if our chosen internal standards would behave similarly to analytes native to
352 tomato. Tomato alkaloids alpha-tomatine ($100.8\% \pm 13.1$) and tomatidine ($93\% \pm 6.8$) as well as
353 the potato-derived internal standards alpha solanine ($94.3\% \pm 3.4$) and solanidine ($99.7\% \pm 7.1$)
354 were efficiently extracted using our method (Table 3). These data indicate that our method is able
355 to effectively extract aglycone and glycosylated steroidal alkaloid species from tomato and our
356 internal standards extract similarly to native analytes.

357

358 *Intra/Interday Variability:* Experiments to determine intra/interday variability were conducted to
359 determine analytical variability in our extraction and analysis methods. A single operator extracted
360 six tomato samples and analyzed them by UHPLC-MS/MS. This experiment was repeated twice
361 more by the same operator. Our data indicate that our methods are reliable with most analytes
362 having coefficient of variations for both intra and interday variability below 5% (Table 4). As
363 expected, interday variability was higher than intraday variability for all analytes reflecting day-
364 to-day variability in the MS.

365

366 *12-hour Stability Experiment:* Tomato phytochemicals typically analyzed, such as carotenoids, are
367 subject to oxidation and need to be run in small batches to minimize experimental error due to
368 degradation (Kopeck et al., 2012). However, relatively little is known about the stability of tSGAs
369 compared to the above phytochemical classes. We hypothesized that due to the known heat
370 stability of chemically analogous potato steroidal glycoalkaloids, extracted tSGAs would be stable
371 over time. A 12-hour stability study demonstrated that both alpha-tomatine and tomatidine did not
372 degrade over time in an autosampler maintained at 20 °C. While there is currently no published
373 literature investigating the stability of tSGAs, some data exists in chemically analogous potato
374 glycoalkaloids. Often, potato glycoalkaloids are often extracted at 100 °C temperatures to disrupt
375 cell walls and otherwise weaken the sample matrix (Rodriguez-Saona et al., 1999) and processing
376 studies have shown that these compounds are stable up to 180 °C (Chungcharoen, 1988).
377 Therefore, tSGAs may also have similar heat tolerance attributes and we speculate that these
378 analytes may remain unchanged in autosamplers well beyond the 12-hour time period we tested.

379

380 3.3 Application of Extraction and UHPLC-MS/MS Method

381

382 *Grocery Store Survey*: To test our extraction and quantification method, we surveyed several
383 commonly consumed tomato-based products available at grocery stores. The purpose was twofold:
384 to test applicability of our method, and to report comprehensive and quantitative values of tSGAs
385 in commonly consumed tomato products. These products included an assortment of fresh
386 tomatoes, ketchup, pasta sauce, pizza sauce, tomato soup, tomato paste, tomato juice, and whole
387 peeled tomatoes (Table 5). Values are reported per serving to normalize between tomato products
388 subjected to varying degrees of concentration. While there are some reports of tSGA
389 concentrations in fresh tomatoes using modern methods (Baldina et al., 2016), concentrations in
390 tomato-based products are not well reported in the literature. We found that tSGAs varied
391 depending on type of product. High standard deviations likely reflect differences in geographic
392 origin, harvest time, and processing conditions. Of note, many of our tSGAs varied by up to three
393 orders of magnitude among different analytes and tomato products. This finding indicates a broad
394 range of tSGA concentrations in tomato-based products.

395 Alpha-tomatine, the first tSGA in the biosynthesis pathway, was found to be in the highest
396 concentration in processed tomato products such as paste, pasta sauce, and soup (Table 5). The
397 discrepancy between fresh and whole peeled tomatoes is hypothesized to be due to genetic and
398 environmental conditions that influenced the chemical profile of the tomatoes prior to processing.
399 Analyte groups like dehydrolycoperoside F, G or A, lycopersides F, G, or esculeoside A and
400 acetoxytomatine (commonly referred to as lycopersides A, B, or C) were not detectable in most
401 tomato products except for some fresh varieties and ketchup. Interestingly, lycopersides F, G, or
402 esculeoside A are typically the most abundant tSGA in fresh tomatoes. This observation raises
403 questions about the effects of processing on tSGAs where few studies have been conducted to date
404 (Tomas et al., 2017). While the chemically analogous potato glycoalkaloids are considered to be
405 heat stable, high temperatures, pressures, and any combination thereof might be detrimental to
406 some tSGAs or cause shifts in chemical profiles.

407 Concentrations of tSGAs in tomato products were normalized for serving size to
408 contextualize how much might be ingested in a given meal. Other tomato phytochemicals, such as
409 lycopene, tend to be found in concentrations ranging from 0.09 to 9.93 mg/100g FW in fresh
410 tomatoes (Dzakovich et al., 2019). Compared to major carotenoids found in tomato, tSGA
411 concentrations were comparable (0.7 to 3.4 mg/serving) (Cooperstone, 2020) This finding
412 contradicts a long-standing misconception that tSGAs are degraded during ripening (Friedman,
413 2002). Rather, tSGAs such as alpha-tomatine are biochemically transformed during ripening into
414 glycosylated and acetylated forms. Overall, our methods were able to efficiently extract and
415 analyze many types of tSGAs and generate the first quantitative concentration reports of these
416 analytes in commonly consumed tomato products. Moreover, we found that tSGAs can be found
417 in similar concentrations to other major phytochemicals in tomatoes such as carotenoids.

418 We have developed and described the first comprehensive extraction and analysis method
419 for tSGAs. Our extraction method was able to quickly and efficiently extract tSGAs and allowed
420 for high-throughput workflows (16 samples per ~20 min) to be utilized. Our UHPLC-MS/MS
421 method was able to separate and quantify 16 tSGAs representing 9 different tSGA masses, as well
422 as two internal standards, in 13 minutes. Limits of quantification for commercially available tSGAs
423 were 1.09 and 0.34 femtomoles on column for alpha-tomatine and tomatidine, respectively. This
424 corresponds to 0.8 and 0.25 $\mu\text{g}/100\text{g}$ of alpha-tomatine and tomatidine in tomato, respectively,
425 given our extraction procedures. Relative quantification for tSGAs and aglycones that did not have
426 commercially available standards was performed using alpha-tomatine and tomatidine,

427 respectively. Our methods were able to successfully profile tSGAs in a comprehensive array of
428 commonly available tomato-based products. These values are among the first to be reported in the
429 literature and can serve as benchmarks for future studies investigating tSGAs in a variety of
430 contexts. Our extraction and UHPLC-MS/MS method will allow researchers to rapidly and
431 accurately generate data about tSGAs and overcomes a major limitation hampering this field and
432 allow for the field to advance.

433

434 **Conflict of Interest**

435 The authors declare that the research was conducted in the absence of any commercial or
436 financial relationships that could be construed as a potential conflict of interest.

437

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448

449 **Data Availability Statement**

450 The raw data supporting the conclusions of this manuscript will be made available by the
451 authors, without undue reservation, to any qualified researchers.

452

453 **Works Cited:**

- 454 Alseikh, S., Tohge, T., Wendenberg, R., Scossa, F., Omranian, N., Li, J., Kleessen, S.,
455 Giavalisco, P., Pleban, T., Mueller-Roeber, B., Zamir, D., Nikoloski, Z., Fernie, A.R., 2015.
456 Identification and mode of inheritance of quantitative trait loci for secondary metabolite
457 abundance in tomato. *Plant Cell* 27, 485–512. doi:10.1105/tpc.114.132266
- 458 Baldina, S., Picarella, M.E., Troise, A.D., Pucci, A., Ruggieri, V., Ferracane, R., Barone, A.,
459 Fogliano, V., Mazzucato, A., 2016. Metabolite Profiling of Italian Tomato Landraces with
460 Different Fruit Types. *Front. Plant Sci.* 7, 664. doi:10.3389/fpls.2016.00664
- 461 Ballester, A.-R., Tikunov, Y., Molthoff, J., Grandillo, S., Viquez-Zamora, M., de Vos, R., de
462 Maagd, R.A., van Heusden, S., Bovy, A.G., 2016. Identification of Loci Affecting
463 Accumulation of Secondary Metabolites in Tomato Fruit of a *Solanum lycopersicum* ×
464 *Solanum chmielewskii* Introgression Line Population. *Front. Plant Sci.* 7, 1428.
465 doi:10.3389/fpls.2016.01428
- 466 Caprioli, G., Cahill, M.G., James, K.J., 2014. Mass Fragmentation Studies of α -Tomatine and
467 Validation of a Liquid Chromatography LTQ Orbitrap Mass Spectrometry Method for Its
468 Quantification in Tomatoes. *Food Anal. Methods* 7, 1565–1571. doi:10.1007/s12161-013-
469 9771-9
- 470 Cárdenas, P.D., Sonawane, P.D., Heinig, U., Bocobza, S.E., Burdman, S., Aharoni, A., 2015.
471 The bitter side of the nightshades: Genomics drives discovery in Solanaceae steroidal
472 alkaloid metabolism. *Phytochemistry* 113, 24–32. doi:10.1016/j.phytochem.2014.12.010
- 473 Cárdenas, P.D., Sonawane, P.D., Pollier, J., Vanden Bossche, R., Dewangan, V., Weithorn, E.,
474 Tal, L., Meir, S., Rogachev, I., Malitsky, S., Giri, A.P., Goossens, A., Burdman, S.,
475 Aharoni, A., 2016. GAME9 regulates the biosynthesis of steroidal alkaloids and upstream
476 isoprenoids in the plant mevalonate pathway. *Nat. Commun.* 7, 10654.
477 doi:10.1038/ncomms10654
- 478 Cayen, M.N., 1971. Effect of dietary tomatine on cholesterol metabolism in the rat. *J. Lipid Res.*
479 12, 482–490.
- 480 Choi, S.H., Ahn, J.-B., Kozukue, N., Kim, H.-J., Nishitani, Y., Zhang, L., Mizuno, M., Levin,
481 C.E., Friedman, M., 2012. Structure–Activity Relationships of α -, β 1 -, γ -, and δ -Tomatine
482 and Tomatidine against Human Breast (MDA-MB-231), Gastric (KATO-III), and Prostate
483 (PC3) Cancer Cells. *J. Agric. Food Chem.* 60, 3891–3899. doi:10.1021/jf3003027
- 484 Chungcharoen, A., 1988. Glycoalkaloid content of potatoes grown under controlled
485 environments and stability of glycoalkaloids during processing. PhD Thesis. University of
486 Wisconsin, Madison, WI.
- 487 Cichon, M.J., Riedl, K.M., Wan, L., Thomas-Ahner, J.M., Francis, D.M., Clinton, S.K.,
488 Schwartz, S.J., 2017. Plasma Metabolomics Reveals Steroidal Alkaloids as Novel
489 Biomarkers of Tomato Intake in Mice. *Mol. Nutr. Food Res.* 61, 1700241.
490 doi:10.1002/mnfr.201700241
- 491 Cooperstone, J.L., 2020. Lycopene, in: Wildman, E.C., Bruno, R.S. (Eds.), *Handbook of*
492 *Nutraceuticals and Functional Foods*. CRC Press, pp. 37–53. doi:10.1201/9780429195594-3

- 493 Cooperstone, J.L., Tober, K.L., Riedl, K.M., Teegarden, M.D., Cichon, M.J., Francis, D.M.,
494 Schwartz, S.J., Oberyshyn, T.M., 2017. Tomatoes protect against development of UV-
495 induced keratinocyte carcinoma via metabolomic alterations. *Sci. Rep.* 7, 5106.
496 doi:10.1038/s41598-017-05568-7
- 497 Del Giudice, R., Raiola, A., Tenore, G.C., Frusciante, L., Barone, A., Monti, D.M., Rigano,
498 M.M., 2015. Antioxidant bioactive compounds in tomato fruits at different ripening stages
499 and their effects on normal and cancer cells. *J. Funct. Foods* 18, 83–94.
500 doi:10.1016/J.JFF.2015.06.060
- 501 Dzakovich, M.P., Gas-Pascual, E., Orchard, C.J., Sari, E.N., Riedl, K.M., Schwartz, S.J., Francis,
502 D.M., Cooperstone, J.L., 2019. Analysis of Tomato Carotenoids: Comparing Extraction and
503 Chromatographic Methods. *J. AOAC Int.* 102, 1069–1079. doi:10.5740/jaoacint.19-0017
- 504 Eltayeb, E.A., Roddick, J.G., 1984. Changes in the Alkaloid Content of Developing Fruits of
505 Tomato (*Lycopersicon esculentum* Mill.) II. *J. Exp. Bot.* 35, 261–267.
506 doi:10.1093/jxb/35.2.261
- 507 Etalo, D.W., De Vos, R.C.H., Joosten, M.H.A.J., Hall, R.D., 2015. Spatially resolved plant
508 metabolomics: Some potentials and limitations of laser-ablation electrospray ionization
509 mass spectrometry metabolite imaging. *Plant Physiol.* 169, 1424–1435.
510 doi:10.1104/pp.15.01176
- 511 Fontaine, T.D., Irving, G.W., Ma, R., Poole, J.B., Doolittle, S.P., 1948. Isolation and partial
512 characterization of crystalline tomatine, an antibiotic agent from the tomato plant. *Arch.*
513 *Biochem.* 18, 467–75.
- 514 Friedman, M., 2002. Tomato glycoalkaloids: role in the plant and in the diet. *J. Agric. Food*
515 *Chem.* 50, 5751–5780.
- 516 Friedman, M., Levin, C.E., 1998. Dehydrotomatine content in tomatoes. *J. Agric. Food Chem.*
517 46, 4571–4576. doi:10.1021/jf9804589
- 518 Friedman, M., Levin, C.E., 1992. Reversed-phase high-performance liquid chromatographic
519 separation of potato glycoalkaloids and hydrolysis products on acidic columns. *J. Agric.*
520 *Food Chem.* 40, 2157–2163. doi:10.1021/jf00023a023
- 521 Fujiwara, Y., Takaki, A., Uehara, Y., Ikeda, T., Okawa, M., Yamauchi, K., Ono, M., Yoshimitsu,
522 H., Nohara, T., 2004. Tomato steroidal alkaloid glycosides, esculeosides A and B, from ripe
523 fruits. *Tetrahedron* 60, 4915–4920. doi:10.1016/j.tet.2004.03.088
- 524 Hövelmann, Y., Jagels, A., Schmid, R., Hübner, F., Humpf, H.-U., 2019. Identification of
525 potential human urinary biomarkers for tomato juice intake by mass spectrometry-based
526 metabolomics. *Eur. J. Nutr.* 1–13. doi:10.1007/s00394-019-01935-4
- 527 Iijima, Y., Fujiwara, Y., Tokita, T., Ikeda, T., Nohara, T., Aoki, K., Shibata, D., 2009.
528 Involvement of Ethylene in the Accumulation of Esculeoside A during Fruit Ripening of
529 Tomato (*Solanum lycopersicum*). *J. Agric. Food Chem.* 57, 3247–3252.
530 doi:10.1021/jf8037902
- 531 Iijima, Y., Nakamura, Y., Ogata, Y., Tanaka, K.K., Sakurai, N., Suda, K., Suzuki, T., Suzuki, H.,
532 Okazaki, K., Kitayama, M., Kanaya, S., Aoki, K., Shibata, D., 2008. Metabolite annotations

- 533 based on the integration of mass spectral information. *Plant J.* 54, 949–62.
534 doi:10.1111/j.1365-313X.2008.03434.x
- 535 Iijima, Y., Watanabe, B., Sasaki, R., Takenaka, M., Ono, H., Sakurai, N., Umemoto, N., Suzuki,
536 H., Shibata, D., Aoki, K., 2013. Steroidal glycoalkaloid profiling and structures of
537 glycoalkaloids in wild tomato fruit. *Phytochemistry* 95, 145–157.
538 doi:10.1016/J.PHYTOCHEM.2013.07.016
- 539 Irving, G.W., Fontaine, T.D., Doolittle, S.P., 1945. Lycopersicin, a Fungistatic Agent from the
540 Tomato Plant. *Science* (80-). 102, 9–11.
- 541 Itkin, M., Heinig, U., Tzfadia, O., Bhide, A.J., Shinde, B., Cardenas, P.D., Bocobza, S.E., Unger,
542 T., Malitsky, S., Finkers, R., Tikunov, Y., Bovy, A., Chikate, Y., Singh, P., Rogachev, I.,
543 Beekwilder, J., Giri, A.P., Aharoni, A., 2013. Biosynthesis of antinutritional alkaloids in
544 solanaceous crops is mediated by clustered genes. *Science* 341, 175–9.
545 doi:10.1126/science.1240230
- 546 Itkin, M., Rogachev, I., Alkan, N., Rosenberg, T., Malitsky, S., Masini, L., Meir, S., Iijima, Y.,
547 Aoki, K., de Vos, R., Prusky, D., Burdman, S., Beekwilder, J., Aharoni, A., 2011.
548 GLYCOALKALOID METABOLISM1 Is Required for Steroidal Alkaloid Glycosylation
549 and Prevention of Phytotoxicity in Tomato. *Plant Cell* 23, 4507–4525.
550 doi:10.1105/tpc.111.088732
- 551 Keukens, E.A.J., Hop, M.E.C.M., Jongen+, W.M.F., 1994. Rapid High-Performance Liquid
552 Chromatographic Method for the Quantification of-Tomatine in Tomato, *Food Chem.*
- 553 Kopec, R.E., Cooperstone, J.L., Cichon, M.J., Schwartz, S.J., 2012. Analysis Methods of
554 Carotenoids, in: *Analysis of Antioxidant-Rich Phytochemicals*. Wiley-Blackwell, Oxford,
555 UK, pp. 105–148. doi:10.1002/9781118229378.ch4
- 556 Kozukue, N., Friedman, M., 2003. Tomatine, chlorophyll, Beta-carotene and lycopene content in
557 tomatoes during growth and maturation. *J. Sci. Food Agric.* 83, 195–200.
558 doi:10.1002/jsfa.1292
- 559 Kozukue, N., Han, J.-S., Lee, K.-R., Friedman, M., 2004. Dehydrotomatine and α -Tomatine
560 Content in Tomato Fruits and Vegetative Plant Tissues. *J. Agric. Food Chem.* 52, 2079–
561 2083. doi:10.1021/jf0306845
- 562 Lawson, D.R., Erb, W.A., Miller, A.R., 1992. Analysis of Solanum Alkaloids Using Internal
563 Standardization and Capillary Gas Chromatography. *Food Chem* 40, 2186–2191.
- 564 Lee, K.-R., Kozukue, N., Han, J.-S., Park, J.-H., Chang, E., Baek, E.-J., Chang, J.-S., Friedman,
565 M., 2004. Glycoalkaloids and Metabolites Inhibit the Growth of Human Colon (HT29) and
566 Liver (HepG2) Cancer Cells. *J. Agric. Food Chem.* 52, 2832–2839. doi:10.1021/jf030526d
- 567 Manabe, H., Fujiwara, Y., Ikeda, T., Ono, M., Murakami, K., Zhou, J.R., Yokomizo, K., Nohara,
568 T., 2013. Saponins esculeosides B-1 and B-2 in Italian canned tomatoes. *Chem. Pharm.*
569 *Bull.* 61, 764–767. doi:10.1248/cpb.c13-00202
- 570 Mintz-Oron, S., Mandel, T., Rogachev, I., Feldberg, L., Lotan, O., Yativ, M., Wang, Z., Jetter,
571 R., Venger, I., Adato, A., Aharoni, A., 2008. Gene Expression and Metabolism in Tomato
572 Fruit Surface Tissues. *PLANT Physiol.* 147, 823–851. doi:10.1104/pp.108.116004

- 573 Moco, S., Bino, R.J., Vorst, O., Verhoeven, H.A., de Groot, J., van Beek, T.A., Vervoort, J., de
574 Vos, C.H.R.H.R., 2006. A liquid chromatography-mass spectrometry-based metabolome
575 database for tomato. *Plant Physiol.* 141, 1205–18. doi:10.1104/pp.106.078428
- 576 Nohara, T., Fujiwara, Y., Zhou, J.R., Urata, J., Ikeda, T., Murakami, K., El-Aasr, M., Ono, M.,
577 2015. Saponins, esculeosides B-1 and B-2, in tomato juice and sapogenol, esculeogenin B 1.
578 *Chem. Pharm. Bull.* 63, 848–850. doi:10.1248/cpb.c15-00449
- 579 Ökmen, B., Etalo, D.W., Joosten, M.H.A.J., Bouwmeester, H.J., de Vos, R.C.H., Collemare, J.,
580 De Wit, P.J.G.M., 2013. Detoxification of α -tomatine by *Cladosporium fulvum* is required
581 for full virulence on tomato. *New Phytol.* 198, 1203–1214. doi:10.1111/nph.12208
- 582 Ono, H., Kozuka, D., Chiba, Y., Horigane, A., Isshiki, K., 1997. Structure and Cytotoxicity of
583 Dehydrotomatine, a Minor Component of Tomato Glycoalkaloids. *J. Agric. Food Chem.*
584 doi:10.1021/jf970253k
- 585 Rick, C.M., Uhlig, J.W., Jones, D., 1994. High alpha-tomatine content in ripe fruit of Andean
586 *Lycopersicon esculentum* var. *cerasiforme*: developmental and genetic aspects. *Proc. Natl.*
587 *Acad. Sci.* 91, 12877–81.
- 588 Rodriguez-Saona, L.E., Wrolstad, R.E., Pereira, C., 1999. Glycoalkaloid Content and
589 Anthocyanin Stability to Alkaline Treatment of Red-Fleshed Potato Extracts. *J. Food Sci.*
590 64, 445–450. doi:10.1111/j.1365-2621.1999.tb15060.x
- 591 Schlösser, E., Gottlieb, D., 1966. Mode of hemolytic action of the antifungal polyene antibiotic
592 filipin. *Zeitschrift fur Naturforsch. - Sect. B J. Chem. Sci.* 21, 74–77. doi:10.1515/znB-
593 1966-0120
- 594 Sonawane, P.D., Heinig, U., Panda, S., Gilboa, N.S., Yona, M., Pradeep Kumar, S., Alkan, N.,
595 Unger, T., Bocobza, S., Pliner, M., Malitsky, S., Tkachev, M., Meir, S., Rogachev, I.,
596 Aharoni, A., 2018. Short-chain dehydrogenase/reductase governs steroidal specialized
597 metabolites structural diversity and toxicity in the genus *Solanum*. *Proc. Natl. Acad. Sci. U.*
598 *S. A.* doi:10.1073/pnas.1804835115
- 599 Tajner-Czopek, A., Rytel, E., Aniołowska, M., Hamouz, K., 2014. The influence of French fries
600 processing on the glycoalkaloid content in coloured-fleshed potatoes. *Eur. Food Res.*
601 *Technol.* 238, 895–904. doi:10.1007/s00217-014-2163-6
- 602 Tomas, M., Beekwilder, J., Hall, R.D., Sagdic, O., Boyacioglu, D., Capanoglu, E., 2017.
603 Industrial processing versus home processing of tomato sauce: Effects on phenolics,
604 flavonoids and in vitro bioaccessibility of antioxidants. *Food Chem.*
605 doi:10.1016/j.foodchem.2016.09.201
- 606 Zhu, G., Wang, S., Huang, Z., Zhang, S., Liao, Q., Zhang, C., Lin, T., Qin, M., Peng, M., Yang,
607 C., Cao, X., Han, X., Wang, X., van der Knaap, E., Zhang, Z., Cui, X., Klee, H., Fernie,
608 A.R., Luo, J., Huang, S., 2018. Rewiring of the Fruit Metabolome in Tomato Breeding. *Cell*
609 172, 249-261.e12. doi:10.1016/j.cell.2017.12.019
- 610
- 611

612 **Figure Captions**

613

614 **Fig. 1. Structural and isomeric variation in selected tomato steroidal alkaloids.** Steroidal
615 glycoalkaloids found in tomato (tSGAs) are spirosolane-type saponins with variations in a singular
616 double-bond (C5:6), F-ring decorations (C22-C27), F-ring rearrangement (resulting in a change in
617 stereochemistry at C22), and C3 glycosylation (typically a four-sugar tetrasaccharide,
618 lycotetraose). The undecorated SA backbone is shown first with relevant carbons numbered and
619 ring names (A-F). Steroidal alkaloids (SAs) were grouped based on structural similarity with bonds
620 of varying stereochemistry denoted by wavy bonds and varying C5:6 saturation status denoted by
621 a dashed bond. Structural variation, along with the monoisotopic mass, molecular formula, and
622 common name are displayed alongside structures for each group. R-groups were used to denote
623 status of C3 glycosylation in all groups (R₁ and R₂) and possible positions of glucosylation on
624 glucosylated (dehydro)acetoxytomatine (R₃, R₄). All possible isomers and derivatives are not
625 shown, just those quantitated in this method.

626

627 **Fig. 2. Chromatogram of tSGAs found in red ripe tomatoes measured by our UHPLC-MS/MS**
628 **method. Peaks are identified as follows: 1a-c: Esculeoside B1-3; 2a-d: Hydroxytomatine; 3:**
629 **Dehydrolycoperoside F, G, or Dehydroesculeoside A; 4a,b: Lycoperoside F, G, or Esculeoside A;**
630 **5a-c: Acetoxytomatine; 6a,b: Dehydrotomatine; 7: Alpha-tomatine; 8: Alpha-solanine; 9:**
631 **Solanidine; 10: Tomatidine; 11: Tomatidenol**

632

633

634 Table 1. LC-MS/MS MRM parameters of steroidal glycoalkaloids quantified by our method.

Analyte	Retention Time (min)	Parent Mass [M+H] _a	Product Ions	Cone Voltage (V)	Collision Energy (eV)
Esculeoside B	2.55, 2.67, 2.74	1228.6	254.9*, 1048.8	75	65, 40
Hydroxytomatine	3.02, 3.28, 3.37, 3.57	1050.6	254.9, 1032.7	55	55, 30
Dehydrolycoperoside F, G, or Dehydroesculeoside A	3.08	1268.6	252.9*, 1208.9	80	65, 35
Lycoperoside F, G, or Esculeoside A	3.11, 3.26	1270.6	1048.8, 1210.9	70	60, 30
Acetoxytomatine _b (I)	4.28	1092.6	84.7, 1032.7	40	65, 35
Dehydrotomatine _c	5.09, 5.49	1032.5	84.7, 252.9*	70	80, 50
Acetoxytomatine (II)	5.42, 5.66	1092.6	144.7, 162.8	40	50, 45
Alpha-tomatine _c	5.45	1034.6	84.7*, 160.8	70	85, 60
Alpha-solanine _{c,d}	5.64	869.1	97.8*, 399.1	70	85, 65
Solanidine _{c,d}	7.22	398.7	80.7, 97.8*	70	55, 35
Tomatidine _c	7.30	416.4	160.8, 254.9	50	30, 30
Tomatidenol _c	7.36	414.3	125.8, 270.7	40	30, 20

635 _aAnalytes were quantified using the following settings: Mass span: 0.3 Da, Capillary voltage: 0.5

636 kV, extractor voltage: 5 V, RF Lens voltage: 0.5 V, source temperature: 150°C, desolvation

637 temperature: 500°C, desolvation flow rate: 1000 L/hr, cone gas flow rate: 50 L/hr.

638 _bCommonly referred to as lycoperosides A, B, or C

639 _cIndicates that analyte was confirmed by authentic standard

640 _dIndicates analyte used as an internal standard.

- 641 *Indicates quantifying ion; other ions used for qualifying purposes. Compounds with no
642 indicated quantifying ion were quantified using the sum of both MRM transitions.

643 Table 2. UHPLC-QTOF-MS Confirmation of tSGA Identities

Tentative Identification	Molecular Formula	Retention Time (min)	Monoisotopic Mass	Observed Mass [M+H]	Mass Error (Δ ppm)	Common MS/MS Fragments ^a
Esculeoside B	C ₅₆ H ₉₃ NO ₂₈	2.24	1227.5884	1228.5989	2.20	1048.5380, 273.2120, 255.2016, 163.0509, 145.0404, 85.0205
		2.34		1228.5967	0.41	
		2.45		1228.5966	0.33	
Hydroxytomatine	C ₅₀ H ₈₃ NO ₂₂	2.84	1049.5407	1050.5500	1.43	1032.5385, 273.2213, 255.2203, 161.1318, 145.0489, 85.0279
		3.22		1050.5513	2.67	
		3.29		1050.5506	2.00	
		3.50		1050.5501	1.52	
Dehydrolycoperoside F, G, or Dehydroesculeoside A	C ₅₈ H ₉₃ NO ₂₉	2.41	1267.5828	1268.5930	1.89	1208.5714, 1046.5175, 271.2054, 253.1951, 163.0600, 85.0284
Lycoperoside F, G, or Esculeoside A	C ₅₈ H ₉₅ NO ₂₉	2.44	1269.5985	1270.6076	1.02	1210.5900, 1048.5324, 273.2213, 255.2108, 163.0600, 85.0285
		3.06		1270.6095	2.52	
Acetoxytomatine (I)	C ₅₂ H ₈₅ NO ₂₃	4.22	1091.5507	1092.5614	2.65	1032.5386, 273.2216, 255.2112, 161.1326, 145.0497, 85.0287
Dehydrotomatine ^b	C ₅₀ H ₈₁ NO ₂₁	4.99	1031.5301	1032.5388	0.87	1014.5274, 271.2054, 253.1951, 145.0495, 85.0284, 57.0337
		5.20		1032.5373	0.58	
Acetoxytomatine (II)	C ₅₂ H ₈₅ NO ₂₃	5.32	1091.5507	1092.5619	3.11	1032.5404, 273.2216, 255.2114, 161.1328, 145.0499, 85.0288
		5.39		1092.5608	2.11	
Alpha-tomatine ^b	C ₅₀ H ₈₃ NO ₂₁	5.35	1033.5457	1034.5557	2.13	1016.5449, 416.3523, 273.2217, 255.2112, 145.0498, 85.0287
Tomatidine ^b	C ₂₇ H ₄₅ NO ₂	6.95	415.3450	416.3531	0.72	398.3414, 273.2208, 255.2101, 161.1318, 126.1271, 81.0693

Tomatidenol ^b	C ₂₇ H ₄₃ NO ₂	6.98	413.3294	414.3371	0.24	396.3260, 271.2053, 253.1949, 161.1322, 126.1275, 81.0695
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644 ^aMS/MS product ions generated at 70 eV and 45 eV for glycosylated and aglycone species, respectively. Other source parameters

645 were previously enumerated.

646 ^bIdentification confirmed by authentic standard

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648 Table 3. Extraction efficiency of commercially available tSGAs and potato-derived internal
649 standards.

Analyte	Sample Size	Extraction Efficiency (%)	LOD (femtomoles injected)	LOQ (femtomoles injected)
Alpha-tomatine	n=6	100.8 ± 13.1	1.0988	0.3296
Alpha-solanine ^a	n=6	94.3 ± 3.4	N/A ^b	N/A
Tomatidine	n=6	93.0 ± 6.8	0.3354	0.1006
Solanidine ^a	n=6	99.7 ± 7.1	N/A	N/A

650 ^aAnalyte used as an internal standard with no calibration curve

651 ^bNot applicable due to its use as an internal standard

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666 Table 4. Intraday and interday coefficient of variation values for analytes quantified by our
667 UHPLC-MS/MS method

Analyte	Intraday Coefficient of Variation (%) ^a	Interday Coefficient of Variation (%) ^b
Esculeoside B	4.46	6.84
Hydroxytomatine	4.00	5.60
Dehydrolycoperoside F, G, or Dehydroesculeoside A	8.42	8.03
Lycoperoside F, G, or Esculeoside A	3.35	4.21
Acetoxytomatine (I)	3.56	3.89
Dehydrotomatine	4.25	7.11
Acetoxytomatine (II)	7.57	7.70
Alpha-tomatine	3.92	6.42
Tomatidine	11.78	13.73
Tomatidenol	11.69	13.61

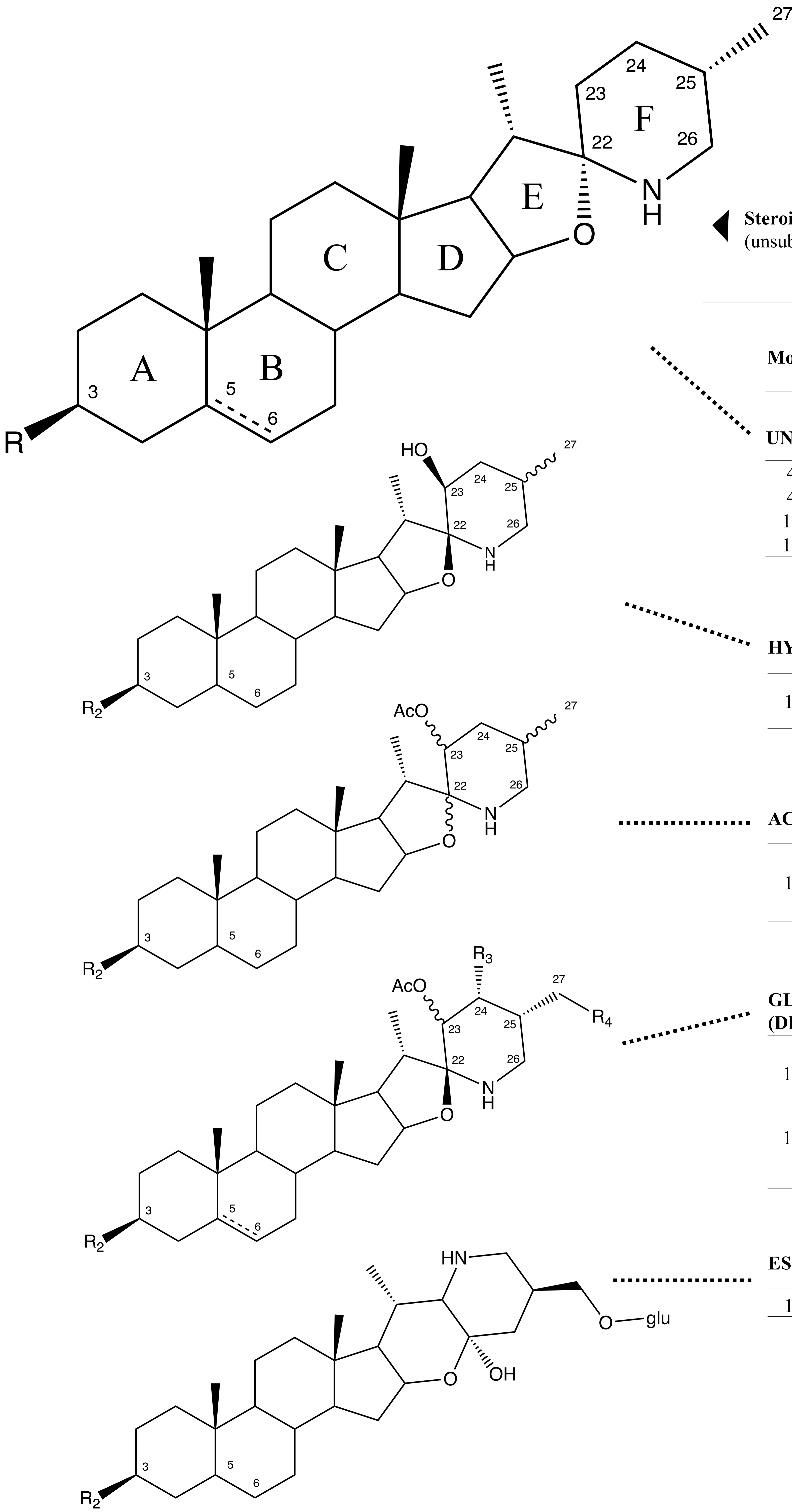
668 ^aAverage coefficient of variation within a day of six samples extracted and run by a single
669 operator. The experiment was repeated over three days.

670 ^bAverage coefficient of variation over a three-day period of 18 samples extracted and run by a
671 single operator.

672 Table 5. Survey of tSGAs in common tomato-based products reported in µg per serving size

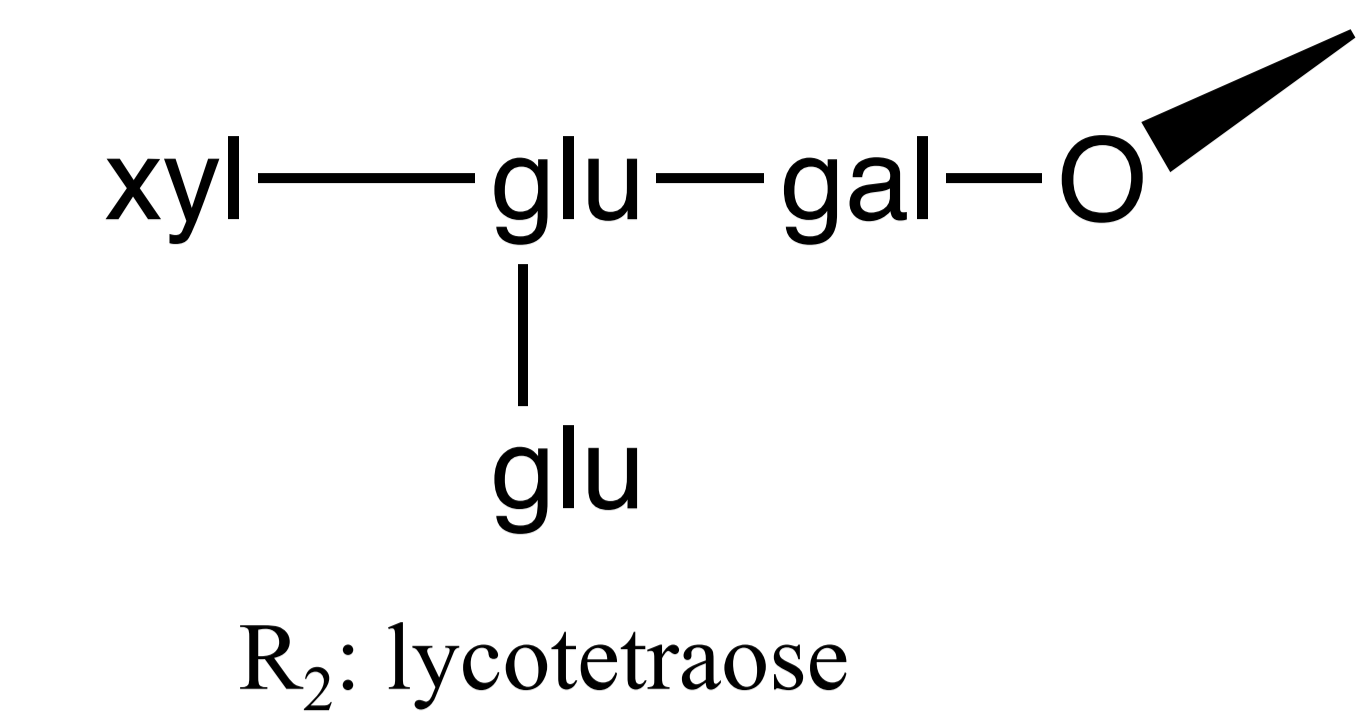
Analyte	Fresh market (n = 7)	Juice (n = 3)	Ketchup (n = 3)	Pasta sauce (n = 3)	Paste (n = 3)	Pizza sauce (n = 3)	Soup (n = 3)	Whole peeled (n = 3)
Serving size (g):	126	228.5	17	126	33	62	126	126
Esculeoside B	4.3±9.7 _a	3.3±3.00	0.3±0.6	5.9±7.2	3.6±0.8	1.8±0.6	2.0±2.1	21.8±10.3
Hydroxytomatine	297.9±248.1	54.3± 25.0	12.1±1.3	80.4±14.5	57.9±13.2	26.4±4.9	42.0±7.3	50.4±3.7
Dehydrolycoperoside F, G, or Dehydroesculeoside A	7.0±12.1	N.D. _b	N.D.	N.D.	N.D.	N.D.	N.D.	N.D.
Lycoperoside F, G or Esculeoside A	1589.4±1738.3	N.D.	N.D.	N.D.	N.D.	N.D.	N.D.	N.D.
Acetoxytomatine	30.6±31.6	17.4±8.6	3.1±2.0	20.3±15.8	25.0±4.1	9.4±3.0	10.0±4.8	1.8±3.2
Dehydrotomatine	4.1±3.0	41.0±29.0	5.7±0.7	41.3±14.9	28.2±3.9	19.4±2.0	31.6±7.8	11.3±5.9
Alpha-tomatine	64.5±56.0	1083.5±747.4	156.3±9.7	1109.9±390.8	889.5±119.4	524.7±85.5	964.3±62.5	338.4±156.5
Tomatidine	N.Q. _c	N.Q.	0.4±0.3	1.7±1.3	0.8±0.0	0.8±0.1	1.5±0.5	0.42±0.2
Tomatidenol	N.Q.	N.Q.	N.Q.	0.2±0.1	0.1±0.0	N.Q.	N.Q.	N.Q.
Total	3376.0±2886.3	1307.7±823.7	191.7±23.4	1541.9±410.3	1135.1±285.9	736.5±166.6	1126.3±34.4	1101.3±116.5

673 _aMean ± standard deviation674 _bNot detected675 _cNot quantified



R-GROUPS

R : R₁ or R₂
 R₁ : hydroxyl
 R₂ : lycotetraose
 R₃ : H or O-glucose
 R₄ : H or O-glucose



Monoisotopic Mass	Molecular Formula	tSGA	Variations in Isomerization and Functional Groups								
UNDECORATED F-RING			C3	C5:6 alkene	C22	C23	C24	C25	F-ring Decoration		
413.3294	C ₂₇ H ₄₃ NO ₂	Tomatidenol	R ₁	Y	S	-	-	S	-		
415.3450	C ₂₇ H ₄₅ NO ₂	Tomatidine	R ₁	Y	S	-	-	S	-		
1031.5301	C ₅₀ H ₈₁ NO ₂₁	Dehydrotomatine	R ₂	N	S	-	-	S	-		
1033.5457	C ₅₀ H ₈₃ NO ₂₁	Alpha-tomatine	R ₂	N	S	-	-	S	-		
HYDROXYTOMATINE			C3	C5:6 alkene	C22	C23	C24	C25	F-ring Decoration		
1049.5407	C ₅₀ H ₈₅ NO ₂₂	Neorickioside A	R ₂	N	S	S	-	R	OH (C23)		
		Neorickioside B	R ₂	N	S	S	-	S	OH (C23)		
ACETOXYTOMATINE			C3	C5:6 alkene	C22	C23	C24	C25	F-ring Decoration		
1091.5507	C ₅₂ H ₈₅ NO ₂₃	Lycoperoside A	R ₂	N	R	R	-	S	OAc (C23)		
		Lycoperoside B	R ₂	N	S	S	-	R	OAc (C23)		
		Lycoperoside C	R ₂	N	S	S	-	S	OAc (C23)		
GLUCOSYLATED (DEHYDRO)ACETOXYTOMATINE			C3	C5:6 alkene	C22	C23	C24	C25	F-ring Decoration		
1269.5985	C ₅₈ H ₉₅ NO ₂₉	Esculeoside A	R ₂	N	S	S	-	S	OAc (C23)	H (C24)	Oglu (C27)
		Lycoperoside F	R ₂	N	S	R	-	S	OAc (C23)	H (C24)	Oglu (C27)
		Lycoperoside G	R ₂	N	S	S	R	S	OAc (C23)	Oglu (C24)	H (C27)
1267.5828	C ₅₈ H ₉₃ NO ₂₉	Dehydroesculeoside A	R ₂	Y	S	S	-	S	OAc (C23)	H (C24)	Oglu (C27)
		Dehydrolycoperoside F	R ₂	Y	S	R	-	S	OAc (C23)	H (C24)	Oglu (C27)
		Dehydrolycoperoside G	R ₂	Y	S	S	R	S	OAc (C23)	Oglu (C24)	H (C27)
ESCUELOSDIE B			C3	C5:6 alkene	C22	C23	C24	C25	F-ring Decoration		
1227.5884	C ₅₆ H ₉₃ NO ₂₂	Esculeoside B	R ₂	N	-	-	-	-	F-ring rearrangement		

