Rapid assessment of ligand binding to the SARS-CoV-2 main protease by saturation transfer

difference NMR spectroscopy

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Abstract

Severe acute respiratory syndrome coronavirus 2 (SARS-CoV-2) is the etiological cause of the

coronavirus disease 2019, for which no effective therapeutics are available. The SARS-CoV-2 main

protease (M^{pro}) is essential for viral replication and constitutes a promising therapeutic target.

Recently, crystallographic fragment screening uncovered a large number of molecules binding at the

M^{pro} active site as well as at sites of potential allostery. However, no information on fragment affinity

is available. Here, we describe an efficient protocol for screening chemical compounds for binding to

M^{pro} using saturation transfer difference nuclear magnetic resonance (STD-NMR) spectroscopy. We

characterised the relative affinity of M^{pro}-binding fragments using STD-NMR, revealing a ~200-fold

difference in binding potential. Combined with crystallographic structures of these fragments bound

to M^{pro}, this information can assist ongoing drug design efforts.

Introduction

Infections by the severe acute respiratory syndrome coronavirus 2 (SARS-CoV-2) have resulted in over 400,000 deaths as of early June 2020 (1) and led to the coronavirus 2019 (COVID-19) pandemic (2-4). SARS-CoV-2 is a zoonotic betacoronavirus highly similar to SARS-CoV-1 and MERS-CoV, which caused outbreaks in 2002 and 2012, respectively (5-7). Much of SARS-CoV-2 biology has been described thus far by analogy to previously studied coronaviruses, including SARC-CoV-1 and MERS-CoV. SARS-CoV-2 encodes its proteome in a single, positive-sense, linear RNA molecule of ~30 kb length, the majority of which (~21.5 kb) is translated into two polypeptides, pp1a and pp1ab, via ribosomal frame-shifting (8, 9). Key viral enzymes and factors, including most proteins of the reverse-transcriptase machinery, inhibitors of host translation and molecules signalling for host cell survival, are released from pp1a and pp1ab via post-translational cleavage by two viral cysteine proteases (10). These proteases, a papain-like enzyme cleaving pp1ab at three sites, and a 3C-like protease cleaving the polypeptide at 11 sites, are primary targets for the development of antiviral drugs.

The 3C-like protease of SARS-CoV-2, also known as the viral main protease (M^{pro}), has been the target of intense study owing to its centrality in viral replication. M^{pro} studies have benefited from previous structural analyses of the SARC-CoV-1 3C-like protease and the development of putative inhibitors (11-14). The active sites of these proteases are highly conserved, and at least one family of SARS-CoV-1 3C-like protease inhibitors, comprising peptidomimetic α-ketoamides, has been shown to be effective against M^{pro}, thereby offering a promising lead for antiviral development (15). However, to date no M^{pro}-targeting inhibitors have been validated in clinical trials. Recently, structures of M^{pro} in complex with 74 fragments were resolved as part of a high-throughput screening using X-ray crystallography (16). Hits included covalently and non-covalently bound molecules targeting the protease active site, as well as the M^{pro} dimerisation interface and surface epitopes. These structures offer opportunities for the development of potent compounds that would directly inhibit M^{pro} activity or allosterically regulate it, for example by disrupting M^{pro} dimerisation, which is known to reduce activity of coronavirus 3C-like proteases (17). However, crystallographic structures of protein – ligand complexes are silent on compound affinities, which is key information to select the most promising leads. Further, the combined high concentration of protein (solid crystals) and of compounds (50 mM typical) in crystallographic fragment screening assays raises the possibility that some very weak binders may have been selected, which could lead inhibitor development towards blind alleys.

Here, we show that saturation transfer difference nuclear magnetic resonance (STD-NMR) spectroscopy can rapidly assess compound binding to M^{pro}. We use this method to derive the relative

58   affinities of non-covalent interactors identified by crystallographic fragment screening. This

59   information can aid selection of the most promising fragment 'hits' for further development.

Materials and Methods

Protein production and purification

A SARS-CoV-2 M^{pro} genetic construct including the viral protease, a glutathione S-transferase N-terminal tag, and a C-terminal His$_6$-tag (16) was transformed into *E. coli* strain Rosetta(DE3). Transformed clones were pre-cultured at 37 °C in lysogeny broth supplemented with appropriate antibiotics and grown to exponential phase for approximately 5 h. Starter cultures were used to inoculate 1 L of Terrific Broth Autoinduction Media (Formedium) supplemented with 10% v/v glycerol and appropriate antibiotics. Cell cultures were grown at 37 °C for 5 h and then cooled to 18 °C for 12 h. Bacterial cells were harvested by centrifugation at 5,000 x g for 15 min.

Cell pellets were resuspended in 50 mM trisaminomethane (Tris)-Cl pH 8, 300 mM NaCl, 10 mM imidazole buffer, incubated with 0.05 mg/ml benzonase nuclease (Sigma Aldrich) and lysed by sonication on ice. Lysates were clarified by centrifugation at 50,000 x g at 4 °C for 1 h. Lysate supernatants were loaded onto a HiTrap Talon metal affinity column (GE Healthcare) pre-equilibrated with lysis buffer. Column wash was performed with 50 mM Tris-Cl pH 8, 300 mM NaCl and 25 mM imidazole, followed by protein elution using the same buffer and an imidazole gradient from 25 to 500 mM concentration. The His$_6$-tag was cleaved using home-made recombinant human rhinovirus (HRV) 3C protease. The HRV 3C protease, His$_6$-tag and further impurities were removed by a reverse HiTrap Talon column. Flow-through fractions were concentrated and applied to a Superdex75 26/600 size exclusion column (GE Healthcare) equilibrated in NMR buffer (see below).

Nuclear magnetic resonance (NMR) spectroscopy

All NMR experiments were performed using the Oxford Biochemistry 950 MHz (22.3 T) solution-state instrument comprising an Oxford Instruments superconducting NMR magnet, Bruker Avance III console and TCI probehead. A Bruker SampleJet sample changer was used for sample manipulation. Experiments were performed and data processed using TopSpin (Bruker). Samples comprised 10 μM M^{pro} alone, 0.8 mM of compounds alone, or 10 μM M^{pro} mixed with 0.8 mM compounds, formulated in NMR buffer (150 mM NaCl, 20 mM Na$_2$HPO$_4$ pH 7.4) supplemented with 10% v/v D$_2$O and deuterated dimethyl sulfoxide (D$_6$-DMSO, 99.96% D, Sigma Aldrich) to 5% v/v final D$_6$-DMSO concentration. Sample volume was 140 μL and samples were loaded in 3 mm outer diameter SampleJet NMR tubes (Bruker) placed in 96-tube racks. NMR tubes were sealed with POM balls.

STD-NMR experiments were performed at 10 °C using a pulse sequence described previously (18) and an excitation sculpting water-suppression scheme (19). Protein signal were suppressed in STD-NMR by the application of a 30 msec spin-lock pulse. We collected time-domain data of 16,384

complex points and 41.6 µsec dwell time (12.02 kHz sweepwidth). Data were collected in an interleaved pattern, with on- and off-resonance irradiation data separated into 16 blocks of 16 transients each (256 transients per irradiation frequency). Transient recycle delay was 4 sec and on- or off-resonance irradiation was performed using 0.1 mW of power for 3.5 sec at 0.5 ppm or 26 ppm, respectively, for a total experiment time of approximately 45 minutes. Reconstructed time-domain data from the difference of on- and off-resonance irradiation (STD spectra) or only the off-resonance irradiation (reference spectra) were processed by applying a 2 Hz exponential line broadening function prior to Fourier transformation. Phasing parameters were derived for each sample from the reference spectra and copied to the STD spectra. ^{1}H peak intensities were integrated in TopSpin using a local-baseline adjustment function. For the calculation of STD parameters and for estimating ligand concentrations we integrated ^{1}H peaks originating from ligand aromatic groups, as peak dispersion in this area of the spectra allowed accurate consideration of the number of underlying ^{1}H atoms. Ligand concentrations are reported relative to the most abundant compound.

Ligand binding was quantified by calculating the ratio of integrated signal for a peak or peaks in the ^{1}H STD spectrum over that of the reference spectrum (STD_{ratio}). From this the STD amplification factor (STD_{amp}) is derived as $STD_{amp} = STD_{ratio} \; x \; ligand \; excess$. STD_{amp} is inversely proportional to the ligand dissociation constant, K_d, as $STD_{amp} \propto \frac{[L]}{K_d+[L]}$, where [L] is the ligand concentration.

Notes

Compounds were procured from Enamine (Riga, Latvia) in the form of a plated library (DSI-poised), dissolved in D_6-DMSO at 500 mM nominal concentration. 1 µL of dissolved compounds was aspirated from the library and immediately mixed with 9 µL of D_6-DMSO for final fragment concentration of 50 mM, from which NMR samples were formulated. Crystallographic structures of M^{pro} with bound fragments were visualised using PyMOL (20) and correspond to Protein Data Bank entries 5R81 (ligand x0195), 5REC (x0390), 5R83 (x0434) and 5RGQ (x1086) (16).

Results

STD-NMR assays of M^{pro} ligand binding

M^{pro} forms dimers in crystals (15), likely with a sub-μM solution dissociation constant (K$_d$) based on

analogy to previously studied 3C-like coronavirus proteases (17). At the 10 μM protein concentration

of our NMR assays M^{pro} is thus expected to be dimeric with an estimated molecular weight of

approximately 68 kDa. Despite the relatively large size of M^{pro} for NMR, we were able to record one-

dimensional ^{1}H spectra of M^{pro} alone showing the presence of multiple up-field shifted (<0.5 ppm)

resonances, likely corresponding to methyl groups in the protein hydrophobic core, as well as down-

field shifted resonances (>8.5 ppm) originating from hydrogen-bonded amides (Fig. 1A). In addition to

demonstrating that M^{pro} is folded under the conditions tested, these spectra allowed us to identify

multiple resonance frequencies where irradiation for STD-NMR experiments might be applied in a

protein-specific manner.

As the vast majority of compounds assessed feature aromatic groups (see below), we chose to

irradiate M^{pro} at up-field shifted frequencies in STD-NMR experiments. Trials revealed that irradiation

at 0.5 ppm frequency produced unambiguous ^{1}H signals in STD spectra from ligands bound to M^{pro},

while simultaneously avoiding ligand excitation that would yield false-positive ^{1}H signals in the

absence of M^{pro} (Fig. 1B). Further, contributions of small molecules not binding to M^{pro}, such as DMSO,

to the STD spectra were pseudo-dispersive in nature, due to imperfect signal subtraction, while true

ligands produced peaks in STD with absorptive ^{1}H lineshapes. We conclude that STD-NMR provides an

efficient method for detecting ligand binding to M^{pro}.

Relative affinities of M^{pro}-binding fragments

To better evaluate the information provided by STD-NMR, we assessed using this method the

binding of compounds previously seen complexed to M^{pro} in crystallographic fragment screening

experiments (16). As STD-NMR signal relies on ligand exchange between protein-bound and -free

states, we focused on non-covalent M^{pro} interactors. 39 such compounds are part of the DSI-poised

fragment library to which we were given access, comprising 17 active site binders, two compounds

targeting the M^{pro} dimerisation interface and 20 molecules binding elsewhere on the protein surface.

We initially recorded STD-NMR experiments in the absence of M^{pro} to confirm that under these

conditions we obtained no or minimal STD signal from the ligands, and to verify ligand identity from

the reference ^{1}H spectra. Five ligands were either absent from our DSI-poised library or produced

reference ^{1}H spectra inconsistent with the compound chemical structure; these ligands were not

evaluated further. Samples of 10 μM M^{pro} and 0.8 mM nominal ligand concentration were formulated

from the remaining 34 compounds, and STD-NMR spectra were recorded. The folded state of M^{pro} in

the presence of each ligand was verified by collecting ^{1}H NMR spectra similar to that of Fig. 1A from

all samples ahead of STD-NMR experiments.

We observed large variation in ^{1}H STD signal intensity in the presence of M^{pro} across ligands, with

many ligands producing little or no STD signal, suggesting substantial differences in compound affinity

for the protease (Table 1). However, we also recorded up to ~15-fold differences in the per-^{1}H

integrated intensity of reference ligand spectra, indicating significant variation in ligand

concentrations in solution. Such concentration differences may arise from errors in sample

formulation or from concentration inconsistencies in the ligand library. To evaluate the former we

integrated the residual ^{1}H signal of deuterated DMSO in our reference spectra, which should have

been constant as DMSO was added alongside ligands in the samples, and found it to vary by less than

35% across any pair of samples (11% average deviation). Thus, although sample formulation did

introduce ligand concentration errors by up to ~1/3, the extend of these errors did not account for

the up to ~15-fold differences in relative ligand concentrations.

To factor in these concentration differences we calculated the STD amplification ratio (STD$_{amp}$)

upon M^{pro} binding for all compounds, which is related to ligand affinity. Strikingly, M^{pro} binding varied

by ~200-fold across ligands (Fig. 2A), with the strongest interactions recorded for compounds x0434,

x0390, x1086 and x0195 (Table 1). Large differences in ligand affinity were also readily apparent by

incubating M^{pro} with mixtures of two different compounds, where STD signal for only one compound

was visible (Fig. 2B). We surmise that the relative affinities of small molecules towards M^{pro} can be

readily assessed by STD-NMR, and that M^{pro}-binding fragments identified by crystallography vary

substantially in their interaction affinities for this protease.

Discussion

Fragment-based screening is a tried and tested method for reducing the number of compounds that need to be screened against a specific target in order to sample chemical space (21). Combined with X-ray crystallography, which provides information on the target site and binding pose of ligands, identified fragments are thought of as 'bricks' that, when combined, can yield potent and specifically-interacting compounds. However, crystallographic structures do not report on ligand affinity, and obtaining fragment affinities is often difficult by biophysical methods, such as calorimetry, due to the weak nature of these interactions. However, NMR methods, such as STD-NMR, provide highly powerful and sensitive tools for identifying ligand binding to macromolecules, even when interaction affinities are in the mM K_d range (22). Here, we showed that STD-NMR is a suitable method for characterising the binding of fragments to M^{pro}, an essential protease of SARC-CoV-2.

Despite its considerable molecular weight (~68 kDa dimer) and the low protein concentration (10 μM) of samples in our study, M^{pro} yielded ^{1}H NMR spectra of high quality and with strong signal/noise in relatively little acquisition time (Fig. 1A). Importantly, we observed no protein precipitation during sample preparation even in M^{pro} concentrations of hundreds of μM, and samples were stable over time yielding highly reproducible protein ^{1}H spectra during the course of several days. These observations suggest that this viral protease is amenable to further NMR studies, such as experiments to characterise the M^{pro} dynamic behaviour during catalysis.

STD-NMR allowed us to screen for ligand binding to M^{pro} using relatively small amounts of protein and in under one hour of experiment time per ligand. Ligands interacting strongly with M^{pro} yielded high signal/noise STD data (Fig. 1B), which suggests that further improvements in our protocol are possible. For example, we anticipate that halving the amount of M^{pro} used in STD-NMR experiments, or reducing the experiment time to ¼ of its current duration, will still provide us with sufficient signal to detect interacting compounds. Although we refrained from doing so in this study, as our goal was to characterise the interaction of even weekly-binding M^{pro} ligands, such modifications should be considered if STD-NMR was used in a high-throughput manner to identify compounds binding to this protease.

Our work showed that fragments identified by previous crystallography-based assays as binding to M^{pro} in a non-covalent manner (16) do so with a wide-range of relative affinities. Four ligands, x0195, x0390, x0434 and x1086, were shown to be most promising on the basis of their M^{pro} interaction strength in solution. Of these, both x0195 and x0434 bind at the M^{pro} active site (Fig. 2C). However, the x0196 and x0434 binding poses overlap only in part, with x0434 being in contact with the catalytic cysteine residue and x0195 in the active site periphery. Thus, a likely direction towards the synthesis

of a potent M^{pro} inhibitor would be combining these two fragments, thereby exploiting a larger
fraction of the active site interface.

Interestingly, both x0390 and x1086 bind at the M^{pro} surface at the junction between the protease
N-terminal catalytic subdomain and its C-terminal subdomain (Fig. 2D). Previous work showed that
coronavirus 3C-like proteases are extensively regulated by the C-terminal subdomain, likely as this
subdomain enhances protease dimerisation that, in turn, increases catalytic activity by several fold
(23). The x1086 binding site locates proximal to the interface between the two M^{pro} protomers; thus,
further development of this ligand holds potential for directly disrupting the M^{pro} dimer and reducing
protease activity. However, x0390-derivatives may also yield means towards the allosteric control of
M^{pro}, by altering the relative conformation of the catalytic and regulatory subdomains, thereby
impacting on protease function.

In conclusion, we presented here a protocol for the rapid assessment of SARS-CoV-2 M^{pro} ligands
using STD-NMR spectroscopy and evaluated the relative affinities of fragments seen to bind this
protease. Although development of novel antivirals to combat COVID-19 is still at an early stage, we
hope that this information will prove valuable to groups working towards such treatments.

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Figures and captions

Figure 1: 1D and STD-NMR spectra of SARS-CoV-2 M^pro. A) 1D ^{1}H NMR spectrum of recombinant SARS-CoV-2 M^pro, recorded from a 10 µM protein concentration sample in a 5 mm NMR tube at 25 °C using an excitation sculpting water-suppression method (19). 512 acquisitions with recycle delay of 1.25 sec were averaged, for a total experiment time of just over 10 min. Data were processed with a quadratic Sine function prior to Fourier transformation. The broad areas where ^{1}H from protein chemical groups resonate are indicated by brackets. The residual signals of H_2O and Tris, from the protein purification, as well as the position where on-resonance irradiation was applied for STD experiments, are indicated. B) Vertically offset ^{1}H STD-NMR spectra from ligand x0434 binding to M^pro. The reference spectrum is shown on top (black) with the x0434, H_2O and DMSO ^{1}H resonances indicated. The STD spectrum of x0434 in the presence of M^pro is shown in the middle (red), while that in the absence of M^pro is at the bottom (green). STD spectra are scaled up 64x compared to the reference spectrum.

Figure 2: Assessment of non-covalent ligand binding to M^{pro}. A) Plot of the STD amplification factor (STD$_{amp}$) versus relative ligand concentration for non-covalent fragments seen as binding to M^{pro} in crystallographic screening (16). STD$_{amp}$ is inversely proportional to K$_d$ for a given ligand concentration. The strongest interacting ligands are boxed. B) Magnification of the aromatic ^{1}H area of STD-NMR spectra from a mixture of ligands (x1237 and x1132, each at 0.8 mM concentration) in the presence of M^{pro}. Both ligands interacted with the same M^{pro} site in crystallographic structures (16). Reference spectrum is shown on top (black) and STD spectrum at the bottom (red, scaled up by 64x). The resonances originating from each ligand are indicated. Whereas x1237 binds to M^{pro} under our solution conditions and gives rise to STD peaks with absorptive lineshapes, x1132 does not and only produces dispersive peak artefacts. C) Magnification of the M^{pro} active site and position of ligands x0434 and x0195 within it from crystallographic structures (16). The sidechain of the catalytic cysteine residue is shown as spheres; ligands are shown as sticks. D) View of a M^{pro} protomer (cyan) with the catalytic (N-terminal) and regulatory (C-terminal) subdomains indicated. The second M^{pro} protomer in the dimer is shown in grey. Ligands x0390 and x1086 (sticks) bind at the junction of the two protease subdomains, with x1086 facing the dimerisation interface.

 Tables

329 *Table 1: M^{pro}-binding fragments assessed by STD-NMR*

Ligand	M^{pro} site	SMILES and structure	Relative ligand concentration	STDamp ($\times 10^{-3}$)
x0072	Active	CS(=O)(=O)NCCC=1C=CC=CC1	0.68	4.24
x0107	Active	CC(=O)NC=1C=NC=CC1C	0.71	3.98
x0161	Active	COC(=O)C=1C=CC(=CC1)S(=O)(=O)N	0.66	3.69
X0165	Surface	CN1CCN(CC1)C(=O)NC=2C=CC=CC2	0.29	0.07
X0177	Surface	NC(=O)C1CCN(CC1)C(=O)NC=2C=CC=CC2	0.27	1.19
x0194	Surface	CC(=O)NC=1C=CC(OC=2N=CC=CN2)=CC1	0.06	1.97
x0195	Active	CN1CCCC=2C=CC(=CC12)S(=O)(=O)N	0.20	9.09

x0305	Active	CCNC=1C=CC(C#N)=CN1	0.61	5.29
X0336	Surface	CC(=O)NCC=1C=CC(=CC1)S(=O)(=O)N	0.44	0.25
X0350	Surface	FC=1C=CC=C(CNCC2=CC=CO2)C1	0.20	2.49
x0354	Active	CN1CCN(CC1)C(=O)COC=2C=CC(C)=CC2	0.66	2.03
X0376	Surface	O=C(N1CCCCCC1)C=2C=CC=3OCOC3C2	1.00	3.35
x0387	Active	OC1CCN(CC=2C=CSC2)CC1	0.76	2.11
x0390	Surface	OC=1C=CC=CC1CNC2=NC=3C=CC=CC3N2	0.28	12.86
x0395	Active	CC1=NN=C(CN2CCC=C(F)C2)S1	0.99	3.13

x0397	Active	CN(CC=1C=C(C)ON1)C(=O)NC2CC2	0.67	1.24
x0398	Surface	C(CN1CCOCC1)SC=2C=CC=CC2	0.16	1.19
x0426	Active	FC=1C=CC=CC1C(=O)NCCC=2C=CN=CC2	0.21	3.60
x0434	Active	O=C(NC=1C=CC=CC1)NC=2C=CC=NC2	0.29	15.03
x0464	Surface	CC1C(O)CCCN1CC=2C=CC=CC2	0.46	0.47
x0478	Surface	COC(=O)C=1C=CC=C(NS(=O)(=O)C)C1	0.39	2.56
X0499	Surface	NC(=O)C=1C=CC(NC(=O)[C@@H]2CCCO2)=CC1	0.20	0.52
x0540	Active	O=C(NCCC=1C=CN=CC1)NC2CCCCC2	0.21	3.59

x0669	Surface	ClC=1C=CC=C(CN2CCOCC2)C1	0.16	1.31
x1086	Dimer	CC=1C=C(F)C=CC1CS(=O)(=O)N	0.45	12.18
x1101	Surface	CC(C)N(C)C=1N=CN=C2N(C)N=CC12	0.22	0.57
x1119	Surface	NC=1C=CC(=CC1)S(=O)(=O)NC=2C=CC=CN2	0.43	2.64
x1132	Surface	CC1CN(CCO1)C(=O)CN2C=CC=N2	0.39	1.13
x1163	Surface	OCC1CN(CC=2C=CC=CC2)CCO1	0.42	0.94
x1187	Dimer	CN1C=CC(=N1)C(=O)NC[C@@H]2CCCO2	0.30	2.04
x1226	Surface	CCNCC1=CN(C)N=N1	0.19	1.99

x1235	Surface	COC(=O)NC=1SC(C)=NC1C=2C=CC=CC2	0.38	3.88
x1237	Surface	CS(=O)(=O)CC1=NC=2C=CC=CC2N1	0.42	6.41
x1249	Active	O=C(NCC=1C=CC(C#N)=CC1)N2CCOCC2	0.25	1.04