Unifying framework for the diffusion of microscopic particles in mucus

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Abstract

Mucus is a fluid that protects animals against pathogens while promoting interactions 10 with commensal microbes. Changes in the diffusivity of particles in mucus alter viruses 11 infectivity, the efficiency of bacterial pathogens to invade a host, and the effectivity of drug 12 delivery. Multiple physicochemical properties modulate the diffusion of microscopic particles 13 in mucus, but their combined effect is unclear. Here, we analyzed the impact of particle size, 14 15 charge, chemistry, anomalous diffusion exponent, and mucus composition in the diffusivity of particles from 106 published experiments. We used a time window sampling of one second 16 to define a consistent, effective diffusion across experiments. The effective diffusion spanned 17 seven orders of magnitude from 10^{-5} to $10^2 \,\mu\text{m}^2/\text{s}$. The anomalous exponent was the strongest 18 predictor among all variables tested. It displayed an exponential relationship with the effective 19 diffusion that explained 90% of the empirical data variance. We showed that the relationship 20 and dominance of the anomalous diffusion exponent resulted from a general mathematical 21 relationship obtained from first-principles for any subdiffusion mechanism. Our derivation 22 demonstrated that the generalized diffusion coefficient is not a measurable physical quantity 23 and must be replaced by the length and time scales associated with the underlying mobility 24 mechanisms. This led us to a fundamental reformulation of the classic subdiffusion equation, 25 which calls for a reinterpretation of anomalous diffusion in physical systems. We also discussed 26 how our results impact the characterization of microscopic particle diffusion in mucus and other 27 hydrogels. 28

Mucus is a complex fluid secreted by animals. It protects organs against the invasion of 29 pathogens and promotes the interaction with commensal microbes (Bäckhed et al. 2005; Bakshani 30 et al. 2018; Silveira and Rohwer 2016). Mucins—a characteristic component of mucus—are glyco-31 proteins that form a polymeric mesh in mucus (Spagnolie 2015). Changes in the mucin network 32 alter the diffusion of microscopic particles in mucus with disparate outcomes for the animal host. 33 Low pH thickens mucus, reducing, for example, the diffusion and infection rate of viruses like HIV 34 (Lai et al. 2009). Interaction with mucins also alters the diffusivity of particles in mucus. Com-35 mensal viruses that infect bacteria and reside in the gut, for instance, display immunoglobulin-like 36 domains that are attracted to mucins, which reduces their diffusivity and increase their infectivity 37 against bacteria (Barr et al. 2013, 2015). The regulation of particle diffusion in mucus is thus paramount for animal health, and the enhancement of diffusivity is also key for the delivery of 39 medical drugs in the body (Cone 2009). The combined effect of these different physicochemical factors, however, remains puzzling. 41

Small biomelecules and biomolecular complexes tend to diffuse more readily through mucus, 42 while larger particles are caught in the mucin network (Amsden and Turner 1999; Cone 2009). On 43 the other hand, non-adhesive polystyrene particles with a diameter of 500 nm diffuse faster than 44 smaller particles (200 nm) of the same type (Lai et al. 2007). Therefore, parameters other than 45 particle size must play a role. Neutrally charged particles, for instance, display higher diffusivity 46 than negatively and positively charged particles of the same size in mucus with a net negative 47 charge (Abdulkarim et al. 2015; Arends et al. 2013; Hansing et al. 2016; Lieleg et al. 2010; Li et al. 48 2013). Increase of salt concentration shields charged particles, leading to diffusivities similar to neutrally charged particles (Arends et al. 2013; Lieleg et al. 2010; Hansing et al. 2016). Instead, 50 low pH increases the distribution of negative charges in mucins altering the electrostatics as well 51 as viscoelasticity of mucus, reducing diffusivity for most particles (Celli et al. 2009; Lai et al. 2009; 52 Lieleg et al. 2010; Spagnolie 2015; Suk et al. 2011). Each of these studies show how different 53 physical properties can controll the diffusion of microscopic particles in mucus, but the emerging 54 picture is complex, and it is unclear if any of these physical factors is more dominant than others. 55

To assess the combined impact of each factor, we studied twenty-three published articles measuring the diffusion of particles in mucus or mucus-like hydrogels. Ten studies contained diffusion data that could be compared at the same time scale for spherical nanoparticles (Abdulkarim et al. 2015; Barr et al. 2015; Lai et al. 2007, 2009; Lieleg et al. 2010; Olmsted et al. 2001; Newby et al. 2017; S.Schuster et al. 2013; Suk et al. 2011; Yildiz et al. 2015). Using WebPlotDigitizer (Rohatgi 2019), we extracted 106 measurements of effective diffusion, measured at a window time of one second, that is,

$$D_{eff} = \frac{\langle MSD \rangle}{2k\,\Delta t_{eff}} \ . \tag{1}$$

Here $\langle MSD \rangle$ was the ensamble mean squared displacement for each particle tracking experiment, k was the dimensions of the particle diffusivity, and $\Delta t_{eff} = 1$ sec was the effective sampling time window (Huang et al. 2013). In all measurements, we estimated particle hydrodynamic diameter, particle type, mucus source, dominant mucin expression, and temperature. When possible, we extracted or derived the anomalous diffusion exponent (α), particle charge, mucus pH, mucus salt concentration, and mucin concentration. The anomalous exponent was obtained from the classic subdiffusion equation:

$$\langle MSD(\Delta t) \rangle = 2kD_{\alpha}\Delta t^{\alpha} \quad , \tag{2}$$

⁵⁶ Here D_{α} is the generalized diffusion constant and Δt is the sampling time window (Metzler et al. ⁵⁷ 2014).

Table 1 displays the ranges obtained for each physical parameter, and Table S.1 contains the full data set. The effective diffusions spanned seven orders of magnitude, from $\sim 10^{-2} \ \mu m^2/s$ to

 $10^5 \ \mu m^2/s$. Particle diameter (d) spanned three orders of magnitude, from 1 nm to 1,300 nm; 60 three quarters of the particles (n=80, 75%) had diameters greater than 100 nm. The anomalous 61 exponent (α) ranged from strongly subdiffusive ($\alpha \approx 0.1$) to purely diffusive ($\alpha \approx 1$), but it was 62 obtained only for a third of the dataset (n = 33, 30%). The zeta potential (ζ) measured the 63 effective surface charge of particles in solution (Kumar and Dixit 2017). The values ranged from 64 -70 mV to +40 mV and were obtained for half of the dataset (n = 57, 52%). The temperature 65 ranged was narrow, 295 K to 310 K. The pH ranged from mildly acidic (pH = 3.0) to slightly 66 alkaline (pH = 7.4). Mucus experiments were associated to cell lines from four sources: human 67 respiratory, human cervix, pig gastric, pig intestines. The dominant mucins were MUC5B, MUC2, 68 and MUC5C. A third of the experiments were conducted in artificial mucus-like hydrogels. 69

The non-parametric statistical method random forest was applied to identify the most relevant 70 variables impacting the effective diffusion. The variables pH, mucus concentration, salt concen-71 tration, and MUC5B were omitted because they were missing for most data in the multivariate 72 analysis. The selection of variables was obtained in two rounds, discarding not statistically sig-73 nificant variables (p-value > 0.05) in each round. This led to five significant variables (Figure 1). 74 The anomalous diffusion exponent (α) was the most dominant variable with an average percentage 75 increase in mean square error (%MSE) of $22.4(\pm 3.2)$ % (S.D.) (p-value = 0.0099). The second 76 most dominant variable was particle type, followed by zeta potential, mucus source, and dominant 77 78 mucin.

When analyzing the selected variables individually, the anomalous diffusion exponent displayed 79 by far the strongest correlation with the effective diffusion (non-parametric Spearman correlation 80 ho = 0.93, p < 2.2E-16***, n = 39). The effective diffusion increased exponentially with the 81 anomalous diffusion exponent (Figure 2a). The linear regression for the transformed data (log-82 linear) explained 89% of the variance (slope= 5.3 ± 0.3 , p-value $< 2.0 \cdot 10^{-16 \star \star \star}$, $R^2 = 0.89$, 83 least-squares method). The anomalous exponent, however, was only reported for $\sim 37\%$ (n=33) of 84 the data, which included carboxylated, pegylated, and viral particles. An inverse statistical model 85 was fitted to estimate the mean anomalous exponents as a function of the effective diffusions for 86 the remaining 63% of the data, corresponding to amine, chitosan, antibodies, and proteins particles (n = 67). Particles with effective diffusions above $D_{eff} > 10 \ \mu m^2/s$ were predicted to display regular diffusion ($\alpha \sim 1$). 89

To elucidate the physical origin of the dominance of the anomalous exponent (α), its relationship with the effective diffusion, D_{eff} , was derived from Eqs. (1) and (2):

$$D_{eff} = \frac{D_{\alpha}}{\Delta t_{eff}} \Delta t_{eff}^{\alpha} \quad . \tag{3}$$

The effective diffusion, thus, has a factor that varies exponentially with the anomalous exponent 90 (α). However, it also depends on the generalized diffusion (D_{α}), which is implicitly a function of 91 the anomalous diffusion exponent as well as particle and fluid properties, and its functional form 92 changes depending on the specific underlying subdiffusion mechanism (Metzler et al. 2014; Joiner 93 et al. 2019). Our meta-analysis contained a broad range of data (Table 1), including particles with 94 different chemistry, mucus of different types, different physicochemical conditions, and independent 95 groups carrying different experimental implementations. Therefore, it was not obvious how the 96 generalized diffusion would be changing in each case, and Eq. (3) was not sufficient to justify the 97 dependence and dominance of α in determining the mobility of particles in mucus. To understand 98 this phenomenon, the generalized diffusion had to be analyzed further. 99

The units of the generalized diffusion constant, D_{α} , depend on the anomalous exponent. In our study, these units were $\mu m^2/s^{\alpha}$. The anomalous diffusion exponent, as any other physical quantity, has an associated uncertainty (error or standard deviation) (Taylor 1997). Thus, the units of D_{α} are uncertain. In other words, the generalized diffusion constant is not measurable. The fact that D_{α} is not a physical quantity has been previously overlooked and mandates a revision of the classic subdiffusion equation, Eq. (2).

To reformulate the subdiffusion equation, we split the generalized diffusion into the characteristic length scale (L_D) and characteristic time scale (t_D) associated to the physical mechanism responsible for the mobility of the particle:

$$D_{\alpha} = \frac{L_D^2}{t_D^{\alpha}} \quad . \tag{4}$$

This ansatz was combined with the classic subdiffusion equation, Eq. (2), obtaining:

$$\langle MSD(\Delta t) \rangle = 2kL_D^2 \left(\frac{\Delta t}{t_D}\right)^{\alpha}$$
 (5)

This reformulated subdiffusion equation is valid for window times larger than the characteristic mobility time scale, $\Delta t \ge t_D$. For smaller window times, the underlying mobility mechanism will dominate, requiring a different formulation for the displacement (Joiner et al. 2019).

The reformulated subdiffusion equation, Eq. (5), was combined with the definition of the effective diffusion, Eq. (1), obtaining

$$D_{eff}(\alpha) = \frac{L_D^2}{\Delta t_{eff}} \left(\frac{\Delta t_{eff}}{t_D}\right)^{\alpha} .$$
(6)

The effective diffusion, thus, depends exponentially on the anomalous diffusion exponent, α , explaining the relationship observed empirically for the effective diffusion of particles in mucus (Figure

2). To investigate the origin of the dominance of the anomalous diffusion exponent in the variation of the effective diffusion across multiple scales, we investigated the logarithm of the effective diffusion equation, Eq. (3):

$$\log D_{eff}(\alpha) = \log \frac{L_D^2}{\Delta t_{eff}} + \alpha \log \frac{\Delta t_{eff}}{t_D} .$$
(7)

For a fix time window, Δt_{eff} , the change of the effective diffusion with respect the anoma-109 lous diffusion is $\partial \log D_{eff}/\partial \alpha = \log (\Delta t_e f f/t_D)$, while the impact of the change in the 110 characteristic mobility length and time scales are, respectively, $\partial \log D_{eff} / \partial \log L_D = 2$ and 111 $\partial \log D_{eff} / \partial \log t_D = -\alpha$. The changes were evaluated with respect the logarithms of the length 112 and time scales to obtain results independent of the measuring units. The change with respect 113 the length scale is constant with a value of two, while the change with resepct the time scale is 114 bounded within an absolute value of one. Thus, for sampling time windows that are more than 115 two orders of magnitude larger than the characteristic mobility time scale, $\Delta t_{eff}/t_D \gg 10^2$, the 116 anomalous diffusion would be the dominant physical factor determining the change in the effective 117 diffusion. 118

This hypothesis was confirmed for the collected diffusion data in mucus, Eq. (7), by extracting 119 the average mobility length and time scales from the empirical data (Figure 2a). This led to $L_D \sim 3$ 120 nm and $T \sim 5 \,\mu s$. The sampling window time was $\Delta t_{eff} = 1 \, s$. Therefore, $\Delta t_{eff}/t_D \sim 10^6 \gg 10^2$, 121 satisfying the condition for the dominance of the anomalous diffusion exponent derived above. To 122 justify the values obtained for the average length, L_D and time scales t_D , it was necessary to 123 look into the underlying mechanisms fueling the mobility of the particles. A given mechanism 124 would propel the particles with a velocity v_D for the characteristic time t_D . This defines the 125 characteristic length scale $L_D \sim v_D t_D$. In all experiments analyzed, the particles were passive, 126 acquiring a transient velocity fueled by the transfer of kinetic energy from the thermal buffeting 127 of the fluid, that is, $v_D^2 \sim k_B T/m$, where k_B is the Boltzmann constant, and m is the mass of 128 the particle. Mucus is a viscous fluid, and this velocity will dissipate with a characteristic time 129 $t_D \sim m/\gamma$, where γ is the friction coefficient. Not surprisingly, this leads to the Stokes-Einstein 130 equation for the underlying characteristic diffusion, $D \sim L_D^2/t_D \sim k_B T/\gamma$. For the typical mid-size 131 particle in the data analyzed, $d \sim 100$ nm, the relaxation time is $t_D \sim 1 \ \mu s$ and the microscopic 132 diffusion is $D \sim 1 \ \mu^2/s$. This leads to the characteristic mobility length scale $L_D \sim \sqrt{D t_D} \sim 1$ 133 nm (Joiner et al. 2019). Therefore, the estimated values for L_D and t_D are consistent with the 134 average empirical values obtained for Eq. (7), supporting our reformulated subdiffusion framework. 135 Thus, the problem of characterizing the diffusion of a particle in mucus reduces to identifying 136 the physical factors that regulating the anomalous exponent. These factors depend on the specific 137

mechanism hindering the regular diffusion (Metzler et al. 2014). There are at least two mechanism that may play an important role in mucus. First, microscopic particles can bind to the mucin fibers that constitute mucus leading to subdiffusion (Barr et al. 2015). Second, mucin fibers form a polymeric mesh that can trap particles as observed in other hydrogels (Wong et al. 2004). Below we discuss the physical factors that control the subdiffusion exponent in each case.

Binding to mucins in mucus does not necessarily lead to subdiffusion. If a particle has a single binding site with a characteristic binding time t_b , this will elongate the characteristic diffusion time, $t_D \sim t_r + t_b$ leading to the microscopic diffusion $D v_D^2 t_r^2 / t_D \sim f_r k_B T/m$. Therefore, the diffusion will be rescaled by the fraction of time spent in the relaxation of the thermal energy, $f_r = t_r/(t_r + t_b)$, without altering the anomalous exponent. However, if more than one region of the particle can bind to mucins simultaneously and the number of number of regions bound to mucines vary stochastically, an increase of the binding time beyond the sampling time, $t_b \gg \Delta t_{eff}$, would lead to an effective power law distribution of binding times with no apparent characteristic binding time Xu et al. 2011. The emergence of long-tailed attachment time distributions leads to subdiffusion. The anomalous exponent, α , is equal to the exponent, ν , of the asymptotic approximated power-law distribution of attachment times (Metzler et al. 2014; Joiner et al. 2019). The generalized subdiffusion constant is given by

$$D_{\alpha} = \frac{D \tau_D}{\tau_D^{\alpha}} \quad . \tag{8}$$

This was obtained using a continuous-time random walk approximation (Joiner et al. 2019). Here, D is the diffusion of the particle in the absence of interactions with mucins, tau_D is the average diffusion time of a particle before attaching again to a mucin fiber. This result is consistent with the ansatz that we introduced in Eq. (4). The anomalous exponent can also be related to the average minimum time of a particle attached to a mucin fiber (τ_0):

$$sinc^{1/\alpha}\alpha = \frac{\tau_0}{\tau_D} \quad . \tag{9}$$

This mechanism indicates that particle-mucin affinity will dominate the effective diffusion of a particle in mucus. Unfortunately, the experiments analyzed did not explored the particle affinities to mucus explicitly.

The microenvironment trapping mechanism was observed in F-actin networks, where microscopic tracers were shown to follow anomalous diffusion. The anomalous exponent was a linear function of the ratio between the particle size (d) and network's mesh size (ξ) (Wong et al. 2004). The empirical dependency obtained was $\alpha \approx 1$ for $d/\xi < 0.1$, $\alpha \approx -1.25 d/\xi + 1.38$ for $0.1 < d/\xi < 1.1$, and $\alpha \approx 0.1$ for $d/\xi > 1.1$. Thus, particles with a size that is 10% of the

mesh size or smaller diffused normally, while particles with a size similar or larger to the mesh or 151 displayed a reduced diffusitivity with a low anomalous exponent. This phenomenon was justified 152 qualitatively assuming an elastic energy threshold that is eventually large enough to overcome the 153 free energy barrier and push the particle over a new microenvironment. The specific parameters 154 of the relationship were not derived from first principles, but one would expect a similar behavior 155 in mucus. This mechanism indicates that the effective diffusion of relatively large particles will 156 be severely affected independently on particle-mucin interactions, that is, the particle chemistry. 157 Unfortunately, the mesh size was not measured or reported in most experiments reviewed in our 158 study. 159

Particle size was not an apparent significant predictor in the random forest analysis (Figure 160 1a), but the subdiffusion mechanisms discussed above indicated that it should be relevant when 161 approaching the typical mesh size of mucus. The analysis of the effective diffusion as a func-162 tion of particle diameters indicated a clear threshold around $d^{\star} \sim 100$ nm (Figure 3a). Larger 163 particles, d > 100 nm, displayed lower effective diffusion values but with no apparent statistical 164 correlation with size ($\rho = -0.24$, p= 0.19). The associated empirical and predicted anomalous 165 exponents ranged from 0.15 to 1, indicating that factors other than particle size are influencing 166 the subdiffusion. Smaller particles, d < 100 nm, displayed an effective diffusion with a significant 167 statistical correlation (Figure 3a). In particular, those particles that had been predicted to display 168 regular diffusion were inversely dependent with particle size, that is, slope $m \sim -1$ (Figure 3). As 169 predicted by the microenvironment trapping mechanism, small particles in mucus displayed regu-170 lar diffusion, midsize particles were subject to subdiffusion (although the attachment-mechanism 171 cannot be discarded), and large particles display a variety of outputs probably dependent on the 172 mesh size (and potential interaction with mucus). The average mucus in humans has a typical 173 mesh size between 100 to 1000 nm (Cone 2009), which explain the diffusion behavior for particles 174 around 100 nm, or greater, in Figure 3b. 175

Particle type was selected as the second most relevant variable to predict the effective diffusion 176 based on the random Forest analysis (Figure 1). Comparing the effective diffusion for the different 177 particles confirms this prediction (Figure S.1a). Antibodies and proteins displayed the fastest 178 effective diffusion with a mean of $48.9 \ \mu m^2/s$ (Figure S.1). Viruses were the second fastest group 179 with a mean effective diffusion an order of magnitude smaller, $3.5 \ \mu m^2/s$. Pegylated and amine 180 particles formed the third group. They displayed statistically similar effective diffusions with means 181 (medians) 0.99 μ m²/s and 2· 10⁻² μ m²/s. This was followed by COOH particles, mean (median) 182 $3 \cdot 10^{-2} \ \mu m^2/s$, and finally chitosan $4 \cdot 10^{-3} \ \mu m^2/s$. Difference in particle size could explain the 183 reduction in effective diffusion for antibodies/proteins, viruses, and PEG particles (Figure S.1b). 184 They had, respectively, median sizes of ~ 10 nm, ~ 100 nm, and ~ 1000 nm. It is unclear what 185

were the physico-chemical factors behind the slower diffusion of Amine, COOH, and Chitosan particles (Figure S.1).

The third predictor for effective diffusion was particle charge, express as the zeta-potential 188 (Figure 1). Particles with negative zeta potential displayed a positive correlation with the effective 189 diffusion constant with a Spearman correlation of $\rho = 0.6$ (p = 0.002***, n = 36) (Figure 4a). The 190 relationship was approximated by an exponential function, $D_{eff} \sim exp(m\xi)$. The potential rate, 191 m, was $m = 0.024 \pm 0.006$ (p ~ 0.0002***) obtained from a least-square linear regression using 192 the log-linear data. This exponential model accounted for 30 % of the variance $(R^2 = 0.30)$. The 193 largest effective diffusions were achieved at neutral zeta potentials. Positive zeta potentials (n=21)194 had lower values but did not display a statistically significant correlation the effective diffusion. 195 Particle size or other properties did not seem to explain the trend observed for negatively charged 196 zeta potentials. These particles, however, displayed a linear positive correlation with the anomalous 197 diffusion (Figure 4b). Based on the two mechanisms discussed above, one interpretation could be 198 that given the negative charge of the mucin fibers, an increasing negative charge of a partice will 199 increase its effective radius, increasing the particle size to network mesh ratio and thus reducing 200 the diffusivity. Alternatively, the presence of negative charges could compete for ions with respect 201 the mucin fibers exposing hydrophobic regions that could interact with the particles as has been 202 observed in carboxilated particles forming bundles with mucus (Lai et al. 2007, 2009). Based on 203 the attachment-mechanism, the increase in negative zeta potential would be proportional to the 204 attachment time induced. The positive zeta potential would be expected to interact with mucin 205 fibers. That could explain the reduction inf effective diffusion with respect neutral structures, but 206 it did not display any apparent correlation with the anomalous exponent. 207

The mucus source and dominant mucin were the last two significant predictor of effective 208 diffusion. The effective diffusion was faster in human cervix samples with a median $\sim 10 \ \mu m^2/s$, 209 although the values spanned six orders of magnitude, from $\sim 10^{-4}$ to $\sim 10^2 \ \mu m^2/s$ (Figure S.1a). 210 The effective diffusion was the slowest in mucus from human lung (median $\sim 10^{-2} \mu^2/s$) and 211 pig intestine (median $\sim 10^{-2} \ \mu m^2/s$). The median particle size in empirical data from human 212 cervix mucus was more than an order of magnitude smaller, ~ 10 nm, than for the empirical data 213 from the other sources. The median pH for the empirical data from human cervix mucus was 214 significantly lower pHs (median 5.5) compared to the other sources (median 7). Lower pH tends 215 to thicken mucus (Hwang et al Rheological Properties of Mucus 1969), thus expecting a slower 216 effective diffusion. But the particle size may have offset this trend. The transcription analysis 217 identified MUC5B, which is dominant in human cervix, displaying the largest effective diffusion 218 (median $\sim 10 \ \mu m^2/s$) compared to the other dominant mucins, MUC2 common in respiratory 219 mucus (median diffusion $\sim 10^{-1} \ \mu m^2/s$), and MUC5AC common in intestinal mucus (median 220

diffusion $\sim 10^{-2} \ \mu m^2/s$) (Figure S.3).

Thus, our meta-analysis discovered that the anomalous exponent is the dominant factor regu-222 lating the effective diffusion microscopic particles in mucus, explaining 90% of the variance across 223 6 orders of magnitude (Figure 2a). However, less than 40% of the empirical data had measured 224 the anomalous exponent, indicating an important gap in the field about the dominance of this 225 factor in the diffusion of particles in mucus. To empirically validate our finding, we provide pre-226 dictions of the anomalous exponent in Figure 2b. Remarkably, those particles predicted to display 227 regular diffusion showed an inverse dependence between the effective diffusion and particle size, as 228 expected from standard Brownian motion (Figure 3). 229

Our first-principles analysis of subdiffusion provided a general equation that explained the 230 exponential relationship between anomalous diffusion and the effective diffusion as well as the 231 dominance of the anomalous exponent, independently of the underlying subdiffusion mechanism, 232 Eq. (6). Our analysis indicated that the generalized diffusion constant is not a well-defined phys-233 ical quantity, and it must be replaced by characteristic length and time scales associated to the 234 underlying mobility mechanisms, Eq. (4). This led to a reformulation of the subdiffusion equation, 235 Eq. (5), which applies to any physical system. We conclude that the physical factors regulating 236 the anomalous exponent is key to characterize and control the diffusivity of particles. In mucus 237 and hydrogels, in particular, we propose that the attachment and microenvironment mechanisms 238 should be used as a guide. Measuring the attachment time distributions and mucus mesh size are key factors regulating the anomalous exponent in these mechanisms. But they had not been 240 measured even among the mucus experiments that measured anomalous exponents. Our work, 241 thus, fills a gap that would guide a more effective and insightful study of effective diffusions in 242 mucus and other complex fluids. Our study also mandates a reinterpretation of the generalized 243 diffusion constant in any physical system. 244

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³⁵⁹ Figures and Tables

Property	Symbol	Range	Data points
Effective diffusion	$D_{\rm eff}$	$3.1 \cdot 10^{-5}$ to $1.3 \cdot 10^2 \ \mu m^2/s$	106
Anomalous exponent	α	0.16 to 1.02	39
Diameter	d	3.5 to $1280.0~\mathrm{nm}$	106
Zeta potential	ζ	-73.0 to $+33.3$ mV	57
Temperature	Т	295 to $310~{\rm K}$	106
pН	$_{\rm pH}$	3.0 to 7.4	63
Mucus source		Hydrogel, human lung, human cervix,	106
		pig stomach, pig intestines	
Mucin type		MUC2, MUC5AC, MUC5B	103

Table 1: Summary of empirical data. D_{eff} : The effective diffusion was obtained for a common time window of 1 second. For references that shared relative diffusion with respect diffusion in water, the effective diffusion constant was scaled using the Stokes-Einstein equation using the temperature and hydrodynamic particle diameter reported (Miller 1924). Room temperature (298 K) was assumed if temperature was not reported in the study. Particle type data was obtained by classifying particles as COOH, PEG, virus, amine, antibody/protein, or chitosan. This was chose as a qualitative measure of particle-mucin bonds. The dominant mucin composition from each mucus source was obtained by evaluating the expression levels of mucin genes from the genome bioinformatics portal Ensembl (Zerbino et al. 2018). Mucins were identified assuming the tissue/organ associated with each mucus, or closely associated tissues. Expression levels were collected by taking the average of reported median of transcript per million (TPM) RNA-sequence and the most explicitly stated expression levels of low, medium, and high. Based on potential gene expression of mucins with reported levels of below cutoff, TPM measured below the minimum (0.05 TPM) is distinguished from experiments with no data due to possible gene expression. An expression level of low, medium or high was obtained over reports of below cutoff in the same tissue. The dominant mucin was determined by the highest expression level then, if necessary, by the highest average of median TPM. Identification of mucin expression based on tissues was associated with each mucus: human respiratory mucus and human cystic fibrosis mucus were associated with the human lung mucin genes; human cervical mucus and cervicovaginal mucus were associated with human cervix or uterus mucin genes; pig intestinal mucus was originally from jejunum part of the small intestine, however, due to a lack of reports for jejunum tissue, the associated mucin genes were taken as the average of the median of TPM of pig duodenum and pig ileum parts of the small intestine based on the close proximity to the jejunum; pig ileum intestinal mucus were associated with ileum tissue mucin genes; pig gastric mucus were collected from pig stomach mucin genes.

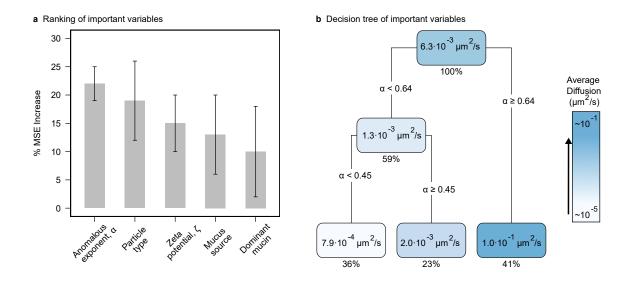


Figure 1: Selected variables impacting effective diffusion. **a**, Average percentage increase of meansquare error (% MSE) for the selected variables. These variables were investigated in permutations of three using random forest (R package rfPermute by Archer 2019). The error bars correspond to the standard deviation. **b**, Decision tree for the most important variables. Each node contains the predicted average D_{eff} and percentage of data predicted. The gradient display diffusion values from ~ $10^{-5} \text{ }\mu\text{m}^2/\text{s}$ (white) to ~ $10^{-1} \text{ }\mu\text{m}^2/\text{s}$ (blue).

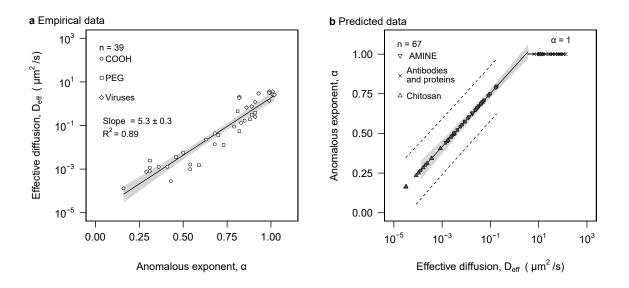


Figure 2: Effective diffusion and anomalous exponent analysis. **a**, effective diffusion was plotted as a function of anomalous exponent. The solid line represents the regression model. The grey area represents the 95% confidence interval. Statistically significant slope and R^2 of linear regression is displayed. **b**, anomalous exponent was predicted based on the model found empirically in **a**. The solid line designates the predicted linear model. The grey area represents the 95% confidence interval of the predicted linear model. The dashed line represents a 95% prediction interval **a-b**, distinguished particle types are represented in the legend of both panels.

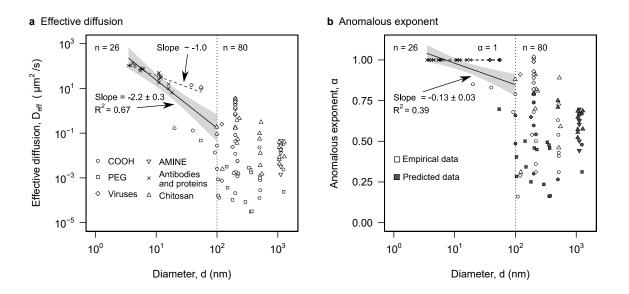


Figure 3: Particle size analysis. **a**, effective diffusion was plotted against particle size, d. **b**, anomalous exponent was plotted as a function of particle size. **a-b**, different particle types with different symbols are indicated in **a**'s legend. Separate analysis was conducted for particles smaller than 100 nm represented by the dotted line at d = 100 nm. Solid line corresponds to significant linear regression and the grey area represents the 95% confidence interval. Significant slope and R^2 of each linear regression are displayed.

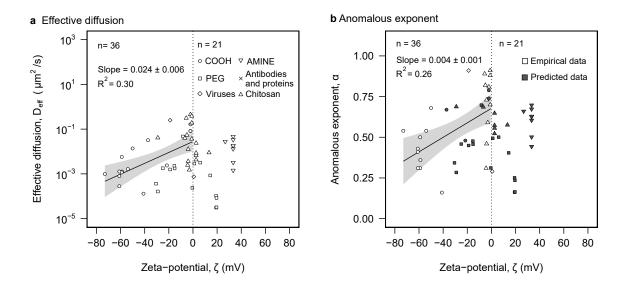


Figure 4: Electrostatic analysis. **a**, effective diffusion was plotted against zeta potential. **b**, anomalous exponent as a function of zeta potential. Classification of empirical and predicted data is represented in the legend. **a-b**, Different particle types are distinguished in **a** legend. Separate tests based on charge sign was conducted and designated by the dotted line at $\zeta = 0$. Solid line represents significant linear regression. Grey area represents the 95% confidence interval of the linear regression. Significant slope and R^2 of each linear regression are displayed.

360 Supplementary material

Common core.csv

Table S.1: Supplementary material of common core file of all reviewed data. Common_core. The first row is a header with designated column names fitted to physical properties collected. Particle name, particle type, zeta potential, particle size, effective diffusion constant at 1s, anomalous exponent, diffusion in water, ratio between effective diffusion constant at 1s and diffusion in water, temperature, pH, dosing medium, salt type used, salt concentration, mucus used in the experiment, mucus concentration, mucus purification, mucin gene expression level, and dominant mucin gene is denoted as Particle, Surface_Chemistry, Zeta, Diameter, D_w, Diffusion_constant, alpha, Ratio_Diffusion, Temperature, pH, Dosing Medium, Salt_type, Salt_Concentration, Mucus_Type, Muc_Con, Purification, 'mucin gene name'_EL, and Dominate_Mucin, respectively.

Mucin_expression_level.csv

Table S.2: Supplementary material on dominant mucin classification based on mucus source. Mucin_expression_level. The first row is a header and first column designates mucus type. Represented expression levels are denoted as 'mucin gene name'_EL. Expression levels are classified as below cutoff, low, medium, high, or unavailable. The average median Transcripts per million (TPM) for each mucin gene is designated as 'mucin gene name'_TPM_median_avg. See methods for more information.

Dependent	Independent						
variable	variable	Slope	Intercept	R^2	p -value \parallel	$rho\ddagger$	p-value‡
Effective Diffusion $\P \land$	Diameter∧∘	-2.1 ± 0.3	3.5 ± 0.4	0.67	3.5E-7***	-0.9	1.7E-12***
Effective Diffusion $\P \land$	Negative [†]	0.024 ± 0.006	-1.6 ± 0.2	0.30	0.0006^{***}	0.6	0.0002^{***}
Effective Diffusion $\P \land$	Positive [†]	0.01 ± 0.02	-2.8 \pm 0.4	0.03	0.5	0.3	0.2
Effective Diffusion $\P \land$	Alpha	5.3 ± 0.3	-5.0 ± 0.2	0.89	$<\!\!2.0\text{E-}16^{***}$	0.9	$<\!\!2.2\text{E-}16^{***}$
Anomalous Exponent	$\mathrm{Diameter} \wedge \circ$	-0.13 ± 0.03	1.11 ± 0.04	0.39	0.0007^{***}	-0.6	0.001
Anomalous Exponent	Negative [†]	0.004 ± 0.001	0.68 ± 0.04	0.26	0.002***	0.5	0.0007***
Anomalous Exponent	Positive [†]	0.003 ± 0.003	0.44 ± 0.07	0.04	0.4	0.3	0.2

Table S.3: Supplementary material of linear Analysis of Nanoparticles' Mobility Through Mucus and Biohydrogels. || Simple Linear Regression and Pearson's p-value for the slope. \ddagger Spearman analysis. ¶ Effective Diffusion coefficient ($\mu m^2/s$). \land Logarithmic of base 10 (log_{10}). \circ Diameter less than 100 nm. *** Data has strong significance. \ddagger Zeta potential (mV). Overview table of values associating with statistically significant simple linear regression along with spearson and pearson analysis.

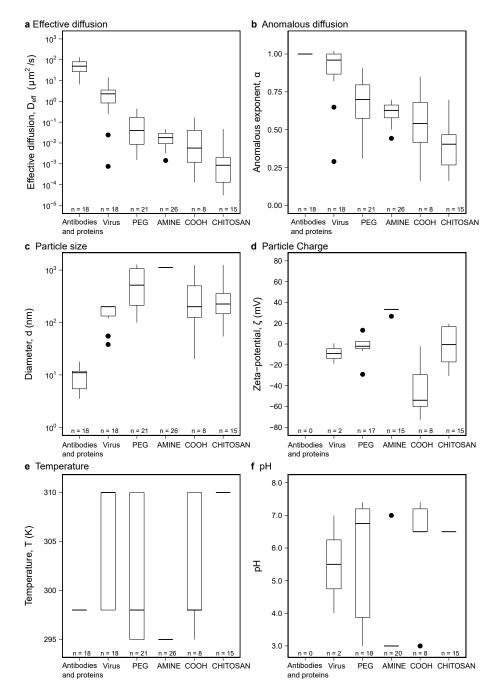


Figure S.1: Transport capabilities based on particle type. **a**, effective diffusion constant at one second based on particle type **b**, anomalous exponent based on particle type. **c**, .particle size based on particle type. **d**, particle net charge based on particle type. **e**, mucus temperature based on particle type. **f**, mucus pH based on particle type. **a**-**f**, box plots are ranked by effective diffusion from high to low. The total amount of data points for each particle type is designated as n aligned with their respected particle type for each panel.

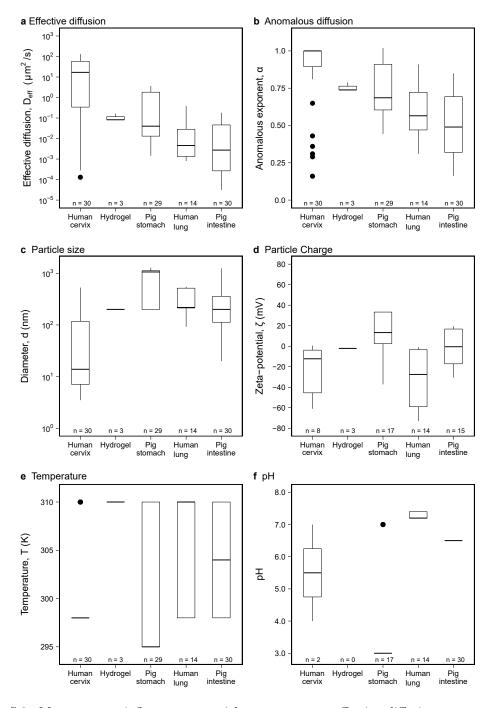


Figure S.2: Mucus source influence on particle transport. **a**, effective diffusion constant at one second based on mucus source. **b**, anomalous exponent based on mucus source. **c**, particle size based on mucus source. **d**, particle net charge based on mucus source. **e**, mucus temperature based on mucus source. **f**, mucus pH based on mucus source. **a**-**f**, box plots are ranked by effective diffusion from high to low. The total amount of data points for each mucus source is designated as n aligned with their respected mucus source for each panel.

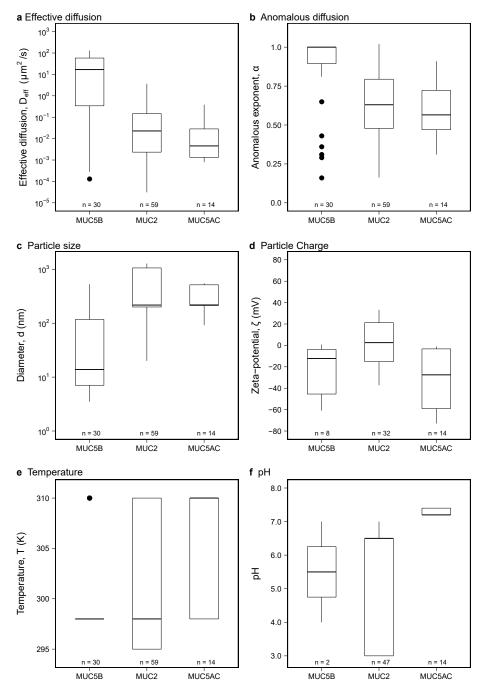
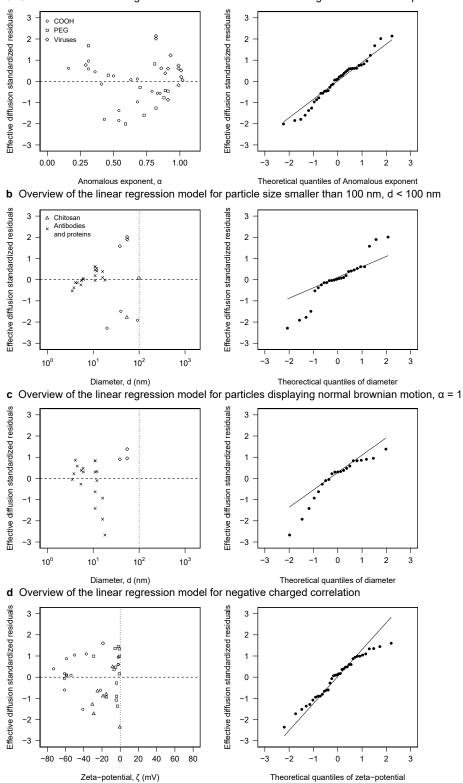


Figure S.3: Impact of dominant mucin genes on particle transport. **a**, effective diffusion constant at one second based on dominate mucin. **b**, anomalous exponent based on dominate mucin. **c**, particle size based on dominate mucin. **d**, particle net charge based on dominate mucin. **e**, mucus temperature based on dominate mucin. **f**, mucus pH based on dominate mucin. **a**-**f**, box plots are ranked by effective diffusion from high to low. The total amount of data points for each dominate mucin is designated as n aligned with their respected dominate mucin for each panel.



a Overview of the linear regression model for effective diffusion against anomaous exponent

Figure S.4: In-depth overview of significant linear regression models. **a**, standardized residual and normal probability of standardized residuals for effective diffusion as a function of anomalous exponent. **b**, standardized residual and normal probability of standardized residuals for effective diffusion as a function of particle size for sizes less than 100 nm. Dotted line is a visualization marker for particles smaller than 100 nm. **c**, standardized residual and normal probability of standardized residuals for effective diffusion as a function. Dotted line is a visualization marker for particles displaying normal brownian motion. Dotted line is a visualization marker for particles smaller than 100 nm. **d**, standardized residual and normal probability of standardized residuals for effective diffusion as a function of zet potential for negatively charged particles. Dotted line is a visualization marker for negatively charged particles. **a-d**, different particle types with corresponding symbols are designated in **a** and **b**'s legend.