Development of Remdesivir as a Dry Powder for Inhalation by Thin Film Freezing

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Abstract: Remdesivir exhibits in vitro activity against SARS-CoV-2 and was granted approval for Emergency Use. To maximize delivery to the lungs, we formulated remdesivir as a dry powder for inhalation using thin film freezing (TFF). TFF produces brittle matrix nanostructured aggregates that are sheared into respirable low-density microparticles upon aerosolization from a passive dry powder inhaler. In vitro aerodynamic testing demonstrated that drug loading and excipient type affected the aerosol performance of remdesivir. Remdesivir combined with optimal excipients exhibited desirable aerosol performance (up to 93.0% FPF; 0.82μm MMAD). Remdesivir was amorphous after the TFF process, which benefitted drug dissolution in simulated lung fluid. TFF remdesivir formulations are stable after one-month storage at 25 °C/60%RH. In vivo pharmacokinetic evaluation showed that TFF-remdesivir-leucine was poorly absorbed into systemic circulation while TFF-remdesivir-Captisol® demonstrated increased systemic uptake compared to leucine. Remdesivir was hydrolyzed to the nucleoside analog GS-441524 in lung, and levels of GS-441524 were greater in lung with the leucine formulation compared to Captisol®. In conclusion, TFF technology produces high potency remdesivir dry powder formulations for inhalation suitable to treat patients with COVID-19 on an outpatient basis and earlier in the disease course where effective antiviral therapy can reduce related morbidity and mortality.

Keywords: Remdesivir, SARS-CoV-2, COVID-19, dry powder for inhalation, thin film freezing, brittle matrix powder.

1. Introduction

The coronavirus disease 2019 (COVID-19) is an ongoing worldwide pandemic. As of September 2020, laboratory-confirmed cases have been reported in 213 countries and territories with more than 30 million reported cases and close to 1 million reported deaths [1]. Although this disease is asymptomatic to mild in most people, in some cases, it can develop into pneumonia, acute

respiratory distress syndrome (ARDS) and multi-organ dysfunction [2]. While effective treatments for

COVID-19 are urgently needed, including therapeutics and vaccines, currently no drugs are approved

to treat COVID-19. Only one drug, remdesivir that is administered by intravenous (IV) injection, has

been granted emergency use authorization in adult and pediatric patients hospitalized with severe

disease. Therefore, additional therapeutics and routes of administration are needed to treat this

respiratory virus.

Remdesivir (GS-5734), an investigational broad-spectrum antiviral agent, was developed by

Gilead Sciences Inc. [3]. Remdesivir is a monophosphoramide prodrug that is intracellularly

metabolized into an adenosine analog (GS-441524) [4]. Both GS-441524 and remdesivir (GS-5734)

are metabolized into the active nucleoside triphosphate (GS-443902) by the host [5]. The adenosine

nucleoside analog can inhibit viral RNA polymerases, evade proofreading by viral exonucleases, and

subsequently decreases viral RNA production [4]. Previous studies demonstrated that remdesivir has

broad-spectrum activity against members of the filovirus (e.g., EBOV, MARV) [6], CoVs (e.g., SARS-

CoV, MERS-CoV) [7, 8], and paramyxovirus (e.g., respiratory syncytial virus, Nipah virus, and Hendra

virus) [9].

Remdesivir was initially developed for the treatment of EBOLA virus disease [3]. Recently, in

vitro studies showed that remdesivir was effective against COVID-19 in the human airway epithelial

cell [10]. Short term outcomes of remdesivir were reported for 53 patients infected with COVID-19

treated in over 20 hospitals [11]. It was reported that overall clinical improvement was observed in 36

of 53 patients (68%) receiving a 10-day regimen of IV remdesivir [11]. However, 32 patients (60%) in

the study experienced at least one adverse event, and 12 patients (23%) reported serious adverse

events [11]. The most common adverse events included diarrhea, rash, septic shock, acute kidney

injury, renal impairment, and hypotension [11, 12]. Several studies are ongoing, including two clinical

studies evaluating the efficacy and safety of remdesivir for the treatment of COVID-19 respiratory

disease in humans [13, 14]. In one study, 308 patients with mild to moderate COVID-19 were given

remdesivir as a 200 mg loading dose on day 1, then 100 mg IV once-daily for 9 days [13]. In another

study, the same regimen was given to 452 patients with severe clinical manifestations of COVID-19

[14].

The feasibility to deliver remdesivir via different administration routes (e.g., intravenous (IV)

injection, intramuscular (IM) injection and oral administration) has been investigated [3]. Unfortunately,

remdesivir is not suitable for oral delivery since the drug is mostly metabolized and cleared by first-

pass metabolism, resulting in poor hepatic stability and poor bioavailability [3]. The delivery of

remdesivir by the IM route also faces the challenge of variable release from muscle and slow acting

of the active compound in peripheral blood mononuclear cells [3]. Therefore, remdesivir is

administrated by IV injection since this route allows wide distribution to most tissues including kidney,

kidney medulla, and liver [3].

For IV administration, remdesivir was developed in two dosage forms, a concentrate solution

intended for dilution and administration by infusion and a lyophilized powder for reconstitution and

dilution intended for administration by infusion [3]. Since remdesivir is practically insoluble in water,

both formulations contain sulfobutylether-beta-cyclodextrin (SBECD) as a solubility enhancer by

complexation [3, 15-17]. Since the excipient, SBECD, is cleared by the kidneys, remdesivir is

contradicted in patients with several renal impairment [3]. A pharmacokinetic study showed that both

formulations are interchangeable based on similar plasma concentrations [3]. Lyophilized remdesivir

powder is stable when stored below 30 °C and has a shelf-life of 51 months, while remdesivir

concentrate solution for dilution requires the product to be stored at freezing storage conditions

(−25 °C to −10 °C) for a shelf-life of 48 months [3].

Currently, remdesivir is administered by injection into a vein as an infusion to treat patients

with COVID-19 only in hospitals. However, due to the pandemic, many patients are not able to be

hospitalized, and they do not have access to injectable administration of remdesivir. To provide

remdesivir for other patients besides those most severely ill, more convenient and accessible dosage

forms for different routes of administration must be quickly developed and tested so that patients have

more options to get treated.

Intriguing strategies for future improvement of remdesivir include the development of inhaled

dry powder of remdesivir for direct administration to the primary site of infection, the lungs. The

mechanism of coronavirus infection is mostly in the respiratory tract, especially in the deep lungs.

Hence, the inhalation route must be immediately pursued as the most promising route of

administration to maximize direct delivery to the target site without first-pass metabolism, boost local

antiviral activity in the lungs, and limit the potential for systemic side effects [18]. In addition, the cost

of the drug can be reduced, and the supplies of the drug can be maximized, thus treating more

patients due to less dose required by inhalation as compared to injectable forms. The treatment cost

can also be decreased when administered by inhalation, since patients may not need to visit hospitals

as is required to administer the IV injectable dose. Therefore, more affordable and early stage

treatment can be provided to patients with inhaled remdesivir.

Nebulization of the current IV formulation in a diluted form is a potential method of pulmonary

administration; however, the drug is prone to degrade by hydrolysis in aqueous solution to form the

nucleoside monophosphate, which has difficulty penetrating cell membranes, thereby minimizing the

antiviral activity in the lung cells [19]. Another concern is the use of SBECD as an excipient in

inhalation dosage forms. Although cyclodextrin is not an excipient in an approved FDA inhaled

product [20], several reports demonstrated the use cyclodextrin in inhaled formulations. Tolman et al.

[21, 22] previously demonstrated that aerosolized voriconazole nebulized solution that contained a

diluted form of the commercially available Vfend®, an IV voriconazole dosage form containing

SBECD, is capable of producing clinically relevant lung tissue and plasma concentrations of the drug

in animals. However, in this form, voriconazole in lung tissue was not able to be detected 6-8 h after

administration of a single inhaled dose in the animal study by Tolman et al. [21], while voriconazole

dry powder lasted longer in the lungs (up to 24 h) in an animal study by Beinborn et al [23]. Therefore,

a dry powder formulation for inhalation can be more favorable with a smaller number of daily doses.

The advantages of dry powders for inhalation over liquids administered by nebulizers are not

limited only to improve the stability of the drug during inhalation, but also dry powder inhalers provide

the ease of maintenance and having a short administration time [24]. The inconvenience of nebulized

inhalation solutions includes the use of a large nebulizer device requiring electric power and water,

user manipulation for cleaning and operating, and long nebulization times. Also, the cost per dose

and initial cost of the nebulizer are higher than a DPI device [24]. Hence, dry powder inhalation is an

ideal dosage form for the treatment of COVID-19 in an outpatient setting, which would minimize the

risk of spreading the virus to healthcare professionals.

Although several techniques have been used to prepare inhalable powders, including

mechanical milling and spray drying, the advantages of thin film freezing (TFF) over other techniques

rely on the ability to produce aerosolizable particles composed of brittle matrix, nanostructured

aggregates. These are high surface area powders that are ideally suited for dry powder inhalation.

TFF employs ultra-rapid freezing (on the order of 100-1,000 K/sec) such that precipitation (either as

a crystalline nanoaggregate or amorphous solid dispersion) and particle growth of the dissolved

solute can be prevented [25]. Subsequently, nanostructured aggregates are formed as a low-density

brittle matrix powder [26], which is efficiently sheared into respirable low-density microparticles by a

dry powder inhaler upon inhalation by the patient [27]. Despite a large geometric diameter (>10

microns), low-density microparticles composed of these nanostructured aggregates can be delivered

to the deep lung with optimal aerodynamic diameters of <3 to 4 microns [27]. Additionally, it has been

reported that the engineered particles with a geometric diameter greater than 10 microns can extend

drug retention time in the lungs due to the ability of the deposited particles to escape from

macrophage phagocytosis [28]. According to a recent study by Longest et al., computational models

demonstrated that nanoaggregates are favorable for higher drug absorption efficiency and dose

uniformity in the lungs, compared to microparticles [29]. Furthermore, TFF can produce amorphous

high surface area powders with submicron primary particles, which can enhance the dissolution rate

and subsequently improve the bioavailability of poorly water-soluble drugs like remdesivir in the lungs

[30].

This work aims to apply the TFF technology to develop remdesivir dry powder formulations

for inhalation administered by a commercially available dry powder inhaler. We hypothesize that the

ultra-rapid freezing rate of the TFF technology will produce low-density, high porosity brittle matrix

powders of remdesivir, which are aerosolizable by the shear forces generated from the passive dry

powder inhaler, and thus allow high doses of remdesivir to be administered to the lung.

2.   Materials and Methods

2.1   Materials

Remdesivir for formulation was purchased from Medkoo Biosciences (Research Triangle

Park, NC, USA). Remdesivir, GS-441524, and its heavy isotope internal standards were purchased

from Alsachim (Illkirch-Graffenstaden, France). Lactose monohydrate, leucine, polysorbate 20,

acetonitrile (HPLC grade), and trifluoracetic acid were purchased from Fisher Scientific (Pittsburgh,

PA, USA). D-Mannitol was bought from Acros Organic (Fair lawn, NJ, USA).

Dipalmitoylphosphotidylcholine (DPPC) was purchased from Avanti Polar Lipid, Inc. (Alabaster,

Alabama, USA). Cholesterol, albumin, transferrin, ascorbic acid and Hanks' Balanced Salt solution

(HBBS) were purchased from Sigma-Aldrich (St.Louis, MO, USA). Sulfobutylether-beta-cyclodextrin

(SBECD, Captisol®) was kindly provided by CyDex Pharmaceuticals, Inc. High-resistance Monodose

RS00 dry powder inhalers were kindly provided by Plastiape S.p.A. (Osnago, Italy).

2.2 Preparation of dry powder for inhalation using thin film freezing

Remdesivir and excipients (i.e., Captisol®, mannitol, lactose and leucine) were dissolved in

a mixture of either acetonitrile/water (50/50 v/v) or 1,4-dioxane/water (50/50 v/v) at the solid contents

shown in Table 1. The solutions were stored in the refrigerator at 2-8°C before the thin film freezing

process. The solution was passed through an 18-gauge syringe needle and dropped from a height of

approximately 10 cm onto a rotating cryogenically cooled stainless-steel drum. The solutions were

frozen at a drum surface temperature of −100 °C. The frozen samples were collected in a stainless-

steel container filled with liquid nitrogen and then transferred into a −80 °C freezer before drying in a

lyophilizer. The frozen samples were primary dried at −40 °C for 20 hours, and ramped to 25 °C over

20 hours, and then secondary dried at 25 °C for 20 hours. The vacuum pressure was set at 100 mTorr

for the whole drying cycle.

Table 1. List of TFF remdesivir compositions and parameters used for their preparation.

Formulation	Drug loading (% w/w)	Excipient	Solid content (% w/v)	Solvent (v/v)	Processing temperature (°C)
F1	5	Captisol®	1.00	Acetonitrile/water (50/50)	−100
F2	20	Captisol®	1.00	Acetonitrile/water (50/50)	−100
F3	20	Mannitol	1.00	Acetonitrile/water (50/50)	−100
F4	20	Lactose	1.00	Acetonitrile/water (50/50)	−100
F5	20	Leucine	0.75	Acetonitrile/water (50/50)	−100
F6	50	Captisol®	0.30	Acetonitrile/water (50/50)	−100
F7	50	Mannitol	0.30	Acetonitrile/water (50/50)	−100
F8	50	Lactose	0.30	Acetonitrile/water (50/50)	−100
F9	50	Leucine	0.30	Acetonitrile/water (50/50)	−100

F10	80	Captisol®	0.30	Acetonitrile/water (50/50)	−100
F11	80	Mannitol	0.30	Acetonitrile/water (50/50)	−100
F12	80	Lactose	0.30	Acetonitrile/water (50/50)	−100
F13	80	Leucine	0.30	Acetonitrile/water (50/50)	−100
F14	100	-	0.25	Acetonitrile/water (50/50)	−100
F15	100	-	0.50	1,4-dioxane/water (50/50)	−100
F16	100	-	1.00	1,4-dioxane/water (50/50)	−100
F17	80	Captisol®	1.00	1,4-dioxane/water (50/50)	−100
F18	80	Mannitol	1.00	1,4-dioxane/water (50/50)	−100
F19	80	Lactose	1.00	1,4-dioxane/water (50/50)	−100
F20	80	Leucine	1.00	1,4-dioxane/water (50/50)	−100

2.3   Drug quantification (HPLC)

The content of remdesivir was determined through analysis with a Thermo Scientific™

Dionex™ UltiMate™ 3000 HPLC system (Thermo Scientific, Sunnyvale, CA, USA) at a wavelength

of 246 nm. A Waters Xbridge C18 column (4.6 × 150 mm, 3.5 μm) (Milford, MA) was used at 30 °C

and a flow rate of 0.8 mL/min. The isocratic method was performed for 4.5 minutes using a mobile

phase of 50:50 (%v/v) water –acetonitrile containing 0.05% (v/v) TFA. The retention time of remdesivir

was approximately ~3.5 minutes. Dimethyl sulfloxide: acetonitrile: water (10:60:30, v/v) was used as

diluent. All analyses exhibited linearity in the range tested of 0.2–250 μg/mL.   All chromatography

data were processed by Chromeleon Version 7.2.10 software (Thermo Scientific, Sunnyvale, CA,

USA).

For quantity and quality analysis of stability samples of TFF-remdesivir powders, Agilent 1220

Infinity II LC system equipped with Agilent 1290 Infinity II evaporative light scattering detector (ELSD)

was utilized. Waters XBridge BEH C18 column (4.6 x 150 mm, 3.5 μm) (Milford, MA) was used at

30 °C and a flow rate of 0.8 mL/min. The gradient method with a mobile phase ratio of 5 to 95 %

acetonitrile in water with 0.1% trifluoroacetic acid was used for total running time of 20 minutes. 10

μL of each sample was injected. The stability samples were monitored with 246 nm UV and ELSD to

detect possible degradants. For ELSD, evaporator and nebulizer temperatures were set at 60 °C, and

nitrogen gas flow was 1.6 L/min. The chromatography data were processed by Chemstation version

C.01.10.

2.4   In vitro aerosol performance

The aerodynamic properties of the TFF powder samples were determined using a Next

Generation Pharmaceutical Impactor (NGI) (MSP Corp, Shoreview, MN) connected to a High

Capacity Pump (model HCP5, Copley Scientific, Nottingham, UK) and a Critical Flow Controller

(model TPK 2000, Copley Scientific, Nottingham, UK). The pre-separator was not used in this study.

To avoid particle bounce, the NGI collection plates were coated with 1.5% w/v polysorbate 20 in 100%

methanol and then let the plates dry before analysis. A high resistance Plastiape® RS00 inhaler

(Plastiape S.p.A, Osnago, Italy) containing size #3 HPMC capsules (V-Caps® Plus, Lonza,

Morristown, NJ) and 1-3 mg of powder was attached to a USP induction port by a molded silicon

adapter, and the powder was dispersed to the NGI at the flow rate of 60 L/min for 4 s per each

actuation. The deposited powders from the capsule, inhaler, adapter, induction port, stages 1-7, and

the micro-orifice collector (MOC) were collected by diluting with a mixture of DMSO/water/acetonitrile

(10:30:60 v/v). The remdesivir content in the deposited powders was determined using the HPLC

method as described in Section 2.3.

Copley Inhaler Testing Data Analysis Software (CITDAS) Version 3.10 (Copley Scientific,

Nottingham, UK) was used to calculate the mass median aerodynamic diameter (MMAD), the

geometric standard deviation (GSD), the fine particle fraction (FPF), and the emitted fraction (EF).

The EF was calculated as the total amount of remdesivir emitted from the device as a percentage of

total amount of remdesivir collected. The FPF of recovered dose was calculated as the total amount

of remdesivir collected with an aerodynamic diameter below 5 µm as a percentage of the total amount

of remdesivir collected. The FPF of delivered dose was calculated as the total amount of remdesivir

collected with an aerodynamic diameter below 5 µm as a percentage of the total amount of remdesivir

deposited on the adapter, the induction port, stages 1- 7 and MOC.

2.5   X-ray powder diffraction (XRPD)

The crystallinity of the powders was evaluated using a benchtop X-ray diffraction instrument

(Rigaku Miniflex 600 II, Woodlands, TX, USA) equipped with primary monochromated radiation (Cu

K radiation source, λ = 1.54056 Å). The instrument was operated at an accelerating voltage of 40 kV

at 15 mA. Samples were loaded in the sample holder and scanned in continuous mode with a step

size of 0.02º over a 2θ range of 5–40 ° at a scan speed of 2 °/min, and a dwell time of 2 s

2.6 Modulated differential scanning calorimetry (mDSC)

Thermal analysis of powder samples was conducted using a differential scanning calorimeter

Model Q20 (TA Instruments, New Castle, DE) equipped with a refrigerated cooling system (RCS40,

TA Instruments, New Castle, DE). Samples of 2-3 mg were weighed and loaded into a T-zero pan.

The pan with a T-zero hermetic lid were crimped, and a hole was drilled in the lid before placing the

pan in the sample holder. To evaluate the glass transition and glass forming ability type of remdesivir

unprocessed powder, samples were heat at a heating ramp rate of 10 °C/min from 25 °C to 150 °C,

then cooled down to −40 °C, and then heated again to 250 °C. To investigate crystallinity of the TFF

formulations, samples were heated from 25 °C to 350 °C with a heating ramp rate of 5 °C/min. The

scans were performed with a modulation period of 60 s and a modulated amplitude of 1 °C. Dry

nitrogen gas at a flow rate of 50 mL/min was used to purge the DSC cell throughout the analysis.

Data were processed by TA Instruments Trios V.5.1.1.46572 software.

2.7 Scanning electron microscopy (SEM)

225 Scanning electron microscopy (Zeiss Supra 40 C SEM, Carl Zeiss, Heidenheim an der Brenz,

226 Germany) was used to identify surface particle morphology of the TFF remdesivir powder

227 formulations. A small amount from each of the powder formulations was placed on a carbon tape. A

228 sputter coater was used to coat all samples with 15 nm of 60/40 Pd/Pt before capturing the images.

2.8 Nuclear Magnetic Resonance (NMR)

^{1}H-NMR spectra were obtained using a Varian® NMR 400 MHz Spectrometer (Agilent Inc.,

Palo Alto, CA, USA) at 25 °C and were used to confirm the purity of remdesivir in the TFF powders,

and to study interactions between excipients and remdesivir of TFF powders. Formulations F7, F8,

F9, F10, F14 and unprocessed L-leucine and remdesivir drug substance were dissolved in dimethyl

sulfoxide-d6 (DMSO-d6). Solutions were then transferred to 5 mm NMR tubes for NMR data

acquisition. Chemical shifts were referenced to a residual solvent, DMSO, at 2.50 ppm. The NMR

spectrum peaks of remdesivir in the formulations produced by TFF were compared to those of

remdesivir unprocessed powder as received from the vendor. Excipients peaks in TFF-remdesivir

powders were also compared to those of unprocessed excipients.

2.9   Dissolution test

Simulated lung fluid (SLF) was used as a dissolution medium. SLF contained

dipalmitoylphosphotidylcholine (DPPC), cholesterol, albumin, transferrin, ascorbic acid and Hank's

balanced salt solution (HBSS). The SLF was prepared by two steps including a preparation of

liposomal dispersion and freeze drying. In the first step, a liposomal dispersion was prepared by

mixing 2.12 mL of 25 mg/mL DPPC and 5 μL of 200 mg/mL cholesterol in chloroform. Then, the

solution was dried using an evaporator until the dry film was obtained. The thin film of lipid was

dispersed by adding 8 mL of 57 mg/mL bovine serum albumin in water, 1 mL of 15 mg/mL transferrin

in water, and 88.5 µL of 10 mM ascorbate solution. The mixture was sonicated until it fully dispersed.

The dispersion was adjusted to 10 mL by adding 1M HBSS). The dispersion was frozen at -20 °C for

6 hours, and primary dried in the lyophilizer at -40 °C for 30 hours and ramp to 25 °C for 12 hours

and finally secondary dried at 25 °C for 20 hours. The lyophilized SLF was stored at 2-8 °C and

reconstituted with purified water upon dissolution testing.

Dissolution testing of F10, F11, F12, F13, F14 and unprocessed crystalline remdesivir was

performed under sink conditions in SLF. An NGI (MSP Corp, Shoreview, MN) connected to a High

Capacity Pump (model HCP5, Copley Scientific, Nottingham, UK) and a Critical Flow Controller

(model TPK 2000, Copley Scientific, Nottingham, UK) was used to load aerosolized powder. Three

24 mm diameter filter papers were placed on stage 4 on the NGI to capture aerosolized powder to be

used in the dissolution test. The transwell systems were modified as described in Rohrschneider et

al. The filter papers were loaded in the donor compartment of the modified transwell systems.

Reconstituted SLF was heated up to 37 °C before loading 1.5 mL in the six-well plate. The experiment

was initiated after placing the donor compartment in the plate and adding 0.1 mL of SLF onto the filter

papers. The modified transwell systems was placed on an orbital shaker at 100 rpm and 37 °C.

Samples of 0.1 mL were collected at 10, 20, 30, 45, 60, 90, 120, 180, 240, 360, and 540 mins.

Samples were centrifuged and diluted with DMSO/acetonitrile (10/90) to separate proteins in SLF.

Supernatants were collected and analyzed by HPLC as described in section 2.3. Removed volumes

were refilled with fresh SLF to maintain the dissolution volume. At the last time point, the filter

membranes were removed and washed with DMSO/acetonitrile (10/90) to analyze the amount of

undissolved drug.

2.10 Stability study

TFF powder of F10, F12 and F13 was transferred to a borosilicate glass vial. The vials were

dried in a lyophilizer at 25 °C and 100 mTorr for 6 h, then backfilled with nitrogen gas before stoppering

inside the lyophilizer. The vials were sealed with aluminum caps before packing in an aluminum foil

bag with a desiccant. All samples were stored at 25 °C/60%RH. After one month, samples were

collected and analyzed to determine physical stability and chemical stability. Additionally, samples

were analyzed to determine aerosol performance.

2.11 In vivo pharmacokinetic study

2.11.1 Single-dose dry powder insufflation

In vivo pharmacokinetic study was conducted using Sprague–Dawley rats in compliance with

the Institutional Animal Care and Use Committee (IACUC; Protocol Number AUP-2019-00253)

guidelines at The University of Texas at Austin. Rats weighing 200 to 250 g (average weight of 214.3

g) were housed in a 12-h light/dark cycle with access to food and water ad libitum and were subjected

to one week of acclimation time to the housing environment.

Intratracheal administration was carried out using the dry powder insufflator (DP-4M model,

Penn-Century Inc., Philadelphia, PA) connected to the air pump (AP-1 model, Penn-Century Inc.,

Philadelphia, PA). TFF powder was passed through a No. 325 sieve (45 μm aperture) to break down

large aggregates. TFF remdesivir F10 and F13 were given to 5 rats per group. A precisely weighed

quantity of sieved TFF powder was introduced into the sample chamber of the insufflator device. The

dose was targeted at 10 mg/kg. Each rat was placed on its back at a 45 ° angle, and its incisors were

secured with a rubber band. A laryngoscope was used to visualize the trachea and insert the

insufflator device into the trachea. The sieved TFF powder was actuated into the lung using 3-5

pumps (200 μL of air per pump). The mass of powder delivered was measured by weighing the

chamber before and after dose actuations.

292   Following insufflation, whole blood was collected at each time point (5, 15, 30, 2, 4, 8, and 24

293 hours) by cardiac puncture into heparinized vials. The blood samples were centrifuged at 10,000 rpm

294 for 3 mins to obtain plasma. Lung were harvested after collecting blood at the last time point. Plasma

295 samples and lungs were frozen and stored at -20 °C until analysis.

296 **2.11.2 Analysis of remdesivir and its metabolites in plasma and lung tissue**

297   Plasma samples were prepared by combining 100 µL of plasma with 100 µL of methanol that

298 contained 100 ng/mL of the heavy labeled internal standards for remdesivir and GS-441524, followed

299 by vortexing and centrifuging at 12,000 rpm for 15 minutes. The supernatant was placed in a 96 well

300 plate for LC/MS/MS analysis.

301   Lung tissue samples were cut in small pieces, added into a 2 mL tube with 3.5 g of 2.3 mm

302 zirconia/silica beads (BioSpec Producs, Bartlesville, OK), and homogenized at 4800 rpm for 20

303 seconds. After homogenization, 1000 µL of methanol, containing 100 ng/mL internal standards for

304 remdesivir and GS-441524 was added to the tube, and the tube was vortexed and centrifuged at

305 12,000 rpm for 15 minutes. The supernatant was placed in a 96 well plate for LC/MS/MS analysis.

306 Calibration standards were prepared for plasma and lung tissue in the same fashion, spiking

307 remdesivir and GS-441524 standard solutions into the blank plasma and lung tissue to obtain matrix

308 matched calibrations. The calibration range was 0.1 – 1000 ng/ml for plasma and 50 – 10,000 ng/ml

309 for lung tissue. The calibration range was chosen to bracket sample levels measured.

310   Remdesivir and GS-441524 were separated on an Agilent Poroshell column (2.1 x 50 mm, 2.7

311 µm) (Agilent, Santa Clara, CA) using a gradient of 0 to 90.25% of acetonitrile with 0.025%

312 trifluoroacetic acid in 5 minutes at a flow rate of 0.35 mL/min, and the column temperature of 40°C.

313 10 µL of each sample was injected to analyze with an Agilent 6470 triple quadrupole LC/MS/MS

314 system (Agilent, Santa Clara, CA).

315 **2.11.3 Pharmacokinetic analysis**

316   The concentration of remdesivir and GS-441524 in plasma versus time was plotted. The data

317 are presented as mean ± standard deviation (SD). The plots were used to compare the

pharmacokinetic behavior of F10 and F13. A one compartmental analysis was performed to

determine key pharmacokinetic parameters using Microsoft Excel 2016.

2.12 Statistical analysis

The statistical significance of experimental results was conducted using Student's t-test in

JMP 10. Alpha level was set at 0.05.

3.   Results

3.1   Physical properties of TFF remdesivir

The particle morphology of TFF remdesivir powder formulations containing different drug

loading concentrations and different excipients was determined using SEM. All formulations exhibited

a brittle matrix structure of highly porous particles (Figure 1). For acetonitrile/water solvent system,

Captisol®- and leucine-based formulations containing the same drug loading showed relatively

smaller nanostructure aggregates, compared to mannitol- and lactose-based formulations. Different

cosolvent systems resulted in different brittle matrix powder morphologies as observed by SEM. At

the same drug load, mixture prepared from 1,4-dioxane/water produced smaller nanostructure

aggregates, compared to an equivalent mixture prepared from acetonitrile/water (Figure 1B).

Figure 1. Brittle matrix particle morphology of TFF remdesivir compositions; (A) formulations

prepared in acetonitrile/water (50/50) co-solvent mixture; (B) formulations prepared in 1,4-

dioxane/water (50/50) co-solvent mixture.

Figure 2 shows the x-ray diffraction patterns of TFF remdesivir powder formulations. No sharp

peaks of remdesivir were observed in any of formulations indicating remdesivir was amorphous after

the TFF process. The drug loading and type of co-solvent did not affect the morphology of remdesivir.

For excipients, sharp peaks of mannitol (13.5, 17, 18.5, 20.2,21, 22, 24.5, 25, 27.5, and 36 degree

two-theta) were observed in F3, F7 and F11, indicating that mannitol in these formulations remained

crystalline as a mixture of δ and α form [31]. Similarly, some peaks of leucine (6 and 19 degree two-

theta) were observed in the TFF remdesivir – leucine formulations, F5, F9, F13, and F20, indicating

that leucine remained crystalline after the process. In contrast, Captisol® and lactose in the

compositions were amorphous after the TFF process.

Figure 2. XRPD diffractograms of TFF remdesivir compositions; (A) formulations prepared in

acetonitrile/water (50/50) solvent mixture; (B) formulations prepared in 1,4-dioxane/water (50/50)

solvent mixture.

mDSC was employed to identify the glass-forming ability of remdesivir and determine the

glass transition temperature (Tg) of remdesivir in each formulation. Figure 3A shows a mDSC

thermogram of remdesivir unprocessed powder. The first heating cycle demonstrated remdesivir was

crystalline with a melting point of ~133 °C. The cooling and second heating cycle showed that the

glass transition temperature of remdesivir was about 60 °C. No recrystallization peak was observed

in any cycle.

Figure 3B shows the physical state of remdesivir and excipients. Despite the different

formulation compositions, no melting peak of remdesivir was observed in any formulations, it

remained amorphous. These results agree with XRD diffractograms showing that remdesivir was

amorphous after the TFF process. The mDSC thermograms showed that the Tg of remdesivir in all

formulations was in the range 56-61 °C. Captisol® in both F6 and F9 showed no endothermic peaks,

demonstrating that Captisol® was in amorphous form. Endothermic peak of mannitol in F7 and F11

was observed at ~165 °C and ~160 °C, respectively, indicating mannitol was crystalline after TFF

process. For lactose-based formulations, two glass transition temperatures were observed, including

Tg of remdesivir (~59 °C) and Tg of lactose (~117 °C), indicating lactose and remdesivir were not

molecularly miscible. For leucine-based formulations, an endothermic peak was observed in F9 and

F13 at ~227 °C and ~203 °C, respectively, which are not identical with the melting peak of

unprocessed leucine (~300 °C) [32]. Therefore, the melting point depression of leucine can be

attributed to an interaction of remdesivir and leucine.

(A)

Figure 3. mDSC thermograms; (A) remdesivir unprocessed powder; (B) TFF remdesivir powder formulations.

3.2 Aerodynamic properties of TFF remdesivir

The aerodynamic particle size distribution of TFF remdesivir formulations was evaluated using a Plastiape® RS00 high resistance DPI and NGI apparatus. Figure 4 and Table 2 show aerodynamic diameter distribution and aerosol performance of TFF remdesivir formulations, respectively. In vitro aerodynamic testing revealed that drug loading affected the aerosol performance of the TFF formulations. Despite using different excipients, the aerosol performance generally increased as the drug loading was increased. The FPF (of recovered dose) of Captisol®-based formulations containing 5%, 20%, 50% and 80% of remdesivir (F1, F2, F6, F10, respectively) were $30.73 \pm 1.11\%$, $45.84 \pm 6.03\%$, $65.19 \pm 1.18\%$ and 68.29%, respectively. The MMAD of these TFF remdesivir powder formulations was 3.10 ± 0.04 µm, 2.59 ± 0.15 µm, 2.22 ± 0.14 µm, and 2.16 ± 0.21µm, respectively. For lactose- and mannitol-based formulations, there were significant increases in FPF (of recovered dose) when the drug loading was increased from 20% to 50%; however, FPF did not significantly change when the drug loading was increased further from 50% to 80% with these materials. Interestingly, FPFs of leucine-based formulations containing 20%, 50% and 80% of remdesivir were not significantly different. However, MMADs of these leucine-based formulations at

drug loadings of 20%, 50% and 80% increased from 0.74 ± 0.06 μm to 0.82 ± 0.07 μm to 1.45 ± 0.07

μm, respectively. When the drug loading was increased to 100% (i.e., excipient free), F14 exhibited

an FPF of 62.33% (of recovered drug) and an MMAD of 2.06 μm.

An effect of the excipient on aerosol performance was also observed. At the same drug

loading, leucine-based formulations exhibited significantly greater FPF (81-85% FPF), and smaller

MMAD (0.7–1.5 μm), compared to other excipients, which indicates that leucine has distinct

advantages over other compositions in terms of aerosol performance.

We also investigated the effect of co-solvent mixture and solid content on the aerosol

performance of the formulations. For 100% drug loading (i.e., no excipient present in composition),

significant differences in MMAD, FPF, and EF were observed between the formulations prepared

from the acetonitrile/water (F14) and a 1,4-dioxane/water (F15 and F16) co-solvent systems. F15 and

F16 exhibited smaller MMAD, larger FPF and EF, compared to F14. Comparing the formulations

prepared in the same solvent system, F16 showed better aerosol performance although it was

prepared at higher solid content.

Similar trends were observed in the Captisol®, mannitol, lactose, and leucine-based

formulations. Captisol®, mannitol, lactose, and leucine-based formulations that were prepared in a

1,4-dioxane/water co-solvent (F17, F18, F19, F20, respectively) showed smaller MMAD, compared

to the same compositions prepared in an acetonitrile/water cosolvent system (F10, F11, F12 and F13,

respectively).

(B)

Figure 4. Aerodynamic diameter distribution of TFF remdesivir formulations when emitted from a

Plastiape® RS00 high resistance DPI at a flow rate of 60 L/min (n=3); (A) formulations prepared in

acetonitrile/water (50/50) co-solvent mixture; (B) formulations prepared in 1,4-dioxane/water (50/50)

co-solvent mixture

Table 2. Aerosol performance of TFF remdesivir powders using a Plastiape® RS00 high resistance

DPI at a flow rate of 60 L/min (n=3). (MMAD, mass median aerodynamic diameter; GSD, geometric

standard deviation; FPF, fine particle fraction; EF, emitted fraction)

Formulations	MMAD (μm)	GSD	FPF (%, recovered dose)	FPF (%, delivered dose)	EF (%, recovered dose)
F1	3.10 ± 0.04	2.88 ± 0.03	30.73 ± 1.11	36.87 ± 0.78	83.35 ± 1.83
F2	2.59 ± 0.15	2.56 ± 0.28	45.84 ± 6.03	55.00 ± 5.44	83.32 ± 6.70
F3	2.03 ± 0.10	2.76 ± 0.07	56.21 ± 2.16	62.27 ± 2.72	90.28 ± 0.67
F4	2.02 ± 0.37	2.69 ± 0.10	52.38 ± 4.65	60.98 ± 3.16	85.86 ± 5.40
F5	0.74 ± 0.06	3.50 ± 0.13	86.03 ± 2.82	92.10 ± 1.76	93.40 ± 1.73
F6	2.22 ± 0.14	2.73 ± 0.22	65.19 ± 1.18	72.17 ± 2.04	90.37 ± 2.49
F7	2.32 ± 0.07	2.57 ± 0.03	65.90 ± 0.91	73.08 ± 0.22	90.17 ± 0.97
F8	2.40 ± 0.39	2.28 ± 0.11	68.43 ± 4.33	75.74 ± 4.60	90.34 ± 0.82
F9	0.82 ± 0.07	3.19 ± 0.10	85.83 ± 3.96	92.99 ± 1.11	92.28 ± 3.23
F10	2.16 ± 0.21	2.42 ± 0.06	71.48 ± 5.52	78.08 ± 5.51	91.48 ± 2.09
F11	2.44 ± 0.06	2.53 ± 0.06	64.21 ± 3.53	71.62 ± 3.10	89.61 ± 1.30
F12	2.03 ± 0.11	2.51 ± 0.12	70.32 ± 3.39	77.24 ± 2.94	91.02 ± 1.55
F13	1.45 ± 0.07	2.17 ± 0.01	82.71 ± 2.54	89.68 ± 0.91	92.21 ± 1.94
F14	2.09 ± 0.07	2.79 ± 0.13	65.80 ± 2.68	74.44 ± 1.29	88.37 ± 2.16
F15	1.53 ± 0.16	2.70 ± 0.10	81.66 ± 0.80	85.11 ± 1.20	95.95 ± 0.56
F16	1.42 ± 0.20	2.77 ± 0.18	84.25 ± 1.87	86.77 ± 1.67	97.10 ± 0.40

F17	1.55 ± 0.16	2.99 ± 0.43	74.59 ± 7.03	78.76 ± 6.75	94.67 ± 0.97
F18	2.03 ± 0.20	2.48 ± 0.07	75.34 ± 1.06	79.94 ± 2.72	94.30 ± 1.94
F19	1.28 ± 0.10	2.92 ± 0.17	82.40 ± 1.18	87.28 ± 0.57	94.41 ± 0.83
F20	1.29 ± 0.11	3.15 ± 0.18	82.26 ± 3.86	86.09 ± 3.51	95.53 ± 0.60

3.3   Interactions between TFF remdesivir and excipients

1H-NMR was performed to identify interactions between remdesivir and excipients. Figure 5B

demonstrates an expansion of 1H-NMR spectra for selected TFF remdesivir powder formulations and

remdesivir unprocessed powder. While the peak at 6.03 ppm is sharp and does not show any

differences from the presented samples, the peaks at 5.37 and 6.33 ppm of F9 exhibited broader

peak. Also, these peaks were slightly shifted to downfield. Figure 5C shows an expansion of 1H-NMR

spectra of F9 and leucine unprocessed powder. The peak at 3.09 ppm of leucine was also shifted

downfield to 3.13 ppm in F9.   These results indicate that remdesivir and leucine form interactions

during the TFF process.

3.4   Stability study of TFF remdesivir

3.4.1 Chemical stability of TFF remdesivir compositions

Chemical stability of remdesivir after TFF processing was confirmed by 1H-NMR. Figure 5A

presents the 1H-NMR spectra of selected TFF remdesivir powder formulations and remdesivir

unprocessed powder. Remdesivir peaks from the TFF remdesivir powder formulations consisting of

50% (w/w) remdesivir with mannitol, lactose, and leucine (F7, F8, and F9, respectively) are identical

to the peaks of remdesivir unprocessed powder. The 80% (w/w) remdesivir with Captisol® (F10) and

100% remdesivir without excipient (F14) powders also presented indistinguishable peaks

corresponding to remdesivir.   Therefore, neither the aqueous-organic cosolvent mixture nor the

conditions used in the TFF process affected the chemical stability of remdesivir.

After one-month storage at 25 °C/60% RH, samples were analyzed for chemical stability by

NMR and HPLC. NMR spectra demonstrated that no chemical shift of remdesivir peaks and no new

peaks were observed in F10, F12 and F13. Additionally, no degradant peak was observed in the

HPLC chromatograms (data not shown), which agrees with NMR spectra. Both analyses indicated

that remdesivir was chemically stable without degradation after storage.

Figure 5. ^{1}H-NMR spectra; (A) TFF remdesivir powder formulations (F7, F8, F9, F10, F14) at initial condition, TFF remdesivir powder formulations (F10, F12, F13) at 25°C/60%RH at 1 month, and remdesivir unprocessed powder (Remdesivir); (B) expansion of TFF remdesivir powder formulations (F7, F8, F9, F10, F14) and remdesivir unprocessed powder (Remdesivir); (C) expansion of TFF remdesivir powder (F9) and leucine unprocessed powder (Leucine).

3.4.2 Physical stability of TFF remdesivir compositions

Samples after storage at 25°C/ 60%RH were analyzed for physical stability by XRD. XRD diffractogram demonstrates that no remdesivir peak was observed in any samples after storage (Figure 6), indicating that remdesivir in F10, F12 and F13 remained amorphous after one-month storage.

Figure 6. XRD diffractograms of F10, F12, and F13 after storage at 25 °C/60% RH.

3.4.3 Aerosol performance after storage

The aerosol performance of F10, F12 and F13 was analyzed after one-month storage at 25 °C/60%RH. Overall, MMAD and FPF (of recovered dose) of F10, F12, and F13 was in the range of 1-2 μm and 74-83%, respectively (Table 3). Although MMAD was slightly decreased in all three formulations, there was no significant change in aerosol performance after one-month storage at 25 °C/60%RH.

Table 3. Aerosol performance of TFF remdesivir powders after one-month storage at 25 °C/60%RH using a Plastiape® RS00 high resistance DPI at a flow rate of 60 L/min (n=3). (MMAD, mass median aerodynamic diameter; GSD, geometric standard deviation; FPF, fine particle fraction; EF, emitted fraction)

Formulation	Condition	MMAD (µm)	GSD	FPF (%, recovered dose)	FPF (%, delivered dose)	EF (%, recovered dose)
F10	Initial	1.99 ± 0.19	2.49 ± 0.13	74.84 ± 4.70	80.22 ± 3.08	93.22 ± 2.71
	25 °C/60%RH, 1 M	1.70 ± 0.16	2.67 ± 0.13	75.85 ± 1.45	80.40 ± 1.79	94.35 ± 0.54
F12	Initial	1.78 ± 0.23	2.72 ± 0.30	76.88 ± 6.37	81.79 ± 3.70	93.88 ± 3.61
	25 °C/60%RH, 1 M	1.41 ± 0.20	2.65 ± 0.16	81.46 ± 2.32	85.89 ± 1.16	94.84 ± 1.52
F13	Initial	1.33 ± 0.10	2.54 ± 0.42	83.07 ± 1.35	88.19 ± 2.43	94.22 ± 1.09
	25 °C/60%RH, 1 M	1.19 ± 0.22	2.96 ± 0.16	81.23 ± 4.16	87.60 ± 1.82	92.70 ± 3.02

Figure 7. Aerosol performance of TFF remdesivir powder formulations after one-month storage at 25

°C/60%RH when emitted from a Plastiape® RS00 high resistance DPI at a flow rate of 60 L/min (n=3).

3.5 Drug release in simulated lung fluid

The in vitro dissolution profile of remdesivir in SLF media under sink conditions are shown

in Figure 8. The dissolution rates of TFF remdesivir formulations were significantly higher than the

unprocessed crystalline remdesivir ($p < 0.05$). More than 80% of remdesivir in F10, F11, F12 and F13

was released and dissolved within 90 mins, while 80% remdesivir in F14 was dissolved in 180 mins.

Figure 8. Dissolution profiles of TFF remdesivir dry powder compositions in simulated lung fluid.

3.6 *In vivo* **pharmacokinetic study**

The plasma concentration-time profiles of remdesivir and GS-441524 following a single

inhalation dose of TFF remdesivir formulations are shown in Figures 9A and 9B, respectively.

Remdesivir was detected in plasma at 5 min (0.068 ng/mL) and reached a maximum 30 min (1.149

ng/mL) following pulmonary dosing of F10. A much smaller amount of remdesivir (< 0.01 ng/mL) was

detected at 5 and 15 mins following pulmonary dosing of F13. Remdesivir concentrations were below

the lower limit of quantitation in plasma after 2 hours and after 30 minutes following pulmonary dosing

of F10 and F13, respectively.

Figure 9B shows a comparison of plasma GS-441524 concentration-time profile from F10

and F13, and the calculated pharmacokinetic parameters following pulmonary administration are

presented in Table 4. The plasma GS-441524 profiles following pulmonary dosing of F10 and F13

has peak concentrations (C_{max}) of 220.4 ± 118.0 ng/mL at 2 hours and 264.3 ± 88.5 ng/mL at 4 hours,

respectively, before plasma concentrations started decreasing. AUC_{0-24} and AUC_{inf} of F13 were

significantly greater than those of F10 ($p<0.05$).

Table 4 shows the amount of remdesivir and GS-441524 that remained in the lung tissues of

treated animals at 24 hours following pulmonary administration of F10 and F13. Despite no statistical

difference, F10 exhibited greater amounts of remdesivir in the lungs at 24 hours, while F13 showed

greater amounts of GS-441524 in the lungs at 24 hours.

Figure 9. Plasma concentration - time profiles of F10 (Remdesivir-Captisol ®; 80/20 w/w) and F13

(Remdesivir-Leucine; 80/20 w/w) after a single inhalation administration in rats; (A) Remdesivir; (B)

GS-441524.

Table 4. *In vivo* pharmacokinetic parameters for the plasma GS-441524 concentrations of F10 and

F13 following a single inhalation administration.

Formulations	T_{max} (h)	C_{max} (ng/mL)	AUC_{0-24} (ng·h/mL)	AUC_{inf} (ng·h/mL)
F10	2	220.4 ± 118.0	2115.3 ± 979.6	2397.8 ± 1178.3
F13	4	264.3 ± 88.5	2788.5 ± 857.1	3204.9 ± 967.6

503 **Table 5.** Amount of remdesivir and GS-441524 in the lungs at 24 hours following a single inhalation

504 administration of F10 and F13.

Formulation	Rat	Wet Lung (mg)	Remdesivir		GS-441524	
			Total Weight (µg)	In wet Lung (ng/mg)	Total Weight (µg)	In wet Lung (ng/mg)
F10	A	1164.5	16.5	14.2	6.2	5.3
	B	1494.8	11.3	7.5	6.4	4.3
	C	1255.2	13.6	10.9	10.8	8.6
	D	1194.0	0.1	0.1	0.2	0.2
	E	1176.3	16.2	13.8	12.2	10.4
	Average ± SD	-	11.6 ± 6.7	9.3 ± 5.8	7.2 ± 4.7	5.8 ± 4.0
F13	F	1025.1	12.4	11.2	12.5	11.3
	G	1126.8	7.7	7.6	15.3	14.9
	H	1254.5	4.0	3.5	18.7	16.6
	I	1263.1	3.4	2.7	18.7	14.9
	J	1164.5	0.6	0.5	1.2	1.0
	Average ± SD	-	5.6 ± 4.5	5.1 ± 4.3	13.4 ± 7.3	11.7 ± 6.3

4.   Discussion

4.1   Thin film freezing produces high potency remdesivir dry powders for inhalation with
high aerosol performance

We investigated the feasibility of using the thin film freezing process to prepare inhalable

remdesivir powder formulations. Different excipients including Captisol®, mannitol, lactose, and

leucine were evaluated in this study. Mannitol and lactose are presently contained in FDA approved

inhalation dosage forms, while leucine has gained recent interest for pulmonary delivery and is being

used in an inhaled product in FDA clinical trials [33-35]. Captisol® was selected in this study since it

is a solubilizer used in both the solution concentrate for dilution and infusion and the lyophilized

powder for reconstitution/dilution and infusion of current remdesivir products [3].

The RS00 high resistance monodose DPI is a capsule-based DPI device that is available for

commercial product development, and it functions to disperse the powder based on impaction force.

A previous study confirmed that this impact-based DPI can disperse low-density brittle matrix powders

made by TFF process into respirable particles better than a shear-based DPI (e.g., Handihaler®) [36].

Another study also evaluated the performance of different models of the monodose DPI (RS01 and

RS00) on the aerosol performance of brittle matrix powders containing voriconazole nanoaggregates

prepared by TFF [37]. It was shown that the RS00 device exhibited better powder shearing and

deaggregation through smaller holes of the capsule created by the piercing system of the RS00

device [37]. Therefore, the RS00 high resistance Plastiape® DPI was selected for this study.

We found that excipient type and drug loading affect the aerosol performance of TFF

remdesivir powder compositions. Overall, the aerosol performance of TFF remdesivir powders

increased as the drug loading was increased, a highly desirable feature. This trend is obvious for the

Captisol®-, lactose-, and mannitol-based formulations when the drug loading was increased from 20%

to 50%. Furthermore, high potency TFF remdesivir powder without excipients (F14 and F15) also

exhibited high FPF and small MMAD, which indicates remdesivir itself has a good dispersing ability

without the need of a dispersing excipient when prepared using the TFF process. This shows that the

TFF technology can be used to minimize the need of excipient(s) in the formulation, thus maximizing

the amount of remdesivir being delivered to the lungs by dry powder inhalation.

The aerosol performance of the leucine-based formulations did not significantly change when

the drug loading was increased from 20% to 80%, and these formulations exhibited superior aerosol

performance compared to the other excipient-based formulations studied in this paper. This is likely

attributed to the surface modifying properties of leucine. Several papers report that leucine can

minimize the contact area and distance between particles [38, 39]. This decreases Van der Waal

forces between drug particles and subsequently increases aerosol performance.

Additionally, different co-solvent systems affected the aerosol performance of TFF remdesivir

powders. The formulations, except for the TFF remdesivir-leucine, prepared in a 1,4-dioxane/water

co-solvent system exhibited smaller MMAD and higher FPF than the formulations prepared in an

acetonitrile/water co-solvent system. Interestingly, the co-solvent systems did not affect the aerosol

performance of the TFF remdesivir-leucine formulation.   This agrees with SEM figures showing that

the formulations prepared in a 1,4-dioxane/water co-solvent system has smaller nanostructured

aggregates than the formulations prepared in an acetonitrile/water cosolvent system. This may be

due to the difference in viscosity of the solvent system. The viscosity of ACN/water (50/50 v/v) was

lower than that of 1,4-dioxane/water (50/50 v/v) (0.81 vs. 1.62 mPa·s) [40, 41]. Our results agree with

previous studies by Beinborn et al. [42] and Moon et al. [37] showing that the viscosity of the solvent

system has an impact on the aerosol performance of TFF powder. The higher viscosity of the co-

solvent minimizes the movement of molecules during the ultra-rapid freezing step, resulting in more

homogenous distribution in the frozen state [37]. On the contrary, the lower-viscosity solvent allows

molecules to move more readily, which increases the chance of molecular aggregation and

subsequently decreases the aerosol performance [37].

4.2   Physical and chemical stability of remdesivir dry powder produced using TFF

Both XRD diffractograms and DSC thermograms showed that remdesivir was amorphous

after the TFF process. An amorphous form of the drug generally provides for faster dissolution rate

than its crystalline form. Since the drug needs to be absorbed and then penetrate through the cell

membrane before it is hydrolyzed to nucleoside monophosphate GS-441524, TFF remdesivir

powders provide benefits for the dissolution of the deposited powder in the lung fluid that leads to

improvement in the bioavailability and efficacy of the drug when administered by inhalation.

One potential concern related to the amorphous drug is physical instability due to its high

energy state. According to criteria described by Wyttenbach et al., our study confirmed that remdesivir

is categorized as a class III glass-forming drug (i.e., it is a stable glass former) [43], because no

crystallization peak was observed in both the cooling and heating cycles on DSC. It was reported that

the glass forming ability (GFA) has an influence on the physical stability of drug [44]. The GFA class

III drugs have a strong tendency to transform into its amorphous state, resulting in the highest physical

stability compared to class I and class II APIs [44, 45].

Additionally, NMR spectra demonstrated that Captisol® and leucine have interactions with

remdesivir, which may help to stabilize remdesivir during storage. Also, DSC thermograms indicated

that lactose and remdesivir were not molecularly dispersed in one single amorphous phase. To

investigate an effect of drug and excipient interaction on physical and chemical stability, we selected

F10, F12 and F13 in the stability study.

We found that the properties of a class III glass-forming drug agree with the physical stability

of remdesivir. After one-month storage at 25°C/60% RH, the results showed remdesivir in all three

formulations was physically stable. Different types of excipients did not affect the physical stability of

remdesivir. Even though lactose was not molecularly dispersed with remdesivir, it's use did not affect

the physical stability of remdesivir. Lastly, these results indicate that amorphous remdesivir was

physically stable without a stabilizer.

In addition to the physical stability, remdesivir, as a prodrug, is prone to degrade by hydrolysis

in aqueous solution. Since an organic/water co-solvent system is required to dissolve the drug and

excipients in the TFF process, chemical stability is another concern during preparation. NMR spectra

demonstrated that remdesivir did not chemically degrade as a result of the TFF process. Even though

remdesivir was exposed to binary co-solvent systems consisting of water during the process, the

entire TFF process used to produce remdesivir dry powder inhalation formulations did not induce

chemical degradation of the parent prodrug. Furthermore, remdesivir was chemically stable by HPLC

(data not shown) and NMR after one-month storage at 25°C/60%RH.

4.3   Remdesivir in TFF powder compositions can be dissolved, absorbed and metabolized to

GS-441524 in the lungs.

Since remdesivir is a poorly water-soluble drug, its dissolution may be a critical factor of drug

release in the lung fluid, especially in high drug load formulations. Undissolved particles can be

cleared by mucociliary clearance or macrophage uptake, causing lower drug concentration, lung

irritation, and inflammatory response [46]. Therefore, a dissolution test was evaluated in this study.

The dissolution profile demonstrated that physical form of drug appears to have a significant effect

on the dissolution rate. Amorphous remdesivir prepared using TFF had a much faster dissolution rate

in simulated lung fluid as compared to crystalline remdesivir. Moreover, the high porosity and high

surface area of the TFF powders also contribute to faster wetting, thereby enhancing the dissolution

rate [30, 47].

The presence of excipients also affected the dissolution rate of remdesivir. The dissolution

rate of 100% drug load formulation (i.e., no excipient was used), even though it is amorphous, was

slower, compared to 80% drug loading formulations. The enhancement of dissolution rate is likely

attributed to improved wetting from the use of excipients. Lactose and mannitol are hydrophilic

carriers that can wet the powder formulations rapidly, promoting faster exposure of the drug particles

to the dissolution medium [48]. In the case of Captisol®, cyclodextrins and their derivatives have been

widely used as solubility and dissolution modifying excipients. In our study, the molar ratio of

remdesivir to Captisol® in the high drug load composition (80/20 w/w) was 10:1, meaning that only

about 10% of remdesivir was present in a complex with Captisol® in the solubilized form, the

remaining remdesivir was not complexed with Captisol® and was present as amorphous powder in

the matrix. It is known that certain drugs form a complex with Captisol® by hydrophobic interactions

[15-17]. The inclusion complexation between cyclodextrin and drug can increase the aqueous

solubility of drug, and increase the dissolution rate [49]. It was previously reported that the solubility

of remdesivir can be increased by Captisol®, and Veklury®, the commercial product containing

remdesivir, includes Captisol® to solubilize remdesivir for the IV solution formulation [15]. Interestingly,

the TFF leucine-based formulation exhibited a higher dissolution rate than the TFF remdesivir

formulations that contain mannitol or lactose, despite its hydrophobic properties. The faster

dissolution rate of the leucine-based formulation is likely attributed to interactions (e.g., hydrogen

bonding) between remdesivir and leucine. Peaks corresponding to acidic protons (-–OH or -–NH) of

remdesivir from F9, a leucine-based formulation, became broader than those from the other

formulations without leucine, and shifted downfield. These wider peaks result from inter-molecular

interactions [50]. A peak shift in NMR spectra indicates magnetic field changes. The downfield shift

of the acidic protons from F9 is caused by deshielded proton, reduced electron density, indicating

hydrogen bonding at the protons [51, 52]. Along with a leucine corresponding peak shift from F9, it

was concluded that remdesivir favorably interacts with leucine in hydrogen bonding. The molar ratio

of remdesivir to leucine of the high drug load TFF compositions (80/20 w/w) was 1:1.5, meaning that

leucine in this composition was in excess in terms of its ability to interact with remdesivir. Similarly,

leucine interactions have been reported in Mangal's study to enhance the solubility of azithromycin

[53]. Specifically, leucine formed a complex with azithromycin through hydrogen bonding [53].

Generally, even though leucine is a hydrophobic amino acid, it is a water-soluble excipient that has

higher solubility than poorly water-soluble drugs like remdesivir. Therefore, the interactions between

drug and leucine at the molecular level can increase the dissolution rate of poorly water-soluble drug

[53, 54], like that observed in our study .

Significantly higher plasma levels of GS-441524 than remdesivir were observed in our rat

pharmacokinetic study. The half-life of remdesivir is reportedly much shorter than that of GS-441524.

While half-life of GS-441524 is approximately 24.5 hours, for remdesivir it is only about 1 hour in

humans following multiple once-daily IV administrations [3]. In rats, the half-life of remdesivir in

plasma is reported to be less than 0.9 minute, which is much shorter than that in humans, due to

higher esterase activity in rodent plasma [3]. Therefore, with a T_{max} of 30 minutes and a level of

remdesivir in plasma that is lower than the detection limit at 2 hours, F10 presented an initial rapid

systemic absorption of remdesivir that was complete at less than 2 hours. T_{max} of GS-441524 plasma

concentration (2 hours) also supports this observation. In comparison, remdesivir plasma

concentrations of F13 were lower than those of F10 at 5 and 15 minutes, and it was lower than the

detection limit at the other time points, indicating that the initial systemic absorption of remdesivir from

lung was much slower with F13. With the later T_{max} of GS-441524 plasma concentration, however,

more sustained release of remdesivir from lung to plasma was observed.

The initial faster systemic absorption of F10 from the lungs is likely due to the fact that F10

contains about 10% remdesivir (on a molar basis) that is complexed with Captisol® in a solubilized

form. Higher concentrations of both remdesivir and GS-441524 were detected in plasma following

administration of F10, which contains 20% w/w of Captisol®. Captisol® (i.e., sulfobutylether-β-

cyclodextrin), is known to enhance stability, solubility, and bioavailability of drugs [55]. When

Captisol® forms a complex by hydrophobic interactions with drugs, these interactions are even

greater than with simple physical mixtures of drug and Captisol® [16, 17]. Although F10 and F13

consist of amorphous remdesivir in the TFF remdesivir powder formulations, inclusion of Captisol®

in F10 likely increased the absorption of the complexed remdesivir into systemic circulation due to its

improvement in solubilization [15, 56]. Even though F10 incorporates a high molar ratio of drug to

Captisol® (10:1), and not all of remdesivir in F10 complexes with Captisol®, partial remdesivir-

Captisol® complex will still dissolve faster in lung fluid, rapidly becoming a solution and absorbed

faster into systemic circulation than the non-complexed remdesivir fraction with Captisol® of F10.

Eriksson et al. described that drugs in solution can be absorbed faster than dry powder in lungs in

the isolated perfused lung model [57]. Enhanced absorption of inhaled drug complexed with Captisol®

from lung to systemic circulation was reported in inhaled voriconazole studies. A solution form of

nebulized voriconazole formulation with Captisol® resulted in an 8-fold higher AUC_{plasma} to AUC_{lung}

ratio as compared to nanostructured amorphous voriconazole powder formulations when delivered

to the lung directly (0.64 vs. 0.08 respectively) [21, 23].

Lastly, and related to potential use in COVID-19 therapy, the 50% maximal effective

concentration (EC_{50}) value is important to evaluate antiviral efficacy of drugs. While there are studies

reported to evaluate remdesivir potency to inhibit SARS-CoV-2 replication, only one study has

reported antiviral efficacy of GS-441524 against SARS-CoV-2. Pruijssers et al. determined inhibition

of SARS-CoV-2 replication in established cell lines by GS-441524, and reported that EC_{50} is 0.62 μM

(180.6 ng/mL) in Calu3 2B4 cells, and 0.47 μM (136.9 ng/mL) in Vero E6 cells [58]. The target dose

of 10 mg/kg by dry powder insufflation in our rat pharmacokinetic study achieved these EC_{50} values

necessary to reach EC_{50} to inhibit SARS-CoV-2. The plasma concentrations of GS-441524 in our

pharmacokinetic study also achieved higher than reported EC_{50} of 0.18 μM (52.4 ng/mL) in SARS-

CoV-infected HAE cells [59].

5.   Conclusions

We have demonstrated that low density, highly porous brittle matrix particles of remdesivir

powder formulations made by thin film freezing are highly aerosolizable and suitable for inhaled

delivery to the lung, the site of SARS-CoV-2 replication. The properties of a good glass former, like

remdesivir, supports that remdesivir maintains its physical stability during storage if protected from

moisture. The thin film freezing process is suitable to prepare remdesivir dry powder for inhalation

without causing chemical degradation. Dry powder administration can deliver remdesivir to the lungs,

which appears to be converted to its nucleoside analog GS-441524 in the lung before entering system

circulation based on the data showing negligible levels of remdesivir in the blood at all times for the

TFF-remdesivir-leucine formulation. Therefore, dry powder remdesivir for inhalation produced using

thin film freezing technology can provide a dry powder inhalation product for the treatment of patients

with COVID-19 that can be administered on an outpatient basis and earlier in the disease course

where effective antiviral therapy has a greater potential to reduce the morbidity and mortality associated with the disease.

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References

1.   COVID-19 Dashboard by the Center for Systems Science and Engineering (CSSE) at Johns Hopkins University (JHU). Johns Hopkins University: 2020.

2.   Singhal, T., A Review of Coronavirus Disease-2019 (COVID-19). *Indian J Pediatr* **2020,** *87* (4), 281-286.

3.   *Summary on compassionate use: Remdesivir Gilead*; European Medicines Agency: 03 April 2020, 2020.

4.   Ko, W. C.; Rolain, J. M.; Lee, N. Y.; Chen, P. L.; Huang, C. T.; Lee, P. I.; Hsueh, P. R., Arguments in favour of remdesivir for treating SARS-CoV-2 infections. *Int J Antimicrob Agents* **2020,** *55* (4), 105933.

5.   Amirian, S. E.; Levy, J. K., Current knowledge about the antivirals remdesivir (GS-5734) and GS-441524 as therapeutic options for coronaviruses. *One Health* **2020,** *9*, 100128.

6.   Warren, T. K.; Jordan, R.; Lo, M. K.; Ray, A. S.; Mackman, R. L.; Soloveva, V.; Siegel, D.; Perron, M.; Bannister, R.; Hui, H. C.; Larson, N.; Strickley, R.; Wells, J.; Stuthman, K. S.; Van Tongeren, S. A.; Garza, N. L.; Donnelly, G.; Shurtleff, A. C.; Retterer, C. J.; Gharaibeh, D.; Zamani, R.; Kenny, T.; Eaton, B. P.; Grimes, E.; Welch, L. S.; Gomba, L.; Wilhelmsen, C. L.; Nichols, D. K.; Nuss, J. E.; Nagle, E. R.; Kugelman, J. R.; Palacios, G.; Doerffler, E.; Neville, S.; Carra, E.; Clarke, M. O.; Zhang, L.; Lew, W.; Ross, B.; Wang, Q.; Chun, K.; Wolfe, L.; Babusis, D.; Park, Y.; Stray, K. M.; Trancheva, I.; Feng, J. Y.; Barauskas, O.; Xu, Y.; Wong, P.; Braun, M. R.; Flint, M.; McMullan, L. K.; Chen, S. S.; Fearns, R.; Swaminathan, S.; Mayers, D. L.; Spiropoulou, C. F.; Lee, W. A.; Nichol, S. T.; Cihlar, T.;

Bavari, S., Therapeutic efficacy of the small molecule GS-5734 against Ebola virus in rhesus

monkeys. *Nature* **2016,** *531* (7594), 381-5.

7.   Gordon, C. J.; Tchesnokov, E. P.; Feng, J. Y.; Porter, D. P.; Gotte, M., The antiviral compound

remdesivir potently inhibits RNA-dependent RNA polymerase from Middle East respiratory syndrome

coronavirus. *J Biol Chem* **2020,** *295* (15), 4773-4779.

8.   Williamson, B.; Feldmann, F.; Schwarz, B.; Meade-White, K.; Porter, D.; Schulz, J.; Van

Doremalen, N.; Leighton, I.; Yinda, C. K.; Perez-Perez, L., Clinical benefit of remdesivir in rhesus

macaques infected with SARS-CoV-2. *BioRxiv* **2020**.

9.   Lo, M. K.; Jordan, R.; Arvey, A.; Sudhamsu, J.; Shrivastava-Ranjan, P.; Hotard, A. L.; Flint, M.;

McMullan, L. K.; Siegel, D.; Clarke, M. O.; Mackman, R. L.; Hui, H. C.; Perron, M.; Ray, A. S.; Cihlar,

T.; Nichol, S. T.; Spiropoulou, C. F., GS-5734 and its parent nucleoside analog inhibit Filo-, Pneumo-

, and Paramyxoviruses. *Sci Rep* **2017,** *7*, 43395-43395.

10. Agostini, M. L.; Andres, E. L.; Sims, A. C.; Graham, R. L.; Sheahan, T. P.; Lu, X.; Smith, E. C.;

Case, J. B.; Feng, J. Y.; Jordan, R.; Ray, A. S.; Cihlar, T.; Siegel, D.; Mackman, R. L.; Clarke, M. O.;

Baric, R. S.; Denison, M. R., Coronavirus Susceptibility to the Antiviral Remdesivir (GS-5734) Is

Mediated by the Viral Polymerase and the Proofreading Exoribonuclease. *mBio* **2018,** *9* (2).

11. Grein, J.; Ohmagari, N.; Shin, D.; Diaz, G.; Asperges, E.; Castagna, A.; Feldt, T.; Green, G.;

Green, M. L.; Lescure, F. X.; Nicastri, E.; Oda, R.; Yo, K.; Quiros-Roldan, E.; Studemeister, A.;

Redinski, J.; Ahmed, S.; Bernett, J.; Chelliah, D.; Chen, D.; Chihara, S.; Cohen, S. H.; Cunningham,

J.; D'Arminio Monforte, A.; Ismail, S.; Kato, H.; Lapadula, G.; L'Her, E.; Maeno, T.; Majumder, S.;

Massari, M.; Mora-Rillo, M.; Mutoh, Y.; Nguyen, D.; Verweij, E.; Zoufaly, A.; Osinusi, A. O.; DeZure,

A.; Zhao, Y.; Zhong, L.; Chokkalingam, A.; Elboudwarej, E.; Telep, L.; Timbs, L.; Henne, I.; Sellers,

S.; Cao, H.; Tan, S. K.; Winterbourne, L.; Desai, P.; Mera, R.; Gaggar, A.; Myers, R. P.; Brainard, D.

M.; Childs, R.; Flanigan, T., Compassionate Use of Remdesivir for Patients with Severe Covid-19. *N*

Engl J Med **2020**.

12. Ferner, R. E.; Aronson, J. K., Remdesivir in covid-19. *BMJ* **2020,** *369*, m1610.

13. B, C., Mild/moderate 2019-nCoV remdesivir RCT - Full Text View - ClinicalTrials.gov. 2020.

14. B, C., Severe 2019-nCoV remdesivir RCT - Full Text View - ClinicalTrials.gov. 2020. 2020.

15. Stella, V. J.; Rajewski, R. A., Sulfobutylether-beta-cyclodextrin. *Int J Pharm* **2020,** *583*, 119396.

16. Pal, A.; Roy, S.; Kumar, A.; Mahmood, S.; Khodapanah, N.; Thomas, S.; Agatemor, C.; Ghosal,

K., Physicochemical Characterization, Molecular Docking, and In Vitro Dissolution of Glimepiride-

Captisol Inclusion Complexes. *ACS Omega* **2020,** *5* (32), 19968-19977.

17. Beig, A.; Agbaria, R.; Dahan, A., The use of captisol (SBE7-beta-CD) in oral solubility-enabling

formulations: Comparison to HPbetaCD and the solubility-permeability interplay. *Eur J Pharm Sci*

2015, *77*, 73-8.

18. Labiris, N. R.; Dolovich, M. B., Pulmonary drug delivery. Part I: Physiological factors affecting

therapeutic effectiveness of aerosolized medications. *British Journal of Clinical Pharmacology* **2003,**

56 (6), 588-599.

19. Sun, D., Remdesivir for Treatment of COVID-19: Combination of Pulmonary and IV

Administration May Offer Aditional Benefit. *AAPS J* **2020,** *22* (4), 77.

20. Inactive Ingredient Search for Approved Drug Products. U.S. Food and Drug Administration:

2020.

21. Tolman, J. A.; Nelson, N. A.; Son, Y. J.; Bosselmann, S.; Wiederhold, N. P.; Peters, J. I.;

McConville, J. T.; Williams, R. O., 3rd, Characterization and pharmacokinetic analysis of aerosolized

aqueous voriconazole solution. *Eur J Pharm Biopharm* **2009,** *72* (1), 199-205.

22. Tolman, J. A.; Nelson, N. A.; Bosselmann, S.; Peters, J. I.; Coalson, J. J.; Wiederhold, N. P.;

Williams, R. O., 3rd, Dose tolerability of chronically inhaled voriconazole solution in rodents. *Int J*

Pharm **2009,** *379* (1), 25-31.

23. Beinborn, N. A.; Du, J.; Wiederhold, N. P.; Smyth, H. D.; Williams, R. O., 3rd, Dry powder

insufflation of crystalline and amorphous voriconazole formulations produced by thin film freezing to

mice. *Eur J Pharm Biopharm* **2012,** *81* (3), 600-8.

24. Gardenhire, D. S.; Burnett, D.; Strickland, S.; Myers, T. R., A guide to aerosol delivery devices

for respiratory therapists. 4th ed.; American Association for Respiratory Care, 2017.

25. Overhoff, K. A.; Johnston, K. P.; Tam, J.; Engstrom, J.; Williams, R. O., Use of thin film freezing

to enable drug delivery: a review. *Journal of Drug Delivery Science and Technology* **2009,** *19* (2), 89-

98.

26. Wang, Y. B.; Watts, A. B.; Williams, R. O., Effect of processing parameters on the

physicochemical and aerodynamic properties of respirable brittle matrix powders. *Journal of Drug*

Delivery Science and Technology **2014,** *24* (4), 390-396.

27. Watts, A. B.; Wang, Y. B.; Johnston, K. P.; Williams, R. O., 3rd, Respirable low-density

microparticles formed in situ from aerosolized brittle matrices. *Pharm Res* **2013,** *30* (3), 813-25.

28. Wang, Y. B.; Watts, A. B.; Peters, J. I.; Liu, S.; Batra, A.; Williams, R. O., 3rd, In vitro and in vivo

performance of dry powder inhalation formulations: comparison of particles prepared by thin film

freezing and micronization. *AAPS PharmSciTech* **2014,** *15* (4), 981-93.

29. Longest, P. W.; Hindle, M., Small Airway Absorption and Microdosimetry of Inhaled

Corticosteroid Particles after Deposition. *Pharm Res* **2017,** *34* (10), 2049-2065.

30. Overhoff, K. A.; Engstrom, J. D.; Chen, B.; Scherzer, B. D.; Milner, T. E.; Johnston, K. P.;

Williams, R. O., 3rd, Novel ultra-rapid freezing particle engineering process for enhancement of

dissolution rates of poorly water-soluble drugs. *Eur J Pharm Biopharm* **2007,** *65* (1), 57-67.

31. Cares-Pacheco, M. G.; Vaca-Medina, G.; Calvet, R.; Espitalier, F.; Letourneau, J. J.; Rouilly, A.;

Rodier, E., Physicochemical characterization of D-mannitol polymorphs: the challenging surface

energy determination by inverse gas chromatography in the infinite dilution region. *Int J Pharm* **2014,**

475 (1-2), 69-81.

32. Molina, C.; Kaialy, W.; Chen, Q.; Commandeur, D.; Nokhodchi, A., Agglomerated novel spray-

dried lactose-leucine tailored as a carrier to enhance the aerosolization performance of salbutamol

sulfate from DPI formulations. *Drug Deliv Transl Res* **2018,** *8* (6), 1769-1780.

33. Pilcer, G.; Amighi, K., Formulation strategy and use of excipients in pulmonary drug delivery. *Int*

J Pharm **2010,** *392* (1-2), 1-19.

34. ANNUAL REPORT PURSUANT TO SECTION 13 OR 15(d) OF THE SECURITIES EXCHANGE

ACT OF 1934 For the fiscal year ended December 31, 2019

Form 10-K Savara Inc: United States Securities and Exchange Commission, 2020.

35. A Study of AeroVanc for the Treatment of MRSA Infection in CF Patients. ClinicalTrials.gov: no.

NCT03181932, 2017.

36.　Sahakijpijarn, S.; Moon, C.; Ma, X.; Su, Y.; Koleng, J. J.; Dolocan, A.; Williams, R. O., Using thin

film freezing to minimize excipients in inhalable tacrolimus dry powder formulations. *International*

Journal of Pharmaceutics **2020,** *586*, 119490.

37.　Moon, C.; Sahakijpijarn, S.; Koleng, J. J.; Williams, R. O., Processing design space is critical for

voriconazole nanoaggregates for dry powder inhalation produced by thin film freezing. *Journal of*

Drug Delivery Science and Technology **2019,** *54*.

38.　Paajanen, M.; Katainen, J.; Raula, J.; Kauppinen, E. I.; Lahtinen, J., Direct evidence on reduced

adhesion of Salbutamol sulphate particles due to L-leucine coating. *Powder Technology* **2009,** *192*

(1), 6-11.

39.　Mangal, S.; Park, H.; Nour, R.; Shetty, N.; Cavallaro, A.; Zemlyanov, D.; Thalberg, K.; Puri, V.;

Nicholas, M.; Narang, A. S.; Zhou, Q. T., Correlations between surface composition and

aerosolization of jet-milled dry powder inhaler formulations with pharmaceutical lubricants. *Int J*

Pharm **2019,** *568*, 118504.

40.　Thompson, J. W.; Kaiser, T. J.; Jorgenson, J. W., Viscosity measurements of methanol-water

and acetonitrile-water mixtures at pressures up to 3500 bar using a novel capillary time-of-flight

viscometer. *J Chromatogr A* **2006,** *1134* (1-2), 201-9.

41.　Besbes, R.; Ouerfelli, N.; Latrous, H., Density, dynamic viscosity, and derived properties of binary

mixtures of 1,4 dioxane with water at T=298.15 K. *Journal of Molecular Liquids* **2009,** *145* (1), 1-4.

42.　Beinborn, N. A.; Lirola, H. L.; Williams, R. O., 3rd, Effect of process variables on morphology and

aerodynamic properties of voriconazole formulations produced by thin film freezing. *Int J Pharm* **2012,**

429 (1-2), 46-57.

43.　Wyttenbach, N.; Kuentz, M., Glass-forming ability of compounds in marketed amorphous drug

products. *Eur J Pharm Biopharm* **2017,** *112*, 204-208.

44.　Blaabjerg, L. I.; Bulduk, B.; Lindenberg, E.; Lobmann, K.; Rades, T.; Grohganz, H., Influence of

Glass Forming Ability on the Physical Stability of Supersaturated Amorphous Solid Dispersions. *J*

Pharm Sci **2019,** *108* (8), 2561-2569.

45.　Panini, P.; Rampazzo, M.; Singh, A.; Vanhoutte, F.; Van den Mooter, G., Myth or Truth: The

Glass Forming Ability Class III Drugs Will Always Form Single-Phase Homogenous Amorphous Solid

Dispersion Formulations. *Pharmaceutics* **2019,** *11* (10).

46. Jones, R. M.; Neef, N., Interpretation and prediction of inhaled drug particle accumulation in the

lung and its associated toxicity. *Xenobiotica* **2011,** *42* (1), 86-93.

47. Sinswat, P.; Overhoff, K. A.; McConville, J. T.; Johnston, K. P.; Williams, R. O., 3rd, Nebulization

of nanoparticulate amorphous or crystalline tacrolimus--single-dose pharmacokinetics study in mice.

Eur J Pharm Biopharm **2008,** *69* (3), 1057-66.

48. Jiang, N.; Wang, S.; Cheng, Z.; Liu, W., In vitro and in vivo evaluation of porous lactose/mannitol

carriers for solubility enhancement of poorly water-soluble drugs. *Drying Technology* **2020,** *38* (7),

889-902.

49. van der Merwe, J.; Steenekamp, J.; Steyn, D.; Hamman, J., The Role of Functional Excipients

in Solid Oral Dosage Forms to Overcome Poor Drug Dissolution and Bioavailability. *Pharmaceutics*

2020, *12* (5).

50. Maity, S.; Gundampati, R.; Kumar, T., NMR Methods to Characterize Protein-Ligand

Interactions. *Nat Prod Commun* **2019**, 1-17.

51. Hibbert, F.; Emsley, J., Hydrogen Bonding and Chemical Reactivity. *Adv Phys Org Chem* **1990,**

26, 255-379.

52. Konrat, R.; Tollinger, M.; Kontaxis, G.; Krautler, B., NMR Techniques to Study Hydrogen Bonding

in Aqueous Solution. *Monatshefte fur Chemie* **1999,** *130,* 961-982.

53. Mangal, S.; Nie, H.; Xu, R.; Guo, R.; Cavallaro, A.; Zemlyanov, D.; Zhou, Q. T., Physico-

Chemical Properties, Aerosolization and Dissolution of Co-Spray Dried Azithromycin Particles with L-

Leucine for Inhalation. *Pharm Res* **2018,** *35* (2), 28-28.

54. Löbmann, K.; Grohganz, H.; Laitinen, R.; Strachan, C.; Rades, T., Amino acids as co-amorphous

stabilizers for poorly water soluble drugs--Part 1: preparation, stability and dissolution enhancement.

European journal of pharmaceutics and biopharmaceutics : official journal of Arbeitsgemeinschaft fur

Pharmazeutische Verfahrenstechnik e.V **2013,** *85* (3 Pt B), 873-881.

55. Lockwood, S. F.; O'Malley, S.; Mosher, G. L., Improved aqueous solubility of crystalline

astaxanthin (3,3'-dihydroxy-beta, beta-carotene-4,4'-dione) by Captisol (sulfobutyl ether beta-

cyclodextrin). *J Pharm Sci* **2003,** *92* (4), 922-6.

56. Yang, K., What do we know about remdesivir drug interactions? *Dlin Transl Sci* **2020,** *13,* 842-

844.

57. Eriksson, J.; Thorn, H.; Sjogren, E.; Holmsten, L.; Rubin, K.; Lennernas, H., Pulmonary

Dissolution of Poorly Soluble Compounds Studied in an ex Vivo Rat Lung Model. *Mol Pharm* **2019,**

16 (7), 3053-3064.

58. Pruijssers, A. J.; George, A. S.; Schafer, A.; Leist, S. R.; Gralinksi, L. E.; Dinnon, K. H., 3rd;

Yount, B. L.; Agostini, M. L.; Stevens, L. J.; Chappell, J. D.; Lu, X.; Hughes, T. M.; Gully, K.; Martinez,

D. R.; Brown, A. J.; Graham, R. L.; Perry, J. K.; Du Pont, V.; Pitts, J.; Ma, B.; Babusis, D.; Murakami,

E.; Feng, J. Y.; Bilello, J. P.; Porter, D. P.; Cihlar, T.; Baric, R. S.; Denison, M. R.; Sheahan, T. P.,

Remdesivir Inhibits SARS-CoV-2 in Human Lung Cells and Chimeric SARS-CoV Expressing the

SARS-CoV-2 RNA Polymerase in Mice. *Cell Rep* **2020,** *32* (3), 107940.

59. Yan, V. C.; Muller, F. L., Advantages of the Parent Nucleoside GS-441524 over Remdesivir for

Covid-19 Treatment. *ACS Med Chem Lett* **2020,** *11* (7), 1361-1366.

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