Development of novel MOG analogues with increased stability to explore

MCT2 and α-ketoglutarate biology *in vivo*

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ABSTRACT

α-ketoglutarate (αKG) is a central metabolic node with far-reaching influence on cellular

physiology. The αKG analogue *N*-oxalylglycine (NOG) and its membrane-permeable pro-drug

derivative dimethyloxalylglycine (DMOG) have been broadly used as tool compounds both *in*

vitro and *in vivo* to study αKG-dependent processes. In cell culture media, DMOG is rapidly

converted to MOG, a substrate of the monocarboxylate transporter MCT2. The expression

level of MCT2 determines the intracellular concentration of NOG, and, as such, influences the

molecular targets NOG engages with. Here we show that DMOG and MOG are highly unstable

also in mouse blood. We therefore designed and characterised a series of MOG analogues

with two aims: to improve pharmacokinetic properties, and to explore the pharmacophore of

MCT2, a relatively understudied member of the SLC16 family. We report MOG analogues that

maintain MCT2-dependent uptake, including NOG-generating compounds that replicate the

metabolic effects of MOG in a concentration-dependent manner. One such analogue, IPOG,

shows significantly increased blood stability, and an improved overall pharmacokinetic profile,

leading to increased NOG accumulation in MCT2-expressing tumours versus isogenic

controls.

KEYWORDS

INTRODUCTION

The study of metabolism has long been aided by the use of metabolite analogues that allow the rapid and reversible inhibition of enzymes and pathways in different experimental settings[1]. In the field of cancer metabolism, analogue compounds such as 2-deoxyglucose, 6-diazo-5-oxo-L-norleucine (DON) and dichloroacetate (DCA) continue to complement genetic approaches to dissect the strengths and vulnerabilities associated with oncogene-driven metabolic changes in tumours[2–4]. Metabolite analogues are also among some of the most important clinically used chemotherapeutic compounds: from gemcitabine and 5-fluoro-uracil (5-FU), nucleoside analogues used as therapies in pancreatic and colorectal cancer; to methotrexate and pemetrexed, folate analogues administered to treat a range of malignancies[5,6]. The development and refinement of metabolite analogues for *in vivo* studies therefore provides valuable tools to facilitate improved mechanistic understanding of both metabolism and tumorigenesis.

α-ketoglutarate (αKG) is a key metabolic node and understanding its complex biology has been significantly facilitated by the structural analogue *N*-oxalylglycine (NOG), which has been extensively used *in vitro* along with its cell-permeable derivative, dimethyl-oxalylglycine (DMOG), to inhibit a range of αKG-dependent enzymes[7–9]. Most commonly, DMOG is utilised to elicit hypoxia signalling by inhibiting prolyl hydroxylase domain (PHD) enzymes leading to stabilisation of the transcription factor Hypoxia Inducible Factor 1α (HIF1α)[8,10]. HIF1α stabilisation is a therapeutic aim in conditions ranging from ischaemia and anaemia to inflammatory diseases[11,12], and previous studies using DMOG have demonstrated the potential benefits of inhibiting PHDs to achieve these goals[13,14].

As well as being a cofactor for αKG-dependent enzymes, αKG is also the entry point for glutamine carbons into the TCA cycle, a substrate for a large number of transaminase reactions, and has also been shown to influence mechanistic target of rapamycin (mTOR) activity[15]. Through regulation of ten-eleven translocation (TET) hydroxylases and Jumonji demethylases, αKG can modify epigenetic profiles during development[9] and in pathogenic contexts[16]. Additionally, through an as-yet unclear mechanism, αKG can influence aging[17], highlighting that there is still much to be discovered about the physiological functions of this metabolite.

DMOG is selectively toxic to some cancer cell lines, in a manner that strongly correlates to the expression level of the monocarboxylate transporter MCT2[18]. DMOG is unstable in cell culture media and is non-enzymatically converted to the monocarboxylate methyl-oxalylglycine (MOG) with a half-life of 10 min. MOG is a substrate for MCT2, the expression of which determines the concentration of NOG that accumulates intracellularly

([NOG]$_{IC}$). In cells with high MCT2 expression, [NOG]$_{IC}$ can reach millimolar levels sufficient

to engage low-affinity αKG-binding enzymes such as isocitrate dehydrogenase (IDH) and

glutamate dehydrogenase (GDH) to severely disrupt metabolism and cause cytotoxicity.

DMOG has been extensively used *in vivo* for pre-clinical studies where, typically, it is

administered at concentrations that far exceed those required to inhibit the intended

intracellular targets[11,19,20]. Instability as a result of chemical or enzymatic de-esterification and

a subsequent loss of cell-permeability could explain the disparity in potency observed between

in vitro (purified enzyme) and *in vivo* studies, particularly in light of the high level of esterase

activity in blood. In support of this hypothesis, here we show that, in mouse blood, DMOG

degradation leads to a transient accumulation of MOG, which is rapidly converted to NOG.

With the aim of generating a more stable compound for *in vivo* exploration of αKG biology, we

designed and characterised a series of MOG analogues, identified those that recapitulate the

metabolic effects of MOG in cells, and then tested them in mice. The creation of a range of

analogues additionally enabled us to explore the pharmacophore of MCT2 ligands and thereby

increase understanding of this lesser-studied member of the *SLC16* transporter family.

RESULTS

Design and synthesis of MOG analogues

Given its relative instability in cell culture media[18] we reasoned that DMOG, and its

derivative MOG, may also have a short half-life *in vivo* (Fig. 1a). To test this idea, we examined

DMOG stability in whole mouse blood using liquid chromatography-mass spectrometry (LC-

MS) and found that it was almost instantaneously converted to MOG, with a half-life of less

than 1 min. (Fig. 1b). The MOG formed did not accumulate and was rapidly de-esterified to

NOG, with a half-life of 4 min. Synthetic MOG, which only contains the glycinate methyl ester

group, was converted to NOG (Fig. 1c) with a half-life mirroring the conversion of DMOG-

derived-MOG to NOG, suggesting that the oxoacetate methyl group of DMOG is lost first and

more rapidly, followed by the glycinate methyl group. The half-lives of these compounds in

mouse blood are significantly shorter than those previously measured in aqueous solution[18],

which we attribute to the well-documented high level of blood esterase activity.

Low stability of DMOG or MOG in blood prevented further well-controlled studies with

either compound *in vivo*. In order to overcome these issues, we designed a number of MOG

analogues, with two main objectives. Our first and major aim was to improve compound

stability while maintaining the ability to produce NOG intracellularly, in order to improve the *in*

vivo utility of this compound. Secondly, since MOG is a substrate for MCT2, by generating a

chemically diverse range of analogues, we aimed to broaden understanding of the chemical

space accommodated by this transporter.

We initiated our compound design using MOG as a chemical scaffold (**1,** Fig. 1a).

Based on our previous findings and the fact that esters are typically used as pro-drugs for

poorly absorbed carboxylic acid drugs,[21–24] we focused on the methyl ester of MOG as the

primary cause of compound instability in plasma. Furthermore, as the pharmacophore

required for MCT2-driven transport is unknown, substitutions were kept relatively

conservative. Therefore, we replaced the glycinate ester on MOG with i) bulkier alkyl esters

such as ethyl or isopropyl (compounds **2**-**3**), ii) esters possessing methyl substituents at the α

position (compounds **4**-**6**), iii) a ketone (compound **7**), or iv) 5-membered aromatic

heterocycles (compounds **8**-**10**). Importantly, compounds **1**-**6** are predicted to be de-esterified

to form NOG or methyl-substituted NOG, and therefore able to engage intracellular targets for

functional studies.

Esters **2** and **3** were designed to minimally increase the steric hindrance of the ester

substituent, which, would be expected to increase both chemical and metabolic stability[25–29].

We also explored branched esters (**4**-**6**), as substitutions on the α-carbon can increase

chemical stability[30] and cellular esterases can exhibit surprising selectivity toward complex

esters[31–33]. Ketones and amides (**7**) are classical ester isosteres[34], with the amide typically

used to improve the stability of drugs[35]. Notably, such small changes can have a significant

effect on the binding affinity of the compounds to their targets[36]. Finally, 5-membered ring

heterocycles, such as oxadiazoles (**10**) or oxazoles (**8**), have also been successfully used as

ester bioisosteres[37–40].

MCT2-dependent uptake is maintained by a subset of MOG analogues

To determine the MCT2-dependence of MOG analogue uptake into cells, we utilised

HCC1569 cells, a human breast cancer line that expresses very low levels of MCT2 and is

naturally resistant to MOG-induced toxicity[18]. We compared compound uptake in these cells

transduced with a control 'empty vector' plasmid (HCC1569-EV) to an isogenic line expressing

exogenous MCT2 (HCC1569-MCT2) (Fig. 2a, b). After incubation of both cell lines with each

analogue for 4 h, all tested compounds or derivatives thereof could be detected intracellularly.

In the case of compounds **7**-**10**, we detected the intact parent compound, however, as

expected, compounds **1**-**6** were all de-esterified intracellularly and therefore in these cases

we quantified NOG, or the NOG analogues that were formed. Intracellular accumulation of

these compounds occurred to varying extents, with the bulkier alkyl esters (compounds **2** and

3) and α-methyl substituents (compounds **4**-**6**) reaching concentrations in the millimolar range

(Fig. 2c). The de-esterification of these compounds within cells further decreases their

membrane permeability, which could effectively trap them inside cells and may explain the

higher concentrations observed. We defined MCT2-dependent uptake as a two-fold increase

in compound accumulation in HCC1569-MCT2 cells compared with HCC1569-EV cells[41]. The

MCT2-dependence of uptake varied between groups, but was maintained in both of the bulkier

alkyl esters (with the fold-change in uptake of compound **2** equivalent to MOG), as well as in

all 3 of the 5-membered aromatic heterocycles (compounds **8**-**10**, Fig. 2d). Although there was

a modest increase in the uptake of the α-methyl and ketone analogues in MCT2-expressing

cells compared to controls, these compounds did not meet the two-fold cut off criterion and so

were not further studied.

MCT2-dependent analogues replicate the effects of MOG in a NOG-concentration dependent

manner

MOG inhibits cell proliferation and leads to apoptosis in MCT2-expressing cells by

disrupting metabolism[18]. To test the relative efficacy of the bulkier alkyl esters and 5-

membered aromatic heterocycles in eliciting metabolic changes and death in cells, we first

assayed their effect on cell proliferation at varying concentrations, utilising our

HCC1569±MCT2 cell model. Over 96 h, MOG inhibited cell mass accumulation in a

concentration-dependent manner. This inhibition was markedly higher in MCT2-expressing

cells and was associated with increased $[NOG]_{IC}$ and increased apoptosis (Fig. 3a). The

bulkier alkyl esters (**2**,**3**) were also capable of slowing cell proliferation in a manner

proportional to the observed $[NOG]_{IC}$ achieved after dosing with these compounds (Fig. 3b).

Aligned with this observation, only the highest concentrations of compounds **2** and **3** were

able to increase apoptosis in MCT2-expressing cells. The 5-membered aromatic heterocycles

(**8**-**10**) did not affect proliferation of cells, either in the presence or absence of MCT2, except

in the case of compound **9**, where the highest dose led to a small decrease in proliferation

and a concomitant increase in apoptosis.

The most common use of DMOG as a tool compound is to inhibit PHDs and thereby

promote the accumulation of the transcription factor HIF1α. Similar to the effects observed on

cell growth and apoptosis, after 4 h of treatment with 1 mM of each analogue (the

concentration commonly used for treatment of cells with DMOG) either of the alkyl esters (**2**,

3), which produce NOG intracellularly, but not the aromatic heterocycles (**8**-**10**), induced

HIF1α stabilisation (Fig. 3c) in a $[NOG]_{IC}$-dependent manner.

A subset of metabolic effects of MOG are maintained by the bulkier alkyl ester analogues

As well as inhibiting prolyl hydroxylases, upon MCT2-driven accumulation to high

concentrations, NOG is able to engage other, lower-affinity αKG-binding enzymes. The effects

of MOG treatment on metabolism are rapid and extensive, and at high $[NOG]_{IC}$, targets include

αKG-binding enzymes GDH and IDH[18]. Inhibition of these targets can be determined by

monitoring levels of the citrate m+4 and m+5 isotopologues derived from glutamine uniformly

labelled with ^{13}C ([U^{13}C]-glutamine). Citrate m+4 is formed from the oxidative use of glutamine

carbons in the canonical TCA cycle (Fig. 3d), and decreases as a result of GDH inhibition by

NOG[18]. Citrate m+5 is produced by the reductive carboxylation of αKG by IDH (Fig. 3d).

Reductive carboxylation is important in a range of physiological and pathological settings,

including cancer, to enable the use of glutamine-derived carbons for the production of fatty

acids, and to maintain redox balance[42–45].

We analysed the metabolism of HCC1569±MCT2 cells treated for 4 h with MOG

analogues in the presence of [U-^{13}C]-glutamine. The depletion of TCA metabolites that had

previously been observed after MOG treatment was partially recapitulated by compounds **2**

and **3**, again in a [NOG]$_{IC}$ -dependent manner (Fig. 3e). Citrate m+5 was significantly

decreased after treatment with either of the bulkier alkyl esters (**2**, **3**), whereas citrate m+4

was not (Fig. 3f), in line with the fact that IDH is inhibited by lower [NOG]$_{IC}$ than GDH[18]. These

data therefore demonstrate that bulky alkyl ester MOG analogues are able to disrupt reductive

carboxylation independently of effects on the oxidative TCA cycle. This differential inhibition

of IDH and GDH relative to the [NOG]$_{IC}$ is reflective of their relative *in vitro* K_d values[18,46] and

is consistent with our previous findings[18].

As expected, the 5-membered aromatic heterocycles (**8**-**10**, which do not form NOG)

had no effect on the concentrations of TCA intermediates. One of the other key metabolites

affected by MOG is α-glycerophosphate (αGP). Interestingly, even in DMSO-treated cells, a

modest increase in αGP was observed in the MCT2-expressing cells suggesting that this

transporter may lead to some alteration of cellular metabolism in HCC1569 cells (Fig. 3d).

However, even relative to the MCT2-dependent increase, αGP levels were more than doubled

after MOG treatment, with modest potentiation also seen with compounds **2**, **3** and **9**

(Supplementary Fig. 1a).

Together, our data showed that compounds **2** and **3** entered cells and recapitulated

the metabolic effects of MOG, in a manner that was proportional to the [NOG]$_{IC}$. We therefore

chose to examine the stability of these two compounds in mice to enable more in-depth *in vivo*

studies.

Higher alkyl ester analogues of MOG have improved pharmacokinetic profile

To test whether the alkyl ester substitutions changed stability for *in vivo* use, we

measured the conversion of compounds **2** and **3** to NOG in mouse whole blood. In blood, the

half-life of compound **2** was marginally lower than that of MOG (95% confidence intervals (CI)

of 2.27-2.96 min and 3.01-3.69 min, respectively), however, compound **3** had a half-life of 14.4

min (95% CI = 10.18-21.24 min), a 4-fold improvement in blood stability compared with MOG
(Fig. 4a).   As predicted, the three aromatic heterocycles were all stable in blood
(Supplementary Fig. 2a).

Based on its improved stability in blood, we decided to further characterise compound
3 to better understand its pharmacokinetic and distribution profile. After intra-peritoneal (i.p.)
injection at 100 mg/kg into *Nod* SCID gamma (NSG) mice, the measured blood concentration
of the intact analogue was higher in compound **3**-dosed animals at both 15 and 60 min post-
injection than in MOG-dosed animals, with a corresponding increase in exposure (Fig. 4b). In
a small number of animals examined after 2 hours post-injection with MOG or (**3**) most NOG
was found in the liver and kidney, the two major sites of drug metabolism and excretion, with
lower levels observed in the spleen and at detectable, albeit very low, levels in heart, skeletal
muscle and brain (Supplementary Fig. 2b).

Since compound **3** demonstrated an improved pharmacokinetic profile relative to
MOG, it was taken forward into *in vivo* studies, and will henceforth be referred to as isopropyl-
oxalylglycine (IPOG).

MCT2-dependent transport of IPOG in vivo

To explore the utility of IPOG for *in vivo* experiments, we tested whether it would
selectively accumulate in MCT2-expressing tumours. To this end, we established an
orthotopic xenograft assay in which HCC1569 cells±MCT2, were injected into mammary fat
pads on opposing flanks of female NSG mice (Fig. 5a). Tumours grew slowly, as previously
reported[47], and there was no significant difference between the size of tumours from
HCC1569-EV and HCC1569-MCT2 cells (Fig. 5b).

Once tumours had reached 5 mm in diameter, mice were acutely dosed i.p. with either
MOG or IPOG at 100 mg/kg. At 2 hours post injection, mice were sacrificed and tumours were
collected. Analysis by LC-MS showed that the overall concentration of NOG in tumours from
mice dosed with IPOG was higher than in those dosed with MOG (Fig. 5c). Importantly,
tumours expressing MCT2 accumulated significantly more NOG when dosed with IPOG than
paired control tumours, whereas this was not the case in MOG-dosed animals.

Since there was no MCT2-specific accumulation of NOG in MOG-dosed animals, and
MOG is so rapidly converted to NOG in blood, we considered whether the observed NOG in
these tumours could be the result of direct NOG uptake. Indeed, it has been shown that NOG
is a substrate for the organic anion transporter OAT1[48], so to control for this possibility, we
also dosed animals with NOG directly. The concentrations of NOG in both EV and MCT2
tumours were similar to those dosed with MOG (Supplementary Fig. 2c, Fig. 5c), suggesting
that direct NOG uptake is likely the main source of NOG accumulation in this case, rather than

uptake of MOG itself. We can therefore infer that the increased stability of IPOG compared

to MOG drives the increased NOG concentrations observed in MCT2-expressing tumours.

In addition to LC-MS analysis of NOG levels, we extracted metabolites from the paired,

treated tumours and analysed them using gas chromatography mass spectrometry (GC-MS)

metabolomics. Within the time-frame tested, there were minimal metabolic effects seen within

the tumours after dosing with either compound, compared to the metabolic effects of MOG in

cells[18]. Although there was a relative decrease in αKG and increase in αGP in MCT2-

expressing cells upon IPOG dosing (which was not observed with MOG or NOG treatment)

these changes were not statistically significant (Fig. 5d, Supplementary Fig. 3d).

Together, these data demonstrate that as a more stable pro-drug, IPOG enabled

increased accumulation of NOG in MCT2-expressing tissues and could be used as a scaffold

for further chemical iterations to develop additional prodrugs with improved utility as *in vivo*

tool compounds.

DISCUSSION

Metabolism has far reaching effects on cellular physiology which extend beyond

biomass accumulation, energy production and redox balance. A prototypical metabolite that

illustrates this concept is α-KG; a central metabolic node which is not only the entry point of

glutamine-derived carbons into the TCA cycle, but also a metabolite with important regulatory

roles for key signalling proteins such as mTOR and HIF1α[8,15]. The effects of fluctuations in its

concentration can be long lasting, with αKG also acting as a cofactor for DNA and chromatin

modifying enzymes such as TET hydroxylases[9] and Jumonji demethylases[7], thereby

influencing epigenetic processes within the cell. The αKG mimic NOG has been used to help

understand many of the established roles of this important molecule. DMOG is a membrane-

permeable, esterified analogue of NOG, which is rapidly de-esterified in cell culture media to

the monocarboxylate MOG, a substrate of MCT2. The expression level of MCT2 determines

the $[NOG]_{IC}$, which at high levels, inhibits a number of low affinity metabolic targets such as

GDH and IDH, leading to toxicity in MCT2-expressing cancer cells[18].

Although DMOG continues to be used *in vivo*[49,50], it is dosed at very high

concentrations, significantly beyond those required to inhibit intended targets *in vitro*[11]. Here,

we show that DMOG and MOG are both rapidly converted to NOG in blood, each with a half-

life of less than 5 minutes. We generated a series of analogues to improve the stability of

MOG, allowing us to further increase its potential utility *in vivo*, while also exploring the

structure-activity relationship of MCT2-dependent transport.

Insights into the MCT2 pharmacophore

As well as MOG, MCT2 transports endogenous monocarboxylates ranging from
pyruvate and lactate to larger ketone bodies such as β-hydroxybutyrate, acetoacetate, α-
ketoisovalerate and α-ketoisocaproate[51] (Supplementary Fig. 3) with a higher affinity than the
other SLC16 family members[9]. Although less well studied, MCT2 plays important physiological
roles including the uptake of astrocyte-secreted lactate into neurons within the brain [20,21].
MCT2 is highly expressed in some human cancers (Supplementary Fig. 4) and has been
proposed as a biomarker for prostate cancer[22], as well as having pro-tumourigenic [23] and pro-
metastatic roles[24] in breast cancer. A more detailed mechanistic understanding of this
transporter could therefore open up new therapeutic opportunities, and provide the basis for
further studies to generate *in vivo* imaging tools. By generating a series of MOG analogues,
we therefore also aimed to further explore the structure-activity relationship (SAR) between
MCT2 and its ligands, beyond what has been established based on endogenous substrates[51].

Interestingly, we observed very little tolerance for the α-methyl substitutions (**4**-**6**), all
of which failed to meet our 2-fold threshold for MCT2-dependent uptake. These three
analogues bare some structural similarity to α-ketoisocaproate (Fig. 1d, Supplementary Fig.
3), for which MCT2 has a K_m of 100 μM[51]. Their lack of transport therefore suggests that the
combination of an α-methyl substitution with a carboxyl-ester group cannot be accommodated
(Fig. 2c, d). Unexpectedly, we also observed no MCT2-dependent transport of the ketone
analogue (**7**), potentially indicating that the oxygen in the MOG ester participates in an
interaction with the substrate recognition pocket of MCT2 that is required for substrate binding
and transport.

MCT2-dependent transport was maintained in the bulkier alkyl esters. We found that
replacement of the methyl-ester leaving group with an ethyl-ester (**2**) was well tolerated by
MCT2, with an almost eight-fold increase in uptake by MCT2-expressing cells compared to
the control line. While transport was maintained with an isopropyl-ester substitution (**3**, IPOG),
it occurred at a decreased level (a 2-fold increase in MCT2-expressing cells, Fig. 2d) indicating
that the increased size led to some steric hindrance within the transporter.

The 5-membered aromatic heterocycles (**8**-**10**) were stable in blood (Supplementary
Fig. 2a) and were also all transported in an MCT2-dependent manner with between a 2- and
4-fold enrichment in MCT2-expressing cells. Given the increasing interest in the role of MCT2
in cancer[52–54], this finding provides a useful set of scaffolds that could be built upon for the
design of PET ligands to image MCT2 activity *in vivo*.

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Transporter-dictated effective concentration of compound determines target engagement

Even in the case of very selective drugs, intracellular concentrations significantly

beyond those required to engage the intended target could lead to off-target effects and

toxicity. Here, we were able to demonstrate that transporter-mediated uptake heavily

influences compound efficacy and toxicity by determining intracellular concentration. Although

both the bulkier alkyl esters (**2**,**3**) are converted to NOG intracellularly, the $[NOG]_{IC}$ achieved

in MCT2-expressing cells varied widely (48.1, 16.4 and 4.65 mM for MOG, **2** and **3**

respectively), despite each analogue being dosed at the same concentration. The $[NOG]_{IC}$

achieved by each compound determined their effects within cells, reflected by their relative

impact on cell growth and apoptosis (Fig. 3a, b). PHD engagement by **2** and **3** was lower than

that by MOG based on the observed stabilisation of HIF1α (Fig. 3b). Similarly, compounds **2**

and **3** only partially inhibited reductive carboxylation, but they did not significantly suppress

the oxidative production of citrate (Fig. 3e) via inhibition of GDH[18]; as such only minimally

depleted TCA intermediates relative to that seen when dosing with MOG (Fig. 3c). Together,

our observations suggest that the decrease in proliferation that is observed with **2** and **3** is

modest because these compounds result in a $[NOG]_{IC}$ only sufficient to engage a subset of

the NOG targets that are collectively required for the maximal effects seen with MOG.

Trade-off between blood stability and MCT2-dependent uptake

The uptake of compound **2** into cells was higher than IPOG, however the ethyl-ester

substitution failed to improve its stability in blood relative to MOG. IPOG, on the other hand,

had a blood half-life more than four-fold higher than MOG. As a proof of principle, we were

able to show that its increased metabolic stability was sufficient to promote a significant

increase in NOG accumulation in MCT2-expressing tumours compared to control tumours

within the same animal (Fig. 5c). Therefore, while the MCT2-mediated uptake of IPOG

analogues would ideally be improved with further design iterations, IPOG, exemplifies a fine

balance between compound metabolic stability and MCT2 transporter-mediated uptake

towards future *in vivo* applications.

Building new tools to probe αKG-dependent processes

Given the extensive roles of αKG in cells, it is widely appreciated that NOG, as an αKG

mimic, is a promiscuous compound. Significant efforts have been made to generate tool

compounds and potential clinical leads to inhibit a number of its targets more selectively, in

particular the prolyl hydroxylases[11]. For other groups of αKG-dependent dioxygenases, there

are far fewer specific chemical inhibitors available. The TET enzymes are of particular interest,

given their well-established roles in regulating DNA methylation during early embryonic

development. More recently, it has become clear that they also enable cellular metabolic state

to impact upon epigenetic regulation[55,56] which can influence differentiation in cancer[16].

Isoform-specific TET inhibitors have yet to be developed, and so 'bump and hole'-based

approaches[57] have been employed to allow individual isoform targeting via engineered

enzyme isoforms with expanded active sites which can accommodate bulkier NOG

analogues[58]. The studies presented here could aid in the creation of cell-permeable

derivatives of these NOG analogues, enabling the study of specific TET enzymes both in cells

and potentially also *in vivo.*

Finally, although our analogue series were designed to mimic MOG to enable the study

of its MCT2-specific interactions *in vivo*, our findings could also be of use for the many labs

using DMOG as an *in vivo* tool more generally. Further development of IPOG with analogues

that also feature bulkier oxoacetate carboxyl ester groups will likely enhance blood stability

while maintaining general membrane permeability, improving the pharmacokinetic profile of

this tool compound.

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AUTHOR CONTRIBUTIONS

S.C. and D.H. designed (with input from L.F. and D.A.) and synthesised MOG analogues, and

advised on mouse dosing experimental design; P.N. performed orthotopic cell injections and

assisted with compound dosing in mice; N.B. assisted with cell line work; E.S. assisted with

and advised on compound stability measurements; M.S.d.S. and J.I.M. assisted with and

advised on metabolomics experiments; L.F. designed and performed all other experiments,

analysed and interpreted data. D.A. supervised the study, designed experiments and

interpreted data. L.F. wrote the first draft of the manuscript and developed it with support from

D.A. and input from S.C. and D.H. All authors reviewed and commented on the manuscript.

MATERIALS AND METHODS

Chemical Synthesis

Please see Supplementary Methods.

Cell lines and cell culture

HCC1569 cells were obtained from the American Type Culture Collection. Cells were cultured

in RPMI 1640 medium (Gibco, 31840) containing 10% fetal calf serum (FCS), 2 mM glutamine

and 100 U/ml penicillin/streptomycin, and were incubated in a humidified incubator at 37 °C

and 5% CO_2. All cell lines tested mycoplasma-free, and identity was confirmed by short-

tandem-repeat profiling (Francis Crick Institute Cell Services Science Technology Platform).

Generation of 'HCC1569-MCT2' over-expressing cells was achieved using retroviral

transduction of HCC1569 cells with a pBabe-puro vector containing the *SLC16A7* cDNA

sequence, as described previously[18]. 'HCC1569-EV' cells transduced with an empty pBabe-

puro vector were used as controls.

Western blotting

Cells on cell culture dishes were washed twice with PBS, before scraping in SDS sample

buffer (without beta-mercaptoethanol or bromophenol blue) and boiled for 5 min at 95 °C.

Protein was quantified using a BCA assay before adding beta-mercaptoethanol and

bromophenol blue and resolving by SDS–PAGE. Proteins were transferred to nitrocellulose

membranes by electroblotting, before blocking with 5% milk in Tris-buffered saline (50 mM

Tris-HCl, pH 7.5, and 150 mM NaCl) containing 0.05% Tween 20 (TBS-T). Membranes were

then incubated with the primary antibody overnight at 4 °C, washed with TBS-T and incubated

with horseradish peroxidase-conjugated secondary antibody for 1 h at RT in 5% milk TBS-T.

Antibodies were visualized by chemiluminescence and imaged using an Amersham

Imagequant 600 RGB.

Primary antibodies: HIF1α 1:500 (BD Biosciences 610958), MCT2 1:500 (generated by

Anastasiou lab). The secondary antibodies were as follows: goat anti–rabbit IgG conjugated

to HRP.

Cell proliferation and apoptosis measurements

Cell proliferation and apoptosis were measured in real time using an IncuCyteZoom (Essen

Bioscience). HCC1569 EV or MCT2 cells were seeded in 96-well plates at 9,000 cells per

well, in the presence of Incucyte Caspase 3/7 Green Apoptosis Assay Reagent (Essen

Bioscience). MOG analogues were added at doses indicated 16–20 h after seeding. Cells

were imaged (phase and fluorescence) at 3 h intervals, and automated image analysis was

used to determine confluence and number of apoptotic cells.

Stable Isotope labelling and metabolite extraction for metabolomics

Cells were seeded 1 day prior to the experiment in 6-cm dishes in RPMI medium (as described

above), containing dialysed FCS (3,500-Da MWCO). Medium was replaced with fresh at t = –

1 h. At t = 0, medium was replaced again to medium containing [U-^{13}C]-glutamine (2 mM) and

the MOG analogue of interest (1 mM) or 0.1% DMSO (vehicle control). Treatment/Labeling

was carried out for 4 h unless otherwise stated. Four or five technical-replicate plates were

used per condition and two or three additional plates of each cell line were counted to use for

normalisation of metabolite measurements. Cell diameter was also recorded for calculation of

cell volumes in order to determine intracellular concentrations. Cell diameter and number were

measured using a Nexcelcom Bioscience Cellometer Auto T4. At the end of the experiment,

plates were washed twice with ice-cold PBS, before quenching cells with the addition of 725 µl

dry-ice-cold methanol. Each plate was then scraped on ice, and samples were transferred to

a microcentrifuge tube containing 160 µl CHCl$_3$ and 180 µl H$_2$O (containing 2 nmol of scyllo-

inositol as an internal standard). Plates were scraped once more with an additional 725 µl of

cold MeOH, which was then added to the microcentrifuge tube containing the rest of the

sample. Samples were sonicated for 3 × 8 min in a water bath, and metabolites were extracted

overnight at 4 °C. Precipitated material was removed by centrifugation and samples were

subsequently dried and resuspended in 3:3:1 (vol/vol/vol) MeOH/H$_2$O/CHCl$_3$ (350 µl total), to

separate polar and apolar metabolites into an upper aqueous phase and lower organic phase,

respectively.

Analogue uptake assays

Cells were incubated with MOG or MOG analogues (1 mM) for 4 hours. Cells were then

washed with ice-cold PBS, and extracted as described for GC–MS above. Samples of the

polar phases were diluted 50-fold in 1:1 (vol/vol) MeOH/H$_2$O (containing 5 µM [U-^{13}C,^{15}N]-

valine as an internal standard) and analysed by LC–MS as described below.

Gas Chromatography-Mass Spectrometry

For GC–MS analysis, 150 µl of the aqueous phase was dried down in a vial insert, before

washing twice with 40 µl MeOH and drying again. Samples were methoximated (20 µl of

20 mg/ml methoxyamine in pyridine, RT overnight) before derivatising with 20 µl of N,O-

bis(trimetylsilyl)trifluoroacetamide + 1% trimethylchlorosilane (Sigma, 33148) for ≥1 h. An

Agilent 7890B-5977A GC-MS system was use to perform metabolite analysis. Splitless

injection (injection temperature 270 °C) onto a 30 m + 10 m × 0.25 mm DB-5MS + DG column

(Agilent J&W) was used, using helium carrier gas, in electron-impact ionization (EI) mode.

Initial oven temperature was 70 °C (2 min) with a subsequent increase to 295 °C at 12.5

°C/min, then to 320 °C at 25 °C/min (before holding for 3 min). Metabolite identification and

quantification was performed using MassHunter Workstation software (B.06.00 SP01, Agilent

Technologies) by comparison to the retention times, mass spectra and responses of known

amounts of authentic standards. Fractional labelling of individual metabolites is reported after

correction for natural abundance.

Liquid Chromatography-Mass Spectrometry

The LC–MS method was adapted from ref[59]. Samples were injected into a Dionex UltiMate

LC system (Thermo Scientific) using a ZIC-pHILIC (150 mm × 4.6 mm, 5-µm particle) column

(Merck Sequant). A 15-min elution gradient was used (80% solvent A to 20% solvent B),

followed by a 5-min wash (95:5 solvent A to solvent B) and 5-min re-equilibration; solvent A

was 20 mM ammonium carbonate in water (Optima HPLC grade, Sigma Aldrich) and solvent

B was acetonitrile (Optima HPLC grade, Sigma Aldrich). Flow rate, 300 µl/min; column

temperature, 25 °C; injection volume, 10 µl; and autosampler temperature, 4 °C. MS was

performed with positive/negative polarity switching using a Q Exactive Orbitrap (Thermo

Scientific) with a HESI II (heated electrospray ionization) probe. MS parameters: spray

voltage, 3.5 kV and 3.2 kV for positive and negative modes, respectively; probe temperature,

320 °C; sheath and auxiliary gases, 30 and 5 arbitrary units, respectively; full scan range, 70

to 1,050 m/z with settings of AGC target and resolution as 'balanced' and 'high' (3 × 106 and

70,000), respectively. Xcalibur 3.0.63 software (Thermo Scientific) was used to record data.

Prior to analysis, mass calibration was performed for both ESI polarities using the standard

Thermo Scientific Calmix solution. Calibration stability was enhanced by application of lock-

mass correction to each analytical run using ubiquitous low-mass contaminants. Parallel

reaction monitoring acquisition parameters: resolution, 17,500; auto gain control target,

2 × 105; maximum isolation time, 100 ms; isolation window, m/z 0.4; and collision energies,

set individually in high-energy collisional-dissociation mode. Equal volumes of each sample

were pooled to provide quality-control samples and were analysed throughout the run, thereby

providing a measurement of the stability and performance of the system. Xcalibur Qual

Browser and Tracefinder 4.1 software (Thermo Scientific) were used to perform qualitative

and quantitative analysis respectively, according to the manufacturer's workflows.

Blood stability assay

Blood was collected from euthanised NSG female mice into heparinised tubes, and used

immediately for experiments. To test stability, compounds were incubated at a final

concentration of 100 µM in blood (600 µL total volume), pre-warmed to 37 °C Samples were

collected in triplicate at 0 (compound added in to sample after extraction), 5, 15, 30 and 60
minutes. At the indicated time, 30 μl was taken and added to a tube on ice containing 100 μl
chloroform, 300 μl MeOH and 270 μl H_2O. [$U^{13}C$-^{15}N] valine was present in the aqueous phase
at a final concentration of 5 μM. Immediately after collection, samples were vortexed and
placed on ice. After all samples had been collected, they were sonicated for 3 x 8 mins and
incubated for 1h at 4 °C to allow extraction to proceed. Samples were then centrifuged (10
min, 4°C, full speed) and the aqueous phase transferred to a new tube, and stored at -80 until
they were ready to run on the LC-MS (Q Exactive) system, as described above. Samples were
quantitated against a 7-point standard curve of compound in mouse blood extract to minimise
matrix effects.
516
Short term analogue dosing in vivo
Adult female NSG mice were dosed by intra-peritoneal injection with MOG or MOG analogues
at 100 mg/kg. Blood samples were collected from the saphenous vein onto a Mitra micro-
sampling device (10 μl capacity, 100601-A, neoteryx) prior to injection (t=0) and at 15 minutes,
1h and 2h post injection. After two hours, dosed animals were euthanised by cervical
dislocation and tissues were harvested immediately and snap-frozen in liquid nitrogen before
storing at -80 °C. For blood samples, the micro-sampler 'sponge' tip was transferred into a
microfuge tube containing 33.3 μl CHCl3, 100 μl MeOH and vortexed thoroughly. [$U^{13}C$-^{15}N]-
valine was included to a final concentration of 5 μM in aqueous phase. Samples were
sonicated 3 x 8 mins at 4 °C in sonicating water bath and stored at -80 until required. Phases
were split by addition of 90 μl H_2O before spinning at 4 °C full speed for 5 mins (since MOG
analogues are susceptible to aqueous hydrolysis, H_2O was added immediately prior to
running). Aqueous phase was run on LC-MS as described above.
Extraction from tissues is described below.
531
Orthotopic Xenograft assays
6-8 week old female NSG mice were anaesthetised with 1.5-2% isofluorane (with oxygen),
and 1 million cells resuspended in Matrigel were injected into the mammary fat pad,
underneath the nipple. Two injections were carried out into mammary fat pads on opposing
flanks, using HCC1569 cells expressing an empty pBabe-puro vector ('HCC1569-EV') on one
side and pBabe-puro-MCT2 ('HCC1659-MCT2') on the other. Cells were tested for
mycoplasma and murine norovirus prior to injection. Tumours were monitored by palpation
and animal weight was measured twice per week. Length and width of tumours were
measured using callipers and volume calculated as 0.5*(length x width^2). Short-term dosing
experiments were carried out when tumours reached approximately 0.5 cm diameter.
542

All animal studies were ethically reviewed and carried out in accordance with Animals

(Scientific Procedures) Act 1986 and the GSK Policy on the Care, Welfare and Treatment of

Animals.

Metabolite/drug extraction from tissue

Snap frozen tissues were lyophilised and when dry (approx. 24h) they were ground in a pestle

and mortar. Powdered tissue was weighed out, and weight noted (~7 mg), so that after the

final biphasic extraction, samples can be diluted to ensure there are 2 mg/ 100 µl aqueous

phase. 600 µl 2:1 $CHCl_3$/MeOH was added to weighed tissue and vortexed thoroughly.

Samples were then sonicated in a sonicating water bath for 3 x 8 mins with 8 mins rest

between. Samples were spun full speed in microfuge for 5 mins to pellet insoluble material

and supernatant transferred to a new tube and dry down in a vacuum concentrator (with cold

trap). 600 µl 2:1 MeOH/H_2O was added to the remaining pellets and vortexed again.

Sonication (3 x 8 mins) was repeated before leaving to extract for at 4 for at least an hour.

Samples were then spun full speed in a microfuge for 5 mins to pellet insoluble material and

supernatant transferred to the tube containing the previous (apolar) supernatant, before drying

completely in a vacuum concentrator. Dried tubes can be stored at -80. A biphasic extraction

was then carried out on each sample (1:3:3 $CHCl_3$/MeOH/H_2O, 350 µl total) with aqueous

phase containing 5 µM [U-^{13}C-^{15}N]-valine as an internal standard, before vortexing thoroughly

and centrifuging at full speed, 4 °C for 5 mins to separate phases. For LC-MS analysis, the

volume of aqueous phase corresponding to 2 mg of dried tissue was added into a vial insert,

and made up to 100 µl with 1:1 MeOH/H_2O (with aqueous phase containing 5 µM [U-^{13}C-^{15}N]-

valine as an internal standard) so that equivalent tissue amounts are analysed. For GC-MS

analysis, *scyllo*-inositol was used as an internal standard, and samples were dried down after

biphasic before continuing with derivatisation as described for cell samples above.

FIGURE LEGENDS

Figure 1 – Creating MOG analogues to improve blood stability and explore MCT2 pharmacophore

a) Schematic illustrating the conversion of DMOG to MOG, and subsequently NOG, which is the active αKG analogue.

b) LC-MS analysis of DMOG stability in whole mouse blood over time. DMOG is very rapidly converted to MOG, which is also unstable and subsequently forms NOG.

c) LC-MS analysis of synthetic MOG stability in whole mouse blood, table showing half-lives of DMOG conversion to MOG and subsequently NOG.

d) MOG methyl ester replacement analogues. i) bulkier alkyl esters, ii) α-methyl substituents, iii) ketone analogue, iv) 5-membered aromatic heterocycles.

e) Synthetic route for the preparation of MOG analogues 2-10.

Figure 2 – MCT2-dependent uptake is maintained by alkyl ester and aromatic heterocycle analogues

a) Schematic to illustrate the cell system used to measure relative uptake of MOG analogues: HCC1569 breast cancer cells transduced with either an empty pBabe vector control (EV) or pBabe-MCT2.

b) Western blot demonstrating expression level of MCT2 in HCC1569-EV or HCC1569-MCT2 cells.

c) Concentration of each of the analogues in EV or MCT2 expressing HCC1569 cells after 4 hours of incubation with each analogue at 1 mM (mean ± SD).

d) Fold-change of analogue concentration in MCT2 expressing cells relative to EV cells. A 2-fold increase (dashed line) was considered the threshold for MCT2-dependent uptake (mean ± SD).

Figure 3 – Analogues replicate MOG effects in cells in a NOG-dose-dependent manner

a) Growth curves illustrating the dose-dependent effects of MOG analogues on cell proliferation and apoptosis at different compound concentrations in either EV or MCT2-expressing HCC1569 cells (mean ± SD).

b) Linear relationship between the inhibition of cell proliferation after treatment with 1 mM of the indicated analogue (defined as the % increase in confluence at 96 h after treatment, relative to untreated control cells, data from panel A) and the intracellular NOG accumulated by that analogue. Error bars represent ± standard deviation.

c) Western blot showing the stabilisation of HIF1α after 4 hours of incubation with MOG
analogues in either EV or MCT2-expressing HCC1569 cells.
d) Schematic to demonstrate different routes of citrate synthesis and subsequent
labelling patterns from [U-^{13}C]-Glutamine.
e) Metabolic changes in EV or MCT2-expressing HCC1569 cells after 4 hours of
incubation with MOG analogues at 1 mM. Significance tested by multiple t-tests with
Holm-Sidak correction for multiple comparisons. (* = p< 0.05, ** = p< 0.01, *** = p<
0.001).
f) Labelling from [U-^{13}C]-Glutamine in EV or MCT2-expressing cells after 4 h of
incubation with analogues at 1 mM. M+4 citrate is derived from oxidative use of
glutamine carbons through the TCA cycle, whereas m+5 is formed as a result of
reductive carboxylation. Significance tested by multiple t-tests with Holm-Sidak
correction for multiple comparisons. (* = p< 0.05, ** = p< 0.01, *** = p< 0.001).
616

617 **Figure 4 – Increasing size of the alkyl ester substituent improves pharmacokinetics**
618 a) LC-MS analysis of MOG analogue stability in whole mouse blood over time (and
619   accumulation of NOG). Data points represent mean ± SD.
620 b) Concentration of MOG or compound **3** (IPOG) in blood at different time points after IP
621   injection at 100 mg/kg, and mean area under the curve (±SEM).
622

Figure 5 – Compound 3 (IPOG) has improved MCT2-dependent uptake *in vivo*
a) Schematic illustrating orthotopic xenograft approach, in which HCC1569 cells
expressing either EV or MCT2 were injected into mammary fat pads on either flank of
the mouse to allow for paired analysis of uptake.
b) Growth rates of HCC1569 orthotopic xenograft tumours expressing either EV or MCT2
(no significant difference in growth, as determined by multiple t-tests at different time
points, with holm-sidak correction for multiple comparisons).
c) NOG accumulation in EV and MCT2 tumours 2 hours after 100 mg/kg IP injection with
either MOG or IPOG. Significance tested by paired t-tests. (* = p< 0.05)
d) Levels of α-ketoglutarate and α-glycerophosphate in EV and MCT2 tumours 2 hours
after 100 mg/kg IP injection with either MOG or IPOG. Significance tested by paired t-
tests.

a

b **c**

Parent compound	Species	Half Life (min)	R^2
DMOG	DMOG	0.61	0.99
	MOG (accumulation)	0.54	0.82
	MOG (degradation)	4.01	0.97
	NOG	3.49	0.95
MOG	MOG	3.54	0.98
	NOG	3.46	0.97

d

e

Fets *et al*; Figure 1

Fets *et al*; Figure 2

Fets *et al*; Figure 3

Fets *et al*; Figure 4

Fets *et al*; Figure 5

SUPPLEMENTARY FIGURE LEGENDS

Supplementary Figure 1

a) Levels of αGP in MCT2-expressing HCC1569 cells after 4 hours of incubation with
MOG analogues at 1 mM. Significance tested by ANOVA with Dunnet's test for multiple
comparisons, relative to DMSO control sample. (* = p< 0.05, ** = p< 0.01, *** = p<
0.001, **** = p< 0.0001).

Supplementary Figure 2

a) LC-MS analysis of MOG analogue stability in whole mouse blood over time (mean ±
SD).
b) Small scale analysis of NOG tissue distribution after IP injection of MOG or compound
3 (2h incubation, 100 mg/kg dose).
c) NOG accumulation in EV and MCT2 tumours 2 hours after 100 mg/kg IP injection with
NOG. Significance tested by paired t-tests. (* = p< 0.05)
d) Levels of α-ketoglutarate and α-glycerophosphate in EV and MCT2 tumours 2 hours
after 100 mg/kg IP injection with NOG. Significance tested by paired t-tests.

Supplementary Figure 3

Chemical structures of endogenous MCT2 substrates

Supplementary Figure 4

Ridge plots showing the range of expression levels of *SLC16A1* (MCT1), *SLC16A7*
(MCT2) and *SLC16A3* (MCT4) in different human cancer types. Data source: The Cancer
Genome Atlas, SKCM – Skin cutaneous melanoma; PAAD – Pancreatic Adenocarcinoma;
LUSC – Lung squamous cell carcinoma; LUAD – Lung adenocarcinoma; LGG – Low grade
glioma; GBM – Glioblastoma; ESCA – Oesophageal carcinoma; BRCA – Breast Invasive
Carcinoma.

a

Fets *et al*; Suppl. Fig 1

Fets *et al*; Suppl. Fig 2

Lactate

Pyruvate

Acetoacetate

β-Hydroxybutyrate

α-Ketoisovalerate

α-Ketoisocaproate

Fets *et al*; Suppl. Fig 3

Fets *et al*; Suppl. Fig 4

674   **Supplementary Table 1.**

675   Mass spectrometry parameters for detection of MOG and MOG analogues

Compound Name/Number	m/z (negative mode)	Fragments	HCD
DMOG	174.0408	-	
NOG	146.0095	74.0245	35
1 (MOG)	160.0254	84.009, 59.013, 72.993, 100.004	35
2	174.0405	74.024, 84.009, 100.004	30
3 (IPOG)	188.0567	74.024, 84.009, 100.004, 85.029	30
4	174.0405	98.025, 114.020, 72.993, 59.013	30
5	174.0405	98.025, 114.020, 72.993, 59.013	30
6	188.0559	102.0559, 128.0354	30
7	144.0302	99.926	35
8	169.0255	68.014, 82.029	30
9	236.0171	135.0058,114.9996, 164.0323	30
10	184.0358	89.0429, 68.9955	30

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