Integration of plant carbohydrate dynamics by Fourier polynomials

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Highlight

A balance equation model is developed to quantify effects of transient heat exposure on plant carbon assimilation. The model is based on Fourier polynomials for quantitative assessment of system dynamics.

Abstract

Quantification of system dynamics is a central aim of mathematical modelling in biology. Defining experimentally supported functional relationships between molecular entities by mathematical terms enables the application of computational routines to simulate and analyse the underlying molecular system. In many fields of natural sciences and engineering, trigonometric functions are applied to describe oscillatory processes. As biochemical oscillations occur in many aspects of biochemistry and biophysics, Fourier analysis of metabolic functions promises to quantify, describe and analyse metabolism and its reaction towards environmental fluctuations. Here, Fourier polynomials were developed from experimental time-series data and combined with block diagram simulation of plant metabolism to study heat shock response of photosynthetic CO_2 assimilation and carbohydrate metabolism. Findings suggest that increased capacities of starch biosynthesis stabilize photosynthetic CO_2 assimilation under transient heat exposure. Among soluble sugars, fructose concentrations were observed to fluctuate least under heat exposure which might be the consequence of high respiration rates under elevated temperature. Finally, Col-0 and two mutants of *Arabidopsis thaliana* with deficiencies in starch and sucrose metabolism were discriminated by fundamental frequencies of Fourier polynomials across different experiments. This suggests balance modelling based on Fourier polynomials as a suitable approach for mathematical analysis of dynamic plant-environment interactions.

Key words

metabolism, dynamics, Fourier series, block diagram, simulation, photosynthesis, carbohydrate metabolism, heat stress response

Introduction

Capturing dynamics in biological systems by mathematical terms is the general aim of biomathematical modelling. Differential equations represent an adequate strategy to describe dynamics over space and time. Ordinary differential equations (ODEs) and partial differential equations (PDEs) have been successfully applied to reveal biological system dynamics and to develop predictive models of growth rates, transcription, translation or metabolic processes (Lopatkin and Collins, 2020). In a metabolic context, ODE models are frequently applied to simulate enzyme kinetic reactions and, by this, to explain dynamics of observed metabolite concentrations. Kinetic models, based on ODEs, have frequently been applied a broad field of biological research, e.g. in context of metabolic engineering of microbial systems and strain design (Costa *et al.*, 2016), disease research (Ramos *et al.*, 2019), and plant metabolism (Feldman-Salit *et al.*, 2019; Rohwer, 2012; Weiszmann *et al.*, 2018). In contrast to the high diversity of application fields, the principle of ODE kinetic modelling remains conserved: based on genome sequence information or biochemical evidence from literature a biochemical reaction network is established, substrate and product concentrations are quantified, and enzyme kinetics are applied to calculate reaction rates within a metabolic system. Enzyme kinetic parameters, e.g., velocity under substrate saturation (V_{max}) or substrate affinity (K_M) of Michaelis-Menten equations are experimentally determined and used for computationally assisted parameter estimation. A well-defined and experimentally validated kinetic ODE model enables computational simulation and prediction of complex system behaviour. A clear limitation of such an approach, however, is the requirement of kinetic parameters which are (frequently) difficult and/or expensive to quantify. Initiatives like KiMoSys, a public repository of published experimental data, summarise and concentrate data on metabolites, protein abundance and fluxes providing a solid database for model construction and initial development (Mochão *et al.*, 2020). Yet, under highly dynamic conditions, e.g., in a fluctuating environment, it still remains a challenge to resolve system dynamics on an enzyme kinetic level. This is due to the high dynamics of metabolites, transcripts, protein levels and enzyme activities (Espinoza *et al.*, 2010; Nägele and Heyer, 2013).

Although being laborious, development and optimization of ODE kinetic models provides an important and informative mathematical method to study biochemical

system behaviour. Simultaneously, however, diverse problems might occur with solving and applying such models due to uncertainties about parameters, model structure, kinetic rate laws or parameter sensitivities (Gutenkunst *et al.*, 2007; Schaber *et al.*, 2009). Depending on the research question focused by a study, an explicit knowledge about enzymatic activities and their dynamics might not be essential to derive a mathematical description of metabolite dynamics. For example, metabolic fluxes might be estimated by tracing labelled atoms or molecules in a metabolic pathway system (Crown *et al.*, 2016). While it is not possible to derive information about single enzyme activities or kinetics from flux estimations, they still provide comprehensive insights into metabolic states and pathway activities, also on a large scale (Basler *et al.*, 2018). Beyond, algorithms and user interfaces have been developed which enable the combination of flux data with relative metabolite levels (Sajitz-Hermstein *et al.*, 2016).

For estimating metabolic functions, i.e., the sum of synthesising and degrading/consuming reactions of metabolite pools, under dynamic environmental conditions we have previously suggested a method for implicit estimation of metabolic functions (Nägele *et al.*, 2016). Similar to flux analysis, dynamics of metabolite concentrations in time-series experiments were used in this approach to derive a time-continuous mathematical function to identify regulatory cascades in metabolic pathways. This approach made use of spline interpolations which were composed by cubic polynomials which were fitted to adjacent pairs of data points in a time-series data set. While such an approach is suitable for accurate data fitting, underlying mathematical functions are mostly not related to biological function and, thus, are less predictive than enzyme kinetic models. In the present study, we applied Fourier analysis to develop a mathematical model based on Fourier polynomials to simulate and analyse dynamics of photosynthesis and carbohydrate metabolism under transient heat exposure. Our approach suggests that diurnal and stress-induced dynamics of plant carbohydrate fixation are reflected by harmonic oscillations which provides a mathematical toolbox to study consequences of a changing environment on plant metabolism.

Materials and Methods

Plant cultivation and stress treatment

Plants of *Arabidopsis thaliana*, accession Columbia-0 (Col-0), *spsa1*(AT5G20280, SALK line 148643C) and *pgm1* (AT5G51820; TAIR stock CS3092) were grown on a 1:1 mixture of GS90 soil and vermiculite in a climate chamber under short-day conditions (8 h/16 h light/dark; 100 µmol m^{-2} s^{-1}; 22 °C/18 °C; 60% relative humidity). The *spsa1* line was confirmed via PCR to be homozygous (data not shown) and activity was found to be decreased to 30-50% of the wildtype Col-0 (Fig. S1). The *pgm1* mutant had a dwarf phenotype and starch content was below the detection limit. After 4 weeks, plants were transferred to a growth cabinet (Conviron®, www.conviron.com) and grown for two further weeks under short day conditions with same settings as in the climate chamber. After 6 weeks, at the day of sampling, temperature in the growth cabinet was kept at 22°C during the first two hours in the light (0h→2h, 22°C). Then, in three independent experiments, temperature was increased to (i) 32°C, (ii) 36°C or (iii) 40°C for a total of 4 hours (2h → 6h, temperature increase). In the control experiment, temperature was set constantly to 22°C. Between 6h and 8h, i.e., until the end of the light period, temperature was set to 22°C in all experiments. A graphical representation of the experimental setup is provided in Fig. S2. Plants were sampled at each timepoint (0h, 2h, 6h, 8h) by cutting the full leaf rosette at the hypocotyl. Samples were immediately frozen in liquid nitrogen and stored at -80°C until further use.

Quantification of net CO_2 uptake

Rates of net photosynthesis were recorded within the Conviron® growth cabinet using a WALZ® GFS-3000FL system equipped with measurement head 3010-S (Heinz Walz GmbH, Effeltrich, Germany, https://www.walz.com/). Temperature, light and humidity control of the measurement head was set to follow ambient conditions, i.e., to follow surrounding growth cabinet conditions. A summary of recorded temperature, light and humidity curves is provided in the supplement (Fig. S3). For each genotype and growth condition, i.e., temperature setup, three independent samples were measured.

 Extraction and quantification of carbohydrates

Plant material was ground to a fine powder under constant freezing with liquid nitrogen. The powder was lyophilized for three days and subsequently used for carbohydrate analytics. Starch and soluble carbohydrates were extracted and photometrically determined as described before (Kitashova $et\ al.$, 2021). Plant powder was incubated with 80% ethanol at 80°C for 30min. After centrifugation, the supernatant was transferred to a new tube and extraction was repeated with the pellet. Supernatants were unified and dried in a desiccator. The starch-containing pellet was hydrolyzed with 0.5 M NaOH for 45min at 95°C. After acclimation to room temperature, 1 M CH_3COOH was added and the suspension was digested with amyloglucosidase solution, finally releasing glucose moieties from starch granules. Glucose was photometrically determined applying a coupled glucose oxidase/peroxidase/o-dianisidine assay.

Soluble sugars sucrose, glucose and fructose were determined from dried ethanol extracts after dissolving in water. After incubation with 30% KOH at 95 °C, sucrose was quantified using an anthrone assay. Anthrone was dissolved in 14.6 M H_2SO_4 (0.14% w/v), incubated with the prepared sample for 30 min at 40 °C and absorbance was determined photometrically at 620 nm. Glucose amount was determined photometrically by a coupled hexokinase/glucose 6-phosphate dehydrogenase assay resulting in $NADPH + H^+$ at 340 nm. For fructose quantification, phosphoglucoisomerase was added to the reaction mixture after glucose determination.

Quantification of SPS activity

Activity of sucrose phosphate synthase (SPS) was determined using the anthrone assay as described previously (Kitashova $et\ al.$, 2021). In brief, freeze-dried leaf tissue was suspended in extraction buffer containing 50 mM HEPES–KOH (pH 7.5), 10 mM MgCl2, 1 mM EDTA, 2.5 mM DTT, 10% (v/v) glycerol and 0.1% (v/v) Triton-X-100. Following incubation on ice, extracts were incubated for 30 min at 25 °C with a reaction buffer containing 50 mM HEPES–KOH (pH 7.5), 15 mM MgCl2, 2.5 mM DTT, 35 mM UDP-glucose, 35 mM F6P and 140 mM G6P. Reactions were stopped by adding 30%KOH and heating to 95 °C. Sucrose was determined photometrically after incubation with anthrone in H_2SO_4.

 Statistics, data fitting and modelling

195 Statistical analysis was performed in R and R Studio (www.r-project.org) (R Core
196 Team, 2019). Fourier series fitting was done within MATLAB®
197 (www.themathworks.com), and block diagram models were created in Simulink®
198 (www.themathworks.com).
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Results

A block diagram model based on Fourier polynomials for carbon balancing of plant
metabolism

A block diagram model of the central carbohydrate metabolism of plants was
developed to integrate experimental data on net photosynthesis (NPS), starch and
sugar metabolism (Fig. 1). The net carbon input block, i.e., NPS block, represented a
Fourier polynomial describing NPS dynamics depending on genotypes and
environments (see Fig. 2, solid lines). This input flux was multiplied by the
stoichiometric factor 1/6 to enable quantitative summation with starch and sugar
fluxes (unit: μmol C6 h^{-1} gDW^{-1}). Carbon balance equation 1 (BE$_1$) comprised the
summation of NPS rates and negative starch rates, balance equation 2 (BE$_2$)
additionally comprised summation of negative sugar rates (Fig. 1). As a result, BE$_1$
revealed the net carbon flux (in C6 equivalents per hour and gram dry weight) which
was left from photosynthetically assimilated CO_2 after starch synthesis, e.g., for
sugar biosynthesis or biomass production. Further, BE$_2$ revealed residual net carbon
flux after additional sugar biosynthesis.

Rates of net starch and sugar biosynthesis were determined by differentiating
Fourier polynomials of starch and sugar dynamics with respect to time. Hence, the
Fourier polynomial balance models comprised three input functions, FP$_{input}$ (Eqs. 1-
3), and two balance equations, BE$_{1,2}$ (Eqs. 4,5).

$$FP_{input,NPS} = a_{0,NPS} + \sum_{k=1}^{n} \left[a_{k,NPS} \cos(k\omega_{NPS}t) + b_{k,NPS} \sin(k\omega_{NPS}t) \right] \qquad (Eq.\,1)$$

$$FP_{input,Starch} = a_{0,Starch}$$

$$+ \sum_{k=1}^{n} \left[a_{k,Starch} \cos(k\omega_{Starch}t) + b_{k,Starch} \sin(k\omega_{Starch}t) \right] \qquad (Eq.\,2)$$

$$FP_{input,Sugars} = a_{0,Sugars}$$

$$+ \sum_{k=1}^{n} \left[a_{k,Sugars} \cos(k\omega_{Sugars}t) + b_{k,Sugars} \sin(k\omega_{Sugars}t) \right] \qquad (Eq.\,3)$$

$$BE_1 = \left(\frac{1}{6}\right) \cdot FP_{input,NPS} - \frac{d(FP_{input,Starch})}{dt} \qquad (Eq.\,4)$$

$$BE_2 = BE_1 - \frac{d(FP_{input,Sugars})}{dt} \qquad (Eq.\,5)$$

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Here, a_k and b_k represent the Fourier coefficients for NPS, starch and sugar
equations. ω is the fundamental frequency of the signal ($\omega = 2\pi/T$, where T is the
period). This Fourier polynomial-based balance equation model was applied to
simulate dynamics of carbohydrate metabolism in plants of *Arabidopsis thaliana*,
accession Columbia-0, under transient heat exposure. Additionally, NPS and
carbohydrate dynamics were recorded and simulated in a starch-deficient mutant
pgm1 and a mutant with deficiency in sucrose biosynthesis capacity, *spsa1*.
Coefficients of Fourier polynomials are provided in the supplements (Table S1).

Fourier polynomials reflect dynamics of net CO_2 assimilation rates
During the first 30-45 min of the light period, rates of net CO_2 assimilation increased
steeply in all genotypes and reached a first plateau at ~1250 µmol CO_2 h^{-1} gDW^{-1}
which was stable during the first half of the light period before it slightly increased
until the end of the day at 22°C (Fig. 2, grey coloured lines). No significant difference
was observed between genotypes, yet *spsa1* had slightly higher assimilation rates

compared to Col-0 while rates of *pgm1* were slightly lower (Fig. 2 a, d, g).
Temperature increase from 22°C to 32°C resulted in a drop of assimilation rates
during the first hour of the treatment before the rates stabilized again and reached
similar values than in the control (22°C) experiment (Fig. 2 a, d, g). At 32°C, starch
deficient *pgm1* plants were most susceptible and mean values differed most from
22°C rates (Fig. 2 d). During the last 2 hours of the light period in which temperature
was decreased to 22°C, all genotypes increased assimilation rates to control rates
again. A similar scenario was observed within the 36°C experiment for *pgm1* and
spsa1 while Col-0 had significantly decreased assimilation rates during the last two
hours of temperature treatment compared to the control experiment (Fig. 2 b). In
pgm1, assimilation rates dropped significantly during the first 30 min of the recovery
phase, i.e., between 6h and 6.5h, when temperature was decreased from 36°C to
22°C (Fig. 2 e). Such a significant recovery drop was also observed for both *pgm1*
and *spsa1* mutants in the 40°C experiment, but not for Col-0 which showed again
significantly decreased CO_2 assimilation rates between the last two hours of the
temperature treatment, i.e. between 4h and 6h of the light period (Fig.2 c, f, i).
Experimentally determined mean values of CO_2 assimilation rates and, by this, all
described effects were covered by Fourier polynomials with $R^2 > 0.94$ (exception:
pgm1, 32°C, $R^2=0.8177$).

Transient heat exposure significantly affects dynamics of starch and soluble carbohydrates

The exposure to transient heat lead to a significant change in starch dynamics in
Col-0 and *spsa1* (Fig. 3). Starch amount in *pgm1* was below the detection limit of the
applied photometric detection method (Fig. 3 e-h). Starch concentration dropped
significantly after transient heat exposure (6h) in comparison to control conditions in
Col-0 and *spsa1* ($p<0.001$). This drop did not change significantly among the
different temperatures. In the recovery phase after the heat shock (8h), plants
increased their starch content depending on the temperature they were subjected to.
Plants of both genotypes treated with 32°C increased their starch concentration by
approximately 40% between 6h and 8h (Fig. 3 b, j), whereas plants treated with 36°C
increased it by over 90% (Fig. 3 c, k). Col-0 subjected to the 36°C transient heat
exposure was even able to reach the level of the control plants at the 8h time point.

An effect of transient heat exposure on sucrose levels was only detectable for higher

temperatures, i.e., within 36°C and 40°C experiments (Fig. 4). When subjected to

32°C of transient heat, no significant change in sucrose levels was detected in all

genotypes (Fig. 4 b, f, j). In Col-0, only heat exposure of 40°C resulted in a

significant increase in sucrose concentration at the 6h time point ($p < 0.001$), but no

further change was observed after the recovery phase at 8h (Fig. 4 d). Under all

conditions, *pgm1* accumulated more sucrose over the course of the light phase

compared to Col-0 and *spsa1* (Fig. 4 e-h). Nevertheless, heat treatment with 36°C

and 40°C reduced the amount of sucrose in the *pgm1* plants almost significantly

($p = 0.05$). In the recovery phase, the sucrose concentration in the heat treated *pgm1*

plants returned to a level comparable to control conditions (Fig. 4 e-h). Due to high

variance in the sucrose measurements of *spsa1* plants, there was no significant

difference between Col-0 and *spsa1* in the control plants and the plants exposed to

32°C and 36°C of transient heat (Fig. 4 i-k). Only at 40°C, *spsa1* showed significantly

higher sucrose levels than Col-0 at 40°C before and after the recovery phase (6h

and 8h, $p < 0.001$) or *spsa1* under control conditions (6h: $p < 0.004$, 8h: $p < 0.01$), whilst

exhibiting very low variance (Fig. 4 l).

In *pgm1*, glucose and fructose dynamics differed significantly from Col-0 (Fig. 5).

Starting at equal hexose content at the start of the light period, the difference

between *pgm1* and Col-0 and *spsa1* increased almost tenfold over the course of the

day. Whilst *pgm1* increased hexoses steeply under control conditions, reaching a

plateau at 6h, hexose concentrations in Col-0 and *spsa1* peaked at 2h, decreasing

again afterwards.

Col-0 and *spsa1* showed a significant drop in glucose levels after heat exposure to

32°C and 36°C ($p < 0.001$, Fig. 5 a). Within the 32°C experiment, glucose levels at 8h

after recovery did not differ significantly from control conditions. Within the 36°C and

40°C experiments, however, the recovery phase after heat exposure resulted in a

significant increase in glucose concentration compared to control conditions

($p < 0.001$). Additionally, in Col-0 glucose levels were increasing above the level of

control plants already during heat exposure to 40°C at 6h ($p < 0.001$). Fructose

dynamics in Col-0 were similar to the glucose dynamics in Col-0 (Fig. 5 b). In *spsa1*,

however, fructose dynamics did not change significantly in response to transient heat

exposure. The only differences were observable after the recovery phase at 8h in

36°C and 40°C. Here, a significantly higher fructose content could be measured

compared to 22°C (p<0.001). In *pgm1,* hexose levels decreased significantly after
temperature treatment (6h, p<0.001), with the lowest values being reached at 40°C.
After recovery (8h), hexose levels did not change significantly from the 6h time point
in the 32°C plants. After exposure to 36°C and 40°C, however, hexose levels
increased significantly from 6h to 8h (p<0.001).

Numerical differentiation and integration of carbon balance equations reveals
genotype-dependent system fluctuations due to transient heat exposure

To reveal how dynamics of carbon balance equations, which combine net
photosynthesis, starch (BE_1) and sugar metabolism (BE_2), are affected by transient
heat exposure, derivatives were built with respect to time (Fig. 6). In Col-0 and
spsa1, increasing temperature resulted in decreased fluctuation of BE_1 and BE_2
derivatives (Fig. 6 a, b, e, f). Particularly in 36°C and 40°C experiments, oscillations
of derivatives were significantly damped during the second half of the temperature
stress period, i.e. between 4h and 6h of the light period. In Col-0, this dampening
effect was strongest at 40°C, in *spsa1* at 36°C. In both genotypes, the 32°C
experiment had only little effect on derivatives, i.e., on absolute changing rates of
Fourier polynomials (yellow lines in Fig. 6 a, b, e, f). In *pgm1*, fluctuation of
derivatives increased for both BE_1 and BE_2 under 32°C and 36°C compared to 22°C
(Fig. 6 c, d; grey (22°C), yellow (32°C) and orange (36°C) lines). Within the 40°C
experiment, fluctuations of derivative fluctuations were slightly damped in *pgm1*, and
rates became negative in BE_2 during the second half of heat exposure, i.e., between
4h and 6h before they increased again until 7h in the light period (Fig. 6 d; red lines).
In addition to derivative functions which revealed the absolute changing rates of
balance equations, fundamental frequencies of Fourier polynomials of BE_1 and BE_2
further suggested a differential effect of high temperature on genotypes' carbon
balances (Fig. 7). In Col-0, frequencies of BE_1 and BE_2 under 32°C and 36°C were
almost double as high as in the 22°C experiment before they dropped in the 40°C
experiment. This decrease was more emphasized in BE_2 than in BE_1 indicating a
contribution of sugar dynamics. In *pgm1*, frequencies of BE_1 constantly increased
with temperature in experiments (Fig. 7 a). Frequencies of BE_2 peaked under 32°C
and, finally, were lower under 40°C than under 22°C (Fig. 7 b). In *spsa1*, dynamics of
BE_1 and BE_2 frequencies across experiments were similar, yet more pronounced in

BE$_2$. In contrast to Col-0 and *pgm1*, lowest frequency of both balance equations was
observed for the 36°C experiment.

Numerical integration of NPS rates over time period in light revealed a decreased
amount of assimilated carbon due to heat exposure in Col-0 (Fig. 8 a) and *pgm1*
(Fig. 8 d). A detailed summary of numerical values of integrals is provided in the
supplements (Table S2). In Col-0, transient heat effects on NPS rates became
strongest after 2h of temperature treatment, i.e. after 4h in the light period. Between
4h and 6h, particularly the amount of carbon assimilated at 36°C and 40°C deviated
clearly from the 22°C experiment (Fig. 8 a). In *pgm1*, this effect was observed two
hours earlier, i.e., during the first two hours of heat treatment between 2h and 4h in
the light period (Fig. 8 d). Surprisingly, however, the 40°C effect on carbon
assimilation was not as strong as observed for 32°C and 36°C. In *spsa1*, carbon
assimilation under heat was most robust and similar to control conditions, i.e., 22°C
(Fig. 8 g).
Integrals of BE$_1$, which in addition to NPS rates also considered starch dynamics,
revealed that starch dynamics in Col-0 were adjusted proportionally to affected NPS
rates during transient heat exposure (Fig. 8 b). In particular, integrals of 32°C and
40°C experiments became similar to the control experiment (22°C). Due to starch
deficiency, this effect was not observed in *pgm1* (Fig. 8 e) while heat exposure
resulted in larger integrals of BE$_1$ in *spsa1* (Fig. 8 h). These heat-induced effects
became more pronounced in BE$_2$ integrals which additionally considered net carbon
flux into soluble sugar biosynthesis. In Col-0, discrepancy of integrals between heat
and control experiments was minimized during the first half of the light period, i.e.,
within the first two hours of heat exposure (Fig. 8 c). During the second half of the
light period, discrepancy increased for 36°C and 40°C experiments. In *pgm1*, net
carbon flux into sugar biosynthesis was reduced in a temperature-dependent manner
which resulted in an (over-)compensation of reduced CO_2 assimilation rates under
36°C and 40°C (Fig. 7 f). Also, in *spsa1* integrals of BE$_2$ increased under transient
heat but this effect was less pronounced than in *pgm1* and Col-0 (Fig. 8 i).
During the recovery phase, in which temperature was set to 22°C again (6h → 8h of
light period), changes in starch and sugar dynamics became obvious in integrals of
BE$_1$ and BE$_2$ for all genotypes. In Col-0 this effect was most pronounced within 36°C
and 40°C experiments (Fig. 8 b, c). In this phase, curves of integrals showed an

inflection point directing the curve of integrals towards the control samples. In
summary, this indicated reversibility of temperature-induced metabolic effects.

Discussion

In temperate regions, plants are frequently exposed to a changing temperature
regime, and these changes might occur over short and relatively long time scales.
For example, temperature typically changes between day and night, and beyond,
temperature might also change transiently within the diurnal light and dark period.
While temperature acclimation of plants typically can be observed after days of
exposure to non-lethal cold or heat (Atanasov *et al.*, 2020; Garcia-Molina *et al.*,
2020), transient temperature changes and plant stress response occurs within
minutes or hours. Interestingly, *Arabidopsis thaliana* was found to memorize already
5 minutes of heat stress which indicates a tightly regulated molecular network
involved in heat stress response (Oyoshi *et al.*, 2020). High temperature, e.g.,
between 35°C – 40°C, is well known to result in a reduced rate of photosynthesis
(Sharkey, 2005) which has also been observed in the present study. While in Col-0
and *spsa1*, 32°C resulted in only slightly decreased NPS rates, higher temperatures
of 36°C and 40°C resulted in a significantly decreased NPS rate during the second
half of the heat exposure period only in Col-0 (see Fig. 2). As previously
summarised, a decreased NPS rate is not due to photosystem damage, but rather
due to rubisco deactivation (Sharkey, 2005). Consistent with previous findings which
show a decreased rubisco activation at leaf temperature >35°C (Crafts-Brandner and
Salvucci, 2000), the effect of 32°C on NPS rates was much less significant than at
36°C and 40°C. In *spsa1*, however, NPS rates were less affected by heat than in
Col-0 which might have several reasons. First, compared to Col-0, *spsa1* might have
had a reduced rate of photorespiration and/or mitochondrial respiration during heat
exposure. While it remains speculation from our study, a higher starch accumulation
rate in *spsa1* might result in a lowered respiration rate under heat because carbon
equivalents can be fixed more efficiently. The observation of a destabilized NPS rate
in starchless *pgm1* plants would support the stabilizing role of starch biosynthesis
under transient heat exposure. However, previous reports under ambient conditions
have shown that SPS knockout mutants have rather enhanced than lowered dark
respiration rates which does not directly support this hypothesis (Bahaji *et al.*, 2015).

Another explanation might be a secondary effect of the *spsa1* mutation on rubisco
and/or rubisco activase which, to our knowledge, has not been shown in current
literature but which needs to be excluded in future studies.
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Fructose concentration profiles are more conserved than glucose and sucrose
profiles during transient heat exposure

While sucrose and glucose metabolism showed a dynamic and differential
accumulation profile between 32°C, 36°C and 40°C experiments, dynamics of
fructose concentrations were most conserved across all temperature treatments and,
remarkably, also across genotypes (see Fig. 5). An initial accumulation within the
first 2h of the light period was followed by a significant decrease until the end of heat
exposure and an accumulation during the recovery phase between 6h and 8h of the
light period. Only *pgm1* showed a differential pattern at 22°C and 32°C but became
similar in its fructose profile to Col-0 and *spsa1* under 36°C and 40°C. In mature
Arabidopsis leaves, fructose levels are significantly affected by invertases which
catalyze hydrolysis of sucrose and release free hexoses (Wan *et al.*, 2018), and by
fructokinase catalyzing ATP-dependent phosphorylation which yields fructose-6-
phosphate (Claeyssen and Rivoal, 2007). As fructose and glucose profiles differed in
the present study, this cannot (solely) be explained by invertase reactions which
release equimolar concentrations of both hexoses. However, differential regulation of
hexokinase and fructokinase could explain the different hexose profiles.
Fructokinase yields the direct substrate for glycolysis, TCA cycle and mitochondrial
respiration. In a previous study which analysed transcript levels in *Arabidopsis*
thaliana under combined drought and heat stress found increased transcripts for
both hexokinase and fructokinase (Rizhsky *et al.*, 2004). Although the experimental
design differed significantly from the present study, together with other findings this
suggests a central role of hexose phosphorylation in heat stress response and
acclimation (Atanasov *et al.*, 2020). As leaf respiration rates typically increase under
elevated temperature (Dusenge *et al.*, 2019), observed consistent fructose dynamics
might be due to a relatively high rate of glycolytic consumption under transient heat
exposure.
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Data integration by Fourier polynomials reveals dynamic properties of plant carbon
metabolism in a changing environment

Integrating net CO_2 assimilation rates with starch and sugar turnover allows for
balancing of the central carbohydrate metabolism. In this context, Fourier
polynomials support the functional and time-continuous estimation of dynamics of
metabolism. Integration and differentiation of Fourier polynomials is straight forward,
and, at the same time, provides a comprehensive mathematical framework which is
applied in diverse fields of natural sciences and engineering (Li *et al.*, 2018; Plonka
et al., 2018; Wang *et al.*, 1997). As described in the block diagram model (see Fig.
1), metabolic dynamics were simulated by addition of Fourier polynomials comprising
input functions (NPS rates) and consuming functions (starch and sugar dynamics).
With such a design, dynamics of plant carbon balancing become traceable without
the need for application of composed spline functions. Further, properties of Fourier
polynomials can reveal further insight into metabolic regulation and consequences of
environmental changes. For example, in the present study both amplitude and
frequency of derivatives of balancing equations differed with regard to genotype and
environment. A different pattern was observed in Col-0 and *spsa1* than in *pgm1*,
indicating that the starchless mutant has a less buffered metabolic response towards
heat stress than both other genotypes. This was supported by the comparison of
fundamental frequencies of BE_1 and BE_2 Fourier polynomials. Here, a genotype-
specific pattern was observed which reflected the impact of starch deficiency in
pgm1 and comparatively high metabolic dynamics in *spsa1* within the 40°C
experiment (see Fig. 4 and 5). Thus, summarizing effects of transient heat on NPS
rates and carbohydrate metabolism resulted in characteristic Fourier polynomials
which enabled the discrimination of genotypes by their derivatives and fundamental
frequencies. This discrimination was rather enabled by the dynamics of frequencies
across different temperature experiments than their absolute value.
Fourier analysis and spectra of frequencies have been applied before in a different
context, e.g., to analyse gene-expression data time-series data (Dong *et al.*, 2011).
These authors coupled Fourier analysis to supervised learning algorithms to
discriminate between housekeeping genes and non-housekeeping genes in HeLa
cells. This example provides evidence for the suitability of Fourier analysis to be
combined with machine learning algorithms which is of particular interest for large-
scale data sets. However, also data sets with only a relatively low number of
variables may need mathematical functions for quantitative analysis and integration,
e.g., as shown in the present study. This is due to the need for combining dynamics

of variables rather than steady-state values under one condition. Here, Col-0, *pgm1*
and *spsa1* could successfully be discriminated by the dynamics of fundamental
frequencies of Fourier polynomials across different experiments rather than by one
absolute value of a frequency. This observation further emphasizes the need for a
functional mathematical description of experimentally observed system dynamics
because underlying attributes, e.g., monotonicity or curvature, can be derived from
such a description. These attributes provide valuable information about system
properties like stability or predictability which need to be essentially addressed for
predictive modelling (Angeli, 2013; Sontag, 2007). Conclusively, Fourier analysis
provides a mathematical approach which can essentially support nonlinear modelling
of metabolism, and which might, in future studies, even serve as a mathematical
framework to connect oscillations in metabolism with quantum theory (lotti *et al.*,
2010; Jaffe *et al.*, 2020).
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488

Supplementary data

490 **Figure S1.** SPS maximum activity under substrate saturation in Col-0 and *spsa1*.
491 **Figure S2.** Schematic representation of the experimental design used in this study.
492 **Figure S3.** Summary of actual growth cabinet parameters during sampling period.
493 **Table S1.** Fourier polynomials and balance equations.
494 **Table S2.** Results of numerical integration of net carbon assimilation.
495

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Author contributions

502 All authors performed experiments. CS and TN performed statistics and modelling
503 and wrote the paper. All authors approved the paper.
504

Data availability statement

506 The datasets used and/or analysed during the current study are provided in the
507 manuscript and supplements.

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Figure Legends

Figure 1. Block diagram applied for Fourier polynomial balance modelling. Input functions are marked in grey (NPS), green (starch amount) and light blue (sugar amount) coloured blocks (left side). Arrows indicate the direction of flux and connect input blocks via multiplication ('x') summation ('+/-'), differentiation ('d/dt') and integration ('∫') with output blocks.

Figure 2. Rates of net CO_2 uptake during short day transient heat exposure. Error bars represent mean values +/- SD of experimental data (n=3), lines represent Fourier series fits. (a) – (c): Col-0, (d) – (f): *pgm1*, (g) – (i): *spsa1*. Grey lines: 22°C experiment; yellow lines: 32°C experiment; orange lines: 36°C experiment; red lines: 40°C experiment. Temperature was set to 22°C between 0h-2h and 6h-8h. Temperature was transiently increased between 2h and 6h. Temperature curves recorded during the experiments are illustrated in Figure S3a. A summary of Fourier polynomial coefficients is provided in the supplements (Table S1).

Figure 3. Starch amounts during short day transient heat exposure in glucose equivalents. Error bars represent mean values +/- SD of experimental data. (a) - (d): Col-0 (n≥5); (e) - (h): *pgm1* (n≥3); (i) - (l): *spsa1* (n≥5). Grey bars: 22°C experiment; yellow bars: 32°C experiment; orange bars: 36°C experiment; red bars: 40°C experiment. Temperature was set to 22°C between 0h-2h and 6h-8h. Temperature was transiently increased between 2h and 6h.

Figure 4. Sucrose concentrations during short day transient heat exposure. Error bars represent mean values +/- SD of experimental data. (a) - (d): Col-0 (n≥5); (e) - (h): *pgm1* (n≥3); (i) - (l): *spsa1* (n≥5). Grey bars: 22°C experiment; yellow bars: 32°C experiment; orange bars: 36°C experiment; red bars: 40°C experiment. Temperature was set to 22°C between 0h-2h and 6h-8h. Temperature was transiently increased between 2h and 6h.

Figure 5. Hexose concentrations during short day transient heat exposure. Error bars represent mean values +/- SD of experimental data. Range of the y-axes differs for *pgm1* due to high difference in concentration. (a) Glucose concentrations; (b) Fructose concentrations. Top row: Col-0 (n≥5); middle row: *pgm1* (n≥3); bottom row: *spsa1* (n≥5). Grey bars: 22°C experiment; yellow bars: 32°C experiment; orange bars: 36°C experiment; red bars: 40°C experiment. Temperature was set to 22°C between 0h-2h and 6h-8h. Temperature was transiently increased between 2h and 6h.

Figure 6. Derivatives of carbon balance equations with respect to time. Derivatives of balance equations were built for the experiments "22°C" (control; grey lines), "32°C" (yellow lines), "36°C" (orange lines) and "40°C" (red lines). Left panel represents derivatives of balance equation 1 (BE_1) for Col-0 (a), *pgm1* (c) and *spsa1* (e). Right panel represents derivatives of balance equation 2 (BE_2) for Col-0 (b), *pgm1* (d) and *spsa1* (f).

Figure 7. Fundamental frequencies of Fourier polynomials of BE_1 (a) and BE_2 (b). Frequencies were calculated as the ratio of ($2\pi/T$) where T represents periods.

Bar colour indicates different experiments. Grey bars: 22°C experiment; yellow bars: 32°C experiment; orange bars: 36°C; red bars: 40°C experiment.

Figure 8. Integrals of carbon balance rates during transient heat exposure. NPS rates and rates derived from BE_1 and BE_2 were integrated over time to reveal the total sum of net carbon gain during the light period. **(a) – (c)**: Col-0 integrals of NPS rates **(a)**, BE_1 **(b)** and BE_2 **(c)**. **(d) – (f)**: *pgm1* integrals of NPS rates **(d)**, BE_1 **(e)** and BE_2 **(f)**. **(g) – (i)**: *spsa1* integrals of NPS rates **(g)**, BE_1 **(h)** and BE_2 **(i)**. Numerical values of integrals are provided in the supplements (Table S2).

Figure 1. Block diagram applied for Fourier polynomial balance modelling. Input functions are marked in grey (NPS), green (starch amount) and light blue (sugar amount) coloured blocks (left side). Arrows indicate the direction of flux and connect input blocks via multiplication ('x') summation ('+/-'), differentiation ('d/dt') and integration ('∫') with output blocks.

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