

1 **Discovery and characterization of H_v1-type proton channels in reef-building corals.**

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32 **Abstract**

33 Voltage-dependent proton-permeable channels are membrane proteins mediating a number
34 of important physiological functions. Here we report the presence of a gene encoding for Hv1
35 voltage-dependent, proton-permeable channels in two species of reef-building corals. We
36 performed a characterization of their biophysical properties and found that these channels
37 are fast-activating and modulated by the pH gradient in a distinct manner. The biophysical
38 properties of these novel channels make them interesting model systems. We have also
39 developed an allosteric gating model that provides mechanistic insight into the modulation
40 of voltage-dependence by protons. This work also represents the first functional
41 characterization of any ion channel in scleractinian corals. We discuss the implications of the
42 presence of these channels in the membranes of coral cells in the calcification and pH
43 regulation processes and possible consequences of ocean acidification related to the function
44 of these channels.

45

46 **Introduction**

47 Scleractinian or stony corals are organisms in the phylum Cnidaria that deposit calcium
48 carbonate in the form of aragonite to build an exoskeleton. Stony corals are the main
49 calcifying organisms responsible for the construction of coral reefs, which are major
50 ecosystems hosting numerous and diverse organisms. Coral reefs also act as natural barriers
51 from strong ocean currents, waves and tropical storms, providing coastal protection. This
52 protection centers on the ability of scleractinian corals to produce enough calcium carbonate
53 (CaCO_3). The increase in atmospheric CO_2 concentrations as a result of human activity poses
54 threats to coral-reef building organisms due to rising sea surface temperatures (Hoegh-
55 Guldborg, 1999) and because CO_2 is taken up by the ocean, dangerously lowering the pH of
56 the sea water (Caldeira and Wickett, 2003).

57 It is known that precipitation of the aragonitic form of calcium carbonate is facilitated at
58 elevated pH values, at very low concentrations of protons. Calcification by scleractinian
59 corals is a process that has been shown to be modulated by the pH of the solution in which
60 calcium carbonate is precipitated (Allemand et al., 2011). To this end, corals produce a
61 specialized compartment between the ectoderm and the external substrate or skeleton
62 called calicoblastic compartment, which contains a fluid derived from the surrounding sea

63 water. The composition of this calciblastic fluid or liquor is strictly regulated by the coral to
64 maintain both an elevated pH, often close to one unit higher than the surrounding sea water,
65 and an increased concentration of Ca^{2+} and carbonates. The molecular details of pH regulation
66 in the calciblastic fluid are not understood completely. Involvement of proton-pumps has
67 been postulated and is likely to be part of proton transport in corals. Both P-type and V-type
68 hydrogen pumps are present in coral transcriptomes and are known to play roles in the
69 physiology of coral-algal symbiosis (Tresguerres et al., 2017). V-type H^+ -ATPases have also
70 been shown to be involved in calcification in foraminifera (Toyofuku et al., 2017). If a proton
71 pump is involved in lowering proton concentration in the calciblastic fluid to maintain high
72 calcification rates, protons will be transported to the cytoplasm of the ectodermal cells that
73 constitute the calciblastic epithelium, producing a profound acidification of the cytoplasmic
74 pH (pH_i). Although measurements of the (pH_i) in corals indicate values of 7.13-7.4 (Venn et
75 al., 2009), it is unknown how coral cells regulate pH_i . Thus, an efficient pH-regulatory
76 mechanism is to be expected to be present in corals. We hypothesized that proton channels
77 might be fundamental to this physiological process and also required for calcification in hard
78 corals.

79 Although a number of studies have delineated the physiological roles of H_v1 voltage-gated
80 proton channels in vertebrate cells (DeCoursey, 2013), less is known about their role in
81 invertebrates. These channels are potential mediators in processes that are critically
82 dependent on proton homeostasis. As an example, they have been shown to be involved in
83 regulating the synthesis of the calcium carbonate skeleton in coccolithophores, calcifying
84 unicellular phytoplankton (Taylor et al., 2011).

85 The range of voltages over which channel activation occurs is strongly modulated by the
86 transmembrane proton gradient, characterized by: $\Delta\text{pH} = \text{pH}_o - \text{pH}_i$ i.e., the difference
87 between external and internal pH. In the majority of known H_v1 channels, the voltage at
88 which half the channels are activated, the $V_{0.5}$ or the apparent threshold for channel opening
89 (V_{Thr}), shifts by roughly 40 mV per unit of ΔpH . Thus, the pH gradient strongly biases the
90 voltage-independent free energy of channel activation (Cherny et al., 1995). With few
91 exceptions, channel activation occurs at voltages that are more positive than the reversal
92 potential for protons, implying that protons are always flowing outward under steady-state

93 conditions. The fact that most H_v1s mediate outward currents is the reason these channels
94 are mostly involved in reversing intracellular acidification or producing voltage-dependent
95 cytoplasmic alkalization (Lishko and Kirichok, 2010; DeCoursey, 2013).
96 Here we report the presence of genes encoding for H_v1 channels in two species of reef-
97 building corals. We cloned and characterized the biophysical properties of these channels in
98 an expression system using patch-clamp electrophysiology. The demonstration of the
99 presence of voltage-gated proton channels in corals is an initial step to a deeper
100 understanding of coral calcification and its dysregulation under ocean acidification
101 conditions. We show that some of the coral H_v1's biophysical properties are different from
102 other known proton channels and this behavior makes them interesting models to try to
103 understand some basic biophysical mechanisms in these channels. To explain this behavior,
104 we develop a novel activation model to describe voltage- and pH-dependent gating that has
105 general applicability to H_v1 channels.

106

107 **Materials and methods**

108 *Identification of H_v1 sequences and cloning.* Blast searches of the transcriptome of the Indo-
109 Pacific coral *Acropora millepora* (Moya et al., 2012) detected four sequences that we
110 identified as belonging to a putative proton-permeable channel. The GenBank accession
111 numbers for these are: XM_015907823.1, XM_015907824.1, XM_029346499.1 and
112 XM_029346498.1. We designed two pairs of oligonucleotides to amplify two of these
113 sequences (Table 1). Total RNA was extracted from tissue obtained from a fragment of
114 *Acropora millepora* acquired from a local salt-water aquarium provider. RNA was extracted
115 by dipping the whole fragment for 2 min in 5 ml of solution D (4 M guanidinium thiocyanate,
116 25 mM sodium citrate, 5 % sarkosyl and 0.1 M 2-mercaptoethanol). After incubation, tissue
117 was removed by gently pipetting the solution for 2 min. At this point, the calcareous skeleton
118 was removed and RNA extraction continued according to (Chomczynski and Sacchi, 1987).
119 Total RNA (1 µg) from *A. millepora* was used for RT-PCR, employing oligo dT and
120 SuperScripII reverse transcriptase (Invitrogen, USA). Complementary DNA obtained from
121 RT-PCR was used in three PCR reactions using oligos: 1) AcHv1Nter5' and 3'; 2) AcHv1Cter5'
122 and 3' and, 3) AcHv1Nter5' and AcHv1Cter3' (Table 1). The Platinum Pfx DNA polymerase
123 (Invitrogen) was used for amplification according to the manufacturer's instructions. 1 µl of

124 Taq DNA polymerase (Invitrogen, USA) was used for 10 min at 72 °C to add a poly A tail at 5'
125 and 3' ends and facilitate cloning into the pGEM-T vector.

126 The PCR reaction 3 gave rise to a full open reading frame (ORF) containing AmH_v1. New
127 oligos AcHv1Nter5' and AcHv1Cter3' containing restriction sites Kpn1 and Not1 respectively
128 were used to re-amplify the ORF in pGEM-T and subclone it into pcDNA3.1 for heterologous
129 expression.

130 The H_v1 channel from *A. palmata* was cloned from a fragment of an adult specimen collected
131 in the Limones Reef off of Puerto Morelos, Mexico. RNA extraction from small coral pieces
132 was carried out by flash freezing in liquid nitrogen and grinding the frozen tissue. All other
133 cloning procedures were as for *A. millepora*. All clones were confirmed by automatic
134 sequencing at the Molecular Biology Facility of the Instituto de Fisiología Celular at UNAM.

135
136 *Heterologous expression of AmH_v1.* The cloned AmH_v1 was expressed in HEK293 cells. Cells
137 were grown on 100 mm culture dishes with 10 ml of Dulbecco's Modified Eagle Medium
138 (DMEM, Invitrogen) containing 10 % fetal bovine serum (Invitrogen, USA) and 100 units/ml-
139 100 µg/ml of penicillin-streptomycin (Invitrogen, USA), incubated at 37°C in an incubator
140 with 5.2 % CO₂ atmosphere. When cells reached 90 % confluence, the medium was removed,
141 and the cells were treated with 1 ml of 0.05 % Trypsin-EDTA (Invitrogen, USA) for 5 min.
142 Subsequently, 1 ml of DMEM with 10 % FBS was added. The cells were mechanically
143 dislodged and reseeded in 35 mm culture dishes over 5x5 mm coverslips for
144 electrophysiology or in 35 mm glass bottom dishes for FRET experiments. In both cases, 2
145 ml of complete medium were used. Cells at 70 % confluence were transfected with
146 pcDNA3.1-AmH_v1 prepared from a plasmid midiprep, using jetPEI transfection reagent
147 (Polyplus Transfection, France). For patch-clamp experiments, pEGFP-N1 (BD Biosciences
148 Clontech, USA) was cotransfected with the channel DNA to visualize successfully transfected
149 cells via their green fluorescence. Electrophysiological recordings were done one or two days
150 after transfection.

151
152 *FRET measurement of stoichiometry.* In order to measure the stoichiometry of subunit
153 interaction employing FRET, we constructed fusion proteins between AmH_v1 and mCerulean

154 and mCitrine fluorescent proteins (FPs), to be used as donor and acceptor, respectively. The
155 FPs were fused to the N-terminus of the channel in order to disrupt as little as possible the
156 C-terminus mediated interaction. These constructs were transfected into HEK293 cells as
157 described above. The apparent FRET efficiency between FP-containing constructs, E_{app} , was
158 measured via sensitized emission of the acceptor, employing the spectral-FRET method (De-
159 la-Rosa et al., 2013; Zheng et al., 2002). Fluorescence was measured in a home-modified TE-
160 2000U inverted epifluorescence microscope (Nikon, Japan). The excitation light source was
161 an Argon Ion laser (Spectra-Physics, Germany) mainly producing light at 458, 488 and 514
162 nm; the laser beam is focused and then collimated using a 3 mm ball lens and a 50 mm focal
163 length planoconvex lens. Collimated light is steered with a mirror and then is focused into
164 the objective back focal plane by a 300 mm focal length achromatic lens.

165 Cells were imaged with a Nikon 60X oil immersion objective (numerical aperture 1.4). The
166 detection arm of the microscope is coupled to a spectrograph (Acton Instruments, USA) and
167 an EM-CCD camera (Ixon, Andor, Ireland) controlled by Micromanager software (Edelstein
168 et al., 2014). Excitation was achieved with appropriate excitation filters (Chroma, Vermont,
169 USA) for mCerulean (458 nm) and mCitrine (488 nm). The emission filter was a long-pass
170 filter in order to collect the full emission spectrum of the FRET pair. Apparent FRET
171 efficiency is plotted as a function of the fluorescence intensity ratio ($I_{donor}/I_{acceptor}$). This
172 relationship can be fitted with models of subunit association with fixed stoichiometry,
173 according to (De-la-Rosa et al., 2013).

174

175 *Electrophysiology.* All chemicals for solutions were acquired from Sigma-Aldrich (Mexico).
176 Proton current recordings were made from HEK293 cells expressing pCDNA3.1-AmH_v1 in
177 the inside-out, whole-cell and outside-out configurations of the patch-clamp recording
178 technique. For whole-cell and inside-out recordings, the extracellular solution (bath and
179 pipette, respectively) was (in mM): 80 TMA-HMESO₃, 100 buffer (MES: pH 5.5, 6.0 and 6.5;
180 HEPES: pH 7.0, 7.5), 2 CaCl₂, 2 MgCl₂ and pH adjusted NMDG/TMAOH and HCl. The
181 intracellular solution (pipette and bath respectively) was (in mM): 80 TMA-HMESO₃, 100
182 buffer (MES: pH 5.5, 6.0 and 6.5; HEPES: pH 7.0, 7.5), 1 EGTA and pH adjusted NMDG/TMAOH
183 and HCl.

184 *Conditions for recording zinc effects.* The effect of zinc was evaluated in outside-out patches
185 at a ΔpH of 1. The bath solution composition was (in mM): 100 TMA-HMESO₃, 100 HEPES, 8
186 HCl, 2 CaCl₂, 2 MgCl₂, and the indicated concentration of ZnCl₂. The pipette solution was (in
187 mM): 100 TMA-MESO₃, 100 MES, 8 HCl, 10 EGTA and 2 MgCl₂. Both solutions were adjusted
188 to pH 7 and pH 6 respectively with TMA-OH/HCl. Patches were placed in front of a perfusion
189 tube that was gravity-fed with the appropriate solution. Tubes were changed with a home-
190 built rapid perfusion system.

191 Macroscopic currents were low-pass filtered at 2.5 kHz, sampled at 20 kHz with an Axopatch
192 200B amplifier (Axon Instruments, USA) using an Instrutech 1800 AD/DA board (HEKA
193 Elektronik, Germany) or an EPC-10 amplifier (HEKA Elektronik, Germany). Acquisition
194 control and initial analysis was done with PatchMaster software. Pipettes for recording were
195 pulled from borosilicate glass capillaries (Sutter Instrument, USA) and fire-polished to a
196 resistance of 4-7 M Ω when filled with recording solution for inside- and outside-out
197 recordings and 1-3 M Ω for whole-cell. The bath (intracellular) solutions in inside-out patches
198 were changed using a custom-built rapid solution changer. For whole-cell recordings all the
199 bath solution was exchanged to manipulate pH. In some recordings, linear current
200 components were subtracted using a p/4 subtraction protocol.

201
202 *Data analysis.* Conductance, G , was calculated from I-V relations assuming ohmic
203 instantaneous currents, according to:

$$204 \quad I(V) = G \cdot (V - V_{rev})$$

205 The normalized conductance-voltage (G - V) relations were fit to a Boltzmann function
206 according to equation 1:

$$207 \quad \frac{G}{G_{max}} = \frac{1}{1 + \exp\left(\frac{q(V - V_{0.5})}{K_B T}\right)} \quad \text{Equation 1.}$$

208 Here, $V_{0.5}$ is the voltage at which $G/G_{max} = 0.5$, q is the apparent gating charge (in elementary
209 charges, e_0) and K_B is the Boltzmann constant and T temperature in Kelvin (22°C).

210 The time constant of activation was estimated via a fit of the second half of currents to the
211 equation:

212
$$I(t) = I_{ss} \cdot \left(1 - e^{\left(\frac{-(t-\delta)}{\tau}\right)}\right)$$
 Equation 2.

213 Where I_{ss} is the amplitude of the current at steady-state, δ is the delay of the exponential with
214 respect to the start of the voltage pulse and τ is the time constant, both with units of ms. The
215 voltage-dependence of δ and τ were estimated from a fit to equation:

216
$$k(V) = k(0)e^{(-Vq_i/K_B T)}$$
 Equation 3.

217 Where i stands for δ or τ and $k(0)$ is the value of either parameter at 0 mV.

218 Currents in the presence of zinc were normalized to the current before application of the ion
219 to obtain a normalized fraction of current blocked as: $F_B = 1 - I/I_{max}$. The zinc dose response
220 curve was fitted to Hill's equation in the form:

221
$$F_B = \frac{1}{1 + \left(\frac{K_D}{[Zn]_o}\right)^{n_H}}$$
 Equation 4.

222 K_D is the apparent dissociation constant, $[Zn^{2+}]_o$ is the extracellular zinc concentration and
223 n_H is the Hill coefficient.

224

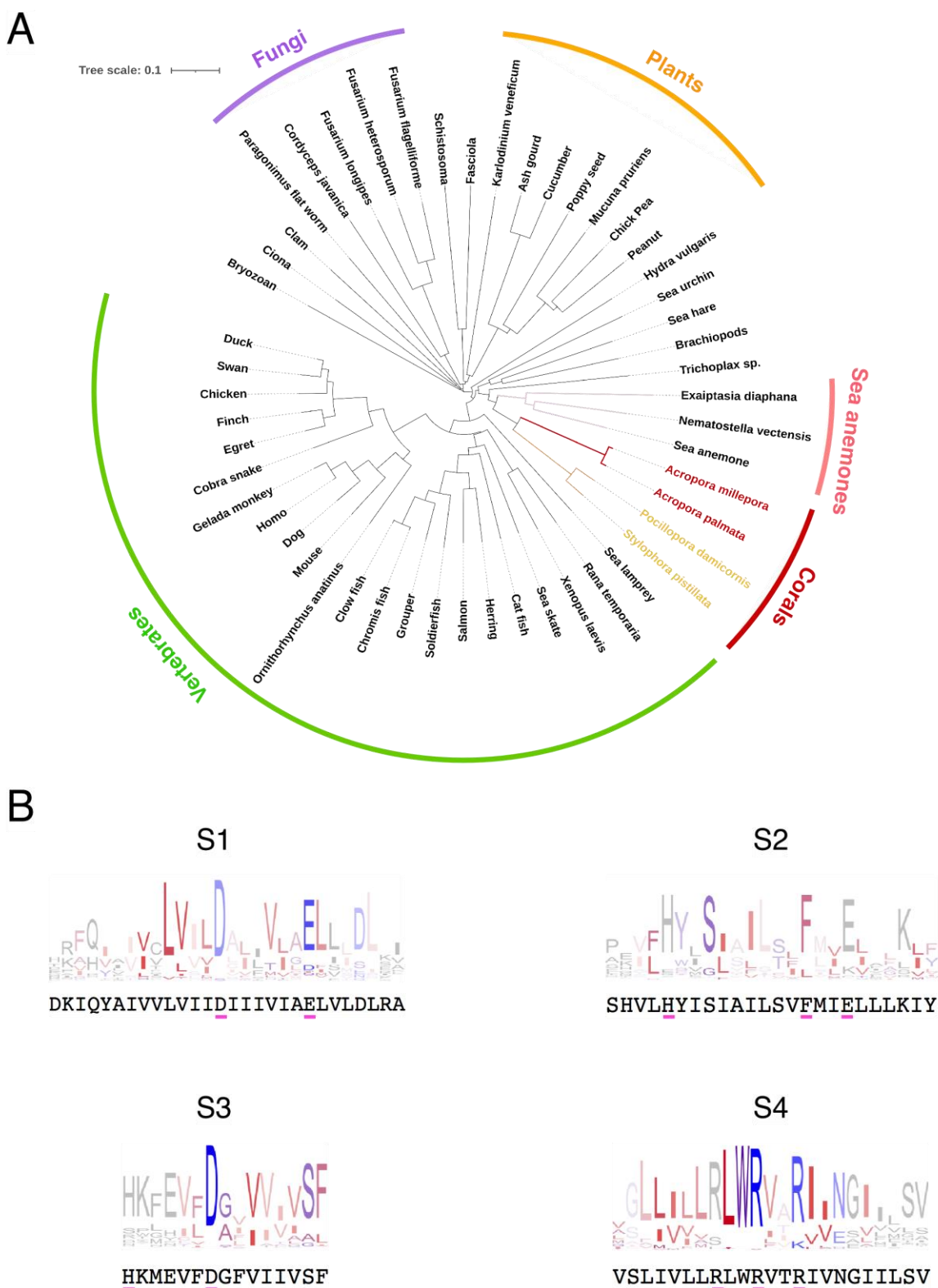
225 Results

226 Ion channels have not been studied in corals. In an effort to initiate their study in these
227 organisms, we searched the transcriptome of the Indo-Pacific coral *Acropora millepora*
228 (Moya et al., 2012) for sequences coding for putative voltage-sensing residues present in
229 canonical H_v1 channels with the form: RxxRxxRIx, which corresponds to the S4 segment of
230 H_v1 channels and is also found in other voltage-sensitive membrane proteins. Blast searches
231 detected four sequences that seem to correspond to a gene encoding the H_v1 voltage-
232 activated proton-selective ion channel (Ramsey et al., 2006; Sasaki et al., 2006). *Acropora*
233 *millepora* is one of the most widely studied species of scleractinian corals and is well
234 represented in the commercial coral trade (Cleves et al., 2018; Ying et al., 2019). We
235 proceeded to clone this gene from a small specimen of *A. millepora* obtained from a local
236 aquarium (Reef Services, Mexico City). As indicated in the methods section, total RNA was
237 extracted from tissue and mRNA was retrotranscribed to obtain cDNA. We managed to
238 obtain a full-length clone and refer to this sequence as Am H_v1 or H_v1 -type proton channel of
239 *Acropora millepora*.

240 We were interested in knowing if the same gene is present in a closely related species from
241 the Caribbean Sea. Thus, we used the same primers to clone the H_v1 channel from *Acropora*
242 *palmata*, a widespread coral in the same family and which we call ApH_v1. The amino acid
243 sequence is almost identical to AmH_v1 (Figure 1-Supplement1A), the greatest divergence is
244 found between a few amino acid residues in the C-terminal region. This result suggests that
245 despite the large biogeographic difference, these two genes have not diverged significantly.
246 The ApH_v1 sequence also gives rise to fast-activating voltage-gated proton currents (Figure
247 1-Supplement1B).

248 The most diagnostic feature of the H_v1 protein is the sequence of the fourth transmembrane
249 domain or S4, which contains three charged amino acids in a characteristic triplet repeat.
250 The presence of these repeats in our sequence allowed us to initially identify our clone as an
251 H_v1 channel. However, we decided to compare our sequence to those of several
252 H_v1 orthologues. We selected a list of 130 H_v1 protein sequences that are well curated in the
253 Gene Bank (<https://www.ncbi.nlm.nih.gov/>), representing several branches of the
254 eukaryotes, from unicellular plants to mammals. As expected, the protein sequence of AmH_v1
255 has similarity to several other H_v1 genes from varied organisms (Figure 1A). The identity
256 varies from 98% when compared to other putative coral and anemone sequences, to less
257 than 30% when compared to plant and nematode sequences. In spite of this variability, the
258 putative transmembrane domains of all these proteins show high conservation and
259 consensus sequences logos can detect the presence of highly conserved individual amino
260 acid sequences that can be considered characteristic of H_v1 channels. Figure 1B compares
261 these transmembrane domain consensus logos with our AmH_v1 sequence. It can be gleaned
262 that AmH_v1 contains the highly conserved residues that form the voltage-sensing amino acid
263 residues in S4 as well as their acidic pairs present in S2 and S3. The extracellular histidine
264 residues involved in Zn²⁺ coordination are also present. These results suggest that our
265 sequence is that of a *bona fide* H_v1 voltage-sensing domain (VSD).

266



268 **Figure 1. Conservation and phylogenetic relationships of H_v1 channels.** A) Tree
269 obtained from a multiple sequence alignment from H_v1 channels in CLUSTAL-O. Highlighted
270 in red and yellow are the branches containing coral H_v1 sequences. B) Consensus logo
271 sequences of transmembrane domains of H_v1 channels. The color code indicates the
272 hydrophobicity of each residue, where blue indicates charged residues, red indicates non-
273 polar residues and other colors indicate either non-polar or charged residues with less
274 conservation.

275
276 Apart from canonical voltage-gated channels, several other proteins contain VSDs. Examples
277 are the voltage-sensing phosphatases like VSPs (Iwasaki et al., 2008) and TPTE and TPTE2
278 (Halaszovich et al., 2012) proteins (transmembrane protein with Tensin homology) and
279 genes like TMEM266. These proteins are relevant to us since some TPTEs have been shown
280 to also mediate proton currents and TMEM266 can be modulated by Zn²⁺ (Papp et al., 2019).
281 We compared the sequence of AmH_v1 with several orthologues of TPTEs and TMEM266.
282 Although there is some similarity within transmembrane domains (Figure 1-Supplement 2),
283 the overall sequence comparison shows that AmH_v1 and these VSD-containing proteins are
284 different.

285 As mentioned before, we performed a multiple sequence alignment with 130 H_v1 sequences.
286 In Figure 2 we show the detailed sequence alignment of AmH_v1 with five of these sequences,
287 which represent some of the best studied H_v1 genes. It can be seen that there is a high degree
288 of identity, especially in the transmembrane domains. The least degree of conservation
289 appears when comparing this sequence to the dinoflagellate *Karlodinium veneficum* H_v1
290 channel (Figure 2A). A search of available transcriptomes from several coral species allowed
291 us to detect the presence of sequences that are found in H_v1 channels. This suggests that H_v1
292 proton channels might be found in many families of scleractinian corals (Figure 2-
293 Supplement 1), as has also been recently shown (Capasso et al., 2021).

294 Secondary-structure prediction suggests that AmH_v1 is a canonical H_v1 channel formed by a
295 voltage-sensing domain with four transmembrane segments. The protein sequence was used
296 for 3D modeling using the SWISS MODEL server (Waterhouse et al., 2018), which produced
297 models based on the H_v1 chimera structure (Takeshita et al., 2014a) and the Kv1.2 potassium
298 channel voltage-sensing domain (Long et al., 2005). This structural model is shown in Figure

299 2B. The predicted model indicates a shortened N-terminal region, four transmembrane
 300 domains and a long C-terminal helix.

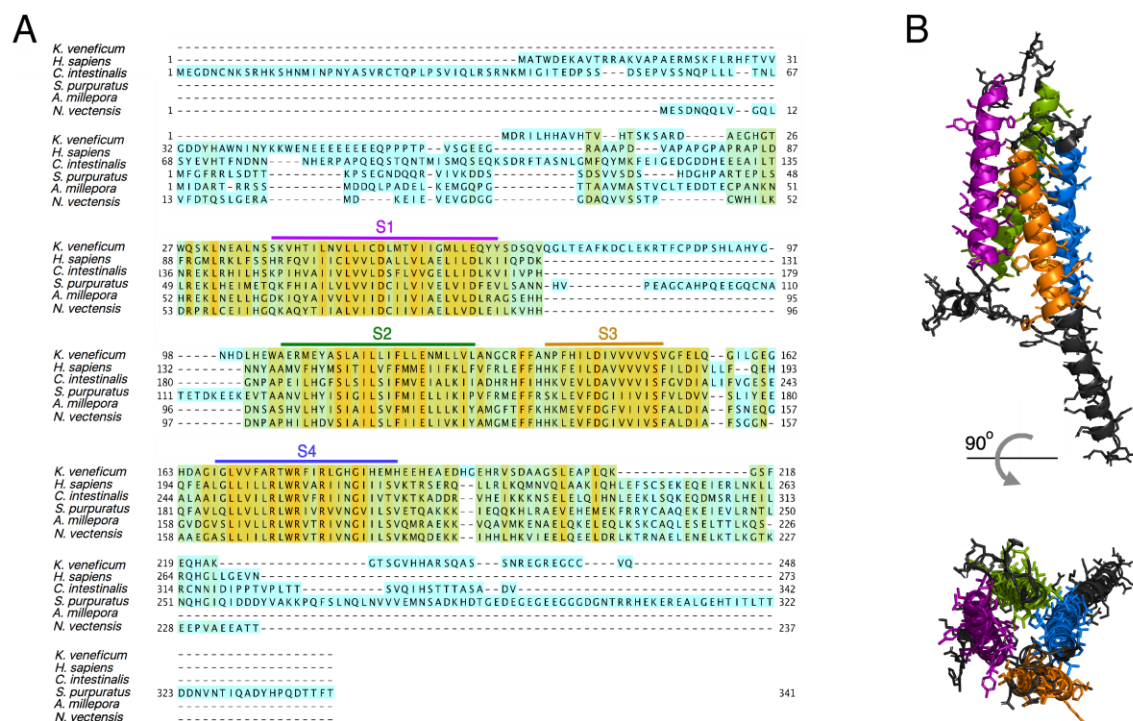


Figure 2

301
 302
 303 **Figure 2. Protein sequence alignment of the *Acropora millepora* Hv1 (AmHv1) channel**
 304 **with selected Hv1s from other organisms.** A) Amino acid sequence alignment of AmHv1
 305 with other known Hv1 orthologues provided by the CLUSTAL-O algorithm. The predicted
 306 transmembrane domains are shown by the colored horizontal lines and letters. The colors
 307 highlighting the sequence indicate sequence identity. Orange indicates identical amino acids,
 308 blue indicates no identity. B) Predicted structural topology of AmHv1. Transmembrane
 309 domains are colored to correspond with the sequences in A. The top panel is the view parallel
 310 to the membrane while the bottom panel is the view from the top (extracellular) side.

311
 312

313 Voltage-gated proton channels from *Ciona* (Sasaki et al., 2006) and humans (Lee et al.,
314 2008a) have been shown to express as dimers in the plasma membrane and this dimeric
315 form is understood to be the functional unit of these proton channels. The dimer is stabilized
316 by a coiled-coil interaction mediated by an alpha helical C-terminal domain. As shown by the
317 model in Figure 2, AmH_v1 has a long C-terminal helix, which is predicted to engage in a
318 coiled-coil (Paircoil2 (McDonnell et al., 2006). We calculated the probability per residue to
319 form a coiled-coil for all the C-terminal residues, both for human and AmH_v1 channels, using
320 the program COILS (Lupas et al., 1991). Figure 3A shows that the coiled-coil probability for
321 AmH_v1 C-terminus is at least as high or higher than that for hH_v1, an established dimer,
322 strongly suggesting that coral H_v1s might also form dimers.

323 In order to study the oligomeric state of the coral H_v1, we performed FRET experiments with
324 AmH_v1 channel tagged with fluorescent proteins as a FRET pair. Figure 3B shows that there
325 is significant FRET efficiency between fluorescent protein-tagged subunits, indicating a very
326 close interaction between monomers. The measured apparent FRET efficiency vs. the
327 fluorescence intensity ratio can be fitted to a model where the subunits assemble as a dimer.
328 From this fit, we can estimate a distance between fluorophores of ~60 Å, which is compatible
329 with AmH_v1 being a dimer, at least in HEK293 cells.

330

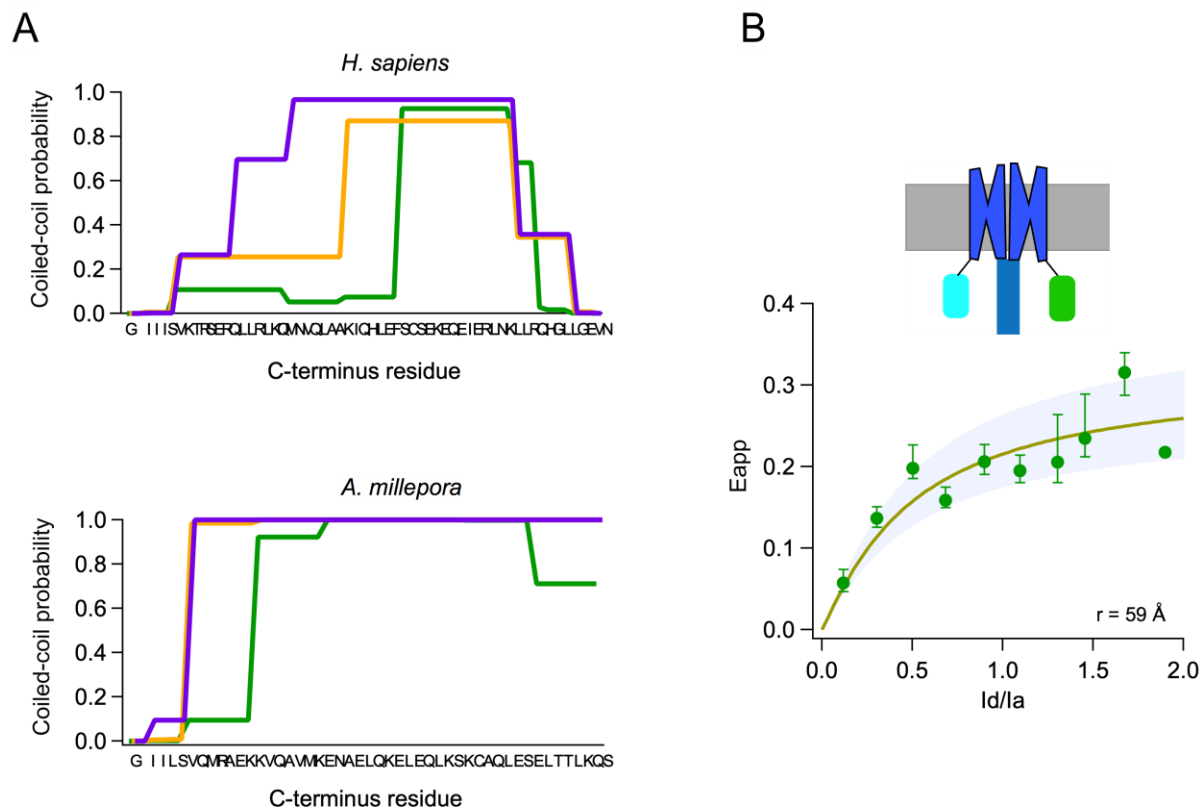


Figure 3

331
 332 **Figure 3. Subunits of the *Acropora millepora* Hv1 (AmHv1) channel associate to form**
 333 **dimers.** A) Probability of coiled-coil formation per amino acid residue of the C-terminus
 334 domain of hHv1 (top) and AmHv1 (bottom). The different colors correspond to the three
 335 seven-residue windows used by the program to calculate the score. The sequence of the C-
 336 terminus is shown in the x-axis. B) FRET measurement of dimer formation. The apparent
 337 FRET measured from 134 cells is plotted as a function of the ratio of donor to acceptor
 338 fluorescence (I_d/I_a). Shown is the average and s.e.m. for data in I_d/I_a windows of 0.1. The
 339 continuous curve is the fit of the data to the prediction of a model that considers random
 340 assembly of donor- and acceptor-tagged subunits into a dimer. The separation of the FRET
 341 pair in a dimer is $\sim 60 \text{ \AA}$, according to the model. The upper panel depicts a cartoon of the
 342 presumed FP-tagged dimer in the membrane.

343
 344
 345

346 ***Functional expression of AmH_v1. Voltage-dependence and kinetics.***

347 The cDNA of AmH_v1 was cloned in the pcDNA3 expression vector and transfected into
348 HEK293 cells. Under whole-cell conditions we recorded large voltage-dependent outward
349 currents. Figure 4A shows a family of such currents. The data suggest that these currents
350 were carried mostly by protons, since the reversal potential, measured from a tail current
351 protocol, closely followed the equilibrium potential for protons, as given by the Nernst
352 equation (Figure 4B).

353 The voltage-dependence of channel gating was estimated from a fit of the normalized
354 conductance vs. voltage to equation 1. The steepness of the curve corresponds to an apparent
355 charge of $\sim 2 e_0$, comparable to other H_v1's under similar recording conditions (Figure 4C).
356 Interestingly, these channels seem to activate rapidly. This is apparent from the current
357 traces, which reach a steady-state within a few hundred ms (Figure 4A), as quantified in
358 Figure 4D. Equation 3 estimates two parameters, an activation time constant (τ) and a delay
359 (δ). Both the time constant and the delay are similarly voltage-dependent at positive
360 potentials. The existence of a delay in the time course implies that activation is a multiple
361 state process. The delay magnitude is smaller than the time constant at all voltages, which
362 can be interpreted to mean that the rate limiting step for opening comes late in the activation
363 pathway (Schoppa and Sigworth, 1998).

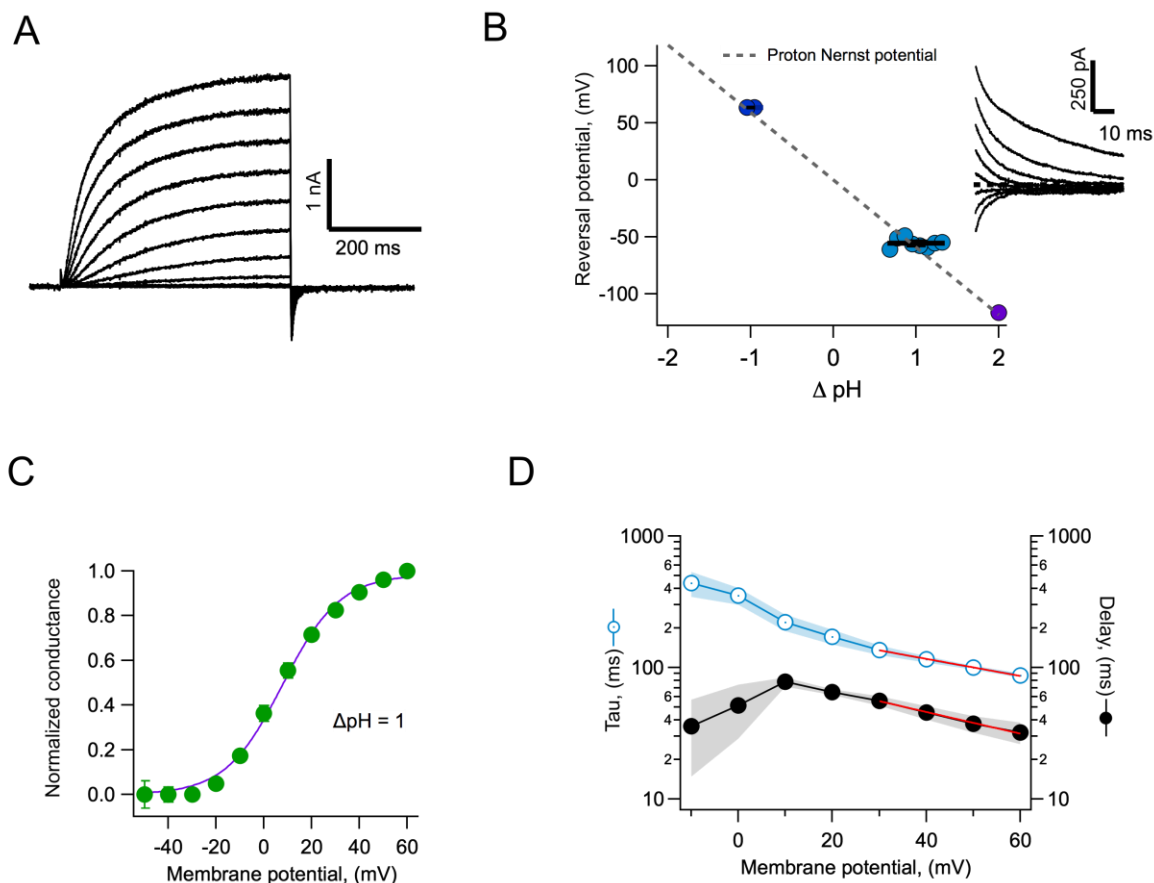


Figure 4

364
 365 **Figure 4. Proton currents mediated by AmH_v1 expressed in HEK 293 cells.** A) Typical
 366 proton current family elicited by depolarizing pulses from -50 to 60 mV in 10 mV intervals.
 367 The duration of the pulses is 500 ms. Linear current components have been subtracted. B)
 368 Reversal potential of currents as a function of the pH gradient. Symbols are individual data
 369 and the black horizontal lines are the mean. The dotted line is the expected reversal potential
 370 as predicted by the Nernst equation. The inset shows a tail current family from which
 371 instantaneous IV curves were extracted to measure the reversal potential. Recordings
 372 shown in A and B were obtained in the whole-cell configuration. C) Normalized conductance-
 373 voltage curve at Δ pH = 1. The red curve is the fit to equation 1 with parameters $V_{0.5} = 7.85$
 374 mV, $q = 2.09 e_0$. Circles are the mean and error bars are the s.e.m. ($n = 7$). D) Kinetic
 375 parameters of activation. Activation time constant and delay estimated from fits of current
 376 traces to equation 2. Circles are the mean and the s.e.m. is indicated by the shaded areas (n
 377 = 6). The voltage-dependence of the delay and tau of activation were estimated from a fit to

378 equation 3, which appears as the red curve. Parameters are: $\delta(0) = 98.2$ ms, $q_{\delta} = 0.47 e_0$. The
379 voltage-dependence parameters for tau are: $\tau(0) = 212$ ms, $q_{\tau} = 0.37 e_0$.

380

381 ***Comparison to human H_v1 channel properties.***

382 Human H_v1 is probably the best characterized of the voltage-gated proton channels
383 (Musset et al., 2008), so we compared some of the properties of AmH_v1 with hH_v1. AmH_v1
384 channels activate faster than their human counterpart. Figure 5 compares the activation
385 kinetics of these two channels under the same conditions. Steady-state is apparently
386 reached sooner after a voltage pulse in AmH_v1 (Figure 5A) when compared to hH_v1 (Figure
387 5B). The slower kinetics of the human orthologue is also evidenced in the more sluggish
388 deactivation tail currents (Figure 5B). The range of voltages over which activation happens
389 is also different between the two channels, with the coral H_v1 channel activating 40 mV
390 more negative than the human clone (Figure 5C. Notice that the proton gradient is the same
391 in these recordings). Even though AmH_v1 activates at more negative voltages, the
392 activation range is still more positive than the proton reversal potential, thus coral proton
393 currents activated by depolarization, in the steady state and at least as expressed in
394 HEK293 cells, are always outward.

395 The faster kinetics of AmH_v1 is clearly evidenced when the time constant of activation, τ ,
396 estimated using fits of the activation time course to equation 2, is compared for coral and
397 human H_v1 channels. AmH_v1 is almost 10-fold faster at 0 mV and over a range of positive
398 voltages (Figure 5D).

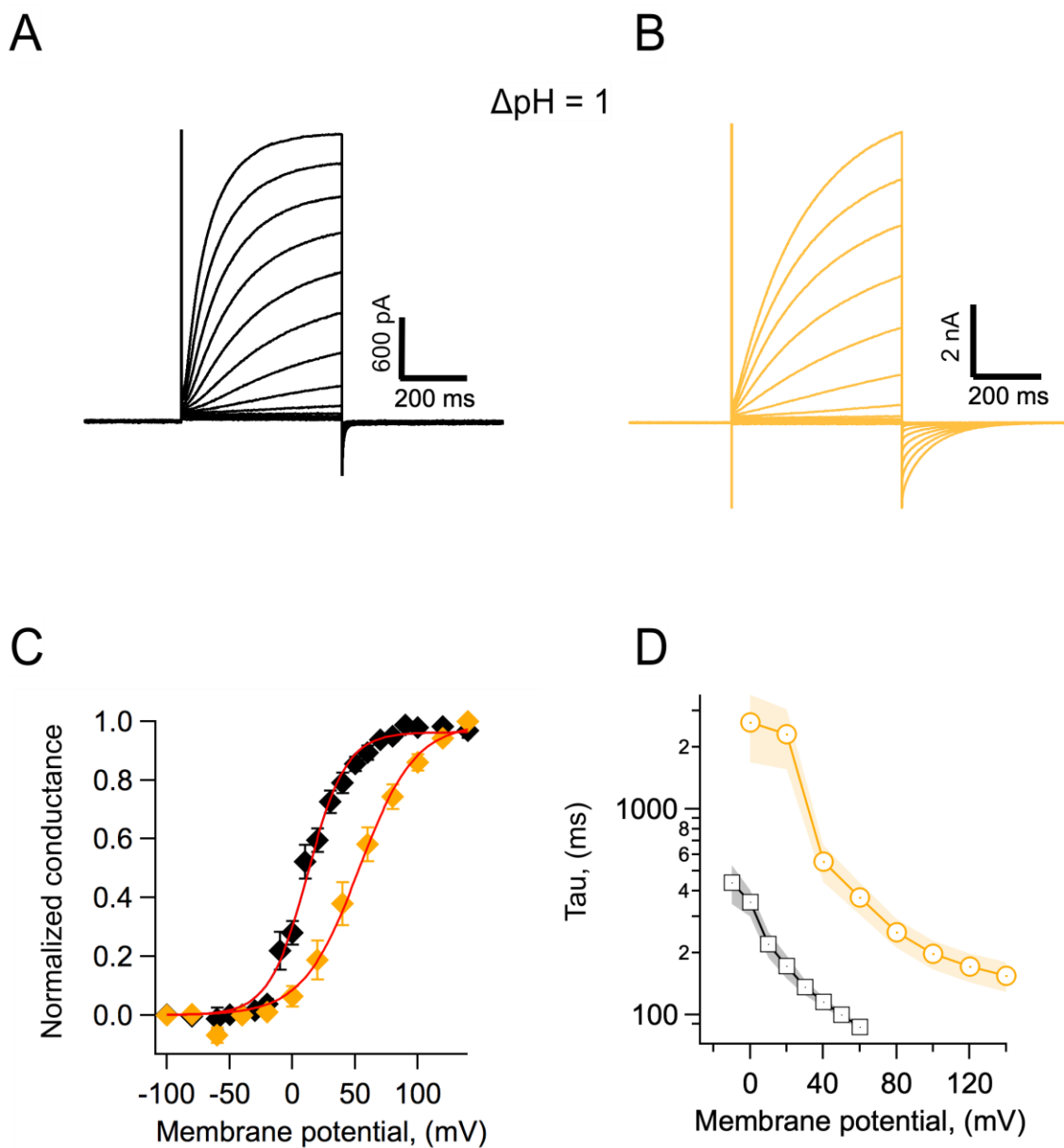


Figure 5

399
400 **Figure 5. Coral H_v1 channels are faster and activate more readily than their human**
401 **counterpart.** A) AmH_v1 currents in response to voltage-clamp pulses from -100 to 120 mV.
402 B) Currents through hH_v1 channels in response to voltage-clamp pulses from -100 to 120
403 mV. Recordings shown in A and B were obtained in whole-cell the configuration. C)
404 Comparison of the conductance-voltage relationship for both channels. Black diamonds are
405 the mean G/G_{\max} for AmH_v1 and yellow diamonds for hH_v1. The error bars are the s.e.m. (n=

406 3, for both channels). The continuous red curves are fits to equation 1. The fitted parameters
407 are: AmH_{v1} , $q = 1.62 e_0$, $V_{0.5} = 12.2$ mV; hH_{v1} , $q = 1.11 e_0$, $V_{0.5} = 53.1$ mV. D) The activation time
408 constant estimated from fits of currents to equation 2. Circles are the mean for hH_{v1} and
409 squares for AmH_{v1} . The shaded areas are the s.e.m. ($n = 3$, for both channels).

410

411 ***Effects of the pH gradient on gating***

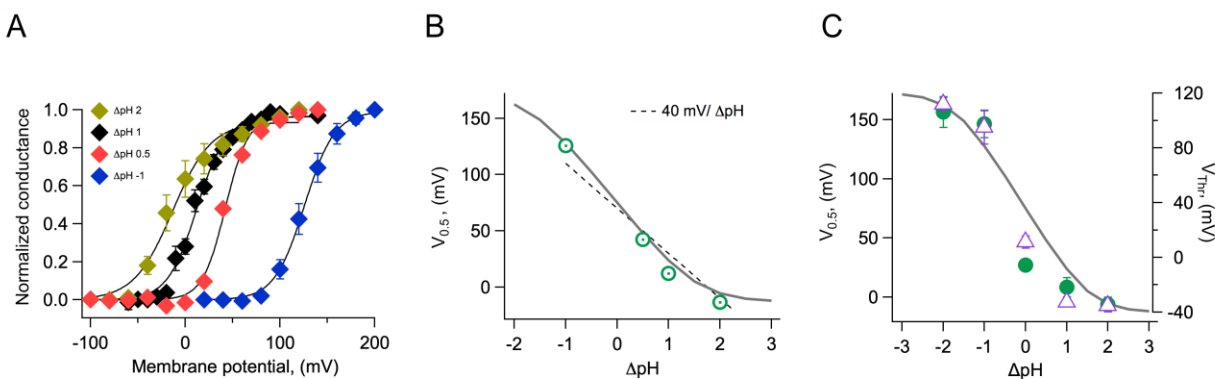
412 Both native and cloned voltage-gated proton channels are characteristically modulated by
413 the pH gradient (Cherny et al., 1995; Sasaki et al., 2006; Ramsey et al., 2006). We carried out
414 experiments to investigate the modulation of the coral H_{v1} channels by different pH
415 gradients. We first recorded whole-cell currents at various ΔpH and estimated the voltage-
416 dependence of the conductance. These G-V curves were fitted to equation 2 to obtain the
417 voltage of half activation, $V_{0.5}$ and apparent gating charge, q , that determines the steepness
418 of the fit. As is the case with other H_{v1} channels, the $V_{0.5}$ shifts to negative voltages when ΔpH
419 is greater than 0 and to positive voltages when $\Delta pH < 0$ (Figure 6A). When we plot the $V_{0.5}$ as
420 a function of ΔpH the relationship seems to be mostly linear over the range of ΔpH -1 to 2.
421 This relationship is somewhat steeper than the generally observed -40 mV/ ΔpH (Figure 6B).
422 We tried to obtain recordings over an extended range of ΔpH values. To this end, we
423 performed inside-out recordings in which the composition of solutions can be better
424 controlled, tend to be more stable and the size of currents is smaller. However, recordings
425 were unstable at extreme pH values and we only managed to reliably extend the data to ΔpH
426 of -2. Figure 6C shows the summary of the inside-out recordings. We have plotted both the
427 $V_{0.5}$ and the threshold voltage, V_{Thr} . To obtain this last parameter, we fitted the exponential
428 rise of the G-V curve to a function of the form:

$$429 \quad G(V) = G' \cdot \exp^{qV/K_B T}$$

430 V_{Thr} was calculated as the voltage at which the fit reaches 10 % of the maximum conductance.
431 The parameter V_{Thr} should be less sensitive than $V_{0.5}$ to the possible change in the proton
432 gradient that can occur with large currents. It is clear from these data that at extreme values
433 the dependence of $V_{0.5}$ or V_{Thr} on ΔpH deviates from a simple linear relationship and instead
434 it appears to saturate with increasing ΔpH .

435

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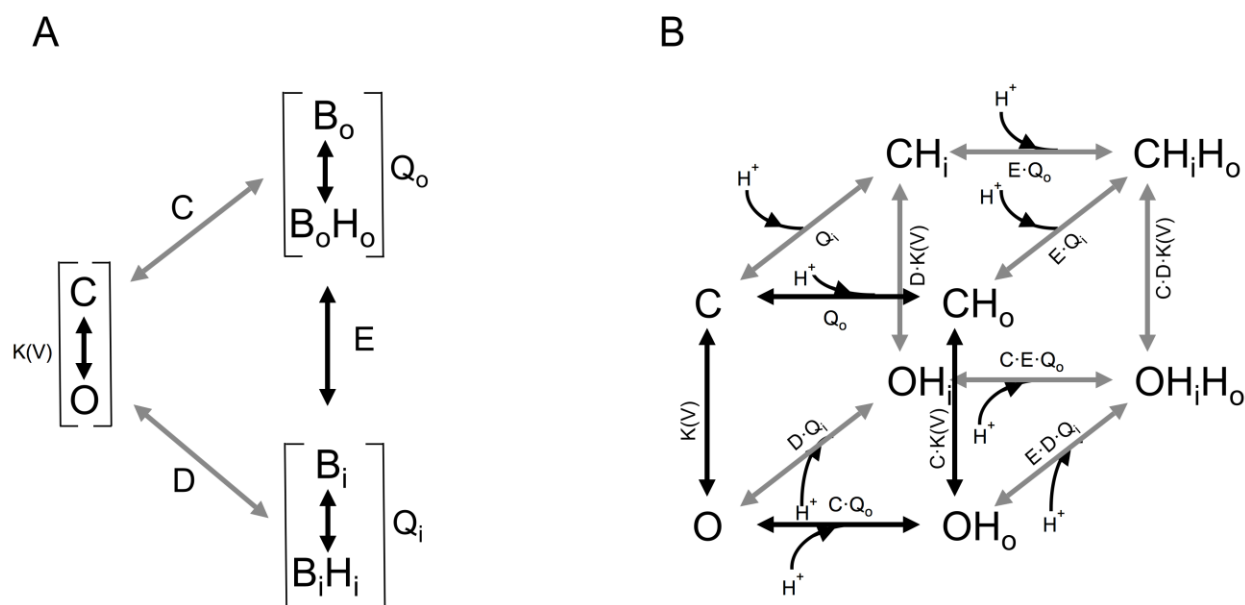
439 **Figure 6. Modulation of channel activation by the pH gradient.** A) Conductance vs.
440 voltage relationships obtained at the indicated ΔpH values, obtained from whole-cell
441 recordings of AmH_v1 proton currents. Continuous lines are fits to equation 1. B) The
442 parameter $V_{0.5}$ was obtained from the fits in A and is displayed as a function of ΔpH . The
443 dotted line is the 40 mV/ ΔpH linear relationship. The continuous grey curve is the prediction
444 of the allosteric model. C) Parameters $V_{0.5}$ (green circles) and V_{Thr} (purple triangles) obtained
445 from a different set of inside-out current recordings. Data are mean \pm s.e.m. The continuous
446 grey curve is the same prediction of the allosteric model shown in B. The model parameters
447 used to generate the theoretical curve are: $E=5 \times 10^5$, $D=10^5$, $C=0.0002$, $K_v(0)=0.00005$,
448 $q_g=1.0 e_0$, $pK_o=3.4$, $pK_i=7$.

449

450 ***Allosteric model of voltage and pH-dependent gating***

451 Currently, there are is only one quantitative model that has been used to explain ΔpH gating
452 of H_v1 channels (Cherny et al., 1995). However, this model is euristic and does not provide
453 mechanistic insight into the process of proton modulation of the voltage dependence of
454 proton permeable channels. In order to explain the modulation of the range of activation by
455 the proton gradient, parameterized by the $V_{0.5}$, we developed a structurally-inspired
456 allosteric model of voltage and proton activation. As many voltage-sensing domains, H_v1 has
457 two water-occupied cavities exposed to the extracellular and intracellular media (Ramsey et
458 al., 2010; Islas and Sigworth, 2001; Ahern and Horn, 2005). Recent evidence suggests that
459 these cavities function as proton-binding sites through networks of electrostatic interactions

460 (De La Rosa et al., 2018). In our model, we propose that these two proton-binding sites, one
461 intracellular and one extracellular, allosterically modulate the movement of the voltage-
462 sensing S4 segment and thus channel activation in opposite ways. The extracellular site is
463 postulated as inhibitory, while the intracellular site is excitatory, facilitating voltage sensor
464 movement. As a first approximation, we employ a simplified allosteric formalism based on a
465 Monod-Wyman-Changeux (MWC) style model (Horrigan and Aldrich, 2002; Changeux,
466 2012). As a simplifying assumption, in this model we assume that the voltage sensor moves
467 in a single voltage-dependent activation step. We assume the external and an internal
468 proton-binding sites have simple protonation given by a single pK_a value. These sites operate
469 as two allosteric modules and are coupled to the voltage sensor according to coupling factors
470 C and D, respectively. These binding sites in turn interact with each other through the
471 coupling factor E. The modular representations of the model are illustrated in Figure 7A,
472 while the full model depicting all open and closed states with all permissible transitions and
473 the corresponding equilibrium constants for each transition is shown in Figure 7B. Full
474 details of equations derived from these schemes are given in supplementary data.
475 This allosteric model represents a first attempt at producing a quantitative mechanistic
476 understanding of the interaction of the voltage sensor and protons in H_v1 channels.
477 From the data shown in Figure 6C, it can be seen that the model is capable of reproducing
478 the very steep dependence of $V_{0.5}$ on ΔpH and importantly, the saturation of this relationship
479 at extreme values. Some H_v1 channels from other organisms show a linear dependence of
480 gating over a large range of ΔpH values, while others show a reduced dependence and even
481 saturation over some range of ΔpH (Thomas et al., 2018). Our model can explain these
482 different behaviors as different channels having distinct values of pK_a s for the internal or
483 external sites, differences in coupling factors or differences in the voltage-dependent
484 parameters (Figure6 -Supplement 1).
485



486

487 **Figure 7. Gating scheme I.** A) Modular representation of a simple MWC model; the channel
 488 opening transition is voltage-dependent, with equilibrium constant $K(V)$. B_o and B_i are the
 489 unbound states of the extracellular and intracellular proton-binding sites, respectively and
 490 B_oH_o and B_iH_i are the proton bound states of these binding sites. Q_o and Q_i are equilibrium
 491 constants that depend on the pK_a of each of these binding states. C, D and E are the coupling
 492 constants between each of the indicated modules. B) All the individual states implied in A
 493 are depicted, along with proton-binding states and the appropriate equilibrium constants. C,
 494 closed states, O, open states. OH_x , OH_xH_x and CH_x , CH_xH_x are single or doubly proton-occupied
 495 states, where x can be o for outside or i for inside-facing binding sites.

496

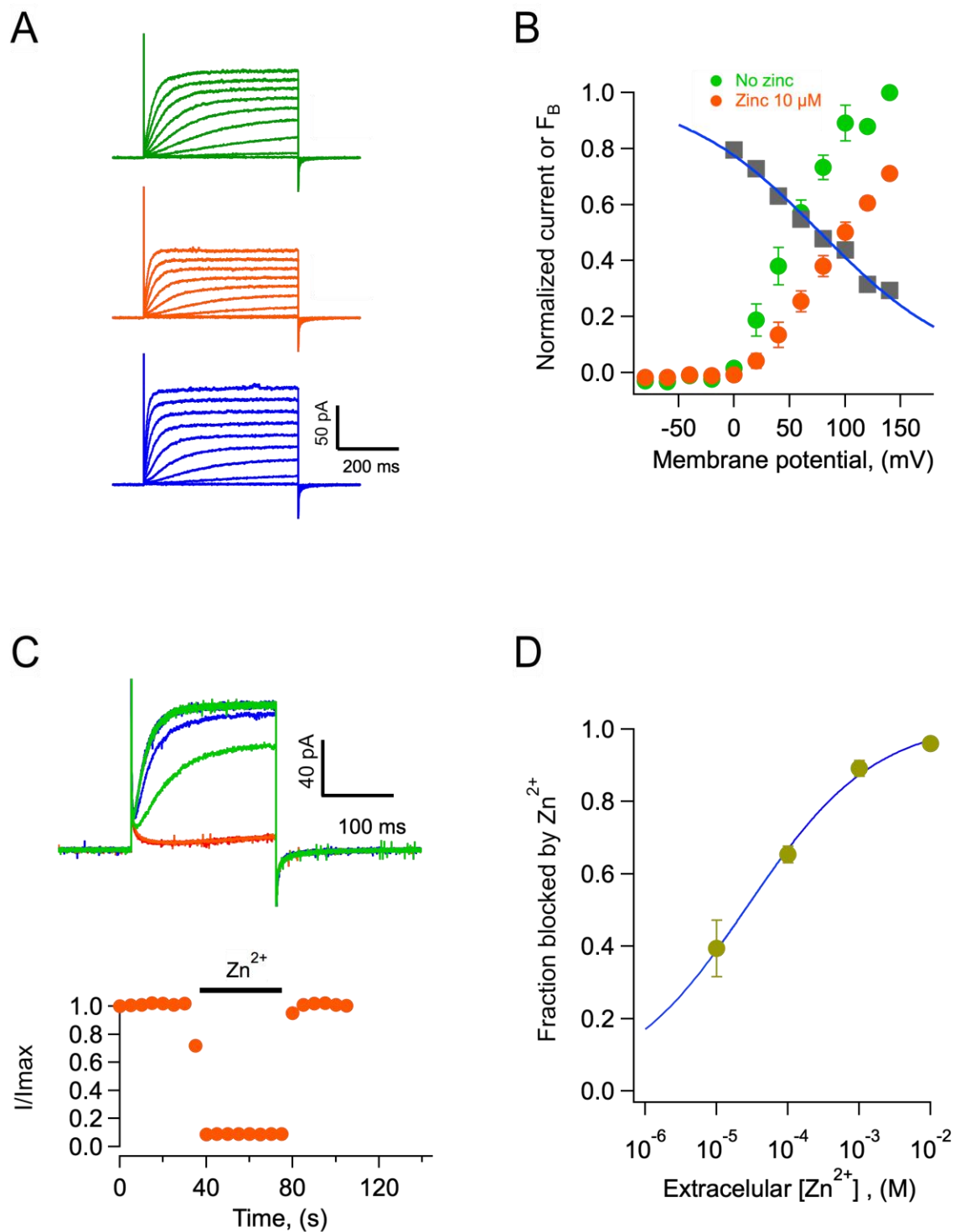
497

498 **Block by Zn^{2+}**

499 The best characterized blocker of proton channels is the divalent ion zinc (Cherny et al.,
 500 2020; De La Rosa et al., 2018; Qiu et al., 2016). We performed experiments to determine if
 501 Am H_v1 channels are also inhibited by zinc. We found that indeed, extracellular application
 502 of zinc in outside-out patches produced inhibition of the channels, reflected in reduced

503 current amplitude (Figure 8A). Figure 8B shows average current-voltage relationships in the
504 absence and presence of 10 μM external zinc. It can be seen that the fraction of current
505 blocked is not the same at every voltage, indicating that this inhibition might be voltage-
506 dependent. The fraction of blocked channels was calculated and is plotted at each voltage
507 along with the I-V curves (Figure 8B). It can be clearly seen that inhibition by Zn^{2+} is voltage-
508 dependent. A simple mechanism for voltage-dependent blockage was proposed by
509 (Woodhull, 1973). This model postulates that a charged blocker molecule interacts with a
510 binding site in the target molecule that is located within the electric field. Fitting the data
511 according to this model, and given that zinc is a divalent ion, its apparent binding site is
512 located at a fraction $\delta = 0.2$ of the membrane electric field from the extracellular side (Figure
513 8B).

514 Zinc blockage proceeds very fast. At 1 mM the channels are blocked almost instantaneously,
515 and the inhibition washes off very fast as well (Figure 8C). Finally, we report the dose
516 response curve (Figure 8D). The inhibition dose response curve can be fit by a Hill equation
517 (Equation 4) with a slope factor of near 0.5 and an apparent dissociation constant, K_D of 27
518 μM .



519

520 **Figure 8. Block of AmHv1 channels by extracellular zinc.** A) AmHv1-mediated currents
521 from an outside-out patch in the absence (top), presence of 10 μM zinc (middle) and after
522 washing of zinc (bottom). The scale bars apply to the three current families. B) Normalized
523 current voltage relationships before and in the presence of 10 μM zinc from 4 patches as in

524 A. The grey squares are the ratio $I_{\text{zinc}}(V)/I(V)$, which gives the voltage-dependence of the
525 blocking reaction. The blue curve is the fit to the Woodhull equation:

526
$$F_B = \frac{1}{1 + e^{-\delta z(V - V_{0.5})/K_B T}}$$
, where F_B is the fraction of current blocked, δ is the fraction of the

527 electric field where the blocker binds, z is the valence of the blocker, $V_{0.5}$ is the potential
528 where half of the current is blocked, K_B is Boltzmann's constant and T the temperature in
529 Kelvin. The fitting parameters are: $\delta=0.19$, $V_{0.5}=77.6$ mV. C) The effect of zinc is fast.
530 Application of 1 mM zinc to an outside-out patch produces almost instantaneous block of
531 ~90 % of the current. The effect also washes off quickly upon removal of zinc. Trace colors
532 are as in A. Voltage pulse was 100 mV applied every 5 sec. D) Dose-response curve of zinc
533 block of AmHv1 obtained at 100 mV. The continuous curve is a fit of the data to equation 4
534 with apparent $K_D = 27.4$ μM and $n = 0.48$.

535

536

537 **Discussion and conclusions**

538 A few ion transport mechanisms in reef-building corals have been described, but up to now,
539 no ion channels have been characterized from any scleractinian species. Here we show that
540 voltage-gated proton-permeable channels formed by the Hv1 protein are present in corals.
541 In particular, we have cloned these channels from two species of the genus *Acropora*, *A.*
542 *millepora* and *A. palmata*. It is interesting that the protein sequence of these proteins shows
543 a very high degree of conservation, suggesting that, even when the two species are found in
544 different oceans, they haven't had time to diverge substantially or alternatively, selective
545 pressures on these channels are very similar in both species. The presence of Hv1 sequences
546 in many other species of corals from disparate clades, suggest that Hv1 plays an important
547 role in coral physiology.

548 Hv1 channels are formed by a protein fold that is structurally equivalent to the voltage-
549 sensing domains (VSDs) of canonical voltage-gated channels (Sasaki et al., 2006; Ramsey et
550 al., 2006). The VSD is formed by a bundle of four antiparallel alpha helices (Takeshita et al.,
551 2014b). In some species, it has been shown that Hv1 channels are dimeric (Lee et al., 2008b;
552 Mony et al., 2020; Lee et al., 2008a). Accordingly, we also show here that the AmHv1 is a

553 dimer. Our FRET results are consistent with the high propensity to form a coiled-coil shown
554 by its C-terminal domain.

555 H_v1 channels are different from canonical voltage-gated channels in that both voltage-
556 sensing and permeation are mediated through a single protein domain. Voltage-sensing is
557 thought to occur through the interaction of charged amino acid side chains with the electric
558 field, leading to outward movement of the fourth domain or S4, in a similar fashion to other
559 voltage-sensing domains (Carmona et al., 2018; De La Rosa and Ramsey, 2018). This outward
560 movement of the S4 is coupled to protons moving through the VSD in a manner that is not
561 completely understood (Randolph et al., 2016). Most proton permeable channels seem to
562 have evolved to extrude protons from the cell, and towards this end, their voltage
563 dependence is tightly modulated by the proton gradient between extracellular and
564 intracellular solutions (Cherny et al., 1995).

565 Our electrophysiology experiments show that these coral channels give rise to proton
566 currents when expressed in HEK293 cells and that they retain the functional characteristics
567 that have been shown to define the class in other species, such as very high selectivity for
568 protons, activation by voltage and modulation of this activation by the proton gradient. The
569 new channels reported here activate faster than the human H_v1 channel. It has been known
570 that different orthologs of H_v1 activate with varying kinetics. For example, sea urchin,
571 dinoflagellate and recently, fungal H_v1 channels activate rapidly, while most mammalian
572 counterparts have slow activation rates (Musset et al., 2008; Smith et al., 2011; Zhao and
573 Tombola, 2021). A comparative study suggests that two amino acids in the S3
574 transmembrane segment are important determinants of kinetic differences between sea
575 urchin and mouse H_v1 (Sakata et al., 2016). The authors suggest that the time course of
576 activation is slow in channels containing a histidine and a phenylalanine at positions 164 and
577 166, respectively (mouse sequence numbering). The AmH_v1 has a histidine at equivalent
578 position 132 and a methionine at 134. It is possible that this last amino acid in AmH_v1 confers
579 most of the fast kinetics phenotype. A separate work showed that a lack of the amino-
580 terminal segment in human sperm H_v1 also produced fast-activating channels (Berger et al.,
581 2017). Interestingly, the *Acropora* channels have a shorter amino-terminal sequence, which
582 could also contribute to their fast kinetics.

583 One of the most interesting characteristics found in these new proton channels is their
584 modulation by the proton gradient. As opposed to other H_v1 channels, we can observe a
585 trend towards saturation of the $V_{0.5}$ for activation as a function of ΔpH at extreme values of
586 this variable. A tendency towards saturation of the $V_{0.5}$ - ΔpH relationship has been observed
587 in mutants of the hH_v1 channel (Cherny et al., 2015) or at negative values of ΔpH for a snail
588 H_v1 (Thomas et al., 2018), but it seems it can be fully appreciated in AmH_v1 . Since our model
589 explains the observation of saturation of voltage gating at extreme values of ΔpH as a
590 consequence of the existence of two saturable sites for proton binding, we attribute this
591 behavior, to the large separation of pK_a values for the extracellular and intracellular proton
592 binding sites.

593 The strength of allosteric coupling of these sites and the voltage sensor will determine if
594 saturation is observed over a short or extended range of ΔpH values and the range of values
595 of $V_{0.5}$ that a particular channel can visit. Our model should provide a framework to better
596 understand gating mechanisms in future work.

597 It is clear that more complicated models, with a larger number of voltage dependent and
598 independent steps (Villalba-Galea, 2014) and coupling to protonation sites should be the
599 next step to improve data fitting and explore voltage-and proton-dependent kinetics. In
600 particular, these types of models can help explain mutagenesis experiments exploring the
601 nature of the protonation sites.

602 H_v1 proton channels seem fundamental in handling fluctuations in intracellular pH and take
603 part in several well-characterized physiological processes that depend on proton
604 concentration changes, such as intracellular pH regulation, sperm flagellum beating, reactive
605 oxygen species production and bacterial killing in immune cells, initiation of
606 bioluminescence in single-celled algae, etc. (Castillo et al., 2015).

607 What is the function of voltage-gated proton channels in corals? The deposition of a $CaCO_3$
608 exoskeleton is one of the main defining characteristics of scleractinians, however, the ionic
609 transport mechanisms involved in this process are mostly unknown. In order for aragonite
610 precipitation to occur favorably, the pH of the calicoblastic fluid, right next to the skeleton is
611 maintained at high levels, between 8.5 and 9 and above the pH of sea water (Le Goff et al.,
612 2017). It has been posited that corals control this pH via vectorial transport of protons to the

613 gastrodermal cavity (Jokiel, 2013). Since proton transport away from the site of calcification
614 would incur a drastically lower intracellular pH in the cells of the aboral region, we propose
615 that, given their ability to rapidly regulate intracellular pH (De la Rosa et al., 2016), H_v1
616 proton channels contribute by transporting protons from the cells. Thus, these proton
617 channels would be a major component of the mechanisms of intracellular pH regulation.
618 Given that the activation range of H_v1 is controlled by the pH gradient, a large intracellular
619 acidification would facilitate opening of these channels at the resting potential of cells, which
620 is presumably negative.

621 The finding that coral H_v1 channels retain their sensitivity to Zn²⁺, opens the possibility of
622 using this ion as a pharmacological tool to study the role of proton channels in pH
623 homeostasis. It is interesting that a recent report has shown detrimental effects of zinc
624 supplementation in coral growth (Tijssen et al., 2017), a result that could be explained by
625 zinc inhibition of H_v1.

626 The physiological role of H_v1 channels in corals might be essential in the response of these
627 organisms to ocean acidification. We theorize that as the pH of sea water acidifies, gating of
628 H_v1 should require stronger depolarization, thus hindering its capacity to transport protons
629 from the cell. This will contribute to a diminished calcification rate and less aragonite
630 saturation of the CaCO₃ skeleton. It would be interesting and important to study the effects
631 of acidification on H_v1 physiology and pH regulation in corals in vivo. Essentially nothing is
632 known about the electrophysiological properties of coral cells. This report represents the
633 first time that an ion channel has been cloned and characterized in any coral and should open
634 a new avenue of research, such as uncovering the cellular and possible subcellular
635 localization of these channels and carefully measuring their physiological role in vivo.

636

637
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642 Author contributions. G.R-Y, performed cloning, performed heterologous expression,
643 performed experiments, read the paper. C.C., M.A.C.-R., E.S-D., performed cloning, expression

644 and electrophysiology and FRET experiments. L.D.I. obtained funding, conceived research,
645 procured animals, analyzed data, wrote the paper. A.B. and E.M. procured collection permits
646 and specimens, performed RNA extraction, revised and edited the paper. I.S.R. contributed
647 ideas, revised and edited the paper. T.R. analyzed data, wrote and edited the paper,
648 contributed ideas.

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814 **Table 1.**

Oligo name	Sequence
<u>AcHvNt5'</u>	ATGATTGATGCAAGAACCAGACGATCGAGCATGGATGAT
<u>AcHvNt3'</u>	TGATCCTGCTCTCAAGTCAAGAACCAACTCAGCAATGAC
<u>AcHvCt5'</u>	ATGGGATTCACATTTTCAAGCACAAATGGAGGTGTTT
<u>AcHvCt3'</u>	TCAGCTTTGTTTTAATGTTGTCAATTCAGACTCCAACCTG

815 Oligonucleotides used to clone amino and carboxy terminal partial sequences of AmH_v1 from
816 total reverse-transcribed mRNA from *A. millepora*.

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839 **Supplementary materials.**

A

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Ap MIDARTRRSSMDDQLPADELKEMGQPGTTAAVMASVCLTEDDTECPANKNHREKLNELL
Am MIDARTRRSSMDDQLPADELKEMGQPGTTAAVMASVCLTEDDTECPANKNHREKLNELL
*****

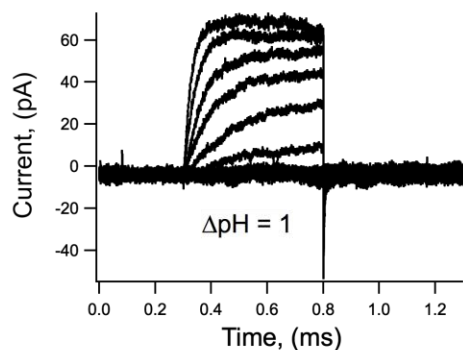
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Am HGDKIQYAIIVVLVIIDIIIVIAELVLDLRAGSEHHDNSASHVLHYISIAILSVFMIELL
*****

Ap KIYAMGFTFFKHKMEVFDGFVIIVSFALDIAFSNEQGGVDGVSLIVLLRLWRVTRIVNGI
Am KIYAMGFTFFKHKMEVFDGFVIIVSFALDIAFSNEQGGVDGVSLIVLLRLWRVTRIVNGI
*****

Ap ILSVQMRARERKVQAVTKENAELEQLKSKCAQLESELTTLKQS
Am ILSVQMRARERKVQAVTKENAELEQLKSKCAQLESELTTLKQS
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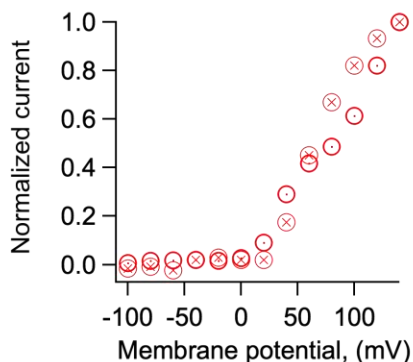
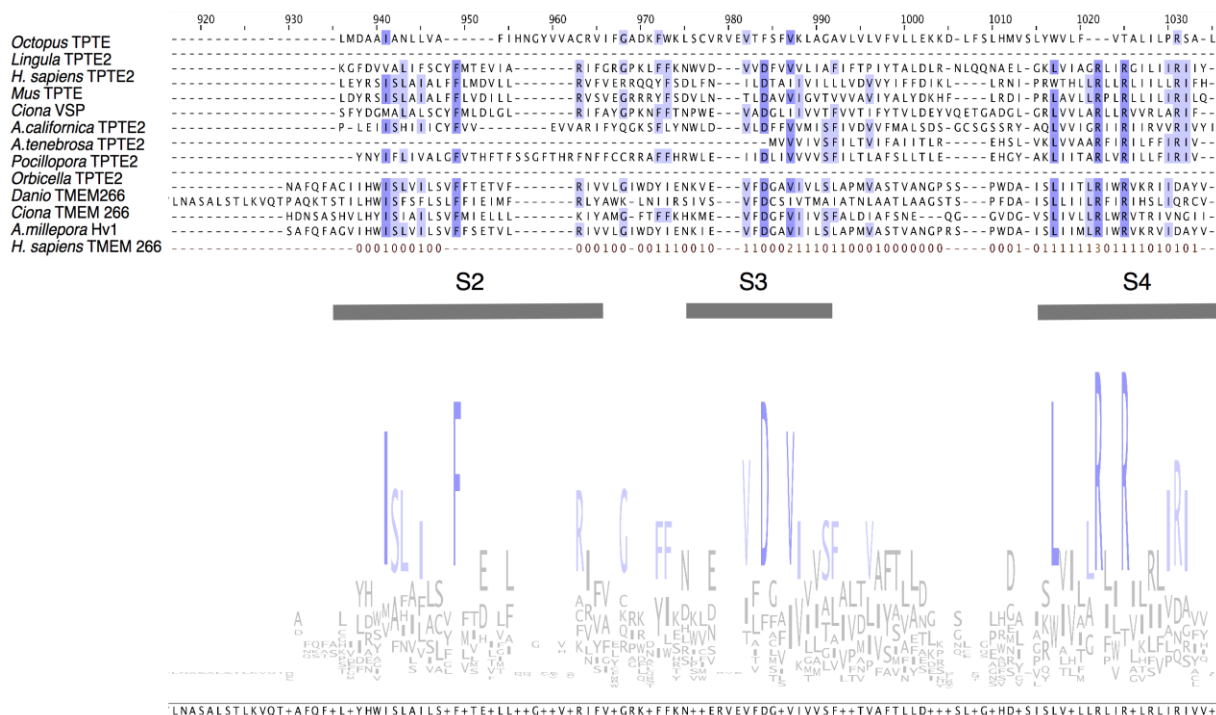
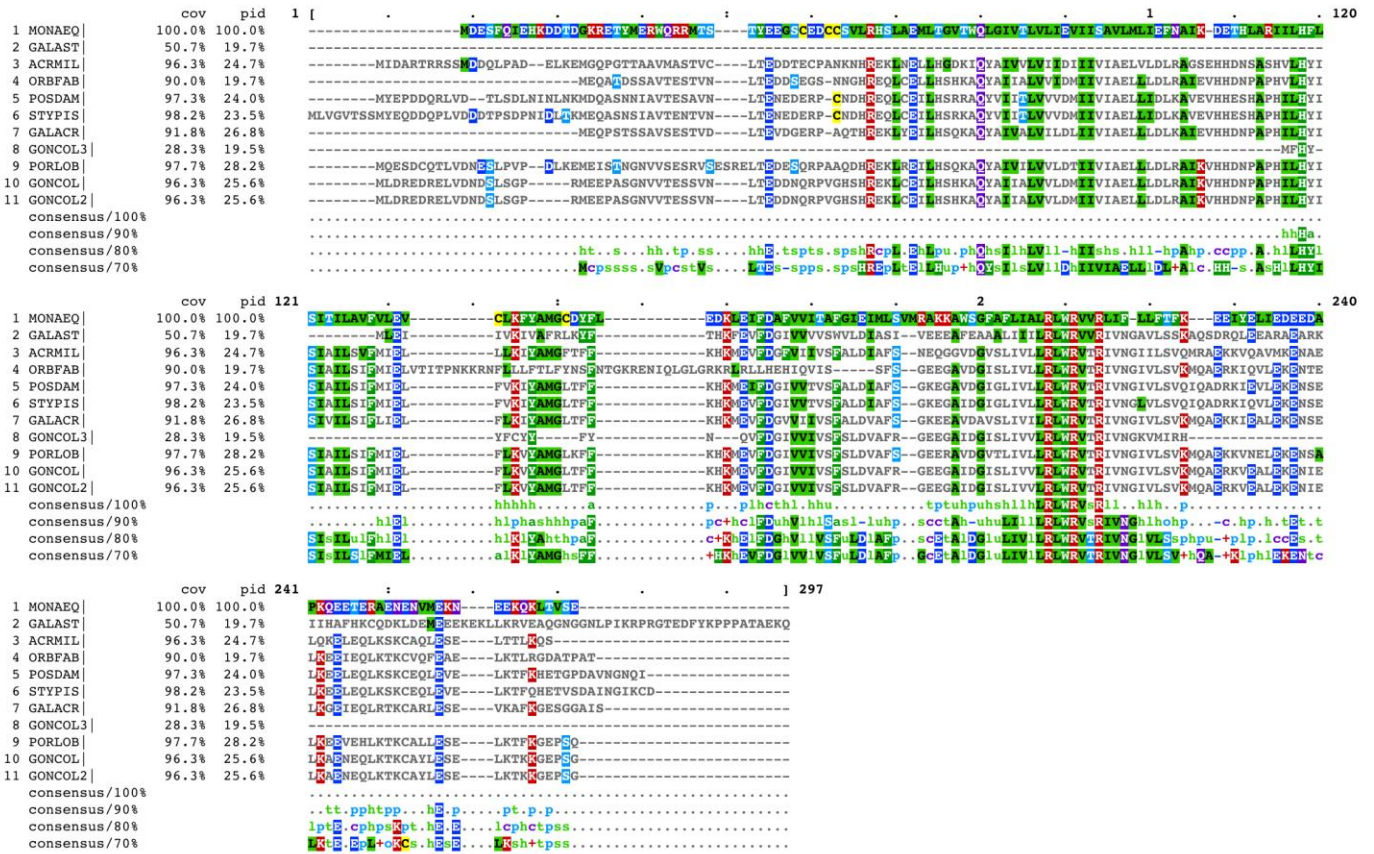


Figure 1- Supplement 1

840
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 842 **Figure 1- Supplement 1.** Some characteristics of the H_v1 from *Acropora palmata*. A)
 843 Comparison of the amino acid sequence of H_v1s from *Acropora millepora* (Am) and *Acropora*
 844 *palmata* (Ap). The asterisks bellow each residue indicate identity. B) Currents elicited from
 845 an inside-out patch obtained from a HEK293 cell expressing ApH_v1. Voltage pulses were
 846 from -100 to 140 mV in 20 mV steps. The ΔpH was 1. C) Normalized current-voltage
 847 relationships of two patches obtained as in B.



848
849 **Figure 1- Supplement 2.** Comparison of the sequence of AmHv1 to other voltage-sensing
850 proteins. Voltage-sensing phosphatases live CiVSP, TPTE and TPTE2 membrane proteins
851 contain a voltage-sensing domain (VSD). Other proteins such as TMEM 266 also contain
852 VSDs. Sequence similarity can be detected in the putative transmembrane domains. We show
853 the sequence alignment of the region with highest similarity between the selected sequences
854 of diverse organisms. The logo consensus sequence shows conservation, specially of the S4
855 segment. Other amino acid residues common in VSDs are also conserved in S2 and S3.
856 However, TPTE and TMEM266 proteins are very different from *Acropora millepora* Hv1.
857



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859

860 **Figure 2-Supplement 1.** Comparison of the AmH_v1 protein sequence with similar
 861 sequences found in other coral species. MONAEQ: *Montastrea*. GALAST: *Galastrea*. ACRMIL:
 862 *Acropora millepora*. ORBFAB: *Orbicela faveolata*. POSDAM: *Posillopora damicornis*. STYPIS:
 863 *Stylophora pistilata*. GALACR: *Galaxea*. GONCOL: *Goniopora*. PORLOB: *Porites lobata*.

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866 **Figure 6- Supplement 1. Model equations and simulations.**

867

868 The full complement of discrete states in our model is shown in Figure 7B. This allosteric
 869 model predicts that the open probability, $P(V, pH)$ is dependent on voltage and pH according
 870 to the following equations:

871

$$P(V, pH) = \frac{K_V \Omega}{K_V \Omega + \Gamma}$$

872

873 Where: $\Omega = 1 + Q_o C + Q_i D + Q_o Q_i iDEC$

874 $\Gamma = 1 + Q_o + Q_i + Q_o Q_i iE$

875 $K(V) = K(0) \cdot e^{(q_g V / K_B T)}$

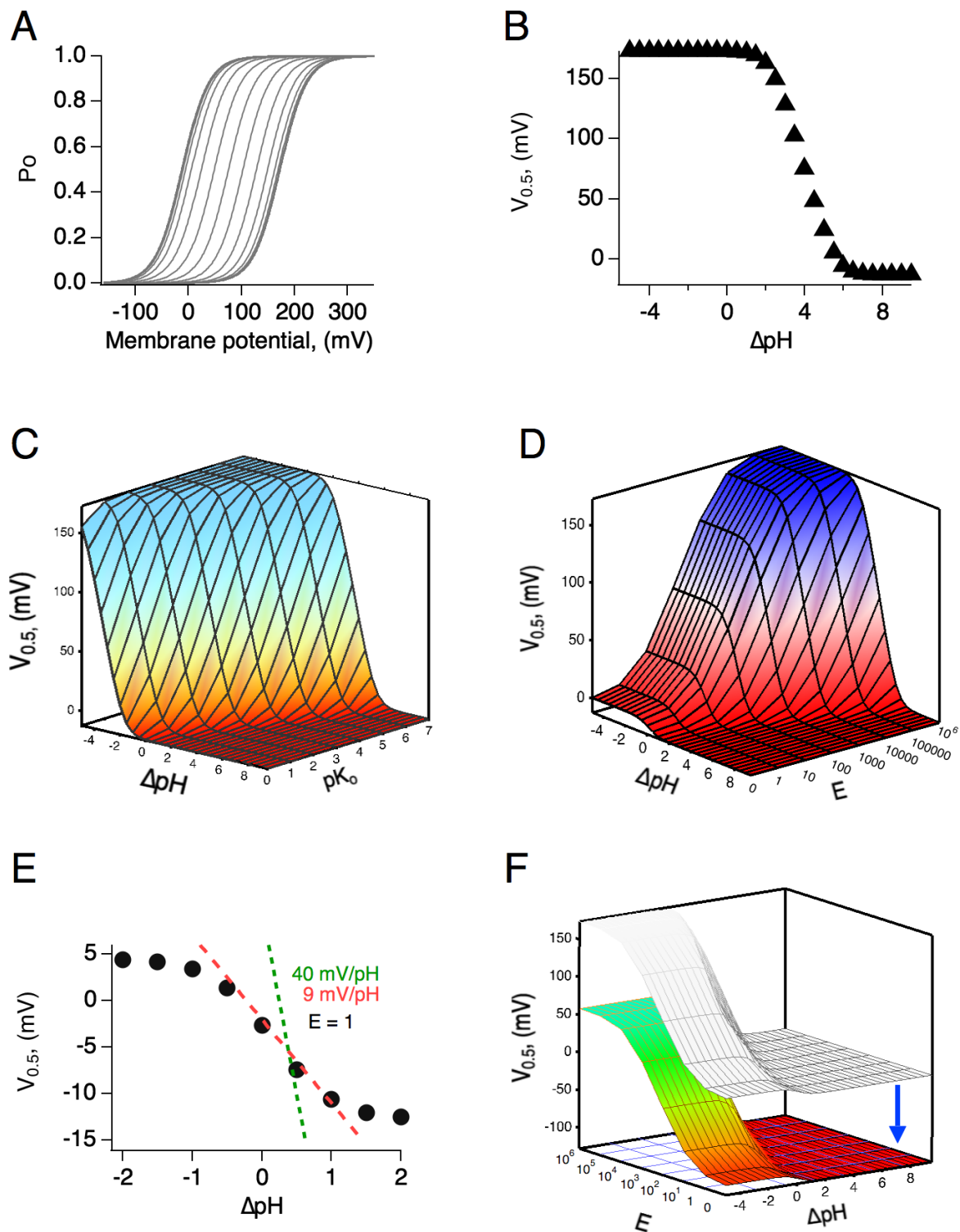
876 $Q_o = \frac{1}{1 + 10^{(pH_o - pK_o)}}$

877 $Q_i = \frac{1}{1 + 10^{(pH_i - pK_i)}}$

878 The voltage of half activation is given by:

879 $V_{0.5} = \frac{K_B T}{q_g} \cdot \ln \left(\frac{\Gamma}{K(0) \cdot \Omega} \right)$

880 pK_o and pK_i are the pK_a values of the extracellular and intracellular proton binding sites,
881 respectively.



882

883 **Supplementary data-Figure 1.** Simulations of the voltage- and pH-dependent behavior
884 predicted by the allosteric model. A) Calculated G-V curves and B) $V_{0.5}$ as a function of ΔpH .

885 Model parameters are: pK_o , $pK_i = 7$, $K(0) = 0.00005$, $q = 1 e_o$, $E = 10^6$, $C = 0.0002$, $D = 10^5$. C)
886 $V_{0.5}$ as a function of ΔpH calculated for different values of the pK_o . D) $V_{0.5}$ as a function of ΔpH
887 calculated for different values of the coupling factor E , which determines the allosteric
888 communication between external and internal protonation sites. Note that lack of coupling
889 between sites ($E = 1$), results in channels with very shallow modulation by pH , as illustrated
890 in E. All other parameters are as in A and B. E) A slice of the surface in D, with $E = 1$. The two
891 lines show the expected dependence of $V_{0.5}$ on ΔpH for a fully modulated channel ($E > 100$).
892 Also shown is the dependence of $9 \text{ mV}/\Delta pH$ unit. F) The range of $V_{0.5}$ is dependent on the
893 value of the voltage-dependent equilibrium constant at 0 mV . An increase to $K(0) = 0.005$
894 shifts the whole surface by approximately -110 mV . All other parameters are as in D.