Directional Proton Conductance in Bacteriorhodopsin Is Driven by

Concentration Gradient, Not Affinity Gradient

Zhong Ren

Department of Chemistry, University of Illinois at Chicago, Chicago, IL 60607, USA

Renz Research, Inc., Westmont, IL 60559, USA

zren@uic.edu

ORCID 0000-0001-7098-3127

Keywords

proton pump, retinal, serial crystallography, singular value decomposition, X-ray free

electron laser

Abstract

It is widely spread that microorganisms can harvest energy from sun light to establish

electrochemical potential across cell membrane by pumping protons outward. Light

driven proton pumping against a transmembrane gradient entails exquisite electronic

and conformational reconfiguration at fs to ms time scales. However, transient

molecular events along the photocycle of bacteriorhodopsin are difficult to comprehend

from noisy electron density maps obtained from multiple experiments when the

intermediate populations coexist and evolve as a function of 13 decades of time. Here I

report an in-depth meta-analysis of the recent time-resolved datasets collected by

several consortiums. This analysis deconvolutes the observed mixtures, thus

substantially improves the quality of the electron density maps, and provides a clear

visualization of the isolated intermediates from I to M. The primary photoproducts

revealed here suggest a proton transfer uphill against 15 pH units is accomplished by

the same physics that governs the tablecloth trick. While the Schiff base is displaced at

the beginning of the photoisomerization within ~30 fs, the proton stays due to its

inertia. This affinity-independent early deprotonation builds up a steep proton

concentration gradient that drives the directional proton conductance toward the

extracellular medium. This mechanism fundamentally deviates from the widely

adopted assumption based on equilibrium processes driven by light-induced changes of

proton affinities. The method of a numerical resolution of concurrent events from

mixed observations is also generally applicable.

Introduction

Bacteriorhodopsin (bR) captures photon energy to pump protons outward from the

cytoplasm (CP) against the concentration gradient, thus converts light into

electrochemical energy. This integral membrane protein of 28 kDa singlehandedly

achieves the exact same goal of photosynthesis and cellular respiration combined. This

simplicity is widely exploited in optogenetics and bioelectronics (Fenno et al., 2011; Li et

al., 2018). A trimeric form of bR on the native purple membrane shares the retinal

chromophore and the same protein fold of seven transmembrane helices A-G (Fig. S1)

with large families of microbial and animal rhodopsins (Ernst et al., 2014; Kandori,

2015). An all-*trans* retinal is covalently connected to Lys216 of helix G through a

protonated Schiff base (SB) pointing toward the extracellular side (EC) in the resting

state, of which the double bond $C_{15}=N_\zeta$ is also in *trans* (traditionally also noted as *anti*

(McCarty, 1970)) Upon absorption of a visible photon, the retinal isomerizes efficiently

and selectively to adopt the 13-*cis* configuration (Govindjee et al., 1990). Extensive

structural events during the ultrafast isomerization sampling and stereochemical

selection were recently revealed by the companion study of this work (Ren, 2021).

Photocycle

It has been shown that the species I or a collection of species prior to the

photoisomerization arises before 30 fs and remains in 13-*trans* instead of a near 90°

configuration about $C_{13}=C_{14}$ (Zhong et al., 1996). A red-shifted intermediate species J in

13-*cis* forms around 450-500 fs and decays at 3 ps to a less red-shifted species K

(Applebury et al., 1978; van den Berg et al., 1990), which lasts longer than five decades

of time therefore becomes the most "stable" intermediate throughout the entire

photocycle on a log-time scale. The blue-shifted L state emerges from K at 1-2 μs and

converts to strongly blue-shifted M states at ~50 μs (Lozier et al., 1975). The L → M

transition has been considered as the step for the proton acceptor, the carboxylate group

of Asp85, to become neutralized by receiving a proton (Fahmy et al., 1992). This

established view has been largely based on peak assignments in resonance Raman and

Fourier transform infrared spectroscopy (FTIR) on wildtype bR and various mutants

(Braiman et al., 1988, 1991; Gerwert et al., 1990; Lewis et al., 1974). However, the event

of SB deprotonation has been mistakenly inferred from the observed protonation of the

carboxylate of Asp85. It has been widely quoted in the literature that the proton

transfer from the SB to the proton accepter occurs as the M formation. This study finds

that this proton transfer spans ten decades in time instead of a single event. The

electron density maps of I to M are unscrambled from one another in this analysis.

They clearly show extensive conformational changes leading to and developing from

the isomerization of the chromophore (Fig. 1).

Structure

Although the first low resolution map of bR was revealed in 1975 (Henderson and

Unwin, 1975), the majority of structural information awaited technology advances until

the 1990s in cryo electron microscopy (Henderson et al., 1990), electron diffraction

(Grigorieff et al., 1996), and X-ray crystallography (Pebay-Peyroula et al., 1997). Finally,

the atomic resolution of a bR model was achieved (Luecke et al., 1999a). The main

technique to study intermediates along the photocycle used to be cryo trapping,

including both light illumination of crystals at room temperature followed by rapid

freezing and illumination at elevated cryo temperatures. The 13-*cis* retinal and its

associated H-bond network have been observed in M state with extensive

conformational changes in the anchor helix G (Luecke et al., 1999b), FG loop, EC half of

helix C (Takeda et al., 2004), and an enlarged CP pocket (Sass et al., 2000). Numerous

attempts have been made to capture intermediates other than M with inconsistent

results (Wickstrand et al., 2015).

A new phase began with time-resolved serial crystallography conducted at X-ray

free electron lasers (XFELs). The ultrashort X-ray laser pulses open the opportunity to

capture transient structural species in the photocycle as short-lived as fs. It is equally

important that serial crystallography enables photo triggering and data collection at

room temperature. Authentic signals of conformational changes of greater amplitudes

are expected at room temperature compared to cryo trapping. Several international

consortiums carried out large operations and accumulated abundant time points along

the photocycle that spans more than 13 decades of time. Judged only from the

extensiveness of the reported signals, time-resolved XFEL data at room temperature

unfortunately do not surpass those captured by cryo trapping, which hints at much

needed improvements in experimental protocols and data analysis methods. Nogly et

al. captured retinal isomerization to 13-*cis* at 10 ps (Nogly et al., 2018). More

importantly, they observed that the SB water moves before the isomerization and the SB

itself has already displaced at the earliest time point of 49-406 fs, which is the key

contribution to this analysis. Kovacs et al. contributed datasets at many short time

delays (Kovacs et al., 2019). Nango et al. found an ordered water in a newly established

H-bond network involving the SB in the 13-*cis* retinal pointing to the CP on ns-μs time

scale, which led to their conclusion of the proton path (Nango et al., 2016). These are

three major sources of data analyzed in this study (Table S1). Several synchrotron

datasets at ms delays also contribute to this analysis (Weinert et al., 2019).

Structural heterogeneity is the common difficulty to both cryo trapping and time-

resolved serial crystallography at room temperature. To what extend a specific species

can be enriched in crystals depends on the reaction kinetics under a condition of

illumination, such as wavelength and temperature. A more stable species K or M can

reach a higher fractional concentration at a specific time point due to a greater ratio

between the rates going into and exiting from that species. Others will be populated

poorly because of their transient nature. This structural heterogeneity causes the

practical difficulties in the interpretation of electron density maps and the refinement of

these intermediate structures (Ren et al., 2013). An assumption in nearly all previous

studies has been that each dataset, at a cryo temperature or at a time delay, is derived

from a mixture of a single photoinduced species and the ground state. Therefore, the

difference map reveals a pure intermediate structure. This assumption leads to the

more time points the more overinterpretation rather than the more overdetermination.

Previous studies using cryo trapping and time-resolved crystallography could not

evaluate clear electron density maps of pure intermediate species without a proper

deconvolution of heterogeneous species. Because none of the experimental conditions

at a cryo temperature or a time point can guarantee an experimental map that does not

involve heterogeneous mixture (Ren et al., 2013; Yang et al., 2011). This work is yet

another case study to test our general strategy of dynamic crystallography, that is, from

the outset every observation at a time point or a temperature setting is treated as a

mixture of unknown number of structural species with unknown populations

(Methods). The structures of all intermediate species are also unknown except the

ground state. The analytical protocol is responsible for a reliable structural

interpretation by overdetermination of these unknowns from consistent observations

(Ren, 2019; Ren et al., 2013; Yang et al., 2011). Inconsistent observations are identified

and isolated.

Mechanism

Understanding the mechanism of proton pumping uphill against gradient has been

centered on two entangled aspects: Where are the proton uptake and release pathways?

How does the retinal photoisomerization adjust the chemical groups' ability to hold

proton along these pathways? The ability of a chemical group to hold a proton is

measured by its value of pK_a, where p stands for $-\log_{10}$ function and K_a is the acid

dissociation constant, one form of the equilibrium constant of a chemical reaction. It is

clear that a proton previously retained by the SB is accepted by the carboxylate group of

Asp85 in helix C and eventually released to the EC medium. Each of Arg82, Asp85, 212,

Glu194, 204, and some waters could play a role (Balashov et al., 1997; Govindjee et al.,

1996; Richter et al., 1996). However, the sequence of events during this proton

conductance is unclear thus often mistakenly inferred.

At the resting state of bR, the SB holds a proton tightly with a pK_a of 13.3 pointing

toward the EC channel (Druckmann et al., 1982; Sheves et al., 1986). The nearby proton

acceptor Asp85 bridged to the SB by a water stays quite acidic with its pK_a spanning 2-3

range (Balashov et al., 1996). That water, often numbered as 402, has been identified as

the centerpiece of the mechanism (Wickstrand et al., 2015). Therefore, the SB proton is

associated with both N and O simultaneously in N_ζ:H^+:OH_2 with both interactions much

weaker than an ordinary single bond (Fig. 2c). One of the competing hypotheses is

willing to consider that the proton is first brought to the CP side by the SB

isomerization to 13-*cis*, and then conducted through Thr89 and a newly found water

other than Wat402 on the CP side, and finally reaches Asp85 (Nango et al., 2016). The

difficulty encountered by this hypothesis is the rollercoasting pK_a values that the proton

has to go through. The other competing hypothesis relies on a H-bond network long

after photoisomerization but largely similar to that in the resting state to translocate the

proton on the SB. The H-bond from the SB to Wat402 has to be reconciled with two

double bonds $C_{13}=C_{14}$ and $C_{15}=N_\zeta$, both highly twisted away from *trans* and *cis*

configurations to near 90°, until a successful proton translocation to Asp85 (Lanyi and

Schobert, 2007). This interpretation of these highly twisted double bonds is a direct

consequence of the inability to unscramble a mixed observation of multiple

conformations.

Despite intense studies, fundamental questions on the operating mechanism of this

proton pump remain puzzling and some unanswered. Q1) How does the protonated

SB with a pK_a of 13.3 transfer its proton to the acidic acceptors and when? Q2) Which

molecular event, the isomerization, or another event, switches the accessibility of the SB

from EC to CP and how does the SB avoid reprotonation on the wrong side? Q3) How

are protons conducted > 15 Å through the rollercoasting pK_a values and released into

the EC medium from the SB of a high pK_a through aspartic acids of low pK_a, arginine

(strictly speaking, its guanidinium ion) and tyrosines of high pK_a, and glutamic acids of

low pK_a again? Q4) How is a gradient-driven proton backflow prevented in the resting

state and during the pumping motions? No convincing evidence can directly

demonstrate a photoinduced reversal of the drastically different proton holding abilities

along the EC half channel to facilitate a spontaneous proton conductance that would

require an ascending order of pK_a's. The fundamental flaw in the previous line of

thinking is multiple equilibrium steps driven by proton affinities as the values of pK_a

describe. A proton is spontaneously transferred from one chemical group with a less

proton affinity, that is, smaller pK_a, to another with a greater affinity, that is, larger pK_a

(Jardetzky, 1966; Stoeckenius et al., 1979). Such proton conductance was presumed to

be accomplished by the means of abundant thermal events of protonation and

deprotonation, that is, equilibrium. If a slope of affinity does not satisfy the directional

flow of protons, it has been commonly assumed that a photoinduced reversal of proton

affinity must have occurred during the photocycle (Stoeckenius, 1999). This study

shows that equilibrium is not how the light-driven proton pump bR works, and the

directional proton conductance is not driven by increasing affinities but by a

concentration gradient of protons.

Here the ultrafast molecular events revealed from the serial crystallographic

datasets demonstrate that the photochemical reaction of bR transfers a proton held by

the SB to form a strong acid that barely has any ability to hold a proton. However, this

proton transfer is not achieved by equilibrium therefore not governed by pK_a's. Instead,

before the isomerization occurs, this proton transfer takes place due to the SB

displacement within ~30 fs in a single molecular event. The proton's inertia prevents it

from accelerating as fast as the SB under the same physics of the tablecloth trick (Q1).

Rapid photoisomerization flips the already deprotonated SB at 500 fs, thus protects the

SB from being reprotonated on the wrong side (Q2). Proton conductance follows a

descending order of the proton concentration gradient established in the EC half

channel (Q3). A built-in seal constructed by an irregular helix reacts to the

isomerization and ensures no proton leakage back to the CP channel during the motions

of proton pumping (Q4). The inner EC channel is strictly surrounded by chemical

groups of high pK_a's so that protons can only flow out but cannot flow back in even

during dark (Q4). See Supplementary Information (SI) for coherent answers to these

questions.

Results and Discussion

Results and Discussion

A total of 42 datasets and 36 time points analyzed in this study are divided into two

groups: 18 short delays up to 10 ps and 18 long delays ranging from ns to ms, which

unfortunately leaves a gap longer than three decades not surveyed (Fig. 1 and Table S1).

Difference Fourier maps among these time points and with respect to their

corresponding dark datasets are calculated according to protocols previously described

(Methods). Singular value decomposition (SVD) of difference maps and the subsequent

Ren rotation in a multi-dimensional Euclidean space established by SVD (Ren, 2016,

2019, 2021) are performed separately to 126 difference maps of the short delays and 101

maps of the long delays (Methods). This method of "decomposition and

deconvolution" results in a clean separation of electron density changes due to

coexisting photoexcited species mixed in the observed datasets. The main findings are

summarized in a Gantt chart (Fig. 1). Eight intermediate structures along the

photocycle, now isolated from one another, are refined against reconstituted structure

factor amplitudes (Methods; Table S2) (Ren, 2021). The assignments of the previously

identified intermediates I to M to these refined structures are largely based on the time

stamps of the observed signals. No absorption signatures are directly associated with

the X-ray data.

Intermediates I', I, and Schiff base deprotonation

Three decomposed map components U_{10}, U_{14}, and U_{17} of the short delays contain

extraordinary structural signals in terms of their extensiveness and quality (Ren, 2021).

These signals originate exclusively from a few time points of Nogly et al., too few to fit

the time dependency with exponentials. Instead, a spline fitting through these time

points estimates the coefficients c_{10}, c_{14}, and c_{17} in the linear combination of $c_{10}\boldsymbol{U}_{10}$ + $c_{14}\boldsymbol{U}_{14}$

+ $c_{17}\boldsymbol{U}_{17}$ for the early states I, J, and their respective precursors I′, J′ (Ren, 2021). The

precursor prior to I state is located on the spline trajectory from the origin, that is, the

ground state bR at the time point of 0- before the photon absorption, to the first time

point of 49-406 fs (PDB entry 6g7i). It would be appropriate to name this location on the

trajectory I′ as a precursor leading to I state judged only by the time point at ~30 fs.

The all-*trans* retinal in bR is largely flat except C_{15} (Fig. 2b 2nd panel). However, two

single bonds of Lys216 are highly twisted to near 90° (Fig. 2b 4th panel), which forms a

corner at C_ε that deviates from the flat plane of the all-*trans* retinal, protrudes inboard,

and makes a van der Waals contact at 3.4 Å with Thr89 in helix C (Fig. 2b 2nd panel).

This contact is identified in this study as a seal between the extracellular (EC) and

cytoplasmic (CP) half channels. The retinal in I′ remains in near perfect all-*trans*

configuration, including the Schiff base (SB) double bond $C_{15}=N_\zeta$ (Fig. S2), while it is

creased into an S-shape in an expanded retinal binding pocket together with other

features revealing the isomerization sampling (Ren, 2021). The slit between Thr89 and

the SB increases to 4.3 Å that may break the seal between two half channels, which is

inconsequential as argued below. The C_{20} methyl group swings 1.1 Å to the outboard

direction, a motion previously known but at much later time. These motions are

rotations around the long axis of the polyene chain instead of translations (Fig. 2b

bottom panel). It is important that the displaced Wat402, now a hydronium ion H_3O^+

(see below), remains H-bonded to both Asp residues in I′. However, its association

with the SB increases to 3.3 Å that barely qualifies as a H-bond.

Due to the availability of the time stamp, a conformational change captured in a

time-resolved experiment can be converted to the velocity of an atomic displacement

that would further hint the acceleration and force required to achieve the observed

conformation changes at the time of the measurement. It can be shown that a force >

500 pN is required to keep the proton accelerating as fast as N_ζ that has been displaced

by 0.7 Å within 30 fs. This is not achievable given the charge on the proton and its H-

bond association with Wat402. Compared to the regular single bond N-H in an amine

group at a pK_a of 40, the coordination bond N_ζ:H^+ in the protonated SB at its pK_a of 13.3

is much weaker. In addition, a proton acceleration with the SB would have to break its

interaction with Wat402, nearly as costly as to break the coordination bond N_ζ:H^+ (Fig.

2c). Therefore, which of the two weak interactions N_ζ:H^+:O402 survives the abrupt tare

before the photoisomerization is determined by the inertia of the proton. I' captures the

starting moment of a reliable deprotonation from the SB that does not depend on

thermal equilibrium. Here the 0.6 Å increase from the SB to Wat402 is attributed to the

separation of the proton from the SB rather than from Wat402. Instead of accelerating

with the SB, the proton stays with Wat402 to form a hydronium H_3O^+ due to its inertia

(Q1; Figs. 1 inset and 2c). Since several water molecules have been observed in the

inner EC channel, possibilities of other proton complexes also exist, such as $H_5O_2^+$ (Marx

et al., 1999; Mathias and Marx, 2007). The hydronium 402 has been shown possible by

magic angle spinning NMR (Friedrich et al., 2020).

This conjecture here, however, is not based on an observation of the scattered X-rays

by the proton as the proton does not scatter X-ray. Direct evidence supporting this

water cluster with excess protons reside in FTIR observations. Complex kinetic

behaviors of several broad band continua at the frequencies > 1700 cm^{-1} were recorded

in FTIR experiments and interpreted as protonated water clusters (Garczarek et al.,

2004, 2005; Lorenz-Fonfria et al., 2017; Wang and El-Sayed, 2001). However, detailed

assignments of these continua remain debatable. For example, the removal of one

terminal carboxylate near the exit of the EC channel (Fig. S1) in the mutant E204Q

certainly could affect proton release. But the observed change in some FTIR continua

cannot be easily pinpointed to a water cluster in the outer EC channel (Garczarek et al.,

2004), since shutting down one proton release pathway could affect the kinetics of the

upstream water cluster in the inner EC channel. Furthermore, the FTIR continua most

likely originate from a super-acidic water cluster in the inner EC channel because a

well-known proton barrier surrounds the inner EC channel but no proton barrier for the

outer EC channel (see below).

The hydronium 402 is likely short-lived. Either Asp85 or 212 "quickly" accepts

another proton from the hydronium and returns it back to Wat402 (Fig. 1 inset). The

lifetime of hydronium 402 is unclear from X-ray diffraction. Nevertheless, this lifetime

does not have to contradict the commonly adopted view that these aspartic acids are

protonated later in tens of μs to ms during M formation (Braiman et al., 1988, 1991;

Gerwert et al., 1990), given that the pK_a difference between the aspartic acid and

hydronium is < 4 pH units. As argued below, a super-acidic water cluster harboring
excess protons in the inner EC channel must be established before proton release.
Reliable deprotonations from the SB in consecutive photocycles continuously
overcharge the inner EC channel. The deprotonation event cannot be directly inferred
from the vibrational signatures due to the protonation of the carboxylic proton
acceptors in the Asp residues. In other words, the "proton transfer" from the SB to the
proton acceptor, as commonly stated in the literature, is not a single event. Instead, it
spans ten decades in time.
314
The retinal in I state remains in near perfect all-*trans*, including the SB. However,
the major difference from the precursor is that the single bond N_ζ-C_ε is now in a perfect
syn conformation, and the anchor Lys216 has largely returned to its resting
conformation (Figs. 2b 4th panel and S3). The SB makes a sharp U-turn toward inboard
before 500 fs (Fig. S2) and closes the slit with Thr89 to 3.8 Å and ends the breakage in
the seal that is too brief to result in a proton leakage (Q3). After the U-turn, the
deprotonated SB is pointing toward inboard while Wat402, currently a hydronium, is
displaced outboard but remains H-bonded to both Asp85 and 212. The SB and Wat402
are not only beyond a H-bond distance, but also in a bad geometry for proton exchange
(Fig. S3d). The accessibility of the SB in I state is already very limited while the SB is
reoriented toward inboard despite that the isomerization is yet to occur (Q2). This
ultrafast U-turn is how the SB gets rid of the proton by a mechanism of the same
physics that governs the tablecloth trick (Q1; Fig. S2). Holding Wat402 in place by
Asp85 and 212 symmetrically from inboard and outboard, respectively, is as important
as the ultrafast shaking of the SB to its successful deprotonation. See SI for mutant
functions when the symmetric holding of Wat402 is disrupted. Here the tablecloth trick
is not an analogy to the SB deprotonation. Both are governed by the exact same
Newton mechanics: A certain mass, such as a planetary body, a dinning set, or a
proton, does not accelerate fast enough unless a sufficient force is exerted on it. The U-
turn of the SB marks the peak power, or the maximum rate of enthalpy change dH/dt,
of the entire photochemical reaction (see Discussion). The spectrogram derived from
the transient absorption or transmittance clearly supports that the photocycle reaches its
peak power within the first tens of fs (Kahan et al., 2007; Kobayashi et al., 2001). If a
deprotonation is unsuccessful at the peak power of the photocycle, there will be no
further opportunity to translocate a proton from a fair base to a decent acid uphill

against 11 pH units under equilibrium. The same is true that a proton transfer is
energetically impossible from the protonated SB at a pK_a of 13.3 to form a hydronium, a
strong acid of a pK_a of -1.7, under equilibrium. The deprotonated SB is extremely
vulnerable at this moment in I state despite its limited accessibility, as it will be quickly
reprotonated if it lingers on in the EC channel to allow an equilibrium. As shown
below, the isomerization to a 13-*cis* retinal prevents such an equilibrium in a futile
photocycle. However, some of the photocycles are indeed unsuccessful as its quantum
yield indicates if a reprotonation event does occur before the isomerization.
348
349   Incidentally, all previous hypotheses on the mechanism of SB deprotonation imply a
350   formation of hydronium at one point or another (Lanyi and Schobert, 2007; Nango et al.,
351   2016). It would require a donor group with a pK_a more negative than -1.7 to reliably
352   form a hydronium by the means of equilibrium. None of the previous hypotheses have
353   successfully explained how this is possible.
354
Intermediates J', J and traveling water or hydronium 402
Near perfect 13-*cis* is successfully refined in both structures of J' at ~700 fs and J at ~20
ps with a contracted retinal binding pocket (Fig. 2b 4th panel) (Ren, 2021). However, the
SB double bond $C_{15}=N_\zeta$ is momentarily distorted from the *trans* configuration in J' with
a torsion angle of 133°. The *trans* double bond $C_{15}=N_\zeta$ is promptly restored in J (Fig. 2b
4th panel). The SB N_ζ is rotating clockwise looking from the proximal to distal direction
in the entire process of the isomerization I' → I → J' → J (Fig. 2a). If the deprotonated
SB isomerizes to 13-*cis* at ~500 fs, this photocycle is productive. Otherwise, a photocycle
could be wasted if reprotonation occurs prior to the isomerization as it could happen
sometimes even in wildtype. The isomerization completes the accessibility switch for
the SB from the EC half channel to the CP (Q2).
366
Due to the 13-*cis* configuration, C_{20} methyl group protrudes to a CP direction that is
further tilted toward outboard from its original direction. No additional consequence is
immediately obvious until the retinal flattens and returns to its original plane at later
intermediates. At this moment, Trp182 and the retinal anchor Lys216 stay in their
resting positions. Wat402, perhaps currently still a hydronium ion, is well defined in
both Js (Ren, 2021). Contrary to the previous conclusion (Wickstrand et al., 2015), this
traveling water is never disordered throughout the surveyed portion of the photocycle

up to M_2 (see below) despite its large trajectory spanning greater than 3 Å (Fig. 2a).

Wat402 is observed with both strong difference electron density and its refined electron

density. Its association with Asp85 and 212 remains unwavering throughout the

photocycle except one increase to 4.1 Å from Asp212 during J', the first moment after

isomerization. This observation demonstrates that Wat402, a hydronium through a

large part of the photocycle, is an integral part of the moving cluster that accepts the

proton from the SB. Most likely, this water will remain ordered throughout the rest of

the photocycle not yet surveyed.

The main cause to this contradiction is that a light induced change at any given time

point consists of a mixture of multiple conformations. As the most straightforward

example, the dramatic motion of Wat402 appears to be a disorder in an experimental

map prior to deconvolution. An experimental map is a spatial average over the

diffracting crystals and a temporal average over the X-ray exposures. In case of serial

crystallography, the range of this average is an extensive pool of crystals potentially

with diverse individualities (Ren et al., 2018). The only common feature among

multiple species of bR intermediates is the negative density on the original location of

Wat402 due to its departure. Multiple positive features at various locations are of low

occupancies without a proper deconvolution. In this analysis, a linear combination of

several SVD components unscrambles the mixture so that each state along the

photocycle presents a pure conformational species, a time-independent structure. Is

there an experimental solution to this problem of structural heterogeneity? Would a

better time resolution improve an experimental difference map so that a traveling water

molecule would always be temporally resolved? The answer is both surprisingly and

conclusively no, even if, for the sake of argument, an ultimate time resolution of truly

zero is achievable by two zero-width impulses of pump and probe (Strictly speaking,

each crystal must also be a spatial impulse, that is, zero size.). Because at no time

throughout the entire photocycle, the traveling water is synchronized throughout the

entire crystal, even if every retinal in the crystal absorbs a visible photon at the exact

same time zero. There is no difference between crystal and solution in this aspect,

which has been long recognized as Moffat laid the foundation of time-resolved

crystallography (Moffat, 1989, 2001). In conclusion, a numerical resolution of structural

heterogeneity, as exemplified here, is the last and the only resort (Ren et al., 2013).

Five-decade long-lived K and global change in L

This work continues with another joint analysis of the time-resolved datasets at long

delays > 10 ns collected mainly by Nango et al. with contributions from Nogly et al.,

Kovacs et al., and Weinert et al. (Table S1). These datasets at the long delays are

relatively noisier and contain systematic errors from one experiment to another. The

analytical procedure of SVD and the subsequent Ren rotation (Ren, 2016, 2019, 2021) is

capable of isolating time-resolved signals from these error and noise components. SVD

of 101 difference Fourier maps identifies 19 major components (Fig. S4a). Four of them

contain outstanding signals of structural changes (Fig. S5). Two others also carry

structural signals but without a clear time dependency (Fig. S6). The number of

datasets at long delays warrants an exponential fitting to the SVD coefficients (Fig. S7)

(Schmidt et al., 2010), which models the reaction scheme of the photocycle better

compared to the spline fitting applied to the limited number of short delays above. The

time constants obtained from the exponential fitting agree well with the transition times

between intermediates K, L, M_1, and M_2 known from spectroscopy. Coefficients for

relatively pure states are estimated and the refinement protocol based on reconstituted

difference maps again produce atomic coordinates of these states (Methods) (Ren, 2021).

The refined structures of K and L are discussed in SI (Figs. S8 and S9). A flattened

13-*cis* retinal is finally achieved in L with many global changes. The L state represents

the end goal of the global conformational changes due to the retinal isomerization.

Additional changes after L are no longer global (see below). C_δ-C_ε of the anchor moves

outboard and could break the seal with Thr89 once again. Last time, a potential

breakage occurs in I' at tens of fs that is too brief for any proton leak (Fig. 2b 2nd panel).

This time, the ns-μs time scale allows the protein to react. It is clearly observed that

Thr89 and helix C always follow C_δ-C_ε outboard in J, K, and L and maintain a distance

no more than 3.6 Å because of its tendency to be straightened (Figs. 3f and S10a). This

motion was also captured previously in cryo-trapped intermediates (Royant et al.,

2000). Therefore, the seal between two half channels is kept intact (Q4). Here the

mechanism of these motions of helices is attributed to several irregularities in these

helices (Fig. S10), the kinked helix C (Grigorieff et al., 1996), the stretched helix F, and

the π helical segment in G (Pebay-Peyroula et al., 1997). Contrary to the previous belief

(Luecke et al., 1999a), it is our general conclusion also from studies of other protein

mechanisms that irregularities in secondary structures not merely fulfill some structural

requirements but offer a wide variety of mechanical properties to achieve the functions
of protein nanodevices (Ren et al., 2016; Shin et al., 2019). These motions of the helical
segments collectively result in a tighter EC channel and an open CP channel in L (Fig.
S11).
446
Intermediates M_1, M_2 and proton release
The refined structures of M_1 and M_2 are described in SI (Figs. S12 and S13). Thr89 is
kept 3.1 Å away from the SB in L, M_1, and M_2 constantly so that the seal is well
maintained between two half channels (Q4). Compared to the slightly longer distance
between Thr89 and the SB at 3.6 Å in J and K, it is obvious that Thr89 and helix C react
to the retreating SB during J and K, but the SB is actively pushing Thr89 and helix C
from L to M (Fig. 4c). As discussed below, during pumping activity, drastic proton
concentration gradient is present across this seal that divides two half channels. A well-
maintained seal is crucial to the efficiency of the pump. Various mutants with
substitutions of Thr89 that could result in a proton leak retain ~2/3 of the pumping
activity compared to the wildtype (Marti et al., 1991).
458
The motions over the EC half of the molecule from tens of μs to ms are directly
related to the pathway of the proton release (Q3). Residues in the EC half channel are
not arranged in a monotonic order of their pK_a's. Quite the contrary, acidic residues
Asp85 and 212 with the smallest pK_a's are at the end of the half channel next to the SB
with the greatest pK_a. Another layer of acidic residues Glu9, 194, and 204 are at the EC
surface. It is puzzling that Arg82, Tyr57, and 83 are located between these two acidic
layers (Figs. 5 and S1). It had been depicted multiple times by similar illustrations in
the past that a high pK_a layer is sandwiched between two acidic layers with low pK_a's in
the EC half channel. Yet, the question remains: How are protons conducted through
the proton barrier of the guanidinium group of Arg82 that is already protonated and
the phenol hydroxyl groups of the Tyr residues that cannot be protonated anymore
(Brown et al., 1993)? No evidence shows that such rollercoasting pK_a values would
change in any meaningful way by the light-induced conformations. Therefore, the
previous proposal on an affinity-driven proton conductance is fundamentally flawed.
473
Proton conductance after SB deprotonation through the EC half channel not
necessarily operates on the basis of one event of proton release per photocycle. It was

previously suggested that a proton pumped in one photocycle could be released in the
next or subsequent photocycles (Braiman et al., 1991). However, a flawed common
assumption persists that the pumping and releasing of each proton are somehow
scheduled in different periods of a photocycle. Here a concentration-driven, instead of
affinity-driven, proton conductance is hypothesized. Proton release becomes
synchronized to the photocycle only when a super-acidic inner EC channel is
established. The first a few photocycles build up a proton concentration at the end of
the EC half channel. Multiple groups in the inner EC channel, including various waters,
could be all protonated after consecutive photocycles, that is, a puddle of protons
establishes a super-acidic inner EC channel as long as the seal of Thr89 is well
maintained. This is possible because the proton pumping step is a physical process
instead of a chemical equilibrium (see above). Although the first a few photocycles are
productive, they are not capable of releasing any proton before an established proton
concentration gradient. An established super-acidic inner EC channel triggers a
concentration-driven proton conductance. Proton release from the wildtype bR was
observed before proton uptake as if it is coupled with, if not faster than, the protonation
of the proton acceptor Asp85 (Braiman et al., 1988, 1991; Gerwert et al., 1990; Zimanyi et
al., 1992), which strongly suggests that the newly pumped proton is not the one
released to the EC medium in the same photocycle. Here the concentration-driven
proton conductance argues that the newly pumped proton causes a proton overflow
from the already super-acidic inner EC channel. Therefore, the observed "early release"
is actually a very late release during a subsequent photocycle, several cycles after the
very proton has been pumped into the inner EC channel.

The inner EC channel has been called a proton cage at the active site of bR (Friedrich
et al., 2020), which reflects the same fact of the proton barrier (Fig. 5). Here I argue that
the super-acidic inner EC channel causes a second protonation of the guanidinium
group because of three electron-rich nitrogens. The observation to corroborate this
speculation is the motions of Arg82, Glu194, and 204 during K $\rightarrow$ L $\rightarrow$ M$_1$ $\rightarrow$ M$_2$. The
monovalent guanidinium cation during the early photocycle has no salt bridge to
Glu194 and 204 until M state. It appears that Asp85 and 212 are continuously
protonated by proton pumping thus neutralized in the inner EC channel. Several
hydronium ions make the inner EC channel positively charged. The observed kinetics
of FTIR continua could explain excess protons harbored by the water cluster in the

inner EC channel (Garczarek et al., 2004, 2005; Lorenz-Fonfria et al., 2017; Wang and El-
Sayed, 2001). An extra positive charge on the short-lived divalent guanidinium in M
drives itself away from the inner EC channel and toward the negative charged Glu
residues in the outer EC channel (Fig. 5a). The EC segments of helices C and G move
closer to each other in the refined M structures, where Arg82 is sandwiched in between
(SI; Fig. 4). Molecular dynamics simulations found that the orientation of Arg82 is
correlated with the charge difference between the inner and outer acids (Ge and
Gunner, 2016). The contact among Glu194, 204, and the guanidinium discharges the
extra proton to the outer EC channel and the EC medium. A monovalent guanidinium
cation is unable to discharge its proton to the carboxylates of the Glu residues due to
their low proton affinity. Therefore, Arg82 is a unidirectional proton valve that
conducts protons outward, but only under sufficient proton concentration in the inner
EC channel. During dark, Arg82 prevents a concentration-driven proton leak back into
the cell, because the EC medium cannot become super acidic as long as sufficient water
is available. See SI for mutant functions when the proton barrier is breached by
substitutes for Arg82.
526

Discussion
Under the guidance of Jardetzky's allostery model of transmembrane pump (Jardetzky,
1966) and the common line of thinking led by Stoeckenius (Stoeckenius et al., 1979),
evidences were sought in the past half century to explain how the absorbed photon
energy is used to alter the proton affinity landscape along its pathway so that protons
can flow in one direction along an ascending order of pK_a values. Neither a hydraulic
powered noria or an animal powered sakia in the old time, nor an electric powered
fluid pump in the modern days, works under such an operating principle that requires
energy for a reconstruction of the landscape. Quite the contrary, the landscape is a
given constant and energy is absorbed by the to-be-pumped substance to gain its
potential energy, such as water is lifted by a noria or pressurized by an electric pump.
The energized substance will then flow freely along the largely constant landscape
governed by thermodynamics. In the meanwhile, it is critical to prevent a backflow
along a far more steep gradient than the forward-driving gradient. The molecular
pump bacteriorhodopsin (bR) for protons works under the exact same principle that
fundamentally differs from what Jardetzky and Stoeckenius once imagined. Other
transmembrane molecular pumps and transporters likely work the same way at this

philosophical level disregarding any molecular detail. Alteration of affinities to the
pumped substance cannot be an effective strategy because it is against the chemical
nature along the translocation pathway and therefore energetically difficult to achieve.
For example, the pK_a values differ more than 11 pH units at the resting state between
the protonated SB and the Asp acids, the proton acceptors. That is to say, a
photoinduced reduction of one million times to the proton affinity of the SB must occur
simultaneously with an increase of one million times to the proton affinity of the
carboxylates. Even if these changes in proton affinity are achievable, there will be 10%
failed attempts of proton transfers at a prolonged equilibrium. The proton levee,
consisting of a few tyrosine residues, main chain carbonyls, and most importantly,
Arg82, is strategically constructed around the inner EC channel to prevent proton
backflow (Fig. 5). Their high pK_a values are not meant to be changed even though small
changes could be detected. In this study, the assumption of an affinity-driven proton
conductance is replaced by a mechanism of concentration-driven proton conductance
that is not necessarily synchronous to the early photocycles.

All chemical reactions involve energy transfer or conversion, known as enthalpy
change ΔH. This flow of energy in a unit time is the temporal rate of enthalpy change
dH/dt or power as a function of time. This function is usually far from a constant. The
peak power of a chemical reaction could be orders of magnitude greater than its
average power. This is particularly true for each individual molecular event compared
to a molecular ensemble. Consider the photochemical reaction of the retinal
isomerization in bR. Its peak power is reached at the U-turn of the SB (Figs. 2a and S2),
which accomplishes the initial deprotonation of each photocycle. The protein structure
of bR prevents a photochemical reaction from reaching its equilibrium as the
photoisomerization of the retinal switches the accessibility of the Schiff base to a
different reaction space. This is usually unachievable in chemical reactions of small
molecules.

Acknowledgements
This work is supported in part by the grant R01EY024363 from National Institutes of
Health. The following database and software are used in this work: CCP4 (ccp4.ac.uk),
Coot (www2.mrc-lmb.cam.ac.uk/Personal/pemsley/coot), dynamiX™ (Renz Research,

Inc.), gnuplot (gnuplot.info), PDB (rcsb.org), PHENIX (phenix-online.org), PyMOL

(pymol.org), Python (python.org), and SciPy (scipy.org).

Competing interests

ZR is the founder of Renz Research, Inc. that currently holds the copyright of the

computer software dynamiX™.

References

Adams, P.D., Afonine, P.V., Bunkóczi, G., Chen, V.B., Davis, I.W., Echols, N., Headd, J.J., Hung, L.-W.,
Kapral, G.J., Grosse-Kunstleve, R.W., et al. (2010). PHENIX: a comprehensive Python-based system for
macromolecular structure solution. Acta Crystallogr. D Biol. Crystallogr. *D66*, 213–221.

Applebury, M.L., Peters, K.S., and Rentzepis, P.M. (1978). Primary intermediates in the photochemical
cycle of bacteriorhodopsin. Biophys. J. *23*, 375–382.

Balashov, S.P., Govindjee, R., Kono, M., Imasheva, E., Lukashev, E., Ebrey, T.G., Crouch, R.K., Menick,
D.R., and Feng, Y. (1993). Effect of the arginine-82 to alanine mutation in bacteriorhodopsin on dark
adaptation, proton release, and the photochemical cycle. Biochemistry *32*, 10331–10343.

Balashov, S.P., Imasheva, E.S., Govindjee, R., and Ebrey, T.G. (1996). Titration of aspartate-85 in
bacteriorhodopsin: what it says about chromophore isomerization and proton release. Biophys. J. *70*, 473–
481.

Balashov, S.P., Imasheva, E.S., Ebrey, T.G., Chen, N., Menick, D.R., and Crouch, R.K. (1997). Glutamate-
194 to cysteine mutation inhibits fast light-induced proton release in bacteriorhodopsin. Biochemistry *36*,
8671–8676.

van den Berg, R., Du-Jeon-Jang, Bitting, H.C., and El-Sayed, M.A. (1990). Subpicosecond resonance
Raman spectra of the early intermediates in the photocycle of bacteriorhodopsin. Biophys. J. *58*, 135–141.

Braiman, M.S., Mogi, T., Marti, T., Stern, L.J., Khorana, H.G., and Rothschild, K.J. (1988). Vibrational
spectroscopy of bacteriorhodopsin mutants: light-driven proton transport involves protonation changes
of aspartic acid residues 85, 96, and 212. Biochemistry *27*, 8516–8520.

Braiman, M.S., Bousche, O., and Rothschild, K.J. (1991). Protein dynamics in the bacteriorhodopsin
photocycle: submillisecond Fourier transform infrared spectra of the L, M, and N photointermediates.
Proc. Natl. Acad. Sci. *88*, 2388–2392.

Brown, L.S., Bonet, L., Needleman, R., and Lanyi, J.K. (1993). Estimated acid dissociation constants of the
Schiff base, Asp-85, and Arg-82 during the bacteriorhodopsin photocycle. Biophys. J. *65*, 124–130.

Chandonia, J.-M., and Brenner, S.E. (2006). The impact of structural genomics: expectations and
outcomes. Science *311*, 347–351.

Dickopf, S., Alexiev, U., Krebs, M.P., Otto, H., Mollaaghababa, R., Khorana, H.G., and Heyn, M.P. (1995).
Proton transport by a bacteriorhodopsin mutant, aspartic acid-85-->asparagine, initiated in the
unprotonated Schiff base state. Proc. Natl. Acad. Sci. *92*, 11519–11523.

Druckmann, S., Ottolenghi, M., Pande, A., Pande, J., and Callender, R.H. (1982). Acid-base equilibrium of
the Schiff base in bacteriorhodopsin. Biochemistry *21*, 4953–4959.

Ernst, O.P., Lodowski, D.T., Elstner, M., Hegemann, P., Brown, L.S., and Kandori, H. (2014). Microbial
and animal rhodopsins: Structures, functions, and molecular mechanisms. Chem. Rev. *114*, 126–163.

Fahmy, K., Weidlich, O., Engelhard, M., Tittor, J., Oesterhelt, D., and Siebert, F. (1992). Identification of
the proton acceptor of Schiff base deprotonation in bacteriorhodopsin: A Fourier-transform-infrared
study of the mutant Asp85 → Glu in its natural lipid environment. Photochem. Photobiol. *56*, 1073–1083.

Fenno, L., Yizhar, O., and Deisseroth, K. (2011). The development and application of optogenetics. Annu.
Rev. Neurosci. *34*, 389–412.

Friedrich, D., Brünig, F.N., Nieuwkoop, A.J., Netz, R.R., Hegemann, P., and Oschkinat, H. (2020).
Collective exchange processes reveal an active site proton cage in bacteriorhodopsin. Commun. Biol. *3*, 4.

Garczarek, F., Wang, J., El-Sayed, M.A., and Gerwert, K. (2004). The Assignment of the Different Infrared
Continuum Absorbance Changes Observed in the 3000–1800-cm−1 Region during the Bacteriorhodopsin
Photocycle. Biophys. J. *87*, 2676–2682.

Garczarek, F., Brown, L.S., Lanyi, J.K., and Gerwert, K. (2005). Proton binding within a membrane protein
by a protonated water cluster. Proc. Natl. Acad. Sci. *102*, 3633–3638.

Ge, X., and Gunner, M.R. (2016). Unraveling the mechanism of proton translocation in the extracellular
half-channel of bacteriorhodopsin: Proton translocation in bacteriorhodopsin. Proteins Struct. Funct.
Bioinforma. *84*, 639–654.

Gerwert, K., Hess, B., Soppa, J., and Oesterhelt, D. (1989). Role of aspartate-96 in proton translocation by
bacteriorhodopsin. Proc. Natl. Acad. Sci. *86*, 4943–4947.

Gerwert, K., Souvignier, G., and Hess, B. (1990). Simultaneous monitoring of light-induced changes in
protein side-group protonation, chromophore isomerization, and backbone motion of bacteriorhodopsin
by time-resolved Fourier-transform infrared spectroscopy. Proc. Natl. Acad. Sci. *87*, 9774–9778.

Glynn, C., and Rodriguez, J.A. (2019). Data-driven challenges and opportunities in crystallography.
Emerg. Top. Life Sci. ETLS20180177.

Govindjee, R., Balashov, S.P., and Ebrey, T.G. (1990). Quantum efficiency of the photochemical cycle of
bacteriorhodopsin. Biophys. J. *58*, 597–608.

Govindjee, R., Misra, S., Balashov, S.P., Ebrey, T.G., Crouch, R.K., and Menick, D.R. (1996). Arginine-82
regulates the pKa of the group responsible for the light-driven proton release in bacteriorhodopsin.
Biophys. J. *71*, 1011–1023.

Grigorieff, N., Ceska, T.A., Downing, K.H., Baldwin, J.M., and Henderson, R. (1996). Electron-
crystallographic refinement of the structure of bacteriorhodopsin. J. Mol. Biol. *259*, 393–421.

Hackett, N.R., Stern, L.J., Chao, B.H., Kronis, K.A., and Khorana, H.G. (1987). Structure-function studies
on bacteriorhodopsin. V. Effects of amino acid substitutions in the putative helix F. J. Biol. Chem. *262*,
9277–9284.

Henderson, R., and Unwin, P.N.T. (1975). Three-dimensional model of purple membrane obtained by
electron microscopy. Nature *257*, 28–32.

Henderson, R., Baldwin, J.M., Ceska, T.A., Zemlin, F., Beckmann, E., and Downing, K.H. (1990). Model
for the structure of bacteriorhodopsin based on high-resolution electron cryo-microscopy. J. Mol. Biol.
213, 899–929.

Henry, E.R., and Hofrichter, J. (1992). Singular value decomposition: Application to analysis of
experimental data. In Numerical Computer Methods, (Academic Press), pp. 129–192.

Jardetzky, O. (1966). Simple allosteric model for membrane pumps. Nature *211*, 969–970.

Jung, Y.O., Lee, J.H., Kim, J., Schmidt, M., Moffat, K., Šrajer, V., and Ihee, H. (2013). Volume-conserving
trans–cis isomerization pathways in photoactive yellow protein visualized by picosecond X-ray
crystallography. Nat. Chem. *5*, 212–220.

Kahan, A., Nahmias, O., Friedman, N., Sheves, M., and Ruhman, S. (2007). Following photoinduced
dynamics in bacteriorhodopsin with 7-fs impulsive vibrational spectroscopy. J. Am. Chem. Soc. *129*, 537–
546.

Kandori, H. (2015). Ion-pumping microbial rhodopsins. Front. Mol. Biosci. *2*.

Kobayashi, T., Saito, T., and Ohtani, H. (2001). Real-time spectroscopy of transition states in
bacteriorhodopsin during retinal isomerization. Nature *414*, 531–534.

Kovacs, G.N., Colletier, J.-P., Grünbein, M.L., Yang, Y., Stensitzki, T., Batyuk, A., Carbajo, S., Doak, R.B.,
Ehrenberg, D., Foucar, L., et al. (2019). Three-dimensional view of ultrafast dynamics in photoexcited
bacteriorhodopsin. Nat. Commun. *10*, 3177.

Lanyi, J.K., and Schobert, B. (2007). Structural changes in the L photointermediate of bacteriorhodopsin. J.
Mol. Biol. *365*, 1379–1392.

Lewis, A., Spoonhower, J., Bogomolni, R.A., Lozier, R.H., and Stoeckenius, W. (1974). Tunable laser
resonance Raman spectroscopy of bacteriorhodopsin. Proc. Natl. Acad. Sci. *71*, 4462–4466.

Li, Y.-T., Tian, Y., Tian, H., Tu, T., Gou, G.-Y., Wang, Q., Qiao, Y.-C., Yang, Y., and Ren, T.-L. (2018). A
review on bacteriorhodopsin-based bioelectronic devices. Sensors *18*, 1368.

Liebschner, D., Afonine, P.V., Baker, M.L., Bunkóczi, G., Chen, V.B., Croll, T.I., Hintze, B., Hung, L.-W.,
Jain, S., McCoy, A.J., et al. (2019). Macromolecular structure determination using X-rays, neutrons and
electrons: recent developments in Phenix. Acta Crystallogr. Sect. Struct. Biol. *75*, 861–877.

Lorenz-Fonfria, V.A., Saita, M., Lazarova, T., Schlesinger, R., and Heberle, J. (2017). pH-sensitive
vibrational probe reveals a cytoplasmic protonated cluster in bacteriorhodopsin. Proc. Natl. Acad. Sci.
114, E10909–E10918.

Lozier, R.H., Bogomolni, R.A., and Stoeckenius, W. (1975). Bacteriorhodopsin: a light-driven proton
pump in Halobacterium Halobium. Biophys. J. *15*, 955–962.

Luecke, H., Schobert, B., Richter, H.-T., Cartailler, J.-P., and Lanyi, J.K. (1999a). Structure of
bacteriorhodopsin at 1.55 Å resolution. J. Mol. Biol. *291*, 899–911.

Luecke, H., Schobert, B., Richter, H.-T., Cartailler, J.-P., and Lanyi, J.K. (1999b). Structural changes in
bacteriorhodopsin during ion transport at 2 angstrom resolution. Science *286*, 255–260.

Luecke, H., Schobert, B., Cartailler, J.-P., Richter, H.-T., Rosengarth, A., Needleman, R., and Lanyi, J.K.
(2000). Coupling photoisomerization of retinal to directional transport in bacteriorhodopsin. J. Mol. Biol.
300, 1237–1255.

Marti, T., Otto, H., Mogi, T., Rösselet, S.J., Heyn, M.P., and Khorana, H.G. (1991). Bacteriorhodopsin
mutants containing single substitutions of serine or threonine residues are all active in proton
translocation. J. Biol. Chem. *266*, 6919–6927.

Marx, D., Tuckerman, M.E., Hutter, J., and Parrinello, M. (1999). The nature of the hydrated excess proton
in water. Nature *397*, 601–604.

Mathias, G., and Marx, D. (2007). Structures and spectral signatures of protonated water networks in
bacteriorhodopsin. Proc. Natl. Acad. Sci. *104*, 6980–6985.

McCarty, C.G. (1970). Chapter 9 syn-anti isomerizations and rearrangements. In The Chemistry of the
Carbon-Nitrogen Double Bond, (John Wiley & Sons, Ltd), p. 363.

Moffat, K. (1989). Time-Resolved Macromolecular Crystallography. Annu. Rev. Biophys. Biophys. Chem.
18, 309–332.

Moffat, K. (2001). Time-resolved biochemical crystallography: A mechanistic perspective. Chem Rev *101*,
1569–1582.

Mogi, T., Stern, L.J., Marti, T., Chao, B.H., and Khorana, H.G. (1988). Aspartic acid substitutions affect
proton translocation by bacteriorhodopsin. Proc. Natl. Acad. Sci. *85*, 4148–4152.

Nango, E., Royant, A., Kubo, M., Nakane, T., Wickstrand, C., Kimura, T., Tanaka, T., Tono, K., Song, C.,
Tanaka, R., et al. (2016). A three-dimensional movie of structural changes in bacteriorhodopsin. Science
354, 1552–1557.

Nogly, P., Weinert, T., James, D., Carbajo, S., Ozerov, D., Furrer, A., Gashi, D., Borin, V., Skopintsev, P.,
Jaeger, K., et al. (2018). Retinal isomerization in bacteriorhodopsin captured by a femtosecond x-ray laser.
Science *361*, eaat0094.

Pebay-Peyroula, E., Rummel, G., Rosenbusch, J.P., and Landau, E.M. (1997). X-ray structure of
bacteriorhodopsin at 2.5 Angstroms from microcrystals grown in lipidic cubic phases. Science *277*, 1676–
1681.

Ren, Z. (2013a). Reaction trajectory revealed by a joint analysis of Protein Data Bank. PLoS ONE *8*, e77141.

Ren, Z. (2013b). Reverse engineering the cooperative machinery of human hemoglobin. PLoS ONE *8*,
e77363.

Ren, Z. (2016). Molecular events during translocation and proofreading extracted from 200 static
structures of DNA polymerase. Nucleic Acids Res. *6*, 1–13.

Ren, Z. (2019). Ultrafast structural changes decomposed from serial crystallographic data. J. Phys. Chem.
Lett. *10*, 7148–7163.

Ren, Z. (2021). Photoinduced isomerization sampling of retinal in bacteriorhodopsin (bioRxiv
doi:10.1101/2021.09.16.460656).

Ren, Z., Perman, B., Srajer, V., Teng, T.-Y., Pradervand, C., Bourgeois, D., Schotte, F., Ursby, T., Kort, R.,
Wulff, M., et al. (2001). A molecular movie at 1.8 Å resolution displays the photocycle of photoactive
yellow protein, a eubacterial blue-light receptor, from nanoseconds to seconds. Biochemistry *40*, 13788–
13801.

Ren, Z., Chan, P.W.Y., Moffat, K., Pai, E.F., Royer, W.E., Šrajer, V., and Yang, X. (2013). Resolution of
structural heterogeneity in dynamic crystallography. Acta Cryst *D69*, 946–959.

Ren, Z., Ren, P.X., Balusu, R., and Yang, X. (2016). Transmembrane helices tilt, bend, slide, torque, and
unwind between functional states of rhodopsin. Sci. Rep. *6*, 34129.

Ren, Z., Ayhan, M., Bandara, S., Bowatte, K., Kumarapperuma, I., Gunawardana, S., Shin, H., Wang, C.,
Zeng, X., and Yang, X. (2018). Crystal-on-crystal chips for in situ serial diffraction at room temperature.
Lab. Chip *18*, 2246–2256.

Richter, H.T., Needleman, R., and Lanyi, J.K. (1996). Perturbed interaction between residues 85 and 204 in
Tyr-185-->Phe and Asp-85-->Glu bacteriorhodopsins. Biophys. J. *71*, 3392–3398.

Royant, A., Edman, K., Ursby, T., Pebay-Peyroula, E., Landau, E.M., and Neutze, R. (2000). Helix
deformation is coupled to vectorial proton transport in the photocycle of bacteriorhodopsin. Nature *406*,
645–648.

Sass, H.J., Büldt, G., Gessenich, R., Hehn, D., Neff, D., Schlesinger, R., Berendzen, J., and Ormos, P. (2000).
Structural alterations for proton translocation in the M state of wild-type bacteriorhodopsin. Nature *406*,
649–653.

Schaffer, J.E., Kukshal, V., Miller, J.J., Kitainda, V., and Jez, J.M. (2021). Beyond X-rays: an overview of
emerging structural biology methods. Emerg. Top. Life Sci. ETLS20200272.

Schmidt, M., Rajagopal, S., Ren, Z., and Moffat, K. (2003). Application of singular value decomposition to
the analysis of time-resolved macromolecular X-ray data. Biophys. J. *84*, 2112–2129.

Schmidt, M., Graber, T., Henning, R., and Srajer, V. (2010). Five-dimensional crystallography. Acta
Crystallogr. A *66*, 198–206.

Sheves, M., Albeck, A., Friedman, N., and Ottolenghi, M. (1986). Controlling the pKa of the
bacteriorhodopsin Schiff base by use of artificial retinal analogues. Proc. Natl. Acad. Sci. *83*, 3262–3266.

Shin, H., Ren, Z., Zeng, X., Bandara, S., and Yang, X. (2019). Structural basis of molecular logic OR in a
dual-sensor histidine kinase. Proc. Natl. Acad. Sci. *116*, 19973–19982.

Šrajer, V., Ren, Z., Teng, T.-Y., Schmidt, M., Ursby, T., Bourgeois, D., Pradervand, C., Schildkamp, W.,
Wulff, M., and Moffat, K. (2001). Protein conformational relaxation and ligand migration in myoglobin: A
nanosecond to millisecond molecular movie from time-resolved Laue X-ray diffraction. Biochemistry *40*,
13802–13815.

Stoeckenius, W. (1999). Bacterial rhodopsins: Evolution of a mechanistic model for the ion pumps. Protein
Sci. *8*, 447–459.

Stoeckenius, W., Lozier, R.H., and Bogomolni, R.A. (1979). Bacteriorhodopsin and the purple membrane
of halobacteria. Biochim. Biophys. Acta BBA - Rev. Bioenerg. *505*, 215–278.

Takeda, K., Matsui, Y., Kamiya, N., Adachi, S., Okumura, H., and Kouyama, T. (2004). Crystal structure of
the M intermediate of bacteriorhodopsin: Allosteric structural changes mediated by sliding movement of
a transmembrane helix. J. Mol. Biol. *341*, 1023–1037.

Ursby, T., and Bourgeois, D. (1997). Improved estimation of structure-factor difference amplitudes from
poorly accurate data. Acta Crystallogr. A *53*, 564–575.

Wang, J., and El-Sayed, M.A. (2001). Time-Resolved Fourier Transform Infrared Spectroscopy of the
Polarizable Proton Continua and the Proton Pump Mechanism of Bacteriorhodopsin. Biophys. J. *80*, 961–
971.

Weinert, T., Skopintsev, P., James, D., Dworkowski, F., Panepucci, E., Kekilli, D., Furrer, A., Brünle, S.,
Mous, S., Ozerov, D., et al. (2019). Proton uptake mechanism in bacteriorhodopsin captured by serial
synchrotron crystallography. Science *365*, 61–65.

Wickstrand, C., Dods, R., Royant, A., and Neutze, R. (2015). Bacteriorhodopsin: Would the real structural
intermediates please stand up? Biochim. Biophys. Acta *1850*, 536–553.

Yang, X., Ren, Z., Kuk, J., and Moffat, K. (2011). Temperature-scan cryocrystallography reveals reaction
intermediates in bacteriophytochrome. Nature *479*, 428–432.

Zhong, Q., Ruhman, S., Ottolenghi, M., Sheves, M., Friedman, N., Atkinson, G.H., and Delaney, J.K.
(1996). Reexamining the primary light-induced events in bacteriorhodopsin using a synthetic C13=C14-
locked chromophore. J Am Chem Soc *118*, 12828–12829.

Zimanyi, L., Varo, G., Chang, M., Ni, B., Needleman, R., and Lanyi, J.K. (1992). Pathways of proton
release in the bacteriorhodopsin photocycle. Biochemistry *31*, 8535–8543.

(2012). The Protein Data Bank at 40: Reflecting on the Past to Prepare for the Future. Structure *20*, 391–
396.

Directional Proton Conductance in Bacteriorhodopsin Is Driven by

Concentration Gradient, Not Affinity Gradient

Zhong Ren

Department of Chemistry, University of Illinois at Chicago, Chicago, IL 60607, USA

Renz Research, Inc., Westmont, IL 60559, USA

zren@uic.edu

ORCID 0000-0001-7098-3127

Figures and Legends

Ren: Concentration-driven proton conductance

Time (s)

| 10^{-14} | 10^{-13} | 10^{-12} | 10^{-11} | 10^{-10} | 10^{-9} | 10^{-8} | 10^{-7} | 10^{-6} | 10^{-5} | 10^{-4} | 10^{-3} | 10^{-2} | 10^{-1} |

bR (568 nm)
O (640 nm)
N (560 nm)
M₂ (410 nm)
M₁ (410 nm)
L (550 nm)
K (610 nm)
J (625 nm)
I (460 nm)
bR (568 nm)

CP channel opens.
EC channel contracts.

Retinal binding pocket contraction
Retinal binding pocket expansion
Stimulated Raman gain change

Protonated SB
Deprotonated SB
Protonated SB
Single bonds deviated from *anti* signaling charge separation
Charge separation on retinal at Frank–Condon point
13–*cis* retinal
Creased and shorter retinal; Distal segment inbound; Proximal segment outbound
All–*trans* retinal
M₂: SB continues to move inboard.
M₁: Several ordered waters in enlarged CP channel
L: Most flattened retinal–lysine chain fully erected
K: C₂₀ methyl group swinging back inboard in a long process
J: Retinal fully isomerized to 13–*cis* in a contracted binding pocket
J': Isomerization to 13–*cis*; C₂₀ methyl group protrudes to CP.
I: Retinal binding pocket expands due to isomerization sampling. SB U–turns back to resting position.
I': Several single bonds deviate from *anti*; C₂₀ methyl group and SB swing outbound.
1ˢᵗ period in oscillation of 61 cm⁻¹
1ˢᵗ period in oscillation of 150 cm⁻¹
1ˢᵗ period in oscillation of 224 cm⁻¹
1ˢᵗ period in oscillation of 334 cm⁻¹
1ˢᵗ period in oscillation of 396 cm⁻¹

Long delays
Gap of missing data > 3 decades
Short delays

- 33 ms 6ga3
- 10–15 ms 6rph
- 8.33 ms 6g7l
- 0–5 ms 6rnj
- 1.725 ms 5b6z
- 657 ∝s 5h2p
- 250 ∝s 5h2o
- 95.2 ∝s 5h2n
- 36.2 ∝s 5b6y
- 13.8 ∝s 5h2m
- 5.25 ∝s 5h2l
- 2 ∝s 5h2k
- 760 ns 5b6x
- 290 ns 5h2j
- 110 ns 5h2i
- 40 ns 5h2h
- 16 ns 5b6w

Laser pulse (Nango et al.)

Decomposed structural events (this & companion work (Ren, 2021))
Previously known events
Laser and XFEL pulses (Nango et al.)
Laser and XFEL pulses (Nogly et al.)
Laser and XFEL pulses (Kovacs et al.)
Laser and X–ray pulses (Weinert et al.)

- 10 ps 6g7k
- 10 ps 6ga6
- 3 ps 6ga5
- 1 ps 6ga4
- 740 fs 6gai
- 680 fs 6gah
- 630 fs 6gag
- 590 fs 6gaf
- 560 fs 6gae
- 457–646 fs 6g7j
- 530 fs 6gad
- 490 fs 6gac
- 460 fs 6gab
- 430 fs 6gaa
- 390 fs 6ga9
- 330 fs 6ga8
- 49–406 fs 6g7i
- 240 fs 6ga7
Excitation laser pulse (Nogly et al.)
Excitation laser pulse (Kovacs et al.)

| 10fs | 100fs | 1ps | 10ps | 100ps | 1ns | 10ns | 100ns | 1∝s | 10∝s | 100∝s | 1ms | 10ms | 100ms |

Time

Ren: Concentration-driven proton conductance

Figure 1. Gantt chart summarizing analyzed datasets and results. (Inset) Proposed

scheme of deprotonation from SB. Deprotonation occurs at I′ around ~30 fs to form a

transient hydronium ion (Fig. 2c). Another proton is transferred to either Asp85 or 212,

between which an equilibrium is maintained over time. A second deprotonation in the

next photocycle causes both Asp residues to be neutralized. One or more waters in the

inner EC channel become overcharged with excess protons during the subsequent

photocycles. A super-acidic inner EC channel then causes a second protonation of

Arg82 to form a divalent guanidinium cation. The positive charged guanidinium group

is attracted to the negative charged Glu residues in the outer EC channel and swings

outward to form a salt bridge. The extra proton is discharged to the EC medium

through the Glu residues and several waters in the outer EC channel, known as the

proton release complex. The monovalent guanidinium returns for the next shuttling

(Fig. 5).

Figure 2. Trajectories of bacteriorhodopsin photocycle. (a) Two orthographical views of

refined atomic trajectories. bR structure in the resting state is rendered in white ribbon

and sticks. The photocycle trajectories of several atoms of the chromophore C_{14}, C_{20}, C_{ε},

N_ζ, and Wat402 are rendered as small spheres. Their positions in the ground state bR

are in black. The intermediates I′, I, J′, J, K, L, M_1, and M_2 are colored differently. This

color scheme (key in b) is consistently applied to all figures in this paper. Other small

spheres along the atomic trajectories are in white for C, blue for N, and red for O. (b)

Conformational parameters calculated from the refined chromophore. The chemical

structure of the chromophore on top is aligned to the horizontal axis. Double bonds are

shaded in gray. Atomic displacements of each intermediate from the resting state show

greater changes in the proximal segment (top panel). A plane is least-squares fitted to

C_4 through C_{14} of the resting state. This retinal plane is largely parallel to the three-fold

axis of bR trimer. Therefore, one side of the plane where the three-fold axis is located is

called inboard, and the opposite side of the plane away from the three-fold axis is called

outboard (Ren, 2021). The distances of all atoms to this plane in the inboard and

outboard directions show the curvature of the chromophore. The creased retinal in the

early intermediates and the inboard protruding corner at C_ε in the resting state are

clearly shown (2nd panel). Distances to atom C_4 are calculated for all refined

chromophores. Changes in these distances with respect to the resting state show the

shortened chromophore in I′ and I. Once isomerization to 13-*cis* occurs, the segment

from C_{15} to C_δ around the SB becomes significantly closer to the β-ionone ring, while the

distal segment of the retinal from C_{14} and beyond stretches (3rd panel). The torsion

angles of single and double bonds quantify *anti/syn* or *trans/cis* for the ground state and

all intermediates. Only a single bond can be twisted with its torsion angle near 90°. A

double bond would spend very little time at a torsion angle near 90° compared to *trans*

or *cis* configurations thus not easily observed (4th panel). Each double bond is least-

squares fitted with a plane. The interplanar angle between one double bond and the

corresponding one in the ground state measures the local tilting of the retinal (bottom

panel). (c) Deprotonation from the Schiff base. In the resting state, a proton is situated

between the SB N_ζ and the O of Wat402. Two lone pairs of electrons from N and O,

respectively, are interacting with the proton. During thermal dissociation to the left, the

proton stays with the SB with greater probability due to its greater proton affinity. This

thermal process does not cause deprotonation from the SB, and it could revert. On the

other hand, when the SB is abruptly displaced by the photoisomerization as observed

here at ~30 fs (a), the proton inertia tips the balance of these two interactions. The

proton cannot overcome its association with the water and accelerate fast enough with

the SB. As a result, it stays with Wat402 to form a hydronium ion. This deprotonation

from the SB to the right is not governed by the proton affinities of two competing

moieties based on chemical equilibrium. In addition, this process does not revert as the

photoisomerization makes the SB no longer accessible from the EC half channel.

Figure 3. Intermediates J, K, and L. (a) Cross sections of component map U_2 of the long

delays > 10 ns. The middle cross section is an integration ±0.2 Å around the surface

through the retinal. The top cross section is an integration 1.4-1.8 Å outboard from the

retinal surface and the bottom is an integration 0.6-1 Å inboard. Green and red indicate

electron density gain and loss, respectively. Stronger signals are around the proximal

segment of the retinal, including Wat402. Positive and negative sheets of densities flank

the distal segment of the retinal from inboard and outboard sides, respectively. This

component is nearly constant in all long-delay difference maps with a dark dataset as

reference (Fig. S4). (b) Overview of the component map U_2 of the long delays contoured

at ±3σ. The extraordinary signal-to-noise level is shown as the association with specific

elements of the structure. Some parts of the structure are omitted to show the interior.

(c) Two orthographical views of the reconstituted difference map K – bR from U_1, U_2,

and U_3 of the long delays (Fig. S5abc). The map is contoured at ±2.5σ in green and red

mesh, respectively. Atomic displacements are indicated by the arrows marking the

negative and positive pairs of densities. (d) Reconstituted difference map L – bR from

U_2, U_3, and U_6 of the long delays (Fig. S5bcd). The map is contoured at ±3.5σ in green

and red mesh, respectively. The strongest signals are concentrated on the irregularities

of helices F, G, and C (Fig. S10). (e and f) Two orthographical views of the refined

retinal in J, K, and L compared with the resting state in white. The retinal is less creased

in K and flattened in L compared to the previous intermediates (Fig. 2b 2nd panel). C_{20}

methyl group is less and less tilted from its orientation in the resting state (Fig. 2b

bottom panel). Wat402 with good electron density (Figs. S8 and S9) remains its H-

bonds with both Asp85 and 212 in both K and L. Thr89 follows the motion of the SB

toward outboard.

Figure 4. Intermediates L, M_1, and M_2. (a, b, and c) Three orthographical views of EC

half channel. The refined structures of L, M_1, and M_2 in pink, yellow, and orange,

respectively, are compared to the resting state in white. Several arrows mark the

motions from the ground state bR to these intermediates or from these intermediates

back to bR. The SB is moving more and more inboard since L and finally establishes the

corner at C_ε protruding inboard in the ground state by isomerization back to all-*trans*.

Meanwhile, Thr89 and helix C are pushed away. (d) Atomic displacements in the main

chain around the EC half channel from the resting state to L and M are marked with

arrows 10× as long as the actual displacements. The EC segments of helices C and G

move closer to each other as Arg82 swings out. The same portion is displayed as in (a-

c).

898
899

Figure 5. H-bond network in EC half channel. The EC half channel is decorated with

acidic residues, such as Asp85 and 212 at the end, called inner EC channel shaded in

red, and Glu9, 194, and 204 at the EC surface, called outer EC channel also shaded in

red. Arg82, Tyr57, and 83 are located between these two layers that form a proton

barrier shaded in blue (Fig. S1). (a) The refined structures of K in blue, L in pink, and

M_1 in yellow are rendered in the framework of the resting state in white. Several

arrows mark the motions of K → L → M_1. (b and c) The flattened networks in ground

state and M state. Dashed lines mark H-bonds and salt bridges. Hydrogens, except the

proton on the SB, are omitted due to the ambiguity in their associations with other

atoms. 400 numbers are waters. The guanidinium group of Arg82 is already

protonated with a positive charge at a neutral pH. The phenol hydroxyl groups of Tyr

cannot be protonated anymore. These residues form a proton barrier shaded in blue

between the inner aspartic acids and the outer glutamic acids shaded in red. The H-

bond network in the EC half channel connects the inner acidic residues with the outer

ones through the proton barrier of Arg82, Tyr57, and the main chain carbonyls of Ala81

and Thr205. Proton conductance through the proton barrier occurs only when excess

protons are pumped into the inner EC channel, which becomes super-acidic. The

guanidinium group of Arg82 is protonated for a second time under the extraordinary

proton concentration.

Directional Proton Conductance in Bacteriorhodopsin Is Driven by Concentration Gradient, Not Affinity Gradient

Zhong Ren

Department of Chemistry, University of Illinois at Chicago, Chicago, IL 60607, USA

Renz Research, Inc., Westmont, IL 60559, USA

zren@uic.edu

ORCID 0000-0001-7098-3127

Supplementary Information

Intermediates K and L

Emerged from J around 10 ps, K lasts more than five decades into μs, which suggests that it is a struggle to achieve the next state L over a significant energy barrier. The creased retinal ever since I' at the very first time point (Ren, 2021) is relaxing in K (blue; Figs. 3ef and S8). This slow and difficult relaxation could be driven in part by the energy stored in the creased 13-*cis* retinal and leads to global changes over the entire molecule. First, C_{20} methyl group is tilting less at 20° from its resting orientation (Fig. 2b bottom panel). Although protruded toward the CP direction due to the 13-*cis* configuration, the tilted C_{20} in J is not pointing to Trp182 directly (Fig. 3ef). Therefore, Trp182 is little affected in J. As C_{20} swings back in K, it starts to push Trp182 toward the CP and causes displacements in the main chain (Fig. 3cdef). The positive density that indicates the position of C_{20} in the reconstituted difference map K – bR forms an arc shape centered on the polyene chain (Fig. 3c). This observation strongly suggests that K is a process, rather than a discrete state, with C_{20} distributed along the arc leading toward the next state L, in which the methyl group is fully erected (Fig. 3ef). Second, the entire chromophore in K, except the most distal segment beyond C_8, is in a flat conformation with the double bond $C_{13}=C_{14}$ in *cis* and the single bond $C_\varepsilon–C_\delta$ in *syn*. 12 consecutive torsion angles from C_8 to C_α deviate no more than 20° from *trans/cis* or *anti/syn*. The root-mean-squared deviation (rmsd) is 8.6° (Fig. 2b 4th panel). To achieve such energetically favorable conformation in the chromophore, the retinal anchor Lys216 is driven toward the EC direction, which causes a significant shift of the main chain in the vicinity. Third, again due to the 13-*cis* configuration, Trp86 on the EC side

moves with the proximal segment of the retinal toward the CP direction, which also

causes movement in the main chain. All three developments above are the progresses

in a long and hard work toward the goal to establish the state L with global

consequences.

After five decades of time since tens of ps, the large energy barrier is finally

overcome around several µs as described by the component map U_3 (Fig. S5c) in the

global transition of K → L. This transition is modeled here at a time constant of 2 µs

(Fig. S7). The potential energy in the creased retinal established within tens of fs after

photon absorption is spent to trigger this transition, at least in part. Other sources of

potential energy, such as that stored in the bent helix C (Fig. S10a), could also drive the

global transition. Strong signals in the reconstituted difference map L – bR (Fig. 3d)

support the refined structure of L (pink; Fig. S9). The refined structure of L shows that

the 13-*cis* retinal is finally flattened as its potential energy is consumed. 12 consecutive

torsion angles from C_5 to C_δ, including the SB, deviate from *trans/cis* or *syn/anti* no more

than 23° with an rmsd of 10° (Fig. 2b 4th panel). As a small cost, a couple of single bonds

of Lys216 have to deviate from the perfect *syn* or *anti* conformation like those in the flat

all-*trans* retinal in the resting state. The plane of the flattened 13-*cis* retinal has largely

returned to the original plane of the all-*trans* retinal (Fig. 2b 2nd panel). C_{20} methyl

group is pointing directly to Trp182, which causes helix F to displace (Figs. 3d and

S10b). Leu93 in helix C is also affected by the protruding C_{20} methyl group back into

the original retinal plane. On the other hand, the fully erected C_{20} displaces the anchor

helix G with equally strong signals (Figs. 3d and S10c).

Intermediates M_1 and M_2

An increase of the component map U_1 separating the long delays > 10 µs from the

earlier ones is modeled as the L → M transition at a time constant of 27 µs. The M state

could be split into two by an increase of the component map U_6 at a time constant of 290

µs (Fig. S7). Signals in both U_1 and U_6 are distributed over the EC half of the molecule

instead of globally (Fig. S5ad). Two similar structures M_1 and M_2 (yellow and orange)

are refined according to the reconstituted difference maps M_1 – bR and M_2 – bR (Figs.

S12 and S13). Since the L state, the SB is moving inboard mainly by twisting the torsion

angle around N_ζ-C_ε (Fig. 2b 4th panel). C_ε in M_1 and M_2 has moved 0.6 and 0.8 Å,

respectively, from its position in L. It seems that the conformation is approaching a

point poised to isomerize back to all-*trans*. In the meanwhile, Thr89 and helix C are

being pushed back toward inboard, which restores the kink of helix C in the resting

state (Fig. 4abc). The new development in the retinal displaces helix G further toward

the EC direction as previously observed (Takeda et al., 2004). On the other hand, the

stretched helix F in L springs back in M and becomes shorter than that in the resting

state. The EC segment of helix F turns around to move outboard since L (Fig. 4d).

However, the EC segments of helices C and G squeeze even tighter, while Arg82 swings

out of the way as observed previously (Luecke et al., 2000; Sass et al., 2000).

Mutant functions

Under the mechanism of proton conductance driven by concentration gradient

presented here, a number of mutant functions are explained. How mutants behave, in

return, further support these hypotheses. Proton pumping activity is abolished in

D85N at neutral pH (Mogi et al., 1988). The remaining proton acceptor Asp212 is H-

boned to the phenol hydroxy group of Tyr57 (Fig. S14). After two successful

photocycles, Asp212 and Wat402 are both protonated. However, proton conductance is

stopped at the carbonyl groups =O of both side chains of Asn85 and Asp212, the amide

group -NH$_2$ of Asn85, and the hydroxy groups -OH of Tyr57 and 185, since none of

them can be protonated further. The lack of a proton outlet causes the SB to be

reprotonated quickly during I' or I before isomerization. The proton pump stalls with

repeated futile photocycles. It has been shown that the proton pumping activity of

D85N can be somewhat restored at an alkaline pH (Dickopf et al., 1995), which is

consistent with the concentration-driven proton conductance proposed here. The

phenol hydroxy group of Tyr57, a weak acid, could start to conduct proton to Wat403

and Arg82 only at an alkaline pH by deprotonation. On the other hand, some activity

remains in D212N, which suggests a proton outlet going through Asp85, Wat401, 400,

Arg82, and beyond (Fig. S14). In addition, Asp to Glu mutants are harder to assess

because of their severe structural impacts. Proton pumping activity is also abolished in

D212A (Mogi et al., 1988) perhaps due to a more mobile Wat402 or a hydronium that

reprotonates the SB before isomerization or fails to withhold the proton during the SB

U-turn. This mutant function further underscores the importance that Wat402 or a

hydronium ion maintains its H-bonds to both Asp 85 and 212 throughout the

photocycle as observed in this study despite its large traveling trajectory (Fig. 2a).

1022

Mutants at the site of Arg82 are crucial tests to the mechanisms proposed here. M

formation in the wildtype is pH-dependent at a few μs under alkaline pH and 85 μs at

neutral pH (Balashov et al., 1993; Zimanyi et al., 1992), which nicely illustrates the

concentration-gradient-driven proton conductance proposed here as faster proton

release takes place driven by greater proton gradient. The proton barrier is breached in

R82A mutant (Fig. 5). Therefore, the EC half channel must be filled with a continuous

water network. Very little proton gradient can be established along the water filled

channel during continuous pumping activity. R82A exhibits the same rate of proton

uptake as the wildtype, but a rapid M formation without a pH dependency between

neutral pH and pH10. Such rapid M formation is even faster than the wildtype under

alkaline condition (Balashov et al., 1993). This is obviously due to the inability to

establish the super-acidic inner EC channel without an intact proton barrier so that the

protonation of Asp85 is accelerated compared to the wildtype. More interestingly, the

fast phase of proton release is absent in R82A. Instead, a slow relaxation of 30 ms after

proton uptake was observed (Balashov et al., 1993). Under the mechanism of a

concentration-gradient-driven proton conductance, this slow relaxation is the

spontaneous movement of the newly pumped proton in the same photocycle driven by

a small gradient. Therefore, the "early release" in the fast phase of proton release

cannot be observed without a well-established super-acidic inner EC channel in R82A.

The "early release" observed from the wildtype is actually a much slower proton

movement, in which a proton pumped into the super-acidic inner channel several

photocycles ago is "squeezed out" by a newly pumped proton. This proton released

from the wildtype appears to be coupled with or even faster than the protonation of

Asp85.

1047

Similar to R82A, another mutant R82Q also features rapid, biphasic formation of M

at 1 and 160 μs but delayed proton release after uptake at neutral pH (Govindjee et al.,

1996). However, unlike R82A, proton release from R82Q shows its strong pH-

dependency. At neutral pH, the majority of proton release occurs late at 23 ms. If the

pH increases to 8, most proton release precedes proton uptake at a few ms. The amide

group on the side chain of the neutral residue Gln82 could be protonated by

hydroniums in the inner EC channel. However, the amide group is not a sufficient

proton barrier with its pK_a of -1.4 that could only establish a proton puddle less acidic

compared to that of the wildtype. Another important factor to a smaller proton puddle

is that Gln is two bonds shorter than Arg. As a result, rapid proton release at a few ms

is possible with a sufficient proton gradient under high pH. However, proton release is

delayed to 23 ms as the proton gradient decreases under neutral pH. Again, the proton

apparently released early is one of the protons accumulated in the inner EC channel, not

the one freshly pumped in the same photocycle.

1062

Supplementary Q&A

This study addresses four key questions regarding the operating mechanisms of the

light-driven proton pump bR in addition to the question of isomerization sampling

(Ren, 2021). Answers to these questions are somewhat scattered in the main text while

evidence are presented to support these answers. Here the answers are recapped

without the interruption of the technical details of the supporting data.

1069

1070   Q1) How does the protonated SB with a pK_a of 13.3 transfer its proton to the acidic

1071   acceptors and when?

1072

A1) Which of the two weak interactions N_ζ:H$^+$:O402 survives an abrupt tare while the

retinal prepares to undergo photoisomerization is determined by the inertia of the

proton (Fig. 2c). It is decisive that the proton dissociates from the SB due to its inertia at

the beginning of the retinal photoisomerization. The SB nitrogen makes a rapid U-turn

at ~30 fs (Figs. 2a and S2). A force > 500 pN is required to accelerate the proton as fast

as the SB. Breaking the H-bond H$_2$O402:H$^+$ is nearly as costly as breaking the

association N_ζ:H$^+$ (Figs. 1 inset and 2c). Therefore, the proton joins Wat402 to form a

hydronium ion due to its inertia instead of keeping up to maintain the SB protonated.

Such a proton transfer uphill over 15 pH units within tens of fs is not achieved by a

chemical equilibrium, thus not governed by pK_a values. This proton transfer requires a

high peak power dH/dt localized at the SB. On the other hand, holding Wat402 in place

by both Asp85 and 212 symmetrically from inboard and outboard is as important as the

ultrafast shaking of the SB. This proton transfer uses the same mechanism of the

tablecloth trick. The early deprotonation from the SB is a separate event from the

protonation of Asp85 ten decades later at the formation of M state. Vibrational

signatures from the neutralization of the carboxylates previously observed do not imply

SB deprotonation. This light-driven reaction of a proton pump differs markedly from a

thermal driven chemical equilibrium (Fig. 2c).

1091

1092   Q2) Which molecular event, the isomerization, or another event, switches the

1093   accessibility of the SB from EC side to CP side and how does the SB avoid reprotonation

1094   on the wrong side?

1095

A2) Immediately before isomerization in the I state, the SB is reoriented toward the

inboard while Wat402, a hydronium at this moment, is displaced to the outboard. This

geometry is very bad for proton exchange (Fig. S3). Therefore, the accessibility of the

SB is already limited even before isomerization. However, accessibility switch of the SB

is finally accomplished by the photoisomerization during I → J′. The

photoisomerization must be quick enough to prevent a chemical equilibrium in the EC

half channel. If the equilibrium is reached before isomerization, the SB is reprotonated

from the EC side, and the quantum yield of the proton pump is sacrificed (Govindjee et

al., 1990).

1105

1106   Q3) How are protons conducted > 15 Å through the rollercoasting pK_a values and

1107   released into the EC medium from the SB of a high pK_a through aspartic acids of low

1108 pK_a, arginine (strictly speaking, its guanidinium ion) and tyrosine of high pK_a, and

1109   glutamic acids of low pK_a again?

1110

A3) The EC half channel is obviously not lined with residues of pK_a values in a

monotonic order (Figs. 5 and S1). Despite numerous observations of pK_a changes, no

sufficient evidence supports photoinduced changes to form an ascending order of pK_a

values either. Proton is therefore not conducted along an ascending order of pK_a values

driven by the proton affinity (Stoeckenius, 1999; Stoeckenius et al., 1979). Instead,

proton conductance is driven by concentration gradient, not by affinity gradient. The

guanidinium ion of the arginine and tyrosines of high pK_a, together with several main

chain carbonyls, are arranged to function as a proton levee so that protons could only

overflow outward from a super-acidic inner EC channel established after several

successful cycles of proton pumping. No proton is released during the first a few cycles

of productive pumping until a sufficient proton gradient is established. Under a super-

acidic condition in the inner EC channel, the guanidinium group of Arg82 is protonated

for a second time to form a divalent cation. While it swings outward, the extra proton is

discharged to Glu194, 204, and a water cluster in the outer EC channel, the proton

release complex. Protons could never flow backward into the cell as long as sufficient

water is available in the EC medium. The rollercoasting pK_a values ensure the

concentration-driven outward conductance of protons and prevent the concentration-

driven inward conductance during dark, that is, a unidirectional valve.

1129

1130　Q4) How is a gradient-driven proton leak prevented in the resting state and during the

1131　pumping motions?

1132

A4) A potential proton leak back into the cell during dark is prevented by the exact

mechanism that conducts proton outward (A3). However, the chance of proton leak

increases dramatically during the molecular motions of proton pumping because the

inner EC channel becomes super acidic and the retinal and its surrounding are in

constant motions. Lys216 forms a corner at C_ε that protrudes inboard and makes a van

der Waals contact with Thr89 in helix C (Figs. 2b and S10a). Pressed by the corner at C_ε,

helix C is kinked and bends away. This contact is a seal between two half channels.

This seal could be broken briefly in I′ state but inconsequential. C_δ-C_ε moves outboard

in L state once again due to a flattened 13-*cis* retinal (Figs. 2b and S10a). But this motion

is slow enough to allow the spring-loaded helix C to react. Helix C straightens and

maintains the contact between the chromophore and Thr89 to prevent a concentration-

gradient-driven proton leak back into the CP (Figs. 3f and 4c). During the M states, the

corner at C_ε is reestablished. It pushes Thr89 and helix C to restore its kink.

1146

Methods

From the outset, the key presumption is that every crystallographic dataset, at a given

temperature and a given time delay after the triggering of the photochemical reaction,

captures a mixture of unknown number of intermediate species at unknown fractions.

Needless to say, all structures of the intermediates are also unknown except the

structure at the ground state that has been determined and well refined by static

crystallography. A simultaneous solution of all these unknowns requires multiple

datasets that are collected at various temperatures or time delays so that a common set

of intermediate structures are present in these datasets with variable ratios. If the

number of available datasets is far greater than the number of unknowns, a linear

system can be established to overdetermine the unknowns with the necessary

stereochemical restraints (Ren et al., 2013). This analytical strategy is recapped below.

1159

The methodological advance in this work and the companion work (Ren, 2021) is the

refinement of each pure intermediate structure that has been deconvoluted from

multiple mixtures. Structure factor amplitudes of a single conformation free of

heterogeneity are overdetermined. Unlike the conventional crystallography, such a

deconvoluted dataset of structure factor amplitudes are not observed directly from any

crystal but computed from many experimental datasets. It could be considered as an

extrapolated dataset "on steroids" if compared to the traditional extrapolation of small

differences, such as, 3Fo-2Fc map, a technique often used to overcome a partial

occupancy of an intermediate structure, except that this deconvolution method is an

interpolation among many experimental datasets rather than an extrapolation. The

standard structural refinement software is taken full advantage of with the built-in

stereochemical constraints, e.g. PHENIX (Adams et al., 2010; Liebschner et al., 2019). In

case that the computed deconvolution has not achieved a single pure structural species,

the structural refinement is expected to make such indication.

1174

Difference Fourier maps

A difference Fourier map is synthesized from a Fourier coefficient set of F_{light}-$F_{reference}$

with the best available phase set, often from the ground state structure. Before Fourier

synthesis, F_{light} and $F_{reference}$ must be properly scaled to the same level so that the

distribution of difference values is centered at zero and not skewed either way. A

weighting scheme proven effective assumes that a greater amplitude of a difference

Fourier coefficient F_{light}-$F_{reference}$ is more likely caused by noise than by signal (Ren et al.,

2001, 2013; Šrajer et al., 2001; Ursby and Bourgeois, 1997). Both the dark and light

datasets can serve as a reference in difference maps. If a light dataset at a certain delay

is chosen as a reference, the difference map shows the changes since that delay time but

not the changes prior to that delay. However, both the dark and light datasets must be

collected in the same experiment. A cross reference from a different experimental

setting usually causes large systematic errors in the difference map that would swamp

the desired signals. Each difference map is masked 3.5 Å around the entire molecule of

bacteriorhodopsin (bR). No lipid density is analyzed.

1190

Meta-analysis of protein structures

Structural meta-analysis based on singular value decomposition (SVD) has been

conducted in two forms. In one of them, an interatomic distance matrix is calculated

from each protein structure in a related collection. SVD of a data matrix consists of

these distance matrices enables a large-scale joint structural comparison but requires no

structural alignment (Ren, 2013a, 2013b, 2016). In the second form, SVD is performed

on a data matrix of electron density maps of related protein structures (Ren, 2019; Ren

et al., 2013; Schmidt et al., 2003, 2010). Both difference electron density maps that

require a reference dataset from an isomorphous crystal and simulated annealing omit

maps that do not require the same unit cell and space group of the crystals are possible

choices in a structural meta-analysis (Ren, 2019; Ren et al., 2013). The distance matrices

or the electron density maps that SVD is performed on are called core data. Each

distance matrix or electron density map is associated with some metadata that describe

the experimental conditions under which the core data are obtained, such as

temperature, pH, light illumination, time delay, mutation, etc. These metadata do not

enter the SVD procedure. However, they play important role in the subsequent

interpretation of the SVD result. This computational method of structural analysis takes

advantage of a mathematical, yet practical, definition of conformational space with

limited dimensionality (Ren, 2013a). Each experimentally determined structure is a

snapshot of the protein structure. A large number of such snapshots taken under a

variety of experimental conditions, the metadata, would collectively provide a survey of

the accessible conformational space of the protein structure and reveal its rection

trajectory. Such joint analytical strategy would not be effective in early years when far

fewer protein structures were determined to atomic resolution. Recent rapid growth in

protein crystallography, such as in structural genomics (Chandonia and Brenner, 2006,

2012) and in serial crystallography (Glynn and Rodriguez, 2019; Schaffer et al., 2021),

has supplied the necessarily wide sampling of protein structures for a joint analytical

strategy to come of age. The vacancies or gaps in a conformational space between well-

populated conformational clusters often correspond to less stable transient states whose

conformations are difficult to capture, if not impossible. These conformations are often

key to mechanistic understanding and could be explored by a back calculation based on

molecular distance geometry (Ren, 2013a, 2016), the chief computational algorithm in

nucleic magnetic resonance spectroscopy (NMR), and by a structure refinement based

on reconstituted dataset, a major methodological advance in this work (see below).

These structures refined to atomic resolution against reconstituted datasets may reveal

short-lived intermediate conformation hard to be captured experimentally.

Unfortunately, a protein structure refined against a reconstituted dataset currently

cannot be recognized by the Protein Data Bank (PDB). Because crystallographic

refinement of a macromolecular structure is narrowly defined as a correspondence from

one dataset to one structure. A never-observed dataset reconstituted from a collection

of experimental datasets does not match the well-established crystallographic template

of PDB; let alone a refinement of crystal structure with the NMR algorithm.

1233

A distance matrix contains M pairwise interatomic distances of a structure in the

form of Cartesian coordinates of all observed atoms. An everyday example of distance

matrix is an intercity mileage chart appended to the road atlas. Differences in the

molecular orientation, choice of origin, and crystal lattice among all experimentally

determined structures have no contribution to the distance matrices. Due to its

symmetry, only the lower triangle is necessary. A far more intimate examination of

protein structures in PDB is a direct analysis of their electron density maps instead of

the atomic coordinates. M such (difference) electron densities, often called voxels in

computer graphics, are selected by a mask of interest. In the case of difference maps,

only the best refined protein structure in the entire collection supplies a phase set for

Fourier synthesis of electron density maps. This best structure is often the ground state

structure determined by static crystallography. Other refined atomic coordinates from

the PDB entries are not considered in the meta-analysis. That is to say, a meta-analysis

of difference electron density maps starts from the X-ray diffraction data archived in

PDB rather than the atomic coordinates interpreted from the diffraction data, which

removes any potential model bias.

1250

Singular value decomposition of (difference) electron density maps

An electron density map, particularly a difference map as emphasized here, consists of

density values on an array of grid points within a mask of interest. All M grid points in

a three-dimensional map can be serialized into a one-dimensional sequence of density

values according to a specific protocol. It is not important what the protocol is as long

as a consistent protocol is used to serialize all maps of the same grid setting and size,

and a reverse protocol is available to erect a three-dimensional map from a sequence of

M densities. Therefore, a set of N serialized maps, also known as vectors in linear

algebra, can fill the columns of a data matrix $\mathbf{A}$ with no specific order, so that the width

of $\mathbf{A}$ is N columns, and the length is M rows. Often, $M \gg N$, thus $\mathbf{A}$ is an elongated

matrix. If a consistent protocol of serialization is used, the corresponding voxel in all N

maps occupies a single row of matrix $\mathbf{A}$. This strict correspondence in a row of matrix

$\mathbf{A}$ is important. Changes of the density values in a row from one structure to another

are due to either signals, systematic errors, or noises. Although the order of columns in

matrix $\mathbf{A}$ is unimportant, needless to say, the metadata associated with each column

must remain in good bookkeeping.

1267

SVD of the data matrix $\mathbf{A}$ results in $\mathbf{A} = \mathbf{UWV}^T$, also known as matrix factorization.

Matrix $\mathbf{U}$ has the same shape as $\mathbf{A}$, that is, N columns and M rows. The N columns

contain decomposed basis components $\boldsymbol{U}_k$, known as left singular vectors of M items,

where $k = 1, 2, \ldots, N$. Therefore, each component $\boldsymbol{U}_k$ can be erected using the reverse

protocol to form a three-dimensional map. This decomposed elemental map can be

presented in the same way as the original maps, for example, rendered in molecular

graphics software such as Coot and PyMol. It is worth noting that these decomposed

elemental maps or map components $\boldsymbol{U}_k$ are independent of any metadata. That is to

say, these components remain constant when the metadata vary. Since each left

singular vector $\boldsymbol{U}_k$ has a unit length due to the orthonormal property of SVD (see

below), that is, $|\boldsymbol{U}_k| = 1$, the root mean squares (rms) of the items in a left singular

vector is $1/\sqrt{M}$ that measures the quadratic mean of the items.

1280

The second matrix $\mathbf{W}$ is a square matrix that contains all zeros except for N positive

values on its major diagonal, known as singular values w_k. The magnitude of w_k is

considered as a weight or significance of its corresponding component $\boldsymbol{U}_k$. The third

matrix $\mathbf{V}$ is also a square matrix of $N \times N$. Each column of $\mathbf{V}$ or row of its transpose $\mathbf{V}^T$,

known as a right singular vector V_k, contains the relative compositions of $\boldsymbol{U}_k$ in each of

the N original maps. Therefore, each right singular vector V_k can be considered as a

function of the metadata. Right singular vectors also have the same unit length, that is,

$|V_k| = 1$. Effectively, SVD separates the constant components independent of the

metadata from the compositions that depend on the metadata.

1290

A singular triplet denotes 1) a decomposed component $\boldsymbol{U}_k$, 2) its singular value w_k,

and 3) the composition function V_k. Singular triplets are often sorted in a descending

order of their singular values w_k. Only a small number of n significant singular triplets

identified by the greatest singular values w_1 through w_n can be used in a linear

combination to reconstitute a set of composite maps that closely resemble the original

ones in matrix $\mathbf{A}$, where $n < N$. For example, the original map in the ith column of

matrix $\mathbf{A}$ under a certain experimental condition can be closely represented by the ith

composite map $w_1v_{1i}\boldsymbol{U}_1 + w_2v_{2i}\boldsymbol{U}_2 + \ldots + w_nv_{ni}\boldsymbol{U}_n$, where $(v_{1i}, v_{2i}, \ldots)$ is from the ith row of

matrix $\mathbf{V}$. The coefficient set for the linear combination is redefined here as $c_{ki} =$

$w_kv_{ki}/\sqrt{M}$. The rms of the values in a map component, or the average magnitude

measured by the quadratic mean, acts as a constant scale factor that resets the modified

coefficients c_{ki} back to the original scale of the core data, such as Å for distance matrices

and $e^-/Å^3$ for electron density maps if these units are used in the original matrix $\mathbf{A}$.

Practically, an electron density value usually carries an arbitrary unit without a

calibration, which makes this scale factor unnecessary. In the linear combination $c_{1i}\boldsymbol{U}_1 +$

$c_{2i}\boldsymbol{U}_2 + \ldots + c_{ni}\boldsymbol{U}_n$, each component $\boldsymbol{U}_k$ is independent of the metadata while how much of

each component is required for the approximation, that is, c_{ki}, depends on the metadata.

1308

Excluding the components after $\boldsymbol{U}_n$ in this approximation is based on an assumption

that the singular values after w_n are very small relative to those from w_1 through w_n. As

a result, the structural information evenly distributed in all N original maps is

effectively concentrated into a far fewer number of n significant components, known as

information concentration or dimension reduction. On the other hand, the trailing

components in matrix $\mathbf{U}$ contain inconsistent fluctuations and random noises.

Excluding these components effectively rejects noises (Schmidt et al., 2003). The least-

squares property of SVD guarantees that the rejected trailing components sums up to

the least squares of the discrepancies between the original core data and the

approximation using the accepted components.

1319

1320 　　However, no clear boundary is guaranteed between signals, systematic errors, and

1321 noises. Systematic errors could be more significant than the desired signals. Therefore,

1322 excluding some components from 1 through n is also possible. If systematic errors are

1323 correctly identified, the reconstituted map without these significant components would

1324 no longer carry the systematic errors.

1325

The orthonormal property of SVD

The solution set of SVD must guarantee that the columns in $\mathbf{U}$ and $\mathbf{V}$, the left and right

singular vectors $\mathbf{U}_k$ and $\mathbf{V}_k$, are orthonormal, that is, $\mathbf{U}_h \bullet \mathbf{U}_k = \mathbf{V}_h \bullet \mathbf{V}_k = 0$ (ortho) and $\mathbf{U}_k \bullet \mathbf{U}_k$

$= \mathbf{V}_k \bullet \mathbf{V}_k = 1$ (normal), where $h \neq k$ but both are from 1 to N. The orthonormal property

also holds for the row vectors. As a result, each component $\mathbf{U}_k$ is independent of the

other components. In other words, a component cannot be represented by a linear

combination of any other components. However, two physical or chemical parameters

in the metadata, such as temperature and pH, may cause different changes to a

structure. These changes are not necessarily orthogonal. They could exhibit some

correlation. Therefore, the decomposed components $\mathbf{U}_k$ not necessarily represent any

physically or chemically meaningful changes (see below).

1337

Due to the orthonormal property of SVD, an N-dimensional Euclidean space is

established, and the first n dimensions define its most significant subspace. Each

coefficient set $c_i = (c_{1i}, c_{2i}, \ldots, c_{ni})$ of the ith composite map is located in this n-dimensional

subspace. All coefficient sets for $i = 1, 2, \ldots, N$ in different linear combinations to

approximate the N original maps in a least-squares sense can be represented by N

points or vectors $c_1, c_2, \ldots, c_N$ in the Euclidean subspace. This n-dimensional subspace is

essentially the conformational space as surveyed by the jointly analyzed core data. The

conformational space is presented as scatter plots with each captured structure

represented as a dot located at a position determined by the coefficient set c_i of the ith

observed map. When the subspace has greater dimensionality than two, multiple two-

dimensional orthographical projections of the subspace are presented. These scatter

plots are highly informative to reveal the relationship between the (difference) electron

density maps and their metadata.

1351

If two coefficient sets $c_i \approx c_j$, they are located close to each other in the

conformational space. Therefore, these two structures i and j share two similar

conformations. Two structures located far apart from each other in the conformational

space are distinct in their conformations, and distinct in the compositions of the map

components. A reaction trajectory emerges in this conformational space if the temporal

order of the core data is experimentally determined (Figs. S4 and S7). Otherwise, an

order could be assigned to these structures based on an assumed smoothness of

conformational changes along a reaction trajectory (Ren, 2013a, 2013b, 2016). Causation

and consequence of structural motions could be revealed from the order of the

structures in a series, which may further lead to structural mechanism. In addition, an

off-trajectory location in the conformational space or a location between two clusters of

observed structures represents a structure in a unique conformation that has never been

experimentally captured. Such a hypothetical structure can be refined against a

reconstituted distance matrix using molecular distance geometry (Ren, 2013a, 2013b,

2016) or a reconstituted electron density map with the method proposed below.

1367

Rotation in SVD space

Dimension reduction is indeed effective in meta-analysis of protein structures when

many datasets are evaluated at the same time. However, the default solution set of SVD

carries complicated physical and chemical meanings that are not immediately obvious.

The interpretation of a basis component $\boldsymbol{U}_k$, that is, "what-does-it-mean", requires a

clear demonstration of the relationship between the core data and their metadata. The

outcome of SVD does not guarantee any physical meaning in a basis component.

Therefore, SVD alone provides no direct answer to "what-does-it-mean", thus its

usefulness is very limited to merely a mathematical construction. However, the

factorized set of matrices $\boldsymbol{U}$, $\boldsymbol{W}$, and $\boldsymbol{V}$ from SVD is not a unique solution. That is to say,

they are not the only solution to factorize matrix $\boldsymbol{A}$. Therefore, it is very important to

find one or more alternative solution sets that are physically meaningful to elucidate a

structural interpretation. The concept of a rotation after SVD was introduced by Henry

& Hofrichter (Henry and Hofrichter, 1992). But they suggested a protocol that fails to

preserve the orthonormal and least-squares properties of SVD. The rotation protocol

suggested by Ren incorporates the metadata into the analysis and combines with SVD

of the core data. This rotation achieves a numerical deconvolution of multiple physical

and chemical factors after a pure mathematical decomposition, and therefore, provides

a route to answer the question of "what-does-it-mean" (Ren, 2019). This rotation shall

not be confused with a rotation in the three-dimensional real space, in which a

molecular structure resides.

1389

1390   A rotation in the *n*-dimensional Euclidean subspace is necessary to change the

1391 perspective before a clear relationship emerges to elucidate scientific findings. It is

1392 shown below that two linear combinations are identical before and after a rotation

1393 applied to both the basis components and their coefficients in a two-dimensional

1394 subspace of h and k. That is,

1395

$$c_h \boldsymbol{U}_h + c_k \boldsymbol{U}_k = f_h \boldsymbol{R}_h + f_k \boldsymbol{R}_k, \tag{1}$$

1396

1397 where c_h and c_k are the coefficients of the basis components $\boldsymbol{U}_h$ and $\boldsymbol{U}_k$ before the

1398 rotation; and f_h and f_k are the coefficients of the rotated basis components $\boldsymbol{R}_h$ and $\boldsymbol{R}_k$,

1399 respectively. The same Givens rotation of an angle θ is applied to both the components

1400 and their coefficients:

1401

$$\begin{cases} \boldsymbol{R}_h = \boldsymbol{U}_h \cos\theta - \boldsymbol{U}_k \sin\theta; \\ \boldsymbol{R}_k = \boldsymbol{U}_h \sin\theta + \boldsymbol{U}_k \cos\theta. \end{cases} \tag{2}$$

1402

1403 Obviously, the rotated components $\boldsymbol{R}_h$ and $\boldsymbol{R}_k$ remain mutually orthonormal and

1404 orthonormal to other components. And

1405

$$\begin{cases} f_h = s_h t_h = c_h \cos\theta - c_k \sin\theta; \\ f_k = s_k t_k = c_h \sin\theta + c_k \cos\theta. \end{cases} \tag{3}$$

1406

Here $s_{h|k} = \sqrt{\Sigma f_{h|k}^2}$ are the singular values that replace w_h and w_k, respectively, after the

rotation. They may increase or decrease compared to the original singular values so

that the descending order of the singular values no longer holds. $T_{h|k} = (t_{h|k1}, t_{h|k2}, \ldots,$

$t_{h|kN}) = (f_{h|k1}, f_{h|k2}, \ldots, f_{h|kN})/s_{h|k}$ are the right singular vectors that replace V_h and V_k,

respectively. T_h and T_k remain mutually orthonormal after the rotation and

orthonormal to other right singular vectors that are not involved in the rotation.

1413

1414   Eq. 1 holds because the dot product of two vectors does not change after both

1415 vectors rotate the same angle. To prove Eq. 1 in more detail, Eqs. 2 and 3 are combined

1416 and expanded. All cross terms of sine and cosine are self-canceled:

1417

1418 $f_h \boldsymbol{R}_h + f_k \boldsymbol{R}_k = (c_h \cos\theta - c_k \sin\theta)(\boldsymbol{U}_h \cos\theta - \boldsymbol{U}_k \sin\theta) + (c_h \sin\theta + c_k \cos\theta)(\boldsymbol{U}_h \sin\theta + \boldsymbol{U}_k \cos\theta)$

1419 $\qquad = c_h \boldsymbol{U}_h \cos^2\theta + c_k \boldsymbol{U}_k \sin^2\theta + c_h \boldsymbol{U}_h \sin^2\theta + c_k \boldsymbol{U}_k \cos^2\theta \pm c_h \boldsymbol{U}_k \sin\theta\cos\theta \pm c_k \boldsymbol{U}_h \sin\theta\cos\theta$

1420 $\qquad = c_h \boldsymbol{U}_h (\cos^2\theta + \sin^2\theta) + c_k \boldsymbol{U}_k (\sin^2\theta + \cos^2\theta)$

1421 $\qquad = c_h \boldsymbol{U}_h + c_k \boldsymbol{U}_k$

1422

A rotation in two-dimensional subspace of h and k has no effect in other dimensions,

as the orthonormal property of SVD guarantees. Multiple steps of rotations can be

carried out in many two-dimensional subspaces consecutively to achieve a multi-

dimensional rotation. A new solution set derived from a rotation retains the

orthonormal property of SVD. The rotation in the Euclidean subspace established by

SVD does not change the comparison among the core data of protein structures. Rather

it converts one solution set $\mathbf{A} = \mathbf{UWV}^{\mathrm{T}}$ to other alternative solutions $\mathbf{A} = \mathbf{RST}^{\mathrm{T}}$ so that

an appropriate perspective can be found to elucidate the relationship between the core

data and metadata clearly and concisely.

1432

For example, if one physical parameter could be reoriented along a single dimension

k but not involving other dimensions by a rotation, it would be convincing to show that

the left singular vector $\mathbf{U}_k$ of this dimension illustrates the structural impact by this

physical parameter. Before this rotation, the same physical parameter may appear to

cause structural variations along several dimensions, which leads to a difficult

interpretation. Would a proper rotation establish a one-on-one correspondence from all

physical or chemical parameters to all the dimensions? It depends on whether each

parameter induces an orthogonal structural change, that is, whether structural

responses to different parameters are independent or correlated among one another. If

structural changes are indeed orthogonal, it should be possible to find a proper rotation

to cleanly separate them in different dimensions. Otherwise, two different rotations are

necessary to isolate two correlated responses, but one at a time.

1445

For another example, if the observed core datasets form two clusters in the

conformational space, a rotation would be desirable to separate these clusters along a

single dimension k but to align these clusters along other dimensions. Therefore, the

component $\mathbf{U}_k$ is clearly due to the structural transition from one cluster to the other.

Without a proper rotation, the difference between these clusters could be complicated

with multiple dimensions involved. A deterministic solution depends on whether a

clear correlation exists between the core data and metadata. A proper rotation may

require a user decision. A wrong choice of rotation may select a viewpoint that hinders

a concise conclusion. However, it would not alter the shape of the reaction trajectory,

nor create or eliminate an intrinsic structural feature. A wrong choice of rotation cannot

eliminate the fact that a large gap exists between two clusters of observed core datasets

except that these clusters are not obvious from that viewpoint. A different rotation may
reorient the perspective along another direction. But the structural conclusion would be
equivalent. See example of before and after a rotation in (Ren, 2016).
1460

This rotation procedure finally connects the core crystallographic datasets to the
metadata of experimental conditions and accomplishes the deconvolution of physical or
chemical factors that are not always orthogonal to one another after a mathematical
decomposition. SVD analysis presented in this paper employs rotations extensively
except that no distinction is made in the symbols of components and coefficients before
and after a rotation except in this section. This method is widely applicable in large-
scale structural comparisons. Furthermore, Ren rotation after SVD is not limited to
crystallography and may impact other fields wherever SVD is used. For example, SVD
is frequently applied to spectroscopic data, images, and genetic sequence data.
1470

A preliminary analysis of SVD involving all difference maps (Table S1) shows that
the short and long delays are in two different subspaces (Fig. S15). That is to say,
difference maps in the time regimes before 10 ps and after 10 ns share no common
features. This conclusion cannot be reached without a proper rotation of the
components. The separation between the short and long delays into different subspaces
can only be viewed clearly from this specific perspective shown in Fig. S15. Therefore,
these time regimes are analyzed separately.
1478

1479 *Exponential fitting of SVD coefficients*
1480 The time dependencies of the SVD coefficients $c_1(t)$, $c_3(t)$, and $c_6(t)$ for the long delays >
1481 10 ns are modeled with exponential functions (Fig. S7). Each time-dependent coefficient
1482

$$c_k(t) = b_k + p_k \exp\left(-\frac{t}{\tau_1}\right) + q_k \exp\left(-\frac{t}{\tau_3}\right) + r_k \exp\left(-\frac{t}{\tau_6}\right) \tag{4}$$

1483

where k = 1, 3, or 6. The time constant τ_k associated with the major change in each
coefficient models each step in the reaction scheme K → L → M$_1$ → M$_2$. Whether the
time-dependent coefficients can be successfully fitted with exponentials is also a good
test for a proper rotation. Because the fact that the fractional concentrations of
intermediate species can be expressed as exponentials indicate that the reaction follows
a regular kinetic scheme.

1490

Structural refinement against reconstituted dataset

The linear combination $\Delta\rho(t) = f_1(t)R_1 + f_2(t)R_2 + \ldots + f_n(t)R_n$ after a rotation reconstitutes

one of the observed difference maps at a specific time point t. This time-dependent

difference map depicts an ever-evolving mixture of many excited species. A

reconstituted difference map $\Delta\rho(E)$ for a time-independent, pure, excited species E =

intermediate I′, I, J′, and J deconvoluted from many mixtures would take the same form

except that only one or very few coefficients remain nonzero if a proper rotation has

been found (Table S2). In order to take advantage of the mature refinement software for

macromolecular structures with extensive stereochemical restraints, a set of structure

factor amplitudes is needed. Therefore, it is necessary to reconstitute a set of structure

factor amplitudes that would produce the target difference map $\Delta\rho(E)$ based on a

known structure at the ground state. First, an electron density map of the structure at

the ground state is calculated. This calculated map is used as a base map. Second, this

base map of the ground state is combined with the positive and negative densities in the

target difference map $\Delta\rho(E)$ so that the electron densities at the ground state are skewed

toward the intermediate state. Third, structure factors are calculated from the

combined map. Finally, the phase set of the calculated structure factors is discarded,

and the amplitudes are used to refine a single conformation of the intermediate species

E that $\Delta\rho(E)$ represents.

1510

This protocol following the SVD and Ren rotation of components achieves a

refinement of a pure structural species without the need of alternative conformations.

Several points are noteworthy. First, the minimization protocol in this refinement is

performed against a numerically reconstituted amplitude set that has never been

directly measured from a crystal. This reconstituted dataset could be considered as an

extrapolated dataset "on steroids" if compared to the traditional extrapolation of small

differences, such as, the Fourier coefficient set to calculate a 3Fo-2Fc map, a technique

often used to overcome a partial occupancy of an intermediate structure. An

extrapolation of small differences is not directly observed either but computed by an

exaggeration of the observed difference based on an assumption that the intermediate

state is partially occupied, such as the doubling of the observed difference in 3Fo-2Fc =

Fo + 2(Fo-Fc). In contrast to the conventional technique of extrapolation, the

deconvolution method applied here is an interpolation among many experimental

datasets rather than an extrapolation. Secondly, the deconvolution is a simultaneous

solution of multiple intermediate states mixed together instead of solving a single

excited state.

1527

Second, a map calculated from the ground state structure is chosen as the base map

instead of an experimental map such as Fo or 2Fo-Fc map. If the second step of the

protocol is skipped, that is, no difference map is combined with the ground state map,

the refinement would result in an R factor of nearly zero, since the refinement is

essentially against the calculated structure factors (bR in Table S2). This is to say, the

residuals of the refinement are solely due to the difference component instead of the

base map. This is desirable since errors in the static structure of the ground state are

gauged during its own refinement. On the other hand, if an experimental map is

chosen as a base map, the refinement R factors would reflect errors in both the base map

and the difference map, which leads to a difficulty in an objective evaluation of this

refinement protocol.

1539

Third, the combination of the base map and a difference map is intended to

represent a pure intermediate species. Therefore, alternative conformations in

structural refinement that model a mixture of species would defeat this purpose.

However, this combined map could be very noisy and may not represent a single

species without a proper rotation. This is particular the case, if the target difference

map $\Delta\rho$ is not derived from an SVD analysis and Ren rotation. The SVD analysis

identifies many density components that are inconsistent among all observed difference

maps and excludes them, which greatly reduces the noise content. Therefore, this

refinement protocol may not be very successful without an SVD analysis. Another

source of noise originates from the phase set of the structure factors. Prior to the

refinement of the intermediate structure, the phase set remains identical to that of the

ground state. This is far from the reality when an intermediate structure involves

widespread changes, such as those refined in this study. If the rotation after SVD is not

properly selected, the target difference map would remain as a mixture minus the

ground state. Therefore, the refinement of a single conformation would encounter

difficulty or significant residuals, as judged by the R factors, the residual map, and the

refined structure. A proper solution to this problem is a better SVD solution by Ren

rotation rather than alternative conformations. A successful refinement of near perfect

trans or *cis* double bonds is a good sign to indicate that the reconstituted amplitude set

after a rotation reflects a relatively homogeneous structure. If a double bond could not

be refined well to near perfect *trans* or *cis* configuration, the dataset of structure factor

amplitudes is likely from a mixture of heterogeneous configurations, which occurred

frequently in previous studies of bR and photoactive yellow protein (Jung et al., 2013;

Lanyi and Schobert, 2007; Nogly et al., 2018). It has been a great difficulty in

crystallographic refinement in general that a heterogeneous mixture of conformations

cannot be unambiguously refined even with alternative conformations. This difficulty

becomes more severe when a mixture involves more than two conformations or when

some conformations are very minor.

1568

Lastly, the refinement protocol proposed here could be carried out in the original

unit cell and space group of the crystal at the ground state. However, this is not always

applicable as the original goal of the meta-analysis is a joint examination of all available

structures from a variety of crystal forms. It would be highly desirable to evaluate

difference maps of the same or similar proteins from non-isomorphous crystals together

by SVD. Alternatively, the refinement protocol could also be performed in the space

group of P1 with a virtual unit cell large enough to hold the structure, which is the

option in this study (Table S2). This is to say, the entire analysis of SVD-rotation-

refinement presented here could be extracted and isolated from the original crystal

lattices, which paves the way to future applications to structural data acquired by

experimental techniques beyond crystallography, most attractively, to single particle

reconstruction in cryo electron microscopy.

1581

1582

1583 **Supplementary Tables**

1584

Table S1. Datasets analyzed in this work

Publication	PDB	Label	Resolution	Main conclusions	New findings in this work
Nango et al. Science 354, 1552, 2016	5b6v	dark5	1.77 Å	A newly ordered water on the CP side bridges the SB to Thr89 after isomerization, thus further conducts a proton to Asp85.	These datasets contribute to the overdetermination of the structures K → L → M_1 → M_2. The early creased retinal is flattened; C_{20} methyl group returns into the original plane of the resting retinal as K transitions to L. The most flattened chromophore is achieved in L. The EC half channel contracts in L while the CP half channel starts to open. Reprotonation of the SB starts in L. Several ordered waters are observed in the enlarged CP half channel in M_1. The SB move toward inboard in M_1 and M_2.
	5b6w	16ns	2.0 Å		
	5h2h	40ns	2.0 Å		
	5h2i	110ns	2.0 Å		
	5h2j	290ns	2.0 Å		
	5b6x	760ns	2.0 Å		
	5h2k	2µs	2.0 Å		
	5h2l	5.25µs	2.0 Å		
	5h2m	13.8µs	2.0 Å		
	5b6y	36.2µs	2.0 Å		
	5h2n	95.2µs	2.0 Å		
	5h2o	250µs	2.0 Å		
	5h2p	657µs	2.0 Å		
	5b6z	1.725ms	2.0 Å		
Nogly et al. Science 361, eaat0094, 2018	6g7h	dark6	1.5 Å	Retinal fully isomerizes at 10 ps. But the SB water dissociates earlier.	The short-delay datasets contribute to the structures of I' → I → J' → J. Photoisomerization in J'; retinal binding pocket expansion before 1 ps in I and contraction at 10 ps in J
	6g7i	49-406fs	1.9 Å		
	6g7j	457-646fs	1.9 Å		
	6g7k	10ps	1.9 Å		
	6g7l	8.33ms	1.9 Å		
Kovacs et al. Nat. Commun. 10, 3177, 2019	6ga1	dark1	1.7 Å	The exceedingly high power density of the pump laser causes two-photon absorption. Vibrational motions were observed.	The sub-ps datasets exhibit extensive vibrations at various frequencies. The vibrational signals are widespread over the entire bR molecule and not associated with any structural elements. Therefore, it is concluded that these global vibrations are intrinsic properties of bR induced by short laser pulses. The vibrational signals are more prominent under higher power density of the laser pulses. However, these vibrations are irrelevant to the light-driven proton pumping function of bR.
	6ga2	dark2	1.8 Å		
	6rmk	dark3	1.8 Å		
	6ga7	240fs	1.8 Å		
	6ga8	330fs	1.8 Å		
	6ga9	390fs	1.8 Å		
	6gaa	430fs	1.8 Å		
	6gab	460fs	1.8 Å		
	6gac	490fs	1.8 Å		
	6gad	530fs	1.8 Å		
	6gae	560fs	1.8 Å		
	6gaf	590fs	1.8 Å		
	6gag	630fs	1.8 Å		
	6gah	680fs	1.8 Å		
	6gai	740fs	1.8 Å		
	6ga4	1ps	1.8 Å		
	6ga5	3ps	1.9 Å		
	6ga6	10ps	1.8 Å		
	6ga3	33ms	2.1 Å		
Weinert et al. Science 365, 61, 2019	6rqp	dark	1.8 Å	Helices E and F swing outward after 10 ms.	These datasets contribute to the overdetermination of M_2, in which an open CP half channel is observed. The SB moves toward inboard.
	6rnj	0-5ms	2.6 Å		
	6rph	10-15ms	2.6 Å		
	6rqo	activated	2.0 Å		

1585

Table S2. Refinement statistics

Intermediate		bR	I′	I	J′	J	K	L	M₁	M₂
Time period		0-	< 50 fs	40-700 fs	0.5-2 ps	1-30 ps	3 ps-2 μs	1-100 μs	10-1000 μs	0.5-5 ms
Short delay components	U_{10}			3,300		-4,200				
	U_{14}		2,000	2,700	2,700	2,000				
	U_{17}		3,000		-1,300	-300				
Long delay components	U_1						-700		2,200	1,000
	U_2						1,800	1,800	1,800	1,200
	U_3						-1,000	2,100	1,500	900
	U_6							-600		1,500
Starting model		PDB 6g7h								
Resolution range		50-2.1 Å								
Space group		P1								
Unit cell		a = 62.32 Å; b = 62.32 Å; c = 111.10 Å; α = 90°; β = 90°; and γ = 120°								
Unique reflections		80,354 in working set + 4,236 in test set = 84,590 total								
Completeness		95% in working set + 5% in test set = 100% reconstituted								
R (%)		1.8	29.4	31.0	29.1	30.0	26.0	27.7	28.4	28.5
R_{free} (%)		1.9	31.1	32.4	30.4	30.7	26.3	28.8	29.6	29.7
Refined content		230 protein residues + 1 retinal + water molecules								
Number of atoms		1,798	1,795	1,798	1,796	1,795	1,798	1,794	1,799	1,796
Water molecules		8	5	8	6	5	8	4	9	6
RMSD bonds (Å)		0.005	0.009	0.009	0.009	0.009	0.009	0.009	0.009	0.009
RMSD angles (°)		0.793	1.206	1.105	1.085	1.068	1.006	1.074	1.120	1.081
Rama. favored (%)		98.7	96.5	95.6	96.1	96.5	97.8	96.5	97.4	96.1
Rama. outliers (%)		0.0	0.0	0.4	0.4	0.4	0.4	0.0	0.0	0.4
Clash score		4	9	5	4	6	6	7	10	7

1586

1587

1588   **Supplementary Figures and Legends**

1589

1590

1591

Figure S1. Orthographical views of EC and CP half channels. Parts of the structure are

omitted to reveal the interior. The acidic residues in the half channels are highlighted in

red. Those in the EC half channel are deprotonated. Asp96 in the hydrophobic CP half

channel could be protonated (Gerwert et al., 1989). Arg82 in blue is protonated at

neutral pH to form a positive charged guanidinium ion with a pK_a of 12.5. Tyr57 and 83

with a pK_a of 10 in blue could be deprotonated at alkaline pH.

1598

1599
1600

Figure S2. Two orthographical views of early intermediates. The refined conformations of the retinal and several key side chains are rendered in stick models of various colors. These early intermediates are compared with the resting state in white ribbon and stick model. Wat402 is rendered in small spheres. A zoom-in view of the SB is presented in the inset. N_ζ is shown in blue spheres. Its trajectory marked by a spline curve shows a

1606   sharp U-turn at I′ state. The SB moves toward the distal direction during I and J due to

1607   a creased retinal (Fig. 2b).

1608

1609

Figure S3. Refined structure of I state. (a, b, and c) Three orthographical views of retinal in I state. The refined retinal conformation in purple is compared with the resting state in white. The SB is pointing toward inboard with a perfect *syn* conformation at the single bond N_ζ-C_ε. C_{20} methyl group remains tilted toward outboard. Wat402 with good electron density is displaced toward outboard and remains H-bonded with both Asp85 and 212. (d) Zoom-in view of SB and Wat402. Wat402 is 3.7 Å away from the SB N_ζ. The would-be H-bond directions of the SB and Wat402 are pointing toward the opposite directions as marked by the arrows. This geometry makes not only a H-bond impossible but also any proton transfer difficult.

1623

Figure S4. SVD analysis of difference Fourier maps at long delays > 10 ns. SVD

analyses of difference Fourier maps result in time-dependent coefficients $c_k(t)$, where $k =$

1, 2, …, each corresponding to a time-independent components U_k. Each raw difference

map at a time delay t can be closely represented by a linear combination of these

components, $c_1(t)U_1 + c_2(t)U_2 + …$, that is called a reconstituted difference map. Each of

these components U_k and the reconstituted difference maps can be rendered in the same

way as an observed difference map. The coefficient set $c_k(t)$ is therefore a trace of the

photocycle trajectory, when these time-dependent functions are plotted in a multi-

dimensional space or plotted together against the common variable t. See Methods for

detail. (a) Singular values before and after rotation (Ren, 2019, 2021). 19 components

stand out. (b-g) Time-dependent coefficients of four components U_1, U_2, U_3, and U_6.

Each pair of the adjacent orthographical views, horizontal or vertical, can be folded

along a straight line between them to erect a three-dimensional space. Each colored

trace represents difference maps in a time series calculated with a common reference.

Those time series with a dark reference are plotted with thick lines with dots. Other

series are in thin lines without dots.

1640

1641

Figure S5. Component maps U_1, U_2, U_3, and U_6 from long delays. The main chain and

side chains of the protein are rendered with ribbons and sticks, respectively. The retinal

and Lys216 are in purple sticks. Several key waters are in blue spheres. Parts of the

structure are omitted to reveal more of the interior. The maps are contoured at $\pm3\sigma$ in

green and red, respectively. (a) Two orthographical views of U_1. The signals are

distributed over the EC half of the molecule. (b) Two orthographical views of U_2. The

signals are concentrated around the proximal segment of the retinal. (c) Three

orthographical views of U_3. Widespread signals are associated with all seven helices.

(d) Two orthographical views of U_6. The signals are over the EC half of the molecule.

1653

1654
1655
Figure S6. Component maps U_4 and U_5 from long delays. The main chain and side
chains of the protein are rendered with ribbons and sticks, respectively. The retinal and
Lys216 are in purple sticks. Several key waters are in blue spheres. Parts of the
structure are omitted to reveal more of the interior. The maps are contoured at $\pm 3\sigma$ in
green and red, respectively. These components are not associated with a clear time
dependency. Therefore, they are isolated but not used in structure refinement. (a) Two
orthographical views of U_4. Some signals are distributed over the EC half of the

molecule. (b) Two orthographical views of U_5. Some signals are located around the CP

segments of helices E and F.

1665

1666

Figure S7. Exponential fitting of SVD coefficients of long delays. (a, b, and c) The time dependencies of the SVD coefficients $c_1(t)$, $c_3(t)$, and $c_6(t)$ of the long delays are modeled with exponential functions (Methods). The time constant τ_k associated with the major change in each coefficient models a step in the reaction scheme K → L → M_1 → M_2. τ_3 = 2.1±1 µs or a rate of 470 ms^{-1} is the rate of increase in $c_3(t)$. τ_1 = 27±10 µs or a rate of 37 ms^{-1} describes the jump in $c_1(t)$. τ_6 = 290±100 µs or a rate of 4 ms^{-1} seems to separate M_1 and M_2. The fitted exponential functions are plotted as the gray curves through the black dots derived from SVD in three orthographical views of the three-dimensional subspace of c_1, c_3, and c_6. The large colored dots mark the relatively pure states of K, L, M_1, and M_2. (d) SVD coefficients and their exponential fittings as functions of time. The relatively pure states are marked as color bars at the approximate time. The result of exponential fitting shows the time-dependency clearer. The second component map U_2 seems to be a constant component without a time-dependent trend (Fig. S4cde). It describes significant changes near the SB and the retinal anchor (Figs. 3ab and S5b). The

third component map U_3 increases sharply and consistently around 1 μs (Figs. S4beg).

This component contains widespread signals associated with all seven helices (Fig. S5c),

therefore describes the global changes during K → L transition. The first component U_1

increases at 10 μs (Figs. S4bdf), which could reflect the changes during L → M

transition. U_1 clearly contains signals over the EC half of the molecule (Fig. S5a), which

indicates events related to the proton release. Similarly, the sixth component U_6 also

shows additional signals on the EC half to a lesser extent (Fig. S5d). This component

seems to increase at a rather late time around hundreds of μs (Figs. S4cfg), which splits

the M state into M_1 and M_2.

1692

1693
1694
1695   Figure S8. Two orthographical views of the 2Fo-Fc map of K contoured at 6σ. Here Fo
1696   is the reconstituted structure factor amplitudes rather than observed amplitudes (Table
1697   S2). Fc is the structure factor amplitudes calculated from the refined structure
1698   (Methods). The same applies to the other 2Fo-Fc maps.

1699
1700
Figure S9. Two orthographical views of the 2Fo-Fc map of L contoured at 3.5σ. Here Fo
is the reconstituted structure factor amplitudes rather than observed amplitudes (Table
S2). Fc is the structure factor amplitudes calculated from the refined structure
(Methods).

1705

1707

Figure S10. Irregularities in helices C, F, and G. bR in the resting state and L are

rendered in white and pink, respectively. (a) Kinked helix C. All main chain H-bonds

are regular except that Pro91 in helix C causes a kink. This kink was noticed as soon as

the bR structure was determined to a reasonable resolution (Grigorieff et al., 1996). In

the resting retinal, the corner of C_ε makes a contact with Thr89 and pushes the helix

inboard. In bR, J, K, and L, C_ε moves more and more outboard (Fig. 3f). So does Thr89.

The opposite motions occur in L, M_1, M_2, and back to bR (Fig. 4c). Thr90 also

contributes to the seal. Pro91 is located exactly at right place to facilitate this spring-

loaded bend. (b) Stretched helix F. Pro186 removes one H-bond in helix F. Several

would-be H-bonds in the main chain around Pro186 are stretched to 3.6-4.2 Å in the

resting state, such as Val180O-Ala184N, Leu181O-Tyr185N, and Ser183O-Val187N.

Pro186 and these stretched H-bonds marked by dashed lines effectively breaks helix F

into two segments. In J, K, and L, Trp182 is pushed by C_{20} methyl group more and

more toward the CP. The breakage in helix F stretches even more. P186L mutant is

likely to feature an intact α helix F with uninterrupted main chain H-bonds. Its proton

pumping activity is severely affected (Hackett et al., 1987). A more rigid helix F would

increase the difficult to displace Trp182 toward the CP direction and hinder the opening

of the CP half channel for proton uptake. (c) π helical segment in helix G. The retinal

anchor is in a π helical segment with two main chain H-bonds Val213O-Gly218N and

Ser214O-Phe219N. Unlike the regular α helix, a π helix features H-bonds from the

carbonyl of residue *n* to the amide N of residue *n*+5. During the transition from the α

helix to the π helical segment, Val217N does not have a H-bond partner. As the π

helical segment transitions back to the regular α helical conformation, Ala215O does not

join the H-bond pattern for helical formation. Loss of main chain H-bonds on both

sides of the retinal anchor allows the anchor to move 1.2 Å in L. In addition, two

nearby Gly218 and 220 increase the flexibility of the anchor.

1734

1735

1736

Figure S11. Tighter EC half channel in L. (a-c) Three orthographical views of the EC
half channel. The resting state and L state are rendered in white and pink, respectively.
(d) Atomic displacements in the main chain around the EC half channel from the
resting state to L are marked with arrows 10× as long as the actual values.

1741

1742

1743

1744 Figure S12. Two orthographical views of the 2Fo-Fc map of M_1 contoured at 3σ. Here

1745 Fo is the reconstituted structure factor amplitudes rather than observed amplitudes

1746   (Table S2). Fc is the structure factor amplitudes calculated from the refined structure

1747   (Methods).

1748

1749

Figure S13. Two orthographical views of the 2Fo-Fc map of M2 contoured at 5σ. Here Fo is the reconstituted structure factor amplitudes rather than observed amplitudes (Table S2). Fc is the structure factor amplitudes calculated from the refined structure (Methods).

1756

1757
1758

Figure S14. H-bond network at end of EC half channel in I state. Two curved arrows

mark two possible proton conductance pathways. At more acidic EC pH, the pathway

through Asp85 is in use and that through Asp212 is blocked by Tyr57. At more alkaline

EC pH, Tyr57 could deprotonate thus the pathway through Asp212 may help to

increase the proton flow. Both pathways conduct protons only under a concentration

gradient of protons.

1765

Figure S15. Short and long delays in two orthogonal subspaces. A preliminary SVD analysis shows that the difference Fourier maps at short and long delays do not share common features among the major components. This is an example to show that the third and eighth components are in two orthogonal subspaces. Each colored trace represents difference maps in a time series calculated with a common reference. Those time series with a dark reference are plotted in thick lines with dots. Other series are in thin lines without dots. Short delays are in cool colors and long delays are in warm colors.