Structural and functional properties of a magnesium transporter of the

SLC11/NRAMP family

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Abstract

Members of the ubiquitous SLC11/NRAMP family catalyze the uptake of divalent transition metal

ions into cells. They have evolved to efficiently select these trace elements from a large pool of

Ca^{2+} and Mg^{2+}, which are both orders of magnitude more abundant, and to concentrate them in the

cytoplasm aided by the cotransport of H^+ serving as energy source. In the present study, we have

characterized a member of a distant clade of the family found in prokaryotes, termed NRMTs, that

were proposed to function as transporters of Mg^{2+}. The protein transports Mg^{2+} and Mn^{2+} but not

Ca^{2+} by a mechanism that is not coupled to H^+. Structures determined by cryo-EM and X-ray

crystallography revealed a generally similar protein architecture compared to classical NRAMPs,

with a restructured ion binding site whose increased volume provides suitable interactions with

ions that likely have retained much of their hydration shell.

Introduction

Introduction

Divalent cations constitute important factors in numerous biological processes and they are thus

essential nutrients that need to be imported into cells in the required amounts. From a chemical

perspective, we distinguish the alkaline earth metals Ca^{2+} and Mg^{2+} from transition metal ions such

as Fe^{2+} and Mn^{2+}, which show distinct properties owing to their size and electronic structure. As

major constituent of bone, Ca^{2+} has to be taken up by vertebrates in high amount. Although its

largest part is immobilized within the body, free Ca^{2+} is present in the extracellular medium in low

millimolar concentrations. In contrast, cytoplasmic Ca^{2+} serves as second messenger and its

distribution in the cell thus requires tight control (Carafoli and Krebs, 2016). Mg^{2+} on the other

hand is a ubiquitous ligand of nucleotides and nucleic acids (de Baaij et al., 2015; Maguire and

Cowan, 2002). Despite the similar extracellular concentration, its abundance in the cytoplasm is

several orders of magnitude higher compared to Ca^{2+}. In their interaction with other molecules,

these alkaline earth metal ions are coordinated by electronegative hard ligands, preferentially

oxygens, which compensate their positive charges by either full or partial countercharges (Carafoli

and Krebs, 2016; Payandeh et al., 2013). In contrast to Ca^{2+} and Mg^{2+}, which both constitute minor

elements in the human body, transition metals are trace elements that are found at much lower

concentration as they are required for specialized processes. Transition metals are distinguished

by their incompletely filled d-orbitals. This property permits coordinative interactions with soft

ligands containing free electron pairs and allows them to change their oxidation state, which makes

them important cofactors in redox reactions. The transition metal ion Fe^{2+} plays a central role in

oxygen transport and, together with Mn^{2+}, it is a cofactor of enzymes catalyzing redox reactions.

Due to their low abundance in the extracellular environment, the uptake of transition metals has

thus to proceed with high selectivity to prevent competition with several orders of magnitude more

abundant alkaline earth metals, which would prohibit their efficient accumulation. This challenge is overcome by specific transmembrane transport proteins, which catalyze the accumulation of transition metal ions inside cells. Among these proteins, members of the SLC11/NRAMP family play an important role. They are expressed in all kingdoms of life where they facilitate the transmembrane transport of different transition metal ions by a secondary active process that involves the cotransport of H^+ as energy source (Courville et al., 2006; Nevo and Nelson, 2006). In animals, these proteins are used for the transport of Fe^{2+} (Montalbetti et al., 2013), whereas in prokaryotes the primary substrate is Mn^{2+} (Makui et al., 2000). Besides the transport of both transition metals, plant NRAMPs are also involved in detoxification processes by transporting noxious metal ions such as Cd^{2+} (Huang et al., 2020a).

The transport properties of SLC11/NRAMP proteins have been investigated in numerous functional studies (Gunshin et al., 1997; Mackenzie et al., 2006; Makui et al., 2000), supported by the structures of prokaryotic transporters determined in different conformations (Bozzi et al., 2016b; Bozzi et al., 2019b; Ehrnstorfer et al., 2014; Ehrnstorfer et al., 2017). With respect to its fold, the family is distantly related to other secondary active transporters that include the amino-acid transporter LeuT (Ehrnstorfer et al., 2014; Yamashita et al., 2005), although there is little relationship on a sequence level, SLC11/NRAMP structures provide detailed insight into substrate preference and transport mechanisms. Characterized family members show an exquisite selectivity for transition metals over alkaline earth metal ions, but they poorly discriminate between transition metals (Ehrnstorfer et al., 2014; Gunshin et al., 1997). This is accomplished by a conserved ion binding site that is located in the center of the protein (Ehrnstorfer et al., 2014). Several residues coordinate the largely dehydrated metal ion with hard, oxygen-containing ligands, which would equally well interact with other divalent cations. Additionally, the site allows interaction with the

free electron pair of the thioether group of a conserved methionine residue, which serves as soft

ligand that permits coordinative interactions with transition metal- but not alkaline earth metal ions

(Ehrnstorfer et al., 2014). The mutation of this methionine to alanine in one of the bacterial

homologues had a comparably small effect on transition metal ion transport but instead converted

Ca^{2+} into a transported substrate, which emphasizes the role of the residue in conferring ion

selectivity (Bozzi et al., 2016a).

Although, throughout kingdoms of life, the majority of SLC11/NRMP transporters share similar

functional properties, there are few family members that were described to transport with distinct

substrate preference. Among these family members are prokaryotic proteins, termed NRAMP-

related magnesium transporters (NRMTs), that were identified as uptake systems for Mg^{2+}. In the

bacterium *Clostridium acetobutylicum*, the prototypic NRMT (CabNRMT) has permitted growth

at limiting Mg^{2+} concentrations, even in absence of any alternative transport pathways (Shin et al.,

2014). The altered substrate preference is remarkable since, owing to its small ionic radius and the

consequent high charge density, the interaction of Mg^{2+} with its surrounding solvent is unusually

strong, which makes the dehydration of the ion energetically costly and thus requires unique

features for its transport (Chaudhari et al., 2020; Maguire and Cowan, 2002). The uptake of Mg^{2+}

into cells is thus accomplished by few selective transport systems (Payandeh et al., 2013) that

include the bacterial proteins CorA (Lunin et al., 2006), CorC (Huang et al., 2021), MgtE (Tanaka

et al., 2007) and their eukaryotic homologues (Schaffers et al., 2018; Schweigel-Rontgen and

Kolisek, 2014), the TRP channel TRPM7 (Huang et al., 2020b) and Mg^{2+}-selective P-type pumps

(Maguire, 1992). In the present study we became interested in the mechanism of Mg^{2+} transport

by SLC11 proteins and we thus set out to investigate the functional and structural basis of this

process by combining transport experiments with structural studies. We identified a biochemically

well-behaved NRMT, which allowed us to characterize its transport behavior using *in vitro*

transport experiments. Structure determination by cryo-electron microscopy and X-ray

crystallography revealed an expanded substrate binding site that binds the divalent cation as

complex with most of its hydration shell, which permits selection against the larger Ca^{2+}, which is

not a transported substrate.

Results

Phylogenetic analysis of NRAMP transporters

In light of the proposed altered substrate preference of certain prokaryotic SLC11 proteins, we

became interested in the phylogeny of the family. For that purpose, we have identified numerous

SLC11 homologues in BLAST searches of pro- and eukaryotic sequence databases using human

DMT1, the prokaryotic Mn^{2+} transporters EcoDMT and ScaDMT, the putative NRAMP-related

aluminium transporter (NRAT) from *Orzya sativa* and the proposed Mg^{2+} transporter CabNRMT

as queries. In this way sequences of 1,100 eukaryotic and 447 prokaryotic NRAMP proteins and

745 prokaryotic NRMTs were aligned and subjected to phylogenetic characterization. The

resulting phylogenetic tree displayed in Figure 1 shows a distributed organization with different

family members clustering into separated clades. Branches of this phylogenetic tree group proteins

from the animal kingdom including human DMT1 and NRAMP1, which both function as H^+

coupled Fe^{2+} transporters (Forbes and Gros, 2003; Gunshin et al., 1997), prokaryotic homologues

including proteins of known structure such as ScaDMT (Ehrnstorfer et al., 2014), EcoDMT

(Ehrnstorfer et al., 2017) and DraNRAMP (Bozzi et al., 2019b), whose primary function is H^+

coupled Mn^{2+} transport and plant proteins ascribed to be involved in the transport of various

transition metal ions such as the toxic metal Cd^{2+} (Huang et al., 2020a). Within the plant kingdom,

a small but separated branch encompasses putative Al^{3+} transporters classified as NRATs

(Chauhan et al., 2021; Xia et al., 2010). Finally, a large and distant clade of the family containing

homologues of the prototypic CabNRMT defines a group of transporters which, although clearly

being part of the family, appear to have evolved to serve a different purpose (Figure 1B) (Shin et

al., 2014). The proposed functional distinction between different clades is manifested in the

sequence of the substrate binding site that was identified in structural studies of prokaryotic SLC11

family members to coordinate divalent transition metals (Ehrnstorfer et al., 2014) (Figure 1A,

Figure 1—figure supplement 1). This site is defined by a signature containing a DxxN motif in the

unwound center of α-helix1 and an A/CxxM motif in the unwound center of α-helix 6. The

signature is fully conserved in Fe^{2+} transporters from animals, bacterial Mn^{2+} transporters and most

transporters of plant origin, which all exhibit a strong selectivity against alkaline earth metal ions

but poor discrimination between transition metal ions. Conversely, the motif is replaced by a

DxxN-AxxT motif in NRATs and a DxxG-TxxA/T motif in NRMTs thus indicating that latter

clades would exhibit a different substrate preference (Figure 1—figure supplement 1).

Functional characterization of EcoDMT mutants

To gain further insight into the presumed substrate preference, we investigated how mutations of

ion binding site residues would alter the selectivity of a classical NRAMP protein by characterizing

the functional properties of the bacterial Mn^{2+} transporter EcoDMT. To do so, we employed

previously established proteoliposome-based transport assays with appropriate fluorophores to

detect the uptake of specific ions into vesicles (Ehrnstorfer et al., 2017).

Classical NRAMP proteins efficiently transport divalent transition metal ions with low micromolar

affinity while discriminating against alkaline earth metal ions such as Ca^{2+} and Mg^{2+} (Ehrnstorfer

et al., 2017; Mackenzie et al., 2006). These functional properties are reflected in transport studies

of EcoDMT, which employ the fluorophore calcein as efficient reporter to monitor micromolar

concentrations of Mn^{2+} whereas it is not responsive to either Ca^{2+} or Mg^{2+} even at much higher

concentrations. In our experiments, we find a concentration-dependent quenching of the

fluorescence due to Mn^{2+} influx into the vesicles containing the trapped fluorophore (Figure 2A,

B), which is neither inhibited by the addition of Ca^{2+} nor Mg^{2+} to the outside solution (Figure 2C,

D). Latter would be expected if these ions would compete for the binding of Mn^{2+}. As suggested

from previous studies (Bozzi et al., 2016a; Ehrnstorfer et al., 2014), we expected a key role of the

methionine (Met 235) in the binding site of the protein as major determinant of this pronounced

selectivity. We thus initially investigated whether it was possible to alter the substrate preference

of EcoDMT by shortening the methionine side chain in the mutant EcoDMT M235A. This

construct was still capable of transporting Mn^{2+} with a threefold increase of its K_m (Figure 2A, B,

E, G), whereas now also Ca^{2+} became a permeable substrate as illustrated by the competition of

the ion with Mn^{2+} transport (Figure 2F, G) and the direct detection of Ca^{2+} transport using the

fluorophore Fura2 (Figure 2H). A similar result was previously obtained for the equivalent

mutation of the homologous DraNRAMP (Bozzi et al., 2016a). In contrast to Ca^{2+}, we were not

able to detect any interaction with Mg^{2+} (Figure 2I), indicating that the energetic requirements to

transport this smaller divalent cation are higher than for Ca^{2+}.

Finally, we attempted to confer Mg^{2+} transport properties to EcoDMT by creating the triple mutant

M235A/N54G/A232T, which converts its entire binding site to the residues found in prokaryotic

NRMTs. As for the mutant M235A, this construct folded into a stable protein that retained its

capability of Mn^{2+} transport while remaining unable to interact with Mg^{2+} (Figure 2J-L). In

summary, the mutation of the binding site residues of EcoDMT towards an NRMT was well

tolerated with the resulting constructs retaining their capability for Mn^{2+} transport. This suggests

that the structural and energetic requirements for interactions with this transition metal are lower

than for alkaline earth metals. We also confirmed the role of the binding site methionine as an

important determinant for the counterselection of Ca^{2+}, which due to its much higher concentration

would outcompete Mn^{2+} transport in a physiological environment. However, neither the

replacement of the methionine nor the conversion of the binding site to residues observed in

NRMTs was sufficient to facilitate Mg^{2+} transport, demonstrating that structural features beyond

the immediate binding site residues are required to transport this small divalent cation.

Consequently, we decided to turn our attention towards the characterization of prokaryotic NRMTs

to gain further insight into Mg^{2+} transport by SLC11 homologues.

169   Functional characterization of EleNRMT

Initially, our attempts towards the characterization of NRMTs were focused on the identification

of a suitable candidate for structural and functional studies. Towards this end we cloned and

investigated the expression properties of 82 representative homologues selected from the pool of

745 sequences of putative NRMTs identified in BLAST searches (Figure 1B). Within this subset,

we were able to single out a candidate from the bacterium *Eggerthella lenta* termed EleNRMT

with promising biochemical properties (Figure 3—figure supplement 1). After expression and

purification in the detergent n-dodecyl-beta-D-Maltoside (DDM), we found this protein to elute as

monodisperse peak from a size exclusion chromatography column. (Figure 3—figure supplement

2A). Although much better behaved than the other investigated homologues, EleNRMT was still

comparably instable in detergent solution and difficult to reconstitute into liposomes. We thus

attempted to improve its stability by consensus mutagenesis (Cirri et al., 2018) by converting 9

positions identified in sequence alignments to their most abundant residues in NRMTs to create

the construct EleNRMT^{ts} (Figure 3—figure supplement 1A). This approach improved the thermal

stability of the monomeric protein and the efficiency of reconstitution without affecting its

functional properties (Figure 3, Figure 3—figure supplement 2B-D). Due to the sensitivity of the

assay and the fact that binding site mutants of EcoDMT retained their ability to transport Mn^{2+},

we initially investigated transport properties of EleNRMT for this divalent transition metal ion and

found robust concentration-dependent activity that saturates with a K_m of around 120 µM for either

WT or the thermostabilized mutant (Figure 3A, B, D, E). For both constructs, we found Mg^{2+} to

inhibit Mn^{2+} transport in a concentration-dependent manner, thus indicating that Mg^{2+} would

compete with Mn^{2+} for its binding site (Figure 3C, F). We subsequently quantified binding of

either ion to detergent solubilized protein by isothermal titration calorimetry and found a K_d of

around 100 µM for Mn^{2+}, which corresponds well with the measured K_m of transport of the same

ion and a somewhat lower affinity of around 400 µM for Mg^{2+} (Figure 3G, H, Figure 3—figure

supplement 2E, F). Together, transport and binding experiments demonstrate the capability of

EleNRMT to interact with Mg^{2+} thus providing evidence that it might indeed catalyze Mg^{2+} uptake

into bacteria.

To further characterize Mg^{2+} transport by EleNRMT, we have quantified Mn^{2+} uptake into

proteoliposomes in presence of Mg^{2+} and found a strong interference (Figure 4A, B). By directly

assaying Mg^{2+} with the selective fluorophore Magnesium Green, we observed a concentration-

dependent increase of the transport rate (Figure 4C), with the lack of saturation at high substrate

concentrations reflecting the poor affinity of the dye for the ion. Since our experiments have

confirmed Mg^{2+} as substrate of EleNRMT, we wondered whether the protein would also transport

Ca^{2+} and thus investigated the inhibition of Mn^{2+} uptake by Ca^{2+} and the import of Ca^{2+} into

proteoliposomes monitored with the fluorophore Fura-2 (Figure 4D, E). However, in contrast to

Mg^{2+}, we did in no case find strong evidence for the recognition of this larger alkaline earth metal

ion, which has become a transported substrate for EcoDMT mutants lacking a conserved Met in

the binding site (Figure 2F-H). Latter finding is remarkable since the presumed metal ion binding

site only consists of hard ligands that would also coordinate Ca^{2+} and it thus emphasizes the

presence of distinct structural features of the pocket facilitating Mg^{2+} interactions (Figure 3—

figure supplement 1B).

After confirming the capability of EleNRMT to specifically transport Mn^{2+} and Mg^{2+}, we were

interested whether the uptake of metal ions would be coupled to H^+ as observed for previously

characterized transition metal ion transporters of the family. To this end we employed the

fluorophore ACMA to monitor pH changes that are induced by metal ion symport (Figure 4F-H).

Whereas such concentration-dependent changes are readily observed for EcoDMT (Figure 4H),

there was no acidification of proteoliposomes containing EleNRMT as a consequence of either

Mg^{2+} or Mn^{2+} uptake (Figure 4F, G), thus suggesting that metal ion transport by EleNRMT is not

coupled to H^+. Collectively our functional experiments confirm the role of EleNRMT to function

as uncoupled Mg^{2+} transporter. Due to the high sequence conservation in this subclade of the

SLC11 family, it is safe to assume that this property would likely also be shared by other NRMTs.

Structural characterization of EleNRMT

Following the characterization of the transport properties of EleNRMT, we became interested in

the molecular determinants underlying its distinct selectivity and thus engaged in structural studies

by X-ray crystallography and cryo-EM. Since the monomeric protein, with a molecular weight of

only 47 kDa and lacking pronounced domains extending from the membrane, is itself not a suitable

target for cryo-EM, we initially attempted structural studies by X-ray crystallography but were

unable to obtain crystals, despite extensive screening. To enlarge the size of the protein and

facilitate crystallization, we generated specific nanobodies by immunization of alpacas with

purified EleNRMT[ts] followed by the assembly of a nanobody library from blood samples and

selection by phage display. These experiments allowed us to identify three distinct binders

recognizing the protein termed Nb1-3[EleNRMT] (short Nb1-3) (Figure 4—figure supplement 1A).

Despite the increase of the hydrophilic surface, which was frequently found to facilitate

crystallization, all attempts to obtain crystals of EleNRMT[ts] in complex with a single nanobody

were unsuccessful. Similarly, efforts towards the structure determination of the same complexes

by cryo-EM did not permit a reconstruction at high resolution. We thus attempted to further

increase the size of the complex and found that Nb1 and either one of the two closely related

nanobodies Nb2 or Nb3 were able to bind concomitantly to the protein to assemble into a ternary

complex (Figure 4—figure supplement 1B-D). This complex showed improved crystallization

properties and was of sufficient size to allow structure determination by single particle averaging.

The structure of the tripartite EleNRMT[ts] nanobody complex was determined by cryo-EM in

absence of additional Mg^{2+} (EleNRMT[ts]-Nb1,2) and in buffer containing 10 mM Mg^{2+}

(EleNRMT[ts]Mg^{2+}-Nb1,2). Datasets collected from respective samples allowed the 3D-

reconstruction of a map at 3.5 Å for EleNRMT[ts]-Nb1,2 and at 4.1 Å for EleNRMT[ts]Mg^{2+}-Nb1,2

(Figure 5—figure supplements 1 and 2, Table 1). Both maps display equivalent conformations of

the complex and permitted its unambiguous interpretation by an atomic model defining the features

of an NRMT (Figure 5A, B, Figure 5—figure supplement 3). The structures show a protein

complex consisting of the transporter EleNRMT^{ts} and two nanobodies binding to non-overlapping

epitopes on the extracellular side (Figure 5A and B, Figure 5—figure supplement 4). This complex

structure was subsequently used as search model for molecular replacement for a dataset collected

from a crystal of the EleNRMT^{ts}-Nb1,2 complex grown in presence of 50 mM Mg^{2+} and extending

to 4.1 Å (Figure 5—figure supplement 5A, Table 2). In the X-ray structure, both copies of the

complex in the asymmetric unit closely resembled the cryo-EM model, thus providing an ideal

system to determine the location of bound ions by anomalous scattering experiments (Figure 5—

figure supplement 5A-C). Towards this end, we have soaked our crystals in solutions where Mg^{2+}

was replaced by Mn^{2+} in the expectation that latter would occupy the binding site as suggested

from functional experiments (Figure 3) and collected data at the anomalous absorption edge of the

transition metal ion. This data allowed the unambiguous localization of Mn^{2+} in the anomalous

difference density, which showed a single strong peak per complex located in the equivalent

position as previously identified for the binding site of ScaDMT (Ehrnstorfer et al., 2014) (Figure

5C, Figure 5—figure supplement 5B, C). In similar location, we find residual density in our cryo-

EM datasets thus suggesting that Mg^{2+} and Mn^{2+} would both occupy the same site of the transporter

(Figure 5—figure supplement 5D, E).

EleNRMT structure and ion coordination

The structure of EleNRMT^{ts} shows a transporter that displays the general organization of the

SLC11 family. It encompasses 11 membrane-spanning segments, with the first 10 segments

organized as two structurally related repeats of 5 α-helices that are inserted in the membrane with

opposite orientations (Figure 6A, Figure 3—figure supplement 1). The protein adopts an inward-

facing conformation that is illustrated in the comparison with the equivalent conformations of the

transporters ScaDMT and DraNRAMP, which both superimpose with RMSDs of 2.2 and 2.0 Å,

respectively (Figure 6B). This structure contains a large and continuous water-filled cavity that

extends from the cytoplasm and leads to a pocket located in the center of the protein at the unwound

parts of α-helices 1 and 6, which together form the ion binding site (Figures 5C and 6C). The

location of the bound ion is defined in the anomalous difference density of Mn^{2+} and its position

presumably also corresponds to the binding position of Mg^{2+} (Figure 5C, Figure 5—figure

supplement 5B-E). Although the bound ion is found at an equivalent position as in the transporter

ScaDMT, the binding pocket of EleNRMT is expanded as a consequence of the about 1.5 Å shift

in the location of backbone atoms and differences in the volume of interacting side chains (Figure

7A-C). In both, Mg^{2+} and Mn^{2+} transporters, an Asp on α-helix 1 and a main chain interaction at

the site of the unwound part of α-helix 6a remain a common feature of their binding sites, whereas

the replacement of an alanine (Ala 223) in ScaDMT by a threonine (Thr 224) in EleNRMT

introduces a side chain containing a hydroxyl group whose oxygen atom serves as additional ligand

for ion interactions (Figure 7B). Further changes of other residues of the binding site alter its shape

and volume. In ScaDMT, the bound ion is to a large extent excluded from the aqueous environment

and instead tightly surrounded by residues of the binding site, whereas in EleNRMT the

replacement of the asparagine side chain on α1 with a glycine and the methionine residue on α6 to

alanine in conjunction with differences in the main chain positions widen the cavity to leave

sufficient space for interacting water molecules (Figure 7A-C), suggesting that Mg^{2+} binds to the

transporter in a partly solvated state retaining much of its first hydration shell. Thus, the EleNRMT

structure implies that it is a combination of both, changes in the local environment conferred by

the chemical nature of side chains and more delocalized changes causing an expansion of the

binding site, which together are responsible for the altered selectivity.

Similar to the altered substrate preference, also the second functional hallmark of EleNRMT

concerning the absent coupling of metal ion transport to protons, is manifested in its structure

(Figure 7D, E). Proton coupling in classical NRAMP transporters was assigned to structural

features extending from the ion binding site (Bozzi et al., 2019a; Bozzi et al., 2020; Ehrnstorfer et

al., 2017; Mackenzie et al., 2006; Pujol-Gimenez et al., 2017). These include a conserved histidine

on α-helix 6b located two residues downstream of the binding site methionine and a continuous

path of acidic and basic residues on α-helices 3 and 9 surrounding a narrow aqueous cavity, which

together appear to constitute an intracellular H^+ release pathway. These residues are altered in

NRMTs where the histidine on α6 is replaced by a tryptophane, which cannot accept protons, and

where the polar pathway is filled by hydrophobic residues (Figure 7B, D. Figure 3—figure

supplement 1B).

Functional properties of EleNRMT binding site mutants

After identifying the structural determinants of ion interactions in NRMTs, we were interested in

the consequence of mutations of residues of the binding site on selectivity and thus investigated

the transport properties of mutants. Consistent with experiments on other family members, the

mutation of the aspartate in the binding site (Asp55), which is a universally conserved position in

all SLC11 proteins, strongly compromises transport (Figure 8A, B). Similarly, the mutation of Gln

379 on α10, a conserved position that was proposed to be involved in ion coordination in the

occluded state of DraNRAMP (Bozzi et al., 2019b), exerts a similarly strong effect (Figure 8A, C,

Figure 3–figure supplement 1B), thus suggesting that EleNRMT undergoes an equivalent

conformational change. In contrast, the mutation of Thr 224 to alanine, the amino acid found in

the equivalent position of NRAMP transporters, has smaller but still detectable impact on Mn^{2+}

transport, which is reflected in an at least two-fold increase of its K_m (Figures 8A, D, E). In contrast

to Mn^{2+}, this change has a much more pronounced effect on Mg^{2+} interactions, which is manifested

in the absence of appreciable inhibition of Mn^{2+} transport in presence of increasing Mg^{2+}

concentrations and the non-detectable Mg^{2+} transport when directly assaying this alkaline earth

metal ion (Figure 8F, G). Finally, by replacing Gly 88 to asparagine in the background of the

T224A mutant, we create an EleNRMT construct with a similar binding site signature as the

EcoDMT mutant M235A, which showed increased Ca^{2+} permeability. We thus investigated

whether an equivalent conversion would also change the properties of EleNRMT with respect to

its ability to transport Ca^{2+}, which is not a substrate of the WT protein (Figure 4D, E). In line with

the pronounced effect already observed for T224A (Figure 8D, E), we see a further reduction of

Mn^{2+} transport in the double mutant T224A/G88N (Figure 8H). In contrast to the compromised

Mn^{2+} transport properties, this double mutant shows detectable Ca^{2+} transport activity, thus

illustrating that the conversion has introduced structural determinants that facilitate Ca^{2+}

interactions and that change the properties of the binding site to more closely resemble NRAMP

proteins.

Discussion
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In our study we have investigated the mechanistic basis of ion selectivity in the SLC11/NRAMP

family. Whereas the bulk of the family efficiently transports divalent transition metal ions such as

Fe^{2+} or Mn^{2+} from a high background of alkaline earth metal ions, there is an extended clade of

the family in prokaryotes that has evolved as transport systems of the divalent cation Mg^{2+} and

that were thus termed NRAMP related Mg^{2+} transporters (NRMTs) (Shin et al., 2014) (Figure 1).

These proteins are characterized by a distinct signature of residues constituting the ion binding

site, located in the center of the protein in the unwound parts of the pseudo-symmetry related

transmembrane helices α1 and α6 (Figs. 1, 7). We thus initially investigated whether we would be

able to confer the capability to transport Mg^{2+} to a transition metal ion transporter by mutation of

its ion binding site.

The most striking difference in the ion binding site of both clades of the NRAMP family concerns

the position of a conserved methionine that serves as a soft ligand for the interaction with transition

metal ions, which is replaced by an alanine in NRMTs (Figure 1). The mutation of this residue in

the classical NRAMP EcoDMT only moderately affected its capability to transport Mn^{2+} and

instead exerted its pronounced effect by converting Ca^{2+} into a transported substrate (Fig2 A-H).

This emphasizes the primary role of this residue to disfavor binding of this ubiquitous alkaline

earth metal ion, which was already proposed in a previous study (Bozzi et al., 2016a). Similarly,

an EcoDMT mutant where all three deviating residues of the binding site were mutated to their

correspondents found in NRMTs was still efficient in Mn^{2+} transport (Figure 2E), thus suggesting

that, among divalent cations, the transport of transition metal ions might not depend on stringent

structural prerequisites. In contrast to Ca^{2+}, neither of the mutants converted EcoDMT into a Mg^{2+}

transporter (Figure 2D, K), thus suggesting that the transport of latter might require specific

features to cope with the unique chemical properties of this ion. Among divalent cations of

biological relevance, Mg^{2+} is distinguished by its small ionic radius, which results in an unusually

strong interaction with the surrounding water molecules that would render its dehydration during

transport energetically costly (Persson, 2010). The first hydration shell of Mg^{2+}, which coordinates

the ion in an octahedral geometry (Figure 9A), was thus suggested to remain bound to the ion in

its initial encounter with the binding site and probably also during transport. A similar feature was

indeed observed in the structure of the unrelated bacterial Mg^{2+} transport proteins MgtE (Takeda

et al., 2014) and CorA (Guskov et al., 2012) whereas protein interactions in the membrane-inserted

domain of the transporter CorC are established with a fully dehydrated ion (Huang et al., 2021).

The distinct requirements for Mg^{2+} transport are reflected in the properties of EleNRMT

investigated in this study. Its capability to transport Mn^{2+} (Figure 3A-D, G) underlines the

described promiscuity of this ion in its interaction with proteins. In contrast, our studies have

shown transport of Mg^{2+} but not Ca^{2+} (Figs. 3E, F, H and 4A-E) thus illustrating that both ions rely

on distinct features of the protein which mediate their interaction. These are illustrated in the

structure of EleNRMT in an inward-facing conformation determined by cryo-EM (Figure 5 and

6). In this structure, a wide aqueous cavity leads to an ion binding site whose spacious volume

would preclude simultaneous interaction with a completely dehydrated ion (Figs 6C and 7B, C).

The larger size of the site is an immediate consequence of the decreased side chain-volume of the

surrounding residues concomitant with local differences in the backbone positions (Figure 7A, B),

which explains why it was impossible to convert EcoDMT into a Mg^{2+} transporter upon mutation

of respective residues (Figure 2D, I, K). The position of a Mn^{2+} ion bound to EleNRMT was

defined by its anomalous difference density in a crystal structure of the protein (Figure 5C). Since

Mn^{2+} shows a similar octahedral coordination as Mg^{2+}, although with weaker affinity to the

surrounding waters, we assume Mg^{2+} to bind in an equivalent manner, which is also reflected in

the presence of residual density in a dataset determined at high Mg^{2+} concentration by cryo-EM

(Figure 5-figure Supplement 5E). After placement of an octahedral Mg^{2+}-water complex, it is

apparent that only one of the solvent molecules would be replaced by a direct interaction with the

oxygen of the amide backbone group of Thr 224 whereas other interactions with the conserved

binding site would be mediated by coordinating waters (Figure 9B, C). The detailed structural

properties of this cavity are expected to change upon a transition from and inward-open into an

occluded intermediate conformation, which was observed in the corresponding structure of

DraNRAMP, where the access of the binding site to the cytoplasm is closed (Bozzi et al., 2019b).

However, in contrast to DraNRAMP, where the bound ion becomes tightly encircled by protein

residues, the distinct structural features of EleNRMT and the lower side chain volume of residues

of its binding site suggest that this cavity would be sufficiently big for Mg^{2+} to retain much of its

surrounding water during transport (Figure 9D-F) thus emphasizing that it is the larger hydrated

complex of Mg^{2+} rather than the naked ion that is relevant for transport.

A second pronounced feature that distinguishes EleNRMT and likely other NRMTs from classical

SLC11 proteins, concerns the absent coupling of metal ion transport to an energy source, which in

case of NRAMPS is conferred by the cotransport of protons (Ehrnstorfer et al., 2017; Gunshin et

al., 1997) (Figure 4F-H). This property is reflected in the location of ionizable residues that are

conserved within the proton- coupled part of the family and that are absent in EleNRMT (Figure

7E, F). It thus appears that in light of the high extracellular Mg^{2+} concentration and the inside

negative membrane potential such coupling may not be required for efficient uptake in the natural

habitat of the bacterium.

Collectively, the presented study has provided detailed mechanistic insight into the structural basis

of Mg^{2+} transport in homologues of the SLC11/NRAMP family and it has confirmed previous

results that have proposed this role based on biochemical considerations (Shin et al., 2014). Our

work also illustrates the promiscuity of transition metal ion transport, which contrasts the tight

requirements for the alkaline earth metal ions such as Ca^{2+} and Mg^{2+}, latter of which is transported

in a largely hydrated state. The observation that the previously investigated NRMT homologue did

not permit growth upon Mn^{2+} -limitation (Shin et al., 2014) could be a consequence of the lower

affinity to this ion, the competition with abundant Mg^{2+} and the absence of an energy source, which

might be mandatory for efficient accumulation of the scarce substrate. Our study illustrates the

specific requirements for the transport of Mg^{2+} which are not met by classical NRAMP

transporters. In respect, it appears unlikely that the human transporter SLC11A1, which resides in

the phagosomes of macrophages and is involved in the immune response against invading

microbes, would itself permit Mg^{2+} transport as suggested in a recent study (Cunrath and Bumann,

2019).

Materials and Methods

Key resources table

Reagent type	Designation	Source or reference	Identifiers	Additional information
Chemical compound, drug	1-palmitoyl-2-oleoyl-sn-glycero-3-phospho-(1'rac-glycerol) (18:1 06:0 POPG)	Avanti Polar Lipids	840457C	
Chemical compound, drug	E.coli polar extract	Avanti Polar Lipids	100600P	
Chemical compound, drug	1-palmitoyl-2-oleoyl-sn-glycero-3-phosphoethanolamine (18:1 06:0 POPE)	Avanti Polar Lipids	850757C	

Chemical compound, drug	Triton X-100	Sigma	Cat#T9284	
Chemical compound, drug	Lysozyme	Applichem	Cat#A3711	
Chemical compound, drug	Benzamidine	Sigma	B6506	
Chemical compound, drug	Chloroform	Fluka	25690	
Chemical compound, drug	DM	Anatrace	D322	
Chemical compound, drug	DDM	Anatrace	D310S	
Chemical compound, drug	Diethyl ether	Sigma	296082	
Chemical compound, drug	DNase I	AppliChem	A3778	
Chemical compound, drug	glycerol 99%	Sigma	G7757	
Chemical compound, drug	HCl	Merck Millipore	1.00319.1000	
Chemical compound, drug	HEPES	Sigma	H3375	
Chemical compound, drug	Imidazole	Roth	X998.4	
Chemical compound, drug	L-(+)-arabinose	Sigma	A3256	

Chemical compound, drug	Leupeptin	AppliChem	A2183	
Chemical compound, drug	Pepstatin	AppliChem	A2205	
Chemical compound, drug	Valinomycin	Thermofischer Scientific	V1644	
Chemical compound, drug	Calcimycin	Thermofischer Scientific	A1493	
Chemical compound, drug	Ionomycin	Thermofischer Scientific	I24222	
Chemical compound, drug	Fura2	Thermofischer Scientific	F1200	
Chemical compound, drug	Magnesium Green	Thermofischer Scientific	M3733	
Chemical compound, drug	ACMA	Thermofischer Scientific	A1324	
Chemical compound, drug	CCCP	Sigma	C2759	
Chemical compound, drug	Calcein	Thermofischer Scientific	C481	
Chemical compound, drug	Phosphate buffered saline	Sigma	D8537	
Chemical compound, drug	PEG400	Sigma	91893	
Chemical compound, drug	Manganese acetate	Sigma	330825	

Chemical compound, drug	Magnesium acetate	Fluka	63047	
Chemical compound, drug	Calcium Chloride	Sigma	223506	
Chemical compound, drug	Manganese Chloride	Fluka	31422	
Chemical compound, drug	Magnesium Chloride	Fluka	63065	
Chemical compound, drug	PMSF	Sigma	P7626	
Chemical compound, drug	Potassium chloride	Sigma	746346	
Chemical compound, drug	Sodium chloride	Sigma	71380	
Chemical compound, drug	Terrific broth	Sigma	T9179	
Commercial assay or kit	4–20% Mini-PROTEAN® TGX™ Precast Protein Gels, 15-well, 15 µl	BioRad Laboratories	4561096DC	
Commercial assay or kit	Amicon Ultra-4 Centrifugal Filters Ultracel 10K, 4 ml	Merck Millipore	UFC801096	
Commercial assay or kit	Amicon Ultra-4 Centrifugal Filters Ultracel 50K, 4 ml	Merck Millipore	UFC805096	
Commercial assay or kit	Amicon Ultra-4 Centrifugal Filters Ultracel 100K, 4 ml	Merck Millipore	UFC810024	

Commercial assay or kit	Biobeads SM-2 adsorbents	BioRad Laboratories	152-3920	
Commercial assay or kit	Avestin Extruder kit	Sigma	Cat#Z373400	
Commercial assay or kit	400nm polycarbonate filters	Sigma	Cat#Z373435	
Commercial assay or kit	96-well black walled microplates	Thermofischer Scientific	Cat#M33089	
Commercial assay or kit	Ni-NTA resin	ABT Agarose Bead Technologies	6BCL-NTANi-X	
Commercial assay or kit	QuantiFoil R1.2/1.3 Au 200 mesh	Electron Microscopy Sciences	Q2100AR 1.3	
Commercial assay or kit	Superdex 200 10/300 GL	Cytiva	17517501	
Commercial assay or kit	Superdex 200 Increase 3.2/300	Cytiva	28990946	
Commercial assay or kit	Superdex 200 Increase 5/150 GL	Cytiva	28990945	
Commercial assay or kit	Superdex 75 10/300 GL	Cytiva	17517401	
Other	BioQuantum Energy Filter	Gatan	NA	
Other	HPL6	Maximator	NA	
Other	K3 Direct Detector	Gatan	NA	

Other	Titan Krios G3i	ThermoFisher Scientific	NA	
Other	Viber Fusion FX7 imaging system	Witec	NA	
Other	TECAN M1000 Infinite	TECAN	NA	
Other	Vitrobot Mark IV	ThermoFisher Scientific	NA	
Other	µDAWN MALS Detector	Wyatt Technology	NA	
recombinant DNA	gDNA Eggerthella lenta	DSMZ	2243	
recombinant DNA	gDNA Eremococcus coleocola	DSMZ	15696	
recombinant DNA	bacterial expression vector with C-terminal 3C cleavage site, GFP-tag and His-tag	Dutzler laboratory	NA	
recombinant DNA	bacterial expression vector with N-terminal His-tag and 3C cleavage site	Dutzler laboratory	NA	
recombinant DNA	bacterial expression vector with N terminal pelB sequence, His-tag, MBP, 3C cleavage site	Dutzler laboratory	NA	
recombinant protein	HRV 3C protease	Expressed (pET_3C) and purified in Dutzler laboratory	NA	
Software, algorithm	3DFSC	Tan et al., 2017	https://3dfsc.salk.edu/	

Software, algorithm	ASTRA7.2	Wyatt Technology	https://www.wyatt.com/products/software/astra.html	RRID: SCR_016255
Software, algorithm	Chimera v.1.15	Pettersen et al., 2004	https://www.cgl.ucsf.edu/chimera/	RRID: SCR_004097
Software, algorithm	ChimeraX v.1.1.1	Pettersen et al., 2021	https://www.rbvi.ucsf.edu/chimerax/	RRID: SCR_015872
Software, algorithm	Coot v.0.9.4	Emsley et al., 2010	https://www2.mrc-lmb.cam.ac.uk/personal/pemsley/coot/	RRID: SCR_014222
Software, algorithm	cryoSPARC v.3.0.1/v.3.2.0	Structura Biotechnology Inc.	https://cryosparc.com/	RRID: SCR_016501
Software, algorithm	DINO		http://www.dino3d.org	RRID: SCR_013497
Software, algorithm	EPU2.9	ThermoFisher Scientific	NA	
Software, algorithm	Phenix	Liebschner et al., 2019	https://www.phenix-online.org/	RRID: SCR_014224
Software, algorithm	RELION 3.0.7	Zivanov et al., 2018	https://www3.mrc-lmb.cam.a	RRID: SCR_016274

			c.uk/relion/	
Software, algorithm	WEBMAXC calculator	Bers et al., 2010	https://somapp.ucdmc.ucdavis.edu/pharmacology/bers/maxchelator/webmaxc/webmaxcS.htm	RRID: SCR_018807
Software, algorithm	JALVIEW	Waterhouse et al., 2009		
Software, algorithm	Muscle	Edgar, 2004		
Strain	E. coli MC1061	ThermoFisher Scientific	C66303	

Phylogenetic Analysis of the SLC11 family

The protein sequences of ScaDMT (UniprotKB A0A4U9TNH6), EcoDMT (UniprotKB E4KPW4), hDMT1 (UniprotKB P49281), OsNRAT1 (Uniprot KB Q6ZG85), CabNRMT-c0685 (UniprotKB Q97L77) and CabNRMT-c3329 (UniprotKB Q97DZ1) were respectively used as query sequences in BLASTp searches in diverse sequence databases. In that manner, 1100 eukaryotic and 447 prokaryotic standard SLC11 transporters and 745 prokaryotic NRMTs were selected and aligned with CLC Main Workbench (Quiagen). The identity of the 745 prokaryotic

NRMTs was confirmed by individual inspection of the proposed ion binding site. A multiple

sequence alignment was used to generate a phylogenetic tree using the neighbor joining cluster

generation method. Protein distances were measured using a Jukes-Cantor model and a

Bootstrapping analysis was performed with 100 replicates.

Expression and purification of EcoDMT

EcoDMT and its mutants were expressed and purified as described previously (Ehrnstorfer et al.,

2017). Briefly, the vector EcoDMT-pBXC3GH (Invitrogen) was transformed into *E. coli* MC1061

cells and a preculture was grown in TB-Gly medium overnight using ampicillin (100 µg/ml) as a

selection marker for all expression cultures. The preculture was used at a 1:100 volume ratio for

inoculation of TB-Gly medium. Cells were grown at 37 °C to an OD_{600} of 0.7-0.9, after which the

temperature was reduced to 25 °C. Protein expression was induced by addition of arabinose to a

final concentration of 0.004% (w/v) for 12-14 h at 18 °C. Cells were harvested by centrifugation

for 20 min at 5,000 g, resuspended in buffer RES (200 mM NaCl, 50 mM KPi pH 7.5, 2 mM

$MgCl_2$, 40 µg/ml DNAseI, 10 µg/ml lysozyme, 1µM leupeptin, 1 µM pepstatin, 0.3 µM aprotinin,

1mM benzamidine, 1mM PMSF) and lysed using a high pressure cell lyser (Maximator HPL6).

After a low speed spin (10,000 g for 20 min), the supernatant was subjected to a second

centrifugation step (200,000 g for 45 min). The pelleted membrane was resuspended in buffer EXT

(200 mM NaCl, 50 mM KPi pH 7.5, 10% glycerol) at a concentration of 0.5 g of vesicles per ml

of buffer. The purification of EcoDMT and its mutants was carried out at 4°C. Isolated membrane

fractions were diluted 1:2 in buffer EXT supplemented with protease inhibitors (Roche cOmplete

EDTA-free) and 1% (w/v) n-decyl-β-D-maltopyranoside (DM, Anatrace). The extraction was

carried out under gentle agitation for 2 h at 4 °C. The lysate was cleared by centrifugation at

200,000 g for 30 min. The supernatant supplemented with 15 mM imidazole at pH 7.5 was

subsequently loaded onto NiNTA resin and incubated for at least 1 h under gentle agitation. The

resin was washed with 20 column volumes (CV) of buffer W (200 mM NaCl, 20 mM HEPES pH

7, 10% Glycerol, 50 mM imidazole pH 7.5, 0.25% (w/v) DM) and the protein was eluted with

buffer ELU (200 mM NaCl, 20 mM HEPES pH 7, 10% Glycerol, 200 mM imidazole pH 7.5,

0.25% (w/v) DM). Tag cleavage proceeded by addition of HRV 3C protease at a 3:1 molar ratio

and dialysis against buffer DIA (200 mM NaCl, 20 mM HEPES pH7, 10% Glycerol, 0.25% (w/v)

DM) for at least 2 h at 4 °C. After application to NiNTA resin to remove the tag, the sample was

concentrated by centrifugation using a 50 kDa molecular weight cutoff concentrator (Amicon) and

further purified by size exclusion chromatography using a Superdex S200 column (GE Healthcare)

pre-equilibrated with buffer SEC (200 mM NaCl, 20 mM HEPES pH 7, 0.25% (w/v) DM); The

peak fractions were pooled, concentrated, and used directly without freezing.

Expression screening of NRMT homologues

The DNA coding for NRMTs from 82 bacterial and archaeal strains were cloned from gDNA

(obtained from DSMZ) into pBXNHG3 or pBXC3H vectors (Invitrogen) using FX cloning

(Geertsma and Dutzler, 2011) providing a final construct where an NRMT contains a 10-His tag,

a GFP and a HRV 3C protease cleavage site fused to its N-terminus (pBXNHG3) or a HRV 3C

protease cleavage site and a 10-His tag fused to its C-terminus (pBXC3H). Both vectors contain

an arabinose promoter and an ampicillin resistance gene. Consequently, all cultures were

supplemented with 100 μg/ml ampicillin as selection marker. The vectors were transformed into

E. coli MC1061 cells and precultures were grown in Terrific Broth (TB) medium overnight. The

small-scale expression was performed in 24 well plates with gas permeable adhesive covers in TB

medium supplemented with 0.5% glycerol (TB-Gly). Each culture was inoculated at a 1:100

volume ratio. The cells were grown at 37 °C to an OD_{600} of 0.7-0.9, before the temperature was

reduced to 25 °C. The expression was induced by addition of arabinose to a final concentration of

0.001% (w/v) for 12-14 h at 18 °C. Cells were harvested by centrifugation for 20 min at 5,000 g,

resuspended in 400 µl of lysis buffer (200 mM NaCl, 50 mM KPi pH 7.5, 2 mM $MgCl_2$, 40 µg/mL

DNAseI, 1 µM leupeptin, 1 µM pepstatin, 0.3 µM aprotinin, 1 mM benzamidine, 1 mM PMSF).

Cells were lysed by disruption using sharp 500 µm glass beads. The supernatant was supplemented

with DDM at a final concentration of 1 % (w/v) and incubated for 1 h at 4 °C under gentle agitation.

The whole-cell extract was clarified by centrifugation at 200,000 g for 15 min and expressed

proteins not containing a fusion to a fluorescent protein were detected by Western blots. Constructs

fused to a fluorescent protein were detected by in-gel fluorescence imaging (Geertsma et al.,

2008a) and fluorescent size exclusion chromatography (FSEC) (Kawate and Gouaux, 2006) using

a Superdex S200 column connected to an Agilent HPLC system (GE Healthcare) (λ_{ex}=489 nm /

λ_{em}= 510 nm). The NRMT homologue (WP_114552427.1) from the prokaryote *Eggerthella lenta*

(DSM 2243, EleNRMT) was selected as the biochemically best-behaved homologue based on the

profile of the eluted peak. The stability of EleNRMT was further improved by mutagenesis. To

this end consensus amino acids were identified using JALVIEW (Waterhouse et al., 2009) from

aligned sequences of representative NRMT homologs, using Muscle (Edgar, 2004), and reported

criteria (Cirri et al., 2018).

Expression and Purification of EleNRMT

For large-scale overexpression, the DNA coding for the NRMT from the prokaryote *Eggerthella*

lenta (EleNRMT) was cloned into the pBXNH3 vector (Invitrogen) using the FX cloning

technique (Geertsma and Dutzler, 2011) providing a construct where *EleNRMT* contains a 10-His

tag and a HRV 3C protease cleavage site at its N-terminus. All mutants were generated using the

Quickchange site-direct mutagenesis method (Agilent). The vector was transformed into *E. coli*

MC1061 cells and a preculture was grown in TB-Gly medium overnight using ampicillin (100

µg/ml) as a selection marker for all expression cultures. The preculture was diluted at 1:100 volume

ratio into TB-Gly medium. The cells were grown at 37 °C to an OD_{600} of 0.7-0.9, before the

temperature was lowered to 25 °C. The expression was induced by addition of arabinose to a final

concentration of 0.0045% (w/v) for 12-14 h at 18 °C. Cells were harvested by centrifugation for

20 min at 5,000 g, resuspended in buffer RES and disrupted using a high-pressure cell lyser

(Maximator HPL6). After low spin centrifugation (10,000 g for 20min) to remove cell debris, the

supernatant was subjected to an ultracentrifugation step (200,000 g for 45 min) and the pelleted

membrane was resuspended in buffer EXT at a concentration of 0.5 g of vesicles per ml of buffer.

The purification of EleNRMT and its mutants was carried out at 4 °C. Suspended membrane

vesicles were diluted 1:2 in buffer EXT supplemented with protease inhibitors (Roche cOmplete

EDTA-free) and 1% (w/v) detergent, depending on the purpose of the experiments. DDM

(Anatrace) was used as detergent for proteoliposome-reconstitution and cryoEM sample

preparation and DM (Anatrace) for ITC and crystallization. The extraction was carried under

gentle agitation for 2 h at 4 °C. The lysate was cleared by centrifugation at 200,000 g for 30 min.

The supernatant supplemented with 15 mM of imidazole at pH 7.5 was loaded onto NiNTA resin

and incubated for at least 1 h under gentle agitation. During later stages of purification, the

detergent concentration was reduced to 0.04% (w/v) for DDM or 0.25% (w/v) for DM. The resin

was washed with 20 CV of buffer W and protein was eluted with buffer ELU, both containing the

respective detergent. The tag was cleaved by addition of HRV 3C protease at a 3:1 molar ratio and

dialyzed against buffer DIA containing the respective detergent for at least 2h at 4 °C. After

addition of NiNTA resin to remove the cleaved tag, the sample was concentrated by centrifugation

using a 50 kDa molecular weight cut-off concentrator (Amicon) and further purified by size

exclusion chromatography on a Superdex S200 column (GE Healthcare) pre-equilibrated with

buffer SEC containing the respective detergent. The peak fractions were pooled, concentrated, and

used immediately for further experiments.

Reconstitution of EleNRMT and EcoDMT into proteoliposomes

EleNRMT, EcoDMT and all of their mutants were reconstituted into detergent-destabilized

liposomes (Geertsma et al., 2008b). The lipid mixture made of POPE, POPG (Avanti Polar Lipids)

at a weight ratio of 3:1: was washed with diethylether and dried under a nitrogen stream followed

by exsiccation overnight. The dried lipids were resuspended in 100 mM KCl and 20 mM HEPES

at pH7. After 3 freeze-thaw cycles, the lipids were extruded through a 400 nm polycarbonate filter

(Avestin, LiposoFast-BAsic) and aliquoted into samples with a concentration of 45 mg/ml. The

extruded lipids were destabilized by adding Triton X-100 and the protein was reconstituted at a

lipid to protein ratio of 50. After several incubation rounds with biobeads SM-2 (Biorad), the

proteoliposomes were harvested the next day, resuspended in 100 mM KCl and 20 mM HEPES

pH7 and stored at -80 °C.

Isothermal titration calorimetry

A MicroCal ITC200 system (GE Healthcare) was used for Isothermal titration calorimetry

experiments. All titrations were performed at 6 °C in buffer SEC (200 mM NaCl, HEPES pH7,

0.25% DM). The syringe was filled with 3 mM $MnCl_2$ or $MgCl_2$ and the titration was initiated by

the sequential injection of 2 μl aliquots into the cell filled with EleNRMT or mutants at a

concentration of 70 μM. Data were analyzed with the Origin ITC analysis package and fitted

assuming a single binding site for metal ions and a stoichiometry of N = 1. Each experiment was

repeated twice from independent protein purifications with similar results. Experiments with

buffer not containing any protein were performed as controls.

Fluorescence-based substrate transport assays

The proteoliposomes were incubated with buffer IN (100 mM KCl, 20 mM HEPES pH 7,

containing either 250 μM Calcein (Invitrogen), 100 μM Fura-2 (ThermoFischer Scientific), or 400

μM MagGreen (ThermoFischer Scientific) depending on the assayed ion. After 3 cycles of freeze-

thawing, the liposomes were extruded through a 400 nm filter and the proteoliposomes were

centrifuged and washed 3 times with buffer IN not containing any fluorophores. The

proteoliposomes were subsequently diluted to 0.025 mg/ml in buffer OUT (100 mM NaCl, 20 mM

HEPES pH7) and aliquoted in 100 μl batches into a 96-well plate (ThermoFischer Scientific) and

the fluorescence was measured every 4 seconds using a Tecan Infinite M1000 fluorimeter. A

membrane potential was generated by addition of valinomycin to a final concentration of 100 nM.

After 20 cycles of incubation, the substrate was added at different concentrations. The transport

reaction was terminated by addition of calcimycin or ionomycin at a final concentration of 100

nM. The initial rate of transport was calculated by linear regression within the initial 100 seconds

of transport after addition of the substrate. The calculated values were fitted to a Michaelis-Menten

equation. Fura-2 was used to assay the transport of Ca^{2+}. For detection, the ratio between calcium

bound Fura-2 (λ_{ex}= 340 nm ; λ_{em}= 510 nm) and unbound Fura-2 (λ_{ex}= 380 nm ; λ_{em}= 510 nm) was

monitored during kinetic measurements. Calcein was used to assay the transport of Mn^{2+} (λ_{ex}=

492 nm ; λ_{em}= 518 nm) and Magnesium Green was to assay the transport of Mg^{2+} (λ_{ex}= 506 nm ;

λ_{em}= 531 nm).

For assaying coupled proton transport, the proteoliposomes were mixed with buffer 1 (100 mM

KCl, 5 mM HEPES pH 7, 50 μM ACMA). After clarification by sonication, the proteoliposomes

were diluted to a concentration of 0.15 mg/ml in buffer 2 (100 mM NaCl, 5 mM HEPES pH7) and

aliquoted into batches of 100 μl into a flat black 96-well plate (ThermoFischer Scientific). The

fluorescence was measured every 4 seconds using a fluorimeter Tecan Infinite M1000 (λ_{ex}= 412

nm; λ_{em}= 482 nm). The transport reaction was initiated after addition of the substrate and

valinomycin at a final concentration of 100 nM. The reaction was terminated by addition of CCCP

at 100 nM.

Generation of nanobodies specific to EleNRMT

EleNRMT specific nanobodies were generated using a method described previously (Pardon et al.,

2014). Briefly, an alpaca was immunized four times with 200 μg of detergent solubilized

EleNRMT[ts]. Two days after the final injection, the peripheral blood lymphocytes were isolated

and the RNA total fraction was extracted and converted into cDNA by reverse transcription. The

nanobody library was amplified and cloned into the phage display pDX vector (Invitrogen). After

two rounds of phage display, ELISA assays were performed on the periplasmic extract of 198

individual clones. 3 nanobodies were identified. For all steps of phage display and ELISA, an

enzymatically biotinylated version of EleNRMT with an additional Avi tag on its N-terminus (Avi-

10His-3C) was used (Fairhead and Howarth, 2015) and bound to neutravidin coated plates. The

biotinylation was confirmed by total mass spectrometry analysis.

Expression and purification of nanobodies and preparation of the EleNRMT^{ts}-Nb1,2 complex

All nanobodies were cloned into the pBXNPHM3 vector (Invitrogen) using the FX cloning technique. The expression construct contains the nanobody with a fusion of a pelB sequence, a 10-His tag, a maltose-binding protein (MBP) and a HRV 3C protease site to its N-terminus. The vector was transformed into *E. coli* MC1061 cells and a preculture was grown in TB-Gly medium overnight using ampicillin (100 µg/ml) as a selection marker in all expression cultures. The preculture was used for inoculation of TB-Gly medium at 1:100 volume ratio. Cells were grown to an OD$_{600}$ of 0.7-0.9 and the expression was induced by addition of arabinose to a final concentration of 0.02% (w/v) for 4 h at 37 °C. Cells were harvested by centrifugation for 20 min at 5,000 g, resuspended in buffer A (200 mM KCl, 50 mM KPi pH 7.5, 10% Glycerol, 15 mM imidazole pH 7.5, 2 mM MgCl$_2$, 1µM leupeptin, 1 µM pepstatin, 0.3 µM aprotinin, 1mM benzamidine, 1mM PMSF) and lysed using a high pressure cell lyser (Maximator HPL6). The purification of all nanobodies was carried out at 4 °C. The lysate was cleared by centrifugation at 200,000 g for 30 min. The supernatant was loaded onto NiNTA resin and incubated for at least 1h under gentle agitation. The resin was washed with 20 CV of buffer B (200 mM KCl, 20 mM Tris.HCl pH8, 10% Glycerol, 50 mM imidazole, pH 7.5). The protein was eluted with buffer C (200 mM KCl, 20 mM Tris.HCl pH 8, 10% Glycerol, 300 mM imidazole, pH 7.5). The tag was cleaved by addition of HRV 3C protease at a 3:1 molar ratio and dialyzed against buffer D (200 mM KCl, 20 mM Tris.HCl pH8, 10% Glycerol) for at least 2 h at 4 °C. After application of NiNTA resin to remove the tag, the sample was concentrated by centrifugation using a 10 kDa molecular weight cut off concentrator (Amicon) and further purified by size exclusion chromatography using a Superdex S200 column (GE Healthcare) pre-equilibrated with buffer E (200 mM KCl, 20 mM

Tris.HCl pH 8). The peak fractions were pooled, concentrated and flash-frozen and stored at -20

°C for subsequent experiments.

Following the purification of EleNRMT in DDM (for cryoEM data collection) or DM (for

crystallization) as described, the purified membrane protein and nanobodies Nb1 and Nb2 were

incubated at a 1:1.3:1.3 molar ratio for 30 min and the complex was further purified by size

exclusion chromatography on a Superdex S200 column (GE Healthcare), pre-equilibrated in the

same SEC buffer as used for EleNRMT purification. The sample was subsequently concentrated

to 3.5 mg/ml (for cryoEM) or 8-12 mg/ml (for crystallization).

Cryo-EM sample preparation and data collection

For sample preparation for cryo-EM, 2.5 μl of the complex were applied to glow-discharged holey

carbon grids (Quantifoil R1.2/1.3 Au 200 mesh). Samples were blotted for 2-4 s at 4 °C and 100%

humidity. The grids were frozen in liquid propane-ethane mix using a Vitrobot Mark IV (Thermo

Fisher Scientific). For the dataset in presence of Mg^{2+}, the sample was supplemented with 10 mM

$MgCl_2$ and incubated on ice for 20 min before grid preparation. All datasets were collected on a

300 kV Titan Krios (ThermoFischer Scientific) with a 100 μm objective aperture and using a post-

column BioQuantum energy filter with a 20 eV slit and a K3 direct electron detector in super-

resolution mode. All datasets were recorded automatically using EPU2.9 with a defocus ranging

from -1 to 2.4 μm, a magnification of 130,000x corresponding to a pixel size of 0.651 Å per pixel

(0.3255 Å in super resolution mode) and an exposure of 1.01s (36 frames). Three datasets were

collected for EleNRMT-Nb1,2 without $Mg^{2+}$with respective total doses of 69.725, 61 and 69.381

$e^-/Å^2$ and for EleNRMT-Nb1,2 with Mg^{2+} one dataset with a total dose of 69.554 $e^-/Å^2$

Cryo-EM data processing

Datasets were processed using Cryosparc v3.2.0 (Punjani et al., 2017) following the same

processing pipeline (Figure 5—figure supplements 1, 2). All movies were subjected to motion

correction using patch motion correction with a fourier crop factor of 2 (pixel size of 0.651 Å/pix).

After patch CTF estimation, high quality micrographs were identified based on relative ice

thickness, CTF resolution and total full frame motion and micrographs not meeting the specified

criteria were rejected. For the EleNRMT^{ts}-Nb1,2 complex, a partial dataset (981 micrographs) was

used to generate a low-resolution 3D *ab initio* model. For this purpose, particles were selected

using a blob picker with a minimum particle diameter of 160 Å and a minimum inter-particle

distance of 64 Å. The selected particles were extracted with a box size of 360 pixels and binned

4x (pixel size of 2.64 Å per pixel). These particles were initially used to generate 2D classes and

later for template-driven particle picking. After extraction and 2D classification, an initial 3D *ab*

initio model was generated in Cryosparc. This model was subsequently used to generate 2D

templates for particle picking on the remainder of the dataset.

The 2D classes generated from particles selected from the entire dataset were extracted using a

box size of 360 pix and down-sampled to 180 pixels (pixel size of 1.32 Å/pix). These particles

were used for generation of 4 *ab initio* classes. Promising *ab initio* models were selected based on

visual inspection and subjected to heterogenous refinement using one of the selected models as

'template' and an obviously bad model as decoy model. After several rounds of heterogenous

refinement, the selected particles and models were subjected to non-uniform refinement (input

model filtering to 8 Å) followed by local CTF refinement and another round of non-uniform

refinement. Finally, the maps were sharpened using the sharpening tool from the Cryosparc

package. The quality of the map was evaluated validated using 3DFSC (Tan et al., 2017) for FSC

validation and local resolution estimation.

Cryo-EM model building and refinement

Model building was performed in Coot (Emsley and Cowtan, 2004). Initially, the structure of

ScaDMT (PDB 5M94) was rigidly fitted into the densities and served as template for map

interpretation. The quality of the map allowed for the unambiguous assignment of residues 43-

438. The structure of Nb16 (PDB 5M94) was used to build Nb1 and Nb2. The model was

iteratively improved by real space refinement in PHENIX (Afonine et al., 2018) maintaining

secondary structure constrains throughout. Figures were generated using ChimeraX (Pettersen et

al., 2021) and Dino (http://www.dino3d.org). Surfaces were generated with MSMS (Sanner et al.,

1996).

Crystallization and X-ray structure determination of the EleNRMT[ts]-Nb1,2 complex

The EleNRMT[ts]-Nb1,2 complex was purified in DM as described before, supplemented with *E.*

coli polar lipids solubilized in DM to a final concentration of 100 µg/ml and used for the

preparation of vapor diffusion crystallization experiments in sitting drops by mixing 150 nl of the

concentrated protein with 150 nl of the mother liquor containing 50 mM MgAc, 50 mM HEPES

pH 7.2-7.6 and 25-30% PEG400. Crystals grew after 4 days and reached a maximum size of 0.2 x

0.3 x 0.3 mm after 12 days. The crystals were fished, cryoprotected by soaking into mother liquor

supplemented with PEG400 to a final concentration of 35% and flash frozen into liquid nitrogen.

All data sets were collected on frozen crystals on either the X06SA or the X06DA beamline at the

Swiss Light Source of the Paul Scherrer Institute (SLS, Villingen) on an EIGER 16M or PILATUS

2M-F detector (Dectris), respectively. Anomalous data were collected at the absorption edge of

manganese (1.896 Å). Integration and scaling was performed using XDS (Kabsch, 1993) and data

were further processed with CCP4 programs (CCP4, 1994). Phases were obtained by molecular

replacement in Phaser (McCoy et al., 2007). For the ternary complex EleNRMT- Nb1-Nb2, the

refined cryoEM structure was used as search model for the respective datasets. Structures were

refined in PHENIX (Liebschner et al., 2019) maintaining tight coordinate restraints and models

were inspected and modified in Coot (Emsley and Cowtan, 2004).

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Author Contributions

K.R. cloned, expressed and purified proteins, performed transport assays and ITC experiments,

prepared samples for X-ray crystallography and cryo-EM. M.S.S. collected cryo-EM data. K. R.

processed cryo-EM data and built models. C.M. and K.R. collected and processed ITC and X-ray

data. K.R., C.M. and R.D. jointly planned experiments, analyzed data and wrote the manuscript.

Data availability

The cryo-EM density maps of the EleNRM-Nb complex in absence and presence of Mg^{2+} have

been deposited in the Electron Microscopy Data Bank under ID codes EMD-XXX and EMD-

YYY, respectively. The coordinates for the atomic model of the EleNRM-Nb complex in absence

of Mg^{2+} refined against the 3.4 Å cryo-EM density and the coordinates of the EleNRMT^{ts}-Nb1,2

complex in presence of Mg^{2+} refined against the 4.1 Å cryo-EM density have been deposited in

the Protein Data Bank under ID codes XXX and YYY. The coordinates and structure factors of

the EleNRMT^{ts}-Nb1,2 complexes in Mg^{2+} and Mn^{2+} have been deposited in the Protein Data Bank

with the accession codes XXX and YYY.

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Competing financial interests

726   The authors declare no competing financial interests.

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Figures

Figure 1. Phylogenetic analysis of the SLC11/NRAMP family. (**A**) Model of the consensus ion

binding site of NRAMP transporters as obtained from the X-ray structure of ScaDMT (PDBID

5M95) in complex with Mn^{2+}. Regions of α1 and α6 are shown as Cα-trace, selected residues

coordinating the ion as sticks. (**B**) Phylogenetic tree of SLC11 residues with different clades of the

family highlighted. The consensus sequence of the respective ion binding site of each clade is

shown.

The following figure supplements are available for figure 1:

Figure supplement 1. Sequence alignment of selected protein regions of SLC11 family members.

		1a	1b		6a	6b

prokaryotic NRAMPs

	1a / 1b	6a / 6b
prNRAMP1	PGLLVSVGYMDPGNWATAIEAGSRFG	YLAIGILGATVMPHNLYLHSSVV
prNRAMP2	PGLLVAVGYMDPGNWITSMQGGAQYG	YIALGIIGATIMPHNLYLHSSIV
prNRAMP3	PGALVAVGYMDPGNWSTSITGGQNFQ	SASLGIIGATIMPLNLFLHSAIS
prNRAMP4	PAFVAAIAYVDPGNVAANVSSGAQFG	LLAAAIMGATVMPHAVYLHSGLA
prNRAMP5	PAFIAAIAYIDPGNFATNISAGSKYG	LLAAAIMGATVMPHAVYLHSGLA
prNRAMP6	PAVIASIAYMDPGNFATNIEGGARYG	YLAVGIIGATVMPHVIYLHSALT
prNRAMP7	PAFIAAIGYIDPGNFATNIQAGASFG	FLAAGVLGATIMPHVIYLHSSLT
prNRAMP8	PAFIAAIGYIDPGNFATNIQAGASFG	FLAAGVLGATIMPHVIYLHSSLT

animal NRAMPs

anNRAMP1	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP2	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP3	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP4	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP5	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP6	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP7	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP8	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP9	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV
anNRAMP10	PGFLMSIAYLDPGNIESDLQSGAVAG	EQAVGIVGAVIMPHNMYLHSALV

plant NRAMPs

plNRAMP1	PGFLMSIAFLDPGNLEGDLQAGAIAG	QQAVGIVGCIIMPHNVFLHSALV
plNRAMP2	PGFLMSIAFLDPGNLEGDLQAGAIAG	RQAVGVVGCVIMPHNVYLHSALV
plNRAMP3	PGFLMSIAFLDPGNLEGDLQAGAIAG	QQAVGVVGCIIMPHNVFLHSALV
plNRAMP4	PGFLMSIAFLDPGNLEGDLQSGAIAG	QQAVGVVGCVIMPHNVYLHSALV
plNRAMP5	PGFLMSIAFLDPGNLEGDLQSGAIAG	QQAVGVVGCIIMPHNVFLHSALV
plNRAMP6	PGFLMSIAFLDPGNLEGDLQSGAIAG	QQAVGVVGCIIMPHNVFLHSALV
plNRAMP7	PGFLMSIAFLDPGNLEGDLQSGAIAG	QQAVGVVGCIIMPHNVFLHSALV
plNRAMP8	PGFLMSIAFLDPGNLEGDLQAGAIAG	QQAVGVVGCLIMPHNVFLHSALV
plNRAMP9	PGFLMSIAFLDPGNLEGDLQAGAIAG	QQAVGVVGCLIMPHNVFLHSALV
plNRAMP10	PGFLMSIAFLDPGNLEGDLQAGAIAG	QQAVGVVGCLIMPHNVFLHSALV

plant NRATs

plNRAT1	PGALVAIGFLDPSNIETDMQAGAEFK	TNAIALFGSIITPYNLFLHSALV
plNRAT2	PGALVAIGFLDPSNLETDMQAGADFK	ANAIALFGAIITPYNLFLHSALV
plNRAT3	PGALVAIGFLDPSNLETDMQAGADFK	ANAIALFGAIITPYNLFLHSALV
plNRAT4	PGALVAIGFLDPSNLETDMQAGADFK	ANAIALFGAIITPYNLFLHSALV
plNRAT5	PGALVAIGFLDPSNLETDMQAGADFK	ANAIALFGAIITPYNLFLHSALV
plNRAT6	PGALVAIGFLDPSNMETDMQAGADFK	ANAIALFGAIITPYNLFLHSALV
plNRAT7	PGALVAIGFLDPSNLETDMQAGADFK	ANAIALFGAIITPYNLFLHSALV

prokaryotic NRMTs

prNRMT1	PGIVTAMAGNDAGGISTYSTVGAKFG	SLVIAMIGTTIAPWMMFFNQSNV
prNRMT2	PGIIAASAGNDAGGITTYAMVGAELQ	TLFVGTVGTTIAPWMQFYLQSSI
prNRMT3	PGLVTATADNDASGIATYAMVGAMFG	LIFIAMIGTTITPWMQFYLQSSV
prNRMT4	PGLITVNAGNDAGGIATYATVGASYG	LTLIGMIGTTITPYMQFYLQSSV
prNRMT5	PGVITANAGNDAGGIATFASVGAEHG	FTLVAVIGTTISPYMQVFIQSSV
prNRMT6	PGFITAVVDNDSGGIFTYSAAGARFG	YMLIGMVGTTIAPWMQFYLQAAV
prNRMT7	PGIITANVDNDAGGITTYSITGATTG	SILIGFIGTTITPWMQFYIQSST
prNRMT8	PGIITANVDNDAGGITTYSLAAANYG	MTLIGIVGTTIAPWMQFYQQSAV
prNRMT9	PGIITSNVDNDAGGITTYSLAGAEYG	TMAIGLVGTTIAPWMQFYLQSSV
prNRMT10	PGIITANVDNDAGGITTYSVAGAQFG	MMIIGVVGTTIAPWMQFYLQSAV
prNRMT11	PGIITASVDNDAGGITTYSLAGARFG	LMLIGLVGTTIAPWMQFYIQSAI

Figure 1—figure supplement 1. Sequence alignment of selected protein regions of SLC11 family

members. Residues constituting the ion binding site are highlighted in green. Secondary structure

elements are indicated on the top. Prokaryotic NRAMPSs (prNRAMP): 1 *D. radiodurans*

(Uniprot: Q9RTP8), 2 *S. capitis* (Uniprot: A0A4U9TNH6), 3 *E. coelocola* (Uniprot: E4KPW4), 4

E.   coli   (UniProt: sp|*P0A769*|MNTH_ECOLI),   5   *M.   tuberculosis*   (UniProt:

sp|P9WIZ5|MNTH_MYCTU), 6 *B. subtilis* (UniProt: sp|P96593|MNTH_BACSU), 7 *S.*

typhimurium   (Uniprot: sp|Q9RPF4|MNTH_SALTY),   8   *P.   aeruginosa*   (Uniprot:

sp|Q9RPF3|MNTH1_PSEAE).

Animal NRAMPs (anNRAMP): 1 *H. sapiens* NRAMP2 (NCBI: NP_001366375.1), 2 *P. paniscus*

NRAMP2 (NCBI: XP_008949491.1), 3 *C. sabaeus* NRAMP2 (NCBI: XP_037853046.1), 4 *R.*

roxellana NRAMP2 (NCBI: XP_010383634.1), 5 *M. mulatta* NRAMP2 (NCBI: EHH20731.1), 6

E. caballus NRAMP2 (NCBI: XP_023499318.1), 7 *C. canadensis* NRAMP2 (NCBI:

XP_020035491.1), 8 *H. hyaena* NRAMP2 (NCBI: XP_039097283.1), 9 *C. ferus* NRAMP2

(NCBI: XP_032348726.1), 10 *B. indicus* NRAMP2 (NCBI: XP_019816126.1).

Plant NRAMPs (plNRAMP): 1 *P. vulgaris* (NCBI: XP_007155869.1), 2 *T. subterraneum* (NCBI:

GAU12542.1), 3 *G. max* (NCBI: NP_001344702.1), 4 *C. arietinum* (NCBI: XP_004511877.1), 5

D. chrysotoxum (NCBI: KAH0462125), 6 *T. cacao* (NCBI: XP_007047433.1), 7 *V. unguiculata*

(NCBI: XP_027914252.1), 8 *G. soja* (NCBI: KAG4933418.1), 9 *M. truncatula* (NCBI:

XP_003611648.1), 10 *A. annua* (NCBI: PWA53748.1).

Plant NRATs (plNRAT): 1 *O. sativa* (NCBI: XP_015625418.1), 2 *D. oligosanthes* (NCBI:

OEL35611.1), 3 *S. italica* (NCBI: XP_004952002.1), 4 *S. bicolor* (NCBI: XP_002451480.2), 5 *P.*

miliaceum (NCBI: RLM80353.1), 6 *P. virgatum* (NCBI: XP_039819678.1), 7 *Z. mays* (NCBI:

PWZ19830.1).

Prokaryotic NRMTs (prNRMT): 1 *P. propionicus* (NCBI: WP_011735082.1), 2 *M. australiensis*

(NCBI: WP_013780282.1), 3 *C. saccharolyticus* (NCBI: WP_011916762.1), 4 *A. bacterium*

(NCBI: PYS45348.1), 5 *G. bemidjiensis* (NCBI: WP_012531668.1), 6 *T. narugense* (NCBI:

PMP85905.1), 7 *C. acetobutylicum* (NCBI: WP_034583260.1), 8 *M. paludicola* (NCBI:

BAI61876.1), *9 S. aciditrophicus* (NCBI: WP_011418107.1), 10 *S. usitatus* (NCBI:

WP_011688372.1), 11 *E. laentae* (NCBI: WP_009305646.1).

Figure 2. Functional characterization of EcoDMT mutants. (**A**) EcoDMT mediated Mn^{2+} transport into proteoliposomes. Data display mean of 3 experiments from 3 independent reconstitutions. (**B**) Mn^{2+} concentration dependence of transport. Initial velocities were derived from individual traces of experiments displayed in **A**, the solid line shows the fit to a Michaelis–Menten equation with an apparent K_m of 13.5 μM. (**C**) Mn^{2+} transport in presence of Ca^{2+}. Data display mean of 2 to 7 experiments from 2 independent reconstitutions, except for the measurement without divalent ions (0/0, green, mean of 2 experiments). (**D**) Mn^{2+} transport into EcoDMT proteoliposomes in presence of Mg^{2+}. Data display mean of 3 (0/0, green), 4 (50/0, blue) and 9 experiments (50/500, orange and 50/2000, red) from 3 independent reconstitutions. (**E**) Mn^{2+} transport into proteoliposomes containing the EcoDMT mutant M235A. Data display mean of 8 experiments from 3 independent reconstitutions. (**F**) Inhibition of Mn^{2+} transport in the mutant M235A by Ca^{2+} (3 experiments from 3 independent reconstitutions). (**G**) Mn^{2+} concentration dependence of transport into M235A proteoliposomes. For Mn^{2+}, initial velocities were derived from individual traces of experiments displayed in **E**, for Mn^{2+} in presence of 2mM Ca^{2+}, data show mean of 3 experiments from 3 independent reconstitutions. The solid lines are fits to a Michaelis–Menten equation with apparent K_m of 41 μM (Mn^{2+}) and 107 uM (Mn^{2+} in presence of 2mM Ca^{2+}). (**H**) Ca^{2+}-transport into M235A proteoliposomes assayed with the fluorophore Fura-2. Data display mean of 3 experiments from 3 independent reconstitutions. (**I**) Mn^{2+} transport in presence of Mg^{2+}. Data display mean of 6 experiments from 3 independent reconstitutions. (**J**) Mn^{2+} transport into proteoliposomes containing the EcoDMT triple mutant M235A/N54G/A232T. Data display mean of 3 experiments from 3 independent reconstitutions. (**K**) Inhibition of Mn^{2+} transport in the mutant M235A/N54G/A232T by Mg^{2+} (6 experiments from 3 independent reconstitutions). (**L**) Mn^{2+} concentration dependence of transport into M235A/N54G/A232T proteoliposomes. For Mn^{2+},

initial velocities were derived from individual traces of experiments displayed in **J**, for Mn^{2+} in

presence of 2mM Mg^{2+}, data show mean of 3 (without Mg^{2+}, red) and 6 experiments (with Mg^{2+},

blue) from 3 independent reconstitutions. The solid lines are fits to a Michaelis–Menten equation

with apparent K_m of 31 µM (Mn^{2+}) and 38 µM (Mn^{2+} in presence of 2mM Mg^{2+}). **A, C, D, E, F, I,**

J, K Uptake of Mn^{2+} was assayed by the quenching of the fluorophore calcein trapped inside the

vesicles. **A, C, D, E, F, H, I, J, K** Averaged traces are presented in unique colors. Fluorescence is

normalized to the value after addition of substrate (t=0). Applied ion concentrations are indicated.

A-L, Data show mean of the indicated number of experiments, Errors are s.e.m.

Figure 3. Transport properties of EleNRMT. (**A**) EleNRMT mediated Mn^{2+} transport into

proteoliposomes. Data display mean of 3 experiments from 2 independent reconstitutions. (**B**)

Mn^{2+} concentration dependence of transport. Initial velocities were derived from individual traces

of experiments displayed in **A**, the solid line shows the fit to a Michaelis–Menten equation with

an apparent K_m of 119 μM. (**C**) Mn^{2+} transport in presence of Mg^{2+}. Data display mean of 3 (0/0,

green), 6 (50/0, blue), 5 (50/500, orange) and 9 (50/2000, red) experiments from 2 independent

reconstitutions. (**D**) Mn^{2+} transport into proteoliposomes mediated by the thermostabilized mutant

EleNRMT^{ts}. Data display mean of 4 experiments from 2 independent reconstitutions, except for

the measurement with 25 µM Mn^{2+} (mean of 3 experiments). (**E**) Mn^{2+} concentration dependence

of transport. Initial velocities were derived from individual traces of experiments displayed in **D**,

the solid line shows the fit to a Michaelis–Menten equation with an apparent K_m of 124 µM. (**F**)

Mn^{2+} transport in presence of Mg^{2+}. Data display mean of 4 (0/0, green), 9 (50/0, blue), 11 (50/500,

orange) and 8 (50/2000, red) experiments from 2 independent reconstitutions. **A, C, D, F**. Uptake

of Mn^{2+} was assayed by the quenching of the fluorophore calcein trapped inside the vesicles.

Averaged traces are presented in unique colors. Fluorescence is normalized to the value after

addition of substrate (t=0). Applied ion concentrations are indicated. **A-F**, Data show mean of the

indicated number of experiments, Errors are s.e.m.. **G-H**, Binding isotherms obtained from

isothermal titrations of Mn^{2+} (**G**) and Mg^{2+} (**H**) to EleNRMT^{ts}. The data shown for two biological

replicates per condition was fitted to a model assuming a single binding site with the binding

isotherm depicted as solid line. Errors represent fitting errors.

Figure supplement 1. EleNRMT sequence and topology

Figure supplement 2. Stabilization and biochemical characterization of EleNRMT.

Figure 3—figure supplement 1. EleNRMT sequence and topology. (**A**) Sequence of EleNRMT (NCBI: WP_009305646.1) with secondary structure elements indicated below. Residues mutated in the thermostabilized EleNRMTts are highlighted with replacements shown on top. (**B**) Comparison of functionally relevant regions of EleNRMT and the Mn^{2+} transporter ScaDMT (Uniprot: A0A4U9TNH6) with secondary structure elements shown below. Dots indicate ion binding site (red), a residue interacting with the ion in an occluded state (orange) and positions relevant for H^+ transport in NRAMPs on α6 (blue) and α3, 9 (green).

Figure 3—figure supplement 2. Stabilization and biochemical characterization of EleNRMT. Gel

filtration profiles on a Superdex 200 column of WT EleNRMT (**A**) and the thermostabilized

construct of ElenRMT^{ts} (**B**). Insets show SDS-PAGE gel of peak fractions with molecular weight

marker indicated (kDa). (**C**) Thermal stability assay of EleNRMT and EleNRMT^{ts}. Data show

averages of four technical replicates, errors are s.e.m.. (**D**) SEC/MALS experiment of purified

EleNRMT^{ts} with obtained molecular weight corresponding to a monomeric transporter (measured

molecular weight of 45.6 kDa versus theoretical molecular weight of 46.9 kDa). Thermograms of

Mn^{2+} (**E**) and Mg^{2+} (**F**) binding to EleNRMT^{ts} obtained from isothermal titration calorimetry

experiments shown in Figure 3G, H. E, F Panels show titration against buffer (left) and traces of

two biological replicates (center, right)

Figure 4. Ion selectivity and substrate coupling. (**A-E**) Metal ion transport by EleNRMT^{ts}. (**A**) EleNRMT^{ts} mediated Mn^{2+} transport into proteoliposomes in presence of 2 mM Mg^{2+}. Data display mean of 8 experiments from 3 independent reconstitutions, except for the measurement with 200 μM Mn^{2+} (mean of 7 experiments). (**B**) Mn^{2+} concentration dependence of transport into EleNRMT^{ts} proteoliposomes. For Mn^{2+}, data are as displayed in Figure 3E, for Mn^{2+} in presence

of 2mM Mg^{2+}, initial velocities were derived from individual traces of experiments displayed in

A. The solid lines are fits to a Michaelis–Menten equation. (**C**) Mg^{2+}-transport into EleNRMT^{ts}

proteoliposomes assayed with the fluorophore Magnesium Green. Data display mean of 5 (0 μM

Mg^{2+}), 8 (800 μM and 1250 μM Mg^{2+}), 4 (2500 μM Mg^{2+}) and 3 (5000 μM Mg^{2+}) experiments

from 2 independent reconstitutions. (**D**) Mn^{2+} transport in presence of Ca^{2+}. Data display mean of

4 to 11 experiments from 3 independent reconstitutions. **A, D** Uptake of Mn^{2+} was assayed by the

quenching of the fluorophore calcein trapped inside the vesicles. (**E**) Ca^{2+}-transport into

EleNRMT^{ts} proteoliposomes assayed with the fluorophore Fura-2. Data display mean of 3 to 12

experiments from 2 independent reconstitutions. The small increase of the fluorescence at high

Ca^{2+} concentrations is likely from non-specific leak into proteoliposomes. **F-H** Assay of H^+

transport with the fluorophore ACMA. Experiments probing metal ion coupled H^+ transport into

proteoliposomes containing EleNRMT^{ts} upon addition of Mn^{2+} (8 experiments from 2 independent

reconstitutions) (**F**) and Mg^{2+} (8 experiments from 2 independent reconstitutions) (**H**) do not show

any indication of H^+ transport. Mn^{2+} coupled H^+ transport into EcoDMT proteoliposomes (3

experiments from 3 independent reconstitutions and for the negative control (no protein, 400 μM

Mn^{2+}), 4 measurements) is shown for comparison (**I**). **A, C-H** Averaged traces are presented in

unique colors. Fluorescence is normalized to the value after addition of substrate (t=0). Applied

ion concentrations are indicated. **A-H**, Data show mean of the indicated number of experiments,

Errors are s.e.m.

1012

1013   **Figure supplement 1.** Nanobody selection.

1014

Figure 4—figure supplement 1. Nanobody selection. (**A**) Sequence alignment of the three

EleNRMT binders Nb1-Nb3. Secondary structure is shown above. The location of complementary

determining regions (CDRs) are indicated, residues in Nb1 and Nb2 making contacts with

EleNRMT[ts] are highlighted in green. SEC profiles of purified (**B**) Nb1, (**C**) Nb2, and (**D**) the

EleNRMT[ts]-Nb1,2 complex assembled with a stoichiometric excess of both nanobodies. Inset

shows SDS-PAGE gel with molecular weights (kDa) indicated.

1021

Figure 5. Structural characterization of EleNRMT^{ts} by Cryo-EM and X-ray crystallography. (**A**)

Cryo-EM density (contoured at 5 σ) of EleNRMT^{ts} in complex with Nb1 and Nb2 at 3.5 Å viewed

from within the membrane with the extracellular side on top and (**B**), ribbon representation of the

complex in the same view. Proteins are shown in unique colors. (**C**) Anomalous difference density

of Mn^{2+} obtained from the crystal structure of the EleNRMT^{ts}-Nb1-Nb2 complex (calculated at 5

Å and displayed as black mesh contoured at 4.5 σ). Model of EleNRMT^{ts} encompassing the

unwound center of helices α1 and α6 is shown as ribbon.

1029

Figure supplement 1. Cryo-EM reconstruction of the EleNRMT^{ts}-nanobody complex.

Figure supplement 2. Cryo-EM reconstruction of EleNRMT^{ts}-nanobody in presence of Mg^{2+}.

Figure supplement 3. Cryo-EM densities of EleNRMT^{ts}-nanobody complexes.

Figure supplement 4. Nanobody interfaces.

Figure supplement 5. X-ray structures of EleNRMT^{ts}-nanobody complexes in presence of Mg^{2+}

and Mn^{2+}.

Figure 5—figure supplement 1. Cryo-EM reconstruction of the EleNRMT^{ts}-nanobody complex.

(**A**) Representative micrograph (out of a total of 22 117 images) of the complex acquired with a

Titan Krios G3i microscope equipped with a K3 camera. (**B**) 2D class averages of the EleNRMT^{ts}-

Nb1-Nb2 dataset. (**C**) Data processing workflow. After extraction and one round of 2D

classification, a single *ab initio* reconstruction with four classes was performed. The particle

distribution is indicated in %. The two most promising classes, together with a decoy class were

subjected to nine rounds of heterogeneous refinement. Non-uniform (NU) refinement, local CTF-

refinement and a second round of NU-refinement with C1 symmetry yielded a map at a resolution

of 3.5 Å. (**D**) Angular distribution of particle orientations. The heatmap displays the number of

particles for a given viewing angle. (**E**) Directional and global FSC plots. The global FSC is shown

in black. Dashed line indicates 0.143 cut-off and the resolution at which the FSC curve drops below

0.143 is indicated. The directional FSC curves providing an estimation of anisotropy of the dataset

are shown for directions x, y, and z. (**F**) Final 3D reconstruction of the EleNRMT^{ts}-Nb1-Nb2

complex in indicated orientations, colored according to the local resolution, estimated in

cryoSPARC v.3.2.0.

1052

Figure 5—figure supplement 2. Cryo-EM reconstruction of EleNRMT^{ts}-nanobody in presence

of Mg^{2+}. (**A**) Cryo-EM reconstruction of the EleNRMT^{ts}-nanobody complex in presence of Mg^{2+}.

(**A**) Representative micrograph (out of a total of 12 427 images) of the complex acquired with a

Titan Krios G3i microscope equipped with a K3 camera. (**B**) 2D class averages of the EleNRMT^{ts}-

Nb1-Nb2 dataset. (**C**) Data processing workflow. After extraction and one round of 2D

classification, a single *ab initio* reconstruction with four classes was performed. The particle

distribution is indicated in %. The particles from the two most promising classes, together with a

decoy class were subjected to 22 rounds of heterogeneous refinement. Non-uniform (NU)

refinement, local CTF-refinement and a second round of NU-refinement with C1 symmetry

yielded a map at a resolution of 4.1 Å. (**D**) Angular distribution of particle orientations. The

heatmap displays the number of particles for a given viewing angle. (**E**) Directional and global

FSC plots. The global FSC is shown in black. Dashed line indicates 0.143 cut-off and the resolution

at which the FSC curve drops below 0.143 is indicated. The directional FSC curves providing an

estimation of anisotropy of the dataset are shown for directions x, y, and z. (**F**) Final 3D

reconstruction of the EleNRMT^{ts}-Nb1-Nb2 complex in indicated orientations, colored according

to the local resolution, estimated in cryoSPARC v.3.2.0.

A

EleNRMT/Nb1/Nb2 no Mg²⁺ 3.5 Å

B

EleNRMT/Nb1/Nb2 Mg²⁺ 4.1 Å

1069

1070   **Figure 5—figure supplement 3.** Cryo-EM densities of EleNRMT-nanobody complexes. (**A**)

1071   EleNRMT[ts]-nanobody complex in absence of Mg^{2+} at 3.5 Å. Electron density (contoured at 6σ,

1072   grey mesh) is shown superimposed on indicated protein regions. (**B**) EleNRMT[ts]-nanobody

1073   complex in presence of Mg^{2+} at 4.1 Å. Electron density (contoured at 6σ, grey mesh) is shown

1074   superimposed on indicated protein regions.

1075

1076

1077   **Figure 5—figure supplement 4.** Nanobody binding interfaces. (**A**) Contact region of EleNRMT[ts]

1078   and nanobodies Nb1 and Nb2. (**B**) Binding interface of Nb2. (**C**) Binding interface of Nb1. **C, D,**

1079   Side chains of residues mediating binding interactions between the transporter and the nanobodies

1080   are displayed as sticks. **A-C** Proteins are shown as ribbons with respective regions labeled.

1081

Figure 5—figure supplement 5. X-ray structures of EleNRMT-nanobody complexes in presence

of Mg^{2+} and Mn^{2+}. (**A**) Structure of the EleNRMT[ts]-Nb1-Nb2 complex in Mg^{2+} at 4.1 Å (**B**)

Structure of the EleNRMT[ts]-Nb1-Nb2 complex soaked in Mn^{2+} at 4.6 Å. A, B 2Fo-Fc density

(contoured at 1 σ) is shown as green mesh superimposed on the model. B, C, Anomalous difference

electron density (calculated at 6 Å and contoured at 4 σ) from the dataset displayed in B is depicted

as black mesh. A-C, shown is one of two complexes in the asymmetric unit as Cα trace with protein

chains displayed in unique colors. Phases for 2Fo-Fc and anomalous difference density maps were

obtained from molecular replacement using the cryo-EM model improved by constrained minimal

refinement. Residual density in the cryo-EM maps of the EleNRMT[ts]-Nb1-Nb2 complex in

1091   absence (**D**) and presence of Mg^{2+} (**E**) in vicinity to the Mn^{2+} binding position (indicated by black

1092   sphere). Protein is represented as sticks with water (red) or Mg^{2+} (green) as spheres. Maps are as

1093   displayed in Figure 5—figure supplement 3.

1094

Figure 6. EleNRMT structure. (**A**) Ribbon representation of EleNRMT viewed from within the

membrane with α1-α5 colored in beige, α6- α10 in blue and α11 in magenta. the position of a

bound ion is indicated by a black sphere. (**B**) Superposition of EleNRMT and ScaDMT (with

modeled α1a), both showing equivalent inward-open conformations. Ribbon representation of

ScaDMT is shown in grey, the representation of EleNRMT is as in A. (**C**) Surface representation

of EleNRMT. Left, view from the cytoplasm and, right, slice of the surface of EleNRMT viewed

from within the membrane, both showing the aqueous cavity leading to the ion binding site. The

position of a bound ion is represented by a black sphere and with a red asterisk. A-C Selected

helices are labeled. Membrane boundaries (A, C) and relationship between views (A-C) are

indicated.

1105

Figure 7. Comparison of regions of functional relevance between EleNRMT and ScaDMT. (**A**)

Superposition of the metal ion binding regions of EleNRMT and ScaDMT. The coloring is as in

Figure 6B. (**B**) Metal ion binding sites of EleNRMT (**B**) and ScaDMT (**C**). B, C, inset (right)

shows molecular surface surrounding the bound ion with indicated relationship. (**D**) Region of

EleNRMT on α3 and α9 implicated in proton transport in NRAMPs and corresponding region in

ScaDMT (**E**). D, E the bound ion with the conserved binding site Asp on α1 and a residue

corresponding to the α6b His in NRAMPs, which was identified as potential H$^+$ acceptor are shown

for reference. A-E The proteins are depicted as Cα-trace, selected residues as sticks. The position

of bound metal ions is represented as a black sphere. B-E selected secondary structure elements

and residues are indicated.

1116

Figure 8. Functional characterization of binding site mutants. (**A**) Model of the ion binding site of

EleNRMT with ion (black sphere) showing bound Mn^{2+}. Regions of α1 and α6 are shown as Cα-

trace, selected residues coordinating the ion as sticks. (**B-D**) Mn^{2+} transport mediated by

EleNRMT[ts] binding site mutants. (**B**) D55A (2 (0 μM Mn^{2+}), 6 (50 μM Mn^{2+}) and 7 (400 μM

Mn^{2+})experiments from 2 independent reconstitutions), (**C**) Q379A (2 (0 µM Mn^{2+}), 6 (50 µM

Mn^{2+}) and 7 (400 µM Mn^{2+}) experiments from 2 independent reconstitutions), (**D**) T224A (5

experiments from 2 independent reconstitutions). (**E**) Mn^{2+} concentration dependence of transport

of T224A. Initial velocities were derived from individual traces of experiments displayed in **D**, the

solid line shows the fit to a Michaelis–Menten equation with an apparent K_m exceeding 290 µM.

(**F**) Mn^{2+} transport into T224A proteoliposomes in presence of Mg^{2+} (4 experiments from 2

independent reconstitutions). (**G**) Mg^{2+}-transport into T224A proteoliposomes assayed with the

fluorophore Magnesium green (4 experiments from 2 independent reconstitutions). H-I Transport

mediated by the $EleNRMT^{ts}$ binding site mutant T224A/G58N. (**H**) Mn^{2+} transport into

proteoliposomes containing T224A/G58N (2 (0 µM Mn^{2+}), 6 (50 µM Mn^{2+}) and 7 (400 µM Mn^{2+})

experiments from 2 independent reconstitutions). (**I**) Ca^{2+}-transport into T224A/G58N

proteoliposomes assayed with the fluorophore Fura-2 (9 (0 µM Ca^{2+}), 10 (50 µM Ca^{2+}), 11 (150

µM Ca^{2+}) and 10 (500 µM Ca^{2+}) experiments from 2 independent reconstitutions). **B-D, F- H**

Uptake of Mn^{2+} was assayed by the quenching of the fluorophore calcein trapped inside the

vesicles. **B-D, F-I** Averaged traces are presented in unique colors. Fluorescence is normalized to

value after addition of substrate (t=0). Applied ion concentrations are indicated. **B-I**, Data show

mean of the indicated number of experiments, Errors are s.e.m.

Figure 9. Features of Mg^{2+} transport by NRMTs. Schematic depiction of (**A**), the octahedral

coordination of the first hydration shell surrounding a Mg^{2+} ion and (**B**), the same octahedral

coordination within the binding site of a NRMT with five of the six water molecules remaining

bound to the ion and where the bulk of protein interactions with the ion are mediated via

coordinated water. (**C**) Model of Mg^{2+} bound to the site of EleNRMT in the inward-facing

conformation. The coordinates of the hydrated Mg^{2+} were obtained from the high-resolution

structure of the Mg^{2+} transporter MgtE (PDBID 4U9L). The Mg^{2+} ion was placed on the position

of Mn^{2+} obtained from the anomalous difference density defined in the X-ray structure. A single

water molecule with a backbone oxygen atom was removed and the remaining complex was

oriented to maximize interactions of coordinating waters with protein residues. Regions of the

binding site are represented as Cα-trace and selected interacting residues as sticks. The molecular

surface surrounding the binding site of the apo transporter is shown. (**D**) Schematic depiction from

an -inward-facing to an occluded conformation of a NRMT with bound hydrated Mg^{2+} indicated.

1152   (**E**) Approximate model of the transition of EleNRMT from an inward-facing (grey) to an occluded

1153   conformation (colored) that was based on the occluded conformation of DraNRAMP (PDBID:

1154   6C3I). (**F**) Close-up of the modeled binding pocket with bound Mg^{2+}-$5H_2O$ complex. E-F, binding

1155   pocket was calculated from EleNRMT modeled in and occluded conformation.

 Table 1. Cryo-EM data collection, refinement and validation statistics.

	Dataset 1 EleNRMT-NB (EMDB-) (PDB)	Dataset 2 EleNRMT-NB Mg^{2+} complex (EMDB-) (PDB)
Data collection and processing		
Microscope	FEI Titan Krios	FEI Titan Krios
Camera	Gatan K3 GIF	Gatan K3 GIF
Magnification	130'000	130'000
Voltage (kV)	300	300
Electron exposure (e$^-$/Å^2)	69.725/ 61/ 69.381	69.554
Defocus range (μm)	-1.0 to -2.4	-1.0 to -2.4
Pixel size* (Å)	0.651 (0.3255)	0.651 (0.3255)
Initial number of micrographs (no.)	22117	12427
Initial particle images (no.)	4 139 894	2 582 066
Final particle images (no.)	453 950	100 176
Symmetry imposed	C1	C1
Map resolution (Å) FSC threshold	3.5 0.143	4.1 0.143
Map resolution range (Å)	2.9-7.0	3.3-8.3
Refinement		
Model resolution (Å) FSC threshold	3.5 0.5	4.1 0.5
Map sharpening b-factor (Å^2)	-184.4	-159.2
Model vs Map CC (mask)	0.79	0.79

Model composition
    Non-hydrogen atoms
    Protein residues
    Ligand
    Water
	
	4846
	640
	0
	3
		
		4844
		640
		1 (Mg)
		0

B factors (Å2)
    Protein
    Ligand
    Water
	
	39.92
	0
	22.87
		
		44.04
		28.38
		0

R.m.s. deviations
    Bond lengths (Å)
    Bond angles (°)
	
	0.003
	0.641
		
		0.003
		0.652

Validation
  MolProbity score   2.08   2.17
  Clashscore   11.93   13.05
  Poor rotamers (%)   0   0

Ramachandran plot
  Favored (%)   91.96   90.06
  Allowed (%)   8.04   9.94
  Disallowed (%)   0   0

*Values in parentheses indicate the pixel size in super-resolution.

 Table 2. X-ray data collection and refinement statistics

	EleNRMT-Nb complex	EleNRMT-Nb complex Mn^{2+}-soak
Data collection		
Space group	P21	P21
Cell dimensions		
a, b, c (Å)	93.3, 122.2, 149.1	92.9, 115.9, 149.1
α, β, γ (°)	90, 107.8, 90	90, 107.6, 90
Wavelength (Å)	1.000	1.896
Resolution (Å)	50-4.1 (4.2-4.1)	50-4.6 (4.7-4.6)
R_{merge}	12.5 (166.8)	15.4 (147.6)
$I / \sigma I$	11.6 (2.4)	10.8 (1.8)
Completeness (%)	99.1 (98.9)	99.8 (99.6)
Redundancy	25.1 (27.1)	14.0 (14.1)
Refinement		
Resolution (Å)	12-4.1	12-4.6
No. reflections	23700	15750
R_{work} / R_{free}	27.1 / 29.6	27.0 / 31.8
No. atoms		
Protein	9536	9536
solvent	6	6
B-factors		
Protein	286.4	272.4
solvent	247.9	288.7
R.m.s deviations		
Bond lengths (Å)	0.005	0.005
Bond angles (°)	0.66	0.66

 Values in parentheses are for highest-resolution shell.