

Investigating Increased CO₂ concentration on the pH of various plant species

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Abstract

The concept of bioremediation is quickly becoming the norm in the resolution of environmental issues. The steady increase in carbon dioxide (CO₂) levels, as documented by NASA, inspired scientists to engineer plants to absorb excess CO₂ from the atmosphere. Here, we have explored the consequences of the uptake of excess CO₂ by select plants. Carbon dioxide dissolves in H₂O to produce H₂CO₃, which dissociates to yield H⁺ ions. We hypothesized that increased CO₂ absorption results in decrease in pH of plant sap. Three plants (*Byophyllum pinnatum*, Romaine Lettuce and Nevada Lettuce), exposed to increased CO₂ concentrations (15%), demonstrated a consistent increase in pH towards alkalinity compared to control plants. Based on the outcome being opposite of what we have hypothesized, our results suggest *Byophyllum pinnatum*, Romaine lettuce and Nevada lettuce, all have a unique homeostatic system to prevent over-absorption of CO₂ in a CO₂-rich environment.

Introduction

This work is inspired by the concept of bioremediation; the use of naturally occurring or genetically engineered organisms to solve environmental problems. Carbon dioxide levels have steadily increased over the years, raising concerns for individuals who want to solve this problem. In Scientists quests to address this problem, the idea of engineering plants to increase the capacity of CO₂ absorption from the environment is still in the early stages. Others have pursued the use of a semi-synthetic rubisco, which has the ability to increase the rate of photosynthesis [2]. We ask the question, what will be the effect of excess CO₂ on the pH of plant sap? Theoretically, carbon dioxide dissolves in water to produce carbonic acid according to equation [1].



According to equation [2], dissolved CO₂ in the form of H₂CO₃ may lose up to two protons through the acid equilibrium. The relatively small amounts of H⁺ produced, when built up, is anticipated to decrease the pH of plant cytosol. We hypothesize that the exposure of plants to increased CO₂ levels results in a decrease in the pH of plant cytosol.

Objectives

- Monitor pH of plant sap at increased CO₂ concentrations
- Determine whether bioremediation is a safe solution to combat CO₂ pollution based on pH data
- Further understand the relationship between CO₂ and pH of plant sap

Materials

- pH meter
- Garlic press

- CO₂ incubator
- Lamp for light source
- Plant soil heater

Method

Plants (e.g. *Byophyllum pinnatum*) were incubated in the presence of increased CO₂ with a light source or under normal atmospheric conditions with a light source. The lighting schedule, temperature and watering of the two plants were kept the same, with the only variable being the difference in CO₂ concentration. The control plant was maintained in atmospheric CO₂ concentrations and the test plant was maintained in 15% CO₂ concentration. The light sources were on for approximately ten hours per day and the plants were watered once per week. The same set up was observed for 6 stalks each of *Romaine lettuce* and *Nevada lettuce*. Every 2 to 3 days, leaves from each plant was homogenized for sap extraction and pH testing. The results are as follows:

Results: *Byophyllum pinnatum*

Table 1: pH readings vs exposure time in *Byophyllum pinnatum*

Results: *Byophyllum Pinnatum*

Figure 1: The results were not what we expected. The plant exposed to excess CO₂ became more alkaline instead of more acidic, while the control plant remained well under pH=5. This suggests *Bryophyllum pinnatum* has a homeostatic system to prevent over-absorption of CO₂ in a CO₂-rich environment.

Results: *Romaine lettuce (Lactuca sativa, variety longifolia)*

Table 2: pH readings vs exposure time (Days) in Romaine lettuce

Figure 2: The results were not what we expected. The plant exposed to excess CO₂ became more alkaline instead of more acidic, while the control plant remained mostly under pH=6.3. This suggest *Romaine lettuce* also has a homeostatic system to prevent over-absorption of CO₂ in a CO₂-rich environment.

Results: *Nevada lettuce (Lactuca sativa)*

Table 3: pH readings vs exposure time (Days) in Nevada lettuce

Figure 3: The results were not what we have expected. The plant exposed to excess CO₂ became more alkaline instead of more acidic, while the control plant remained well under pH=6.3. This suggest *Nevada lettuce* also has a homeostatic system to prevent over-absorption of CO₂ in a CO₂-rich environment.

Discussion and Conclusion

At the end of our experiment, we found pH levels moving in the opposite direction of what we had hypothesized. That is, the pH level of plants exposed to an increased level of CO₂ increased, compared to the control plants. The more CO₂ the plant was exposed to, the higher the pH of its internal environment became. We reject our initial hypothesis that a plant exposed to an increased CO₂ will decrease the pH of its cytosol. Based on these results, we believe plant systems have unique homeostatic mechanism that drives out excess CO₂ which results in the loss of normal levels of CO₂ and causes the plant to run in a CO₂ deficit. This would explain the increase in pH. This homeostatic system needs further investigation.

After performing a 2-Way ANOVA, we can conclude that statistically, there is no difference in the pH between the two different plants ($p=1$); but there is a difference in the pH between plants exposed to higher CO₂ and normal CO₂ ($p=0.012$); there is no interaction between type of plant and CO₂ level ($p=0.65$).

Further Hypotheses

We further hypothesize that the exposure of plants to increased CO₂ accelerates the rate of plant glucose breakdown.

Acknowledgment

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References

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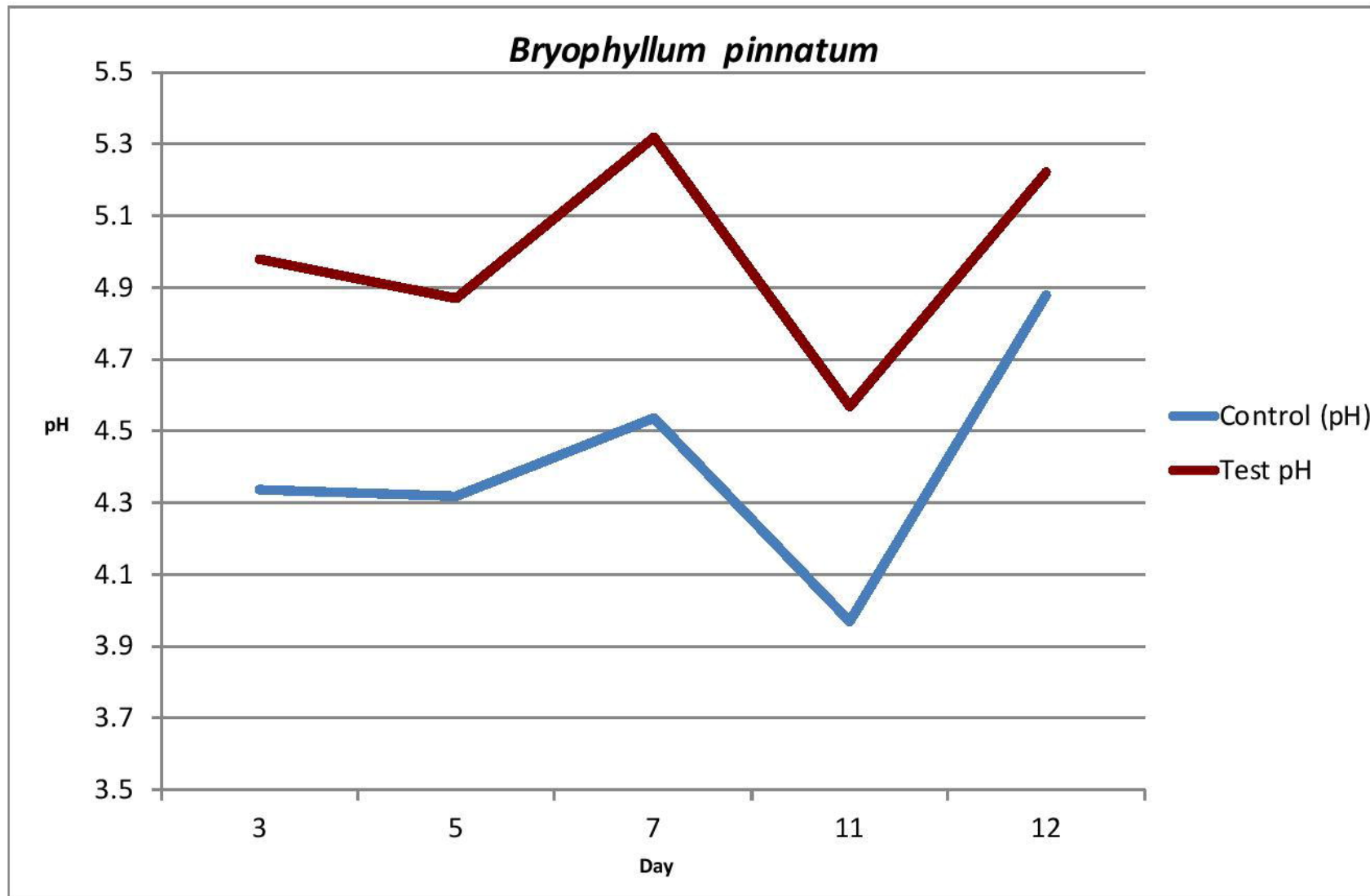


Figure 1

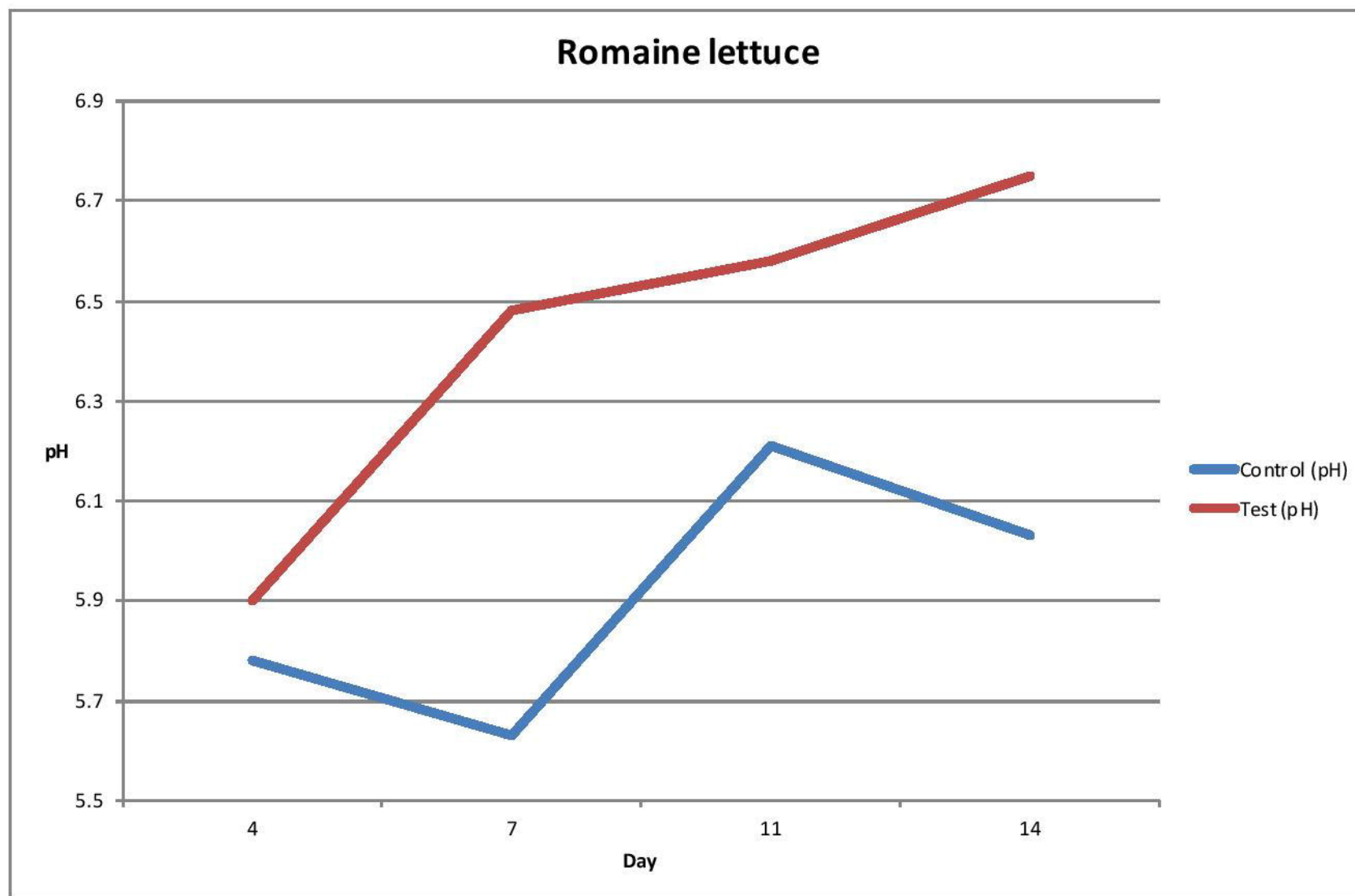


Figure 2

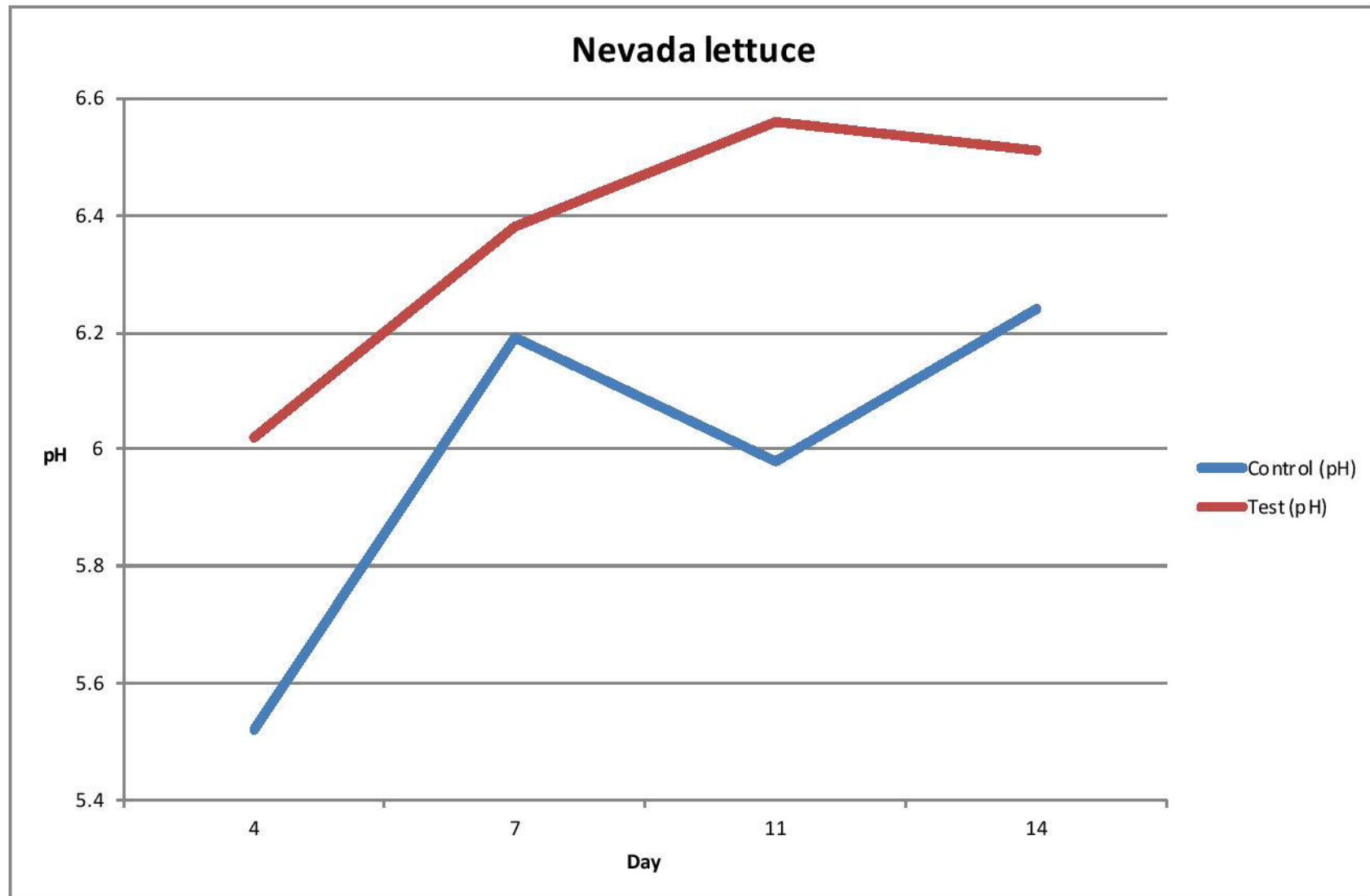


Figure 3

Table 1

Day	Control (pH)	Test (pH)
3	4.34	4.98
5	4.32	4.87
7	4.54	5.32
11	3.97	4.57
12	4.88	5.22

Table 2

Day	Control (pH)	Test (pH)
4	5.78	5.9
7	5.63	6.48
11	6.21	6.58
14	6.03	6.75

Table 3

Day	Control (pH)	Test (pH)
4	5.52	6.02
7	6.19	6.38
11	5.98	6.56
14	6.24	6.51