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Stable Potassium Isotopes (^{41}K/^{39}K) Track Transcellular and Paracellular Potassium Transport in Biological Systems

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Abstract: As the most abundant cation in archaeal, bacterial, and eukaryotic cells, potassium (K^+) is an essential element for life. While much is known about the machinery of transcellular and paracellular K transport – channels, pumps, co-transporters, and tight-junction proteins - many quantitative aspects of K homeostasis in biological systems remain poorly constrained. Here we present measurements of the stable isotope ratios of potassium ($^{41}K/^{39}K$) in three biological systems (algae, fish, and mammals). When considered in the context of our current understanding of potential mechanisms of K isotope fractionation and K^+ transport in these biological systems, our results provide evidence that the fractionation of K isotopes depends on transport pathway and transmembrane transport machinery. Specifically, we find that passive transport of K^+ down its electrochemical potential through channels and pores in tight-junctions at favors ^{39}K, a result which we attribute to a kinetic isotope effect associated with dehydration and/or size selectivity at the channel/pore entrance. In contrast, we find that transport of K^+ against its electrochemical gradient via pumps and co-transporters is associated with less/no isotopic fractionation, a result that we attribute to small equilibrium isotope effects that are expressed in pumps/co-transporters due to their slower turnover rate and the relatively long residence time of K^+ in the ion pocket. These results indicate that stable K isotopes may be able to provide quantitative constraints on transporter-specific K^+ fluxes (e.g. the fraction of K efflux from a tissue by channels vs. co-transporters) and how these fluxes change under different physiological states. In addition, precise determination of K isotope effects associated with K^+ transport through channels, pumps, and co-transporters may provide unique constraints on the mechanisms of K transport that could be tested with steered molecular dynamic simulations.

Keywords: potassium, homeostasis, stable isotope, physiology

1. Introduction

Archaeal, bacterial, and eukaryotic cells all concentrate potassium (K^+). In fact, the ubiquity of elevated intracellular K^+ has been hypothesized to provide insight into the environments (high K/Na ratios) where the first cells emerged (Mulkidjanian et al., 2012). In bacteria, intracellular K^+ is at >100 mM by the active pumping of K^+ from extracellular fluid (ECF) into the intracellular fluid (ICF) by transporters that use ATP or the proton motive force (Stautz et al., 2021). In eukaryotic cells, intracellular K^+ is maintained at similarly elevated levels by the active pumping of K^+ by plasma membrane Na,K-ATP$_{ase}$, an energetically intensive process which can account for ~20% of basal metabolism in mammals (Rolfe and Brown, 1997). In all cells, the active transport of K^+ into the cell establishes a steep transmembrane gradient that is a key determinant of membrane potential and a source of energy to drive action potentials, control muscle contractility,

maintain cell turgor, contribute to pH homeostasis, and power ion transporters (Stautz et
al., 2021; Youn and McDonough, 2009).

The regulation of K^+ at both the cellular and organismal level, K^+ homeostasis, is
ultimately accomplished at the molecular level by transcellular and paracellular
transporters (including ion pumps, channels, co-transporters) that move K^+ across
membranes and between cells (Agarwal et al., 1994; Furukawa et al., 2012; Gierth and
Mäser, 2007; Youn and McDonough, 2009). Although life has evolved a diversity of
molecular machines capable of K^+ transport across cell membranes, certain aspects of
cellular K^+ transport are conserved across the major domains. For example, all known K^+
channels are members of a single protein family whose amino acid sequence contains a
highly conserved segment (the K^+ channel signature sequence (MacKinnon, 2003)). The
shared molecular machinery of K^+ transport in biological systems reflects the
fundamental role of the electrochemical potential generated by biologically maintained
gradients of K^+ across cell membranes. K^+ pumps establish and maintain concentration
gradients across cell membranes by moving K^+ (from ECF-ICF) and Na or H (from ICF to
ECF) "uphill" against their electrochemical potentials coupled to and driven by the
hydrolysis of ATP. K^+ co-transporters couple uphill K^+ transport to "downhill" transport of
another ion (e.g., Na^+ or Cl^-). Finally, K^+ channels and the pores in tight-junction proteins
provide an energetically favorable pathway for rapid, yet highly selective, transport of K^+
down its electrochemical potential.

While the identity and structure of the molecular machines that maintain K^+
homeostasis in biological systems are well-known, many quantitative aspects of K^+
homeostasis and the specific mechanisms of K^+ transport by pumps, channels, co-
transporters, and through tight-junction proteins are unconstrained or debated. For
example, structural, functional and computational studies of K^+ channels produce
conflicting results as to whether the molecular mechanism of K^+ permeation is water
mediated or occurs by direct knock-on of cations (Mironenko et al., 2021). In addition,
while studies of gene expression in cells and tissues can be used to identify which K
transporters are active, quantitative constraints on the fluxes of K^+ through the different
types of transporters *in vivo* are non-existent.

In natural systems potassium is made up of two stable (potassium-39 and
potassium-41) and one radioactive (potassium-40) isotope. The two stable isotopes of K,
^{39}K and ^{41}K, constitute 93.258% and 6.730% of the total, respectively, resulting in a ratio
of $^{41}K/^{39}K$ in nature of ~0.07217. Recent advances in inductively coupled plasma mass
spectrometry (ICP-MS) now permit the precise quantification of deviations from the
terrestrial ratio resulting from the biogeochemical cycling of potassium in nature with a
precision of 1 part in 10,000 (Morgan et al., 2018; Wang and Jacobsen, 2016). Here we
apply this analytical tool to study K^+ homeostasis in three biological systems – aquatic
green alga *Chlamydomonas reinhardtii* (*C. reinhardtii*), a suite of marine fish that include

species that can tolerate a wide range in salinities (euryhaline) as well as those that
have a restricted salinity range (stenohaline), and the terrestrial mammal *Rattus*
norvegicus (*R. norvegicus*). The results in each system are interpreted as reflecting K$^+$
homeostasis under normal (optimal) growth conditions. The data are presented using
standard delta notation in parts per thousand (‰).

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$$\delta^{41/39}K = \left(\frac{R^{41K/39K}_{sample}}{R^{41K/39K}_{standard}} - 1 \right) \cdot 10^3$$

where $R^{41K/39K}_{sample}$ is the ratio of $^{41}K/^{39}K$ in a sample and $R^{41K/39K}_{standard}$ is the $^{41}K/^{39}K$ of the
standard). For plants the standard is the $^{41}K/^{39}K$ of the growth medium, whereas for
marine teleosts it is the $^{41}K/^{39}K$ of seawater. For *R. norvegicus*, we report the data in one
of two ways. To explore total body K$^+$ homeostasis we report the $^{41}K/^{39}K$ of K excretory
losses normalized to the $^{41}K/^{39}K$ of the diet whereas to examine the partitioning of K
isotopes between extracellular and intracellular compartments we report the $^{41}K/^{39}K$ of
tissues relative to the $^{41}K/^{39}K$ of blood plasma or cerebrospinal fluid (CSF). K isotope
fractionation factors are calculated as ratios of isotope ratios (e.g. $\alpha_{medium/cell} =$
$[^{41}K/^{39}K_{medium}] / [^{41}K/^{39}K_{cell}]$) and reported as ε values (in ‰) where $\varepsilon_{medium/cell} = \alpha_{medium/cell} -$
1.

2. **Materials and Methods**

C. reinhardtii cultures: The CMJ030 wild type strain was obtained from the
Chlamydomonas culture collection www.chlamycollection.org). Tris Phosphate (TP)
medium was prepared according to: Gorman, D.S., and R.P. Levine (1965) Proc. Natl.
Acad. Sci. USA 54, 1665-1669. The culture at an initial density of 0.5 x10^5 cells mL^{-1} was
grown in TP under continuous illumination (100 µmol photons m^{-2} s^{-1}) and shaking for
four days. Samples (in triplicate) containing 2 x10^7 cells (~800 mg) were harvested and
washed twice in 5 mM HEPES and 2 mM EDTA before collection and air-drying of the
cell pellet. Pelletized cells (~30mg) were digested in screw-capped teflon vials on a hot
plate at elevated temperatures (~75 °C) using a 5:2 mixture of HNO$_3$ (68-70 vol.%) and
H$_2$O$_2$ (30 vol.%).
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Euryhaline and stenohaline marine teleosts: Samples of teleost muscle tissue were
sourced from fish markets (Nassau Seafood and Trader Joe's in Princeton, NJ and the
Fulton Fish Market in Brooklyn, NY) and research cruises (NOAA NEFSC Bottom Trawl
Survey, Fall 2015 and Spring 2015). All teleosts were caught in seawater which has a
uniform $\delta^{41}K$ value of +0.12‰ relative to SRM3141a (Ramos et al., 2020). Samples of
white dorsal muscle (100 to 3000 mg) were digested on a hot plate at elevated
temperatures (~75 °C) or in a high-pressure microwave system (MARS 6) using HNO$_3$
(68-70 vol.%) H$_2$O$_2$ (30 vol.%) in a ratio of 5:2. Major/minor element analyses for
digested samples were carried out at Princeton University using a quadrupole inductively

coupled plasma mass spectrometer (Thermo Scientific iCap Q). Concentrations and
elemental ratios were determined using externally calibrated standards and average
uncertainties (element/element) are ~10%.

R. norvegicus experiments: All rat experiments were approved by the Institutional Animal
Care and Use Committees of the University of Southern California. Two series were
conducted. Series #1: Male Wistar rats (n=3, 250-275g body weight, Envigo,
Indianapolis, IN) were housed in a climate controlled (22-24°C) environment with a 12
hr: 12hr light/dark cycle, and fed casein based normal K^+ diet TD.08267 (Envigo,
Indianapolis, IN) and water ad libitum for 11 days. At day 8, rats were placed overnight
into metabolic cages (Techniplast, Buguggiate, Italy) with food and water ad libitum for
16-hour collection of urine and feces. On termination day (1:30-3:30PM), rats were
anesthetized with an intramuscular (IM) injection of ketamine (80 mg/kg, Phoenix
Pharmaceuticals, St. Joseph, MO) and xylazine (8 mg/kg, Lloyd Laboratories,
Shenandoah, IA) in a 1:1 ratio. Through a midline incision, the liver, kidneys, heart, fat
pads, and stomach (flushed of contents) were removed; blood was collected via cardiac
puncture, spun down to separate plasma from RBCs. Then gastrocnemius, soleus, TA,
and EDL skeletal muscles were dissected. All tissues were washed in ice-cold TBS to
remove excess blood, weighed and snap frozen in liquid nitrogen. Series #2: Male
Sprague Dawley rats (n=4, 250-300g, Envigo, Indianapolis, IN) were housed in a climate
controlled (22-24°C) environment with a 12 hr: 12hr light/dark cycle and fed grain-based
vivarium chow (LabDiet 5001, labdiet.com). CSF extraction procedures are as reported
previously (Noble et al., 2018). In brief, Rats were deeply anesthetized using a cocktail
of ketamine 90mg/kg, xylazine, 2.8 mg/kg, and acepromazine 0.72 mg/kg by
intramuscular injection. A needle was lowered to below the caudal end of the occipital
skull and the syringe plunger pulled back slowly, allowing the clear CSF to flow into the
syringe. After extracting ~100-200 μl of CSF, the needle was raised quickly (to prevent
suction of blood while coming out of the cisterna magna) and the CSF dispensed into a
microfuge tube and immediately frozen in dry ice and then stored at -80 °C until time of
analysis. Following CSF extraction and decapitation, whole brains with 10-15mm spinal
cord extension were rapidly removed and immediately flash frozen and stored in -80°C
until dissection into spinal cord, cerebrum, and cerebellum for subsequent digestion and
K isotopic analysis.

Ion chromatography and isotope ratio mass-spectrometry: K was purified for isotopic
analyses using an automated high-pressure ion chromatography (IC) system. The IC
methods utilized here followed those previously described in (Morgan et al., 2018;
Ramos et al., 2018). The accuracy of our chromatographic methods was verified by
purifying and analyzing external standards (SRM3141a and SRM70b) alongside
unknown samples. Purified aliquots of K were analyzed in 2% HNO_3 for their isotopic
compositions on a Thermo Scientific Neptune Plus multi-collector inductively coupled
plasma mass spectrometer (MC-ICP-MS) at Princeton University, using previously
published methods (Morgan et al., 2018; Ramos et al., 2020). The external
reproducibility of our protocols (chromatography and mass spectrometry) was
determined through replicate measurements of international standards. Measured

values of SRM70b, reported relative to SRM3141 ($\delta^{41}K_{SRM3141}$) are -0.57 ± 0.17‰ (2σ; N

= 59), indistinguishable from published values (Ramos et al., 2018; Ramos et al., 2020;

Wang and Jacobsen, 2016). For samples analyzed once (chromatography and mass

spectrometry), reported errors are the 2σ uncertainties of the external standard. In cases

where samples were analyzed multiple times, reported errors in are twice the standard

error of the mean (2 S.E. or 95% confidence).

3. Results

Results for the freshwater algae *C. reinhardtii* are shown in Figure 1. Measured

$\delta^{41}K$ values of the whole cells are 1.2±0.07‰ lower than the $\delta^{41}K$ value of the growth

medium (0‰ by definition) (95% confidence; $P = 4.5 \times 10^{-5}$).

Results for *R. norvegicus* are shown in Figures 2-4. Measured $\delta^{41}K$ values of

both urine and feces, normalized to the $\delta^{41}K$ value of the rat diet ($\delta^{41}K_{diet}$), indicate

preferential net uptake of ^{39}K relative to ^{41}K across the gut epithelium leading to a

positive $\delta^{41}K_{diet}$ value in feces (+0.19±0.09‰, 95% confidence; $P = 0.037$; Fig. 2) and

slightly negative $\delta^{41}K_{diet}$ value in urine (-0.09±0.10‰, 95% confidence; $P = 0.1895$).

Measurements of K isotopes of various tissues in *R. norvegicus* (except brain

tissues) are normalized to the $\delta^{41}K$ value of the blood plasma ($\delta^{41}K_{plasma}$), as this

represents the extracellular K concentration bathing the tissues. For brain tissues,

measured $\delta^{41}K$ values are normalized to CSF ($\delta^{41}K_{CSF}$) for the same reason. The overall

range in $\delta^{41}K_{plasma}$ values between different tissues and fluids is ~1‰ (Figs. 3,4); $\delta^{41}K$

values of the following tissues are elevated in ^{41}K relative to blood plasma (positive

$\delta^{41}K_{plasma}$ values): red blood cells (+0.40±0.08‰, $P = 0.005$), heart (+0.55±0.15‰, $P =$

0.0047), liver (+0.30±0.05‰, $P = 0.01$), and soleus muscle (+0.17±0.03‰, $P = 0.05$).

Similarly, brain tissues are elevated in ^{41}K relative to CSF (positive $\delta^{41}K_{CSF}$ values):

cerebrum (+0.34±0.15‰, $P = 0.012$), spinal cord (+0.30±0.17‰, $P = 0.026$), and

cerebellum (+0.21±0.11‰, $P = 0.037$). In contrast, urine (-0.50±0.10‰, $P = 0.0019$) and

CSF (-0.59±0.12‰, $P = 6.2 \times 10^{-5}$) are characterized by negative $\delta^{41}K_{plasma}$ values. In

another category are tissues with $\delta^{41}K_{plasma}$ values that are statistically indistinguishable

from zero including: kidneys (+0.07±0.07‰, $P = 0.31$), adipose tissue (+0.01±0.08‰, P

= 0.92), extensor digitorum longus (EDL, +0.17±0.11‰, $P = 0.099$) gastrocnemius

(+0.01±0.15‰, $P = 0.92$), and tibialis anterior (TA, +0.09±0.1‰, $P = 0.30$).

Results for white muscle tissue from a suite of stenohaline and euryhaline marine

fish, reported relative to the $^{41}K/^{39}K$ of seawater ($\delta^{41}K_{seawater} = 0$‰) are shown in Figure

5a. The total range in muscle $\delta^{41}K$ values is ~2‰ (+1‰ to -1‰). Stenohaline species

including *Gadus morhua* (Atlantic Cod), *Peprilus striacanthus* (Butterfish), Xiphias

gladius (Swordfish), *Pseudopleuronectes americanus* (Winter Flounder), and

Hippoglossus stenolepis (Pacific Halibut) are characterized by $\delta^{41}K_{seawater}$ values that are
uniformly negative whereas the euryhaline species *Oncorhynchus kisutch* (Coho
Salmon), *Oncorhynchus tshawytscha* (King Salmon), and *Oncorhynchus nerka*
(Sockeye Salmon) are characterized by $\delta^{41}K_{seawater}$ values that are close to zero or
positive. When species are grouped by salinity tolerance, average measured $\delta^{41}K_{seawater}$
values of the two groups are -0.58±0.09‰, and 0.29±0.18‰ for stenohaline and
euryhaline species, respectively (95% confidence; $P = 1.4 \times 10^{-10}$).

4. **Discussion**

The shared machinery of K^+ transport across the domains of life suggests that
the variability in K isotopes observed in biological systems can be reduced to a
consideration of 1) how different classes of K^+ transporters - tight junctions, channels,
pumps, and co-transporters - discriminate between ^{39}K and ^{41}K and 2) how these
transporters are assembled into a homeostatic system. The extent to which K^+
transporters fractionate K isotopes will, in turn, depend on the mechanics and selectivity
of the transporter and whether the binding of dissolved K^+ to transporter is the rate-
limiting step. In particular, K^+ channels and pores in tight-junction proteins are both
associated with selective and rapid (near the limit of diffusion) transport of K^+ where the
rate-limiting step is the binding of dissolved K^+ to the transporter/selectivity filter. These
conditions favor kinetic isotope effects which may arise from either partial or full
desolvation of K^+ as it binds to the selectivity filter (Berneche and Roux, 2003; Hofmann
et al., 2012; Kopec et al., 2018; Yu et al., 2010), differences in ionic radius of ^{39}K and ^{41}K
(size selectivity (Christensen et al., 2018)), or some combination of the two. Critically,
the magnitude of the kinetic isotope effects associated with both desolvation and/or size
selectivity appear to be large, with estimates from laboratory experiments and molecular
dynamic simulations ranging from ~1 to 2.7‰ (Hofmann et al., 2012).

In contrast to channels and pores in tight-junction proteins, transport of K^+ by
pumps and co-transporters is relatively slow (per unit transporter, (Roux, 2017)),
requires the simultaneous binding of multiple ions (e.g. 2 K^+ ions in the case of Na,K-
ATPase), and is associated with mechanisms of self-correction that prevent the pump
cycle from proceeding if incorrect ions bind to the ion pocket (Rui et al., 2016). As a
result, although the binding of K^+ to the active site of a pump or cotransporter may also
involve desolvation, it is not the rate-limiting step in K^+ transport. In this case, the
isotopic composition of K^+ that is transported by pumps and co-transporters may reflect
isotopic equilibration between K^+ in the ion pocket and the fluid due to the longer
residence time of K^+ in the ion pocket prior to occlusion. As equilibrium K isotope effects
tend to be small (Li et al., 2017; Ramos et al., 2018; Zeng et al., 2019), we hypothesize
that K^+ transport by pumps and co-transporters will be associated with less isotopic
fractionation.

In the following sections we show how our results for the three studied biological
systems (algae, fish, mammals) can be interpreted within a single framework where K$^+$
transport by channels and pores in tight-junction proteins are associated with large
(kinetic) K isotope effects whereas K$^+$ transport by pumps and/or co-transporters is
associated with smaller (equilibrium) K isotope effects. In simple biological systems (e.g.
single celled alga) linking K isotope fractionation to the machinery of K$^+$ transport is
relatively straightforward due to the small number of pathways involved and the
unidirectional nature of K$^+$ transport. In more complex biological systems (e.g. higher
plants, fish, and mammals) discerning transporter-specific fractionation of K isotopes is
more difficult for at least two reasons. First, our understanding of K homeostasis at both
the tissue and organism level is incomplete. Second, our measured K isotopic
differences (e.g. between plasma/CSF/seawater and tissue) only provide information on
net K isotope fractionation associated with bi-directional K$^+$ transport at steady-state. In
spite of these limitations, our results support a framework for K isotope fractionation in
biological systems that can provide new quantitative insights into the mechanisms of K$^+$
transport and the dynamics of K homeostasis.

4.1 Algae. As an essential macronutrient in all plants and the most abundant cation in
the cytoplasm, K$^+$ contributes to electrical neutralization of anionic groups, membrane
potential and osmoregulation, photosynthesis, and the movements of stomata (Gierth
and Mäser, 2007; Marchand et al., 2020). It is well-established that K$^+$ channels play
prominent roles in K$^+$ uptake (Lebaudy et al., 2007). For example, under normal growth
conditions ([K$^+$]$_{ext}$ ~1 mM; (Epstein et al., 1963; Kochian and Lucas, 1982; Malhotra and
Glass, 1995)) K$^+$ uptake in plants is dominated by transport via inward-rectifying K$^+$
channels electrically balanced by the ATP-driven efflux of H$^+$ (Britto and Kronzucker,
2008; Hirsch et al., 1998). Considered in this context our results for whole cells of *C.*
reinhardtii can be interpreted as reflecting K isotope fractionation associated with uptake
via K$^+$ channels ($\varepsilon_{in} = \delta^{41}K_{medium} - \delta^{41}K_{cell}$ = ~1.2‰; Figure 1). Although we cannot
guarantee that K uptake occurs exclusively via K$^+$ channels, this result represents our
best estimate of a transporter-specific K isotope effect (ε_{ch}). The magnitude of this effect
is smaller than modeled estimates of the kinetic isotope effect associated with ion
desolvation (Hofmann et al., 2012) and of the same order as hypothesized kinetic
isotope effects associated with size-selectivity (Christensen et al., 2018).

4.2 Higher Plants. Our result for K isotope fractionation associated with K$^+$ uptake in *C.*
reinhardtii is similar in sign, though somewhat larger in magnitude, than the K isotope
effect associated with K$^+$ uptake estimated by Christensen et al. (2018) in experiments
involving higher plants. Those authors analyzed the δ^{41}K values of the roots, stems, and
leaves of *Triticum aestivum* (wheat), *Glycine max* (soy) and *Oryza sativa* (rice) grown
under hydroponic conditions and observed systematic differences in the δ^{41}K value of
the different reservoirs. In particular, roots, stems, and leaves exhibited increasingly

negative $\delta^{41}K$ values (Figure S1). Compared to the freshwater alga *C. reinhardtii*,
quantifying transport-specific K isotope effects in higher plants is more complex as the
$\delta^{41}K$ value of each individual compartment (root, stem, leaf) reflects a balance between
isotopic sources and sinks (e.g. the $\delta^{41}K$ value of the root will depend on K isotope
effects associated with both net K^+ uptake as well as translocation and recycling). Using
a model of K isotope mass balance that includes assumptions regarding plant growth
and the partitioning of K^+ fluxes between translocation and recycling, Christensen et al.
(2018) estimated large K isotope effects for uptake (ε_{in} ~0.7 to 1.0‰) and translocation
from root to stem (~0.6‰), but smaller isotopic effects (0 to 0.2‰) for translocation of K^+
from stem to leaf and recycling of K^+ from leaf to root. Interestingly, in addition to their
important role in K^+ uptake, K^+ channels have also been shown to be involved in
translocation (root to stem; (Gaymard et al., 1998)) providing a potential explanation for
the relatively large K isotope effect observed by Christensen et al. (2018) for this
process. In contrast, translocation and recycling of K^+ from leaf to root, a process that
also involves K^+ channels (Lacombe et al., 2000; Pilot et al., 2001), does not appear to
result in significant fractionation of K isotopes. This could be due to the increased
importance of other types of K^+ transporters in this process, a reduction in the
expression of K isotope fractionation by channels due to rapid internal recycling of K^+
between stem and leaf (compared to K^+ transport from root to stem), or some
combination of the two.

4.3 Terrestrial Mammals. K^+ homeostasis in terrestrial mammals reflects the balance
between K^+ gained from diet and K^+ lost in urine and feces (Figure 2, Figure 3a). This
balance is largely achieved by the kidneys and colon, which possess a remarkable
ability to sense a change in K^+ in the diet and then appropriately adjust K^+ loss in
response. ECF (which includes blood plasma), the reservoir through which internal K^+ is
exchanged, represents only 2% of total body K^+. Though small, this reservoir is tightly
regulated to maintain membrane potential as indicated by the narrow range of normal
ECF $[K^+]$ (~ 3.5 – 5 mEq/L,)(Boron, 2005; McDonough and Youn, 2017). Of the
remaining 98% of total body K^+, 75% resides in muscle tissues ($[K^+]$ ~ 130 mEq/L) and
23% in non-muscle tissues. Some tissues, particularly skeletal muscle, are critical to K^+
homeostasis by providing a buffering reservoir of K^+ that can take up K^+ after a meal and
altruistically donate K^+ to ECF to maintain blood plasma levels during fasting.
Cerebrospinal fluid (CSF), the body fluid that surrounds the brain and spinal cord of all
vertebrates, merits special mention as its K^+ content is even more tightly regulated than
ECF (Bradbury and Davson, 1965).

4.3.1. External K isotope mass balance in terrestrial mammals. Roughly ~80-90% of K in
the diet is lost in urine with the remainder lost in feces (Agarwal et al., 1994). Uptake of
K^+ across gut endothelial cells occurs largely paracellularly through pores in tight-
junction proteins like caudin-15 (Garcia-Hernandez et al., 2017) as a result of the

transmucosal electrical potential difference (Barnaby and Edmonds, 1969). However,
transcellular K$^+$ uptake (via pumps) and secretion (via channels) has also been shown to
occur in gut epithelial cells (Foster et al., 1984). As a result, K isotope fractionation
associated with net K uptake in the gut will reflect the flux-weighted average of the K
isotope fractionation associated with both transcellular and paracellular K$^+$ transport. The
observation that feces is characterized by a positive $\delta^{41}K_{diet}$ value and urine with a
slightly negative $\delta^{41}K_{diet}$ value (Figure 2) indicates that the net K isotope effect
associated with K uptake under these conditions (ε_{gut}) is ~0.2‰ (Figure 3b). Relative to
the diet, internal tissues of *R. norvegicus* exhibit a 1‰ range from -0.04‰ for the
cerebellum to +1.05‰ for the heart (Figure 2). Assuming an internal K$^+$ distribution that
is 75/15/10 muscle/adipose/other tissues, a reasonable average whole-body $\delta^{41}K_{diet}$
value of *R. norvegicus* is ~+0.5‰. As net uptake of K$^+$ in the gut prefers ^{39}K, the elevated
average whole-body $\delta^{41}K_{diet}$ of *R. norvegicus* requires that there be even greater K
isotope fractionation (favoring ^{39}K) associated with K$^+$ loss in the urine (see below).

4.3.2 Internal K isotope mass balance in terrestrial mammals. As shown in Figures 3a
and 4, measured δ^{41}K values of the 16 different tissues and fluids analyzed in *R.*
norvegicus fall into 3 distinct categories relative to ECF ($\delta^{41}K_{plasma}$ or $\delta^{41}K_{CSF}$): those with
positive $\delta^{41}K_{plasma/CSF}$ values (red blood cells, heart, liver, soleus muscle and brain
tissues), those with $\delta^{41}K_{plasma/CSF}$ values that are close to 0 (stomach, adipose tissue,
kidney, gastrocnemius, EDL and TA muscles), and those with negative $\delta^{41}K_{plasma/CSF}$
values (urine and CSF).

The timescale for K$^+$ turnover in the studied tissue/cell/fluid reservoirs is rapid
(e.g. 0.9-10%/min; (Jones et al., 1977; Terner et al., 1950)). As a result, for all measured
tissues/cells, the δ^{41}K values in Figure 3a can be interpreted as reflecting steady-state K
isotope mass balance between the tissue/cell/fluid and the relevant fluid (blood
plasma/ECF or CSF). At isotopic steady-state, the δ^{41}K value of K$^+$ entering the reservoir
must equal to δ^{41}K value of K$^+$ leaving the reservoir. As a result, reservoirs with
$\delta^{41}K_{plasma/CSF}$ values that differ from 0‰ require that net K$^+$ transport in one direction
results in greater fractionation of K isotopes than net K$^+$ transport in the opposite
direction. For example, reservoirs with positive $\delta^{41}K_{plasma}$ values including red blood cells,
heart, liver, and soleus require that K$^+$ transport from ICF to plasma or ECF is associated
with a larger K isotope effect than K$^+$ transport from ECF to ICF. The same relationship
is observed between brain tissues and CSF (Figure 4). Similarly, reservoirs with
$\delta^{41}K_{plasma}$ values that are close to 0‰ require that net K$^+$ transport in both directions does
not fractionate K isotopes, i.e., the isotope effects must be of equal magnitude and sign
and cancel. These include kidney, adipose tissue, stomach, and gastrocnemius and TA
muscles. Finally, the negative $\delta^{41}K_{plasma}$ values for CSF requires that net K$^+$ transport
from ECF to CSF through the choroid plexus and blood brain barrier (BBB) is

characterized by a K isotope effect that is ~0.6‰ greater than the K isotope effect
associated K$^+$ transport from CSF to ECF (Figure 4).

Quantitatively linking the observed isotopic differences between ECF and ICF of
various tissues/cells to the machinery of K$^+$ homeostasis in complex biological systems
is complicated by both an incomplete understanding of the machinery involved and
limitations of steady-state isotopic mass balance. In particular, although much is known
about the identity and molecular structure of the machinery that maintains K$^+$
homeostasis, quantitative information on how each transporter contributes to the gross
fluxes of K$^+$ between ICF and ECF at the resting membrane potential is lacking. For
example, in red blood cells, probably the best understood with regard to the machinery
of K$^+$ homeostasis due to the fundamental role it plays in the regulation of cell volume
and longevity (Tosteson and Hoffman, 1960), elevated intercellular K$^+$ is believed to be
maintained by a 'pump-leak' mechanism where the pump is Na,K-ATP$_{ase}$ and the leak is
K$^+$ channels (Gardos, 1958). However, red blood cells also show activity for co-
transporters such as NKCC (Duhm, 1987; Sachs, 1971) and KCC (Franco et al., 2013),
as well as inward-rectifying K$^+$ channels (Hibino et al., 2010), and exactly how much K$^+$
is transported by which transporter at the resting membrane potential is not well known.

In order to evaluate whether or not differences in measured δ^{41}K$_{plasma}$ values
reflect quantitative differences in K$^+$ fluxes through various transporters, we developed a
generic model of K homeostasis (Figure 3b) that assumes 1) the tissue/cell is at the
resting membrane potential and K is at isotopic steady-state between ECF and ICF, 2)
all channels fractionate K isotopes identically (ε_{ch}), 3) all pumps and co-transporters
fractionate K isotopes identically ($\varepsilon_{p/co}$), and 4) pumps, co-transporters, and channels
may transport K$^+$ in either direction ($F_{p/co}^{in}$, $F_{p/co}^{out}$, F_{ch}^{in}, and F_{ch}^{out}):

$$F_{p/co}^{in} + F_{ch}^{in} = F_{p/co}^{out} + F_{ch}^{out} \tag{1}$$

$$(\delta^{41}K_{ECF} - \varepsilon_{p/co}) \cdot F_{p/co}^{in} + (\delta^{41}K_{ECF} - \varepsilon_{ch}) \cdot F_{ch}^{in} = (\delta^{41}K_{ICF} - \varepsilon_{p/co}) \cdot F_{p/co}^{out} + (\delta^{41}K_{ICF} - \varepsilon_{p/co}) \cdot F_{ch}^{out} \tag{2}$$

393 where ε_{ch}, and $\varepsilon_{p/co}$ are the K isotope fractionation factors associated with channels and
394 pumps/co-transporters, respectively. Solving for the $\Delta\delta^{41}$K$_{plasma}$ value of the ICF (δ^{41}K$_{ICF}$ -
395 δ^{41}K$_{ECF}$) yields:

$$\delta^{41}K_{plasma}\ (‰) = \Delta f_{ch} \cdot (\varepsilon_{ch} - \varepsilon_{p/co}) \tag{3}$$

397 where $\Delta f_{ch} = f_{ch}^{out} - f_{ch}^{in}$, $f_{ch}^{in/out} = F_{ch}^{in/out}/(F_{ch}^{in/out} + F_{p/co}^{in/out})$. Assuming, based on our
398 results for *C. reinhardtii*, that $\varepsilon_{ch} \sim 1.2$‰ and speculating, based on analogues with
399 equilibrium K isotope effects associated with mineral precipitation in aqueous systems
400 (Li et al., 2017; Ramos et al., 2018; Zeng et al., 2019), that $\varepsilon_{p/co} \sim 0$‰, reduces eqn (1)
401 to:

402                    $$\delta^{41}K_{plasma}\ (\permil) = 1.2 \cdot \Delta f_{ch} \tag{4}$$

As $f_{ch}^{in/out}$ represent fractions of a flux, the value of Δf_{ch} is bounded at ±1; +1
representing a perfect 'pump-leak' mechanism, -1 representing a perfect 'leak-pump'
mechanism, and 0 consistent with either a 'pump-pump' or 'leak-leak' mechanism.
Accordingly, $\Delta\delta^{41}K_{plasma}$ values are bounded at ±1.2‰, with +1.2‰ representing a
perfect 'pump-leak' mechanism, -1.2‰ representing a perfect 'leak-pump' mechanism,
and 0‰ representing either a 'pump-pump' or 'leak-leak' mechanism. The observation
that the $\Delta\delta^{41}K_{plasma}$ values of all internal reservoirs are ≥ 0 suggests that $\Delta f_{ch} \geq 0$, a result
that is qualitatively consistent with the idea of elevated intercellular K^+ being maintained
by a 'pump-leak' mechanism. However, $\delta^{41}K_{plasma}$ values of the analyzed reservoirs also
exhibit a considerable range, from ~0‰ for many skeletal muscles to +0.55‰ for the
heart. In the context of eqn. (1), these differences can be explained by an increase in the
relative importance of K^+ loss via channels. Skeletal muscles appear to maintain K
homeostasis by something closer to a 'pump-pump' or 'leak-leak' mechanism whereas
red blood cells, the liver and the heart approach ~50% of a perfect 'pump-leak'
mechanism.

418   Two lines of evidence provide additional independent support for our
interpretation of $\delta^{41}K_{plasma}$ values as quantitative indicators of the relative importance of
K^+ loss from ICF via channels (eqn. (3)). First, measured $\delta^{41}K_{plasma}$ values are correlated
with the activity of NaKATP$_{ase}$ (umol P/h per g tissue) in tissues where this activity is
related to the maintenance of intercellular $[K^+]$. In particular, NaKATP$_{ase}$ activity
increases from 91±15 in skeletal muscles, to 253±8 in the liver, and 351±23 in the heart
(Gick et al., 1993). All else held constant, this increase in NaKATP$_{ase}$ activity (K^+ sources
to the ICF via pumps) will lower f_{ch}^{in} and increase Δf_{ch}, in agreement with observations.
Second, measured $\delta^{41}K_{plasma}$ values also correlate with intercellular $[K^+]$ (Figure 4c).
Because intercellular $[K^+]$ reflects the balance between K^+ the sources and sinks to the
ICF, one possible explanation for lower intercellular $[K^+]$ is a reduction in K^+ supply
and/or an increase in K^+ loss. As K^+ supply from NaKATP$_{ase}$ increases (see above), the
most likely alternative explanations are a decline in K^+ supply from inward rectifying K^+
channels and/or an increase in K^+ loss via outward rectifying K^+ channels. Both of these
processes increase Δf_{ch} and result in higher $\delta^{41}K_{plasma}$ values, again in agreement with
observations.

4.3.2.1 Urine. Unlike the internal K^+ reservoirs discussed above, all of which are
interpreted as independent homeostatic systems at isotopic steady-state with respect to
ECF (plasma), the loss of K^+ through the urine represents the end product of a series of
steps each of which may contribute to the observed net K isotope fractionation (e.g.
Figure 3b; $\varepsilon_{kidney} \sim 0.50\permil$). These are 1) glomerular filtration, 2) reabsorption along the
proximal tubule and the thick ascending limb of the loop of Henle, and 3) secretion and

reabsorption by principal and intercalated collecting duct cells. Although a detailed
description of the potential K isotope effects associated with each step is beyond the
scope of this manuscript, a brief description follows. Glomerular filtration is not expected
to fractionate K isotopes as the slit diaphragms freely filter ions and small molecules. A
large fraction (~85%) of the filtered K^+ is subsequently reabsorbed via both paracellular
and transcellular routes in the proximal tubule and thick ascending limb (TAL). The
residual K^+ is passed along to the collecting ducts where K^+ is added prior to excretion
as urine. The addition of K^+ in the collecting ducts occurs transcellularly, via ROMK
channels in principal cells and BK channels in intercalated cells (McDonough and Youn,
2017). Although we expect K^+ channels to be associated with a large K isotope effect
(ε_{ch} = 1.2‰), the extent to which this determines the $\delta^{41}K_{plasma}$ value of urine depends on
1) the $\delta^{41}K_{plasma}$ value of the residual K^+ exiting the TAL (after reabsorption but before
secretion) and 2) internal K^+ cycling within the principal and intercalated cells (e.g. eqn
(3)). The observation that ε_{kidney} is only ~ 0.50‰ suggests that either the residual K^+
exiting the TAL has an elevated $\delta^{41}K_{plasma}$ value due to preferential removal of ^{39}K during
reabsorption and/or the $\delta^{41}K_{plasma}$ value of ICF in principal and intercalated cells
elevated, for example due to high values of Δf_{ch}.

4.3.2.2 Cerebrospinal fluid (CSF). K^+ in CSF reflects a balance between paracellular
and transcellular K^+ transport across endothelial cells at the blood-brain-barrier (BBB)
and paracellular K^+ transport of across epithelial cells of the choroid plexus (Hladky and
Barrand, 2016). Gross fluxes of K^+ into the brain across the BBB are 4x larger than
those associated with the choroid plexus (Katzman et al., 1965), suggesting that the
observed net K isotope effects may be largely due to fractionation associated with
transport across the BBB. However, K^+ transport from ECF to CSF through the choroid
plexus is thought to occur by paracellular routes through pores in tight junctions (Figure
4), a process that we expect to fractionate K isotopes and thus may contribute to the
observed negative $\delta^{41}K_{plasma}$ values for CSF. With regards to K^+ transport across the
BBB, both paracellular (through tight-junction pores) and transcellular (through BBB
endothelial cells) routes may be important (Hladky and Barrand, 2016). Again, we expect
paracellular K^+ transport through tight junction pores to be associated with a larger K
isotope effect whereas any K isotope effects associated with transcellular transport will
depend on the internal cycling of K^+ (and associated isotope effects) within endothelial
BBB cells (pumps, (Betz et al., 1980); co-transporters, (Foroutan et al., 2005); and
channels, (Van Renterghem et al., 1995)). Overall, the observation of large K isotope
fractionation associated with the transport of K^+ from plasma to CSF (Fig. 5; ε_{CSF} =
0.59 ± 0.12‰) requires that either 1) there is a large K isotope effect associated with
transcellular K^+ transport across endothelial BBB cells or 2) transport of K^+ from plasma
to CSF is dominated by paracellular routes in both the choroid plexus and across the
BBB.

4.4 Stenohaline and Euryhaline Marine Fish. K$^+$ homeostasis in both euryhaline and
stenohaline marine fish is linked to ionic and osmotic regulation (Figure 5). While K$^+$
sources include ingestion of seawater and diet (Hickman Jr, 1968), by far the largest K$^+$
source is transport across the gills (Maetz, 1969), which are permeable to monovalent
cations (Na$^+$, K$^+$) and anions (Cl$^-$). Fish balance this salt intake by actively secreting Na$^+$,
K$^+$ and Cl$^-$ through mitochondria-rich cells (MRCs) of the gills and paracellularly through
tight-junction proteins (claudins) in the gill epithelium (Evans et al., 2005; Kolosov et al.,
2013). K$^+$ loss is not well-understood but recent discoveries of apical ROMK channels
indicate that transcellular secretion through MRCs is likely an important pathway
(Furukawa et al., 2012). Both stenohaline and euryhaline marine fish possess this
capability but euryhaline marine fish have evolved the ability to adapt this machinery to a
wide range of water salinities by adjusting expression of ROMK and tight junction
claudins (Furukawa et al., 2012; Furukawa et al., 2014; Furukawa et al., 2015; Kolosov
et al., 2013).

In seawater adapted stenohaline and euryhaline marine fish, K$^+$ homeostasis can
be approximated as a balance between gain of K$^+$ across tight-junctions in the gill
epithelium and loss of K$^+$ through apical ROMK channels in MRCs (Figure 5b). Other
potential sources and sinks of K$^+$ including the ingestion of seawater, diet, and excretion
are either small compared to the fluxes of K$^+$ across the gills (Maetz, 1969) or transient
in nature and unlikely to explain the systematic difference we observe between the
$\delta^{41}K_{seawater}$ values of stenohaline and euryhaline fish. Furthermore, as most of the total K$^+$
content of fish resides in muscle tissue, the $\delta^{41}K_{seawater}$ value of the muscle can be used
as a reasonable approximation of the $\delta^{41}K_{seawater}$ value of the whole organism (Figure
5b). At steady-state ($F_{gill}^{in} = F_{MRC}^{out}$), the $\delta^{41}K_{seawater}$ value of the whole organism will reflect
the balance between K isotope fractionation associated with K$^+$ sources (paracellular
transport across the gills, ε_{gill}) and K$^+$ sinks (transcellular transport through ROMK
channels in MRCs, ε_{ch}):

506
$$(\delta^{41}K_{seawater} - \varepsilon_{gill}) \cdot F_{gill}^{in} = (\delta^{41}K_{MRC} - \varepsilon_{ch}) \cdot F_{MRC}^{out} \tag{5}$$

507 which reduces to:

508
$$\delta^{41}K_{seawater}^{MRC} = (\varepsilon_{ch} - \varepsilon_{gill}) \tag{5a}$$

where $\delta^{41}K_{seawater}^{MRC}$ is the difference between the δ^{41}K value of the MRCs and
seawater. The δ^{41}K value of MRCs can be related to the δ^{41}K value of the whole fish by
combining equation (5a) with equation (3) and assuming 1) that the flux of K$^+$ out of
MRCs via apical ROMK channels is small compared to the gross fluxes of K$^+$ into/out of
MRCs associated with K homeostasis and 2) the δ^{41}K value of plasma is similar to

muscle (a reasonable assumption for most muscles in *R. norvegicus*, Figure 3b) . In this
case:

$$\delta^{41}K_{seawater}^{plasma/muscle} = (\varepsilon_{ch} - \varepsilon_{gill}) - \Delta f_{ch} \cdot (\varepsilon_{ch} - \varepsilon_{p/co}) \qquad (6)$$

where Δf_{ch} is the relative difference in K loss via channels in MRCs. Although we do not
independently know the value of ε_{gill} and suspect that it may vary systematically between
stenohaline and euryhaline marine fish, we speculate that ~0.2 to 1.2‰ represents a
plausible range based on results for epithelial cells in the gut of *R. norvegicus* (Figure
3b) and the maximum K isotope fractionation expected for channels (ε_{ch}). Values of ε_{ch}
and $\varepsilon_{p/co}$ are assumed to be 1.2‰, and 0‰, respectively. Given these values, inspection
of equation (6) indicates both positive and negative $\delta^{41}K_{seawater}^{plasma/muscle}$ are possible. For
example, as $\Delta f_{ch} \to 0$, $\delta^{41}K_{seawater}^{plasma/muscle} \to (\varepsilon_{ch} - \varepsilon_{gill})$ a result which yields $\delta^{41}K_{seawater}^{plasma/muscle} = 0$
to +1‰, in agreement with measured $\delta^{41}K_{seawater}^{plasma/muscle}$ values in euryhaline marine fish.
Alternatively, as $\Delta f_{ch} \to 1$, $\delta^{41}K_{seawater}^{plasma/muscle} \to (\varepsilon_{p/co} - \varepsilon_{gill})$ a result which yields
$\delta^{41}K_{seawater}^{plasma/muscle} = -0.2$ to -1.2‰, in agreement with measured $\delta^{41}K_{seawater}^{plasma/muscle}$ values in
stenohaline marine fish.

But is there any independent evidence that K homeostasis in MRCs in
stenohaline marine fish operates close to a perfect 'pump-leak' ($\Delta f_{ch} \sim 1$) mechanism
whereas the MRCs of euryhaline marine fish operate much closer to a 'leak-leak' or
'pump-pump' mechanism ($\Delta f_{ch} \sim 0$)? The gill epithelium of stenohaline marine fish is
considered to be more permeable than stenohaline freshwater fish due to the presence
of tight-junctions connecting gill accessory cells with MRCs (Chasiotis et al., 2012). This
increased permeability to NaCl must be balanced by increased NaCl secretion, a
process that has been shown to be linked to the activity of NaKATP$_{ase}$ in MRCs;
activities of NaKATP$_{ase}$ is higher in marine species than freshwater species and increase
when euryhaline species are acclimated to seawater (Epstein et al., 1967; Jampol and
Epstein, 1970). As an increase in the activity of NaKATP$_{ase}$ activity will, all else held
constant, lower f_{ch}^{in} and increase Δf_{ch}, one way of interpreting the apparent difference in
Δf_{ch} between MRCs of stenohaline and euryhaline marine fish is that the more
permeable gills of stenohaline marine fish require greater NaCl secretion and higher
activities of NaKATP$_{ase}$ in MRCs (e.g. Figure 3c).

5. Conclusions: The results presented here demonstrate that K+ homeostasis in
biological systems is associated with systematic variability in $^{41}K/^{39}K$ ratios and strongly
suggests that K+ transport through channels and tight-junction proteins is associated
with greater fractionation of K isotopes than transport via pumps and co-transporters.
Using results from 3 biological systems (algae, mammals, and fish) we developed a
simple framework for interpreting differences in measured $\delta^{41}K$ values between

intercellular fluid (ICF) and extracellular fluid (ECF) and between different internal fluid
reservoirs (e.g. blood plasma and cerebrospinal fluid). If correct, this framework permits
quantification of the relative importance of K+ channels and pump/cotransporters in
different physiological states – e.g. K depletion/excess - *in situ* (e.g $\delta^{41}K_{plasma}^{heart}$) and in
some cases *in vivo* (e.g $\delta^{41}K_{plasma}^{urine}$). However, with the exception of *C. Reinhardtii* where
the observed isotopic difference between media and the whole cell can be attributed to a
single transport mechanism, our results do not directly constrain the magnitude of the
individual K isotope effects associated with the machinery of K+ transport. Quantifying
machinery-specific K isotope effects through a combination of laboratory and numerical
approaches (14) is therefore a high-priority for future research.

Data Availability Statement. The data that support the findings of this study are available in the methods and/or supplementary material of this article.

Conflict of Interest Statement. The authors declare no conflict of interest.

Author Contributions JAH, AAM, JHY, DPS, and CS designed the research. JAH, DPS, DH, SK, SG, and CS carried out the experiments and associated analytical measurements. JAH, AAM, JHY, DPS, and CS interpreted the data and wrote the manuscript. All authors were involved in the revising of the manuscript.

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763 **Table S1.**

764 *C. Reinardtii*

Sample ID	$\delta^{41}K_{medium}$	2 S.E.	N	K (pg/cell)	Notes
TP Medium	0	0.1	3		Tris Phosphate (TP) growth medium
Wild Type 1	-1.14	0.06	3	0.258	wild type
Wild Type 2	-1.1	0.14	3	0.278	wild type
Cre50 mutant	-1.24	0.18	3	0.210	mitrochondrial Mg transport mutant

766 *Stenohaline and Euryhaline Marine Teleosts*

Sample ID	Type	Source	[K$^+$] mM	K/Na (g/g)	$\delta^{41}K_{seawater}$	2σ / 2 S.E.	N
Oncorhynchus kisutch_1	Euryhaline	Nassau Seafood, Princeton, NJ	85	2.50	0.27	0.17	1
Oncorhynchus kisutch_2	Euryhaline	Fulton Fish Market, NY, NY	84	1.87	0.63	0.11	2
Oncorhynchus kisutch_3	Euryhaline	Fulton Fish Market, NY, NY	87	2.02	0.93	0.17	1
Oncorhynchus tshawytscha_1	Euryhaline	Nassau Seafood, Princeton, NJ	79	3.76	-0.1	0.17	1
Oncorhynchus tshawytscha_2	Euryhaline	Fulton Fish Market, NY, NY	74	2.44	0.2	0.17	1
Oncorhynchus tshawytscha_3	Euryhaline	Nassau Seafood, Princeton, NJ			0.17	0.17	1
Oncorhynchus nerka_1	Euryhaline	Trader Joe's, Princeton, NJ	89	2.50	0.22	0.17	1
Oncorhynchus nerka_2	Euryhaline	Fulton Fish Market, NY, NY	99	1.87	0.25	0.17	1
Oncorhynchus nerka_3	Euryhaline	Fulton Fish Market, NY, NY	106	2.02	0.2	0.13	2
Oncorhynchus nerka_4	Euryhaline	Fulton Fish Market, NY, NY	111	3.76	0.12	0.17	1
Gadus morhua_1	Stenohaline	NOAA Cruise ID201504	92	2.50	-0.64	0.05	4
Gadus morhua_2	Stenohaline	NOAA Cruise ID201504	100	1.87	-0.47	0.05	3
Gadus morhua_3	Stenohaline	Fulton Fish Market, NY, NY	102	2.02	-0.54	0.06	2
Gadus morhua_4	Stenohaline	Fulton Fish Market, NY, NY	105	3.76	-0.68	0.04	2
Peprilus striacanthus_1	Stenohaline	NOAA Cruise ID201504	128	2.50	-0.72	0.04	2
Peprilus striacanthus_2	Stenohaline	NOAA Cruise ID201504	152	1.87	-0.48	0.16	2
Peprilus striacanthus_3	Stenohaline	NOAA Cruise ID201504	124	2.02	-0.7	0.17	1
Peprilus striacanthus_4	Stenohaline	NOAA Cruise ID201504	112	3.76	-0.33	0.17	1
Xiphias gladius_1	Stenohaline	Trader Joe's, Princeton, NJ	99	2.50	-0.58	0.17	1
Xiphias gladius_2	Stenohaline	Trader Joe's, Princeton, NJ	104	1.87	-0.32	0.04	2

Xiphias gladius_3	Stenohaline	Fulton Fish Market, NY, NY	109	2.02	-0.55	0.17	1
Xiphias gladius_4	Stenohaline	Fulton Fish Market, NY, NY	113	3.76	-0.74	0.17	1
Pseudopleuronectes americanus_1	Stenohaline	NOAA Cruise ID201504	139	2.50	-0.39	0.05	2
Pseudopleuronectes americanus_2	Stenohaline	NOAA Cruise ID201504	121	1.87	-0.68	0.01	2
Pseudopleuronectes americanus_3	Stenohaline	NOAA Cruise ID201504	137	2.02	-0.37	0.17	1
Pseudopleuronectes americanus_4	Stenohaline	NOAA Cruise ID201504	134	3.76	-0.47	0.17	1
Pseudopleuronectes americanus_5	Stenohaline	NOAA Cruise ID201504	124	2.44	-0.36	0.17	1
Hippoglossus stenolepis_1	Stenohaline	Fulton Fish Market, NY, NY	132	3.76	-0.62	0.17	1
Hippoglossus stenolepis_2	Stenohaline	Fulton Fish Market, NY, NY	116	2.44	-0.97	0.04	2
Hippoglossus stenolepis_3	Stenohaline	Nassau Seafood, Princeton, NJ			-1.02	0.17	1

768   *R. norvegicus (no brain)*

Sample	$\delta^{41}K_{diet}$	$\delta^{41}K_{plasma}$	2σ / 2 S.E.	N	$[K^+]$ mM	K/Na (mol/mol)
Diet$_{average}$	**0.00**	**-0.35**	**0.04**	**10**		
Urine$_1$	-0.08	-0.43	0.06	6	381	1.60
Urine$_2$	-0.06	-0.59	0.01	2	280	1.97
Urine$_3$	-0.12	-0.47	0.05	8	216	2.02
Urine$_{average}$	**-0.09**	**-0.50**	**0.10**	**3**	**293**	**1.80**
Feces$_1$	0.25	-0.10	0.05	2		
Feces$_2$	0.22	-0.31	0.06	2		
Feces$_3$	0.11	-0.24	0.17	1		
Feces$_{average}$	**0.19**	**-0.22**	**0.12**	**3**		
Plasma$_1$	0.35	0.00	0.06	3	3.95	0.03
Plasma$_2$	0.53	0.00	0.03	4	4.13	0.03
Plasma$_3$	0.35	0.00	0.04	6	6.77	0.05
Plasma$_{average}$	**0.41**	**0.00**	**0.12**	**3**	**4.95**	**0.04**
Red blood cells$_1$	0.81	0.46	0.07	2	102	7.23
Red blood cells$_2$	0.93	0.40	0.03	5	92	6.02
Red blood cells$_3$	0.68	0.33	0.08	3	95	6.46
Red blood cells$_{average}$	**0.81**	**0.40**	**0.08**	**3**	**96**	**6.56**
Heart$_1$	1.05	0.70	0.14	2	74	1.07
Heart$_2$	0.99	0.46	0.01	2	78	1.41
Heart$_3$	0.84	0.49	0.06	2	79	1.24
Heart$_{average}$	**0.96**	**0.55**	**0.15**	**3**	**77**	**1.23**
Liver$_1$	0.69	0.34	0.03	2	95	3.81
Liver$_2$	0.85	0.32	0.04	2	82	2.57
Liver$_3$	0.60	0.25	0.03	4	98	3.16
Liver$_{average}$	**0.71**	**0.30**	**0.05**	**3**	**92**	**3.13**
Soleus$_1$	0.49	0.14	0.01	2	104	3.10
Soleus$_2$	0.71	0.18	0.04	5	93	3.23
Soleus$_3$	0.54	0.19	0.03	4	101	3.20
Soleus$_{average}$	**0.58**	**0.17**	**0.03**	**3**	**100**	**3.17**

TA$_1$	0.34	-0.01	0.11	2	116	4.18
TA$_2$	0.64	0.11	0.02	2	120	3.82
TA$_3$	0.51	0.16	0.05	3	113	3.15
TA$_{average}$	**0.50**	**0.09**	**0.10**	**3**	**117**	**3.67**
EDL$_1$	0.53	0.18	0.10	3		
EDL$_2$	0.60	0.07	0.07	2	104	4.14
EDL$_3$	0.61	0.26	0.04	3	107	3.63
EDL$_{average}$	**0.58**	**0.17**	**0.11**	**3**	**106**	**3.86**
Gastrocnemius$_1$	0.48	0.13	0.03	2	120	5.07
Gastrocnemius$_2$	0.56	0.03	0.05	2	117	3.74
Gastrocnemius$_3$	0.22	-0.13	0.07	2	118	4.20
Gastroc$_{average}$	**0.42**	**0.01**	**0.15**	**3**	**118**	**4.28**
Stomach$_1$	0.54	0.19	0.10	2		
Stomach$_2$	0.52	-0.01	0.03	2		
Stomach$_3$	0.39	0.04	0.06	3		
Stomach$_{average}$	**0.48**	**0.07**	**0.12**	**3**		
Kidney$_1$	0.42	0.07	0.07	2		
Kidney$_2$	0.66	0.13	0.04	2		
Kidney$_3$	0.36	0.01	0.07	2		
Kidney$_{average}$	**0.48**	**0.07**	**0.07**	**3**		
Adipose tissue$_1$	0.44	0.09	0.10	2		
Adipose tissue$_2$	0.49	-0.04	0.07	2		
Adipose tissue$_3$	0.33	-0.02	0.17	1		
Adipose tissue$_{average}$	**0.42**	**0.01**	**0.08**	**3**		

770 *R. norvegicus (brain)*

Sample	$\delta^{41}K_{diet}$	$\delta^{41}K_{plasma}$	$\delta^{41}K_{CSF}$	2 S.E.	N
Plasma$_1$	0.36	0.00	0.54	0.05	4
Plasma$_2$	0.38	0.00	0.56	0.12	5
Plasma$_3$	0.59	0.00	0.77	0.14	3
Plasma$_4$	0.33	0.00	0.51	0.1	6
Plasma$_{average}$	**0.41**	**0.00**	**0.59**	**0.12**	**4**
Cerebrospinal fluid (CSF)$_1$	-0.13	-0.54	0.00	0.07	4
Cerebrospinal fluid (CSF)$_2$	-0.15	-0.56	0.00	0.18	4
Cerebrospinal fluid (CSF)$_3$	-0.36	-0.77	0.00	0.09	3
Cerebrospinal fluid (CSF)$_4$	-0.10	-0.51	0.00	0.15	4
Cerebrospinal fluid (CSF)$_{average}$	**-0.18**	**-0.59**	**0.00**	**0.12**	**4**
Cerebellum$_1$	0.06	-0.35	0.19	0.09	4
Cerebellum$_2$	-0.07	-0.48	0.09	0.15	4
Cerebellum$_3$	0.00	-0.41	0.36	0.04	7
Cerebellum$_4$	0.12	-0.29	0.21	0.12	5
Cerebellum$_{average}$	**0.03**	**-0.38**	**0.21**	**0.11**	**4**
Spinal cord$_1$	0.01	-0.40	0.14	0.07	6
Spinal cord$_2$	0.04	-0.37	0.20	0.06	6
Spinal cord$_3$	0.16	-0.25	0.52	0.1	6
Spinal cord$_4$	0.23	-0.18	0.33	0.09	7
Spinal cord$_{average}$	**0.11**	**-0.30**	**0.30**	**0.17**	**4**
Cerebrum$_1$	0.26	-0.15	0.39	0.07	6
Cerebrum$_2$	0.11	-0.30	0.26	0.05	6
Cerebrum$_3$	0.17	-0.24	0.53	0.09	6
Cerebrum$_4$	0.08	-0.33	0.18	0.09	3
Cerebrum$_{average}$	**0.16**	**-0.25**	**0.34**	**0.15**	**4**

771   **Figures.**

Figure 1. The difference in $\delta^{41}K$ values between external and intracellular K^+ in growth experiments of *C. reinhardtii* and a diagram of K^+ transport in single-celled algae grown under optimal conditions (See Methods) where K uptake occurs via channels that fractionate K isotopes (ε_{ch}) by ~1.2‰ . $P = 4.5\times10^{-5}$ by one-way ANOVA in Matlab.

Figure 2. Whole-body K isotope mass balance in *R. norvegicus* on a controlled diet. Measured $\delta^{41}K$ values of urine ($P = 0.19$) and feces ($P = 0.037$) relative to diet indicate a preferential uptake of ^{39}K in the gut. The total range in $\delta^{41}K_{diet}$ values in the tissues of *R. norvegicus* is ~1‰ with an estimated whole-body average $\delta^{41}K_{diet}$ value of approx. +0.5‰, consistent with preferential loss of ^{39}K in urine.

Figure 3. a) K isotopic composition of tissues/fluids *Rattus norvegicus* normalized to plasma ($\delta^{41}K_{plasma} = 0‰$). $\delta^{41}K_{plasma}$ values that are positive (●) indicate that net K⁺ transport out of the tissue/cell/fluid is enriched in ^{39}K relative to K⁺ transport into the tissue/cell/fluid. $\delta^{41}K_{plasma}$ values that are indistinguishable from 0 (◉) indicate that the K isotope fractionation associated with K⁺ transport into and out of the cell/tissue/fluid are either 0 or equal in magnitude (and thus cancel). $\delta^{41}K_{plasma}$ values that are negative (⬤) indicate that net K⁺ transport into the cell/tissue/fluid is enriched in ^{39}K relative to K⁺ transport out of the cell/tissue/fluid. **b)** A diagram of K isotope mass balance including transport by (1) K-channels, (2) K-pumps/co-transporters, (3) through pores in tight-junctions, and (4) glomerular filtration. Equation (3) solves for the $\delta^{41}K$ value of a tissue/cell relative to blood plasma assuming steady-state K isotope mass balance with bi-directional K⁺ transport by both channels and pumps/co-transporters. Variables include $f_{ch}^{out/in}$, the fraction of total K⁺ transport into/out of the tissue/cell that occurs through K⁺ channels, Δf_{ch}, the difference in the fraction of K transport via channels ($f_{ch}^{out} - f_{ch}^{in}$), ε_{ch}, K isotope fractionation associated with K channels, and $\varepsilon_{p/co}$, K isotope fractionation associated with pumps/co-transporters. K isotope fractionation associated with K⁺ uptake in the gut (ε_{gut}) and K⁺ loss in urine (ε_{kidney}) reflect the flux-weighted average of both paracellular and transcelluar K⁺ transport. For example, K isotope fractionation in the kidney will be associated with both re-absorption and secretion of K⁺ in the proximal tubule, the thick ascending limb (TAL), and cells in the collecting ducts of the kidney. **c)** Correlation between intercellular [K⁺] and $\Delta\delta^{41}K_{plasma}$ values for a subset of tissues from *R. norvegicus*. Also shown are the measured activities of NaKATPase (μmol P h⁻¹ g⁻¹) from (49) and values of Δf_{ch} from eqn (3).

Figure 4. K isotopic composition of cerebrospinal fluid from *R. norvegicus* raised on a controlled diet normalized to blood plasma ($\delta^{41}K_{plasma}$ = 0‰) and brain tissues normalized to cerebrospinal fluid ($\delta^{41}K_{CSF}$ = 0‰) together with a diagram of K homeostasis across the choroid plexus and blood brain barrier (BBB) from (50) including transport by (1) K-channels, (2) K-pumps/co-transporters, and (3) transcellularly through pores in tight-junctions. $\delta^{41}K_{CSF}$ values that are positive (●) indicate that net K transport out of brain tissues (spinal cord, cerebellum, and cerebrum) and into CSF is enriched in ^{39}K relative to K transport from CSF into these tissues. The negative $\delta^{41}K_{plasma}$ values (●) for CSF indicate that net K transport into CSF from blood plasma is enriched in ^{39}K relative to K transport from CSF to blood plasma and reflects the flux-weighted average of the tranporters that dominate K exchange across both the choroid plexus and BBB.

Figure 5. a) The difference in $\delta^{41}K$ values between seawater and dorsal white muscle of various stenohaline and euryhaline marine fish. Muscle [K⁺] and $\delta^{41}K_{seawater}$ values are listed in Table S1. $P = 1.4 \times 10^{-10}$ for the difference in $\delta^{41}K_{seawater}$ value between stenohaline and euryhaline teleosts. **b)** A model of of K isotope mass balance in a marine fish. K⁺ is supplied across the gill epithelium via transport through pores in tight-junction proteins and lost from mitochondria-rich cells (MRCs) through channels (equation 5). K⁺ cycling within MRCs is modeled using eqn. (3). Combining eqn. (3) and eqn. (5) yields eqn. (6), the $\delta^{41}K$ value of plasma/muscle relative to seawater as a funciton of ε_{ch}, ε_{gill}, Δf_{ch} of MRC's and $\varepsilon_{p/co}$. As $\Delta f_{ch} \to 0$, $\delta^{41}K$ of fish muscle $\to \varepsilon_{ch} - \varepsilon_{gill}$ ($\geq$ 0‰, euryhaline), whereas as $\Delta f_{ch} \to 1$, $\delta^{41}K$ of fish muscle $\to \varepsilon_{p/co} - \varepsilon_{gill}$ (< 0‰, stenohaline).

$$\delta^{41}K_{seawater}^{MRC} (‰) = \varepsilon_{ch} - \varepsilon_{gill} \qquad (5a)$$

$$\delta^{41}K_{plasma}^{MRC} (‰) = \Delta f_{ch} \cdot (\varepsilon_{ch} - \varepsilon_{p/co})$$

$$\delta^{41}K_{seawater}^{plasma/muscle} (‰) = \varepsilon_{ch} - \varepsilon_{gill} - \Delta f_{ch} \cdot (\varepsilon_{ch} - \varepsilon_{p/co}) \qquad (6)$$

$$\Delta f_{ch} \to 0$$

$$\delta^{41}K_{seawater}^{plasma/muscle} (‰) \to \varepsilon_{ch} - \varepsilon_{gill} = 0 \text{ to } +1‰$$

$$\Delta f_{ch} \to 1$$

$$\delta^{41}K_{seawater}^{plasma/muscle} (‰) \to \varepsilon_{p/co} - \varepsilon_{gill} = -0.2 \text{ to } -1.2‰$$