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On the evolution of the plant phytochrome chromophore

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running head: Phytochrome chromophore biosynthesis in streptophytes

Abstract

Phytochromes are biliprotein photoreceptors present in plants, algae, certain bacteria and fungi. Land plant phytochromes use phytochromobilin (PΦB) as the bilin chromophore. Phytochromes of streptophyte algae, the clade from which land plants evolved, employ phycocyanobilin (PCB), leading to a more blue-shifted absorption spectrum. Both chromophores are synthesized by ferredoxin-dependent bilin reductases (FDBRs) starting from biliverdin IXα (BV). In cyanobacteria and chlorophyta, BV is reduced to PCB by the FDBR phycocyanobilin:ferredoxin oxidoreductase (PcyA), whereas, in land plants, BV is reduced to PΦB by phytochromobilin synthase (HY2). However, phylogenetic studies proved the absence in streptophyte algae of any ortholog of PcyA and the presence of only PΦB biosynthesis related genes (*HY2*). The HY2 of the early diverging streptophyte alga *Klebsormidium nitens* (formerly *Klebsormidium flaccidum*) was already indirectly indicated to be involved in PCB biosynthesis. To provide the direct evidence of an intrinsic FDBR activity of this enzyme, we overexpressed and purified a His_6-tagged variant of *K. nitens* HY2 (KflaHY2) in *E. coli*. In anaerobic bilin reductase activity assays, we found that KflaHY2 is a functional FDBR yielding mostly 3(*Z*)-PCB as the reaction product. Via coupled phytochrome assembly assays, we identified 3(*Z*)-phytochromobilin and $18^1,18^2$-dihydrobiliverdin ($18^1,18^2$-DHBV) as intermediates of the reaction. With the help of site-directed mutagenesis two aspartate residues were identified to be critical for catalysis. Overall, our study gives insight into the evolution of the HY2 lineage of FDBRs.

Introduction

Photosynthesis is arguably one of the most significant biological processes on earth. To efficiently harvest light, photosynthetic organisms possess chlorophyll light-harvesting antennae incorporated in their photosystems. Moreover, some of these organisms are also capable to sense the quality and intensity of the harvested light. To do that, they employ several types of photoreceptors, which are able to absorb photons and to modulate biological activities based on this information (Möglich et al., 2010). Among these photoreceptors, phytochromes can be found in plants, algae, certain bacteria and fungi (Sharrock, 2008). Phytochromes are capable to perceive red and far-red light, by employing open-chain tetrapyrroles as light-sensing chromophores. Phytochrome's chromophore composition is dependent on the organism's habitat. Interestingly, even if phylogenetically closely related, land plants and streptophyte algae employ different bilins in their phytochromes, marking the impact of terrestrialization. Indeed, land plants use the dark blue pigment phytochromobilin (PΦB), whereas streptophyte algae employ the cyan pigment phycocyanobilin (PCB) (Wu et al., 1997). The different bound chromophore leads to a blue shift of the photocycles in the phytochromes of streptophyte algae compared to the ones of land plants. The light-sensing open-chain tetrapyrroles employed in the photoreceptors are all derived from heme. The first step in the biosynthesis of the bilins is the cleavage of the heme macrocycle at the α-methine bridge mediated by heme oxygenases (HOs). This reaction yields the open-chain tetrapyrrole biliverdin IXα (BV), liberating H_2O, CO and Fe^{2+} (Cornejo and Beale, 1988). BV is then subsequently reduced to the specific bilins by a class of enzymes called ferredoxin-dependent bilin reductases (FDBRs) (Figure S1) (Frankenberg et al., 2001; Frankenberg and Lagarias, 2003). In land plants, the enzyme phytochromobilin synthase (HY2) reduces BV in a two-electron reduction at the A-ring $2,3,3^1,3^2$-diene system to the dark blue pigment PΦB (**Figure 1**) (Kohchi et al., 2001; McDowell and Lagarias, 2001). In cyanobacteria and chlorophyta, phycocyanobilin:ferredoxin oxidoreductase (PcyA) catalyzes the four-electron reduction of BV to the cyan pigment PCB via the intermediate $18^1,18^2$-dihydrobiliverdin ($18^1,18^2$-DHBV) (**Figure 1**) (Frankenberg and Lagarias, 2003). Interestingly, the streptophyte alga *Klebsormidium nitens* is known to employ PCB in its phytochromes but only possesses a HY2 homolog (Rockwell and Lagarias, 2017). Using coexpression studies, it was suggested that the HY2 of *K. nitens* (named KflaHY2) is able to synthesize PCB and not PΦB, like its higher plant homolog (Rockwell et al., 2017). Here we provide direct biochemical evidence that the recombinant KflaHY2 is able to reduce BV to PCB. In addition, we investigated the reaction in more detail, identifying important catalytic residues. Overall, our data give more insight into the evolution of this interesting enzyme family and also on the evolution of land plants phytochromes.

Figure 1 Overview of the reactions catalyzed by the FDBRs HY2 and PcyA. Biliverdin IXα (BV) is the substrate of
both FDBRs. "P" indicates the propionate side chains. HY2 catalyzes the two-electron reduction of BV to
phytochromobilin (PΦB). PcyA catalyzes the four-electron reduction of BV to phycocyanobilin (PCB) via the
intermediate $18^1,18^2$-dihydrobiliverdin (DHBV). Sites of reduction are highlighted in grey.

Materials and Methods

Chemicals

All chemicals used in this study were ACS grade or better. The purification resin TALON® Superflow™ was purchased from Cytiva. Most of the components for the anaerobic bilin reductase activity tests were purchased from Sigma, except from biliverdin IXα, which was purchased from Frontier Scientific. Ferredoxin (PetF) and FNR (PetH) were recombinantly expressed and purified as described elsewhere (Dammeyer et al., 2008; Busch et al., 2011). Sep-Pak® C18 Plus Light cartridges were obtained from Waters. HPLC-grade acetone, formic acid and acetonitrile were purchased from VWR Chemicals.

Plasmids

The plasmid pET28a_KflaHY2 was gifted by Prof. J. Clark Lagarias (UC Davis, CA, USA). Site-directed mutants of KflaHY2 were generated in the pET28a_KflaHY2 plasmid using the QuikChange® Lightning kit (Agilent) employing primers listed in Table S1. The variant proteins were produced and purified according to the protocol used for KflaHY2 unless stated otherwise. The full list of employed constructs can be found in Supplemental Materials (Table S2).

Production and purification of recombinant proteins

For the production of recombinant KflaHY2 and variants, 2 L of LB medium containing the appropriate antibiotics was inoculated 1:100 with an overnight culture of *E. coli* BL21(DE3) carrying the respective plasmids. Cells were grown at 37°C and 100 rpm (Innova® 44, New Brunswick Scientific) to an OD_{600} of 0.4-0.6. The temperature was decreased to 17°C and gene expression was induced by the addition of isopropyl-β-D-thiogalactopyranoside (IPTG) to a final concentration of 1 mM (for pET28a_KflaHY2) or 0.5 mM (for pET28a_KflaHY2_N105D, pET28a_KflaHY2_D122N, pET28a_KflaHY2_D242N, pET28a_KflaHY2_N105D_D122N, pET28a_KflaHY2_N105D_D242N). The cultures were incubated under shaking for additional 19 h and harvested at 4°C by centrifugation for 10 min at 17.000 g (Sorvall™ LYNX™ 6000 centrifuge, Rotor T9). The pellets were resuspended in "binding buffer" (20 mM sodium phosphate pH 7.4; 500 mM NaCl). After the addition of a spatula tip of DNaseI (Applichem) and lysozyme (Sigma Aldrich), the suspension was kept on ice for 30 min. The cells were disrupted by sonication for 5 min (Bandelin Sonopuls HD 2200; tip KE76; 5'' pulses, 10'' pauses; cycle 6/10; ≈ 40% power output) and centrifuged for 45 min at 4°C and 50.000 g (Sorvall™ LYNX™ 6000 centrifuge, rotor T29-8). The crude extract was loaded onto a gravity flow column containing 2 mL of TALON® Superflow™ resin (Cytiva). The column was washed with 10 column volumes (CV) of binding buffer and the elution was performed using 4 CV of "elution buffer" (20 mM sodium phosphate pH 7.4; 500

mM NaCl; 500 mM imidazole). Protein containing fractions were pooled and dialyzed

overnight against TES-KCl buffer (25 mM TES/KOH pH 7.5; 100 mM KCl; 10% glycerol). The

concentration of the purified protein was determined based on its absorbance at 280 nm.

Extinction coefficients were calculated as described by Gill and von Hippel (Gill and von

Hippel, 1989).

GST-tagged PcyA and GST-tagged mHY2 were produced and purified as described

previously (Kohchi et al., 2001; Frankenberg and Lagarias, 2003). Apo-Cph1 and apo-BphP

were produced as described elsewhere (Tasler et al., 2005; Fiege and Frankenberg-Dinkel,

2020).

Anaerobic bilin reductase assay

The anaerobic bilin reductase activity assays were performed as described previously with

minor modifications (Ledermann et al., 2016). In this study, recombinant ferredoxin PetF from

the cyanophage P-SSM2 was used as the electron donor in a final concentration of 1 µM.

PetF was reduced using recombinant Ferredoxin-NADP$^+$ reductase (FNR) from

Synechococcus sp. PCC7002, PetH, at a final concentration of 0.01 µM (Busch et al., 2011).

The reaction was started by the addition of a NADPH regenerating system consisting of 65

mM glucose-6-phosphate, 8.2 µM NADP+ and 11 U/ml glucose-6-phosphate

dehydrogenase. For the assays employing defined electron equivalents, the NADPH

regenerating system was replaced with a NADPH solution of a specific concentration. The

photometric measurements were carried out using an Agilent 8453 spectrophotometer. The

reaction was stopped by the dilution 1:10 in ice cold 0.1% TFA. Products were isolated via

solid phase extraction using Sep-Pak® C18 Plus Light cartridges (Waters) and freeze-dried

using an Alpha 2-4 LSC plus lyophilizer (Martin Christ GmbH).

HPLC analyses

The products of the anaerobic bilin reductase assays were analyzed using an Agilent 1100

series chromatograph equipped with a Luna® 5µm reversed phase C18-column

(Phenomenex). The mobile phase consisted of 50% (v/v) acetone and 50% (v/v) 20 mM

formic acid at a flow rate of 0.6 mL/min. Reaction products were identified by comparison of

the retention time with known standards as well as whole spectrum analysis of the elution

peaks. Products of interest were collected directly at the outlet of the UV-Vis detector, frozen

at -80°C and freeze-dried.

Coupled phytochrome assembly assays

In vitro chromophore assembly of the phytochromes Cph1 and BphP with different bilins was

carried out using the lysate of a pET_*cph1* or a pASK_*bphP* overexpression after

centrifugation and filtration. The assembly was tested employing 50 µL lysate incubated

either with 40 µM standard bilin or with 4 µL of bilin solutions of unknown concentration for 30

min at room temperature, in the dark. Afterwards the volume was adjusted to 500 µL with

PBS. Absorbance spectra were recorded after incubation for 3 min with red light (636 nm –

Pr spectrum) and after incubation for 3 min with far red light (730 nm for Cph1; 750 nm for

BphP – Pfr spectrum). Difference spectra were calculated by the subtraction of the Pfr from

the Pr spectrum.

Phytofluor analyses

157 Phytofluor formation was evaluated via fluorescence measurements (Murphy and Lagarias,

158 1997). Briefly, 50 µL lysate of a pET_*cph1* or a pASK_*bphP* overexpression, after

159 centrifugation and filtration, was incubated with 4 µL of the KflaHY2_N105D reaction product,

160 for 30 min, at room temperature, in the dark. Afterwards the volume was adjusted to 500 µL

161 with PBS. Fluorescence spectra were recorded using a Jasco FP-8300 spectrofluorometer.

162 Results

163 KflaHY2 catalyzes the reduction of BV to PCB

It has been known for quite some time that the phytochromes of streptophyte algae use PCB

as their chromophore instead of the PΦB found in higher plants phytochrome (Wu et al.,

1997; Rockwell et al., 2017). Nevertheless, the streptophyte alga *K. nitens* possesses an

FDBR homolog to higher plant HY2. KflaHY2 has 39% amino acid sequence homology to the

HY2 of *Arabidopsis thaliana*, and experiments conducted by Rockwell *et al.* suggest that

PCB is the enzymatic product of its reaction. In order to biochemically investigate this, the

reductase was recombinantly overproduced in *E. coli* and purified by affinity chromatography.

The activity of the purified protein was tested in an anaerobic bilin reductase activity assay

(Figure 2).

Figure 2 Time-resolved UV-Vis spectra of an anaerobic bilin reductase activity test of recombinant KflaHY2
employing BV as the substrate. Total reaction time was 5 min and spectra were recorded every 30s. The course
of the spectra is marked by arrows. For reasons of clarity, only relevant spectra are shown. The dashed spectrum
was acquired without starting the reaction. The dotted-dashed spectrum represents the first acquired after starting
the reaction. The dotted curve represents the end spectrum.

Interestingly, the reductase and the substrate formed an intense turquoise colored complex

with an absorbance maximum at 622 nm, suggesting that the reductase was folded correctly

and was able to bind BV (Figure 2). The presence of a shoulder at ~ 750 nm indicates the

presence of BV in its protonated form (Frankenberg and Lagarias, 2003; Tu et al., 2004).

After the reaction was started, the absorbance decreased instantly. Moreover, a rapid

formation of a product with an absorbance maximum at 670 nm, which can be linked to the

formation of PCB, was observed. Subsequent HPLC analyses, employing the reaction

products of PcyA as standards, revealed that KflaHY2 forms 3(Z)-PCB as the main reaction

product (**Figure 3**).

Figure 3 HPLC analyses of the reaction products of KflaHY2 with BV as substrate. The products were analyzed
using a reversed-phase 5 µm C18 Luna column (Phenomenex) as stationary phase. The mobile phase consisted
of 50% (v/v) acetone and 50% (v/v) 20 mM formic acid flowing at 0.6 mL/min. Products of an anaerobic turnover
of BV to PCB mediated by PcyA served as standards. Absorbance was continuously measured at 650 nm.

KflaHY2 reaction can proceed via two intermediates

As the KflaHY2 reaction product PCB is a tetrahydro-BV, the reaction should proceed via a

two-electron reduced intermediate: either the A-ring reduced PΦB, or the D-ring reduced

$18^1,18^2$-DHBV. Intermediates of the reaction were isolated using the products of an

anaerobic bilin reductase activity assay conducted with one electron equivalent of NADPH.

HPLC analyses of the products showed that KflaHY2 forms two intermediates (**Figure 4** –

marked as "Inter 1" and "Inter 2").

Figure 4 HPLC analyses of the reaction products of an anaerobic activity assay employing equimolar amounts of KflaHY2 and BV with one electron equivalent of NADPH. Products were resolved using a reversed-phase 5 µm C18 Luna column (Phenomenex) as stationary phase and a mixture of 50% (v/v) acetone and 50% (v/v) 20 mM formic acid as mobile phase. The products of an anaerobic turnover of BV to PΦB mediated by HY2 derived from *A. thaliana* were used as standards. Absorbance was continuously measured at 650 nm.

The intermediates possessed the same retention times as the products 3(*E*)-PΦB (~ 19 min) and 3(*Z*)-PΦB (~ 27 min) of the recombinant reductase HY2 from *A. thaliana* (mHY2). However, $18^1,18^2$-DHBV, the intermediate of the cyanobacterial PcyA reaction towards PCB, and 3(*E*)-PΦB are roughly characterized by the same retention time. Therefore, "Inter 1" could have been either of the two (Frankenberg and Lagarias, 2003).

The intermediates formed by KflaHY2 are $18^1,18^2$-DHBV and 3(*Z*)-PΦB

To confirm the nature of the intermediates identified by HPLC analyses, coupled phytochrome assembly assays of the isolated intermediates with the apo-phytochromes Cph1 from *Synechocystis* sp. PCC 6803 and BphP from *Pseudomonas aeruginosa* were conducted (Figure 5A). Cph1 finds its natural chromophore in PCB, while BphP uses BV. However, phytochromes are also able to accommodate different bilins as long as they possess the same A-ring configuration as the natural chromophore. Thus, Cph1 is able to bind bilins possessing an ethylidene group at the A-ring, whereas BphP only binds bilins with an A-ring endovinyl group (Li and Lagarias, 1992; Tasler et al., 2005). Apo-Cph1 incubated with "Inter 1" did not produce a photoactive adduct since no characteristic difference

spectrum was obtained, indicating the lack of the ethylidene group at the A-ring C3 side

chain (Figure 5A – dashed line) (Xu et al., 2019). This led to the conclusion that "Inter 1"

could not be 3(E)-PΦB. Incubation of apo-Cph1 with "Inter 2" (Figure 5A – dotted line) led to

a difference spectrum like it was obtained for the incubation of Cph1 with 3(Z)-PΦB (Figure

5A – solid line). These results, as well as the results of the HPLC analyses (Figure 4), prove

that "Inter 2" is 3(Z)-PΦB.

Figure 5 Red-far red-light induced difference spectra of coupled phytochrome assembly assays employing the
intermediates of the KflaHY2 reaction. Absorbance spectra were recorded after incubation for 3 min with red light
(636 nm – Pr spectrum) and after incubation for 3 min with far red light (730 nm for Cph1; 750 nm for BphP – Pfr
spectrum). Difference spectra were calculated by the substraction of the Pfr from the Pr spectrum. All calculated
difference spectra were smoothed applying a 20 pt. Savitzky-Golay filter. A. Coupled phytochrome assembly
assays employing apo-Cph1 and "Inter 1" (dashed line); apo-Cph1 and "Inter 2" (dotted line); apo-Cph1 and 3(Z)-
PΦB (solid line) B. Coupled phytochrome assembly assay employing apo-BphP and "Inter 1" (dashed line); apo-
BphP and "Inter 2" (dotted line); apo-BphP and $18^1,18^2$-DHBV (solid line).

On the other hand, "Inter 2" failed to form a photoactive adduct with apo-BphP (Figure 5B –

dotted line), whereas, "Inter 1" formed a photoactive adduct with apo-BphP with a difference

spectrum (**Figure 5**5B – dashed line) similar to the difference spectrum obtained by the

incubation of BphP with $18^1,18^2$-DHBV (**Figure 5**5B – solid line). This revealed that "Inter 1"

is indeed $18^1,18^2$-DHBV.

To clarify whether the intermediates are only artifacts of the *in vitro* assay or if they are
actually suitable in KflaHY2-catalyzed BV reduction, both compounds were used as
substrates for KflaHY2 in anaerobic bilin reductase activity assays. Subsequent HPLC
analyses revealed that both intermediates are transformed to mostly 3(*Z*)-PCB (Figure 6).
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Figure 6 HPLC analyses of the reaction products of activity tests employing 3(*Z*)-PΦB and $18^1,18^2$-DHBV as
substrates for KflaHY2. Products were analyzed using a 5 μm reversed-phase C18 Luna column (Phenomenex)
as stationary phase. The mobile phase consisted of 50% (v/v) acetone and 50% (v/v) 20 mM formic acid flowing
at 0.6 mL/min. Reaction products of a PcyA mediated reduction of BV to PCB were used as standards.
Absorbance was continuously measured at 650 nm.

Asp^{122} and Asp^{242} are essential for the catalytic activity

In order to gain further insight on the unusual behavior of KflaHY2 and to learn about the
evolution of HY2, an amino acid sequence alignment employing different FDBRs was
constructed (Figure S2). Three residues caught our attention: Asn^{105}, Asp^{122} and Asp^{242}.
KflaHY2 Asn^{105} represents the homolog to *Synechocystis sp.* PCC 6803 PcyA His^{88}, which
was proven to be involved in protonation and substrate positioning in the active site
(Hagiwara et al., 2010). The substitution of the His to an Asn is not unusual since it can also
be found in other FDBRs like PebA, PebB and PebS. On the other hand, land plants HY2s
show an Asp in the corresponding position which was also proven to be essential for
substrate positioning in *Arabidopsis thaliana* HY2 and involved in catalysis in *Solanum*

lycopersicum HY2 (SlHY2) (Tu et al., 2008; Sugishima et al., 2020). KflaHY2 Asp^{122} finds a

homolog in all the FDBRs except for the ones from land plants. In PcyA and PebS, it is

located in the central β-sheet of the binding pocket and was shown to be the initial proton

donor (Hagiwara et al., 2006; Busch et al., 2011). Finally, Asp^{242} finds a homolog in Asp^{256} of

A. thaliana HY2, Asp^{256} of *S. lycopersicum* HY2 (SlHY2), Asp^{206} of PebS and Asp^{219} of

Guillardia theta PEBB. All of these Asp residues were shown to be involved in the reduction

of the A-ring 2,3,3^{1},3^{2}-diene system (Tu et al., 2008; Busch et al., 2011; Sommerkamp et al.,

2019; Sugishima et al., 2020). On the basis of these considerations, site-directed

mutagenesis was performed. To investigate the influence of ionization on catalysis, Asp^{122}

and Asp^{242} were replaced by asparagine residues whereas Asn^{105} was replaced by an

aspartate residue. Moreover, two double variants were generated. In particular, one was

constructed for KflaHY2 to resemble a plant HY2, the other to prove the presence of two Asp

residues is enough for reducing BV to PCB. The activity of the purified variant enzymes was

investigated *in vitro* via FDBR assays and subsequent HPLC analyses (Figure 7).

Figure 7 Activity assay and characterization of reaction products of recombinant KflaHY2 variants. A. UV-Vis
spectra of the anaerobic FDBR Assay performed using KflaHY2 variants and BV as the substrate. Total reaction
time was 20 minutes for KflaHY2_D122N, KflaHY2_D242N, KflaHY2_N105D_D122N and 60 min for
KflaHY2_N105D, KflaHY2_N105D_D242N. Spectra were recorded every 30 seconds. For reasons of clarity, only
relevant spectra are shown. The arrow indicates the course of absorbance during the reaction. B. HPLC elution
profiles of the reaction products of KflaHY2 variants. The stationary phase consisted in a Luna 5µm reversed
phase C18 column (Phenomenex). The mobile phase consisted of a mixture of 50% (v/v) acetone and 50% (v/v)
20mM formic acid flowing at 0.6 mL/min. Absorbance was continuously measured at 560 or 650 nm. BV shows
change in the retention time due to substitution of the mobile phase between the different HPLC measurements
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The D122N variant was able to bind BV and use it as a substrate, since the formation of a
peak at ~ 720 nm was observed (Figure 7A – D122N). Intermediates were isolated from the
reaction mix at certain time points and analyzed via HPLC. Time-course HPLC analyses
revealed the formation of small amounts of 3(*Z*)-PΦB after 5 min from the start of the
reaction (Figure 7B – D122N). This variant partially retains the capability to reduce BV A-ring
but is at all impaired in reducing the D-ring, thus proving the importance of D122 for D-ring
reduction.
The D242N variant still retained the ability to bind BV but failed in its reduction. The increase
of the absorption at ~ 460, ~ 740 and ~ 790 nm suggested the formation of the radical BVH•
(Figure 7A – D242N) (Tu et al., 2004; Tu et al., 2008; Busch et al., 2011). The HPLC elution
profiles were compatible with BV, therefore confirming the inactivity (Figure 7B – D242N).
This result is coherent with what was found for mHY2_D256N and SlHY2_D263N indicating
the importance of this residue for BV reduction (Tu et al., 2008; Sugishima et al., 2020).
The N105D variant showed an unexpected behavior. The assay was initially performed for
20 min but strangely revealed the formation of a violet-colored product with a peak ~ 570 nm,
suggesting production of either phycoerythrobilin (PEB) or 15,16-dihydrobiliverdin (15,16-
DHBV) (data not shown). In order to potentially accumulate more of this violet compound, we
decided to let the reaction run for a longer time, 60 min. This second assay showed the
formation of a final product with an absorbance peak at ~ 540 nm (Figure 7A – N105D).
Time-based HPLC analyses revealed that the bilin produced after 5 min from the start of the
reaction was indeed PCB, whereas a pattern compatible with the unidentified compound and
mostly 3(*Z*)-PCB was obtained for the 20 min product (data not shown). As for the final 60
min product, HPLC analysis revealed a double peak with the retention time of 10.5 and 10.8
min, not matching the retention time of any known bilin and therefore leading to uncertainty
(Figure 7B – N105D). To identify the unknown bilin, its integration in the phytochromes Cph1
and BphP was evaluated. PEB and 15,16-DHBV lack the C15-16 double bond required for
the photoconversion of the phytochrome between the two isoforms. Therefore, phytochromes
binding bilins lacking the C15-16 double bond are locked in the red-light absorbing form. In
this condition, there is no complete de-excitation of the bilin, leading to the formation of a
fluorescent product named phytofluor (Murphy and Lagarias, 1997). The bilin product
isolated from the reaction catalyzed by KflaHY2_N105D was then incubated in parallel with
the two apo-phytochromes and fluorescence was measured. The formation of a
characteristic phytofluor was only observed for the incubation with apo-Cph1, thus
suggesting the product is PEB (Figure 8).

Figure 8 Fluorescence spectra of the apo-Cph1-KflaHY2_N105D reaction product adduct. The isolated bilin
produced by KflaHY2_N105D was incubated for 30 min in the dark with the lysate of a pET_*cph1* overproduction.
Excitation spectra was obtained at 530 nm. Fluorescence emission was recorded after excitation at 560 nm.
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The double variant N105D_D122N, constructed to mimic a plant HY2, surprisingly retained

the native activity, revealing the conversion of BV to PCB instead of the hypothesized 2e⁻

reduction to PΦB (Figure 7 – N105D_D122N). This result led to the hypothesis that the

presence of two aspartate residues in the active site, even if in different position than in the

native enzyme, is enough for the reduction of BV to PCB. To further investigate it, the double

variant N105D_D242N was also generated. Interestingly, performing the assay for 60 min

revealed the formation of a pink product with an absorbance at ~ 600 nm (Figure 7A –

N105D_D242N). Via HPLC analysis, the final product was discovered to be 3(*E*)-PEB

(Figure 7B – N105D_D242N), thus showing a different activity than expected.

Discussion

The evolution of land plants finds its roots in streptophyte algae (de Vries and Archibald,

2018). Studies on this clade are particularly significant since they could enlighten the process

of colonization of land by embryophytes (Delwiche and Cooper, 2015). Streptophyte algae

can mostly be found in freshwater habitats but genera of the Chlorokybophyceaeae and

Klebsormidiophyceae can also thrive on land. Recent research suggests that freshwater

lifestyle paved the way for a pre-adaptation to the terrestrial environment (Becker and Marin,

2009). As photosynthetic organisms, both land plants and streptophyte algae endow

complexes to efficiently sense and harvest light. Among them, phytochromes are bilin-

photoreceptors which control growth and development (Hu et al., 2013). The impact of

terrestrialization is evident when considering the different phytochromes' bilin composition of

the two genera. Land plants use PΦB as the bilin chromophore whereas streptophyte algae

were shown to employ PCB (Wu et al., 1997; McDowell and Lagarias, 2001; Rockwell et al.,

2017). These bilins are characterized by different spectral properties: while PCB absorbs

orange and red light (λ_{PCB} = ~ 620 - 670 nm), PΦB absorbs far-red light ($\lambda_{PΦB}$ = ~ 660 - 685

nm). Both chromophores are synthesized by a class of enzymes named FDBRs. In land

plants, PΦB is synthesized by the FDBR HY2; in cyanobacteria and chlorophyta, PcyA

synthesizes PCB. In streptophyte algae, phylogenetic studies proved the existence of a HY2

homolog (Hori et al., 2014). However, previous research indirectly pointed out to a PcyA-like

activity of streptophyte algae HY2 (Rockwell et al., 2017). In this study, we found that the

HY2 of *Klebsormidium nitens* is indeed a functional FDBR, which, in contrast to the

phylogenetic classification as an HY2, catalyzes the reduction of BV to PCB. Furthermore,

we identified $18^1,18^2$-DHBV and 3(*Z*)-PΦB as intermediates occurring under the chosen

assay conditions. These results were somehow unexpected, as they lead to the conclusion

that KflaHY2 can mediate the reduction of BV to PCB employing two different reaction

mechanisms. One pathway implies the reduction of BV D-ring leading to $18^1,18^2$-DHBV,

which is then reduced at the A-ring to PCB. The other starts with the reduction of BV A-ring

ethylidene group to 3(*Z*)-PΦB. The D-ring exovinyl group of 3(*Z*)-PΦB is then reduced,

leading to PCB. The formation of 3(*Z*)-PΦB as an intermediate during PCB biosynthesis is

also tying well with previous phytochrome studies in the Zygnematophycea *Mesotaenium*

caldariorum (Wu et al., 1997). Additionally, according to evolution, one could argue the

preferred way is the latter, meaning that the HY2 of land plants lost the ability to perform the

last reduction step, to yield 3(*Z*)-PΦB as the final product. In this context, studies on even

closer relatives to embryophytes and early diverging land plants could provide the answer.

Using a site-directed mutagenesis approach, we also proved that the residues Asp[122] and

Asp[242] are of crucial importance. This conclusion fitted with the hypothesis made based on

sequence homology among FDBRs. To further corroborate the thesis, a model structure of

KflaHY2 was obtained using AlphaFold (Jumper et al., 2021). The obtained model structure

showed an overall similar folding of KflaHY2 to other FDBRs. Additionally, two structure

alignments were constructed using PyMOL: one with the crystal structure of the HY2 of

Solanum lycopersicum, with which KflaHY2 shares high sequence homology, the second

with the crystal structure of PcyA of *Synechocystis* sp. PCC 6803 (SynPcyA), catalyzing BV

to PCB reduction as KflaHY2 (Figure 9) (Hagiwara et al., 2006; DeLano, 2020; Sugishima et

al., 2020). The obtained alignments revealed a higher similarity of the KflaHY2 active site to

the one of SlHY2. As a matter of fact, the two bottom β-sheets residues completely

superimpose to each other, whereas only a slight shift in the upper aspartate residue is

appreciable (Figure 9A). On the other hand, the relative residue in SynPcyA, Asp[220], is facing

outside of the active site and was proven to be uninvolved in catalysis, whereas the A-ring

reducing residue was found to be Asp[105] (Figure 9B) (Hagiwara et al., 2006). However, if this

residue is conserved among KflaHY2 and SlHY2, the same is not true for the other. Indeed,

based on the biochemical data, we propose KflaHY2 Asp[122] to be involved in BV D-ring

reduction, but this corresponds to Asn[140] in SlHY2. The additional proton donor of SlHY2 was

found to be Asp[123], which, on the other hand, corresponds to an asparagine residue (Asn[105])

in KflaHY2. Contrastingly, in SynPcyA, the reduction of the D-ring exovinyl group was proven

to be catalyzed by the residue Glu[76], which does not find a homolog in KflaHY2 (Hagiwara et

al., 2006). The specificity of reduction is furthermore influenced by the substrate binding

mode of BV which was shown to be flipped in the PebB/HY2 lineage of FDBRs

(Sommerkamp et al., 2019; Sugishima et al., 2020). We propose this is also true for KflaHY2

as two critical amino acid residues responsible for the stabilization of BV in the ZZZssa

flipped binding mode are conserved. Overall, the data indicate that the ability of KflaHY2 to

synthesize PCB has evolved independently from the biosynthesis catalyzed by PcyA.

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Figure 9 Structures alignment of KflaHY2 with SlHY2 and SynPcyA. The model structure of KflaHY2 (displayed in
violet cartoon) was obtained using AlphaFold. The alignments were generated using PyMOL (DeLano, 2020). A.
Overlay of the active sites of KflaHY2 and SlHY2. The crystal structure of SlHY2 (PDB accession code: 6KME) is
displayed as gold cartoon. Biliverdin IXα, in the conformation assumed in the active site of SlHY2, is shown as
displayed in green with A-ring being labeled. Amino acid residues of interest are shown as labeled sticks and colored as the
belonging structure. B. Overlay of the active sites of KflaHY2 and SynPcyA. The crystal structure of SynPcyA
(PDB accession code: 2D1E) is displayed as pink cartoon. Biliverdin IXα, in the conformation assumed in the
active site of SlHY2, is shown in green with A-ring being labeled. Amino acid residues of interest are shown as
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Surprisingly, KflaHY2_N105D_D122N, generated for KflaHY2 to resemble a plant HY2, did

not catalyze the hypothesized 2e⁻ reduction of BV to PΦB. Product specificity could then be

determined by the architecture of the active site and other residues in the surroundings of the

binding pocket. Recently, studies on PcyA proved the replacement of two residues at the rim

of the substrate binding pocket determines the accumulation of the 2e⁻ reduced $18^1,18^2$-

DHBV as the final product (Miyake et al., 2022). A sequence alignment revealed KflaHY2

only possesses a homolog to one of these two PcyA residues. As a following step, the
possible involvement of this residue in accumulation of the intermediate in the KflaHY2
catalyzed reaction will be investigated. Furthermore, the unexpected outcome of the
KflaHY2_N105D_D122N reaction led to the initial assumption that the presence of two
aspartate residues in the active site, even if in different position than in the native enzyme, is
enough for the reduction of BV to PCB. This hypothesis was disproved by studying the
KflaHY2_N105D_D242N variant which, similarly to KflaHY2_N105D, revealed reductase-
isomerase activity leading to PEB production. Looking at the model structures of the variants,
the formation of either PCB or PEB seemed to be determined by the position of the two
aspartate residues in the binding pocket (Figure 10). Indeed, in KflaHY2_N105D and
KflaHY2_N105D_D242N only, where the two aspartate residues are laying on the same side
of the active site, reductase-isomerase activity could be observed. On the other hand, in
KflaHY2 and KflaHY2_N105D_D122N, the two aspartates are opposite to each other,
yielding PCB. Ultimately, X-ray crystallography will provide more insights on the catalytic
mechanism   of   KflaHY2.

Figure 10 Schematical representation of the active sites and activity of KflaHY2 and its variants. The red box highlights the position of the two aspartate residues. Exchanged residues are underlined. The production of either PCB or PEB is determined by the position of aspartate couple in the active site. Variants in which the aspartate residues are opposite to each other are catalyzing BV reduction to PCB. Variants in which the aspartates are laying on the same side show reductases-isomerase activity, first catalyzing BV reduction to PCB and then isomerizing it to PEB.

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Author contributions

FF, BL and NFD designed research, FF, BL, JH performed research, EPP and EH contributed analytical tools, FF, BL and NFD wrote the paper. All authors read and approved the final version of the manuscript.

Conflict of Interest

The authors declare no conflict of interest.

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555   Figure legends

557   **Figure 1** Overview of the reactions catalyzed by the FDBRs HY2 and PcyA. Biliverdin IXα (BV) is the substrate of
558   both FDBRs. "P" indicates the propionate side chains. HY2 catalyzes the two-electron reduction of BV to
559   phytochromobilin (PΦB). PcyA catalyzes the four-electron reduction of BV to phycocyanobilin (PCB) via the
560   intermediate $18^1,18^2$-dihydrobiliverdin (DHBV). Sites of reduction are highlighted in grey.

Figure 2 Time-resolved UV-Vis spectra of an anaerobic bilin reductase activity test of KflaHY2 employing BV as the substrate. Total reaction time was 5 min and spectra were recorded every 30s. The course of the spectra is marked by arrows. For reasons of clarity, only relevant spectra are shown. The dashed spectrum was acquired without starting the reaction. The dotted-dashed spectrum represents the first acquired after starting the reaction. The dotted curve represents the end spectrum.

Figure 3 HPLC analyses of the reaction products of KflaHY2 with BV as substrate. The products were analyzed using a reversed-phase 5 µm C18 Luna column (Phenomenex) as stationary phase. The mobile phase consisted of 50% (v/v) acetone and 50% (v/v) 20 mM formic acid flowing at 0.6 mL/min. Products of an anaerobic turnover of BV to PCB mediated by PcyA served as standards. Absorbance was continuously measured at 650 nm.

Figure 4 HPLC analyses of the reaction products of an anaerobic activity assay employing equimolar amounts of
KflaHY2 and BV with one electron equivalent of NADPH. Products were resolved using a reversed-phase 5 μm
C18 Luna column (Phenomenex) as stationary phase and a mixture of 50% (v/v) acetone and 50% (v/v) 20 mM
formic acid as mobile phase. The products of an anaerobic turnover of BV to PΦB mediated by HY2 derived from
A. thaliana were used as standards. Absorbance was continuously measured at 650 nm.

Figure 5 Red-far red-light induced difference spectra of coupled phytochrome assembly assays employing the
intermediates of the KflaHY2 reaction. Absorbance spectra were recorded after incubation for 3 min with red light
(636 nm – Pr spectrum) and after incubation for 3 min with far red light (730 nm for Cph1; 750 nm for BphP – Pfr
spectrum). Difference spectra were calculated by the subtraction of the Pfr from the Pr spectrum. All calculated
difference spectra were smoothed applying a 20 pt. Savitzky-Golay filter. A Coupled phytochrome assembly
assays employing apo-Cph1 and "Inter 1" (dashed line); apo-Cph1 and "Inter 2" (dotted line); apo-Cph1 and 3(Z)-
PΦB (solid line) B Coupled phytochrome assembly assay employing apo-BphP and "Inter 1" (dashed line); apo-
BphP and "Inter 2" (dotted line); apo-BphP and $18^1,18^2$-DHBV (solid line).

Figure 6 HPLC analyses of the reaction products of activity tests employing 3(Z)-PΦB and $18^1,18^2$-DHBV as
substrates for KflaHY2. Products were analyzed using a 5 μm reversed-phase C18 Luna column (Phenomenex)
as stationary phase. The mobile phase consisted of 50% (v/v) acetone and 50% (v/v) 20 mM formic acid flowing
at 0.6 mL/min. Reaction products of a PcyA mediated reduction of BV to PCB were used as standards.
Absorbance was continuously measured at 650 nm.

Figure 7 Activity assay and characterization of reaction products of KflaHY2 variants. A UV-Vis spectra of the anaerobic FDBR Assay performed using KflaHY2 variants and BV as the substrate. Total reaction time was 20 minutes for KflaHY2_D122N, KflaHY2_D242N, KflaHY2_N105D_D122N and 60 min for KflaHY2_N105D, KflaHY2_N105D_D242N. Spectra were recorded every 30 seconds. For reasons of clarity, only relevant spectra are shown. The arrow indicates the course of absorbance during the reaction. B HPLC elution profiles of the reaction products of KflaHY2 variants. The stationary phase consisted in a Luna 5μm reversed phase C18 column (Phenomenex). The mobile phase consisted of a mixture of 50% (v/v) acetone and 50% (v/v) 20mM formic acid flowing at 0.6 mL/min. Absorbance was continuously measured at 560 or 650 nm. BV shows change in the retention time due to substitution of the mobile phase between the different HPLC measurements (*).

Figure 8 Fluorescence spectra of the apo-Cph1-KflaHY2_N105D reaction product adduct. The isolated bilin
produced by KflaHY2_N105D was incubated for 30 min in the dark with the lysate of a pET_*cph1* overproduction.
Excitation spectra was obtained at 530 nm. Fluorescence emission was recorded after excitation at 560 nm.

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