Denitrification by bradyrhizobia under feast and famine and the role of the bc1 complex in

securing electrons for N$_2$O reduction

Yuan Gao, Magnus Øverlie Arntzen, Morten Kjos, Lars R. Bakken and Åsa Frostegård

Faculty of Chemistry, Biotechnology and Food Sciences, Norwegian University of Life Sciences,

N-1432 Ås, Norway

Abstract

Rhizobia living as microsymbionts inside nodules have stable access to carbon substrates, but

also have to survive as free-living bacteria in soil where they are starved for carbon and energy

most of the time. Many rhizobia can denitrify, thus switch to anaerobic respiration under low

O$_2$ tension using N-oxides as electron acceptors. The cellular machinery regulating this

transition is relatively well-known from studies under optimal laboratory conditions, while little

is known about this regulation in starved organisms. It is, for example, not known if the strong

preference for N$_2$O- over NO$_3^-$-reduction in bradyrhizobia is retained under carbon limitation.

Here we show that starved cultures of a *Bradyrhizobium* strain with respiration rates 1-18% of

well-fed cultures, reduced all available N$_2$O before touching provided NO$_3^-$. Proteomics showed

similar abundance of Nap (periplasmic NO$_3^-$ reductase) and NosZ (N$_2$O reductase), suggesting

that competition between electron pathways to Nap and NosZ favoured N$_2$O reduction also in

starved cells, similar to well-fed cultures. This contrasts the general notion that NosZ activity

diminishes under carbon limitation. The results suggest that bradyrhizobia carrying NosZ can

act as strong sinks for N$_2$O under natural conditions and that this criterion should be considered

in the development of biofertilizers.

Importance

Legume cropped farmlands account for substantial N_2O emissions globally. Legumes are

commonly inoculated with N_2-fixing bacteria, rhizobia, to improve crop yields. Rhizobia

belonging to *Bradyrhizobium*, the micro-symbionts of several economically important legumes,

are generally capable of denitrification but many lack genes encoding N_2O reductase and will be

N_2O sources. Bradyrhizobia with complete denitrification will instead act as sinks since N_2O-

reduction efficiently competes for electrons over nitrate reduction in these organisms. This

phenomenon has only been demonstrated under optimal conditions and it is not known how

carbon substrate limitation, which is the common situation in most soils, affects the

denitrification phenotype. Here we demonstrate that bradyrhizobia retain their strong

preference for N_2O under carbon starvation. The findings add basic knowledge about

mechanisms controlling denitrification and support the potential for developing novel methods

for greenhouse gas mitigation based on legume inoculants with the dual capacity to optimize

N_2-fixation and minimize N_2O emission.

Introduction

Bacteria in most natural and engineered environments are faced with fluctuating availability of

nutrients and need to adapt to a "feast and famine" lifestyle. While many soil types are rich in

total organic carbon, the concentration of bioavailable carbon substrate is low, particularly in

non-rhizosphere soil where lack of substrate is a major factor limiting the growth of

heterotrophic bacteria (1). It is likely that bacteria in soil are starved most of the time (2) and

only experience infrequent episodes of ample provision of carbon substrate, for example as

exudates from a root or organic material released during decay of dead (micro)organisms.

Bacteria have developed several strategies to survive extended periods of starvation, such as

the development of high-affinity uptake systems to scavenge alternative carbon sources from

the surroundings, as well as changes in cell morphology, condensation of the nucleoid and

decreased protein synthesis to adapt to a low metabolic activity (3). Many bacteria produce

carbon-rich storage materials such as PHA (poly-3-hydroxyalkanoates) and glycogen during

periods of substrate availability, which can be utilized to sustain a minimum of metabolic

activity when deprived of carbon and energy (4,5).

Several of the microbially mediated processes in the global nitrogen cycle are closely linked

to the carbon (C) metabolism of the organisms. One example is denitrification, which is the

reduction of NO_3^- to N_2 through anaerobic respiration where the N-oxides are used as terminal

electron acceptors when O_2 becomes scarce. This process can be performed by a diverse range

of heterotrophic bacteria, archaea and fungi, which use various forms of organic compounds as

electron donors to obtain energy (6) or, in some cases, H_2 (7). The last step of denitrification is

the reduction of N_2O, a strong climate gas, to harmless N_2, catalyzed by the N_2O reductase

(Nos) (8,9). It is found in a diverse range of prokaryotic organisms but has not been reported in

eukaryotes. Some denitrifying prokaryotes can perform all steps of denitrification, others only

some, and lack of the last step is common due to absence of the *nosZ* gene coding for NosZ , or

lack of other essential gene(s) in the *nos* operon (10,11), but the amounts of N_2O released from

denitrification in relation to N_2 (the N_2O/N_2 product ratio) is also influenced by transcriptional

and post-transcriptional control mechanisms and by various environmental factors (6,12-19).

Denitrification in agricultural soils is a major source of N_2O, accounting for more than 60% of

the global anthropogenic emissions (20,21). A steady increase in atmospheric N_2O has been

recorded since the start of industrialization, largely driven by increasing and excessive use of

synthetic fertilizers (22,23) and these emissions are predicted to continue to increase unless

novel mitigation options are developed (24,25).

Although the addition of reactive nitrogen compounds via synthetic fertilizers accounts for

the main part of the N_2O emissions from agricultural soil, the N_2O emitted from legume

cropped fields is far from negligible. A compilation of data from ca 70 legume cropped fields

estimated ca 1.29 kg N_2O-N ha^{-1} during one growing season, while the corresponding data for

N-fertilized crops and pastures showed emissions of 3.22 kg N_2O-N ha^{-1} (26). Legumes do not

have to rely on uptake of reactive N such as NH_4^+ or NO_3^- but can acquire N through their

symbiotic relationship with some groups of bacteria, collectively called rhizobia, that elicit the

production of root nodules on the plant in which they fix atmospheric N_2. In this process, the

rhizobia reduce N_2 to NH_3 and the plant cells reduce the NH_3 further to glutamine which they

use to produce amino acids and eventually proteins (27). When this N-rich plant material is

degraded, organic N is released and mineralized to NH_3/NH_4^+ which will be oxidized to NO_3^- by

nitrifying organisms. The O_2 consumption by the nitrifiers, together with the production of NO_3^-

and the availability of organic compounds from the degraded plants, creates conditions that are

conducive to denitrification. A novel approach to minimize N_2O emissions from agricultural soil

is to enhance the populations of N_2O reducing bacteria (25). In the case of legume crops, which

are often inoculated with rhizobia to optimize the N_2-fixation, there are a few promising studies

reporting decreased N_2O emissions from soybean fields inoculated with rhizobia with the dual

capability of efficient N_2 fixation and N_2O reduction (28,29).

Based on this, selection of rhizobial strains for development of commercial inoculants

should, ideally, take both these aspects into account. One problem is, however, that far from all

rhizobia carry the *nosZ* gene that encodes Nos. There are relatively few surveys of

denitrification genes in different groups of rhizobia. Complete denitrification, which includes all

four reduction steps of NO_3^- to N_2, has so far mainly been reported for the genus

Bradyrhizobium, which is the microsymbiont of a range of economically important legume

crops such as soybean, cow pea and peanut (30,31). A full set of denitrification reductases in

bradyrhizobia include, with few exceptions, the periplasmic NO_3^- reductase Nap; the Cu-

containing NO_2^- reductase NirK, the *bc*-type NO reductase cNor and a NosZ belonging to clade I

(19,32). We recently investigated the denitrification capacity of bradyrhizobia from two strain

collections, one obtained from nodules of legume trees and herbs growing in Ethiopia, the

other mainly consisting of strains isolated from nodules of peanut growing in China (18,19). In

these collections, 50 and 37% of the isolates, respectively, were complete denitrifiers, while the

others generally lacked NosZ and thus were potential N_2O sources. Common to all strains with

complete denitrification was a strong preference for N_2O- over NO_3^- reduction. Transcription

analysis and proteomics showed comparable expression levels of Nap and Nos, suggesting a

control mechanism at the metabolic level where the electron pathway to Nos, which receives

electrons from cytochrome *c* via the *bc1* complex, competes very efficiently with the electron

pathway to Nap which goes via the membrane bond NapC.

The results presented in (18,19) were based on experiments with organisms provided with ample amounts of C substrate (electron donor), which is likely to reflect the situation in legume nodules where the microsymbiont receives C from the plant. It can, however, be expected that rhizobia, which may survive for many years in soil (33), spend a large part of their life cycle as free-living organisms in soil where they will experience lack of available C substrate most of the time (2). Rhizobial inoculants that carry NosZ are thus potentially important sinks for N_2O produced both by themselves and by other soil microbes. It is, however, not known if the competition for electrons favoring N_2O reduction in well-fed cultures is retained during substrate limitation ("starvation"). Here we exposed cultures of *Bradyrhizobium* strain HAMBI 2125, also studied in (19), to shorter and extended periods of starvation and analyzed the denitrification kinetics, including the electron flow rates to the individual denitrification reductases. We also quantified the cellular abundancies of Nap, Nir and Nos. The results have practical implications, supporting that these organisms can act as sinks for N_2O under natural conditions. Moreover, the results are ecologically interesting since they show that cultures exposed to extended starvation divided into two opposite denitrification phenotypes, one with very slow metabolism, the other with retained metabolism, possibly reflecting a strategy to increase the chances for survival during periods of starvation.

Results

Denitrification kinetics in cultures prepared following Bioassay 1. Two bioassays were developed for starvation experiments, Bioassay 1 and 2 (Figs. 1A and B). In a first experiment, the denitrification gas kinetics, concentrations of NO_3^- and NO_2^- and electron flow rates to the

different reductases were compared for well-fed *vs* starved cultures (Figs. 2A-D). For this, the starved cultures were prepared following Bioassay 1 (Fig. 1A), in which the cultures were allowed to synthesize the denitrification reductases in the presence of ample amounts of substrate. Cultures were raised under fully oxic conditions in YMB medium, after which the headspace was made hypoxic to allow transition to anaerobic respiration in response to a gradual depletion of oxygen. After centrifugation/washing, pellets were pooled and inoculated into flasks containing YMB (9.9E+08 cells flask^{-1}) or buffer (5.0E+09 cells flask^{-1}), supplemented with 1 mM KNO_3 and 0.25 mM KNO_2, and with He plus 1 ml N_2O (around 80 µmol N) in headspace.

The initial O_2 concentration, which was 0.5 µM in cultures with YMB and 0.2 µM in cultures with buffer, was depleted within the first 5 h in both treatments (insets in Figs. 2A and B). The provided NO_2^- and N_2O were reduced simultaneously from the beginning of the incubation in both treatments. The NO_3^- was left untouched in the well-fed cultures until the exogenous N_2O was reduced (Fig. 2A), also seen from lack of electron flow to Nap except for a small peak in electron flow early in the anoxic incubation (Fig. 2C). No NO_3^- reduction took place in the starved cultures throughout the entire incubation period, as seen from the lack of electron flow to Nap (Fig. 2D). The negative V_{eNap} estimated for the starved cultures (inset in Fig. 2D) and the initial phase of the well-fed cultures are probably due to minor errors in calibration of N-gas measurements as well as in parameters used to calculate rates of N-transformation (sampling loss and N_2-leakage, see Molstad et al. (34)), which amounts to substantial errors in the estimates of NO_3^- reduction rates because they are based on N-mass balance (explained in detail by Lim et al. (35)). This implies a relatively high detection limit for NO_3^- reduction, and the

negative values cannot be taken as evidence for the complete absence of any electron flow to

NO_3^-. However, the measured NO concentrations lend some support to the claim that V_{eNap} was

~0 until depletion of the externally provided N_2O. In the well-fed culture, the concentration of

NO declined to zero in response to depletion of NO_2^-, and increased soon after, as V_{eNap}

increased (Figs. 2A and C). Likewise, the NO concentration declined to very low values in

response to depletion of NO_2^- in the starved culture (Fig. 2B).

Fig. 2C and D show the calculated electron flow rates per cell to the individual reductases,

and their sum, illustrating their competition for electrons. This shows that the well-fed cells

(Fig. 2C) sustained a nearly constant total electron flow rate around 9 fmol e^- $cell^{-1}$ h^{-1}

throughout, but allocated to different reductases depending on the availability of electron

acceptors: As long as NO_2^- and N_2O were both present, Nos captured around 50% of the

electrons ($V_{eNos} \sim V_{eNir} + V_{eNor}$), increasing to 100% when NO_2^- was depleted after 6-7 hours,

while the electron flow rate to Nap remained insignificant until the externally provided N_2O

became depleted. The starving cells (Fig. 2D) had an order of magnitude lower total electron

flow rate per cell, declining gradually throughout the incubation, and here Nos captured >> 50%

of the electrons during the first 10 h ($V_{eNos} >> V_{eNir} + V_{eNor}$), but declined to ~50% towards the

end.

Bioassay 1 was also used in a second experiment, but in this case the cultures were

incubated anoxically ([O_2] <0.21 ±0.04 µM) in buffer containing only one of the N-oxides (NO_3^-,

NO_2^- or N_2O) as initial electron acceptor (Figs. 3A-C). This experiment established that starved

cultures readily reduced NO_3^- from the start when no other N-oxide was provided (Fig. 3A). It

also demonstrated that the cell specific electron flow to N-oxides was practically unaffected by

the type of electron acceptor provided, except for the slightly lower rates initially for flasks with

NO_2^-. For all treatments, the cell specific electron flow rate decreased gradually throughout,

and the levels are very similar to those observed in the first experiment (Fig. 3D).

Cell size and PHA accumulation of starved vs well-fed cultures. The morphologies of single

bacterial cells from different treatments (starved/well-fed and anoxic/oxic) were analyzed by

phase contrast microscopy (Fig. S1A). By quantifying cell dimensions, we found marginal but

statistically significant effects (Mann-Whitney test, $p < 0.01$) of starvation: while there was no

significant difference in cell area of individual cells (Fig. S1B), starved cells were on average

~12% longer and ~5% thinner than well-fed cells (Fig. S1B). By further calculating cell volumes, a

slight reduction in average cell volume was observed upon starvation of aerobic cells ($p=0.057$),

but not for cells grown under anoxic conditions (Fig. S1C). Furthermore, a qualitative analysis of

the presence of PHA granules was done by staining the cells with Nile Red, a lipophilic dye with

high affinity for PHA (36). Large foci corresponding to PHA granules were seen in all conditions.

It should be noted that, as a lipophilic dye, Nile Red will also bind non-specifically to lipids and

membrane, and based on the imaging performed here, it could not be concluded whether there

were any significant differences in PHA content between the conditions.

Denitrification kinetics after providing starving cultures with an artificial electron donor.

An experiment was then performed in which TMPD plus ascorbate was added to starved

cultures (Bioassay 1), to provide cytochrome *c* with an excess of electrons (37,38). Since

cultures treated according to Bioassay 1 were able to provide electrons for denitrification (10-

18% of the electron flow in well-fed cells during the first 5 h, then decreasing to ca 4%; Fig. 2D),

we expected that the TMPD treatment would reduce or eliminate the oxidation of quinol by the

$bc1$ complex and that this would allow electrons to flow to Nap via NapC, resulting in NO_3^-

reduction. We also expected that loading cytochrome c with electrons would relieve or weaken

the competition for electrons between Nir, Nor and NosZ. The results (Fig. 4) lend little support

to the former since the electron flow to Nap remained insignificant until N_2O had been

depleted. But the results confirm an effect of TMPD on the competition between Nos and Nir:

V_{eNir}, V_{eNor} and V_{eNos} were similar and high (4-5 fmol e$^-$ cell^{-1} h^{-1}) until NO_2^- was depleted. Thus,

Nir and Nos competed equally well when the cytochrome c pool was fully reduced by TMPD.

After depletion of NO_2^-, V_{eNos} leaped to its maximum level, 13-14 fmol N cell^{-1} h^{-1}, and kept this

rate until all the exogenous N_2O was depleted.

Denitrification kinetics and reductase abundancies of cultures exposed to extended

starvation following Bioassay 2. While Bioassay 1 successfully induced starvation in terms of a

downshift in respiration, the rates did not reach a stable level during the assay but declined

gradually throughout, apparently approaching more stable low levels after 20 h. On this

background, we introduced a more severe starvation assay (Bioassay 2, Fig. 1B) by including a

20 h aerobic incubation in buffer prior to the starvation-denitrification assay, in order to reach a

lower and more stable rate of respiration than in the first experiment. In addition, Bioassay 2

was designed to force the cells to synthesize the denitrification proteome while starving. In

several preliminary experiments with Bioassay 2 (results not shown), we found a conspicuous

variability between flasks regarding the cell specific respiration rate where one or two out of

three replicate flasks had four to six times higher cell respiration rates than the other(s). At first,

we suspected that it could be due to impurities of flasks or magnets, but meticulous acid

washing failed to remove the stochastic variation. Being convinced that the stochasticity

reflects a real regulatory switch of the cultures, we performed a final experiment in which 15

replicate flasks were monitored for denitrification kinetics (Fig. 5).

The cultures received initially 1 mM NO_3^- in the buffer (but no NO_2^-), and He plus 1 ml N_2O

(80 μmol N) in headspace. The O_2 concentration at the time of inoculation was <0.5 μM in the

liquid. The flasks separated into two distinct denitrification phenotypes (Figs. 5A and B). Nine

flasks showed "low" cell specific respiration rates (total electron flow maximum 0.27 fmol e^-

$cell^{-1}$ h^{-1}), corresponding to approximately 2.7% compared to well-fed cultures (Fig. 5A), while

the other six showed a "fast" respiration rate (total electron flow 1.0-1.8 fmol e^- $cell^{-1}$ h^{-1}). Both

phenotypes reduced N_2O from the beginning of the incubation, showing a strong preference for

N_2O over NO_3^-. In the flasks with fast respiration, the cells started to reduce NO_3^- in response to

depletion of the externally provided N_2O. In the flasks with slow respiration, N_2O was not

depleted within the time frame of the experiment, and NO_3^- reduction remained negligible.

To investigate if cell lysis, and thus release of available C, could explain why some cultures

showed the "fast" growth phenotype, we compared the OD_{600} and the number of viable cells in

"slow" *vs* "fast" cultures. The samples were taken after incubation for 3.1 h in anoxic buffer,

when the two phenotypes were clearly distinct but when some possible growth of cells in the

"fast" cultures would not yet hide if lysis had occurred. The OD_{600} spanned from 0.12-0.13 with

no statistical difference (p>0.3) between cultures with "fast" and "slow" respiration. Similarly,

no difference (p=0.4) was found for the viable counts which showed 1.21E+10 to 1.33E+10 CFU

flask^{-1} for the "fast" cultures and 1.23E+10 to 1.34E+10 CFU flask^{-1} for the "slow" cultures (Fig.

S2).

We also tested the metabolic integrity of the cells in the flasks with slow respiration by

injecting C-substrates (YMB) to the flasks after 24.8 hours, which proved their capacity to

quickly regain activity approaching that of well-fed cells (Fig. 5A). To investigate if the observed

divergencies in phenotypes were due to differences in denitrification reductase abundancies,

we quantified the relative abundances of Nap, NirK and NosZ in samples taken at different time

points throughout the incubations (Figs. 5E and F), together with the corresponding

denitrification kinetics (Figs. 5A and B) and cell specific electron flow to reductases (Figs. 5C and

D). The membrane-bound NO reductase (cNor) could not be extracted quantitatively (the

results showed 1000 times lower abundancies than for the other denitrification reductases) and

is therefore not shown. The inoculum had been cultured aerobically for 3-4 generations, never

permitting the OD$_{600}$ to exceed 0.1, to secure that any denitrification enzymes would be diluted

to extinction by aerobic growth, assuming that the transcription of all genes is effectively

repressed by oxygen. This strategy was apparently successful since Nap and Nir were

undetectable at the start of the anoxic incubation. NosZ, on the other hand, was detected also

in the aerobic cultures, suggesting that the *nosZ* gene is constitutively transcribed at low levels

in these organisms.

After transferring the cells to anoxic buffer, the abundance of all three reductases increased

both in cultures with "slow" and "fast" respiration. Cultures with "slow" respiration rate

synthesized less denitrification reductases than those with "fast" respiration rate during the

first 20 h. The relative abundances of the different reductases also differed between the two

phenotypes. In cultures with "slow" respiration, NosZ was significantly more abundant than

Nap and Nir ($p<0.01$), which were comparable (Fig. 5E). Cultures with "fast" respiration instead

contained higher abundancies of NosZ and Nir compared to Nap at 5 h. After this, the

abundance of Nir increased more than that of the others and at 20 h the LFQ of Nir was 0.20 ±

0.07, while the abundancies of NosZ and Nap were approximately half of that (Fig. 5F). After

addition of carbon substrate to the cultures with "slow" respiration, a rapid synthesis of all

three reductases took place. This synthesis was most prominent for Nir which, after a couple of

hours, had increased three-fold, reaching a relative abundance that was twice as high as Nap

and NosZ (Fig. 5E), resembling the abundance profile of the cultures with "fast" respiration (Fig.

5F).

Discussion

Detailed eco-physiological studies during the past decades have revealed several aspects of

how the denitrification process is regulated in different organisms (6,13,39). Most of this

knowledge is, however, based on laboratory studies where cultures have been grown under

optimal conditions, while less attention has been paid to understanding how denitrification,

and particularly the release of N_2O, is affected if cells are starved, which is the normal state of

cells in most natural environments. We focused on how starvation, i. e. lack of carbon substrate,

affects denitrification and N_2O release. The general notion has been that N_2O reductase is less

competitive for electrons than the other denitrification reductases, leading to emissions of N_2O

when the availability of electron donors is low. This is largely based on a single study of

Alcaligenes faecalis (40) and has been supported by some studies of complex communities but

contested by others (41-43). *A. faecalis* can perform partial denitrification reducing NO_2^- to N_2

using NirK, cNor and NosZ clade I, but this organism lacks dissimilatory NO_3^- reductases (Nar or

Nap). The study by Schalk-Otto (40), performed in continuous cultures, showed increased N_2O

release under low substrate concentrations, and it was suggested that N_2O reductase did not

compete successfully with the other reductases for electrons from cytochrome *c*, possibly due

to lower affinity for the electron donor. This conclusion needs, however, further verification by

more detailed studies of the mechanism involved. Moreover, such studies need to be extended

to other groups of denitrifying microorganisms and should include organisms carrying a

complete denitrification pathway (thus with Nar and/or Nap). Research over the past decades

has revealed diverse denitrification phenotypes among even closely related bacteria, with

implications for their accumulation of denitrification intermediate products (15,44).

The phenotype described for a range of taxonomically diverse *Bradyrhizobium* strains with a

complete denitrification pathway is characterized by a strong preference for N_2O over NO_3^-

when grown in full-strength YMB medium under denitrifying conditions (18,19). The present

study provides compelling evidence that this preference is retained when the organisms are

starved for carbon and energy, with practically no electron flow to Nap when N_2O was present

(Figs. 2A-D). This was not due to too low levels of Nap since the cultures showed well-

functioning Nap activity if NO_3^- was provided as the only initial electron acceptor (Fig. 3A).

Further evidence for this was obtained from the proteomics results, which showed that the

cultures, even when grown under more severe starvation conditions using Bioassay 2, were

able to produce comparable amounts of Nap, Nir and Nos (Figs. 5E and F). Therefore, the lack of

NO_3^- reduction during the period when there was N_2O in the system could not be explained by a

lack of Nap molecules. Instead, the results suggest the same metabolic-level phenomenon as

found for well-fed cultures in this study (Fig. 2A) and earlier (18,19), where NosZ outcompetes

Nap for electrons, leaving Nap virtually without electrons so long as exogenous N_2O is available.

The attempt to tweak the electron flow toward Nap by the addition of TMPD and ascorbate

did not result in measurable NO_3^- reduction in the starved cultures (Fig. 4A). A recent study by

Mania et al. (18) demonstrated that well-fed Bradyrhizobium cells did reduce some NO_3^- in the

presence of N_2O, if provided with TMPD and ascorbate. This suggested that NosZ (as well as Nir

and Nor) received electrons from strongly reduced cytochrome c, thus bypassing the electron

flow via the bc1 complex, which allowed Nap to receive electrons from quinol, delivered from

the TCA cycle. The specificity of the electron delivery from TMPD to cytochrome c cannot be

taken for granted, however, and the result by Mania et al. (18) could instead reflect a minimum

of electron flow from TPMD to quinon or to NapC, directly, or indirectly. In the present

experiment with starved cells, TMPD + ascorbate failed to induce measurable electron flow to

Nap in the presence of N_2O (Fig. 4B). This probably reflects the marginal electron flow from the

TCA cycle to the quinon/quinol pool due to starvation. A separate experiment supported this,

showing that when carbon substrate (YMB) was added to starved cultures, this provided

enough electrons to support some Nap activity, although NosZ activity still dominated (Fig. S3).

At the same time, the result refutes the concerns regarding unspecific electron delivery of

electrons from TMPD to quinon. Functional Nap was apparently present, and NO_3^- reduction

started when N_2O was almost depleted, with V_{eNap} being 2 fmol e$^-$ cell^{-1} h^{-1} which was twice as

high as $V_{eNir/Nor}$ (Fig. 4B). The latter is as expected, since the reduction of 1 mole of NO_3^- to NO_2^-

requires 2 mole electrons, while reduction of NO_2^- and NO requires 1 mole electrons.

Furthermore, the electron flow to NosZ was the same as to Nir and Nor, suggesting that Nir and

NosZ competed equally well for electrons from cytochrome c. The maximum total electron flow

of 13-14 fmol e$^-$ cell^{-1} h^{-1} when the cytochrome c pool was saturated with electrons from TMPD

is likely to be close to the maximum capacity of the denitrification pathway. This electron flow

rate is higher than the total electron flow rate in the well-fed cells, which was 8-10 fmol e$^-$ cell^{-1}

h^{-1} (Fig. 2C). The factor limiting V_{eNos} to 13-14 fmol cell^{-1} h^{-1} is plausibly the rate of electron

delivery from cytochrome c and/or k_{cat} for Nos.

In another experiment we added YMB medium to "slow respiration" phenotypes of the

cultures that had been exposed to extended starvation (Figs. 5A, C and E). These cultures had

some NosZ activity (V_{eNos} 0.1-0.3 fmol e$^-$ cell^{-1} h^{-1}), while Nap and Nir activities could not be

detected. Addition of the electron donor (YMB) led to an immediate upshoot in the activities of

all reductases. This could not have been the case if the reductases were not present already,

which was proven by the proteomics results. Taken together, the results support that the

absolute preference for N_2O over NO_3^- was due to competition for electrons, also under severe

starvation conditions.

It is well known that bacteria have developed a range of physiological responses to tackle

starvation. Some survive by forming spores, but most bacteria survive by strongly reduced

metabolic rates and minimizing the synthesis of some proteins while upregulating others such

as genes for high-affinity transporters, essential repair mechanisms and alternative energy

sources (3,45,46). Some such changes have been observed in rhizobia belonging to *Rhizobium*

leguminosarum, which stayed viable for long periods (55 days) of C starvation (47). Changes in

cell size are common during long-term starvation, sometimes leading to the formation of small

or even ultra-microcells which may have increased tolerance to antibiotics and other stresses

(2,46). Another way for many bacteria, including rhizobia, to survive is to use carbon stored as

polyhydroxyalkanoates (PHA) or glycogen, formed during periods of ample nutrient abundance

(48). In the present study, the metabolic activity, measured as anaerobic respiration rate,

decreased to between 1 and 18% that of well-fed cultures, depending on which starvation

bioassay was used. We did not, however, detect any obvious decrease in cell size when

comparing cells starved for 24 h, although cell morphologies were slightly altered (Figs. S1A and

B). PHA was observed in the starved cells as well as well-fed ones, as shown with Nile red

staining (Fig. S1A) and no obvious reduction in PHA could be observed during starvation from

our assays. Since this was a comparably short period of starvation, it is conceivable that the

cells would make use of the stored carbon if the starvation was prolonged. It could be

speculated that one reason for not using the PHA reserves (and glycogen) immediately, is that

these are saved to be used during bacteroid formation (48). To clarify these issues, more

detailed studies are however needed.

A striking separation into two phenotypes during starvation, as shown in Fig. 5, was

observed in repeated experiments (Bioassay 2), each time with about two thirds of the flasks

showing a "fast" and the others "slow" respiration: 1.0-1.8 fmol e$^-$ cell^{-1} h^{-1} vs 0.1-0.3 fmol e$^-$

cell^{-1} h^{-1}, respectively. Phenotypic heterogeneity has been observed in single-strain cultures of

various bacteria when exposed to carbon substrate deficiency and may, or may not, be due to

mutations during starvation for one week or more (3,49). Mutations in the entire population in

several replicate flasks are unlikely and cannot, however, have caused the rapid diversification

into "slow"- or "fast" respiration in the present study. It may instead reflect a stochastic

phenomenon, or that the culture contained different subpopulations. We speculated that the

"fast respiration" phenotype may be due to a fraction of the cells dying, allowing the other cells

to survive on nutrients released from lysed cells, as seen for other bacteria (3). However, this

would require lysis of a substantial fraction of the cells, which is refuted by the observation that

the flasks with "slow" and "fast" phenotypes had practically identical numbers of viable cells

(Fig. S2). Thus, further studies are needed to understand this phenomenon of different

respiration rates.

Although the starvation bioassays developed for this study cannot be regarded as a close

mimicking of the conditions in natural environments, it is conceivable from the experiments

that these organisms are potentially strong sinks for N_2O when living in soil under fluctuating

availability of carbon substrate. Bioassay 1 is probably closer to a "real-world" situation than

Bioassay 2, since it is likely that denitrifying bacteria experience regular fluctuations in oxygen

and thus are not devoid of denitrification reductases if they enter starvation. On the other hand,

Bioassay 2, where the cells had to produce the denitrification proteome in the absence of

external electron donors (C-substrate), showed that even under these conditions N_2O reduction

strongly dominated over NO_3^- reduction.

Materials and methods

Bacterial strain and culture preparations. *Bradyrhizobium* strain HAMBI 2125, originally

isolated from nodules of *Arachis hypogaea* growing in Sichuan, China (50), was used in all

experiments. This strain, which is closely related to *Bradyrhizobium ottawaense*, contains the

genetic set-up for complete denitrification (19). A culture was raised from one single colony

after streaking on agar plates. After checking the purity by sequencing the 16S rRNA gene (19),

portions were preserved in 15% glycerol at -80 °C. Cultures for all the experiments were started

from the -80 °C stocks and raised under fully oxic conditions in 120 ml serum flasks containing

50 ml Yeast Mannitol Broth (YMB): 10 g l^{-1} D-Mannitol, 0.5 g l^{-1} K_2HPO_4, 0.2 g l^{-1} $MgSO_4 \cdot 7H_2O$,

0.1 g l^{-1} NaCl and 0.5 g l^{-1} yeast extract (51). All incubations were done at 28 °C. Medical flasks

(120 ml) were used in all experiments. A magnet in each flask secured vigorous stirring (600-

700 rpm) to avoid cell aggregation and to optimize the gas exchange between liquid and gas

phases (18). To prevent that the cells experienced anoxia during this oxic incubation, and thus

to avoid de novo synthesis of denitrification reductases, portions were regularly transferred to

new flasks containing fresh medium, so that the OD_{600} was never allowed to exceed 0.1 (19).

These aerobically grown cultures were used as inoculants in the "starvation bioassays"

described below.

Flasks for cultures incubated under hypoxic conditions were prepared as described in (18).

Briefly, flasks (120 ml) containing 50 ml buffer (or medium) were capped with sterilized, gas

tight butyl rubber septa (Matrix AS, Norway). The air was removed by applying vacuum

repeatedly (6 x 360 s) and He was then filled for 30 s, after which the overpressure was

released. The flasks were left for two days to equilibrate the gases between the headspace and

liquid (52). Then, 0.7 ml or 1 ml O_2 (equal to 1 or 1.5 vol %) and 1 ml N_2O (70 to ~80 µmol N

$flask^{-1}$) was injected into the headspace and sterile filtered solutions of KNO_3 (and sometimes

KNO_2) were added to the liquid reaching initial, desired concentrations (0.25 -1 mM).

Starvation bioassays. Starvation bioassays were established, which followed one of the

procedures described in Fig. 1. In Bioassay 1 "Mild starvation", cultures were allowed to make

the transition to denitrification in full-strength YMB before being exposed to starvation. Oxically

grown, well-fed cultures were incubated for 48 h after which the headspace was replaced by He

and 1% O_2, and 1 mM KNO_3 was added to the medium. When O_2 was depleted, the cultures

were centrifuged (10 000 × g at 4 °C for 10 min) and washed twice in sterile ddH_2O. The pellets

(triplicates) were pooled to reduce bias in the form of variations due to the

centrifugation/washing. Each pellet was divided into three and used to inoculate anoxic flasks

containing 50 ml C-free buffer supplemented with 1 mM KNO_3 and 0.25 or 0.5 mM KNO_2 and

with He and/or 1 ml N_2O (around 80 µmol N flask^{-1}) in the headspace. When incubated in the

buffer, the respiration rate of the cultures was 10-18% that of well-fed cultures during the first

15 h, then it decreased to about 4%.

In Bioassay 2 ("extended starvation"), denitrification was instead induced after having

exposed the cells to starvation for 20 h. The cultures were raised in fully oxic flasks containing

YMB medium. When OD_{600} reached ~0.1, the cultures were centrifuged (10 000 × g at 4 °C for

10 mins) and washed twice in sterile ddH_2O. The pellets (triplicates) were pooled after which

they were evenly divided and used to inoculate fully oxic flask containing C-free "starvation

buffer". These cultures were incubated for 20 h, then centrifuged after which the pellets were

pooled and divided evenly before being inoculated into flasks containing C-free "starvation

buffer" with 1 mM KNO_3, and with He and 1 ml N_2O (around 80 µmol N flask^{-1}) in headspace.

The respiration rate of cultures exposed to "extended starvation" was ca 1-4% compared to

that of well-fed cultures. Results from the different starvation bioassays were compared to

those from well-fed cultures. To avoid biases, the treatments, which were in all cases set up at

least in triplicates (n ≥ 3), were the same regarding centrifugations and washings until the last

step when pooled cells were inoculated to anoxic flasks containing either YMB or buffer. The

carbon source in the YMB medium comprised >200 times surplus of electron donor compared

to electron acceptors throughout all incubations, thus ensuring that the electron donor was not

depleted. All cultures were incubated at 28°C, and with vigorous stirring (600-700 rpm).

Addition of YMB medium or TMPD as electron donors to starved cultures. Experiments

were performed to investigate how starved cultures of *Bradyrhizobium* strain HAMBI 2125

responded to the addition of an electron donor, either provided as an artificial electron donor

(Fig. 4) or as YMB medium (2 ml mannitol /yeast solution providing them with a substrate

concentration corresponding to half-strength YMB, thus 5 g mannitol and 0.25 g yeast l^{-1}, Fig. 5).

As artificial electron donor we used sodium TMPD (N, N, N´, N´-tetramethyl-p-

phenylenediamine) which, in the presence of ascorbate, donates electrons to cytochrome c

thus providing electrons to Nir and NosZ (18,37). Ascorbate and TMPD (both from Sigma-

Aldrich®, Germany) were dissolved in ddH_2O or 96% ethanol, respectively, and filter sterilized.

The solutions were added to the incubation flasks 10-15 mins before gas sampling. The effect of

different concentrations of TMPD (100, 250, and 500 µM in the culture buffer) combined with

10 mM ascorbate on N_2O reduction was checked prior to the main experiments. This showed

that 500 µM TMPD had an obvious inhibition effect on N_2O reduction, while 100 µM and 250

µM showed no inhibitory effect (not shown). To minimize other, possible effects we used 100

µM TMPD for the experiments (as done in Mania et al. (18)). In another experiment, shown in

Fig. S3, 4 ml of a mannitol/yeast solution was added to flasks containing 50 ml starved cultures,

providing them with a substrate concentration corresponding to full strength YMB (10 g

mannitol and 0.5 g yeast l^{-1}).

Monitoring of gas kinetics, NO_3^- and NO_2^- concentrations and electron flow rates. The

culture flasks were placed in a robotized incubation system and the headspace gas was sampled

frequently for N_2, N_2O, NO and O_2 measurements (34). Gas losses caused by sampling were

taken into account when calculating the production and consumption of gases, as described by

Molstad et al. (34) and Mania et al. (18). The concentration of O_2 in the anoxic incubation flasks

was <0.6 µM at the start of the experiment, which was well below the level for initiating

denitrification (4.6 µM; (19)).

The NO_2^- concentrations were monitored as described in Mania et al. (18). Briefly, samples

(0.1- 0.5 ml) were taken every 1 or 2 hours (n=3) from the liquid phase through the septum of

the flasks using a sterile syringe. To avoid that the gas kinetics was affected by the sampling, a

set of flasks was dedicated to NO_2^- measurements and a parallel set was left untouched for gas

measurements. NO_2^- was determined using a chemoluminescence NOx analyzer (Sievers NOA^{TM}

280i, GE Analytical Instruments) after at first reducing the NO_2^- to NO by adding 10 µl liquid

sample into a purging device containing a reducing agent (50% acetic acid with 1% (w/v) NaI)

(53,54). The NO_2^- concentrations were determined against a standard curve (range 0-2 mM

NO_2^-; r^2= 0.999).

The cell specific electron flow rates (V_e, mol e^- $cell^{-1}$ h^{-1}) for each time increment between

two gas samplings were calculated as $V_{eflask}(t)/(N(0)+E_{cum}(t)*Y)$, where $V_{eflask}(t)$ is the electron

flow rate in the flask (mol e$^-$ flask^{-1} h^{-1}) calculated from measurements, *N(0)* is number of cells in

the flask at time=0, $E_{cum}(t)$ is the cumulated electron flow (mol e$^-$ flask^{-1}) at time=t , and Y is the

yield per mol electron (cells mol^{-1} e$^-$) for HAMBI 2125 as measured previously (19). The

conversion factor 5.8E8 cells ml^{-1} *$OD_{600}$$^{-1}$ was used to convert OD_{600} to cell numbers (19).

Proteomics. The abundances of Nap, Nir and NosZ were quantified in starved cultures

treated as described for Bioassay 2 (extended starvation). Altogether eighteen flasks were

prepared according to Fig. 1B. After the 20 h aerobic incubation in C-free buffer, three flasks

were harvested for proteomics analysis and, following centrifugation/washing, the three cell

pellets from these flasks were frozen individually at -20 °C. The cultures in the other flasks were

pooled three by three and after centrifugation/washing, each cell pellet was divided in three

and used to inoculate new flasks containing C-free buffer with NO_3^- in the liquid and with He

and 1 ml N_2O in the headspace (Fig. 1B). These flasks (fifteen in total) were placed in the robotic

incubation system for monitoring of gas kinetics and NO_2^- concentrations as described above.

The entire culture volume (50 ml) was harvested from each of triplicate flasks at different time

points during the anoxic incubation. The cultures in six of the flasks showed a "fast" respiration

rate (total electron flow 1.0-1.8 e$^-$ cell^{-1} h^{-1}) and were harvested at 5 and 20 h of incubation in

anoxic buffer (triplicates at each sampling point). The cultures in the other nine flasks showed a

"slow" respiration rate (maximum total electron flow rate was 0.27 fmol e$^-$ cell^{-1} h^{-1}). Six of

them were harvested in triplicates after 10 and 20 h in anoxic buffer. The remaining three flasks

from the cultures with "slow" respiration rate received a portion of YMB at 24.8 hand were

harvested at 27 h. Harvested cell cultures were centrifuged and stored as pellets at -20 °C. The

protein extraction was as described in Gao et al. (19). Briefly, the thawed cell pellets were

resuspended in lysis buffer (20 mM Tris–HCl pH 8, 0.1% v/v Triton X-100, 200 mM NaCl, 1 mM

DTT). They were then subjected to bead beating (3 × 45 s) with glass beads (particle size ≤ 106

µm; Sigma) using a MP Biomedicals™ FastPrep-24™ (Thermo Fischer Scientific) at maximum

power and with cooling on ice between the cycles. After centrifugation to remove cell debris

(10 000 × g; 5 min), the supernatant, containing water soluble proteins, was used for proteomic

analysis using an Orbitrap mass spectrometer (described in (19,55)). Quantification was based

on LFQ (label-free quantification) in MaxQuant (56), and the relative abundance of the

individual reductases was calculated as percentages of the sum of all protein abundances for

each time point. The nitric oxide reductases NorB/C were not measured since only a small

fraction of these membrane-bound enzymes can be accurately obtained with this extraction

protocol (approximately 1/1000th the quantity of the other reductases).

Viable counts and microscopy. The number of viable cells in starved cultures showing slow

and fast respiration rates, observed in flasks prepared according to Bioassay 2, was determined

by plating dilutions of the cultures on yeast-mannitol agar (YMA). The morphology of cells from

well-fed cultures and starved cultures was compared using phase contrast microscopy and a

qualitative determination (presence/absence) of PHA was done by staining with Nile Red

(Sigma-Aldrich) followed by fluorescence microscopy. Microscopy was performed on a Zeiss

AxioObserver with an Orca-Flash4.0 CMOS camera (Hamamatsu Photonics) controlled by the

ZEN Blue software. Images were taken with a 100x phase contrast objective. A HPX-120

Illuminator was used as light source for fluorescence microscopy. Images were prepared using

ImageJ and analysis of cell sizes was done using the ImageJ-plugin MicrobeJ (57).

Acknowledgment

This project was supported by Kingenta Ecological Engineering Group Co., LTD and by the

project PASUSI financed by the EC Horizon2020 ERA-NET Cofund Programme, Grant agreement

No. 727715 and by the Research Council of Norway, projects No. 290488 and 325770. Yuan Gao

is grateful to the China Scholarship Council (CSC) for financial support. We thank Gro Stamsås

for help with microscopy.

References

1. Hobbie JE, Hobbie EA. 2013. Microbes in nature are limited by carbon and energy: The

starving-survival lifestyle in soil and consequences for estimating microbial rates. Front

Microbiol 4:324.

2. Morita RY. 1990. The starvation-survival state of microorganisms in nature and its

relationship to the bioavailable energy. Experientia 46:813–817.

3. Bergkessel M, Basta DW, Newman DK. 2016. The physiology of growth arrest: Uniting

molecular and environmental microbiology. Nat Rev Microbiol 14:549–562.

4. Ratcliff WC, Kadam SV, and Denison, R.F. 2008. Poly-3-hydroxybutyrate (PHB) supports

survival and reproduction in starving rhizobia. FEMS Microbiol Ecol 65:391–399.

5. Müller-Santos M, Koskimäki JJ, Alves LPS, de Souza EM, Jendrossek D, Pirttilä AM. 2021. The

protective role of PHB and its degradation products against stress situations in bacteria.

FEMS Microbiol Rev 45:fuaa058.

6. Zumft WG. 1997. Cell biology and molecular basis of denitrification. Microbiol Mol Biol Rev

61:533–616.

7. Duffner C, Kublik S, Fösel B, Frostegård Å, Schloter M, Bakken L, Schulz S. 2022. Genotypic

and phenotypic characterization of hydrogenotrophic denitrifiers. Environ Microbiol

24:1887-1901.

8. Schneider LK, Wüst A, Pomowski A, Zhang L, Einsle O. 2014. No laughing matter: The

unmaking of the greenhouse gas dinitrogen monoxide by nitrous oxide reductase. Met Ions

Life Sci 14:177–210.

9. Hein S, Simon J. 2019. Bacterial nitrous oxide respiration: electron transport chains and

copper transfer reactions. Adv Microb Physiol 75:137–175.

10. Shapleigh JP. 2013. Denitrifying prokaryotes: Prokaryotic Physiology and Biochemistry. In

Rosenberg E, DeLong EF, Lory S, Stackebrandt E, Thompson F. (eds), The prokaryotes, pp.

405-425. Berlin, Heidelberg: Springer.

11. Graf DRH, Jones CM, Hallin S. 2014. Intergenomic comparisons highlight modularity of the

denitrification pathway and underpin the importance of community structure for N_2O

emissions. PLoS ONE 9:e114118.

12. Richardson D, Felgate H, Watmough N, Thomson A, Baggs E. 2009. Mitigating release of the

potent greenhouse gas N_2O from the nitrogen cycle - could enzymic regulation hold the key?

Trends Biotechnol 27:388-97.

13. Spiro S. 2012. Nitrous oxide production and consumption: regulation of gene expression by

gas-sensitive transcription factors. Philos Trans R Soc Lond B Biol Sci 367:1213-25.

14. Liu B, Frostegård Å, Bakken LR. 2014. Impaired reduction of N_2O to N_2 in acid soils is due to

a posttranscriptional interference with the expression of *nosZ*. mBio 5:e01383-14.

15. Liu B, Mao Y, Bergaust L, Bakken LR, Frostegård A. 2013. Strains in the genus *Thauera*

exhibit remarkably different denitrification regulatory phenotypes. Environ Microbiol

15:2816-28.

16. Gaimster H, Alston M, Richardson DJ, Gates AJ, Rowley G. 2018. Transcriptional and

environmental control of bacterial denitrification and N_2O emissions. FEMS Microbiol Lett

365:fnx277.

17. Lycus P, Soriano-Laguna MJ, Kjos M, Richardson DJ, Gates AJ, Milligan DA, Frostegård Å,

Bergaust L, Bakken LR. 2018. A bet-hedging strategy for denitrifying bacteria curtails their

release of N_2O. Proc Natl Acad Sci 115:11820–11825.

18. Mania D, Woliy K, Degefu T, Frostegård Å. 2020. A common mechanism for efficient N_2O

reduction in diverse isolates of nodule-forming bradyrhizobia. Environ Microbiol 22:17–31.

19. Gao Y, Mania D, Mousavi SA, Lycus P, Arntzen M, Woliy K, Lindström K, Shapleigh JP, Bakken

LR, Frostegård Å. 2021. Competition for electrons favours N_2O reduction in denitrifying

Bradyrhizobium isolates. Environ Microbiol 23:2244–2259.

20. Butterbach-Bahl K, Baggs EM, Dannenmann M, Kiese R, Zechmeister-Boltenstern S. 2013.

Nitrous oxide emissions from soils: How well do we understand the processes and their

controls? Philos Trans R Soc Lond B Biol Sci 368:20130122.

21. Thompson RL, Lassaletta L, Patra PK, Wilson C, Wells KC, Gressent A, Koffi EN, Chipperfield

M, Winiwarter W, Davidson EA, Tian H, Canadell JG. 2019. Acceleration of global N_2O

emissions seen from two decades of atmospheric inversion. Nat Clim Change 9:993–998.

22. Tian H, Lu C, Ciais P, Michalak AM, Canadell JG, Saikawa E, Huntzinger DN, Gurney KR, Sitch

S, Zhang B, Yang J, Bousquet P, Bruhwiler L, Chen G, Dlugokencky E, Friedlingstein P, Melillo

J, Pan S, Poulter B, Prinn R, Saunois M, Schwalm CR, Wofsy SC. 2016. The terrestrial

biosphere as a net source of greenhouse gases to the atmosphere. Nature 531:225–228.

23. Tian H, Xu R, Canadell JG, Thompson RL, Winiwarter W, Suntharalingam P, Davidson EA,

Ciais P, Jackson RB, Janssens-Maenhout G, Prather MJ, Regnier P, Pan N, Pan S, Peters GP,

Shi H,   Tubiello FN,   Zaehle S, Zhou F, Arneth A, Battaglia G,   Berthet S,   Bopp L,

Bouwman AF,   Buitenhuis ET, Chang J, Chipperfield MP, Dangal, SRS, Dlugokencky E,

Elkins JW, Eyre BD, Fu B, Hall B,   Ito A,   Joos F, Krummel PB, Landolfi A, Laruelle GG,

Lauerwald R, Li W, Lienert S, Maavara T, MacLeod M, Millet DB, Olin S, Patra PK, Prinn RG,

Raymond PA, Ruiz DJ, van der Werf GR, Vuichard N, Wang J, Weiss RF, Wells KC, Wilson

C, Yang J, Yao Y. 2020. A comprehensive quantification of global nitrous oxide sources and

sinks. Nature 586: 248–256.

24. Davidson EA, Kanter D. 2014. Inventories and scenarios of nitrous oxide emissions. Environ

Res Lett 9:105012.

25. Bakken LR, Frostegård Å. 2020. Emerging options for mitigating N_2O emissions from food

production by manipulating the soil microbiota. Curr Opin Environ Sustain 47:89–94.

26. Jensen ES, Peoples MB, Boddey RM, Gresshoff PM, Henrik HN, Alves BJR, Morrison MJ. 2012.

Legumes for mitigation of climate change and the provision of feedstock for biofuels and

biorefineries. A review. Agron Sustain Dev 32:329-364.

27. Poole P, Ramachandran V, Terpolilli J. 2018. Rhizobia: From saprophytes to endosymbionts.

Nat Rev Microbiol 16:291–303.

28. Hénault C, Revellin C. 2011. Inoculants of leguminous crops for mitigating soil emissions of

the greenhouse gas nitrous oxide. Plant Soil 346:289–296.

29. Itakura M, Uchida Y, Akiyama H, Hoshino YT, Shimomura Y, Morimoto S, Tago K, Wang Y,

Hayakawa C, Uetake Y, Sánchez C, Eda S, Hayatsu M, Minamisawa K. 2013. Mitigation of

nitrous oxide emissions from soils by *Bradyrhizobium japonicum* inoculation. Nat Clim

Change 3:208–212.

30. Jaiswal SK, Dakora FD. 2019. Widespread distribution of highly adapted *Bradyrhizobium*

species nodulating diverse legumes in Africa. Front Microbiol 10:310.

31. Woliy K, Degefu T, Frostegård Å. 2019. Host Range and Symbiotic Effectiveness of N_2O

Reducing Bradyrhizobium Strains. Front Microbiol 10:2746.

32. Bedmar EJ, Robles EF, Delgado MJ. 2005. The complete denitrification pathway of the

symbiotic, nitrogen-fixing bacterium *Bradyrhizobium japonicum*. Biochem Soc Trans 33:141-

144.

33. Narozna D, Pudełko K, Króliczak J, Golińska B, Sugawara M, Mądrzak CJ, Sadowsky MJ. 2015.

Survival and competitiveness of *Bradyrhizobium japonicum* strains 20 years after

introduction into field locations in Poland. Appl Environ Microbiol 81:5552–5559.

34. Molstad L, Dörsch P, Bakken LR. 2007. Robotized incubation system for monitoring gases

(O_2, NO, N_2O, N_2) in denitrifying cultures. J Microbiol Methods 71:202–211.

35. Lim NYN, Frostegård Å, Bakken LR. 2018. Nitrite kinetics during anoxia: The role of abiotic

reactions versus microbial reduction. Soil Biol Biochem 119:203–209.

36. Spiekermann P, Rehm BH, Kalscheuer R, Baumeister D, Steinbüchel A. 1999. A sensitive,

viable-colony staining method using Nile red for direct screening of bacteria that

accumulate polyhydroxyalkanoic acids and other lipid storage compounds. Arch Microbiol

171:73-80.

37. Kimelberg HK, Nicholls P. 1969. Kinetic studies on the interaction of TMPD with cytochrome

c and cytochrome *c* oxidase. Arch Biochem Biophys 133:327–335.

38. Bueno E, Richardson DJ, Bedmar EJ, Delgado MJ. 2009. Expression of *Bradyrhizobium*

japonicum cbb3 terminal oxidase under denitrifying conditions is subjected to redox control.

FEMS Microbiol Lett 298:20–28.

39. Torres MJ, Simon J, Rowley G, Bedmar EJ, Richardson DJ, Gates AJ, Delgado MJ. 2016.

Nitrous Oxide Metabolism in Nitrate-Reducing Bacteria: Physiology and Regulatory

Mechanisms. Adv Microb Physiol 68:353-432.

40. Schalk-Otte S, Seviour R, Kuenen J, Jetten MSM. 2000. Nitrous oxide (N_2O) production by

Alcaligenes faecalis during feast and famine regimes. Water Res 34:2080-2088.

41. Itokawa H, Hanaki K, Matsuo T. 2001. Nitrous oxide production in high-loading biological

nitrogen removal process under low COD/N ratio condition. Water Res 35:657-664.

42. Lu H, Chandran K. 2010. Factors promoting emissions of nitrous oxide and nitric oxide from

denitrifying sequencing batch reactors operated with methanol and ethanol as electron

donors. Biotechnol Bioeng 106:390-8.

43. Pan Y, Ni BJ, Bond PL, Ye L, Yuan Z. 2013. Electron competition among nitrogen oxides

reduction during methanol-utilizing denitrification in wastewater treatment. Water Res

47:3273-81.

44. Lycus P, Bøthun KL, Bergaust L, Shapleigh JP, Bakken LR, Frostegård Å. 2017. Phenotypic and

genotypic richness of denitrifiers revealed by a novel isolation strategy. ISME J 11:2219–

2232.

45. Kjelleberg S, Albertson N, Flärdh K, Holmquist L, Jouper-Jaan A, Marouga R, Ostling J,

Svenblad B, Weichart D. 1993. How do non-differentiating bacteria adapt to starvation?

Antonie Van Leeuwenhoek 63:333-41.

46. Gray DA, Dugar G, Gamba P, Strahl H, Jonker MJ, Hamoen LW. 2019. Extreme slow growth

as alternative strategy to survive deep starvation in bacteria. Nat Commun 10:890.

47. Thorne SH, Williams HD. 1997. Adaptation to nutrient starvation in Rhizobium

leguminosarum bv. phaseoli: analysis of survival, stress resistance, and changes in

macromolecular synthesis during entry to and exit from stationary phase. J Bacteriol

179:6894-901.

48. Lodwig EM, Leonard M, Marroqui S, Wheeler TR, Findlay K, Downie JA, Poole PS. 2005. Role

of polyhydroxybutyrate and glycogen as carbon storage compounds in pea and bean

bacteroids. Mol Plant Microbe Interact 18:67-74.

49. Chen H, Chen CY. 2014. Starvation induces phenotypic diversification and convergent

evolution in *Vibrio vulnificus*. PLoS One 9:e88658.

50. Zhang X, Nicki G, Kaijalaineni S, Terefework Z, Paulin L, Tighe SW, Graham PH, Lindström K.

(1999) Phylogeny and Diversity of *Bradyrhizobium* Strains Isolated from the Root Nodules of

Peanut (Arachis hypogaea) in Sichuan, China. Syst Appl Microbiol 22:378-386.

51. Schwartz W. 1972. Vincent, J.M. A Manual for the Practical Study of the Root-Nodule

Bacteria (IBP Handbuch No. 15 des International Biology Program, London). XI u. 164 S., 10

Abb., 17 Tab., 7 Taf. Oxford-Edinburgh 1970: Blackwell Scientific Publ., 45 s. Z Allg Mikrobiol

12:1521-4028.

52. Mania D, Heylen K, van Spanning RJ, Frostegård Å. 2016. Regulation of nitrogen metabolism

in the nitrate-ammonifying soil bacterium *Bacillus vireti* and evidence for its ability to grow

using N_2O as electron acceptor. Environ Microbiol 18:2937-50.

53. Cox RD. 1980. Determination of nitrate and nitrite at the parts per billion level by

chemiluminescence. Anal Chem 52:332–335.

54. MacArthur PH, Shiva S, Gladwin MT. 2007. Measurement of circulating nitrite and S-

nitrosothiols by reductive chemiluminescence. J Chromatogr B Analyt Technol Biomed Life

Sci 851:93–105.

55. Conthe M, Lycus P, Arntzen MØ, Ramos da Silva A, Frostegård Å, Bakken LR, Kleerebezm R,

van Loosdrecht MCM. 2019. Denitrification as an N_2O sink. Water Res 151:381–387.

56. Cox J, Mann M. 2008. MaxQuant enables high peptide identification rates, individualized

p.p.b.-range mass accuracies and proteome-wide protein quantification. Nat Biotechnol

26:1367-72.

57. Ducret A, Quardokus EM, Brun YV. 2016. MicrobeJ, a tool for high throuput bacterial cell

detection and quantitative analysis. Nat Microbiol 20:16077.

Figure legends

FIG 1 Bioassays for assessing the effect of starvation on electron flow to denitrification enzymes.

A: Bioassay 1: starvation of cells with a previously expressed denitrification proteome. Cells

were raised from stock cultures in fully oxic flasks containing YMB medium. When OD_{600}

reached ~0.1, the flasks were made anoxic by He-washing, then supplemented with 1% O_2 in

the headspace and 1 mM NO_3^- in the liquid. The cultures were allowed to deplete the O_2 and to

initiate denitrification when growing in YMB medium, after which they were centrifuged (10

000 × g at 4 °C for 10 mins) and washed twice in sterile ddH_2O. The pellets from triplicate flasks

were pooled and used to inoculate flasks containing either C-free buffer or YMB medium (well-

fed control), in both cases supplemented with 1 mM KNO_3 and 0.25 or 0.5 mM KNO_2, and with

He and 1 ml N_2O (around 80 μmol N flask^{-1}) in headspace. The starving cells, incubated in buffer,

had a low respiratory electron flow rate (mol electrons cell^{-1} h^{-1}), initially being 10-18 % that of

the well-fed cultures, and then decreasing to reach around 4% after 20 h. Results from

experiments using Bioassay 1 are presented in Figs. 2, 3 and 4.

B: Bioassay 2: denitrification induced during starvation. Cells were raised from stock cultures in

fully oxic flasks containing YMB medium. When OD_{600} reached ~0.1, the cultures were

centrifuged (10 000 × g at 4 °C for 10 mins) and washed twice in sterile ddH_2O. The pellets from

triplicate flasks were pooled, after which they were evenly divided and used to inoculate fully

oxic flasks containing C-free buffer. These cultures were incubated for 20 h, then centrifuged

after which the pellets were pooled and divided evenly before being inoculated into flasks

containing C-free buffer-provided with 1 mM KNO_3, and with He and 1 ml N_2O (around 80 µmol

N flask^{-1}) in headspace. The respiration rate (mol electrons cell^{-1} h^{-1}) of the starving cultures was

1-4% compared to that of well-fed cultures. All cultures in Bioassays 1 and 2 were incubated at

28°C, and with vigorous stirring (650 rpm). Results from experiments using Bioassay 1 are

presented in Figs. 5 and S2.

FIG 2 Denitrification kinetics as affected by starvation in cultures with a complete denitrification

proteome (Bioassay 1). Cells were allowed to develop a full denitrification proteome under well-

fed conditions and were then washed twice in buffer prior to inoculation to flasks with YMB

(well-fed control) or buffer (starved), supplemented with 1 mM KNO_3 and 0.25 mM KNO_2, and

with He plus 1 ml N_2O (around 80 µmol N) in headspace. A larger inoculum was given to the

flasks with buffer (9.9E+08 cells flask^{-1}) than to the flasks with YMB (5.0E+09 cells flask^{-1}), to

secure measurable activity in the starved cells and adequate time resolution of the

denitrification kinetics in the well-fed cells. <u>Panels A and B</u> show the denitrification kinetics in

well-fed and starved cultures, respectively. The flasks were practically anoxic from the start with

an initial O_2 concentration in the liquid of <0.52 µM which decreased to approximately 0 µM

(insets in A & B). <u>Panels C and D</u> show the cell specific electron flow rate. $V_{e\ total\ den}$ designates

the total electron flow to the denitrification reductases and $V_{e\ total\ incl\ O2}$ the total electron flow,

including that to denitrification and to O_2. The electron flow rate to the individual reductases is

designated as V_{eNap}, V_{eNir}, V_{eNor} and V_{eNos}. V_{eNir} and V_{eNor} were practically identical and cannot be

distinguished from one another in the figure. Inserted panels show V_{eNap} throughout, including

negative values which are due to slight errors in determination of N_2 and N_2O (V_{eNap} was

calculated by N-mass balance). Bars in all graphs show standard deviation (n=3).

FIG 3 Bioassay 1 with single nitrogen oxides**.** Experimental conditions as for Fig. 2, but the

starving cells (in buffer) were provided with either NO_3^-, NO_2^- or N_2O in individual flasks (n=3 for

each treatment). <u>Panels A-C</u> show the gas kinetics in flasks provided with 1 mM NO_3^- (A), 0.5

mM NO_2^- (B), or 70 µmol N_2O-N added to the headspace (C). <u>Panel D</u> shows the cell-specific

electron flow rate (V_e) measured in buffer supplemented with NO_3^-, NO_2^- or N_2O; the O_2

concentration (left y-axis); and V_e as percentage of the rates in well-fed cultures (right y-axis).

The number of cells inoculated into the incubation flasks at the start (0 h of incubation in anoxic

buffer) was 3.6E+09 for all treatments. Bars show standard deviation (n=3).

FIG 4 Denitrification in starved cells after addition of TMPD as an external electron donor.

Preparation of the cultures followed Bioassay 1 (Fig. 1A), except that 100 µM TMPD and 10 mM

ascorbate were injected into the flasks with starving cells 15 min before the first gas sampling.

Each flask was inoculated with 5.28E+08 cells (n=4 replicate flasks). The initial O_2 concentration

was 0.35 µM and decreased to approximately 0 within 5 h (inset in A). Bars show standard

deviation (n=4). <u>Panel A</u>: kinetics of NO_2^-, NO, N_2O and N_2 (O_2 in inserted panel). <u>Panel B:</u>

Calculated cell specific electron flow rate (V_e, fmol cell^{-1} h^{-1}) to each of the reductases Nap, Nir,

Nor, and NosZ, and the total electron flow rate (V_{etotal}). The electron flow to Nir was practically

identical to the electron flow to Nor (miniscule amounts of NO), and the two are shown as a

single graph ($V_{eNir/Nor}$). Inset plots show V_{eNap} throughout, including negative values which are

due to slight errors in determination of N_2 and N_2O (V_{eNap} was calculated by N-mass balance).

FIG 5 Bioassay 2: stochasticity of starvation response, response to input of organic C and

quantification of denitrification enzymes. Altogether fifteen flasks were prepared following

Bioassay 2 (see Fig. 1B). All flasks were anoxic (<0.5 µM at the start of the incubation) and

contained 50 ml buffer supplemented with 1 mM KNO_3 and with He plus 1 ml N_2O in the

headspace. The NO_2^- concentrations, measured during the first 5 hours after incubation in the

buffer, were approximately 0.59±0.20 µM (not shown). The cultures separated into two distinct

phenotypes: nine flasks had "slow" respiration (panel A) and 6 flasks had "fast" respiration

(panel B). The electron flow to the individual reductases (fmol e^- $cell^{-1}$ h^{-1}) for the flasks with

slow and fast respiration are shown in panels C and D, respectively. Cultures with "slow"

respiration rate had a total electron flow rate of maximum 0.27 e^- $cell^{-1}$ h^{-1} (C) and cultures with

"fast" respiration rate had a total electron flow rate of 1.0-1.8 fmol e- $cell^{-1}$ h^{-1} (D). The inset plot

in panel C shows the electron flow to the individual denitrification reductases after the carbon

addition. The cultures (entire flasks, 50 ml) were sampled for proteomics analyses at different

time points (panels E and F). At timepoints throughout, marked by dashed vertical lines in

panels A and B, three flasks were harvested for proteomics analysis (including 0 h). The six flasks

with "fast" respiration were harvested for proteomics analysis at 5 and 20 h of incubation in

these anoxic buffers (triplicates at each sampling point). Of the nine flasks with "slow"

respiration, three were harvested at 10 and three at 20 h. The remaining three "slow"

respiration flasks were supplemented with YMB at 24.8 h, at a concentration which made the

buffer a half-strength YMB medium, containing 5 g mannitol l^{-1} and 0.25 yeast extract l^{-1}, and

then harvested for proteomics at 27 h. The gas measurements and electron flows are averages

from the three flasks (n=3) for each phenotype, that were left untouched until the end of the

incubation when they were sampled for proteomics. Proteomics analyses were done for

triplicate flasks (n=3) at each sampling point. Standard deviations are indicated as bars in all

graphs (in several cases not visible due to low variation).

Bioassay 1: Denitrification induced before starvation

Bioassay 2: Denitrification induced during starvation

FIG 1 Bioassays for assessing the effect of starvation on electron flow to denitrification enzymes.
A: Bioassay 1: starvation of cells with a previously expressed denitrification proteome. Cells were raised from stock cultures in fully oxic flasks containing YMB medium. When OD_{600} reached ~0.1, the flasks were made anoxic by He-washing, then supplemented with 1% O_2 in the headspace and 1 mM NO_3^- in the liquid. The cultures were allowed to deplete the O_2 and to initiate denitrification when growing in YMB medium, after which they were centrifuged (10 000 × g at 4 °C for 10 mins) and washed twice in sterile ddH_2O. The pellets from triplicate flasks were pooled and used to inoculate flasks containing either C-free buffer or YMB medium (well-fed control), in both cases supplemented with 1 mM KNO_3 and 0.25 or 0.5 mM KNO_2, and with He and 1 ml N_2O (around 80 µmol N flask^{-1}) in headspace. The starving cells, incubated in buffer, had a low respiratory electron flow rate (mol electrons cell^{-1} h^{-1}), initially being 10-18 % that of the well-fed cultures, and then decreasing to reach around 4% after 20 h. Results from experiments using Bioassay 1 are presented in Figs. 2, 3 and 4.
B: Bioassay 2: denitrification induced during starvation. Cells were raised from stock cultures in fully oxic flasks containing YMB medium. When OD_{600} reached ~0.1, the cultures were centrifuged (10 000 × g at 4 °C for 10 mins) and washed twice in sterile ddH_2O. The pellets from triplicate flasks were pooled, after which they were evenly divided and used to inoculate fully oxic flasks containing C-free buffer. These cultures were incubated for 20 h, then centrifuged after which the pellets were pooled and divided evenly before being inoculated into flasks containing C-free buffer-provided with 1 mM KNO_3, and with He and 1 ml N_2O (around 80 µmol N flask^{-1}) in headspace. The respiration rate (mol electrons cell^{-1} h^{-1}) of the starving cultures was 1-4% compared to that of well-fed cultures. All cultures in Bioassays 1 and 2 were incubated at 28°C, and with vigorous stirring (650 rpm). Results from experiments using Bioassay 1 are presented in Figs. 5 and S2.

FIG 2 Denitrification kinetics as affected by starvation in cultures with a complete denitrification proteome (Bioassay 1). Cells were allowed to develop a full denitrification proteome under well-fed conditions and were then washed twice in buffer prior to inoculation to flasks with YMB (well-fed control) or buffer (starved), supplemented with 1 mM KNO_3 and 0.25 mM KNO_2, and with He plus 1 ml N_2O (around 80 µmol N) in headspace. A larger inoculum was given to the flasks with buffer (9.9E+08 cells flask^{-1}) than to the flasks with YMB (5.0E+09 cells flask^{-1}), to secure measurable activity in the starved cells and adequate time resolution of the denitrification kinetics in the well-fed cells. <u>Panels A and B</u> show the denitrification kinetics in well-fed and starved cultures, respectively. The flasks were practically anoxic from the start with an initial O_2 concentration in the liquid of <0.52 µM which decreased to approximately 0 µM (insets in A & B). <u>Panels C and D</u> show the cell specific electron flow rate. $V_{e\ total\ den}$ designates the total electron flow to the denitrification reductases and $V_{e\ total\ incl\ O2}$ the total electron flow, including that to denitrification and to O_2. The electron flow rate to the individual reductases is designated as V_{eNap}, V_{eNir}, V_{eNor} and V_{eNos}. V_{eNir} and V_{eNor} were practically identical and cannot be distinguished from one another in the figure. Inserted panels show V_{eNap} throughout, including negative values which are due to slight errors in determination of N_2 and N_2O (V_{eNap} was calculated by N-mass balance). Bars in all graphs show standard deviation (n=3).

FIG 3 Bioassay 1 with single nitrogen oxides. Experimental conditions as for Fig. 2, but the starving cells (in buffer) were provided with either NO_3^-, NO_2^- or N_2O in individual flasks (n=3 for each treatment). <u>Panels A-C</u> show the gas kinetics in flasks provided with 1 mM NO_3^- (A), 0.5 mM NO_2^- (B), or 70 µmol N_2O-N added to the headspace (C). <u>Panel D</u> shows the cell-specific electron flow rate (V_e) measured in buffer supplemented with NO_3^-, NO_2^- or N_2O; the O_2 concentration (left y-axis); and V_e as percentage of the rates in well-fed cultures (right y-axis). The number of cells inoculated into the incubation flasks at the start (0 h of incubation in anoxic buffer) was 3.6E+09 for all treatments. Bars show standard deviation (n=3).

FIG 4 Denitrification in starved cells after addition of TMPD as an external electron donor. Preparation of the cultures followed Bioassay 1 (Fig. 1A), except that 100 µM TMPD and 10 mM ascorbate were injected into the flasks with starving cells 15 min before the first gas sampling. Each flask was inoculated with 5.28E+08 cells (n=4 replicate flasks). The initial O_2 concentration was 0.35 µM and decreased to approximately 0 within 5 h (inset in A). Bars show standard deviation (n=4). <u>Panel A</u>: kinetics of NO_2^-, NO, N_2O and N_2 (O_2 in inserted panel). <u>Panel B:</u> Calculated cell specific electron flow rate (V_e, fmol cell^{-1} h^{-1}) to each of the reductases Nap, Nir, Nor, and NosZ, and the total electron flow rate (V_{etotal}). The electron flow to Nir was practically identical to the electron flow to Nor (miniscule amounts of NO), and the two are shown as a single graph ($V_{eNir/Nor}$). Inset plots show V_{eNap} throughout, including negative values which are due to slight errors in determination of N_2 and N_2O (V_{eNap} was calculated by N-mass balance).

FIG 5 Bioassay 2: stochasticity of starvation response, response to input of organic C and quantification of denitrification enzymes. Altogether fifteen flasks were prepared following Bioassay 2 (see Fig. 1B). All flasks were anoxic (<0.5 µM at the start of the incubation) and contained 50 ml buffer supplemented with 1 mM KNO₃ and with He plus 1 ml N₂O in the headspace. The NO₂⁻ concentrations, measured during the first 5 hours after incubation in the buffer, were approximately 0.59±0.20 µM (not shown). The cultures separated into two distinct phenotypes: nine flasks had "slow" respiration (panel A) and 6 flasks had "fast" respiration (panel B). The electron flow to the individual reductases (fmol e⁻ cell⁻¹ h⁻¹) for the flasks with slow and fast respiration are shown in panels C and D, respectively. Cultures with "slow" respiration rate had a total electron flow rate of maximum 0.27 e⁻ cell⁻¹ h⁻¹ (C) and cultures with "fast" respiration rate had

a total electron flow rate of 1.0-1.8 fmol e- cell^{-1} h^{-1} (D). The inset plot in panel C shows the electron flow to the individual denitrification reductases after the carbon addition. The cultures (entire flasks, 50 ml) were sampled for proteomics analyses at different time points (panels E and F). At timepoints throughout, marked by dashed vertical lines in panels A and B, three flasks were harvested for proteomics analysis (including 0 h). The six flasks with "fast" respiration were harvested for proteomics analysis at 5 and 20 h of incubation in these anoxic buffers (triplicates at each sampling point). Of the nine flasks with "slow" respiration, three were harvested at 10 and three at 20 h. The remaining three "slow" respiration flasks were supplemented with YMB at 24.8 h, at a concentration which made the buffer a half-strength YMB medium, containing 5 g mannitol l^{-1} and 0.25 yeast extract l^{-1}, and then harvested for proteomics at 27 h. The gas measurements and electron flows are averages from the three flasks (n=3) for each phenotype, that were left untouched until the end of the incubation when they were sampled for proteomics. Proteomics analyses were done for triplicate flasks (n=3) at each sampling point. Standard deviations are indicated as bars in all graphs (in several cases not visible due to low variation).