

1 Denitrifying bradyrhizobia retain strong N₂O reduction during periods of starvation

2 Yuan Gao, Magnus Øverlie Arntzen, Morten Kjos, Lars R. Bakken and Åsa Frostegård

3 Faculty of Chemistry, Biotechnology and Food Sciences, Norwegian University of Life Sciences,

4 N-1432 Ås, Norway

5

6 **Abstract**

7 Rhizobia living as microsymbionts inside nodules have stable access to carbon substrates, but
8 also have to survive as free-living bacteria in soil where they are starved for carbon and energy
9 most of the time. Many rhizobia can denitrify, thus switch to anaerobic respiration under low
10 O₂ tension using N-oxides as electron acceptors. The cellular machinery regulating this
11 transition is relatively well-known from studies under optimal laboratory conditions, while little
12 is known about this regulation in starved organisms. It is, for example, not known if the strong
13 preference for N₂O- over NO₃⁻-reduction in bradyrhizobia is retained under carbon limitation.
14 Here we show that starved cultures of a *Bradyrhizobium* strain with respiration rates 1-18% of
15 well-fed cultures, reduced all available N₂O before touching provided NO₃⁻. Proteomics showed
16 similar abundance of Nap (periplasmic NO₃⁻ reductase) and NosZ (N₂O reductase), suggesting
17 that competition between electron pathways to Nap and NosZ favoured N₂O reduction also in
18 starved cells, similar to well-fed cultures. This contrasts the general notion that NosZ activity
19 diminishes under carbon limitation. The results suggest that bradyrhizobia carrying NosZ can
20 act as strong sinks for N₂O under natural conditions and that this criterion should be considered
21 in the development of biofertilizers.

22

23 **Importance**

24 Novel biotechnological approaches are needed to curb the escalating emissions of the
25 greenhouse gas N₂O from agricultural soils. One possibility is inoculation of N₂O-reducing
26 organisms into soil. Rhizobial inoculants are used at large scale to enhance N₂-fixation of
27 legumes. The genus *Bradyrhizobium* encompasses N₂-fixing symbionts of economically
28 important legumes. Most bradyrhizobia can also denitrify, but the commercially available
29 inoculants generally lack the last denitrification step, reduction of N₂O. Recent results revealed
30 that bradyrhizobia with complete denitrification strongly favor N₂O over nitrate, making them
31 potential strong sinks for N₂O. In those studies the cultures received ample amounts of
32 substrate, while bacteria in soil generally starve. Here we demonstrate that bradyrhizobia
33 retain their strong preference for N₂O also under carbon starvation. The findings add basic
34 knowledge about mechanisms controlling denitrification and support the potential for
35 developing new bradyrhizobial legume inoculants with the dual capacity to optimize N₂-fixation
36 and minimize N₂O emission.

37

38 **Introduction**

39 Bacteria in most natural and engineered environments are faced with fluctuating availability of
40 nutrients and need to adapt to a “feast and famine” lifestyle. While many soil types are rich in
41 total organic carbon, the concentration of bioavailable carbon substrate is low, particularly in
42 non-rhizosphere soil where lack of substrate is a major factor limiting the growth of
43 heterotrophic bacteria (1). It is likely that bacteria in soil are starved most of the time (2) and
44 only experience infrequent episodes of ample provision of carbon substrate, for example as

45 exudates from a root or organic material released during decay of dead (micro)organisms.
46 Bacteria have developed several strategies to survive extended periods of starvation, such as
47 the development of high-affinity uptake systems to scavenge alternative carbon sources from
48 the surroundings, as well as changes in cell morphology, condensation of the nucleoid and
49 decreased protein synthesis to adapt to a low metabolic activity (3). Many bacteria produce
50 carbon-rich storage materials such as PHA (poly-3-hydroxyalkanoates) and glycogen during
51 periods of substrate availability, which can be utilized to sustain a minimum of metabolic
52 activity when deprived of carbon and energy (4,5).

53 Several of the microbially mediated processes in the global nitrogen cycle are closely linked
54 to the carbon (C) metabolism of the organisms. One example is denitrification, which is the
55 reduction of NO_3^- to N_2 through anaerobic respiration where the N-oxides are used as terminal
56 electron acceptors when O_2 becomes scarce. This process can be performed by a diverse range
57 of heterotrophic bacteria, archaea and fungi, which use various forms of organic compounds as
58 electron donors to obtain energy (6) or, in some cases, H_2 (7). The last step of denitrification is
59 the reduction of N_2O , a strong climate gas, to harmless N_2 , catalyzed by the N_2O reductase
60 (Nos) (8,9). It is found in a diverse range of prokaryotic organisms but has not been reported in
61 eukaryotes. Some denitrifying prokaryotes can perform all steps of denitrification, others only
62 some, and lack of the last step is common due to absence of the *nosZ* gene coding for NosZ, or
63 lack of other essential gene(s) in the *nos* operon (10,11), but the amounts of N_2O released from
64 denitrification in relation to N_2 (the $\text{N}_2\text{O}/\text{N}_2$ product ratio) is also influenced by transcriptional
65 and post-transcriptional control mechanisms and by various environmental factors (6,12-19).
66 Denitrification in agricultural soils is a major source of N_2O , accounting for more than 60% of

67 the global anthropogenic emissions (20,21). A steady increase in atmospheric N₂O has been
68 recorded since the start of industrialization, largely driven by increasing and excessive use of
69 synthetic fertilizers (22,23) and these emissions are predicted to continue to increase unless
70 novel mitigation options are developed (24,25).

71 Although the addition of reactive nitrogen compounds via synthetic fertilizers accounts for
72 the main part of the N₂O emissions from agricultural soil, the N₂O emitted from legume
73 cropped fields is far from negligible. A compilation of data from ca 70 legume cropped fields
74 estimated ca 1.29 kg N₂O-N ha⁻¹ during one growing season, while the corresponding data for
75 N-fertilized crops and pastures showed emissions of 3.22 kg N₂O-N ha⁻¹ (26). Legumes do not
76 have to rely on uptake of reactive N such as NH₄⁺ or NO₃⁻ but can acquire N through their
77 symbiotic relationship with some groups of bacteria, collectively called rhizobia, that elicit the
78 production of root nodules on the plant in which they fix atmospheric N₂. In this process, the
79 rhizobia reduce N₂ to NH₃ and the plant cells reduce the NH₃ further to glutamine which they
80 use to produce amino acids and eventually proteins (27). When this N-rich plant material is
81 degraded, organic N is released and mineralized to NH₃/NH₄⁺ which will be oxidized to NO₃⁻ by
82 nitrifying organisms. The O₂ consumption by the nitrifiers, together with the production of NO₃⁻
83 and the availability of organic compounds from the degraded plants, creates conditions that are
84 conducive to denitrification. A novel approach to minimize N₂O emissions from agricultural soil
85 is to enhance the populations of N₂O reducing bacteria (25). In the case of legume crops, which
86 are often inoculated with rhizobia to optimize the N₂-fixation, there are a few promising studies
87 reporting decreased N₂O emissions from soybean fields inoculated with rhizobia with the dual
88 capability of efficient N₂ fixation and N₂O reduction (28,29).

89 Based on this, selection of rhizobial strains for development of commercial inoculants
90 should, ideally, take both these aspects into account. One problem is, however, that far from all
91 rhizobia carry the *nosZ* gene that encodes Nos. There are relatively few surveys of
92 denitrification genes in different groups of rhizobia. Complete denitrification, which includes all
93 four reduction steps of NO_3^- to N_2 , has so far mainly been reported for the genus
94 *Bradyrhizobium*, which is the microsymbiont of a range of economically important legume
95 crops such as soybean, cow pea and peanut (30,31). A full set of denitrification reductases in
96 bradyrhizobia include, with few exceptions, the periplasmic NO_3^- reductase Nap; the Cu-
97 containing NO_2^- reductase NirK, the *bc*-type NO reductase cNor and a NosZ belonging to clade I
98 (19,32). We recently investigated the denitrification capacity of bradyrhizobia from two strain
99 collections, one obtained from nodules of legume trees and herbs growing in Ethiopia, the
100 other mainly consisting of strains isolated from nodules of peanut growing in China (18,19). In
101 these collections, 50 and 37% of the isolates, respectively, were complete denitrifiers, while the
102 others generally lacked NosZ and thus were potential N_2O sources. Common to all strains with
103 complete denitrification was a strong preference for N_2O - over NO_3^- reduction. Transcription
104 analysis and proteomics showed comparable expression levels of Nap and Nos, suggesting a
105 control mechanism at the metabolic level where the electron pathway to Nos, which receives
106 electrons from cytochrome *c* via the *bc1* complex, competes very efficiently with the electron
107 pathway to Nap which goes via the membrane bond NapC.

108 The results presented in (18,19) were based on experiments with organisms provided with
109 ample amounts of C substrate (electron donor), which is likely to reflect the situation in legume
110 nodules where the microsymbiont receives C from the plant. It can, however, be expected that

111 rhizobia, which may survive for many years in soil (33), spend a large part of their life cycle as
112 free-living organisms in soil where they will experience lack of available C substrate most of the
113 time (2). Rhizobial inoculants that carry NosZ are thus potentially important sinks for N₂O
114 produced both by themselves and by other soil microbes. It is, however, not known if the
115 competition for electrons favoring N₂O reduction in well-fed cultures is retained during
116 substrate limitation (“starvation”). Here we exposed cultures of *Bradyrhizobium* strain HAMBI
117 2125, also studied in (19), to shorter and extended periods of starvation and analyzed the
118 denitrification kinetics, including the electron flow rates to the individual denitrification
119 reductases. We also quantified the cellular abundancies of Nap, Nir and Nos. The results have
120 practical implications, supporting that these organisms can act as sinks for N₂O under natural
121 conditions. Moreover, the results are ecologically interesting since they show that cultures
122 exposed to extended starvation divided into two opposite denitrification phenotypes, one with
123 very slow metabolism, the other with retained metabolism, possibly reflecting a strategy to
124 increase the chances for survival during periods of starvation.

125

126 **Results**

127 **Denitrification kinetics in cultures prepared following Bioassay 1.** Two bioassays were
128 developed for starvation experiments, Bioassay 1 and 2 (Figs. 1A and B). In a first experiment,
129 the denitrification gas kinetics, concentrations of NO₃⁻ and NO₂⁻ and electron flow rates to the
130 different reductases were compared for well-fed vs starved cultures (Figs. 2A-D). For this, the
131 starved cultures were prepared following Bioassay 1 (Fig. 1A), in which the cultures were
132 allowed to synthesize the denitrification reductases in the presence of ample amounts of

133 substrate. Cultures were raised under fully oxic conditions in YMB medium, after which the
134 headspace was made hypoxic to allow transition to anaerobic respiration in response to a
135 gradual depletion of oxygen. After centrifugation/washing, pellets were pooled and inoculated
136 into flasks containing YMB ($9.9\text{E}+08$ cells flask⁻¹) or buffer ($5.0\text{E}+09$ cells flask⁻¹), supplemented
137 with 1 mM KNO₃ and 0.25 mM KNO₂, and with He plus 1 ml N₂O (around 80 μmol N) in
138 headspace.

139 The initial O₂ concentration, which was 0.5 μM in cultures with YMB and 0.2 μM in cultures
140 with buffer, was depleted within the first 5 h in both treatments (insets in Figs. 2A and B). The
141 provided NO₂⁻ and N₂O were reduced simultaneously from the beginning of the incubation in
142 both treatments. The NO₃⁻ was left untouched in the well-fed cultures until the exogenous N₂O
143 was reduced (Fig. 2A), also seen from lack of electron flow to Nap except for a small peak in
144 electron flow early in the anoxic incubation (Fig. 2C). No NO₃⁻ reduction took place in the
145 starved cultures throughout the entire incubation period, as seen from the lack of electron flow
146 to Nap (Fig. 2D). The negative V_{eNap} estimated for the starved cultures (inset in Fig. 2D) and the
147 initial phase of the well-fed cultures are probably due to minor errors in calibration of N-gas
148 measurements as well as in parameters used to calculate rates of N-transformation (sampling
149 loss and N₂-leakage, see Molstad et al. (34)), which amounts to substantial errors in the
150 estimates of NO₃⁻ reduction rates because they are based on N-mass balance (explained in detail
151 by Lim et al. (35)). This implies a relatively high detection limit for NO₃⁻ reduction, and the
152 negative values cannot be taken as evidence for the complete absence of any electron flow to
153 NO₃⁻. However, the measured NO concentrations lend some support to the claim that V_{eNap} was
154 ~0 until depletion of the externally provided N₂O. In the well-fed culture, the concentration of

155 NO declined to zero in response to depletion of NO_2^- , and increased soon after, as V_{eNap}
156 increased (Figs. 2A and C). Likewise, the NO concentration declined to very low values in
157 response to depletion of NO_2^- in the starved culture (Fig. 2B).

158 Fig. 2C and D show the calculated electron flow rates per cell to the individual reductases,
159 and their sum, illustrating their competition for electrons. This shows that the well-fed cells
160 (Fig. 2C) sustained a nearly constant total electron flow rate around $9 \text{ fmol e}^- \text{ cell}^{-1} \text{ h}^{-1}$
161 throughout, but allocated to different reductases depending on the availability of electron
162 acceptors: As long as NO_2^- and N_2O were both present, Nos captured around 50% of the
163 electrons ($V_{eNos} \sim V_{eNir} + V_{eNor}$), increasing to 100% when NO_2^- was depleted after 6-7 hours,
164 while the electron flow rate to Nap remained insignificant until the externally provided N_2O
165 became depleted. The starving cells (Fig. 2D) had an order of magnitude lower total electron
166 flow rate per cell, declining gradually throughout the incubation, and here Nos captured $\gg 50\%$
167 of the electrons during the first 10 h ($V_{eNos} \gg V_{eNir} + V_{eNor}$), but declined to $\sim 50\%$ towards the
168 end.

169 Bioassay 1 was also used in a second experiment, but in this case the cultures were
170 incubated anoxically ($[\text{O}_2] < 0.21 \pm 0.04 \mu\text{M}$) in buffer containing only one of the N-oxides (NO_3^- ,
171 NO_2^- or N_2O) as initial electron acceptor (Figs. 3A-C). This experiment established that starved
172 cultures readily reduced NO_3^- from the start when no other N-oxide was provided (Fig. 3A). It
173 also demonstrated that the cell specific electron flow to N-oxides was practically unaffected by
174 the type of electron acceptor provided, except for the slightly lower rates initially for flasks with
175 NO_2^- . For all treatments, the cell specific electron flow rate decreased gradually throughout,
176 and the levels are very similar to those observed in the first experiment (Fig. 3D).

177

178 **Cell size and PHA accumulation of starved vs well-fed cultures.** The morphologies of single
179 bacterial cells from different treatments (starved/well-fed and anoxic/oxic) were analyzed by
180 phase contrast microscopy (Fig. S1A). By quantifying cell dimensions, we found marginal but
181 statistically significant effects (Mann-Whitney test, $p < 0.01$) of starvation: while there was no
182 significant difference in cell area of individual cells (Fig. S1B), starved cells were on average
183 ~12% longer and ~5% thinner than well-fed cells (Fig. S1B). By further calculating cell volumes, a
184 slight reduction in average cell volume was observed upon starvation of aerobic cells ($p=0.057$),
185 but not for cells grown under anoxic conditions (Fig. S1C). Furthermore, a qualitative analysis of
186 the presence of PHA granules was done by staining the cells with Nile Red, a lipophilic dye with
187 high affinity for PHA (36). Large foci corresponding to PHA granules were seen in all conditions.
188 It should be noted that, as a lipophilic dye, Nile Red will also bind non-specifically to lipids and
189 membrane, and based on the imaging performed here, it could not be concluded whether there
190 were any significant differences in PHA content between the conditions.

191

192 **Denitrification kinetics after providing starving cultures with an artificial electron donor.**

193 An experiment was then performed in which TMPD plus ascorbate was added to starved
194 cultures (Bioassay 1), to provide cytochrome *c* with an excess of electrons (37,38). Since
195 cultures treated according to Bioassay 1 were able to provide electrons for denitrification (10-
196 18% of the electron flow in well-fed cells during the first 5 h, then decreasing to ca 4%; Fig. 2D),
197 we expected that the TMPD treatment would reduce or eliminate the oxidation of quinol by the
198 *bc1* complex and that this would allow electrons to flow to Nap via NapC, resulting in NO_3^-

199 reduction. We also expected that loading cytochrome *c* with electrons would relieve or weaken
200 the competition for electrons between Nir, Nor and NosZ. The results (Fig. 4) lend little support
201 to the former since the electron flow to Nap remained insignificant until N₂O had been
202 depleted. But the results confirm an effect of TMPD on the competition between Nos and Nir:
203 V_{eNir} , V_{eNor} and V_{eNos} were similar and high (4-5 fmol e⁻ cell⁻¹ h⁻¹) until NO₂⁻ was depleted. Thus,
204 Nir and Nos competed equally well when the cytochrome *c* pool was fully reduced by TMPD.
205 After depletion of NO₂⁻, V_{eNos} leaped to its maximum level, 13-14 fmol N cell⁻¹ h⁻¹, and kept this
206 rate until all the exogenous N₂O was depleted.

207

208 **Denitrification kinetics and reductase abundancies of cultures exposed to extended**
209 **starvation following Bioassay 2.** While Bioassay 1 successfully induced starvation in terms of a
210 downshift in respiration, the rates did not reach a stable level during the assay but declined
211 gradually throughout, apparently approaching more stable low levels after 20 h. On this
212 background, we introduced a more severe starvation assay (Bioassay 2, Fig. 1B) by including a
213 20 h aerobic incubation in buffer prior to the starvation-denitrification assay, in order to reach a
214 lower and more stable rate of respiration than in the first experiment. In addition, Bioassay 2
215 was designed to force the cells to synthesize the denitrification proteome while starving. In
216 several preliminary experiments with Bioassay 2 (results not shown), we found a conspicuous
217 variability between flasks regarding the cell specific respiration rate where one or two out of
218 three replicate flasks had four to six times higher cell respiration rates than the other(s). At first,
219 we suspected that it could be due to impurities of flasks or magnets, but meticulous acid
220 washing failed to remove the stochastic variation. Being convinced that the stochasticity

221 reflects a real regulatory switch of the cultures, we performed a final experiment in which 15
222 replicate flasks were monitored for denitrification kinetics (Fig. 5).

223 The cultures received initially 1 mM NO_3^- in the buffer (but no NO_2^-), and He plus 1 ml N_2O
224 (80 $\mu\text{mol N}$) in headspace. The O_2 concentration at the time of inoculation was $<0.5 \mu\text{M}$ in the
225 liquid. The flasks separated into two distinct denitrification phenotypes (Figs. 5A and B). Nine
226 flasks showed “low” cell specific respiration rates (total electron flow maximum 0.27 fmol e^-
227 $\text{cell}^{-1} \text{ h}^{-1}$), corresponding to approximately 2.7% compared to well-fed cultures (Fig. 5A), while
228 the other six showed a “fast” respiration rate (total electron flow $1.0\text{-}1.8 \text{ fmol e}^- \text{ cell}^{-1} \text{ h}^{-1}$). Both
229 phenotypes reduced N_2O from the beginning of the incubation, showing a strong preference for
230 N_2O over NO_3^- . In the flasks with fast respiration, the cells started to reduce NO_3^- in response to
231 depletion of the externally provided N_2O . In the flasks with slow respiration, N_2O was not
232 depleted within the time frame of the experiment, and NO_3^- reduction remained negligible.

233 To investigate if cell lysis, and thus release of available C, could explain why some cultures
234 showed the “fast” growth phenotype, we compared the OD_{600} and the number of viable cells in
235 “slow” vs “fast” cultures. The samples were taken after incubation for 3.1 h in anoxic buffer,
236 when the two phenotypes were clearly distinct but when some possible growth of cells in the
237 “fast” cultures would not yet hide if lysis had occurred. The OD_{600} spanned from 0.12-0.13 with
238 no statistical difference ($p>0.3$) between cultures with “fast” and “slow” respiration. Similarly,
239 no difference ($p=0.4$) was found for the viable counts which showed $1.21\text{E}+10$ to $1.33\text{E}+10$ CFU
240 flask^{-1} for the “fast” cultures and $1.23\text{E}+10$ to $1.34\text{E}+10$ CFU flask^{-1} for the “slow” cultures (Fig.
241 S2).

242 We also tested the metabolic integrity of the cells in the flasks with slow respiration by
243 injecting C-substrates (YMB) to the flasks after 24.8 hours, which proved their capacity to
244 quickly regain activity approaching that of well-fed cells (Fig. 5A). To investigate if the observed
245 divergencies in phenotypes were due to differences in denitrification reductase abundancies,
246 we quantified the relative abundances of Nap, NirK and NosZ in samples taken at different time
247 points throughout the incubations (Figs. 5E and F), together with the corresponding
248 denitrification kinetics (Figs. 5A and B) and cell specific electron flow to reductases (Figs. 5C and
249 D). The membrane-bound NO reductase (cNor) could not be extracted quantitatively (the
250 results showed 1000 times lower abundancies than for the other denitrification reductases) and
251 is therefore not shown. The inoculum had been cultured aerobically for 3-4 generations, never
252 permitting the OD₆₀₀ to exceed 0.1, to secure that any denitrification enzymes would be diluted
253 to extinction by aerobic growth, assuming that the transcription of all genes is effectively
254 repressed by oxygen. This strategy was apparently successful since Nap and Nir were
255 undetectable at the start of the anoxic incubation. NosZ, on the other hand, was detected also
256 in the aerobic cultures, suggesting that the *nosZ* gene is constitutively transcribed at low levels
257 in these organisms.

258 After transferring the cells to anoxic buffer, the abundance of all three reductases increased
259 both in cultures with “slow” and “fast” respiration. Cultures with “slow” respiration rate
260 synthesized less denitrification reductases than those with “fast” respiration rate during the
261 first 20 h. The relative abundances of the different reductases also differed between the two
262 phenotypes. In cultures with “slow” respiration, NosZ was significantly more abundant than
263 Nap and Nir ($p < 0.01$), which were comparable (Fig. 5E). Cultures with “fast” respiration instead

264 contained higher abundancies of NosZ and Nir compared to Nap at 5 h. After this, the
265 abundance of Nir increased more than that of the others and at 20 h the LFQ of Nir was $0.20 \pm$
266 0.07 , while the abundancies of NosZ and Nap were approximately half of that (Fig. 5F). After
267 addition of carbon substrate to the cultures with “slow” respiration, a rapid synthesis of all
268 three reductases took place. This synthesis was most prominent for Nir which, after a couple of
269 hours, had increased three-fold, reaching a relative abundance that was twice as high as Nap
270 and NosZ (Fig. 5E), resembling the abundance profile of the cultures with “fast” respiration (Fig.
271 5F).

272

273 Discussion

274 Detailed eco-physiological studies during the past decades have revealed several aspects of
275 how the denitrification process is regulated in different organisms (6,13,39). Most of this
276 knowledge is, however, based on laboratory studies where cultures have been grown under
277 optimal conditions, while less attention has been paid to understanding how denitrification,
278 and particularly the release of N_2O , is affected if cells are starved, which is the normal state of
279 cells in most natural environments. We focused on how starvation, i. e. lack of carbon substrate,
280 affects denitrification and N_2O release. The general notion has been that N_2O reductase is less
281 competitive for electrons than the other denitrification reductases, leading to emissions of N_2O
282 when the availability of electron donors is low. This is largely based on a single study of
283 *Alcaligenes faecalis* (40) and has been supported by some studies of complex communities but
284 contested by others (41-43). *A. faecalis* can perform partial denitrification reducing NO_2^- to N_2
285 using NirK, cNor and NosZ clade I, but this organism lacks dissimilatory NO_3^- reductases (Nar or

286 Nap). The study by Schalk-Otto (40), performed in continuous cultures, showed increased N₂O
287 release under low substrate concentrations, and it was suggested that N₂O reductase did not
288 compete successfully with the other reductases for electrons from cytochrome *c*, possibly due
289 to lower affinity for the electron donor. This conclusion needs, however, further verification by
290 more detailed studies of the mechanism involved. Moreover, such studies need to be extended
291 to other groups of denitrifying microorganisms and should include organisms carrying a
292 complete denitrification pathway (thus with Nar and/or Nap). Research over the past decades
293 has revealed diverse denitrification phenotypes among even closely related bacteria, with
294 implications for their accumulation of denitrification intermediate products (15,44).

295 The phenotype described for a range of taxonomically diverse *Bradyrhizobium* strains with a
296 complete denitrification pathway is characterized by a strong preference for N₂O over NO₃⁻
297 when grown in full-strength YMB medium under denitrifying conditions (18,19). The present
298 study provides compelling evidence that this preference is retained when the organisms are
299 starved for carbon and energy, with practically no electron flow to Nap when N₂O was present
300 (Figs. 2A-D). This was not due to too low levels of Nap since the cultures showed well-
301 functioning Nap activity if NO₃⁻ was provided as the only initial electron acceptor (Fig. 3A).
302 Further evidence for this was obtained from the proteomics results, which showed that the
303 cultures, even when grown under more severe starvation conditions using Bioassay 2, were
304 able to produce comparable amounts of Nap, Nir and Nos (Figs. 5E and F). Therefore, the lack of
305 NO₃⁻ reduction during the period when there was N₂O in the system could not be explained by a
306 lack of Nap molecules. Instead, the results suggest the same metabolic-level phenomenon as

307 found for well-fed cultures in this study (Fig. 2A) and earlier (18,19), where NosZ outcompetes
308 Nap for electrons, leaving Nap virtually without electrons so long as exogenous N₂O is available.

309 The attempt to tweak the electron flow toward Nap by the addition of TMPD and ascorbate
310 did not result in measurable NO₃⁻ reduction in the starved cultures (Fig. 4A). A recent study by
311 Mania et al. (18) demonstrated that well-fed *Bradyrhizobium* cells did reduce some NO₃⁻ in the
312 presence of N₂O, if provided with TMPD and ascorbate. This suggested that NosZ (as well as Nir
313 and Nor) received electrons from strongly reduced cytochrome *c*, thus bypassing the electron
314 flow via the *bc1* complex, which allowed Nap to receive electrons from quinol, delivered from
315 the TCA cycle. The specificity of the electron delivery from TMPD to cytochrome *c* cannot be
316 taken for granted, however, and the result by Mania et al. (18) could instead reflect a minimum
317 of electron flow from TPMD to quinon or to NapC, directly, or indirectly. In the present
318 experiment with starved cells, TMPD + ascorbate failed to induce measurable electron flow to
319 Nap in the presence of N₂O (Fig. 4B). This probably reflects the marginal electron flow from the
320 TCA cycle to the quinon/quinol pool due to starvation. A separate experiment supported this,
321 showing that when carbon substrate (YMB) was added to starved cultures, this provided
322 enough electrons to support some Nap activity, although NosZ activity still dominated (Fig. S3).

323 At the same time, the result refutes the concerns regarding unspecific electron delivery of
324 electrons from TMPD to quinon. Functional Nap was apparently present, and NO₃⁻ reduction
325 started when N₂O was almost depleted, with V_{eNap} being 2 fmol e⁻ cell⁻¹ h⁻¹ which was twice as
326 high as $V_{eNir/Nor}$ (Fig. 4B). The latter is as expected, since the reduction of 1 mole of NO₃⁻ to NO₂⁻
327 requires 2 mole electrons, while reduction of NO₂⁻ and NO requires 1 mole electrons.
328 Furthermore, the electron flow to NosZ was the same as to Nir and Nor, suggesting that Nir and

329 NosZ competed equally well for electrons from cytochrome *c*. The maximum total electron flow
330 of 13-14 fmol e⁻ cell⁻¹ h⁻¹ when the cytochrome *c* pool was saturated with electrons from TMPD
331 is likely to be close to the maximum capacity of the denitrification pathway. This electron flow
332 rate is higher than the total electron flow rate in the well-fed cells, which was 8-10 fmol e⁻ cell⁻¹
333 h⁻¹ (Fig. 2C). The factor limiting V_{eNos} to 13-14 fmol cell⁻¹ h⁻¹ is plausibly the rate of electron
334 delivery from cytochrome *c* and/or k_{cat} for Nos.

335 In another experiment we added YMB medium to “slow respiration” phenotypes of the
336 cultures that had been exposed to extended starvation (Figs. 5A, C and E). These cultures had
337 some NosZ activity (V_{eNos} 0.1-0.3 fmol e⁻ cell⁻¹ h⁻¹), while Nap and Nir activities could not be
338 detected. Addition of the electron donor (YMB) led to an immediate upshoot in the activities of
339 all reductases. This could not have been the case if the reductases were not present already,
340 which was proven by the proteomics results. Taken together, the results support that the
341 absolute preference for N₂O over NO₃⁻ was due to competition for electrons, also under severe
342 starvation conditions.

343 It is well known that bacteria have developed a range of physiological responses to tackle
344 starvation. Some survive by forming spores, but most bacteria survive by strongly reduced
345 metabolic rates and minimizing the synthesis of some proteins while upregulating others such
346 as genes for high-affinity transporters, essential repair mechanisms and alternative energy
347 sources (3,45,46). Some such changes have been observed in rhizobia belonging to *Rhizobium*
348 *leguminosarum*, which stayed viable for long periods (55 days) of C starvation (47). Changes in
349 cell size are common during long-term starvation, sometimes leading to the formation of small
350 or even ultra-microcells which may have increased tolerance to antibiotics and other stresses

351 (2,46). Another way for many bacteria, including rhizobia, to survive is to use carbon stored as
352 polyhydroxyalkanoates (PHA) or glycogen, formed during periods of ample nutrient abundance
353 (48). In the present study, the metabolic activity, measured as anaerobic respiration rate,
354 decreased to between 1 and 18% that of well-fed cultures, depending on which starvation
355 bioassay was used. We did not, however, detect any obvious decrease in cell size when
356 comparing cells starved for 24 h, although cell morphologies were slightly altered (Figs. S1A and
357 B). PHA was observed in the starved cells as well as well-fed ones, as shown with Nile red
358 staining (Fig. S1A) and no obvious reduction in PHA could be observed during starvation from
359 our assays. Since this was a comparably short period of starvation, it is conceivable that the
360 cells would make use of the stored carbon if the starvation was prolonged. It could be
361 speculated that one reason for not using the PHA reserves (and glycogen) immediately, is that
362 these are saved to be used during bacteroid formation (48). To clarify these issues, more
363 detailed studies are however needed.

364 A striking separation into two phenotypes during starvation, as shown in Fig. 5, was
365 observed in repeated experiments (Bioassay 2), each time with about two thirds of the flasks
366 showing a “fast” and the others “slow” respiration: $1.0\text{-}1.8 \text{ fmol e}^- \text{ cell}^{-1} \text{ h}^{-1}$ vs $0.1\text{-}0.3 \text{ fmol e}^-$
367 $\text{cell}^{-1} \text{ h}^{-1}$, respectively. Phenotypic heterogeneity has been observed in single-strain cultures of
368 various bacteria when exposed to carbon substrate deficiency and may, or may not, be due to
369 mutations during starvation for one week or more (3,49). Mutations in the entire population in
370 several replicate flasks are unlikely and cannot, however, have caused the rapid diversification
371 into “slow”- or “fast” respiration in the present study. It may instead reflect a stochastic
372 phenomenon, or that the culture contained different subpopulations. We speculated that the

373 “fast respiration” phenotype may be due to a fraction of the cells dying, allowing the other cells
374 to survive on nutrients released from lysed cells, as seen for other bacteria (3). However, this
375 would require lysis of a substantial fraction of the cells, which is refuted by the observation that
376 the flasks with “slow” and “fast” phenotypes had practically identical numbers of viable cells
377 (Fig. S2). Thus, further studies are needed to understand this phenomenon of different
378 respiration rates.

379 Although the starvation bioassays developed for this study cannot be regarded as a close
380 mimicking of the conditions in natural environments, it is conceivable from the experiments
381 that these organisms are potentially strong sinks for N₂O when living in soil under fluctuating
382 availability of carbon substrate. Bioassay 1 is probably closer to a “real-world” situation than
383 Bioassay 2, since it is likely that denitrifying bacteria experience regular fluctuations in oxygen
384 and thus are not devoid of denitrification reductases if they enter starvation. On the other hand,
385 Bioassay 2, where the cells had to produce the denitrification proteome in the absence of
386 external electron donors (C-substrate), showed that even under these conditions N₂O reduction
387 strongly dominated over NO₃⁻ reduction.

388

389 **Materials and methods**

390 **Bacterial strain and culture preparations.** *Bradyrhizobium* strain HAMBI 2125, originally
391 isolated from nodules of *Arachis hypogaea* growing in Sichuan, China (50), was used in all
392 experiments. This strain, which is closely related to *Bradyrhizobium ottawaense*, contains the
393 genetic set-up for complete denitrification (19). A culture was raised from one single colony
394 after streaking on agar plates. After checking the purity by sequencing the 16S rRNA gene (19),

395 portions were preserved in 15% glycerol at -80 °C. Cultures for all the experiments were started
396 from the -80 °C stocks and raised under fully oxic conditions in 120 ml serum flasks containing
397 50 ml Yeast Mannitol Broth (YMB): 10 g l⁻¹ D-Mannitol, 0.5 g l⁻¹ K₂HPO₄, 0.2 g l⁻¹ MgSO₄·7H₂O,
398 0.1 g l⁻¹ NaCl and 0.5 g l⁻¹ yeast extract (51). All incubations were done at 28 °C. Medical flasks
399 (120 ml) were used in all experiments. A magnet in each flask secured vigorous stirring (600-
400 700 rpm) to avoid cell aggregation and to optimize the gas exchange between liquid and gas
401 phases (18). To prevent that the cells experienced anoxia during this oxic incubation, and thus
402 to avoid de novo synthesis of denitrification reductases, portions were regularly transferred to
403 new flasks containing fresh medium, so that the OD₆₀₀ was never allowed to exceed 0.1 (19).
404 These aerobically grown cultures were used as inoculants in the “starvation bioassays”
405 described below.

406 Flasks for cultures incubated under hypoxic conditions were prepared as described in (18).
407 Briefly, flasks (120 ml) containing 50 ml buffer (or medium) were capped with sterilized, gas
408 tight butyl rubber septa (Matrix AS, Norway). The air was removed by applying vacuum
409 repeatedly (6 x 360 s) and He was then filled for 30 s, after which the overpressure was
410 released. The flasks were left for two days to equilibrate the gases between the headspace and
411 liquid (52). Then, 0.7 ml or 1 ml O₂ (equal to 1 or 1.5 vol %) and 1 ml N₂O (70 to ~80 μmol N
412 flask⁻¹) was injected into the headspace and sterile filtered solutions of KNO₃ (and sometimes
413 KNO₂) were added to the liquid reaching initial, desired concentrations (0.25 -1 mM).

414

415 **Starvation bioassays.** Starvation bioassays were established, which followed one of the
416 procedures described in Fig. 1. In Bioassay 1 “Mild starvation”, cultures were allowed to make

417 the transition to denitrification in full-strength YMB before being exposed to starvation. Oxically
418 grown, well-fed cultures were incubated for 48 h after which the headspace was replaced by He
419 and 1% O₂, and 1 mM KNO₃ was added to the medium. When O₂ was depleted, the cultures
420 were centrifuged (10 000 × g at 4 °C for 10 min) and washed twice in sterile ddH₂O. The pellets
421 (triplicates) were pooled to reduce bias in the form of variations due to the
422 centrifugation/washing. Each pellet was divided into three and used to inoculate anoxic flasks
423 containing 50 ml C-free buffer supplemented with 1 mM KNO₃ and 0.25 or 0.5 mM KNO₂ and
424 with He and/or 1 ml N₂O (around 80 μmol N flask⁻¹) in the headspace. When incubated in the
425 buffer, the respiration rate of the cultures was 10-18% that of well-fed cultures during the first
426 15 h, then it decreased to about 4%.

427 In Bioassay 2 (“extended starvation”), denitrification was instead induced after having
428 exposed the cells to starvation for 20 h. The cultures were raised in fully oxic flasks containing
429 YMB medium. When OD₆₀₀ reached ~0.1, the cultures were centrifuged (10 000 × g at 4 °C for
430 10 mins) and washed twice in sterile ddH₂O. The pellets (triplicates) were pooled after which
431 they were evenly divided and used to inoculate fully oxic flask containing C-free “starvation
432 buffer”. These cultures were incubated for 20 h, then centrifuged after which the pellets were
433 pooled and divided evenly before being inoculated into flasks containing C-free “starvation
434 buffer” with 1 mM KNO₃, and with He and 1 ml N₂O (around 80 μmol N flask⁻¹) in headspace.
435 The respiration rate of cultures exposed to “extended starvation” was ca 1-4% compared to
436 that of well-fed cultures. Results from the different starvation bioassays were compared to
437 those from well-fed cultures. To avoid biases, the treatments, which were in all cases set up at
438 least in triplicates (n ≥ 3), were the same regarding centrifugations and washings until the last

439 step when pooled cells were inoculated to anoxic flasks containing either YMB or buffer. The
440 carbon source in the YMB medium comprised >200 times surplus of electron donor compared
441 to electron acceptors throughout all incubations, thus ensuring that the electron donor was not
442 depleted. All cultures were incubated at 28°C, and with vigorous stirring (600-700 rpm).

443
444 **Addition of YMB medium or TMPD as electron donors to starved cultures.** Experiments
445 were performed to investigate how starved cultures of *Bradyrhizobium* strain HAMBI 2125
446 responded to the addition of an electron donor, either provided as an artificial electron donor
447 (Fig. 4) or as YMB medium (2 ml mannitol /yeast solution providing them with a substrate
448 concentration corresponding to half-strength YMB, thus 5 g mannitol and 0.25 g yeast l⁻¹, Fig. 5).
449 As artificial electron donor we used sodium TMPD (N, N, N', N'-tetramethyl-p-
450 phenylenediamine) which, in the presence of ascorbate, donates electrons to cytochrome c
451 thus providing electrons to Nir and NosZ (18,37). Ascorbate and TMPD (both from Sigma-
452 Aldrich®, Germany) were dissolved in ddH₂O or 96% ethanol, respectively, and filter sterilized.
453 The solutions were added to the incubation flasks 10-15 mins before gas sampling. The effect of
454 different concentrations of TMPD (100, 250, and 500 µM in the culture buffer) combined with
455 10 mM ascorbate on N₂O reduction was checked prior to the main experiments. This showed
456 that 500 µM TMPD had an obvious inhibition effect on N₂O reduction, while 100 µM and 250
457 µM showed no inhibitory effect (not shown). To minimize other, possible effects we used 100
458 µM TMPD for the experiments (as done in Mania et al. (18)). In another experiment, shown in
459 Fig. S3, 4 ml of a mannitol/yeast solution was added to flasks containing 50 ml starved cultures,

460 providing them with a substrate concentration corresponding to full strength YMB (10 g
461 mannitol and 0.5 g yeast l⁻¹).

462

463 **Monitoring of gas kinetics, NO₃⁻ and NO₂⁻ concentrations and electron flow rates.** The
464 culture flasks were placed in a robotized incubation system and the headspace gas was sampled
465 frequently for N₂, N₂O, NO and O₂ measurements (34). Gas losses caused by sampling were
466 taken into account when calculating the production and consumption of gases, as described by
467 Molstad et al. (34) and Mania et al. (18). The concentration of O₂ in the anoxic incubation flasks
468 was <0.6 μM at the start of the experiment, which was well below the level for initiating
469 denitrification (4.6 μM; (19)).

470 The NO₂⁻ concentrations were monitored as described in Mania et al. (18). Briefly, samples
471 (0.1- 0.5 ml) were taken every 1 or 2 hours (n=3) from the liquid phase through the septum of
472 the flasks using a sterile syringe. To avoid that the gas kinetics was affected by the sampling, a
473 set of flasks was dedicated to NO₂⁻ measurements and a parallel set was left untouched for gas
474 measurements. NO₂⁻ was determined using a chemoluminescence NOx analyzer (Sievers NOA™
475 280i, GE Analytical Instruments) after at first reducing the NO₂⁻ to NO by adding 10 μl liquid
476 sample into a purging device containing a reducing agent (50% acetic acid with 1% (w/v) NaI)
477 (53,54). The NO₂⁻ concentrations were determined against a standard curve (range 0-2 mM
478 NO₂⁻; r²= 0.999).

479 The cell specific electron flow rates (V_e , mol e⁻ cell⁻¹ h⁻¹) for each time increment between
480 two gas samplings were calculated as $V_{eflask}(t)/(N(0)+E_{cum}(t)*Y)$, where $V_{eflask}(t)$ is the electron
481 flow rate in the flask (mol e⁻ flask⁻¹ h⁻¹) calculated from measurements, $N(0)$ is number of cells in

482 the flask at time=0, $E_{cum}(t)$ is the cumulated electron flow ($\text{mol e}^- \text{flask}^{-1}$) at time=t, and Y is the
483 yield per mol electron ($\text{cells mol}^{-1} \text{e}^-$) for HAMBI 2125 as measured previously (19). The
484 conversion factor $5.8\text{E}8 \text{ cells ml}^{-1} * \text{OD}_{600}^{-1}$ was used to convert OD_{600} to cell numbers (19).

485

486 **Proteomics.** The abundances of Nap, Nir and NosZ were quantified in starved cultures
487 treated as described for Bioassay 2 (extended starvation). Altogether eighteen flasks were
488 prepared according to Fig. 1B. After the 20 h aerobic incubation in C-free buffer, three flasks
489 were harvested for proteomics analysis and, following centrifugation/washing, the three cell
490 pellets from these flasks were frozen individually at -20°C . The cultures in the other flasks were
491 pooled three by three and after centrifugation/washing, each cell pellet was divided in three
492 and used to inoculate new flasks containing C-free buffer with NO_3^- in the liquid and with He
493 and 1 ml N_2O in the headspace (Fig. 1B). These flasks (fifteen in total) were placed in the robotic
494 incubation system for monitoring of gas kinetics and NO_2^- concentrations as described above.
495 The entire culture volume (50 ml) was harvested from each of triplicate flasks at different time
496 points during the anoxic incubation. The cultures in six of the flasks showed a “fast” respiration
497 rate (total electron flow $1.0\text{-}1.8 \text{ e}^- \text{ cell}^{-1} \text{ h}^{-1}$) and were harvested at 5 and 20 h of incubation in
498 anoxic buffer (triplicates at each sampling point). The cultures in the other nine flasks showed a
499 “slow” respiration rate (maximum total electron flow rate was $0.27 \text{ fmol e}^- \text{ cell}^{-1} \text{ h}^{-1}$). Six of
500 them were harvested in triplicates after 10 and 20 h in anoxic buffer. The remaining three flasks
501 from the cultures with “slow” respiration rate received a portion of YMB at 24.8 h and were
502 harvested at 27 h. Harvested cell cultures were centrifuged and stored as pellets at -20°C . The
503 protein extraction was as described in Gao et al. (19). Briefly, the thawed cell pellets were

504 resuspended in lysis buffer (20 mM Tris–HCl pH 8, 0.1% v/v Triton X-100, 200 mM NaCl, 1 mM
505 DTT). They were then subjected to bead beating (3 × 45 s) with glass beads (particle size ≤ 106
506 μm; Sigma) using a MP Biomedicals™ FastPrep-24™ (Thermo Fischer Scientific) at maximum
507 power and with cooling on ice between the cycles. After centrifugation to remove cell debris
508 (10 000 × g; 5 min), the supernatant, containing water soluble proteins, was used for proteomic
509 analysis using an Orbitrap mass spectrometer (described in (19,55)). Quantification was based
510 on LFQ (label-free quantification) in MaxQuant (56), and the relative abundance of the
511 individual reductases was calculated as percentages of the sum of all protein abundances for
512 each time point. The nitric oxide reductases NorB/C were not measured since only a small
513 fraction of these membrane-bound enzymes can be accurately obtained with this extraction
514 protocol (approximately 1/1000th the quantity of the other reductases).

515

516 **Viable counts and microscopy.** The number of viable cells in starved cultures showing slow
517 and fast respiration rates, observed in flasks prepared according to Bioassay 2, was determined
518 by plating dilutions of the cultures on yeast-mannitol agar (YMA). The morphology of cells from
519 well-fed cultures and starved cultures was compared using phase contrast microscopy and a
520 qualitative determination (presence/absence) of PHA was done by staining with Nile Red
521 (Sigma-Aldrich) followed by fluorescence microscopy. Microscopy was performed on a Zeiss
522 AxioObserver with an Orca-Flash4.0 CMOS camera (Hamamatsu Photonics) controlled by the
523 ZEN Blue software. Images were taken with a 100x phase contrast objective. A HPX-120
524 Illuminator was used as light source for fluorescence microscopy. Images were prepared using
525 ImageJ and analysis of cell sizes was done using the ImageJ-plugin MicrobeJ (57).

526

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533

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693

694 **Figure legends**

695 **FIG 1** Bioassays for assessing the effect of starvation on electron flow to denitrification enzymes.

696 A: Bioassay 1: starvation of cells with a previously expressed denitrification proteome. Cells

697 were raised from stock cultures in fully oxic flasks containing YMB medium. When OD₆₀₀

698 reached ~ 0.1 , the flasks were made anoxic by He-washing, then supplemented with 1% O₂ in
699 the headspace and 1 mM NO₃⁻ in the liquid. The cultures were allowed to deplete the O₂ and to
700 initiate denitrification when growing in YMB medium, after which they were centrifuged (10
701 000 × g at 4 °C for 10 mins) and washed twice in sterile ddH₂O. The pellets from triplicate flasks
702 were pooled and used to inoculate flasks containing either C-free buffer or YMB medium (well-
703 fed control), in both cases supplemented with 1 mM KNO₃ and 0.25 or 0.5 mM KNO₂, and with
704 He and 1 ml N₂O (around 80 μmol N flask⁻¹) in headspace. The starving cells, incubated in buffer,
705 had a low respiratory electron flow rate (mol electrons cell⁻¹ h⁻¹), initially being 10-18 % that of
706 the well-fed cultures, and then decreasing to reach around 4% after 20 h. Results from
707 experiments using Bioassay 1 are presented in Figs. 2, 3 and 4.

708 B: Bioassay 2: denitrification induced during starvation. Cells were raised from stock cultures in
709 fully oxic flasks containing YMB medium. When OD₆₀₀ reached ~ 0.1 , the cultures were
710 centrifuged (10 000 × g at 4 °C for 10 mins) and washed twice in sterile ddH₂O. The pellets from
711 triplicate flasks were pooled, after which they were evenly divided and used to inoculate fully
712 oxic flasks containing C-free buffer. These cultures were incubated for 20 h, then centrifuged
713 after which the pellets were pooled and divided evenly before being inoculated into flasks
714 containing C-free buffer-provided with 1 mM KNO₃, and with He and 1 ml N₂O (around 80 μmol
715 N flask⁻¹) in headspace. The respiration rate (mol electrons cell⁻¹ h⁻¹) of the starving cultures was
716 1-4% compared to that of well-fed cultures. All cultures in Bioassays 1 and 2 were incubated at
717 28°C, and with vigorous stirring (650 rpm). Results from experiments using Bioassay 1 are
718 presented in Figs. 5 and S2.

719

720 **FIG 2** Denitrification kinetics as affected by starvation in cultures with a complete denitrification
721 proteome (Bioassay 1). Cells were allowed to develop a full denitrification proteome under well-
722 fed conditions and were then washed twice in buffer prior to inoculation to flasks with YMB
723 (well-fed control) or buffer (starved), supplemented with 1 mM KNO_3 and 0.25 mM KNO_2 , and
724 with He plus 1 ml N_2O (around 80 $\mu\text{mol N}$) in headspace. A larger inoculum was given to the
725 flasks with buffer ($9.9\text{E}+08$ cells flask $^{-1}$) than to the flasks with YMB ($5.0\text{E}+09$ cells flask $^{-1}$), to
726 secure measurable activity in the starved cells and adequate time resolution of the
727 denitrification kinetics in the well-fed cells. Panels A and B show the denitrification kinetics in
728 well-fed and starved cultures, respectively. The flasks were practically anoxic from the start with
729 an initial O_2 concentration in the liquid of <0.52 μM which decreased to approximately 0 μM
730 (insets in A & B). Panels C and D show the cell specific electron flow rate. $V_{e \text{ total den}}$ designates
731 the total electron flow to the denitrification reductases and $V_{e \text{ total incl O}_2}$ the total electron flow,
732 including that to denitrification and to O_2 . The electron flow rate to the individual reductases is
733 designated as $V_{e\text{Nap}}$, $V_{e\text{Nir}}$, $V_{e\text{Nor}}$ and $V_{e\text{Nos}}$. $V_{e\text{Nir}}$ and $V_{e\text{Nor}}$ were practically identical and cannot be
734 distinguished from one another in the figure. Inserted panels show $V_{e\text{Nap}}$ throughout, including
735 negative values which are due to slight errors in determination of N_2 and N_2O ($V_{e\text{Nap}}$ was
736 calculated by N-mass balance). Bars in all graphs show standard deviation ($n=3$).

737

738 **FIG 3** Bioassay 1 with single nitrogen oxides. Experimental conditions as for Fig. 2, but the
739 starving cells (in buffer) were provided with either NO_3^- , NO_2^- or N_2O in individual flasks ($n=3$ for
740 each treatment). Panels A-C show the gas kinetics in flasks provided with 1 mM NO_3^- (A), 0.5
741 mM NO_2^- (B), or 70 $\mu\text{mol N}_2\text{O-N}$ added to the headspace (C). Panel D shows the cell-specific

742 electron flow rate (V_e) measured in buffer supplemented with NO_3^- , NO_2^- or N_2O ; the O_2
743 concentration (left y-axis); and V_e as percentage of the rates in well-fed cultures (right y-axis).
744 The number of cells inoculated into the incubation flasks at the start (0 h of incubation in anoxic
745 buffer) was $3.6\text{E}+09$ for all treatments. Bars show standard deviation ($n=3$).

746

747 **FIG 4** Denitrification in starved cells after addition of TMPD as an external electron donor.
748 Preparation of the cultures followed Bioassay 1 (Fig. 1A), except that $100\ \mu\text{M}$ TMPD and $10\ \text{mM}$
749 ascorbate were injected into the flasks with starving cells 15 min before the first gas sampling.
750 Each flask was inoculated with $5.28\text{E}+08$ cells ($n=4$ replicate flasks). The initial O_2 concentration
751 was $0.35\ \mu\text{M}$ and decreased to approximately 0 within 5 h (inset in A). Bars show standard
752 deviation ($n=4$). Panel A: kinetics of NO_2^- , NO, N_2O and N_2 (O_2 in inserted panel). Panel B:
753 Calculated cell specific electron flow rate (V_e , $\text{fmol cell}^{-1}\ \text{h}^{-1}$) to each of the reductases Nap, Nir,
754 Nor, and NosZ, and the total electron flow rate ($V_{e\text{total}}$). The electron flow to Nir was practically
755 identical to the electron flow to Nor (miniscule amounts of NO), and the two are shown as a
756 single graph ($V_{e\text{Nir/Nor}}$). Inset plots show $V_{e\text{Nap}}$ throughout, including negative values which are
757 due to slight errors in determination of N_2 and N_2O ($V_{e\text{Nap}}$ was calculated by N-mass balance).

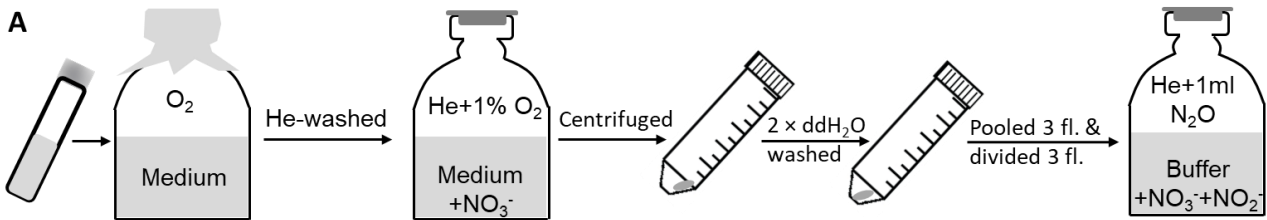
758

759 **FIG 5** Bioassay 2: stochasticity of starvation response, response to input of organic C and
760 quantification of denitrification enzymes. Altogether fifteen flasks were prepared following
761 Bioassay 2 (see Fig. 1B). All flasks were anoxic ($<0.5\ \mu\text{M}$ at the start of the incubation) and
762 contained 50 ml buffer supplemented with $1\ \text{mM}$ KNO_3 and with He plus 1 ml N_2O in the
763 headspace. The NO_2^- concentrations, measured during the first 5 hours after incubation in the

764 buffer, were approximately $0.59 \pm 0.20 \mu\text{M}$ (not shown). The cultures separated into two distinct
765 phenotypes: nine flasks had “slow” respiration (panel A) and 6 flasks had “fast” respiration
766 (panel B). The electron flow to the individual reductases ($\text{fmol e}^- \text{cell}^{-1} \text{h}^{-1}$) for the flasks with
767 slow and fast respiration are shown in panels C and D, respectively. Cultures with “slow”
768 respiration rate had a total electron flow rate of maximum $0.27 \text{ e}^- \text{cell}^{-1} \text{h}^{-1}$ (C) and cultures with
769 “fast” respiration rate had a total electron flow rate of $1.0\text{-}1.8 \text{ fmol e}^- \text{cell}^{-1} \text{h}^{-1}$ (D). The inset plot
770 in panel C shows the electron flow to the individual denitrification reductases after the carbon
771 addition. The cultures (entire flasks, 50 ml) were sampled for proteomics analyses at different
772 time points (panels E and F). At timepoints throughout, marked by dashed vertical lines in
773 panels A and B, three flasks were harvested for proteomics analysis (including 0 h). The six flasks
774 with “fast” respiration were harvested for proteomics analysis at 5 and 20 h of incubation in
775 these anoxic buffers (triplicates at each sampling point). Of the nine flasks with “slow”
776 respiration, three were harvested at 10 and three at 20 h. The remaining three “slow”
777 respiration flasks were supplemented with YMB at 24.8 h, at a concentration which made the
778 buffer a half-strength YMB medium, containing 5 g mannitol l^{-1} and 0.25 yeast extract l^{-1} , and
779 then harvested for proteomics at 27 h. The gas measurements and electron flows are averages
780 from the three flasks ($n=3$) for each phenotype, that were left untouched until the end of the
781 incubation when they were sampled for proteomics. Proteomics analyses were done for
782 triplicate flasks ($n=3$) at each sampling point. Standard deviations are indicated as bars in all
783 graphs (in several cases not visible due to low variation).

784

Bioassay 1: Denitrification induced before starvation



Bioassay 2: Denitrification induced during starvation

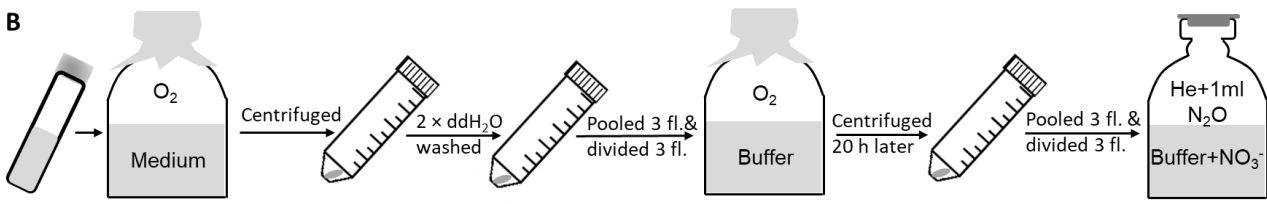


FIG 1 Bioassays for assessing the effect of starvation on electron flow to denitrification enzymes.

A: Bioassay 1: starvation of cells with a previously expressed denitrification proteome. Cells were raised from stock cultures in fully oxic flasks containing YMB medium. When OD₆₀₀ reached ~0.1, the flasks were made anoxic by He-washing, then supplemented with 1% O₂ in the headspace and 1 mM NO₃⁻ in the liquid. The cultures were allowed to deplete the O₂ and to initiate denitrification when growing in YMB medium, after which they were centrifuged (10 000 × g at 4 °C for 10 mins) and washed twice in sterile ddH₂O. The pellets from triplicate flasks were pooled and used to inoculate flasks containing either C-free buffer or YMB medium (well-fed control), in both cases supplemented with 1 mM KNO₃ and 0.25 or 0.5 mM KNO₂, and with He and 1 ml N₂O (around 80 μmol N flask⁻¹) in headspace. The starving cells, incubated in buffer, had a low respiratory electron flow rate (mol electrons cell⁻¹ h⁻¹), initially being 10-18 % that of the well-fed cultures, and then decreasing to reach around 4% after 20 h. Results from experiments using Bioassay 1 are presented in Figs. 2, 3 and 4.

B: Bioassay 2: denitrification induced during starvation. Cells were raised from stock cultures in fully oxic flasks containing YMB medium. When OD₆₀₀ reached ~0.1, the cultures were centrifuged (10 000 × g at 4 °C for 10 mins) and washed twice in sterile ddH₂O. The pellets from triplicate flasks were pooled, after which they were evenly divided and used to inoculate fully oxic flasks containing C-free buffer. These cultures were incubated for 20 h, then centrifuged after which the pellets were pooled and divided evenly before being inoculated into flasks containing C-free buffer-provided with 1 mM KNO₃, and with He and 1 ml N₂O (around 80 μmol N flask⁻¹) in headspace. The respiration rate (mol electrons cell⁻¹ h⁻¹) of the starving cultures was 1-4% compared to that of well-fed cultures. All cultures in Bioassays 1 and 2 were incubated at 28°C, and with vigorous stirring (650 rpm). Results from experiments using Bioassay 1 are presented in Figs. 5 and S2.

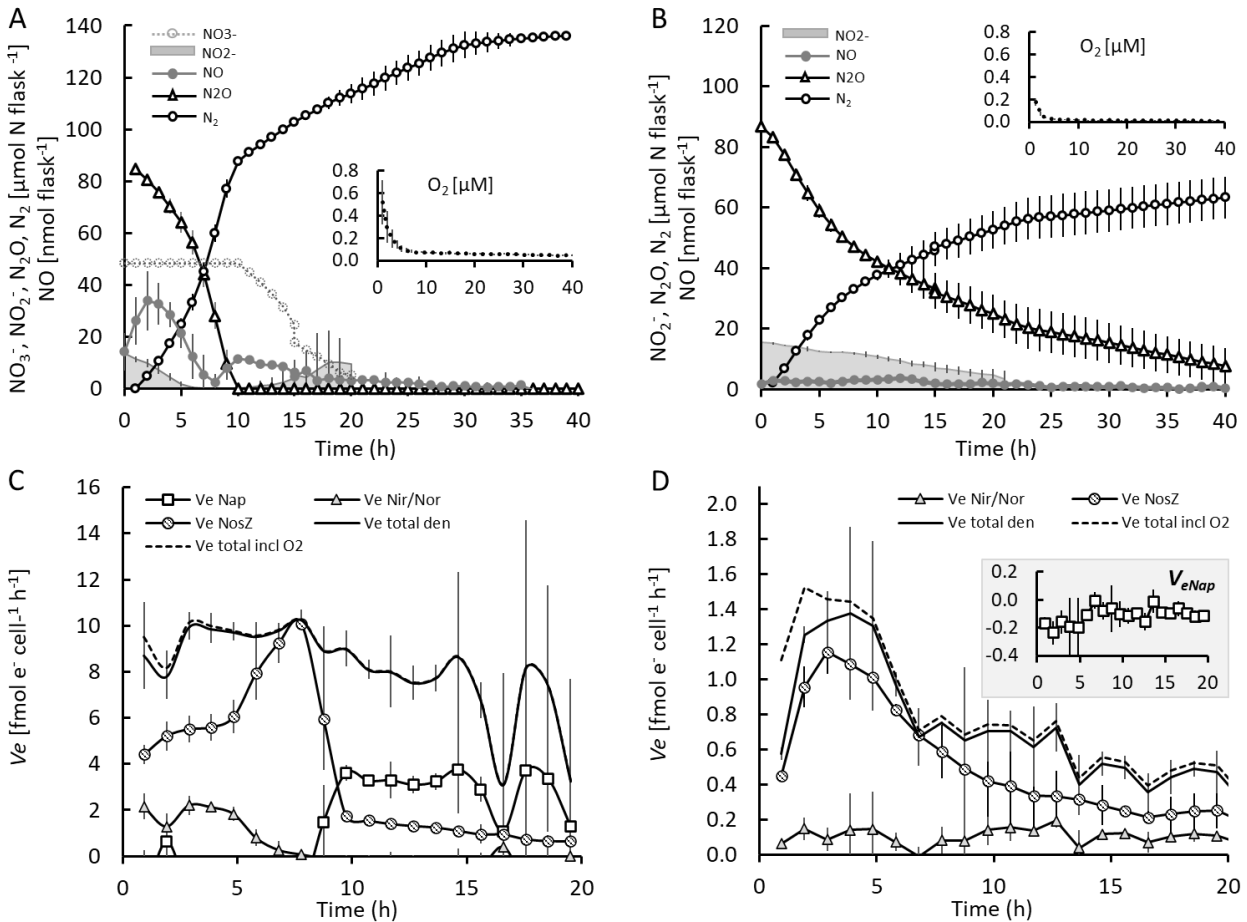


FIG 2 Denitrification kinetics as affected by starvation in cultures with a complete denitrification proteome (Bioassay 1). Cells were allowed to develop a full denitrification proteome under well-fed conditions and were then washed twice in buffer prior to inoculation to flasks with YMB (well-fed control) or buffer (starved), supplemented with 1 mM KNO_3 and 0.25 mM KNO_2 , and with He plus 1 ml N_2O (around 80 $\mu\text{mol N}$) in headspace. A larger inoculum was given to the flasks with buffer ($9.9\text{E}+08$ cells flask^{-1}) than to the flasks with YMB ($5.0\text{E}+09$ cells flask^{-1}), to secure measurable activity in the starved cells and adequate time resolution of the denitrification kinetics in the well-fed cells. Panel A and Panel B show the denitrification kinetics in well-fed and starved cultures, respectively. The flasks were practically anoxic from the start with an initial O_2 concentration in the liquid of $<0.52 \mu\text{M}$ which decreased to approximately $0 \mu\text{M}$ (insets in A & B). Panel C and Panel D show the cell specific electron flow rate. $V_{e \text{ total den}}$ designates the total electron flow to the denitrification reductases and $V_{e \text{ total incl O}_2}$ the total electron flow, including that to denitrification and to O_2 . The electron flow rate to the individual reductases is designated as $V_{e \text{ Nap}}$, $V_{e \text{ Nir}}$, $V_{e \text{ Nor}}$ and $V_{e \text{ NosZ}}$. $V_{e \text{ Nir}}$ and $V_{e \text{ Nor}}$ were practically identical and cannot be distinguished from one another in the figure. Inserted panels show $V_{e \text{ Nap}}$ throughout, including negative values which are due to slight errors in determination of N_2 and N_2O ($V_{e \text{ Nap}}$ was calculated by N-mass balance). Bars in all graphs show standard deviation ($n=3$).

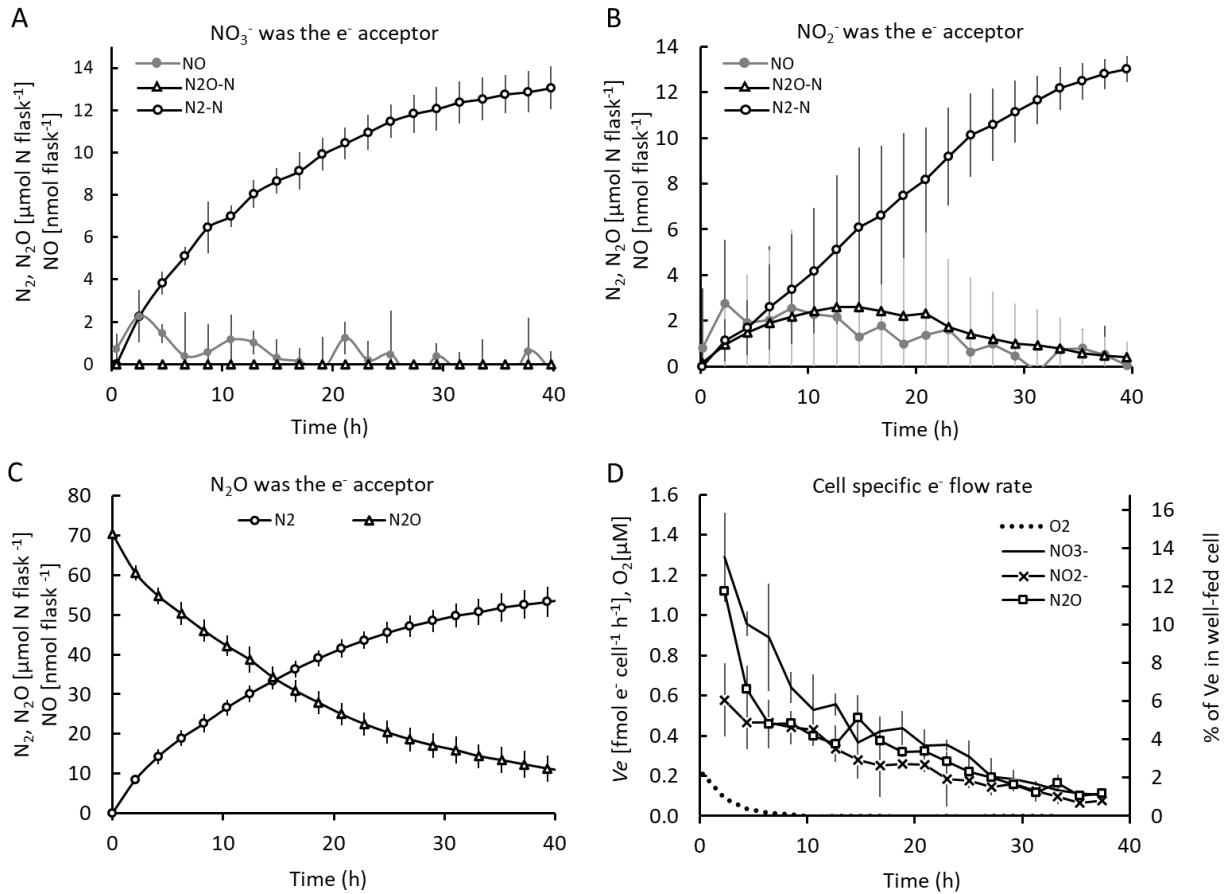


FIG 3 Bioassay 1 with single nitrogen oxides. Experimental conditions as for Fig. 2, but the starving cells (in buffer) were provided with either NO_3^- , NO_2^- or N_2O in individual flasks ($n=3$ for each treatment). Panel A-C show the gas kinetics in flasks provided with 1 mM NO_3^- (A), 0.5 mM NO_2^- (B), or 70 $\mu\text{mol N}_2\text{O-N}$ added to the headspace (C). Panel D shows the cell-specific electron flow rate (V_e) measured in buffer supplemented with NO_3^- , NO_2^- or N_2O ; the O_2 concentration (left y-axis); and V_e as percentage of the rates in well-fed cultures (right y-axis). The number of cells inoculated into the incubation flasks at the start (0 h of incubation in anoxic buffer) was $3.6\text{E}+09$ for all treatments. Bars show standard deviation ($n=3$).

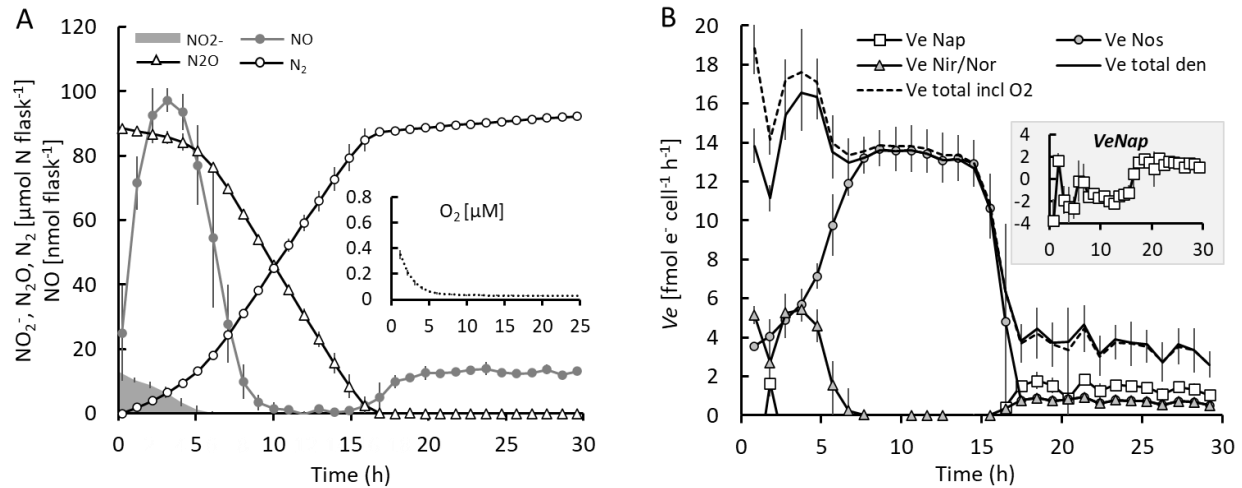


FIG 4 Denitrification in starved cells after addition of TMPD as an external electron donor. Preparation of the cultures followed Bioassay 1 (Fig. 1A), except that 100 μM TMPD and 10 mM ascorbate were injected into the flasks with starving cells 15 min before the first gas sampling. Each flask was inoculated with $5.28\text{E}+08$ cells ($n=4$ replicate flasks). The initial O_2 concentration was 0.35 μM and decreased to approximately 0 within 5 h (inset in A). Bars show standard deviation ($n=4$). Panel A: kinetics of NO_2^- , NO , N_2O and N_2 (O_2 in inserted panel). Panel B: Calculated cell specific electron flow rate (V_e , $\text{fmol cell}^{-1} \text{h}^{-1}$) to each of the reductases Nap, Nir, Nor, and NosZ, and the total electron flow rate ($V_{e\text{total}}$). The electron flow to Nir was practically identical to the electron flow to Nor (miniscule amounts of NO), and the two are shown as a single graph ($V_{e\text{Nir/Nor}}$). Inset plots show $V_{e\text{Nap}}$ throughout, including negative values which are due to slight errors in determination of N_2 and N_2O ($V_{e\text{Nap}}$ was calculated by N-mass balance).

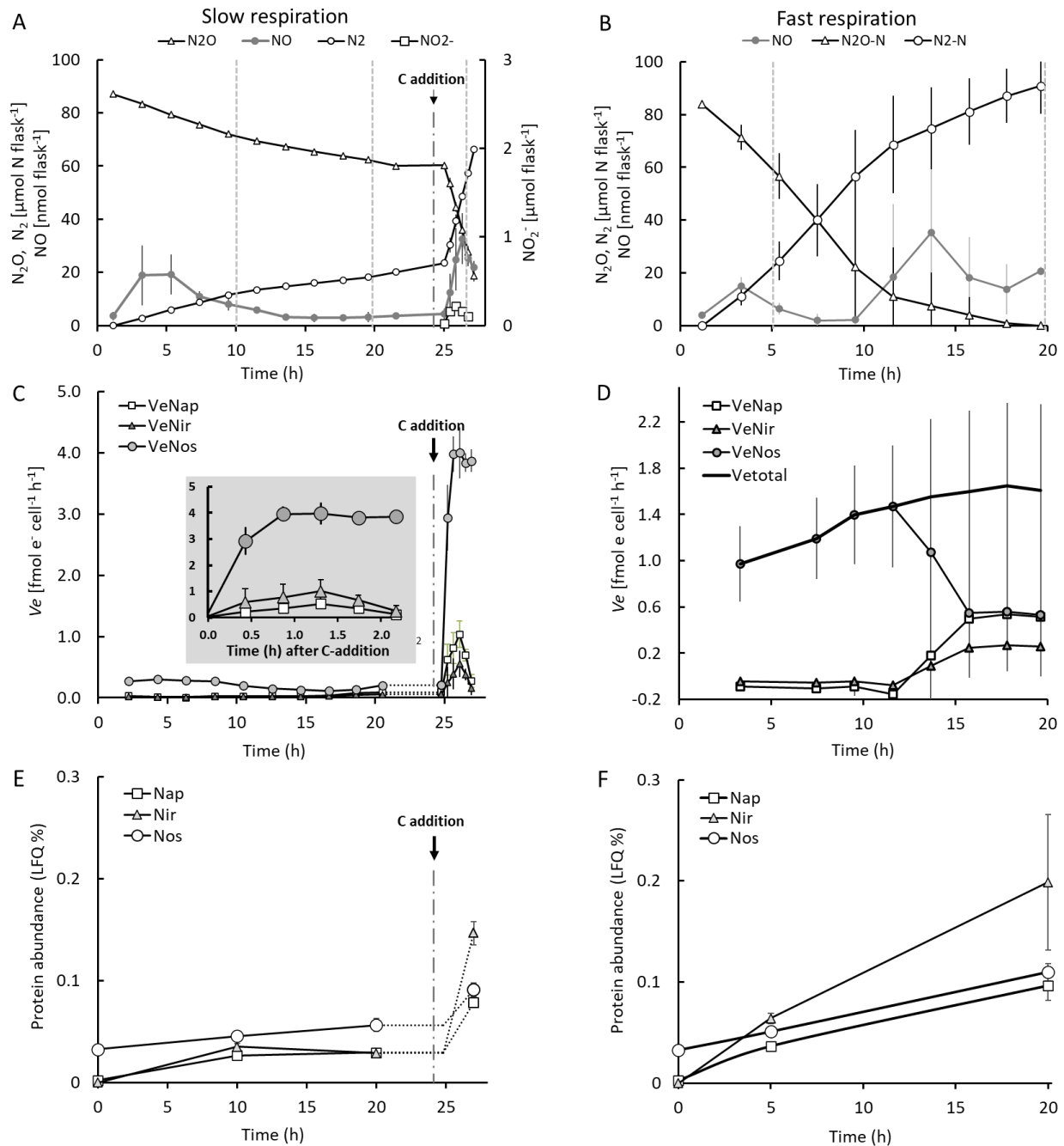


FIG 5 Bioassay 2: stochasticity of starvation response, response to input of organic C and quantification of denitrification enzymes. Altogether fifteen flasks were prepared following Bioassay 2 (see Fig. 1B). All flasks were anoxic ($<0.5 \mu\text{M}$ at the start of the incubation) and contained 50 ml buffer supplemented with 1 mM KNO_3 and with He plus 1 ml N_2O in the headspace. The NO_2^- concentrations, measured during the first 5 hours after incubation in the buffer, were approximately $0.59 \pm 0.20 \mu\text{M}$ (not shown). The cultures separated into two distinct phenotypes: nine flasks had “slow” respiration (panel A) and 6 flasks had “fast” respiration (panel B). The electron flow to the individual reductases ($\text{fmol e}^- \text{cell}^{-1} \text{h}^{-1}$) for the flasks with slow and fast respiration are shown in panels C and D, respectively. Cultures with “slow” respiration rate had a total electron flow rate of maximum $0.27 \text{ e}^- \text{cell}^{-1} \text{h}^{-1}$ (C) and cultures with “fast” respiration rate had

a total electron flow rate of 1.0-1.8 fmol e⁻ cell⁻¹ h⁻¹ (D). The inset plot in panel C shows the electron flow to the individual denitrification reductases after the carbon addition. The cultures (entire flasks, 50 ml) were sampled for proteomics analyses at different time points (panels E and F). At timepoints throughout, marked by dashed vertical lines in panels A and B, three flasks were harvested for proteomics analysis (including 0 h). The six flasks with “fast” respiration were harvested for proteomics analysis at 5 and 20 h of incubation in these anoxic buffers (triplicates at each sampling point). Of the nine flasks with “slow” respiration, three were harvested at 10 and three at 20 h. The remaining three “slow” respiration flasks were supplemented with YMB at 24.8 h, at a concentration which made the buffer a half-strength YMB medium, containing 5 g mannitol l⁻¹ and 0.25 yeast extract l⁻¹, and then harvested for proteomics at 27 h. The gas measurements and electron flows are averages from the three flasks (n=3) for each phenotype, that were left untouched until the end of the incubation when they were sampled for proteomics. Proteomics analyses were done for triplicate flasks (n=3) at each sampling point. Standard deviations are indicated as bars in all graphs (in several cases not visible due to low variation).