1	Differential effects of cobalt ions <i>in vitro</i> on gill (Na ⁺ , K ⁺)-ATPase kinetics
2	in the blue crab Callinectes danae (Decapoda, Brachyura)
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21	Running title : Cobalt ions and gill (Na ⁺ , K ⁺)-ATPase kinetics
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23	Keywords : Cobalt ions; K ⁺ -phosphatase activity; <i>Callinectes danae</i> ; gill (Na ⁺ ,K ⁺)-ATPase; <i>p</i> -
24	nitrophenylphosphate
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33 Abstract

34 To evaluate the crustacean gill (Na^+, K^+) -ATPase as a molecular marker for toxic 35 contamination by heavy metals of estuarine and coastal environments, we provide a comprehensive analysis of the effects of Co²⁺ in vitro on modulation of the K⁺-phosphatase 36 37 activity of a gill (Na⁺, K⁺)-ATPase from the blue crab *Callinectes danae*. Using *p*-nitrophenyl phosphate as a substrate, Co^{2+} can act as both stimulator and inhibitor of K⁺-phosphatase 38 activity. Without Mg²⁺, Co²⁺ stimulates K⁺-phosphatase activity similarly but with a \approx 4.5-fold 39 greater affinity than with Mg^{2+} . With Mg^{2+} , K⁺-phosphatase activity is almost completely 40 inhibited by Co^{2+} . Substitution of Mg^{2+} by Co^{2+} slightly increases enzyme affinity for K⁺ and 41 NH_4^+ . Independently of Mg^{2+} , ouabain inhibition is unaffected by Co^{2+} . Mg^{2+} displaces bound 42 Co²⁺ from the Mg²⁺-binding site in a concentration dependent mechanism. However, at 43 saturating Mg^{2+} concentrations, Co^{2+} does not displace Mg^{2+} from its binding site even at 44 elevated concentrations. Saturation by Co^{2+} of the Mg²⁺ binding site does not affect *pNPP* 45 recognition by the enzyme. Given that the interactions between heavy metal ions and enzymes 46 are particularly complex, their toxic effects at the molecular level are poorly understood. Our 47 findings elucidate partly the mechanism of action of Co^{2+} on a crustacean gill (Na⁺, K⁺)-48 49 ATPase.

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52 Highlights

53 1. Without Mg^{2+} , cobalt ions stimulate the gill (Na⁺, K⁺)-ATPase

54 2. Co^{2+} has a 4.5-fold greater affinity for the gill (Na⁺, K⁺)-ATPase than does Mg²⁺

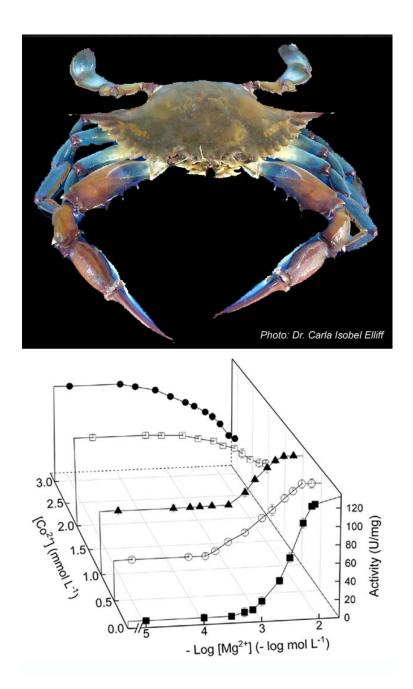
55 3. Mg^{2+} displaces Co^{2+} from the Mg^{2+} -binding site in a concentration dependent manner

56 4. Ouabain inhibition with Co^{2+} or Mg^{2+} is identical

57 5. Saturation by Co^{2+} of Mg^{2+} -binding sites does not affect substrate recognition

59 Graphical Abstract

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- 62

63 Graphical abstract (synopsis)

Using a crab gill (Na⁺, K⁺)-ATPase, we demonstrate that Co^{2+} inhibits K⁺-phosphatase activity with Mg²⁺, which is stimulated without Mg²⁺. Mg²⁺ displaces Co^{2+} from the Mg²⁺binding site but Co^{2+} cannot displace Mg²⁺. Ouabain inhibition is unaffected by Co^{2+} , independently of Mg²⁺. The molecular mechanism of Co^{2+} toxicity is partly elucidated.

69 1. INTRODUCTION

70 Estuarine and coastal environments accumulate toxic contaminants owing to natural 71 phenomena and/or anthropogenic activities [1]. Such pollutants include a wide variety of heavy metal ions, organic compounds and various micro/nano-particles [2-4]. Transition 72 73 metals stand out specifically as the most abundant contaminants found in aquatic 74 environments and contribute largely to toxicity [5,6]. Heavy metal ions are particularly 75 harmful to organisms as they are not degradable and accumulate acutely or chronically in 76 cells and tissues, altering biochemical and physiological homeostatic mechanisms, which can 77 lead to demise at the organismal level [7].

78 The uptake and toxicity of heavy metals in aquatic organisms is determined by various 79 ambient parameters like pH, temperature and salinity [8]. The biotic ligand model is the model 80 most used to predict metal toxicity in aquatic environments [9]. A profusion of ecotoxicological 81 studies has highlighted the importance of heavy metal toxicity in aquatic environments [e.g., 82 10,11,12,13]. Most studies of heavy metal toxicity in aquatic organisms concern their 83 bioaccumulation in different tissues [1,14,15] and effects on physiological and biochemical processes like osmoregulatory capability and aerobic and oxidative stress metabolism [1,11,16]. 84 85 The toxic effects of heavy metals at the molecular level are poorly known since the 86 interactions between such metal ions and enzymes are complex [17,18]. The scant information 87 available regarding the molecular mechanisms of heavy metal toxicity is limited mainly to fish 88 and crustaceans [19–22].

89 While cobalt is a trace element indispensable for various physiological processes 90 [21,23,24] it is toxic at high intracellular concentrations [25]. In humans, excessive exposure to 91 cobalt results in complex health deficits involving heme oxidation, cytotoxicity, oxidative 92 stress, apoptosis, altered membrane permeability, calcium channel blockage and DNA 93 damage [26–28]. Active Ca²⁺ transport is inhibited in freshwater fish gills [22] while 94 metabolic pathways are affected in freshwater algae [29].

95 Although cobalt concentrations in open oceanic waters are less than 7 ng L^{-1} [30], anthropogenic activities have led to progressive contamination of estuarine and coastal 96 97 environments [see 25 for review]. To illustrate, mean cobalt levels measured in surface waters near an industrial plant in an Iberian Mediterranean estuarine system are $\approx 700 \ \mu g \ Co^{2+} \ L^{-1}$ and 98 reach up to $\approx 2,800 \ \mu g \ Co^{2+} \ L^{-1}$ [31]. Cobalt titers in brachyuran crabs from marine/estuarine 99 ecosystems range from 0.26 to 0.43 μ g Co²⁺ g⁻¹ soft tissue dry mass in *Callinectes sapidus* 100 [32]; 0.91 to 1.2 μ g Co²⁺ g⁻¹ hepatopancreas dry mass in male and female *Portunus segnis* 101 [33]; and 0.72 to 0.86 μ g Co²⁺ g⁻¹ dry mass in whole *Carcinus maenas* [34]. The lack of 102

molecular information regarding harm to aquatic organisms, including the effects of Co^{2+} , has impaired our overall comprehension of environmental damage and particularly of toxicity owing to bioaccumulation in marine organisms [16,23,25,35].

106 Estuarine and coastal organisms are powerful indicators of the health of the marine 107 environment [36,37]. The portunid blue crab Callinectes danae is a crustacean species 108 recommended as an environmental monitor of biological responses in contaminated estuarine 109 and coastal areas [13]. The crab is a euryhaline osmoregulator, tolerant of exposure to wide 110 range of salinities [38], and of great commercial value. It occurs from Florida (USA) to the 111 southern Brazilian coast [39,40] and can be found in muddy estuaries and mangroves, on 112 sandy and muddy shores, and in coastal waters up to 75 meters depth, including biotopes in 113 which salinity varies from brackish to sea water [40,41].

114 In crustaceans, the gills provide a selective interface between the external environment 115 and internal fluids, contributing to osmotic, ionic, excretory, and acid-base homeostasis. They 116 are also an important route of entry of heavy metal ions [1,2,42]. Metal toxicity can thus vary depending on osmoregulatory strategy while alterations in salinity can affect the availability 117 118 of metal ion species in the water column [36,43]. In brachyuran crabs, the three posterior gill pairs are specialized in ion transport [42], exhibiting a thickened epithelium [42,44] and 119 increased expression and activity of ion transporters, including the (Na⁺, K⁺)-ATPase 120 [1,45,46]. The presence of the (Na⁺, K⁺)-ATPase in the gill tissue and its role in physiological 121 122 homeostasis renders this enzyme a suitable bioindicator to evaluate the kinetic effects of heavy metal ions like Co^{2+} . 123

The (Na^+, K^+) -ATPase is a transmembrane enzyme that mediates the coupled transport 124 125 of three Na^+ from the cytosol into the extracellular fluid and of two K^+ into the cytosol per 126 ATP molecule hydrolyzed. The enzyme consists of two main subunits: a catalytic α -subunit, 127 responsible for ion transport driven by ATP hydrolysis, and a highly glycosylated, non-128 catalytic β -subunit that modulates the transport properties of the enzyme [47,48]. The 129 subunits are often associated with an FXYD peptide or γ -subunit that modulates (Na⁺, K⁺)-ATPase activity by altering the enzyme's apparent affinity for Na⁺, K⁺ and ATP [49]. Briefly, 130 131 the catalytic cycle involves alternation between the phosphorylated E1 and E2 conformations that show high affinity for Na⁺ and K⁺, respectively [47,48]. Magnesium is also essential 132 although the ion is not transported during the catalytic cycle. Rather, Mg²⁺participates as the 133 true substrate (a Mg•ATP complex) and plays a regulatory role, interfering with the 134 conformational transitions of the enzyme [50,51]. At high concentrations, Mg^{2+} can inhibit 135 Na^+ and K^+ binding by occupying a second specific inhibitory site outside the α -subunit 136

membrane domains [50–52] or binding to the protein surface near the access channel of the ion binding sites [51,53,54]. Various divalent cations like Ca^{2+} , Mn^{2+} , Ba^{2+} and Sr^{2+} can substitute for Mg^{2+} by binding to the Mg^{2+} regulatory site, activating the enzyme [55–57]. The presence of both ATP and Na^{+} induces immediate enzyme phosphorylation such that Mg^{2+} binding and the phosphorylation reaction cannot be examined separately [51].

142 While the (Na^+, K^+) -ATPase exhibits high specificity for ATP it also catalyzes the 143 ouabain-sensitive hydrolysis of other nucleoside triphosphates [58] and various non-144 nucleotide substrates such as *p*-nitrophenyl phosphate, acetyl phosphate, 2,4-dinitrophenyl phosphate, β -(2-furyl)-acryloyl phosphate, O-methyl fluorescein phosphate and 4-azido-2-145 nitrophenylphosphate [59–64]. The activity corresponding to the hydrolysis of non-nucleotide 146 substrates is known as the K^+ -phosphatase activity and requires Mg^{2+} and K^+ but is inhibited 147 by Na⁺. Such activity represents a partial reaction of the (Na⁺, K⁺)-ATPase in which the E2 148 form is the main conformational state involved in pNPP hydrolysis [58]. The use of such non-149 150 nucleotide substrates has disclosed important kinetic characteristics of the (Na^+, K^+) -ATPase and, under most experimental conditions, these are better substrates than ATP itself [59,61– 151 63,65]. Differently from ATP, p-nitrophenyl phosphate hydrolysis by the gill enzyme 152 153 involves only a single substrate binding site [61,66]. Likewise, the (Na^+, K^+) -ATPase is also 154 phosphorylated by p-nitrophenyl phosphate and other acyl phosphates [56,60,67,68]. K^+ 155 occlusion does not participate in phosphatase turnover [69] although, in the absence of Na⁺, K^+ stimulates K^+ -phosphatase activity [70]. The use of p-nitrophenyl phosphate as a substrate 156 thus facilitates the study of Mg^{2+} binding [51]. 157

Given the paucity of information on the molecular effects of Co^{2+} on aquatic crustaceans from marine environments contaminated by heavy metals, in this study we provide a comprehensive analysis of the differential effects of Co^{2+} *in vitro* on the steady state kinetic properties of the K⁺-phosphatase activity of a gill (Na⁺, K⁺)-ATPase from the blue crab *Callinectes danae*.

163

164 2. MATERIAL AND METHODS

165 Material

Millipore MilliQ (Merck KGaA, Darmstadt, Germany) ultrapure, apyrogenic deionized water was used to prepare all solutions. Chemicals of the highest purity commercially available were purchased from Sigma Chemical Co. (St. Louis, MO, USA) or Merck (Darmstadt, Germany). All salts were used as chlorides. The homogenization buffer consisted of 20 mmol L⁻¹ imidazole (pH 6.8), 250 mmol L⁻¹ sucrose and a proteinase inhibitor

171 cocktail (1 mmol L^{-1} benzamidine, 5 µmol L^{-1} antipain, 5 µmol L^{-1} leupeptin,5 µmol L^{-1} 172 phenyl-methane-sulfonyl-fluoride, and 1 µmol L^{-1} pepstatin A). Analytical estimation of the 173 stock CoCl₂ solution concentration (100 mmol L^{-1}) was performed employing inductively 174 coupled mass spectrometry (Perkin Elmer Avio 200 optical emission spectrometer, Boston 175 MA, USA). NH₄⁺ was removed from the crystalline ammonium sulfate suspensions of LDH 176 and PK according to Fabri et al. [71]. When necessary, enzyme solutions were concentrated 177 on YM-10 Amicon Ultra filters.

178

179 **2.1. Crab collection**

180 Adult specimens of *Callinectes danae* of ≈ 9 cm carapace width were collected at low tide using seine nets or baited hand nets from Barra Seca beach (23° 25' 01.5" S, 45° 03' 181 182 01.0" W), Ubatuba, São Paulo State, Brazil (SISBIO/ICMBio/IBAMA authorization 183 02027.002342/98-04, permit #29594-18 to John C. McNamara). The crabs were held briefly 184 in plastic boxes containing 30 L aerated seawater from the collection site during transport to the laboratory where they were immediately anesthetized by chilling in crushed ice. After 185 186 bisecting and removal of the carapace, all three posterior gill pairs (120 gills/preparation, ≈ 5 g 187 wet mass) were rapidly dissected out and frozen in liquid nitrogen in homogenization buffer 188 in Falcon tubes.

189

190 **2.2. Preparation of the gill microsomal fraction**

For each of the three (N=3) gill homogenates prepared, the gills frozen in the 191 192 homogenization buffer were thawed, diced and homogenized (20 mL homogenization 193 buffer/g wet tissue) at 600 rpm in a Potter homogenizer in a crushed ice bath. The (Na^+, K^+) -194 ATPase-rich microsomal fraction was prepared by stepwise differential centrifugation (20,000 195 and $100,000 \times g$) of the gill homogenate [38]. The resulting pellet was resuspended in 20 mmol L⁻¹ imidazole buffer (pH 6.8) containing 250 mmol L⁻¹ sucrose (15 mL buffer/g wet 196 tissue). Aliquots (0.5 mL) were frozen in liquid nitrogen, stored at -20 °C and used within 197 three-month's storage provided that at least 95% (Na⁺, K⁺)-ATPase activity was present. 198

199

200 **2.3. Estimation of** *p***-nitrophenyl phosphatase activity**

201 *p*-Nitrophenyl phosphate (*p*NPP) was used as the enzyme substrate. The *p*-nitrophenyl 202 phosphatase activity (*p*NPPase) of the microsomal fraction was estimated continuously at 25 203 °C, following the release of the *p*-nitrophenolate ion (*p*NP⁻) at 410 nm (ϵ 410nm, pH 7.5= 204 13,160 M⁻¹ cm⁻¹) in a Shimadzu UV-1800 spectrophotometer (Vernon Hills IL, USA)

equipped with thermostatted cells. The standard incubation medium contained 50 mmol L⁻¹ HEPES buffer, pH 7.5, and appropriate *p*NPP, Mg^{2+} , K⁺ or NH₄⁺ concentrations (see Results and Figure legends for substrate and specific ionic concentrations) and 9 µg alamethicin, in final volume of 1 mL. Activity was estimated with or without 7 mmol L⁻¹ ouabain, the difference corresponding to the K⁺-phosphatase activity of the gill (Na⁺, K⁺)-ATPase. The reaction was always initiated by the addition of the enzyme.

Because tissue homogenization usually results in the formation of small vesicles in the microsomal preparation that can occlude the catalytic site of the enzyme, *p*NPPase activity also was estimated after 10 min pre-incubation with 9 μ g alamethicin, a membrane poreforming antibiotic, to verify the presence of vesicles showing ATPase activity.

Controls without added enzyme were used to estimate the spontaneous hydrolysis of the substrate under the assay conditions. The kinetic measurements were carried out in duplicate, and substrate hydrolysis was accompanied over the shortest possible period to guarantee initial velocity measurements (<5% of substrate hydrolyzed during the reaction period). One unit (U) of enzyme activity was defined as the amount of enzyme that hydrolyses 1.0 nmol of *p*NPP per minute at 25 °C.

221

222 **2.4. Protein**

Protein concentration was measured in duplicate aliquots of the microsomal preparations using the Coomassie Blue G dye-binding assay employing bovine serum albumin as a standard. Assays were read at 595 nm using a Shimadzu UV-1800 spectrophotometer [71].

227

228 **2.5. Estimation K⁺-phosphase activity with cobalt ions**

The effect of cobalt ions on *p*NPPase activity was estimated as above using cobalt concentrations between 10^{-5} and 2×10^{-2} mmol L⁻¹. Our study on the effect of Co²⁺ on the interaction of the different ligands with the (Na⁺,K⁺)-ATPase was performed using 3 mmol L⁻ 1 Co²⁺ (50.8 µg L⁻¹ Co²⁺) which inhibits K⁺-phosphatase activity by ≈50%.

233

234 2.6. Measurement of ATP hydrolysis

To provide a direct comparison with previous microsomal preparations, initial rates of ATP hydrolysis also were estimated continuously at 25 °C, using a pyruvate kinase/lactic dehydrogenase coupling system in which ATP hydrolysis is coupled to NADH oxidation [72]. The specific activity of (Na⁺, K⁺)-ATPase for ATP of 296.5 \pm 10.2 nmol Pi min⁻¹ mg⁻¹ protein corresponding to $125.5 \pm 2.3 \text{ nmol } p\text{NP}^- \text{min}^{-1} \text{ mg}^{-1}$ protein of K⁺-phosphatase activity is comparable to our previous findings [73].

241

242 **2.7. Estimation of kinetic parameters**

243 The kinetic parameters V_M (maximum velocity), $K_{0.5}$ (apparent dissociation constant) 244 and the n_H value (Hill coefficient) for pNPP hydrolysis were calculated using SigrafW 245 software ([74]; freely available from http://portal.ffclrp.usp.br/sites/fdaleone/downloads). The apparent dissociation constant of the enzyme-inhibitor complex (K_I) was estimated using the 246 Dixon plot in which the reaction rate corresponding to the K^+ -phosphatase activity was 247 248 corrected for residual activity at high inhibitor concentrations [75]. Data points (mean \pm SD) 249 in the figures representing each substrate/ligand concentration are mean values of duplicate 250 aliquots from the same preparation and were used to fit the saturation curves that were 251 repeated using three different microsomal preparations (N= 3). The kinetic parameters (V_M , K_M , K_I or $K_{0.5}$) shown in the tables are calculated values and represent the mean (± SD) derived 252 253 from values estimated for each of three (N=3) microsomal preparations.

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255 **3. RESULTS**

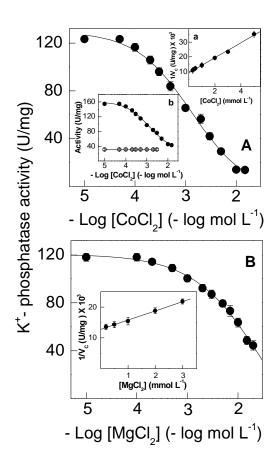
256 Estimation of *p*NPPase activity with alamethic revealed that the gill microsomal preparation from fresh caught C. danae includes $\approx 25\%$ sealed, pNPPase-containing vesicles 257 $(164.5 \pm 3.8 \text{ and } 122.5 \pm 1.8 \text{ nmol } p\text{NP}^- \text{min}^{-1} \text{ mg}^{-1}$ protein with or without alamethicin, 258 259 respectively). Thus, pNPPase activity was always estimated using 9 µg alamethicin. Seven mmol L⁻¹ ouabain decreased pNPPase activity from 164.5 ± 3.8 to 39.2 ± 1.9 nmol pNP⁻ min⁻ 260 ¹ mg⁻¹ protein, indicating that $\approx 75\%$ of the total phosphohydrolyzing activity (125.5 \pm 2.3 261 nmol pNP^{-} min⁻¹ mg⁻¹ protein) corresponds to the K⁺-phosphatase activity of the gill (Na⁺, 262 K⁺)-ATPase. Measurements using ATP as a substrate showed a (Na⁺, K⁺)-ATPase activity of 263 296.5 ± 10.2 nmol Pi min⁻¹ mg⁻¹ protein. 264

265

3.1. Effect of Co²⁺ on K⁺-phosphatase activity

Under optimal assay conditions (see legends to Fig. 1A and 1B), increasing Co²⁺ concentrations from 10⁻⁵ to 2×10^{-2} mol L⁻¹ in the incubation medium inhibited K⁺⁻ phosphatase activity by 90% (Fig. 1A and Table 1) with K_I= 2.77 ± 0.33 mmol L⁻¹ (inset a to Fig. 1). K⁺-phosphatase activity decreased following a single titration curve to 13.7 ± 3.6 nmol pNP⁻ min⁻¹ mg⁻¹ protein. The ouabain-insensitive *p*NPPase activity of ≈38 nmol pNP⁻ min⁻¹ mg⁻¹ protein was unaffected by increasing Co²⁺ concentrations (inset b to Fig. 1). On

- fixing Co^{2+} at 3 mmol L⁻¹, increasing Mg²⁺ concentrations (10⁻⁵ to 2×10⁻² mol L⁻¹) inhibited
- 274 K⁺-phosphatase activity by $\approx 60\%$ (Fig. 1B and Table 1). A K_I= 4.81 ± 0.71 mmol L⁻¹ was
- 275 calculated for inhibition of K⁺-phosphatase activity by Mg^{2+} in the presence of Co^{2+} (inset to
- 276 Fig. 1B).
- 277



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Figure 1. Effect of cobalt or magnesium ions in the presence of each other on the K⁺phosphatase activity of *C. danae* gill (Na⁺, K⁺)-ATPase

Activity was assayed continuously at 25 °C in 50 mmol L⁻¹ HEPES buffer, pH 7.5, containing 281 10 mmol L^{-1} pNPP, 15 mmol L^{-1} KCl, 9 µg alamethicin and the metal ions in a final volume 282 of 1 mL. The mean activity of duplicate aliquots of the same microsomal preparation (\approx 15 µg 283 protein) was used to fit the saturation curve which was repeated using three different 284 microsomal preparations (± SD, N= 3). Where lacking, error bars are smaller than the 285 symbols used. A- with 7 mmol L^{-1} MgCl₂. Inset a- Dixon plot for estimation of K_I in which v_c 286 is the K⁺-phosphatase activity corrected for residual pNPPase activity found at high Co^{2+} 287 concentration. Inset b- total pNPPase activity (\bullet) and ouabain-insensitive pNPPase activity 288 (O). **B**- with 3 mmol L⁻¹ CoCl₂. Inset- Dixon plot for estimation of K_I in which v_c is the K⁺-289 phosphatase activity corrected for residual K^+ -phosphatase activity found at high Mg^{2+} 290 291 concentrations.

Cobalt ions can substitute for Mg^{2+} , stimulating gill K⁺-phosphatase activity (Fig. 2A 293 and Table 1). Under optimal assay conditions (see legends to Fig. 2A and 2B) without Mg^{2+} , 294 Co^{2+} stimulated K⁺-phosphatase activity to a maximum rate of V_M= 122.5 ± 3.1 nmol pNP⁻ 295 min⁻¹ mg⁻¹ protein with $K_{0.5}=0.69\pm0.23$ mmol L⁻¹, showing a single saturation curve 296 obeying cooperative kinetics ($n_{\rm H}$ = 1.4). Ouabain-insensitive *p*NPPase activity was stimulated 297 to 46.4 ± 3.7 nmol pNP⁻ min⁻¹ mg⁻¹ protein over the same Co²⁺ concentration range (inset to 298 Fig. 2A). K⁺-phosphatase activity was inhibited by Co^{2+} concentrations above 10^{-2} mol L⁻¹. In 299 the absence of Co^{2+} , Mg^{2+} (10⁻⁴ to 2×10⁻² mol L⁻¹) stimulated K⁺-phosphatase activity to a 300 maximum rate of V_{M} = 135.1 ± 5.0 nmol pNP⁻ min⁻¹ mg⁻¹ protein with $K_{0.5}$ = 2.98 ± 0.59 301 mmol L^{-1} and cooperative kinetics ($n_{\rm H}=2.2$), following a single saturation curve (Fig. 2B and 302 Table 1). Stimulation by Mg^{2+} of ouabain-insensitive pNPPase activity was negligible over 303 the concentration range used (inset to Fig. 2B). The \approx 4.5-fold lower K_{0.5} suggests that Co²⁺ 304 binds more tightly to the Mg²⁺-binding sites than does Mg²⁺ itself. 305 306

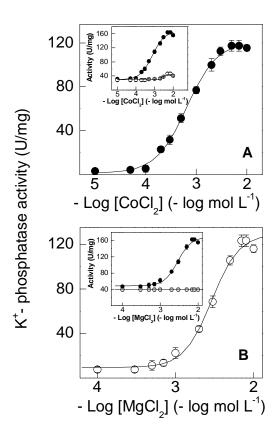


Figure 2. Estimulation by cobalt or magesium ions of K^+ -phosphatase activity of *C*. *danae* gill (Na⁺, K⁺)-ATPase

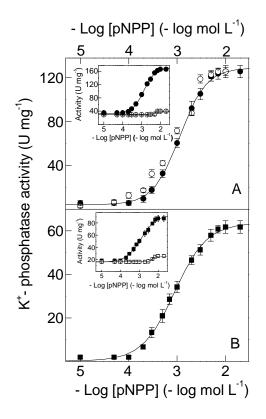
- Activity was assayed continuously at 25 °C in 50 mmol L^{-1} HEPES buffer, pH 7.5, containing
- 311 10 mmol L⁻¹ pNPP, 15 mmol L⁻¹ KCl, 9 μ g alamethicin in a final volume of 1 mL. The mean

activity of duplicate aliquots of the same microsomal preparation ($\approx 15 \ \mu g$ protein) was used to fit the saturation curve which was repeated using three different microsomal preparations (\pm SD, N= 3). Where lacking, error bars are smaller than the symbols used. A- with cobalt ions. B- with magnesium ions. Inset to figures- total *p*NPPase activity (\bigcirc); ouabaininsensitive *p*NPPase activity (\bigcirc).

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319 **3.2. Effect of Co²⁺ on pNPP hydrolysis**

Under optimal assay conditions (see legends to Fig. 2A and 2B), increasing pNPP 320 concentrations stimulated K⁺-phosphatase activity to a maximum rate of V_{M} = 138.1 ± 4.2 321 nmol pNP⁻¹ min⁻¹ mg⁻¹ protein with $K_{0.5} = 1.76 \pm 0.49$ mmol L⁻¹, following cooperative 322 kinetics ($n_{H}=1.3$) and showing a single saturation curve (Fig. 3A and Table 1). Substitution of 323 Mg^{2+} with 3 mmol L⁻¹ Co²⁺ also gave a single saturation curve, overlapping that for Mg^{2+} , 324 showing a maximum rate of V_{M} = 128.2 ± 4.4 nmol pNP⁻ min⁻¹ mg⁻¹ protein with $K_{0.5}$ = 1.15 ± 325 0.61 mmol L⁻¹ (Fig. 3A). With Mg²⁺, the ouabain-insensitive pNPPase activity of \approx 30 nmol 326 pNP^{-} min⁻¹ mg⁻¹ protein was not affected by increasing pNPP concentrations (not shown). 327 However, with 3 mmol L⁻¹ Co²⁺ this activity was stimulated by \approx 35% over the same *pNPP* 328 concentration range (inset to Fig. 3A). With both Mg^{2+} and Co^{2+} , under the same saturating 329 ionic and substrate concentrations, K⁺-phosphatase activity decreased to a maximum rate of 330 $V_{M} = 63.1 \pm 3.8 \text{ nmol } p\text{NP}^{-1} \text{ min}^{-1} \text{ mg}^{-1}$ protein with $K_{0.5} = 0.92 \pm 0.28 \text{ mmol } L^{-1}$ also obeying 331 cooperative kinetics (Fig. 3B and Table 1). This $\approx 50\%$ inhibition of K⁺-phosphatase activity 332 was accompanied by a 2-fold decrease in $K_{0.5}$ compared to Mg²⁺ (Table 1). The ouabain-333 insensitive pNPPase activity of $\approx 20 \text{ nmol } pNP^- \text{ min}^{-1} \text{ mg}^{-1}$ protein was stimulated $\approx 60\%$ over 334 335 the same *p*NPP concentration range (inset to Fig. 3B). 336



337

Figure 3. Effect of cobalt or magnesium ions on the modulation by *p*NPP of the K⁺phosphatase activity of *C. danae* gill (Na⁺, K⁺)-ATPase

Activity was assayed continuously at 25 °C in 50 mmol L⁻¹ HEPES buffer, pH 7.5, containing 340 15 mmol L⁻¹ KCl, 9 μ g alamethicin and the metal ion (7 mmol L⁻¹ Mg²⁺ or 3 mmol L⁻¹ Co²⁺) 341 in a final volume of 1 mL. The mean activity of duplicate aliquots of the same microsomal 342 343 preparation (\approx 15 µg protein) was used to fit the saturation curve which was repeated using three different microsomal preparations (\pm SD, N= 3). Where lacking, error bars are smaller 344 than the symbols used. A- with magnesium (\bigcirc) or cobalt ions (\bigcirc) . B- with both magnesium 345 and cobalt ions. **Inset to figures-** total *p*NPPase activity (\bullet , \blacksquare); ouabain-insensitive *p*NPPase 346 activity (\bigcirc, \Box) for Co²⁺only. 347

348

349 **3.3. Effect of Co²⁺ on Mg²⁺ stimulation**

Magnesium can displace Co^{2+} bound to the gill (Na⁺, K⁺)-ATPase below 2 mmol L⁻¹ Co²⁺ (Fig. 4). Increased Mg²⁺ (10⁻⁵ to 5×10⁻² mol L⁻¹) displaces bound Co²⁺ (0.5 and 1 mmol L⁻¹), stimulating K⁺-phosphatase activity to a maximum rate of ≈130 nmol *p*NP⁻ min⁻¹ mg⁻¹ protein with similar K_{0.5} (Fig. 4 and Table 1). However, at 2 mmol L⁻¹ Co²⁺ and above no displacement was seen and increasing Mg²⁺ concentrations inhibited K⁺-phosphatase activity with K_I= 4.41 ± 0.69 and 4.81 ± 0.71 mmol L⁻¹ for 2 and 3 mmol L⁻¹ Co²⁺, respectively (Table 1).

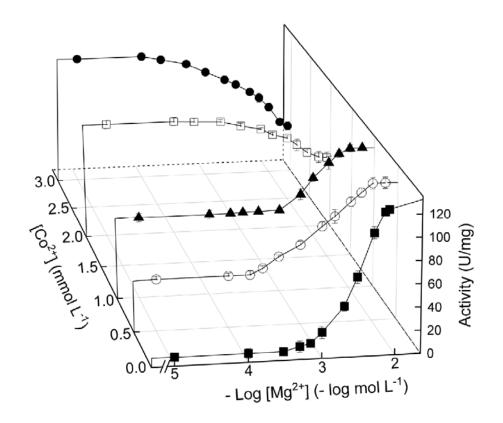




Figure 4. Effect of magnesium ions on the modulation of K⁺-phosphatase activity at different cobalt ion concentrations in *C. danae* gill (Na⁺, K⁺)-ATPase.

Activity was assayed continuously at 25 °C in 50 mmol L⁻¹ HEPES buffer, pH 7.5, containing 10 mmol L⁻¹ *p*NPP, 15 mmol L⁻¹ KCl and 9 µg alamethicin in a final volume of 1 mL. The mean activity of duplicate aliquots of the same microsomal preparation (\approx 15 µg protein) was used to fit the saturation curve which was repeated using three different microsomal preparations (\pm SD, N= 3). Where lacking, error bars are smaller than the symbols used. (\blacksquare) without Co²⁺. (O) 0.5 mmol L⁻¹ Co²⁺. (\blacktriangle) 1 mmol L⁻¹ Co²⁺. (\square) 2 mmol L⁻¹ Co²⁺. (\bigcirc) 3 mmol L⁻¹ Co²⁺.

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370 3.4. Effect of Co²⁺ on K⁺ stimulation

Under saturating ionic and substrate concentrations (see legends to Fig. 5A and 5B) in 371 the absence of Co^{2+} , increasing K⁺ stimulated K⁺-phosphatase activity to a maximum rate of 372 $V_{M} = 134.2 \pm 4.5 \text{ nmol } pNP^{-1} \text{ min}^{-1} \text{ mg}^{-1}$ protein with $K_{0.5} = 9.60 \pm 2.04 \text{ mmol } L^{-1}$ (Fig. 5A and 373 Table 1) following cooperative kinetics ($n_{\rm H}=2.2$). Substitution of Mg²⁺ by 3 mmol L⁻¹ Co²⁺ 374 also gave a single saturation curve overlapping with that for Mg^{2+} and showing a maximum 375 rate of V_{M} = 133.3 ± 3.3 nmol pNP⁻ min⁻¹ mg⁻¹ protein with $K_{0.5}$ = 6.00 ± 1.50 mmol L⁻¹ (Fig. 376 5A and Table 1). Ouabain-insensitive pNPPase activity was not stimulated by either metal ion 377 over the concentration range used (inset to Fig. 5A). With Co^{2+} plus Mg^{2+} , K⁺-phosphatase 378 activity decreased to V_{M} = 59.5 ± 4.0 nmol pNP⁻ min⁻¹ mg⁻¹ protein with K_{0.5} = 2.79 ± 0.41 379 mmol L^{-1} following a single titration curve (Fig. 5B and Table 1), obeying cooperative 380

kinetics ($n_{\rm H}$ = 1.5). K⁺-phosphatase activity at K⁺ concentrations below 10⁻⁴ mol L⁻¹ was ≈15 nmol *p*NP⁻ min⁻¹ mg⁻¹ protein. Ouabain-insensitive *p*NPPase activity was stimulated to ≈22 nmol *p*NP⁻ min⁻¹ mg⁻¹ protein over the same K⁺ concentration range (inset to Fig. 5B). K⁺-

384 phosphatase activity was not synergically stimulated by K^+ plus NH_4^+ (not shown).

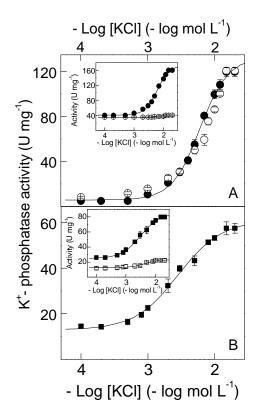




Figure 5. Effect of cobalt or magnesium ions on the modulation by potassium ions of K^+ phosphatase activity of *C. danae* gill (Na⁺, K⁺)-ATPase

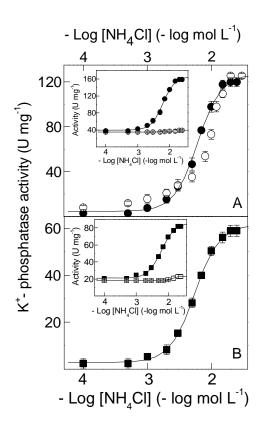
Activity was assayed continuously at 25 °C in 50 mmol L⁻¹ HEPES buffer, pH 7.5, containing 10 mmol L⁻¹ *p*NPP, 9 µg alamethicin and the metal ion (7 mmol L⁻¹ Mg²⁺ or 3 mmol L⁻¹ Co²⁺) in a final volume of 1 mL. The mean activity of duplicate aliquots of the same microsomal preparation (\approx 15 µg protein) was used to fit the saturation curve which was repeated using three different microsomal preparations (\pm SD, N= 3). Where lacking, error bars are smaller than the symbols used. A- K⁺-phosphatase activity with Mg²⁺ (O) or Co²⁺ (\bullet). B- K⁺-phosphatase activity with both Mg²⁺ and Co²⁺. Insets to figures- total *p*NPPase activity (\bullet) and ouabain-insensitive *p*NPPase activity (O) Co²⁺.

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397 3.5. Effect of Co²⁺ on NH₄⁺ stimulation

Under saturating ionic and substrate concentrations (see legends to Fig. 6A and 6B) in the absence of Co^{2+} , a maximal rate of V_{M} = 122.2 ± 5.2 nmol *p*NP⁻ min⁻¹ mg⁻¹ protein with $K_{0.5}$ = 9.02 ± 2.51 mmol L⁻¹ was estimated for NH₄⁺ concentrations increasing from 10⁻⁴ to 5×10^{-2} mol L⁻¹ (Fig. 6A and Table 1). Substitution of Mg²⁺ by 3 mmol L⁻¹ Co²⁺ also gave a

single saturation curve with a maximum rate of V_{M} = 127.9 ± 4.2 nmol pNP⁻ min⁻¹ mg⁻¹ 402 protein with $K_{0.5} = 6.00 \pm 1.10$ mmol L⁻¹, overlapping with that for Mg²⁺ (Fig. 6A and Table 403 1). Stimulation of ouabain-insensitive pNPPase activity by Mg^{2+} and Co^{2+} was negligible over 404 the NH_4^+ concentration range used (inset to Fig. 6A). With Co^{2+} plus Mg^{2+} , K^+ -phosphatase 405 activity decreased to $V_{M}= 61.9 \pm 3.7 \text{ nmol } p\text{NP}^{-1} \text{ min}^{-1} \text{ mg}^{-1}$ protein with $K_{0.5}= 5.46 \pm 0.64$ 406 mmol L^{-1} (Fig. 6B and Table 1). Stimulation of the ouabain-insensitive pNPPase activity was 407 <10% with Co²⁺ plus Mg²⁺ (inset to Fig. 6B). K⁺-phosphatase activity was not stimulated 408 synergically by K^+ plus NH_4^+ (not shown). 409



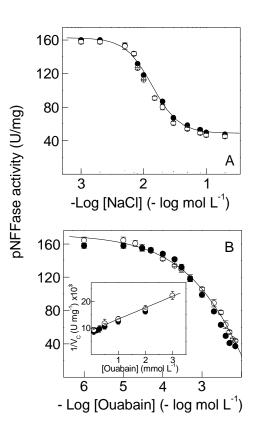
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Figure 6. Effect of cobalt or magnesium ions on the modulation by ammonium ions of K⁺-phosphatase activity of *C. danae* gill (Na⁺, K⁺)-ATPase

Activity was assayed continuously at 25 °C in 50 mmol L⁻¹ HEPES buffer, pH 7.5, containing 10 mmol L⁻¹ *p*NPP, 9 µg alamethicin and the metal ion (7 mmol L⁻¹ Mg²⁺ or 3 mmol L⁻¹ Co²⁺) in a final volume of 1 mL. The mean activity of duplicate aliquots of the same microsomal preparation (\approx 15 µg protein) was used to fit the saturation curve which was repeated using three different microsomal preparations (\pm SD, N= 3). Where lacking, error bars are smaller than the symbols used. A- K⁺-phosphatase activity with Mg²⁺ (\bigcirc) or Co²⁺ (\bigcirc). B- K⁺-phosphatase activity with both Mg²⁺ and Co²⁺. Insets to figures- total *p*NPPase activity (\bigcirc) and ouabain-insensitive *p*NPPase activity (\bigcirc) Co²⁺.

421 **3.6. Effect of Co²⁺ on inhibition by Na⁺ and ouabain**

Cobalt does not affect inhibition of *p*NPPase activity by Na^+ or ouabain (Fig. 7). 422 Sodium concentrations from 10^{-3} to 0.2 mol L⁻¹ inhibited *pNPPase* activity by $\approx 70\%$ either 423 with or without Co^{2+} (Fig. 7A and Table 1). The IC₅₀ estimated for Na⁺ inhibition of *p*NPPase 424 activity was $\approx 16 \text{ mmol } \text{L}^{-1}$ (Table 1). Over the ouabain concentration range from 10^{-6} to 10^{-2} 425 mol L^{-1} , the Na⁺ and ouabain inhibition curves overlapped independently of Co²⁺ (Fig. 7B and 426 Table 1). Their monophasic behavior with very similar inhibition constants (K_{I} = 2.27 ± 0.62 427 mmol L⁻¹ and K_I= 2.13 \pm 0.97 mmol L⁻¹ for Mg²⁺ and Co²⁺ respectively) (inset to Fig. 7B) 428 suggests a single ouabain binding site. 429



430

Figure 7. Effect of cobalt or magnesium ions on the inhibition by sodium and ouabain on pNPPase activity of *C. danae* gill (Na⁺, K⁺)-ATPase

Activity was assayed continuously at 25 °C in 50 mmol L⁻¹ HEPES buffer, pH 7.5, containing 433 10 mmol L⁻¹ pNPP, 15 mmol L⁻¹ KCl and 9 μ g alamethicin in a final volume of 1 mL. The 434 mean activity of duplicate aliquots of the same microsomal preparation (\approx 15 µg protein) was 435 used to fit the saturation curve which was repeated using three different microsomal 436 preparations (\pm SD, N= 3). Where lacking, error bars are smaller than the symbols used. A-437 sodium ion. B- ouabain. Inset to Fig. 8B- Dixon plot for estimation of K_I in which v_c is the 438 pNPPase activity corrected for residual pNPPase activity found at high ouabain 439 concentrations. ($\overset{\circ}{O}$) with 7 mmol L⁻¹ Mg²⁺. ($\overset{\bullet}{\bullet}$) with 3 mmol L⁻¹ Co²⁺. 440

441	Table 1. Kinetic parameters calculated for the effect of pNPP, Mg^{2+} , K^+ , NH_4^+ , Na^+ , Co^{2+} and ouabain on K^+ -phosphatase activity of
442	(Na ⁺ , K ⁺)-ATPase in a gill microsomal preparation from <i>Callinectes danae</i> .

Ligand	Co ²⁺ mmol L ⁻¹	<i>p</i> NPP mmol L ⁻¹	Mg ²⁺ mmol L ⁻¹	K ⁺ mmol L ⁻¹	NH4 ⁺ mmol L ⁻¹	Na ⁺ mmol L ⁻¹	V _M U mg ⁻¹	K _{0.5} mmol L ⁻¹	n _H	K _I mmol L ⁻¹	IC_{50} mmol L ⁻¹
Co^{2+}	Variable	10	-	15	-	-	122.5 ± 3.1	0.69 ± 0.23	1.4	-	-
Co^{2+}	Variable	10	7	15	-	-	-	-	-	2.77 ± 0.33	-
<i>p</i> NPP	-	Variable	7	15	-	-	138.1 ± 4.2	1.76 ± 0.49	1.3	-	-
<i>p</i> NPP	3	Variable	-	15	-	-	128.2 ± 4.4	1.15 ± 0.61	1.7	-	-
<i>p</i> NPP	3	Variable	7	15	-	-	63.1 ± 3.8	0.92 ± 0.28	1.4	-	-
\mathbf{K}^+	-	10	7	Variable	-	-	134.2 ± 4.5	9.60 ± 2.04	2.2	-	-
\mathbf{K}^+	3	10	-	Variable	-	-	133.3 ± 3.3	6.00 ± 1.50	2.0	-	-
\mathbf{K}^+	3	10	7	Variable	-	-	59.5 ± 4.0	2.79 ± 0.41	1.5		
$\mathrm{NH_4}^+$	-	10	7	-	Variable	-	122.2 ± 5.2	9.02 ± 2.51	3.1	-	-
$\mathrm{NH_4}^+$	3	10	-	-	Variable	-	127.9 ± 4.2	6.00 ± 1.10	2.2	-	-
$\mathrm{NH_4}^+$	3	10	7	-	Variable	-	61.9 ± 3.7	5.46 ± 0.64	2.1	-	-
Mg^{2+} Mg^{2+} Mg^{2+} Mg^{2+} Mg^{2+} Mg^{2+}	-	10	Variable	15	-	-	135.1 ± 5.0	2.98 ± 0.59	2.2	-	-
Mg^{2+}	0.5	10	Variable	15	-	-	129.6 ± 3.0	2.25 ± 0.85	1.2	-	-
Mg^{2+}	1	10	Variable	15	-	-	126.4 ± 3.7	2.99 ± 0.66	2.6	-	-
Mg^{2+}	2	10	Variable	15	-	-	-	-	-	4.41 ± 0.69	-
Mg^{2+}	3	10	Variable	15	-	-	-	-	-	4.81 ± 0.71	-
Na ⁺	-	10	7	15	-	Variable	-	-	-	-	16.7 ± 3.65
Na^+	3	10	-	15	-	Variable	-	-	-	-	15.2 ± 3.42
Ouabain	-	10	7	15	-	-	-	-	-	2.27 ± 0.62	-
Ouabain	3	10	-	15	-	-	-	-	-	2.13 ± 0.97	-

448 **4. DISCUSSION**

We provide a comprehensive analysis of the effects of Co^{2+} on the modulation *in* 449 *vitro* of the K⁺-phosphatase activity in a gill (Na⁺, K⁺)-ATPase from the blue crab 450 Callinectes danae. Depending on Mg²⁺, Co²⁺serves as both stimulator and inhibitor of 451 K⁺-phosphatase activity. Without Mg^{2+} , Co^{2+} stimulates activity as does Mg^{2+} , although 452 with a \approx 4.5-fold greater affinity. With Mg²⁺, activity is almost completely inhibited by 453 Co^{2+} , while ouabain inhibition is unaffected. Substitution of Mg²⁺ by Co^{2+} slightly 454 increases enzyme affinity for K^+ and NH_4^+ . Mg^{2+} displaces bound Co^{2+} from the Mg^{2+} -455 binding site in a concentration dependent manner; however, Co²⁺ does not displace 456 bound Mg^{2+} even at elevated concentrations. Saturation by Co^{2+} of the Mg^{2+} -binding 457 site does not affect substrate recognition by the enzyme. 458

K-phosphatase activities estimated with Mg^{2+} or Co^{2+} are similar and their 459 overlapping pNPP saturation curves show comparable cooperative effects; their similar 460 $K_{0.5}$ values suggest that Co^{2+} saturation of the $\mathrm{Mg}^{2+}\text{-binding}$ site does not affect 461 substrate recognition. The high stability constants for the Co^{2+} -pNPP (130 mol L⁻¹, [76]) 462 and Mg^{2+} -pNPP (170 mol L⁻¹, [77]) complexes suggest that negligible free metal ions 463 464 are present at millimolar metal ion concentrations, i.e., the metal-pNPP complex is the true enzyme substrate. The lower apparent dissociation constant for Co^{2+} (K_{0.5}= 1.15 ± 465 0.61 mmol L⁻¹), close to that for Mg²⁺ (K_{0.5}= $1.76 \pm 0.49 \text{ mmol L}^{-1}$), is comparable to 466 the enzyme from Cancer pagurus axonal membranes despite its 2-fold greater 467 maximum *p*NPP hydrolysis rate [78]. 468

Millimolar Mg^{2+} concentrations are required for K⁺-phosphatase activity of the 469 C. danae (Na⁺, K⁺)-ATPase and, like the mammalian enzyme [58,79], no detectable 470 activity can be measured without this ion. The millimolar Mg^{2+} or Co^{2+} concentrations 471 necessary for maximum K⁺-phosphatase activity exclude the likelihood of metal binding 472 other than Mg^{2+} or Co^{2+} to the Mg^{2+} -binding site during the catalytic cycle [56]. The 473 inhibition by free Mg^{2+} or Co^{2+} of K⁺-phosphatase activity may result from competition 474 with K^+ for the Mg²⁺-binding site [56,61,80] or to excess Mg²⁺ bound during the E2K 475 conformation, decreasing affinity for *pNPP* [52,56,81]. 476

477 Co^{2+} can substitute for Mg^{2+} , stimulating K⁺-phosphatase activity more efficiently 478 (K_{0.5} ≈4.5-fold lower). However, Co²⁺ does not displace Mg²⁺ from the Mg²⁺-binding 479 site of the *C. danae* enzyme. Inhibition by excess Co²⁺ likely results from Co²⁺ binding at 480 a site different from the Mg²⁺-binding site. Two distinct Mg²⁺ binding sites are known

for crustacean [73] and mammalian [50] (Na^+ , K^+)-ATPases, although only one is 481 relevant for pNPPase and ATPase activities [82]. Like Co^{2+} , Mn^{2+} also stimulates sheep 482 kidney (Na⁺, K⁺)-ATPase activity, the $K_D = 0.88 \ \mu mol \ L^{-1}$ for Mn^{2+} binding being very 483 similar to the kinetic constant for ATP hydrolysis [83]. Co^{2+} stimulates dog kidney outer 484 medulla (Na⁺, K⁺)-ATPase activity by substituting for Mg^{2+} [20]. Differently from Co^{2+} , 485 Cu²⁺ inhibits the gill (Na⁺, K⁺)-ATPase activity of rainbow trout Oncorhynchus mykiss 486 [84], and the K⁺-phosphatase and (Na⁺, K⁺)-ATPase activities of rabbit kidney [85] by 487 directly interfering with Mg²⁺ binding, affecting Mg•ATP hydrolysis. 488

The similar V_M and $K_{0.5}$ values for stimulation by K^+ or NH_4^+ of K^+ -phosphatase 489 activity with Mg^{2+} or Co^{2+} suggests that NH_4^+ binds to the same site as K^+ during the 490 catalytic cycle, independently of Mg^{2+} or Co^{2+} bound to the Mg^{2+} -binding site. While 491 K⁺-phosphatase activity is inhibited to same degree by Mg^{2+} and Co^{2+} , the 2-fold greater 492 $K_{0.5}$ for NH₄⁺ (5.46 mmol L⁻¹) compared to K⁺ (2.79 mmol L⁻¹) suggests that Co²⁺ 493 binding to a different Mg^{2+} -binding site affecting enzyme interaction with NH_4^+ . 494 Without Na⁺, stimulation by K^+ of K^+ -phosphatase activity involves two K^+ binding 495 496 sites: one that regulates pNPP access to the phosphatase site, the other increasing 497 catalytic activity [86]. ATP binding to the low-affinity substrate binding site induces a conformational change in the cytoplasmic domain of the enzyme attributed to the E2 to 498 E1 transition; the subsequent binding of Mg^{2+} to the enzyme•ATP complex induces a 499 new conformational change that facilitates the E1 to E2 transition [87]. Like Mg²⁺, Co²⁺ 500 may induce a conformational change, stimulating the enzyme during *p*NPP hydrolysis. 501

502 K⁺-phosphatase activity can be stimulated by Tl^+ , Rb⁺ or NH₄⁺ to rates similar to K^+ stimulation while Cs⁺ and Li⁺ exhibit lower stimulation (5-30%) [88,89]. For 21‰ 503 (low salinity)-acclimated C. danae, Rb⁺ stimulates gill pNPPase activity by 1.5-fold 504 compared to K^+ [63]. NH₄⁺ may sustain ATP hydrolysis by replacing K^+ [89,90] and is 505 actively transported by crustacean and vertebrate enzymes [91,92]. Together with K^+ , 506 NH_4^+ synergically stimulates ATP hydrolysis by the C. danae (Na⁺, K⁺)-ATPase 507 through an additional increment, strongly influenced by Mg²⁺ and Na⁺, underlying 508 ammonia excretion in crustaceans [90]. The E2 conformation is the main state 509 responsible for pNPP hydrolysis [58] and is likely the reason that K^+ -phosphatase 510 activity is not synergically stimulated by K⁺ and NH₄⁺. Synergic stimulation using 511 512 pNPP as a substrate is known exclusively for the shelling crab C. ornatus [63]. When 513 using ATP as a substrate, species-specific synergic stimulation is found in various crustaceans [73]. 514

The inhibition by Na⁺ of C. danae pNPPase activity independently of Mg^{2+} or 515 Co²⁺ is seen also for (Na⁺, K⁺)-ATPases from various sources [73], and reflects 516 competition by Na⁺ for the cytoplasmic K⁺-binding sites, favoring the E1 conformation 517 [50,56,58,93-95]. Na⁺ inhibition also may involve events other than the simple binding 518 of the ion. To illustrate, the synergistic stimulation by Na⁺ and K⁺ (3%) of pNPPase 519 activity in the electric organ of *Electrophorus electricus* [86] suggests that both ions 520 bind to different sites. Both 10 mmol L^{-1} Na⁺ and 15 mol L^{-1} K⁺ inhibit the *C. danae* 521 pNPPase activity by $\approx 40\%$, as seen in the freshwater shrimp Macrobrachium olfersii 522 523 [61] and crabs *Cancer pagurus* [95] and *C. ornatus* [63] under identical assay conditions. Na⁺ inhibits pNPPase activity allosterically in various mammalian and 524 crustacean (Na⁺, K⁺)-ATPases [61,63,86,89,95] including M. olfersii [61] and C. 525 526 *pagurus* [95]. Thus, considering a two K^+ -binding site model [86], at low (10-fold less than K^+) concentrations, Na⁺ competes for the high-affinity K^+ binding site, leading to 527 allosteric effects. At high concentrations (similar to K^+), Na⁺ competes for the low-528 affinity K⁺-binding site, reducing maximum hydrolysis rate. 529

Co²⁺ does not affect ouabain binding to the *C. danae* gill (Na⁺, K⁺)-ATPase, the single inhibition curve and K_I being very similar to Mg²⁺. Most species, except the red river crab *Dilocarcinus pagei* [96], exhibit a single ouabain inhibition curve independently of substrate [62,63,72,97,98]. The K_I for ouabain inhibition with Co²⁺ lies in the range for various (Na⁺, K⁺)-ATPases [73]. It should be noted that only for the cerebromicrovascular (Na⁺, K⁺)-ATPase, Pb²⁺ and Al³⁺ caused selective alterations in ATP hydrolysis inhibiting and stimulating ouabain binding, respectively [99].

Our findings reveal that without Mg^{2+} , Co^{2+} stimulates the gill (Na⁺, K⁺)-ATPase 537 to levels similar to Mg^{2+} . Without Mg^{2+} , Co^{2+} stimulates K⁺-phosphatase activity 538 similarly although with a \approx 4.5-fold greater affinity than with Mg²⁺, which almost 539 completely inhibits K^+ -phosphatase activity. Mg^{2+} displaces Co^{2+} from the Mg^{2+} -binding 540 sites in a concentration dependent manner. Ouabain inhibition is identical with Co^{2+} or 541 Mg^{2+} . Saturation by Co^{2+} of the Mg^{2+} -binding sites does not affect substrate recognition 542 by the enzyme. Given the complex interactions between heavy metal ion contaminants 543 and enzymes, their toxic effects at the molecular level are poorly understood. Our 544 findings contribute to elucidate partly the mechanism of action of Co^{2+} on a crustacean 545 gill (Na⁺, K⁺)-ATPase. 546

547

549 Acknowledgements

The authors thank the Instituto Chico Mendes de Conservação da Biodiversidade, Ministério do Meio Ambiente (ICMBio/MMA) for authorizing collecting permit #29594-18 to JCM, and INCT-ADAPTA II (Instituto Nacional de Ciência e Tecnologia para Adaptações da Biota Aquática da Amazônia, ADAPTA-II) with which FAL's laboratory is integrated, and the Amazon Shrimp Network (Rede de Camarão da Amazônia).

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922 923 Funding information

This investigation was financed by research grants from the Fundação de 924 925 Amparo à Pesquisa do Estado de São Paulo (FAPESP 2016/25336-0 and 2019/21899-926 8), Fundação de Amparo à Pesquisa do Estado de Minas Gerais (FAPEMIG APQ-01893-16), Conselho de Desenvolvimento Científico e Tecnológico (CNPq 927 458246/2014-0), and in part by INCT ADAPTA II (CNPq 465540/2014-7) and the 928 929 Fundação de Amparo à Pesquisa do Estado do Amazonas (FAPEAM 062.1187/2017). FAL (302072/2019-7), and JCM (303613/2017-3) received Excellence in Research 930 scholarships from CNPq. LMF, CMM and MICC received scholarships from the 931 932 Coordenação de Aperfeiçoamento de Pessoal de Nível Superior (CAPES, Finance code 933 001).

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936 Francisco A. Leone: Conceptualization, Formal analysis, Resources, Funding 937 acquisition, Methodology, Supervision, Project administration. Writing original draft, 938 Review & Editing. Leonardo М. Fabri: Methodology, Investigation, Conceptualization, Writing original draft, Review & Editing. Cintya M. Moraes: 939 940 Methodology, Investigation, Writing original draft. Maria I. C. Costa: Methodology, Investigation, Writing original draft. Daniela P. Garcon: Conceptualization, 941 Methodology, Formal analysis, Funding acquisition, Writing original draft, Review & 942

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946 **Data availability**

947 The datasets generated and/or analyzed during this study are available from the 948 corresponding author on reasonable request.

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950 **Declaration of competing interests**

All authors certify that they have no affiliations with or involvement in any
organization or entity with any financial or non-financial interest in the subject matter or
materials discussed in this manuscript.

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955 Ethical approval studies in animals

This investigation complies with all local, state, federal and international
guidelines as regards the use of invertebrate animals in scientific research. This study
also complies with the ARRIVE guidelines.