Multiple Intermediates in the Detergent-Induced Fusion of Lipid Vesicles

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Abstract

Detergent-induced vesicle interactions, critical for applications including virus inactivation, varies according to the detergent type and membrane composition, but the underlying mechanistic details remain underexplored. Here, we use a lipid mixing assay based on Förster resonance energy transfer (FRET), and single-vesicle characterization approaches to identify that sub-micron sized vesicles are induced to fuse by the non-ionic detergent Triton-X 100. We demonstrate that the process is a multi-step mechanism, characterized by discrete values of FRET efficiency between membrane-embedded fluorophores, and involves permeabilization, vesicle docking, hemi-fusion and full lipid mixing at sub-solubilizing detergent concentrations. We also dissect the kinetics of vesicle fusion to surface-tethered vesicles using a label-free quartz-crystal microbalance with dissipation monitoring approach, opening a platform for biotechnology applications. The presented strategies provide mechanistic insight into the dynamics of vesicle fusion and have implications for applications including drug delivery and sensor development where transport and manipulation of encapsulated cargo is essential.

Introduction

Detergent-membrane interactions are critical for biotechnological applications including membrane-protein extraction[1, 2], cell lysis[3], targeted drug delivery[4], and virus inactivation[5, 6]. In this context, the widely used non-ionic detergent Triton X-100 (TX-100) plays a key role by efficiently and effectively solubilizing membrane bilayers. Open questions regarding the interaction, which involves membrane fusion as a key component[7, 8, 9], include elucidating the detergent's precise molecular-level mode of action, its role in lipid-lipid interactions, and the nature and morphology of the fused species.

Thermodynamic approaches aimed at characterizing the TX-100-membrane interaction have typically used highly-controllable model-membrane systems such as supported bilayers or synthetic unilammelar vesicles (UVs) to unveil concentration requirements necessary for solubilization[10], and to disentangle the influence of membrane phase, composition, charge, and detergent: lipid ratio on the interaction[11, 12, 13]. In this context, UVs are classified as small (SUVs), large (LUVs) or giant (GUVs) reflecting vesicle sizes of < 100 nm, 100 nm - 1 μm and > 1 μm respectively. GUVs in particular have played a key role in interrogating the TX-100 membrane interaction[14] with measurements broadly confirming that TX-100 induces solubilization at concentrations above its critical micellar concentration (CMC). However, dynamic light scattering (DLS) and turbidity approaches have also revealed an intriguing growth in vesicle size under sub-solubilizing conditions[7, 8, 9, 15], pointing towards morphological changes such as membrane swelling and fusion[8] that could be harnessed for biotechnological applications.

To assess the swelling aspect, single-vesicle characterization approaches based on the fluorescence imaging of GUVs have revealed detergent-induced surface-area enhancements concurrent with the formation of dynamic pores[16, 17, 18]. Similar work also revealed that the membrane tension on both micron- and 100 nm-diameter vesicles decreases with TX-100 concentration, and the detergent introduces phase separation events in vesicles incorporating cholesterol[13, 18]. More recently, we demonstrated that variations in the efficiency of Förster resonance energy transfer (FRET) between fluorescently labelled lipids in LUVs quantitatively reports on their expansion upon TX-100 addition[19], and combinations of molecular dynamics simulations and fluorescence approaches have revealed dynamic membrane bending events and the formation of local invaginations all prior to micellization[20].

These results, and others[21], are broadly supportive of a dynamic model for TX-100 induced membrane solubilization that involves detergent saturation on the membrane surface, the formation of mixed detergent-lipid micelles, structural changes within the intact membrane, and the release of mixed micelles to solution. A more quantitative extension of this model involves the formation of phase- and composition-sensitive pores, content leakage, variations in membrane curvature and local changes to the bilayer fluidity[22]. However, the molecular mechanisms underpinning another major class of event - namely TX-100 induced vesicle fusion - are not well characterized owing in part to a lack of experimental techniques capable of capturing the entire process.

In the context of vesicle fusion, optical microscopy experiments based on the imaging of unlabelled GUVs incubated with fluorescently tagged native membranes at sub-solubilizing TX-

100 concentrations have indicated the presence of fusion by the appearance of fluorescence at
the GUV site[23]. Using similar approaches, fusion between proteoliposomes and native
membranes has been harnessed to enable reconstitution of transmembrane proteins into GUV
bilayers after TX-100 addition[7, 8, 9, 24]. Phase contrast imaging and DLS measurements also indicate
that the degree of fusion is dependent on the lipid composition, suggesting that the TX-100-
membrane interaction may be a much more complex and multifaceted process. Confocal-based
fluorescence experiments have also been employed to assess permeable GUV fusion, and
although fusion was rarely observed, the results point towards a situation where fusion can occur
due to casual contact between pores on adjacent membranes[17]. Unfortunately, conventional
microscopy techniques are limited in their ability to resolve structural changes in objects smaller
than 5-10 μm, can only quantify macroscopic changes in size and packing density and they lack
the ability to follow molecular-level structural rearrangements on the nanoscale. There is
therefore demand for alternative methods that can monitor the fusion of individual vesicles with
sizes comparable to LUVs. In this regard, molecular dynamics simulations have alluded to a stock-
pore mechanism of spontaneous bilayer fusion where initial vesicle docking precedes the
formation of an hourglass-shaped connection between two adjoining membranes before hemi-
fusion and full lipid mixing[17, 25, 26]. Whether such a mechanism occurs in detergent-rich media,
and whether these states can be controllably harnessed remains an open question.

Inspired by these insights and in search of the mechanistic origins of TX-100 induced vesicle
fusion, we adopted a range of ensemble- and single-vesicle characterization approaches to
interrogate the structural integrity of highly curved LUVs composed of POPC (1-palmitoyl-2-
oleoyl-glycero-3-phosphocholine) in response to TX-100. The LUVs used here differ from low-
curvature GUV species by displaying a maximal principal curvature[27] of ~10^7m^{-1}, which is an order
of magnitude greater than those typically observed in GUVs. By employing particle sizing
approaches including DLS and fluorescence correlation spectroscopy (FCS), we first confirm an
enhancement in vesicle size under sub-solubilizing conditions. We next showcase a steady-state
and time-resolved assay based on FRET[28, 29], to quantitatively assess the degree of TX-100 induced
lipid mixing between vesicles. Based on the magnitude of the observed FRET efficiency between
fused vesicles, coupled with morphological changes observed via electron microscopy
approaches, we provide evidence that the detergent-induced fusion process is a time-dependent
and multi-step process comprising vesicle docking, hemifusion and full lipid mixing prior to
solubilization. While we have previously used FRET-based methods to quantitatively report on
vesicle swelling[19, 30, 31], here we take advantage of enhancements in the FRET efficiency to
quantify the magnitude and state of fusion. To further characterize the interaction, we also
deploy a calcium-sensing assay to confirm TX-100 induces vesicle permeabilization during the
process. Finally, to demonstrate the potential of TX-100 to facilitate vesicle fusion of freely-
diffusing vesicles to surface-immobilized vesicles, and to extract kinetic details of the process, we
deployed an acoustic-sensing approach based on quartz-crystal microbalance with dissipation
monitoring (QCM-D). In this regard, the QCM-D approach offers highly sensitive real-time
monitoring of surface mass and viscoelasticity, without the need for fluorescent labels. This not
only opens exciting possibilities for the controllable delivery of encapsulated cargo to a substrate
but may also support a range of bionanotechnological applications including diagnostic sensing,
drug delivery and the formation of biomaterials.

Our discovery of a multi-step and controllable fusion process in both freely diffusing and surface-
tethered vesicles at sub-solubilizing detergent concentrations provides clues to the underlying

TX-100-membrane interaction and may have important implications for the transport of
membrane proteins such as proton pumps and transport receptors[23]. We also expect the
presented techniques to have general applicability for assessing interactions between highly
curved lipid vesicles and a wide range of fusogens, such as neurotoxic proteins with important
biomedical significance.

Results

Triton X-100 Alters the Morphology of Vesicles at Sub-Solubilizing Concentrations

LUVs composed of POPC lipids were prepared by the extrusion method and confirmed to be ~200
nm in diameter by DLS (**Supplementary Figure 1**). We first employed FCS and DLS to assess
morphological changes as TX-100 was progressively added to a solution containing freely diffusing
vesicles. Both techniques are sensitive to the diffusion coefficient of vesicles in solution, and thus
quantitatively report on their hydrodynamic radius via the Stokes-Einstein relationship[32]. The FCS
correlation curves associated with freely diffusing LUVs labelled with 0.1 % of the fluorescent
membrane dye DiI progressively shifted towards longer lag times at detergent concentrations
below the reported CMC (~0.2-0.3 mM)[3, 33] (**Figure 1a**) - a finding that warranted further attention
given non-ionic surfactants typically only achieve membrane solubilization once above the CMC.
Here, a reduction of 40 % in vesicle diffusion coefficient, from an initial starting value of 1.28 ±
0.01 $\mu m^2 s^{-1}$ (±SD) in the absence of TX-100, was observed, corresponding to an increase in
hydrodynamic diameter (d_H) from 168 ± 1 nm to 290 ± 5 nm. In all cases, the FCS curves were best
fitted to a single diffusing component model (see **Methods**) with low residuals, and notably lacked
a significant population of micelles exhibiting higher diffusion coefficients. In line with previous
work[19], we attributed these observations to vesicle swelling, docking, fusion, aggregation, or
combinations of these effects in solution. At TX-100 concentrations beyond the CMC, the particle
radius reduced to ~3.5 nm which we assigned to solubilization and the formation of mixed
detergent-lipid micelles. DLS also confirmed a positive shift in size distribution, though we note
in these experiments, the lipid concentration was increased by two orders of magnitude relative
to the FCS work, which we hypothesised could facilitate fusion. Here, correlation curves obtained
from the intensity of scattered light also progressively shifted towards longer lag times (**Figure
1b**). Without detergent, the vesicles displayed a lognormal size distribution, as expected for freely
diffusing species, peaking at 182 nm with a low polydispersity index of 0.16 ± 0.01 (**Figure 1c,
Supplementary Figure 2**). At detergent concentrations approaching the CMC, the correlation
curves then revealed a 1.7-fold increase in vesicle size (**Figure 1c**) and ~4-fold increase in
polydispersity index (**Figure S2**), suggestive of mixed species in the ensemble. At 0.5 mM TX-100,
most particles displayed d_H in the range 6 – 10 nm, consistent with mixed detergent-lipid micelles.
However, we also observed objects with sizes ranging from 250 – 480 nm, in line with expanded,
fused and/or aggregated species, and to a lesser extent, particles with intermediate diameters,
which could be indicative of bicelles and cylindrical micelles[34] (**Figure S2**). When intensity
distributions were converted into volume distributions, similar trends were observed (**Figure S2**).
Caution should be taken in interpreting such distributions due to underlying assumptions
related to vesicle morphology and noise amplification, but below the CMC, the distributions
broadened with TX-100 addition, and a mixture of species was exposed at 0.5 mM, in line with
our intensity-based analysis.   While a direct comparison between FCS and DLS datasets is not

entirely straightforward owing to variations in the initial lipid concentration, both analyses are
broadly supportive of a progressive increase in particle size below the CMC.

Triton X-100 Induces Vesicle Fusion

To directly test for fusion, we next employed a lipid mixing assay to monitor interactions between
LUVs containing 2 % DiI (FRET donor) and those containing 2 % DiD (FRET acceptor) (**Figure 2a**).
Direct excitation of DiI in the absence of fusion thus yields low sensitized emission from the
acceptor, however, upon vesicle fusion, lipid mixing brings the FRET pairs into close proximity,
triggering enhanced sensitized acceptor emission as a consequence of non-radiative energy
transfer[28, 29, 35]. A key benefit of the approach is that the total number of fluorophores remains
constant throughout the fusion process, allowing for intermediate states, corresponding to
different levels of lipid mixing, to be observed by quantifiable variations in the FRET efficiency
(Equation 2). Ensemble FRET measurements were performed as an initial step to quantify
interactions between vesicles containing DiI and DiD in response to TX-100. Here, we measured
the efficiency of energy transfer using the *RatioA* parameter (Equation 3) which compares the
amount of sensitized acceptor emission with that obtained by direct excitation of the acceptor.
In the absence of TX-100 the intensity of sensitized acceptor emission obtained from a 1: 3
mixture of DiI: DiD vesicles was minimal, then progressively increased upon TX-100 addition
towards the CMC and was anti-correlated with quenching of the donor (**Figure 2b**,
Supplementary Figure 3). The anti-correlated nature of the DiI and DiD signals represents the
classic hallmark of FRET[36], providing assurance that the observed changes are due to energy
transfer. Beyond the CMC, the apparent FRET efficiency then decreased towards its initial value,
reflected by an enhancement of the donor signal and corresponding reduction in sensitized
acceptor emission (**Figure 2b, Supplementary Figure 3**). The increase in *RatioA* observed below
the CMC indicates a progressive decrease in the mean donor-acceptor distance, supporting a
model where progressive addition of TX-100 shifts the vesicle population from intact to fused
species prior to solubilization and the formation of mixed micelles (**Figure 2c**). Indeed, this is
reflected in the *RatioA* response, where a progressive increase towards a peak of 0.49 ± 0.03 at
the CMC was observed, suggestive of morphological changes in the ensemble that results in close
proximity of the dyes, followed by a reduction towards its initial value, indicative of solubilization
(**Figure 2d**). To provide confidence that the observed increase in *RatioA* was due to lipid mixing,
and not lipid exchange between separated vesicles, we also assessed the response of the labelled
vesicles in the presence of various cyclodextrins. As previously highlighted, methyl-β-cyclodextrin
(MβCD) and 2-hydropropyl-β-cyclodextrin (HPβCD) induce lipid exchange at sub-solubilizing
concentrations[37], and both were added here to a solution of vesicles identical to those shown in
Figure 2b. We observed that the sensitized acceptor emission (and corresponding *RatioA* and
E_{FRET} signals) increased slightly above the baseline (E_{FRET} ~ 0.09) upon both MβCD and HPβCD
addition (**Supplementary Figure 4**), and that the FRET signal was enhanced when DiI and DiD were
replaced with Lipo-Cy3-CO and Lipo-Cy5-N$_3$ (**Supplementary Figure 4**). As previously reported,
the Lipo derivatives are structurally similar to DiI and DiD (**Supplementary Figure 4**), but their
FRET is enhanced upon cycloaddition[38]. We interpreted the combined data as evidence that both
cyclodextrins can, to some extent, exchange DiI and DiD, however, this may not be as efficient as
the exchange of unlabelled lipids. To explicitly test for DiI and DiD exchange, we also performed
an experiment whereby 200 nm sized POPC vesicles labelled with 2 % DiI were separated from
vesicles containing 2 % DiD by a dialysis membrane (MWCO = 3 kDa) (**Supplementary Figure 5**).
In this case, we prevented DiI and DiD labelled vesicles from directly interacting, but allowed low

molecular weight species, including free lipids, to diffuse across the membrane (**Supplementary**
Figure 5). However, after 5 hours of dialysis in the absence and presence of TX-100, we observed
no evidence of DiD absorption or sensitized emission after observing the solution containing DiI.
We did, however, observe increased levels of scattering upon TX-100 addition, which we
interpreted as a signature of fusion. Taking the combined data together, the implication is that
the data shown in **Figure 2b** is more likely due to DiI and DiD mixing as opposed to DiI and DiD
exchange.

When additional control experiments involving the addition of TX-100 to a 1:3 mixture of LUVs
lacking fluorophores and those containing 2 % DiD were performed, similar results were obtained
(**Supplementary Figure 6**), indicating further that the observed increase in *RatioA* shown in **Figure**
2b is due to energy transfer and sensitized acceptor emission, as opposed to photophysical
phenomenon such as fluorescence de-quenching. We also note that the amounts of DiI and DiD
per vesicle and the DiI: DiD ratio were optimized (1: 3, 2 % of each dye) to maximize the magnitude
of the FRET response across the interaction (**Supplementary Figure 7**).

To confirm the presence of an energy transfer mechanism between DiI and DiD during the
interaction, we also evaluated the amplitude-weighted fluorescence lifetime of DiI in the absence,
τ_D, and presence, τ_{DA}, of DiD. In all cases, the lifetime decays displayed bi-exponential behaviour
after deconvolution with the instrument response function (**Supplementary Figure 8**) (Equation
4). As TX-100 was progressively titrated at concentrations below the CMC, τ_D progressively
increased from an initially self-quenched value of 0.53 ± 0.03 ns to 1.36 ± 0.02 ns, representing
the unquenched DiI lifetime (**Supplementary Figure 9**).We also found that τ_{DA} remained largely
unchanged from its initial starting value of ~0.54 ns, then sharply increased towards ~1.2 ns as
the concentration of TX-100 increased beyond the CMC (**Supplementary Figure 9**). We note that
τ_D and τ_{DA} displayed similar initial starting values which we interpreted as evidence of minimal
spontaneous fusion prior to injection of TX-100. The corresponding variation in the efficiency of
energy transfer, *E* (Equation 5), which takes into account variations in τ_{DA} relative to τ_D, then
displayed several common features to the *RatioA* measurements, including (i) a progressive
increase as the TX-100 concentration approached the CMC, (ii) a turning point at the CMC and
(iii) a progressive reduction towards its initial value during the solubilization step (**Figure 2e**). The
observed increase in *RatioA* and *E*, which effectively remove contributions from DiI de-quenching,
provide confidence of a TX-100-induced solubilization mechanism comprising vesicle fusion (high
FRET states) below the CMC, followed by solubilization and mixed micelle formation above the
CMC (low FRET states) (**Figure 1c**).

We also note the time-dependent increase in sensitized DiD emission following TX-100 injection
followed bi-exponential kinetics, pointing further towards a fusion process involving multiple
stages. In the absence of TX-100, DiD emission remained largely invariant across the observation
time window, also indicating a lack of spontaneous fusion, however, upon TX-100 addition, bi-
exponential kinetics were monitored where the amplitude weighted time constant was 273 ± 6 s
in the presence of 0.1 mM TX-100, and 396 ± 11 s in the presence of 0.25 mM (**Figure 2f,**
Supplementary Table 1). While the precise nature of the bi-exponential kinetics required further
exploration, we hypothesised that the various components are more likely to represent fusion
steps (for example hemi-fusion and full lipid mixing) versus combinations of lipid exchange and
fusion.

To further characterize the interaction and assess the degree of membrane permeabilization and
solution exchange during the TX-100 LUV interaction, we also implemented an approach based
on the measurement of Ca^{2+} entry into vesicles encapsulating the fluorescent calcium indicator
Cal-520. This approach is similar in nature to other permeabilization assays[30, 39] whereby Ca^{2+} flux
into intact, surface-tethered vesicles triggers a concentration dependent intensity enhancement
of Cal-520 at ~520 nm, as illustrated in **Supplementary Figure 10**. Prior to experimentation,
nonencapsulated molecules were removed by size exclusion chromatography as detailed in the
Methods. Immobilized vesicles were then imaged in buffer solution containing 10 mM Ca^{2+} with
and without addition of TX-100. We imaged multiple fields of view per condition, enabling
quantification of intensity distributions from several thousand vesicles before adding buffer
solution containing TX-100 and Ca^{2+}. In the absence of TX-100 and Ca^{2+}, we identified ~100
vesicles per 25 x 50 μm field of view, and the intensity distribution fitted to a log-normal model
with a peak position of 135 counts / 100 ms. When 10 mM Ca^{2+} was flushed across the same
vesicles, the number of fluorescent foci remained invariant, and the intensity distribution was
comparable, peaking at 147 counts/ 100 ms. A similar dataset was observed when the vesicles
were incubated with 0.025 mM TX-100 and 10 mM Ca^{2+}, however in the presence of 0.05 – 0.1
mM TX-100, we observed brighter foci and clearly discernible shifts in the distributions from low-
to-high intensities. In such cases, a 2-fold or greater increase in the probe's emission intensity
was observed, in line with those previously identified[30, 40]. Beyond 0.1 mM, the number of spots
decreased ~10-fold and each had an intensity below the baseline level, indicating Cal-520 leakage
from the LUVs. Control experiments performed independently on free Cal-520 in solution
indicated that the dye's fluorescence is insensitive to TX-100, and that it responds to Ca^{2+} in
detergent-rich media (**Supplementary Figure 11**), providing confidence that the observed
intensity enhancements are due to membrane permeabilization and Ca^{2+} transport across the
membrane. Importantly, the number of surface-immobilized vesicles on the surface remained
largely unchanged as TX-100 up to 0.1 mM was introduced. When compared with the FRET
measurements shown in **Figure 2d and 2e**, which indicate substantial levels of lipid mixing at
comparable TX-100 concentrations, the implication is that combinations of membrane
permeabilization and content exchange occur concurrently with fusion events in the ensemble.
However, whether content can be controllably recovered after release from the vesicles during
the permeabilization and/or solubilization stages remains an open question.

The Fusion Mechanism involves Docking, Hemifusion and Full Lipid Mixing

To further explore the observed structural changes, and to dissect the mechanism of fusion, we
next performed single-vesicle imaging via a custom-built wide-field objective-type total internal
reflection fluorescence microscope that enables the simultaneous imaging of DiI and DiD
emission[41]. Here, POPC vesicles incorporating 2 % DiI were induced to fuse with LUVs
incorporating 2 % DiD and 1 % biotin-PE. Fused species were then tethered to a glass coverslip
functionalized with biotinylated BSA and NeutrAvidin (**Figure 3a**). Vesicles were added to a
surface-functionalized coverslip after incubation with TX-100, and DiI and DiD fluorescence
trajectories were collected with 50 ms time integration. Fused vesicles were then identified by
the appearance of co-localized diffraction limited spots across both detection channels (**Figure
3b**) and lipid mixing between vesicles was observed via an appearance or enhancement in the
mean FRET efficiency per vesicle. In the absence and presence of 0 – 0.2 mM TX-100, the traces
obtained from > 500 individual FRET-active species were photostable over the duration of the
measurement (**Figure 3c, Supplementary Figure 12)** and the population distributions of FRET

efficiencies showed a major peak at ~0.1 (**Figure 3d**), suggestive of close contact or docking
between DiI and DiD-labelled vesicles, but without substantial lipid mixing. As suggested
elsewhere, a combination of factors, including Van der Waals, electrostatic and hydrophobic
interactions may be responsible for some spontaneous vesicle-vesicle interactions, even in the
absence of TX-100[42, 43]. As the TX-100 concentration then increased towards 0.5 mM, the traces
remained photostable (**Supplementary Figure 12**) and the low FRET population decreased in
favour of a broad distribution of high FRET states that reached a value of 0.7 at 0.5 mM, suggestive
of substantial lipid mixing events (**Figure 3d**). We note the observed shift in the distributions from
low to high FRET efficiencies broadly reflect the changes observed in the ensemble and the
observation of FRET states between 0.1 and 0.7 reflect the morphological heterogeneity of
intermediates that are structurally different from the fully fused state in which both leaflets have
fully mixed. We also note that as the TX-100 concentration progressively increased across the
titration, the total intensity per vesicle, defined as the sum of DiI and DiD emission intensities,
increased > 2-fold from 1325 ± 75 counts/ 50 ms at 0.1 mM TX-100 to 3070 ± 182 counts / 50 ms.
Since the particle size scales with the number of fluorophores[44], the data shown in **Figure 3d**
arises from the presence of large immobilized fusion products, as opposed to smaller mixed
micelles. We therefore assigned E_{FRET} = 0.7 as an indicator of full fusion, in which both leaflets of
the vesicle fully mixed, whereas intermediate states were assigned to hemi-fusion, where only
partial lipid mixing was achieved. In all cases, tri-Gaussian functions were fitted to the FRET
efficiency histograms, and used to identify thresholds of E_{FRET} < 0.25, 0.25 ≥ E_{FRET} ≤ 0.55 and E_{FRET}
> 0.55 for classification of docked, hemi-fused and fully fused species respectively. As shown in
Figure 3e, the percentage of species exhibiting full fusion increased 5-fold across the titration,
whereas the number of docked species progressively diminished. While the current experiments
do not inform on whether TX-100 promotes vesicle docking, its insertion into intact vesicles,
coupled with its hydrophobicity and ability to form mixed detergent-lipid regions, could trigger
vesicle-vesicle interactions via a route that involves vesicle destabilization as they dynamically
interact[15]. Nevertheless, the imaging of single fused species via FRET provides evidence for the
presence of intermediate states along the fusion process, with each state characterized by
discrete values of FRET efficiency and thus DiI-DiD separation distance.

To provide further evidence for the presence of multiple intermediates, we also acquired
scanning electron microscopy (SEM) micrographs of vesicles in the absence and presence of
fusion-promoting detergent concentrations. Without TX-100, the vesicles (N = 242) were
predominantly spherical, exhibiting a mean circularity, ϕ, of 0.84 ± 0.15, and a mean particle size,
d, of 246 ± 61 nm, with little evidence of fusion (**Figure 4a, Figure 4f, Supplementary Figure 13**).
Upon incubation with TX-100, the mean particle size then increased 2-fold (d = 528 ± 56), the
majority of species were elliptical (ϕ = 0.62 ± 0.22) (**Figure 4f**) and four major species were
identified. First, we observed adjoined vesicles without evidence of clefting, which we interpreted
as early fusion products that involve vesicle docking (**Figure 4b, Supplementary Figure 14**).
Further examination revealed vesicles within close proximity but with a clear cleft between
adjacent surfaces (**Figure 4c, Supplementary Figure 15**). Based on the similarity of these
structures to those observed in previous studies of protein-induced vesicle fusion[45, 46], we
assigned these events to intermediate hemi-fusion products involving two or more vesicles.
Finally, we observed the appearance of much larger micron-sized ellipses (**Figure 4d,**
Supplementary Figure 16) and larger conglomerates (**Figure 4e, Supplementary Figure 17**) which
we interpreted as evidence of multiple fusion events. While we previously identified that TX-100
leads to a 20 % increase in vesicle size through swelling[19], the ellipses observed here were more

than double the size of a single vesicle, and are therefore more likely to represent fully fused
species as opposed to single expanded vesicles. Closer inspection of the surface topography of
such species also indicated variations in texture, protrusions, and indentations which we
speculate could be indicative of pore formation and/or dissociation of membrane regions (**Figure**
4c).

Having demonstrated TX-100 induced morphological changes by SEM, we next used Cryo-TEM to
visualize interactions between adjacent membranes. Here, the majority of isolated vesicles were
intact and unilamellar, and in line with our prior SEM measurements, they exhibited a mean
particle size of 249 ± 78 nm and were predominantly spherical (ϕ = 0.95 ± 0.04) (**Supplementary**
Figure 18). In agreement with previous observations[47], we also identified lower populations (< 40
%) of vesicles containing multiple bilayers (**Supplementary Figure 19**). Based on the prior FRET
and SEM analysis, and in line with models predicted for extracellular vesicle fusion[48], we
hypothesized that TX-100 induced fusion intermediates consist of (i) docked vesicles showing
close contact between the lipid bilayers, but without lipid mixing, (ii) progressive fusion of the
outer and inner leaflets leading to a fusion pore and (iii) development of the fused pore into a
fully fused state where both leaflets had fully mixed. Indeed, we identified examples of where the
membranes of opposing vesicles were in parallel and within close proximity (< 5 nm), but with a
clear barrier separating the vesicle interiors (**Figure 4g**). We also identified structures where the
bilayer leaflets had merged at a ~25 nm diameter contact pore, and where the two bilayers
transitioned continuously from one vesicle to the other (**Figure 4h**). The fully fused species, after
expansion of the fusion pore and upon complete content mixing, was then captured as a larger
fusion product containing a single bilayer, with no evidence of a barrier preventing content mixing
(**Figure 4i**). In some cases, we also observed examples of where further vesicles had either docked
to or budded from the fully fused form (**Figure 4i**). Those species classified as docked, hemifused,
or fully fused comprised 31 % of the population in 0.15 mM TX-100 and had a mean calliper
diameter of 568 ± 86 nm, which we note represents a 2.3-fold increase relative to the intact
vesicles (**Supplementary Figure 18**). The intermediates also displayed a shift towards lower
circularity values, consistent with a global morphological change from spherical intact vesicles to
larger, elliptical fusion products (**Supplementary Figure 18**). While the current Cryo TEM analysis
cannot differentiate between further docking or budding events, we note that the smaller vesicles
are connected to the main vesicle by a narrow aqueous channel pointing towards docked species.
TX-100 has, however, been shown to induce vesicle budding in GUVs of similar composition[49, 50,]
[51], and the release of smaller species cannot be ruled out. Nevertheless, these results further
demonstrate the heterogeneous nature of the TX-100 induced fusion products.

TX-100 Facilitates Fusion between Freely-Diffusing and Surface-Immobilized Vesicles

Controllable vesicle fusion to surface-tethered vesicles has vast potential for a range of
applications including biosensing and drug delivery[52]. We therefore explored whether TX-100
could controllably fuse vesicles in solution with vesicles pre-immobilized on a solid surface,
allowing for real-time content delivery and solution exchange. In contrast to our TIRF
measurements, here we assessed the kinetics of accumulated mass transfer and variations in
surface viscoelasticity as vesicles from solution interacted with those already pre-immobilized to
a substrate via QCM-D. POPC vesicles incorporating 1 % biotin-PE were first tethered to a bovine
serum albumin coated SiO_2 substrate incorporating 2 % biotin via NeutrAvidin as described in

Figure 3a. The frequency and dissipation signals, reflecting surface mass and viscoelasticity,
respectively, were then recorded under equilibrium and exposure to a constant flow rate.
Supplementary Figure 20 shows representative QCM-D kinetics recorded upon injection of BSA-
Biotin, Neutravidin and biotinylated vesicles. After the sensors were saturated with vesicles, the
surface was washed with 50 mM Tris (pH 8) buffer to remove unattached species. Importantly,
the frequency and dissipation responses from the sensor showed little recover after the wash
step indicating successful surface tethering. Next, vesicles lacking biotin-PE at a final lipid
concentration of 0.16 mg/mL were flushed across the sensor surface in the absence and presence
of TX-100. Negligible vesicle deposition was noted in the absence of TX-100, indicated by the lack
of variation in both the frequency and dissipation responses (**Supplementary Figure 20**). When a
solution containing LUVs and 0.1 mM TX-100 was injected across the layer, an initial 20 Hz
reduction in frequency, concurrent with a 1.2-fold increase in dissipation was recorded, indicative
of (i) a mass gain over the period of the injection lasting tens of minutes and (ii) structural
rearrangements on the surface (**Figure 5a**). We assigned the bi-exponential nature of the
frequency decay to combinations of detergent interactions with the immobilized layer and
vesicle-vesicle interactions on the surface; however, the QCM-D traces are unable to clearly
distinguish between docked, hemi-fused and fully fused species. Experiments performed
independently to assess the interaction between freely diffusing TX-100 with surface-tethered
vesicles indicated only a 4-5 Hz shift upon injection (**Supplementary Figure 21**)[19], consistent with
a small mass gain on the sensor surface which we attribute to TX-100 incorporation into the
vesicle bilayers until saturation is reached[19]. While the QCM-D data does not report on TX-100
flip-flop across the bilayer, we expect this to occur rapidly, and simultaneously with its
incorporation[15]. Irrespective, the small shift observed here, indicates that the observed frequency
change shown in **Figure 5a** is mostly due to TX-100 induced vesicle-vesicle interactions on the
surface, as opposed to TX-100 incorporation into the surface-immobilized vesicles.

When the concentration of TX-100 was doubled, similar rate constants were observed (334 ± 4
s[-1]) for the frequency decay though we note that the magnitude of the relative frequency and
dissipation responses increased 1.5- and 2.5-fold respectively pointing towards an abundance of
fused states on the surface (**Figure 5b**). When both surfaces were washed with solution, only a
partial recovery in ΔF and ΔD was observed indicating that most of the deposited material
irreversibly attached to the immobilized layer (**Figures 5a and 5b**). However, when detergent
concentrations above the CMC were in the injectant, the deposited mass returned towards the
initial frequency and dissipation levels during the wash step, suggesting rupture of the freely
diffusing vesicles upon interaction with the surface, and the subsequent removal of lipid mass
(**Figure 5c**). Indeed, these assertions were further confirmed when variations in frequency were
plotted against dissipation across the timescale of the fusion interaction (**Supplementary Figure
22**). The slope of the dissipation changes versus frequency shown in **Supplementary Figure 22**
reflect variations in the viscoelastic properties of the immobilized layer across time. For detergent
concentrations below the CMC, these changes broadly indicate that mass gain coupled with
viscoelasticity enhancements on the surface precede minor levels of mass loss upon washing. At
concentrations above the CMC, the vesicles in solution are susceptible to lysis and dissociation: a
finding that we note is in line with our previous ensemble FRET approaches.

Discussion

Several studies provide support for the hypothesis that TX-100 induces fusion of sub-micron sized
vesicles. For instance, optical microscopy experiments have almost exclusively demonstrated
GUV fusion to fluorescently-tagged native membranes at sub-solubilizing TX-100 concentrations[23]
and similar approaches have been employed to achieve TX-100-induced fusion between
proteoliposomes and native membranes, resulting in the reconstitution of transmembrane
proteins into GUVs[7, 8, 24]. In confocal-based fluorescence experiments, fusion was observed
between permeable GUVs, suggestive that the process can occur through an initial contact
between pores on adjoining vesicles[17]. While the mechanisms of GUV- versus LUV-detergent
interactions have yet to be fully worked out, previous work indicates bilayer fluctuations,
solubilization kinetics, detergent concentration requirements and detergent packing densities
depend on the vesicle size and curvature[14, 25, 30, 53]. Thus, the development of techniques to dissect
the multi-step mechanism of membrane fusion, especially at the higher end of the membrane
curvature space are required.

To overcome this challenge, we implemented a robust in vitro fusion assay based on the
measurement of energy transfer between lipophilic membrane dyes during lipid mixing. While
FRET-based lipid mixing assays are not without shortcomings[54] (for example, our observation that
DiI is initially self-quenched suggests absolute FRET values may be underestimated and caution
should be taken for accurate distance determinations) we demonstrate that the assay unveils
multiple fusion intermediates as recorded by changes in the FRET efficiency and support our
observations by a range of ensemble and single-particle measurements. Indeed, the cumulative
data strongly supports vesicle fusion over alternative processes such as lipid exchange, evident
by the marked increase in particle dimensions observed via DLS, SEM and Cryo-EM. Under
symmetric TX-100 induced lipid exchange and dynamic equilibrium, the vesicle size distribution
should remain relatively constant, but our experimental observations reveal significant particle
size variations, pointing towards a fusion mechanism as the dominant process rather than lipid
exchange. We do however acknowledge that we cannot completely exclude the possibility that
some degree of POPC lipid exchange occurs, especially as vesicles in close proximity begin to
interact. In this context, molecular dynamics simulations probing the interaction may be valuable.
Nevertheless, the specific detergent employed and the speed through which the detergent leads
to fusion and solubilization, categorized as "fast" or "slow" based on the detergent's flip-flop rate
are highly likely to govern the underlying mechanism[15]. In instances of slow solubilization, the
process involves lipid extraction and/or the generation of micelles that are subsequently released
and exchanged with enclosed vesicles. Conversely, in the context of TX-100, the rapid flip flop
rate leads to the equal distribution of detergent molecules across both membrane leaflets,
saturation of the membrane surface, and a pathway towards solubilization that occurs via open
vesicular intermediates, which could facilitate fusion[25]. Here, examples of fused species induced
by TX-100 were visualized by SEM and Cryo-TEM, both of which provide information on the
morphology of the intermediates, and their occurrence under varying TX-100 conditions, and the
combination of experiments strongly implies that the fusion process is unidirectional, precedes
solubilization, and occurs via a mechanism that involves permeabilization, docking, hemifusion,
and full lipid mixing. This is striking for two reasons: first, the fusion process occurs at sub-
solubilizing detergent concentrations below the CMC and second, the fusion products observed
here are remarkably similar to those observed during protein-induced vesicle fusion and viral
fusion, raising the tantalising possibility that the observed intermediates are common, and share
structural similarities, regardless of the fusogen.
470

A direct quantitative comparison of our findings on LUVs to previous studies on both LUVs and
GUVs is challenging due to differences in membrane composition, vesicle size, the detergent-to-
lipid ratio used and the precise measurement method. Nevertheless, it is clear the fused LUVs
studied here, and those studied previously[13, 17, 23, 55], share common attributes. This includes
similar TX-100 concentration requirements to achieve fusion, observations of particle size and
surface area enlargements upon TX-100 addition below the CMC, and vesicle permeabilization
(which may suggest that TX-100 induced pores on adjacent vesicles facilitate fusion). Moreover
our observation of nm-sized mixed detergent-lipid micelles under solubilizing conditions is in line
with previous findings. Our studies on highly curved sub-micron sized vesicles are therefore
complementary to previous optical microscopy and light scattering measurements and open a
platform for exploring a matrix of environmental variables that may regulate the interaction.
482
Remarkably, our QCM-D results also show discrete steps of vesicle fusion and serve to highlight
the benefits of using TX-100 to controllably fuse vesicles to surfaces containing immobilized
vesicles: an important observation that may find direct relevance in a wide range of
biotechnological applications. For example, in targeted drug delivery, vesicles encapsulating
therapeutic agents are required to fuse to target sites, controllably releasing the encapsulated
molecule. Vesicle fusion to surfaces can also be employed to create biomimetic systems that
mimic natural cellular structures, or controllably modify and functionalize materials for
enhancement of properties including conductivity and chemical reactivity. There is also scope for
the fusion products to be used or manipulated in the context of sensor development, for instance
in the detection of specific analytes, or in microfluidic devices where vesicle fusion can be
employed to create well-defined membrane structures or facilitate the controllable transport and
manipulation of small liquid volumes. We previously suggested that TX-100, along with other non-
ionic surfactants induce swelling of lipid vesicles, but the absence of fusion observed with
detergents such as Tween-20, suggest that the fusion efficacy is detergent-dependent
(**Supplementary Figure 23**). Nevertheless, we expect the presented tools and techniques will be
used to further define the fusion mechanism when different vesicle subpopulations are
introduced, and to disentangle the impact of vesicle curvature, presence of lipid rafts and other
environmental factors (pH, ionic strength) which may regulate the interaction. Indeed, the
combined approaches and developed tools unambiguously dissect the underlying mechanism of
TX-100-induced vesicle fusion and could become an indispensable toolbox for further
characterizing the influence of fusogens, including those with important biomedical significance.
Moreover, dissecting the mechanism of LUV fusion is critical, not only for expanding our
fundamental knowledge of vesicle biology, but for the development of biocompatible and tissue-
specific delivery systems.

A major aspect of our results that deserves attention is that sub-solubilizing concentrations of TX-
100 were required to produce the fusion states described. While techniques such as isothermal
titration calorimetry have been applied extensively to quantify the thermodynamics of detergent
binding and partitioning below the CMC[12, 22, 56, 57], our work, which quantifies single vesicle
morphologies, implies that individual TX-100 monomers play a key role in initiating vesicle fusion.
This is particularly remarkable because non-ionic detergents typically achieve structural changes
only once above their CMCs. While further work is required to elucidate the precise role of the
molecule in initiating fusion, we hypothesise that sufficiently high local detergent densities on the
membrane may act as a nucleation site for docking, potentially via a pathway that involves
creation of pores. Our approaches enabled fusion intermediates to be observed up to detergent:

lipid ratios of ~ 5: 1, suggesting that morphological rearrangements and solution exchange are
initiated during LUV-detergent saturation. While membrane saturation and insertion of TX-100
into the bilayer may lead to local invaginations and permeabilization within intact vesicles prior
to fusion, it is possible that bilayer bending due to vesicle swelling could also facilitate the process.
It is also worth re-emphasizing a key benefit of the FRET-based strategy: unlike conventional
phase contrast or fluorescence-based imaging, where only a 2-dimensional plane is captured, the
FRET response reports on the dye separation distance across the entire 3-dimensional volume of
the fused species.

525 Conclusions

The approaches described here provide a general framework for dissecting the precise molecular
level events that lead to TX-100 induced LUV fusion in vitro. TX-100 dynamically alters the
conformation and integrity of both freely diffusing and surface-immobilized vesicles via a
mechanism involving vesicle permeabilization, docking, hemifusion, and full lipid mixing, prior to
solubilization and lysis. These observations provide mechanistic insights for how the widely used
TX-100 detergent dynamically perturbs and modulates highly curved membrane structures, even
at concentrations below the CMC. We anticipate that our current demonstration will be a starting
point for addressing many important aspects of vesicle fusion, including but not limited to, real-
time analysis of single-vesicle fusion events, and the influence of environmental factors, including
cholesterol and lipid content, and may be directly relevant to key biotechnological applications
where targeted vesicle fusion, especially to surfaces, is critical. We also expect that the developed
approaches will find general utility for unveiling the multi-step mechanism of vesicle fusion in
response to additional perturbative agents, including membrane-disrupting proteins and anti-
viral agents, and could have important implications for evaluating drug delivery efficiencies.

540 Methods

541 Materials

1-palmitoyl-2-oleoyl-glycero-3-phosphocholine (POPC) and 1-oleoyl-2-(12-
biotinyl(aminododecanoyl))-sn-glycero-3-phosphoethanolamine (biotin-PE) lipids in chloroform
were purchased from Avanti Polar Lipids and used without additional purification. TX-100 was
purchased from Sigma Aldrich and freshly suspended in 50 mM Tris buffer (pH 8) prior to each
use. 1,1'-Dioctadecyl-3,3,3',3'-Tetramethylindocarbocyanine (DiI) and 1,1'-Dioctadecyl-3,3,3',3'-
Tetramethylindodicarbocyanine (DiD)) were purchased from ThermoFisher Scientific and used
without additional purification. All lipid stock solutions were stored in chloroform at -20°C prior
to use, whereas DiI and DiD stock solutions from the manufacturer were stored at 4°C. Lipo-Cy3-
CO and Lipo-Cy5-N$_3$ were prepared as described elsewhere[38].

551 Preparation of vesicles incorporating DiI and DiD

Large unilamellar vesicles composed of lipid, biotin-PE and either DiI or DiD were prepared
immediately prior to each experiment by extrusion. Briefly, unlabelled lipids and fluorescently
labelled lipids were mixed in chloroform at the ratios specified in the main text prior to
evaporation of the solvent by application of a gentle nitrogen stream. The resulting lipid film was

resuspended in 50 mM Tris buffer (pH 8) and mixed well by vortex. Final solutions were then
extruded through a polycarbonate membrane filter with size cut off of 200 nm prior to use. The
same procedure was used to incorporate Lipo-Cy3-CO and Lipo-Cy5-N$_3$.

Preparation and purification of vesicles encapsulating Cal-520 and Ca^{2+}

Vesicles encapsulating either Cal-520 or Ca^{2+} were prepared as previously described[30]. Briefly,
chloroform was removed from POPC lipid stocks under gentle nitrogen flow, and the resulting
thin lipid film was dissolved and vortexed in 50 mM Tris buffer (pH 8) containing either 100 mM
Cal-520 or 10 mM Ca^{2+}. The solutions were then extruded at least 21 times through a 200 nm cut
off polycarbonate membrane filter (Avanti Polar Lipids). Vesicles were then filtered using PD-10
desalting columns (Sigma Aldrich) to separate non-encapsulated molecules from the loaded
vesicles.

Fluorescence Correlation Spectroscopy

568 FCS experiments were performed using a Zeiss LSM 880 microscope equipped with a 514 nm
569 excitation line and GaAsP detector. LUVs incorporating 99 % POPC and 0.1 % DiI were diluted in
570 50 mM Tris buffer (pH 8) to a final lipid concentration of 6 μM, pipetted onto a pre-cleaned
571 microscope slide, and sandwiched to a 1.5 glass coverslip via silica. Typical excitation powers were
572 4 μW as measured at the sample plane. Correlation curves were fitted to an equation of the form

$$573 \quad G(\tau) = y_0 + \frac{1}{n} \frac{1}{(1+\tau/\tau_{\text{diff}})} \frac{1}{\sqrt{1+\tau/SP^2\tau_{\text{diff}}}} \left(1 + \frac{Te^{\tau/\tau_{\text{trip}}}}{1-T}\right) \text{(Equation 1)}$$

where y_0, n, τ, τ_{diff}, SP, T and τ_{trip} are the offset, effective number of particles in the confocal
volume, lag time, residence time in the confocal volume, structure parameter (which describes
the shape of the detection volume and is equal to ω_{xy}/ω_z where ω is the width of the laser spot
in the x, y and z directions), fraction of particles in the triplet state and residence time in triplet
state, respectively. Diffusion coefficients were then determined and the hydrodynamic radius was
estimated via the Stokes-Einstein relationship.

Steady State FRET Vesicle Fusion Assay

200nm sized POPC vesicles incorporating 2 % DiI were suspended in 50 mM Tris buffer (pH 8) with
those incorporating 2 % DiD at a DiI-LUV: DiD-LUV ratio of 1: 3 and final lipid concentration of 88
μM. The solution was then gently stirred by magnetic stirring at 1.5 Hz. Fluorescence emission
spectra were then acquired using a HORIBA Fluoromax-4 spectrophotometer with excitation
wavelengths of 520 nm and 620 nm. All emission spectra were corrected for background and
acquired under magic angle conditions. Apparent FRET efficiencies were estimated via

$E_{FRET} = I_A/(I_A+I_D)$ (Equation 2)

where I_D (λ_{ex} = 520 nm; λ_{em} = 525 − 639 nm) and I_A (λ_{ex} = 520 nm; λ_{em} = 640 − 800 nm) represent
the integrated fluorescence emission intensities of DiI and DiD, respectively. We also measured
the efficiency of energy transfer using the *RatioA* method which yields a value proportional to *E*
by comparing the magnitude of sensitized acceptor emission after removal of donor

bleedthrough to the emission obtained under direct excitation of the acceptor (I_A^*) (λ_{ex} = 620 nm;
λ_{em} = 640 – 800 nm). Here,

RatioA = I_A/I_A^* (Equation 3)

Plots of *E* versus *[TX-100]* are reported as the mean and standard error from three experimental
runs.

Time Correlated Single Photon Counting

Time-resolved fluorescence decays were acquired using a FluoTime 300 instrument equipped
with a pulsed 530 nm excitation line (80 MHz) and PMA Hybrid 07 photomultiplier tube
(Picoquant). Briefly, DiI and DiD loaded POPC vesicles were mixed and magnetically stirred at 1.5
Hz. Fluorescence lifetime decays at an emission wavelength of 565 nm were then collected under
magic angle conditions until 10^4 photon counts accumulated at the decay maximum. Decay curves
were then fitted by iterative reconvolution of the instrument response function and the observed
intensity decay, I_t, using a multi-exponential decay of the form

$$I_t = \sum_{i=1}^{n} a_i e^{-\frac{t}{\tau_i}} \text{ (Equation 4)}$$

where a_i and t_i represent the fractional amplitudes and lifetimes of the i'th decay components.
FRET efficiencies, *E*, were estimated via

$$E = 1 - \frac{\tau_{DA}}{\tau_D} \text{ (Equation 5)}$$

where τ_{DA} and τ_D represent the lifetime of DiI in the presence and absence of DiD, respectively.
All experiments were performed in 50 mM Tris buffer (pH 8) using DiI-LUV: DiD-LUV ratios as
specified in the main text and final lipid concentrations of 88 μM.

Dynamic Light Scattering

613 Hydrodynamic radii of vesicles in TX-100-rich solutions were estimated using a Zetasizer mV DLS
614 system (Malvern Panalytical) equipped with a 632.8 nm laser line. Briefly, POPC LUVs at a final
615 lipid concentration of 88 μM were suspended in 50 mM Tris buffer (pH 8, n = 1.33) and TX-100
616 was added at the desired concentration. Correlation functions, g(τ), produced via the intensity of
617 scattered light were fitted to a single-species model of the form
618
619 $$g(\tau) = \beta(e^{-\Gamma\tau}\left[1 + \frac{1}{2}\mu_2\tau^2\right])^2 \text{ (Equation 6)}$$
620
621 or a multi-species model of the form
622
623 $$g(\tau) = \sum_n \left(\beta_n(e^{-\Gamma_n\tau})\right)^2 \text{(Equation 7)}$$
624
625 where β is the coherence factor, μ is the central moment of the distribution of decay rates and Γ
626 is the decay rate . Diffusion times, τ, were then related to the diffusion coefficients, *D*, via
627
$\Gamma=Dq^2$ (Equation 8)

where q is the wave factor. Hydrodynamic radii were then extracted using the Stokes-Einstein
equation as discussed elsewhere[58]. All DLS measurements were performed using 178°
backscattering, and intensity-size distributions were fitted to a lognormal distribution of the form

$$I = y_0 + \frac{A}{\sqrt{2\pi}wx} e^{-\frac{\left[\ln\left(\frac{x}{x_c}\right)\right]^2}{2w^2}}$$ (Equation 9)

where y_0 is the offset, x_c is the centre of the distribution, w is the log standard deviation and A is
the area.

Single Vesicle FRET Spectroscopy

Microfluidic flow cells were constructed[41] and coated with 0.1 mg/mL BSA-Biotin, 1 mg/mL BSA
and 0.2 mg/mL NeutrAvidin. Between each addition, the flow cells were washed with buffer to
remove any unbound molecules. Biotinylated POPC vesicles incorporating 2 % DiD were mixed
with non-biotinylated vesicles containing 2 % DiI (DiI-vesicle: DiD vesicle = 1: 3) at a final lipid
concentration of 88 μM in TX-100 rich media (50 mM Tris buffer, pH 8). The fused species were
then added to the functionalized coverslip for FRET-based imaging. Unbound vesicles were then
removed by washing the flowcell with buffer. Objective-based TIRF microscopy was then
performed on an inverted Nikon Eclipse Ti microscope containing a 100 x NA 1.49 oil immersion
objective lens (Nikon)[41]. Excitation was achieved from a continuous wave 532 nm laser line (Obis,
Coherent) with the diameter of the illuminated region in the field of view calculated to be 26 μm.
To mitigate against photobleaching, typical excitation intensities were 8 μWcm^{-2}. DiI and DiD
emission was spatially separated using a DualView Image Splitter (Photometrics) containing a
dichroic (T640LPXR, Chroma) and band pass filters (ET585/65M and ET700/75M, Chroma) and
imaged in parallel on a back-illuminated Prime 95b CMOS camera cooled to -30°C (Photometrics).
Fluorescence and FRET trajectories from immobilized vesicles were acquired with 50 ms exposure
time and recorded images were analysed in MATLAB (R2019a) using iSMS FRET microscopy
software[59]. Briefly, co-localized DiI and DiD emission trajectories, termed I_D and I_A respectively,
were obtained by integration of the intensity within the area containing the vesicle of interest at
each time point, and apparent FRET efficiencies, E_{FRET}, were estimated as previously described. In
line with previous work[28], tri-Gaussian fits were applied to the distributions shown in **Figure 3d**
and based on the crossing points of the individual components, thresholds for deciding if the
species were docked (E_{FRET} < 0.25), partially fused (0.25 ≥ E_{FRET} ≤ 0.55) or fully fused (E_{FRET} > 0.55)
were used.

Quartz Crystal Microbalance with Dissipation Monitoring

QCM-D experiments were performed using a Q-Sense E4 system (Biolin Scientific) instrument.
Briefly, SiO$_2$ coated substrates (Biolin Scientific) with a fundamental frequency of 5 MHz were first
cleaned by UV ozone for 10 minutes, then sonicated in 2 % Hellmanex III for 10 minutes and
ultrapure Milli-Q water for 20 minutes. The sensors were then dried under nitrogen before being
treated with UV ozone for an additional 30 minutes. The sensors were then immersed in 4 % v/v
ATES/IPA solution for 16 hours to produce an amine monolayer and dried under nitrogen before
being installed in the sensor housing module. All sensors were washed with 50 mM Tris (pH 8)
under a flow rate of 20 μL per minute until a stable baseline, defined as < 5 Hz shift per minute
was achieved. At the start of each experiment, the sensor surfaces were functionalized

sequentially with 0.1 mg mL^{-1} biotinylated bovine serum albumin (BSA), 1 mg mL^{-1} BSA and 0.2
mg mL^{-1} NeutrAvidin as per the single-vesicle TIRF experiments. The sensors were also washed
with buffer after each addition to remove any unbound molecules. Vesicles incorporating 1 %
Biotin-PE were then flushed across the surface at a final lipid concentration of 0.25 mg mL^{-1} until
saturation of the surface was reached, followed by another buffer wash step. Next, POPC vesicles
lacking Biotin-PE at a final lipid concentration of 0.16 mg mL^{-1} were mixed with TX-100 at the
specified concentrations and immediately washed over the sensor surface. Unlike the TIRF
measurements, here we monitored the fusion of vesicles in solution to vesicles pre-tethered to
the sensor substrate.

Scanning Electron Microscopy

SEM images were acquired using a JEOL JSM 7800-F system operating at 5kV. Vesicles were
induced to fuse in 50 mM Tris (pH 8) conditions containing TX-100 as specified in the main text,
diluted, and deposited onto a silicon substrate. The solution was then evaporated, and the
immobilized vesicle layer was sputtered with a 5 nm layer of Pt/Pd. We note that the final
concentration of vesicles applied to the substrate was sub-nM to prevent the formation of
conglomerates that are not directly correlated to the effects of TX-100. This was confirmed by the
fact that for low TX-100 concentrations, and for vesicles imaged in the absence of detergent, we
did not observe any vesicle clusters. Vesicle diameters were determined using ImageJ, where
automated analysis of black and white binary images enabled separation of regions of white pixels
(vesicles) against a dark background. Vesicle circularity, ϕ, was measured via

$\phi = 4p(A/p^2)$ (Equation 10)

where A is the 2-dimensional surface area of the vesicle and p is the vesicle perimeter.

Cryo-TEM

Cryo-TEM was used to directly visualize fused vesicle species[47]. Briefly, Quantifoil copper R 1.2/1.3
200 mesh grids (Electron Microscopy Sciences) were prepared by glow discharging at 20 mA and
0.26 mbar for 1 minute in a Pelco easiGlow glow discharge system. Small volumes of vesicles pre-
incubated with TX-100 were than applied to the carbon side of the EM grid in 90 % humidity.
Excess liquid was blotted off and the grids were plunge frozen into precooled liquid ethane using
a Vitrobot system (Thermo Scientific), allowing fused species to be embedded and preserved
within a thin layer of amorphous ice. Micrographs were then obtained using a Thermo Scientific
Glacios Cryo-TEM microscope using an operating voltage of 200 kV and 120,000x magnification.
Vesicle diameters and circularities were determined using in-house computational analyses.

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Data Availability

The datasets generated during the current study are available from the corresponding author on
reasonable request.

Author Contributions

S. D. Q. designed the research and wrote the manuscript with the help of all other co-authors. L.
G. D., C. K., D. C., C. M. H., N. S, and S. D. Q. collected the data and performed the research. L. G.
D., C. K., D. C., C. M. H., N. S., L. C., J. C. P., M. C. L., and S. D. Q. analyzed the data. S. D. Q. was the
project administrator.

Competing Interests

The authors declare no competing interests.

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Figure 1. Characterization of vesicle-detergent interactions at sub-solubilizing TX-100 concentrations. (a) Top:
Normalized variation in FCS correlation curves (solid lines) and fits (dotted lines) associated with interactions
between LUVs and 0 (blue), 0.1 (green), 0.2 (red) and 0.5 (grey) mM TX-100 at 21°C. Bottom: corresponding residuals
of the fits. (b) Top: Normalized variation in DLS correlation curves (solid lines) and fits (dotted lines) associated with
interactions between LUVs and 0 (blue), 0.1 (green), 0.2 (red) and 0.25 (orange) mM TX-100 at 21°C. Bottom:
residuals of the fits. (c) Corresponding hydrodynamic diameter distributions obtained from LUVs in the absence (blue)
and presence of 0.15 (green) and 0.2 (red) mM TX-100. The solid black lines represent lognormal fits with peak d_H
values of 182.6 ± 0.1 nm (Chi-squared χ_r^2 = 0.02) (top), 226.8 ± 0.2 nm (χ_r^2 = 0.02) (middle) and 311.2 ± 0.7 nm (χ_r^2 =
0.01) (bottom), respectively.

Figure 2. TX-100 induces vesicle fusion. (a) Schematic illustration of the lipid mixing assay. POPC LUVs containing 2
% DiI (green) and 2 % DiD (red) are incubated at a 1: 3 ratio. (b) Representative variation in fluorescence emission
spectra (λ_{ex} = 520 nm) obtained from LUVs in the presence of 0 – 0.3 mM (top panel) and 0.3 mM – 2 mM TX-100
(bottom panel) at 21°C. (c) Mixing LUVs with TX-100 results in FRET enhancements assigned to vesicle docking and
fusion prior to FRET reductions attributed to micellization. (d) Corresponding variation in *RatioA*. (e) Variation in FRET
efficiency, *E*, as estimated from changes in τ_D and τ_{DA} across the titration. Data points represent the mean and
standard error of the mean from three separated experimental runs. (f) Time-dependent variation in DiD emission
upon injection of TX-100 at 21°C (λ_{ex} = 520 nm). The grey dashed line corresponds to a linear fit, coloured dashed
lines correspond to bi-exponential fits and the shaded areas correspond to the SE. The arrow indicates the point of
TX-100 injection.

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Figure 3. TX-100 induces vesicle docking, hemifusion and full lipid mixing. (a) Schematic illustration of the
immobilization scheme. LUVs incorporating 2 % DiD and 1 % Biotin-PE were fused with LUVs incorporating 2 % DiI at
21°C and attached to a coverslip via BSA, biotin and NeutrAvidin. (b) Representative TIRF image obtained from
surface-tethered LUVs in the presence of TX-100, showing co-localized DiI and DiD regions (yellow circles) (λ_{ex} = 532
nm). Insets: 3D intensity profiles of DiI and DiD emission obtained from a single fusion product. (c) Representative
DiI (green) and DiD (red) intensity traces obtained from individually fused LUVs in the absence and presence of TX-
100. (d) Histograms of the mean apparent FRET efficiency obtained from N > 500 surface immobilized vesicles before
and after incubation with TX-100 rich solutions. Solid black lines represent tri-Gaussian fits. (e) Variation in the
number of docked (E_{FRET} < 0.25) versus fully fused (E_{FRET} > 0.55) species observed as a function of TX-100.

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Figure 4. Confirmation of fused vesicles by SEM and Cryo-TEM. (a) SEM micrographs of intact vesicles in the absence and (b-e) presence of 0.15 mM TX-100 showing intact immobilized vesicles (yellow dashed regions), docked vesicles (white dashed regions), and large elliptical species (red dashed regions) assigned to full fusion. Vesicles were prepared and incubated with TX-100 at 21°C. (f) Size distribution histograms obtained from vesicles in the absence (blue) and presence (orange) of 0.15 mM TX-100. Inset: comparative box plots summarizing the corresponding variation in particle circularity. The boxes represent the interquartile ranges, the upper and lower whiskers represent the 5th and 95th percentiles, and the solid lines inside the boxes represent the medians. (g) Schematic (left panel) and representative cryo-TEM image (right panel) of docked, (h) hemi-fused and (i) fully fused species. Insets: regions of interest at higher magnification showing bilayers defined by the luminal grey level distribution (bottom panels). The green arrows indicate further docking or budding events.

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Figure 5. QCM-D analysis of TX-100 induced vesicle fusion on solid surfaces. Schematic illustrations (top panels) and
normalized variation in frequency (red) and dissipation (blue) (bottom panels) associated with the 7th (solid lines)
and 9th (dashed lines) harmonics upon injection of LUVs and (a) 0.1 mM, (b) 0.2 mM and (c) 0.3 mM TX-100 to a
surface functionalized with BSA-biotin, NeutrAvidin and biotinylated LUVs. As illustrated, TX-100 induces a time-
dependent increase in mass on the surface due to vesicles in solution merging with anchored vesicles, and minimal
material is removed during surface washing when the TX-100 concentration is below the CMC. In contrast, tethered
species lyse and dissociate when the detergent concentration exceeds the CMC. Injection of TX-100 and freely
diffusing LUVs is denoted by the yellow shaded areas and was performed after washing the vesicle saturated surface
with 50 mM Tris buffer (pH 8). The blue shaded areas correspond to the final wash step when post-interaction, the
surface was rinsed with buffer. All experiments were performed at 21ºC.

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