Membrane permeability selection drove the stereochemistry of life.

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Abstract

Early in the evolution of life a proto-metabolic network was encapsulated within a membrane compartment. The permeability characteristics of the membrane determined several key functions of this network by determining which compounds could enter the compartment and which compounds could not. One key feature of known life is the utilisation of right-handed D- ribose and deoxyribose sugars and left-handed L- amino acid stereochemical isomers (enantiomers), however, it is not clear why life adopted this specific chirality. We previously demonstrated that an archaeal and an intermediate membrane mimic, bearing a mixture of bacterial and archaeal lipid characteristics (a 'hybrid' membrane), display increased permeability compared to bacterial-like membranes. Here, we investigate if these membranes can drive stereochemical selection on pentose sugars, hexose sugars and amino acids. Using permeability assays of homogenous unilamellar vesicles, we demonstrate that both membranes select for D- ribose and deoxyribose sugars while the hybrid membrane uniquely selects for a reduced alphabet of L- facing amino acids. This repertoire includes alanine, the plausible first L- amino acid utilised. We conclude such compartments could provide stereochemical compound selection thereby demonstrating a solution to the chirality problem during the evolution of life.

Introduction

Life on Earth is defined by a curious and universal stereochemical asymmetry [1, 2]. Specifically, the pentose sugars utilised for DNA and RNA (deoxyribose and ribose) possess three stereocenters imposing a D- (right-handed) stereochemistry. In contrast, 19 of the 20 universal proteinogenic amino acids used possess one or two stereocenters imposing L- (left-handed) stereochemistry on polypeptides. A number of abiotic processes have been proposed for generation of homochiral states (e.g., [1-4]). Yet importantly, autocatalytic polymerisation of mononucleotides proceeds with efficacy in mixtures with high relative ratios of one enantiomer [5]. Similar catalytic processes have been demonstrated for amino acids which polymerise stereo-selectively in abiotic conditions [3, 6]. This demonstrates Wald's conjecture [1], which set out how the polymerisation characteristics of ribonucleotides and amino acids would lead life to preferentially adopt a single stereochemical form. While this explains why life might use homochiral states, it does not explain the conditions under which specific homochiral mixes arose in a biologically proximate system allowing this bias to become hardwired into the central metabolism of life.

Ideas about the origin of life have been shaped by the demonstration that abiotic chemical conditions can generate many compounds utilised by living systems. Compounds generated by these abiotic processes are equal mixes of both enantiomers (racemic). Experimentally simulated prebiotic conditions have been shown to form ribose through the formose reaction [7]. The distribution of variant glycolysis pathways across the tree of life [8, 9] also suggests that glycolysis was an ancestral feature of the last universal common ancestor (LUCA) [10, 11]. Consistent with the phylogenetic age of these pathways, experimental chemical simulations of the early Earth Ocean led to the formation of glycolysis intermediates [12], suggesting pathway intermediates were available for catabolism and as precursors of ribonucleotides and phospholipids. Additional work has suggested that gluconeogenesis preceded glycolysis [13] and, therefore, glucose, and possibly other hexose sugars, were available for early life forms.

Similarly, the Miller experiments [14, 15] demonstrate that a subset of amino acids will form in simulated early Earth conditions. Importantly, these experiments demonstrate glycine and alanine are generated at much higher relative concentration than any other proteinogenic amino acids. Similar amino acid composition has been detected in meteorite-

58 derived materials [16]. The meteorite data provide important 'outside the lab'
59 demonstrations that chemical compounds relevant for proto-metabolic function can form in
60 abiotic conditions.

61 A systematic review of 40 different criteria has been used to reconstruct the relative
62 temporal order of amino acid utilisation as life evolved, providing a consensus that glycine
63 and alanine were the first amino acids utilised [17]. Glycine does not possess any
64 stereocenters, while alanine has one. As such, the use of L- alanine over D- alanine putatively
65 imposed L- stereochemistry on all future biogenic proteins. However, the amino acids
66 generated from the Miller experiments and detected in meteorite samples are racemic,
67 possessing equal shares of both enantiomers [14, 16], a mixture of compounds that would
68 restrict the evolution of early metabolic functions [1, 3, 6] without some source of enantiomer
69 selection.

70 J. B. S. Haldane discussed how life could have emerged 'enclosed in an oily film' [18],
71 a scenario which would have separated early biological systems from abiotic mixtures present
72 in the surrounding environment. Independent of whether life started as a heterotrophic form
73 [19, 20] (i.e., the 'Oparin-Haldane' theory [21]) or an autotrophic form [12, 22-25], a critical
74 step in the history of life involved the encapsulation of metabolic functions within a
75 membrane [26, 27]. Acquisition of a membrane bestowed a proto-cellular form with the
76 capacity for metabolite accumulation [28], generation of chemical gradients [29], and co-
77 association of interacting systems necessary for co-function [27]. Compartmentalisation also
78 generated divisible units, allowing variation in biochemical functions including different
79 reaction kinetics, upon which Darwinian selection could act [30, 31]. Cell membranes have
80 been known to be semi-permeable since the late 18th century [32, 33], forming a barrier that
81 can selectively allow some compounds to pass through while excluding others. The
82 permeability of proto-cellular membranes must therefore have determined the degree to
83 which internal membrane space is a chemically distinct environment and therefore the
84 evolution of early metabolic functions [28, 34, 35].

85 Fatty acid membrane compartments of various compounds are useful model systems
86 [36] for understanding protocell evolution. For example, they have been shown to assemble
87 from meteorite extracts [37] and can be generated from clays in a process which can
88 encapsulate RNA templates [38]. Such membranes are also permeable to chemical precursors

89 necessary for core cellular functions [19, 28, 35, 39] and can 'grow' in a process driven by
90 vesicle content [30, 40] including internalised RNA replication [19]. Nonetheless, there are
91 chemical limitations to fatty acid membranes as proto-cellular systems; chiefly, such vesicles
92 would not maintain integrity in environments with high concentrations of divalent cations
93 (e.g., seawater) [31]. Similarly, they could not host the high concentrations of Mg^{2+} thought
94 necessary for ribozyme-catalysed polymerase function, which has been suggested to act as
95 an early replicator in the RNA world hypothesis [19, 41, 42]. These observations suggest that
96 a phase of evolution using an alternative membrane chemistry was necessary.

97 The membrane of all extant cellular life is built from variant phospholipids [43]
98 Experiments have demonstrated that vesicles of eukaryotic membrane analogs (POPC and
99 DPPC) are permeable to a range of pentose sugars including ribose [35], while vesicles of
100 bacterial membrane chemical analogs (EPC and DMPC) are permeable to amino acids [34].
101 Phospholipid vesicles can encapsulate macromolecules upon dehydration-hydration cycles
102 [44], making them important models for understanding how proto-cellular compartments
103 could internalise metabolic compounds, templates or catalysts. Furthermore, phospholipids
104 can form in abiotic conditions [45], can be used to encapsulate RNA polymerisation [46], can
105 perform budding and division in the presence of *ad hoc* precursors [47], and can withstand
106 ion concentration gradients [48].

107 A range of phospholipid forms are used by microbes, often in heterogenous and
108 varying mixes. Despite this diversity, there are two fundamental core membrane
109 phospholipid chemistries present in cellular life [43]: bacteria and eukaryotes possess
110 membranes predominantly composed of fatty acid chains bonded to a glycerol-3-phosphate
111 (G3P) backbone via ester bonds; a phospholipid with a right-handed stereochemistry. In
112 contrast, most archaeal membranes are predominantly constructed from diether lipids with
113 isoprenoid chains containing methyl branches bonded to a glycerol-1-phosphate (G1P)
114 backbone via ether bonds; a phospholipid with a left-handed stereochemistry [43, 48, 49].
115 This pattern demonstrates that deep within the tree of life two different lipid forms were
116 adopted on distinct branches [43]. It is not clear if this transition was from: i) one form to the
117 other, ii) an intermediate form (for example, a phospholipid with a combination of bacterial
118 and archaeal characteristic i.e., a 'hybrid' membrane), iii) the fatty acid membranes similar to
119 the types discussed above, iv) the common ancestor of Archaea and Bacteria possessed a

120 mixed heterochiral membrane, or v) an intermediate that possessed no membrane; with the
121 two membrane types adopted later and independently [24, 50, 51]. There is currently limited
122 consensus on the placement of the last universal common ancestor (LUCA) which would
123 enable us to polarise the transition in membrane chemistry. Some analyses suggests that the
124 root of the tree of life may reside between the Bacteria and the Archaea [52-57], whilst
125 alternative analyses suggest that the root lies within the Bacteria [58-60]. The placement of
126 the root determines the minimum number and nature of transitions in membrane chemistry
127 during the evolution of the prokaryotic domains, however, neither rooting scenario excludes
128 presence of hybrid membrane variants deep within either the Bacteria or the Archaea or at
129 their common ancestor. Such hybrid forms may have had a critical impact on permeability
130 characteristics during the early evolution of cellular forms.

131 Using unilamellar vesicle microfluidic permeability assays, we have previously shown
132 that archaea-like G1PC phospholipid membranes demonstrate higher permeability to core
133 metabolites than bacterial-like membranes. These core metabolites include: nucleobases
134 (that possess no stereochemistry), ribose, deoxyribose and six proteinogenic amino acids
135 [61]. These experiments also included comparisons with a plausible intermediate 'hybrid'
136 phospholipid membrane consisting of diether lipids with isoprenoid chains containing methyl
137 branches (archaeal traits) bonded to a glycerol-3-phosphate (G3P) with right-handed
138 stereochemistry (bacterial traits). Experiments using this hybrid membrane demonstrated
139 increased permeability compared to both bacterial and archaeal membranes for deoxyribose,
140 ribose and glycine. Alternative hybrid membrane forms with different combinations of
141 chemical characteristics showed reduced permeability in comparison to both the hybrid and
142 archaeal membrane. We therefore suggest that the permeability characteristics of archaeal-
143 like membranes might have driven the early evolution of proto-cellular metabolism [61].

144 Historically, few experiments have compared L- and D- enantiomer permeability
145 through membranes. For example, Nagano in 1902 [62, 63] compared the absorption rate of
146 a range of sugars across the intestinal membrane of dogs, an experiment that included
147 comparisons of D- hexose to L- pentose sugars. More recently, Sacerdote and Szostak
148 compared absorption of L- and D- xylose through four homogeneous vesicle types, including
149 three different fatty acids and one phospholipid (POPC, a compound similar to eukaryotic
150 membranes) [35]. This experiment demonstrated no difference in the enantiomer

151 permeability of xylose. Here, we expand such comparisons to a range of compounds useful
152 for proto-metabolism and focus on archaeal and hybrid membrane mimics. We focus on these
153 two membrane mimics because our previous experiments show they display the highest
154 permeability [61]. The aim of this work is to test the hypothesis that membrane chemistry
155 imposes permeability selection on enantiomers. We compare the enantiomer permeability
156 of four pentose sugars, two hexose sugars and twelve amino acids, demonstrating strong
157 selection for D- ribose and D- deoxyribose in both membrane types. In contrast only the
158 hybrid membrane shows permeability selection for a handful of L- amino acids including
159 strong selection for L- alanine. We therefore suggest that early utilisation of a 'hybrid'
160 membrane chemistry –a possible precursor form to either or both bacterial or archaeal
161 membrane chemistry– would have allowed enrichment of specific enantiomers and therefore
162 hardwired a proto-cellular form to utilise L- form amino acids and D- form ribonucleotides.

163 **Results**

164 **Pentose sugar permeability shows selection for D- ribose and D- deoxyribose**

165 To explore the possibility that an archaeal or hybrid membrane imposes permeability
166 selection on different enantiomers we compared permeability of four pentose sugars:
167 deoxyribose, ribose, xylose and arabinose (Figure 1). Both membranes demonstrate
168 permeability to ribose, xylose and arabinose and, importantly, show increased permeability
169 to D- ribose compared to L- ribose ($P < 0.01$). Stereochemical selection was not evident for
170 xylose or arabinose.

171 As discussed above, permeability for D- ribose has been reported for both fatty acid
172 membranes and some eukaryotic-like phospholipids, but L- and D- rates were not compared
173 as part of these experiments. However, those experiments did include a comparison of
174 permeability for L- and D- xylose [35], demonstrating: i) reduced permeability for both
175 enantiomers of xylose compared to D- ribose, and ii) no difference in permeability between
176 the two xylose enantiomers through POPC eukaryotic-like phospholipids [35] similar to our
177 results for xylose in both the membranes tested here (Figure 1). Our experiments therefore
178 also produce data consistent with the idea that ribose was favoured over xylose or arabinose
179 by membrane selection (Figure 1, & S1).

180 L- deoxyribose showed very low average permeability through either membrane type
181 but, in contrast, D- deoxyribose showed significantly increased permeability through both
182 membrane types ($P < 0.01$ and $P < 0.0001$ for the archaeal and hybrid membrane,
183 respectively). These results demonstrate that both the archaeal and hybrid membrane can
184 drive stereochemical selection generating a 'passive sorting mechanism' [35] for D- ribose
185 and D- deoxyribose. Equivalent comparisons of bacterial membrane mimics do not show
186 stereochemical selection (Figure S1). The patterns of stereochemical selection observed
187 through the archaeal and hybrid membranes are consistent with the utilisation of these
188 compounds as the chemical building blocks for DNA and RNA, the molecules central to the
189 hereditary, transcription and translational systems of all life [64, 65].

190 **Hexose sugar permeability through archaeal and hybrid membranes**

191 As discussed above, another common function that likely emerged early in the evolution of
192 life is the glycolysis pathway [11, 12, 49]. This pathway provides chemical precursors for both
193 RNA and phospholipid compounds, suggesting a varied role in the evolution of life [50].
194 Although there is considerable differentiation in the glycolysis pathways across the tree of
195 life, and in many cases enzymes of the pathway are not encoded by homologous genes [8, 9],
196 both archaea and bacteria can use glucose and fructose as a primary carbon source [66-68].

197 To understand how early cellular forms could make use of abiotically derived glucose
198 or fructose [12, 13] (both with four stereocenters), we performed permeability selection
199 assays through the archaeal and hybrid membranes. These data show both hexoses can
200 permeate through these membrane forms; but a minor ($P < 0.01$), selection for D- fructose
201 over L- fructose enantiomers was observed through the archaeal membrane (Figure 2). We
202 found no selection for L- or D- fructose enantiomers through a bacterial membrane (Figure
203 S1). These data suggest utilisation of the archaeal membrane could have generated a bias for
204 a D- fructose precursor providing a D- source metabolite bias for glycolytic metabolism.

205 **Amino acid permeability through archaeal and hybrid membranes**

206 We previously demonstrated that archaeal membranes have a higher permeability to glycine
207 and alanine compared to bacterial membranes [61]. To explore if the archaeal and the hybrid
208 membranes impose selection on alanine, we compared permeability of L- and D- alanine
209 enantiomers. We found no differential selection through the archaeal membrane, but strong

210 selection for the L- alanine enantiomer in comparison to the effective exclusion of D- alanine
211 in the hybrid membrane ($P < 0.0001$, Figure 3). We also note that we had previously
212 demonstrated that L- alanine does not permeate through bacterial membranes [61]. This
213 result demonstrates a passive L- alanine sorting mechanism consistent with the amino acid
214 stereochemical asymmetry early in the evolution of amino acid proteinogenic utilisation, but
215 only in the hybrid membrane and not the archaeal membrane mimic tested.

216 To further explore stereochemical permeability selection, we performed permeability
217 assays on 11 additional proteinogenic amino acids. These 11 amino acids included seven
218 amino acids (V, D, E, S, L, T, I) that were shown to be produced from the Miller experiments
219 but at much lower molar concentrations than alanine or glycine [14, 15]. The amino acids
220 analysed include compounds with two stereocenters (T, I) and a range of differing physical-
221 chemical properties including: three non-polar (V, L, I, Figure 3), five polar (S, T, N, C, Q, Figure
222 4), one positively and two negatively charged amino acids (R, D, E, respectively, Figure 5).
223 These assays confirmed that all 14 amino acids tested (when including glycine and tryptophan
224 previously analysed [61]) permeated through the archaeal membrane (Figures 3-5).
225 Furthermore, focusing on amino acids where enantiomers were compared: I, S, C, R and D
226 amino acids demonstrated enantiomer selection through the archaeal membrane. However,
227 these results were mixed, with only arginine (R) showing L- selection while the other four (D,
228 C, S, I) showed D- selection (Figures 3-5). This pattern is not consistent with how life evolved
229 to use amino acid enantiomers.

230 In contrast, stereochemical permeability assays through the hybrid membrane (a
231 phospholipid with the opposite D- stereochemistry than the archaeal membrane [43])
232 demonstrated fewer amino acids could permeate this membrane. Specifically, C, E, S, T and
233 V showed no permeability in contrast to the archaeal membrane (Figures 3-5). Consistent
234 with the alanine results discussed above, the hybrid membrane showed L- enantiomer
235 selection for L, Q, R, and D amino acids (Figures 3-5), demonstrating how this membrane can
236 generate passive selection for a reduced alphabet of L- amino acids. These results show that
237 specific chemical characteristics of the membrane phospholipid can determine membrane
238 permeability function. We also note, as the hybrid membrane effectively excluded multiple
239 amino acid enantiomers, these experiments confirm that the amino acid selectivity

240 represents true differences in membrane permeability function (see also Figure S3 for further
241 controls).

242 To test for enantiomer selection through the bacterial membrane we compared L- and
243 D- permeability of R, C, and L amino acids through a bacterial mimic. These amino acids were
244 chosen as they showed very different enantiomer permeability through archaeal and hybrid
245 membranes (Figure 3-5). These experiments demonstrate no evidence of enantiomer
246 selection through the bacterial membrane (Figure S2).

247 **Discussion**

248 Using controlled unilamellar vesicle experiments, we demonstrate that a model archaeal
249 membrane and one hybrid membrane form drive stereochemical permeability selection
250 (summarised in Table 1). These data suggest an ancient phase of evolution based on either of
251 these two-membrane chassis could have driven selection for the stereochemical
252 characteristics of proto-metabolism. Specifically, an evolutionary bottleneck dependent on
253 either membrane chemistry tested would have generated a high level of selection for D-
254 ribose and D- deoxyribose, hard-wiring the use of right-handed ribonucleotides. Strikingly, a
255 cell-type with a hybrid membrane -a possible precursor of both archaeal and bacterial
256 membrane types- could have hardwired both the use of D- ribonucleotide oligonucleotides
257 and L- form amino acids, although initially with a reduced alphabet of amino acids. Given
258 current data, we advocate for a phase of evolution prior to, or at the last common ancestor,
259 which was a hybrid membrane consisting of diether lipids with isoprenoid chains containing
260 methyl branches (archaeal traits) bonded to a glycerol-3-phosphate (G3P) with right-handed
261 stereochemistry (bacterial traits). We note that this scenario only represents one phase of
262 membrane evolution in a long history of possible membrane transitions between the first
263 'life' form and the last common ancestor. As such this scenario is compatible with many
264 previously suggested ancestries for the prokaryotic domains (e.g., [24, 50, 51]).

265 Different phylogenomic analyses have suggested that the universal common ancestor
266 lies between the Archaea and the Bacteria [52-57], or potentially within the Bacteria [58-60].
267 Our results do not directly inform this debate, as multiple transitions in membrane chemistry
268 are possible; indeed, the transition through the hybrid membrane we advocate could have
269 occurred before the derivation of either the Archaea or the Bacteria. It is also important to
270 mention that a pure form membrane, as used here, is implausible as an early proto-cellular

271 chassis, especially if they were derived from an abiotic source [45] and occurred prior to the
272 emergence of a dedicated metabolic network for the generation of membrane forming
273 compounds. It is not clear, at this stage, if mixed membranes encompassing the types of
274 phospholipids used here are viable, although experiments with *E. coli* suggest glycerol-1-
275 phosphate and glycerol-3-phosphate lipids can form a heterochiral membrane [69].
276 Furthermore, if it is possible to form mixed phospholipid unilamellar vesicles, would such
277 membranes generate similar patterns of permeability selection identified here. Future works
278 should focus on assessing the stability and permeability function of mixed membranes.

279 The results reported here also demonstrate that additional bottlenecks through an
280 archaeal type membrane could have hard-wired bias for D- hexose sugars for metabolic
281 utilisation such as glycolysis, as demonstrated by our result of increased archaeal membrane
282 permeability to D- fructose (Figure 2). However, this was not a strong result, and the
283 possibility of two bottlenecks with differing membrane chemistries leaves some confusion
284 regarding the relative inferred order of membrane utilisation relative to the tree of life. We
285 also note that glycolytic catabolism of hexose sugars does not impose similar constraints on
286 enantiomer utilisation as would the emergence of the anabolic polymerisation of ribonucleic
287 acids and amino acid polypeptides [1]. This is because hexose compounds are primarily
288 broken down for metabolic benefit and not polymerised for storage or translation of
289 information in the form of DNA, RNA and amino acid polypeptides. Differential utilisation, i.e.,
290 polymerisation in the case of amino acids and ribonucleotides vs catabolic metabolism in the
291 case of glycolysis of hexoses, therefore suggests that the stereochemical permeability
292 selection on precursors of glycolysis is not as an important a factor as precursors of the
293 compounds utilised in the central dogma.

294 We acknowledge that wider analysis of membrane types, including bacterial-mimics
295 may reveal additional permeability selection traits informative for understanding proto-
296 cellular membrane evolution. We have partly investigated this possibility in a bacterial
297 membrane mimic for three pentose sugars, one hexose sugar and three amino acids (Figure
298 S1-2). These were chosen as they showed a wide variation of permeability characteristics in
299 the archaeal and the hybrid membranes tested (Table 1). In all these tests the bacterial
300 membrane mimic demonstrated no stereochemical selection, an outcome that could be a
301 function of their lower permeability identified on our experimental platform [61].

302 Here we have discussed published examples of chemical polymerisation of
303 nucleotides and amino acids in abiotic conditions [1, 3, 5, 6]. In these examples both types of
304 polymerisation are limited when conducted in racemic mixes of substrate due to enantiomer
305 compatibility and the reducing availability of compatible enantiomers as polymerisation
306 progresses in a closed pool of chemical precursors. The membrane permeability selection
307 demonstrated here would solve this problem by providing a compartment with specific
308 enantiomer enrichment, allowing polymerisation to progress without inhibition generated by
309 racemic mixes of the constituent subunits. Such a system would have multiple advantages in
310 competition over protocells without enantiomer selection. Specifically, both permeability and
311 polymerisation selection would act synergistically to generate competitive advantage and
312 therefore Darwinian selection [30, 31]. Understanding how life began is a jigsaw puzzle [70];
313 the utilisation of a hybrid phospholipid membrane to derive stereochemical selection for
314 deoxyribose, ribose and a reduced alphabet of amino acids (Table 1) is a potential piece of
315 that puzzle. These observations are valuable information towards explaining the chirality
316 problem as part of the origin of life, however we acknowledge that these observations do
317 need testing with alternative methods.

318 **Methods**

319 All methods are adaptations of those reported in Łapińska et al. 2023 [61].

320 **Preparation of materials**

321 All chemicals were purchased from Merck. All Lipids were purchased from Avanti Polar Lipids
322 [i.e. 1,2-di-O-phytanyl-sn-glycero-1-phosphocholine, 1,2-di-O-phytanyl-sn-glycero-3-
323 phosphocholine, 1,2-dioleoyl-sn-glycero-3-phosphoethanolamine, 1,2-dioleoyl-sn-glycero-3-
324 phospho-(1'-rac-glycerol) and 1,3-bis(sn 3'-phosphatidyl)-sn-glycerol)]. Indium tin oxide (ITO)-
325 coated glass slides were purchased from VisionTek Systems. The fluorescent probe 5(6)-
326 carboxyfluorescein (CF), (MW=376 g/mol), was dissolved in absolute ethanol at a stock
327 concentration of 10 mg/mL. In order to perform all permeability experiments at physiological
328 pH (7.4), the washing buffer was prepared by dissolving sucrose (MW = 342 g/mol) in 5 mM
329 4-(2-hydroxyethyl)-1-piperazineethanesul-fonic acid (HEPES) at pH=7.4 at a final sucrose
330 concentration of 195 mM. All metabolites were dissolved in the washing buffer to a final
331 concentration of 1 mM.

332 **Preparation of synthetic vesicles**

333 Giant unilamellar vesicles were electroformed using a Vesicle Prep Pro (Nanion) [71]. Three
334 different lipid stocks; archaeal lipid mimic (G1PC) (2,3-di-O-phytanyl-sn-glycero-1-
335 phosphocholine), hybrid lipid mimic (G3PC) (1,2-di-O-phytanyl-sn-glycero-3-phosphocholine)
336 and bacterial lipid mimic mixed of three lipids: 1,2-dioleoyl-sn-glycero-3-
337 phosphoethanolamine, 1,2-dioleoyl-sn-glycero-3-phospho-(1'-rac-glycerol) and 1,3-bis(sn-3'-
338 phosphatidyl)-sn-glycerol were prepared to a final concentration of 10 mM dissolved in
339 chloroform (Table S1.). Next, 10 µL of each lipid solution was spread evenly onto the
340 conductive side of an ITO-glass slide and vacuum desiccated to dry the solution. Meanwhile,
341 495 µL of washing buffer was degassed and mixed with 5 µL of 10 mg/ml of 5(6)-CF solution
342 to a final concentration of 0.266 mM. The lipid coated ITO-glass slide was loaded into the
343 Nanion and a greased rubber O-ring was placed over the dried layer. The area inside the O-
344 ring was filled with 300 µL of 0.266 mM CF solution and another ITO-glass slide conductive
345 side down was placed over, avoiding the formation of bubbles. Different electroformation
346 protocols were implemented depending on the lipid used (see Table S2.). The final suspension
347 contains CF fluorescent vesicles and free CF molecules, which were stored at 4°C for 2 days
348 maximum.

349 **Fabrication of the microfluidic device**

350 The microfluidic device is described in detail in Łapińska, 2023 [61]. In short, the device
351 contains two inlets connecting to a main chamber split into 4 channels, these contain
352 hydrodynamic traps (also known as coves) with the chambers finally leading to a single outlet.
353 Using a mould created by multilevel photolithography [72], polydimethylsiloxane (PDMS)
354 replicas of the device were produced. In summary, a 10:1 (base:curing agent) PDMS mixture
355 (SYLGARD 184 Silicone Elastomer Kit, Dow) was poured over the mould and degassed for 30
356 mins, this was cured for 2 hr at 70°C in an oven. The PDMS was peeled from the mould and a
357 1.5 mm biopsy punch was used to create fluidic access. To irreversibly seal the PDMS replica
358 to a glass slide, oxygen plasma treatment was used (10 sec exposure to 30W plasma power,
359 Plasma etcher, Diener Electronic GmbH).

360 **Microfluidic permeability assay**

361 Washing buffer was injected into the buffer inlet until the device was flooded completely and
362 escaping the metabolite inlet and outlet. The metabolite inlet was then sealed with tape and
363 10 µL of the fluorescent vesicle suspension was pipetted into the washing inlet. The tape
364 prevented fluorescent vesicle suspension entering the metabolite inlet. The chip was
365 mounted onto an inverted epifluorescence microscope (Olympus IX73) equipped with a
366 sCMOS camera (Zyla 4.2, Andor, used at an exposure time of 0.1s), a blue LED (CoolLED
367 pE300white, used at 30% of its intensity) and a FITC filter. Prepared 1 mL syringes were filled
368 with washing buffer, and the second with 1 mM metabolite of interest were connected to 23-
369 gauge needles (Becton Dickinson) and Masterflex Transfer Tygon tubing with 0.5 mm inner
370 and 1.5 mm outer diameter (Cole-Parmer Instrument). These were controlled by a Nemesys
371 pump operated via the Qmix Elements software (Centoni). The tape was removed from the
372 metabolite inlet and the tubing containing the metabolite of interest was inserted flowing at
373 a rate of 0.5 µL/hr. Then, the tubing containing the washing buffer was inserted into the
374 remaining inlet, flowing at a rate of 5 µL/hr. These flow rates were increased in a stepwise
375 fashion at steps of 0.5 µL/hr up to 2 µL/hr for the metabolite solution and 5 µL/hr up to 25
376 µL/hr for the washing buffer. These were kept constant for 20 mins to remove any free CF
377 molecules from the microfluidic device. After this time the metabolite solution was increased
378 to 25 µL/hr and the washing buffer solution was reduced to 1 µL/hr. Brightfield and
379 fluorescent imaging of an area containing 14 coves were acquired every 30 sec for 3 mins.

380 **Vesicle free carboxyfluorescein and metabolite interaction assays**

381 To explore the possibility that compound stereochemical variation altered carboxyfluorescein
382 function, we ran separate microplate reader experiments demonstrating that the
383 stereochemistry of the metabolic compound does not have a differential effect on the
384 fluorescence of carboxyfluorescein (Figure S3), the reporter used throughout these
385 experiments. These data provide a further control for our permeability selection data
386 reported above.

387 Washing buffer was used to dissolve CF to a concentration of 0.532 mM and 100 µL
388 was added to wells of a 96-well plate (Costar 96, Fisher Scientific). Then, 100 µL of 1 mM
389 metabolite solution was added to each well and the plate placed within a Clariostar Plus plate
390 reader (BMG Labtech, UK) for fluorescence data acquisition over 30 mins (Excitation 483-14,

Emission 596, with a FITC filter). Three independent experiments were carried out for each metabolite.

Image and data analysis

Images for each membrane/metabolite combination tested, at every timepoint, were exported into ImageJ for quantitative comparative analysis. Only single unilamellar vesicles were retained for analysis and a circle was drawn around each vesicle. A second identical circle was drawn in an area 10 µm upstream of the vesicle, which was then used to produce background-subtracted values for intra-vesicle fluorescence. The initial intra-vesicle fluorescence value (t=0) was subtracted. The values were then corrected for the impact of both the delivery of the washing buffer solution and photobleaching on the intra-vesicle CF fluorescence signal (see methods detailed in [73]). The permeability of each membrane mimic was measured in three independent experiments. Statistical comparisons were carried out using the Welch's t test.

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Competing interests

The authors have declared that no competing interests exist.

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624 **Table**

	Archaeal permeability L/D	Hybrid permeability L/D	Archaeal enantiomer selection	Hybrid enantiomer selection
Ribose	++/+++++	+++/+++++	**	**
Deoxyribose	+/++++	+/+++++	**	****
Xylose	+++/+++	+++/+++		
Arabinose	+++/+++	++/+++		
Fructose	+++/++++	+++/+++	**	
Glucose	+++/+++	+++/+++		
Alanine	+++/+++	+++/+		****
Valine	+++/+++	+/+		
Leucine	++/+++	+++/++		**
Isoleucine	+++/++++	+/++	**	
Serine	+++/+++++	+/+	*	
Threonine	+++/++++	-/+		
Glutamine	++/+	+++++/+		*
Cysteine	+++/++++	+/+	****	
Asparagine	+++/+++	++/+++		
Arginine	+++/++++	++/-	***	***
Aspartic acid	++/+++	+++/++	***	*
Glutamic acid	+++/+++	+/+		

625

626 **Table 1. Summary of permeability and enantiomer selection data for the eighteen**
627 **substrates tested against the two membranes.** Carboxyfluorescein (CF) fluorescence after 3
628 minutes of exposure to each substrate at a concentration of 1 mM is summarised as follows:
629 <0 = -; 0-5 a.u. = +, 5-10 a.u. = ++, 10-15 a.u. = +++, 15-20 a.u. = ++++, >20 a.u. = +++++.
630 Statistical support for different enantiomer selection is indicated: ****: p-value < 0.0001, ***:
631 p-value < 0.001, **: p-value < 0.01,*: p-value < 0.05. Enantiomer selection consistent with the
632 stereochemical bias of life is coloured blue. Enantiomer selection that contradicts the
633 stereochemical bias of life is coloured orange.

634

Figure Legends

Figure 1. Membrane permeability to pentose sugars shows bias for D- ribose and D-deoxyribose. Temporal dependence of average carboxyfluorescein (CF) fluorescence in the archaea-like G1PC phospholipid membrane (G1PC, filled symbols) and in the 'hybrid' phospholipid membrane (G3PC, open symbols) during the exposure to 1 mM of variant substrates delivered to the microfluidic coves at constant rate. Mean (symbols) and standard deviation (error bars) were calculated from at least 10 single-vesicle measurements across three independent experiments. Lines connecting points are provided as a guide. N is the number of single vesicles investigated for each substrate exposure and each type of vesicle. N varies across different substrate experiments investigated due to technical constraints. However, care has been taken to obtain the same N for each substrate experiment across the two different enantiomer treatments in order to ensure reliable statistical comparisons. Such comparisons have been carried out via Welch's t-tests between the distributions of CF fluorescence values at t = 3 min for each enantiomer pair and are shown with corresponding violin plots next to each time-course graph (C, F, I, L). ****: p-value < 0.0001, **: p-value < 0.01. Numerical values of CF fluorescence in individual vesicles for each lipid type during the delivery of each substrate are provided in S1 File and S2 File.

Figure 2. Hexose sugar permeability through archaeal and hybrid membranes. Temporal dependence of average carboxyfluorescein (CF) fluorescence in the archaea-like G1PC phospholipid membrane (G1PC, filled symbols) and in the 'hybrid' phospholipid membrane (G3PC, open symbols) during the exposure to 1 mM of variant substrates delivered to the microfluidic coves. Mean (symbols) and standard deviation (error bars) were calculated from at least 10 single-vesicle measurements across three independent experiments. Data are processed and illustrated as described in Figure 1 legend. N is the number of single vesicles investigated for each substrate exposure and each type of vesicles. **: p-value < 0.01. Numerical values of CF fluorescence in individual vesicles for each lipid type during the delivery of each substrate are provided in S1 File and S2 File.

Figure 3. Archaeal and hybrid membrane permeability to non-polar amino acids. Temporal dependence of average carboxyfluorescein (CF) fluorescence in the archaea-like G1PC phospholipid membrane (G1PC, filled symbols) and in the 'hybrid' phospholipid membrane (G3PC, open symbols) during the exposure to 1 mM of variant substrates delivered to the

666 microfluidic coves. Data are processed and illustrated as described in Figure 1 legend. Mean
667 (symbols) and standard deviation (error bars) were calculated from at least 10 single-vesicle
668 measurements across three independent experiments. N is the number of single vesicles
669 investigated for each substrate exposure and each type of vesicles. ****: p-value < 0.0001,
670 **: p-value < 0.01. Numerical values of CF fluorescence in individual vesicles for each lipid
671 type during the delivery of each substrate are provided in S1 File and S2 File.

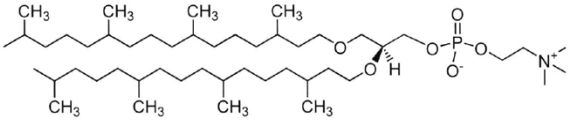
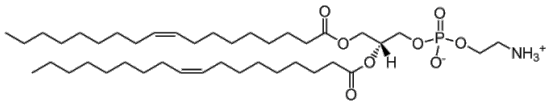
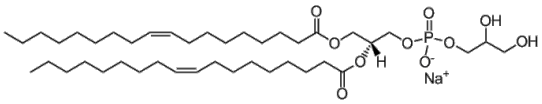
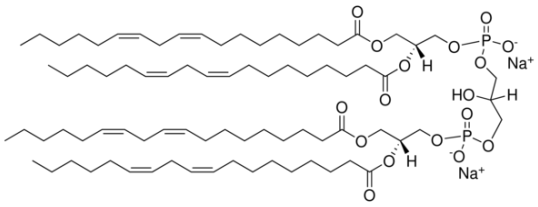
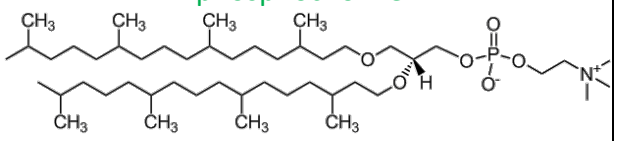
672 **Figure 4. Polar amino acid permeability through archaeal and hybrid membranes.** Temporal
673 dependence of average carboxyfluorescein (CF) fluorescence in the archaea-like G1PC
674 phospholipid membrane (G1PC, filled symbols) and in the 'hybrid' phospholipid membrane
675 (G3PC, open symbols) during the exposure to 1 mM of variant substrates delivered to the
676 microfluidic coves. Data are processed and illustrated as described in Figure 1 legend. Mean
677 (symbols) and standard deviation (error bars) were calculated from at least 10 single-vesicle
678 measurements across three independent experiments. N is the number of single vesicles
679 investigated for each substrate exposure and each type of vesicles. ****: p-value < 0.0001, *:
680 p-value < 0.05. Numerical values of CF fluorescence in individual vesicles for each lipid type
681 during the delivery of each substrate are provided in S1 File and S2 File.

682 **Figure 5. Charged amino acid permeability through archaeal and hybrid membranes.**
683 Temporal dependence of average carboxyfluorescein (CF) fluorescence in the archaea-like
684 G1PC phospholipid membrane (G1PC, filled symbols) and in the 'hybrid' phospholipid
685 membrane (G3PC, open symbols) during the exposure to 1 mM of variant substrates delivered
686 to the microfluidic coves. Data are processed and illustrated as described in Figure 1 legend.
687 Mean (symbols) and standard deviation (error bars) were calculated from at least 10 single-
688 vesicle measurements across three independent experiments. N is the number of single
689 vesicles investigated for each substrate exposure and each type of vesicles. ***: p-value <
690 0.001, *: p-value < 0.05. Numerical values of CF fluorescence in individual vesicles for each
691 lipid type during the delivery of each substrate are provided in S1 File and S2 File.

692

693 ## Supplementary Tables and Figure Legends

694 **Table S1. Details of membrane phospholipid compounds used for vesicle synthesis.**

Lipid	Lipid chemical structure/ IUPAC name	Mimic type/Avanti name	MW [g/mol]
1.	2,3-di-O-phytanyl-sn-glycero-1-phosphocholine 	Archaeal 4ME 16:0 Diether G1PC	818
2	1,2-dioleoyl-sn-glycero-3-phosphoethanolamine 	Bacterial Diester G3PE-PG-CA: 67%: 18:1 (Δ9-Cis)	744
	1,2-dioleoyl-sn-glycero-3-phospho-(1'-rac-glycerol) 	Diester G3PE 23.2%: 18:1 (Δ9-Cis)	797
	1,3-*bis*(*sn*-3'-phosphatidyl)-*sn*-glycerol 	Diester G3PG 9.8%: Cardiolipin, sodium salt	1494
3.	1,2-di-O-phytanyl-sn-glycero-3-phosphocholine 	4ME 16:0 Diether G3PC	818

695

696

697 **Table S2. Parameters used for electroformation of vesicles**

Lipid	Frequency [Hz]	Amplitude [V]	Temp. [°C]	Time-rise [min]	Time-main [min]	Time-fall [min]
1. & 3.	5	3	37	5	120	5
2.	10	1.6	37	5	160	5

698

699 **Figure S1. Sugar enantiomer permeability through bacterial membranes.** Temporal
700 dependence of average carboxyfluorescein (CF) fluorescence in the bacteria-like phospholipid
701 membrane during the exposure to 1 mM of variant substrates delivered to the microfluidic
702 coves. Mean (symbols) and standard deviation (error bars) were calculated from at least 10
703 single-vesicle measurements across three independent experiments. Lines are guides for the
704 eye. N is the number of single vesicles investigated for each substrate exposure. N varies
705 across different substrate experiments investigated due to technical constraints. However,
706 care has been taken to obtain the same N for each substrate experiment across the two
707 different enantiomers treatments in order to ensure reliable statistical comparisons. Such
708 comparisons have been carried out via Welch's t tests between the distributions of CF
709 fluorescence values at t = 3 min for each enantiomer and are shown with corresponding violin
710 plots next to each time-course graph (B, D, F, H). ****: p-value < 0.0001, **: p-value < 0.01.
711 Numerical values of CF fluorescence in individual vesicles during the delivery of each
712 substrate are provided in S3 File.

713 **Figure S2. Amino acid enantiomer permeability through bacterial membranes.** Temporal
714 dependence of average carboxyfluorescein (CF) fluorescence in the bacteria-like phospholipid
715 membrane during the exposure to 1 mM of variant substrates delivered to the microfluidic
716 coves. Mean (symbols) and standard deviation (error bars) were calculated from at least 10
717 single-vesicle measurements across three independent experiments. Lines are guides for the
718 eye. N is the number of single vesicles investigated for each substrate exposure. N varies
719 across different substrate experiments investigated due to technical constraints. However,
720 care has been taken to obtain the same N for each substrate experiment across the two
721 different enantiomers in order to ensure reliable statistical comparisons. Such comparisons
722 have been carried out via Welch's t tests between the distributions of CF fluorescence values

723 at t = 3 min for each enantiomer and are shown with corresponding violin plots next to each
724 time-course graph (B, D, F). ****: p-value < 0.0001, **: p-value < 0.01. Numerical values of
725 CF fluorescence in individual vesicles during the delivery of each substrate are provided in S3
726 File.

727 **Figure S3. Metabolite stereochemistry does not have an impact on carboxyfluorescein**
728 **fluorescence in solution.** Temporal dependence of (A) carboxyfluorescein (CF) fluorescence
729 alone or in the presence of D- or L- enantiomers (red squares or blue circles, respectively) of
730 (B-E) sugars, or (F-J) amino acids. Mean (symbols) and standard deviation (error bars) were
731 calculated from three independent experiments performed in 96-well plates with
732 fluorescence measured via a plate reader (note the reduced scale of the Y axis compared to
733 the other plots reported). 100 µL of 0.532 mM CF was added to 100 µL of 1 mM metabolite;
734 these were the concentrations used in the permeability experiments above. Numerical values
735 of CF fluorescence in individual experiments in the presence of each metabolite are provided
736 in S4 File.