A comparison of lipid diffusive dynamics in monolayers and bilayers in the context of interleaflet coupling

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Abstract

Cellular membranes are composed of lipids typically organized in a double-leaflet structure. Interactions between these two leaflets – often referred to as interleaflet coupling – play a crucial role in various cellular processes. Despite extensive study, the mechanisms governing such interactions remain incompletely understood. Here, we investigate the effects of interleaflet coupling from a specific point of view, i.e. by comparing diffusive dynamics in bilayers and monolayers, focusing on potential lipid-specific interactions between opposing leaflets. Through quantitative fluorescence microscopy techniques, we characterize lipid diffusion and mean molecular area in monolayers and bilayers composed of different lipids. Our results suggest that the observed decrease in bilayer lipid diffusion compared to monolayers depends on lipid identity. Furthermore, our analysis suggests that lipid acyl chain structure and spatial configuration at the bilayer may strongly influence interleaflet interactions and dynamics in bilayers. These findings provide insights into the role of lipid structure in mediating interleaflet coupling and underscore the need for further experimental investigations to elucidate the underlying mechanisms.

Abbreviations: OMPC: 1-oleoyl-2-myristoyl-sn-glycero-3-phosphocholine, DOPC: 1,2-dioleoyl-sn-glycero-3-phosphocholine, POPC: 1-palmitoyl-2-oleoyl-glycero-3-phosphocholine, SOPC: 1-stearoyl-2-oleoyl-sn-glycero-3-phosphocholine, mSM: Milk Sphingomyelin, Rh-PE: Rhodamine PE, TF-PC: TopFluor PC, BSA: Bovine serum albumin, PBS: Phosphate buffer saline, NaP: Sodium phosphate, D: Diffusion coefficient, MMA: Mean molecular area, lsFCS: line-scanning fluorescence correlation spectroscopy, RICS: Raster image correlation spectroscopy, PTFE: Polytetrafluoroethylene, LD: Lipid droplet, MD: Molecular dynamics, SDHPW: Saffman-Delbrück-Hughes-Pailthorpe-White, SEM: Standard error of the mean, SD: Standard deviation, $MMA_{expected}$: Expected Mean molecular area, N_{Av}: Avogadro's number, r: radius of the monolayer chamber, C: Concentration of lipid solution in molar units, V: Volume, w_0: Excitation beam waist, τ_D: Diffusion time, A_{eff} : Effective illuminated area, S: Structural parameter, MMA_b: Mean molecular area of lipid in the bilayer, D_b: Diffusion coefficient of lipids in the bilayer, N: Number of fluorescent particles, N_{total}: Total number of all lipid particles, $MMA_{m\text{-}oil}$: Mean molecular area of lipid in oil-water interface, D_m: Diffusion coefficient of lipids in the air-water interface, $D_{m\text{-}oil}$: Diffusion coefficient of lipids in oil-water interface, MMA_m: Mean molecular area of lipid in the air-water interface, $C_{\%TF\text{-}PC}$: Concentration of TF-PC molecules , a: Radius of diffusing lipid probe head, h_m: Monolayer membrane thickness, μ_m: Air-water monolayer bulk viscosity, $\mu_{m\text{-}oil}$: Oil-water monolayer bulk viscosity, μ_w: Viscosity of water, μ_s: Viscosity of sucrose, μ_a: Viscosity of air, η_m: Monolayer surface viscosity, ε: Reduced radius, h_b: Bilayer membrane thickness, h_l: Thickness of a leaflet in a bilayer membrane, μ_l: Bulk viscosity of a leaflet in a bilayer membrane, $\mu_{midplane}$: Bilayer midplane viscosity, μ_b: Bilayer bulk viscosity.

Keywords: Fluorescence Correlation Spectroscopy, Raster Image Correlation Spectroscopy, Lipids, Monolayer, Bilayer, Interleaflet Coupling.

1 Introduction

Cell membranes consist of bilayers[1] whose structural integrity and function depend on the self-assembly of a plethora of structurally diverse lipid molecules[2,3]. Furthermore, cellular lipid bilayers possess nano- to microscale heterogeneities (i.e., domains) with different characteristics than the surrounding bulk membrane[4–6]. Fluorescence microscopy studies of nano- (< ca. 200 nm) and microdomains (> ca. 200 nm) in model membranes have shown that these entities do not exist just in one leaflet, but rather span the whole bilayer[4,7–9]. In this context, the interaction between the two leaflets in lipid membranes (i.e., interleaflet coupling[10–12]) has been shown to be important for various cellular processes[13], including e.g. signal transduction[14–18], registered lipid-protein domain formation[10,19] and pathogen-host interactions[14,15,20–22]. A common feature of these phenomena is that lipid-lipid interactions (e.g., packing, phase separation, diffusive dynamics) are somehow communicated to or shared between the opposing leaflets in a bilayer.

Interleaflet coupling has been experimentally characterized from different points of view and within different systems[13]. For example, several groups have investigated how lipid domain formation in one leaflet can induce analogous phase separation in the opposing leaflet[23–26] or, more in general, how phase transitions are coupled between leaflets[27]. Alternatively, the friction or the shear stress between the two leaflets have been analysed both theoretically[28–30] and experimentally[31–33]. Finally, several efforts have been made to understand how lipid dynamics in one leaflet affect those in the other leaflet, including investigations via fluorescence spectroscopy[34–37]. In this context, we and others have observed that the interleaflet coupling of lipid diffusive dynamics in symmetric and asymmetric phosphatidylcholine bilayers (e.g. OMPC, POPC and SOPC) might depend on their acyl chain structure[34,38].

Several physical processes have been proposed to mediate interleaflet interactions, including membrane undulations, cholesterol flip-flop, line tension and interdigitation (see the review by Sarmento *et al.*[13] and references within). While the latter mechanism is controversially discussed in the literature[11,39,40], it is reasonable to hypothesize that interleaflet coupling might be indeed mediated by interactions at the bilayer midplane between acyl chains belonging to opposing leaflets[13,34,38,40,41].

Here, we focus on a specific possible manifestation of interleaflet coupling, i.e. the decreased lipid diffusive dynamics observed in bilayers, compared to monolayers. Lipid monolayers (e.g., at air-water[42] interfaces) are simple biophysical models of each leaflet in a lipid bilayer and are characterized in general by fast lipid diffusion (25-35 $\mu m^2/s$)[43–45]. Free-standing lipid bilayers, on the other hand, exhibit lateral diffusion coefficients (D) for lipids with values between ca. 5 and 12 $\mu m^2/s$[34,46,47]. In the context of the Saffman-

Delbrück theory[48], such difference is partially explained by differing thickness and boundary conditions between lipid monolayers and bilayers. Any further decrease of lipid diffusion in bilayers might be attributed to some form of interleaflet interactions[42,45]. In this work, we have performed a systematic analysis of this effect by comparing monolayers and bilayers composed of different lipids. Lipid diffusion and mean molecular area (*MMA*) in monolayers and bilayers were quantified using fluorescence fluctuation microscopy methods (i.e. line-scanning fluorescence correlation spectroscopy (lsFCS)[49] and raster image correlation spectroscopy (RICS)[50]. Both these approaches have been previously used to characterize the physical properties of model membranes, such as giant unilamellar vesicles (GUV)[34,47] and lipid monolayers[43,51].

By analysing the behaviour of lipids which were previously shown to potentially influence interleaflet coupling[34,38], we aimed to verify the hypothesis that diffusive dynamics in lipid bilayers might be determined by lipid-specific interactions between opposing leaflets.

2 Materials and Methods

2.1 Chemicals

For the preparation of lipid monolayers and GUVs, the following lipids were purchased from Avanti Polar Lipids (USA): 1-oleoyl-2-myristoyl-sn-glycero-3-phosphocholine (OMPC), 1,2-dioleoyl-sn-glycero-3-phosphocholine (DOPC), 1-palmitoyl-2-oleoyl-glycero-3-phosphocholine (POPC), 1-stearoyl-2-oleoyl-sn-glycero-3-phosphocholine (SOPC), Sphingomyelin (Milk, Bovine), 1-oleoyl-2-(6-((4,4-difluoro-1,3-dimethyl-5-(4-methoxyphenyl)-4-bora-3a,4a-diaza-s-indacene-2-propionyl)amino)hexanoyl)-sn-glycero-3-phosphocholine (TF-PC) and 1,2-distearoyl-sn-glycero-3-phosphoethanolamine-N-(lissamine rhodamine B sulfonyl-ammonium salt/ Rh-PE). 10X Phosphate Buffer Saline (pH 7.4) for GUV experiments, Glycerine trioleate (Triolein) for the preparation of lipid droplets, Alexa Fluor 488 dye and Alexa Fluor 555 dye for the calibration of excitation beam waist, isopropanol for monolayer glass chamber cleaning were purchased from ThermoFisher Scientific (USA). Sucrose for electroformation was purchased from PanReac Applichem GmbH (Germany). Bovine Serum Albumin (BSA) for GUV chamber coating, Disodium phosphate and Monosodium phosphate for Sodium phosphate buffer (NaP) preparation and ethanol for monolayer glass chamber cleaning were purchased from Carl Roth GmbH (Germany). Hellmanex® for monolayer glass chamber cleaning was purchased from Hellma GmbH & Co. KG (Germany).

2.2 Lipid mixing and air-water monolayer preparation

Lipid solutions were mixed in glass vials using positive-displacement micropipettes with glass capillary tubes. Methanol was used as solvent and a minimum volume of ~1 mL was used for all lipid solutions. These precautions mitigated concentration errors due to pipetting and solvent evaporation. Lipid solutions were stored in glass vials sealed with polytetrafluoroethylene (PTFE) tape and wrapped in aluminium foil.

Lipid monolayers were prepared similarly to what was described before[42,43]. To this aim, lipid solutions were always prepared at 0.1 mg/mL and contained 0.005 mol% TF-PC. The effective concentration of TF-PC was confirmed after each lipid mixture preparation via spectrophotometry measurements of serial dilutions. Then, the solvent was dried with a nitrogen stream and replaced with chloroform. This facilitates the solvent evaporation step needed after spreading the lipid solution at the air-water interface. The monolayer chamber[43] was cleaned using sequentially 2% Hellmanex® in water, isopropanol, ethanol, and Milli-Q® water. The cleaned chamber was dried with compressed air after every rinsing step. To prepare lipid monolayers, 200 µl of autoclaved and filtered Milli-Q® water was added to the chamber and the lipid solution in chloroform was

spread dropwise on the surface. The volume of the solution depended on the desired *MMA* ($MMA_{expected}$) of the monolayer to be obtained, according to the following equation:

$$MMA_{expected} = \frac{Area\ of\ chamber}{Number\ of\ lipid\ molecules} = \frac{\pi r^2}{C\,V N_{Av}} \quad (1)$$

wherein r is the radius of the monolayer chamber, C is the concentration of the 0.1 mg/mL lipid solution in molar units, V is the volume to be spread and N_{Av} is the Avogadro's number. After solvent evaporation (~15 mins), 30-50 µl of water was carefully removed to keep the monolayer within the working distance of the objective[43]. Finally, the chamber was carefully covered with a Teflon coated lid to minimise water evaporation. All measurements were performed at room temperature.

2.3 GUV preparation

GUVs with different lipid compositions were prepared by electroformation[52,53] using cylindrical Teflon chambers containing two platinum wires[52,53]. Lipid solutions (2.5 mg/mL with 0.005 mol% TF-PC) were prepared in ethanol and 2 µL were spread evenly on each wire twice and then dried under nitrogen stream. GUV swelling took place in the Teflon chamber filled with a 50 mM sucrose solution, applying alternating voltage (2 V peak-to-peak, 10 Hz) for 1 hour. To detach the GUVs from the wires, the process was continued for 30 minutes longer, decreasing the frequency to 2 Hz. The entire procedure was performed at room temperature. After electroformation, 100 µL of GUV suspension was transferred to custom-made microscopy chambers, pre-treated with a 10% BSA solution for 15 minutes. The GUV suspension was further diluted with an equal volume of 40 mOsmol/kg PBS containing 0.3% (w/v) low melting agarose to induce a slight inflation and stabilization of GUVs[46].

2.4 Lipid droplet preparation (oil-water monolayers)

Lipid droplets (LDs) were prepared with Triolein as oil-phase according to available protocols[54,55]. A lipid solution doped with 0.001 mol% Rh-PE in methanol was dried in microcentrifuge tubes under a nitrogen stream. Unlike the case of GUVs and air-water monolayers, TF-PC showed a significantly low fluorescence signal in oil-water monolayers if similar excitation powers were used (data not shown). Moreover, higher excitation intensities resulted in extensive photobleaching. For these reasons, the fluorescent head-labelled lipid Rh-PE was used as a probe for experiments in LDs, rather than TF-PC. Pre-warmed triolein was added to the dried film (1:500 mass ratio) and the mixture was extensively vortexed for 15 minutes and sonicated for 30 minutes. The solution was then added to 20 mM NaP buffer of pH 7.4 (1:10 v/v) and sonicated for 10 seconds after short vortexing, thus obtaining an oil in buffer emulsion that contained the LDs. 75 µL of this

LD emulsion was transferred to a single well of a 96-well plate and further diluted with an equal volume of the same NaP buffer prior to imaging. This procedure results in LDs with low mobility (as they appear to be adhering to the glass surface of the observation chamber).

2.5 Line-Scan Fluorescence Correlation Spectroscopy

Line-Scan Fluorescence Correlation Spectroscopy (lsFCS) was performed on GUVs and LDs as previously described[49,56–58]. Briefly, GUVs or LDs were scanned perpendicularly 500,000 times using a pixel dwell time of 0.79 µs and resolution of 256 pixels (total scan time ~4 min). The pixel size was 0.08 µm. A 488 nm argon laser at a power of 1.45 µW was used to excite the GUV samples and a 561 nm laser (6.5 µW) was used to excite the LD samples, while keeping the photobleaching below ca. 10% of the total initial intensity. Each GUV or LD was scanned only once. All measurements for the GUVs were performed on a Zeiss LSM780 system (Carl Zeiss, Oberkochen, Germany) using a Plan-Apochromat 40x/1.1 Korr DIC M27 water immersion objective. Measurements on the LDs were performed on a Zeiss LSM880 system with Plan-Apochromat 40x/1.2 Korr DIC M27 water immersion objective. Calibration of the excitation beam waist (w_0) was performed each day by measuring the fluorescence autocorrelation curve of 0.25 µM Alexa488 dye in 40 mOsmol/kg PBS with 35 mM sucrose for the GUV experiments to correct the variation in laser alignment. The same calibration procedure was performed for LD experiments, using a similar concentration of Alexa 555 dye in 20 mM NaP buffer. A mean of the diffusion times (τ_D) from three independent autocorrelation curves (100 repetitions with acquisition time 5 seconds) was used to calculate w_0 using the reported diffusion coefficient (D) of the fluorescent dyes[59,60] neglecting, as an approximation, the ca. 2% decrease due to the presence of the solute[61,62]:

$$D = \frac{w_0^2}{4 \cdot \tau_D} \tag{2}$$

All measurements were performed at room temperature and with a confocal pinhole size of 39 µm for measurements in GUVs and 45 µm for those in LDs. The collar ring of the objective was adjusted to optimize the signal. Each data set was exported as TIFF files and then analysed in MATLAB (The MathWorks, Natick, MA) using custom-written code[63].

The effective illuminated area (A_{eff}) measured on the GUV or LD during the line scanning was calculated as[49]:

$$A_{eff} = \pi \cdot S \cdot w_0^2 \tag{3}$$

wherein S is the structural parameter (derived from the daily calibration).

Each measurement provided the number of particles (N) in the measured area and τ_D for TF-PC or Rh-PE. A simple calculation yielded the actual number of all (i.e., labelled and unlabelled) lipid molecules in the illuminated area (N_{total}), since the dye was included at a known concentration compared to the other lipids. Finally, the mean molecular area (MMA_b) or area per lipid in the bilayer was calculated using equation 3:

$$MMA_b = \frac{A_{eff}}{\left(\frac{N_{total}}{2}\right)} \tag{4}$$

In LDs, the mean molecular area ($MMA_{m\text{-}oil}$) was calculated as:

$$MMA_{m-oil} = A_{eff}/N_{total} \tag{5}$$

2.6 Raster Image Correlation Spectroscopy

Raster Image Correlation Spectroscopy (RICS) was performed on air-water lipid monolayers as previously reported[50,64]. The microscope setup was the same as it was for the GUV measurements described in the previous section. A total of 30 frames with a resolution of 256 X 256 pixels (pixel size 0.05 µm) were acquired at a pixel dwell time of 1 µs and 2.2 µW laser excitation power. The photobleaching was thus always below 10% of the initial intensity. Due to continuous evaporation of the water subphase, the monolayer moved consistently downwards in the z-directions causing the fluorescence signal to decrease with time. To counteract this, at the beginning of the measurement, the focus was positioned ca. 1 µm below the monolayer to allow the focal plane to move in the opposite z-direction and the measurement time was kept short (ca. 4 seconds) to get a quasi-constant and maximized signal[65]. Each data set was analysed in MATLAB using custom-written code[66]. The analysis directly provided lipid diffusion in monolayers (D_m) and the surface concentration C of TF-PC molecules in the monolayer. As described in the previous section (section 2.2), knowledge of TF-PC concentration relative to lipids ($C_{\%TF\text{-}PC}$) and surface concentration allowed the estimation of the lipid MMA (MMA_m):

$$MMA_m = \frac{C_{\%TF-PC}}{C \cdot 100} \tag{6}$$

The correct formation of a lipid monolayer with the expected properties was also confirmed at this step by comparing MMA_m with the anticipated value of $MMA_{expected}$ obtained from equation 1.

2.7 Molecular dynamics simulations

To calculate the thicknesses of the monolayers and bilayers and to study the effect of acyl chain chemistry on interleaflet interactions, we resorted to all-atom molecular dynamics (MD) simulations. Single-component lipid bilayers containing a total of 1024 POPCs, DOPCs, SOPCs, or OMPCs and solvated by 50 water

molecules per lipid were set up using CHARMM-GUI and equilibrated following the associated protocol[67]. While POPC, DOPC, and SOPC are readily available in the CHARMM lipid library, the OMPC topology was built in-house following the building block approach of CHARMM lipids. These bilayers were then simulated for 100 ns, after which their MMAs had stabilised. We then converted the bilayer systems into monolayer ones by the translation of the coordinates of one of the leaflets and the simultaneous addition of a vacuum slab into the system. The monolayer systems contained two monolayers lining the water slab, and a vacuum phase (~17.5 nm) separating the acyl chains across the periodic boundary conditions.

The lipids were modelled using the CHARMM36 lipid force field[68]. Water was described with the CHARMM-specific TIP3P potential[69,70]. All four bilayers and four monolayers were simulated for 1 µs using GROMACS 2021.5[71], and the last 900 ns were used in all analyses.

The integration of the equations of motion was performed with a leap-frog integrator with a time step of 2 fs. The smooth particle mesh Ewald algorithm was used to incorporate long-range electrostatics with the real space cutoff optimised for each run[72]. The Lennard-Jones potential was truncated at 1.2 nm, and the forces were switched to zero beyond 1.0 nm. Information of atomic neighbours was maintained using buffered Verlet lists[73]. The monolayers were simulated in the canonical (NVT) ensemble, whereas the isothermal–isobaric (NPT) ensemble was used for the bilayers. The temperature of the lipids and water were separately coupled to the Bussi–Donadio–Parrinello thermostat[74]. The target temperature was set to 298.15 K, and the time constant to 1 ps. For bilayers, pressure was maintained at 1 bar using a semi-isotropic coupling scheme (the two dimensions in the plane of the membrane coupled together). The Parrinello–Rahman barostat[75] with a time constant of 5 ps and compressibility of 4.5×10^{-5} bar^{-1} was used. All bonds involving hydrogen atoms were constrained using P-LINCS[76].

In order to estimate monolayer and bilayer thickness values with a single approach, we considered the bilayer and monolayer thickness to be the span of the *z* coordinate (along the bilayer/monolayer normal) in which the lipid density was larger than 10% of its maximum value (see supplementary Fig.S1). These thicknesses were evaluated for the bilayer and the monolayer, as well as for a single leaflet in the bilayer. The density profiles were extracted using the gmx density tool included in the GROMACS distribution.

2.8 Viscosity measurements and interleaflet coupling

The aim of this work is to compare interleaflet interactions for different lipids, by quantifying the difference in lateral diffusion between monolayers and bilayers. However, different lipid compositions result in varying membrane thickness values which, in turn, would affect diffusion dynamics[48,77]. For this reason, we quantify the viscosity of lipid membranes and interleaflet interactions by evaluating the experimentally obtained *D* and thickness values in the context of the Saffman-Delbrück-Hughes-Pailthorpe-White (SDHPW) model[48,77]. To

this aim, we prepared water-air lipid monolayers and LDs (i.e., water-oil monolayers) with the same *MMA* of the corresponding bilayers (i.e. bilayers with the same lipid composition), as mentioned in section 2.1. Such *MMA* values are in the range 60-70 Å^2/lipid (see below). The experimentally obtained lipid diffusion in monolayers (D_m) and the monolayer thickness h_m obtained from simulations were then used to calculate the three-dimensional monolayer bulk viscosity μ_m according to the SDHPW model (see Table S1). Using the same formalism of Vaz *et al*.[78], the reduced radius is defined in this case as:

$$\varepsilon = \frac{(\mu_w + \mu_a)a}{\eta_m} \tag{7}$$

with

$$\eta_m = \mu_m \cdot h_m \tag{8}$$

wherein η_m is the monolayer surface viscosity, a is the radius of the diffusing lipid probe and the viscosities of the bounding fluids on both sides of the sheet (i.e., μ_w and μ_a) are those of water[79] and air[80], respectively. To avoid complications, we have assumed the radius a of the fluorescent lipid probe to be similar to the radius of the non-fluorescent lipid molecules[81,82]. Since, for our experiments, the reduced radius ε is not much smaller than 1 in general, we make use of the analytical approximation introduced by Petrov *et al*.[83] to determine viscosity (μ) from D. Similar calculations are also performed for monolayers at the oil-water interface: from $D_{m\text{-}oil}$, it is thus possible to estimate the bulk viscosity of the monolayer $\mu_{m\text{-}oil}$, considering that the viscosities of the bounding fluids on both sides of such monolayers (i.e., μ_w and μ_t) are those of water and triolein[84], respectively. Of note, only single lipid mixtures resulted in reproducible and stable LD preparations, and, for this reason, oil-water monolayers were prepared exclusively from either DOPC, POPC, SOPC or OMPC samples. For simplicity, we assume here that h_m does not change considerably between the two types of monolayers.

Differently from the case of lipid monolayers, the application of SDHPW-related models to study the diffusion of lipids in bilayers (i.e., objects spanning only one of two leaflets) is less univocally described in the literature. On the one hand, the relationship between the thickness of the sheet (i.e., the bilayer membrane thickness, h_b), the thickness of a leaflet h_l and the length of the diffusing object is not clear[28,85]. Additionally, although continuum models are used with great utility for modelling small solute diffusion[28], it was suggested that other models (e.g., free volume model[78]) might describe lipid diffusion in bilayers better than a continuum fluid hydrodynamic formalism. Nevertheless, it must be noted that these models have several parameters that are difficult to independently quantify in the context of a comparison of lipids with different structures. Most importantly, the goal of this work is not to provide an absolute estimation of viscosity values, but rather to compare the decrease in lipid diffusivity between monolayers and bilayers, for different lipid types. Therefore,

to compare monolayers and their corresponding bilayers (i.e., same lipid composition and similar *MMA*), we used two different simple approaches.

First, following the reasoning of Vaz *et al.*[78], TF-PC in a bilayer is assumed to be embedded and diffuse within a single leaflet with bulk viscosity μ_l. The viscosities of the bounding fluids are assumed to be that of water, μ_w, on one side of the leaflet and that experienced at the bilayer midplane, $\mu_{midplane}$, on the other. Using the experimentally determined values for lipid diffusion in bilayers (D_b) and h_l from MD simulations, we calculated $\mu_{midplane}$, using the analytical solution by Petrov *et al.*[83] and the following definition[78]:

$$\varepsilon = \frac{(\mu_w + \mu_{midplane})a}{\mu_l h_l} \tag{9}$$

For each lipid composition, several replicate measurements on monolayers and bilayers were performed. For each estimation of D_m, one μ_m value was calculated. From each μ_m value, one $\mu_{midplane}$ value was calculated using a single D_b value (i.e., $\langle D_b \rangle$ calculated for the specific lipid composition). For example, as shown in Table S1, in the case of OMPC, 49 measurements of D_m resulted in 49 μ_m values, 49 $\mu_{midplane}$ values and one D_b value (obtained, in turn, as average of 122 measurements on bilayers). Vanishingly small $\mu_{midplane}$ values (i.e., comparable to μ_a) are obtained if the diffusivity in (a leaflet of) the bilayers was approaching that measured in the corresponding air-water monolayer, i.e. if each leaflet was "unaware" of the presence of the other. For this reason, we use $\mu_{midplane}$ as an approximate measure of the interleaflet interaction, under the simplifying assumption that μ_l is not affected by such interactions (i.e., $\mu_l = \mu_m$).

As an alternative, we followed one of the approaches mentioned by Adrien *et al.*[85]. Briefly, in a rough approximation, lipid-like fluorescent probes can be treated as cross-layer particles, since it was often observed that D only weakly depends on the length of the diffusing lipid[28,29,33,86–89]. This implies that the length of the diffusing molecule is simply assumed to be equal to the thickness of the whole bilayer h_b. In the absence of any additional interaction between the leaflets, the bilayer bulk viscosity μ_b should be the same as the bulk viscosity measured in the corresponding monolayer. The bulk viscosities of the bounding fluids are that of water, μ_w, on one side and that of 50 mM sucrose, μ_s, on the other side of the membrane:

$$\varepsilon = \frac{(\mu_w + \mu_s)a}{\mu_b h_b} \tag{10}$$

Using each experimentally derived D_b values and h_b (from sFCS measurements and MD simulations, respectively), we calculated the bilayer bulk viscosity μ_b, for each lipid composition. For simplicity, the bulk viscosity of 40 mOsmol/kg PBS is considered to be same as that of water. If μ_b were in fact equal to μ_m, D_b would be lower than D_m simply due to the increased thickness of the lipid sheet and the presence of water on both its sides (instead of water and air). Any further decrease in D_b (compared to D_m) is taken here as an indication of coupling/interaction between the two leaflets, as previously suggested[42]. Concretely, this is estimated as the ratio $<\mu_b>/\mu_m$ or $<\mu_b>/\mu_{m\text{-}oil}$ (referred to as "coupling factor"). For example, as shown in Table S1, in the case of OMPC, 49 μ_m values were experimentally obtained from measurements in monolayers and 49 values for the corresponding coupling factor were calculated, using a single $<\mu_b>$ value (obtained as the average of 122 measurements on bilayers). Regarding the interpretation of the coupling factor, a value of ca. 1 would correspond to the above-mentioned case in which $D_b< D_m$ only because of increased thickness and surrounding fluid viscosity. Concretely, for the lipid systems investigated here, this would correspond to a ca. 2-fold reduction in D when switching from air-water monolayers to bilayers. A coupling factor value above one would indicate that the leaflets in a bilayer have a higher bulk viscosity than their monolayer counterpart and we attribute this increase to interleaflet interaction.

2.9 Statistical analyses

Box plots in Figures 1, 2 and S4, the mean, median, first and third quartile with standard deviations as whiskers. Bars in Figures 3, S5 and S6 represent mean values with standard deviations. Each point in Figure S3 represents mean values with standard error of the mean as error bars. Statistical significance between different datasets, for all figures, was determined using two-sample t-test. All figures and statistical tests were made using Origin (Pro) version 2023 (OriginLab Corporation, Northampton, MA, USA).

3 Results and Discussion

3.1 Lipid packing and mobility in GUVs are simultaneously quantified via lsFCS

In order to study the influence of chain length and asymmetry on lipid packing and mobility in bilayers, we prepared GUV model membranes with varying lipid compositions. The lipids for the GUV preparation were chosen based on their similar structure (i.e. one saturated and one mono-unsaturated oleoyl chain) and the difference between the sn-1 and sn-2 acyl chain length (i.e., OMPC, POPC and SOPC). Together with DOPC, these lipids have been previously the subject of investigation in the context of interleaflet interactions[34,38]. Apart from single-lipid compositions, we examined a lipid mixture as well i.e., GUVs composed of SOPC+mSM, in order to investigate the effect of acyl chain asymmetry[33,34].

All GUVs were labelled with 0.005 mol% TF-PC[90,91] to enable both membrane visualisation and fluorescence fluctuation measurements[49]. Specifically, we performed lsFCS to determine the τ_d and number (N) of TF-PC molecules within the observation volume, in a single experiment[49,56,57]. These parameters are used to calculate the diffusion coefficient D_b (equation 2) and N_{total}. The latter parameter, in turn, allows the estimation of the MMA_b according to equations 3 and 4.

Figures 1 and S2 show that both MMA_b and D_b values do not vary significantly among the tested lipid compositions, taking into account the experimental uncertainties. All average MMA_b values are between ca. 60 and 70 Å^2/lipid. D_b values are between ca. 6 and 8 µm²/s, with the lowest median values being measured for SOPC and SOPC+mSM and the highest for DOPC (see Table S1).

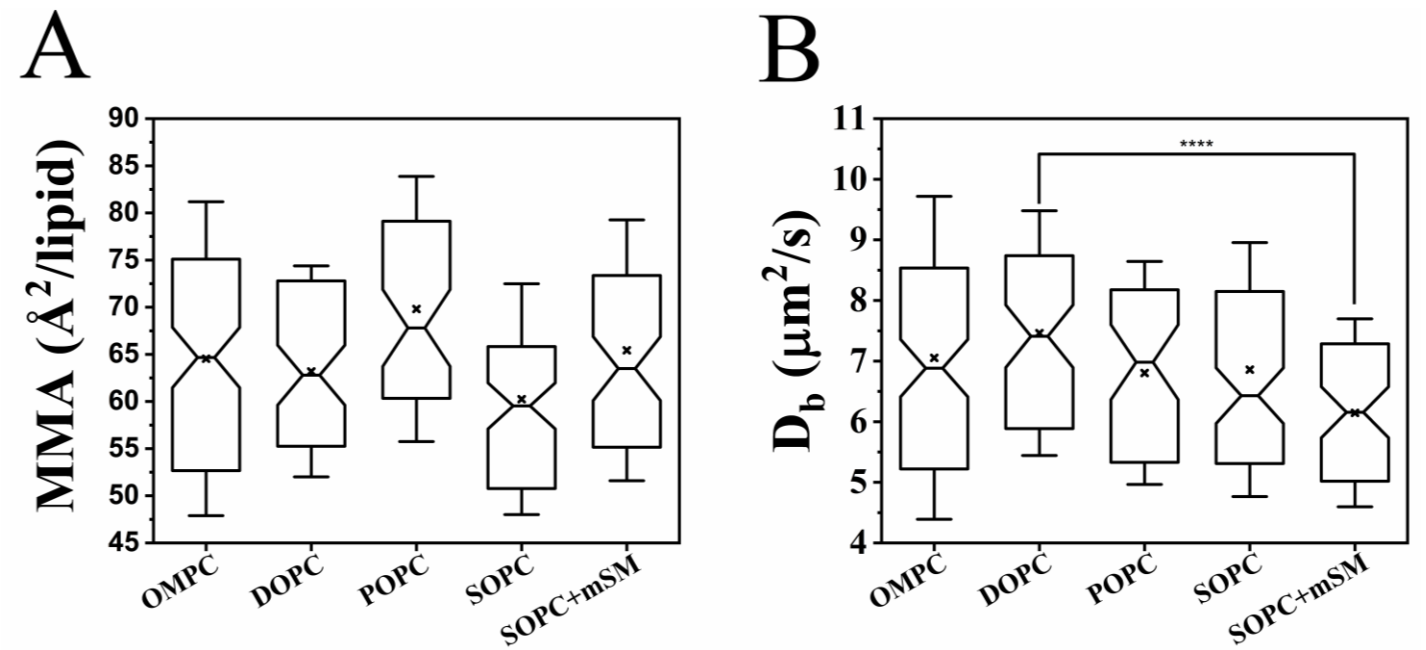

Figure 1: MMA_b and D_b values in GUVs measured via lsFCS as a function of lipid composition.

Each GUV is composed of a particular lipid (i.e., OMPC, POPC, SOPC, DOPC) or a mixture of two lipid species (i.e., SOPC+mSM in 4:1 molar ratio). All GUVs were labelled with 0.005 mol% TF-PC for visualization and lsFCS analysis. (**A**) Box plots of the MMA_b values obtained via lsFCS, as a function of lipid composition. (**B**) Box plots of the D_b values obtained via lsFCS as a function of lipid composition. For each lipid composition, a total of 53 (up to 122) GUVs were analysed from at least 2 (and up to 5) independent sample preparations (Table S3). The box plots originate from 122, 76, 53, 95 and 72 data points for OMPC, DOPC, POPC, SOPC and SOPC+mSM, respectively (Table S1), within 2 (and up to 5) independent sample preparations. In both panels, each box represents the median (notch), first and third quartile, with 'x' marking the mean value. Whiskers indicate standard deviations. ****:p<0.0001.

These results are in good agreement with previously reported data[30,38,92–97]. It is worth mentioning that the data spread observed in these samples might be partially due to slight variations in membrane tension from vesicle to vesicle[98,99], despite the precautions described in section 2.3 employed to obtain GUV samples containing mostly tense/inflated vesicles.

In conclusion, these experiments indicate that packing density and diffusivity in free-standing planar lipid bilayers can be quantified within a single measurement, using lsFCS.

3.2 Diffusive dynamics in monolayers depend on lipid identity and area per lipid

Next, we quantified lipid diffusion in air-water monolayers, i.e. simple models that mimic each leaflet of a bilayer[100,101]. Previous works have reported D_m values for monolayers with few specific lipid compositions[43,44,102–104], however, systematic studies of D_m values as a function of sn-1 and sn-2 acyl chain structure are limited[102]. Here, monolayers were prepared with the same set of lipids as for GUV samples, using a previously described setup[42,43]. TF-PC was added in trace amounts (0.005 mol%) for the purpose of visualization and to obtain the parameters N and τ_d via fluorescence fluctuation analysis. Using the method described by Khmelinskaia *et al.*[42] and Chwastek *et al.*[43], the total amount of lipids was adjusted to explore a large range of MMA values. The resulting (effective) MMA_m and lipid dynamics were quantified using RICS[50,64]. Compared to lsFCS, this approach is more suitable for obtaining the required physical parameters for flat lipid systems (e.g. monolayers or bilayers) parallel to the focal plane[50,65].

As expected, we observed a general correlation between MMA_m and D_m (Fig. S3), in agreement with published data[42,43]. In order to compare lipid dynamics in monolayers and bilayers with the same composition, we focused our analysis on the subset of lipid monolayer preparations which displayed MMA_m values in the range (i.e. within one or two standard errors of the mean) of those measured in average for the corresponding bilayers[105] (Fig. 1A). Concretely, that implies that all the monolayers examined exhibited similar MMA_m values (i.e., between ca. 60 and 70 $Å^2$/lipid). The D_m values obtained for these monolayers are shown in Fig. 2, as a function of lipid composition. For a limited subset of samples (i.e., OMPC and SOPC), we also

measured the diffusion of a different fluorophore, i.e. Rh-PE, and we found similar D_m values (i.e., ~30 µm²/s for OMPC and ~18 µm²/s for SOPC, Table S1) within similar MMA_m range.

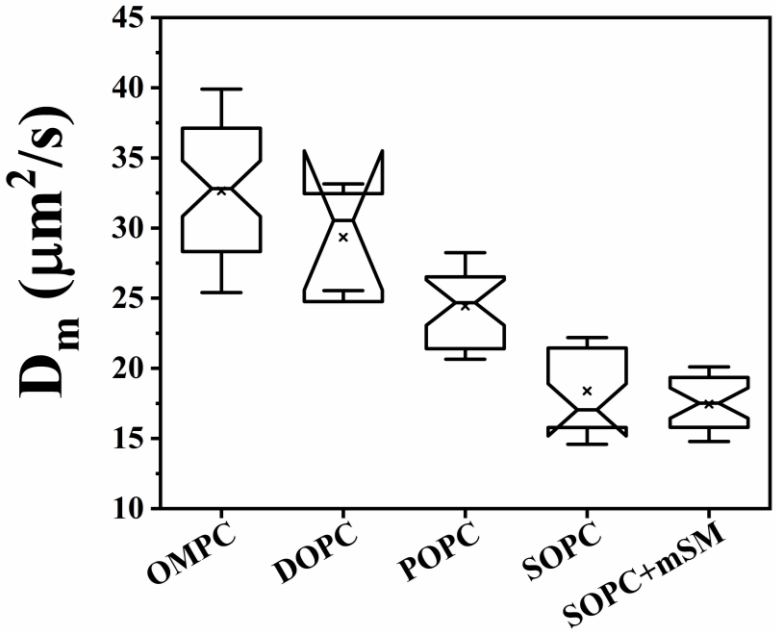

Figure 2: Diffusion coefficients of lipid monolayers measured within specific MMA_m ranges, as a function of lipid composition.
D_m values were obtained via RICS analysis for lipid monolayers composed of OMPC, DOPC, POPC, SOPC and SOPC+mSM (4:1) and labelled with 0.005 mol% TF-PC. All the measured values are shown in Fig. S3. In this figure, the box plots represent the spread of D_m values for monolayers with *MMAs* within the range (i.e., one standard error of the mean) of the MMA_b values measured for the corresponding bilayers (Fig. 1A). For POPC and DOPC, a larger range was used (i.e., two standard errors of the mean) in order to obtain at least 6 data points for each box. The box plots originate from 49, 6, 25, 23 and 27 data points for OMPC, DOPC, POPC, SOPC and SOPC+mSM, respectively, within 2 (and up to 5) independent sample preparations (Table S2). Each box represents the median (notch), first and third quartile with 'x' marking the mean value. Whiskers indicate standard deviations.

In line with previous results[89,106], lipid dynamics are significantly faster in monolayers compared to bilayers by a factor of ca. 2-5. OMPC and DOPC monolayers exhibit the highest D_m values (~30 µm²/s). Reasonably, lipid diffusion is faster in monolayers of fully unsaturated lipids (i.e., DOPC) and slows down for lipids possessing long saturated acyl chains (i.e., D_m for SOPC<POPC<DOPC). Interestingly, within the specific explored MMA_m range, we did not observe significant differences between SOPC monolayers and a more complex mixture including mSM (p=0.3).

Similar experiments were carried out also in LDs, i.e. monolayers at the oil-water interface (Fig. S4 and Table S1), as it was previously suggested that such monolayers are more reliable models of the leaflets in a bilayer[105]. From a qualitative point of view, $D_{m\text{-}oil}$ values vary for the examined lipid compositions following the same trend observed for air-water monolayers (Fig. S4). Interestingly, the absolute lipid diffusion values in LDs are ca. 4 times smaller than those measured in air-water monolayers, as expected, due to the high viscosity of the surrounding media (i.e., water and oil vs. water and air) and as it was recently reported for DOPC and POPC oil-water monolayers[102].

Finally, as described in section 2.8, D_m and $D_{m\text{-}oil}$ can be used to estimate the monolayer bulk viscosities μ_m and $\mu_{m\text{-}oil}$ for these monolayers, according to the continuum fluid hydrodynamic model proposed by Hughes *et al*.[48,77,83]. As shown in Table S1, the measured viscosities in the monolayer systems are in the range ~10-60 mPa·s. Furthermore, we do not observe in general significant differences between the viscosity values obtained for each lipid, in air-water monolayers or in oil-water monolayers (see 95% confidence intervals for μ_m and $\mu_{m\text{-}oil}$ in Table S1). This result supports the simple view according to which the diffusion of a probe in a lipid monolayer is indeed determined by the bulk viscosity of the monolayer itself (which is within comparable ranges for oil-water and air-water monolayers), the bulk viscosity of the surrounding media and the size of the diffusing molecule[85]. Previous results have suggested that lipid dynamics in oil-water monolayers might be additionally influenced by interdigitation between oil molecules and lipid acyl chains[107]. Nevertheless, we do not observe such an effect, in the limits of our precision, as evidenced by the similarity of μ_m and $\mu_{m\text{-}oil}$ for most lipids (Table S1). Also, it has been suggested that the chemical nature of the diffusing probe (and the specific interactions with the environment) might affect the determination of the membrane viscosity[85]. Nevertheless, within the experimental uncertainty, we do not observe this effect, as apparent from i) the similar bulk viscosity values obtained in oil-water monolayers and air-water monolayers and ii) the similar D_m values obtained using different fluorophores, at least in OMPC or SOPC monolayers (see above).

3.3 Leaflet-leaflet interactions depend on lipid identity

Decreased lipid diffusive dynamics in bilayers, as compared to water-air monolayers with the same *MMA*, can be expected in general considering the different thickness and boundary conditions (water-water for bilayers and, e.g., water-air for monolayers)[48,77]. Any additional effect might be attributed to leaflet-leaflet interactions. From a qualitative comparison of the results shown in Figs. 1B and 2, it appears that the dynamics of certain lipid species (e.g., OMPC) decrease drastically when switching from an air-water monolayer to a bilayer system. For other lipids (e.g., SOPC), this effect is much smaller. Of note, the large difference between D_m and D_b observed for instance for OMPC samples does not appear to strongly depend on the specific choice of *MMA* (see Fig. S3). Such a simple comparison, though, does not take into account other important variables, such as the varying thickness of monolayers and bilayers of different lipid compositions. In what follows, we try thus to quantify such an effect considering also the variations in lipid layer thickness and viscosity of surrounding media, while comparing bilayers and different types of monolayers. The goal of this analysis is not to determine absolute viscosity values for the different analysed lipid systems, but rather to quantify the changes in the "apparent" viscosity and diffusivity in monolayers and bilayers, for different lipid structures.
To this aim, as described in detail in section 2.8, we first calculated the bulk viscosity of the investigated bilayers μ_b, using a simplified approach[85], according to which a fluorescent lipid analogue can be

approximatively considered a cross-layer particle (i.e., spanning the whole bilayer length, for the purpose of this calculation)[28,29,33,86–89]. In this case, the viscosities of the surrounding fluids are taken to be that of water on one side and 50 mM sucrose on the other side of the bilayer (see equation 10). As shown in Table S1, the resulting bilayer viscosity values μ_b are between ~50 and 60 mPa·s, in agreement with previous results[30,108,109]. Second, we defined a "coupling factor" as the ratio between the mean bilayer bulk viscosity and the bulk viscosity of each corresponding monolayer, for all lipid compositions, as discussed in detail in section 2.8. Figures S5 (A and B) reports the absolute values of the coupling factors calculated for all lipid samples using bulk monolayer viscosities of either air-water or oil-water monolayers. These values range between ca. 1 and 10, with values close to unity (as, e.g., observed for SOPC) suggesting that the bulk viscosity of a bilayer (or each leaflet of the bilayer) is not too far from its corresponding monolayer model. In other words, a value close to 1 would indicate that the decrease in lipid diffusion observed when switching from monolayers to bilayers can simply be ascribed to an increased layer thickness and a change in the surrounding medium, with no additional contribution from e.g. inter-leaflet interactions. Larger values suggest the presence of "additional" factors that might slow down dynamics in bilayers. In order to simultaneously compare the coupling factors referring to air-water and oil-water monolayers, Fig. 3 shows the normalized coupling factors obtained from both types of monolayer systems. For the case of SOPC/mSM mixtures, only water-air monolayers could be reproducibly obtained. In general, no major differences can be observed in the trend of the results obtained from calculations involving either type of monolayer. As previously suggested[105] though, we propose that coupling factors calculated using oil-water interfaces might be more reliable than those calculated using air-water monolayers. Regardless, our results suggest that the "additional" decrease in bilayer diffusive dynamics follows the trend OMPC>DOPC≈POPC≳SOPC≈SOPC+mSM.

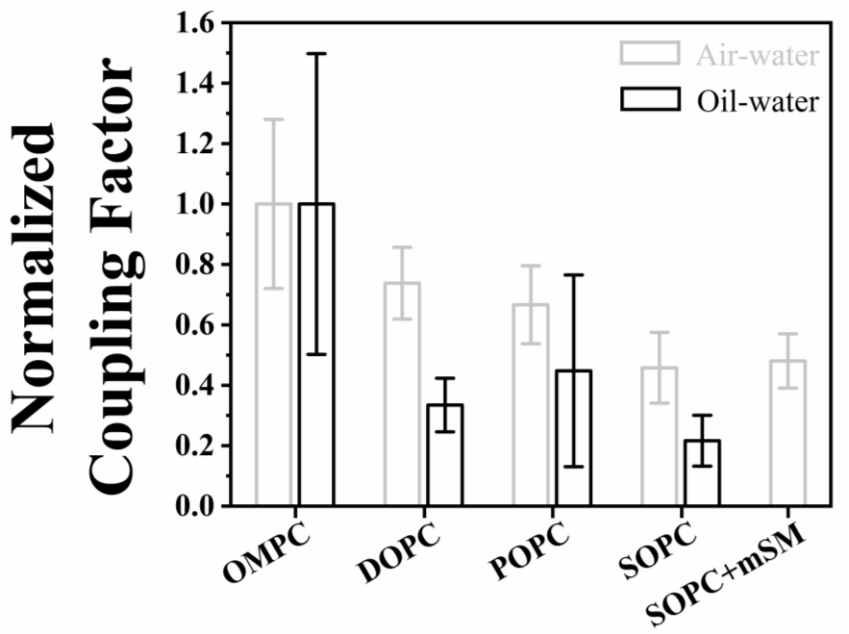

Figure 3: Normalized coupling factors calculated from different monolayer models (air-water and oil-water), as a function of lipid type.

D_b values from Fig. 1B, D_m from Fig. 4 and $D_{m\text{-}oil}$ from Fig. S4 are used to calculate μ_b, μ_m and $\mu_{m\text{-}oil}$ using the continuum fluid hydrodynamic model, under the assumption that TF-PC or Rh-PE can be considered a cross-layer particle (see main text, section 2.8). For each lipid composition, the bars represent the mean values of the normalized ratio between the $<\mu_b>$ value shown in Table S1 and each μ_m (light grey bars) or $\mu_{m\text{-}oil}$ (black bars) value obtained from D_m or $D_{m\text{-}oil}$ values, as represented in Fig. 2 and S4 respectively. The light grey bars originate from 49, 6, 25, 23 and 27 data points for OMPC, DOPC, POPC, SOPC and SOPC+mSM respectively (Table S1). The black bars originate from 10, 5, 47 and 7 data points for OMPC, DOPC, POPC and SOPC respectively (Table S1). The whiskers indicate the standard deviations.

As an alternative characterization of possible interleaflet interactions causing hindered dynamics in lipid bilayers compared to monolayers, we applied the analysis proposed by Vaz *et al.*[78], according to which fluorescent lipids in a bilayer diffuse within a single leaflet. As discussed in detail in section 2.8 (see equation 9), when analysing a single leaflet of a bilayer according to the SPHPW model, the bulk viscosities of the surrounding fluids are that of water on one side of the leaflet and that of the bilayer midplane $\mu_{midplane}$, on the other. Such "interleaflet or midplane viscosity" is shown in Fig. S6 for all the examined lipids, referring alternatively to either μ_m or $\mu_{m\text{-}oil}$. Interestingly, the results obtained for this approach display a trend which is qualitatively similar to that indicated by the "coupling factor" analysis (Fig. 3 and S5). Independently from the type of monolayer used in the analysis, $\mu_{midplane}$ is consistently the highest in OMPC bilayers. This is particularly evident in the case of oil-water monolayers (Fig. S6B). Only minor differences are observed between the other lipid compositions. For example, the addition of mSM to SOPC bilayers does not appear to significantly influence the $\mu_{midplane}$ (p=0.05) values. Since mSM is a mixture characterized by a high degree of acyl chain asymmetry[34,95], this suggests that interdigitation might not play a major role in explaining specifically the differences in diffusive dynamics between monolayers and bilayers. Of interest, we have previously reported that interdigitation might instead play a role in the process by which reduced dynamics in one leaflet of an asymmetric bilayer induce a similar reduction also in the opposing leaflet[34]. These observations confirm that acyl chain interdigitation has a non-trivial role in mediating interleaflet interactions and lipid dynamics[11,39,40].

In line with the results shown here, previous studies from ours and another group[34,38] identified OMPC as a peculiar lipid, in the context of trans-bilayer interactions and spatial organization of the bilayer midplane. The general importance of the spatial distribution of methyl groups at the bilayer midplane in determining interleaflet interaction was also highlighted by other studies[30,31,40,109]. Specifically, Capponi *et al.*[38] proposed that OMPC bilayers are characterized by a "distributed complementarity" of methyl groups, i.e. the peak positions of the *sn*-1 and *sn*-2 methyl distributions in the same leaflet are extraordinarily distant, compared to other lipid bilayers. Of interest, the same behaviour could be reproduced in our simulations that included a ca. 14-fold higher number of lipids in the bilayer (Fig. S7). Such spatial distribution of the acyl chain terminal groups might be associated with enhanced packing and, presumably, stronger inter-leaflet interactions across the bilayer midplane. It is worth noting that such behaviour was observed, although to a lower degree, also for

SOPC bilayers[38]. The discrepancy between the results presented in this work for SOPC and those from previous MD investigations should be an object of future investigation.

4 Conclusions

In order to verify whether lipid acyl chain structure might influence inter-leaflet interactions, we have systematically compared lipid dynamics in monolayers and bilayers with similar *MMAs*. The expected reduction in diffusive dynamics observed for lipid bilayers appears indeed to depend on the component lipid identity. More in detail, we consistently observed the strongest reduction in diffusive dynamics for bilayers composed of OMPC. On the other hand, the presence of asymmetric lipids (which are supposed to induce strong chain interdigitation) does not appear to affect the decrease of lipid diffusion in bilayers, as compared to monolayers. While our results did not indicate a univocal pattern in how this effect depends on specific structural features of the lipid molecules, there might be a connection between the peculiar behaviour of OMPC and the complementary spatial distribution of terminal methyl ends at the bilayer midplane observed via MD. Such a possibility should be explored through an analogous systematic comparison of lipids with different acyl chains performed via alternative experimental approaches.

5 Author Contributions

Titas Mandal: Conceptualization, Methodology, Validation, Formal analysis, Investigation, Data curation, Writing – Original Draft, Writing – Review & Editing, Visualization. **Nadine Brandt**: Methodology, Validation, Investigation, Writing – Review & Editing. **Carmelo Tempra**: Methodology, Software, Validation, Formal analysis, Investigation, Data curation, Writing – Review & Editing, Visualization. **Matti Javanainen**: Methodology, Software, Validation, Formal analysis, Investigation, Resources, Data curation, Writing – Original Draft, Writing – Review & Editing, Visualization, Supervision, Project administration, Funding acquisition. **Balázs Fábián**: Methodology, Software, Validation, Formal analysis, Investigation, Data curation, Writing – Original Draft, Writing – Review & Editing, Visualization. **Salvatore Chiantia**: Conceptualization, Software, Resources, Writing – Original Draft, Writing – Review & Editing, Supervision, Project administration, Funding acquisition.

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Data Availability

Raw data for the figures can be found in the supplementary section.

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