Comparative study of Ergosterol and 7-dehydrocholesterol and their Endoperoxides: Generation, Identification and Impact in Phospholipid Membranes and Melanoma Cells

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ABSTRACT

Melanoma is an aggressive cancer that has attracted attention in recent years due to its
high mortality rate of 80%. Damage caused by oxidative stress generated by radical
oxidation (type I reaction) and singlet oxygen (type II reaction) oxidative reactions may
induce cancer. Thus, studies that aim to unveil the mechanism that drives these oxidative
damage processes become relevant. Ergosterol, an analogue of 7-dehydrocholesterol (7-
DHC), important in the structure of cell membranes, is widely explored in cancer
treatment. However, to date little is known about the impact of different oxidative reactions
on these sterols in melanoma treatment, and conflicting results about their effectiveness
complicates the understanding of their role in oxidative damage. Our results highlight
differences among ergosterol, 7-DHC and cholesterol in membrane properties when
subjected to distinct oxidative reactions. Furthermore, we conducted a comparative study
exploring mechanisms of cell damage by photodynamic treatment in A375 melanoma,
comparing ergosterol, 7-DHC, and their respective endoperoxides. Notably,
endoperoxides generated by singlet oxygen showed superior efficacy in reducing the
viability of A375 cells compared to their precursor molecules. We also describe a step-
by-step process to produce and identify endoperoxides derived from ergosterol and 7-
DHC. While further studies are needed, this work provides new insights for understanding
cancer cell death induced by different oxidative reactions in the presence of biologically
relevant sterols.

INTRODUCTION

Skin cancer was reported to be the fifth most common cancer worldwide in 2020.
It is divided in non-melanoma (basal cell carcinoma and squamous cell carcinoma) and
melanoma. Although melanoma corresponds to only 4% of total skin cancer, it is
considered the most dangerous due to its high death rate of 80% (1). A total of 325,000
new cases of melanoma and 57,000 deaths were estimated for 2020 worldwide, and an
increase of approximately 50% of new cases and 68% deaths are projected for 2040 (2).

Thus, a convenient treatment becomes crucial for the improvement of the survival rate of
this disease.

Sunlight exerts many beneficial effects such as vitamin D synthesis, anti-
inflammatory stimulation, and prevention against several diseases (e.g., certain types of
cancers, diabetes, and multiple sclerosis) (3). However, excess radiation may lead to
negative effects causing oxidative stress and inflammation to skin cells: both UV and
visible light have the potential to promote photoexcitation of endogenous
photosensitizers, which can convert light energy into chemical energy and produce
reactive oxygen species (ROS), leading to redox imbalance in the cells under excessive
radiation (4). Examples of endogenous photosensitizers are porphyrins, flavin, melanin,
and lipofuscin, and the concentration and presence of these photosensitizers influence
the sensitivity of specific tissue or cells to photoinduced damage (5). Photooxidation
reactions are classified as Type I or Type II reactions. Type I reaction generates reactive
radical species, such as superoxide radical anion ($O_2^{\bullet-}$) and hydroperoxyl radicals
(HOO˙), while type II reactions generate singlet molecular oxygen ($^1O_2$) (6). Although
current photoprotection products block UV rays, emerging studies show that
photodamage by visible light is responsible for more than 50% of skin radicals generated
under sun exposure (4, 7) . Among several biomolecules, lipids play essential roles in cell
membrane structure, serve as energy source, and generate metabolites that support
proper cell development and proliferation (8) Lipids are classified into various categories,
including glycerolipids, and sterol lipids. Cholesterol is essential for cell membrane
structure and as a precursor for other steroidal biomolecules such as hormones and
vitamin D (9, 10). When oxidized, cholesterol gives rise to oxidized derivatives, known as
oxysterols, including reactive sterol electrophiles that may disturb cellular function,
leading to several diseases including cancer (11).

Ergosterol (Erg) was first discovered in 1889 in extracts of the ergot fungus
(*Claviceps purpurea*) by Tanret and described as the major sterol of fungi (12). This sterol
is found mainly in the lipid bilayer of the cell membranes of most fungi and is essential for
cell maintenance and growth (13). Erg has a similar function and structure to cholesterol,
differing by two additional double bonds at the C_7 and C_{22} positions and a methyl bond at
the C_{24} position (**Figure 1**). Increasing attention has been given to Erg and its derivatives

given their properties to inhibit cell growth efficacy in several cancer cells, by interacting

with JAK/STAT mechanism or caspase-independent apoptos   (14, 15). Additionally,

studies have indicated that oxidized derivatives of Erg, such as the 5α, 8α-

epidioxiergosta-6, 22-dien-3β-ol (EEP), an Erg peroxide purified from *Ganoderma*

lucidum, induce human hepatocellular carcinoma cell death through the activation of pro-

apoptotic protein Puma and Bax under stimulation of Foxo3 activity by EEP (16).

7-dehydrocholesterol (7-DHC) is a precursor of cholesterol in mammals, with a

very similar structure to Erg (**Figure 1**) (17, 18) . The anticancer effect of 7-DHC in the

melanoma A2058 cell line was demonstrated by the dose-dependent decrease in its

viability. In the absence of light, cells treated with 7-DHC generated ROS, increased pro-

apoptotic Bax protein, and affected mitochondrial permeabilization by reducing

mitochondria membrane potential (19). Additionally, in ovarian cancer, replacing

cholesterol with 7-DHC significantly enhanced the anticancer activity of photosensitizer-

encapsulated liposomes upon irradiation (20). In contrast, other studies demonstrated

that 7-DHC suppressed and regulated cell death by ferroptosis, an iron-dependent non-

apoptotic cell death caused by exacerbated lipid peroxidation (21, 22). It has been shown

that pre-treatment of B16F10 melanoma cells with 7-DHC significantly protected cells

from ferroptosis, suggesting that 7-DHC can increase melanoma resistance against

oxidative stress (21).

Given conflicting results mentioned above and considering the cell death caused

by lipid oxidation reactions, we were interested in comparing the damage to the cell

membrane caused by sterols under oxidation reactions type I and type II. It was also our

interest to compare the cancer cell death efficacy of these products and their respective

endoperoxides. Thus, this work aims to present a step-by-step process to produce

endoperoxide derivatives of Erg and 7-DHC and determine whether they protected lipid

bilayers against Type I or Type II photooxidative damage and their influence on

melanoma A375 cell viability.

Figure 1. Structure of 7-dehidrocholesterol and ergosterol and their oxidation products.

MATERIALS

Phospholipids palmitoyl-oleoyl phosphatidylcholine (POPC, 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine) and egg yolk phosphatidylcholine (EggPC, L-α-phosphatidylcholine) were purchased from Avanti Polar Lipids (Alabaster, USA). Solvents (HPLC grade) were purchased from Merck (Darmstadt, Germany), methylene blue,

128 DMMB and other reagents were purchased from Sigma-Aldrich (St. Louis, USA). Fetal
129 bovine serum, penicillin-streptomycin and DMEM, high glucose, GlutaMAX Supplement,
130 and pyruvate were purchased from Gibco (ThermoFischer Scientific). Deuterium
131 chloroform ($CDCl_3$) was purchased from Sigma Aldrich (Steinheim, Germany)

METHODS

Synthesis and purification of sterol oxidation products

Sterol photo-oxidation

First, 200 mg of sterol (Erg, 7-DHC or cholesterol) were dissolved in 40 mL of
chloroform in a 100 mL round-bottom flask, and 50 µL of the photosensitizer methylene
blue (10 mM in methanol) was added. This solution was placed in an ice bath and an
aliquot was collected at time zero. The mixture of sterol and methylene blue was kept
under oxygen flow and stirred under a 500 W tungsten lamp for 90 minutes (Erg and 7-
DHC) or 5 hours (cholesterol). Aliquots were collected in every 30 minutes for qualitative
analysis by thin layer chromatography (TLC) using a silica plate eluted with ethyl acetate
and isooctane (1:1 v/v). Plates were air-dried, sprayed with 50% sulfuric acid, and heated
to allow visualization of sterols and their oxidized products. (**Figure S1).**

Purification of sterol oxidation products

Sterol photooxidation products were initially separated by flash column
chromatography containing silica gel in hexane. After equilibrating the column with
hexane, a gradient of hexane and ethyl acetate in different volume ratios, 95:5 (50 mL),
90:10 (50 ml), 80:20 (100 ml), 70:30 (100 ml), 60:40 (100 mL), 50:50 (50 ml) and 40:60
(50 ml) was used for elution. Fractions of approximately 10 mL were collected and
checked by TLC (**Figure S2**). Sterol oxidized products were further purified by semi-
preparative HPLC (LC 20AD, Prominence, Shimadzu, Kyoto, Japan) coupled to a Diode
Array Detector (DAD, SPDM20A, Prominence, Shimadzu, Kyoto, Japan) using the

following conditions: Luna C18 column (250 x 10 mm, 5 µm, 100 Å); flow rate 5 mL/min;

40°C and mobile phase consisted of 95% Methanol and 5% water. Each peak was

collected in an amber glass tube using a fraction collector ((FRC10-A, Shimadzu, Kyoto,

Japan) and dried with the rotary evaporator.

Characterization of sterol photooxidation products

Analysis of oxidation products by HPLC-DAD

Sterol oxidation products were analyzed by HPLC (Prominence, Shimadzu, Kyoto,

Japan) coupled to a diode array detector (DAD) using a Gemini C18 column (250 x 4.6

mm; 5 µm, 110 Å), in isocratic elution at a flow rate of 1 mL/min with a mixture of 95 %

methanol and 5% water as mobile phase.

Analysis of oxidation products by mass spectrometry

The analysis of oxidation products was also performed via ultra-high-performance

liquid-chromatography (UHPLC Nexera, Shimadzu, Kyoto, Japan) coupled to an

electrospray ionization time-of-flight mass spectrometry (ESI-TOF-MS, Triple-TOF®

6600, Sciex, Concord, USA). Samples were separated on a C18 column (Acquity BEH

100 x 2.1; 1.7 µm, Waters) using a mobile phase composed of (A) water with 0.1% formic

acid and (B) acetonitrile with 0.1% formic acid. The following solvent gradient was used

for sterol oxidation analysis: 50 to 100 %B in 4.5 min, 100 % from 4.5-8.5 min, 100 to 50

% B from 8.5-9.0 min, and 50% B from 9-13 min. MS data were acquired in positive

ionization mode. All data were obtained in data-dependent acquisition mode (IDA®,

Information Dependent analysis) using Analyst® 1.7.1 software.

Sterols and their photo-oxidation products were diluted in isopropanol and

centrifuged at 1500 rpm for 5 minutes before injection. The following precursor ions were

selected for MS/MS analysis: m/z 411 (ergosterol hydroperoxide, EHP), m/z 429 (5α,8α-

epidioxy-22E-ergosta-6,9(11),22-trien-3β-ol, EEP), m/z 377 ((22E)ergosta-5,7,9,22-

tetraen-3β-ol , DHE) and m/z 380 (Erg) were monitored, and the data were processed

using the Peak View software.

Identification of oxidation products by 1H-NMR:

The ^{1}H NMR (500.13 MHz) spectra were recorded at 0 or 25 °C, using CDCl$_3$ as
solvent. The analyses were performed on a Bruker-Biospin Ascend 500 instrument,
Avance III series, (Rheinstetten, Germany), operating at 11.7 T. The instrument was
equipped with 5-mm trinuclear, inverse detection probe with z-gradient (TXI) or 5mm TCI
CryoProbe. The temperature was controlled by a BCU I accessory. All chemical shifts
were expressed in ppm relative to TMS. The data were acquired and processed using
TOPSPIN 3.5 software (Bruker-Biospin, Rheinstetten, Germany).

Stability test of ergosterol and endoperoxides:

198   Sterol stock solutions and endoperoxide solutions were dried with N$_2$ and
199   resuspended in chloroform maintaining a concentration of 1 mM. These vials were
200   placed in a thermomixer for 5h at 37ºC under stirring at 600 rpm. The aliquots were
201   collected at times zero, 1h, 2h, 3h, 4h and 5h and analysis were carried out by HPLC-
202   DAD.

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Analysis of liposome membrane leakage using a carboxyfluorescein system

Liposome membrane leakage with 1,9-dimethyl-methylene blue (DMMB)

Liposomes containing phosphatidylcholine and sterols were prepared by a method
adapted from Miyamoto *et al.* (2012) (23). Membranes containing 50 mM
carboxyfluorescein (CF) were prepared by mixing POPC (1-palmitoyl-2-oleoyl-sn-
glycero-3-phosphocholine) and sterols (Erg, cholesterol, and 7-DHC) in 10 mM Tris buffer
at pH 8. Four liposomes were prepared: one with 100% POPC as a control and three
others combining 75% POPC with 25% of each steroid. Solutions containing each lipid
mixture were placed in a PYREX$^®$ test tube, vortexed, and dried under N$_2$ stream. The
samples were then left for approximately one hour in a vacuum desiccator to ensure the

complete removal of solvent. Then, the samples were redissolved in 500 µL of Tris buffer

(10 mM, pH 8) containing carboxyfluorescein (50 mM). The resulting mixture was

extruded with Avanti Mini-Extruder kit (Avanti Polar Lipids, Alabaster, US) equipped with

a 100 nm pore membrane to produce large unilamellar vesicles (LUVs) of 100 nm size.

CF-loaded LUVs were subjected to a Sephadex G-50 column purification using 10

mM Tris with 0.3 M NaCl at pH 8 as a running buffer. The CF-loaded fraction was collected

in a microtube and then placed in a 96-well plate at a final concentration of approximately

100 µM of each liposome. 15 µM DMMB was added to each well and irradiated with a

633 nm LED for 90 minutes. CF externalization was monitored over time with Agilent's

Synergy H1 Hybrid Reader equipment, using excitation at λex = 480 nm and emission at

λem = 517 nm. At the end of the reaction, Triton X-100 was added to release all the CF

content from the liposomes, to generate the final reading intensity (24, 25).

Liposome membrane leakage by oxidation using iron/ascorbate

Liposomes containing egg yolk phosphatidylcholine (EggPC) and sterols (Erg,

cholesterol, and 7-DHC) were prepared in Tris-HCl buffer as described in the previous

section, however, the DMMB solution was replaced by Fe (III)/ascorbic acid (Fe/Asc) at

10 µM/100 µM final concentrations. All wells contained the final concentrations of

approximately 100 µM of each liposome. CF release was measured as described above.

Quantification of generated oxidized products by HPLC and Mass spectroscopy

Quantification of oxidation products from liposome membrane leakage test by HPLC

Liposomes were produced following the same protocol described previously

without CF. Briefly, 200 µL of each liposome and 27 µL of DMMB 15 mM were placed in

a well plate and irradiated at 633 nm for 90 minutes. Then, the samples were analyzed

by HPLC with C8 column (Luna Phenomenex, 250 x 10 mm, 5 µm, 100 Å), isocratic mode

(94% MeOH and 6% water); flow rate of 1.0 mL/min at 36ºC and 30 µl sample injection

volume.

Quantification of liposome phospholipids by mass spectrometry

POPC was analyzed by LC/MS in the negative ionization mode, with a C18 column (CORTECS Waters 2.1 mm x 10 mm; 1.6 μM) at 35 °C. The flow rate was 0.2 mL/min; injection volume was 2.0 μL, and the mobile phase was (A) water: acetonitrile (60:40); (B) isopropanol: acetonitrile: water (80:10:2) with ammonium acetate (10 mM) in both phases. The gradient started at 40% and increased to 100% over the first 10 minutes. It was held at 100% from 10 to 12 minutes, then decreased back to 40% between 12 and 13 minutes, and was maintained at 54% from 13 to 20 minutes.

DMPC (PC14:0/14:0) was used as an internal standard to semi-quantify POPC, and its concentration was calculated by the peak area of the compound divided by the peak area of the internal standard with known concentration. Aliquots of the samples (10 μl) were mixed with 20 μL of DMPC (5 mM) and 170 μL of isopropanol.

Quantification of ergosterol, 7-dehydrocholesterol, and their oxidized products by mass spectrometry:

To quantify Erg, a calibration curve was built using 7-dehydrocholesterol (7-DHC) as an internal standard. Working solutions for the calibration curve were prepared at concentrations of Erg ranging from 15.6 μM to 1000 μM. To quantify Erg oxidation products (EEP, EHP), a calibration curve was prepared with concentrations ranging from 0.08 μM to 20 μM, using oxidized standards prepared in the laboratory. 7-DHC endoperoxide (CEP) and 7-DHC hydroperoxide (CHP) were used as internal standards at final concentrations of 7.5 μM and 5 μM, respectively.

Quantification of cholesterol and cholesterol hydroperoxides by iodometry and HPLC

Cholesterol hydroperoxides (ChOOH) synthesized by photo-oxidation with methylene blue were quantified by iodometry (26). The quantification of cholesterol and ChOOH in the liposome after photo-oxidation with DMMB was carried out by HPLC, following methods described in **Supplementary Materials**.

In vitro A375 cell viability assay

Cell cultures

A375 cells were cultured with Dulbecco's Modified Eagle's Medium-high glucose (DMEM, catalog number 10569010, Thermo Fischer Scientific) supplemented with 1% penicillin-streptomycin and 10% FBS. It was maintained at 37 °C under a water-saturated atmosphere containing 5% CO_2, until 80%–90% confluence was reached. Cell counting was performed microscopically with Trypan blue exclusion (27).

Photodynamic treatment with sterols and DMMB

Cells were incubated in a 96-well plate with increasing concentrations of sterol (Erg, 7-DHC, EEP, or CEP) for 22 h. Then, DMMB (0.1 to 5 nM) (Sigma-Aldrich, São Paulo, Brazil) was added and incubated for an additional 2 h. The medium was then replaced with PBS, and the cell plate was irradiated at final dose of 5J/cm^2, using red light with 27.84 mW/cm^2 at 660 nm λ_{max} (light source from Biolambda, Brazil).

Cell viability assay

The medium of the cell plate incubated after 0 h and 24h irradiation were discarded and 0.01 g/L resazurin solution in the medium was added to each cell well. Cell viability was determined by measuring fluorescence using the program SoftMax Pro 5.4.1, at 540 nm excitation wavelength and 590 nm emission wavelength. The viability curve was prepared using GraphPad Prism 8.0.1. Data was expressed as means ± SD of at least three independent experiments.

RESULTS

Photooxidation of ergosterol and 7-dehydrocholesterol yields endoperoxides as the most stable products

Erg and 7-DHC were photo-oxidized in chloroform with methylene blue as a photosensitizer (**Figure S1**). Aliquots of the reaction mixture were analyzed by TLC and HPLC. Nearly all the sterols were consumed after 90 minutes of irradiation, resulting in the formation of 3 major photo-oxidized products. Oxidized products were first purified by flash column chromatography and fractions were checked by TLC (**Figure S2**). The fractions containing the endoperoxide, hydroperoxide, and triene products were dried and resuspended in 3 mL of isopropanol for further purification by HPLC. **Figure S3** shows the peaks corresponding to the three major products that were purified by semi-preparative HPLC and their estimated yields. The photooxidation of ergosterol and 7-dehydrocholesterol primarily produced endoperoxides (EEP and CEP) with yields of approximately 37-38%. This was followed by the formation of hydroperoxides (EHP and CHP) and triene products, which were produced at lower yields.

After purification, each oxidation product appeared as single peaks in the HPLC chromatogram, confirming their successful purification (**Figure 2**). Their identities were further confirmed by high-resolution LC-MS analysis.

Figure 2. HPLC analysis of the purified photooxidation products derived from (A) ergosterol: dehydroergosterol (DHE), ergosterol endoperoxide (EEP), and ergosterol hydroperoxide (EHP); (B) 7-dehydrocholesterol: cholestanetrienol (CTL), 7-dehydrocholesterol endoperoxide (CEP), and 8-dehydrocholesterol hydroperoxide (CHP).

MS/MS spectra with the respective fragmentations of the purified ergosterol oxidation products are shown in **Figure 3**. EEP and DHE protonated ions $[M+H]^+$ were observed at m/z 429.3372, respectively, while for the DHE and EHP a dehydrated protonated ions $[(M-H_2O)+H]^+$ appeared as a major ion at m/z 377.3157 and 411.3257. Characteristic fragments were found at m/z 411.32, 393.31 corresponding to the loss of water and O_2 from EEP; and m/z 375.30 for EHP, corresponding to the loss of three water molecules. Endoperoxides of ergosterol and 7-DHC were further confirmed by 1H-NMR

analysis (**Figure S4**, **Table S1** and **Table S2**).

Figure 3. Extracted ion chromatogram and fragmentation spectrum obtained in MS/MS

mode of Ergosterol photooxidation products: (A) EEP; (B) DHE and (C) EHP.

Next, we investigated the production of endoperoxides and hydroperoxides using

dimethyl-naphthalene endoperoxide (DMNO2) as a clean source of singlet oxygen.

Analysis of the oxidation products by HPLC showed that ergosterol incubation with

DMNO2 generates both EEP and EHP, suggesting that singlet oxygen reacts ergosterol

double bonds through *ene* and cycloadditin reactions (**Figure 4**).

Figure 4. Chromatogram of HPLC analysis of ergosterol reaction with $DMNO_2$ monitored

at 210 nm. The pink line represents the initial time and the black line the final time of the

reaction.

Quantitative analysis of ergosterol oxidation products formed with increasing

concentrations of DMNO2 was performed by mass spectrometry (**Figure 5**). As expected,

ergosterol consumption and EEP formation increased proportionally with the generation

of $^1O_2$. However, higher levels of singlet oxygen generated by DMNO2 led to a decrease

in EHP levels and an increase in the formation of a third product, identified as a di-

hydroperoxide (di-EHP) (**Figure S5**). These results demonstrate that the reaction of

singlet oxygen with ergosterol produces EEP, EHP, and small amounts of di-

hydroperoxides (**Figure 5**). Furthermore, stability analysis showed that EEP is highly

stable, while EHP degrades rapidly (**Figure S6**). This suggests that EHP is formed

through Type II photooxidation but is either converted to di-EHP or degraded during

reaction. Overall, these findings indicate that endoperoxides is the most stable products

of photooxidation.

Figure 5. Concentration of ergosterol and products by reaction with DMNO$_2$ analyzed by HPLC-MS.

Sterols in liposomes exhibit distinct membrane leakage kinetics under type I and II oxidation

Membrane protective effects of Erg and 7-DHC were evaluated under type I and type II oxidation conditions using sterol-containing liposomes loaded with carboxyfluorescein. For type II oxidation (singlet oxygen), liposomes were prepared with POPC (control) or POPC with 25% sterol (Erg, 7-DHC, or Cholesterol) and DMMB as a photosensitizer. A significant reduction in membrane leakage was observed for liposomes containing sterols (**Figure 6A**). While carboxyfluorescein release in the controls reached 75% after 180 minutes of irradiation, liposomes containing Erg or 7-DHC released less than 35% of the probe. Notably, cholesterol provided the highest protection, with minimal release of the probe (<5%). For comparison, membrane leakage studies were conducted

using iron/ascorbate (Fe/Asc) as a radical generator under type I oxidation. In contrast to
the results observed under type II oxidation, Erg and 7-DHC provided greater protection
to the liposome membrane than cholesterol, particularly during the initial phase of
oxidation propagation (**Figure 6B**), which is consistent with their high radical-trapping
activity.

Figure 6. Leakage kinetics of carboxyfluorescein encapsulated in (A) POPC liposome in
the presence of the photosensitizer DMMB and (B) Egg phosphatidylcholine liposome
(EggPC) in the presence of 10 µM Fe/Asc. Liposomes were composed of either 100%
POPC or EggPC, or a mixture of 75% POPC or EggPC with 25% sterols (Erg, 7-DHC or
Cholesterol). Leakage was monitored by measuring fluorescence at 517 nm.

Quantification of phospholipid and sterol photo-oxidation products

Sterols protect liposome membranes against phospholipid oxidation

To investigate the membrane protective effects of Erg against photosensitized
oxidation, we analyzed the kinetics of phospholipid oxidation in the liposome membranes.
Phospholipid hydroperoxides (POPC-OOH), the primary products formed by the reaction
of singlet oxygen with POPC, were quantified by HPLC. As shown, the control decrease

(**Figures 7A)** was accompanied by an increase in POPC-OOH (**Figures 7B)**. In control
liposomes (POPC), hydroperoxide levels reached a plateau at 20 minutes and then
decreased after 40 minutes of irradiation (**Figure 7B**), likely due to the degradation of
POPC-OOH forming truncated phospholipid species responsible for membrane leakage
(19). In contrast, ergosterol (POPC + Erg) and cholesterol (POPC + Chol) protected
membranes against phospholipid oxidation, leading to lower POPC-OOH levels
compared to liposomes without sterols (POPC). Liposomes with Erg (POPC + Erg)
showed the least generation of POPC-OOH (**Figure 7B**).

419 **Figure 7.** Time-dependent analysis of POPC consumption and formation of photo-
420 oxidation products in liposomes containing only POPC (4 mM), and POPC (1 mM) with
421 cholesterol (Chol) or ergosterol (Erg) (3 mM). (A) Percentage of POPC over the different
422 reaction times with control. Generation of POPC oxidized products: (B) hydroperoxide
423 (POPC-OOH), (C) hydroxide (POPC-OH) and (D) ketone (POPC=O)

 Oxidation product consumption kinetics differ among sterols

To determine whether the membrane protective effects of sterols are due to their
action as antioxidants or their membrane stabilization effects, we quantified the
consumption of sterols and the formation of oxidized products. Notably, Erg was almost
totally consumed within the first five minutes of the photooxidation reaction. This
consumption was paralleled by the formation of EEP and EHP, with EEP as predominant
product. (**Figure 8A**). This indicates that Erg acts as a sacrificial antioxidant protecting
membranes from photodamage. Additionally, this data confirms that EEP is the most
stable Erg photooxidation product.

In contrast to Erg, the consumption and formation of cholesterol oxidation products
showed a different profile. While Erg was consumed within the first few minutes,
cholesterol consumption and product formation began only after 20 minutes of reaction.
Significant formation of cholesterol hydroperoxides (ChOOH) occurred only after 60
minutes, showing that cholesterol is not rapidly consumed in the reaction (**Figure 8B**).
These data suggest that cholesterol protects membranes against photooxidation primarily
through its membrane-stabilizing effects, which shield phospholipid acyl chains from
singlet oxygen-mediated oxidation.

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447 **Figure 8.** Consumption of sterols and oxidation products formed in the reaction with
448 POPC liposomes in the presence of DMMB. Values obtained by peak integration of the
449 detected products in UPLC-MS using the PeakView program. (A) Erg: ergosterol; EHP:
450 ergosterol hydroperoxide; EEP: ergosterol endoperoxide. (B) Chol: cholesterol; ChOOH:

451   cholesterol hydroperoxides

Photo-oxidized products of sterols show superior A375 cell viability reduction than their precursor molecules

457   Considering the EEP the most representative endoperoxide generated by $^1O_2$ (type II reaction) as shown previously, we compared the effect of this oxidation product with its precursors: in short, we compared the cell viability of Erg and 7-DHC, with their respective endoperoxides, EEP and CEP. We carried out the assay in four different conditions: (A) PDT, with DMMB and irradiation; (B) with DMMB and without irradiation to evaluate the dark toxicity of DMMB; (C) without DMMB and with irradiation, to evaluate the effect of irradiation; and (D) without DMMB nor irradiation, as a control.

464   **Figure 9** shows the significant reduction in cell viability of A375 cells treated with both endoperoxides, EEP and CEP, in comparison to   their precursor molecules, Erg and 7-DHC, respectively. We did not observe significant efficacy of dark toxicity (**Figure 9B**) or irradiation alone (**Figure 9C**). This experiment evidenced the higher toxicity of endoperoxides, compared to their non-oxidized precursors.

Figure 9. Cell viability effect in A375 with different sterols, ergosterol, ergosterol endoperoxide (EEP Erg), 7-DHC and 7-DHC endoperoxide (CEP): (A) PDT, with DMMB (0.1 nM) and irradiation (5 J/cm^2); (B) with DMMB (0.1 nM) and without irradiation to evaluate the dark toxicity of DMMB; (C) without DMMB and with irradiation (5 J/cm^2); to evaluate the effect of irradiation; and (D) without DMMB and irradiation, as a control

DISCUSSION

Oxidation products of ergosterol by singlet oxygen reaction

Our protocol of type II reaction ($^1O_2$) by DMMB was different compared to other studies. For instance, Böcking *et al.* (2000) (28) used toluene blue as a photosensitizer to generate $^1O_2$ in *S. cerevisiae* and this resulted in the yield of 4.0% DHE followed by 1.5% EEP. Ponce *et al.* (2002) (29) carried out several reactions with rose Bengal and eosin with three types of solvents (pyridine, ethanol, and MTBE), and EEP was the only product generated in all combinations. In their work, Erg + MTBE + eosin generated 16% EEP and, 8% EHP, with trace amounts (not quantifiable) DHE. According to the characterization data of our oxidation product by NMR and mass spectrometry, we

confirm that the products synthesized and purified in our work were the same as those found in the literature. This indicates that the main photooxidation products are the same regardless of solvent or photosensitizer type. As mentioned in both our results and in the literature, EEP was the most abundant product in type II Erg oxidation reaction (28–30). Thus, this molecule is a good marker for singlet oxygen reaction, since it is also a stable product at high temperatures (28, 31).

Membrane leakage study with carboxyfluorescein using DMMB as a photosensitizing agent

In this experiment, the purpose was to investigate membrane leakage caused by the photo-oxidation type II reaction ($^1O_2$) on phospholipid and sterol. Cholesterol-containing liposomes (POPC + Chol) showed lower membrane leakage over time compared to control, Erg- (POPC + Erg) or 7-DHC-liposomes (POPC + Chol) (**Figure 6A**). This result confirmed the different protection efficacy promoted by sterols, corroborating with data of Tsubone *et al.* (2019) and Bacellar *et al.* (2014) (32, 33), which showed that a liposome formed by DOPC with 30% cholesterol decreased permeability by up to 60% under photo-oxidative stress. Additionally, a study on molecular dynamics by Zhang *et al.* (2018) (34) reported the absence of pore formation in cholesterol-rich membranes, since cholesterol possibly formed hydrogen bonds with the oxidized lipids, making it difficult to generate an inclination in the interaction angles of the lipids. We can also observe the better membrane protection of POPC + Erg and POPC + 7-DHC, but less than the POPC + Chol. This is possibly due to the rapid oxidation of Erg and 7-DHC by type II reaction, leading to an inferior membrane stabilization compared to cholesterol.

After some time of reaction, however, all liposomes began to release their internal contents into the external environment. This leakage is possibly due to the formation of truncated phospholipid species bearing aldehyde and carboxylic acid groups from phospholipid hydroperoxides (**Figure 7B**). These truncated lipids favor the deformation of the bilayer and the formation of pores, increasing membrane permeabilization, and generating the release of internal content into the external environment (32).

Membrane leakage study with carboxyfluorescein using Fe/Asc as oxidizing agent

Fe/Asc was used as a membrane oxidant to verify the protective efficacy promoted by sterols in liposomes against damage induced by free radicals (type I reaction). Asc reduces Fe (III) into Fe (II). The formation of Fe (II)   initiates a Fenton-type reaction with peroxides present in liposomes producing alkoxyl radicals (27). Using 10 µM Fe/Asc, Erg, and 7-DHC provided greater membrane protection compared to control and cholesterol-membrane. However, after a certain period, membrane leakage occurred (**Figure 6B**). Thus, Erg and 7-DHC promoted greater protection of the membrane at the initial phase of lipid peroxidation, but at the propagation phase, the difference among sterols is no longer significant.

A study carried out by Freitas *et al.* (2024) (22) also noted that 7-DHC protects the membrane, as it reacts to produce two non-enzymatic 7-DHC oxidation products. This indicates the preference for 7-DHC to be oxidized in place of phospholipids, preventing the formation of truncated species. Although an in-depth study for Erg has not yet been published, the similarity of this molecule to 7-DHC (the presence of the conjugated double-bound present in the sterol B-ring) suggests that Erg may have the same membrane protection function as 7-DHC. Indeed, these authors validated this hypothesis in this same work, in yeast strains with targeted deficiencies of genes important for Erg biosynthesis.

Oxidation products of liposomes generated by DMMB photooxidation

Our results showed that the generation of phospholipid hydroperoxides (POPC-OOH) was higher in control (POPC) than in membranes containing Erg (POPC + Erg) or cholesterol (POPC + Chol) (**Figure 7B**). However, it is known that the formation of hydroperoxides, hydroxides, and ketones in the acyl chain of the fatty acid esterified to phospholipids does not cause membrane disruption. These products only modify the interactions between phospholipids, leaving the membrane larger and thinner. Instead of them, Bacellar *et al.* (2019) demonstrated that the products responsible for disrupting the membrane are those derived from truncated phospholipids such as POPC aldehyde

(AldoPC) or POPC carboxylic acid (PAze PC). These molecules have a different
conformation as compared to phospholipids with two chains. In truncated phospholipids,
one chain is shorter, presenting a more conical conformation, which may lead to
membrane leakage and release of the internal content to the external media (35).

Here, formation of AldoPC was not apparent in these liposomes analyzed by
UPLC-MS analysis. However, in control sample (POPC) it was possible to detect the
maximum formation of POPC-OOH between 20 and 40 minutes, with a subsequent
decline, indicating that POPC-OOH started to be consumed after this reaction time
(**Figure 7B**). The consumption of hydroperoxides in oxidation reactions is an indication
of radical reactions (type I oxidation), caused by the photosensitizer, which can generate
truncated lipids such as AldoPC and PAze PC (28). Although liposome with cholesterol
(POPC + Chol) formed more POPC-OOH than liposome with ergosterol (POPC + Erg)
(**Figure 7B**), there was no indication of consumption of this POPC-OOH in POPC + Chol
over the time (**Figure 8B**). This data differs from POPC + Erg, which showed a slight drop
in the amount of POPC-OOH after 60 minutes of reaction (**Figure 7B**), supporting a
potential initialization of truncated lipids formation. Thus, this difference in the generation
of oxidation products, such as hydroperoxides, truncated lipids, and their consumption,
illustrates the possible mechanism of faster membrane disruption of Erg-containing
liposomes than the cholesterol ones in type II reaction (**Figure 6A**).

Tsubone *et al.* (2019) suggested that cholesterol protection is related to the
formation of hydrogen bonds with oxidized lipids, making it difficult to change the angle
of inclination of the lipids that are interacting, thus maintaining the integrity of the
membrane (32). Other work mentioned that cholesterol can reduce the oxygen
permeability in the membrane, since there is a greater electron density on both sides of
the membrane bilayer when the amount of cholesterol is increased, making the packing
of cholesterol with the phospholipid greater, generating a physical barrier to the diffusion
of oxygen (36, 37). Our data showed slower consumption of cholesterol and formation of
oxidation product (ChOOH) in cholesterol-liposome (**Figures 8A**) than ergosterol in Erg-
liposome (**Figures 8B)**. Interestingly, even with distinct oxidation reactions and liposomes
composition, such as 1,2-dipalmitoyl-sn-glycero-3-phosphocholine (DPPC) and 1-
palmitoyl-2-oleoyl-sn-glycero-3 -phospho-(1'-rac- glycerol) (POPG) in different works (32,

34), they showed the same tendency which confirms the consistency of our results.

As a complement, an interesting work using plant extract also evidenced the

antioxidant and physical adsorption mechanisms of several plants to protect membrane,

measured via CF emission in DMMB-liposomes. Gallic acid and trehalose were used as

standard compounds, and the percentage of membrane protection was linearly

proportional to the concentration of these compounds, highlighting the several

mechanisms of compounds influencing the membrane integrity (38).

Cell viability effect of sterols and photodynamic therapy in the melanoma A375 cell

line

Our results showed significant reduction in cell viability of melanoma A375 cells

treated with sterol endoperoxides. EEP and CEP, the endoperoxides of Erg and 7-DHC,

respectively, showed greater efficacy in killing melanoma cells compared to their

precursors. Higher cytotoxicity of sterol endoperoxides was also reported in other cancer

cells. On A549 lung cancer cells, authors showed greater cell viability inhibition of EEP to

Erg and 9,11-dehydroergosterol endoperoxide. According to their work, EEP induced

caspase-dependent apoptosis via mitochondrial damage, cytochrome c release, and

ROS generation (39). Furthermore, the superior cytotoxicity of endoperoxides derived

from ergosterol and 7-DHC has been also evidenced in lung cancer A547, cervical

carcinoma cells HeLa, breast cancer cells SKOV-3, prostatic carcinoma cells DU145, and

human normal liver cells L-02 (40).

Other interesting work with 7-DHC demonstrated the influence of this sterol on

ferroptosis, an iron-dependent and non-apoptotic cell death induced by accumulation of

membrane-destabilizing truncated phospholipid species. These authors identified 7-DHC

as a suppressor of ferroptosis, due to its high reactivity towards peroxyl radicals, which

protects polyunsaturated fatty acids (PUFAs) from peroxidation. Although the unsaturated

B rings in 7-DHC make it a primary target for peroxidation, acting as a sacrificial

antioxidant, the oxidized 7-DHC metabolites have a lower membrane-destabilizing effect

compared to truncated species. This may provide a membrane-shielding effect that

reduces cell death. Membrane leakage tests showed superior membrane protection of 7-

DHC compared to control and cholesterol against radicals produced by the Fe/Asc system (22). This result is in line with our results of 7-DHC-membrane leakage using Fe/Asc (type I reaction) and carboxyfluorescein.

Another work detected the location of 7-DHC inside the cells, which was both in the mitochondria and the plasma membrane. Authors demonstrated the 7-DHC as suppressor of phospholipid peroxidation to reduce ferroptosis in both locations. They showed that Erg, the analogue of 7-DHC, also suppressed ferroptosis. By adding RSL3 (ferroptosis-inducing agent) to HEK293T cells pre-treated with 7-DHC, Erg, stigmasterol, or cholesterol, they showed that Erg and 7-DHC blocked phospholipid autooxidation and inhibited ferroptosis, unlike other sterols (21).

As the mechanism of ferroptosis relies on the high susceptibility of PUFAs in phospholipid membranes to lipid peroxidation, it leads to the accumulation of lipid peroxides and cell death. Nevertheless, it is known that lipid droplets (LDs), the lipid-rich cellular organelles, incorporate excess PUFAs within their neutral core, controlling PUFA storage in the form of triacylglycerols and sterol esters to reduce potential lipotoxic damage (41). It was shown that the formation of LDs induced by cell cycle arrest promoted resistance to ferroptosis. Authors showed that diacylglycerol acyltransferase (DGAT)-dependent LDs were formed under induction by cell cycle arrest. This LD sequestered excessive PUFAs accumulated in cells to suppress ferroptosis, and the inhibition of DGAT resensitized the cell arrest to ferroptosis (42). Moreover, as LDs disruption can provide the release of excessive lipids to generate increased lipid peroxidation, LDs can act in both prevention or induction of lipotoxicity and cell death, depending on the oxidative stress. Additional information is provided by other authors, indicating that the $^1O_2$ generated by PDT can induce both apoptosis and ferroptosis, according to the $^1O_2$ dose. They described that a moderate dose of singlet oxygen induces apoptosis, releasing cyt c from mitochondria. However, an excessive dose of $^1O_2$ inactivates cyt c and caspase enzymes, inhibiting this pathway and inducing ferroptosis by excess lipid peroxidation combined with iron release from lysosomal membrane disruption (43).

Interestingly, it is reported that Erg can also be stored in LD as sterol ester, and this would prevent lipid peroxidation that leads to ferroptosis (44). A study demonstrated that the mitochondrial concentration of Erg in *S. cerevisiae* was mainly located in the inner

mitochondrial membrane (IMM) than in the outer mitochondrial membrane (OMM), and
this sterol is actively transported into the OMM in an energy-independent way. Since
mitochondria communicate with LDs, endoplasmic reticulum (ER), and peroxisomes (45),
the Erg LD storage might be influenced by the correct function of mitochondria. In
contrast, a work evidenced that the endoperoxide EEP inhibited LD synthesis of
differentiated 3T3-L1 cells (46), indicating the opposite effects of Erg and EEP on LD
formation.

Concerning the role of mitochondria, it is well known that this organelle and
lysosome are the main target organelles for type II reaction ($^1O_2$) by PDT (35). Although
lipid peroxidation in the endoplasmic reticulum and plasma membrane highly influences
ferroptosis, mitochondria have also a contribution to this cell death process (47). It is
described that PDT with DMMB reduced mitochondrial membrane potential and promoted
increased reactive oxygen species production, autophagic dysfunction, and increased
membrane permeability. DMMB easily diffuses through the plasma membrane and, due
to its positive charge, accumulates in the mitochondria, an organelle with an inner
negative charge (48). It is reported that PDT with DMMB promoted loss of membrane
potential of mitochondria in the parasite *Leishmania amazonensis*, and released lipids
into the cytosol. Additionally, LDs increased substantially and accumulated large amounts
of these free lipids. This response of LDs to mitochondrial dysfunction regulates
lipotoxicity and preserves organelle function to prevent cell death. However, a redox
imbalance caused by lipid overload possibly led to constant cellular stress, and the
parasite's final death (49).

As a complement, a clinical test with PDT by intratumorally injection of 2%
methylene blue aqueous solution with 2% lidocaine, in a 92-years-old multiple skin
melanoma patient, showed a tumor remission in five out of six sites of tumors, highlighting
the potential of PDT as an effective and inexpensive treatment for melanoma (50).

All above studies could provide several insights into the cell death mechanism in
A375 cells in this work, as follows:

First, the membrane permeability property changes according to the type of
oxidative damage. In the type I reaction, superior membrane protection against leakage

was observed in Erg and 7-DHC than control and cholesterol in the initiation phase of

lipid peroxidation. This agrees with the results presented by Freitas *et al.* (2024) (22),

indicating the effectiveness of 5,7-unsaturated sterols against type I reaction in the initial

phase of lipid peroxidation. This is possibly due to the radical trapping ability (51) and

reduced formation of truncated lipids in the membranes when these sterols are introduced

(22). This result, however, differs from type II oxidation, where cholesterol protected the

membrane more than Erg and 7-DHC. This fact can be explained by the results shown

in **Figures 8A** and **8B** of our work, where there is a significant difference in sterol

consumption. Erg is almost immediately consumed and generates EEP or EHP, while

cholesterol consumption and generation of its main hydroperoxide (ChOOH) takes much

longer under type II photo-oxidation. As cited previously, cholesterol-rich membranes

reported the absence of pore formation possibly due to the hydrogen bonds formation

with the oxidized lipids (34).

Second, the protection of cells against ferroptosis in Li *et al.* (2024) and Freitas *et*

al. (2024) (21, 22) had a greater effect with 7-DHC than the controls. Our results also

showed a non-significant cell viability reduction in A375 cells treated with 7-DHC and Erg.

Here we can raise some hypotheses. The non-excessive dose of $^1O_2$ by PDT (type II

reaction) in our work led to the formation of a small amount of their respective

endoperoxides, not enough to kill the cells, but enough to protect the membranes from

truncated lipid formation. Another hypothesis might be the high amount of both sterols

that induced the formation of LDs, avoiding lipotoxicity and cell death, but still to be

confirmed.

Third, EEP and CEP promoted opposite effects to their precursor molecules, Erg

and 7-DHC. These endoperoxides led to cell death, while Erg and 7-DHC did not.

According to the work by Wu *et al.* (2018) (39) mentioned previously, EEP induces

caspase-dependent apoptosis via mitochondrial damage, cytochrome c release, and

ROS generation, and induces mitophagy at 20 µM. Interestingly, our work showed a

sudden decline of cell viability mostly after 20 µM EEP (**Figure 9**), which highlights the

need for additional studies to deepen this dose-dependent mechanism. Our results

may also be in line with the description of Fujii *et al.* (2023) (43), which suggests that

a moderate dose of $^1O_2$ mainly induces apoptosis rather than ferroptosis. An additional

explanation for the reduction of cell viability may be the lipotoxicity due to the excess

concentration of EEP. EEP suppressed the lipid droplet synthesis by inhibiting

triglyceride synthesis in differentiated 3T3-L1 cells (46). Another study with 24(S)-

hydroxycholesterol showed that its esterification resulted in the formation of atypical

lipid droplets due to the endoplasmic reticulum membrane disorder. This was caused

by the change of polarity of this sterol by esterification, disturbing membrane bilayer

orientation (52). Although they differ in structure, changes in molecular polarity

induced by enzymatic or non-enzymatic processes may lead to non-controlled

lipotoxicity and increased cell death. This information may inspire future investigations

concerning the influence of oxidation products on the rupture of LDs, to promote

lipotoxicity (49). Another explanation could be the generation of a second metabolite

of EEP and CEP under PDT. However, as shown in **Figure S6**, EEP is highly stable,

suggesting that the effect of PDT on the already-formed EEP may not be highly

significant. All these descriptions strongly highlight that these endoperoxides have

different efficacy compared to their precursor molecules, and dose and stress

conditions may be one of the key triggers to direct cell death pathways.

Although additional studies are still expected to confirm these explanations, the

originality of this work relies on the description of the step-by-step process to generate

and identify endoperoxide through the type II reaction, and it also provides new insights

to elucidate different mechanisms of cell death promoted by Erg, 7-DHC and their

respective endoperoxides, in combination with PDT in A375 melanoma cells.

CONCLUSION

Melanoma is one of the most aggressive oncological diseases that require effective

treatment. Ergosterol is a sterol from a natural source that has attracted attention for its

several benefits. 7-DHC, an analog of Erg present in the human membrane, has also

shown benefits in several diseases, including cancer. However, some ambiguity or

uncertainty remains regarding the mechanism of their protective or deleterious effects.

Focusing on the mechanistic point of view of the cell membrane lipid bilayer, we used the

liposome leakage model to compare the possible causes of membrane rupture in distinct

types of lipid oxidation reactions. We then evaluated the cell viability of A375 melanoma
treated with 7-DHC and Erg, and their respective endoperoxides generated by the type II
oxidative reaction.

The protocol carried out in this work confirmed the generation of Erg endoperoxide
and other sterols. The membrane leakage results demonstrated different behaviors
between liposomes composed of Erg, 7-DHC or cholesterol under distinct oxidation
reactions (type I and II). When correlating our results with the literature, the presence of
truncated lipids generated by oxidative stress appears to be the main cause of membrane
integrity disruption, compromising cell viability. In vitro viability testing of A375 melanoma
cells with PDT showed inhibition of cell viability in endoperoxide-treated cells and no
significant reduction in precursor molecules. These results in combination with data from
the literature pointed out some directions, such as the influence of these sterols and
derivatives combined with singlet oxygen on apoptosis, ferroptosis, mitochondria integrity
and formation of lipid droplets. The change in increased or decreased cell viability
appears to be highly dependent on the dose of oxidative stress and the concentration or
type of sterols.

Thus, in this work, unlike other studies, we explored the possible effects of Erg and
7-DHC oxidation derivatives generated mainly by the type II reaction and highlighted the
different effects of sterols and their endoperoxide derivatives on the cell viability of the
A375 melanoma cell line.

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CONFLICT OF INTEREST

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

SUPPLEMENTARY MATERIALS

See supplementary materials.

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