Influence of 8-oxoguanine on the thermal stability of G-quadruplex DNA and the interactions with the FANCJ AKKQ peptide

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Abstract

Guanine-rich nucleic acid sequences can adopt G-quadruplex (G4) structures, which pose barriers to DNA replication and repair. The FANCJ helicase contributes to genome stability by resolving these structures, a function linked to its G4-binding site that features an AKKQ amino acid motif. This site is thought to recognize oxidatively damaged G4, specifically those containing 8-oxoguanine (8oxoG) modifications. We hypothesize that FANCJ AKKQ recognition of 8oxoG-modified G4s (8oxoG4s) depends on the sequence context, the position of the lesion within the G4, and overall structural stability. Using fluorescence spectroscopy, we measured the binding affinities of a FANCJ AKKQ peptide for G4s formed by $(GGGT)_4$, $(GGGTT)_4$, and $(TTAGGG)_4$ sequences. G4 conformation and thermal stability were assessed by circular dichroism spectroscopy. Each sequence was modified to include a single 8oxoG at the first (8oxo1), third (8oxo3), or fifth (8oxo5) guanine position. In potassium chloride (KCl), the most destabilized structures were $(GGGT)_4$ 8oxo1, $(GGGTT)_4$ 8oxo1, and $(TTAGGG)_4$ 8oxo5. In sodium chloride (NaCl), the most destabilized were $(GGGT)_4$ 8oxo1, $(GGGTT)_4$ 8oxo5, and $(TTAGGG)_4$ 8oxo5. FANCJ AKKQ binding affinities varied according to damage position and sequence context, with notable differences for $(GGGT)_4$ in KCl and $(TTAGGG)_4$ in NaCl. These findings support a model in which FANCJ binding to G4 and

34 8oxoG4 structures is modulated by both the oxidative damage position and the G4 local

35 sequence environment.

Introduction

G-quadruplexes (G4s) are stable secondary structures that are formed by guanine-

rich nucleic-acids. G4 DNA structures arise from intramolecular Hoogsteen hydrogen

bonding within single-stranded (ss) DNA containing four tracts of three or more

consecutive guanines [1]. Monovalent cations further stabilize G4s by centrally

coordinating the guanine tetrads, with potassium (K^+) and sodium (Na^+) ions being the

most biologically relevant. Among them, potassium ions generally confer greater stability

to G4s [2-4].

The canonical sequence motif for G4 formation is $(GGGN_{1-7})_4$, where N denotes

any intervening nucleotides [5]. This has led to an estimate of over 400,000 potential G4

forming sequences being identified in the human genome [6]. G4s can adopt multiple

topologies—parallel, antiparallel, or hybrid—depending on the nucleotide sequence and

solution conditions (Fig 1) [4, 7]. These G4 conformations are defined by how the loop

regions (N_{1-7}) connect the guanine tracts [8, 9], and can be distinguished using circular

dichroism (CD) spectroscopy [10]. The spatial arrangements of G4s are biologically

significant, as many G4-binding proteins have a preference for specific topologies; the

Pif1 helicase, for example, binds and unfolds antiparallel G4s [11-15].

Fig 1. G-quadruplex formation and conformations. (A) Formation of a G4 structure. (B)
Graphic representation of three different intra-strand G4 conformations.

G4 structures are enriched in genomic regions involved in transcriptional

regulation, RNA processing, and telomere maintenance [5, 16-20]. While G4s may play

regulatory roles, their accumulation can be deleterious to cells, as they can impede

replication, repair, and transcription [16]. Consequently, G4s have emerged as attractive

therapeutic targets. The G4-stabilizing compound CX-5461 induces DNA breaks and

induces replication fork collapse, highlighting its potential in targeting tumor cells [21].

A key vulnerability of G4-forming regions is their susceptibility to oxidative stress.

Guanine has the lowest oxidation potential of the four DNA bases, and its propensity to

be oxidized increases when the guanines are stacked or arranged consecutive runs

[22,23]. Oxidation of guanine can produce 8-oxoguanine (8oxoG), a lesion that can

mispair with either adenine or cytosine, leading to G to T transversion mutations if left

unrepaired [24,25]. Although G4s containing 8oxoG (i.e. 8oxoG4s) can still form, the

lesion reduces their stability, rendering them susceptible to nuclease digestion and

potentially leading to telomere shortening [26,27]. Nuclear magnetic resonance (NMR)

studies have shown that 8oxoG incorporation into human telomeric G4s alters their

topology by changing the guanine glycosidic angles from anti to syn configuration [26].

Thus, G4s and 8oxoG4s must be resolved in dividing cells to maintain genomic integrity.

FANCJ (Fanconi anemia complementation group J) is one of several proteins that

can unfold G4s in human cells. As an iron-sulfur helicase, FANCJ participates in the

repair of interstrand DNA crosslinks as part of the Fanconi anemia pathway [28].

Mutations in FANCJ are associated with increased cancer risk and Fanconi anemia [28-

30]. FANCJ contributes to genome stability not only through DNA repair, but also by

resolving G4 structures that impede replication [31-35]. Upon encountering a G4-stalled

replication fork, FANCJ binds and unwinds the G4, allowing DNA replication to resume,

potentially through the recruitment of translesion polymerases such as REV1 [36] (Fig 2).

Fig 2. Molecular models of FANCJ. (A) FANCJ possesses a G4 recognition site capable of binding G4s, and an AKKQ motif similar in structure and sequence to the AKKQ binding motif found in G4-binding protein RHAU [11, 37]. (B) The current model for FANCJ-supported DNA replication in G4-rich sequences. Upon reaching a G4 structure, the DNA polymerase stalls. FANCJ binds to the G4 structure to unwind it and another polymerase is recruited to continue DNA replication.

Interestingly, a related RNA helicase RHAU (also known as DHX36), recognizes

G4 structures via an AKKQ motif, where the tandem lysine residues facility G4-binding

[11, 37]. FANCJ contains a similar G4 binding site with an AKKQ motif centered at lysines

K141 and K142 [33]. This motif enables FANCJ to anchor to both G4 and 8oxoG4 DNA as

shown in our previous work (Fig 3) [36]. In this study, we further characterize the

sequence- and damage-dependent recognition of G4 and 8oxoG4 by the FANCJ AKKQ

motif by exploring how intervening loop length and guanine oxidation at distinct

positions affect FANCJ AKKQ peptide binding and G4 stability.

98 **Fig 3. Map of FANCJ**. The AKKQ sequence (amino acids 129-147) has been highlighted.

Results

G-quadruplex conformation and thermal stability

Circular dichroism (CD) spectroscopy was used to assess the conformation and

thermal stability of twelve G4 sequences in buffers containing either KCl and NaCl. G4

structures exhibit distinct CD signatures based on their topology, enabling discrimination

between parallel, antiparallel, and hybrid configurations (Fig 4A) [9, 10]. CD experiments

were performed as a function of temperature to derive the T_m (melting temperature),

which reflects the thermal stability of each G4 substrate (Fig 4B and 4C) [38].

Fig 4. CD spectroscopy cartoons. (A) Graphic representation of CD spectra for parallel, antiparallel, and hybrid G4 conformations. Parallel conformations have a maximum at 262 nm and a minimum at 240 nm. Antiparallel conformations give two peak signals near 295 nm and 240 nm, and a minimum at 260 nm. Hybrid conformations show a peak at 295 nm, a shoulder around 260 nm, and a minimum at 240 nm. (B) Graphic representation of the CD spectra of a parallel G4 conformation as a function of temperature. (C) Graphic representation of a thermal melt graph of two G4s.

Three G4-forming sequences were selected based on their biological relevance and loop length: (GGGT)$_4$, a compact parallel G4-forming aptamer for HIV-1 integrase that possesses a single thymine base for the ssDNA loop [7, 39]; (TTAGGG)$_4$, is the canonical human telomeric repeat sequence that has a TTA loop between the guanine tracts [19]; (GGGTT)$_4$, an artificial sequence with an intermediate loop length (TT) compared to the other two substrates. Oxidative damage was introduced by substate an 8oxoG at the first (8oxo1), third (8oxo3), or fifth (8oxo5) guanine position of each sequence (Fig 5), thereby targeting each of the three tetrad stacks.

Fig 5. Graphic representation of a parallel G4 conformation. An 8oxoG-modified base has been substituted at the (A) first guanine (8oxo1), (B) third guanine (8oxo3), and (C) fifth guanine (8oxo5).

In 150 mM KCl, all (GGGT)$_4$ sequences formed a parallel quadruplex (Fig 6A), with only subtle changes to the CD spectra when 8oxoG was incorporated. Due to temperature limitations of the instrument, the native sequence was not fully unfolded even at 95 °C, and its T$_m$ was estimated to be > 88.4 ± 8.4 °C (Table 1), consistent with prior reports [7,36]. All 8oxoG substitutions led to reduced thermal stability, with the largest destabilization observed for (GGGT)$_4$ 8oxo1 (ΔT$_m$ = −19.0 ± 8.4 °C) and the smallest for 8oxo3 (ΔT$_m$ = −8.1 ± 8.4 °C).

Table 1. Summary of G4 conformation and melting temperature in 150 mM KCl

Name	Conformation	T$_m$ (°C)	Δ T$_m$ (°C)
(GGGT)$_4$	Parallel	> 88.4 ± 8.4	-
(GGGT)$_4$ 8oxo1	Parallel	69.4 ± 0.6	-19.0 ± 8.4

(GGGT)$_4$ 8oxo3	Parallel	80.3 ± 0.4	-8.1 ± 8.4
(GGGT)$_4$ 8oxo5	Parallel	71.6 ± 0.4	-16.8 ± 8.4
(GGGTT)$_4$	Parallel	83.5 ± 0.7	-
(GGGTT)$_4$ 8oxo1	Parallel	52.0 ± 1.4	-31.5 ± 1.6
(GGGTT)$_4$ 8oxo3	Antiparallel	57.2 ± 0.3	-26.3 ± 0.7
(GGGTT)$_4$ 8oxo5	Parallel	55.9 ± 0.3	-27.6 ± 0.8
(TTAGGG)$_4$	Hybrid	62.5 ± 0.5	-
(TTAGGG)$_4$ 8oxo1	Hybrid-esque	54.5 ± 0.5	-8.0 ± 0.7
(TTAGGG)$_4$ 8oxo3	Antiparallel	49.5 ± 0.3	-13.0 ± 0.6
(TTAGGG)$_4$ 8oxo5	Hybrid-esque	38.9 ± 0.6	-23.6 ± 0.8

Fig 6. Experiments performed in the presence of 150 mM KCl. (A) CD spectra of (GGGT)$_4$, (GGGT)$_4$ 8oxo1, (GGGT)$_4$ 8oxo3, and (GGGT)$_4$ 8oxo5 taken from 320 to 220 nm. (B) Temperature melt of same sequences seen in 6A. Melt taken at 262 nm from 15 to 95 °C. (C) CD spectra of (GGGTT)$_4$, (GGGTT)$_4$ 8oxo1, (GGGTT)$_4$ 8oxo3, and (GGGTT)$_4$ 8oxo5 taken from 320 to 220 nm. (D) Temperature melt of same sequences seen in 6C. Melt taken at 262 nm for all sequences except (GGGTT)$_4$ 8oxo3, which was taken at 295 nm. (E) CD spectra of (TTAGGG)$_4$, (TTAGGG)$_4$ 8oxo1, (TTAGGG)$_4$ 8oxo3, and (TTAGGG)$_4$ 8oxo5 taken from 320 to 220 nm. (F) Temperature melt of the same sequences seen in 6E. Melt taken at 295 nm.

(GGGTT)$_4$ also adopted a parallel conformation in KCl (Fig 6C). While 8oxo1 and

8oxo5 substitutions retained a predominantly parallel fold with minor spectral shifts, the

8oxo3 variant formed an antiparallel quadruplex. The native (GGGTT)$_4$ had a T$_m$ of

83.5 ± 0.7 °C. 8oxoG destabilized the structure in a similar trend, with 8oxo1 producing

the largest drop (ΔT$_m$ = −31.5 ± 1.6 °C) and 8oxo3 the least (ΔT$_m$ = −26.3 ± 0.7 °C).

(TTAGGG)$_4$ folded as a hybrid G4 in KCl (Fig 6E). (TTAGGG)$_4$ 8oxo1 and

(TTAGGG)$_4$ 8oxo5 sequences also formed hybrid conformations, although with

significant structural changes. (TTAGGG)$_4$ 8oxo1 had a significant right shift in the 295

nm peak and a double trough at 260 and 240 nm. (TTAGGG)$_4$ 8oxo5 experienced

depression in its peak at 295 nm and an increase in its shoulder at 260 nm and trough at

240 nm. Similar to (GGGTT)$_4$ 8oxo3, the (TTAGGG)$_4$ 8oxo3 sequence changed to the

antiparallel conformation. (TTAGGG)$_4$ had a T$_m$ of 62.5 ± 0.5 °C (Fig 6F; Table 1).

Consistent with a previous study, (TTAGGG)$_4$ 8oxo5 had the highest decrease in thermal

stability of ΔT$_m$ = -23.6 ± 0.8 °C while (TTAGGG)$_4$ 8oxo1 had the lowest change in thermal

stability of ΔT$_m$ = -8.0 ± 0.7 °C [27].

In NaCl-containing buffer, (GGGT)$_4$ maintained a parallel configuration across all

damaged variants (Fig 7A). The native sequence had a T$_m$ of 70.3 ± 0.7 °C. Substitution

with 8oxoG again decreased thermal stability, most notably at the 8oxo1 position

(ΔT$_m$ = −27.8 ± 0.9 °C), with 8oxo3 being least affected (ΔT$_m$ = −13.9 ± 0.8 °C) (Table 2).

Table 2. Summary of G4 conformation and melting temperature in 150 mM NaCl

Name	Conformation	T_m (°C)	ΔT_m (°C)
(GGGT)$_4$	Parallel	70.3 ± 0.7	-
(GGGT)$_4$ 8oxo1	Parallel	42.5 ± 0.5	-27.8 ± 0.9
(GGGT)$_4$ 8oxo3	Parallel	56.4 ± 0.4	-13.9 ± 0.8
(GGGT)$_4$ 8oxo5	Parallel	44.0 ± 0.3	-26.3 ± 0.8
(GGGTT)$_4$	Parallel	44.8 ± 1.2	-
(GGGTT)$_4$ 8oxo1	Antiparallel	47.2 ± 0.3	2.4 ± 1.2
(GGGTT)$_4$ 8oxo3	Antiparallel	44.6 ± 0.2	-0.2 ± 1.2
(GGGTT)$_4$ 8oxo5	Hybrid-esque	27.5 ± 0.9	-17.3 ± 1.4
(TTAGGG)$_4$	Antiparallel	52.1 ± 0.3	-
(TTAGGG)$_4$ 8oxo1	Antiparallel	47.8 ± 0.6	-4.3 ± 0.7
(TTAGGG)$_4$ 8oxo3	Antiparallel	42.2 ± 0.7	-9.9 ± 0.8
(TTAGGG)$_4$ 8oxo5	Antiparallel	33.3 ± 0.4	-18.8 ± 0.5

Fig 7. Experiments performed in the presence of 150 mM NaCl. (A) CD spectra of (GGGT)₄, (GGGT)₄ 8oxo1, (GGGT)₄ 8oxo3, and (GGGT)₄ 8oxo5 taken from 320 to 220 nm. (B) Temperature melt of same sequences as seen in 7A. Melt taken at 262 nm from 15 to 95 °C. (C) CD spectra of (GGGTT)₄, (GGGTT)₄ 8oxo1, (GGGTT)₄ 8oxo3, and (GGGTT)₄ 8oxo5 taken from 320 to 220 nm. (D) Temperature melt of same sequences as seen in 7C. Melt taken at 262 nm for all sequences except (GGGTT)₄ 8oxo3, which was taken at 295 nm. (E) CD spectra of (TTAGGG)₄, (TTAGGG)₄ 8oxo1, (TTAGGG)₄ 8oxo3, and

175   (TTAGGG)$_4$ 8oxo5 taken from 320 to 220 nm. (F) Temperature melt of the same sequences
176   as seen in 7E. Melt taken at 295 nm.
177

(GGGTT)$_4$ retained a parallel conformation in NaCl, but all three 8oxoG4 variants

underwent structural transitions: 8oxo1 and 8oxo3 shifted to antiparallel conformations,

and 8oxo5 adopted a hybrid-like structure (Fig 7C). Surprisingly, 8oxo1 showed a slight

increase in thermal stability ($\Delta T_m = +2.4 \pm 1.2\,°C$), while 8oxo3 had a negligible effect

($\Delta T_m = -0.2 \pm 1.2\,°C$). In contrast, 8oxo5 was the most destabilizing ($\Delta T_m = -17.3 \pm 1.4\,°C$).

The 8oxo5 melting curve was acquired from $-5\,°C$ to $75\,°C$ due to incomplete folding at

$15\,°C$.

(TTAGGG)$_4$ and its 8oxoG4 variants formed antiparallel conformations in NaCl

(Fig 7E). The damaged substrates all showed slight peak shifts in the CD spectra

compared to the native sequence, indicating slight folding differences. The T_m of

(TTAGGG)$_4$ was determined to be 52.1 ± 0.3 °C (Fig 7F; Table 2). Concurrent with

previous literature and the KCl data above, (TTAGGG)$_4$ 8oxo5 had the highest decrease

in thermal stability of $\Delta T_m = -18.8 \pm 0.5$ °C while (TTAGGG)$_4$ 8oxo1 had the lowest of ΔT_m

$= -4.3 \pm 0.7$ °C [27].

FANCJ AKKQ binding to 8oxoG4s

To assess the binding of the FANCJ AKKQ motif to G4 and 8oxoG4 structures,

fluorescence titrations were performed using a synthetic peptide with a C-terminal

tryptophan (Fig 8A). Tryptophan florescence intensity was quenched upon binding to the

DNA substrate, and the extent of quenching was plotted against G4 concentration to

produce binding isotherms (Fig 8B). The curves were well-described by a 1:1 binding

model (Fig 8C). In 150 mM KCl, the FANCJ AKKQ peptide showed preferential binding

to oxidatively damaged $(GGGT)_4$ sequences (Fig 9; Table 3), with the strongest interaction

to 8oxo1 ($K = 7.7 \pm 0.7$ x 10^5 M^{-1}), followed by 8oxo5 ($K = 4.4 \pm 0.6$ x 10^5 M^{-1}), 8oxo3 ($K = 3.2$

± 0.1 x 10^5 M^{-1}), and weakest to the undamaged native sequence ($K = 2.7 \pm 0.4$ x 10^5 M^{-1}).

Because the $(GGGT)_4$ substrate contained only a single thymine nucleotide loop, we next

examined the AKKQ peptide binding to G4 structures formed by $(GGGTT)_4$ to test the

effects of loop length. All 8oxoG4 variants of this substrate bound to the peptide with

similar affinity values ($K \approx 8$–9×10^5 M^{-1}), which were stronger than its interaction with

the undamaged sequence ($K = 6.0 \pm 0.9$ x 10^5 M^{-1}). For the human telomeric G4

$(TTAGGG)_4$, this substrate contained an incrementally longer TTA loop; the peptide

bound with comparable affinity to all variants of this sequence, including the native

structure ($K \approx 1.3$–1.4×10^6 M^{-1}), suggesting a minimal influence of oxidative damage to

binding.

Table 3. Summary of FANCJ AKKQ-G4 binding by fluorescence spectroscopy in 150

mM KCl

Name	K (M^{-1})	R^2
$(GGGT)_4$	2.7 ± 0.4 x 10^5	0.997

(GGGT)$_4$ 8oxo1	$7.7 \pm 0.7 \times 10^5$	0.998
(GGGT)$_4$ 8oxo3	$3.2 \pm 0.1 \times 10^5$	0.997
(GGGT)$_4$ 8oxo5	$4.4 \pm 0.6 \times 10^5$	0.997
(GGGTT)$_4$	$6.0 \pm 0.9 \times 10^5$	0.997
(GGGTT)$_4$ 8oxo1	$8.2 \pm 1.0 \times 10^5$	0.997
(GGGTT)$_4$ 8oxo3	$9.6 \pm 1.0 \times 10^5$	0.996
(GGGTT)$_4$ 8oxo5	$9.3 \pm 0.8 \times 10^5$	0.997
(TTAGGG)$_4$	$1.6 \pm 0.7 \times 10^6$	0.996
(TTAGGG)$_4$ 8oxo1	$1.3 \pm 0.4 \times 10^6$	0.997
(TTAGGG)$_4$ 8oxo3	$1.2 \pm 0.1 \times 10^6$	0.996
(TTAGGG)$_4$ 8oxo5	$1.4 \pm 0.5 \times 10^6$	0.997

Fig 8. Overview of fluorescence spectroscopy analyses. (A) Molecular model of fluorescence spectroscopy titrations. AKKQ is equipped with a tryptophan residue that fluoresces. When bound to DNA, the residue is quenched and fluorescence intensity decreases. (B) Titration of AKKQ with $(GGGT)_4$. As DNA concentration increases, the peak intensity decreases. (C) Analysis of fluorescence titration. The decrease in integrated peak is plotted as a function of DNA concentration. The resulting isotherm is fit to a 1:1 binding interaction model to generate an equilibrium K value.

Fig 9. Summary of FANCJ AKKQ-G4 binding in KCl by fluorescence spectroscopy. Trials performed in 20 mM HEPES pH 7.5, 150 mM KCl, 5 mM TCEP, and 5% (v/v) glycerol. (A) FANCJ AKKQ equilibrium binding to $(GGGT)_4$ and its respective 8oxoG4 sequences. (B) FANCJ AKKQ binding to $(GGGTT)_4$ and its respective 8oxoG4 sequences. (C) FANCJ AKKQ binding to $(TTAGGG)_4$ and its respective 8oxoG4 sequences.

In NaCl, a similar trend was observed for the $(GGGT)_4$ series (Fig 10; Table 4). The AKKQ peptide bound identically and strongest to 8oxo1 (3.4 ± 0.5 x 10^5 M^{-1}) and 8oxo5 (3.4 ± 0.5 x 10^5 M^{-1}), followed by 8oxo3 (2.9 ± 0.5 x 10^5 M^{-1}) and the native sequence (2.3 ± 0.3 x 10^5 M^{-1}). For $(GGGTT)_4$, all sequences exhibited similar binding affinities ($K \approx 4.9$–5.6×10^5 M^{-1}), except for 8oxo1, which showed slightly weaker binding ($K = 3.9 \pm 1.3 \times 10^5$ M^{-1}). For the human telomeric $(TTAGGG)_4$, the highest affinity was observed for 8oxo5 ($K = 1.1 \pm 0.1 \times 10^6$ M^{-1}), followed by the native and 8oxo1 sequences ($K = 6.5 \times 10^5$ M^{-1}), with 8oxo3 binding the weakest ($K = 5.1 \pm 1.1 \times 10^5$ M^{-1}).

Table 4. Summary of FANCJ AKKQ-G4 binding by fluorescence spectroscopy in 150 mM NaCl

Name	K (M^{-1})	R^2
$(GGGT)_4$	2.3 ± 0.3 x 10^5	0.9997
$(GGGT)_4$ 8oxo1	3.4 ± 0.5 x 10^5	0.9996
$(GGGT)_4$ 8oxo3	2.9 ± 0.5 x 10^5	0.9994
$(GGGT)_4$ 8oxo5	3.4 ± 0.5 x 10^5	0.9997
$(GGGTT)_4$	4.9 ± 2.0 x 10^5	0.9995
$(GGGTT)_4$ 8oxo1	3.9 ± 1.3 x 10^5	0.9993

(GGGTT)₄ 8oxo3	$5.2 \pm 1.7 \times 10^5$	0.9994
(GGGTT)₄ 8oxo5	$5.6 \pm 1.1 \times 10^5$	0.9992
(TTAGGG)₄	$6.5 \pm 0.6 \times 10^5$	0.9991
(TTAGGG)₄ 8oxo1	$6.5 \pm 1.4 \times 10^5$	0.9996
(TTAGGG)₄ 8oxo3	$5.1 \pm 1.1 \times 10^5$	0.9998
(TTAGGG)₄ 8oxo5	$1.1 \pm 0.1 \times 10^6$	0.9974

Fig 10. Summary of FANCJ AKKQ-G4 binding in NaCl by fluorescence spectroscopy.
Trials performed in 20 mM HEPES pH 7.5, 150 mM NaCl, 5 mM TCEP, and 5% (v/v)
glycerol. (A) FANCJ AKKQ binding to (GGGT)₄ and respective 8oxoG4 sequences. (B)
FANCJ AKKQ binding to (GGGTT)₄ and respective 8oxoG4 sequences. (C) FANCJ AKKQ
binding to (TTAGGG)₄ and respective 8oxoG4 sequences.

Discussion

G4 conformation and stability

Our CD data confirm and expand upon prior studies demonstrating that the human telomeric sequence $(TTAGGG)_4$ adopts hybrid and antiparallel conformations in KCl and NaCl, respectively, and that 8oxoG modifications influence both structure and stability in a position-dependent manner [26,27,41]. As seen by Vorlíčková et al, the 8oxo5 substitution produced the greatest thermal destabilization in both KCl and NaCl environments, whereas 8oxo1 was the least disruptive [27].

This positional effect was also evident in the $(GGGTT)_4$ series in NaCl, where the native sequence maintained a parallel conformation while the 8oxoG-modified variants adopted antiparallel or hybrid structures. In contrast, the $(GGGT)_4$ sequences maintained parallel topology across all variants and salt conditions, with 8oxo1 substitutions consistently producing the greatest destabilization and 8oxo3 the least. A similar pattern was observed for $(GGGTT)_4$ in KCl, where only the 8oxo3 variant deviated from the parallel configuration.

Collectively, these findings suggest that the thermal stability of G4 structures in the presence of 8oxoG depends on both the specific guanine position modified and the overall structural context of the G4. In compact, parallel G4s, damage to the first guanine (in the bottom tetrad) was most destabilizing, while damage to the third guanine (top tetrad) had the least effect. Conversely, in more flexible hybrid or antiparallel G4s, the

middle tetrad (fifth guanine) was most susceptible to destabilization, while the first

guanine remained relatively unaffected. This aligns with earlier NMR-based studies

showing significant destabilization and conformational change upon 8oxoG substitution

at the fifth guanine in human telomeric G4s [26]. Our data also reinforce the known effects

of monovalent cations on G4 stability. Overall, thermal stability was higher in KCl than

in NaCl, consistent with the greater stabilizing effect of potassium ions due to their

optimal ionic radius for coordinating G-quartets [2–4,42]. Additionally, a trend emerged

linking loop length with thermal stability in which shorter loop sequences exhibited

higher melting temperatures than longer loops, consistent with previous observations

[4,40]. A similar trend was also present but less evident in NaCl buffer, with $(GGGT)_4$ and

$(GGGTT)_4$ having the highest and lowest T_m values, respectively.

FANCJ AKKQ binding to 8oxoG4s

Our fluorescence titration data indicate that the FANCJ AKKQ motif can recognize

both native and 8oxoG-modified G4 structures. The binding selectivity was influence by

DNA sequence, structural conformation, and position of the 8oxoG.

For the $(GGGT)_4$ series, which formed highly stable parallel G4s, AKKQ showed

the strongest affinity for 8oxoG-modified sequences—particularly 8oxo1—mirroring the

pattern of thermal destabilization. This suggests that the peptide bound preferentially to

the damaged base, possibly recognizing features specific to the damaged tetrad rather

than to the single nucleotide loop. Since CD spectra for these sequences showed minimal

changes in global topology, direct recognition of 8oxoG may drive this binding

preference.

In contrast, the AKKQ peptide bound similarly to all 8oxoG variants of the

(GGGTT)$_4$ sequence, which retained parallel conformations for most of the modifications.

The CD spectra revealed structural perturbations and decreased thermal stability of

8oxo1 and 8oxo5 (Fig 6C), but this had no influence on AKKQ binding. We attribute this

to the peptide having direct access to the intervening sequencing, supporting a ssDNA

loop targeting mechanism rather than direct 8oxoG recognition.

The telomeric G4 (TTAGGG)$_4$ was more conformationally dynamic, and again the

AKKQ peptide bound with the same affinity to both damaged and undamaged forms.

This is again consistent with the TTA ssDNA loops being more accessible to the peptide

in the hybrid and antiparallel configurations, allowing the FANCJ peptide to bind

directly to the loop rather than to the 8oxoG. Notably, in KCl, AKKQ binding to

(TTAGGG)$_4$ was approximately an order of magnitude stronger than for the other

sequences, but this distinction diminished in NaCl, except for the 8oxo5 variant, which

showed enhanced binding.

One possible explanation for the observed differences in binding preferences is

that high sodium concentrations affect the electrostatic interactions between the peptide

and the G4s. Sodium ions are known to provide less stabilization to G4s compared to

potassium [2-4]. We anticipated the FANCJ AKKQ peptide to adopt an alpha helical

structure due to previous NMR studies of the structurally and sequentially similar AKKQ

motif of the G4-binding protein RHAU [11]. Although our CD studies (Supporting

information) did not detect any alpha helical structure in either salt, the addition of

trifluoroethanol revealed a helical-like structure in both salt conditions. However, the

spectra obtained in each salt were not superimposable. This suggests that sodium ions

stabilize the peptide to a different extent than potassium ions.

314   Interestingly, the affinities of the AKKQ peptide for G4 DNA increased with loop

315   length, further supporting direct interactions with the ssDNA loop. Although FANCJ is

316   a helicase involved in DNA repair, our results suggest that the AKKQ binding site may

317   not function as a direct damage sensor but rather as a loop- or structure-selective binding

318   element. However, when the loop is too short to be recognized, FANCJ is able to target

319   the 8oxoG damage preferentially (Fig 11).

A.

(GGGT)₄

B.

(TTAGGG)₄

Fig 11. Proposed binding interaction model between FANCJ AKKQ and 8oxoG4 sequences. (A) For G4 sequences that form compact structures with short intervening sequences, such as $(GGGT)_4$, AKKQ preferentially targets the 8oxoG site and promotes repair. (B) For G4 sequences that expose the intervening sequence, as with antiparallel or hybrid conformations, AKKQ preferentially targets the intervening sequence and ignores the 8oxoG.

Biological implications and future directions

FANCJ is thought to coordinate with the translesion synthesis polymerase REV1 during replication of G4-rich regions [43]. It is plausible that FANCJ and REV1 perform complementary roles, with FANCJ targeting conformationally dynamic G4s with exposed loops, while REV1 preferentially targets the more compact and stable structures. Exploring the G4-binding preferences and structural requirements of REV1 will be essential to delineate their specialized activities during G4 remodeling.

This study examined 8oxoG substitutions at only three guanine positions. Future studies should evaluate a broader set of damage positions, including systematic substitutions across all tetrads, to determine whether the observed positional effects on stability and recognition can be generalized. Additionally, incorporating full-length FANCJ helicase for the binding studies instead of just the AKKQ G4-targeting site would

339 provide insights into whether other FANCJ domains participate in the recognition of

340 8oxoG4s.

Materials and methods

Buffers and reagents

344   All buffers were prepared with reagent-grade chemicals and Type I ultrapure

345 water from a Smart2Pure 6 UV/UF system (ThermoFisher, Waltham, MA). Solutions were

346 sterilized by filtration through 0.22 μm PES membranes prior to use.

Peptide and DNA oligos

FANCJ AKKQ peptide (129-PEKTTLAAKLSAKKQASIW-147) was synthesized

by Genscript (Piscataway, NJ), and the lyophilized powder was dissolved in 20 mM

HEPES pH 7.5, 150 mM KCl or NaCl, 5 mM TCEP, and 5% (v/v) glycerol. Peptide

concentration was determined using a NanoDrop One UV-Vis microvolume

spectrophotometer (ThermoFisher), based on the calculated molar extinction coefficients.

Peptides were aliquoted, frozen, and stored at -20 °C.

Oligodeoxyribonucleotides were purchased from Integrated DNA Technologies

(IDT, Coralville, IA). DNA concentrations were similarly measured by UV absorbance.

All oligonucleotides were stored at 4 °C. A list of the G4 DNA sequences used is provided

below in Table 5.

358   **Table 5. DNA Sequences.**

Name	Sequence (5′→3′)	Molar Extinction Coefficient ($M^{-1}cm^{-1}$)	Molecular Weight (g/mol)
(GGGT)4	GGGTGGGTGGGTGGGT	157200	5105.3
(GGGT)4 8oxo1	T/i8oxodG/GGTGGGTGGGTGGGT	159100	5425.5
(GGGT)4 8oxo3	GG/i8oxodG/TGGGTGGGTGGGT	151600	5121.3
(GGGT)4 8oxo5	GGGTG/i8oxodG/GTGGGTGGGT	151600	5121.3
(GGGTT)4	GGGTTGGGTTGGGTTGGGTT	189600	6322.1
(GGGTT)4 8oxo1	/i8oxodG/GGTTGGGTTGGGTTGGGTT	184000	6338.1
(GGGTT)4 8oxo3	GG/i8oxodG/TTGGGTGGGTTGGGTT	184000	6338.1

(GGGTT)4			
8oxo5	GGGTTG/i8oxodG/GTTGGGTTGGGTT	184000	6338.1
(TTAGGG)4	TTAGGGTTAGGGTTAGGGTTAGGG	244600	7575.0
(TTAGGG)4 8oxo1	TTA/i8oxodG/GGTTAGGGTTAGGGTTA GGG	239000	7590.9
(TTAGGG)4 8oxo3	TTAGG/i8oxodG/TTAGGGTTAGGGTTA GGG	239000	7590.9
(TTAGGG)4 8oxo5	TTAGGGTTAG/i8oxodG/GTTAGGGTTA GGG	239000	7590.9

CD spectroscopy

CD measurements were performed using a JASCO J-815 spectropolarimeter equipped with a PTC-423S Peltier temperature controller and a Koolance liquid cooling system. G4 samples were dialyzed into 20 mM boric acid containing either 150 mM KCl or NaCl. CD spectra were collected from 220 nm to 320 nm at 25 °C using 0.3 mg/mL DNA. Five traces per sample were averaged, and buffer baselines were subtracted.

For thermal denaturation experiments, samples were pre-heated to 95 °C and allowed to cool slowly overnight to 25 °C to ensure proper folding. Melting curves were recorded from 15 °C to 95 °C by monitoring the ellipticity at 262 nm (for parallel

368   conformations) or 295 nm (for antiparallel or hybrid structures). Three replicates were

369   performed for each experimental condition and the resulting data were averaged.

370   Melting temperatures (T_m) were determined in SigmaPlot using a two-state

371   transition model described by Greenfield [44], based on the Gibbs-Helmholtz equation:

$$\Delta G = \Delta H \left(1 - \frac{T}{T_m} \right) - \Delta C_p \left((T_m - T) + T \left(\ln \left(\frac{T}{T_m} \right) \right) \right)$$

$$K = e^{\left(-\frac{\Delta G}{RT} \right)}$$

$$y = \frac{K}{1 + K}$$

$$[\theta]_t = \left(([\theta]_F - [\theta]_U) * y \right) + [\theta]_U$$

where ΔG represents Gibbs free energy, ΔH represents enthalpy, ΔC_P represents the

change in heat capacity, K represents the folding constant, and y represents the fraction

of folded DNA at any given temperature, T. θ_t is the observed ellipticity at any

temperature, θ_F is the observed ellipticity of the folded DNA, and θ_U is the observed

ellipticity of the unfolded DNA. The melting temperature, T_m, of the DNA can then be

calculated as the temperature where y = 0.5.

Fluorescence spectroscopy

Fluorescence titration experiments were performed using a Cary Eclipse

Fluorescence Spectrophotometer (Agilent Technologies, Santa Clara, CA) equipped with

a PCB 1500 Peltier-controlled cuvette holder. Samples were exited at 280 nm, and the

emission spectra were collected from 300 to 450 nm at 25 °C.

A scan of buffer alone, 20 mM HEPES pH 7.5, 150 mM KCl or NaCl, 5 mM TCEP,

and 5% (v/v) glycerol, was used as a baseline. FANCJ AKKQ peptide (5 µM) was titrated

with additions of a DNA substrate. The mixture was allowed to equilibrate for 3 minutes

before taking each spectra scan.

Observed fluorescence quenching (ΔF) was calculated using the following

equation, with F_0 representing total fluorescence from the FANCJ peptide alone and F_i

representing the total fluorescence after the ith addition of DNA:

$$\Delta F = \frac{F_0 - F_i}{F_0}$$

Binding isotherms were calculated by plotting ΔF versus total DNA concentration.

The resulting curve was fit to a 1:1 interaction model using Scientist 2.0 software

(Micromath, St. Louis, MO) to calculate an equilibrium K. The equilibrium K was

calculated using the following formulas, where A was the amplitude of fluorescence

quenching and K was the equilibrium association constant. D_f and D_t were the free and

total concentration of DNA, while P_f and P_t described the free and total peptide

concentration. Fluorescence titrations were performed in triplicate. The reported K values

were determined from an average of three independent data-sets.

$$\Delta F_{obs} = A \left(\frac{K D_f}{1 + K D_f} \right)$$
$$D_t = D_f (1 + K P_f)$$
$$P_t = P_f (1 + K D_f)$$

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Author contributions

CGW designed experiments. LC, KL, and CGW acquired data and interpreted results.

LC, KL, EC, and CGW prepared the manuscript draft. All authors have read and agreed

to the published version of the manuscript.

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Competing interests

Authors have no conflict of interest to disclose.

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Supporting information

S1 Fig. Experiments performed in the presence and absence of 20% (v/v) trifluoroethanol. (A) CD spectra of FANCJ AKKQ taken from 260 to 200 nm in the presence (blue) and absence (red) of trifluoroethanol in 150 mM KCl. (B) CD Spectra of FANCJ AKKQ taken from 260 to 200 nm in the presence (black) and absence (gray) of trifluoroethanol in 150 mM NaCl. Upon the addition of trifluoroethanol in either salt condition, an alpha helical conformation appears.