Sustained Isobutene Production by *Synechocystis* sp. PCC 6803 Entrapped in Polyvinyl Alcohol Hydrogel Beads

Sindhujaa Vajravel, Sanjukta Aravind, Karin Stensjö*

Microbial Chemistry, Department of Chemistry-Ångström Laboratory, Uppsala University,

SE-751 20, Uppsala, Sweden

* Corresponding author E-mail address: karin.stensjo@kemi.uu.se

Highlights:

- Optimized light, carbon, and nitrogen sources enhanced isobutene production.
- Rate of isobutene production increased 10-fold compared to our previous report.
- Nitrogen limitation improved isobutene yield by redirecting carbon from growth.
- PVA-entrapped cells sustained isobutene productivity for nearly 31 days.

Abstract:

Cyanobacteria convert CO_2 into valuable compounds using solar energy, making them ideal for sustainable isobutene production, a key precursor for fuels and chemicals. This study aimed to enhance isobutene production in engineered *Synechocystis* sp. PCC 6803 strains: Syn-*Rn*KICD, producing isobutene from α-ketoisocaproate (KIC) via *Rattus norvegicus* α-ketoisocaproate dioxygenase (*Rn*KICD), and Syn-F336V, expressing a mutant variant of *Rn*KICD with improved KIC specificity. We investigated the effects of varying culture conditions, including light intensity, inorganic carbon, and nitrogen on isobutene production. Nitrogen limitation emerged as a critical factor, improving yields by reducing growth. This likely occurred due to decreased competition from branched-chain amino acid (BCAA) biosynthesis, redirecting carbon toward isobutene synthesis. However, prolonged nitrogen limitation ultimately reduced productivity due to impaired metabolic functions. To address this limitation, we employed a polyvinyl alcohol-sodium alginate (PVA-SA) hydrogel, crosslinked with $B(OH)_4^-$ and Ca^{2+}, to entrap cells. This approach restricted growth while maintaining cell viability and isobutene productivity. Optimizing crosslinking parameters such as time, pH, and the hydrogel-to-cell mass ratio improved bead stability under bicarbonate and nitrate supply. This strategy extended cell viability and isobutene productivity in Syn-*Rn*KICD and Syn-F336V by nearly a month, increasing yields by 60% and 80%, respectively, compared to suspension cells, achieving a maximum yield of 94 mg/g DW. This study underscores the importance of optimizing environmental conditions for isobutene production in *Synechocystis*

and highlights the effectiveness of PVA-SA cell entrapment as a biocatalyst platform for

sustained chemical production.

Keywords

Photosynthesis, cyanobacteria, Isobutene, Nitrogen limitation, Cell entrapment, Polyvinyl

alcohol, Biocatalyst

1. Introduction

As global efforts to reduce reliance on fossil resources intensify, the push for renewable

chemical production has become crucial for achieving sustainability and protecting the

environment. Photosynthetic cyanobacteria are being recognized for their potential in

sustainable chemical production, efficiently using sunlight and converting atmospheric CO_2 to

synthesize valuable organic compounds. These microorganisms exhibit notable solar energy

conversion efficiencies, reaching ~8–10% for solar-to-biomass conversion, surpassing the

typical 4-6% efficiency observed in terrestrial plants (Melis, 2009; Zhu et al., 2008).

Additionally, their rapid growth rates, high lipid content for biofuel production, and ability to

thrive in diverse environments, including polluted and wastewater for potential bioremediation

applications, make cyanobacteria efficient and environmentally friendly resources (Farrokh et

al., 2019).

Cyanobacteria's genetic flexibility offers extensive bioengineering potential to optimize

photosynthetic pathways for efficient conversion of CO_2 into targeted organic molecules.

Advances in synthetic biology and metabolic engineering have greatly expanded their capacity

to produce a diverse array of chemicals. These include fuels such as ethanol and butanol,

petrochemical industrial precursors like isoprene, isobutene, and ethylene, and other high-value

compounds such as 2,3-butanediol, sucrose, lactic acid, isobutyraldehyde, and bisabolene

(Angermayr et al., 2015; Carroll et al., 2018; Santos-Merino et al., 2019). Furthermore,

isoprene produced by photosynthetic cyanobacteria can be converted into kerosene-type fuel

precursors through photochemical dimerization processes, highlighting the significant

potential of these organisms for producing renewable, high-energy-density fuels (Rana et al.,

2022). Additionally, the ability of these small, volatile hydrocarbons to easily diffuse out of

cyanobacterial cell membranes without damaging the cells enhances the economic and energy

efficiency of the product harvesting process (Lindberg et al., 2010; Ducat et al., 2011).

Isobutene (C_4H_8) is a highly versatile hydrocarbon pivotal in the chemical industry,

primarily due to its reactive double bond, which facilitates a variety of photochemical reactions,

including oligomerization and polymerization, resulting in a wide range of industrially

important products. Its oligomerization potential extends to creating high-density fuels,

showcasing its viability in sustainable aviation fuel solutions (Hauge et al., 2005; Taylor et al.,

2010; Peters and Taylor, 2013; Kostjuk 2015; Nicholas, 2017). Remarkably, isobutene can

undergo visible light induced cationic photopolymerization, producing significant polymers

such as plastics and rubbers (Hulnik et al., 2023), underscoring its role in advanced materials

science. These chemical reactions are particularly advantageous because they occur under mild

conditions, offering significant benefits over traditional thermal processes. Despite these

advancements, isobutene is still primarily sourced from fossil fuels, highlighting a significant

opportunity and need for developing renewable isobutene production methods to meet the

demands of the petrochemical industry sustainably.

Global demand for sustainable isobutene is surging, with the market projected to

surpass USD 30.71 billion and 16 million tons by 2024 (Rebolledo-Leiva et al., 2022), reaching

USD 38.4 billion by 2030 (Grand View Research). The highest bioisobutene productivity

reported so far is 0.45 mg L^{-1} h^{-1} by the yeast *Rhodotorula minuta* (Wilson et al., 2018), but to

achieve economic viability, yields of ~2–4 g L^{-1} h^{-1} are necessary (van Leeuwen et al., 2012).

Recent advancements focus on exploiting metabolically modified photosynthetic

cyanobacteria to meet these economic and sustainability targets. Our team has engineered the

cyanobacterium *Synechocystis* sp. PCC 6803 (hereafter *Synechocystis)* to produce isobutene

directly from CO_2 by heterologously expressing the *Rn*KICD, re-routing the central metabolism

from pyruvate through the L-leucine biosynthesis pathway-derived metabolite KIC to

isobutene, achieving a rate of 91 ng l^{-1} OD_{750}^{-1} h^{-1} (Mustila et al., 2021). The mammalian

enzyme KICD, catalyzes the decarboxylation and oxygenation of KIC to form β-hydroxy-β-

methyl butyrate (HMB) in the cytosol of liver cells as part of the L-leucine catabolic pathway

(Sabourin & Bieber 1982). Thus, the *Rn*KICD acts on multiple substrates, catalyzing the

conversion of 4-hydroxyphenylpyruvate (HPP) to homogentisate (HGA) and KIC to either

HMB or isobutene. Building on these findings, our team recently refined *Rn*KICD's substrate

preference for KIC to further boost isobutene production. Through protein engineering and

targeted mutagenesis, a specific mutation, F336V*,* was identified, which shifts *Rn*KICD*'*s

substrate preference toward KIC rather than HPP, significantly enhancing isobutene production

(Schumann et al., 2024). However, further optimizations of favorable environmental culture

conditions are essential to achieve production levels that meet commercial demands.

The efficiency of photoproduction of high-value compounds by microalgae is

significantly influenced by various environmental conditions, including light intensity, pH,

temperature, and the availability of macronutrients like inorganic carbon, nitrogen, phosphorus,

and sulfur (Miller and Colman, 1980; Gordillo et al., 1998; Aboim et al., 2019; Rodrigues et

al., 2023). However, traditional suspension cultures face yield limitations due to excessive

biomass accumulation, which increases nutrient demand, oxidative stress, and self-shading

effects, ultimately reducing photosynthetic efficiency (Chisti, 2007; Kim et al., 2011; Pedruzi

et al., 2019). One promising strategy to address these limitations is the physical entrapment of

microalgal cells within hydrogel polymers. One widely used method involves crosslinking

sodium alginate (SA) with $CaCl_2$ to form stable hydrogel beads through ionic bonds between

Ca^{2+} ions and the carboxylate groups of guluronic and mannuronic acid residues in alginate

chains, resulting in a robust three-dimensional network (Simpson et al., 2004). This whole-cell

entrapment approach improved cyanobacterial viability and productivity, as demonstrated by

their notable cell viability and enhanced production of ammonia, hydrogen, ethylene, β-

phellandrene, and sucrose (Brouers and Hall, 1986; Kosourov et al., 2014; Vajravel et al., 2020;

Valsami et al., 2021; Tóth et al., 2022). These improvements were achieved by effective

biomass growth mangement, nutrient distribution, enhnced cellular communication, and

reduced oxidative stress. However, the use of Ca^{2+} alginate beads in cyanobacterial

photoproduction systems presents challenges, particularly due to the potential formation of

$CaCO_3$, which diminishes bead stability, especially in environments where $NaHCO_3$ is used as

a crucial inorganic carbon source in airtight vessels for small volatile compounds production

(Rissanen et al., 2022). Therefore, it is important to select suitable hydrogel polymers for the

cell entrapment approach that can withstand HCO_3^--supplemented cultures.

Polyvinyl alcohol (PVA) hydrogels are widely used in biomedical applications due to

their flexible crosslinking capabilities, biocompatibility, biodegradability, and cost-effective

(Zhong et al., 2024). PVA can be crosslinked with sodium borate, boric acid, glutaraldehyde,

formaldehyde, UV treatment and freeze-thawing (Wu & Wisecarver, 1992; Gohil et al., 2006;

Bolto et al., 2009; Hua et al., 2010). When $B(OH)_4^-$ ions, formed from H_3BO_3, interact with

PVA, they create reversible covalent bonds with the hydroxyl groups of PVA, forming a stable

yet dynamic hydrogel structure that can break and reform bonds in response to external forces

or environmental changes (Cheng and Rodriguez, 1981; Wu & Wisecarver et al., 1992; Choe
et al., 2024; Wu et al., 2024). This dynamic behavior enhances the material's resilience and
adaptability, enabling repeated deformation and recovery, which improves the hydrogel's
functionality in applications such as drug delivery and tissue engineering.

In this study, we aimed to enhance isobutene production in *Synechocystis* strains, Syn-
*Rn*KICD and Syn-F336V, by systematically exploring various environmental factors,
including light intensity, bicarbonate and nitrogen source availability, and employing borate-
PVA hydrogel for cell entrapment. Our findings indicated that incorporating a stabilizing agent
such as SA, adjusting the crosslinking duration, maintaining neutral pH, and optimizing the
cell biomass-to-hydrogel ratio significantly improved the hydrogel beads' stability and
longevity, where HCO_3 served as a crucial inorganic carbon source. By utilizing this optimized
system, we extended the productive phase of isobutene to one month, achieving maximum
yields of ~58 mg/g DW and 94 mg/g DW in the entrapped Syn-*Rn*KICD and Syn-F336V cells,
respectively.

2. Materials and Methods
2.1. Strains and Growth Conditions
The two *Synechocystis* strains were employed in this study were previously constructed: Syn-
*Rn*KICD, which produces a KICD enzyme from *Rattus norvegicus* (*Rn*KICD) (Mustila et al.,
2021), and Syn-F336V, in which a modified version of *Rn*KICD is generated (Schumann et al.,
2024). *Synechocystis* precultures were cultivated photoautotrophically in BG11 liquid medium
in 6-well polystyrene plates, illuminated with continuous white light at a photosynthetic photon
flux density (PPFD) of 30 µmol photons m^{-2} s^{-1}. For the cultivation of experimental cultures,
250 mL Erlenmeyer flasks (E-flasks) containing 80 mL of medium were buffered with 25 mM
HEPES-NaOH (pH 7.4) and supplemented with 50 µg/mL kanamycin (Km). The cultures were
supplied with 50 mM $NaHCO_3$ and maintained at 30°C with shaking at 100 rpm under
illumination of 50 µmol photons m^{-2} s^{-1}. All precultures were initiated from cryopreserved
cells stored at -80°C in 10% glycerol.

2.2. Isobutene Production and Quantification
For isobutene production, *Synechocystis* cells cultivated in E-flasks were harvested by
centrifugation at 5000 rpm for 5 min upon reaching an optical density (OD_{750}) of ~ 1. Following
centrifugation, the cells were washed with fresh BG11 medium, adjusted to similar OD_{750} for

all cultures, and supplemented with 50 mM $NaHCO_3$ and 50 µg mL^{-1} Km. A 10 mL aliquot of

the suspension cultures was transferred into 20 mL gas-tight vials, which were sealed with

Teflon-coated septa.

The suspension cultures were shaken at 100 rpm under an illumination of 50 µmol

photons m^{-2} s^{-1} (unless stated otherwise). Every 24 h of post light illumination, the headspace

of the vials was sampled through the Teflon septum, and after each injection, the septa were

replaced to minimize any loss of produced isobutene. A 5-min interval was allowed during

septa replacement to facilitate gas exchange.

Isobutene detection and quantification were performed using an Agilent 7890A Gas

Chromatography (GC) system coupled with a 5975 Mass Selective Detector (MSD), equipped

with an HP-1ms Ultra I column. Samples were taken from 20 mL GC vials with Teflon-coated

septa (VWR) and injected using a PAL-RSI85 autosampler with the PAL-HS2500 gas syringe

tool. The GC-MS method for isobutene detection was adapted from Rossoni et al (2015). Prior

to injection, GC vials were incubated for 5 min at 40°C to equilibrate the isobutene

concentration in the headspace. For detection, 100 µL of each sample were injected at an inlet

liner temperature of 150°C with a 100:1 split ratio. The oven temperature was held at 40°C for

3 min before increasing by 30°C/min for 2 min. The retention time for isobutene was ~1.94

min with an authentic standard (99% v/v C_4H_8, AGA, Sweden). The MS detector operated in

selected ion monitoring (SIM) mode to enhance sensitivity, specifically detecting isobutene

signals at m/z 41 and 56. Signals within the chromatogram were analyzed for peak area through

autointegration. The quantity of isobutene was determined from a calibration curve established

using various dilution percentages of a commercial gas standard (99% v/v C_4H_8, AGA,

Sweden).

Isobutene productivity was normalized to the initial OD values for suspension cultures

and initial cell dry weight (DW) was used for entrapped cells. For a direct comparison of

isobutene productivity between suspension and entrapped cells, the initial dry weight (DW) of

cells was set to be similar across all samples. The cumulative productivity of isobutene was

calculated by summing the isobutene produced at 24-h intervals. The data were averaged from

3 independent measurements taken from 3 replicates (±SD).

2.3. Induction of Nitrogen Starvation

Nitrogen (N) starvation in *Synechocystis* cells was induced by omitting $NaNO_3$ from the BG-

11 medium, the sole N source for the non-nitrogen-fixing cyanobacteria in the E-flask allowing

for adaptation as mentioned in the section (2.1). In the experimental cultures, varying N

starvation periods were tested to assess the effects of both short-term (24 h) and long-term (72

h) N starvation on *Synechocystis*. After the starvation period the cells were washed with N-free

BG-11 medium by centrifuging at 5000 rpm for 5 min (2x). The cells were then prepared for

isobutene production by resuspending them in N-free medium and transferring them into

tightly sealed vials. They were subsequently adapted under the conditions described in Section

2.2.

2.4. Cell entrapment procedures

Syn-*Rn*KICD and Syn-F336V cells were cultivated under experimental conditions as described

in Section 2.2. Once the cells reached OD_{750} of 1, the cells were harvested by centrifugation at

5000 rpm for 5 min. Subsequently, the cyanobacterial cells were entrapped in hydrogel for

entrapment, as described by Brouers and Hall (1986), with some modifications. We employed

a PVA hydrogel crosslinked with H_3BO_3 using a droplet-based dispersion technique via a 2 mL

syringe. To form a distinct hydrogel beads, 0.5% SA was incorporated into the 6% PVA

hydrogel and crosslinking was achieved using 2% $CaCl_2$ and 4% H_3BO_3 solutions. We used

varying formulations of wet cell biomass and PVA-SA hydrogel, with ratios ranging from

5:6:0.5 to 5:36:3 (cell:PVA:SA) for cell entrapment.

Hydrogel droplets were manually dispensed into the crosslinking agents using a 2 mL

syringe. After dispensation, the plates were sealed with Parafilm, and the crosslinking duration

was optimized between 5 to 20 h. The instantly formed beads were incubated at 30°C to

produce stable beads in the presence of crosslinking agents, under illumination at 30 µmol

photons m^{-2} s^{-1}. After the desired duration, the beads were thoroughly rinsed with sterile

deionized water to remove excess crosslinkers and minimize ionic stress on the cells. The

entrapped cells in hydrogel beads were then transferred into 20 mL gas-tight vials containing

BG11 medium (with and without $NaNO_3$, as stated) with 50 mM $NaHCO_3$ and subjected to the

isobutene production conditions described in Section 2.2.

2.5. Measurement of cell density and pigment content

The growth of the cell cultures was assessed spectrophotometrically (Varian 50 Bio UV-

Visible Spectrophotometer) by measuring the OD_{750}. The chlorophyll *a* (Chl) content was

quantified by measuring the absorbance at 665 nm following extraction with 90% methanol (Lichtenthaler, 1987). Additionally, the absorption spectra of the whole cell suspensions were recorded. Prior to measurement, all cultures were standardized to a uniform cell density of $OD_{750} \sim 0.5$.

3. Results and Discussion

3.1. Optimizing isobutene production in Syn-*Rn*KICD: The interplay of light intensity and inorganic carbon

The main strain used in this study; Syn-*Rn*KICD, expresses the *Rn*KICD gene in *Synechocystis* that encodes KICD, an enzyme that has shown to convert KIC to isobutene *in vitro* and *in vivo* (Mustila et al., 2021). To study how varying light intensities affect isobutene production, we conducted experiments under three different photon flux densities: 35 (low light, LL), 85 (moderate light, ML), and 135 (high light, HL) μmol photons m^{-2} s^{-1}. To ensure adequate inorganic carbon (Ci) availability for photosynthesis in the air-tight vessels, all cultures were supplemented with 50 mM $NaHCO_3$. The optimal light intensity for maximizing isobutene production was determined (Fig 1A).

After 24 h under production conditions, isobutene productivity in Syn-*Rn*KICD cells was observed to be $\sim$11 μg L^{-1} OD_{750}^{-1} under LL intensity (Fig. 1A). Higher productivity levels of $\sim$19 and 20 μg L^{-1} OD_{750}^{-1} were observed under ML and HL, respectively. These results indicate that, while ML and HL light intensities result in higher isobutene production compared to LL, the difference between ML and HL intensities at 24 h was negligible, and their productivity was proportional to the increased cell density as observed in their OD_{750} (Fig. S1A).

At 72 h, isobutene production reached $\sim$25 μg L^{-1} OD_{750}^{-1} under LL, $\sim$33 μg L^{-1} OD_{750}^{-1} under ML, and $\sim$38 μg L^{-1} OD_{750}^{-1} under HL intensities (Fig. 1A). While HL intensity resulted in the highest isobutene production, it also caused cell bleaching and a decrease in cell density by 72 h (Fig. S1A). In contrast, cells exposed to LL and ML showed no bleaching during this period. This suggests that cell bleaching might result from an imbalance between light energy input and Ci assimilation. Prior studies indicate that cell bleaching can impair the growth, photosynthetic efficiency and productivity of *Synechocystis* (Van Alphen et al., 2018; Vajravel et al., 2020). Adequate nutrient supply, including Ci, can mitigate cell bleaching under such light conditions in cyanobacteria (Markou et al., 2014).

Thus, we hypothesized that excessive light energy relative to available Ci could induce oxidative stress, leading to cell bleaching in RnKICD cells. To test this, cells were exposed to constant HL (135 µmol photons m^{-2} s^{-1}) with varying HCO$_3^-$ concentrations (0 to 150 mM). The results showed that cells supplemented with higher concentrations of HCO$_3^-$ (100 and 150 mM) did not exhibit cell bleaching at 72 h, while those with lower HCO$_3^-$ (50 mM) experienced bleaching and reduced growth (Fig. S1B). This suggests that insufficient Ci levels under HL intensity disrupt energy homeostasis, potentially causing oxidative stress and bleaching.

Although isobutene production was observed under all conditions, cells supplemented with HCO$_3^-$ exhibited significantly higher growth and isobutene productivity compared to the non-supplemented control cells. The maximum isobutene productivity (~53 µg L^{-1} OD$_{750}^{-1}$) was achieved with 150 mM HCO$_3^-$. That was nearly 4 times higher than in the control cells (Fig. 1B). Although no significant difference in isobutene productivity was observed between 50 to 150 mM HCO$_3^-$ supplementation (49–53 µg L^{-1} OD$_{750}^{-1}$ at 72 h, Fig. 1B), bleaching occurred earlier in the cultures with low HCO$_3^-$ level (50 mM) potentially impacting long-term isobutene production. Notably, control cells (0 mM HCO$_3^-$), which showed lower isobutene production (14 µg L^{-1} OD$_{750}^{-1}$ at 72 h; Fig. 1B), did not exhibit the bleaching phenotype observed in cells with 50 mM HCO$_3^-$, nor did they show substantial growth (Fig. S1B). Such production likely originated from stored carbon in the control cells. These results highlight the critical balance of HCO$_3^-$ in preventing cell bleaching under HL conditions in Syn-RnKICD cells.

Figure 1: Influence of light intensity and inorganic carbon source on isobutene production in Syn-RnKICD cells. A) Isobutene production was measured in the headspace of cultures cultivated under three light intensities (35, 85, and 135 µmol photons m^{-2} s^{-1}). All

cultures were uniformly supplemented with 50 mM HCO_3. B) Isobutene productivity was

assessed across a gradient of HCO_3 concentrations (0 - 150 mM) under 135 μmol photons m^{-2}

s^{-1}. The cumulative isobutene production was calculated by summing the isobutene produced

at 24-h intervals and normalized to each cell culture's initial OD_{750} values. The initial OD for

all cultures was set similarly across all experiments in this study. KICD refers to Syn-*Rn*KICD.

Data shown are the average of 3 independent measurements from 3 replicates (±SD).

HCO_3^- serves as a key substrate for the Calvin cycle, the primary pathway of CO_2

fixation in cyanobacteria (Bennet, 1991; Miller et al., 1990; Badger and Price 2003). To

balance the supply and demand side of photosynthetic metabolism, HL intensities increase the

demand for CO_2 fixation. When HCO_3^- levels are limiting, the cell growth is competing with

isobutene production pathways. Prior studies also corroborate that excess light can trigger

oxidative stress through reactive oxygen species (ROS) overproduction, decreasing

photosynthetic efficiency and causing bleaching (Miller and Colman, 1980; Miller et al., 1990;

Gordillo et al., 1998; Rodrigues et al., 2023). Our findings emphasize the need to balance light

intensity and Ci source availability to maintain *Synechocystis* cell viability and improve

isobutene production.

3.2. Impact of Nitrogen starvation on the isobutene production in *Synechocystis* cells

3.2.1.   Impact of short-term nitrogen starvation on Syn-*Rn*KICD experimental cultures

Previous research by our group implies that KIC serves as substrates for both isobutene and

BCAA biosynthesis pathways in *Synechocystis* (Mustila et al., 2021; Schumann et al., 2024).

We reasoned that (i) limiting N availability would decrease the metabolic flux toward BCAA

synthesis, thus redirecting more carbon toward isobutene production, (ii) the induction of N

starvation in *Synechocystis* can provide valuable insights into how N manipulation impacts the

partitioning of carbon fluxes between competing biosynthetic pathways in Syn-*RnKICD*. To

test these hypotheses, we subjected Syn-*Rn*KICD cells, incubated with 50 mM HCO_3 under 50

μmol photons m^{-2} s^{-1} light intensity to prevent HL-induced damage, to N starvation and

evaluated its effect on isobutene productivity and on major photosynthetic pigments (Fig. 2).

We observed that isobutene production occurred regardless of N supply (Fig. 2A).

However, N starved cells exhibited chlorosis after 24 h (Fig. 2B), indicating reduced level of

major photosynthetic pigments, particularly phycobilins (PB) and Chl (Fig. 2C). The

absorption (ABS) spectra results showed that a sharp decrease in PB-to-Chl ratio, a known

marker of N deprivation in *Synechocystis,* which was also observed in previous studies in cyanobacteria (Allen and Smith, 1969; Allen et al.,1990). Up to 72 h, there was no significant difference in isobutene production between N supplied (N+) and N starved cells (N-) (Fig. 2A). However, after 96 h, N- cells outperformed N+ cells producing ~46 μg L^{-1} OD$_{750}$$^{-1}$ of isobutene at 120 h, which was ~1.2 times the amount produced by N+ cells (37 μg L^{-1} OD$_{750}$$^{-1}$). At this point, N+ cells showed stagnation in isobutene production (Fig. 2A), along with severe cell bleaching and significant pigments loss (Fig. 2B). On the other hand, N- cells continued to produce isobutene up to 216 h, reaching ~60 μg L^{-1} OD$_{750}$$^{-1}$ (Fig. 2A).

Figure 2: Impact of nitrogen starvation on isobutene production in Syn-*Rn*KICD cells.
A) Isobutene productivity was periodically measured in cell cultures supplemented with 18 mM NaNO$_3$ (N+), and without NaNO$_3$ (N-) in BG11 medium. The experimental cultures were cultivated in an appropriate medium for 24 h before being subjected to isobutene production conditions in tightly sealed vials, which were continuously supplied with an appropriate nitrogen-containing medium. The produced isobutene was measured from the headspace of the culture vials every 24 h. B) Phenotypic changes of the cultures were monitored throughout the

experiment. C) Comparative whole-cell ABS spectra were obtained to a similar OD_{750} of the
cells which showed phycobilin (λmax = 634 nm) and Chl a (λmax = 686 nm) peaks. The black
arrow in Figure A indicates the resuscitation phase, where the reintroduction of $NaNO_3$ into
the N- cultures facilitated a recovery process, and a subsequent cycle of N starvation was
instituted. All the cultures were maintained under 50 μmol m^{-2} s^{-1} with 50 mM HCO_3. Isobutene
productivity was normalized to the initial OD_{750} values of cultures, which was set to similar in
all the experimental setups. These results are the average of 3 independent measurements from
3 replicates ($\pm$SD). KICD refers to Syn-*Rn*KICD.

Although both N+ and N- cells had similar isobutene production rates up to 48 h, N-
cells outperformed N+ cells from 72 h onward (Fig. S2A). Under the tested conditions, the rate
of production was enhanced 10 times as compared to the levels previously reported in
Synechocystis (Mustila et al., 2021). Initial cell growth, as measured by OD_{750}, and Chl content,
was 2 to 4 times higher in N+ cells than in N- cells up to 72 h (Fig. S2B and C). At 96 h, N+
cells showed a rapid decrease in their photosynthetic pigments (Fig. S2D) and no pigment
peaks were detected by 120 h, likely due to severe cell bleaching which led to no further
isobutene production. On the other hand, although N- cells showed reduced pigments levels
initially (Fig. 2B), only a gradual decline in the ABS spectra was observed between 48 and 120
h (Fig. S2E) and these N- cells continued to produce isobutene until 216 h (Fig. 2A).

To explore recovery in N- cells, N was reintroduced in the cultures at 252 h. During
resuscitation phase (R-phase), these cells regenerated their growth and photosynthetic
pigments (Fig. 2B, S2B, C, and F), similar to findings in a previous study on cyanobacteria
(Forchhammer & Schwarz, 2019). After initiating N starvation for the second time at 120 h,
isobutene production resumed, albeit at lower levels than during the first cycle, producing ~84
μg L^{-1} OD_{750}^{-1} of isobutene at 552 h (combining the first and second cycles) (Fig. 2A). This
overall productivity is 2.3-fold higher than that of the N+ cells (37 μg L^{-1} OD_{750}^{-1}). These
results suggest that N limitation could prolong isobutene production by preventing cell growth
and major pigment loss. Furthermore, the recovery after N reintroduction highlights the
potential for cyclic N manipulation to optimize long-term isobutene productivity.

A critical question is why N+ cells underwent bleaching and ceased production of
isobutene despite being maintained under optimal light conditions (50 μmol photons m^{-2} s^{-1})
and with the supply of 50 mM HCO_3. Considering these factors, it is also important to note that

cell bleaching occurred later under these conditions (120 h, Fig. 2B) compared to cells exposed to HL (135 μmol photons m^{-2} s^{-1}), where cell bleaching was observed as early as 72 h (Fig. S1B). The bleaching in N+ cells may be attributed to Ci limitation probably caused by the competition between overgrowth, and isobutene production in air-tight vials with 50 mM HCO_3^- supplementation. In contrast, N+ cells grown in E-flasks did not bleach even after two weeks (data not shown), likely due to the sufficient CO_2 gas exchange through the cotton-sealed flasks.

Surprisingly, however, N- cells, maintained isobutene production over 552 h even with a reduced amount of photosynthetic pigments (Fig. 2A). In correlation with that several previous studies showed that in cyanobacteria N starvation is initiated with the degradation of nitrogen-rich phycobiliproteins, to supply essential nutrients for cell survival (Allen & Smith, 1969; Spat et al., 2018). Further, N starvation induced chlorosis leads to a dormant state in which cell growth is arrested, and low-level photosynthesis is maintained as a long-term survival mechanism (Forchhammer & Schwarz, 2019; Forchhammer & Selim, 2020). Despite this, CO_2 fixation can still occur in N starved cells, albeit at lower levels (Allen et al., 1990). In line with these reports, our results showed a significant growth reduction in N- cells compared to N+ cells (Fig. S2B and C). N+ cells exhibited 4.7-fold increase in OD, while N- cells showed a 2-fold increase after 48 h compared to their initial levels (0 h), despite all cultures having similar initial OD values (Fig. S2B and C). Based on this, we suggest that N- cells could still fix CO_2 and are directing carbon toward isobutene production rather than cell growth. These results reveal that the limited growth under N deprived conditions might prolong cell survival and enable isobutene production for a longer period than in N supplied cells.

3.2.2. Impact of prolonged nitrogen starvation on Syn-*Rn*KICD experimental cultures

To determine if further growth restriction in N starved cells could enhance isobutene productivity within a shorter production period, we extended the N starvation period in Syn-*Rn*KICD experimental cultures. Unlike previous experiments where experimental cultures were subjected to 24 h of N starvation (Fig. S2B), the cells in this experiment underwent 72 h of N starvation. In all N starvation experiments, isobutene production in the tightly sealed vials was consistently supplemented with an appropriate nitrogen-containing medium.

As expected, the cells subjected to prolonged N starvation exhibited higher isobutene production at 48 h (29 μg L^{-1} OD_{750}^{-1}), compared to N+ cells (24 μg L^{-1} OD_{750}^{-1}) (Fig. 3A).

Furthermore, N- cells showed consistency in their rate of isobutene production over 120 h (Fig.

S3A) and OD of these cells slightly increased to1.5 times as compared the initial OD (Fig. S3B

and C). In contrast, cells with only 24 h of adaption to N starvation showed a 2-fold increase

in OD (Fig. S2B). At 120 h, no pigments were detected in N+ cells, while N- cells still displayed

a Chl peak at 686 nm (Fig. S3D and F). Correspondingly, N+ cells showed a decline in

isobutene production by 120 h, whereas N- cells continued producing isobutene until 168 h,

reaching a productivity of 57 µg L^{-1} OD_{750}^{-1}, nearly twice that of N+ cells (30 µg L^{-1} OD_{750}^{-1})

(Fig. 3A). In the cells adapted to nitrogen starvation for 72 h, maximum isobutene productivity

was observed at 168 h (Fig. 3A), while cells with 24 h of N starvation adaptation required 216

h to reach the same level of productivity (Fig. 2A). These results suggest that substantial growth

limitation due to the prolonged N starvation can increase isobutene production within a shorter

time frame, possibly by redirecting carbon flux toward isobutene production, rather than cell

growth.

To assess if severe bleaching in N+ cells was caused by competition for Ci sources

between growth and isobutene production, we tested cells with doubled nitrogen content (2xN),

assuming that this would enhance initial growth on the expense of isobutene production. We

supplemented cultures with 2xN (36 mM $NaNO_3$) and 50 mM HCO_3. The results showed

increased cell growth in cultures supplemented with 2xN by 48 h (Fig. 3B), accompanied by

higher OD and Chl levels at 24 and 48 h, respectively, compared to N+ cells (Fig. S3B and C).

This increased growth potentially contributed to the marginally lower isobutene production as

expected in 2xN cells (28 µg L^{-1} OD_{750}^{-1}) compared to N+ cells (30 µg L^{-1} OD_{750}^{-1}) (Fig. 3A).

Further accelerated growth in these 2xN cells led to faster bleaching, which was observed as

early as 72 h, compared to N+ cells, where bleaching was observed only at 120 h (Fig. 3B).

The ABS spectra also showed that the decrease in photosynthetic pigments was faster in 2xN

cells up to 72 h, with no detectable pigments at 96 h (Fig. S3E), whereas N+ cells still showed

detectable Chl peaks at that time (Fig. S3D). These findings confirm that the relatively

increased growth in 2xN cells resulted in faster pigment degradation, bleaching and relatively

lower isobutene production compared to N+ cells, suggesting that competition between cell

growth and isobutene production can lead to carbon limitation in the tightly sealed vials over

time.

Figure 3: Impact of prolonged nitrogen starvation on isobutene productivity in Syn-
_Rn_KICD cells. A) The cumulative isobutene productivity was evaluated in cultures under
three distinct conditions: N+, N- and 2xN. All experimental cultures were cultivated in 50 mM
HCO_3 supplemented BG11 medium in the presence (N+ and 2xN) and absence (N-) of $NaNO_3$
for 72 h before transferring to tightly sealed vials for determination of isobutene productivity.
The cultures for production conditions were continuously supplied with an appropriate
nitrogen-containing medium. B) Phenotypic changes of the cultures were monitored
throughout the experiment. The cultures were incubated with 50 mM HCO_3 under 50 μmol
photons m^{-2} s^{-1} light intensity. The isobutene productivity was normalized on the initial OD_{750}
values of all the cultures. These results are the average of 3 independent measurements from 3
replicates (±SD). KICD refers to Syn-_Rn_KICD.

3.2.3. Influence of both bicarbonate and nitrate limitation on isobutene production in
Syn-_Rn_KICD cells

To explore whether limiting the cell growth, by reducing Ci availability, would affects
bleaching and isobutene production, Syn-_Rn_KIC were cultivated without HCO_3^-
supplementation. The three culture conditions employed in this experiment were as follows:
N+, 2xN, and N-, all without HCO_3^-. Interestingly, up to 144 h, no noticeable difference in
isobutene productivity was observed across all culture conditions in the Ci-starved state (Fig.
4A). N- cells still exhibited chlorosis-induced pigment loss in the absence of HCO_3^- (Fig. 4B).
Up to 48 h, no growth was observed in the cells under the 2xN condition (Fig. S4A).

In corresponding with that, a significant decrease of photosynthetic pigments was
observed in N+ and 2xN cells after 96 h (Fig. S4B and C). In contrast, N- cells did not show a

decrease in pigment content (Fig. S4D). Interestingly, for an extended period (up to 144 h) all

cells continued to produce almost similar amount of isobutene but at a low level. This suggest

that that in the absence of HCO_3^- no significant difference in isobutene productivity was

observed between N supplied (N+ and 2xN) and N starved (N-) cells (Fig. 4A). However, it is

important to note that without HCO_3^-, overall isobutene productivity dropped significantly to

10 and 9 $\mu g\ L^{-1}\ OD_{750}^{-1}$ at 72 h (Fig. 4A), as compared to 35 and 38 $\mu g\ L^{-1}\ OD_{750}^{-1}$ with HCO_3^-

supplementation in N+ and N- cells, respectively (Fig. 2A). These results suggest that the

differences in isobutene production between N+ and N- cells were only observed in the

presence of HCO_3^-.

To investigate whether the re-addition of HCO_3^- could result in any differences in the

regeneration of cell growth and isobutene production among these Ci-deprived cells, HCO_3^-

was added to all cultures (N+, 2xN, and N-) at 144 h. Surprisingly, all cultures exhibited

growth, as indicated by phenotypic changes observed at 192 h (Fig. 4B). However, increased

isobutene production was only observed in N- cells. In contrast, no significant increase in the

isobutene production was noticed in N supplied cells (N+ and 2xN), which, on the other hand,

show a substantial increase in their cell growth (Fig. 4B), probably leading to Ci depletion,

thus, the ability to produce isobutene in these N supplied cells might be diminished.

Surprisingly, N deprived cells continued producing isobutene for 480 h, reaching ~89 $\mu g\ L^{-1}$

OD_{750}^{-1}. These results indicate that CO_2 fixation using HCO_3 as Ci source is most likely more

directed toward isobutene production since cell growth is restricted in N deprived cells.

Furthermore, the observed difference in isobutene production between N supplied and N

deprived cells clearly to be primarily due to differences in their cell growth, which were highly

influenced by the addition of HCO_3.

Figure 4: Impact of nitrogen and carbon limitation on isobutene production in Syn-*Rn*KICD cells. A) The isobutene productivity was monitored in Syn-*Rn*KICD cells under three distinct conditions: N+, N-, and 2xN without addition of HCO_3. The cultures were adapted to the different nitrogen concentrations for 24 h before transferring to tightly sealed vials for determination of isobutene productivity. The cultures for production conditions were continuously supplied with an appropriate nitrogen-containing medium. B) Phenotypic changes of the cultures were monitored throughout the experiment. C) Cumulative isobutene production was examined under: N+, N-, and 2xN and 50 mM HCO_3 was added after 144 h of production period. The cultures were illuminated with 50 μmol photons m^{-2} s^{-1}. The isobutene productivity was normalized based on the initial OD_{750} values of all the cultures. These results are the average of 3 independent measurements from 3 replicates (±SD). KICD refers to Syn-*Rn*KICD.

Under N limitation, growth is reduced, likely at least partly due to decreased BCAA
biosynthesis. Such limited growth might lead to metabolic redirection of carbon toward
isobutene production. As a result, isobutene formation becomes less dependent on growth, with
carbon flux shifting away from BCAA biosynthesis toward isobutene production.

3.2.4.   Assessment of isobutene production in the Syn-F336V strain under nitrogen
**   starvation**
In a previous report, we showed that by introducing an engineered *Rn*KICD, with enhanced
substrate specificity, to *Synechocystis* (strain Syn-F336V) an increased *in vivo* isobutene
production was achieved (Schumann et al., 2024). Here, we examine the effects of N starvation
on isobutene production in the Syn-F336V cells to determine if the earlier observed results
were specific to the unmodified *Rn*KICD enzyme or represent a broader metabolic shift.
Further, we expect Syn-F336V cells to reach higher production level of isobutene than Syn-
*Rn*KICD under N limitation. This expectation stems from two key factors: the reduced BCAA
synthesis under N limitation, which redirects more carbon precursors toward isobutene
production, and the F336V's improved ability use KIC over HPP as substrate. This approach
can help to clarify how both enzyme and cultivation optimization can synergistically increase
isobutene yields.

Our results showed that the N supplied Syn-F336V cells exhibited a 1.6-fold increase
in isobutene production, reaching 60 $\mu g\ L^{-1}\ OD_{750}^{-1}$ at 96 h (Fig. 5A), as compared to the 37
$\mu g\ L^{-1}\ OD_{750}^{-1}$ by Syn-*Rn*KICD cells (Fig. 2A) under identical conditions. Thus, confirming
that Syn-F336V cells are significantly more efficient at producing isobutene than the Syn-
*Rn*KICD cells. When subjected to N starvation, the Syn-F336V cells showed even higher
isobutene production, reaching ~80 $\mu g\ L^{-1}\ OD_{750}^{-1}$ at 96 h, surpassing the productivity of their
N supplied cells (Fig. 5A). N+ Syn-F336V cells, like N+ Syn-*Rn*KICD cells, displayed
increased cell growth up to 72 h, followed by cell bleaching at 96 h (Fig. 2B and 5B), with a
decline in their isobutene production. Further, under N- conditions, Syn-F336V cells exhibited
chlorosis, a typical symptom of N starved cyanobacterial cells. Despite this, the isobutene
production continued to increase for 72 h in Syn- F336V N- cells, reaching a maximum yield
of ~120 $\mu g\ L^{-1}\ OD_{750}^{-1}$ at 216 h (Fig. 5A). The peak of isobutene productivity in the Syn-
F336V N- cells was 2-fold higher than in the Syn-*Rn*KICD N- cells over the same period (Fig.
2A and 5A). These results suggest that the Syn-F336V cells benefit from reduced BCAA
biosynthesis under N limitation and increased substrate specificity of the *Rn*KICD enzyme

towards KIC, leading to overall enhanced isobutene production than Syn-*Rn*KICD. Both Syn-

F336V and Syn-*Rn*KICD strains showed increased isobutene production under N limitation,

suggesting that this response is specific to the synthetic isobutene production pathway in

Synechocystis cells. In this pathway, branched-chain amino acid (BCAA) biosynthesis and

isobutene production compete for the same substrate, KIC (Fig.11).

Figure 5: Impact of nitrogen starvation on isobutene production in Syn-F336V cells. A)

The cumulative productivity of isobutene was determined in the cells incubated in BG11

medium with and without supplemented $NaNO_3$. B) Phenotypic changes of the cultures were

monitored every 24 h throughout the experiment. The cultures were illuminated with 50 μmol

photons m^{-2} s^{-1} and supplemented with 50 mM HCO_3. The isobutene productivity was

normalized to the initial OD_{750} values of all the cultures. These results are the average of 3

independent measurements from 3 replicates (±SD). 336 refers to Syn-F336V.

Taken together the results suggest that when N+ cells are supplied with HCO_3^- a

competition for carbon may occur between their cell growth and isobutene production. This

competition will eventually lead to carbon limitation, inducing metabolic stress and cell

damage in closed culture vials, as shown in Figures 2 to 5. Conversely, N starvation-induced

growth arrest in N- cells could direct the supplied Ci source primarily towards isobutene

production for an extended period of time, as demonstrated in Figures 2 to 5. Although N

starved cells exhibited prolonged isobutene production as compared to N+ cells under HCO_3^--

supplementation (Figs. 2-5), it is essential to note that by time these N deprived cells

experienced significant nutrient stress, leading to substantial loss of major photosynthetic

pigments (Figs. S2 and S3) and led to a halted isobutene production. Previous studies have

shown that while N starved cells can sustain low levels of photosynthesis (Sauer et al., 2001)

and CO_2 fixation (Allen et al.,1990), this compromised photosynthetic capacity may ultimately

limit long-term cell survival. Thus, utilizing N starved cells may not be ideal for sustainably

maximizing isobutene production. Based on these insights, we propose that long-term

improvements in isobutene production can be achieved by adopting strategies such as

physically restricting cell growth while ensuring the provision of optimal nitrogen and carbon

sources to support robust metabolic function.

3.3. Entrapment of Syn-*Rn*KICD cells in PVA hydrogel beads

In this section, we employed a cell entrapment strategy for prolonged cell viability and

enhanced isobutene production in Syn-*Rn*KICD and Syn-F336V cells by limiting cell growth

under the controlled availability of Ci and N sources. For this approach, we used 6% PVA

hydrogel crosslinked with 4% H_3BO_3 to create a entrapped cell beads. The rationale behind this

approach is that physically entrapping cyanobacterial cells can limit their growth, redirecting

energy and carbon toward product formation rather than biomass accumulation, as supported

by previous studies (Kosourov et al., 2014; Seibert et al., 2018; Vajravel et al., 2020). During

the droplet-based bead formation process, using a 5:6 ratio of cell wet mass to PVA hydrogel,

we encountered a challenge: the PVA hydrogel tended to adhere, preventing the formation of

distinct beads in the presence of the crosslinking agent $B(OH)_4^-$. This issue might have resulted

from insufficient electrostatic repulsion or steric stabilization among the PVA hydrogel

particles, leading to clumping (Wu & Wisecarver et al 1992). To address this, we incorporated

a small amount of SA (0.5%) into the PVA hydrogel mix. Because SA can facilitate bead

formation due to its ability to form crosslinked structures via ionic interactions, enhancing

stabilization. Distinct hydrogel beads were promptly generated following this modification: the

cell-PVA-SA mixture, at a 5:6:0.5 ratio, was introduced into a solution of crosslinking agents,

H_3BO_3 (4%) and $CaCl_2$ (2%).

We first evaluated the effects of entrapping Syn-*Rn*KICD cells in PVA hydrogel beads'

stability in the presence of HCO_3. After crosslinking duration of 5 h at 30 °C, the beads initially

appeared stable, but, upon the addition of HCO_3^-, bead-disintegration was observed at 24 h,

resulting in a cell suspension (Fig. 6A). The shorter crosslinking duration time may have

contributed to the early disintegration of the beads. Thus, we extended the crosslinking duration

from 5 h to 20 h without changing the temperature, aiming to strengthen the interactions

between the hydrogel and crosslinking agents, thereby creating more stable hydrogel beads.

Although the extended crosslinking time (20 h) showed an increased stability of PVA

entrapped cell beads initially in presence of HCO_3, the beads began to disintegrate after 96 h

of HCO_3^- supplementation (Fig. 6B). These results suggest that while longer crosslinking

period can postpone bead breakdown in the presence of HCO_3, it cannot completely prevent it.

Figure 6: Impact of crosslinking duration on the stability of PVA-SA entrapped Syn-

***Rn*KICD cell beads.** A) After 5 h of crosslinking, distinct beads were observed, but they

disintegrated within 24 h after the addition of 50 mM HCO_3^-. B) With a crosslinking duration

of 20 h, bead stability was maintained even after the addition of HCO_3^- for up to 96 h. A 5:6:0.5

ratio of cell wet mass: PVA: SA was employed for droplet-based entrapped cell beads using 4

% H_3BO_3 and 2% $CaCl_2$ crosslinking agents. the entrapped cells were refreshed with BG11

medium, transferred into tightly sealed vials, illuminated with 50 μmol photons m^{-2} s^{-1}. KICD

refers to Syn-*Rn*KICD.

We also observed a significant increase in pH, from ~ 7.4 to 9, in the entrapped cell cultures.

Despite initially supplementing the entrapped cells with a buffered medium, the addition of 50

mM HCO_3 likely contributed to the pH rise. An increase in alkaline pH can destabilize PVA-

SA beads crosslinked by $B(OH)_4^-$ and Ca^{2+} ions by altering the ionization states, weakening

crosslink density, and potentially causing excessive swelling or hydrolysis. This destabilization

may have led to bead disintegration and the release of free cells into the medium.

3.3.1. Effect of pH on the stability of PVA-SA entrapped Syn-*Rn*KICD beads

Our previous results indicated that the stability of PVA-SA hydrogel beads may be compromised under alkaline conditions (Fig. 6). Although *Synechocystis* typically thrives in a pH range of 7.0 to 9.0, maintaining a neutral pH (~7) could be essential for preventing bead disintegration and ensuring cell viability. By adjusting the medium's pH to neutral every 24 h, we observed that the hydrogel beads remained stable without disintegration, even after 96 h of incubation with HCO_3^- (Fig. 7A). These results underscore the critical importance of maintaining a neutral pH for the stability of entrapped cells in the PVA-SA hydrogel beads. Next, we investigated the isobutene-producing capacity of PVA-SA entrapped Syn-*Rn*KICD cells. 120 h after bead formation, the entrapped Syn-*Rn*KICD cells were transferred into tightly sealed vials and placed under illumination at 50 μmol m^{-2} s^{-1} (post adaptation conditions). We monitored isobutene productivity in the headspace of the vials at 24 h intervals. Remarkably, isobutene production was observed at 144 h after bead formation (24 h of post-adaptation conditions), as shown in Fig. 7B. The isobutene productivity reached ~40 mg/g DW after 216 h of post-adaptation under continuous illumination (336 h after bead formation).

Figure 7: Impact of neutral pH on the stability of PVA-SA-entrapped Syn-*Rn*KICD cell beads. A) The phenotypic changes of the entrapped cell beads were observed in the presence of 50 mM HCO_3, with the pH level adjusted to ~7 – 7.4 with 5N HCl every 24 h. B) Entrapped Syn-*Rn*KICD cells were transferred to sealed vials at 120 h after bead formation and illuminated at 50 μmol photons m^{-2} s^{-1} (post-adaptation conditions). Isobutene production was detected in the vial headspace at 144 h (24 h of post-adaptation). The cumulative isobutene production was monitored for 336 h. The cell entrapment was performed using the following formulation: 5:6:0.5 wet cell mass:PVA:SA and crosslinked with 4% H_3BO_3 and 2% $CaCl_2$ for 20 h. After 20 h of crosslinking duration, the entrapped cells were washed with BG11 medium with $NaNO_3$, illuminated with 50 μmol photons m^{-2} s^{-1}. The productivity of isobutene was normalized on the initial DW of the cells. The pH level adjusted to ~7 – 7.4 every 24 h throughout the experiment. KICD refers to Syn-*Rn*KICD.

This observation is particularly notable when compared to the suspension cultures, which exhibited rapid bleaching and cessation of isobutene production within 120 h (Fig. 2). In contrast, entrapped cells within the PVA-SA hydrogel beads did not show any cell bleaching, even after 216 h of incubation in tightly sealed vials, as evidenced by their phenotypic morphology (Fig. 7A). However, after 336 h, partial bead disintegration was observed, but the cells did not exhibit any bleaching phenotype. Such non-bleaching cells likely contributed to the extended period of isobutene production by the entrapped cells (Fig. 7B). These results demonstrate that maintaining a neutral pH in the entrapped cell cultures can enhance the structural integrity of the PVA-SA hydrogel beads for a longer period, which, in turn, could prevent excess growth, cell bleaching and promotes long-term isobutene production. However, despite maintaining a neutral pH over an extended period, the beads showed disintegration and cell suspensions were observed, and distinct beads were no longer visually detectable after 336 h (Fig. 7A). The disintegration of the beads could be attributed to excessive cell growth over time under the supplied nitrogen and HCO_3^- condition. This highlights the eventual limitations of hydrogel bead stability and emphasizes the need to optimize the cell-to-hydrogel ratio to strictly limit cell growth, thereby enhancing bead stability.

3.3.2.   Impact of doubling hydrogel-to-cell ratio on the stability of PVA-SA entrapped cell beads and isobutene productivity

In our continued efforts to improve the stability of PVA-SA-entrapped cell beads and enhance isobutene productivity, we focused on the impact of increasing the hydrogel-to-biomass ratio.

Considering that denser hydrogel may further restrict cell expansion and proliferation, we

adjusted the cell wet mass to hydrogel ratio (cell:PVA:SA) from 5:6:0.5 to 5:12:1, while

maintaining a neutral pH throughout the experiment. This experiment was designed to

investigate how an increase in the hydrogel-to-biomass ratio, combined with neutral pH

adjustments, influences PVA-SA-entrapped cell beads' stability and isobutene productivity in

the presence of HCO_3^- supplementations. Additionally, we examined the entrapped cells under

both N supplied and N deprived conditions. This was motivated by our earlier experiment,

which showed that nitrogen-supplied cell suspensions exhibited significant cell growth but

decreased isobutene production, whereas nitrogen-deprived cells limited growth and increased

isobutene production. Such comparison between N+ and N- conditions can help us better

understand the different growth and production dynamics of PVA-SA-entrapped cells. Before

bead formation, Syn-*Rn*KICD experimental cells were acclimated to N+ and N- cultivation

conditions. The 20-hour crosslinking phase resulted in distinct and stable bead formation, as

evidenced by their phenotypic morphology (Fig. S5). Isobutene production in the entrapped

cells was then monitored at 24 h intervals in the headspace of tightly sealed vials.

The results showed no significant difference in isobutene productivity between N+ and

N- entrapped cells up to 72 h (Fig. 8A). However, at 96 h, N- cells exhibited higher isobutene

production, reaching ~28 mg/g DW, while N+ cells produced ~19 mg/g DW. Although the

beads remained stable up to 144 h under both N+ and N- conditions, at 192 h partial beads'

disintegration was observed under N+ conditions, as shown by their phenotypic morphology

(Fig. 8B). However, the rate of isobutene production continued to be monitored under both

conditions but at a lower level. At 216 h, 25 mM HCO_3^- was added to test if the isobutene

production could be further increased. This addition slightly improved isobutene production in

both N+ and N- cells from 264 h onwards.

At 412 h, the N- entrapped cells still exhibited stable beads and higher isobutene

productivity, reaching ~41 mg/g DW, compared to N+ entrapped cells (~30 mg/g DW) (Fig.

8). No bead disintegration was observed in N- cells at this point (Fig. 8B), suggesting that the

combination of N limitation and entrapment-mediated cell growth limitation did not

compromise bead stability, cell viability and isobutene production. Further under N-

conditions, the entrapped cells not only prolonged their isobutene production but also showed

delayed chlorosis response up to 412 h (Fig. 8B), whereas N starved cell suspensions exhibited

chlorosis within 24 h (Fig. 2B). These observations indicate that entrapping cells in PVA-SA

hydrogel beads could further enhance cell viability and isobutene production for an extended

period under N deprived conditions.

Figure 8: Enhanced bead stability in Syn-*Rn*KICD cells with a 2-fold higher PVA-SA

hydrogel-entrapped cell system under N- conditions. A) The cumulative productivity (lines)

and rate of isobutene (bars) were monitored in the entrapped cells. B) The physical

characteristics of the entrapped cell beads were observed throughout the experiment. The

hydrogel beads were prepared with a 5:12:1 ratio of cell wet mass: PVA:SA using 4% H_3BO_3

and 2% $CaCl_2$ crosslinking. The pH of the cultures was monitored every 24 h and adjusted to

neutral ~7.4. Isobutene production was determined in entrapped cells under N+ and N- culture

conditions, illuminated with 50 μmol photons m^{-2} s^{-1}, and supplemented with 50 mM HCO_3 at

day 1, and additional 25 mM HCO_3 was provided at 216 h. The isobutene productivity was

normalized on the initial DW of the cells used for entrapment. These results are the average of

3 independent measurements from 3 replicates (±SD).

On the other hand, bead disintegration was clearly visible under N supplied conditions

starting at 192 h, with gradual leakage of cells into the medium, as evidenced in Fig. 8B.

Despite this disintegration, isobutene production from the entrapped N+ cells continued for up

to 412 h (Fig. 8A), significantly longer than in suspension cells, which only produced isobutene

up to 120 h (Fig. 2A). However, the disintegration of beads eventually led to cell suspension

and subsequent cell bleaching by 450 h (Fig. 8B). This disintegration of beads could be due to

excessive cell growth, as N+ cells tend to exhibit strong growth responses in the presence of

HCO_3^- supplementation, as seen in the suspension cells (Fig. 2A). These results indicate that

while the 2-fold increase in hydrogel-to-cell ratio and maintaining neutral pH prolonged bead

stability, the cells under N supplied conditions still experienced delayed bleaching due to

delayed beads' disintegration to cell suspension. Therefore, further optimization of the cell

entrapment procedure is needed, particularly for cells under N supplementation, to achieve

prolonged isobutene production in this case.

3.3.3.   Impact of a sixfold increase in hydrogel-to-cell ratio on the stability of PVA-SA

entrapped cell beads and isobutene productivity

Though increasing 2x higher hydrogel to biomass improved beads' stability for a certain

period, our attempts to effectively limit cell growth and prolong isobutene production under N

and HCO_3 conditions were less effective, as seen in bead disintegration, cell leakage, gradual

bleaching, and relatively lower isobutene production (Fig. 8). To address this issue, we further

experimented with a six-fold increase in the hydrogel-to-cell ratio, changing it from 5:12:1 to

5:36:3 (cell:PVA:SA), aiming to limit cell growth and prevent bead disintegration. This

modification in the cell-to-hydrogel ratio led to visually stable and distinct beads for Syn-

*Rn*KICD and Syn-F336V cells (Fig. 9).

Up to 196 h, there was no significant difference in isobutene production observed in

the entrapped Syn F336V cell beads under N supplied and N starved conditions (Fig. 10A).

Interestingly, at 216 h, a notable increase in isobutene production (~67 mg/g DW) was observed

in Syn-F336V N+ cells. Conversely, no further increase was observed in the N- cells, which

reached a maximum productivity of ~57 mg/g DW. On the other hand, Syn-F336V N+ cells

continued to increase their isobutene production, achieving ~91 mg/g DW at 408 h. A similar trend was observed in entrapped Syn-*Rn*KICD cells under the same production conditions, with only slight differences in isobutene productivity between N+ (~38 mg/g DW) and N- (~45 mg/g DW) cells up to 216 h (Fig. 10B). After 216 h, no further increase in productivity was observed in N- cells. At this point, a significant increase in isobutene production was observed in N+ Syn-*Rn*KICD cells, which was continuing to rise until 408 h and reached ~54 mg/g DW. These results showed that both Syn-F336V and Syn-*Rn*KICD strains exhibited higher isobutene production under N supplied conditions than N starved conditions after 216 h (Fig. 10A and B).

Figure 9: Extended bead stability in Syn-F336V and Syn-*Rn*KICD cells with a 6-fold higher PVA-SA hydrogel-entrapped cell system under N+ and N- conditions. The hydrogel beads were prepared with the following ratio between the cell wet mass: PVA: SA was 5:36:3 using H_3BO_3 and $CaCl_2$ crosslinking. pH was adjusted to neutral throughout the experiments. The entrapped cells were illuminated under 50 μmol photons m^{-2} s^{-1} with addition of 50 mM HCO_3. A similar amount of initial biomass was used across all cultures. 336 refers to Syn-F336V, and KICD refers to Syn-*Rn*KICD.

To further enhance isobutene production, we refreshed the cultures with BG11 medium and added 25 mM HCO_3^- at 408 h. Under N+ conditions, the entrapped cell beads remained

stable without disintegration (Fig. 9) and continued to produce more isobutene than N-deprived cells for nearly 744 h (Fig. 10). Although the beads remained stable throughout this period, as shown by their phenotypic morphology, only a slight increase in isobutene production was observed in both entrapped Syn-F336V and Syn-*Rn*KICD cells, peaking at ~94 mg/g DW and ~58 mg/g DW, respectively, at 744 h (Fig. 10). These findings suggest that the six-fold increase in hydrogel-to-biomass ratio further improved bead stability and isobutene productivity by restricting cell growth, even under N+ conditions where cells tend to grow more rapidly and are more prone to cell bleaching.

Figure 10: Extended bead stability and isobutene productivity in Syn-F336V and Syn-*Rn*KICD cells with a 6-fold higher PVA-SA hydrogel-entrapped cell system under N+

conditions. The cumulative productivity (lines) and rate of isobutene (bars) in the entrapped A) Syn-F336V and B) Syn-RnKICD cells under N+ and N- culture conditions. The beads were prepared with the ratio: 5:36:3 for the wet cell weight: PVA: SA, crosslinked with H_3BO_3 and $CaCl_2$. pH was adjusted to neutral throughout the experiments. A similar amount of initial cell mass was used across all cultures. The productivity of isobutene was normalized to the initial DW of the cells. Inlets in Figures A and B showed an enlarged view of the prolonged production of isobutene. The cells were illuminated with 50 μmol photons $m^{-2}\,s^{-1}$ with addition 50 mM HCO_3 at the start of each experiment, additional 25 mM $NaHCO_3$ was added at 408 h. These results are the average of 3 independent measurements from 3 replicates (±SD). 336 refers to Syn-F336V, and KICD to Syn-RnKICD.

To further demonstrate the effectiveness of entrapped cells' isobutene producing capacity under N+ conditions, we compared suspension and entrapped cells with similar initial biomass loading under the same production conditions. Although up to 144 h no significant difference in isobutene productivity was observed between suspension and entrapped N+ cells (Fig. S6A), the suspension cells exhibited early bleaching as it was witnessed from 144 h onward (Fig. S6B), leading to no further increase in isobutene production. In contrast, the entrapped cells showed no signs of bleaching and continued to increase isobutene production up to 744 h, achieving productivity above 1.5- times higher than their suspension-cell counterparts (Fig. S6A). These results further confirm that cell viability and isobutene producing capacity in PVA-SA entrapped cells were enhanced as compared to the corresponding suspension cells. Further, when using entrapped cell beads to limit growth under N+ and N- conditions, we observed that N+ cells exhibited higher isobutene production than N- cells which is contrast to what we observed in the suspension cultures.

Therefore, we suggest that the nitrogen supplied entrapped cells could effectively distribute the carbon flux between BCAA biosynthesis and isobutene production, allowing for the production of essential amino acids required for cellular maintenance while achieving high isobutene output in the cells since cell growth can be limited in this case. The Syn-F336V cells showed higher isobutene production than Syn-RnKICD cells (Fig.10A and B), indicating that the level of substrate available for isobutene formation is lower in the Syn-RnKICD cells as compared to the Syn-F336V cells which is in line to what has been shown (Schumann et al., 2024). In contrast, under N deprived conditions, both strains experience a more pronounced metabolic shift, with carbon being redirected primarily toward isobutene production at the

expense of BCAA biosynthesis. The inability to produce sufficient BCAAs under nitrogen-limited conditions impairs essential cellular functions. As a result, at ~240 h, the strains under N- conditions experienced a sharp decline in isobutene production (Fig. 10), which could be due to reduced cell viability, correlating with their phenotypic morphology (Fig. 9), and probably due to a lack of essential amino acids. These results highlight three competing metabolic processes in the isobutene producing *Synechocystis* cells; one supporting cell growth and BCAA biosynthesis, one driving isobutene production and one producing homogentisate from HPP (Fig.11).

Figure 11: Schematic representation of *Rn*KICD's dual role in the endogenous homogentisate and heterologous isobutene biosynthesis pathways. *Rn*KICD converts α-ketoisocaproate (KIC) into isobutene and ρ-hydroxyphenylpyruvate (HPP) into homogentisate. The black arrows indicate the native pathways, while the blue arrows represent the non-native pathways.

In the nitrogen supplied entrapped cells, both the Syn-F336V and the Syn-*Rn*KICD could maintain a balance between these three processes, resulting in prolonged cell viability and higher isobutene production, up to ~96 mg/g DW. However, N limitation forces the cells to prioritize isobutene synthesis, at the cost of cell viability. However, this reprogramming of metabolism over time reduced productivity due to the depletion of BCAAs. This underscores

the importance of nitrogen in regulating the balance between growth and isobutene production in *Synechocystis.*

4. Conclusions

This study demonstrated that optimizing varying environmental conditions, including light intensity and bicarbonate and nitrate concentrations, significantly enhanced isobutene production in modified *Synechocystis* cells. Under these conditions, the production rate improved over 10-fold compared to our previous report (Mustila et al., 2021). We further demonstrated that PVA-SA hydrogel-based cell entrapment was highly effective, sustaining enhanced isobutene production for nearly a month. PVA-SA hydrogel beads, crosslinked with $B(OH)_4^-$ and Ca^{2+} ions, provided stability in HCO_3-rich environments, enabling long-term production. Optimization of key factors, including crosslinking time, pH balance, and the biomass-to-hydrogel ratio, improved bead durability, restricted cell growth, and prevented cell bleaching, redirecting metabolism toward isobutene production, particularly under bicarbonate and nitrate supply. Nitrogen availability emerged as a critical factor, with entrapped Syn-*Rn*KICD and Syn-F336V cell beads showing significantly higher isobutene production compared to nitrogen-limited conditions. This underscores the importance of nitrogen in balancing BCAA biosynthesis and maximizing carbon flow toward isobutene production in entrapped cells. Overall, the combination of PVA-SA hydrogel-based cell entrapment and precise nutrient management presents an effective strategy for optimizing isobutene production, providing a promising approach for sustainable bio-based chemical production.

5. CRediT authorship contribution statement

Sindhujaa Vajravel: Writing – original draft, Methodology, Investigation, Formal analysis, Data curation, Conceptualization. Sanjukta Aravind: Writing – original draft, Methodology, Investigation, Formal analysis, Data curation. Karin Stensjö: Writing – review & editing, Project administration, Conceptualization.

Acknowledgements

We greatly acknowledge FORMAS—A Swedish Research Council for Sustainable Development (project no. 2021-01669), The Carl Trygger Foundation for Scientific Research (CTS:23:2849), and the Swedish Energy Agency (project no. 52576-1).

Declaration of competing interest

The authors declare no competing financial interest.

Appendix A. Supplementary data

Supplementary material

Data availability

Data will be made available on request.

References

1. Aboim, J. B., Oliveira, D. T. D., Mescouto, V. A. D., Dos Reis, A. S., da Rocha Filho,
G. N., Santos, A. V., ... & do Nascimento, L. A. S. (2019). Optimization of light
intensity and $NaNO_3$ concentration in Amazon cyanobacteria cultivation to produce
biodiesel. Molecules, 24(12), 2326. https://doi.org/10.3390/molecules24122326

2. Allen, M. M., Law, A., & Evans, E. H. (1990). Control of photosynthesis during
nitrogen depletion and recovery in a non-nitrogen-fixing cyanobacterium. Archives of
Microbiology, 153, 428-431. https://doi.org/10.1007/BF00248422

3. Allen, M. M., & Smith, A. J. (1969). Nitrogen chlorosis in blue-green algae. Archiv für
Mikrobiologie, 69(2), 114–120. https://doi.org/10.1007/BF00409755

4. Angermayr, S. A., Rovira, A. G., & Hellingwerf, K. J. (2015). Metabolic engineering
of cyanobacteria for the synthesis of commodity products. Trends in Biotechnology,
33(6), 352-361. http://dx.doi.org/10.1016/j.tibtech.2015.03.009

5. Badger, M. R., & Price, G. D. (2003). CO_2 concentrating mechanisms in cyanobacteria:
molecular components, their diversity and evolution. Journal of Experimental Botany,
54(383), 609-622. https://doi.org/10.1093/jxb/erg076

6. Bolto, B., Tran, T., Hoang, M., & Xie, Z. (2009). Crosslinked poly (vinyl alcohol)
membranes. Progress in Polymer Science, 34(9), 969-981.
https://doi.org/10.1016/j.progpolymsci.2009.05.003

7. Brouers, M., & Hall, D. O. (1986). Ammonia and hydrogen production by immobilized
cyanobacteria. Journal of Biotechnology, 3(5-6), 307-321.
https://doi.org/10.1016/0168-1656(86)90012-X

8. Carroll, A. L., Case, A. E., Zhang, A., & Atsumi, S. (2018). Metabolic engineering
tools in model cyanobacteria. Metabolic Engineering, 50, 47-56.
https://doi.org/10.1016/j.ymben.2018.03.014

9. Cheng, A. T. Y., & Rodriguez, F. (1981). Mechanical properties of borate crosslinked poly (vinyl alcohol) gels. Journal of Applied Polymer Science, 26(11), 3895-3908. https://doi.org/10.1002/app.1981.070261134

10. Chisti, Y. (2007). Biodiesel from microalgae. Biotechnology Advances, 25(3), 294-306. https://doi.org/10.1016/j.biotechadv.2007.02.001

11. Choe, S., You, S., Park, K., Kim, Y., Park, J., Cho, Y., ... & Myung, J. (2024). Boric acid-crosslinked poly (vinyl alcohol): biodegradable, biocompatible, robust, and high-barrier paper coating. Green Chemistry, 26, 8230-8241. DOI: 10.1039/D4GC00618F

12. Ducat, D. C., Way, J. C., & Silver, P. A. (2011). Engineering cyanobacteria to generate high-value products. Trends in Biotechnology, 29(2), 95-103. DOI: 10.1016/j.tibtech.2010.12.003

13. Farrokh, P., Sheikhpour, M., Kasaeian, A., Asadi, H., & Bavandi, R. (2019). Cyanobacteria as an eco-friendly resource for biofuel production: a critical review. Biotechnology Progress, 35(5), e2835. https://doi.org/10.1002/btpr.2835

14. Forchhammer, K., & Schwarz, R. (2019). Nitrogen chlorosis in unicellular cyanobacteria–a developmental program for surviving nitrogen deprivation. Environmental Microbiology, 21(4), 1173-1184. https://doi.org/10.1111/1462-2920.14447

15. Forchhammer, K., & Selim, K. A. (2020). Carbon/nitrogen homeostasis control in cyanobacteria. FEMS Microbiology Reviews, 44(1), 33-53. https://doi.org/10.1093/femsre/fuz025

16. Gohil, J. M., Bhattacharya, A., & Ray, P. (2006). Studies on the crosslinking of poly (vinyl alcohol). Journal of Polymer Research, 13, 161-169. https://doi.org/10.1111/1462-2920.14447

17. Gordillo, F. J., Goutx, M., Figueroa, F. L., & Niell, F. X. (1998). Effects of light intensity, CO_2 and nitrogen supply on lipid class composition of Dunaliella viridis. Journal of Applied Phycology, 10, 135-144. https://doi.org/10.1023/A:1008067022973

18. Hauge, K., Bergene, E., Chen, D., Fredriksen, G. R., & Holmen, A. (2005). Oligomerization of isobutene over solid acid catalysts. Catalysis Today, 100(3-4), 463-466. https://doi.org/10.1016/j.cattod.2004.08.017

19. Hua, S., Ma, H., Li, X., Yang, H., & Wang, A. (2010). pH-sensitive sodium alginate/poly (vinyl alcohol) hydrogel beads prepared by combined Ca^{2+} crosslinking and freeze-thawing cycles for controlled release of diclofenac sodium. International

Journal   of   Biological   Macromolecules,   46(5),   517-523.
https://doi.org/10.1016/j.ijbiomac.2010.03.004

20. Hulnik, M., Trofimuk, D., Nikishau, P. A., Kiliclar, H. C., Kiskan, B., & Kostjuk, S. V.
(2023). Visible-light-induced cationic polymerization of isobutylene: a route toward
the synthesis of end-functional polyisobutylene. ACS Macro Letters, 12(8), 1125-1131.
https://doi.org/10.1021/acsmacrolett.3c00384

21. Kim, H. W., Vannela, R., Zhou, C., & Rittmann, B. E. (2011). Nutrient acquisition and
limitation for the photoautotrophic growth of *Synechocystis* sp. PCC6803 as a
renewable biomass source. Biotechnology and Bioengineering, 108(2), 277-285.
https://doi.org/10.1002/bit.22928

22. Kosourov, S., Leino, H., Murukesan, G., Lynch, F., Sivonen, K., Tsygankov, A. A., ...
& Allahverdiyeva, Y. (2014). Hydrogen photoproduction by immobilized N_2-fixing
cyanobacteria: understanding the role of the uptake hydrogenase in the long-term
process. Applied and Environmental Microbiology, 80(18), 5807-5817.
https://doi.org/10.1128/AEM.01776-14

23. Kostjuk, S. V. (2015). Recent progress in the Lewis acid co-initiated cationic
polymerization of isobutylene and 1,3-dienes. RSC Advances, 5(17), 13125–13144.
DOI: 10.1039/C4RA15313H

24. Lichtenthaler, H. K. (1987). [34] Chlorophylls and carotenoids: pigments of
photosynthetic biomembranes. In Methods in Enzymology (Vol. 148, pp. 350-382).
Academic Press. https://doi.org/10.1016/0076-6879(87)48036-1

25. Lindberg, P., Park, S., & Melis, A. (2010). Engineering a platform for photosynthetic
isoprene production in cyanobacteria, using *Synechocystis* as the model organism.
Metabolic Engineering, 12(1), 70-79. https://doi.org/10.1016/j.ymben.2009.10.001

26. Markou, G., Vandamme, D., & Muylaert, K. (2014). Microalgal and cyanobacterial
cultivation: the supply of nutrients. Water Research, 65, 186-202.
https://doi.org/10.1016/j.watres.2014.07.025

27. Melis, A. (2009). Solar energy conversion efficiencies in photosynthesis: minimizing
the chlorophyll antennae to maximize efficiency. Plant Science, 177(4), 272-280.
https://doi.org/10.1016/j.plantsci.2009.06.005

28. Miller, A. G., & Colman, B. (1980). Evidence for HCO_3^- transport by the blue-green
alga (cyanobacterium) Coccochloris peniocystis. Plant Physiology, 65(2), 397-402.
https://doi.org/10.1104/pp.65.2.397

29. Miller, A. G., Espie, G. S., & Canvin, D. T. (1990). Physiological aspects of CO_2 and HCO_3^- transport by cyanobacteria: a review. Canadian Journal of Botany, 68(6), 1291-1302. https://doi.org/10.1139/b90-165

30. Mustila, H., Kugler, A., & Stensjö, K. (2021). Isobutene production in *Synechocystis* sp. PCC 6803 by introducing α-ketoisocaproate dioxygenase from Rattus norvegicus. Metabolic Engineering Communications, 12, e00163. https://doi.org/10.1016/j.mec.2021.e00163

31. Nicholas, C. P. (2017). Applications of light olefin oligomerization to the production of fuels and chemicals. Applied Catalysis A: General, 543, 82-97. https://doi.org/10.1016/j.apcata.2017.06.011

32. Pedruzi, G. O., Amorim, M. L., Santos, R. R., Martins, M. A., & Vaz, M. G. (2019). Biomass accumulation-influencing factors in microalgae farms. Revista Brasileira de Engenharia Agrícola e Ambiental, 24, 134-139. https://doi.org/10.1590/1807-1929/agriambi.v24n2p134-139

33. Peters, M. W., & Taylor, J. D. (2013). U.S. Patent No. 8,373,012. Washington, DC: U.S. Patent and Trademark Office.

34. Rana, A., Gomes, L. C., Rodrigues, J. S., Yacout, D. M., Arrou-Vignod, H., Sjölander, J., ... & Ottosson, H. (2022). A combined photobiological–photochemical route to C 10 cycloalkane jet fuels from carbon dioxide via isoprene. Green Chemistry, 24(24), 9602-9619. DOI: 10.1039/D2GC03272D

35. Rebolledo-Leiva, R., Moreira, M. T., & González-García, S. (2022). Offsetting the environmental impacts of single or multi-product biorefineries from wheat straw. Bioresource Technology, 361, 127698. https://doi.org/10.1016/j.biortech.2022.127698

36. Rissanen, V., Vajravel, S., Kosourov, S., Arola, S., Kontturi, E., Allahverdiyeva, Y., & Tammelin, T. (2021). Nanocellulose-based mechanically stable immobilization matrix for enhanced ethylene production: a framework for photosynthetic solid-state cell factories. Green Chemistry, 23(10), 3715-3724. DOI: 10.1039/D1GC00502B

37. Rodrigues, J. S., Kovács, L., Lukeš, M., Höper, R., Steuer, R., Červený, J., ... & Zavřel, T. (2023). Characterizing isoprene production in cyanobacteria–insights into the effects of light, temperature, and isoprene on *Synechocystis* sp. PCC 6803. Bioresource Technology, 380, 129068. https://doi.org/10.1016/j.biortech.2023.129068

38. Rossoni, L., Hall, S. J., Eastham, G., Licence, P., & Stephens, G. (2015). The putative mevalonate diphosphate decarboxylase from Picrophilus torridus is in reality a mevalonate-3-kinase with high potential for bioproduction of isobutene. Applied and

1009   Environmental Microbiology, 81(7), 2625-2634. https://doi.org/10.1128/AEM.04033-
1010   14

39. Sabourin, P. J., & Bieber, L. L. (1982). Purification and characterization of an alpha-ketoisocaproate oxygenase of rat liver. *Journal of Biological Chemistry*, *257*(13), 7460-7467. DOI: 10.1016/S0021-9258(18)34400-4

40. Santos-Merino, M., Singh, A. K., & Ducat, D. C. (2019). New applications of synthetic biology tools for cyanobacterial metabolic engineering. Frontiers in Bioengineering and Biotechnology, 7, 33. https://doi.org/10.3389/fbioe.2019.00033

41. Sauer, J., Schreiber, U., Schmid, R., Völker, U., & Forchhammer, K. (2001). Nitrogen starvation-induced chlorosis in *Synechococcus* PCC 7942. Low-level photosynthesis as a mechanism of long-term survival. Plant Physiology, 126(1), 233-243. https://doi.org/10.1104/pp.126.1.233

42. Seibert, M., Kosourov, S. N., He, M., & Allahverdiyeva, Y. (2018). Immobilization of Microalgae as a Tool for Efficient Light Utilization in H_2 Production and Other Biotechnology Applications (No. NREL/CH-2A00-72362). National Renewable Energy Lab. (NREL), Golden, CO (United States). https://doi.org/10.1039/9781849737128-00355

43. Simpson, N. E., Stabler, C. L., Simpson, C. P., Sambanis, A., & Constantinidis, I. (2004). The role of the $CaCl_2$–guluronic acid interaction on alginate encapsulated βTC3 cells. Biomaterials, 25(13), 2603-2610. https://doi.org/10.1016/j.biomaterials.2003.09.046

44. Spat, P., Klotz, A., Rexroth, S., Macek, B., & Forchhammer, K. (2018). Chlorosis as a developmental program in cyanobacteria: the proteomic fundament for survival and awakening. Molecular & Cellular Proteomics, 17(9), 1650-1669. DOI: 10.1074/mcp.RA118.000699

45. Schumann, C., Kugler, A., Berggren G., Land, H., Blikstad, C., & Stensjö, K. (2024). Structure-guided engineering of α-ketoisocaproate dioxygenase increases isobutene production in *Synechocystis* sp. PCC 6803. bioRxiv 2024-12. https://doi.org/10.1101/2024.12.20.629387

46. Taylor, J. D., Jenni, M. M., & Peters, M. W. (2010). Dehydration of fermented isobutanol for the production of renewable chemicals and fuels. Topics in Catalysis, 53, 1224-1230. https://doi.org/10.1007/s11244-010-9567-8

47. Tóth, G. S., Siitonen, V., Nikkanen, L., Sovic, L., Kallio, P., Kourist, R., ... & Allahverdiyeva, Y. (2022). Photosynthetically produced sucrose by immobilized

Synechocystis sp. PCC 6803 drives biotransformation in *E. coli.* Biotechnology for Biofuels and Bioproducts, 15(1), 146. https://doi.org/10.1186/s13068-022-02248-1

48. Van Alphen, P., Abedini Najafabadi, H., Branco dos Santos, F., & Hellingwerf, K. J. (2018). Increasing the photoautotrophic growth rate of *Synechocystis* sp. PCC 6803 by identifying the limitations of its cultivation. Biotechnology Journal, 13(8), 1700764. https://doi.org/10.1002/biot.201700764

49. van Leeuwen, B. N., van der Wulp, A. M., Duijnstee, I., van Maris, A. J., & Straathof, A. J. (2012). Fermentative production of isobutene. Applied Microbiology and Biotechnology, 93, 1377-1387. https://doi.org/10.1007/s00253-011-3853-7

50. Vajravel, S., Sirin, S., Kosourov, S., & Allahverdiyeva, Y. (2020). Towards sustainable ethylene production with cyanobacterial artificial biofilms. Green Chemistry, 22(19), 6404-6414. DOI: 10.1039/D0GC01830A

51. Valsami, E. A., Pateraki, A., Melis, A., & Ghanotakis, D. F. (2021). Heterologous β-phellandrene production by alginate immobilized *Synechocystis* sp. PCC 6803. Journal of Applied Phycology, 33, 2157-2168. https://doi.org/10.1007/s10811-021-02451-x

52. Wilson, J., Gering, S., Pinard, J., Lucas, R., & Briggs, B. R. (2018). Bio-production of gaseous alkenes: ethylene, isoprene, isobutene. Biotechnology for Biofuels, 11, 1-11. https://doi.org/10.1186/s13068-018-1230-9

53. Wu, J., Liu, R., Zhang, W., Zhong, Q., Lei, Y., & Huang, L. (2024). Enhancement of mechanical behaviors of the 3D-printed polyvinyl alcohol–based scaffold by boric acid crosslinking. Journal of Polymer Engineering, 44(4), 251-262. https://doi.org/10.1515/polyeng-2023-0305

54. Wu, K. Y. A., & Wisecarver, K. D. (1992). Cell immobilization using PVA crosslinked with boric acid. Biotechnology and Bioengineering, 39(4), 447-449. https://doi.org/10.1002/bit.260390411

55. Zhong, Y., Lin, Q., Yu, H., Shao, L., Cui, X., Pang, Q., ... & Hou, R. (2024). Construction methods and biomedical applications of PVA-based hydrogels. Frontiers in Chemistry, 12, 1376799. https://doi.org/10.3389/fchem.2024.1376799

56. Zhu, X. G., Long, S. P., & Ort, D. R. (2008). What is the maximum efficiency with which photosynthesis can convert solar energy into biomass? Current Opinion in Biotechnology, 19(2), 153-159. https://doi.org/10.1016/j.copbio.2008.02.004