Chemical diversity promotes ecosystem function

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Abstract

Nature benefits people in many ways that rely on how microorganisms and abiotic processes

interact with organic matter to influence ecosystem function (EF). Despite many studies

evaluating how biodiversity affects EF, most variation in EF remains unexplained. Here, we

tested how the variety of organic compounds in the environment, termed chemodiversity,

influences EF. Using a laboratory experiment and 101-lake survey spanning Europe, we

discovered that chemodiversity, not biodiversity, consistently explained variation in multiple

functions related to carbon cycling and ecosystem metabolism. We attributed positive

chemodiversity-ecosystem multifunctionality relationships to the accumulation of compounds

that could be biochemically transformed in more diverse ways at higher chemodiversity. We

estimated chemodiversity will decline by 2100 due to changes in organic matter sources,

causing European lakes to emit, on average, 1.6-times more carbon. These findings reframe

our understanding of ecosystem functioning and suggest chemodiversity is key to sustain the

benefits that nature delivers to people.

Introduction

Nature benefits society such as by regulating the climate and providing food and clean water[1,2]. These services are in part delivered by the multiple functions undertaken by microorganisms that break down and transform different organic substrates[3]. For example, as microbes assimilate detritus and increase in biomass[4], they support the growth of higher trophic levels[5], as well as modify organic compounds into forms that sequester carbon longer in the environment and thereby contribute to climate regulation[6]. Consequently, ecosystem function is enhanced by both a greater diversity of metabolic capabilities in microbial communities[3] and the environmental conditions that favour this biodiversity[7]. However, large amounts of variation in the capacity of ecosystems to provide multiple functions simultaneously, hereafter termed ecosystem multifunctionality[1], remain unexplained. Many large-scale studies have found that biodiversity only explains between 2 to 20% of the variation in ecosystem multifunctionality[3,8,9], though there are some exceptions[10].

The organic substrates that microbes use to influence biological, geochemical, and physical processes within ecosystems are an overlooked source of variation in past studies that predict ecosystem multifunctionality[6,11,12]. Natural organic matter in aquatic and terrestrial ecosystems contains tens of thousands (and likely millions) of unique molecules of varying origin and composition, making it one of the most complex mixtures on Earth[13]. This variety of molecules within organic matter and their characteristics, such as their bioavailability and stoichiometry, are together referred to as chemodiversity[14]. Chemodiversity can be described by measuring the diversity and abundance of molecular formulae and their characteristics, similar to approaches for studying biological mixtures of species[15]. Despite many surveys of chemodiversity in waters and soils[16,17], there has been no attempt to ask how chemodiversity

influences ecosystem function in nature as compared with biodiversity and other environmental

variables.

We hypothesised that chemodiversity influences ecosystem multifunctionality in two main

ways. First, the chemical properties of organic compounds (e.g. structure, elemental

composition, and biological and photochemical reactivity) determine the biotic and abiotic

pathways by which they are degraded and transformed[18,19]. A greater number of compounds

can therefore directly promote ecosystem multifunctionality by increasing the number of ways

that microbial communities and abiotic processes transform organic matter[13]. For example,

adding compounds to a mixture can increase the variety of unique chemical properties and thus

the number of organic compounds that are degraded, subsequently increasing microbial

biomass. Second, chemodiversity can indirectly influence ecosystem multifunctionality by

changing the physical environment. For example, because compounds vary in their ability to

absorb light[20], the addition of light-absorbing compounds to aquatic ecosystems can reduce

photosynthetic rates and limit heat penetration into deeper waters[21,22]. These changes can result

in warmer and more nutrient-rich surface waters that accelerate microbial activity and carbon

cycling between the land surface and atmosphere[6,23]. For these reasons, the number and

characteristics of organic compounds in the environment can provide an integrative measure

across multiple ecosystem functions whose importance can be compared directly with other

variables traditionally used to explain ecosystem multifunctionality, including biodiversity[9]

and climate[3,8]. However, whether chemodiversity explains a relatively large amount of

variation in ecosystem multifunctionality remains untested.

Here we tested the hypothesis that chemodiversity promotes ecosystem multifunctionality by

combining a manipulative lab experiment with the largest field survey of these variables to

date. We focused on lakes unlike past studies that tested associations between biodiversity and

ecosystem multifunctionality in terrestrial ecosystems[8]. Despite covering only 3.7% of the

Earth's surface, the functions of lakes, including carbon cycling and food web production[23,24],

are disproportionately important to many services valued by society, such as climate regulation

and food provisioning. We first used a lab experiment to test how changing chemodiversity

changes ecosystem multifunctionality, and then tested the importance of this association in

natural lakes that span 30° of latitude as compared with known drivers of ecosystem

multifunctionality, namely biodiversity and physicochemical variables. We defined

multifunctionality from a human perspective that values services like climate regulation and

food web productivity, and measured chemodiversity in surface waters using ultrahigh-

resolution Fourier-transform ion cyclotron resonance mass spectrometry (FT-ICR MS) to

identify the diversity of compounds involved in biotic and abiotic reactions. Although the

analytical window of FT-ICR MS can omit low (<150 m/z) and high-molecular-mass (>1,000

m/z) compounds that contribute to biotic and abiotic reactions[25], it is the most sensitive tool

available for characterising chemodiversity and captures most of the organic compounds that

interact with microbes[26]. We then used the field survey and current theories that synthesise how

different types of chemodiversity originate[13,27] to develop a general understanding of how and

where chemodiversity matters for ecosystem multifunctionality. As lakes are more abundant at

northern latitudes[28] that are undergoing large changes in organic matter inputs, we use our

results to forecast the consequences of environmental change for chemodiversity and the

potential impacts on the benefits that lakes provide to society.

Chemodiversity is the main driver of ecosystem multifunctionality

To test if ecosystem multifunctionality increases with increasing chemodiversity, we first

exposed four levels of biodiversity (1, 2, 3, and 6 bacterial species) under controlled laboratory

conditions to three environmental sources of dissolved organic matter (DOM). The three DOM

sources naturally varied in chemodiversity (1090, 1745, and 2564 formulae), which closely

correlated with indices summarising potential biogeochemical functions, such as defined from

the bioavailability, reactivity, and molecular mass of DOM[29] (Fig. S1). The molecular

composition of the three DOM sources was also largely nested (Fig. S2), whereby increasing

chemodiversity primarily consisted of the addition of aromatic-like formulae (Table S1). Our

fully factorial design allowed us to compare the importance of chemodiversity versus

biodiversity for predicting ecosystem multifunctionality from the main effects of

chemodiversity and biodiversity and their interaction. Ecosystem multifunctionality was

defined using changes in cell abundances, dissolved organic carbon (DOC) concentrations, and

total dissolved nitrogen (TDN) concentrations over a 7-day period. We chose these functions

because they are important measures of food web productivity and nutrient cycling[30]. Because

high microbial activity can increase both DOM and TDN concentration by releasing exudates

or transforming particulate organic matter into dissolved compounds, as well as reduce DOM

and TDN concentration by assimilating and respiring organic compounds[31], we considered the

absolute change in DOC and TDN concentration as an indicator of microbial activity. We found

that these three functions were largely orthogonal to each other in our closed system, thereby

representing complementary axes of ecosystem multifunctionality (Fig. S3). We then summed

standardized values (z-scores) of the change in each of the three functions between day 0 and

day 7 into a single index of ecosystem multifunctionality as is common in ecology[1,3]. Larger

values of this index reflect higher ecosystem multifunctionality. We tested the main and

interactive effects of chemodiversity and biodiversity as factors on multifunctionality using a

linear model without any other variables, i.e. only accounting for the experimental design (see

Methods).

We found strong experimental support for our hypothesis that chemodiversity enhances

ecosystem multifunctionality. Increasing chemodiversity led to higher values of both the

multifunctionality index (Fig. 1b) and each individual ecosystem function, with one exception

(Fig. S4). Over the range of observed chemodiversity (1090 to 2563 formulae),

multifunctionality was estimated to increase from a mean (95% confidence interval [CI]) of -

0.90 (-1.49 – -0.32) to a mean of 1.44 (0.10 – 2.77) (Fig. 1a). This mean increase corresponded

to an absolute change of 260% (107 – 966%). In support of these findings, DOM was

increasingly transformed with increasing chemodiversity. We specifically found ten-times as

many compounds were consumed at the highest compared to the lowest level of

chemodiversity, whereas the percentage of compounds that were produced in the incubations,

such as might arise from cell lysis, was relatively stable across chemodiversity treatments (Fig.

S5). Importantly, we found no evidence that chemodiversity was an outcome of greater

multifunctionality. The change in chemodiversity, that is, the difference between

chemodiversity measured at the end of the incubation in control incubations with no microbes

and chemodiversity measured at the end of the incubation in the presence of microbes, was not

correlated with multifunctionality (ρ=-0.22, p=0.203). Chemodiversity alone explained 24% of

the variation in multifunctionality, which is greater than the effect of biodiversity in many other

studies of ecosystem multifunctionality[3,8–10].

Neither ecosystem multifunctionality (Fig. 1b) nor any individual function increased

consistently with biodiversity in our experiment (Fig. S4). Although the two species

community had higher multifunctionality, its taxa were included in the six species community

(Fig. 1b). Thus, if increasing biodiversity increased ecosystem function, we would expect the

six species community to have an equal or higher multifunctionality than the two species

community. However, we observed lower multifunctionality with six species than with two

species (Fig. 1b). Adding biodiversity can sometimes have no added benefit for ecosystem

function, potentially because of competition and redundancy in the functions undertaken by

different species[32,33]. In our experiment, we found that the two species community grew faster

and caused larger absolute change in DOC and TDN than when its members were grown

alongside other taxa in the six species community (Fig. S4), suggesting competitive inhibition

in the latter. Most microbial taxa negatively interact when grown together[34], which reduces

their growth relative to expectations from when taxa are grown by themselves[35].

We also observed an interaction between chemodiversity and biodiversity that further

emphasizes the importance of chemodiversity for ecosystem multifunctionality. Higher

biodiversity increased ecosystem multifunctionality only at the highest chemodiversity level

(Fig. S6). These results suggest that without a sufficient diversity of compounds for microbes

to use, increasing biodiversity alone cannot sustain a higher ecosystem multifunctionality. By

contrast, in environments with low biodiversity and low chemodiversity, increasing

chemodiversity alone can increase ecosystem multifunctionality by enabling the species

already present to perform more functions. Because ecosystem multifunctionality-biodiversity

relationships[3,8–10,32,36–38] are mostly studied in terrestrial environments, where chemodiversity

is far higher than in aquatic environments[39], the importance of chemodiversity may have been

overlooked.

A potential criticism of our experiment is that because we observed an increase in absolute

carbon and nitrogen concentration, changes in ecosystem multifunctionality may have been

due to cell lysis rather than microbes transforming chemodiversity. However, both carbon and

nitrogen could have come from particulate organic matter as well as nitrogen fixation by some

of our study species[40,41]. In support of the former explanation, between 1.6 to 10.6% of the total

organic carbon in each treatment was in particulate rather than dissolved form. We also found

evidence that nitrogen was incorporated into DOM from other pools than cell lysis.

Specifically, we found that a higher proportion of N-containing compounds were produced at

higher chemodiversity, and this result could not have been due simply to cell lysis because

there was a higher overall consumption of compounds (Fig. S5). Taken together, our results

suggest that increasing chemodiversity increases ecosystem multifunctionality by enabling

microbes to use and transform substrates in more ways than is achieved by increasing

biodiversity alone.

To test the generality of our lab experiment, we sampled surface waters from 101 natural lakes

spanning continental Europe (~30° latitude). As in the lab experiment, the proportion of

aromatic compounds increased from 3.1% to 28.4% over the range of chemodiversity observed

in the field from 396 to 3387 molecular formulae (linear model, t=6.54, df=100, p<0.001). The

chemical characteristics of DOM, including its molecular weight, further suggested that

aromatic-like compounds were increasing with chemodiversity like in the lab experiment (Fig.

S7). We also measured different functions in the field than in our experiment to test if

chemodiversity promoted ecosystem multifunctionality irrespective of the specific functions

being measured, that is, if the association between chemodiversity and ecosystem

multifunctionality was generalizable. The functions measured in the field were related to

carbon cycling (CH_4 and CO_2 emissions) and ecosystem metabolism (total respiration,

maximum net primary production, and bacterial protein production). We reversed the sign of

the measured values for CH_4 emissions, CO_2 emissions, and total respiration so that a higher

multifunctionality value corresponded to a greater benefit to society[37] (see Methods). Thus,

larger values of these three functions indicated lakes were both emitting and mineralizing less

carbon. We then summed z-scores of the five functions into a single index as in the lab

experiment. We also compared the importance of chemodiversity with the importance of

microbial (bacteria, fungi, archaea) and eukaryotic biodiversity estimated from shotgun

metagenomic sequencing of environmental DNA in 82 of the study lakes. Our biodiversity

sampling detected more taxa than in other large-scale studies[42,43] with between 97 to 1589

species (median = 610) per lake.

We found a positive association between ecosystem multifunctionality and chemodiversity but

not with biodiversity across 82 of the 101 field sites with data for all these three variables. Over

the range of chemodiversity (396 to 3387 molecular formulae, median = 2445),

multifunctionality was estimated to increase from a mean of -1.27 (95% CI: -2.16 – -0.37) to

0.30 (-0.33 – 0.92) (linear model: t=2.24, p=0.030, df=81; Fig. 1c). This mean increase

corresponded to a change of 124% (43 – 144%). Neither biodiversity nor its interaction with

chemodiversity influenced multifunctionality (t=1.26, p=0.21 and t=-0.02, p=0.99,

respectively, df=81 for both). Much like in our experiment, the effect of biodiversity on

multifunctionality was limited by chemodiversity. We found that the effect of biodiversity on

multifunctionality only became statistically significant at the highest levels of chemodiversity

(Fig. S8). Three of the five measured functions, CH_4 and CO_2 emissions and bacterial protein

production (BPP), also varied individually with chemodiversity whereas only CO_2 emissions

significantly decreased with biodiversity (Fig. S9). At a molecular-level, these functions were

associated with different formulae, potentially reflecting differences in how microbial

functions consumed organic substrates. For example, BPP was negatively correlated with

generally low H:C formulae (Fig. S10), indicative of consumption[4], because these formulae

were relatively more abundant and thus easier for bacteria to find to build cellular structures

than rarer high H:C compounds that may release more energy[4,44]. CH_4 was positively

associated with the proportion of formulae per lake that contained either N or S (ρ=0.24,

p=0.006), which are precursors for oxic methanogenesis[45,46], and CO_2 negatively correlated

with high H:C formulae (Fig. S10), which are preferably respired by bacteria because they

provide the highest energy return[4,44]. Chemodiversity alone explained 5% of the variation in

multifunctionality, which is lower than in our experiment but we considered more functions

than in our experiment and our experiment controlled for environmental noise, e.g. climate.

This result is also comparable to, and in some occasions up to five-times higher than, the effect

of biodiversity reported in influential field studies explaining variation in ecosystem

multifunctionality[8,36]. The choice of functions affects the strength of the relationship between

biodiversity and multifunctionality, so may explain some of these differences among studies.

To test further the importance of chemodiversity relative to published estimates of biodiversity,

we computed multifunctionality based on the exceedance of data-derived thresholds[37]. This

method was specifically designed to facilitate inter-study comparisons[37]. We found that the

reduced explanatory power of biodiversity relative to chemodiversity persisted when we

estimated ecosystem multifunctionality with this threshold approach (Supplementary Note 1).

Our estimated effect of chemodiversity on multifunctionality with this threshold approach was

larger than the effect of biodiversity on ecosystem function that was previously estimated from

47 published studies[9], including freshwaters (Supplementary Note 1). Unlike for biodiversity-

ecosystem multifunctionality, where positive associations have been reported in studies

measuring plant productivity[3,8], and rarely in studies measuring macronutrient cycling[33], the

chemodiversity-ecosystem multifunctionality relationship may be less sensitive to the specific

functions measured. We found consistent positive associations between chemodiversity and

ecosystem multifunctionality across our experimental and field datasets with fundamentally

different functions associated with primary productivity and macronutrient cycling. Together,

our results from both controlled and natural conditions suggest that the large role of

chemodiversity in supporting ecosystem multifunctionality has, to date, been overlooked and

must be considered in efforts aimed at delivering benefits from nature to society.

Recalcitrance underlies positive chemodiversity-multifunctionality associations

Based on theoretical models that predict that the reactivity of DOM varies with its molecular

composition and local environmental conditions[13,27], we hypothesised that the effect of

chemodiversity on ecosystem multifunctionality depends on the source of chemodiversity

itself. By sampling lakes at similar periods of annual heat accumulation, and with a similar

light environment and DOC concentration (see Methods), we found that spatial variation in

chemodiversity at a continental scale primarily reflected differences in climate and surrounding

land use[16,47] (Fig. S11). Specifically, chemodiversity strongly increased from southern to

northern Europe (ρ=0.40, p<0.001), suggesting that the environment in which chemodiversity

occurred and/or its source could explain why chemodiversity varied throughout Europe. Here

we extend these two potential explanations (see Supplementary Note 2 for further discussion)

to contend that they differentially influence how chemodiversity is used by microbes and thus

ecosystem multifunctionality. If chemodiversity was high because of environmental conditions

("environment" hypothesis in Fig. 2a) that slow microbial transformation of DOM, such as

lower mean annual temperature (MAT) or lower macronutrient concentrations, we expected

that ecosystem functions like microbial respiration and bacterial protein production would also

be reduced[48], resulting in a negative correlation between chemodiversity and

multifunctionality. By contrast, if chemodiversity primarily arose because compounds with

greater chemical complexity were imported into lakes, we expected these compounds to

support a greater number of biochemical transformations and increase ecosystem

multifunctionality ("source" hypothesis in Fig. 2a). Complex compounds, such as those with

high aromaticity, are abundant in lakes and their degradation by microbes involves many

biochemical pathways that increase metabolic activity and biomass formation[4,49,50], and

ultimately multifunctionality. By contrast, simpler compounds are expected to decrease

multifunctionality because they are often used to produce energy, which emits CO_2 as a by-

product.

We tested how both the complexity of DOM and the environment in which DOM was degraded

affected microbial transformation of DOM and subsequently the effect of chemodiversity on

ecosystem multifunctionality. We estimated DOM complexity in each lake based on an

aromaticity index[51] (see Methods). We expected this index to reflect terrestrial vegetation[16],

which is a main source of lake DOM[40]. Slower growing vegetation tends to invest more in

protecting biomass, such as with secondary metabolites[52], because it faces a relatively greater

cost to produce biomass. Compounds involved in plant protection tend to be more complex and

have chemical properties, including a higher aromaticity, which require more biochemical

reactions to be transformed[14,27]. We expected complexity to influence multifunctionality by

determining the extent to microbes can transform DOM and we expected complexity itself to

be determined by terrestrial net primary productivity (NPP), i.e., vegetation growth rates, in

surrounding catchments. We also considered nine environmental variables (MAT, annual

precipitation, water temperature, pH, water colour, lake area, biodiversity, trophic state, and

nutrient concentrations) that could influence chemodiversity by changing the diversity of

terrestrial and aquatic organic matter inputs, controlling the physical energy required to break

chemical bonds, and regulating photo-oxidation[53,54,55]. NPP and the environmental variables

were estimated from high-resolution remote sensing data and in-situ measurements (see

Methods). We then averaged the number of times molecular formulae could be transformed in

each lake using a list of 1255 putative biochemical transformations[19] (see Methods). If the

positive association we observed between chemodiversity and ecosystem multifunctionality

arose from the input of complex compounds rather than the limitation of microbial

transformations of DOM, we expected NPP would be more strongly associated with

chemodiversity than the nine other environmental variables.

Our results indicate that the source of DOM explains how and where a positive chemodiversity-

ecosystem multifunctionality relationship can arise. Specifically, we found that the production

of complex compounds by low terrestrial NPP (ρ=-0.70, p<0.001, Fig. S12a) correlated with

both greater chemodiversity (ρ=0.56, p<0.001; Fig. S12b) and a greater number of average

transformations that formulae could undergo in each lake (ρ=0.45, p<0.001, Fig. S12c).

Consequently, chemodiversity correlated positively with the number of transformations per

formulae (ρ=0.40, p<0.001, Fig. 2b). Correlations between chemodiversity and each of

complexity, NPP, and the number of biochemical transformations are predicted by the source

hypothesis. By contrast, all the environmental variables had weaker associations with

chemodiversity, though some were statistically significant (Table S2). Together, our results

indicate that the production of complex compounds by low NPP increases both chemodiversity

and the number of biochemical transformations that each compound can support.

We synthesised the observed correlations among chemical complexity, biochemical

transformations, and chemodiversity to propose a new general understanding of ecosystem

multifunctionality. This new understanding was that the positive effect of chemodiversity on

multifunctionality arises because of the production of complex compounds by plants and

environmental conditions that favour those compounds within DOM pools. We found further

support for this framework with a path analysis explicitly testing presumed causal pathways.

The path analysis modified our original hypotheses to incorporate additional paths identified

by tests of model structure (Fig. S13). With this analysis, we found that effects related to the

production of organic compounds (i.e. NPP) influenced chemodiversity by changing DOM

complexity and subsequently the number of biochemical transformations it can receive (Fig.

2a). Although environmental variables (i.e., water colour and MAT) also influenced DOM

complexity, the direction of these effects suggested that they were associated with reduced in-

lake primary productivity and an increase in the relative importance of terrestrial DOM (Fig.

S13). This interpretation was further supported by the direct negative effect of chlorophyll *a*

concentrations on chemodiversity (Fig. S13).

To test further if ecosystem multifunctionality was enhanced by chemodiversity because of the

properties of complex compounds associated with low terrestrial NPP, we analysed the

composition and putative source of all 8456 molecular formulae identified in the field survey.

We first identified formulae whose relative abundances across our study lakes were correlated

with terrestrial NPP of the surrounding catchment. Positive and negative correlations suggest

formulae likely originated from high and low terrestrial NPP, respectively. We compared these

correlations with the average number of biochemical transformations that each formula was

predicted to undergo in each lake[19]. This analysis allowed us to infer the origins of compounds

that can be highly transformed by biotic and abiotic reactions, which contribute to higher

ecosystem multifunctionality. Compounds that were associated with low NPP appeared more

transformed (t=-8.19, df=6639, p<0.001), and positively correlated with chemodiversity

(t=8.50, df=6639, p<0.001; warmer colours in Fig. 2c). These compounds were more complex,

on average, with a mean (± standard error) aromaticity index of 0.52 (±0.17) compared to 0.32

(±0.24) in the entire dataset (Wilcoxon rank sum test, W=5.4×10^6, p<0.001, Fig. S14). These

results further supported the hypothesis that complex compounds can be degraded in more

ways and support higher multifunctionality. We then selected molecular formulae that most

increased in abundance with decreasing NPP, that is, were in the top 10% of negative

associations based on Spearman rank correlations. We found that these formulae could support,

on average, 1.4 times more known putative biochemical transformations than formulae that

were weakly associated with NPP, i.e. lowest 10% of negative associations (t=5.80, df=860,

p<0.001), and 1.8 times more transformations than formulae strongly positively associated with

NPP, i.e. top 10% of positive correlations (t=8.00, df=1090, p<0.001). Together, these results

demonstrate that the effect of chemodiversity on multifunctionality emerges from the

properties of individual formulae.

We expect chemodiversity will be less important for ecosystem multifunctionality in

environments where microbes cannot access and transform large quantities of organic matter.

For example, DOM concentrations are typically >5 mg C L^{-1} in rivers, most lakes, wetlands, or

mangroves[19,47,48,56], but <1 mg C L^{-1} in the deep ocean[29] or in large (surface area >500 km^2)

deep lakes with centennial residence times[57]. At these concentrations, individual compounds

are more likely to be too dilute to interact with microbes[52], irrespective of the diversity of

compounds in the environment, and may result in a shallower slope between chemodiversity

and ecosystem function. In support of this expectation, we found that compounds transformed

in lakes reacted less frequently when found in the deep ocean. We identified 2745 formulae

shared between our study lakes and a representative deep ocean sample[18], which is likely an

overestimate because we cannot differentiate structural isomers. We then calculated the

number of times that each formula was transformed in each sample using the established list

of putative biochemical transformations[19]. We found that these formulae underwent, on

average, 0.08 (range: 0 to 1.32, standard deviation = 0.17) putative transformations in lakes but

no transformations were detected in the deep ocean. Given that many of the detected

transformations are ubiquitous, such as methylation, a likely explanation for why we did not

find them in the ocean sample is because their substrates occur at concentrations lower than

our femtomole detection limit[58]. Such concentrations would be several order of magnitude

beneath concentrations of the most limiting organic compounds[59]. This observation further

suggests that chemodiversity will have a greater influence on ecosystem multifunctionality

where individual compounds occur at high enough concentrations to support many biochemical

transformations.

Chemodiversity and ecosystem multifunctionality under global change scenarios

NPP is expected to increase because of rising atmospheric CO_2 concentrations[60], and our results

suggest that this increase could reduce the chemodiversity of lakes in ways that are relevant to

global carbon budgets and climate modelling efforts. To predict how the resulting changes in

chemodiversity will influence ecosystem multifunctionality, we first modelled the value of

each ecosystem function that varied with chemodiversity in our field survey (Fig. S9). By

adding chemodiversity to statistical models of CH_4 and CO_2 emissions and BPP that contained

predictors identified as important in previous large-scale studies[23,24], we increased explained

variation of each function by up to 50% as compared with past studies (Table S4). We achieved

the large increase in explained variation despite being unable to measure some prominent

drivers of ecosystem metabolism, such as phosphorus concentrations[61]. Importantly, based on

the properties of linear models, additional predictors are unlikely to change the effect of

chemodiversity on individual functions and will only further reduce unexplained variation. We

then projected our models for each ecosystem function to all European lakes >10 ha (see

Methods). For chemodiversity, we generated a European-scale raster with the model used in

the path analysis (Fig. S13). This analysis further validated our study because we found that

the extrapolated estimates were within bounds of global-scale studies[23,24]; no European-only

studies exist for comparison. Overall, we estimated that European lakes emitted, on average,

22.7 Tg C yr^{-1} as CH_4 (95% CI 11.2 – 32.0; model $R^2 = 0.46$) and 27.8 Tg C yr^{-1} as CO_2 (95%

CI: 9.9 – 37.4; $R^2 = 0.22$) in 2019. Despite these emissions potentially being overestimated

because they only considered summertime fluxes[62], they corresponded to 7% and 19% of

estimated global lake CH_4 and CO_2 emissions[23,24], respectively, and Europe contains ~20% of

the world's total number of lakes[63]. We also estimated that European lakes fixed an average of

1.8 Tg C yr^{-1} (95% CI: 1.0 – 3.3; $R^2 = 0.45$) in 2019 into microbial biomass, that is, from BPP.

To forecast the consequences of changes in NPP for European lakes, we input NPP estimated

in the year 2100 by Earth system models[64] into the linear model that predicted chemodiversity

(Table S3) and then used the predictions to forecast each ecosystem function (Table S4). We

focused solely on changes in NPP, holding other variables constant, because NPP was the best

individual predictor of chemodiversity (Table S2) and had a clear mechanistic link to

chemodiversity (Fig. S13). We used two shared socioeconomic pathways (SSPs) representing

likely (SSP2) and worst case (SSP5) scenarios for future changes in atmospheric CO_2

concentrations. We found that future increases in NPP alone will reduce chemodiversity in

99.1% of lakes, with the median chemodiversity decreasing by 13% and 24% under the SSP2

and SSP5 scenarios, respectively (Fig. 3a). When summed across lakes, we estimated that the

change in chemodiversity will cause lakes to emit an additional mean 28.4 (95% CI: 3.3 –

166.1) and 32.2 (95% CI: 14.0 – 231.2) Tg C yr^{-1} as CH_4 under SSP2 and SSP5 scenarios,

respectively (Figs 3b, S15). Similarly, the changes in chemodiversity will cause lakes to emit

an additional 3.7 (95% CI: 1.3 – 5.1) and 4.0 Tg C yr^{-1} as CO_2 (95% CI: 1.3 – 5.2 Tg C yr^{-1})

under SSP2 and SSP5 scenarios, respectively (Figs 3c, S15). Microbes will, however, fix 4.3

(95% CI: 1.7 – 10.2) and 4.4 Tg C yr^{-1} (95% CI: 1.7 – 12.6 Tg C yr^{-1}) more carbon into biomass

under SSP2 and SSP5 scenarios, respectively (Figs 3d, S15). Together, our estimates suggest

that European lakes may emit, on average, 1.6-times more carbon by 2100, mostly as CH_4,

because of NPP-associated declines in chemodiversity. Although our estimates are higher than

previous estimates of future CH_4 emissions by lakes in absolute terms by 50%, they are

comparable in terms of relative change, i.e. 73-84% in our study versus 58-88%[62]. The

difference in absolute terms is likely due to the lack of winter measurements in our dataset as

lower microbial activity in colder weather typically reduces greenhouse gas emissions[62]. The

similarity in relative terms instead emphasises that chemodiversity is a necessary predictor of

CH_4 emissions.

We forecasted that lakes in southern Europe, and in the south of the Scandinavian peninsula,

will experience the largest changes in ecosystem function (Fig. 3). These lakes are located near

large population centres and, therefore, provide opportunities to improve management of their

surrounding catchments in ways that can enhance the benefits that European lakes provide to

society. For example, our results suggest that chemodiversity and subsequently ecosystem

multifunctionality can be enhanced by promoting slow growing vegetation, such as many

Fabaceae[65]. Slow growing vegetation produce large numbers of complex polymeric

compounds, which undergo many biochemical transformations when leached into water from

litterfall and soils[17]. They also have higher wood density[66], that can increase long-term carbon

storage on land[67], resulting in a win-win for land and water.

Conclusions

Here, we provided experimental and large-scale observational evidence that increasing

chemodiversity delivers beneficial ecosystem functions, especially a stronger carbon sink

capacity corresponding with greater bacterial biomass and lower greenhouse gas emissions.

We then demonstrated that the ways in which ecosystems deliver these functions depends on

the molecular properties of DOM, namely the capacity of individual compounds to undergo

biochemical transformation. The molecular properties of compounds ultimately depend on

sources of DOM and external environmental conditions, such as their concentrations, as found

in many inland waters, wetlands, and mangroves[19,47,48,56]. Here we add to this understanding

by showing the importance of these properties for ecosystem multifunctionality and their

potential to explain ecosystem-specific differences in chemodiversity–multifunctionality

relationships. Our findings also suggest that chemodiversity and its effect on multifunctionality

can be altered across different aquatic ecosystems by manipulating sources of DOM within

surrounding catchments. Finally, our study suggests that accounting for the strong association

between NPP and chemodiversity can improve integration between terrestrial and aquatic

ecosystem models[5]. For example, because chemodiversity is strongly predicted by NPP, which

itself is predictable at high temporal and spatial scale from bioclimatic data, we can readily

incorporate the effect of chemodiversity on greenhouse gas emissions into existing carbon

budgets. More generally, our study demonstrates that sustaining the benefits that lakes provide

to people will require management that extends beyond shorelines and considers the

chemodiversity exported from land.

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Author contributions

478   AJT and JAF designed the study. JAF performed the field and associated lab work with input

479   from AJT and TD. SRS and JAF performed the laboratory experiment. JAF analysed the data

480   with input from AJT. JAF and AJT wrote the manuscript with input from TD and SRS.

Data availability

483   All data are available within the manuscript or upon reasonable request.

Figure 1: Chemodiversity increases ecosystem function. Multifunctionality increased with **(a)** higher chemodiversity but not with **(b)** higher biodiversity under controlled conditions. Multifunctionality was measured as the sum of z-scores of changes in each of cell abundances, and transformation of dissolved organic carbon (DOC) and total dissolved nitrogen (TDN) over 7 days in 36 microcosms represented by individual points. Boxes encompass the inter-quartile range with median denoted by solid horizontal line and whiskers equal to 1.5-times the interquartile range. Different letters indicate statistically significant ($p<0.05$) differences between factor levels with Tukey's pairwise comparison tests. Chemodiversity of each experimental source was estimated from the exponential of the Shannon diversity index. In a separate field study, multifunctionality increased with **(c)** higher chemodiversity but not with **(d)** higher biodiversity across 82 European lakes represented by individual points. Multifunctionality was measured in the field study as the sum of z-scores of CH_4 and CO_2 emissions, total respiration (Resp), maximal net primary production (PP), and bacteria protein production (BPP). Lines in **(c)** and **(d)** are estimated mean associations based on a linear model $\pm$ 95% confidence intervals. For all, darker colours indicate higher chemodiversity (brown points) and biodiversity (blue points).

Figure 2: The origin of chemodiversity influences its effect on ecosystem multifunctionality. (a) We hypothesized that the positive effect of chemodiversity on ecosystem multifunctionality was caused by environmental conditions reducing the transformation of DOM ("Environment" hypothesis) and sources of DOM producing compounds that could be transformed in a greater number of ways ("Source" hypothesis). + and – symbols show hypothesised cause-and-effect of one variable on another, with either positive or negative links between NPP and complexity possible in the Source hypothesis. We tested these hypotheses using path analysis fitted to 101 lake-level observations. To simplify the path analysis, we created a composite variable "environment" that encompassed 9 measured parameters (see Fig. S13 for effect of each parameter), whereas "source" was equal to NPP in the surrounding catchment of each lake. Complexity and transformations corresponded with the intensity-weighted modified aromaticity index and mean number of putative transformations per molecular formula per lake, respectively. Coloured arrows show statistically significant paths with standardized effects. **(b)** The mean number of transformations per formula per lake was closely correlated with chemodiversity (ρ=0.40, p<0.001). Each dot represents a lake (N=101) coloured for chemodiversity. **(c)** We also correlated the number of transformations that each of 8456 molecular formulae underwent across all lakes with the NPP of the lake catchment. Formula (points) associated with low NPP (i.e., negative value along the *y*-axis) generally underwent more transformations (i.e. larger value along the *x*-axis; ρ=-0.10, p<0.001). Numbers of transformations were weighted by the relative abundance of each formula across lakes. Points are coloured according to their correlation with chemodiversity with low (blue) and high (red).

Figure 3: Lakes will emit more carbon by 2100 due to continental-wide declines in chemodiversity. We estimated **(a)** chemodiversity in 2100 while holding all other variables at 2019 levels for all European lakes >10 ha in surface area (R^2 for chemodiversity model = 0.36). Values are the change in chemodiversity compared to 2019. Using future chemodiversity, we then estimated mean changes in **(b)** CO_2 emissions **(c)** bacterial biomass production, and **(d)** CH_4 emissions from pelagic surface waters between 2019 and 2100 using linear models (R^2 = 0.22, 0.45, and 0.46, respectively; Table S4). For all, points indicate N=175,500 individual lakes. Histograms indicate the number of lakes found with corresponding changes. 95% confidence intervals for each function are given in Fig. S15.

Materials and methods

Laboratory experiment

To test if multifunctionality varies with chemodiversity, we incubated bacterial communities of four biodiversity levels with three different levels of chemodiversity. The four biodiversity levels were 1) *Pseudomonas fluorescens* SBW25 2) *Chryseobacterium* sp. and *Sphingobacterium* sp. isolated from natural communities that degrade DOM[68]; 3) *Rahnella* sp., *Raoultella* sp., and *Pantoea* sp. from the same natural communities[68]; 4) all six microbes together. Before starting the experiment, we grew each microbial isolate individually from a frozen archive for five days in R2A media. A control of filtered (0.2 μm, Sterivex-GP, MilliporeSigma, USA) sterile water was included alongside each treatment and indicated no contamination as measured with flow cytometry (see below). For chemodiversity, the three treatment levels were obtained from reverse-osmosis-concentrated water[69] of a temperate wetland (Luther Marsh, Ontario, Canada; 43°57'43"N, 80°23'50"W), boreal lake (Clearwater Lake, Ontario, Canada; 46°22'11"N, 81°03'04"W), and Arctic river (Cameron River, Northwest Territories, Canada; 62°29'33"N, 113°32' 48"W). Using MilliQ water, we diluted each reverse-osmosis concentrate to a final dissolved organic carbon (DOC) concentration of 10 mg C L^{-1}, as measured with a TOC-TNM-L analyser (Shimadzu, Japan), and filled fifteen 250 mL pre-combusted (4 hr at 550°C) Erlenmeyer flasks with 100 mL of each water source (n = 3 replicates for each of 4 biodiversity levels with 3 controls). Before inoculating the DOM sources with each microbial community, we estimated strain-specific cell number with a CLARIOstar fluorometer (BMG Labtech, UK) after 30 min of staining with SYBR Green (10,000× stock solution, Invitrogen, USA) diluted to a 10× final concentration. Based on these data, we diluted an unstained aliquot of each strain to an expected 10,000 fluorescence units and mixed them to form the four communities. This normalisation created a substitutive design that ensured each community had a similar abundance of DNA and thus potential activity.

Then, we added 1 mL of each of the four microbial communities to each chemodiversity level

in triplicate and incubated the flasks at ca. 20°C for seven days in the dark without shaking.

Measuring chemodiversity

We estimated chemodiversity in the experiment from solid-phase extraction of DOM following

a standard protocol[70]. Briefly, 80 mL of each incubation flask was acidified to a pH of 2 by

adding HCl, passed through 1 g of a polar styrene-divinylbenzene polymer (Bond Elut PPL,

Agilent, USA), and eluted with 3.5 mL of ultrapure methanol (LC-MS LiChrosolv, Merck,

Germany). We diluted the extracts to a final concentration of 2.5 ppm DOC in a 1:1 (v:v)

methanol:water solution before injecting 100 µL into a 15 Tesla Solarix FT-ICR MS (Bruker

Daltonic, Germany) with direct infusion electrospray ionization in negative mode. We

collected 200 scans in duplicate for each sample and calibrated the spectra using the

DataAnalysis software (Bruker Daltonic, Germany). Masses ranging from 150 to 1000 m/z

were exported, and we assigned molecular formulas using the online platform ICBM-

OCEAN[71]. We used the N, S, P rule, deleted all singletons, and used isotopic confirmation to

accept/reject a formula. We limited formula attributions to $C_{1-100}H_{2:200}O_{0-70}N_{0-3}P_{0-2}S_{0-2}$ with a

tolerance of 0.2 ppm and retrieved a total of 3873 individual molecular formulas. We computed

chemodiversity as the exponential of the Shannon diversity index using all assigned molecular

formulas[72] and the summed normalised peak intensity per sample. This index represents the

diversity in a community when all formulae are equally common, so corrects for variation in

evenness that can bias raw counts[73] of formulae number. We also computed the ratio of

hydrogen to carbon atoms in each compound as an indicator of bioavailability, as described

elsewhere[44]. The three DOM sources were largely nested with one another with 93% of

formulae present in the Arctic river (low chemodiversity) also present in the boreal lake

(medium chemodiversity), 70% of boreal lake formulae present in the temperate wetland (high

chemodiversity), and 80% of formulae present in the Arctic river present in the temperate

wetland (Fig. S2). The large number of shared formulae among treatments confirmed that

sources with higher chemodiversity contained the vast majority of formulae present in sources

with lower chemodiversity, i.e. were nested. Additionally, the molecular composition of the

DOM used in the experiment differed as much from the lakes sampled in the field study as the

lakes differed among themselves (Fig. S16).

Ecosystem multifunctionality in the laboratory experiment

We estimated the multifunctionality of each replicate community in the experiment by

measuring carbon consumption, nitrogen consumption, and bacterial growth rate. At the start

and end of the incubation, we measured the DOC and TDN concentrations of each flask using

a Shimadzu TOC-TNM-L analyser. Carbon and nitrogen consumption were estimated as the

change in concentration between the start and end of the incubation. To measure growth rate,

we determined the difference in cell abundance between the start and end of the incubation

using flow cytometry. Briefly, we removed 800 µL of water from each flask at the start and

end of the experiment. Samples were incubated with 200 µL of 25% glutaraldehyde in the dark

at 4°C for 30 min before being preserved at -80°C. Within one week, we thawed the samples,

added SYBR Green at a 10× final concentration (10,000× stock solution, Invitrogen, USA),

and incubated the samples overnight at 4°C. Then, we injected 100 µL of each sample into a

flow cytometer (Attune NxT, Thermo Fisher Scientific, USA) at a rate of 250 µL per minute

and used the same instrument settings for all samples. We optimised the gating for each DOM

source based on stained/unstained samples using FlowJo v10.9 (BD Biosciences, USA) and

used the same gating for each community. Each function was standardized to a mean of zero

and standard deviation of one to obtain a *z*-score, which we summed across the three functions

to estimate multifunctionality[8].

Selection of field study sites

We selected lakes to sample across Europe ensuring there were no systematic biases in DOC

concentrations and water chemistry across our latitudinal gradient. Using the EU's Water

Information System for Europe (WISE) database version 4[74], we subsampled all entries

labelled as "lakes" for which DOC was reported in mg C L^{-1}. We also selected entries with a

total organic carbon (TOC) concentration equal or lower than 5 mg C L^{-1} to include more lakes

with low DOC values, which were underrepresented in the WISE database despite being known

to be the most common on Earth[75]. From the 2169 resulting entries, we removed 256 entries

with a DOC/TOC concentration >30 mg C L^{-1}, which is outside the expected values in lakes[75]

and likely represent measurement error or atypical lakes, and 1595 duplicated measurements.

We grouped the remaining 318 lakes into five bins in 5 mg C L^{-1} intervals: <5 mg C L^{-1}, 5-10

mg C L^{-1}, 10-15 mg C L^{-1}, 15-20 mg C L^{-1}, >20 mg C L^{-1} and six 5° latitudinal intervals: <45°N,

45-50°N, 50-55°N, 55-60°N, 60-65°N, >65°N. The last two DOC/TOC bins (i.e. >15 mg C L^{-1}

^{1}) were absent from the first two latitudinal bins. Therefore, we removed all 47 entries with

DOC/TOC concentrations >15 mg C L^{-1}. In a global analysis[75] of lake DOC concentrations,

93% of lakes were predicted to have DOC concentrations <15 C L^{-1}, so this filtering minimises

potential sampling bias while ensuring we still represent the vast majority of lake DOC

concentrations. To standardise the sites further, we considered total nitrogen (TN) and total

phosphorus (TP) concentrations of all entries. Six entries lacked TP values, so we discarded all

but one, which was located between 50-55°N where entries were sparse. Only 10 entries had

TP concentrations higher than 0.2 mg L^{-1}, and 21 entries had a TN value higher than 1 mg L^{-1},

so we removed these to avoid large differences in lake trophic levels.

After reducing the number of entries to lakes with similar water chemistry, entries were

selected based on their distance from one another. Therefore, we discarded 28 entries from

eastern Poland, one in Latvia, and two in Hungary as they were logistically challenging to

reach. We similarly removed 175 entries situated in the United Kingdom and western France.

We then inspected each remaining entry individually using Google Earth satellite imagery,

removing 86 sites that were dammed or visibly artificial. This process removed all entries in

the WISE database situated south of 40° N and north of 69°N. To extend the latitudinal

gradient, we manually added eight sites in southern Italy (<40°N) and seven in northern

Norway (>69°N) without knowledge about their DOC/TOC or trophic state. None of the

manually chosen lakes were dammed or visibly artificial. This selection process resulted in a

total of 109 lakes. During sampling, severe weather conditions, unexpected fences around

lakes, or equipment failure prevented us from sampling eight lakes, and one site in Poland was

substituted for a lake for which no DOC data were available. Ultimately, we sampled 101 lakes

spread over 31 latitudinal degrees. There was no statistically significant correlation between

latitude and DOC concentration at the time of sampling (Spearman correlation, ρ=0.05, *p*=0.57,

Fig. S17).

Timing of field sampling

The order in which lakes were sampled was chosen to minimise differences in annual heat

accumulation and water temperatures across the latitudinal gradient. Sampling dates were

originally chosen to minimise differences in growing degree days (GDD) among lakes. GDD

represents the difference between a given temperature threshold at which primary production

commences and mean daily temperature[76]. Here we used a threshold of 10°C as is classically

employed[76]. To compute GDD, we used the rnoaa package[77] in R v.4.1.1[78] to retrieve all

weather stations within 50 km of the 101 sampled lakes from the US National Oceanic and

658 Atmospheric Administration National Climatic Data Center database. We determined the

659 median GDD for each day of the year based on temperature data collected from the last 10

660 years. For years with 366 days, the GDD on the 29th of February was added to the GDD from

661 the 28th of February. Cumulative GDD (cGDD) – representing the sum of GDD from the 1

662 January until a given date – were obtained by sequential addition for each day of the year for

663 each lake. The computed cGDD indicated that the growing season commenced (i.e. cGDD

664 >10) as early as the 1 January at the southernmost site and ended (i.e. cGDD reached its highest

665 values and no longer increased) as early as the 10 October at the northernmost site.

667 We then selected sampling dates to survey lakes during similar periods of annual heat

668 accumulation. To find the optimal sampling day, we calculated the variation in cGDD along

669 the latitudinal gradient as the ratio between the 3rd and the 1st quartiles of lake-level cGDD,

670 assuming we started sampling the southernmost site and moved northwards to the closest site

671 at a pace of one lake per day. By starting sampling on the 8 April 2019, we could minimise the

672 variation in cGDD across the latitudinal gradient. Overall, the observed maximal difference

673 between the 5° latitudinal bins was 194.4 GDD, which was much less than the 368 that was

674 originally expected based on the historical data (Fig. S17). This value corresponded with

675 sampling the warmest site 10 days later than the coldest site, during which temperatures

676 remained above 20°C. We further measured underwater irradiance at 1 m depth in 60 lakes

677 across the latitudinal gradient using a HOBO light pendant (UA-002-64, Onset, USA) to test

678 for additional biases in irradiance. Pictures representing the latitudinal gradient are presented

679 in Fig. S18.

681 *Dissolved organic matter sampling*

At each lake, we collected 10 L of surface water in an acid-washed plastic carboy. We sampled

from the point with maximum water column depth or the point equidistant to all shorelines

when no bathymetry information was available. We expected these locations would be the best

mixed in the epilimnion and most representative of the lake water quality. Directly after

sampling, 1 L of lake water was passed through a pre-combusted (450°C, 4 hours) glass fibre

filter (0.5 μm, Macherey-Nagel, Germany), and the filtrate dispatched into three 40 mL pre-

combusted amber borosilicate vials and one 500 mL pre-combusted amber borosilicate bottle.

Once on land, we acidified the samples to a pH of 2 by adding HCl. We kept the filtrates in the

dark, and either on ice or at ca. 4°C for up to 15 days before shipping them to Cambridge, UK.

In the laboratory, we stored the samples at 4°C for up to a week and determined the DOC in

each 40 mL vial using a Shimadzu TOC-TNM-L analyser (Shimadzu, Japan). In line with

regulatory standards[79], we averaged DOC concentrations from the three replicates when the

standard deviation was within 10% of the mean and averaged the two closest values otherwise.

We estimated chemodiversity as described for the laboratory experiment but solid-phase

extracted 500 mL of water instead of 80 mL and retrieved a total of 8456 formulae. We also

evaporated 10 μL of extract and redissolved it in 10 mL of Milli-Q water acidified to a pH of

2 using HCl to determine the DOC concentration in the extract and corresponding extraction

efficiency (mean = 62%, SD =15%).

To determine if chemodiversity promoted multifunctionality because it could undergo more

biochemical transformations, we also estimated the number of times that each molecular

formula could be biochemically transformed using an established approach[19]. Fudyma _et al._[19]

compiled 1255 pairs of molecular formulae associated with commonly observed biotic and

abiotic transformations, such as enzymatic reactions that add or subtract CH_2 groups and

physical condensation or dehydration, respectively. We considered such a transformation to

have occurred if we detected each pair of formulae from a reaction in a sample.

We estimated the complexity of each compound using an aromaticity index (AI_{mod})[51]. A

molecule is considered more complex if it possesses more three-dimensional structures and

aromatic rings, which are both characterised by the presence of double bonds. However, atoms

such as nitrogen can be linked to carbon with double bonds without necessarily forming

aromatic rings. By using AI_{mod}, we considered both the number of double bound equivalents

(DBE), which approximates the saturation of each molecular formula, and the number of

heteroatoms, which provides insight into how carbon is bound. The number of DBE and AI_{mod}

were computed within the ICBM-OCEAN pipeline using equations described by Koch &

Dittmar[51,71].

Biodiversity sampling

We estimated biodiversity by sampling environmental DNA in 82 of the study lakes. During

sampling, we passed 1L of water through a 0.2 µm filter cartridge (Sterivex, Millipore, USA).

The cartridges were stored at -20°C until DNA was using an established protocol[80]. Briefly,

filter cartridges were opened in a sterile environment, and we placed the filters in a cryotube

containing silica and zirconia beads (3.0, 0.7, and 0.1 mm diameter). We then vortexed the

tubes at 2,850 rpm for 15 min to detach cells from the filter. To ensure DNA was released into

solution from lysed cells, we added 0.6 mL of phenol-chloroform-isopropanol (25:24:1), 0.6

mL of 5% cetrimonium bromide, 60 µL of 10% sodium dodecyl sulfate, and 60 µL 10% N-

lauroylsarcosine before vortexing the tubes a second time at 2,850 rpm for 15 min. We then

centrifuged the tubes at 16,000 g for 15 min at 4°C and collected the supernatant. To remove

phenol residues, we added 0.6 mL of chloroform-isopropanol (24:1) to the supernatant, gently

mixed the samples by inversion, and centrifuged at 16,000 g for 10 min at 4°C. We again

collected the supernatant (ca. 0.6 mL) and added 1.2 mL of polyethylene glycol with 1.6M

sodium chlorine to precipitate the DNA overnight at 4°C. The following morning, we

centrifuged the samples at 17,000 g for 90 min at 4°C to obtain a DNA pellet that we washed

with ice-cold (-20°C) 70% ethanol. The pellet was dried at 37°C for 10 min before being

dissolved in 50 µL of ultrapure water (Qiagen, Germany). We used the same protocol to extract

DNA from a ZymoBIOMICS™ Microbial Community Standard II (Log Distribution) (Zymo

Research, USA) and ultrapure water (Qiagen, Germany) that served as positive and negative

controls, respectively. The positive control indicated no external contamination, and no DNA

was retrieved from the negative control.

We prepared shotgun metagenomics sequencing libraries with the NEBNext Ultra II FS kit

(New England Biolabs, USA) following the manufacturer's instructions. Briefly, DNA was

digested for 15 minutes to obtain end polished and A-tailed fragments of ~350bp. Then, we

attached Illumina adaptors by incubating the fragments at 37°C for 15 min in presence of a

ligation enhancer. We cleaned the samples using AMPure XP beads (Beckman Coulter, USA)

and attached Illumina-specific indexes (New England Biosciences, USA). We purified the

product using the AMPure XP beads[81], quantified DNA concentration on a Qubit fluorometer

(ThermoFisher, USA), and pooled libraries at equimolar concentrations before sequencing on

a NovaSeq 6000 PE 250 bp (Illumina, USA) at Novogene (Cambridge, UK), who performed

demultiplexing and checked read quality (Q30 ≥ 75% and Q20 = 100%). We obtained a similar

coverage for all samples (1st and 3rd quartiles: 4.35×10^7 and 6.70×10^7 reads). We assigned

taxonomy to the reads using Kraken2[82] and a database containing bacteria, eukaryotes

(including fungi), and archaea. We used Bayesian Reestimation of Abundance with Kraken to

obtain read counts at the genus level[83] and estimated biodiversity as the exponential of the

Shannon index as for chemodiversity.

Ecosystem functioning in the field survey

At each field site, we measured three functions related to ecosystem metabolism (maximum

primary production, total respiration, and bacterial protein production) and two functions

related to carbon cycling (CH_4 and CO_2 emissions) from the lake surface.

Primary production and respiration were obtained based on differences in dissolved O_2 as

previously described[84]. First, we filled six glass serum bottles with 20 mL of lake water. All

bottles were crimped gastight without air bubbles. Two bottles were processed directly, two

bottles were incubated in the dark for 24 hours, and the last two bottles faced a cold light

(5000K) lamp and received ca. 34,000 lux for 3 hours. For each bottle, we measured the

dissolved O_2 concentration using optical fibre sensors (Presens, Germany). We recorded

dissolved O_2 concentration every second for approximately five minutes until we reached a

plateau. Then, we averaged the last 30 values of the plateau to infer the dissolved O_2

concentration inside the serum bottles. Primary production was estimated as the difference in

dissolved O_2 between the start of the incubation and after exposure to light. Respiration was

the difference in dissolved O_2 between the start and after 24 hours of dark incubation. We

averaged duplicates to obtain one value per lake.

Bacteria protein production (BPP) was estimated based on ^{3}H-leucine incorporation[85]. Directly

after sampling, we filled six tubes (Eppendorf, Germany) containing 17nM of ^{3}H-leucine with

1.5 mL of lake water. We previously added 300 µL of trichloroacetic acid to three tubes that

served as blanks. All tubes were immersed inside the lake in a dark container at ca. 0.5 m depth

for exactly 1 hour. We terminated the incubation by adding 300 µL of trichloroacetic acid and

preserved the samples on ice until we reached the lake shore – the journey time varied from 1

to 30 mins. Then, we centrifuged the samples for 10 min at 16,000 g, removed the supernatant,

washed the pellets with 1 mL of 5% trichloroacetic acid, and centrifuged again for 10 min at

16,000 g. We removed the supernatant and let the pellets air-dry for up to 10 min. We added 1

mL of scintillation cocktail (HiSafe, Perkin Elmer, USA) and measured counts per minutes

with a portable liquid scintillation counter (Triathler, Hidex, Finland). To obtain the final BPP,

we converted counts per minutes into degradation per minutes, subtracted the value of the mean

of the corresponding blanks and expressed the results in mg C L^{-1} day^{-1} as previously

described[86]. We averaged the values from the three replicates when the standard deviation was

within 10% of the mean and averaged the two closest values otherwise.

CH_4 and CO_2 emissions were measured using the floating chamber technique[87]. We used a

chamber with a 0.07 m^2 perimeter and 0.002 m^3 volume that was covered in aluminium tape to

prevent heating. We connected the chamber to an ultraportable greenhouse gas analyser (Los

Gatos Research, USA) that uses off-axis integrated cavity output spectroscopy using gastight

PVC tubing. Before starting any measurements, we let the chamber equilibrate with the

atmosphere by waiting until the CH_4 and CO_2 curves plateaued. Then, we placed the chamber

on the water with 3 cm of the chamber immersed underwater. We monitored the concentration

of CH_4 and CO_2 inside the chamber every second for 15 mins. We estimated fluxes as the linear

slope between the gas concentration and time[87]. We repeated the process (atmospheric

equilibration and measurement) twice, and a third time if the two fluxes differed by more than

10%. We then averaged the slopes to estimate CH_4 and CO_2 fluxes.

To calculate a multifunctionality index that reflects human values of nature, we assessed the

societal benefit of each measured function[1]. We considered that society benefits from higher

primary production because it can fuel food web productivity and fix atmospheric carbon into

biomass. Similarly, larger BPP benefits society because it increases the amount of carbon in

the food web and can store carbon for longer inside lakes. However, higher greenhouse gas

(CO_2 and CH_4) emissions negatively impact society due to their effect on climate. Higher

microbial respiration similarly consumes the carbon stored in lakes and emits CO_2 into the

environment, some of which will escape to the atmosphere. We therefore multiplied each of

CO_2 emissions, CH_4 emissions and respiration rate by -1 to ensure that they had the same

directionality of their societal benefits as primary production and BPP, that is, all functions

benefited society more as they increased in value. This decision seemingly contrasted with the

lab experiment where we calculated absolute values because both increases and decreases in

DOC and TDN indicate greater rates of beneficial biogeochemical cycling. However, both

approaches are consistent with the idea that different functions are valued in different ways.

We then calculated multifunctionality as for the laboratory experiment by summing z-scores of

each function.

Predictors of chemodiversity and ecosystem function

Ecosystem functions and chemodiversity are often influenced by physicochemical

variables[16,36] and climate[8,9]. Therefore, during the field study, we recorded water temperature

and pH using a multiprobe (HI-99171, Hanna Instruments, USA), measured water colour as

the absorbance at 254nm using a spectrophotometer (Flame UV-Vis, Ocean Optics, USA), and

measured chlorophyll *a* using a portable fluorometer (FluoroProbe III, bbe Moldaenke,

Germany). In the lab, we measured TDN concentrations concurrent with DOC concentrations

on the Shimadzu TOC-TNM-L analyser. Climate variables were the mean annual temperature

(MAT) and mean annual precipitation (MAP) of each lake at a ca. 31km resolution compiled

by the European Centre for Medium-Range Weather Forecasts ERA5 data product[88]. We

clipped the ERA5 analysis to the area of each catchment retrieved from the Global Lake area,

Climate, and Population dataset[89] and averaged across months for all pixels in a catchment to

obtain MAT and MAP. Because ecosystem function can also depend on the surrounding

catchment, a net primary production (NPP) raster (500m resolution) based on MODIS satellite

images[90] was clipped to each catchment and averaged across all pixels in a catchment to derive

NPP.

Testing the effects of chemodiversity and biodiversity on ecosystem multifunctionality

We first determined which variables influenced multifunctionality in the lab experiment and

field survey by fitting a linear model that only contained biodiversity and chemodiversity. We

then fitted a second linear model that included an interaction between biodiversity and

chemodiversity to test if any effects of chemodiversity on multifunctionality were conditional

on biodiversity. For the lab experiment, chemodiversity and biodiversity were treated as

factors, as we only had ≤4 levels of each variable, whereas we used them as continuous

variables in the field survey. We used the same model for testing the effect of chemodiversity

on multifunctionality as for testing the effect of chemodiversity on each individual function in

the laboratory experiment. However, we added mean annual temperature, water colour, lake

area, and chlorophyll *a* concentration to model each function in the field survey because

previous studies have revealed that these variables also predict ecosystem function[23,24]. All

models were fitted with R version 4.1.1[78].

In addition to the linear models, we tested explicit hypotheses about the drivers of

chemodiversity-ecosystem multifunctionality relationships using a path analysis. Here, we

designed a path analysis explaining variation in multifunctionality, chemodiversity, the number

of biochemical transformations, and DOM complexity. We hypothesized that variation in

multifunctionality was explained by chemodiversity, as well as biodiversity, MAT, and MAP,

which all predict ecosystem multifunctionality in other studies[3,10,38]. We then hypothesized that

chemodiversity was explained by its complexity and potential transformation consistent with

established theories for the reactivity of DOM[27]. Based on the same theories, we hypothesized

that the number of times a molecule was transformed was also influenced by its complexity

and all the environmental variables we measured, including biodiversity. Finally, we

hypothesized that DOM complexity was influenced by terrestrial primary production that

naturally produces many complex compounds that are leached into downstream waters[65].

Although microbial activity also increases complexity[13], we expected its effect to be minimal

because most DOM in lakes originates from terrestrial production[16]. We then fitted this path

analysis to lake-level observations using the piecewiseSEM package[91] in R, which translates

our hypotheses into a set of linear models. After fitting the initial path analysis, we included

missing paths identified by tests of directed separation as described in Fig. S13.

Identifying the production and consumption of individual compounds

For the laboratory experiment, we first identified individual molecular formulae that were

consumed or produced by microbes. We compared the relative abundance of each molecular

formula at the end of the incubation experiment from all biodiversity levels against the relative

abundance of the same formula in the control with a Welch's t-test. We only considered the

final time point of the control to account for compounds that could be produced or consumed

simply due to abiotic processes. We considered compounds were produced and consumed if

their abundance significantly increased and decreased compared to the control, respectively.

We then summed the total number of produced and consumed compounds at each

chemodiversity level.

For the field survey, we tested how the consumption and production of formulae was associated

with each function that varied with chemodiversity across Europe (CH_4 and CO_2 emissions and

bacterial biomass production). We correlated the relative abundance of each formula in each

lake with the value of each function in the corresponding lake using Spearman rank

correlations. We considered statistically significant negative and positive correlations to

indicate that consumption and production of a formula, respectively, was associated with a

given function.

Extrapolating chemodiversity and ecosystem function from our field survey

To extrapolate chemodiversity and each measured function that varied with chemodiversity

across Europe (CH_4 and CO_2 emissions and bacterial biomass production), we first fitted linear

models to these variables. The models included all physio-chemical and environmental

variables measured in the field survey, and we included chemodiversity and biodiversity as

predictors when modelling each function. Then, we simplified the full models with stepwise

variable selection guided by the Akaike information criterion (AIC). We sequentially removed

the least supported variable, that is, most increased AIC when retained. If all variables reduced

AIC when retained, we dropped the variable that least reduced AIC by <2 units[92]. We repeated

this selection process until all variables retained in the model reduced AIC by ≥2 units. The

goal of this analysis was to predict each function as accurately as possible to extrapolate across

Europe rather than simply compare the importance of chemodiversity versus biodiversity as

when modelling multifunctionality. Model selection indicated that chemodiversity was best

explained by NPP, MAP, and chlorophyll *a* concentration (Table S3), which was identical to

the variables that explained variation in chemodiversity within our path analysis (Figs 2, S13).

To extrapolate our linear models across Europe in 2019, we forced them with the same

MAP and NPP rasters described above. The other predictors required in our models were water

colour and chlorophyll *a* concentration (Table S3, S4), which are unavailable at a European-

scale in publicly available datasets. Therefore, we predicted water colour from elevation, MAP,

and lake area using a linear model identical to past global studies predicting water colour[75].

Similarly, we predicted chlorophyll *a* from MAT, elevation, MAP, and lake area using another

linear model. Past global studies have used all these same variables to predict chlorophyll *a*,

except that they included TP, which was also unavailable from publicly available datasets at a

European-scale[23,24]. Although the models explained 25% and 29% of the variation in water

colour and chlorophyll *a*, respectively, the errors introduced by using these models are not

relevant for our calculations focusing on relative changes over time (see section below).

Forecasting the future functioning of lakes

We estimated how each function that varied with chemodiversity (CH_4 and CO_2 emissions and

bacterial biomass production) would change across Europe by 2100 solely because of changes

to chemodiversity. To predict chemodiversity in 2100, we first obtained the IPSL-CM6A[60],

HadGEM3-GC31[93], and ACCESS-ESM1.5[94] climate forecasts for 2100. We selected these

forecasts because they emphasise the relevance of terrestrial-atmosphere exchange and are

frequently used in climate studies[95]. We downloaded rasters of MAT and MAP for likely

(SSP2) and worst case (SSP5) scenarios for future changes in atmospheric CO_2 concentrations

for each forecast from the Copernicus Climate Data Store (https://cds.climate.copernicus.eu/).

NPP predictions for each forecast were downloaded from the Earth System Grid Federation

servers (https://esgf-node.llnl.gov/search/cmip6/). Because no NPP data were available for the

HadGEM3-GC31 model, we used the average of the IPSL-CM6A and ACCESS-ESM1.5

models. We then input these rasters into our linear model of chemodiversity to predict

chemodiversity in 2100 across Europe (Table S3). To generate the predictor of chlorophyll *a*,

we again used the model described in the previous section but with MAT and MAP from the

climate forecasts, as well as elevation and lake area. The chemodiversity raster was then input

into the linear models of CH_4 and CO_2 emissions and bacterial biomass production (Table S4).

All predictors other than chemodiversity were set at their 2019 values. Rather than focus on

absolute values of each function, we calculated the change in each function between 2019 and

2100 when only changing chemodiversity, that is, holding all other variables constant. Because

errors associated with models of water colour and chlorophyll *a* are likely to be systematic, that

is, non-random, the errors should remain similar between time periods with chemodiversity

being the sole difference in the comparison. Furthermore, estimates of the magnitude of

changes of CH_4 and CO_2 for all of Europe were well in line with previous global studies[23,24],

further suggesting that errors in absolute values of water colour and chlorophyll *a* were of

minor importance.

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