Measuring serotonin binding to its receptors via charge transfer to ANAP

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Abstract

Serotonin (5-HT) is a vital intercellular messenger with diverse signaling functions throughout

the human body. We have characterized and implemented a novel, fluorescence-based method

of measuring 5-HT binding to gain a fuller understanding of the interactions between 5-HT and

its receptors. This method involves expression of 5-HT receptor proteins with the fluorescent,

non-canonical amino acid L-3-(6-acetylnaphthalen-2-ylamino)-2-aminopropanoic acid (ANAP)

incorporated into the ligand binding site. ANAP fluorescence was quenched in solution by both

5-HT and dopamine. Time-resolved photoluminescence and transient absorption spectroscopy

confirmed that ANAP quenching by 5-HT involved a rapidly reversible charge-transfer process.

Based on density functional theory calculations, this process likely involved a ANAP reduction

by 5-HT as assessed using whole-cell patch clamp. To test this method on intact receptors in a

cellular context, we expressed 5-HT$_{3A}$ receptors (5-HT-gated ion channels) in HEK293T cells

with ANAP inserted co-translationally into the transmitter binding site. Fluorescently labeled 5-

HT$_{3A}$ receptors were functional and activated by 5-HT. Addition of 5-HT caused a concentration-

dependent quenching of fluorescence from ANAP-tagged channels expressed in intact cells and

unroofed plasma membranes, demonstrating the utility of this method for measuring 5-HT

binding to its receptors. Collectively, these results delineate a technique for measuring

transmitter binding that can be widely adopted to measure 5-HT binding not only to 5-HT$_3$

receptors, but to any 5-HT receptor, transporter, or binding protein.

Introduction.

Serotonin (5-hydroxytryptamine, 5-HT) is a near-ubiquitous intercellular messenger, playing

crucial roles in blood, the gut, and the central nervous system (CNS) (Hensler, 2012). In the

CNS, 5-HT is primarily produced in the raphe nuclei of the brainstem, which project widely

throughout the brain and spinal cord. Through these projections, 5-HT modulates many diverse

neurobiological phenomena including arousal, mood, and pain transmission (Ossipov et al.,

2010; Kandel et al., 2021). Serotonergic dysfunction has been linked to numerous disease

states including major depressive disorder, drug use disorder, and inflammatory bowel disease.

Though a rich appreciation for 5-HT signaling and its interaction with its receptors is crucial to

gain a better understanding of its role in health and disease, existing methods for measuring 5-

HT binding fall short of providing direct, real-time information about receptor occupancy. In this

work, we propose, fully characterize, and employ a novel fluorescence-based method that

directly reports the binding of 5-HT its receptors in real time.

5-HT binds to a diverse group of receptor molecules. In humans, 5-HT receptors are grouped

into seven families (5-HT_1 through 5-HT_7), based on sequence homology (Hensler, 2012). Each

family encompasses several receptor subtypes. With the exception of the 5-HT_3 family of

receptors, all 5-HT receptors are G-protein coupled; binding of 5-HT prompts dissociation of

heterotrimeric G-proteins from its receptors triggering downstream, intracellular signaling

events. In contrast, 5-HT_3 receptors (5-HT_{3A}-5-HT_{3E} in humans) are part of the pentameric

ligand-gated ion channel (PLGIC) superfamily (Maricq et al., 1991; Gibbs and Chakrapani,

2021). Binding of 5-HT opens an aqueous pathway for cation flow across the plasma

membrane, generating an excitatory current. 5-HT_3 receptors are important regulators of brain-

gut signaling and the vomiting response and have been implicated in several disease processes

including inflammatory bowel disease (IBS), schizophrenia, and substance use disorder

(Engleman et al., 2008; Ellenbroek and Prinssen, 2015; Vahora et al., 2020).

To gain a richer understanding of structure-activity relationships in 5-HT receptors, we sought

to develop an assay that would allow for real-time measurement of transmitter binding.

Equilibrium binding assays operate over long time scales and non-physiological conditions.

Biochemical assays of 5-HT binding depend on downstream signaling events and do not

measure binding directly. Even measuring current evoked in voltage-clamp experiments on 5-

HT_3 receptors provides only indirect information regarding ligand binding, as the binding

equilibrium is directly coupled to the gating process.

An ideal ligand binding assay would meet five criteria: 1) it would allow for the measurement

of 5-HT binding in real time, 2) be non-destructive and chemically reversible, 3) permit

measurements to be made on intact, functional channels in living cells, 4) have a short distance

dependance so binding can be measured site-specifically, and 5) be compatible with high-

throughput biochemical assays and (for $5\text{-}HT_3$ receptors) voltage clamp. In previous work, we

developed such an assay to measure nucleotide binding to ATP-sensitive K^+ channels

(K_{ATP})(Puljung et al., 2019; Usher et al., 2020b). This assay involved expression of channels

labeled with the fluorescent, non-canonical amino acid L-3-(6-acetylnaphthalen-2-ylamino)-2-

aminopropanoic acid (ANAP, Fig. 1A) (Chatterjee et al., 2013). ANAP can be introduced site-

specifically into any site on a protein using amber stop-codon suppression. We previously used

Förster resonance energy transfer (FRET) between ANAP and fluorescent ATP derivatives to

measure binding to K_{ATP} (Puljung et al., 2019; Usher et al., 2020b). The strong distance-

dependence of FRET allowed us to measure binding in a site-specific fashion (i.e. we were able

to discriminate binding between three different classes on nucleotide-binding site on K_{ATP}).

We now report a method that allows measurement of 5-HT binding to its receptors and that

satisfies the five criteria stated above. Reasoning that Prodan (structurally analogous to the side

chain of ANAP) fluorescence is quenched by tryptophan and that 5-HT is derived from

tryptophan, we first demonstrate that 5-HT, as well as dopamine (DA), quench ANAP in solution.

We further investigate this process using density functional theory, time-resolved

photoluminescence and transient absorption spectroscopies to demonstrate that quenching

occurs via a short-range electron transfer process that reversibly reduces ANAP. Finally, we

expressed ANAP-labeled 5-HT$_{3A}$ receptors in HEK293T cells and used this quenching to

measure 5-HT binding to channels in the plasma membrane of living cells. These results point

to the broad applicability of ANAP for directly interrogating the binding of 5-HT and dopamine to

relevant receptors, allowing for an expanded understanding of their physiological signaling

roles, as well as how dysfunction in that signaling produces pathological states.

Materials and methods.

UV/Vis absorbance and steady-state fluorometry. Stock solutions of ANAP (ASIS ChemInc.,

Waltham MA) were prepared at 20 mM in DMSO and diluted into DMSO or water. 5-HT and DA

stocks were prepared at 200 mM in water or DMSO. Stocks were stored frozen at -20 °C. To

reduce the impact of oxidation, all 5-HT and DA solutions were prepared fresh the day before

experiments and kept frozen until the time of measurement. The absorbance of each ANAP/5-

HT or ANAP/DA solution was measured using a Red Tide UBS650U spectrometer (Ocean

Optics; Orlando, FL) and LoggerPro software (Vernier; Beaverton, OR). 5-HT and DA blank

solutions were measured separately. All spectra were acquired using a low-volume quartz

cuvette (Azzota Corp.; Claymont, DE).

The fluorescence emission spectrum of 20 µM ANAP in water or DMSO was measured in

different concentrations of 5-HT or DA, as indicated. Spectra were acquired using a Hitachi F-

7000 fluorescence spectrophotometer and FL Solutions software. The samples were excited at

370 nm and emission was measured between 380 nm and 600 nm. Slit widths were 5 nm on

the excitation and emission side and the PMT was set to 700 V. Spectra were blank subtracted

using emission spectra of 5-HT/DA solutions without ANAP. DA contributed very little to the

fluorescence background, whereas 5-HT accounted for 15.2% of the total fluorescence signal at

the peak wavelength for ANAP fluorescence (448 nm) measured for 20 µM ANAP plus 100 mM

5-HT in DMSO.

Emission spectra were corrected for the inner-filter effect (as indicated) using absorbance

values measured as above and the following equation

$$F = F_{obs} * 10^{\frac{(0.2*OD_{ex}+OD_{em})}{2}}.$$

F_{obs} is the measured emission, OD_{ex} is the optical density at the excitation wavelength (370 nm)

and OD_{em} is the optical density at each emission wavelength. The 0.2 term corrects for the low-

volume cuvette, which had a short, 0.2 cm path oriented toward the excitation light. Before

correction, UV/Vis spectra were smoothed in JMP (JMP Statistical Discovery LLC, Cary, NC)

using local kernel smoothing with a local bandwidth of 0.25 nm.

Stern-Volmer plots were fit with a linear equation of the form

$$\frac{F}{F_0} = F_{intercept} + K_{sv} * [quencher],$$

where F is the averaged peak fluorescence between 445 nm and 455 nm for each solution, F_0 is

the average fluorescence in the absence of quencher, and K_{sv} is the Stern-Volmer constant,

reflecting the quenching efficiency. $F_{intercept}$ was the value extrapolated from the linear fit to zero

[quencher]. This value is expected to be one and was found to be close to one in our fits.

Density functional theory calculations. The relative stability of fluorophore/quencher electron

transfer pairs were computed using density functional theory (DFT). Q-Chem 5.2 was used to

computer neutral singlet structures, as well as reduced and oxidized doublet structures, of each

compound using the ωB97XD density functional and aug-cc-pVDZ basis set and a polarizable

continuum model ($\varepsilon = 80$) (Chai and Head-Gordon, 2008; Shao et al., 2015). Doublet

calculations were restricted to prevent spin contamination (Bursch et al., 2022). Following initial

structure optimizations, harmonic frequency analysis confirmed that each structure represented

a true local minimum (no imaginary frequencies) and provided values to determine ΔH and ΔS

corrections at 310 K for the calculation of theoretical ΔG values (Tartakoff et al., 2021).

Thermodynamics comparisons and plotting were performed using Python 3.9.13 with numpy

1.26.2 and matplotlib 3.8.0.

Time-resolved spectroscopies. Time-resolved photoluminescence (TRPL) and transient

absorption spectroscopy (TAS) measurements were performed using an Edinburgh Instruments

LP980 with a Litron Nano 5 Hz pump laser tripled to produce 355 nm, 8 ns pulses (FWHM).

Both TRPL and TAS data were taken with the LP980's PMT detector configured in its highest

time-resolution mode, corresponding to 400 ps time bins. TRPL measurements were performed

with the laser attenuator set to 10 mJ/pulse on solutions of 100 nM ANAP or 1.0 mM 5-HT in

DMSO; emission was measured at 480 nm for 64 averages. For TAS measurements, laser

power was lowered to 1 mJ/pulse and concentrations were increased to 1.0 mM ANAP and 10

mM 5-HT and were followed via ANAP radical anion peak at 440 nm for 500 averages (made

possible by the complete recovery of signals between shots.) Data were fit to a convolution of

the instrument response function (IRF; measurement of elastically scattered laser pulse after

passing through ND = 3 filter) and exponential decay function(s) using Edinburgh Instruments'

L900 software (Kolber and Barkley, 1986). Additional analysis and plotting were performed

using Python (see above).

Molecular cloning. Mouse 5-HT$_{3A}$ in pICherryNeo was a gift from Dr. Sudha Chakrapani.

pCGFP_EU was a gift from Dr. Eric Gouaux (Kawate and Gouaux, 2006). mOrange-N1 was a

gift from Michael Davidson (Addgene plasmid # 54499 ; http://n2t.net/addgene:54499 ;

RRID:Addgene_54499) (Kremers et al., 2009). pANAP was a gift from Peter Schultz (Addgene

plasmid # 48696 ; http://n2t.net/addgene:48696 ; RRID:Addgene_48696) (Chatterjee et al.,

2013).

As a preliminary step, the GFP insert from pCGFP_EU was cloned into the 5-HT$_{3A}$ sequence

in pICherryNeo at a position between amino acids D382 and S384 in the putatively unstructured

loop between transmembrane domains 3 and 4 (Basak et al., 2018). We used PCR

(QuickChange Kit; Agilent Technologies; Santa Clara, CA) to introduce new restriction sites into

the 5-HT$_{3A}$ sequence and inserted the GFP tag using standard subcloning techniques. We

subsequently replaced this GFP tag with mOrange because the emission peak for mOrange

was farther from the expected ANAP peak. Stop-codon mutations were introduced using

QuickChange. All constructs were verified by direct sequencing at Azenta GeneWiz (South

Plainfield, NJ).

Cell culture. HEK-293T cells were purchased from ATCC (Manassas, VA). Frozen stocks were

prepared during early passages and stored in liquid nitrogen. Cells were verified by ATCC. We

performed no additional verification or mycoplasma testing. Cells were grown in T-75 flasks

(Grenier Bio-One; Monroe, NC) in Dulbecco's Modified Eagle Medium (DMEM; Gibco; Grand

Island, NY) supplemented with 10% fetal bovine serum (FBS; Corning; Tewksbury, MA), 100

U/mL penicillin, and 100 µg/mL streptomycin (Gibco) at 37 °C in a 5% CO_2/95% air atmosphere.

Expression of ANAP-tagged proteins. ANAP-tagged proteins were obtained as previously

described (Chatterjee et al., 2013; Puljung et al., 2019; Puljung, 2021). Cells were co-

transfected with 1 µg of 5-HT$_{3A}$ plasmid with an amber stop codon (TAG) replacing the codon

corresponding to amino acid Y234 and downstream mOrange tag, as well as pANAP, which

encodes a tRNA with the appropriate anticodon (CUA) to recognize the amber stop codon and a

synthetase enzyme capable of charging the tRNA with ANAP. Cells were transfected using

TransIT transfection reagent (Mirus; Madison, WI) at a ratio of 3 µL/µg of DNA. Just prior to

transfection, the culture medium was replaced with DMEM supplemented with FBS,

penicillin/streptomycin, and 20 µM ANAP methyl ester (AsisChem Inc.; Waltham, MA).

Experiments were typically performed 2 days post-transfection.

Electrophysiology. Currents were recorded in the whole-cell patch-clamp configuration. Data

were acquired using a Sutter IPA amplifier controlled by SutterPatch software (Sutter

Instruments; Novato, CA). Data were digitized at 10 kHz and low-pass filtered at 1 kHz. Patch

pipettes were pulled from borosilicate glass (World Precision Instruments 1B150F-4) to a

resistance of 8-10 MΩ. The pipette solution contained 140 mM KCl, 1.2 mM MgCl$_2$, 2.6 mM

CaCl$_2$, and 10 mM HEPES, pH 7.4 with N-methyl-D-glucamine. The bath solution contained 140

mM NaCl, 2.8 mM KCl, 2 mM MgCl$_2$, 1 mM CaCl$_2$, and 10 mM HEPES, pH 7.4 with N-methyl-D-

glucamine. Bath solution was exchanged using a MiniStar peristaltic pump (World Precision

Instruments). 5-HT was applied in bath solution (supplemented with 5 mM glutathione) using a

pressure-driven SmartSquirt Micro-Perfusion System (AutoMate Scientific; Berkeley, CA).

Unroofing HEK cells for imaging. Cells were unroofed to obtain isolated plasma membranes

(Heuser, 2000; Usher et al., 2020a). A small fragment of glass coverslip with adherent HEK293T

cells expressing labeled channels was broken off using jeweler's forceps. The fragment was

dipped in a solution of 0.1% poly-L-lysine for 3x 10 s. Fragments were blotted (cell-side down)

on a clean piece of filter paper (Cytiva Whatman Grade 1; Wilmington, DE), leaving behind

adherent plasma membrane fragments.

Microscopy/spectroscopy in living cells/cell fragments. For experiments, cells were plated in 35

mm dishes (Thermo Scientific; Waltham, MA) on glass coverslips (#1 thickness; Chemglass Life

Sciences; Vineland, NJ) or on FluoroDishes (World Precision Instruments; Sarasota, FL) a day

before transfection. Cells were imaged in the same bath solution used for patch clamp using a

Nikon TE2000-U (Melville, NY) or Ti2E microscope equipped with a 60X water-immersion

objective (Nikon Plan Apo VC, 1.20 NA). ANAP was excited using a ThorLabs LED source

(LED4D067; Newton, NJ) with a center wavelength of 385 nm. For imaging ANAP, the filter set

contained a 390/18 nm band-pass filter (MF390-18, ThorLabs), an MD416 dichroic mirror

(ThorLabs) and a 470/40 nm band-pass filter (MF479-40, ThorLabs). To obtain spectra, the

emission filter was replaced with a 400 nm long-pass filter (FEL0400, ThorLabs). To image the

mOrange tag, the same light source was used with a broad LED centered at 565 nm. A 530-30

band-pass filter was used for excitation (Chroma; Bellows Falls, VT) with a T550lpxr dichroic

mirror (Chroma) and 575/50 band pass emission filter (Chroma).

Images and spectra were collected using an IsoPlane81 camera/spectrograph (Teledyne

Princeton Instruments; Princeton, NJ) or IsoPlane160/PIXIS 256 camera (Teledyne Princeton

Instruments). Image/spectrum acquisition was automated using a Sutter IPA amplifier controlled

by SutterPatch software (Sutter Instruments). Exposure times were typically 10 s.

Photobleaching artifacts were corrected as previously described (Puljung et al., 2019; Usher et

al., 2020a). Briefly, five images were acquired in the absence of 5-HT, the peak intensity of the

ANAP spectrum was plotted as a function of exposure time, and the intensity decay was fit with

a single exponential. The resulting fit curve was used to correct subsequent exposures. For

quenching experiments, 5-HT in bath solution (or bath solution + 5 mM glutathione in unroofed

experiments) was applied using a SmartSquirt perfusion system controlled by SutterPatch.

Wash times were 10-30 s to ensure that ligand binding reached a steady state. The FluoroDish

was perfused with bath solution using a MiniStar peristaltic pump. Solutions were diluted from

frozen stocks just prior to the experiment and stored on ice.

Data analysis. Spectra and chromatograms were analyzed and plotted using custom code

written in R version 4.5.0, with plotly, hyperSpec, beepr, ggplot2, gridExtra, and xlsx packages

(Wickham, 2016; Auguie, 2017; Dragulescu and Arendt, 2020; Sievert, 2020; Beleites and

Sergo, 2024; Bååth, 2024; R Core Team, n.d.). Time-resolved spectroscopic data were

analyzed using custom code in Python 3.9.13. Images were analyzed and displayed using Fiji

(Schindelin et al., 2012). Additional plots, and curve fits were carried out using JMP 2025 JMP®

(version 18.2.2; JMP Statistical Discovery LLC, Cary, NC).

To compare concentration-response relationships between cells/membranes, peak

fluorescence from bleach-corrected spectra from each cell/membrane were normalized by

subtracting the minimum fluorescence at saturating concentrations and dividing by the peak at

the lowest applied 5-HT concentration (where no appreciable quenching was evident).

Concentration-response data were fit with curve of the form

$$\frac{(F-F_{min})}{(F_{max}-F_{min})} = \frac{1}{1+10^{h*((\log EC_{50}-\log[5-HT])}},$$

where [5-HT] is the concentration of 5-HT, F is fluorescence, F_{max} is the fluorescence intensity at

the lowest [5-HT] concentration (where we observed no appreciable quenching), F_{min} is the

fluorescence intensity at the highest [5-HT], and EC_{50} is the concentration that produced half-

maximal quenching.

Data presentation and statistics. For solution spectra, data are displayed as the average of

three samples (two for 20 μM ANAP in DMSO). For data from cells and unroofed membranes,

all of the individual data points were displayed. Errors are displayed as mean ± SEM. *n* values

and fit parameters are included in the figure legends. No power analyses were conducted prior

to our experiments to determine the number of repetitions. Protein structures were displayed

using ChimeraX developed by the Resource for Biocomputing, Visualization, and Informatics at

the University of California, San Francisco with support from National Institutes of Health RO1-

GM129325 and the Office of Cyber Infrastructure and Computational Biology, National Institute

of Allergy and Infectious Disease (Pettersen et al., 2021). Chemical structures were created and

formula masses predicted using ChemDraw (Revvity Signals Software; Waltham, MA).

Materials. Unless otherwise noted, all chemicals were obtained from Sigma-Aldrich (St. Louis,

MO). Restriction enzymes were purchased from ThermoFisher and New England Biolabs, Inc.

(Ipswich, MA).

Results.

Steady-state quenching of ANAP by 5-HT and DA in solution. The fluorophore Prodan,

structurally analogous to the side chain of ANAP, is quenched by tryptophan in micelles and

when both molecules are incorporated into cytochrome P450 (Pospíšil et al., 2014). Given its

similarity to Prodan and its utility as a site-specific protein label, we sought to determine whether

ANAP could be quenched by 5-HT, a neurotransmitter derived from tryptophan. We first

measured the quenching of ANAP by 5-HT in aqueous solutions. At high concentrations (10

mM-50 mM), 5-HT caused a blue shift in ANAP fluorescence (from 499 nm to 487 nm) as well

as a 53% reduction in the peak intensity (Fig. S1). The blue shift was likely the result of ANAP's

reported environmental sensitivity (Chatterjee et al., 2013). To reduce any solvatochromatic

shifts and focus solely on quenching, we measured the spectrum of 20 µM at varying 5-HT

concentrations in DMSO. In DMSO, the peak emission wavelength was 451 nm. Thus, this

greatly reduced any additional the shift induced by 5-HT binding.

In DMSO, 5-HT greatly attenuated the peak ANAP fluorescence in a concentration-dependent

manner (Fig. 1B). Quenching was evident at concentrations of 5 mM 5-HT and above. A Stern-

Volmer plot of the quenching data was well fit with a straight line (Fig. 1C; K_{sv} = 13.3 M^{-1},

intercept = 0.98, R^2 = 0.998). The linear slope and concentration-dependence are both

consistent with collisional quenching. At high concentrations, 5-HT absorbs light over the

concentration range at which ANAP is excited and emits light (Fig. S2A). Therefore, the

absorbance spectrum of each solution was measured and used to correct for the inner-filter

effect. Corrected data for 5-HT quenching are shown in Fig. S3A. Whereas there was some

deviation from linearity after applying this correction, the resulting Stern-Volmer plot was still

well fit with a straight line (Fig. S3B; K_{sv} = 4.31 M^{-1}, intercept = 0.98, R^2 = 0.977). At low 5-HT

concentrations, we observed a slight increase in fluorescence rather than quenching,

particularly in the corrected data (Fig. S3A). We attribute this change to a residual

272 solvatochromatic effect, such as the one we observed in water and the one reported by

273 Chatterjee *et al.* which shows an increase in the apparent quantum yield of ANAP

274 accompanying the blue shift in emission (Chatterjee et al., 2013). Consistent with this, we

275 observed a small, reproducible shift the peak wavelength for ANAP emission from 451 nm to

276 447 nm in the presence of 100 mM 5-HT (Fig. 1B).

277  In addition to its significant role as a modulatory neurotransmitter, DA is a weak partial agonist

278 of 5-HT$_3$-type receptors (Lovinger et al., 2000). Therefore, we tested DA's ability to quench

279 ANAP fluorescence in solution. Similar to 5-HT, DA quenched ANAP fluorescence in a

280 concentration-dependent manner (Fig. 1D). The Stern-Volmer plot for DA quenching was linear

281 (Fig. 1E; K_{sv} = 2.70 M^{-1}, intercept = 0.99, R^2 = 0.975). The shallower slope compared to 5-HT

282 implies a less efficient quenching process for DA (Lakowicz, 2006). DA does not absorb light as

283 strongly as 5-HT over the wavelength range at which ANAP absorbs and emits (Fig. S2B).

284 Nevertheless, we corrected for the inner-filter affect using the absorbance spectra measured for

285 each solution (Fig. S3C). The Stern-Volmer plot for corrected data was still linear and produced

286 a fit with similar parameters (Fig. S3D; K_{sv} = 2.71 M^{-1}, intercept = 0.98, R^2 = 0.953). Quenching

287 of ANAP fluorescence by 5-HT and DA is consistent with criterion 1 of 5, suggesting that ANAP

288 can serve as a useful 5-HT detector.

289 *Time-resolved spectroscopies.* Excited-state behaviors of ANAP and 5-HT were further

290 interrogated using time-dependent spectroscopies. Time-resolved photoluminescence

291 measures light emitted per unit time over sub-nanosecond time bins, revealing the fluorescent

292 lifetimes of excited species following pulsed laser excitation. TRPL measurements at 480 nm

293 following excitation with a 355 nm pump laser revealed fluorescence lifetimes shorter than the

294 laser pulse full width at half max (FWHM) of the laser pulse but within range of our detector time

295 resolution (400 ps), producing data most accurately fit by a convolution of IRF and a single

296 exponential decay (Fig. S4); no multiexponential signals were observed. Lifetimes were 2.231 ±

0.001 ns for 5-HT and 2.776 ± 0.002 ns for ANAP, with the latter exhibiting the expected

qualitatively much larger molar absorptivity and emission. The lifetimes were broadly consistent

with published values for ANAP (described by two exponential components of 1.3 ns and 3.3 ns)

and 5-HT (4 ns) (Nag et al., 2008; Zagotta et al., 2021), which were measured in different

solvents. Our TRPL measurements of emission from a mixture of both compounds exhibited the

expected individual 5-HT and ANAP signal components but with decreased overall intensity.

Transient absorption spectroscopy uses a white-light "probe" to quantify the time-dependent

change in UV-visible absorbance of our system following a laser excitation "pump" pulse; TAS

can thus measure the dynamics of the short-lived species that exist in solution after ANAP is

quenched by 5-HT. TAS measurements required increasing fluorophore and quencher

concentrations to reduce fluorescence background; even so, this background overwhelmed the

detector and resulted in artifacts for the few nanoseconds at peak irradiance. Following this

artifact, a clear product signal appears and persists for several nanoseconds. To address

residual electromagnetic interference (EMI; oscillating pattern) remaining in the TAS signal, we

again fit with a convolution of the IRF; a single exponential function provided the best fit

(superior to second-order-kinetics t^{-1} or biexponential) and indicated that this product peak fully

decayed with a lifetime of 3.18 ± 0.08 ns, consistent with a chemically reversible process (Fig.

2A). The slight mismatch between the EMI noise phase in the IRF and TAS data is the result of

a beating effect stemming from the signal math necessary to produce the TAS data:

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$$\Delta absorbance(t) = -log_{10}\left(\frac{I_{pumped}(t) - I_{fluorescence}(t)}{I_{unpumped}(t)}\right)$$

A negative ΔOD signal (bleach) may also be observed in TAS measurements, resulting from a

reduction in concentration of light-absorbing ground-state ANAP. No bleach signal was

observed, though this may be attributable to the strong absorbance of the solution itself in the

area around 350 nm where such a bleach would be expected, resulting in poor signal:noise

obscuring the bleach. The initial conditions were fully recovered between shots.

Further conclusions can be drawn from comparing this TAS signal to that observed from the

separate solutions of ANAP and 5-HT. ANAP shows a permanent increase in absorbance

following the pump pulse, indicating an irreversible change following excitation, whereas 5-HT

showed no significant TAS signals. If it were the case that ANAP and 5-HT did not interact,

summing the signals of separate ANAP and 5-HT solutions should reproduce the signal of the

mixture; instead, the summed signal shows a permanent increase in signal in comparison with

the ANAP/5-HT mixture's nearly complete recovery (Fig. 2B).

While FRET or Dexter quenching mechanisms would result in the fluorophore/quencher pair

being completely returned to their electrically neutral electronic ground states following the

quench, the existence of a longer-lived 3.18 ns signal is consistent with the formation of a

charge-transfer state that persists in solution after the quench event has occurred (Kandori et

al., 1992; Alsam et al., 2015). In conjunction with DFT results (see below), we therefore attribute

the observed product signal to an ANAP$^{\bullet-}$ radical anion resulting from reduction of ANAP by 5-

HT. As we have demonstrated in earlier work, aromatic radical anions typically show visible

absorbances redshifted from those of their neutral parent species due to the additional

occupation of π* orbitals (Hill et al., 2018).

Density Functional Theory. Whereas time-resolved measurements revealed a nanosecond-

lifetime excited state consistent with charge transfer (CT) as the mechanism behind the

quenching of ANAP, DFT calculations provided additional insight into the directionality of the CT

process (Fig. 3). Comparison of the energy of a reduced/oxidized pair (e.g., ANAP$^-$/5-HT$^+$) to

the ground-state combination (ANAP^0/5-HT^0) revealed the relative stability of each and

confirmed that for all combinations of fluorophores (L-ANAP methyl ester, Prodan) and

quenchers (5-HT, dopamine), the CT state was always significantly less stable (>70 kcal/mol)

than the neutral ground state, consistent with the expected behavior of a reversible CT process.

DFT results also consistently showed that electron transfer from quencher to fluorophore (i.e.,

fluorophore reduction) resulted in a comparatively more stable CT state (73.0–77.1 kcal/mol,

versus 93.2–100.5 kcal/mol for fluorophore oxidation), consistent with the expected greater

stability of an electron delocalized in a larger π system (Hoffmann and Olofson, 1966). Energies

required to access the fluorophore oxidation pathway correspond to <300 nm photons

unavailable in our experiments, whereas the energies associated with the quencher oxidation

pathway are in the ~400 nm range accessible with excitation sources employed in this work.

This further supports electron transfer from quencher to fluorophore as the quenching

mechanism.

Measurement of 5-HT binding to intact, functional 5-HT$_{3A}$ receptors. With the mechanism and

dynamics of ANAP quenching by 5-HT established spectroscopically and computationally, we

next expressed ANAP near the ligand binding site of 5-HT$_{3A}$ receptors. 5-HT$_{3A}$ is a PLGIC that is

activated by direct 5-HT binding to its extracellular domain (Fig. 4A). 5-HT$_{3A}$ forms functional,

homomeric receptors in native and heterologous expression systems (Lambert et al., 1989;

Maricq et al., 1991). Typical of PLGICs, 5-HT binds in a conserved ligand-binding pocket formed

at the interface of adjacent subunits, where it is coordinated by a cage of aromatic amino acids

(Fig. 4B). We sought to express 5-HT$_{3A}$ receptors in which one of the aromatic amino acids

(Y234) is replaced with ANAP (5-HT$_{3A}$-Y234ANAP). We chose this site because Y234

coordinates 5-HT directly and the expected quenching interaction (via charge transfer) should

only operate over very short distances (Fig. 4B). Furthermore, as tyrosine is a bulky aromatic

amino acid, substitution with ANAP should be relatively conservative.

To express 5-HT$_{3A}$-Y234ANAP channels, the gene for mouse 5-HT$_{3A}$ in the pICherryNeo

vector was mutated to replace the triplet encoding Y234 with the amber (TAG) stop codon.

When expressed alone in HEK293T cells, the resulting plasmid is expected to result in the

expression of 5-HT$_{3A}$ prematurely truncated at position 233 in the extracellular region preceding

the first transmembrane domain. To obtain full-length, ANAP labeled channels, cells were co-

transfected with pANAP, which encodes a tRNA that recognizes the amber stop codon and a

synthetase enzyme capable of charging this tRNA with ANAP. ANAP was added to the cell

culture media at the time of transfection. When both plasmids and ANAP are present (Fig. 5A),

full-length 5-HT$_{3A}$-Y234ANAP channels can be produced. As an additional control, we inserted

an mOrange protein tag in an unstructured cytoplasmic loop downstream of amino acid 234.

Thus, orange fluorescence (in addition to ANAP fluorescence) indicates successful

incorporation of the label and translation of full-length channels.

Figure 5A shows a series of images of HEK293T cells transfected with the plasmid encoding

5-HT$_{3A}$-Y234ANAP. When this plasmid was co-transfected with pANAP and ANAP was included

in the culture media, there was obvious fluorescence for both ANAP and mOrange, indicating

the presence of full-length, ANAP-labeled channels. When ANAP was omitted from the bathing

medium, we typically observed no fluorescence for either ANAP or mOrange, indicating that

there was no read-through of the amber stop codon without ANAP present. When ANAP was

present, but pANAP was not included in the transfections, again no mOrange fluorescence was

observed. The small, diffuse ANAP fluorescence we observed likely represents unincorporated

ANAP accumulating inside cells. However, we cannot exclude the possibility that ANAP may be

incorporated into other cellular proteins. Regardless, we identified cells for further

experimentation based on the presence of both ANAP and mOrange fluorescence.

Whereas it is necessary to place ANAP very close to the 5-HT in order to observe any short-

range quenching via charge transfer, mutating amino acids so near the ligand binding site runs

the risk of altering the receptor's response to 5-HT. Thus, we used whole-cell patch-clamp

electrophysiology to verify that 5-HT$_{3A}$-Y234ANAP receptors were functional and responded to

5-HT. Figure 5B shows a typical whole-cell current response to 240 µM 5-HT for an HEK293T

cell expressing 5-HT$_{3A}$-Y234ANAP. Bath application of 5-HT produced a desensitizing current

typical of 5-HT$_{3A}$. We observed an average of 41.2 ± 9.9 pA of current in response to

applications of 24 µM 5-HT in5/7 cells that we identified based on mOrange fluorescence. Thus,

we can conclude that 5-HT$_{3A}$-Y234ANAP receptors are present at the plasma membrane and

maintain their response to 5-HT, despite alterations to the ligand binding site.

To measure 5-HT binding to 5-HT$_{3A}$-Y234ANAP, we expressed the receptor in HEK293T cells

and imaged them using an inverted microscope coupled to a spectrograph. This allowed us to

obtain full emission spectra of ANAP-tagged channels in the plasma membrane (Fig. 5C). The

ANAP peak was 495 ± 5 nm, close to the value we measured in aqueous solution, suggesting

that ANAP at position 234 of 5-HT$_{3A}$ was accessible to bulk water. The secondary peak at ~560

nm corresponds to mOrange (emission maximum 562 nm) (Shu et al., 2006). Application of 5-

HT gave rise to a concentration-dependent quenching of ANAP fluorescence. The apparent

reduction in mOrange fluorescence was due to the overlap in ANAP and mOrange spectra. At

saturating concentrations, 5-HT did not quench 100% of the ANAP fluorescence. Quenching at

saturating concentrations was variable, with an average of 66% ± 7%. To compare across cells,

we subtracted the fluorescence at saturating [5-HT] from each trace and then normalized to the

maximum fluorescence at the lowest agonist concentration (Fig. 5D). The resulting

concentration-response relationship was well-fit with a modified Hill equation with half-maximal

quenching at 594 nM 5-HT and a slope factor of -0.46. The inability of 5-HT to quench 100% of

the fluorescence may be a result of the distance between ANAP and 5-HT or the geometry of

the binding site. However, it is possible that the unquenched fluorescence signal arose from

ANAP-tagged trafficking intermediates in the ER/Golgi that would not be accessible to 5-HT

applied extracellularly. Therefore, we repeated the quenching experiment in unroofed plasma

membranes obtained by blotting adherent cells with filter paper (Fig. S5). Blotting removes

intracellular organelles, thus reducing any background from trafficking intermediates. We

observed a similar concentration-response relationship for quenching in unroofed membranes

(EC_{50} = 160 nM, slope factor -0.49). We did not, however, obtain 100% quenching in unroofed

cells. Therefore, the inability of 5-HT to completely quench likely reflects the geometry of the

binding site or photophysics of the ANAP/5-HT interaction, not an inaccessible population of

channels.

Discussion.

Based on our results, we propose that ANAP can be used to detect 5-HT binding to its

receptors. Our assay meets all five criteria set out in the Introduction. By using fluorescence, we

measured binding in real-time to intact, functional channels in living cells. Based on our time-

resolved fluorescence and transient absorption measurements, we observed a reversible

quenching process that recovers with a time constant of 3.2 ns, likely due to back-electron

transfer from the reduced fluorophore. This recovery time for the intermolecular process

represents an upper bound to back-electron transfer within a ligand binding site and is

nonetheless much faster than the expected time course of 5-HT dissociation from the channel.

Thus, 5-HT can quench the excited ANAP molecules without any irreversible bleaching of the

fluorophore. Charge transfer is very sensitive to distance (requiring collision/contact between

the fluorophore and quencher). Therefore, ANAP/5-HT CT allows for detection of binding very

near to the site of ANAP incorporation and would not be sensitive to 5-HT bound elsewhere.

Replacing Y234 in the 5-HT binding site of 5-HT$_{3A}$ still allowed for binding of 5-HT and channel

activation. Thus, with sufficient signal-to-noise ratio, this binding assay is compatible with

simultaneous measurements of channel current so that ligand occupancy can be directly

correlated with changes in function (i.e. activation, desensitization). Finally, our fluorescence-

based assay can be readily scaled for use in a 96 or 384-well format to enable medium/high

throughput data acquisition.

We used ANAP-tagged 5-HT$_{3A}$ receptors in our assay because, as 5-HT$_{3A}$ is an ion channel,

we could record currents to verify that our ANAP-tagged receptors were functional, expressed in

the plasma membrane, and able to respond to 5-HT. However, this technique should be easily

adapted for use in G-protein-coupled 5-HT receptors and 5-HT transporters. Based on our

steady-state measurements, ANAP may also be a promising DA detector when incorporated

into receptors. DA is a partial agonist of 5-HT$_{3A}$ and comparing current responses and binding

measurements between 5-HT and DA may be a promising way to explore the mechanism of

partial agonism in these channels. Furthermore, ANAP can be incorporated into the binding site

of DA receptors to help illuminate their biological role.

The values we obtained from fits to our 5-HT binding data (EC$_{50}$ 594 nM) were in the range of

EC$_{50}$ values obtained from functional measurements of recombinant 5-HT$_{3A}$ activity. The EC$_{50}$

for activation of 5-HT$_{3A}$ in HEK293T cells measured with Ca^{2+} imaging was 348 nM (Dubin et al.,

1999). In two-microelectrode studies in *Xenopus* oocytes half-maximal current activation

occurred at 3.54 µM (Maricq et al., 1991). The differences in the values we obtained may be the

result of mutating the 5-HT binding site. Alternatively, the discrepancy may represent state-

dependent 5-HT binding. Based on the time course of the channel currents we measured (Fig.

5B) and the duration of our 5-HT applications, we assume that we measured the apparent 5-HT

affinity for desensitized receptors. However, one should exercise caution when interpreting EC$_{50}$

values as reflecting any true binding affinity. Even for assays that measure "binding" directly, any

values reported reflect not only the binding equilibrium, but any coupled conformational

equilibria in the channel complex (Colquhoun, 1998).

As presented, our method offers great promise for understanding the relationship between

binding of transmitters like 5-HT and DA and receptor activation. However, there are several

points of caution worth considering. Our primary difficulty is that, even at saturating

concentrations of 5-HT, maximal quenching of 5-HT$_{3A}$-Y234ANAP was only around 66% and

variable. Whereas the signal-to-noise ratio allowed us to generate reproducible binding curves

in whole cells and unroofed membranes, more robust and more reproducible quenching at

saturating 5-HT concentrations is desirable. Replacement of other aromatic amino acids in the

binding site with ANAP may produce better results.

We used long (10-30 s) applications of 5-HT in our experiments and 10 s exposure times for

our spectra. On this time scale, 5-HT$_{3A}$-Y234ANAP receptors were likely in the desensitized

state. To capture shorter-lived gating intermediates, faster solution exchange and shorter

exposure times will be necessary. The latter will be easier to achieve in preparations with

greater signal-to-noise. Alternatively, mutations that reduce sensitization or allosteric modulators

that delay it may help explore binding to the open state.

ANAP, with its sensitivity to both 5-HT and DA, shows great promise as a neurotransmitter

detector. ANAP is easily inserted co-translationally into any site on a target receptor using

amber stop-codon suppression. As ANAP insertion produced functional channels future

experiments will allow us to explore the relationship between ligand binding and channel gating

for both the full agonist 5-HT and the partial agonist DA. In our previous work, simultaneous

measurements of ligand binding and changes in current allowed us to constrain K_{ATP} gating

models and derive meaningful parameters regarding true ligand binding affinity and the

energetic coupling between binding and activation (Usher et al., 2020b). This approach may

also be used to understand how co-assembly of 5-HT$_{3A}$ with other subunits (5-HT$_{3B}$ through 5-

HT$_{3E}$) affects this process as well as how mutations associated with diseases like IBS, drug

abuse disorder, and schizophrenia can affect the channel's biophysical properties. Finally, this

technique can be readily adapted to study other 5-HT and DA receptors and transporters,

offering a real-time window into their structure and activity.

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Figure Legends.

Figure 1. Quenching of ANAP by 5-HT and DA in solution. A. Molecular structures of ANAP, 5-

HT, and DA. B. Emission spectra (excitation 370 nm) of 20 μM ANAP in DMSO in the absence

and presence of 5-HT. C. Stern-Volmer plot for 5-HT quenching of ANAP. The relationship was

fit with a straight line with an intercept of 0.98 and a slope (K_{sv}) of 13.3 M^{-1} (R^2 = 0.998). D.

Emission spectra (excitation 370 nm) of 20 μM ANAP in DMSO in the absence and presence of

DA. E. Stern-Volmer plot for DA quenching of ANAP. The relationship was fit with a straight line

with an intercept of 0.99 and a slope (K_{sv}) of 2.70 M^{-1} (R^2 = 0.975).

Figure 2. Transient absorption of 1.0 mM L-ANAP methyl ester and 10 mM 5-HT solution at 440

nm following excitation at 355 nm, showing (a) the appearance and recovery of a short-lived

product peak attributed to reduced ANAP; convolution fit lifetime of the resulting recovery was

3.18 ± 0.08 ns. (b) ANAP & 5-HT mixture (black; with 21-point second-order Savitzky-Golay

smoothing) exhibits complete recovery (green dashed line) while with a sum of signals from the

two separate solutions (violet) shows a long-lived transient signal (violet dashed line).

Figure 3. DFT-calculated redox ΔG values for electron transfer reactions of L-ANAP methyl

ester or Prodan with dopamine or 5-HT (ωB97XD density functional and aug-cc-pVDZ basis),

showing that fluorophore reduction/quencher oxidation is consistently more thermodynamically

stable. Structures shown above are those of the most redox-stable radical species.

Figure 4. Structure of 5-HT$_{3A}$. A. Cryo-electron microscopy structure of 5-HT$_{3A}$. Subunits are

individually colored. 5-HT is shown as a gray, space-filling model. B. One (of five) 5-HT binding

sites in 5-HT$_{3A}$ highlighting nearby aromatic residues. Structural models are from Basak *et al.*

((Basak et al., 2018), PDB accession number 6DG8).

Figure 5. 5-HT binding to 5-HT$_{3A}$-Y234ANAP. A. Brightfield and fluorescence micrographs

showing cells co-expressing the 5-HT$_{3A}$-Y234TAG plasmid and pANAP and cultured in the

presence (left) and absence (middle) of ANAP. The right panel shows cells expressing the 5-

HT$_{3A}$-Y234TAG plasmid without pANAP and cultured in the presence of ANAP. B.

Representative whole-cell currents in response to bath application of 240 μM 5-HT for a cell

expressing 5-HT$_{3A}$-Y234ANAP. C. Fluorescence emission spectra acquired from a cell

expressing 5-HT$_{3A}$-Y234ANAP in the presence of increasing concentrations of 5-HT. D. Data

from 7 cells expressing 5-HT$_{3A}$-Y234ANAP showing the normalized quenching by 5-HT. The

data were fit to a modified Hill equation (see Materials and Methods) with a slope of -0.46 and

EC$_{50}$ value of 594 nM. The shaded area shows the 95% confidence interval for the fit.

Supplemental Figure S1. Quenching of ANAP by 5-HT in water. Emission spectra (excitation

370 nm) of 20 μM ANAP in water in the absence and presence of 5-HT.

Supplemental Figure S2. Absorbance spectra of ANAP with 5-HT or DA in DMSO. A.

Absorbance spectra for solutions of 20 μM ANAP plus the indicated amount of 5-HT in DMSO.

B. Absorbance spectra for solutions of 20 μM ANAP plus the indicated amount of DA in DMSO.

The smooth lines are the spectra after local kernel smoothing was applied. Each spectrum is the

average of three samples.

Supplemental Figure S3. Inner-filter-corrected spectra. A. Emission spectra (excitation 370 nm)

of 20 µM ANAP in DMSO in the absence and presence of 5-HT, corrected for the inner-filter

effect (see Materials and Methods). B. Stern-Volmer plot for 5-HT quenching of ANAP. Data

were corrected for the inner-filter effect. The relationship was fit with a straight line with an

intercept of 0.98 and a slope (K_{sv}) of 4.31 M^{-1} (R^2 = 0.977). C. Emission spectra (excitation 370

nm) of 20 µM ANAP in DMSO in the absence and presence of DA, corrected for the inner-filter

effect. D. Stern-Volmer plot for DA quenching of ANAP. Data were corrected for the inner-filter

effect. The relationship was fit with a straight line with an intercept of 0.98 and a slope (K_{sv}) of

2.71 M^{-1} (R^2 = 0.953).

Supplemental Figure S4. Time-resolved photoluminescence at 470 nm following excitation at

355 nm of 10 mM 5-HT (top) and 100 nM L-ANAP methyl ester (bottom) with fits (red) calculated

from the convolution of the instrument response function (gray) and exponential decay, resulting

in fluorescence lifetimes of 2.231 ± 0.001 ns and 2.776 ± 0.002 ns, respectively.

Supplemental Figure S5. 5-HT binding to 5-HT$_{3A}$-Y234ANAP in unroofed plasma membranes.

Data from 3 unroofed plasma membranes from cells expressing 5-HT$_{3A}$-Y234ANAP showing the

normalized quenching by 5-HT. The data were fit to a modified Hill equation (see Materials and

Methods) with a slope of -0.49 and EC$_{50}$ value of 160 nM. The shaded area shows the 95%

confidence interval for the fit.

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Figure 1

DFT-calculated radical structures

DA•+

5-HT•+

Prodan•−

ANAP•−

Figure 4