

1 **Multi-hierarchical Profiling the Structure-Activity Relationships of**
2 **Engineered Nanomaterials at Nano-Bio Interfaces**

3 *Xiaoming Cai*[§], *Jun Dong*[‡], *Jing Liu*[#], *Huizhen Zheng*[†], *Chitrada Kaweeteerawat*[⊥],
4 *Fangjun Wang*[#], *Zhaoxia Ji*^{◇‡*}, *Ruibin Li*^{†*}

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6 [†]School for Radiological and interdisciplinary Sciences (RAD-X), Collaborative
7 Innovation Center of Radiological Medicine of Jiangsu Higher Education Institutions,
8 Jiangsu Provincial Key Laboratory of Radiation Medicine and Protection, State Key
9 Laboratory of Radiation Medicine and Protection, Soochow University, Suzhou
10 215123, China;

11 [§]Center for Genetic Epidemiology and Genomics, School of Public Health, Jiangsu
12 Key Laboratory of Preventive and Translational Medicine for Geriatric Diseases,
13 Medical College of Soochow University, Suzhou, Jiangsu 215123, P. R. China;

14 [#]CAS Key Laboratory of Separation Sciences for Analytical Chemistry, National
15 Chromatographic R&A Center, Dalian Institute of Chemical Physics, Chinese
16 Academy of Sciences (CAS), Dalian 116023, China;

17 [‡]Institute of Environmental and Food Safety, Wuhan Academy of Agricultural
18 Science and Technology, Wuhan, Hubei 430000, China;

19 [⊥]California NanoSystems Institute, University of California Los Angeles, California
20 90095, United States;

21 [◇] Living Proof, Inc., Cambridge, Massachusetts, 02142, United States;

22 [⊥]National Nanotechnology Center, Klong Nueng, 12120, Thailand

23

24 * Address correspondence to

25 Dr. Ruibin Li Email: liruibin@suda.edu.cn

26 Dr. Zhaoxia Ji Email: zji@cnsi.ucla.edu

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30 **Abstract**

31 Increasingly raised concerns (nanotoxicity, clinical translation, *etc*) on
32 nanotechnology require breakthroughs in structure-activity relationship (SAR)
33 analyses of engineered nanomaterials (ENMs) at nano-bio interfaces. However,
34 current nano-SAR assessments failed to disclosure sufficient information to
35 understand ENM-induced bio-effects. Here we developed a multi-hierarchical
36 nano-SAR assessment for a representative ENM, Fe₂O₃ by systematically examining
37 cellular metabolite and protein changes. This nano-SAR profile allows visualizing the
38 contributions of 7 basal properties of Fe₂O₃ to their diverse bio-effects. For instance,
39 while surface reactivity is responsible for Fe₂O₃-induced cell migration, the
40 inflammatory effects of Fe₂O₃ nanorods and nanoplates are determined by their aspect
41 ratio and surface reactivity, respectively. We further discovered the detailed
42 mechanisms, including NLRP3 inflammasome pathway and monocyte
43 chemoattractant protein-1 involved signaling. Both effects were further validated in
44 animal lungs. Our findings provide substantial new insights at nano-bio interfaces,
45 which may facilitate the tailored design of ENMs to endow them with desired
46 bio-effects.

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61 The physicochemical properties of engineered nanomaterials (ENMs) have been
62 demonstrated to play a decisive role in nano-bio interactions¹. Given the rapidly
63 increasing number of ENMs as well as their diverse physicochemical properties
64 including size, shape, surface area, surface reactivity, mechanical strength, *etc.*², the *in*
65 *vitro* structure-activity relationship (SAR) studies on ENMs have significantly
66 promoted the development of nanobiotechnology³⁻⁵. In general, nano-SAR analyses
67 have enabled the determination of the key physicochemical properties of ENMs that
68 are responsible for evoking a target bio-effect in the organism^{1,6}, allowed bio-hazard
69 ranking of various new ENMs⁷, and facilitated the engineering design of
70 biocompatible materials by tailored functionalization⁸. However, current nano-SAR
71 analyses only focus on the influence of a single property (size, charge, or surface
72 charge, *etc.*) of ENMs to individual bio-effects (*e.g.* apoptosis, necrosis, autophagy or
73 inflammation, *etc.*)². Considering some increasingly raised bottleneck problems in
74 nanotechnology, *e.g.* various ENM-induced nanotoxicities^{3,4}, and severe clinical
75 translation barriers in nanomedicine¹⁰, there is a demand for tiered views of
76 nano-SARs.

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78 System biology is a new theme in biological science, aiming at system-level
79 understanding of biological organisms. Several omics-based technologies including
80 genomics, proteomics, metabolomics, *etc.*, have been developed for systematic
81 analysis of biomolecules (genes, proteins, metabolites, *etc.*) expressed in cells or
82 tissues¹⁰. Recently, some progresses have been made using omics to investigate
83 protein corona on ENM surfaces¹¹, examine ENM-induced cell signaling changes^{12,13},
84 define the routes of ENM trafficking¹⁴, decipher cytotoxicity mechanisms¹⁵, *etc.*
85 However, so far, no attempts have been made for nano-SAR assessments¹⁶. Since
86 proteins and metabolites are the executors or end products of signaling pathways and
87 multi-omics analyses offer a better view of the global biological changes¹⁷, we
88 hypothesized that multi-hierarchical nano-SAR assessments could be achieved *via*
89 coupling of proteomics and metabolomics analyses.

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91 In this study, we engineered a series of iron oxide nanoparticles to assess their SAR
92 because they are widely used in constructions¹⁸, pigments¹⁹, biomedicine^{20, 21}, *etc*, and
93 their global production had reached to 1.83 billion in 2015. We selected Fe₂O₃
94 nanorods and nanoplates here based on our previous experience that various nanorods
95 like CeO₂, AlOOH and lanthanide materials, or nanoplates (e.g. Ag nanoplates) were
96 demonstrated to be more reactive than other shapes²²⁻²⁴. The metabolomics and
97 proteomics changes induced by Fe₂O₃ particles were examined in THP-1 cells, a
98 macrophage-like cell line, which are the first port of entry for the ENMs exposed to
99 mammalian systems^{7, 25}. A multi-hierarchical nano-SAR profile was established by
100 integration of the physicochemical properties of Fe₂O₃ particles, biological effects and
101 their correlation coefficients. The identified nano-SARs were selectively validated by
102 deciphering the detailed mechanisms *in vitro* and *in vivo*.

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104 **Results**

105 **Preparation and characterization of Fe₂O₃ nanoparticle library**

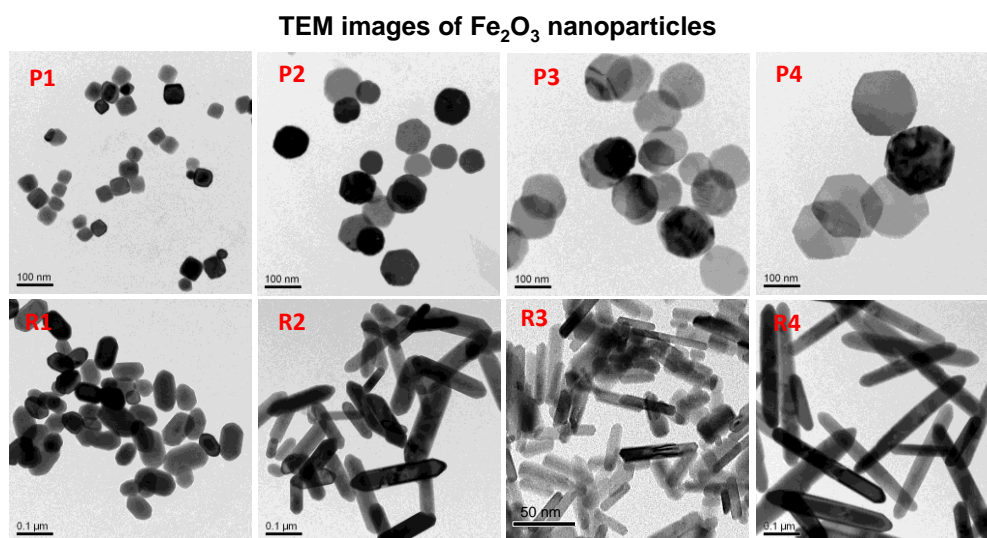
106 To explore the SAR of Fe₂O₃, we synthesized a series of Fe₂O₃ NPs with different
107 morphologies and sizes, including 4 hexagonal nanoplates (P1~P4) with controlled
108 diameters and thicknesses, and 4 nanorods (R1~R4) with systematically tuned lengths
109 and diameters. Transmission electron microscopy (TEM) was used to determine the
110 size and morphology of all Fe₂O₃ particles. Figure 1A shows that the diameters of
111 Fe₂O₃ nanoplates range from 45 to 173 nm and their thicknesses are 16~44 nm, while
112 the lengths and diameters of nanorods are 88~320 and 20~53 nm, respectively. We
113 further calculated the ratios of diameter to thickness for the nanoplates, length to
114 diameter for nanorods, respectively, and denoted them as aspect ratios (ARs). The
115 ARs of Fe₂O₃ nanoplates and nanorods are 1.0~10.8 and 1.7~8.0, respectively. The
116 surface areas were 16~27 m²/g, determined by Brunauer-Egmmett-Teller (BET)
117 method (Table 1).

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119 X-ray diffraction analysis (XRD) was performed to determine the crystal structure of

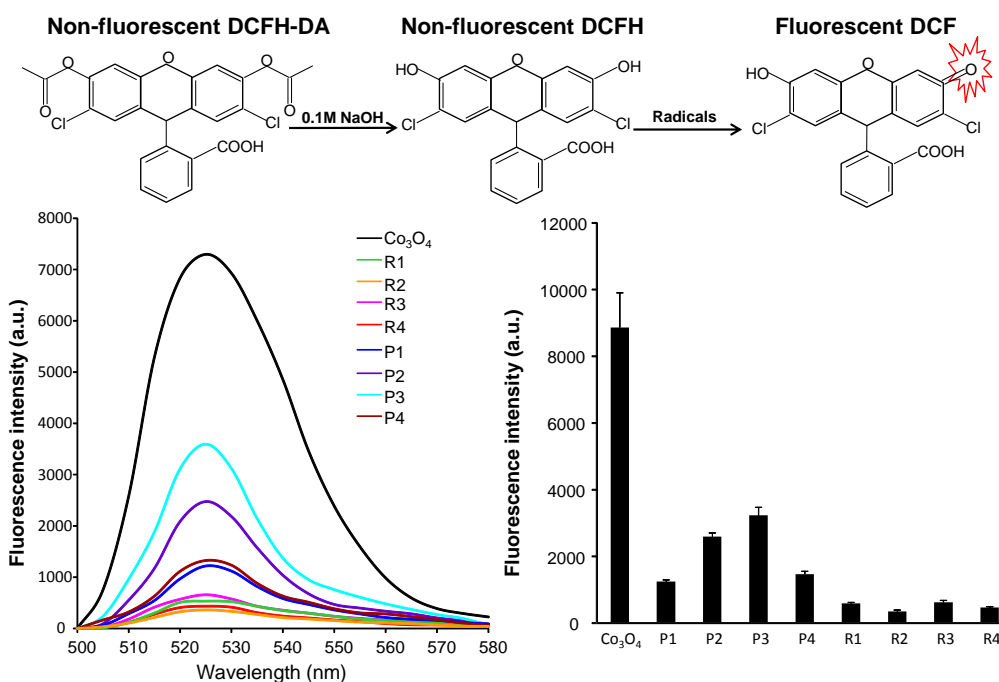
120 Fe₂O₃. Figure S1 shows the XRD patterns for selected nanoplates and nanorods, P2
121 and R2. All diffraction peaks could be indexed to rhombohedral α -Fe₂O₃ phase
122 (JCPDS no. 33-0664) and no other impurity peaks were detected.

123 A



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B



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Figure 1 Characterization of Fe₂O₃ nanoparticles by TEM and DCF assay

129 A) TEM images and B) surface reactivity of Fe₂O₃ nanoparticles. TEM samples were prepared by
130 placing a drop of the particle suspensions (50 μg/mL in DI H₂O) on the grids. To assess the
131 surface reactivity of Fe₂O₃ samples, 95 μL aliquots of 25 ng/mL DCF were added into each well
132 of a 96-multiwell black-bottom plate and mixed with 5 μL of nanoparticle suspensions at 5
133 mg/mL, followed by 2 h incubation. A SpectraMax M5 microplate reader was used to record the
134 fluorescence emission spectra of DCF agent at an excitation wavelength of 490 nm.

135

136 The hydrodynamic sizes in RPMI medium were assessed by dynamic light scattering
 137 (DLS), showing ranges from 170 to 380 nm for plates and 420-540 nm for rods (Table
 138 1). The surface charges of Fe₂O₃ particles were determined by a Zeta potential
 139 analyzer (ZPA), and showed very similar Zeta potential values of 0 to -10 mV in cell
 140 culture media, which reflects the formation of protein corona on particle surface.

141

Table 1 Quantitative Characterization of Fe₂O₃ nanoparticles

Properties	Detection methods	Quantitative Characterization							
		Plates				Rods			
		P1	P2	P3	P4	R1	R2	R3	R4
Length (nm)	TEM	NA	NA	NA	NA	88 ± 8	181 ± 11	116 ± 12	322 ± 26
Diameter (nm)	TEM	45 ± 3	84 ± 5	122 ± 7	173 ± 6	NA	NA	NA	NA
Thickness (nm)	TEM	44 ± 7	23 ± 3	18 ± 2	16 ± 3	53 ± 8	38 ± 5	20 ± 4	40 ± 7
Aspect ratio	Cal.*	1.0 ± 0.1	3.7 ± 0.2	6.8 ± 0.1	10.8 ± 0.3	1.7 ± 0.2	4.5 ± 0.4	5.8 ± 0.5	8.0 ± 0.7
Hydrodynamic size (nm)	DLS	175 ± 10	246 ± 7	366 ± 16	378 ± 8	422 ± 12	463 ± 8	578 ± 8	536 ± 4
Zeta potential (mV)	ZPA	-7.9 ± 0.8	-3.6 ± 0.6	-5.1 ± 0.4	-3.7 ± 0.7	-6.5 ± 0.7	-10.7 ± 1.6	-8.5 ± 0.6	-5.2 ± 2.1
Surface area (m ² /g)	BET	22 ± 1.6	17.9 ± 1.0	17.4 ± 0.6	16.8 ± 0.4	18.3 ± 1.5	20.8 ± 1.8	26.9 ± 2.3	21.3 ± 1.8
Surface activity (a.u.)	DCF	1.25 ± 0.08	2.59 ± 0.11	3.23 ± 0.15	1.46 ± 0.08	0.59 ± 0.04	0.35 ± 0.03	0.63 ± 0.07	0.47 ± 0.03

142 *Calculation (Cal.): Aspect ratio = $\frac{\text{Length (or Diameter)}}{\text{Thickness}}$

143

144 We used the 2',7'-dichlorofluorescein (DCF) assay to investigate the surface reactivity
 145 of Fe₂O₃ nanoparticles. The DCF assay is based on a mechanism that nonfluorescent
 146 2',7'-dichlorodihydrofluorescein (H₂DCF) could be converted to the highly fluorescent
 147 DCF by oxidation. This assay has been widely used to access the radicals or abiotic
 148 reactive oxygen species (ROS) generation on nanoparticle surface^{6, 8}. Co₃O₄
 149 nanoparticles have been demonstrated to exhibit high surface reactivity in DCF assay
 150 and were used as a positive control²⁶. As shown in Figure 1B, Fe₂O₃ nanoplates are
 151 more reactive than nanorods, and P3 exhibits the highest surface oxidative capability.
 152 These differences in surface reactivity may result from their crystal facets. XRD
 153 analysis shows that (104) is the dominant facet in Fe₂O₃ nanoplates as compared to
 154 (110) as the strongest peak for the nanorods (Figure S1). This is consistent with
 155 several earlier studies showing that (104) as the dominant facet of the α-Fe₂O₃

156 nanoplates is highly catalytically active²⁷. Therefore, the high surface reactivity of
157 nanoplates observed here can be attributed to the exposure of more active (104)
158 facets.

159

160 **Using metabolomics to explore Fe₂O₃ induced metabolite changes**

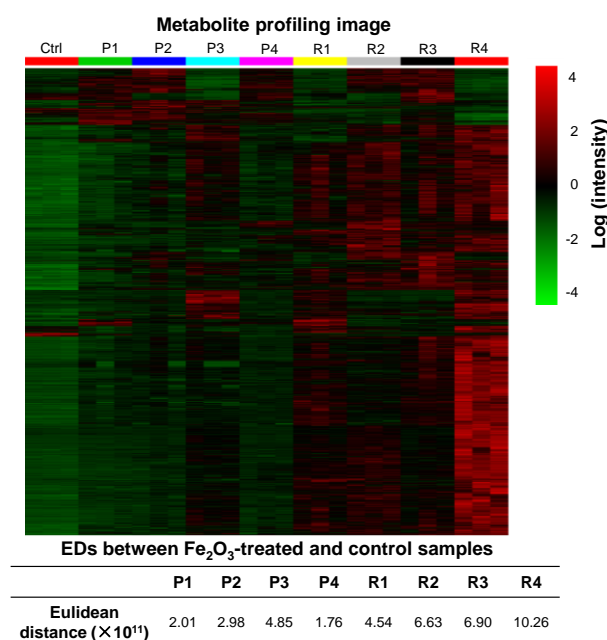
161 Metabolites are the end products of diverse intracellular processes, so the changes of
162 cell metabolome can reflect cell responses to stimuli. We performed a global
163 metabolomics study to explore the metabolic changes induced by Fe₂O₃ nanoparticles
164 in THP-1 cells, a myeloid cell line that is often used as an *in vitro* model for studying
165 the effects of engineered nanoparticles on immune cells²⁸. As described in the
166 experimental section, the metabolites in THP-1 cells after Fe₂O₃ treatment were
167 extracted and subjected to C18 reversed-phase column for nontargeted liquid
168 chromatography-mass spectrometry (LC-MS) analysis on a high-resolution Triple
169 Time of Flight (TOF) mass spectrometry in both positive (ESI+) and negative (ESI-)
170 ionization modes. By use of XCMS software, 8001 and 3479 metabolite features were
171 obtained from the LC-MS data collected in ESI+ and ESI- mode, respectively.
172 One-way analysis of variance (ANOVA) was used to screen metabolite differences
173 associated with Fe₂O₃ treatment. The significance of each feature was determined by
174 its p-value and false discovery rate (FDR) truncated at 0.01 and 0.05, respectively. As
175 a result, 1674 and 1180 discriminating features were detected in the data of ESI+ and
176 ESI- mode, respectively. Figure 2A shows a heat map of the significant features.
177 Compared to the control, Fe₂O₃-treated samples show increases in most of the
178 detected features. R4 induces the most significant metabolic changes in THP-1 cells,
179 R1, R2, R3 and P3 show moderate effects, while P1, P2 and P4 are relatively
180 bio-inert.

181

182 We further performed a hierarchical clustering analysis of metabolites to calculate the
183 Euclidean distances (EDs) between control sample and nanoparticle-treated samples²⁹.
184 This parameter was used to quantitatively describe the global metabolite profile
185 changes by Fe₂O₃, and a longer distance usually means more disruptions to the

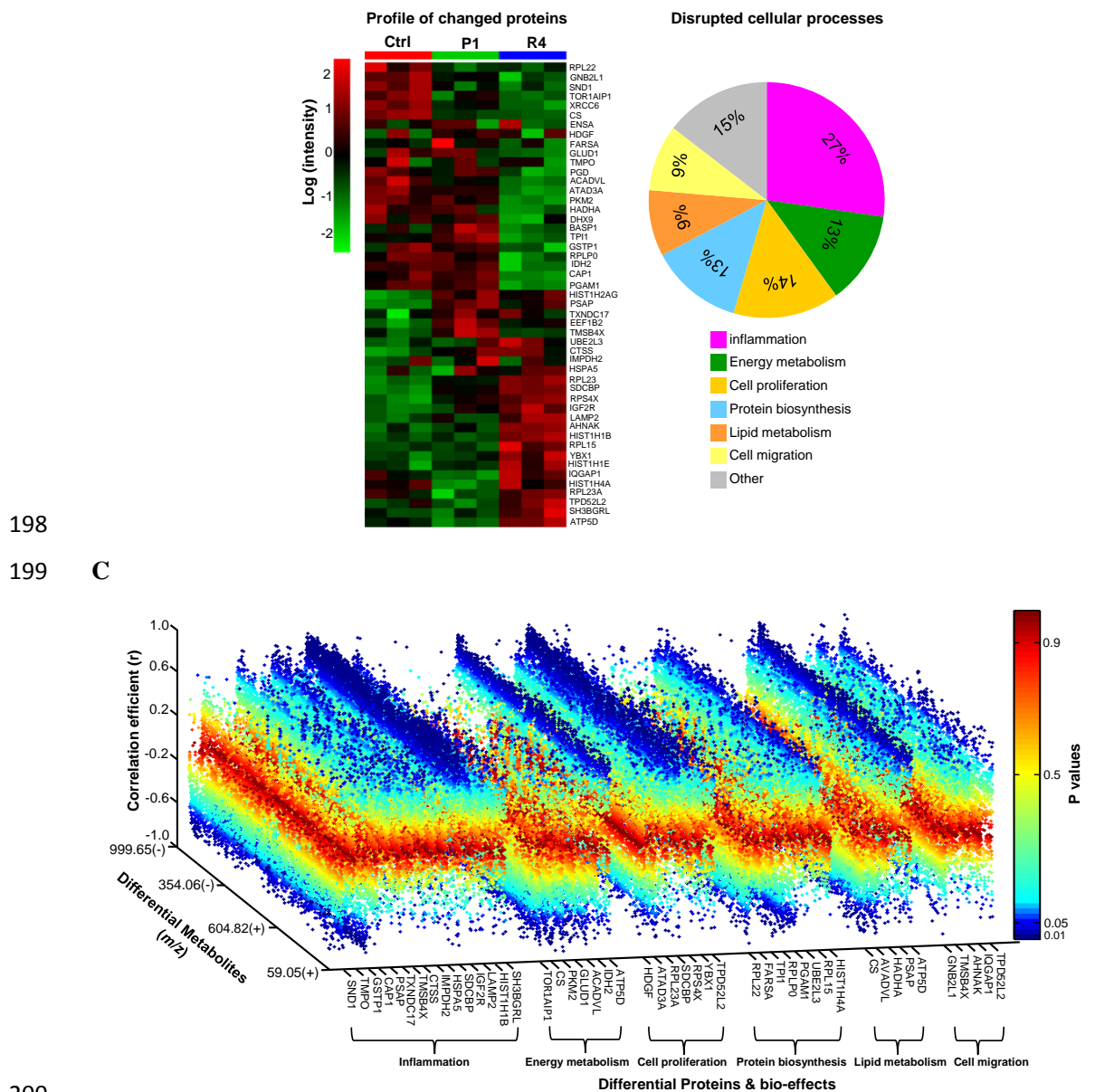
186 homeostasis of cellular metabolism, which could be considered as a bio-activity index
187 of stimuli at systemic levels³⁰. As shown in Figure 2A, the ED ranking of different
188 Fe₂O₃ particles is R4> R3> R2> P3> R1> P2> P1> P4, which is consistent with the
189 observation on metabolite heat map. Linear regression analysis was used to
190 investigate the relationships between EDs and the physical chemical properties of
191 Fe₂O₃ nanoparticles (Figure S2). According to the r² values of regression models,
192 surface reactivity and aspect ratio are the dominant physicochemical properties for the
193 global metabolite changes in THP-1 cells, and account for 88.98% and 98.02% of ED
194 variations in nanoplates and nanorods, respectively.

195 **A**



196

197 **B**



200

201 **Figure 2 Fe₂O₃ induced metabolomics and proteomics changes and their relationships**

202 **A)** Metabolite profile of THP-1 cells exposed to Fe₂O₃ library as well as their Euclidean distances
 203 (EDs). After 24 h treatment, the cell samples (n= 3 independent experiments) were collected to
 204 extract proteins and metabolites. For metabolomics analysis, the log-transformed normalized peak
 205 intensities of metabolites in all the cell samples were expressed using red, black or green colors in
 206 a heat map. EDs between control and nanoparticle-treated samples were calculated to
 207 quantitatively evaluate the global metabolite profile changes induced by Fe₂O₃. **B)** Fe₂O₃
 208 nanoparticle induced protein expression changes and bio-effects. For proteomics analysis, a heat
 209 map was plotted in a similar way as metabolites, showing 49 significantly changed proteins in
 210 proteomics analysis (left panel). The pathways or bio-effects related with these proteins were
 211 determined by KEGG and UniprotKB database (right panel). The percentages of the differential
 212 proteins involved in each specific bio-effects of Fe₂O₃ were shown in the pie chart. **C)** Regression
 213 analysis between differential proteins and metabolites. The regression analysis of the differential
 214 proteins and metabolites was performed in Matlab R2009b. The correlation is considered as

215 statistically significant with correlation coefficient ($r \leq -0.8$ or $r \geq 0.8$, and $p \leq 0.01$).

216

217 **Discovery of the bio-effects of Fe₂O₃ particles by proteomics**

218 Proteins, as a major executor of signaling pathways in biological organisms are
219 involved in many cellular effects. The bio-effects of Fe₂O₃ nanoparticles in cells could
220 be determined by identification of the proteome changes. Since the metabolomics
221 profile suggests that R4 and P1 are the most bio-active and bio-inert materials,
222 respectively, they were selected and exposed to THP-1 cells for proteomics analysis.
223 The protein expression was analyzed by a nanoscale liquid chromatography coupled
224 to tandem mass spectrometry (nano LC-MS/MS) as described in the experimental
225 section, and 785 proteins were identified for statistical analysis.

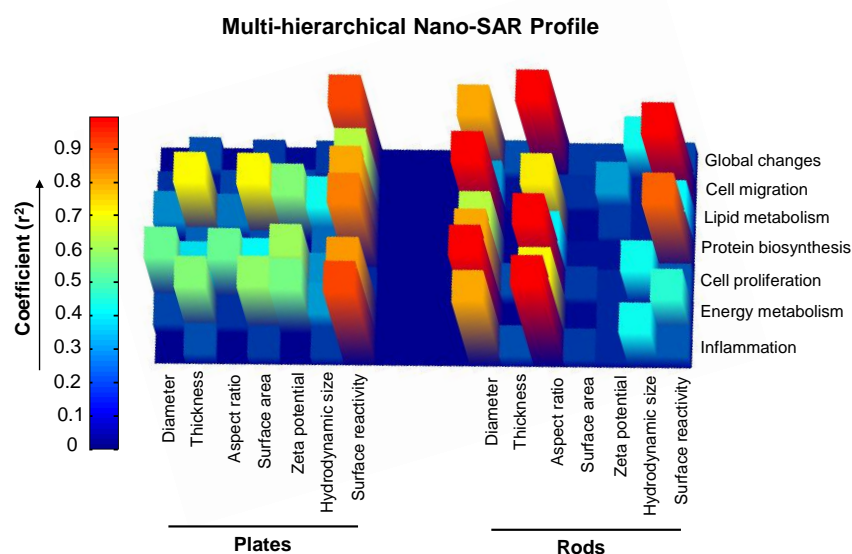
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227 To discover the biomarkers related to the bio-effects of Fe₂O₃, ANOVA analysis was
228 performed. As a result, 49 identified proteins with p values < 0.01 and FDRs < 0.05
229 were considered to be significantly changed after Fe₂O₃ treatment. The data were
230 further integrated into a heat map to visualize the expression levels of these proteins
231 in control, P1 and R4 samples (Figure 2B). While R4 induced significant proteome
232 changes in THP-1 cells including 25 up-regulated and 24 down-regulated proteins, P1
233 had negligible effects. We used KEGG and UniprotKB database to investigate the
234 impacts of the 49 proteins to cell pathways and functions. These proteins were found
235 to mainly participate in 6 biological processes, including inflammation, cell
236 proliferation, energy metabolism, lipid metabolism, protein biosynthesis and cell
237 migration (Figure 2B). Regression analysis was used to explore the relationships
238 between the changed proteins and the metabolites. As shown in Figure 2C,
239 correlations between the bio-effects and metabolite changes could be plotted
240 successfully, which provides an opportunity to explore the relationships between the
241 physicochemical properties of Fe₂O₃ nanoparticles and their bio-effects.

242

243 **Profiling the multi-hierarchical nano-SAR of Fe₂O₃ nanoparticles by a 3D** 244 **heatmap**

245 A 3D heatmap was plotted to quantitatively describe the influences of
246 physicochemical properties to the bio-effects of Fe₂O₃ nanoparticles by regression
247 analysis among the properties of nanoparticles, their metabolite changes and
248 bio-effects (Figure 3). While the zeta potential of Fe₂O₃ nanoplates has some effects
249 on cell proliferations with coefficient r^2 value at 0.64, surface reactivity is the
250 dominant property that impacts other 5 bio-effects as well as global cellular changes.
251 For Fe₂O₃ nanorods, surface reactivity is responsible for the disruption of cell
252 migration ($r^2=0.88$) and protein biosynthesis ($r^2=0.99$); particle length significantly
253 affects the energy and lipid metabolism processes; AR plays a major role in
254 inflammation and cell proliferation. These results suggest that there is one dominant
255 property that best correlates with a specific bio-effect. This is the first time that we
256 determined the contributions of 7 basal physicochemical properties of ENMs to their
257 diverse bio-effects by plotting a nano-SAR profile at multi-hierarchical levels.



258

259 **Figure 3 Multi-hierarchical profiling the nano-SAR of Fe₂O₃ particles**

260 The relationships between the 7 physicochemical properties of Fe₂O₃ nanoparticles and their
261 bio-effects could be visualized by the 3D heatmap, which is established by regression analyses
262 among metabolites, differential proteins and Fe₂O₃ properties.

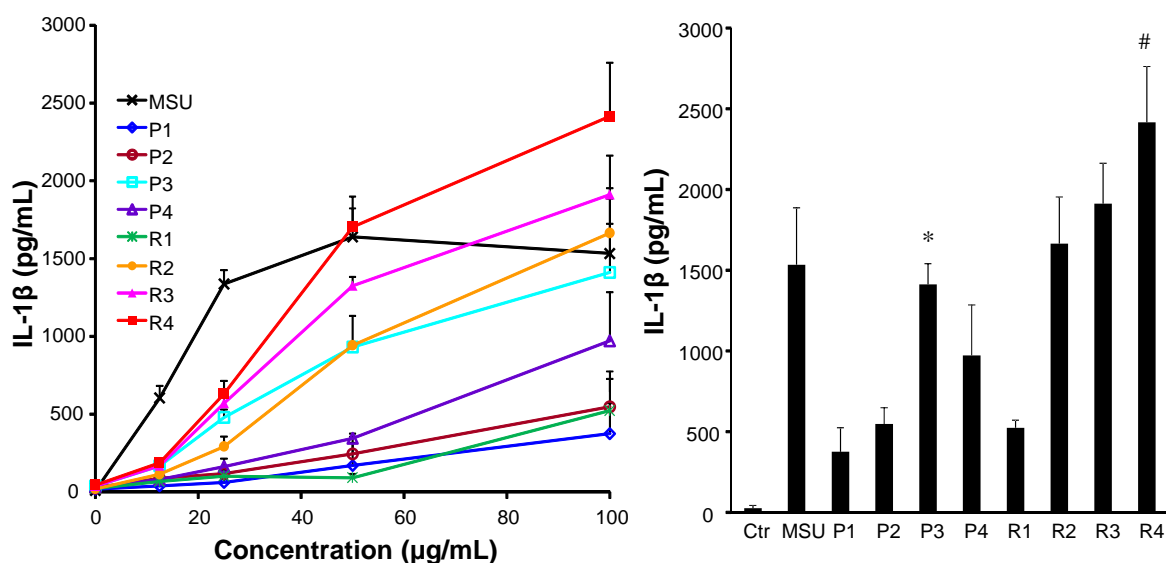
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264 **Exploring the inflammatory effects of Fe₂O₃ nanoparticles in THP-1 cells by** 265 **deciphering the detailed mechanism**

266 The nano-SAR profile indicates that Fe₂O₃ nanoparticles may induce significant

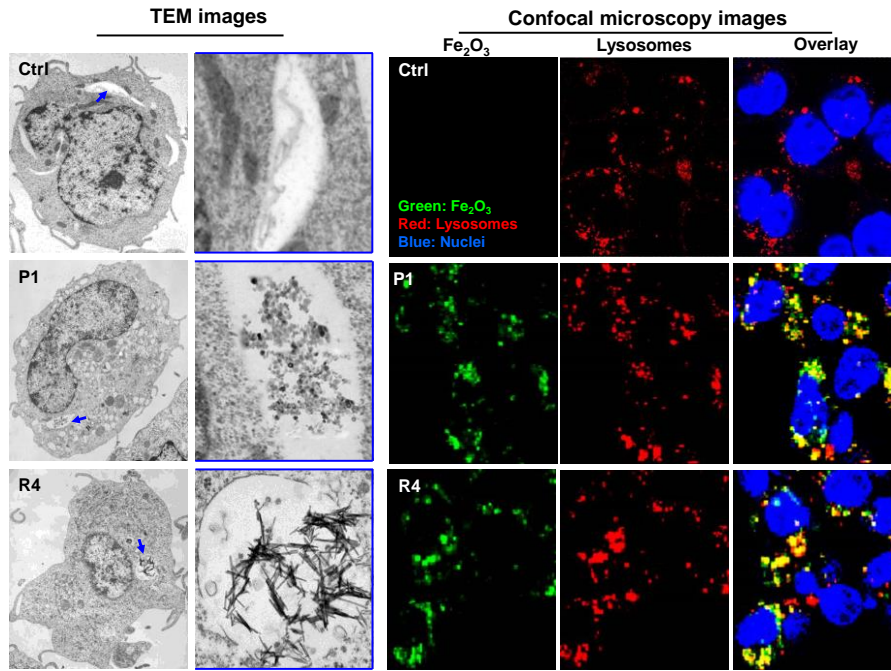
267 inflammatory effects in THP-1 cells. Among the changed proteins, 15 of them are
268 involved in inflammation pathways, *e.g.*, proactivator polypeptide (PSAP), cathepsin
269 S (CTSS), and cation-independent mannose-6-phosphate receptor (IGF2R). These
270 proteins have effects on phagocytosis and lysosomal dysfunction³¹⁻³³, implying a
271 lysosome-involved mechanism. We further investigated this by detecting the
272 pro-inflammatory cytokine release in THP-1 cells. Monosodium urate (MSU) was
273 used as a positive control to evoke inflammatory response. Although Fe₂O₃
274 nanoparticles have little effect on cytokine production in THP-1 cells exposed to
275 0-100 µg/mL particles for 24 h (Figure S3A), all Fe₂O₃-treated cells exhibit
276 significant IL-1β and TNF-α increase in dose-dependent manners at 48 h (Figure 4A
277 and S3B). However, all these particles show little effects in cell viability (Figure S4).
278 At the 100 µg/mL exposure dose, R4 exhibits the highest inflammatory cytokine
279 production, P3, R2 and R3 have moderate effects, while P1, P2, P4 and R1 induce a
280 small amount of cytokine release. This trend can't be explained by the cellular uptake
281 levels of Fe₂O₃ nanoparticles. Although the nanorods in general have relatively higher
282 cellular uptake than the nanoplates, there's no difference among nanorods (or
283 nanoplates) (Figure S5). Consistent with the nano-SAR profile, cytokine productions
284 by rods and plates have good correlation with their aspect ratios and surface reactivity,
285 respectively.

286 A



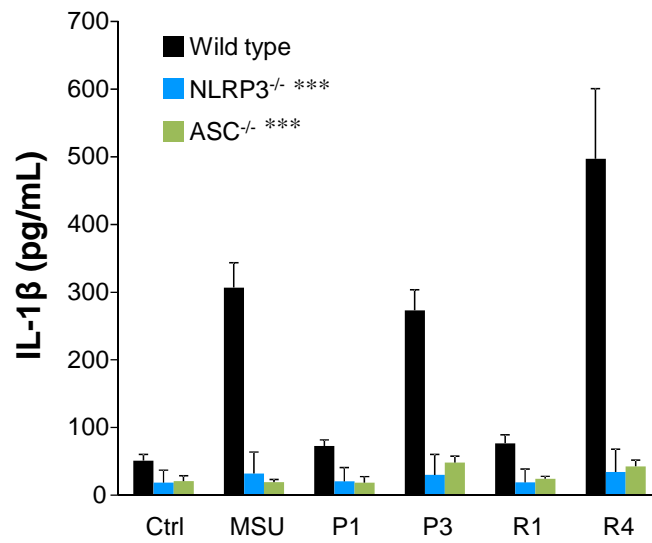
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288 B



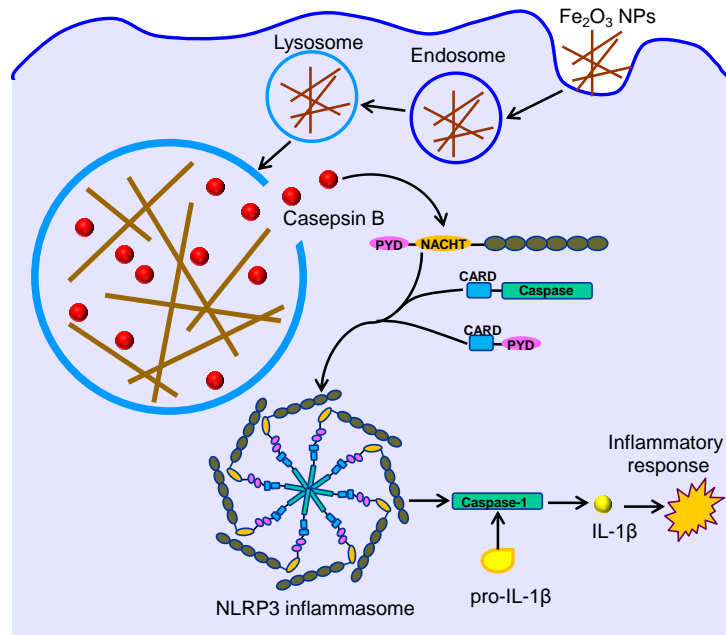
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290 **C**



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292 **D**



293

294 **Figure 4 Determination of the inflammatory pathway of Fe₂O₃ in THP-1 cells**

295 **A)** IL-1β production in THP-1 cells exposed to 0-100 μg/mL Fe₂O₃ nanoparticles for 48 h. *p<
296 0.05 compared to P1, P2 and P4, #p< 0.05 compared to other particles (two-tailed Student's t-test).
297 After 48 h incubation of THP-1 cells with Fe₂O₃ particles, the supernatants were collected to
298 quantify cytokine productions by ELISA. **B)** TEM and confocal microscopy imaging of
299 internalized Fe₂O₃ nanoparticles. THP-1 cells exposed to pristine or FITC-labeled Fe₂O₃
300 nanoparticles were washed, fixed and stained for TEM or confocal microscopy imaging. Hoechst
301 33342 dye (blue) and Alexa fluor 594 labeled antibodies (red) were used to identify nuclei and
302 lysosomes, respectively. **C)** comparison of IL-1β production in Wild-type, NALP3^{-/-}, and ASC^{-/-}
303 THP-1 cells exposed to Fe₂O₃ nanoparticles, ***p< 0.001 compared to particle-treated Wild-type
304 cells (two-tailed Student's t-test). Data in the line and bar graphs are shown as mean ± s.d. from 4
305 independent replicates. **D)** schematic to explain the inflammatory mechanism.

306

307 In order to understand the detailed mechanisms involved in Fe₂O₃-induced
308 inflammatory effect, we further treated THP-1 cells with cytochalasin D, a
309 cytoskeletal inhibitor of endocytosis, before exposure to the Fe₂O₃ particles. All cells
310 treated with cytochalasin D showed a decrease in IL-1β release, suggesting that
311 cellular uptake is essential in generating the inflammatory effect (Figure S6).
312 Confocal microscopy was used to study the cellular uptake of fluorescein
313 isothiocyanate (FITC)-labeled Fe₂O₃ nanoparticles, and we found that most of the
314 labeled nanoparticles co-localized with an Alexa fluor 594-labeled LAMP1-positive
315 compartment with co-localization coefficients ranging from 73-93% by Image J
316 analysis (Figure 4B). This suggests that Fe₂O₃ nanoparticles were mainly taken into

317 the lysosomes of THP-1 cells. TEM data confirmed that Fe₂O₃ nanoparticles are
318 encapsulated into vesicular THP-1 compartments of THP-1 cells. Since lysosome is
319 an acidifying environment, Fe₂O₃ particles tend to aggregate in this intracellular
320 compartment and interact with its membranes. This likely has led to the lysosome
321 membrane damage due to the reactive surface (nanoplates) or geometric shape of
322 Fe₂O₃ nanorods.

323

324 In order to understand the biological impact of Fe₂O₃ nanoparticles in lysosomal
325 compartments, we asked whether that would impact lysosomal function. Confocal
326 microscopy was used to study the subcellular localization of cathepsin B, a lysosomal
327 enzyme capable of cleaving a Magic RedTM-labeled substrate. As shown in Figure
328 S7A, untreated cells show a punctate distribution of Magic RedTM, indicating that the
329 enzyme is contained in intact lysosomes. However, after lysosomal damage by MSU,
330 there is a diffuse cytosolic release of the fluorescence marker. Similarly, P3 and R4
331 nanoparticles induce cathepsin B release, while P1 and R1 nanoparticles are not
332 associated with lysosomal damage. Since cathepsin B is known to contribute to the
333 activation of the NLRP3 inflammasome and IL-1 β production³⁴, this may explain the
334 severe inflammatory cytokine release in Fe₂O₃-treated THP-1 cells. The role of
335 cathepsin B in NLRP3 inflammasome activation was further confirmed by using a
336 cathepsin B inhibitor, CA-074-Me, which shows the inhibitory effect in IL-1 β
337 production (Figure S7B). Moreover, we confirmed that active assembly of the NLRP3
338 inflammasome subunits is required for IL-1 β production by using NLRP3- and ASC-
339 gene knockdowns to show the interference in cytokine release in THP-1 cells (Figure
340 4C).

341

342 Based on the mechanism study, we for the first time deciphered the inflammatory
343 pathway of Fe₂O₃ in THP-1 cells. As shown in Figure 4D, Fe₂O₃ nanoparticles are
344 internalized into lysosomes through endocytosis. Macrophage uptake and lysosomal
345 processing of Fe₂O₃ nanoparticles further lead to the interaction with lysosome
346 membrane. Because of the surface reactivity and geometric shape of Fe₂O₃ nanoplates

347 and nanorods, these particles may induce lysosome damage, cathepsin B release into
348 cytoplasm, recruitment of NLRP3, pro-caspase 1 and ASC subunits, NLRP3
349 inflammasome activation and IL-1 β release from the macrophages. IL-1 β may further
350 participate in a progressive march of inflammation events in organs.

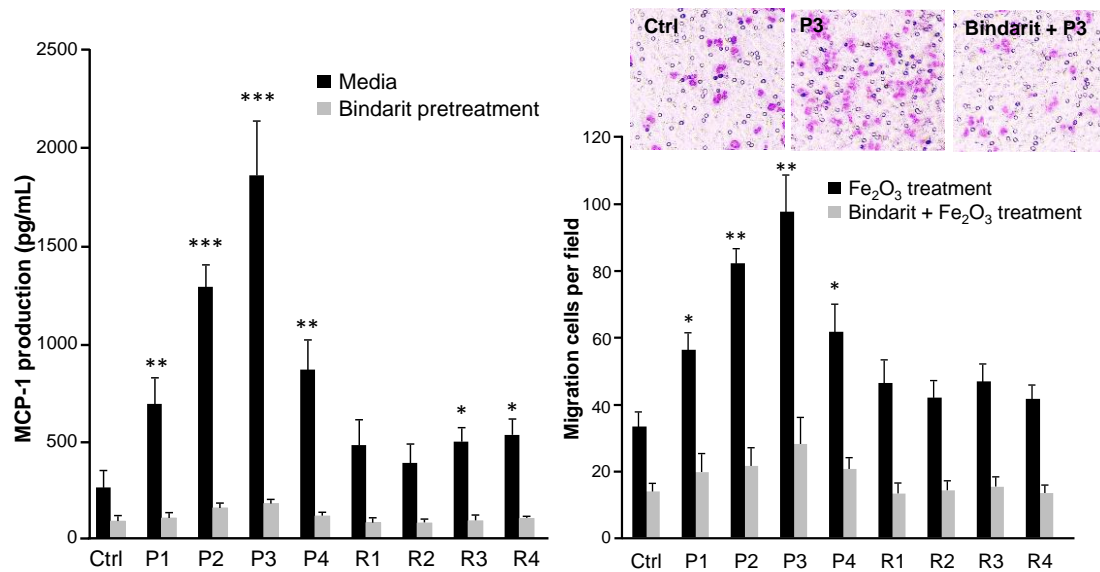
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352 **Examining the effects of Fe₂O₃ nanoparticles in cell migration**

353 The nano-SAR profile also indicates that surface reactivity may be responsible for
354 Fe₂O₃-induced cell migration. Since monocyte chemoattractant protein-1 (MCP-1 or
355 CC-chemokine ligand 2) is widely reported to be a critical factor for mediating arrest
356 of the monocytic cells and directional migration³⁵, we first examined the effects of
357 Fe₂O₃ particles on MCP-1 production. After 24 h exposure to Fe₂O₃ nanoparticles,
358 significant MCP-1 production was detected in the supernatants of THP-1 cells, and
359 Fe₂O₃ plates show higher MCP-1 production than rods (Figure 5A). While P3
360 stimulates the highest MCP-1 secretion in THP-1 cells, P2 and P4 had moderate
361 effects. Bindarit, as an inhibitor of MCP-1 pathway, could effectively block all Fe₂O₃
362 induced MCP-1 productions. Then we transferred the supernatants of Fe₂O₃-treated
363 THP-1 cells to the lower chambers of transwell systems to examine the effects in cell
364 migration. As shown in Figure 5B, Fe₂O₃ plates with high surface reactivity induce
365 significant cell migration, and P3 shows the highest level. To investigate whether the
366 macrophage recruitment is a result of MCP-1 production, we examined the effects of
367 supernatants from THP-1 cells exposed to bindarit and Fe₂O₃ particles on cell
368 migration. Bindarit treatment results in total reduction of cell migration. These results
369 indicate that Fe₂O₃ plates could induce MCP-1 dependent cell migration, and surface
370 reactivity is the dominant property for this effect. Immune cell recruitment is an early
371 statement in acute immune responses and involves transendothelial migration toward
372 the stimulation site to protect health tissues³⁶. Thus, with regard to monocyte or
373 leukocyte migration, Fe₂O₃ nanoplates display immunostimulatory functions and may
374 serve as a modulator to activate immune cell recruitment.

375 **A**

B



376

377 **Figure 5 Effects of Fe₂O₃ nanoparticles on THP-1 cell migration**

378 **A)** MCP-1 production in the supernatants of Fe₂O₃-treated cells, and **B)** transwell migration assays
379 of THP-1 cells. THP-1 cells were pretreated with or without 100 μM bindarit for 3 h before
380 exposure to 100 μg/mL Fe₂O₃ nanoparticles. After 24 h incubation, the supernatants of treated
381 THP-1 cells were collected and divided into 2 portions. One was used for MCP-1 measurement by
382 ELISA, and the other one was added into the lower chambers of transwell systems to incubate
383 with THP-1 cells in upper chamber for 24 h. The migration cells were stained with Diff Quick
384 agent and counted under microscope (n=5 images for each treatment). *p< 0.05, **p< 0.01, ***p<
385 0.001 compared to ctrl (two-tailed Student's t-test).

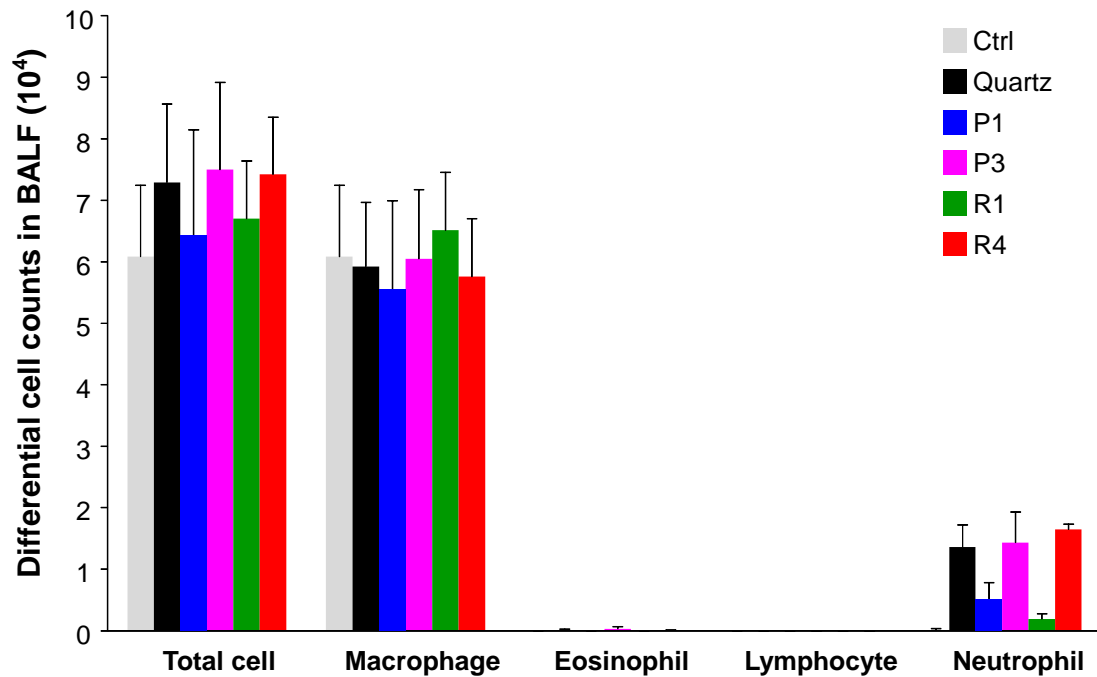
386

387 **Validation of the inflammatory effects of Fe₂O₃ nanoparticles in mouse lungs**

388 In order to further testify the nano-SAR of cellular inflammatory effects induced by
389 Fe₂O₃ particles, we used an acute lung injury model to study the effect of
390 oropharyngeal instilled nanoparticles in the whole lung. This study was performed
391 with a particle dose of 2 mg/kg, which has been previously demonstrated to fall on the
392 linear part of the dose-response curve for pulmonary exposure to metal oxide
393 nanoparticles^{26, 37}. After 40 h exposure, the animals were sacrificed to collect
394 bronchoalveolar lavage fluid (BALF) and lung tissues. The cytokine release in BALF
395 was determined by ELISA. As shown in Figure 6A, most of the immune cells induced
396 by Fe₂O₃ are neutrophils, which are dramatically boosted in P3 and R4 treated animal
397 lungs. In addition, P3 and R4 induced significant cytokine release including IL-1β,
398 TNF-α and LIX (Figure 6B), which is consistent with their *in vitro* inflammatory
399 responses. The migration effect of Fe₂O₃ particles was also validated by the MCP-1
400 production in BALF as well as H&E staining of lung tissues. As shown in Figure 6C,

401 while P3 significantly elevates MCP-1 production and induces massive immune cell
402 recruitment, P1 and R1 have a little effect. Interestingly, R4 exhibits limited effect in
403 MCP-1 production but substantial immune cell recruitment in animal lungs,
404 suggesting there may be other mechanisms involved in animal lungs. All these animal
405 results demonstrated that the nano-SAR in Fe₂O₃-induced inflammatory and
406 migration effects could be validated *in vivo*.

407 A



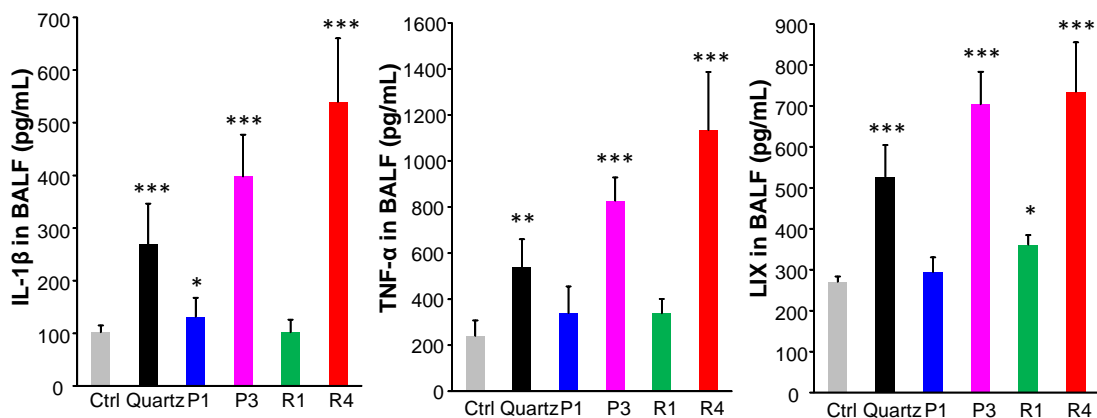
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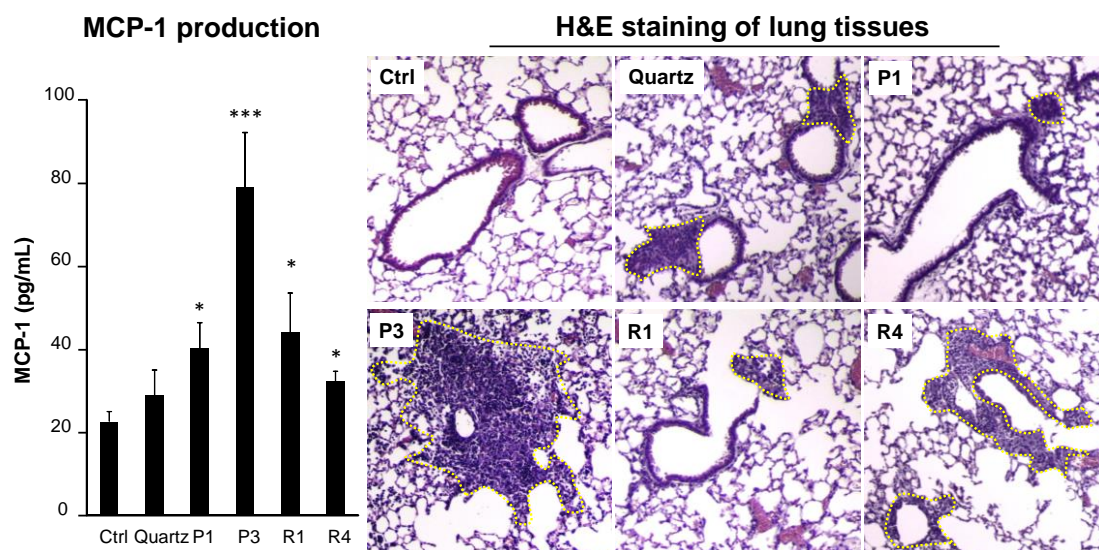
B



412

413

C



414

415 **Figure 6 Assessment of inflammatory effects and cell migration in Fe₂O₃ treated animal**

416 **lungs**

417 **A)** Differential cell counts, **B)** cytokine release in BALF, and **C)** MCP-1 production and H&E
418 staining of lung sections from Fe₂O₃-treated mice. Selected Fe₂O₃ nanoplates (P1, P3) and
419 nanorods (R1, R4) were oropharyngeally administrated at 2 mg/kg (n = 6 mice in each group),
420 while animals received 5 mg/kg quartz exposure were used as positive control. After 40 h, animals
421 were sacrificed to collect BALF for differential cell counting as well as cytokine measurement,
422 including IL-1 β , LIX, TNF- α and MCP-1. Data are shown as mean \pm s.d. from 4 independent
423 replicates. * p < 0.05, ** p < 0.01, *** p < 0.001 compared to vehicle control (two-tailed Student's
424 t-test). The dashed lines in H&E staining images show the immune cell recruitments.

425

426 **Discussions**

427 In this study, we pioneered a multi-hierarchical nano-SAR assessment by
428 simultaneously examining ENM-induced metabolite and protein changes in cells.
429 Unlike traditional method that could only explore the nano-SAR between a specific
430 physicochemical property of ENMs and individual bio-effects²⁻⁴, a 3D heatmap was
431 established in this study to assess the contributions of 7 physicochemical properties of
432 Fe₂O₃ particles to their 6 bio-effects in cells. The nano-SAR investigation could well
433 facilitate the identification of the key physicochemical properties that are responsible
434 for their bio-effects because the biological responses of ENMs have been
435 demonstrated to result from their unique physicochemical properties^{3,38}.

436

437 Recently, Zanganeh *et al.* have demonstrated that iron oxide nanoparticle induced
438 pro-inflammatory macrophage polarization plays an important role in tumor therapy²⁰,

439 however, the nano-SAR information as well as the detailed inflammatory mechanism
440 shrouds in mystery. We answered these questions by setting up a combinatorial Fe₂O₃
441 library with precisely controlled size and shape as well as systematically assessment
442 of their bio-effects. By regression analysis of particle properties, metabolite and
443 protein changes, aspect ratio and surface reactivity were identified as the key
444 physicochemical properties responsible for the inflammatory effects of Fe₂O₃
445 nanorods and nanoplates, respectively. Fe₂O₃-induced cell migration is determined by
446 surface reactivity. These findings were successfully validated in THP-1 cells and
447 animal lungs. *In vitro* experiments further deciphered the MCP-1 dependent cell
448 migration mechanism as well as the NLRP3 inflammatory pathway in Fe₂O₃-treated
449 THP-1 cells.

450

451 Our study successfully achieved tiered view of the nano-SARs of Fe₂O₃ particles in
452 THP-1 cells, providing substantial new insights for tailored design of ENMs by
453 modifying their physicochemical properties to acquire desired bio-effects. This allows
454 us to reduce the toxicities of some hazardous ENMs^{7, 8, 28}, or enhance the effects of
455 nanomedicine^{5, 6, 39}. Besides of Fe₂O₃, the multi-hierarchical nano-SAR assessment
456 method could be potentially extended to other ENMs. This study has far-reaching
457 implications for the sustainable development of nanotechnology.

458

459

460 **Methods**

461 **Materials**

462 Magic Red™ Cathepsin B Assay Kit was purchased from Immunochemistry
463 (Bloomington, MN, USA); Hoechst 33342, H₂-DCFDA, Alexa fluor 594-labeled goat
464 anti-mouse IgG, anti-LAMP1 primary antibody were purchased from Life
465 Technologies (Grand Island, NY, USA); ELISA kits for detection of murine or human
466 IL-1β, TNF-α, MCP-1 and LIX were purchased from BD biosciences (San Jose, CA,
467 USA); MTS assay kit was purchased from Promega (Madison, WI, USA); bindarit
468 was purchased from Abcam (Cambridge, MA, USA); other chemicals unless stated

469 were purchased from Sigma-Aldrich (St. Louis, MO, USA).

470

471 **In-house synthesis of Fe₂O₃ nanoparticles**

472 Both Fe₂O₃ nanorods and nanoplates were prepared using hydrothermal synthesis
473 methods modified from published procedures^{40, 41}. In a typical Fe₂O₃ nanorod
474 synthesis, a 10~16 mL of 0.86 M FeCl₃•6H₂O aqueous solution was prepared in a
475 high density polyethylene bottle, to which 4~10 mL of 1,2-propanediamine solution
476 was added to form a 20 mL synthesis mixture. After stirring for 15 min, the synthesis
477 mixture was transferred to a 23 mL Teflon-lined stainless steel autoclave. The reaction
478 was carried out in an electric oven at 180 °C under autogenous pressure and static
479 conditions. After the reaction was complete, the autoclave was immediately cooled
480 down in a water bath. The fresh precipitate was separated by centrifugation and
481 washed with deionized water and ethanol alternatively for three cycles to remove
482 ionic remnants. The final product was dried at 60 °C overnight under ambient
483 environment. In a typical Fe₂O₃ nanoplate synthesis, 0.4325 g of FeCl₃•6H₂O was
484 dissolved in 16 mL of ethanol with a small amount of water (1.1-4.0 mL) under
485 vigorous stirring. After FeCl₃•6H₂O was completely dissolved, 1.3125 g of sodium
486 acetate was then added and the resulting mixture was mixed for another 15 mins. The
487 reaction and final product collection were carried out the same way as those for the
488 Fe₂O₃ nanorods.

489

490 **Preparation of Fe₂O₃ nanoparticles suspensions in media**

491 The Fe₂O₃ stock solutions were prepared by suspending particle powders in DI H₂O
492 (5 mg/mL) and dispersed in a bath sonicator (Branson, Danbury, CT, USA, model
493 2510; 100 W output power; 42 kHz frequency) for 15 min. To prepare the desired
494 concentrations of Fe₂O₃ suspensions, an appropriate amount of each Fe₂O₃
495 nanoparticle stock solution was added to cell culture media or PBS, and further
496 dispersed using a sonication probe (Sonics & Materials, USA) at 32 W for 10 s before
497 exposure to cultured cells or animals⁴².

498

499 **Physicochemical characterization of Fe₂O₃ nanoparticles**

500 A particle suspension (50 µg/mL in DI H₂O) drop was added on the TEM grids for
501 air-dry at room temperature. The TEM observation was performed on a JEOL 1200
502 EX instrument (accelerating voltage 80 kV). A Philips X'Pert Pro diffractometer
503 equipped with CuK α radiation were used to obtain the XRD spectra. The
504 hydrodynamic diameter and surface charge in water and cell culture media were
505 measured by dynamic light scattering coupled with zeta potential analyzer
506 (Brookhaven Instruments Corporation, Holtsville, NY, USA). DCF assay was used to
507 evaluate the surface reactivity of Fe₂O₃ particles. In detail, 50 µg of H₂DCF-DA were
508 mixed with 280 µL 0.01 M NaOH and incubated for 30 min at room temperature. The
509 resulting solution was diluted with 1720 µL of a sodium phosphate buffer (25 mmol/L,
510 pH = 7.4) to form 25 µg/mL DCF working solution. A 5 µL aliquots of nanoparticle
511 suspension (5 mg/mL) were added into each well of a 96 multiwell black plate (Costar,
512 Corning, NY), and then 95 µL amount of DCF working solution was added to each
513 well, followed by 2 h incubation. DCF fluorescence emission spectra were recorded
514 by a SpectraMax M5 microplate reader at an excitation wavelength of 490 nm⁸.

515

516 **Cell culture and treatment**

517 THP-1 cells and BEAS-2B cells were purchased from ATCC (Manassas, VA, USA),
518 and cultured in RPMI 1640 supplemented with 10% fetal bovine serum and BEGM
519 media, respectively. Before exposure to Fe₂O₃ nanoparticles, THP-1 cells were primed
520 by 1 µg/mL phorbol 12-myristate acetate overnight⁷. For cellular exposure, Fe₂O₃
521 nanoparticles were dispersed in complete RPMI 1640 medium at desired
522 concentrations. Control sample was prepared by replacing the nanoparticle
523 suspensions with pure water.

524

525 **Nontargeted metabolomics via LC-MS**

526 THP-1 cells were exposed to 100 µg/mL Fe₂O₃ nanoparticles for 24 h, and 9 cell
527 samples including control and particle treatments were harvested and lyzed in cold
528 lysis buffer (1.5 mL) by a probe sonication. After centrifugation at 20, 000 g for 10

529 min, the lysis supernatants were collected and extracted by 80% methanol containing
530 0.5mM L-Methionine-(methyl-13C,d3) and 1mM D-Glucose-1,2,3,4,5,6,6-d7 as
531 internal standards. Dried metabolite pellets were re-suspended in 120 μ L of buffer A
532 (0.1% formic acid in 95:5 water/ACN) and 5 μ L aliquots were injected for
533 nontargeted LC-MS on a Shimadzu UFLC-XR system (Shimadzu Corporation, Japan)
534 and an AB SCIEX TripleTOF 5600+ system (AB SCIEX, Foster City, CA). Samples
535 were separated on a C18 reversed phase HPLC column (2.1mm \times 100mm, 100 \AA ,
536 1.7 μ m, Waters, Milford, MA) at 350 μ L/min with a liner gradient of buffer A and
537 buffer B (100% ACN) as follows: isocratic conditions at 100% A (0%B) for 1min, a
538 linear gradient from 100%A (0%B) to 30% A (70%B) over 2 min, a linear gradient
539 from 30%A (70%B) to 0% A (100%B) over 7 min, isocratic conditions at 100% B for
540 2 min. The column temperature maintained at 50 $^{\circ}$ C.

541

542 To acquire the MS data, the mass spectrometry conditions were set as follows: the ion
543 spray voltage was +5.5kV (positive ion mode) or -4.5kV (negative ion mode); turbo
544 spray temperature was 550 $^{\circ}$ C; nebulizer, heater and curtain gases were at 50, 50 and
545 30 psi, respectively; TOF MS was scanned at the mass range of m/z 50 ~ 1200.
546 Analyst v1.6.0 software (AB Sciex) was used to collected raw data, which was further
547 converted into mzXML data format by proteoWizard software (Spielberg Family
548 Center for Applied Proteomics, Los Angeles, CA) for further data processing.

549

550 The XCMS platform (<https://metlin.scripps.edu/xcms/>) was used for peak detection,
551 retention time collection and alignment. All data-collection parameters were set to the
552 “UPLC Triple TOF” default values. Retention times (RT), m/z values and peak
553 intensities of metabolites were exported to an Excel spreadsheet for processing. The
554 peak intensities were normalized to the internal standards: L-Methionine-(methyl-13C,
555 d3) (m/z, 154.077) for positive mode and D-Glucose-1,2,3,4,5,6,6-d7 (m/z, 186.099)
556 for negative mode. Preprocessed data sets were analyzed using Matlab (MathWorks,
557 Natick, MA) and Metaboanalyst (www.metaboanalyst.ca) to perform scatter plot, heat
558 map, cluster and ANOVA analysis. Euclidean distance (ED) is used to measure the

559 dissimilarity of samples with multivariate variables²⁹. Here, the ED between control
560 and nanoparticle-treated samples is defined by the length between the two cluster
561 centers. It was calculated by SPSS using following
562 formula: $ED(C, T) = \sqrt{(t1 - c1)^2 + (t2 - c2)^2 + \dots + (tn - cn)^2}$, where t_n and c_n
563 are the log-transformed normalized peak intensities of metabolite n in
564 nanoparticle-treated sample and control sample, respectively ($n=2854$).

565

566 **Proteomics via nanoflow LC-MS/MS**

567 The cell samples exposed to Fe_2O_3 nanoparticles were harvested and lyzed. After
568 centrifugation at 20000 g, the lysis supernatants (1 mL) were added to 8 mL
569 extraction solute (acetone/ethanol/acetic acid 50:50:0.1), which was pre-cooled at
570 $-20\text{ }^\circ\text{C}$. The mixture was stored at $4\text{ }^\circ\text{C}$ for 24 h. Then the proteins were collected by
571 centrifugation at 20000 g for 30 min. After re-dissolution, the protein concentrations
572 in all the samples were equivalent by a Bradford method, and then the protein samples
573 were denaturalized, digested and desalted according to a reported method⁴⁴. Finally,
574 the resulting peptide samples were re-dissolved in 200 μL 0.1% formic acid for nano-
575 LC-MS/MS analysis.

576

577 A LTQ Orbitrap Velos was equipped with an Accela 600 HPLC system (Thermo, San
578 Jose, CA) to establish the nano-LC-MS/MS system. The peptide samples were
579 injected into a capillary trap column (200 mm i.d. \times 4 cm, 120 \AA , 5 mm), and then
580 separated on an analytical column (15 cm \times 75 mm i.d., 3 mm, 120 \AA). Both of the
581 columns were packed with C18 AQ beads. The separation buffer consisted of 0.1%
582 (v/v) formic acid in DI H_2O (buffer A) and 0.1% (v/v) formic acid in acetonitrile
583 (ACN) as buffer B. The nano-LC-MS/MS analysis was performed based on a
584 previously reported method⁴³. All of the mass spectra were collected in a data
585 dependent mode.

586

587 The resulted raw files were searched in MaxQuant (Version1.3.0.5) using Integrated

588 Uniprot protein fasta database of human. Peptide searching was constrained using
589 fully tryptic cleavage, allowing less than 2 missed cleavages sites for tryptic digestion.
590 Variable modifications included methionine oxidation, acetylation of protein N-term
591 and phosphorylation (STY). Precursor ion and fragment ion mass tolerances were set
592 as 5 ppm and 0.08 Da, respectively. The false discovery rate (FDR) for peptide and
593 protein were less than 1% and peptide identification required a minimum length of six
594 amino acids. The cell pathways and functions related to the identified differential
595 proteins were explored by KEGG (<http://www.kegg.jp/>) and UniprotKB
596 (<http://www.uniprot.org/help/uniprotkb>) database.

597

598 **IL-1 β and TNF-alpha Detection by ELISA**

599 IL-1 β and TNF-alpha productions were detected in the culture media of THP-1 cells
600 using human IL-1 β and TNF-alpha ELISA Kit (BD; San Jose, CA, USA). Briefly,
601 aliquots of 5×10^4 THP-1 cells were seeded in 0.1 mL complete medium and primed
602 with 1 $\mu\text{g}/\text{mL}$ phorbol 12-myristate acetate (PMA) overnight in 96-well plates
603 (Corning; Corning, NY, USA). Cells were treated with the desired concentration of
604 the particle suspensions made up in complete RPMI 1640 medium, supplemented
605 with 10% fetal bovine serum and 10 ng/mL lipopolysaccharide (LPS).

606

607 **Cell migration assay**

608 THP-1 cells were pretreated or not with 100 μM bindarit for 3 h before exposure to
609 100 $\mu\text{g}/\text{mL}$ Fe₂O₃ nanoparticles. After 24 h incubation, the supernatants were
610 collected for MCP-1 detection or cell migration assay, which was performed in
611 24-transwell plates with polycarbonate membranes of 8 μm pores (Corning, NY,
612 USA). Lower wells were filled with 500 μL aliquots of the collected supernatants, and
613 2×10^5 THP-1 cells (100 μL) were seeded into each of the upper wells. After
614 incubation for 6h, nonmigrated cells were scraped off from the upper side of the
615 membrane and cells remaining within the pores or below the membranes were stained
616 with Diff Quick³⁵. Cell numbers were calculated under microscope by randomly
617 selecting at least 5 individual fields for each sample.

618

619 **TEM imaging of Fe₂O₃ particles in THP-1 cells**

620 After exposure to 25 µg/mL Fe₂O₃ for 24 h, THP-1 cells were collected, washed and
621 fixed with 2% glutaraldehyde in PBS. After 1 h post-fixation staining in 1% osmium
622 tetroxide, a dehydration process was performed by treating the cells in a graded series
623 of ethanol, propylene oxide, and finally the cell pellets were embedded in Epon. A
624 Reichert-Jung Ultracut E ultramicrotome was used to cut the TEM sections with
625 approximately 50-70 nm thickness. The sections were further stained with uranyl
626 acetate and Reynolds lead citrate before examining on TEM as previously reported⁴².

627

628 **Confocal microscopy imaging**

629 Leica confocal SP2 1P/FCS microscopes were used to visualize Fe₂O₃ uptake and
630 cathepsin B release in THP-1 cells. High magnification images were obtained under
631 the 63X objective. To visualize the cellular distribution, THP-1 cells were treated with
632 25 µg/mL FITC-labeled Fe₂O₃ nanoparticles for 6 h, fixed and stained with Hoechst
633 33342 and Alexa Fluor 594 labeled antibodies to visualize nuclei and lysosomes,
634 respectively. For cathepsin B imaging, cells exposed to 100 µg/mL Fe₂O₃ particles for
635 16 h were stained with Magic Red™ Cathepsin B kit and Hoechst 33342 for confocal
636 microscopy imaging.

637

638

639 **Inflammation test in mouse lungs**

640 Mice were exposed to nanoparticle suspensions using oropharyngeal aspiration at 2
641 mg/kg. Eight-week-old male C57Bl/6 mice purchased from Soochow University were
642 used for animal experiments. All animals were housed under standard laboratory
643 conditions that have been set up according to Soochow University guidelines for care
644 and treatment of laboratory animals. These conditions were approved by the
645 Chancellor's Animal Research Committee at Soochow University and include
646 standard operating procedures for animal housing (filter-topped cages; room
647 temperature at 23 ± 2 °C; 60% relative humidity; 12 h light, 12 h dark cycle) and

648 hygiene status (autoclaved food and acidified water). Animals were exposed by
649 oropharyngeal aspiration as described by us⁸. Briefly, animals were anesthetized by
650 intraperitoneal injection of ketamine (100 mg/kg)/xylazine (10 mg/kg) in a total
651 volume of 100 μ L. The anesthetized animals were held in a vertical position. 50 μ L
652 aliquots of the nanoparticle suspensions in PBS were instilled at the back of the
653 tongue to allow pulmonary aspiration of a dose of 2 mg/kg. Each experiment included
654 control animals, which received the same volume of PBS. The positive control in each
655 experiment received 5 mg/kg quartz. Each group included six mice. The mice were
656 sacrificed after 40 h exposure. BALF and lung tissue were collected as previously
657 described. The BALF was used for performance of total and differential cell counts
658 and measurement of IL-1 β , TNF- α , MCP-1 and LIX levels. Lung tissue was stained
659 with hematoxylin/eosin.

660

661 **Statistical Analysis**

662 All the experiments were repeated at least thrice with 3-6 replicates. Error bars
663 represent the standard deviations (s.d.). All the cell and animal samples were
664 randomly allocated into experimental groups by drawing lots. All the experiments
665 were repeated at least thrice with 3–6 replicates. Results were expressed as mean \pm
666 SD of multiple determinations from at least three separate experiments. One-way
667 ANOVA or Student t test was used for statistical analysis in MetaboAnalyst 3.0 and
668 excel 2010. The difference is regarded statistically significant with $p \leq 0.01$ and
669 $FDR \leq 0.05$. Correlation analysis of the differential proteins and metabolites was
670 performed in Matlab R2009b. The correlation is considered as statistically significant
671 with correlation coefficient (r) ≤ -0.8 or $r \geq 0.8$, and $p \leq 0.01$. The Euclidean distance
672 (ED) and linear regression analysis of the EDs and particle properties were achieved
673 in SPSS 18.0. The 3D profile of the Structure-Activity relationships was also done in
674 Matlab.

675

676 **Data availability**

677 The data that support the plots within this paper and other findings of this study are

678 available from the corresponding author on reasonable request.

679

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833 **Author contributions**

834 X.C., Z.J. and R.L. conceived and designed the study; X.C. did most experiments; Z.J.
835 synthesized the Fe₂O₃ nanoparticles, and performed TEM, XRD, Zeta potential and
836 hydrodynamic size characterization with H.Z.; J. D. performed the LC-MS analysis
837 for metabolomics study; F. W. and J. L. contributed to the proteomics analysis. The
838 writing of the paper was led by X. C. and R.L with participation from Z.J. and C.K.

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840 **Additional information**

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844 in published maps and institutional affiliations. Correspondence and requests for
845 materials should be addressed to R.L. and Z.J.

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