1	The Abnormal Oswald Ripening of Protein Nanofiber in Myofibrillar
2	Protein Solution
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Abstract: In solutions of myofibrillar protein extracted from giant squid (Dosidicus 23 gigas), the size-coarsening process of protein nanofiber is complex. At high 24 temperature (25°C), nanofiber keeps growth but with two distinguishable patterns, 25 slow rate at the initial stage with $t^{0.2}$ and the fast one at the late stage with $t^{2.3}$. The 26 intersection of these two slopes is around 300 min. Meanwhile, protein concentration 27 in solution enhances as well. These behaviors contradict to the prediction of Ostwald 28 ripening. Thus, we call this process as abnormal. These abnormal behaviors come 29 from the conformation change of some types of constitution protein molecules with 30 chemical potential reduction when they dissolve from nanofiber to solution. On the 31 other hand, low temperature (10°C) depresses this size growth. This observation 32 suggests that temperature regulates protein molecule conformation change in 33 nanofiber. The consequence of this abnormal Ostwald ripening process is that all 34 protein molecules in nanofiber are redistributed. Protein molecules with the absence 35 of conformation change in dissolution accumulate in nanofiber to cause it growing, 36 while the rest concentrates in solution. 37

Keywords: Abnormal Ostwald ripening, myofibrillar protein, molecular conformationchange, nanofiber

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In 1896 Wilhelm Ostwald descried the phenomenon of large particle growth in 50 the cost of small one as Ostwald ripening due to the surface tension which is 51 52 proportional to the particles curvature(1-3). As a result, the solute concentration keeps reduction throughout size-coarsening process. Meanwhile, an essential prerequisite, 53 no molecular conformation change when a molecule transfers from one phase to 54 another, exists but is always ignored. This condition is satisfied in inorganic and 55 organic compounds. However, in protein solutions containing particles with various 56 sizes, it should be cautious to apply Ostwald ripening theory in the size-coarsening 57 process because protein molecules are sensitive to the ambient conditions. Even salt 58 concentration variation could induce protein molecule conformation change(4). Thus, 59 it is worth to verify Ostwald ripening theory in protein solutions. 60

The existence of nanofiber in solutions of myofibrillar protein extracted from giant squid (*Dosidicus gigas*) is verified(5). In order to investigate its size-coarsening process in this solution, 2.73 mg mL⁻¹ myofibrillar protein was diluted 10 times by the buffer solution, and was assessed by the dynamic light scattering (DLS) technique and the fluorescence spectroscopy (FS) technique immediately.

In Fig. 1a, the pattern of light intensity on time depends on the temperature. High temperature $(25^{\circ}C)$ allows the light intensity monotonically growing, while low temperature $(10^{\circ}C)$ depresses this growth with almost constant light intensity throughout the experiment. Due to high activity of enzymolysis in myofibrillar protein at $25^{\circ}C(6)$, it is necessary to estimate its influence in this study. In Fig, 1b, the apparent concentrations of myofibrillar protein in three cases were determined via Bradford protein assay. It displaces two issues: 1) the apparent protein concentration remarkable increase overnight no matter whether ethylenediaminetetra acetic acid (EDTA) is present or absent; 2) the existence of EDTA has minor effect on the apparent concentration determinations. Since EDTA is a efficient agent to depress enzymolysis(7), it could be concluded that the effect of enzymolysis could be ignored in this study.

Another issue of interest in Fig. 1b is the apparent concentration rising with time. 78 Ref. 5 illustrates that the apparent concentration increase via Bradford protein assay 79 hints the more surface of protein molecule exploring to the solvent. In other words, 80 nanofiber dissolves into several pieces with small size. However, the last stage of light 81 intensity observation with fast increase at 25°C in Fig. 1a is against this prediction 82 because the light intensity is proportional to the particle number density and is 83 six-power of the size of particle(8). The light intensity increment due to the number 84 density increase is much smaller than its reduction due to the size shrinking during the 85 dissolution process. In order to clarify this puzzle, the size of corresponding nanofiber 86 was evaluated via DLS technique. Fig. 1c reveals that the decay of the autocorrelation 87 function $g_2(\tau)$ -1 becomes slow throughout the experiment. With the aid of Laplace 88 transform, the corresponding decay time distribution was obtained (Fig. S1). The 89 decay time of the slow mode τ was assessed to construct Fig. S2. In contrast to the 90 case in 10°C in which τ is almost constant throughout experiment (data do not 91 present), the dependency of τ on the observation time has two patterns, $t^{0.2}$ and $t^{2.3}$, 92 with the intersection of these two slopes around 300 min (Fig. S2). In the initial stage 93

of size-coarsening process, 0.2 is close to the value in phase-separation process,
0.212(9, 10). But in the second stage, to the best of our knowledge, 2.3 is larger than
any reported value(11, 12).

Now there is a problem. On the one hand, nanofiber keeps growth. On the other 97 hand, protein concentration in solution increases as well, which contradicts to the 98 prediction of Ostwald ripening. How does this phenomenon occur? Hinted by the fact 99 of tropmysin (Fig. S3) dissolution from F-actin with the expense of partial 100 degradation of helical structure(13), we hypothesized this abnormal Ostwald ripening 101 due to some types of protein molecule conformation change when these molecules 102 dissolve from nanofiber to solution with chemical potential reduction. As a result, 103 these types of protein molecules concentrate in solutions. In order to verify the 104 hypothesis of protein molecule conformation change, FS technique was carried out 105 immediately after 2.15 mg mL⁻¹ myofibrillar protein solution was diluted 10 times. 106 The reason to pick up FS is the high sensitivity of tryptophan to the local environment 107 in intrinsic protein florescence. As a result, change in the emission spectrum of 108 tryptophan is utilized to probe protein conformation change(14, 15). Especially for 109 minor change without secondary structure modification, such as the case of this study, 110 the traditional techniques, such as circular dichroism (CD), are infeasible. The 111 florescent spectra of the corresponding solution at various intervals were displayed in 112 Fig. 1d, in which the fluorescent light intensity keeps reduction with maximum of 113 fluorescent spectra shifting from 309 nm to 307 nm. This observation indicates the 114 occurrence of conformation change of protein molecules. In addition, the interval of 115

remarkable fluorescent light intensity reduction, around 5 h, coincides with the

117	intersection	of the two	slopes	in Fig. S2.

118	Thus, at 25°C the slow growth rate of nanofiber at the initial stage (Fig. S1) may		
119	come from the fact of high activation energy for protein molecule conformation		
120	change, for instance, 430-490 kJ mol ⁻¹ for ovalbumin at pH $7^{(16)}$.		
121	But what factors do dominate protein molecule conformation change? Fig. 1a hints		
122	that it is temperature rather than protein molecule concentration. When temperature is		
123	low as 10°C, nanofiber is stable, which is partially verified by the fact of animal		
124	muscles with less muscle shortening and drip loss around $10 \sim 15^{\circ}$ C(17, 18). Only is		
125	temperature high, such as 25°C, some types of protein molecules commencing		
126	molecular conformation change when they redistribute from nanofiber to solution in		
127	myofibrillar protein. Indeed, it is found that the tropomyosin dissociation from F-actin		
128	has a minimum temperature of 35-40°C(19).		
129	The aforementioned discussions lead to a prediction that in size-coarsening process		
130	protein molecules with the absence of conformation change when dissolving from		
131	nanofiber to solution accumulate to nanofiber while the rest types are condensed in		
132	the solution. In other words, the compositions of solution and nanofiber respectively		
133	become purer after this size-coarsening process, which was tested by the differential		
134	scanning calorimetry (DSC) assessment (Fig. 2 and Table 2), in which solutions with		
135	2.71 mg mL ⁻¹ and 0.26 mg mL ⁻¹ respectively were assessed. The dilution operation		
136	induces the size-coarsening process in myofibrillar protein solutions. Compared with		
137	the high concentration solution, the peak is sharp in the thermogram with the low		

138	denatured temperature and less denatured enthalpy in low concentration solution. The
139	sharp peak indicates the purer composition in nanofiber(20). In addition, the dilution
140	operation causes nanofiber loose, corroborated by fractal dimension d_f assessment
141	with monotonic reduction (Fig S4 and Table S1). This may be the reason for low
142	denatured temperature and less denatured enthalpy in the low concentration solution.
143	From the aforementioned discussion, a crucial conclusion is made that protein
144	molecule conformation change is an essential prerequisite for muscle protein
145	dissolution from the solid state to solution. In addition, this conformation change
146	process is regulated by temperature rather than protein molecule concentration. Just
147	because many food additives have strong interactions with muscle protein molecules,
148	they can significantly affect muscle protein molecules dissolution and thereby can
149	influence the properties of final products made by muscle (21, 22). This observation
150	also sheds light on clinical practices, for instance, the mechanism study of
151	rhabdomyolysis, a complex process associated with morbidity and mortality(23).
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153	SUPPORTING MATERIAL

154 Supporting Materials and Methods, five figures, and one table are available.

155 AUTHOR CONTRIBUTIONS

156 F. N. and R. Z. have equal contribution in this study. F. N., R. Z., and J. F. carried out

the experiments. W. P. designed the study. F. N., R. Z. and W. P. analyzed the data,

discussed and interpreted results. W. P. wrote the manuscript.

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166	The authors declare that they have no known competing financial interests or personal
167	relationships that could have appeared to influence the work reported in this paper.

168 SUPPORTING CITATIONS

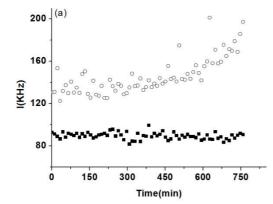
- 169 References (24–30) appear in the Supporting Material.
- 170

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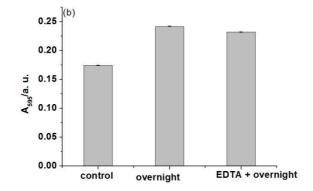
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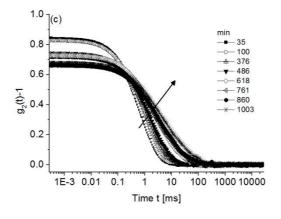
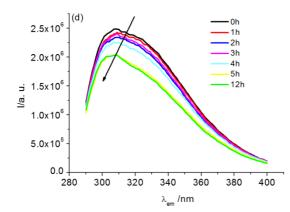
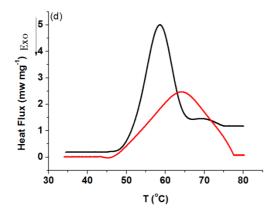


Figure 1 (a) Time evolutions of light intensity at various temperatures. Empty circle



stands for 25°C with 0.27 mg mL⁻¹ myofibrillar protein while solid square does for 249 10°C at the same protein concentration. (b) The influence of enzyme on protein 250 concentration assessment in myofibrillar protein solutions. The control is 0.27 mg 251 mL^{-1} myofibrillar protein solution. While the middle one is the solution same as 252 control but has stayed in a heat bath at 25°C overnight; and the right one is a sample 253 same as the middle one but added EDTA to depress enzyme activity. The Bradford 254 method was applied for absorbance determination. (C)The autocorrelation functions 255 of DLS at 25 ^oC in Fig.1a at some intervals. (d) The fluorescence emission spectrums 256 257 at various intervals, which displace light intensity reduction and maximum of fluorescent spectra shifting from 309 nm to 307 nm. 258 259 260

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Figure 2. DSC determinations of myofibrillar protein solutions at two concentrations 263 2.71 mg mL^{-1} (black line) and 0.26 mg mL^{-1} (red line).

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Table 1 The thermal stability of particles under various dilution ratios			
	Dilution ratio	T _{max} (℃)	$\Delta H (J/s)$
	1	63.9 ^a	0.529 ^a
	10	58.6 ^b	0.362^{b}

266 T_{max} , the denaturation temperature; ΔH , the endotherm enthalpy. Different letters (a-b) indicate

significant (P < 0.05) difference within the same row.