Uncovering mechanisms of global ocean change effects on the Dungeness crab

(*Cancer magister*) through metabolomics analysis

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ABSTRACT

The Dungeness crab is an economically and ecologically important species distributed
along the North American Pacific coast. To predict how Dungeness crab may
physiologically respond to future global ocean change on a molecular level, we
performed untargeted metabolomic approaches on individual Dungeness crab juveniles
reared in treatments that mimicked current and projected future pH and dissolved
oxygen conditions. We found 94 metabolites and 127 lipids responded in a condition-
specific manner, with a greater number of known compounds more strongly responding
to low oxygen than low pH exposure. Pathway analysis of these compounds revealed
that juveniles may respond to low oxygen through evolutionarily conserved processes
including downregulating glutathione biosynthesis and upregulating glycogen storage,
and may respond to low pH by increasing ATP production. Most interestingly, we found
that the response of juveniles to combined low pH and low oxygen exposure was most
similar to the low oxygen exposure response, indicating low oxygen may drive the
physiology of juvenile crabs more than pH. Our study elucidates metabolic dynamics
that expand our overall understanding of how the species might respond to future ocean
conditions and provides a comprehensive dataset that could be used in future ocean
acidification response studies.

INTRO

The continued increase in anthropogenic carbon dioxide emissions is leading to ocean

acidification, with the ocean absorbing an average of 25% of human-caused emissions

annually[1]. If atmospheric carbon dioxide concentration continues to rise at the current rate, the

pH of oceans is predicted to fall 0.3-0.4 units by the end of the century[2,3]. This pH drop could

exacerbate conditions in the U.S. Pacific Northwest, where the ocean pH is lower than that of

the global ocean due to natural oceanographic processes including regional upwelling, and

could pose a greater challenge to the marine inhabitants already coping with this lower pH. An

additional and compounding factor to ocean acidification is ocean deoxygenation, which co-

varies with pH and temperature. Given that global ocean temperatures will rise with global

warming from continued greenhouse gas emissions, hypoxic zones are expected to increase in

duration, intensity, and frequency[4]. It is not certain how future ocean acidification and

deoxygenation environmental stress might affect important Pacific Northwest fisheries like the

Dungeness crab fishery, which is the most lucrative and valued at more than $200 million

annually[5].

Ocean acidification is predicted to have negative indirect effects on Dungeness crab

through loss of prey directly affected by ocean acidification[6], but knowledge of how ocean

acidification might directly impact Dungeness crab is limited. It has been postulated that

Dungeness crab may exhibit limited tolerance for acid-base disturbances given past

observations of Dungeness crab as weak osmoregulators[7–10] and that acid-base balance and

osmoregulation are tightly coupled in decapod crustaceans[11]. However, it was shown that

following a brief two-week exposure to a future-predicted seawater pH of 7.4, adult Dungeness

crab are able to acclimate by increasing hemolymph ion levels (bicarbonate, calcium, chloride,

sulfate, and sodium), and by decreasing both oxygen consumption and nitrogen excretion[12]. On

the other hand, Dungeness crab larvae have shown reduced survival and development rate in

response to a 45-day exposure to the projected future seawater pH of 7.5 and 7.1[13]. Moreover,

adult Dungeness crab have shown behavioral responses to declining oxygen conditions (21-1.5

kPa pO_2 over a 5-hour period), including reduced feeding and a preference for the area with the

highest pO_2 level when placed in a seawater oxygen gradient (2.5-10.5 kPa pO_2 for 1 hour)[14].

Dungeness crab also have shown physiological responses to declining oxygen conditions (18-3

kPa pO_2 over a 6-hour period), including redistributing hemolymph to high-energy-demand

tissues[15]. Despite these clear biological responses to low pH and oxygen, the biochemical

mechanisms underlying Dungeness crab response to combined pH and oxygen stress have not
yet been defined.

One way to gain broad insight into the biochemical processes underlying the
physiological status of organisms is through surveying metabolomes, which is now possible for
nearly any species with the advancement of high-throughput metabolite profiling, also known as
metabolomics. Exploratory untargeted metabolomics approaches can offer unbiased analyses
of the composition of all detectable metabolites for the rapid and quantitative detection of stress
responses, which often leads to the development of targeted approaches and identification of
stress-indicating biomarkers[16,17]. Functional analyses using pathway inference can
subsequently be performed by integrating metabolomics data with databases and other "-omics"
datasets using bioinformatics tools to ultimately establish causal networks between different
experimental conditions and outcomes. Untargeted metabolomics in the context of
understanding ocean acidification has recently been applied to reef-building coral, and revealed
metabolite profiles that were predictive of primary production activity and molecules that could
be used as potential biomarkers of ocean acidification[18].

To better understand which biochemical pathways might be altered in the response of
Dungeness crab to ocean acidification, we applied untargeted metabolomics and lipidomics to
individual Puget Sound juveniles exposed to current pH (7.85) and future pH (7.45) conditions
for an average of 32 days. To account for the dissolved oxygen (DO) levels that naturally co-
vary with pH, we included both ambient oxygen (8.9 mg/L or saturated O_2) and low oxygen (3.0
mg/L or 33% O_2 saturation) treatments with our pH conditions in a factorial design to understand
how pH, oxygen, and/or their interaction might influence metabolite abundances.

RESULTS
Juvenile Dungeness crab general metabolome and lipidome composition
Untargeted lipidomics and metabolomics were carried out on 60 individual juvenile crabs
exposed to ocean acidification treatments (**Table 1** and **Supplementary Table 1**) through the
entire duration of their first juvenile instar until 2 days after molting to their second juvenile instar
(**Supplementary Table 2**). From the lipidomics and metabolomics profiles generated by the
West Coast Metabolomics Core using in-house open-access bioinformatics pipelines[19,20] (see
Methods for additional details), a total of 3113 lipids and 651 general metabolites were detected
of which 88% (3320/3764 total detected compounds) had spectra that did not match LipidBlast[21]
or BinBase[22] records and were classified as unknown compounds (**Supplementary Table 3**
and **4**). Eighteen lipids were detected in fewer than 50% of individual profiles and were therefore

excluded from all downstream analyses. The MS identification techniques found 29/281

metabolite classes among the 246 compounds detected and listed in Human Metabolome

Database (HMDB)[23] and 13/77 lipid classes among 195 compounds detected and listed in LIPID

MAPS Structural Database[24] (**Supplementary Fig. 1**). Among the largest classes represented

in all identified compounds were LIPID MAPS classes glycerophosphocholines, triradylglycerols,

and fatty acids, and HMDB classes organic acids, carboxylic acids, and organic oxygen

compounds, consistent with previously observed *Cancer magister* body composition proportions

of these compounds[25,26].

A combination of statistical methods reveal condition-specific responsive compounds

In general, individual metabolite and lipid profiles showed high variation regardless of

treatment group, with no obvious trends revealed by clustering the relative abundance of all

compounds by treatment groups (**Supplementary Fig. 2**). To identify specific compounds

affected by treatments, we applied both univariate and multivariate statistical approaches,

combining the strengths of different methods[27]. To assess individual effects as well as

interaction effects from pH and DO treatments on individual compounds, we used a two-way

analysis of variance (ANOVA) on the raw abundance data (**Supplementary Table 3 and 4**).

Prior to performing ANOVA, we verified compounds showed homogeneity of variances across

treatment groups (>96% compounds showed a Levene test P value > 0.05), and that metabolite

and lipid abundances were mostly (on average 69% compounds showed a Shapiro-Wilks test P

value > 0.05) normally distributed within treatment groups (**Supplementary Table 5**). Although

about one-third of compounds violated the ANOVA normality assumption, ANOVA can be

considered robust to violations of this assumption when datasets have more than 10 samples

per treatment group[28,29]. Of all compounds analyzed, 56/651 metabolites (including 24/160

known metabolites) and 98/3095 lipids (including 7/284 known lipids) showed overall model

significance at P < 0.1 and at least one model term (pH, DO, and/or pH x DO interaction)

significant at P < 0.05 (**Supplementary Fig. 3 and 4, and Supplementary Table 6**). Because

no metabolite or lipid overall model P values passed their respective 10% false discovery rate

Benjamini-Hochberg P value threshold of 1×10^{-4} and less than 3×10^{-5}, we chose to use their

uncorrected P values.

To identify discriminatory compounds and model the relationship between metabolite

profiles and exposure treatments using a multivariate linear regression approach, we used

partial least square discriminatory analysis (PLS-DA). For facilitating comparisons of metabolite

composition among treatment groups, compound abundances were centered around the mean

and scaled by the reference group (ambient pH and DO) compound standard deviations prior to

multivariate analysis[30]. The PLS-DA model for the metabolite data shows partial separation of

low pH-treated samples from ambient pH-treated samples by the first and second components

accounting for a total of 30.9% (**Fig. 1a**). Moreover, the metabolite data shows partial separation

of low DO-treated samples in the second and third components, explaining an additional 6.8%

of variation (**Fig. 1b**). For the lipid data, the PLS-DA model shows less clear treatment group

separation on the first component (**Fig. 1c**) but shows partial separation of low pH-treated

samples from ambient pH-treated samples on the second and third components explaining

14.7% of variation (**Fig. 1d**). Because the overall PLS-DA model had weak predictive power

(**Supplementary Fig. 5**), we used importance thresholds defined by the point of diminishing

returns in the PLS-DA component loadings plots (**Supplementary Fig 6.** and **Methods**), and

identified a total of 45 metabolites (including 14/160 known metabolites) and 18 unknown lipids

as compounds important in discriminating between treatment groups.

Finally, we used a multivariate nonlinear-based supervised random forest classification

to predict treatment classes from the metabolomic and lipidomic data (**Supplementary Fig. 7a-**

b). A total of 9 metabolites (including 4/160 known metabolites) and 15 lipids (including 1/284

known lipids) were considered compounds important to predicting treatments based on how

much their removal from the model decreased prediction accuracy (**Supplementary Fig. 7c-d**,

Supplementary Table 8, and **Methods**). While no compounds were commonly identified by all

3 statistical methods as was expected given the fundamental differences of the methods, 7

metabolites were commonly identified by ANOVA and PLS-DA, 8 metabolites were in common

between ANOVA and random forest, 1 metabolite overlapped between random forest and PLS-

DA, and 4 lipids overlapped between ANOVA and random forest (**Fig. 2a-b**). This combination

of analyses yielded a comprehensive set of compounds affected by treatments that no one

method was capable of capturing on its own.

Low oxygen more strongly affects compound abundance than pH

Heatmaps of the statistically identified 94 (including 35 known) metabolites and 127

(including 7 known) lipids, show several compounds respond to low pH and oxygen treatments

by commonly increasing or decreasing abundance relative to the ambient treatment (**Fig. 3a-b**).

This is also summarized by violin plots of known compound abundances in **Figure 4a-c**. For

example, 5-methoxytryptamine, butyrolactam, cysteine, cysteine, homoserine, pipecolinic acid,

piperidone, ribose, and xanthine commonly show a decrease in abundance in response to low

DO treatments relative to ambient DO treatments. Glutamic acid, maltose, and maltotriose

commonly show an increase in abundance in response to low DO treatments relative to ambient
DO treatments (**Figure 4a**). Overall, metabolite abundance tends to decrease rather than
increase in response to low DO treatments relative to ambient DO treatments, exemplified by
the mostly blue color of compound abundance averages for the low DO treatment group (green)
in **Figure 3a.** Specific to the low pH treatment, more metabolites show average abundances
similar to ambient treatment signifying pH has a less dramatic effect on metabolites compared
to low DO treatment (**Fig. 3a-b and Fig. 4a-b**). This is also apparent in the combined low pH
and low DO treatment, where metabolites and lipids show average abundances similar to the
low DO treatment signifying low DO has a more dominant effect on compounds than low pH
(**Fig. 3a-b**).

Pathway analysis reveals potential mechanisms of pH and oxygen stress tolerance

To explore the physiological relevance of treatment-responsive compounds, we
performed biochemical pathway analysis on all known compounds. While compounds classified
as sugars and fatty acids were affected by the low pH and DO treatments, most of the
significantly affected compounds were part of the amino acids class, indicating amino acid
metabolism was most significantly altered by the treatments. Abbreviated biochemical pathway
networks focusing on affected amino acids are shown in **Figure 5**, and the complete
biochemical pathway networks affected by low pH, low DO, and the combined low pH and DO
treatments can be found in **Supplementary Figures 8-10.** The trends in amino acid
abundances resulting from either low pH, low DO, or the combined low pH and DO treatment
(**Fig. 5a-c**), suggest different amino acid metabolic pathways are affected in response to each
factor or combination of factors.

Specific to low DO treatment response compared to the normoxic treatment, energy
conservation pathways appear to be most affected (**Fig. 5a**). The increased lysine with
decreased pipecolinic acid and the piperidine derivative piperidone abundance is suggestive of
downregulation of the lysine degradation pathway[31,32]. The decreased abundance of both
cysteine and the cysteine homodimer, cystine, suggests that cysteine catabolism is upregulated.
This coupled with the increased abundance of glutamic acid suggests the glutathione synthesis
pathway could be downregulated, as both cysteine and glutamic acid are precursor molecules in
glutathione synthesis[33,34]. Glutathione itself was not detected, but is typically challenging to
detect by MS techniques in marine animals due to its reactivity[35]. The glycogen intermediates
maltose and maltotriose show an increase in abundance consistent with previously observed
glycogen synthesis pathway upregulation during hypoxic stress[36]. Purine and pyrimidine

metabolic intermediates orotic acid, ribose, and xanthine show decreased abundance
suggesting purine and pyrimidine metabolic pathways could be downregulated[37].

Specific to low pH treatment response compared to ambient pH treatment, ATP
generation pathways appeared most affected (**Fig 5b**). The decreased abundance of aspartic
acid and the subtle decreased abundance of maleic acid may suggest the citric acid cycle
intermediates that they form (oxaloacetate, malic acid, and fumaric acid)[38,39] are favored.
Although, oxaloacetate was not detected (likely due to the instability of the alpha keto acid
compound)[40], and malic acid and fumaric acid did not show a significant difference across
treatments. However, citric acid did show an increase in abundance. Taken all together, the
altered abundance of these compounds suggest that citric acid cycle activity could be
upregulated in response to low pH[41]. An increase in glutaric acid in response to low pH
suggests the catabolism of its parent molecule, glutaryl CoA, which produces ATP in addition to
glutaric acid[42]. Glutaryl CoA catabolism also supports citric acid cycle activity because glutaryl
CoA can uncompetitively inhibit the alpha-ketoglutarate dehydrogenase complex that facilitates
a rate limiting step in the citric acid cycle[43].

Among the compounds affected by the combined low pH and DO treatment,
phosphoethanolamine and phophatidylethanolamine (PE(p-34:2) or PE(o-34:3)) show increased
abundance compared to ambient treatment (**Supplementary Fig. 10)**, suggesting that low pH
and DO might alter the glycerophospholipid synthesis pathway of which
phosphatidulethanolamine is a product and phosphoethanolamine is a substrate intermediate in
a rate limiting step[44]. Alanine shows a decrease in abundance, suggesting that alanine
synthesis is decreased and that its substrate for synthesis, pyruvate, may be limited[45]. The
increased abundance of cystathionine and subtle increase of methionine suggest cysteine and
homoserine synthesis are likely stalled[46] in response to combined low pH and low DO exposure.
This overlaps with the decrease in homoserine abundance and decrease in cysteine synthesis
products (cysteine and cystine) observed in low DO treatment (**Fig. 5b-c**).

DISCUSSION

We used metabolomics to explore how the Dungeness crab might respond to the
simultaneous change in oxygen and carbon dioxide in predicted future climate change
scenarios in the Pacific Northwest. Although the untargeted metabolomics and lipidomics
approaches rapidly identified hundreds of different molecular species, the majority of
compounds identified in our study, including most compounds selected by statistical methods,
lack annotations in existing metabolite databases. Thus, our understanding of their role in

response to low pH and low DO treatment is limited without extensive further validation. Still,

through shedding light on the simultaneous activity of hundreds of known compounds, we were

able to observe a dynamic range of metabolic responses among individuals within treatment

groups that indicates that Dungeness crab have flexibility in how their biochemistry

compensates for environmental change. While we attempted to control for variation between

individuals by collecting animals from the same location within a 2-month period, we were

unable to control for prior environmental exposure or genetic background of the wild-caught

animals we used. Where prior studies have found high genetic variation among individuals

within one sampling site[47,48], we suspect that both genetics and prior environmental exposure

likely contributed to the high variation in metabolite abundances among individuals within

treatment groups, which may have obscured the different treatment effects. Within-sample

variation could be reduced by sampling from a specific tissue rather than using whole animals,

however this may be challenging in early stage juvenile Dungeness crabs given their small size.

We applied three different statistical methods (ANOVA, PLS-DA, and random forest) for

selecting important metabolites in order to combine the strengths of powerful univariate and

multivariate analyses[27]. Although using univariate statistics in the strictest sense with an FDR

correction to control for false discoveries showed no statistically significant variation for

compounds across treatments, not all compounds were identified with the same confidence

level and were subject to the sensitivity of the mass spectrometry method used. Applying an

FDR correction to all detected compounds assumes that all compounds had equal chance for

discovery, which can mask important biology in highly variable datasets[49]. For this reason and

due to our use of wild-caught whole-animal samples, we chose a more liberal focus on all

compounds showing an overall ANOVA model P value < 0.1 and at least one model term P

value of < 0.05, or identified as important in PLS-DA or random forest models in our pathway

analysis.

While we acknowledge a level of uncertainty in our pathway analysis due to liberal

selection of important treatment-responsive compounds, the resulting proposed affected

pathways are consistent with previous observations of pH and hypoxia effects on different

organisms. In general, amino acid metabolism is a well-documented mechanism for stress

tolerance[50,51]. This class of compounds contains versatile chemical structures that serve as

buffering molecules, antioxidants, signaling molecules, and chemical building blocks for the

synthesis of proteins important in stress response (i.e., heat shock proteins, unfolded protein

response proteins, ion channels). Under low oxygen conditions, it is in the best interest of the

animal to limit non-essential energy consuming pathways[52]. One way Dungeness crab may do

this is through reducing the activity of evolutionarily conserved gamma-glutamyl cycle that
synthesizes glutathione and consumes ATP[53,54] by limiting cysteine availability through
catabolism. Interestingly, glutathione reduction in response to hypoxia has been observed in
multiple mammalian cell lines[55–58]. Also under low oxygen conditions in nature, food can be
scarce, and it is theorized that glycogen storage during low oxygen can help prepare cells for
low nutrient conditions. In multiple mouse and human cancer cell lines[36,59,60], increased
glycogen storage has been observed in response to hypoxia, and is induced by highly
evolutionarily conserved hypoxia-inducible factor transcriptional signalling[36]. It seems that
Dungeness crab may also adopt this strategy in combating low oxygen conditions. Under low
pH, adult Dungeness crab initially develop hypercapnia which then abates over time via
elevated hemolymph bicarbonate likely generated through gill restructuring and the upregulation
of energy consuming ion-exchange proteins[12]. Our results indicate that this process may occur
in early juveniles in low pH conditions given that we found metabolite profiles that support
energy generation via increased citric acid cycle activity.

Having parsed out effects from low oxygen and low pH, we found that in combined low
oxygen and pH conditions low oxygen has a more dramatic effect on metabolite abundance.
However, the animals in this treatment group still showed pathway alterations similar to the low
pH treatment group including increased abundance of citric-acid-cycle-related metabolites to
potentially increase ATP production. It is not yet clear what the longer-term consequences are
of these metabolic adjustments or how long these responses could be sustained. Future
avenues of research to expand on these findings should include targeted metabolomics to
confirm the compounds identified in this study as well as capture more antioxidants and citric
acid cycle intermediates that were not detected in this study. Targeted expression profiling
would also be helpful in confirming the biochemical pathway activity predictions from this study.
Ultimately, longer-term exposure experiments on crabs reared over generations might best
reveal the maximal duration that these metabolic responses can be sustained and any long-
term consequences of low pH and oxygen exposure. This exploratory metabolomics and
lipidomics analysis uncovered potential biochemical pathways affected by experiments
simulating ocean acidification and hypoxia, and can now serve as preliminary hypotheses for
deeper investigations of how the Dungeness crab (and even other crustaceans) may tolerate
global ocean change.

METHODS

Animals

Cancer magister megaolopae were collected from a single site in Puget Sound (47.950232, -

122.301784) on several days between the June and September 2016 using light traps that were

set overnight. The contents of each trap were immediately transferred into a 5-gallon bucket that

was, within 5 minutes, pooled into a cooler with ice packs and an air bubbler. After transferring

contents from 7 traps in approximately 1 hour, the cooler containing megalopae was brought

within 5 minutes into the lab and megalopae were individually transferred onto the water flow-

through system described below. The total time for transferring megalopae from the cooler onto

the seawater flow-through system was about 2 hours.

CO_2 exposure experiments

Megalopae were held in individual 250 mL customized jars on Mobile Ocean Acidification

Treatment Systems (MOATS). These systems flowed one-micron-filtered, UV-sterilized, Puget

Sound seawater maintained at 12°C. Prior to flowing through jars, seawater was degassed and

oxygen, nitrogen and carbon dioxide gases were resupplied to finely control dissolved gas

levels. Temperature, pH, and dissolved oxygen were continuously monitored throughout the

duration of the experiment by Omega thermistors, Honeywell Durafet III probes, and Vernier

optical dissolved oxygen probes, respectively. The pH was additionally validated by periodic

sampling of water for dissolved inorganic carbon and total alkalinity, and by bi-weekly

spectrophotometric pH measurements using an Ocean Optics USB 230 2000+ Fiber Optic

Spectometer with SpectraSuite software and a 5mM solution of Sigma Aldritch m-cresol purple

indicator dye. MOATS chemistry parameters were automatically adjusted through a data-driven

feedback system. Megalopae were fed *Artermia salina* (San Francisco Bay brand) at a target

concentration of 1 nauplius per mL every 3 days. Once megalopae transitioned to the first

juvenile instar, they were fed small pieces of squid. Upon transitioning to the second juvenile

instar, crabs were held on the MOATS for an additional 48 hours post-molt in attempt to reduce

variation due to potential stochastic physiological processes associated with molting [61,62].

Juveniles were then immediately frozen and stored at -80°C after lightly blotting with a paper

towel to remove excess seawater. To reduce variation from length of exposure, which in total

ranged from 20-65 days, only 60 crabs with the average exposure time (30-33 days) were

chosen for metabolomics analysis with 15 individuals from each treatment group.

Water chemistry analysis

Mean and standard deviation calculations for pH, DO, and temperature (**Table 1**) were based

on logger data (measured every 10 minutes) with the exception of pH and DO data from

MOATS 5 and 12, where Spectrophotometric pH and Presens DO measurements were instead

used due to incorrectly calibrated pH and DO probes in these MOATS. Temperature logger

readings above 25°C and below 4°C were excluded from the mean and standard deviation

analysis because these were out of the achievable range for the equipment used and were

indicative of a thermistor malfunction. Total alkalinity (TA) and salinity were discrete

measurements, with TA ranging from 2003.4 to 2036.1 µmol/kg and salinity ranging from 29.42

to 29.96 ppt throughout the course of the experiment. The complete data table is included as

Supplementary Table 1.

Sample preparation

Frozen samples were sent to the West Coast Metabolomics Center, Davis, CA for sample

preparation, metabolomics and lipidomics profiling. Whole animals were thawed, weighed, and

derivatized as previously described[63,64] (**Supplementary Table 2**). Briefly, samples were

extracted at -20°C with 2 mL of degassed acetonitrile/isopropanol/water (3:3:2) solution and

solvents were evaporated to complete dryness with a Labconco Centrivap cold trap

concentrator. Membrane lipids and triglycerides were subsequently removed from dried

samples with 50% acetonitrile, and samples were again concentrated to complete dryness. 15

mg of each sample preparation was used for metabolomic profiling. For lipidomic profiling, 15

mg of the same sample preparation was also used to which internal standards, C8-C30 fatty

acid methyl esters were added. Aliquoted samples were derivatized with methoxyamine

hydrochloride (Sigma-Aldrich) in pyridine (Acros Organics) and then by N-methyl-N-

(trimethylsilyl) trifluoroacetamide (Sigma-Aldrich) for trimethylsilylation of acidic protons.

Metabolite and lipid data acquisition

General metabolite and lipid abundances were quantified from derivatized samples by gas-

chromatography, time-of-flight mass spectrometry (GC-TOF/MS) and charged-surface, hybrid-

column, electrospray-quadrupole, time-of-flight mass spectrometry (CSH-ESI QTOF MS/MS),

respectively. For metabolites, an Agilent 6890 gas chromatograph (Santa Clara, CA) was used

with a Leco Pagasus IV time-of-flight mass spectrometer running Leco ChromaTOF software

2.32 (St. Joseph, MI). The following temperature profile was used: 50°C to 275°C final

temperature at a rate of 12°C/s and hold for 3 minutes. Injection volume was 0.5 µl with 10 µl/s

injection speed on a splitless injector with a purge time of 25 seconds. Liner (Gerstel #011711-

010-00) was changed after every 10 samples, (using the Maestro1 Gerstel software vs.
1.1.4.18). Before and after each injection, the 10 µL injection syringe was washed 3 times with
10 µL ethyl acetate. For gas chromatography, a 30 m long, 0.25 mm i.d. Rtx-5Sil MS column
(0.25 µm 95% dimethyl 5% diphenyl polysiloxane film) with additional 10 m integrated guard
column was used (Restek, Bellefonte PA). 99.9999% pure Helium with a built-in purifier (Airgas,
Radnor PA) was set at a constant flow of 1 mL/minute. The oven temperature was held constant
at 50°C for 1 minute and then ramped at 20°C/minute to 330°C at which it is held constant for 5
minutes. The transfer line temperature between gas chromatograph and mass spectrometer
was set to 280°C. Electron-impact ionization at 70V was employed with an ion source
temperature of 250°C. Acquisition rate was 17 spectra/second, with a scan mass range of 85-
500 Da.

For positively charged lipids, an Agilent 6530 QTOF mass spectrometer with resolution
10,000 was used and for negatively charged lipids, an Agilent 6550 QTOF mass spectrometer
with resolution 20,000 was used. Electrospray ionization was used to ionize column elutants in
both positive and negative modes. Compounds were separated using a Waters Acquity ultra-
high-pressure, liquid-chromatography charged surface hybrid column C18 (100 mm length x 2.1
mm internal diameter; 1.7 um particles) using the following conditions: mobile phase A (60:40
acetonitrile:water + 10 mM ammonium formiate + 0.1% formic acid, mobile phase B (90:10
isopropanol:acetonitrile + 10 mM ammonium formiate + 0.1% formic acid), 65°C column
temperature, a flow rate of 0.6 mL/minute, an injection volume of 3 uL, an injection temperature
of 4 C, and a gradient of 0 minutes 15%, 0-2 minutes 30%, 2-2.5 minutes 48%, 2.5-11 minutes
82%, 11-11.5 minutes 99%, 11.5-12 minutes 99%, 12-12.1 minutes 15%, and 12.1-15 minutes
15%. The capillary voltage was set to +3.5 and -3.5 kV, and the collision energy to 25 or 40 eV
for positive and negative modes. Mass-to-charge ratios (m/z) were scanned from 60 to 1700 Da
and spectra acquired every 2 seconds.

Spectral data processing
Acquired metabolite data were processed using UC Davis's BinBase workflow, which performs
data processing including peak detection at signal-to-noise levels of 5:1 throughout the
chromatogram. Resulting apex masses are reported for use in the BinBase algorithm to
facilitate metabolite identification and quantification[65]. Metabolites were identified through
comparison to the BinBase database[22] and peak heights were normalized to total metabolite
content[66].

For acquired lipid data, MassHunter (Qual v. B05.00) was used to find peaks from the

raw data in up to 300 chromatograms. These peaks were imported into Mass Profiler

Professional for alignment to determine which peaks occur in at least 30 % of the

chromatograms. These peaks were then quantified with MassHunter. Resulting accurate mass

data and tandem MS/MS spectra were compared to LipidBlast libraries for compound

identification[21]. All spectra mapping to previously identified compounds are accessible in the

public database MassBank of North America (http://mona.fiehnlab.ucdavis.edu/). Spectra that

did not map to previously identified compounds are accessible by searching their BinBase

identifiers listed in **Supplementary Tables 3 and 4** in the BinVestigate database

(http://binvestigate.fiehnlab.ucdavis.edu).

Statistical analyses

All statistical analyses excluded 18 compounds that were detected in less than half of the

individual crabs surveyed, and were carried out in R, except for PLS-DA and random forest

analyses which were carried out using the MetaboAnalyst web interface.

Univariate statistics

Prior to applying a univariate statistical test, normality and heteroscedasticity of compound

abundances within treatment groups was first assessed using the Shapiro-Wilks test and the

Levene test, respectively (**Supplementary Table 5**). After confirming normality and

heteroscedasticity assumptions were correct, a two-way ANOVA was then applied to each

compound. A Benjamini-Hochberg FDR correction was applied to ANOVA P values to correct

for multiple testing, but corrected P values were not used in selecting important compounds.

Metabolites and lipids were selected for pathway analyses if they had an overall model ANOVA

P value less than 0.1 and an effect P value with less than 0.05 without an FDR correction

applied. Results from ANOVA tests are reported in **Supplementary Table 6**.

Multivariate statistics

Prior to applying multivariate testing, data were normalized by mean-centering and scaling by

the reference-group standard-deviations of each compound in order to make compounds more

comparable to one another[30]. Normalized data was then uploaded to MetaboAnalyst with no

further normalizations applied. The PLS-DA function was run with default settings and a table of

component loadings for each compound were exported from MetaboAnalyst (**Supplementary**

Table 7). Compound loadings were ordered by from greatest variance explained to lowest, and

plotted for PLS-DA components. Importance thresholds were placed at the point of diminishing

returns in the plot curves (**Supplementary Fig. 6**). Specifically for metabolites, loadings were

plotted PLS-DA components 1, 2, and 3 (**Supplementary Fig. 6a-c**) since these PLS-DA

components gave the greatest separation between treatment groups (**Fig. 1a-b**). The points of

diminishing returns that importance thresholds were placed in the metabolite loadings plots

were as follows: component 1, loadings threshold > 0.05; component 2, loadings threshold >

0.1; and component 3, loadings threshold > 0.05. Specifically for lipids, loadings were plotted for

PLS-DA components 2 and 3 (**Supplementary Fig. 6d-e**) since these gave the best separation

between treatment groups, while component 1 could only distinguish the ambient from the

altered treatment groups (**Fig. 1c-d**). The points of diminishing returns that importance

thresholds were placed in the lipids loadings plots were as follows: component 2, loadings

threshold > 0.05; and component 3, loadings threshold > 0.05. For the random forest analysis,

the Random Forest function was run with 2000 decision trees and either 25 predictors for the

metabolite dataset or 56 predictors for the lipid dataset, conforming to the default classification

value being the square root of the number of variables[67]. The complete table of important

variables and their mean decrease in random forest accuracy prediction exported from

MetaboAnalyst (**Supplementary Table 8**). The mean decrease in prediction accuracy for each

compound was ordered from largest to smallest and plotted (**Supplementary Fig. 7c-d**).

Importance thresholds were drawn from the plots at the points of diminishing returns, which for

metabolites was a mean decrease in accuracy > 0.001 and for lipids was a mean decrease in

accuracy > 0.00047.

Heatmaps in **Figure 3** and **Supplementary Figure 2** were made using the heatmap

function in the MetaboAnalyst web interface with Euclidean distance, ward clustering, plotting

auto-scaled compound abundance averages for each treatment group.

Pathway analysis

Pubchem and KEGG identifiers were obtained for metabolites and lipid international chemical

identifier keys provided by the West Coast Metabolomics Core using Chemical Translation

Service[68] and Pubchem Identifier exchange[69]. Metamapp[70] was then used to map metabolites

by chemical and biochemical relationships. The generated SIF file (**Supplementary File 1**) was

imported into Cytoscape[71]. For visualizing the pH, DO, and pH:DO interaction effects, both

mean abundance fold change and Cohen D effect size values were calculated for each

compound within a treatment group relative to the ambient treatment group. Cohen D effect size

$(\frac{\mu_{treatment} - \mu_{ambient}}{s_{pooled}})$ was calculated for each compound raw abundance within a treatment group

relative to the ambient treatment group using the R package effsize[72]. The treatment group $-\log_2$

fold change values and the Cohen D effsize output from R were added to a node attribute file

(**Supplementary Table 9**) that was also imported into Cytoscape. Low pH, low DO, and low

pH:low DO networks were styled using the following settings: node shapes were set according

to statistical method used to select the compound (ANOVA, triangle; PLS-DA, square; random

forest, diamond; both PLS-DA and ANOVA, hexagon; not selected as important by any method,

circle); node border was color by the treatment group mean compound abundance fold change

relative to ambient (1.25X-2X fold change, yellow gradient' -1.25X – 1.25X fold change, grey; -

1.25X - -10X fold change, blue gradient); node fill was colored by the Cohen D effect size of

treatment on compound abundance relative to ambient (effect size of 0.5-1, yellow gradient; -

0.5-0.5, white; and -0.5- -1, blue gradient); node shape size and label size were set to fixed

values (150 height and width with size 50 font for statistically selected compounds, and 30

height and width with size 12 font for compounds not identified by statistics (**Supplementary**

Fig. 8-10).

DATA AVAILABILITY

All mass spectra mapping to previously identified compounds are accessible in the public

database MassBank of North America (http://mona.fiehnlab.ucdavis.edu/). Spectra that did not

map to previously identified compounds are accessible by searching their BinBase identifiers in

the BinVestigate database (http://binvestigate.fiehnlab.ucdavis.edu). Processed metabolite and

lipid spectral data are included as **Supplementary Table 3 and 4**.

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701

702 **AUTHOR CONTRIBUTIONS**

703 K.M.N., P.M., and D.S.B conceived the project. K.M.N., P.M., and D.S.B. advised research.
704 M.M. and D.P. performed experiments. S.A.T. performed statistical analyses and pathway
705 analysis. S.A.T. prepared the manuscript with edits from K.M.N., P.M, and D.S.B.
706

707 **COMPETING INTERESTS**

708 The authors declare no competing interests.
709

treatment	temp ($\circ$C)	Target pH	pH	Target DO	DO (mg/L)
ambient pH ambient DO	12.17 $\pm$ 0.17	7.85	7.83 $\pm$ 0.05	8.9	8.9 $\pm$ 0.2
ambient pH low DO	12.12 $\pm$ 0.12	7.85	7.85 $\pm$ 0.02	8.9	3.0 $\pm$ 0.1
low pH ambient DO	12.14 $\pm$ 0.08	7.45	7.45 $\pm$ 0.05	3.0	8.9 $\pm$ 0.1
low pH low DO	12.15 $\pm$ 0.16	7.45	7.49 $\pm$ 0.11	3.0	3.0 $\pm$ 0.3

Table 1. Summary of water chemistry for experimental treatments showing the mean $\pm$ standard deviation for each parameter. Temperature (temp), pH, and DO were continuously measured by logger probes. Total alkalinity (TA) and salinity were discrete measurements, with TA ranging from 2003.4 to 2043.3 µmol/kg and salinity ranging from 29.42 to 29.96 ppt throughout the course of the experiment. Refer to **Supplementary Table 1** for complete data table.

Figure 1. Summary of PLS-DA analysis. Metabolite data show partial separation of low pH treatment groups (blue and purple) from low DO treatment groups (pink and green) by (**a**) Components 1 and 2 and (**b**) components 2 and 3. Lipid data show partial separation of treatment groups by (**c**) components 1 and 2 and (**d**) components 2 and 3. Shapes correspond to treatment groups (ambient pH and DO,Δ ; ambient pH and low DO, + ; low pH and ambient DO, x ; low pH and low DO, ◊ . Numbers refer to the sample number.

a

ANOVA   PLS-DA

41
(43.6%)   7
(7.4%)   37
(39.4%)

0
(0%)

8
(8.5%)   1
(1.1%)

0
(0%)

Random
Forest

b

ANOVA   PLS-DA

94
(74%)   0
(0%)   18
(14.2%)

0
(0%)

4
(3.1%)   0
(0%)

11
(8.7%)

Random
Forest

Figure 2. Venn diagrams
DA, and/or random forest a

Figure 3. Heatmap plots showing the average compound abundance (average peak intensity) for each treatment group for the (**a**) 94 general metabolites and (**b**) 127 lipids selected by multivariate and univariate statistical methods used to evaluate treatment effects. Individual known and unknown ("unk") compound names are listed over the columns and treatment groups are listed over the rows (orange, ambient pH, ambient O_2; green, ambient pH, low O_2; purple, low pH, ambient O_2; and blue, low pH, low O_2). Compound average abundances are shown as auto-scaled within each compound ("relative abundance").

Figure 4. Compound abundance levels (peak intensities) of known metabolites and lipids selected by univariate and multivariate statistical methods used to evaluate treatment effects. Violin plots with boxplot insets show the distribution of compound abundances within each treatment group for compounds statistically showing (**a**) DO treatment effect, (**b**) pH treatment effect, and (**c**) DO:pH interaction effect. Binbase names for each known compound are listed across the top of each plot. Abundance levels (peak intensities, noted as "quant") are listed on the y-axis while treatments are listed along the x-axis. Statistical methods that each known compound was identified by are noted by shapes in the upper corners of each plot (ANOVA, triangle; PLS-DA, square; and random forest, diamond).

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Figure 5. Differential amino acid network effects from (**a**) low DO, (**b**) low pH, and (**c**) combined low pH and low DO. Compounds are clustered by chemical similarity. Node fill color is colored by Cohen D effect size comparing the treatment group to the ambient pH,ambient DO group, and node borders are colored by the treatment group mean fold change relative to the ambient pH,ambient DO group. Node shapes indicate the statistical method(s) from which the compound was classified as important. Node shape and label size are enlarged if the compound was identified as important by a statistical method. Gray edges indicate nodes sharing a KEGG biochemical pathway(s).