Enhanced biodegradation of naphthalene by *Pseudomonas* sp. consortium

immobilized in calcium alginate beads

Kunal Dutta[1], Sergey Shityakov[2], Ibrahim Khalifa[3,4], Saroj Ballav[1], Debarati Jana[1],
Tuhin Manna[1], Monalisha Karmakar[1], Priyanka Raul[1], Kartik Chandra Guchhait[1] and
Chandradipa Ghosh[1,]*

[1]Microbiology and Immunology Laboratory, Department of Human Physiology with
Community Health, Vidyasagar University, Midnapore -721102, West Bengal, India.

[2]Department of Anaesthesia and Critical Care, University of Würzburg, 97080
Würzburg, Germany.

[3]Food Technology Department, Faculty of Agriculture, 13736 Moshtohor, Benha
University, Egypt.

[4]College of Food Science and Technology, Huazhong Agricultural University, Wuhan
430070, China.
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18 * Corresponding author:
19   Prof. Chandradipa Ghosh
20   Microbiology and Immunology Laboratory
21   Department of Human Physiology with Community Health
22   Vidyasagar University
23   Midnapore - 721102
24   West Bengal,
25   INDIA
26   Email : ch_ghosh@mail.vidyasagar.ac.in
27   Phone : +91 3222 276554/555/557 Extn. 450
28   Fax   : +91 3222 275329
29
30
Kunal Dutta   : orcid.org/0000-0002-0818-8787

Sergey Shityakov   : orcid.org/0000-0002-6953-9771

Ibrahim Khalifa   : orcid.org/0000-0002-7648-2961

Abstract

Polycyclic aromatic hydrocarbons (PAHs) belong to a group of organic

pollutants widespread in nature. PAHs are known as a serious threat to living beings and

their hydrophobic nature make them major contributors to groundwater contamination.

Herein, naphthalene biodegradation potential by *Pseudomonas putida* strain KD10 and

a blend of *Pseudomonas* sp. isolated from different petroleum refinery waste sites are

reported. *Pseudomonas putida* strain KD10 and the *Pseudomonas* sp. consortium were

immobilized in calcium alginate beads (CABs) and naphthalene biodegradation

efficiency was studied in liquid medium with optimized pH and temperature.

Additionally, naphthalene 1, 2-dioxygenase (*nah*Ac) gene sequence analysis reveals two

mutated residues inside the chain A of *nah*Ac those are not seen in the vicinity of the

active site. The gas-phase binding free energy (ΔG_{London}) was found to be -7.10 kcal

mol^{-1} for the mutant *nah*Ac encoded by the *Pseudomonas putida* strain KD10 which

closely resembles that of the wild type variant. HPLC degradation kinetics showed that

KD10 biodegrades naphthalene at the rate of 79.12 mg L^{-1} day^{-1} and it was significantly

elevated up to 123 mg L^{-1} day^{-1} by the immobilized *Pseudomonas* sp. consortium. The

half-life ($t_{1/2}$) for naphthalene biodegradation was 3.1 days with the inhibition constant

(k_i), substrate saturation constant (k_s) and maximum specific degradation rate constant

(q_{max}) of 1268 mg L^{-1}, 395.5 mg L^{-1} and 0.65 h^{-1}, respectively, for the *Pseudomonas*

putida strain KD10. However, the $t_{1/2}$ value was significantly reduced to 2 days along

with k_i, k_s and q_{max} value of 1475 mg L^{-1}, 298.8 mg L^{-1} and 0.71 h^{-1}, respectively, for the

immobilized *Pseudomonas* sp. consortium. The GC-MS data suggest that metabolites of

naphthalene biodegradation by KD10 might step-into the TCA cycle via the meta-

cleavage pathway of catechol biodegradation. It is concluded that naphthalene

biodegradation performance by immobilized *Pseudomonas* sp. consortium was superior

to pure *Pseudomonas putida* KD10. The microbial consortium immobilization could be

a useful tool for water quality management and environmental remediation.

Key words: *Pseudomonas* sp., petroleum wastes, biodegradation, cell immobilization,

mutant naphthalene 1, 2-dioxygenase, rigid-flexible molecular docking.

Highlights

• Superior naphthalene biodegradation by *Pseudomonas* sp. consortium

immobilized in calcium alginate beads.

• A high mutation prone amino acid stretch inside chain A of naphthalene 1, 2-

dioxygenase has been identified.

• A new naphthalene biodegradation pathway by *Pseudomonas putida* strain

KD10 has been proposed.

Polycyclic aromatic hydrocarbons (PAHs) have long been considered as the

potential hazard for living beings (Kumari et al., 2018). Several environmental agencies

including US-EPA, European Union, Environment-Canada, registered PAHs as the

priority pollutants that require urgent actions to clean the environment (de Gannes and

Hickey, 2017; Wang et al., 2018). Physicochemical properties of PAHs makes them

major contributor for soil and ground water contamination and it spread through bio-

magnification, particularly in the heavily industrialized areas (Norris, 2017). According

to the Environmental Health Hazard Assessment, U.S.A., naphthalene (NAP), is not

safe for drinking when the concentration is more than 170 ppb (Bruce et al., 1998).

Different orthodox and expensive techniques of environmental remediation, *viz.,*

incineration, gasification, plasma-gasification had been substituted by green-

technologies such as bioremediation, phytoremediation, nanoremediation, *etc.* (Thomé

et al., 2018).

Bioremediation is considered as one of the most cost-effective and eco-friendly

oil spill management technique (Wilson and Jones, 1993). However, considering the

vast volume of mobile open water system, bioremediation in aquatic environment

stumble upon several limiting factors such as low local concentration of the effective

microorganisms, loss of active microorganisms, *etc.* (Chen et al., 2017). Conversely,

cell immobilization by calcium alginate beads (CABs) is advantageous, owing to

maintaining high local concentration of the effective microorganisms. In addition, this

features of cell immobilization render bacterial cell membrane stability, causes

minimum production cost, can be stored for future use (Bhardwaj et al., 2000), and

overall can substantially improve the stability and efficiency in the bioremediation

process (Mrozik and Piotrowska-Seget, 2010; Tyagi et al., 2011). Cell immobilization

using CAB is a convenient method where maximum cells remain viable and can tolerate

high concentration of toxicant for a long period of time (Lee and Heo, 2000). Moreover,

calcium alginate is nontoxic to the bacterial cell and it has low production cost which

facilities easy reuse (Bhardwaj et al., 2000). For this reason, worldwide initiatives have

been taken to improve overall biodegradation efficiency by development of microbial

consortium, including optimization of other conditions like pH, temperature, *etc.* (Chen

et al., 2017). Biodegradation of a toxicant (complex nutrient for bacteria) by the mixed

bacterial consortium can efficiently enhance biodegradation rate (Kumari et al., 2018).

Different biodegradation pathways of each individual bacterium (Dutta et al., 2018) or a

metabolic intermediates of one bacteria may act as the starting material of other bacteria

(Surkatti and El-Naas, 2018), or different genetic makeup (Woyke et al., 2006) or

synergetic effects of different microbial species, (Ghazali et al., 2004) or by

synthesising different variant of catalytic enzymes.

Microbial consortium are of different types, mainly bacteria-bacteria, bacteria-

fungi (Kumar and Philip, 2006; Yamaga et al., 2010) and it can effectively enhanced the

biodegradation process (Murthy and Manonmani, 2007; Vaidya et al., 2018). Moreover,

it is effective because of its cooperative nature, in particular, they can combine their

metabolic capabilities to utilize the common complex nutrient (Gilbert et al., 2003).

Biodegradation of complex hydrocarbon mixture necessary require the cooperation of

more than one bacterium, which offers broad range of diverse enzymes that finally

enhance the success of the overall processes (Wongwilaiwalin et al., 2010).

The aim of the present study is to evaluate the NAP biodegradation potential of

Pseudomonas putida strain KD10 as free cell and as immobilized in CABs.

Additionally, NAP biodegradation efficiency of the *Pseudomonas* sp. consortium was

studied through the cell immobilization in CABs. In the previous study morphological

changes during biodegradation of NAP was first reported that indicates *Pseudomonas*

putida strain KD9 decrease in size and shape from rod to sphere and their specific

growth rate was also little slower (Dutta et al., 2018). Keeping this in the mind, both

morphological types of the bacteria were applied. Additionally, the naphthalene 1, 2-

dioxygenase (*nah*Ac) was sequenced and analysed. In this present set of study we have

identified that naphthalene biodegradation performance by immobilized *Pseudomonas*

sp. consortium is superior than that of individual free or immobilized *Pseudomonas*

putida strain. Further, a common mutation prone amino acid stretch in chain of A of the

naphthalene 1, 2-dioxygenase have been identified. In addition, the intra and

interspecies comparison of the naphthalene 1, 2-dioxygenase suggested sequence

dissimilarity among different species and in few same bacterial species. Moreover, the

GCMS based metabolite analysis suggested *Pseudomonas putida* KD10 reaches TCA

cycle via D-gluconic acid and it follows meta-cleavage of catechol biodegradation. We

apprehend that, consortium in immobilized form may give rise to efficient models for

biodegradation.

2. Materials and methods

2.1. Chemicals

Naphthalene was purchased from Sigma-Aldrich chemicals Pvt. Ltd. (USA) and

all the chemicals used for the media preparation were procured from the HiMedia

Laboratory (Mumbai, India) together with the GC-MS and HPLC grade solvents were

from the Fisher Scientific (Mumbai, India). Sodium alginate (CAS No. 9005-38-3) of

medium viscosity was purchased from Merck Pvt. Ltd. (USA).

2.2. Microorganisms, growth media, growth condition and consortium preparation

Soil samples were collected from the petroleum refinery waste sites near Indian

Oil (Haldia, West Bengal, India). The enrichment isolation and strain identification was

carried out according to the standard protocol described previously (Dutta et al., 2017).

The carbon deficient minimum medium (CSM), with a pH of 7.1, was used to cultivate

the bacteria. NAP was used as sole source of carbon and energy in carbon-deficient

minimal media (CSM) with following compositions: 0.2 g L^{-1} $MgSO_4$, $7H_2O$; 0.08 g L^{-1}

$Ca(NO_3)_2$, $4H_2O$; 0.005 g L^{-1} $FeSO_4$, $7H_2O$; 4.8 g L^{-1}, K_2HPO_4; 1.2 g L^{-1} KH_2PO_4.

Pseudomonas putida strain KD6 (KX786159.1) and strain KD9 (KX786158.1) and a

newly isolated *Pseudomonas putida* strain KD10 (KX786157.1) with different variant

of naphthalene 1, 2-dioxygenase were used to prepare the blend of *Pseudomonas* sp.

consortium. The consortium was maintained in Luria-Bertani broth at 31°C with 150

rpm in order to grow the bacterial cell in its normal size and shape. Additionally,

bacterial cells were grown in CSM with naphthalene as sole source of carbon and

energy to obtain altered morphological variant.

2.3. Gene sequence analysis

2.3.1. Detection of naphthalene 1, 2-dioxygenase and catechol 2, 3-dioxygenase

The conventional polymerase chain reaction (PCR) for the naphthalene 1, 2-

dioxygenase (*nah*Ac) and catechol 2, 3-dioxygenase (*nah*H) using specific primers

(Table S1) was performed. Additionally, PCR product of *nah*Ac was sequenced and

analysed according to the methods described previously (Dutta et al., 2017).

2.3.2. Clustering and phylogenetic analysis

The evolutionary distance of naphthalene 1, 2-dioxygenase among different

bacterial species was analysed using BLOSUM weighted matrix followed by pairwise

distance computation using MEGA (v7.0) (Kumar et al., 2016). The distance matrix

was then clustered using R programming (Team, 2013) to construct the cladogram and

heatmap plot. The phylogenetic position of the isolated *Pseudomonas putida* strain

KD10 was analysed using previous method (Dutta et al., 2017).

2.3.3. Rigid-flexible molecular docking

The molecular docking studies were conducted using Auto Dock (v4.2.1)

(Morris et al., 1998). Briefly, the centre grid dimensions were set to

20.271×61.989×87.168 with grid spacing of 0.375 Å. The virtual screening was

repeated for 10 times with the same unaltered docking parameters having 2.0 Å cluster

tolerance. Additionally, the rigid-flexible molecular docking was performed using

Molecular Operation Environment (Chemical Computing Group, Montreal Inc.,

Canada). The latter scoring function was employed to identify most favourable docked

poses and to estimate the binding affinity of the protein-ligand complexes and the non-

covalent interactions were analysed using the previous method (Salentin, S., et al.,

2015).

2.4. Enzyme kinetic assay

The enzyme kinetic parameters of the naphthalene 1, 2-dioxygenase $_{I250, V256}$ was

conducted using cell-free extract of the *Pseudomonas putida* strain KD10, grown in 250

mL of CSM with NAP (500 mg L^{-1}) as sole source of carbon and energy as described by

the standard method (Dutta et al., 2017).

2.4. Detection of solvent efflux pumps system

The solvent efflux pump system (*srp*ABC) was amplified using conventional

polymerase chain reaction in a thermal cycler (Mastercycler® nexus gradient,

Eppendrof, (Germany) using the genomic DNA of the KD10 as template and forward,

reverse primers (Table S1). Standard reaction mixtures were prepared by the procedures

described previously (Dutta et al., 2018).

2.5. Cell immobilization

2.5.1. Cell immobilization in calcium alginate beads

Individual bacterial cell and *Pseudomonas* sp. consortium were immobilized

according to the standard protocol described previously (Daâssi et al., 2014). Briefly as,

sodium alginate was dissolved in 0.9 wt. % NaCl (1 gm in 40 mL 0.9 wt. % NaCl) for

24 h and sterilized by autoclaving (121°C for 15 min). Two grams of bacterial cell mass

was added to 8 ml of NaCl solution and again added to the sterile alginate solution. The

mixture was then gently vortex for complete homogenization. The mixture was then

extruded drop wise through a hypodermic syringe into chilled sterile $CaCl_2$ solution

(Figure 2). The beads were hardened in the same solution at room temperature with

gently stirring for 1 h. Finally, the beads were washed several times with 0.9 wt. %

NaCl to remove excess calcium ions and free cells. The beads had an average diameter

of 0.5 mm and stored at 4°C. Sterile beads (without microorganisms) were used to

monitor the abiotic loss of NAP. Sodium alginate of medium viscosity ($\geq$2,000 cP) was

used to prepare the calcium alginate beads (CABs).

2.5.2. Cell viability count

The viable cell enumeration in the CABs was performed by using previous

protocol (Usha et al., 2010). In brief, CABs was washed in saline and keep submerged

for 10 min (for saline soaking). Following soaking, the CABs were shaken with glass

beads for 15 min and 1 g CABs were homogenized in 9 ml saline. The saline was then

used for viable cell enumeration onto a nutrient agar plate. Plating was done with this

treated saline by series dilution method upto to 10^{-5} dilutions on nutrient agar plates.

Plating was also done for the initial CFU count. For every dilution, 10 µl of the solution

was plated. Plating was done by the pour plate method. Plates were incubated at 37°C

for 24 h. (Jain and Pradeep, 2005).

2.6. Scanning electron microscopic (SEM) study

Samples for the SEM analysis were prepared by the protocol described

elsewhere (Dutta et al., 2018). The dried samples were coated in sputter coater

(Quorum-SC7620) under vacuum with a thin gold layer right before SEM analysis

using a scanning electron microscope (Zeiss, EVO 18, Germany) with an accelerating

voltage of 5 kV.

2.7. Biodegradation kinetics

2.7.1. Biodegradation of naphthalene in liquid medium

Naphthalene biodegradation study was conducted by the protocol described

previously (Dutta et al., 2017). The conical flasks were incubated at 31°C with 150 rpm

and uninoculated conical flask were used as control. Culture medium was collected at

regular interval of 72 h for degradation and growth kinetic study. Additionally, the

effect of initial concentration of naphthalene (150-2500 mg L^{-1}) was studied with

different immobilized systems.

2.7.2. Chemical analysis

NAP biodegradation were analysed by using the 1260 infinity series (Agilent, Santa

Clara, CA, USA) HPLC system equipped with Zorbax SB-C18 reversed-phase column

$(4.6 \times 12.5$ mm, 5 µm). The NAP concentration analysis was conducted using isocratic

elution conditions with the mobile phase 80:20 (v/v) methanol: water at a flow rate of 1

ml min^{-1}. The detection was performed at 254 nm according to the protocol described in

the literature (Dutta et al., 2017). The metabolic intermediates of NAP biodegradation

were analyzed using the GC-MS system (GC Trace GC Ultra, MS-Polarisq, Thermo

Scientific India Pvt. Ltd) equipped with a capillary column (TR-WaxMS, 30 m $\times$ 0.25

mm [ID] $\times$ 0.25 µm film thickness) by the protocol described elsewhere (Dutta et al.,

2017). The entire analysis was performed in electron ionization, at full scan mode. The

metabolite identification was based on the mass spectra comparison using the NIST

Mass Spectral library (v2.0, 2008).

2.8. Data analysis

The first-order degradation kinetics model was used to estimate the residual

naphthalene in CSM using equations 1, the algorithms as expressed determine the half-

life ($t_{1/2}$) values of NAP in CSM. The substrate inhibition kinetic parameters were

calculated using equation 2.

$$C_t = C_0 \times e^{-kt} \qquad [1]$$

$$q = q_{max} \frac{S}{K_s + S + S^2/K_i}$$

300   The effect of SCS on growth pattern was measured by calculating the difference

301   in optical density at 600_{nm} and expressed it by slight extension (Equation 4) the

302   Gomperz's sigmoid growth fit equation (Equation 3).

$$y = ae^{-\exp(-k\,(x-x_c))} \qquad [3]$$

$$\Delta X_c = X_{c2} - X_{c1} \qquad [4]$$

306   The growth pattern change by SCS was considered as CFU shift (Equation 5). Where,

307   X_{c2} = Optical density at X_{c2} and X_{c1} = Optical density at X_{c1}. NAP biodegradation was

308   enhanced by 0.5 gm. % sucrose supplementation with CFU $_{shift}$ of 4.4×10^8 (Figure 1).

$$CFU_{shift} = \Delta X_c \qquad [5]$$

The bacterial growth kinetics were analysed by applying the Gomperz's model

using Levenberg-Marquardt algorithm in Origin® 2016 (California, USA) software and

the degradation kinetic by using Graphpad Prism® 6.01 (San Diego, CA, USA).

3. Results and discussion

3.1. Isolation, identification and growth patterns of NAP degrading bacterial strain

Pseudomonas putida strain KD10 was isolated from petroleum refinery waste

with its distinguished colony morphology (Figure S1a, b) on CSM agar plate after the

conventional enrichment isolation protocol. The PCR product of 16S rRNA gene of the

isolated strain was sequenced and deposited at NCBI (https://www.ncbi.nlm.nih.gov/)

under the GenBank accession number KX786157.1. Multiple sequence alignment

(MSA) followed by phylogenetic assessment suggested that the strain is belongs to the

Pseudomonas putida group and it showed 99% sequence similarity with the previously

deposited sequences of *Pseudomonas putida* (Figure S2).

NAP biodegradation property of the *Pseudomonas putida* KD10 was

preliminary confirmed by the catechol test (Figure S1C) followed by growth pattern

analysis in CSM with NAP as sole source of carbon and energy. Additionally, growth

pattern with 0.5 gm % sucrose as a secondary carbon supplement (SCS) was compared

with growth patterns in non-supplemented growth media. The Gompertz's model fit

(Equation 3) was used to analyse the growth curve (Table S4).

3.2. Pseudomonas sp. *consortium, growth pattern and NAP biodegradation*

Growth pattern of the *Pseudomonas* sp. consortium, as shown in Figure 1 was

optimized at 31°C and pH 7.1 suggest a successful cooperation among these three

Pseudomonas putida strain KD6, KD9 and KD10. Further, the growth curve model

(Figure 1) indicates a significant increase of total biomass of these cell population when

cultivated as consortium. The artificial microbial consortium (AMC) by the stain KD6,

KD9 and KD10 showed enhanced tolerance of the high concentration of NAP (Bhatia et

al., 2018).

3.3. Gene sequence and phylogenetic analysis

The naphthalene 1, 2-dioxygenase (*nah*Ac), encoded by *Pseudomonas putida*

strain KD10 have only two point mutations at I250, and V256 which replaced

methionine and glycine respectively (Figure S6, 3A). In previous study, additional

mutations were found in the same chain A at K200, A210, E264, M284 and N334 by

replacing glutamic acid, glycine, aspartic acid, isoleucine, respectively (Dutta et al.

2017, 2018). The three dimensional structural analysis suggests that the single point

mutations at 200, 210, 284 and 334 cause little structural change in comparison to the

wild type variant of the *nah*Ac (Figure 3B). Conversely, the stretch of amino acid

sequences at the close proximity of the active site residues exactly from 248 to 266

suggest a structural mismatch among all three mutant variants of the *nah*Ac (Figure

3B). The active site of an enzyme tends to evolve fast to attain its maximum

performance and functionality in a particular environment (James and Tawfik, 2003).

The occurrence of mutation from amino acid 248 to 266 of chain A, was a

common feature in all three mutant variants of *nah*Ac and this particular stretch of

amino acid is very close to the active site residues (Figure 3C). This implies that the

"248-266" amino acid stretch is highly prone to mutation and it may influence on the

enzymatic efficiency and environmental adaptability. Thus, our study provides a new

insight, which could be beneficial for rational approaches of enzyme redesigning.

Naphthalene 1, 2-dioxygenase is the primary enzyme for naphthalene biodegradation

(Lee et al., 2019). Further, the biodegradation performance of a bacterial strain is linked

with the three dimensional structure of the catalytic enzyme and ligand binding posture

(Singh et al., 2019). Previous studies suggested alteration of a single amino acid in the

catalytic domain of the enzyme caused different ligand binding postures which

eventually lead to an unique biodegradation pathways (Ferraro et al., 2006). In addition,

site directed mutation in the catalytic domain offers superior enzyme activity (Parales et

al., 1999; Parales, 2003). The comparative analysis of binding free energy helps to

comprehend the superior naphthalene biodegradation performance by the *Pseudomonas*

sp. consortium, which is in fact the summative activity of all three mutant variant of

naphthalene 1, 2-dioxygnease.

The evolutionary trace on the *nah*Ac among other species were studied through

pairwise distance matrix analysis (Figure 4), suggests a significant intra and inter

species difference among *Sphingopyxis sp., Croceicoccus naphthovorans, Burkholderia*

multivorans, Burkholderia sp. Massilis sp. and Cycloclasticus sp. Conversely, other

Pseudomonas sp. particularly, *Pseudomonas benzenivorans, Pseudomonas balearica,*

Pseudomonas stutzeri, Pseudomonas kunmingensis, Pseudomonas frederikbergensis

have similarities in their version of *nah*Ac. A few other species such as *Xenophilusa*

zovorans, Ralsoniam annitolilytica, and *Paraburkholderia aromaticivorans* does not

show any significant evolutionary distance in their version of *nah*Ac (Figure 4).

However, a few strain of *Burkholderria multivorans* shows similarities and some other

does not (Figure 4). The source of collection of these two *Burkholderria multivorans*

species may be the reason of such variation in the same species (Li et al., 2007). In a

study by (Su et al., 2016), the epigenetic impacts on the metabolic enzymes have been

suggested. Moreover, the divergence of evolutionary distances of the same enzyme

between different and same bacterial species indicates thrive in a particular

microenvironment. However, further studies are needed to identify underlying cause

and mechanisms for such variation.

3.4. Molecular docking

Molecular docking is a computational technique that samples small molecules as

ligands and macromolecule as receptor, and this virtual screening helps to understand

the theoretical best conformation of the docked molecules (Meng et al., 2011). The

result depicted from the rigid body molecular docking, suggests *nah*Ac-KD10$_{I250, V256}$

mutant has little higher binding free energy than that of mutant *nah*Ac encoded by KD9

but little lower than the *nah*Ac mutant variant encoded by the KD6 (Table 1).

Moreover, the rigid-flexible molecular docking using MOE algorithms showed *nah*Ac

I$_{250}$, G$_{256}$ mutant variant has ΔG_{London} of -7.1 kcal mol^{-1} and $\Delta G_{GBVI/WSA}$ of -1.68 kcal

mol^{-1} and that is little higher in other two mutant versions of *nah*Ac encoded by KD6

and KD9 (Table 1). The interacting amino acid residues in the active site pocket were

confined to be quite same in three mutant variants, except one variation, *i.e.,*His$_{208}$

which is located about 5.49 Å away from the bicyclic ring of NAP (Figure 3D). The

altered binding free energy of the *nah*Ac-KD10$_{I250, V256}$ may be because of the locations

of mutations in a stretch of amino acid "248-266" and His$_{208}$ as a unique interacting

residue of the active site (Figure 3C). Moreover, the occurrence of mutation in an amino

acid stretch "248-266" is common in KD6, KD9 and KD10 suggesting "248-266" as a

common mutation prone location. In addition, the non-covalent interactions of the

*nah*Ac-KD10$_{I250, V256}$ with naphthalene, phenanthrene and anthracene suggest that Phn

352 was common residue and His 208 is involve in π- staking with bond angle of

80.58° and 77.19° respectively for naphthalene and anthracene (Table S5).

3.5. Detection of solvent efflux system

The microenvironment and internal homeostasis of a cell is crucial for

maintaining its normal cellular functionality and cell tends to adapt several strategies in

order to achieve it (Blanco et al., 2016). One such adaptation strategy is the solvent

efflux pump system, that control intracellular toxicity to some extent (Kusumawardhani

et al., 2018). The role of *srp*ABC solvent efflux pump system in the biodegradation of

PAHs are common in literature (Bugg et al., 2000). Besides the other cellular activities,

*srp*ABC also assist in gaining antimicrobial resistance (Schweizer, 2003).

Biodegradation studies on chlorpyrifos indicate that the intermediates formed during

biodegradation, acts as antimicrobial agent to other species, meaning that the metabolic

intermediates may have role on inter-species competition in a particular micro-

environment to thrive in nutrient-limiting condition (Anwar et al., 2009), (Raes and

Bork, 2008). Besides, the role of *srp*ABC on biodegradation enhancement process is

still poorly understood. The presence of *srp*ABC in *Pseudomonas putida* strain KD10

(Figure S5) and its efficient naphthalene biodegradation property can be interlinked

(Dutta et al., 2018).

3.6. Cell immobilization and viability count

The viscosity of CABs determines its efficiency of the immobilization and its

performance of detoxification of environmental pollutant (Young et al., 2006). With this

intention, sodium alginates with medium viscosity were chosen for further studies. The

efficiency of cell immobilization in CABs was evaluated by enumeration of viable cells

(Table S3). Bacteria immobilized in CABs does not significantly lose its viable biomass

after 21 days of incubation at 4°C.

3.7. NAP biodegradation studies

3.7.1 By free Pseudomonas putida strain KD10

Biodegradation kinetics of NAP (500 mg L^{-1}) by the *Pseudomonas putida* strain

KD10 was first tested to determine its NAP biodegradation potential (Figure 4). It was

found that, after 12 days of incubation at 31°C the amount of residual NAP reduced

significantly ($p<0.05$) (Figure 5). Results are also summarised at Table 3 along with

names of the immobilized systems. The first order degradation kinetic data suggest that

the KD10 efficiently decompose 95.22 % NAP in CSM as sole source of carbon and

energy at the end of the incubation. The values of degradation rate constant (k) and half-

life ($t_{1/2}$) were found to be 0.2193 and 3.1 days with R^2 of 0.981. In previous study, 99.1

% of initial NAP was removed within 96 h by strain *Bacillus fusiformis* (BFN).

However, the initial concentration was very low (50 mg L^{-1} of initial NAP) (Lin et al.,

2010).

After confirmation of NAP biodegradation potential of KD10, the stain was

allowed to grow in association with other two strains of *Pseudomonas putida*, namely

Pseudomonas putida strain KD6 and *Pseudomonas putida* strain KD9 (Dutta et al.,

2017; Dutta et al., 2018). These two strains were selected to develop the blend of

Pseudomonas sp. consortium as because they were collected from different isolation

points and they encode different variant of *nah*Ac and they have common optimized

growth parameters (temperature 31°C and pH 7.1) (Dutta et al., 2017 & Dutta et al.,

2018). In addition, *Pseudomonas putida* strain KD6 encodes a six point mutant *nah*Ac

with capability to co-degrade high concentration of NAP, phenanthrene (PHN) and

pyrene (PYR), (500 mg L^{-1} each) and the stain KD9, encodes a four point mutant

variant of *nah*Ac with rhamnolipid production capabilities (Dutta et al. 2018).

In the previous study, $t_{1/2}$ of NAP co-biodegradation with phenanthrene and

pyrene by *Pseudomonas putida* strain KD6 was 4.1 days, which was significantly

reduced to 2.2 days when bacterial cells were allowed to grow as *Pseudomonas* sp.

consortium with only NAP as sole source of carbon and energy (Table 3). Further, this

value was found as 2.7 days for *Pseudomonas putida* KD9, suggesting *Pseudomonas*

putida strain KD6 might faces some sort of substrate inhibition by the co-presence of

other complex PAHs *i.e.* PHN and PYR in the system (Jiang et al., 2018) and the

cooperative nature of the *Pseudomonas* sp. consortium might help to enhance the NAP

bio-utilization. Microbial consortium works several ways, *viz.,* by division of labour,

cross-feeding *etc.* (Smid and Lacroix, 2013). In addition, a successful consortium could

also overcome shortcomings of single bacteria (Bhatia et al., 2018) and metabolically

engineered consortium serves improved functionality (Bernstein et al., 2012). However,

it is a fact that, bacteria select PAH among mixed PAHs based on their structural

simplicity first (Dutta et al., 2017) and the velocity could be optimized by reducing the

initial concentration of the PAHs gradually according to the structural simplicity (Jiang

et al., 2018).

3.7.2. By immobilised Pseudomonas sp. *consortium*

The biodegradation kinetic of NAP by individual *Pseudomonas putida* strain

KD6, KD9, KD10 and *Pseudomonas* sp. consortia immobilized in CABs, depicted

further enhancement of overall NAP bio-utilization (Table 3). Cell immobilization using

hydrogel, such as CAB found to be advantageous rather than free cell (Hameed and

Ismail, 2018). This phenomenon may be attributable to the increased level of tightness

of the cross-linked polymers of the calcium alginate beads that render bacteria adequate

amount of protection from harsh environment (Chen et al., 2013). However, the free

bacterial cell lacks the capabilities to degrade high initial concentration of the toxicant

as because they followed the conventional growth phases (Marrot et al., 2006). In

addition, exposure of free bacterial cells to the high initial concentration of toxicants

may challenge them to experience shock-concentration (Zhao et al., 2006). Conversely,

cells immobilized in calcium alginate beads can tolerate high concentration of the

toxicant and decrease the lag phase duration (Kao et al., 2014). Moreover, the diffusion

limitation natures of the CABs matrix provide a high local concentration of the cell

population (Bezbaruah et al., 2009). Besides this, CABs provides a remarkable stability

and reusable features that effectively reduces the production cost (Daâssi et al., 2014).

Biodegradation of NAP by immobilized *Pseudomonas* sp. consortium

significantly elevated the overall NAP bio-utilization efficiency with $t_{1/2}$ and R^2 values

of 2 days and 0.998, respectively suggesting experimental data well correlated to the

model (Table 3). The individual cell population also displayed the improved

biodegradation efficiency. However, the degradation efficiency was found maximum in

case of KD6, which showed a marked reduction of NAP half-life from 4.1 days to 3.0

days when grown in CSM with NAP as common complex nutrient (Table 3). It

promptly suggests that CABs facilitate KD6 optimized bio-utilization of NAP, which

might be masked up by the co-presence of NAP, PHN, PYR (Jiang et al., 2018).

However, further studies are required to investigate behavioural patterns of consortium

with mixed PAHs.

Bacterial cell grown on CSM with naphthalene as sole source of carbon and

energy showed a delayed growth rate and their total biomass was also low (Figure 1).

Nevertheless, growing cells on CSM prior to immobilization in CABs provide them

essential adaptation periods to the toxicant and mature them for such stress condition

that was found to effective for NAP biodegradation (Table 3). Immobilized cell system

was found to be useful for bioremediation after prior adaptation to the surrounding

environment (Partovinia and Rasekh, 2018). However, growing cells in Luria-Bertani

broth does not adapt cell adequately, and their NAP removal performance was very poor

(Figure S3). Herein, in this present set of study we used sodium alginate of medium

viscosity and the average size of the beads were found to be 0.5 mm (Figure. 2). The

inoculum was normalized (OD600nm = 0.002) for each *Pseudomonas putida* strain in

order to prepare the *Pseudomonas* sp. consortium. In aqueous medium the immobilized

bacterial cell mainly on the surface were exposed to naphthalene and primarily involve

in the naphthalene bio-utilization process. However, due to the micro porous feature of

the calcium alginate beads, microbial cells immobilized other than surface are also

participates in the bio-utilization and bio-sorption process of naphthalene. Further, with

time due the mechanical force generated by shaking, a few microbial cells may release

from the calcium alginate beads. However, to fuel up the cell, it is necessary to uptake

naphthalene for bio-utilization through step by step intra cellular enzymatic reactions

(Lin et al., 2014).

3.7.3. Effect of initial naphthalene concentration on biodegradation kinetics

The effect of different initial concentration of NAP (150 – 2500 mg L^{-1}) on

degradation kinetics was evaluated with immobilized KD6, KD9, and KD10 as pure and

as consortium ~~format~~ (Table S2). The results suggest cell immobilization in CABs

facilitates bacteria to cope with high initial concentration of NAP (Table S2). The

substrate inhibition kinetic parameters, *viz.,* the maximum specific degradation rate

(q_{max}), and inhibition constant (k_i) were found to be high in case of immobilized

Pseudomonas sp. consortium and these values were 0.707 h^{-1} and 1475 mg L^{-1} (Table

S2). However, we did not found any significant change on half saturation constant (k_s),

suggesting the reaction does not depend on its initial concentration and the

biodegradation process follows pseudo-first order reaction kinetics.

In previous study, free *Pseudomonas putida* KD9 in CSM was capable to

tolerate relatively low initial concentration of NAP with k_i value of 1107 mg L^{-1} and

addition of sucrose as SCS provides quite similar potential of NAP tolerance (k_i of 1429

mg L^{-1}) with that of immobilized *Pseudomonas* sp. consortium (Dutta et al., 2018).

However, 0.5 gm. % sucrose as SCS in the mobile open water system might not be

beneficial for bacteria to overcome the high shock concentration of the toxicant, again

suggesting cell immobilization and development of effective microbial consortium as

systematic optimization of biodegradation process.

3.8. Detection metabolic end product

The metabolic pathway of NAP biodegradation that might be followed by

Pseudomonas putida strain KD10 was elucidated through GCMS of the metabolites

(Figure S4). Major metabolites were confirmed by analysing the m/z values with

reference to the NIST library (Figure S4). In the previous study the major compounds

ware restricted to be salicylaldehyde, catechol, D-gluconic acid and pyruvic acid,

suggesting NAP biodegradation via catechol mediated pathway (Dutta et al., 2018).

However, in the present study we have detected salicylic acid as an additional

metabolite (Table 4), which also heads towards the catechol mediated NAP

biodegradation pathway (Heitkamp et al., 1987).

Previous studies on naphthalene biodegradation suggested that catechol step into the

TCA cycle by two possible pathway one is ortho-cleavage and another is meta-cleavage

pathway. Presence of catechol-2, 3-dioxygenase in *Pseudomonas putida* stain KD10

(Figure S5) suggest it follow meta-cleavage of catechol biodegradation pathway. In our

present set of study D-gluconic acid was found as a metabolite of naphthalene

biodegradation by *Pseudomonas putida* strain KD10. We postulate that D-gluconic acid

enters in the bio-conversion of naphthalene possible from glucose as precursor (Figure

6) which finally converted to pyruvate via aldolase. D-gluconic acid have ability to

induce phosphate solubilisation processes (Rodriguez et al., 2004), and seemed to

inhibit fungal growth (Kaur et al., 2006). Hence, suggesting *Pseudomonas putida* KD10

have growth promoting ability (Figure 6). The metabolic intermediates such as salicylic

acid, catechol was reported to have plant-growth promoting activity (Lee et al., 2010).

Further, the antioxidant property of the catechol help and promotes seed germination

(Schweigert et al., 2001). The presence of d-gluconic acid and its subsequent entry in

TCA cycle from glucose as precursor, suggest that bacteria may able to sequestered

catechol upon needs.

4. Conclusion

NAP biodegradation by free and immobilized *Pseudomonas putida* strain KD10

and *Pseudomonas* sp. consortium were studied. Biodegradation of NAP was enhanced

by immobilised *Pseudomonas* sp. *consortium* and 80.1% of initial NAP (500 mg L^{-1})

was utilised after 72 h of incubation. Further, initial NAP tolerance by immobilized

Pseudomonas sp. consortium was found to be highest (1475 mg L^{-1}) with maximum

specific degradation of 0.707 h^{-1}. The sequence analysis of the naphthalene 1, 2-

dioxygenase suggests a significant evolutionary distance among different microbial

species and in few cases, intra-species variation was observed. Moreover, a common

mutation prone amino acid stretch in all three natural *nah*Ac mutants (strain KD6, KD9

and KD10) were found at close proximity of the active site and the rigid-flexible

molecular docking showed better binging free energy than that of wild-type variant of

*nah*Ac. This common mutation prone amino acid stretch could aid the rational

approaches of enzyme redesigning. Overall, this study summarises application of

bacterial cell immobilization in calcium alginate beads and development of the

microbial consortium together for enhanced NAP biodegradation. However, further

studies are required for the systematic optimization of PAHs biodegradation process.

Authors Contributions

KD and CG conceive the main hypothesis. KD design, performed all

experiments, and wrote the manuscript. SS, IK, SB, DJ assisted KD in some

experiments. MK, TM, PR, KCG performed the statistical analysis and wrote the

manuscript. CG critically proofread and wrote the manuscript. All authors read the

manuscript.

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Conflict of Interest

The authors declare that they have no conflict of interests.

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Figure legends

Figure 1. Growth kinetic model of *Pseudomonas putida* and *Pseudomonas* sp.
consortium during naphthalene biodegradation. **A.** Gompertz's growth kinetic
model fit of the biodegradation of naphthalene by *Pseudomonas putida* strain
KD10 and by the *Pseudomonas* sp. consortium. **B.** Effect of 0.5 gm. % sucrose
supplementation on growth pattern of *Pseudomonas putida* strain KD10 and
Pseudomonas sp. consortium.

Figure 2. Cell immobilization in calcium alginate beads (CABs) with different
cell morphology of the *Pseudomonas putida* strain KD10.
830

Figure 3. Sequence and structure analysis of the naphthalene 1, 2-dioxygenase
(*nah*Ac) encoded by *Pseudomonas putida*. **A**. The amino acid sequence of wild
type and three mutant variants of the naphthalene 1, 2-dioxygenase (*nah*Ac).
Mutated residues are highlighted in blue and the common mutation prone amino
acid stretch is highlighted with red box. **B.** Structural mismatches among mutant
variants and wild type naphthalene 1, 2-dioxygenase. Each mutated residues and
its subsequent neighbour residues of all the mutant variant of naphthalene 1, 2-
dioxygenase is overlaid with cartoon representation. The highlighted red circle
indicates local mismatch among the mutant variant of naphthalene 1, 2-
dioxygenase **C.** Different ligand binding posture of the naphthalene 1, 2-
dioxygenase encoded by *Pseudomonas putida* strain KD10. Naphthalene (red),
phenanthrene (cyan) and anthracene (green) binding postures with major
interacting amino acid residues (left). Two-dimensional presentation of the
interacting residues of *nah*Ac$_{I250, V256}$ (right). **D.** Docking pose of naphthalene 1,
2-dioxygenase (Chain A) encoded by *Pseudomonas putida* strain KD10. The
four major naphthalene interacting residues and their molecular distance are
labelled in black.

848
849   **Figure 4.** Distance matrix and cladogram of the naphthalene 1, 2-dioxygenase.
850   The evolutionary distance of the naphthalene 1, 2-dioxygenase I_{250}, V_{255} variant
851   encoded by *Pseudomonas putida* strain KD10 other variants by different
852   microbial species is scaled by colour index (upper-left side).
853

855 **Figure 5.** Naphthalene degradation kinetics of *Pseudomonas putida* strain KD10
856 and *Pseudomonas* sp. consortium free and cell immobilized in calcium alginate
857 beads.
858

859 **Figure 6.** Proposed pathway of naphthalene biodegradation by *Pseudomonas*
860 *putida* strain KD10. Naphthalene (I), salicylic acid (II), salicylaldehyde (III),
861 catechol (IV), pyruvate (V), glucose (VI), d-gluconic acid (VII). The grey arrow
862 indicates alternative pathway via d-gluconic acid.

863
864

865 *Table captions*

866 **Table 1.** Summary of rigid-flexible molecular docking for the analysed mutant
867 naphthalene 1, 2-dioxygenase encoded by different *Pseudomonas putida*
868 strains.
869
870 **Table 2.** Apparent enzyme kinetic parameters of naphthalene 1, 2-dioxygenase
871 $_{I250, V256}$ encoded by *Pseudomonas putida* strain KD10.
872
873 **Table 3.** Parameters of first-order biodegradation kinetics of naphthalene by
874 different free and immobilized *Pseudomonas putida* strains.
875
876 **Table 4.** Major metabolite detected by GCMS analysis during naphthalene
877 biodegradation by *Pseudomonas putida* KD10.
878

Supplementary section

Table S1. Nucleotide sequences used as primer in polymerase chain reactions

Table S2. Parameters of substrate inhibition kinetic model of naphthalene
biodegradation by different *Pseudomonas putida* strains immobilized in calcium
alginate beads.
Table S3. Enumeration of viable cell in calcium alginate beads.

Table S4. Gompertz's growth curve model fit of *Pseudomonas putida* strain
KD10 and *Pseudomonas* sp. consortium.

Table S5. Major amino acid residues of mutant variant of naphthalene 1, 2-
dioxygenase $_{I250, V256}$ involve in hydrophobic interaction with different ligands.

Figure S1. Colony morphology and catechol confirmatory test. **A.** Colony
morphology of *Pseudomonas putida* strain KD10 after 24 h of incubation. **B.**
Colony morphology of *Pseudomonas putida* strain KD10 after one week of
incubation.
Figure S2. Molecular phylogenetic analysis of *Pseudomonas putida* strain
KD10.

Figure S3. Effect of cell adaptation on NAP biodegradation.

Figure S4. Mass spectra of the major metabolites detected during
biodegradation of naphthalene in CSM as sole source of carbon and energy by
Pseudomonas putida KD10.

Figure S5. Agarose gel electrophoresis of PCR products of catechol 2, 3-
dioxygenase (*nah*H) gene and solvent efflux pump (*srp*ABC) system.
A. Catechol 2, 3-dioxygenase (*nah*H). Lane 1 and 2 *nah*H pcr product of
Pseudomonas putida strain KD9 and *Pseudomonas putida* strain KD10. **B.**
Solvent efflux pump (*srp*ABC) system. Lane 1: *Pseudomonas aeruginoasa*
ATCC 9027, Lane 2: *Pseudomonas putida* strain KD10. Lane M: 100bp DNA
ladder (HiMedia, India), Lane 3: Negative control.

Figure S6. Nucleotide sequence chromatogram of the mutant naphthalene 1, 2-
dioxygenase (*nah*Ac), (chain-A) encoded by *Pseudomonas putida* strain KD10.
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Table 1. Summary of rigid-flexible molecular docking of naphthalene and different mutant version of naphthalene 1, 2-dioxygenase encoded by different *Pseudomonas putida* strains

ΔG[*]			LB[4]	IC[5]	IME[6]	VDW[7]	RMSD[c]	RMSD[r]	References
BE[1]	London[2]	GBVI/WSA[3]							
-5.92[a]	-7.18	-4.61	-0.59	46.14	-5.92	-5.91	0.00	77.29	Dutta et al., 2017
-5.41[b]	-7.13	-4.02	-0.54	109.07	-5.41	-5.4	0.00	96.32	Dutta et al., 2019
-5.83[c]	-7.10	-1.68	-0.58	53.13	-5.83	-5.84	0.00	96.08	This study

[*]PubChem ligand CID 931 = Naphthalene, BE[1] = Binding free energy, [2]London free energy, [3]Generalized born volume integral/weighted surface areas, LB[4] = ligand binding, IC[5] = inhibition constant, IME[6] = Intermolecular energy, VDW[7] = Vdw hb desolv energy, RMSD[c] = Cluster RMSD, RMSD[r] = Reference RMSD. Energy unit = kcal mol^{-1}. Inhibition constant unit = µM.at 298.15 K. Naphthalene 1, 2-dioxygenase encoded by (a) *Pseudomonas putida* strain KD6, (b) *Pseudomonas putida* strain KD9 and (c) *Pseudomonas putida* strain KD10.

Table 2. Apparent enzyme kinetic parameters of naphthalene 1, 2-dioxygenase $_{I250,\ V256}$ encoded by *Pseudomonas putida* strain KD10

Substrate	Kinetic parameters*				
	Km (µmol mL^{-1})	Vmax (µmol min^{-1})	Kcat (s^{-1})	Kcat/Km (mL^{-1} mol^{-1} s^{-1})	R^{2**}
Naphthalene	0.867	2.474	123.72	0.142×10^3	0.938

*The kinetic constant were determined at 30°C and pH 7.5 using 0.5-10µM ml^{-1} substrate concentration (each) by Lineweaver–Burk plot.
** Non-liner regression between initial substrate concentration 1/[S] and degradation rate constant 1/V yielded regression equation and regression coefficient (R^2)

Table 3. Parameters of first-order biodegradation kinetics of naphthalene by different free and immobilized *Pseudomonas putida* strains

| CSM | Cells immobilized in CABs | | | free cells | | | |
	k (Days^{-1})	t ½ (Days^{-1})	R^2	k (Days^{-1})	t ½ (Days^{-1})	R^2	References
A	0.228±0.005	3.0±0.077	0.99	0.167±0.012	4.1±0.307	0.98	Dutta et al., 2017
B	0.312±0.014	2.2±0.102	0.93	0.252±0.019	2.7±0.193	0.99	Dutta et al., 2018
C	0.284±0.017	2.4±0.143	0.97	0.219±0.009	3.1±0.137	0.92	This study
D	0.341±0.294	2.0±0.186	0.99	0.309±0.020	2.2±0.141	0.98	This study

[#]Initial naphthalene concentration = 500 mg L^{-1}
A. *Pseudomonas putida* strain KD6, B. *Pseudomonas putida* strain KD9, C. *Pseudomonas putida* strain KD10. D. *Pseudomonas* sp. consortium.
Each figure of the table represents the mean of three replicates.

Table 4. Major metabolites detected by GCMS analysis during naphthalene biodegradation by *Pseudomonas putida* KD10

Metabolite	R_t (min)	Major ion peaks (m/z)	Suggested structure
I	9.41	64.19, 92.17, 120.21, 138.22, 169.14, 191.18, 211.10	Salicylic acid
II	15.27	65.17, 74.41, 104.73, 121.10, 122.57, 142.21, 174.80, 201.22, 249.31	Salicylaldehyde
III	18.71	64.29, 81.35, 91.61, 110.47, 131.47, 149.71	Catechol
IV	31.89	61.21, 73.44, 76.24, 104.18, 117.41, 133.27, 177.23	D-gluconic acid
V	10.47	43.21, 61.27, 71.37	Pyruvic acid

R_t = Retention time

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Figure 1.

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1028   **Figure 2.**

Figure 3.

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1033 **Figure 4.**

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1053 **Figure 5.**

 Figure 6.