

1 Enhanced biodegradation of naphthalene by *Pseudomonas* sp. consortium
2 immobilized in calcium alginate beads

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35

36 *Abstract*

37 Polycyclic aromatic hydrocarbons (PAHs) belong to a large group of organic
38 pollutant which considers as a potential health hazard to living beings. Herein,
39 naphthalene biodegradation potential by free and immobilized *Pseudomonas putida*
40 strain KD10 and *Pseudomonas* sp. consortium were studied. Additionally, naphthalene
41 1, 2-dioxygenase (*nahAc*) was sequenced and analyzed, which reveals two altered
42 amino acid residues. However, the altered amino acid residues are not present in the
43 vicinity of the active site. The gas-phase binding free energy (ΔG_{London}) of the mutant
44 variant of naphthalene 1, 2-dioxygenase was $-7.10 \text{ kcal mol}^{-1}$ which closely resembles
45 the wild type variant. Naphthalene biodegradation rate by *Pseudomonas putida* strain
46 KD10 was $79.12 \text{ mg L}^{-1} \text{ day}^{-1}$ and it was significantly elevated up to $123 \text{ mg L}^{-1} \text{ day}^{-1}$
47 by the immobilized *Pseudomonas* sp. consortium. The half-life ($t_{1/2}$) for naphthalene
48 biodegradation was 3.1 days with the inhibition constant (k_i), substrate saturation
49 constant (k_s) and maximum specific degradation rate constant (q_{max}) of 1268 mg L^{-1} ,
50 395.5 mg L^{-1} and 0.65 h^{-1} , respectively, for the *Pseudomonas putida* strain KD10.
51 However, the $t_{1/2}$ value was significantly reduced to 2 days along with k_i , k_s and q_{max}
52 values of 1475 mg L^{-1} , 298.8 mg L^{-1} and 0.71 h^{-1} , respectively, by the immobilized
53 *Pseudomonas* sp. consortium. The GC-MS data suggest that KD10 might follow D-
54 gluconic acid mediated meta-cleavage pathway of catechol biodegradation. It is
55 concluded that naphthalene biodegradation performance by immobilized *Pseudomonas*
56 sp. consortium was superior to free or immobilized *Pseudomonas putida* KD10.
57 Microbial consortium immobilization could be a useful tool for water quality
58 management and environmental remediation.

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60 *Keywords:*

61 *Pseudomonas* sp., petroleum wastes, biodegradation, cell immobilization, mutant

62 naphthalene 1, 2-dioxygenase, rigid-flexible molecular docking.

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82 *Highlights*

- 83 • Superior naphthalene biodegradation by *Pseudomonas* sp. consortium
84 immobilized in calcium alginate beads.
- 85 • A common mutation prone amino acid stretch inside chain A of naphthalene 1,
86 2-dioxygenase has been identified.
- 87 • A new naphthalene biodegradation pathway by *Pseudomonas putida* strain
88 KD10 has been proposed.

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104 *1. Introduction*

105 Polycyclic aromatic hydrocarbons (PAHs) are considered as the potential health
106 hazard for living beings (Kumari et al., 2018). Environmental agencies including US-
107 EPA, European Union, Environment-Canada, registered PAH as the priority pollutants
108 that require immediate human intervention (Wang et al., 2018). The physicochemical
109 properties of PAH make them a major contributor for soil and groundwater
110 contamination through bio-magnification (Norris, 2017). Moreover, according to the
111 Environmental Health Hazard Assessment, U.S.A., naphthalene (NAP) concentration
112 beyond 170 ppb is not safe drinking (Bruce et al., 1998). Different orthodox and
113 expensive techniques for environmental remediation, *viz.*, incineration, gasification,
114 plasma-gasification have been replaced by green-technologies such as bioremediation,
115 phytoremediation, nanoremediation, *etc.* (Thomé et al., 2018). Bioremediation is
116 considered as the most cost-effective and eco-friendly oil spill management technique
117 (Wilson and Jones, 1993). However, while considering the vast volume of a mobile
118 open water system, bioremediation stumble upon several limiting factors such as the
119 low local concentration of the effective microorganisms, loss of active microorganisms,
120 *etc.* (Chen et al., 2017). Conversely, cell immobilization provides several advantages
121 such as, it helps to retain high local concentration of the effective microorganisms, keep
122 intact bacterial cell membrane stability and can be stored for future reuse (Bhardwaj et
123 al., 2000). Overall, cell immobilization can markedly improve the stability and
124 efficiency of the bioremediation process (Mrozik and Piotrowska-Seget, 2010; Tyagi et
125 al., 2011). Cell immobilization using calcium alginate beads (CABs) is a convenient
126 option where maximum cells remain viable and can tolerate high concentration of the
127 toxicant (Lee and Heo, 2000). Moreover, calcium alginate is nontoxic to the bacterial

128 cell and it has low production cost which facilitates easy reuse (Bhardwaj et al., 2000).
129 Biodegradation of a toxicant (complex nutrient for bacteria) by the microbial
130 consortium can efficiently enhance biodegradation rate (Kumari et al., 2018). This rate
131 enhancement could be achieved by several ways such as, different biodegradation
132 pathways of each individual bacterium (Dutta et al., 2018), a metabolic intermediates of
133 one bacteria may act as the starting material of other bacteria (Surkatti and El-Naas,
134 2018), different genetic makeup (Woyke et al., 2006), synergetic effects of different
135 microbial species, (Ghazali et al., 2004) or by synthesising different variant of catalytic
136 enzymes. Bacterial cells in the microbial consortium can also combine their metabolic
137 capabilities to utilize the common complex nutrient (Gilbert et al., 2003).
138 Biodegradation of complex hydrocarbon mixture by the microbial consortium offers a
139 combination of diverse enzymes which promotes the biodegradation processes
140 (Wongwilaiwalin et al., 2010). However, microbial consortium immobilization for
141 biodegradation of PAHs was not studied previously. The aim of the present study is to
142 evaluate the naphthalene biodegradation performance by *Pseudomonas putida* strain
143 KD10 and *Pseudomonas* sp. consortium as a free and immobilized format. Previous
144 study showed that *Pseudomonas putida* strain KD9 cells decrease in size and shape
145 from rod to sphere during naphthalene biodegradation and their specific growth rate was
146 also little slower (Dutta et al., 2018). Keeping this fact in the mind, both morphological
147 types of bacteria were applied in this present set of study. Additionally, the naphthalene
148 1, 2-dioxygenase (*nahAc*) was sequenced and analyzed.

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151 *2. Materials and methods*

152 *2.1. Chemicals*

153 Naphthalene was purchased from Sigma-Aldrich chemicals Pvt. Ltd. (USA) and
154 all other chemicals used for media preparation were procured from HiMedia Laboratory
155 (Mumbai, India). GC-MS and HPLC grade solvents were procured from Fisher
156 Scientific (Mumbai, India). Sodium alginate (CAS No. 9005-38-3) of medium viscosity
157 was purchased from Merck Pvt. Ltd. (USA).

158

159 *2.2. Microorganisms, growth media, growth condition and consortium preparation*

160 Soil samples were collected from petroleum refinery waste sites near Indian Oil
161 (Haldia, West Bengal, India). The enrichment isolation and strain identification were
162 carried out according to the standard protocol described previously (Dutta et al., 2017).
163 Carbon deficient minimum medium (CSM), with a pH of 7.1, was used to cultivate the
164 bacteria. Naphthalene was used as a sole source of carbon and energy in CSM with
165 following compositions: 0.2 g L⁻¹ MgSO₄, 7H₂O; 0.08 g L⁻¹ Ca (NO₃)₂, 4H₂O; 0.005 g
166 L⁻¹ FeSO₄, 7H₂O; 4.8 g L⁻¹, K₂HPO₄; 1.2 g L⁻¹ KH₂PO₄. *Pseudomonas putida* strain
167 KD6 (KX786159.1) and *Pseudomonas putida* strain KD9 (KX786158.1) and the newly
168 isolated *Pseudomonas putida* strain KD10 (KX786157.1) were used to prepare the
169 blend of *Pseudomonas* sp. consortium. The consortium was maintained in Luria-
170 Bertani broth at 31°C with 150 rpm in order to grow the bacterial cell in its normal size
171 and shape. Alternatively, bacterial cells were grown in CSM with naphthalene as sole
172 source of carbon and energy to obtain altered morphological variant.

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174 *2.3. Gene sequence analysis*

175 *2.3.1. Detection of naphthalene 1, 2-dioxygenase and catechol 2, 3-dioxygenase*

176 The conventional polymerase chain reaction (PCR) for naphthalene 1, 2-
177 dioxygenase (*nahAc*) and catechol 2, 3-dioxygenase (*nahH*) using specific primers
178 (Table S1) were performed. Additionally, PCR product of *nahAc* was sequenced and
179 analyzed according to the methods described previously (Dutta et al., 2017).

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181 *2.3.2. Clustering and phylogenetic analysis*

182 The evolutionary distance of naphthalene 1, 2-dioxygenase among different
183 bacterial species was analyzed using BLOSUM weighted matrix followed by pairwise
184 distance computation using MEGA (v7.0) (Kumar et al., 2016). The distance matrix
185 was then clustered using R (Team, 2013) to create the cladogram and heatmap. The
186 phylogenetic position of the isolated *Pseudomonas putida* strain KD10 was analyzed
187 using the previous method (Dutta et al., 2017).

188

189 *2.3.3. Molecular docking*

190 The rigid-body molecular docking was conducted using Auto Dock (v4.2.1)
191 (Morris et al., 1998). Briefly, the center grid dimensions were set to
192 20.271×61.989×87.168 with a grid spacing of 0.375 Å. The virtual screening was

193 repeated for 10 times with the unaltered docking parameters having 2.0 Å cluster
194 tolerance. Additionally, the rigid-flexible molecular docking was performed using
195 Molecular Operation Environment (Chemical Computing Group, Montreal Inc.,
196 Canada). The latter scoring function was employed to identify the most favourable
197 docked poses and to estimate the binding affinity of the protein-ligand complexes. The
198 non-covalent interactions were analyzed using the previous method (Salentin, S., et al.,
199 2015).

200

201 *2.4. Enzyme kinetic assay*

202 The enzyme kinetic parameters of the naphthalene 1, 2-dioxygenase_{I250, V256} was
203 performed using the cell-free extract of the *Pseudomonas putida* strain KD10, grown in
204 250 mL of CSM with naphthalene (500 mg L⁻¹) as a sole source of carbon and energy as
205 described by previously (Dutta et al., 2017).

206

207 *2.4. Detection of solvent efflux pumps system*

208 The detection of solvent efflux pump system (*srpABC*) in *Pseudomonas putida*
209 strain KD10 was performed using conventional polymerase chain reaction in a thermal
210 cycler (Mastercycler® nexus gradient, Eppendorf, (Germany). Standard reaction
211 mixtures were prepared using forward, reverse primers (Table S1) as described
212 previously (Dutta et al., 2018).

213

214 *2.5. Cell immobilization*

215 *2.5.1. Cell immobilization in calcium alginate beads*

216 *Pseudomonas putida* strain KD10 and *Pseudomonas* sp. consortium were
217 immobilized according to the standard protocol described previously (Daâssi et al.,
218 2014). Briefly as sodium alginate was dissolved in 0.9 wt. % NaCl (1 gm in 40 mL 0.9
219 wt. % NaCl) for 24 h and sterilized by autoclaving (121°C for 15 min). Two grams of
220 bacterial cell mass was added to 8 ml of NaCl solution and again added to the sterile
221 alginate solution. The mixture was then gently vortex for complete homogenization and
222 extruded dropwise through a hypodermic syringe into chilled sterile CaCl₂ solution
223 (Figure 2). The beads were hardened in the same solution at room temperature with
224 gentle stirring for 1 h. Finally, the beads were washed several times with 0.9 wt. %
225 NaCl to remove excess calcium ions and free cells. The beads have an average diameter
226 of 0.5 mm and stored at 4°C. Sterile beads (without microorganisms) were used to
227 monitor the abiotic loss of naphthalene. Sodium alginate of medium viscosity
228 ($\geq 2,000$ cP) was used to prepare the calcium alginate beads (CABs).

229

230 *2.5.2. Cell viability count*

231 The viable cell enumeration in the CABs was performed by using a protocol
232 described previously (Usha et al., 2010). In brief, CABs was washed in saline and keep
233 submerged for 10 min (for saline soaking). Following soaking, the CABs were shaken
234 with glass beads for 15 min and 1 g CABs were homogenized in 9 ml saline. The saline
235 was then used for viable cell enumeration onto a nutrient agar plate. Plating was done

236 with this treated saline by series dilution method up to 10^{-5} dilutions on nutrient agar
237 plates. Plating was also done for the initial CFU count. For every dilution, 10 μ l of the
238 solution was plated. Plating was done by the pour plate method. Plates were incubated
239 at 37°C for 24 h. (Jain and Pradeep, 2005).

240

241 *2.6. Scanning electron microscopic (SEM) study*

242 Samples for SEM analysis were prepared by a protocol described elsewhere
243 (Dutta et al., 2018). The dried samples were coated in a sputter coater (Quorum-
244 SC7620) under vacuum with a thin gold layer right before SEM analysis using a
245 scanning electron microscope (Zeiss, EVO 18, Germany) with an accelerating voltage
246 of 5 kV.

247

248 *2.7. Biodegradation kinetics*

249 *2.7.1. Biodegradation of naphthalene in the liquid medium*

250 Naphthalene biodegradation study was conducted by the protocol described
251 previously (Dutta et al., 2017). The conical flasks were incubated at 31°C with 150 rpm
252 and uninoculated conical flask were used as control. Culture medium was collected at
253 regular interval of 72 h for degradation and growth kinetic study. Additionally, the
254 effect of initial concentration of naphthalene (150-2500 mg L⁻¹) was studied with
255 different immobilized systems.

256

257 *2.7.2. Chemical analysis*

258

259 Naphthalene biodegradation was analyzed by using the 1260 infinity series HPLC
260 system (Agilent, Santa Clara, CA, USA) equipped with Zorbax SB-C18 reversed-phase
261 column (4.6 × 12.5 mm, 5 μm). The analysis of naphthalene concentrations was
262 conducted using isocratic elution conditions with the mobile phase 80:20 (v/v)
263 methanol: water at a flow rate of 1 ml min⁻¹. The detection was performed at 254 nm
264 according to the protocol described in the literature (Dutta et al., 2017). Additionally,
265 the metabolic intermediates of naphthalene biodegradation were analyzed using the GC-
266 MS system (GC Trace GC Ultra, MS-Polarisq, Thermo Scientific India Pvt. Ltd)
267 equipped with a capillary column (TR-WaxMS, 30 m × 0.25 mm [ID] × 0.25 μm film
268 thickness) by the protocol described elsewhere (Dutta et al., 2017). The entire analysis
269 was performed in electron ionization, at full scan mode. The metabolite identification
270 was based on the mass spectra comparison using the NIST Mass Spectral library (v2.0,
271 2008).

272

273 *2.8. Data analysis*

274 The first-order degradation kinetics model was used to estimate the residual
275 naphthalene in CSM using equations 1, the algorithms as expressed determine the half-
276 life ($t_{1/2}$) values of naphthalene in CSM. The substrate inhibition kinetic parameters
277 were calculated using equation 2.

$$C_t = C_0 \times e^{-kt} \quad [1]$$

278

$$q = q_{max} \frac{S}{K_s + S + S^2/K_i} \quad [2]$$

279

280 The effect of SCS on growth pattern was measured by calculating the difference
281 between optical densities at 600_{nm} and expressed by slight extension (Equation 4) of the
282 Gomperz's sigmoid growth fit equation (Equation 3).

283

$$y = ae^{-\exp(-k(x-x_c))} \quad [3]$$

284

$$\Delta X_c = X_{c2} - X_{c1} \quad [4]$$

285

286 The growth pattern change by SCS was considered as CFU shift (Equation 5). Where,
287 X_{c2} = Optical density at X_{c2} and X_{c1} = Optical density at X_{c1} .

$$CFU_{shift} = \Delta X_c \quad [5]$$

288

289 The bacterial growth kinetics were analyzed by applying the Gomperz's model using
290 Levenberg-Marquardt algorithm in Origin[®] 2016 (California, USA) and the degradation
291 kinetics were analyzed by using Graphpad Prism[®] 6.01 (San Diego, CA, USA).

292

293 *3. Results and discussion*

294 *3.1. Isolation, identification and growth patterns of NAP degrading bacterial strain*

295 *Pseudomonas putida* strain KD10 was isolated from petroleum refinery waste
296 with its distinguished colony morphology on CSM agar plate (Figure S1a, b). The PCR
297 product of 16S rRNA gene of the strain KD10 was sequenced. Moreover, the multiple
298 sequence alignment followed by phylogenetic assessment suggests that the strain
299 belongs to the *Pseudomonas putida* group and it showed 99% sequence similarity with
300 the previously deposited sequences of *Pseudomonas putida* (Figure S2). The 16S rRNA
301 gene sequence was deposited at NCBI (<https://www.ncbi.nlm.nih.gov/>) under the
302 GenBank accession number KX786157.1.

303 Naphthalene biodegradation by the *Pseudomonas putida* KD10 was
304 preliminarily confirmed by the catechol test (Figure S1C) followed by growth pattern
305 analysis in CSM with naphthalene as a sole source of carbon and energy. Additionally,
306 growth pattern with 0.5 gm % sucrose as a secondary carbon supplement (SCS) was
307 compared with growth pattern of non-supplemented CSM. Gompertz's model fit
308 (Equation 3) was used to analyze the growth curve (Table S4).

309

310 *3.2. Pseudomonas sp. consortium, growth pattern and NAP biodegradation*

311 The growth pattern of the *Pseudomonas sp.* consortium, as shown in Figure 1
312 was optimized at 31°C and pH 7.1 which reveal successful cooperation among
313 *Pseudomonas putida* strain KD6, KD9 and KD10. Further, the growth curve model
314 indicates a significant increase in total biomass of the *Pseudomonas sp.* consortium and
315 they also showed a high concentration of naphthalene tolerance potential.

316

317 *3.3. Gene sequence and phylogenetic analysis*

318 Naphthalene 1, 2-dioxygenase (*nahAc*), encoded by *Pseudomonas putida* strain
319 KD10 have only two point mutations at I250, and V256 which were replaced by
320 methionine and glycine respectively (Figure 3A, S6). However, additional mutations
321 were found in the same chain A at K200, A210, E264, M284 and N334 by replacing
322 glutamic acid, glycine, aspartic acid, isoleucine, respectively (Dutta et al. 2017, 2018).
323 The three dimensional structural analysis suggests that the mutations at 200, 210, 284
324 and 334 causes a little structural change in comparison to the wild type variant of
325 naphthalene 1, 2-dioxygenase (Figure 3B). Conversely, the amino acids stretch at the
326 close proximity of the active site residues exactly from 248 to 266 showed a structural
327 mismatch among all three mutant and wild type variants of naphthalene 1, 2-
328 dioxygenase (Figure 3B). The active site of an enzyme tends to evolve fast to attain its
329 maximum performance and functionality in a particular environment (James and
330 Tawfik, 2003).

331 The alteration of amino acids bases from 248 to 266 of chain A, was a common
332 feature in all three mutant variants of *nahAc* and this particular stretch of amino acid is
333 very close to the active site residues (Figure 3C). This implies that the “248-266” amino
334 acid stretch is highly prone to mutation and it may influence on the enzymatic efficiency
335 and environmental adaptability. Thus, our study provides a new insight, which could be
336 beneficial for rational approaches of enzyme redesigning.

337 The biodegradation performance of a bacterial strain is linked with the three
338 dimensional structure of the catalytic enzyme and ligand binding posture (Singh et al.,
339 2019). Moreover, alteration of a single amino acid in the catalytic domain of the enzyme
340 caused different ligand binding postures which eventually lead to a unique

341 biodegradation pathways (Ferraro et al., 2006). In addition, site directed mutagenesis in
342 the catalytic domain offers superior enzyme activity (Parales et al., 1999; Parales,
343 2003). The comparative analysis of binding free energy helps to comprehend the
344 superior naphthalene biodegradation performance by the *Pseudomonas* sp. consortium,
345 which is, in fact, the summative activity of all three mutant variants of naphthalene 1, 2-
346 dioxygenase.

347 The evolutionary trace on the naphthalene 1, 2-dioxygenase among different
348 species were studied through pairwise distance matrix analysis (Figure 4), which
349 suggests a significant intra and inter species difference among *Sphingopyxis* sp.,
350 *Croceicoccus naphthovorans*, *Burkholderia multivorans*, *Burkholderia* sp. *Massilis* sp.
351 and *Cycloclasticus* sp. Conversely, other *Pseudomonas* sp. particularly, *Pseudomonas*
352 *benzenivorans*, *Pseudomonas balearica*, *Pseudomonas stutzeri*, *Pseudomonas*
353 *kunmingensis*, *Pseudomonas frederikbergensis* have similarities in their version of
354 naphthalene 1, 2-dioxygenase. A few other species such as *Xenophilusa zovorans*,
355 *Ralsoniam annitolilytica*, and *Paraburkholderia aromaticivorans* does not show
356 significant evolutionary distance in their version of naphthalene 1, 2-dioxygenase
357 (Figure 4). However, a few strains of *Burkholderia multivorans* showed similarities
358 and some other does not (Figure 4). The source of collection of these two *Burkholderia*
359 *multivorans* species may be the reason for such variation (Li et al., 2007). In a study by
360 (Su et al., 2016), the epigenetic impacts on the metabolic enzymes have been suggested.
361 Moreover, the divergence of evolutionary distances of the same enzyme between
362 different and same bacterial species indicates thrive in a particular microenvironment.
363 However, further studies are needed to identify the underlying cause and mechanisms of
364 such variation intra-species variation.

365

366 3.4. Molecular docking

367 The result depicted from the rigid body molecular docking, suggests *nahAc*-
368 KD10_{I250, V256} mutant has a little higher binding free energy than that of mutant *nahAc*
369 encoded by KD9 but little lower than the *nahAc* mutant variant encoded by the KD6
370 (Table 1). However, the rigid-flexible molecular docking using MOE algorithms
371 showed *nahAc* I₂₅₀, G₂₅₆ mutant variant has ΔG_{London} of $-7.1 \text{ kcal mol}^{-1}$ and $\Delta G_{\text{GBVI/WSA}}$
372 of $-1.68 \text{ kcal mol}^{-1}$, which is little higher than other two mutant variants of *nahAc*
373 encoded by KD6 and KD9 (Table 1). The interacting amino acid residues in the active
374 site were confined to be quite same in three mutant variants, except one variation, *i.e.*,
375 His₂₀₈ which is located about 5.49 Å away from the bicyclic ring of naphthalene (Figure
376 3D). The altered binding free energy of the *nahAc*-KD10_{I250, V256} may be due to the fact
377 that the two altered amino acid residues reside in “248-266” and participation of His₂₀₈
378 as a unique interacting residue (Figure 3C). Furthermore, the non-covalent interactions
379 of the *nahAc*-KD10_{I250, V256} with naphthalene, phenanthrene and anthracene suggest that
380 Phn 352 was common residue and His 208 is involve in π - stacking with a bond angle of
381 80.58° and 77.19° respectively for naphthalene and anthracene (Table S5).

382

383 3.5. Enzyme kinetic assay

384 Enzyme activity by cell free extract of *Pseudomonas putida* KD10 was studied
385 using Naphthalene as substrate. The cell free extract of strain KD10 degraded
386 Naphthalene, with R^2 of 0.935 indicating the experimental data are well correlated with
387 the model (Table 2). The performance constant ($K_{\text{cat}}/K_{\text{m}}$) was $0.142 \times 10^3 \text{ ml}^{-1} \text{ mol}^{-1} \text{ s}^{-1}$
388 which further validate the chemical data obtained from chemical analysis (Table 1). In

389 previous study, the performance constant of *Pseudomonas putida* KD6 was little higher
390 than *Pseudomonas putida* strain KD10, suggesting the six altered amino acid residues of
391 the mutant naphthalene 1,2-dioxygenase of the KD6 may be main cause of this
392 difference (Dutta et al. 2017). The performance constant (K_{cat}/K_m) is the indicator of
393 the performance of an enzyme and usually the higher K_{cat}/K_m meaning better
394 enzymatic performance (Koshland Jr DE, 2002).

395

396 3.6. Detection of solvent efflux system

397 The cellular microenvironment and internal homeostasis are crucial for
398 maintaining normal cellular functions and cell tends to adopt several strategies in order
399 to achieve it (Blanco et al., 2016). One such adaptation strategy is the solvent efflux
400 pump system, that control intracellular toxicity to some extent (Kusumawardhani et al.,
401 2018). The role of *srpABC* solvent efflux pump system in the biodegradation of PAHs
402 are common in literature (Bugg et al., 2000). Besides the other cellular activities,
403 *srpABC* also assist in gaining antimicrobial resistance (Schweizer, 2003).
404 Biodegradation studies on chlorpyrifos indicate that the intermediates formed during
405 biodegradation, act as an antimicrobial agent to other species, meaning that the
406 metabolic intermediates may have a role on inter-species competition in a particular
407 micro-environment to thrive in nutrient-limiting condition (Anwar et al., 2009), (Raes
408 and Bork, 2008). Besides, the role of *srpABC* on biodegradation enhancement process
409 is still poorly understood. The presence of *srpABC* in *Pseudomonas putida* strain KD10
410 (Figure S5) and the efficient naphthalene biodegradation property can be interlinked
411 (Dutta et al., 2018).

412

413 3.7. Cell immobilization and viability count

414 The viscosity of CABs determines its efficiency of the immobilization and its
415 performance of detoxification of environmental pollutant (Young et al., 2006).
416 Therefore, sodium alginates of medium viscosity were chosen for this study. The
417 efficiency of cell immobilization in CABs was evaluated by enumeration of viable cells
418 (Table S3). Bacteria immobilized in CABs do not significantly lose its viable biomass
419 after 21 days of incubation at 4°C.

420

421 3.8. Naphthalene biodegradation studies

422 3.8.1 By free *Pseudomonas putida* strain KD10

423 Biodegradation kinetics of naphthalene (500 mg L⁻¹) by the *Pseudomonas putida*
424 strain KD10 was first tested to determine its naphthalene biodegradation potential
425 (Figure 4). It was found that after 12 days of incubation at 31°C the amount of residual
426 naphthalene reduced significantly ($p < 0.05$) (Figure 5). Results are also summarised at
427 Table 3 along with names of the immobilized systems. The first order degradation
428 kinetic data suggest that at the end of the incubation *Pseudomonas putida* strain KD10
429 efficiently decompose 95.22 % naphthalene in CSM as a sole source of carbon and
430 energy. The values of degradation rate constant (k) and half-life ($t_{1/2}$) were 0.2193 and
431 3.1 days with R^2 of 0.981. In the previous study, 99.1 % of initial naphthalene was
432 removed within 96 h by strain *Bacillus fusiformis* (BFN). However, the initial
433 concentration was very low (50 mg L⁻¹ of initial naphthalene) (Lin et al., 2010).

434 After confirmation of naphthalene biodegradation potential of *Pseudomonas*
435 *putida* strain KD10, the stain was allowed to grow in association with other two strains
436 of *Pseudomonas putida*, namely *Pseudomonas putida* strain KD6 and *Pseudomonas*

437 *putida* strain KD9 (Dutta et al., 2017; Dutta et al., 2018). These two strains were
438 selected to develop the blend of *Pseudomonas* sp. consortium, because they were
439 collected from different isolation points, they encode different variant of *nahAc* and
440 they have common optimized growth parameters (temperature 31°C and pH 7.1) (Dutta
441 et al., 2017 & Dutta et al., 2018). Further, *Pseudomonas putida* strain KD6 encodes a
442 six point mutant *nahAc* with the capability to co-degrade high concentration of
443 naphthalene, phenanthrene (PHN) and pyrene (PYR), (500 mg L⁻¹ each). Moreover, the
444 stain KD9 encodes a four point mutant variant of *nahAc* with rhamnolipid production
445 capabilities (Dutta et al. 2018).

446 In the previous study, $t_{1/2}$ of naphthalene co-biodegradation with phenanthrene
447 and pyrene by *Pseudomonas putida* strain KD6 was 4.1 days, which was significantly
448 reduced to 2.2 days when bacterial cells were allowed to grow as *Pseudomonas* sp.
449 consortium with only naphthalene (Table 3). Further, this value was found as 2.7 days
450 for *Pseudomonas putida* KD9, suggesting *Pseudomonas putida* strain KD6 might face
451 some sort of substrate inhibition by the co-presence PHN and PYR in the system (Jiang
452 et al., 2018) and the cooperative nature of the *Pseudomonas* sp. consortium helps to
453 enhance the naphthalene bio-utilization. The microbial consortium works several ways,
454 *viz.*, by the division of labor, cross-feeding, *etc.* (Smid and Lacroix, 2013). Moreover, a
455 successful consortium could also overcome shortcomings of single bacteria (Bhatia et
456 al., 2018). However, it is a fact that, bacteria select PAH among mixed PAHs based on
457 their structural simplicity first (Dutta et al., 2017) and the velocity could be optimized
458 by reducing the initial concentration of the PAHs gradually according to the structural
459 simplicity (Jiang et al., 2018).

460

461 3.8.2. *By immobilized Pseudomonas sp. consortium*

462 The biodegradation kinetic of naphthalene by individual *Pseudomonas putida*
463 strain KD6, KD9, KD10 and *Pseudomonas sp. consortium* immobilized in CABs,
464 depicted further enhancement of overall naphthalene bio-utilization (Table 3). Cell
465 immobilization using hydrogel, such as CAB found to be advantageous rather than free
466 cells (Hameed and Ismail, 2018).

467 This phenomenon may be attributable due to the increased level of tightness of
468 the cross-linked polymers of the calcium alginate beads that render bacteria adequate
469 amount of protection from harsh environment (Chen et al., 2013). However, the free
470 bacterial cell lacks the capabilities to degrade a high initial concentration of the toxicant
471 because they followed the conventional growth phases (Marrot et al., 2006). In addition,
472 exposure of free bacterial cells to the high initial concentration of toxicants may
473 challenge them to experience shock-concentration (Zhao et al., 2006). Conversely, cells
474 immobilized in calcium alginate beads can tolerate high concentration of the toxicant
475 and decrease the lag phase duration (Kao et al., 2014). Moreover, the diffusion
476 limitation natures of the CABs matrix provide a high local concentration of the cell
477 population (Bezbaruah et al., 2009). Furthermore, CABs provides remarkable stability
478 and reusable features that effectively reduces the production cost (Daâssi et al., 2014).

479 Biodegradation of naphthalene by immobilized *Pseudomonas sp. consortium*
480 significantly elevated the overall naphthalene bio-utilization efficiency with $t_{1/2}$ and R^2
481 values of 2 days and 0.998 respectively, suggesting experimental data are well
482 correlated to the model (Table 3). The individual cell population also displayed
483 improved biodegradation efficiency. However, the biodegradation efficiency was
484 maximum by *Pseudomonas putida* strain KD6, which showed a marked reduction of $t_{1/2}$

485 from 4.1 to 3.0 days (Table 3). It promptly suggests that CABs facilitate KD6 optimized
486 naphthalene bio-utilization, that might be masked-up by the co-presence of NAP, PHN,
487 PYR (Jiang et al., 2018). However, further studies are required to investigate
488 behavioural patterns of a consortium with mixed PAH.

489

490 Bacterial cell grown on CSM with naphthalene as a sole source of carbon and
491 energy showed a delayed growth rate and their total biomass was also low (Figure 1).
492 Nevertheless, growing cells on CSM prior to immobilization in CABs provide them
493 essential adaptation to the toxicant and mature them for such stress condition (Table 3).
494 Immobilized cell system was useful for bioremediation of a toxicant after prior
495 adaptation to the surrounding environment (Partovinia and Rasekh, 2018). However,
496 growing cells in Luria-Bertani broth does not adapt cell adequately, and their
497 naphthalene removal performance was very poor (Figure S3). The inoculum was
498 normalized ($OD_{600nm} = 0.002$) for each *Pseudomonas putida* strain in order to prepare
499 the *Pseudomonas* sp. consortium. In aqueous medium, the immobilized bacterial cell
500 mainly on the surface was exposed to naphthalene and primarily involved in the
501 naphthalene bio-utilization process. However, due to the micro-porous feature of the
502 calcium alginate beads, microbial cells immobilized other than surface are also
503 participates in the bio-utilization and bio-sorption process of naphthalene. Further, with
504 time due to the mechanical force generated by shaking, a few microbial cells may
505 release from the calcium alginate beads. However, to fuel up the cell, it is necessary to
506 uptake naphthalene for bio-utilization through step by step intracellular enzymatic
507 reactions (Lin et al., 2014).

508

509

510 *3.8.3. Effect of initial naphthalene concentration on biodegradation kinetics*

511 The effect of different initial concentration of naphthalene (150 – 2500 mg L⁻¹)
512 on degradation kinetics was evaluated with immobilized KD6, KD9, and KD10 as
513 individual and as a consortium (Table S2). The results suggest cell immobilization in
514 CABs facilitates bacteria to cope with a high initial concentration of NAP (Table S2).
515 The substrate inhibition kinetic parameters, *viz.*, the maximum specific degradation rate
516 (q_{\max}), and inhibition constant (k_i) were highest for immobilized *Pseudomonas* sp.
517 consortium and these values were 0.707 h⁻¹ and 1475 mg L⁻¹ respectively (Table S2).
518 However, we did not find any significant change on half saturation constant (k_s),
519 suggesting the reaction does not depend on its initial concentration and the
520 biodegradation process follows pseudo-first order reaction kinetics. In previous study,
521 free *Pseudomonas putida* KD9 in CSM was capable to tolerate relatively low initial
522 concentration of naphthalene with k_i value of 1107 mg L⁻¹ and addition of sucrose as
523 SCS provides quite similar potential of naphthalene tolerance (k_i of 1429 mg L⁻¹) with
524 that of immobilized *Pseudomonas* sp. consortium (Dutta et al., 2018). However, sucrose
525 (0.5 gm. %) in the mobile open water system would not be beneficial for bacteria to
526 overcome the high shock concentration of the toxicant, again suggesting cell
527 immobilization and development of effective microbial consortium as systematic
528 optimization of biodegradation process.

529 *3.9. Detection of metabolic end products*

530 The metabolic pathway of naphthalene biodegradation that might be followed by
531 *Pseudomonas putida* strain KD10 was elucidated through GCMS analysis of the

532 metabolites (Figure S4). In the previous study, the major metabolites were restricted to
533 be salicylaldehyde, catechol, D-gluconic acid and pyruvic acid (Dutta et al., 2018).
534 However, in the present study, we have detected salicylic acid as an additional
535 metabolite (Table 4). Catechol step into the TCA cycle by two possible pathways one is
536 ortho-cleavage and another is meta-cleavage pathway. The meta-cleavage pathway is
537 led by catechol-2, 3-dioxygenase (*nahH*), and presence of *nahH* (Figure S5) suggesting
538 in *Pseudomonas putida* strain KD10 opt the meta-cleavage pathway. Furthermore, we
539 assume that D-gluconic acid enters in the bio-conversion of naphthalene, possibly from
540 glucose as a precursor (Figure 6) and it is finally converted to pyruvate via aldolase. D-
541 gluconic acid implies impotence between bacteria-plant mutualistic association e.g.,
542 induction of phosphate solubilization processes (Rodriguez et al., 2004) which can
543 inhibit fungal growth (Kaur et al., 2006). The metabolic intermediates viz., salicylic
544 acid, catechol has plant-growth promoting activity (Lee et al., 2010) e.g., the antioxidant
545 property of the catechol help and promotes seed germination (Schweigert et al., 2001).
546 The presence of d-gluconic acid as a major metabolite and its subsequent entry in TCA
547 cycle from glucose precursor, suggest that KD10 may be able to sequestered catechol upon
548 needs to promote the plants' growth (Figure 6).

549

550 4. Conclusion

551

552 Naphthalene biodegradation by free and immobilized *Pseudomonas putida*
553 strain KD10 and *Pseudomonas* sp. consortium were studied. HPLC analysis showed
554 80.1% of initial naphthalene (500 mg L⁻¹) was utilized by immobilized *Pseudomonas*

555 sp. consortium after 72 h of incubation. Further, initial naphthalene tolerance by
556 immobilized *Pseudomonas* sp. consortium was highest (1475 mg L⁻¹) with maximum
557 specific degradation of 0.707 h⁻¹. The gene sequence analysis of the naphthalene 1, 2-
558 dioxygenase suggests a significant evolutionary distance among different microbial
559 species and in few cases, intra-species variation was observed. A common mutation
560 prone amino acid stretch inside Chain A of all three natural mutant variants of
561 naphthalene 1, 2-dioxygenase were found at close proximity of the active site. Further,
562 the rigid-flexible molecular docking showed better binding free energy of the mutant
563 variant encoded by *Pseudomonas putida* strain KD10 than that of wild-type variant of
564 naphthalene 1,2-dioxygenase. This common mutation prone amino acid stretch could
565 aid the rational approaches of enzyme redesigning. Overall, this study summarises the
566 application of bacterial cell immobilization in calcium alginate beads and development
567 of the microbial consortium together for enhanced naphthalene biodegradation.
568 However, further studies are required for the systematic optimization of naphthalene
569 biodegradation in a real environment.

570

571 *Authors Contributions*

572 KD and CG conceive the main hypothesis. KD design, performed all
573 experiments, and wrote the manuscript. SS, IK, SB, DJ assisted KD in some
574 experiments. MK, TM, PR, KCG performed the statistical analysis and wrote the
575 manuscript. CG critically proofread and wrote the manuscript. All authors read the
576 manuscript.

577

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588

589 *Conflict of Interest*

590 The authors declare that they have no conflict of interests.

591

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774 *Figure legends*

775 **Figure 1.** Growth kinetic model of *Pseudomonas putida* and *Pseudomonas* sp.
776 consortium during naphthalene biodegradation. **A.** Gompertz's growth kinetic
777 model fit of the biodegradation of naphthalene by *Pseudomonas putida* strain
778 KD10 and by the *Pseudomonas* sp. consortium. **B.** Effect of 0.5 gm. % sucrose
779 supplementation on growth pattern of *Pseudomonas putida* strain KD10 and
780 *Pseudomonas* sp. consortium.

781 **Figure 2.** Cell immobilization in calcium alginate beads (CABs) with different
782 cell morphology of the *Pseudomonas putida* strain KD10.
783

784 **Figure 3.** Sequence and structure analysis of the naphthalene 1, 2-dioxygenase
785 (*nahAc*) encoded by *Pseudomonas putida*. **A.** The amino acid sequence of wild
786 type and three mutant variants of the naphthalene 1, 2-dioxygenase (*nahAc*).
787 Mutated residues are highlighted in blue and the common mutation prone amino
788 acid stretch is highlighted with red box. **B.** Structural mismatches among mutant
789 variants and wild type naphthalene 1, 2-dioxygenase. Each mutated residues and
790 its subsequent neighbour residues of all the mutant variant of naphthalene 1, 2-
791 dioxygenase is overlaid with cartoon representation. The highlighted red circle
792 indicates local mismatch among the mutant variant of naphthalene 1, 2-
793 dioxygenase **C.** Different ligand binding posture of the naphthalene 1, 2-
794 dioxygenase encoded by *Pseudomonas putida* strain KD10. Naphthalene (red),
795 phenanthrene (cyan) and anthracene (green) binding postures with major
796 interacting amino acid residues (left). Two-dimensional presentation of the
797 interacting residues of *nahAc*_{1250, V256} (right). **D.** Docking pose of naphthalene 1,
798 2-dioxygenase (Chain A) encoded by *Pseudomonas putida* strain KD10. The
799 four major naphthalene interacting residues and their molecular distance are
800 labelled in black.

801

802

803 **Figure 4.** Distance matrix and cladogram of the naphthalene 1, 2-dioxygenase.
804 The evolutionary distance of the naphthalene 1, 2-dioxygenase I₂₅₀, V₂₅₅ variant
805 encoded by *Pseudomonas putida* strain KD10 other variants by different
806 microbial species is scaled by colour index (upper-left side).

807
808

809 **Figure 5.** Naphthalene degradation kinetics of *Pseudomonas putida* strain KD10
810 and *Pseudomonas* sp. consortium free and cell immobilized in calcium alginate
811 beads.

812

813 **Figure 6.** Proposed pathway of naphthalene biodegradation by *Pseudomonas*
814 *putida* strain KD10. Naphthalene (I), salicylic acid (II), salicylaldehyde (III),
815 catechol (IV), pyruvate (V), glucose (VI), d-gluconic acid (VII). The grey arrow
816 indicates alternative pathway via d-gluconic acid.

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818

819 *Table captions*

820 **Table 1.** Summary of rigid-flexible molecular docking for the analysed mutant
821 naphthalene 1, 2-dioxygenase encoded by different *Pseudomonas putida*
822 strains.

823

824 **Table 2.** Apparent enzyme kinetic parameters of naphthalene 1, 2-dioxygenase
825 I₂₅₀, V₂₅₆ encoded by *Pseudomonas putida* strain KD10.

826

827 **Table 3.** Parameters of first-order biodegradation kinetics of naphthalene by
828 different free and immobilized *Pseudomonas putida* strains.

829

830 **Table 4.** Major metabolite detected by GCMS analysis during naphthalene
831 biodegradation by *Pseudomonas putida* KD10.

832

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834

835 *Supplementary section*

836

837 **Table S1.** Nucleotide sequences used as primer in polymerase chain reactions

838

839 **Table S2.** Parameters of substrate inhibition kinetic model of naphthalene
840 biodegradation by different *Pseudomonas putida* strains immobilized in calcium
841 alginate beads.

842 **Table S3.** Enumeration of viable cell in calcium alginate beads.

843 **Table S4.** Gompertz's growth curve model fit of *Pseudomonas putida* strain
844 KD10 and *Pseudomonas* sp. consortium.

845

846 **Table S5.** Major amino acid residues of mutant variant of naphthalene 1, 2-
847 dioxygenase _{I250, V256} involve in hydrophobic interaction with different ligands.

848

849 **Figure S1.** Colony morphology and catechol confirmatory test. **A.** Colony
850 morphology of *Pseudomonas putida* strain KD10 after 24 h of incubation. **B.**
851 Colony morphology of *Pseudomonas putida* strain KD10 after one week of
852 incubation.

853 **Figure S2.** Molecular phylogenetic analysis of *Pseudomonas putida* strain
854 KD10.

855 **Figure S3.** Effect of cell adaptation on naphthalene biodegradation.

856 **Figure S4.** Mass spectra of the major metabolites detected during
857 biodegradation of naphthalene in CSM as sole source of carbon and energy by
858 *Pseudomonas putida* KD10.

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860 **Figure S5.** Agarose gel electrophoresis of PCR products of catechol 2, 3-
861 dioxygenase (*nahH*) gene and solvent efflux pump (*srpABC*) system.

862 **A.** Catechol 2, 3-dioxygenase (*nahH*). Lane 1 and 2 *nahH* pcr product of
863 *Pseudomonas putida* strain KD9 and *Pseudomonas putida* strain KD10. **B.**
864 Solvent efflux pump (*srpABC*) system. Lane 1: *Pseudomonas aeruginosa*
865 ATCC 9027, Lane 2: *Pseudomonas putida* strain KD10. Lane M: 100bp DNA
866 ladder (HiMedia, India), Lane 3: Negative control.

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868 **Figure S6.** Nucleotide sequence chromatogram of the mutant naphthalene 1, 2-
869 dioxygenase (*nahAc*), (chain-A) encoded by *Pseudomonas putida* strain KD10.

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Table 1. Summary of rigid-flexible molecular docking of naphthalene and different mutant version of naphthalene 1, 2-dioxygenase encoded by different *Pseudomonas putida* strains

ΔG^*			LB ⁴	IC ⁵	IME ⁶	VDW ⁷	RMSD ^c	RMSD ^r	References
BE ¹	London ²	GBVI/ WSA ³							
-5.92 ^a	-7.18	-4.61	-0.59	46.14	-5.92	-5.91	0.00	77.29	Dutta et al., 2017
-5.41 ^b	-7.13	-4.02	-0.54	109.07	-5.41	-5.4	0.00	96.32	Dutta et al., 2019
-5.83 ^c	-7.10	-1.68	-0.58	53.13	-5.83	-5.84	0.00	96.08	This study

*PubChem ligand CID 931 = Naphthalene, BE¹ = Binding free energy, ²London free energy, ³Generalized born volume integral/weighted surface areas, LB⁴ = ligand binding, IC⁵ = inhibition constant, IME⁶ = Intermolecular energy, VDW⁷ = Vdw hb desolv energy, RMSD^c = Cluster RMSD, RMSD^r = Reference RMSD. Energy unit = kcal mol⁻¹. Inhibition constant unit = μ M.at 298.15 K. Naphthalene 1, 2-dioxygenase encoded by (a) *Pseudomonas putida* strain KD6, (b) *Pseudomonas putida* strain KD9 and (c) *Pseudomonas putida* strain KD10.

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Table 2. Apparent enzyme kinetic parameters of naphthalene 1, 2-dioxygenase_{I250, V256} encoded by *Pseudomonas putida* strain KD10

Substrate	Kinetic parameters*				
	K _m ($\mu\text{mol mL}^{-1}$)	V _{max} ($\mu\text{mol min}^{-1}$)	K _{cat} (s^{-1})	K _{cat} /K _m ($\text{mL}^{-1} \text{mol}^{-1} \text{s}^{-1}$)	R ^{2**}
Naphthalene	0.867	2.474	123.72	0.142×10^3	0.938

*The kinetic constants were determined at 30°C and pH 7.5 using 0.5-10 $\mu\text{M mL}^{-1}$ substrate concentration (each) by Lineweaver–Burk plot.

** Non-linear regression between initial substrate concentration 1/[S] and degradation rate constant 1/V yielded regression equation and regression coefficient (R²)

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Table 3. Parameters of first-order biodegradation kinetics of naphthalene by different free and immobilized *Pseudomonas putida* strains

CSM	Cells immobilized in CABs			free cells			References
	k (Days ⁻¹)	$t_{1/2}$ (Days ⁻¹)	R^2	k (Days ⁻¹)	$t_{1/2}$ (Days ⁻¹)	R^2	
A	0.228±0.005	3.0±0.077	0.99	0.167±0.012	4.1±0.307	0.98	Dutta et al., 2017
B	0.312±0.014	2.2±0.102	0.93	0.252±0.019	2.7±0.193	0.99	Dutta et al., 2018
C	0.284±0.017	2.4±0.143	0.97	0.219±0.009	3.1±0.137	0.92	This study
D	0.341±0.294	2.0±0.186	0.99	0.309±0.020	2.2±0.141	0.98	This study

#Initial naphthalene concentration = 500 mg L⁻¹

A. *Pseudomonas putida* strain KD6, B. *Pseudomonas putida* strain KD9, C. *Pseudomonas putida* strain KD10. D. *Pseudomonas* sp. consortium.

Each figure of the table represents the mean of three replicates.

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Table 4. Major metabolites detected by GCMS analysis during naphthalene biodegradation by *Pseudomonas putida* KD10

Metabolite	R _t (min)	Major ion peaks (m/z)	Suggested structure
I	9.41	64.19, 92.17, 120.21, 138.22, 169.14, 191.18, 211.10	Salicylic acid
II	15.27	65.17, 74.41, 104.73, 121.10, 122.57, 142.21, 174.80, 201.22, 249.31	Salicylaldehyde
III	18.71	64.29, 81.35, 91.61, 110.47, 131.47, 149.71	Catechol
IV	31.89	61.21, 73.44, 76.24, 104.18, 117.41, 133.27, 177.23	D-gluconic acid
V	10.47	43.21, 61.27, 71.37	Pyruvic acid

R_t = Retention time

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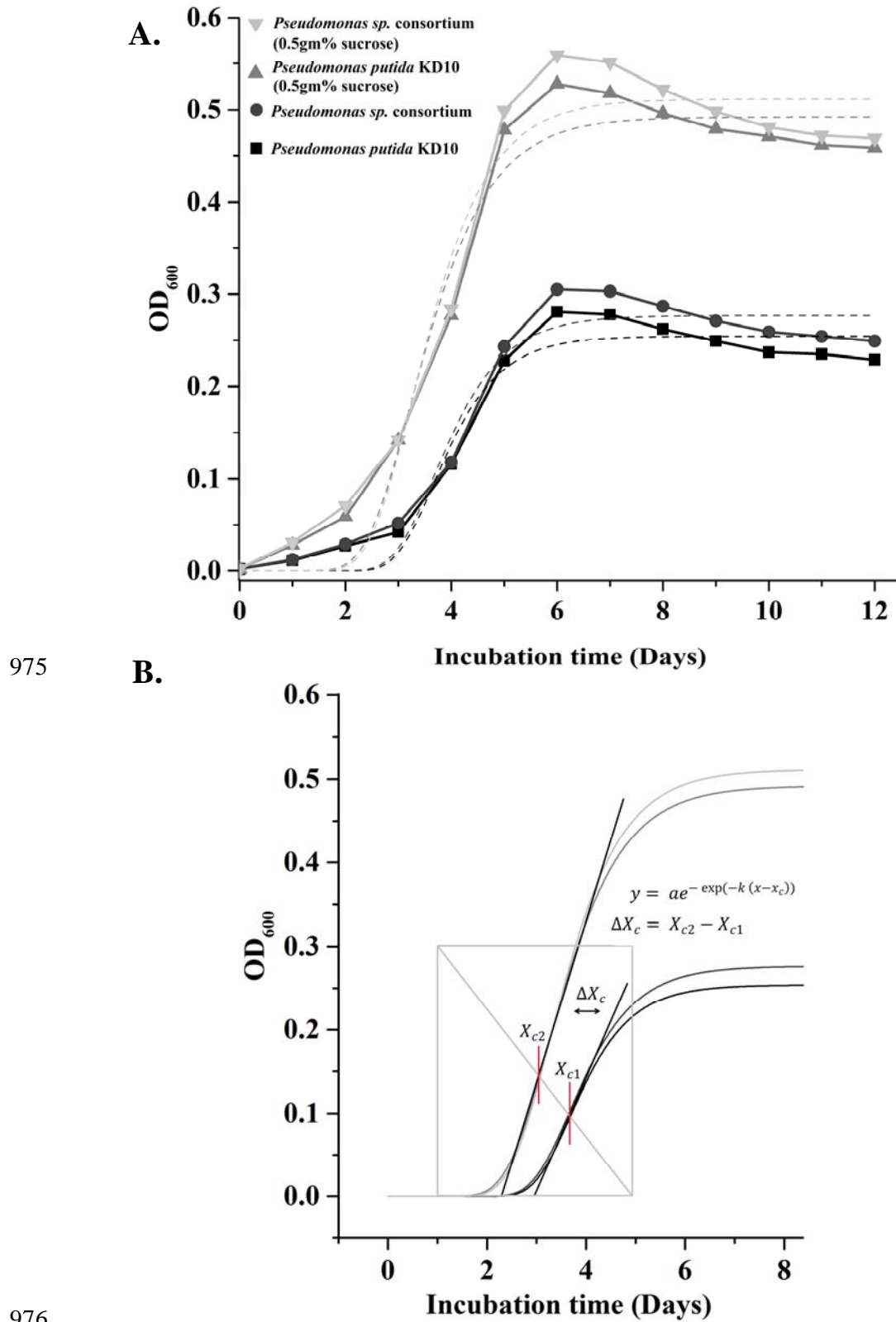
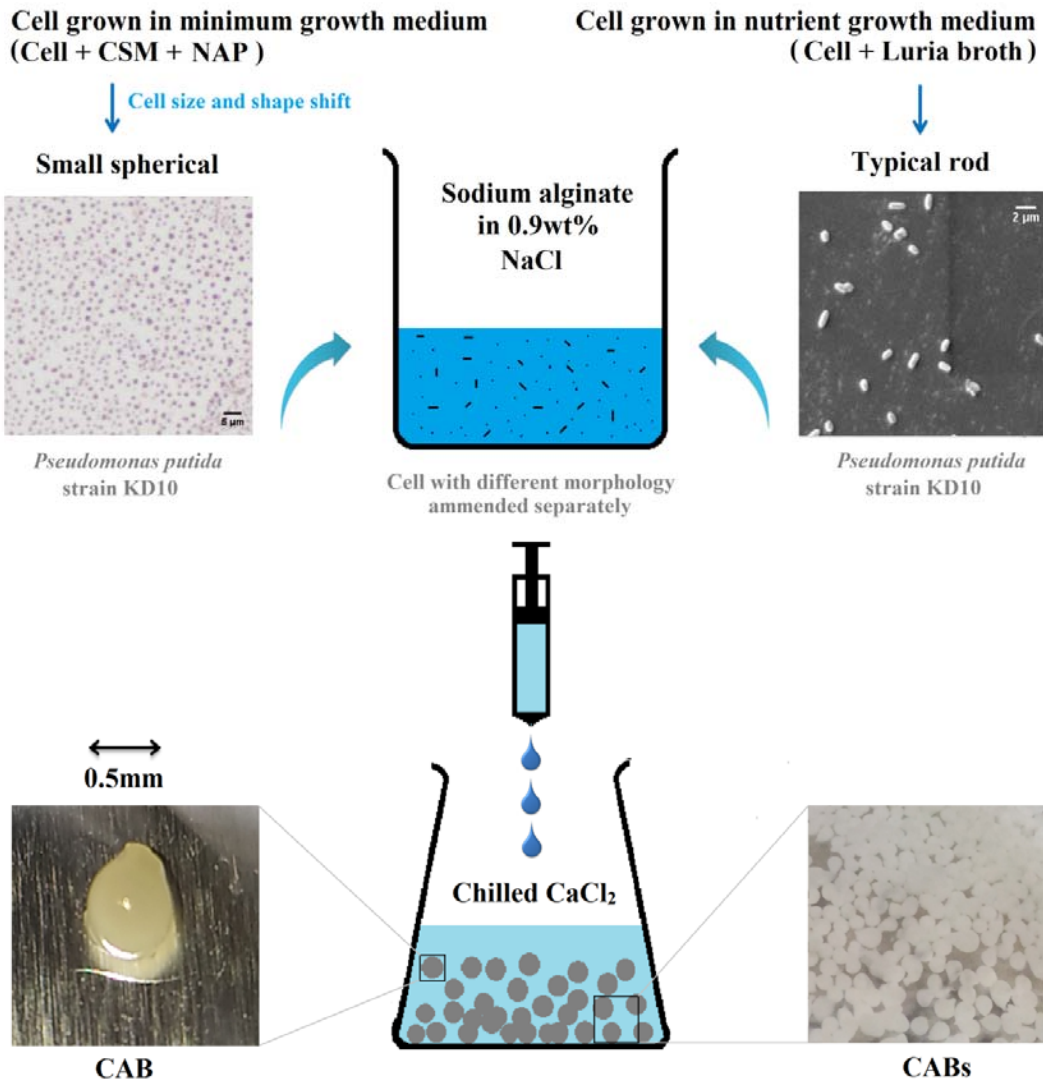


Figure 1.



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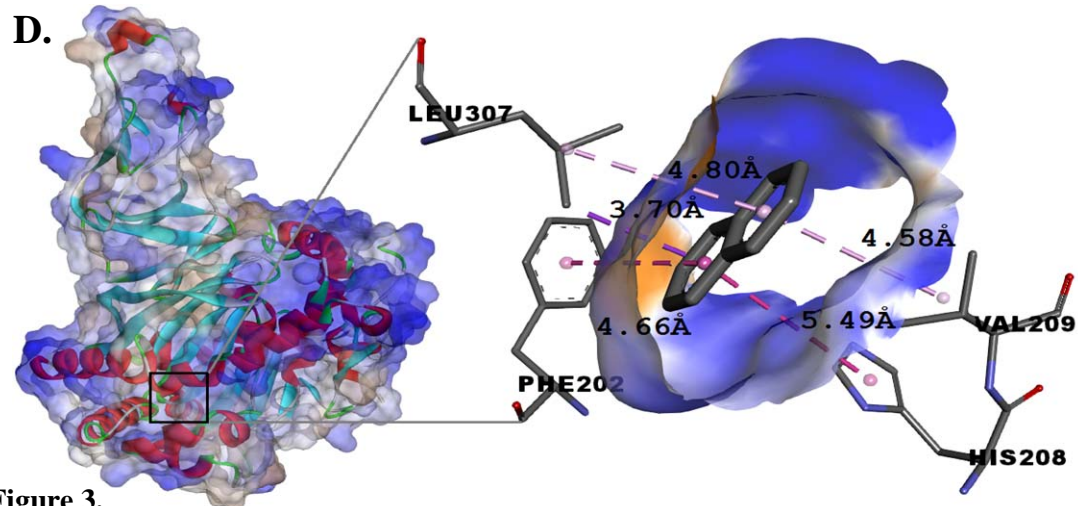
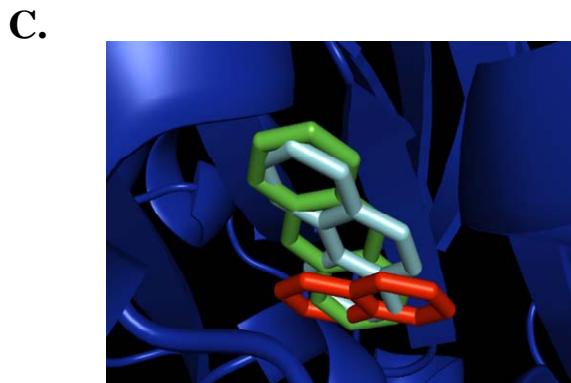
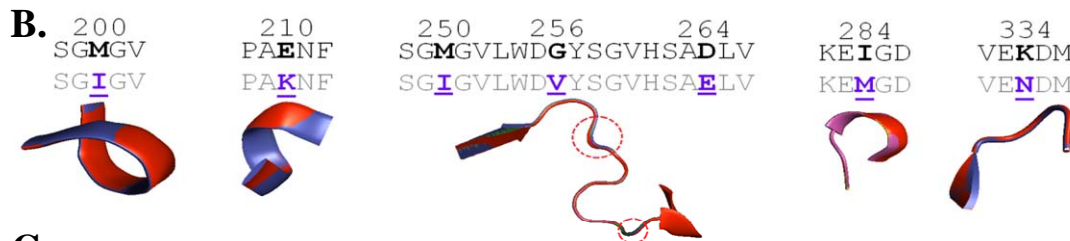
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Figure 2.

A. >Chain A, Naphthalene 1, 2-Dioxygenase

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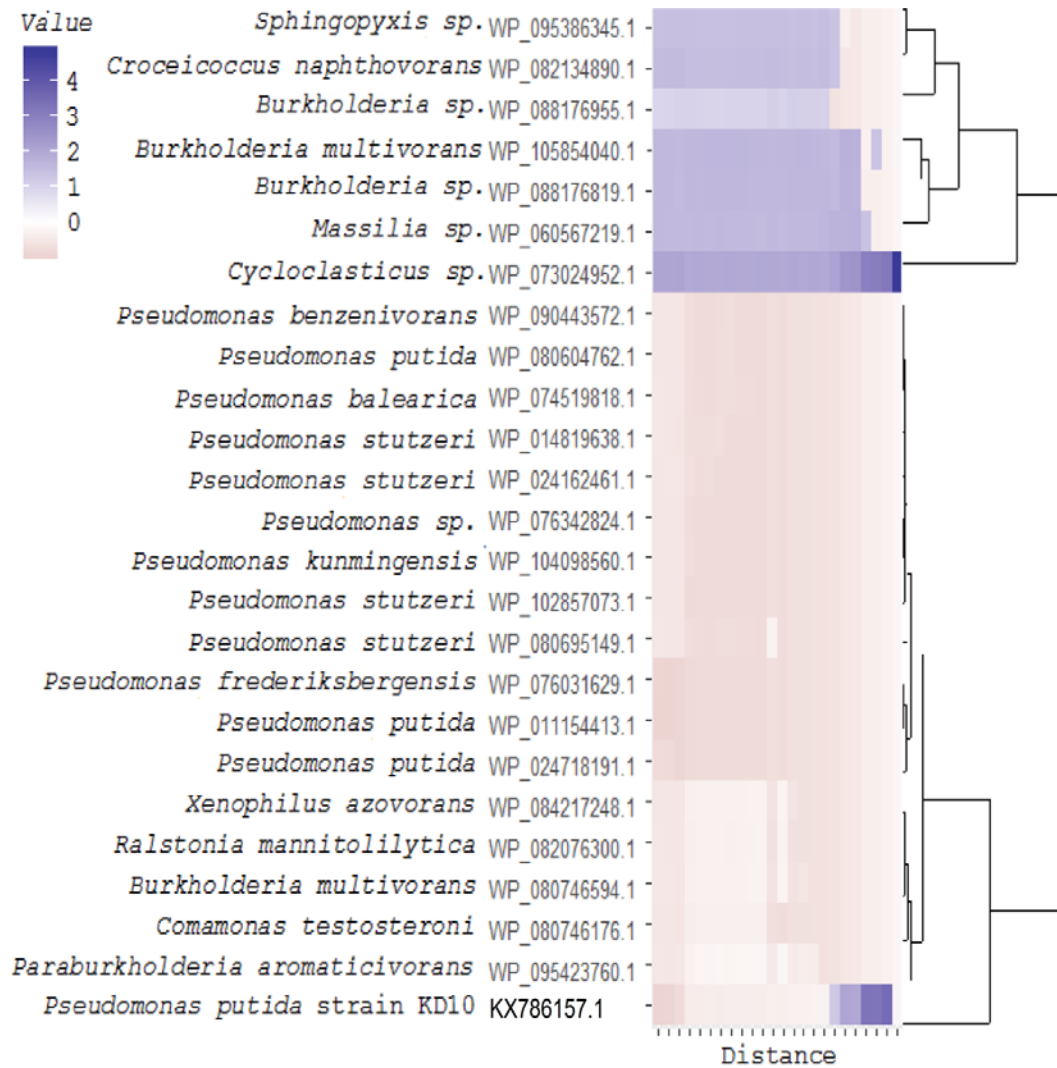
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1NDO  MNYNNKILVSEGLSQKHLIHGDEELFOHELKTIFARNWLFLTHDSLIPAPGDYVTAKMGIDEVIVSRON
      80      90     100     110     120     130     140
1NDO  DGSIRAFLNVCRHRGKTLVSVVEAGNAKGFVCSYHGWFGSNGELQSVPFEKDLYGESLNKKCLGLKEVAR
      150     160     170     180     190     200     210
1NDO  VESFHGFFIYGCFDQEAPPLMDYLGDAAWYLEPMFKHSSGGLELVGPPGKVVIKANWKAPAENFVGDAYHVG
      220     230     240     250     260     270     280
KD6   VESFHGFFIYGCFDQEAPPLMDYLGDAAWYLEPMFKHSSGGLELVGPPGKVVIKANWKAPAENFVGDAYHVG
      220     230     240     250     260     270     280
1NDO  WTHASSLRSGESIFSSLAGNAALPPEGAGLQMTSKYSGMGVLWDGYSGVHSADLVPELMAFGGAKQERL
      290     300     310     320     330     340     350
KD10  WTHASSLRSGESIFSSLAGNAALPPEGAGLQMTSKYSGMGVLWDVYSGVHSADLVPELMAFGGAKQERL
      290     300     310     320     330     340     350
KD9   WTHASSLRSGESIFSSLAGNAALPPEGAGLQMTSKYSGMGVLWDVYSGVHSADLVPELMAFGGAKQERL
      290     300     310     320     330     340     350
KD6   WTHASSLRSGESIFSSLAGNAALPPEGAGLQMTSKYSGMGVLWDVYSGVHSADLVPELMAFGGAKQERL
      290     300     310     320     330     340     350
1NDO  NKEIGDVRARIYRSHLNCTVFPNNSMLTCSGVFKVWNPIDANTTEVWTYAIVEKDMPEDLKRRLADSVQR
      360     370     380     390     400     410     420
KD6   NKEIGDVRARIYRSHLNCTVFPNNSMLTCSGVFKVWNPIDANTTEVWTYAIVEKDMPEDLKRRLADSVQR
      360     370     380     390     400     410     420
KD9   NKEIGDVRARIYRSHLNCTVFPNNSMLTCSGVFKVWNPIDANTTEVWTYAIVEKDMPEDLKRRLADSVQR
      360     370     380     390     400     410     420
1NDO  TFGPAGFWESDDNDNMETASQNGKYQSRDSDLNLSLGFGEDVYGDAVYPGVVGKSAIGETSYRGFYRAY
      430     440     449
1NDO  QAHVSSNWAEFEHASSTWHTELTKTDR
    
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979 Figure 3.

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983 **Figure 4.**

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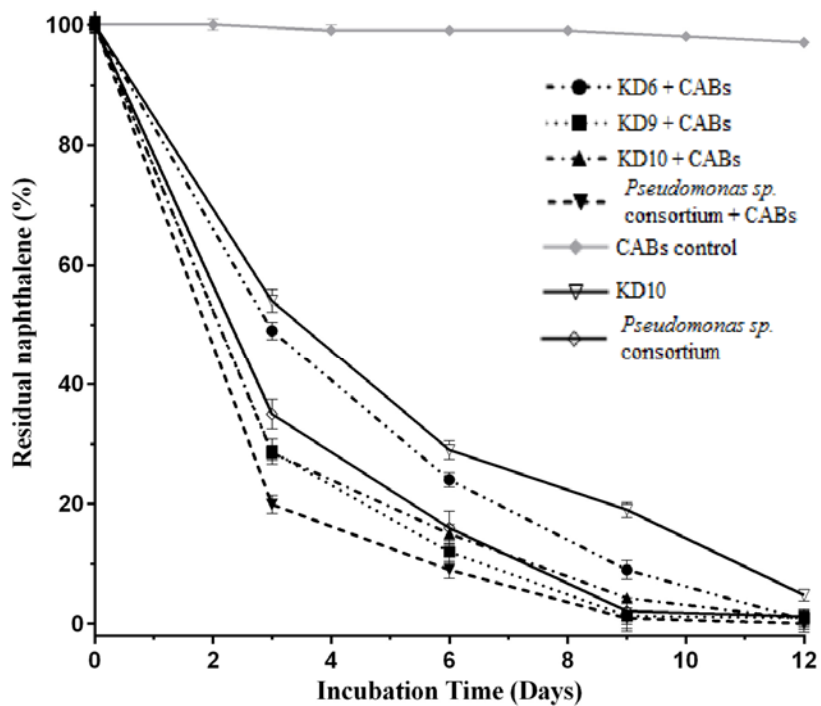
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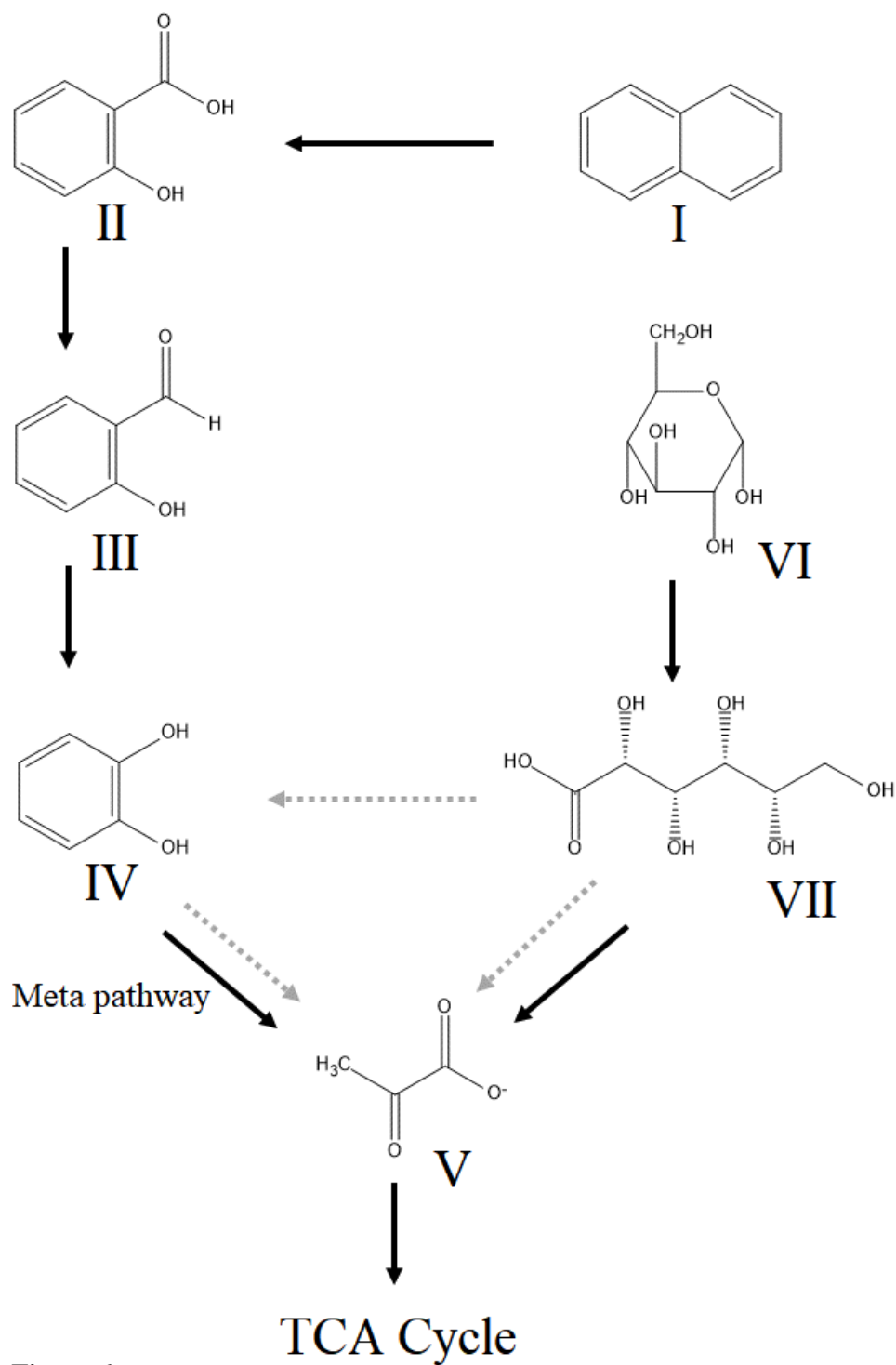
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Figure 5.



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Figure 6.