Exogenous application of KNO₃ elevates the salinity tolerance of *Stevia rebaudiana* through ion homeostasis mechanism

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Highlight

The detrimental effects of moderate and higher salinity levels on growth and dry leaf yield of stevia were observed. However, tolerance level can be elevated through exogenous application of KNO₃.

Abstract

Though relatively little is understood of adaptation, physiological and metabolic changes of *Stevia rebaudiana* under exposure to salinity stress, it is hypothesized that exogenous application of potassium (K⁺) could elevates the salinity tolerance through ions homeostasis. Thus, an experiment was conducted with twenty treatment combinations comprising four salinity levels (irrigation with normal water as control and three level of NaCl at 40, 80 and 120 mM) and five different concentrations of KNO₃ (0.0, 2.5, 5.0, 7.5, and 10.0 g L⁻¹). Dry leaf yield was not negatively affected with mild salinity (40 mM). However, the detrimental effects were observed at moderate and higher salinity levels (80 and 120 mM). The uptakes of K⁺, Ca²⁺, and N were significantly reduced at higher salinity level, whereas accumulations of Na⁺ and Cl⁻ ions in plant tissues were substantially increased. Proline content in leaf was also increased significantly ($P \le 0.05$) in response to salt stress. Among the foliar application, KNO₃ at 5.0 gL⁻¹ registered significantly ($P \le 0.05$) higher dry leaf yield compared with control. Exogenous application of K⁺ under moderate salinity stress maintained ion balance in cytosol, particularly K: Na. Thus, the salinity tolerance of stevia can be elevated to some extent through exogenous application of K⁺.

Keywords: KNO₃, Proline, Salinity stress, *Stevia rebaudiana*, Steviol glycosides, Total phenols,

1 **1. Introduction**

2 The burgeoning rate of diabetic and obesity patients is a serious apprehension in the 3 worldwide. About 346 million people worldwide are in the grip of diabetes whereas in India 4 69.2 million people are diabetic (WHO 2015). The cane sugar is not recommended for 5 consumption to the diabetic and obesity patients. In these circumstances, stevia (Stevia 6 rebaudiana) has emerged to be a natural sweet gift for millions of diabetics as a sweet 7 quotient in their daily life. The stevia leaves contain sweet-tasting and low-calorie 8 diterpenoid steviol glycosides (SGs), which are nearly 300 times sweeter than sucrose 9 (Kinghorn 2002). However, the performance of stevia in saline soil or plant irrigated with 10 saline water is not properly elucidated so far. The global extent of salt-affected land amounts 11 to about 1128 million ha (FAO 2008). The term salt affected refers to the soils that are saline 12 or sodic, which is one of the major stress factors for agriculture in consequence of climate 13 change. Salinity stress alone adversely affects to the production in over 30% of irrigated 14 crops and 7% of dryland agriculture worldwide (Schroeder et al. 2013). Development of 15 agrotechniques to elevate the salt-tolerance within plant or cultivation of salt-tolerant crops 16 may be an efficient approach for better utilization of salt-affected land to address the related 17 issues.

18 The restriction of plant growth in the saline soil is directly linked to osmotic potential of the 19 soil solution and total concentration of soluble salts, and the different mechanisms have been 20 developed by the plants to cope with these situations (Munns 2002; Tavakkoli et al. 2011). 21 The presence of higher concentration of NaCl in soils increases the accumulation of the 22 phytotoxic ions sodium (Na⁺) and chloride (Cl⁻) in plant organs, concurrently reduces the uptake of other essential nutrients, such as potassium (K⁺), calcium (Ca²⁺), magnesium 23 24 (Mg^{2+}) and nitrate (NO_3^{-}) ion (Niu *et al.* 1995; Cornillon and Palloix, 1997; Halperin *et al.* 25 2003; Munns and Tester 2008; James et al. 2011). The imbalance of ions creates the decisive 26 conditions for plant survival by intercepting different plant mechanisms. Thus, plant 27 accumulates various osmolytes (proline, glycine betaine and sugars), secondary metabolites 28 and antioxidants in response to stress which helps in the survival of the plant through 29 maintaining plant turgor (Ashraf and Foolad 2007). 30 Salt stress restricts the N assimilate in plant body through the attenuation of activity of cyto-

- 31 solic NADH nitrate reductase enzyme in nitrate assimilation pathway (Sivasankar and Oaks
- 32 1996; Jabeen and Ahmad 2011). Potassium is another important macronutrient for optimum

33 growth and development of the plant. Potassium also plays important role in maintenance of

34 osmotic adjustment during stress conditions. It is established fact that K⁺ concentration in

35 plant tissue declines as the Na^+ concentration in the root media is increased (Subbarao *et al.*

36 1990; Izzo *et al.* 1991; Perez- Alfocea *et al.* 1996). This disruption of K⁺ further disturbs

37 many metabolic processes in the cytoplasm (Marschner 1995).

38 Thus, the mechanisms to salinity tolerance should be understood properly to elevate salt

39 tolerance for a particular plant species for cultivation in salt affected areas. Exclusion of

40 surplus Na⁺ or compartmentalization and ion homeostasis are the basic mechanisms of salt

41 tolerance for plants (Li et al. 2006; Munns and Tester 2008). So, the exogenous application of

42 nutrients particularly K^+ and NO_3^- through foliar feeding under saline conditions may be an

43 effective mechanism to reduce Na^+ and Cl^- injury and to elevate tolerance against saline to

44 various extents. Foliar spray is an efficient approach, which involves active and passive

45 processes for translocation of nutrients through leaves to other organs of plant when nutrient

46 uptake from soil is disturbed (Fageria *et al.* 2009).

47 However, the degree of salinity tolerance depends upon plant spices. Very few research

48 works have been conducted to understand the effect of salinity on growth, the extent of

49 tolerance and mechanism of tolerance in stevia. Similarly, the attempt toward elevating the

salinity tolerance of stevia has not been initiated so far. Thus, the present experiment was

51 executed with the specific objectives of (i) determining the degree of tolerance and

52 understanding the effects of salinity on the growth, biochemical activities, and steviol

53 glycoside accumulation; (ii) evaluating the effectiveness of exogenous application of K and N

54 to elevate the salinity tolerance in *S. rebaudiana*.

55 56

2. Materials and Methods

57 2.1. Experimental location, plant material and application of treatments

58 A pot experiment was conducted in the open-ended poly-tunnel at CSIR-Institute of

59 Himalayan Bioresource Technology, Palampur, India during two growing seasons of 2016

and 2017. The physico-chemical properties of the experimental soil are presented in **Table 1**.

61 Sixty-days-old stevia seedlings were selected randomly from uniform populations maintained

- 62 in a poly-house under natural light for transplantation into the pot containing 14 kg soil. For
- basal dose, 690 mg N as urea, 136 mg P as single super phosphate, and 311 mg K were

64 applied to the pots. At twentieth-days of transplantation, plants were taped in order to 65 maintain the uniform height. The experiment was carried out in two-factor factorial complete 66 randomized design (CRD) with ten replications. The study includes twenty treatment 67 combinations involving four salinity levels of irrigation water (irrigation with normal water 68 as control and three level of NaCl at 40, 80 and 120 mM) and five different concentrations of KNO_3 (water spray as control, KNO_3 at 2.5, 5.0, 7.5, and 10.0 g L⁻¹). After transplanting, 69 70 plants were irrigated with normal tap water. Thereafter, individual plants were irrigated 71 treatment wise as per requirement throughout the experimental period. The foliar spray of KNO₃ for different treatments was initiated at 10th day after first salinity treatment. Distilled 72 water was used for water spray as control treatment. The foliar spray of KNO₃ solutions took 73 74 place at the interval of 10 days for three times.

75 2.2. Growth and yield

The five plants were randomly harvested after 90 days of transplantation from each treatment.

After recording plant height (cm) and number of branches per plant, the leaves were

reasonable reasonable

⁷⁹ help of Windias-3 Image analysis system (Delta-T Devices Ltd. UK). The fresh weight of

leaf, stem and root were also recorded immediately. Then, the samples were dried at 70 °C \pm 2

^oC in hot air oven until constant weight was attained in order to measure the dry weight.

82 Specific leaf weight (SLW) was also calculated based on leaf area and dry leaf weight, and it

was expressed in mg cm $^{-2}$. The dried leaf, stem and root samples were stored for further

84 chemical and biochemical analysis.

85 2.3. Chlorophyll (Chl) analysis

86 The leaves were collected from each pot at harvesting stage for estimation of Chl content in

the leaf. The Chl was extracted from 200 mg fresh leaf samples with a 25 mL of 80% acetone

(v/v) and was kept under dark conditions at room temperature for 24 h to avoid

89 photooxidation. The absorbance of the extract was recorded at two different wavelengths i.e.

90 645 nm and 663 nm with a UV/Vis spectrophotometer (Shimadzu, UV-2600). The Chl_a , Chl_b

and total Chl content (mg g⁻¹) were determined by using standard equations given by Arnon
(1949).

93 2.4. Determination of Proline

94 The accumulation of proline in leaf was quantified as per the method recommended by Bates 95 et al. (1973). The proline was extracted from 0.5 g fresh leaf samples of stevia with 10 mL of 96 3 % sulphosalicylic acid and then, centrifuged at 12000 g for 10 minutes. 1 mL supernatent 97 was reacted with the equal volume of acid-ninhydrin and glacial acetic acid in a test tube and 98 incubated for 1 h at 100° C. The reaction was stopped by keeping the test tube in an icebath. 99 Then, 2 mL tolune was added to test tube and mixed vigorously and left it undisturbed for 30 100 minutes at room temperature. After that, the sample mixture was separated into two phases. 101 The optical density of the chromophore containing tolune was measured at 520 nm with 102 spectrophotometer (model T 90 C UV/vis, PG Instrument Ltd.). The proline content was 103 determined based on standard curves developed with D-proline.

104 **2.5.** Total phenols

105

For the analysis of total phenols, 0.1 g of leaf sample was suspended in 25 mL solution of 106 70% acetone in a conical flask. The conical flask was kept on shaking water bath at 37° C for

107 2 h. After that, the homogenate was centrifuged at 10000 g for 20 minutes. The supernatant

108 was filtered and collected in a test tube. 1 mL of the extract was taken into a 25 mL

109 volumetric flask, then, 0.5 mL of 1 N Folin-Ciocalteu and 2.5 mL of 35% sodium carbonate

110 were added. The final volume was made up to 25 mL with distilled water. After vortex, the

111 flasks were left in dark conditions at room temperature for 40 minutes for colour

112 development. The intensity of purplish-blue colour was measured at 730 nm with

113 spectrophotometer. A standard curve, prepared with gallic acid, was used for the estimation

114 of total phenols content, and results were expressed as mg gallic acid equivalent (GAE) per g 115 dry leaf.

116 2.6. Steviol glycosides analysis

117 For the estimation of SGs content in stevia leaves, the collected leaf samples were dried at 118 $40\pm 2^{\circ}$ C in a hot air oven until constant weight was achieved. The dried leaf samples were 119 powdered by a grinder. 100 mg of prepared samples were immersed in 10 mL methanol for 120 24 h, then filtered. The filtrate was vacuum dried under reduced pressure, after that defatted 121 with n-hexane. Defatted sample was dissolved in 10 mL high-performance liquid 122 chromatography (HPLC) - grade acetonitrile and water (1:1) mobile phase, then filtered with 123 micro filter (0.22 µm pore diameter). The filtrate was used for the determination of important 124 eight steviol glycosides i.e. stevioside, Reb-A, Reb-F, Reb-C, dulcoside-A, rubusoside, Reb-125 B and steviolbioside (SB) by Ultra HPLC system (LC-MS Shimadzu, 2020 system). The

- system is equipped with LC-30 AD auto detector, quaternary pump, SIL-30 AC autosampler,
- 127 Degaser Unit DGU-20 A5R, reverse phase C18 column and SPD-M 30A diode array
- 128 detector. The temperature of column was set at 30° C. The used mobile phase was the
- 129 combination of A (5 mM ammonium acetate and 12 ppm equivalent v/v of formic acid) and
- 130 B (pure acetonitrile) with a ratio of 70: 30 and flowed at the rate of 0.24 mL min⁻¹ for 20
- 131 minutes. The detection wavelength was 210 nm. The steviol glycosides were quantified based
- 132 on the standard curves developed with the standard sample of stevioside, Reb-A, Reb-F, Reb-
- 133 C, dulcoside-A, rubusoside, Reb-B, and steviolbioside.

134 2.7. Determination of plant mineral ion in different plant organs

- 135 For understanding spatial distribution of N, P, K^+ , calcium (Ca²⁺), Na⁺ and Cl⁻ in different
- 136 plant parts, the dried leaf, stem and root samples were grounded in a grinder to pass through a
- 40 mesh screens. The total N content was estimated by micro-kjeldahl method after the
- 138 digestion of plant samples in concentrated sulphuric acid. For the estimation of other ions (P,
- 139 K^+ , Ca^{2+} , and Na^+), the samples were digested with a mixture of concentrated nitric acid,
- sulphuric acid and perchloric acid (9:4:1). The total P in leaf, stem and root samples was
- 141 determined with a spectrophotometer (model T 90+ UV/ vis, PG Instrument Ltd.), whereas a
- 142 flame photometer (model BWB XP, BWB technologies UK Ltd., UK) was used for the
- estimation of total K^+ , Na⁺, and Ca²⁺ as per the protocols recommended by Prasad *et al.*
- (2006). The Cl⁻ ion was analyzed by the titration method prescribed by Husband and Godden
 (1927).

146 2.8. Statistical analysis

147 The growth, yield and biochemical data obtained from this investigation for consecutive 2

148 years were subjected to analysis of variance (ANOVA) with the help of Statistica 7 software

- 149 (Stat Soft Inc., Tulsa, Oklahoma, USA) for estimating the variance components of main
- 150 effects (salinity level and foliar application of KNO₃) and their reciprocal interactions effects.
- 151 The least significant difference value was used to separate the treatment means when *F*-test
- used to was significant ($P \le 0.05$). A second-degree-polynomial regression model was used to
- establish the relations between salinity level and dry leaf yield, and foliar application of
- 154 KNO₃ and dry leaf yield of stevia. Agronomic traits, SGs profile, biochemical traits and ions
- accumulation were also subjected to the principal component analysis (PCA) to understand
- those, which were largely influenced by the treatment combinations.

157 **3. Results**

158 3.1. Yield attributes

159 The major yield attributes of the stevia viz. plant height, number of branches and LA were 160 significantly ($P \le 0.05$) influenced by the salinity levels and foliar spray of KNO₃ during both 161 the cropping seasons (**Table 2**). Averaged across the KNO₃ levels, plants treated with high 162 concentration of NaCl (\geq 80 mM) showed significantly ($P \leq 0.05$) lower plant height, number 163 of branches and leaf area as compared with the plants irrigated with the non-saline water as 164 control. Compared with control, irrigation of high saline water (NaCl at 120 mM) reduced 165 plant height by 10.86 and 12.58 %, number of branches by 24.26 and 31.29 % and LA by 20.90 and 20.89% in 1st and 2nd cropping seasons respectively. On the same time, application 166 of KNO₃ at 5.0 g L^{-1} registered maximum height (70.18 and 67.58 cm), number of branches 167 (7.17 and 7.33 No. plant⁻¹) and LA (390.92 and 407.66 cm²), which are significantly ($P \le$ 168 169 (0.05) different from the water spray as control in both years. The interaction effects between 170 salinity levels and KNO₃ on plant height and number of branch were insignificant ($P \ge 0.05$),

but significant ($P \le 0.05$) effect was observed in LA during both cropping years (**Table 2**).

172 3.2. Leaf, stem and root biomass yield

173 Averaged across the KNO₃ levels, yield (leaf, stem and root biomass) response to salinity 174 levels varied significantly ($P \le 0.05$) as salinity increased up to NaCl at 120 mM (**Table 3**) 175 during both growing seasons. Irrespective of KNO₃ application, the maximum dry leaf yield 176 of stevia (6.02 and 5.93 g plant⁻¹) was found with the plant irrigated with low salinity water (NaCl at 40 mM) but remained on par ($P \ge 0.05$) with control. In 1st season, the reductions in 177 178 dry leaf yield with high salinity level (NaCl at 120 mM) compared with control and low 179 salinity (NaCl at 40 mM) were 23.80 and 26.07 %, respectively. Irrespective of foliar application of KNO₃, plant irrigated with NaCl at 120 mM reduced root biomass significantly 180 $(P \le 0.05)$ compared with control by about 31 and 36 % in 1st and 2nd cropping season, 181 182 respectively. Averaged across of salinity levels, yield (leaf, stem and root biomass) of stevia 183 in response to foliar application of KNO₃ increased significantly ($P \le 0.05$) as the concentration increased up to at 5.0 g L^{-1} , thereafter decline trend was observed during both 184 growing seasons (**Table 3**). The increases in dry leaf yield with KNO₃ at 5.0 g L^{-1} over water 185

spray as control were 26.59 and 33.04 % in 1^{st} and 2^{nd} years, respectively.

187 Significant ($P \le 0.05$) interactions of salinity levels and KNO₃ application on dry leaf and

- stem yield and root biomass occurred in both seasons (Table 3). At control (irrigated with
- normal water) and low salinity level (NaCl at 40 mM), dry leaf, stem and root biomass
- 190 increased as concentration of KNO₃ increased from 0 to 5.0 g L^{-1} , while the reverse trend was
- 191 observed with further increases in KNO₃ concentration. However, at moderate salinity level
- 192 (NaCl at 80 mM), dry leaf yield was increased with corresponding increase in concentration
- 193 of KNO₃ up to 10.0 g L^{-1} (**Table 3**).

194 3.3. Chlorophyll (Chl) concentration in leaf

- 195 The photosynthetic pigments (Chl_a, Chl_b, total Chl_{a+b}, and Chl_a: Chl_b) in stevia leaf as
- influenced by saline levels and foliar application of KNO₃ are presented in **Fig1 a-d**.
- 197 Averaged over KNO₃ level, there were significant ($P \le 0.05$) differences among the salinity
- levels in concentration of Chl_a, Chl_b, total, and Chl, during both years. The concentrations of
- 199 photosynthetic pigments were gradually decreased at higher salinity level, irrespective of
- 200 foliar spray. Averaged across the salinity levels, the effects of foliar application of KNO₃ on
- 201 Chl_a, Chl_b and total Chl content in leaves were significant ($P \le 0.05$) in both years. The Chl_a
- and total Chl contents in leaves were increased with the increasing concentration of KNO_3 up
- 203 to 5.0 g L^{-1} in both seasons (**Fig. 1a, c**)..
- 204 The interaction effects between salinity levels and foliar application of KNO_3 on Chl_a and
- total Chl content in leaves were found significant ($P \le 0.05$) in both years (**Table 4**). At low
- salinity levels (control and NaCl at 40 mM), although total Chl contents were higher with
- foliar application of KNO₃ at 5.0 g L^{-1} than in water spray control in 2^{nd} year, but these
- differences were statistically insignificant ($P \ge 0.05$). On the other hand, the foliar application
- of KNO₃ at 5.0 g L⁻¹ registered significantly ($P \le 0.05$) higher total Chl content in leaves
- compared with water spray control (Table 4) at higher salinity level (NaCl at 80 and 120
- 211 mM).

212 3.4. Accumulation of total phenols and proline in leaf

Averaged over the KNO₃ levels, at 40 mM NaCl, total phenols (41.53 and 37.07 mg g^{-1})

content in leaves was 19.75 and 15.35 % higher than that of control (34.68 and 31.38 mg

- g^{-1} in 1st and 2nd cropping seasons, respectively (**Fig. 1e**). It has also been observed that a
- significant ($P \le 0.05$) decrease in total phenols content in leaves was occurred in parallel with
- further increase in NaCl concentration, and about 27 and 40 % decreases were registered in

the plants subjected to NaCl at 120 mM compared with the plants treated with NaCl at 40

mM in 1st and 2nd cropping seasons, respectively. Averaged over salinity level, the maximum value (48.29 mg g⁻¹) was observed with KNO₃ at 5.0 g L⁻¹, which was 36.22 % higher than control in 2016. In 2017, by increasing the concentration of KNO₃ from control to 7.5 g L⁻¹,

- the concentration of total phenols was increased from 28.97 to 36.35 mg g^{-1} . As NaCl
- 223 concentration was increased, in parallel accumulation of proline content was increased (Fig.
- 224 **2f**). Averaged across salinity level, there was a significant ($P \le 0.05$) difference in proline
- content between the leaves treated with water spray and higher concentration of KNO₃ (at 5.0
- and 7.5 g L^{-1}). At 7.5 g L^{-1} KNO₃, 31.99 % and 17.54 % increases were observed in proline
- content compared with control in 2016 and 2017, respectively. However, proline content was
- sharply decreased with KNO₃ at 10.0 g L^{-1} (**Fig. 1f**).

218

- Significant ($P \le 0.05$) interaction effects between salinity level and KNO₃ on total phenols
- and proline content in stevia leaves were occurred in both years (**Table 4**). In 1st season, total
- 231 phenols content was significantly increased with KNO_3 at 5.0 g L⁻¹ compared with water
- spray and higher concentration of KNO_3 (10.0 g L⁻¹) under non-saline (control) and low
- saline (NaCl at 40 mM) conditions. Proline content was not significantly ($P \ge 0.05$)
- influenced by the foliar application of KNO₃ at low salinity level (NaCl at 40 mM), whereas
- significantly ($P \le 0.05$) higher amount was recorded with KNO₃ at 10.0 g L⁻¹ compared with
- water spray under non-saline condition in 1^{st} year. On the other hand, at moderate salinity
- 237 level (NaCl at 80 mM), proline content was increased with the corresponding increase in
- 238 concentration of KNO₃ up to 7.5 g L^{-1} (**Table 4**).

239 3.5. Ions distribution in leaf, stem and root

- At harvest stage, ions (N, P, K^+ , Ca^{2+} , Na^+ and Cl^-) accumulation patterns in leaf, stem and
- root as influenced by different salinity levels and foliar applications of KNO₃ are presented in
- **Fig. 2 and 3**. Averaged over KNO₃ level, concentration of N in leaf was significantly ($P \le$
- 243 0.05) decreased from 19.18 and 14.37 mg g^{-1} to 17.36 and 11.8 mg g^{-1} at higher salinity level
- 244 (NaCl at 120 mM) in 2016 and 2017 cropping seasons, respectively (**Fig. 2a**). In root,
- significantly ($P \le 0.05$) higher amount (11.90 and 11.53 mg g⁻¹) of N was recorded with
- plants treated with NaCl at 80 mM compared with NaCl at 120 mM. Regardless of KNO₃
- level, the maximum K⁺ contents in leaf (24.40 and 24.75 mg g^{-1}), stem (25.04 and 23.74 mg
- 248 g^{-1}), and root (19.22 and 21.02 mg g^{-1}) were registered with the plants irrigated with low
- salinity water (NaCl at 40 mM); however, sharply decreased with NaCl at 120 mM during

both growing seasons(**Fig. 2e**). The Ca²⁺ uptake was also reduced significantly ($P \le 0.05$) in the plants treated with high concentration of NaCl (120 mM) (**Fig. 2g**).

252 Averaged over KNO₃ level, the effects of salinity level on the accumulation of Na⁺ and Cl⁻

in leaf, stem and root were found to be significant ($P \le 0.05$) in both years (Fig. 3a,c). As

254 NaCl concentration was increased, in parallel accumulations of Na⁺ and Cl⁻ in leaf, stem and

root were increased. Averaged across KNO₃ level the maximum and lowest values of K⁺: Na⁺

and Ca^{2+} : Na⁺ in all parts (except K⁺: Na⁺ in stem in 2016 and Ca²⁺: Na⁺ in leaf in 2017) were

registered with control (irrigated with non-saline water) and NaCl at 120 mM, respectively

258 (**Fig. 3e, g**).

259 On the other hand, pooled across salinity level, the accumulation patterns of N and Ca^{2+} in

leaf, stem and root were gradually increased with increasing the level of KNO₃ concentration

up to 5.0 g L^{-1} , thereafter decline trend was observed in both the cropping seasons (**Fig. 2b**,

b). On the other hand, pooled across all salinity environments, accumulation of K in leaf was

significantly ($P \le 0.05$) higher with KNO₃ at 5.0 g L⁻¹ compared with rest of the treatments in

both seasons (**Fig. 2f**). Pooled across the salinity level, accumulations of Na^+ and $C\Gamma$ in all

plant organs were sharply declined with corresponding increases of KNO₃ level (Fig. 3b, d).

266 On the other hand, the ratios of K^+ : Na⁺ in leaf, stem and root were influenced significantly

267 $(P \le 0.05)$ by the foliar application of KNO₃, and the lowest values were registered with

water spray as control, which were significantly ($P \le 0.05$) different from the application of

269 KNO₃ at 5.0 g L^{-1} and higher concentration (**Fig. 3f**). In some cases (particularly K⁺ and Na⁺

in leaf), the significant ($P \le 0.05$) interaction effects between salinity and KNO₃ level on ions

accumulation were also observed (Table 4).

272 3.6. Accumulation of Steviol glycosides (SGs):

273 Total eight SGs namely stevioside, Reb-A, Reb-F, Reb-C, dulcoside-A, rubusoside, Reb-B, 274 and steviolbioside were analyzed in this experiment to understand the influence of salinity 275 stress and foliar application of KNO₃. Here, most two prominent SGs, Stevioside and Reb-A, 276 and total SGs are presented in **Fig. 4a**, **b**. In case of total steviol glycosides (TSGs), the sum 277 of eight aforesaid mentioned SGs has been considered. The plants treated with low salinity 278 water (NaCl at 40 mM) accumulated significantly ($P \le 0.05$) higher amount of stevioside (48.43 and 39.35 mg g⁻¹ dry leaf) compared with the plants treated with moderate (NaCl at 80 279 280 mM) and high salinity (NaCl at 120 mM) water in both years (Fig. 4a). Reb-A content was 281 also significantly ($P \ge 0.05$) influenced by salinity level, and high salinity level (NaCl at 120)

mM) produced significant lower Reb-A (17.38 and 18.34 mg g^{-1}) compared with non-salinity

- treatment in both years (**Fig.4b**). In case of TSGs yield (g plant⁻¹), there was no significant (P
- ≥ 0.05) difference between low saline (NaCl at 40 mM) water treated plants and non-saline
- water treated plants in 1^{st} and 2^{nd} cropping season (**Fig.4d**).
- Averaged over salinity levels, there were no significant ($P \le 0.05$) differences among the
- 287 KNO₃ levels for accumulation of stevioside in 1st year. However, significant ($P \le 0.05$)
- differences among the KNO₃ levels were observed in 2^{nd} year, and the maximum (39.58 mg
- g^{-1}) value was recorded with KNO₃ at 5.0 g L⁻¹ (**Fig.4b**). Irrespective of salinity levels, the
- 290 maximum quantity of Reb-A (24.75 and 24.80 mg g^{-1}) was also recorded with KNO₃ at 5.0 g
- 291 L^{-1} , which was significantly (P \leq 0.05) higher than control during both years. Significant (P \leq
- 292 0.05) variations in the TSGs yield (g plant⁻¹) in response to different KNO₃ levels were found
- during both years, and the maximum yield $(0.52 \text{ and } 0.49 \text{ g plant}^{-1})$ was recorded with the
- application of KNO₃ at 5.0 g L^{-1} (**Fig. 4d**).
- 295 The interaction effects between salinity levels and foliar applications of KNO₃ on
- accumulation of stevioside, Reb-A and TSGs and TSGs yield (g $plant^{-1}$) were found to be
- significant ($P \le 0.05$) in both seasons (**Table 5**). Effect of KNO₃ on accumulation of
- stevioside was significant ($P \le 0.05$) under non-saline condition (control) in both years, and
- the maximum value was recorded with KNO₃ at 5.0 g L^{-1} (62.85 and 43.72 mg g⁻¹). Under
- 300 other salinity levels, accumulation patterns of stevioside due to foliar application of KNO₃
- 301 were found to be irregular.

302 3.7. Correlation, regression and principle component analysis

- 303 The statistical relationships among the agronomic traits (Plant height, number of branch, LA,
- 304 SLW, dry root, stem and leaf yield) were presented in a correlation matrix (**Table S1**). The
- data revealed that most of the agronomic traits were positively correlated with each other.
- 306 The relationship of dry leaf yield of stevia with the concentration of KNO₃ was best
- 307 described by the second-degree curve (**Fig.5**) with regression equations as given below:

308
$$Y_{(2016)} = 4.658 + 0.435^* X - 0.0422^* X^2 (R^2 = 0.890^*)$$

- 309 $Y_{(2017)} = 4.517 + 0.493X 0.0491 X^2 (R^2 = 0.738)$
- 310 The dry leaf yield was increased with the corresponding increase in concentration of KNO₃
- 311 up to 5.0 g L^{-1} , and thereafter trend was declined in both the years.
- 312 In this investigation, the principle component analysis (PCA) was conducted on six
- agronomic traits (plant height, branch, leaf area, dry root yield, dry leaf yield, dry stem yield)

and eleven chemical and biochemical parameters (total Chl, total phenols, proline, N, P, K⁺,

315 Ca^{2+} , Na⁺, Cl⁻, K: Na, Ca: Na) of stevia to understand the interaction effects of treatment

316 combination on these traits and their relationship (Fig. 6 a-d). The data revealed that the first

two principal components (PC_1 and PC_2) were cumulatively accounted for 80.32% and 80.07

318 % of the total variations for the 1^{st} and 2^{nd} year, respectively.

Another PCA was carried out on steviol glycosides (stevioside, Reb-A, Reb-F, Reb-C, SB,

TSGs) and biochemical traits (total Chl, total phenols, proline, N, P, K^+ , Ca^{2+} , Na^+ , Cl^-) to

study the relationships among them. The PCA showed that the 1^{st} and 2^{nd} component (PC₁

and PC₂) cumulatively explained 61.62% and 60.77% variations for 1^{st} and 2^{nd} cropping

323 years, respectively (Fig. 7 a-d). The relationships among the variables (different steviol

324 glycosides and biochemical traits) were presented in the space of the first two components

325 (PC₁ and PC₂). During 1^{st} year Reb-F, Reb-C, proline, Na⁺ and Cl⁻ were separated from rest

of variables by the PC_1 and placed in negative coordinate (Fig. 7a). However, the PCA bi-

327 plot (Fig. 7b) did not categorize the treatment combinations in any define clusters.

There were significant ($P \le 0.05$) differences in the soil pH and EC in response to different

salinity levels in both the cropping seasons (**Fig. S1**). he status of available P, K^+ , Ca^{2+} , Na^+

and Cl^{-1} ions in the soil after harvest were significantly (P ≤ 0.05) affected by the salinity

levels in both the cropping seasons, whereas available N content was significantly ($P \le 0.05$)

affected only in 2^{nd} cropping season (**Table S2**).

333 4. Discussion:

In salt-sensitive plant species, the growth inhibition is a common fact under saline conditions.

In this experiment, plant height, number of branches and leaf area were significantly ($P \le P$

336 0.05) reduced by NaCl at 120 mM. This result may be attributed to the accumulation of

337 excess amount of Na⁺ in the older leaves, concurrently it reduces the uptake of other essential

nutrients, such as potassium (K^+), calcium (Ca^{2+}), magnesium (Mg^{2+}) and nitrate (NO_3^-) ions

from the soil. Accumulation of excess amount of Na^+ in the older leaves inhibits growth by

340 accelerating their death consequently decreases the supply of carbohydrates to the

341 meristematic regions (Munns 2002). In the present experiment, accumulation of Na+ in the

leaves was increased by about 33- 40% with NaCl at 120 mM compared with control (Fig.

343 **3a**).

344 The reduction in dry leaf yield with high salinity level (NaCl at 120 mM) was mainly due to 345 the decrease of individual plant height, number of branch and LA per plant. Salinity 346 decreases the ability of a plant to take up water by inducing osmotic stress and thereby 347 reduces photosynthesis rate as a consequence of reduction of leaf expansion and closes 348 stomates (Munns, 2005; Rahnama et al. 2010; Deinlein et al. 2014). Moreover, low biomass 349 yield with high salinity level may be due to the fact that the high salinity induced oxidative 350 stress, which damages membrane lipids, proteins and nucleic acids, and ultimately cell death 351 (PérezLópez et al. 2009; Gill and Tuteja 2010; Del-Rio 2015). In our experiment, LA was 352 reduced drastically at high salinity (NaCl at 120 mM) conditions probably due to the 353 premature senescence of photosynthetic active leaves as a consequence of Na^+ and Cl^- ions 354 toxicity, which ultimately reduced the total dry leaf biomass. It has also been reported that 355 under salt-stress a specific Cl⁻ build-up is observed in the leaves, which triggers 1-356 aminocyclopropane-1-carboxilic acid (ACC) synthesis and its conversion to ethylene, and 357 eventually releases enough quantity of ethylene to hasten leaf abscission (Tudela and Primo-358 Millo 1992; Gómez-Cadenas et al. 1996, 1998; Dodd 2005).

359 The reduction of dry leaf biomass in the present experiment at high salinity stress (NaCl at

120 mM) can be due to the fact that the presence of higher concentration of NaCl in soils

reduces the uptake of potassium (K^+), calcium (Ca^{2+}), and nitrogen (**Fig 2a,e,g**). Restrictions

of N assimilation in plant body due to reduction of activity of cytosolic NADH nitrate

363 reductase enzyme in nitrate assimilation pathway have been reported under salinity stress

364 (Sivasankar and Oaks 1996; Jabeen and Ahmad 2011).

365 In the present study, the yield attributes and dry leaf biomass production of S. Rebaudiana 366 plants did not show any reduction at mild salinity level (40 mM NaCl) rather to some extent 367 increased compared with control (plants irrigated with non-saline water). These results could 368 be due to the fact that Na⁺ played important role as a functional nutrient or manovalent cation 369 in some physiological and metabolic activities and eventually hastened yield attributes of 370 stevia. The functional role Na⁺ in the stomatal physiology of some plants has already been 371 reported by Evans and Sorger (1966). Moreover, Cl⁻ accumulation in stevia plants at low 372 salinity level (40 mM NaCl) may be the suitable amount for functioning in stomatal 373 regulation and osmoregulation.

On the other hand, the rate of increase of dry leaf and stem yield with KNO₃ at 5.0 g L^{-1} might be to the due to availability of adequate amount of K⁺ and its counter ion NO₃⁻ for 376 different biological activities like photosynthesis rate and RuBP carboxylase activity

377 (Ramanujan and Rao 1971; Parasar and Dastane1973; Jabeen and Ahmad 2011). Other

378 probable explanation is higher accumulation of N in leaf, stem and root, which ultimately

induces cytokinin synthesis in root tips and maintains desirable cytokinin and auxin ratio. The

root cell division and differentiation of root are controlled by cytokinin and auxin ratio (Dello

381 Ioio *et al.* 2008), and the events of major cell specification during embryogenesis are

controlled by cytokinin and auxin (Muller and Sheen 2008). The positive effects of the N on

383 yield attributes and yield of stevia plant have also been reported by Pal *et al.* (2013, 2015).

384 The dry leaf yield did not show significant reduction under moderate salinity (NaCl at 80

385 mM) when KNO₃ at 10.0 g L^{-1} was applied compared with absolute control (plant irrigated

386 with non-saline water and without KNO₃). This result could be due to fact that exogenous

application of K+ improved plant water status and maintained ion balance in cytosol,

particularly K: Na and Ca: Na ratio. Higher K: Na ratio was observed with the exogenous

application of K+ compared with water spray under all the salinity levels. Potassium play a

390 key role in plant metabolic processes that improve plant water uptake by regulating the

391 osmotic potential and hydraulic conductivity of membranes under drought or salinity stress

392 (Heinen *et al.* 2009). Thus, it is evident from our investigation that exogenous application of

 K^+ improves salinity tolerance of stevia upto 80 mM NaCl.

Reduction of photosynthetic pigments at higher salinity, such as Chl_a , Chl_b and total Chl in

this experiment may be due to the accumulations of higher amount of Na^+ and Cl^- in leaf,

396 which disrupts the ultra structure of chloroplast and breaks down the Chl. Under salinity

397 stress conditions, Chl degradation is happened due to enhanced activity of chlorophyllase

enzyme (Santos 2004). However, some researcher has explained that Cl⁻ toxicity is the

primary reason for the degradation of chlorophyll in plants (Tavakkoli *et al.* 2010).

400 Chloroplasts also exhibit high permeability for Cl⁻ (Heber and Heldt 1981). On the other

401 hand, the low ratio of Chl_a : Chl_b with NaCl at 40 mM treatment in the present experiment

402 provides evidence that the photosynthetic pigments are not degraded at mild salinity level.

403 During the process of Chl degradation, Chl_a content is increased as a result of conversation

404 from Chl_b to Chl_a (Fang *et al.* 1998; Eckardt 2009). Higher total Chl with KNO₃ at 5.0 g L⁻¹

405 could be due to the fact that right concentration of KNO₃ increased the accumulation of total

406 N and K^+ in the leaf, and eventually increased the Chl content. K^+ checks the decomposition

407 of newly formed Chl and δ -aminolevulinic acid (ALA) formation (Tanaka and Tsuji 1980).

408 The organic osmolytes perform the vital role in maintaining low intracellular osmotic 409 potential of plants and in preventing the detrimental effects of salinity stress (Tarczynski et 410 al. 1993; Verslues et al. 2006). The utmost value of proline with NaCl at 120 mM could be 411 due to fact that stevia plant faces the osmotic stress under salinity condition and proline is 412 produced for osmotic adjustment. However, catabolism of proline is enhanced during 413 recovery (Szekely et al. 2008; Poonlaphdecha et al. 2012; Sharma and Verslues 2010), and 414 during this phase, proline regulates cell proliferation, cell death and expression of stress-415 recovery genes (Szabados and Savoure 2010). On the other hand, significantly ($P \le 0.05$) higher amount of proline was recorded with the foliar application KNO₃ at 5.0 and 7.5 g L^{-1} 416

417 compared with water spray in both the years.

418 It is well known fact that oxidative stress is occurred in salinity stress, mostly because of the 419 generation of excessive ROS (reactive oxygen species) within plant cells. In order to 420 scavenge or detoxify high ROS levels, plants produce phenolic compounds as an antioxidant 421 defense system (Gill and Tuteja 2010; Foyer and Noctor 2011; Petridis et al. 2012; Chawla et 422 al. 2013). Thus, biosynthesis of phenolic compounds is stimulated in salt-exposed plants 423 (Navarro et al. 2006; Bose et al. 2014). However, in the present experiment, the total phenols 424 content in stevia leaves was significantly increased ($P \le 0.05$) at mild salinity level (NaCl at 425 40 mM), whereas significant ($P \le 0.05$) reduction was observed at high salinity level (NaCl at 426 120 mM) compared with control (Fig. 2e). Maximum total phenols content with the foliar 427 application of moderate concentration of KNO₃ under low saline (NaCl at 40 mM) conditions 428 (**Table 4**) may be due to accumulation of higher amount of K^+ and balanced amount of others 429 ions present in plant tissues, which are responsible for the accumulation of higher total 430 phenols content. In the present experiment, higher amount of K^+ in leaf tissues has been 431 recorded with these treatment combinations. The growth strengthening antioxidant system in 432 plants is improved by K⁺ under salinity stress (Zheng *et al.* 2008).

433 Averaged across the KNO₃ levels, the plants treated with low salinity water (NaCl at 40 mM) 434 accumulated higher amount of stevioside, Reb-A and TSGs compared with the plants treated 435 with high salinity (NaCl at 120 mM) water and non-saline water in both the years. The 436 present results are in conformity with the findings of Zeng et al. (2013) and Cantabella et al. 437 (2017). Low SGs accumulation at severe salinity condition may be due to the fact that the 438 energy is utilized for the process of maintaining metabolic homeostasis. It has been reported 439 that when plants survive under high salinity stress conditions, energy is allocated for the 440 synthesis of simple osmolytes and enhancing activities of antioxidant enzymes (Abrol et al.

441 2012). The role of SGs as osmoprotectant molecules under stress has already been established 442 (Geuns and Ceunen 2013). On the other hand, irrespective of salinity levels, higher amounts 443 of stevioside, Reb-A and TSGs in leaves were recorded with the moderate concentration of 444 KNO₃. This result may be due to the fact that exogenous application of KNO₃ increases the accumulations of N, K^+ and Ca^{2+} in plant tissues, which help to develop membrane system of 445 chloroplasts and the content of photosynthetic pigments. Accumulation of steviol glycosides 446 447 in the cells of stevia plant is correlated with the development of the membrane system of 448 chloroplasts and the content of photosynthetic pigments (Ladygin et al. 2008). Moreover, 449 under optimum nutrient availability, carbohydrates are used to increase the amount of SGs

450 (Barbet-Massin *et al.* 2015).

451 Averaged over KNO₃ level, concentration of ions particularly N, K^+ and Ca^{2+} and ratio of K:

452 Na and Ca: Na in plant tissues were drastically reduced at high salinity level (NaCl at 120

453 mM), whereas reversed trends were observed in case of Na^+ and Cl^- . In the saline soil, Na^+

454 ion competes with K^+ for the transporter, since both have the common transport mechanism

455 (Sairam and Tyagi, 2004; Munns and Tester 2008) and similar in charge. However, at mild

salinity (NaCl at 40 mM) level, K⁺ content in leaf was higher in plant tissue, which suggested

that stevia could tolerate moderate salinity stress. This result also suggests that stevia plant

has the ability to do selective K^+ uptake under mild salinity level. Pooled across all salinity

459 environments, the accumulations of N, K^+ and Ca^{2+} in leaf, stem and root were higher with

460 the foliar application of KNO_3 at 5.0 gL⁻¹. However, K⁺ accumulation was declined at higher

- 461 concentration probably due to the destruction of ectodesmata structures (Marschner 1995).
- 462 Moreover, K^+ accumulation leaf tissue was increased with moderate concentration of KNO₃
- under all the salinity level. This may be due to fact that exogenous application of KNO₃

464 increased the K^+ accumulation to cope the adverse effect of salt stress.

465 Thus, it is concluded from the present investigation that the exogenous foliar application of

466 KNO_3 can elevate the salinity tolerance level of stevia through ion homeostasis, osmolyte

467 accumulation, and antioxidant metabolism mechanism.

468 Supplementary data

469 Table S1. Correlation coefficient matrix for agronomic traits of stevia

Table S2. Nutrients status in soil after harvest as influenced by salinity levels and foliar application of
KNO₃

- 472 Fig S1. Influence of irrigation water with different concentrations of NaCl on soil pH (a,b), EC (c,d)
- 473 and organic carbon (e,f).

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Figure legends

Fig.1. Variation in biochemical parameters i.e. (a) chlorophyll a, (b) chlorophyll b, (c) total chlorophyll, (d) chl a/ chl b, (e) total phenols and (f) proline in stevia leaves under different salinity and KNO₃ levels. S₀, S₁, S₂, and S₃ are representing salinity level (NaCl) of 0.0, 40, 80 and 120 mM, respectively, whereas as K₀, K₁, K₂, K₃, K₄ are representing the folia application of KNO₃ at 0.0, 2.5, 5.0, 7.5, and 10.0 g L⁻¹, respectively. Vertical bars represented the mean standard errors (\pm).

Fig. 2. Effects of salinity level and exogenous application of KNO_3 on accumulation of total nitrogen (a, b) phosphorus (c, d), potassium (e, f), and calcium (g, h), in the leaf, stem and root of *Stevia rebaudiana*. S₀, S₁, S₂, and S₃ represents as control, NaCl at 40, 80, and 120 mM, respectively. K₀, K₁, K₂, K₃, K₄ are representing the folia application of KNO₃ at 0.0, 2.5, 5.0, 7.5, and 10.0 g L⁻¹, respectively. The mean standard errors (±) are presented with vertical bars.

Fig. 3. Effects of salinity level and exogenous application of KNO₃ on accumulation of sodium (a, b) chlorine (c, d), K: Na (e, f), and Ca: Na (g, h), in the leaf, stem and root of *Stevia rebaudiana*. S₀, S₁, S₂, and S₃ represents as control, NaCl at 40, 80, and 120 mM, respectively. K₀, K₁, K₂, K₃, K₄ are representing the foliar application of KNO₃ at 0.0, 2.5, 5.0, 7.5, and 10.0 g L⁻¹, respectively. The mean standard errors (\pm) are presented with vertical bars.

Fig. 4. Steviol glycosides accumulation (mg g⁻¹) in leaf as influenced by salinity level and folia application of KNO₃ (a, b). Steviol glycosides yield are represented in g plant⁻¹ (c,d). S₀, S₁, S₂, and S₃ are representing salinity level (NaCl) of 0.0, 40, 80 and 120 mM, respectively, whereas as K₀, K₁, K₂, K₃, K₄ are representing the folia application of KNO₃ at 0.0, 2.5, 5.0, 7.5, and 10.0 g L⁻¹, respectively. The mean standard errors (±) are presented with vertical bars.

Fig. 5. Regression equation between dry leaf yield and salinity level (a). Another regression equation between dry leaf yield and concentration of KNO_3 (b).

Fig. 6. The multivariate analyses of agronomic traits (plant height, branch, leaf area, dry leaf, stem and root yield) and biochemical parameters [total chlorophyll (T-chl), total phenols (TP), proline (PL), N, P, K, Ca, Na, Cl, K: Na, Ca: Na] were conducted through Principal Component analysis (PCA). The first two factors (PC₁ and PC₂) mutually explained 80.32%

and 80.07 % of the total variations for the 1st and 2nd year, respectively (**Fig. 6a-d**). The PCA bi-plots (**Fig. 6b and 6d**) represent the distributions of treatment combinations. The proline (PL), Na and Cl were placed in the positive coordinate of PC₁ in 1st year (**Fig. 6a**) with loading values of 0.82, 0.78 and 0.77, respectively. Similarly, in second cropping season PL, Na, Ca and Cl were also separated by the PC₁ from rest of the agronomic traits and biochemical parameters, and placed in the positive coordinate of PC₁ and PC₂ (**Fig. 6c**). The PCA bi-plot (**Fig. 6b**) separated the treatments S_2K_4 (NaCl at 80 mM with the foliar application of KNO₃ at 10.0 g L⁻¹) and S_3K_4 (NaCl at 120 mM with foliar application of KNO₃ at 10.0 g L⁻¹) by PC₁ and PC₂, and placed in the positive coordinate of both PCs for 1st season. However, no distinct group was formed in the PCA bi-plot (**Fig. 6d**) for 2nd season.

Fig. 7. Principal Component analysis of secondary metabolites and biochemical traits. PC1 and PC₂ jointly explains 61.62% and 60.77% of the total initial variability of the data in the 1st (a, b) and 2nd (c, d) cropping years, respectively (Fig. 7 a-d). During 1st year Reb-F, Reb-C, proline, Na+ and Cl- were separated from rest of variables by the PC1 and placed in negative coordinate (Fig. 7a). However, the PCA bi-plot (Fig. 7b) did not categorize the treatment combinations in any define clusters. Only S₀K₁ (irrigation with normal water X foliar application of KNO₃ at 2.5 g L^{-1}) was separated by PC₁ and PC₂, and situated in the positive coordinate of both PCs. Like 1st cropping season, Reb-F, Reb-C, proline, Na and Cl were again separated by the PC₁ and placed in the negative coordinate (Fig. 7c) for 2^{nd} year. It was also revealed that the loading values three variables such as proline, Na and Cl were quite high with PC_1 . However, PCA bi-plot (Fig. 7d) separated the treatment combinations into two broad distinct clusters by PC1. Non-salinity (control) and low salinity (NaCl at 40 mM) treatments in combination with all foliar spray treatments (total 10 treatment combinations) were separated by PC_1 and placed in positive coordinate (Fig. 7d). In this group, two treatment combinations namely S_1K_2 (NaCl at 40 mM X KNO₃ at 5.0 g L⁻¹) and S_1K_3 (NaCl at 40 mM x KNO₃ at 7.5 g L⁻¹) were further separated from rest of the treatment combinations by both PCs, and placed in positive coordinate of both PCs.

Parameters		Year
T arameters	2016	2017
Sand (%)	42.4	42.7
Silt (%)	42.7	38.0
Clay (%)	17.3	19.3
Texture	Loam	Loam
Bulk density (g cm ⁻³)	1.25	1.17
pH (1:2)	6.72	6.31
EC ₂₅ (1:2)(ds.m ⁻¹)	0.11	0.10
OC (%)	1.53	1.66
N (mg kg ⁻¹)	64.87	122.73
$P(mg kg^{-1})$	14.96	13.21
K^+ (mg kg ⁻¹)	272.80	284.30
$Ca^{2+}(mg kg^{-1})$	925.00	939.80
$Na^+(mg kg^{-1})$	36.80	42.00
Cl ⁻ (mg kg ⁻¹)	3.54	3.12

Table 1. Physico-chemical properties of the experimental soil

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Salinity	Concentration	Plant height		Total branch			area	Specific leaf weight (mg cm ⁻²)		
level	of KNO ₃	(cm)		· 1	$\frac{(\text{No. plant}^{-1})}{2016}$		$\frac{1}{1}$			
(NaCl)		2016	2017	2016	2017	2016	2017	2016	2017	
Control	Control (K_0)	66.00	63.33	6.67	7.33	355.36	350.64	15.42	15.20	
(\mathbf{S}_0)	$2.5 \text{ g } \text{L}^{-1} (\text{K}_1)$	74.13	69.33	7.33	8.33	396.56	393.51	15.31	15.21	
	5.0 g L^{-1} (K ₂)	74.30	71.67	7.67	8.33	460.97	467.04	15.26	15.36	
	7.5 g L^{-1} (K ₃)	66.57	65.00	7.33	7.67	388.20	359.12	15.44	15.09	
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	63.30	62.00	6.67	6.67	325.05	312.85	14.31	13.93	
	Average	68.86	66.27	7.13	7.67	385.83	376.63	15.15	14.96	
40 mM	Control (K ₀)	65.20	60.67	7.00	7.33	357.11	341.10	15.45	15.52	
(\mathbf{S}_1)	2.5 g L^{-1} (K ₁)	76.80	73.67	7.67	8.67	367.12	418.45	16.97	15.75	
	5.0 g L^{-1} (K ₂)	77.13	77.00	8.33	8.67	428.09	488.98	16.52	15.23	
	7.5 g L^{-1} (K ₃)	68.37	64.67	7.00	8.00	440.02	367.36	13.81	15.21	
	$10.0 \text{ g } \text{L}^{-1}(\text{K}_4)$	63.57	62.67	6.67	6.67	372.92	331.35	14.04	14.34	
	Average	70.21	67.73	7.33	7.87	393.05	389.45	15.36	15.21	
80 mM	Control (K ₀)	57.03	55.33	5.33	4.67	268.22	271.29	14.90	14.97	
(S ₂)	2.5 g L^{-1} (K ₁)	63.33	59.67	6.33	6.00	328.79	313.63	13.96	13.43	
	5.0 g L ⁻¹ (K ₂)	63.93	61.33	7.00	6.67	344.77	339.18	14.24	14.35	
	7.5 g L ⁻¹ (K ₃)	64.17	62.33	7.33	7.00	347.81	325.74	14.27	15.43	
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	64.30	62.67	6.00	7.00	306.87	311.54	16.42	16.19	
	Average	62.55	60.27	6.40	6.27	319.29	312.27	14.76	14.88	
120 mM	Control (K ₀)	54.93	52.33	4.67	4.33	251.04	240.17	15.26	14.87	
(S ₃)	$2.5 \text{ g } \text{L}^{-1} (\text{K}_1)$	58.23	57.00	5.33	5.33	324.03	307.94	13.81	13.66	
	5.0 g L^{-1} (K ₂)	65.33	60.33	5.67	5.67	329.86	335.44	14.53	14.37	
	7.5 g L ⁻¹ (K ₃)	65.57	62.00	6.00	6.00	321.34	321.67	15.13	15.39	
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	62.83	58.00	5.33	5.00	297.36	284.54	14.46	14.86	
	Average	61.38	57.93	5.40	5.27	304.72	297.95	14.64	14.63	
Averages	Control (K ₀)	60.79	57.92	5.92	5.92	307.93	300.80	15.26	15.14	
across	2.5 g L^{-1} (K ₁)	68.13	64.92	6.67	7.08	354.13	358.38	15.01	14.51	
salinity	$5.0 \text{ g } \text{L}^{-1} (\text{K}_2)$	70.18	67.58	7.17	7.33	390.92	407.66	15.14	14.83	
level	7.5 g L^{-1} (K ₃)	66.17	63.50	6.92	7.17	374.34	343.47	14.66	15.28	
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	63.50	61.33	6.17	6.33	325.55	310.07	14.81	14.83	
SEm ± (Na	iCl)	1.29	1.12	0.27	0.33	4.11	3.99	0.35	0.45	
LSD ($P=0$.		3.70	3.20	0.78	0.93	11.79	11.45	NS	NS	
SEm ± (KN		1.44	1.25	0.30	0.36	4.60	4.46	0.39	0.50	
LSD $(P=0.$	- /	4.14	3.58	0.87	1.04	13.19	12.81	NS	NS	
$SEm \pm (S \times$	<i>,</i>	2.89	2.49	0.61	0.73	9.19	8.93	0.78	1.00	
LSD $(P=0.$,	NS	NS	NS	NS	26.37	25.61	NS	NS	
LSD $(I = 0.3)$, SA K										

Table 2. Effect of salinity and foliar application of KNO₃ on yield attributes of Stevia rebaudiana

Salinity Concentration		Dry leaf y		Dry stem y	ield (g	Dry root biomass (g		
level	of KNO ₃	plant	-1)	plant ⁻¹)	plant ⁻¹)		
(NaCl)	-	2016	2017	2016	2017	2016	2017	
Control	Control (K ₀)	5.47	5.33	6.35	5.82	1.84	2.05	
(S_0)	2.5 g L^{-1} (K ₁)	6.07	5.99	6.67	6.90	1.87	2.10	
	5.0 g L^{-1} (K ₂)	7.02	7.18	7.36	8.21	2.15	2.38	
	7.5 g L^{-1} (K ₃)	5.99	5.39	6.96	6.58	2.06	2.09	
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	4.66	4.37	5.67	5.21	1.64	1.51	
	Average	5.84	5.65	6.60	6.54	1.91	2.02	
40 mM	Control (K ₀)	5.50	5.30	6.23	5.72	1.72	1.77	
(S ₁)	2.5 g L^{-1} (K ₁)	6.21	6.55	7.29	7.65	2.16	2.41	
	5.0 g L^{-1} (K ₂)	7.07	7.45	7.89	8.64	2.46	2.61	
	$7.5 \text{ g } \text{L}^{-1} (\text{K}_3)$	6.09	5.57	6.95	6.76	1.84	2.03	
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	5.24	4.75	6.73	5.58	1.51	1.36	
	Average	6.02	5.93	7.02	6.87	1.94	2.03	
80 mM	Control (K ₀)	4.01	4.06	4.99	4.83	1.20	1.14	
(S ₂)	2.5 g L^{-1} (K ₁)	4.59	4.22	5.84	5.36	1.39	1.22	
	$5.0 \text{ g } \text{L}^{-1} (\text{K}_2)$	4.92	4.86	6.38	6.15	1.66	1.61	
	$7.5 \text{ g } \text{L}^{-1} (\text{K}_3)$	4.96	5.01	6.44	6.35	1.96	2.08	
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	5.02	5.05	5.50	5.77	1.99	2.09	
	Average	4.70	4.64	5.83	5.69	1.64	1.63	
120 mM	Control (K ₀)	3.83	3.57	4.75	4.22	1.09	1.09	
(S ₃)	2.5 g L^{-1} (K ₁)	4.48	4.21	5.11	4.76	1.32	1.25	
	5.0 g L^{-1} (K ₂)	4.79	4.81	5.63	5.65	1.49	1.29	
	$7.5 \text{ g } \text{L}^{-1} (\text{K}_3)$	4.86	4.95	5.70	5.73	1.62	1.69	
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	4.31	4.21	4.75	4.56	1.08	1.11	
	Average	4.45	4.35	5.19	4.98	1.32	1.29	
	Control (K ₀)	4.70	4.57	5.58	5.15	1.46	1.51	
Averages	2.5 g L^{-1} (K ₁)	5.34	5.24	6.23	6.17	1.69	1.74	
across	5.0 g L^{-1} (K ₂)	5.95	6.08	6.81	7.16	1.94	1.97	
salinity level	$7.5 \text{ g } \text{L}^{-1} (\text{K}_3)$	5.48	5.23	6.51	6.35	1.87	1.97	
level	$10.0 \text{ g L}^{-1}(\text{K}_4)$	4.81	4.60	5.66	5.28	1.56	1.52	
SEm ± (Na	SEm ± (NaCl)		0.15	0.08	0.13	0.06	0.08	
LSD (<i>P</i> =0		0.37	0.42	0.24	0.38	0.18	0.24	
$SEm \pm (KN)$		0.14	0.17	0.09	0.15	0.07	0.09	
LSD $(P=0$		0.41	0.47	0.27	0.43	0.21	0.26	
$SEm \pm (S \times ISD) (P=0)$		0.29	0.33	0.19	0.30	0.15	0.18	
LSD (P=0), 5 × K	0.82	0.95	0.53	0.86	0.41	0.53	

Table 3. Effect of salinity levels and foliar application of KNO₃ on dry leaf yield, dry stem yield and dry root biomass of *Stevia rebaudiana*

Salinity	Concentration		E	Biochemica	l parameter	Ior	n accumula	tion in leaf	f		
levels (NaCl)	of KNO ₃	Total Chl (mg g ⁻¹)			Proline (µg g ⁻¹)		henols g ⁻¹)	Na ⁺ (mg g ⁻¹)		$\frac{K^{+}}{(\text{mg g}^{-1})}$	
		2016	2017	2016	2017	2016	2017	2016	2017	2016	2017
Control (S ₀)	Control (K ₀)	2.73	2.94	23.48	29.07	32.25	28.79	1.30	1.00	22.32	22.80
	$2.5 \text{ g } \text{L}^{-1}(\text{K}_1)$	2.90	2.97	26.57	29.30	39.58	33.92	1.13	1.05	23.05	24.20
	5.0 g L ⁻¹ (K ₂)	2.93	3.12	27.90	30.89	46.21	35.00	1.07	1.00	25.23	25.57
	7.5 g L ⁻¹ (K ₃)	2.64	2.83	36.48	29.15	29.88	29.83	0.90	0.92	23.53	24.40
10 35 (7)	$10.0 \text{ g } \text{L}^{-1}(\text{K}_4)$	2.48	2.51	49.29	28.99	25.50	29.38	0.68	0.92	21.33	23.32
40 mM (S ₁)	Control (K_0)	2.81	2.69	36.10	38.26	38.04	28.13	1.57	1.32	23.92	23.28
	$2.5 \text{ g } \text{L}^{-1}(\text{K}_1)$	2.89	2.83	34.14	36.98	42.42	31.38	1.15	1.07	24.62	25.02
	$5.0 \text{ g } \text{L}^{-1}(\text{K}_2)$	2.79	2.87	35.10	39.26	68.17	43.38	1.15	1.02	26.63	26.98
	7.5 g L ⁻¹ (K ₃)	2.63	2.91	37.52	41.32	41.21	54.04	1.03	0.93	24.85	25.62
	10.0 g L ⁻¹ (K ₄)	1.83	2.72	45.14	41.63	17.83	28.42	0.85	0.95	21.97	22.83
80 mM (S ₂)	Control (K ₀)	1.94	1.66	122.90	111.38	39.50	37.63	1.62	1.48	16.22	16.20
	$2.5 \text{ g } \text{L}^{-1}(\text{K}_1)$	2.12	1.70	164.10	113.47	40.21	35.38	1.52	1.28	17.98	18.45
	$5.0 \text{ g } \text{L}^{-1} (\text{K}_2)$	2.43	2.06	165.10	111.36	45.79	36.79	1.23	1.17	18.67	19.08
	7.5 g L ⁻¹ (K ₃)	2.15	2.28	198.33	136.78	37.08	35.96	1.50	1.10	18.67	19.77
	10.0 g L ⁻¹ (K ₄)	1.33	1.36	118.43	106.47	24.38	18.54	0.97	1.05	18.05	19.65
120 mM (S ₃)	Control (K ₀)	1.49	1.55	177.76	127.79	32.00	21.33	2.07	1.62	15.50	12.53
	$2.5 \text{ g L}^{-1}(\text{K}_1)$	1.74	1.66	150.00	133.45	33.29	20.75	1.55	1.37	14.43	15.18
	$5.0 \text{ g } \text{L}^{-1} (\text{K}_2)$	1.92	2.08	174.81	155.27	33.00	24.17	1.35	1.30	16.77	16.50
	7.5 g L ⁻¹ (K ₃)	1.79	2.12	203.16	146.32	31.92	25.58	1.22	1.23	15.07	15.98
	10.0 g L ⁻¹ (K ₄)	1.32	1.16	122.33	112.56	22.29	19.83	0.95	1.20	13.17	13.48
	for $(S \times K)$	0.09	0.11	6.75	5.46	3.02	2.88	0.07	0.04	0.33	0.48
CD	(<i>P</i> =0.5)	0.26	0.32	19.36	15.67	8.66	8.25	0.19	0.12	0.95	1.39

Table 4. Interaction effects between salinity level and application of KNO3 on biochemical parameters and nutrient accumulation in leaf of Stevia rebaudiana

Salinity Level (NaCl)	Concentration of KNO ₃	Stevioside (mg g ⁻¹)		Rebaudioside-A (mg g ⁻¹)		TSGs (mg g ⁻¹)		TSGs yield (g plant ⁻¹)	
		2016	2017	2016	2017	2016	2017	2016	2017
Control	Control (K ₀)	30.15	31.14	15.39	21.07	70.11	71.83	0.37	0.38
(S_0)	2.5 g L^{-1} (K ₁)	32.63	41.37	43.33	22.37	109.40	85.40	0.66	0.52
	5.0 g L^{-1} (K ₂)	62.85	43.72	22.55	26.37	102.74	86.43	0.73	0.62
	7.5 g L ⁻¹ (K ₃)	60.32	29.16	17.23	19.66	85.71	63.15	0.51	0.34
	$10.0 \text{ g L}^{-1}(\text{K}_4)$	47.75	34.34	10.29	21.56	82.56	68.68	0.39	0.30
40 mM	Control (K ₀)	48.39	35.25	30.50	22.18	89.26	79.11	0.49	0.42
(S ₁)	2.5 g L^{-1} (K ₁)	67.08	37.97	20.19	17.94	93.91	68.41	0.58	0.45
	$5.0 \text{ g L}^{-1} (\text{K}_2)$	36.34	44.89	29.15	33.02	80.44	112.12	0.57	0.84
	7.5 g L ⁻¹ (K ₃)	37.91	48.53	23.21	22.61	70.65	86.77	0.43	0.48
	10.0 g L ⁻¹ (K ₄)	52.43	30.10	10.61	25.84	84.79	71.95	0.44	0.34
80 mM	Control (K ₀)	38.21	35.51	17.50	18.45	65.56	66.05	0.26	0.27
(S ₂)	2.5 g L^{-1} (K ₁)	38.54	34.72	22.08	17.71	81.00	70.34	0.37	0.30
	5.0 g L ⁻¹ (K ₂)	27.18	29.23	16.56	31.13	61.11	78.34	0.32	0.38
	7.5 g L ⁻¹ (K ₃)	40.12	27.39	31.78	29.64	82.16	74.35	0.41	0.3
	10.0 g L ⁻¹ (K ₄)	37.47	27.83	14.10	24.77	78.55	72.09	0.40	0.30
120 mM	Control (K ₀)	39.42	31.50	16.01	22.80	65.43	63.71	0.25	0.23
(S ₃)	2.5 g L ⁻¹ (K ₁)	23.55	34.76	9.23	26.54	43.22	74.61	0.19	0.3
	5.0 g L^{-1} (K ₂)	39.84	40.48	30.72	8.67	98.96	60.09	0.48	0.29
	7.5 g L ⁻¹ (K ₃)	20.51	21.97	8.16	20.96	58.33	62.42	0.28	0.3
	$10.0 \text{ g } \text{L}^{-1}(\text{K}_4)$	13.93	22.11	22.81	12.74	54.54	59.35	0.24	0.25
SEn	$h \pm \text{ for } (S \times K)$	3.55	3.05	2.76	1.46	6.95	3.68	0.05	0.0
С	D (P=0.5)	10.19	8.76	7.91	4.2	19.85	10.57	0.13	0.10

Table 5. The interaction effects between salinity levels and foliar application of KNO₃ on secondary metabolites profile in *Stevia rebaudiana*















