The interplay between molten globule intermediates and heme disassociation characterizes human hemoglobin disassembly

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Abstract

Background: Hemoglobin functions as an oxygen transport protein via its hetero-tetrameric structure, with each subunit containing an iron-cofactor heme. We have developed a global disassembly model for human hemoglobin, linking hemin (ferric heme) disassociation and apo (heme-free) subunits unfolding pathways.

Results: Model evaluation was done using circular dichroism and visible absorbance spectroscopy measurements of guanidine hydrochloride-induced holo (heme-bound) hemoglobin disassembly and our previous measurements of apohemoglobin unfolding. The populations of holo intermediates and equilibrium disassembly parameters were determined quantitatively for adult and fetal hemoglobins. Our results show that the key stages for disassembly into unfolded monomers are characterized by hemichrome intermediates with molten globule characteristics. Hemichromes, which occur when both hemin iron axial sites coordinate amino acids, are not energetically favored in native human hemoglobins. However, we have determined that these hexacoordinate iron complexes are important for preventing hemin disassociation from partially unfolded species during early disassembly. Both our model evaluation and independent small angle X-ray scattering measurements demonstrate that heme disassociation during early disassembly leads to loss of the tetrameric structural integrity.

Our model predicts that dimeric and monomeric hemichrome intermediates mediate the disassembly pathway inside red cells where the hemoglobin concentration is very high. This prediction explains why in the red cells of patients with unstable hemoglobinopathies misassembled hemoglobins often get trapped as hemichromes, accumulate into insoluble

Heinz bodies, become deposited at the cell membranes, and often lead to hemolysis.

Alternatively, when acellular hemoglobin is diluted into blood plasma after red cell lysis, the

disassembly pathway is dominated by early hemin disassociation from the unfolding

intermediates. This leads to the generation of higher fractions of apo subunits and free hemin,

which is known to damage to the integrity of blood vessel walls.

Conclusions: Our model demonstrates the existence of alternate sets of

disassembly/assembly pathways for human hemoglobin, illuminates the pathophysiology of

some hemoglobinopathies and other disease states associated with unstable globins and red

cell lysis, and provides insights into the factors governing hemoglobin assembly during

erythropoiesis.

Keywords: Hemoglobin, protein folding, heme binding, hemoglobinopathies, metalloprotein

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Introduction

An interwoven relationship between protein function, structure, and assembly occurs during hemoglobin assembly through step-wise hemin (Fe(II)-protoporphyrin IX) binding that guides the formation of secondary, tertiary, and quaternary protein structures (Figure 1). We depend on erythrocyte-encapsulated hemoglobin (Hb) to transport oxygen in our cardiovascular system [1]. Hb regulates cooperative binding of oxygen to each of its ferrous heme iron atoms via its quaternary conformations. Functional adult HbA consists of α and β globin subunits interacting at both the $\alpha_1\beta_1$ dimer and $\alpha_1\beta_2$ tetramer interfaces. Each globin subunit has 7 or 8 alpha helical segments labeled A to H, and is folded with a heme molecule sandwiched within a cavity between the E and F helices [2, 3].

Building on our earlier work [4], we have been able to establish a global model for human hemoglobin A (HbA) disassembly that involves reversible hemin and subunit disassociations at different disassembly stages (Figure 1). The concept that multiple assembly/disassembly pathways can occur for protein complexes formed by coupling between two protein subunits or between a protein subunit and a cofactor has been explored for other protein systems but not for hemoglobins [5-8]. Our model predicts that the main disassembly pathway for HbA is characterized by both types couplings and will differ markedly between intracellular red cell and acellular blood plasma environments as a result of marked differences in total protein (and heme) concentrations. In order to develop this model, we measured GdnHCl (guanidine hydrochloride) induced disassembly of holoHb at a wide range of protein concentrations. GdnHCl was chosen as a denaturant to avoid irreversible precipitation of hemin [9-11]. We then developed a multilayered analysis approach for Hb disassembly based on our recently published measurements and mechanism for GdnHCl induced unfolding of apoHb [4] and the new holoHb unfolding data generated in this study (Figure 1). Our model was used to analyze simultaneously circular dichroism (CD) measurements of secondary

structure changes in both apo- and holoproteins and populations of hemin-containing unfolding

species. The later populations were deconvoluted from visible absorbance measurements at

different stages of Hb disassembly. Equilibrium disassembly parameters extrapolated to 0 M

GdnHCl concentrations were obtained from fitting of our model describing these changes and

populations to the measured data.

Background

Fatal complications in patients with disassembled or misassembled hemoglobins can

result from hypoxia, hemolytic anemias, and iron toxicities due to free hemin release into the

blood stream [12-19]. The pathogeneses of genetic Hb misassembly and instability diseases

often originate from accelerated rates of heme loss, disruption of the inter-subunit interfaces,

and accumulation of unstable intermediates states occurring either on- or off-disassembly

pathways (Figure 1) [12, 17, 20].

The heme iron is axially coordinated at its proximal side to the $N\varepsilon2$ atom of the F8

histidine on the F helix. In the reduced state, this pentacoordinate iron can bind the key

physiological ligands O_2, CO, or NO axially at the sixth coordination position on the distal side

of the porphyrin ring [21]. Auto-oxidation of ferrous heme in oxyhemoglobin leads to aqua-

metHb, with H_2O bound to Fe(III) protoporphyrin IX (hemin). This metHb form has $\geq$10,000-fold

higher rates of heme disassociation than the corresponding reduced Hb complexes [22-24].

Thus, the first step in disassembly of hemoglobins involves auto-oxidation to the metHb form,

which is then followed by protein unfolding and hemin loss. Free heme itself is unstable in the

presence of oxygen and auto-oxidizes in milliseconds[21, 25]. Thus, during assembly it is the

oxidized form of heme that probably binds to nascent apo polypeptide chains before reduction

occurs. In our present study we have focused on the mechanisms and intermediates that occur

during aquametHb disassembly.

In their disassembly study of sperm whale holo myoglobin (Mb), which is structurally

similar to each Hb subunit, Culbertson and Olson [11] developed the strategy of expanding on

an apoprotein unfolding scheme to include steps involving hemin dissociation from holoprotein

species. Our overall goal was to apply their approach for monomeric Mb to examine

quantitatively the effects of hemin binding on the unfolding and disassembly of human holoHb,

which is a much more complicated oligomeric system with two types of subunits. We further

applied deconvolution methods on visible absorbance spectra to compute the populations of

native aquametHb, partially unfolded hemichrome intermediates, and free dissociated hemin in

varying mixtures of GdnHCl and holoHb samples. For folded Hbs, hemichromes or

hemochromes (ferrous hemichromes) are not native conformers [26]. Ferric hemichromes

form via bis-axial coordination of hemin iron to amino acids through combinations of $N\varepsilon$ atoms

from histidines, S atoms from either methionine or cysteine, and phenoxy O atoms from

tyrosine [26].

Specific roles and biophysical characteristics of the hemichromes that form during

different stages of Hb disassembly have not been well-characterized [27-30]. Our work shows,

for the first time, that (a) heterodimeric molten globule hemichromes play both a critical role for

$\alpha_1\beta_1$ dimer interface formation and dampening of hemin disassociation during early

disassembly stages; and (b) monomeric and dimeric molten globule hemichromes mediate the

major pathway for Hb disassembly at high Hb concentrations in erythrocytes. Our prediction of

a predominantly hemichrome mediated HbA disassembly pathway in erythrocytes helps to

explain why hemoglobinopathies have been often linked to accumulation of denatured

hemichrome precipitates that aggregate into inclusion bodies or Heinz bodies (Figure 1A) [12,

17, 20].

We also observed that a shift between apo- versus holo intermediate mediated Hb

disassembly could be modulated by mutations at the inter-subunit interfaces. To test of our

disassembly model, we examined fetal holohemoglobin (HbF) because of its intrinsic greater

resistance to denaturation relative to HbA [31]. In HbF, γ subunits are expressed in place of

the β subunits [1, 32], with amino acid residue differences at the dimer but not tetramer

interfaces. The populations of molten hemichrome intermediates during the disassembly of

holoHbF are significantly greater than those of holoHbA, and are explained by the increased

resistance of the $\alpha_1\gamma_1$ dimer interface to disassociation.

Results

Disassembly mechanism of holoHbA

In our model for Hb disassembly, heterotetrameric holoHb TH_4 initially disassociates

into two holoHb dimers (DH_2) without any helical content loss. In addition to undergoing hemin

loss leading to the apoHb dimer (D) and its subsequent unfolding (right sides of Figures 1A

and 1B), the holodimer (DH_2) can unfold into a dimeric holo molten globule intermediate (IH_2),

with a hemichrome absorbance spectrum. Each IH_2 can also further unfold and disassociate

into 2 holo monomers (UH). We obtained much better fitting to our spectral measurements

when we assigned hemichrome absorbance to the UH species. This assignment is supported

by previous experimental studies, which showed that isolated aquamet α and β monomers

form unstable hemichromes [29, 33]. Hemin disassociation from UH leads to formation of the

apo monomer U states, with residual helical content, and then the completely unfolded chains

at very high denaturant concentrations.

Complete chemical extraction of hemin from folded Hbs leads to a heterodimeric apoHb

(D) that has undergone 30% helical content loss[4, 34]. Our previous studies show that this D

state initially unfolds into a molten globule apo dimer (I), which has lost 70% of its original α-

helical content relative to holoHbA. Next, this molten globule apo I state disassociates into

unfolded monomers (U) with only residual helical content (≤ 10 %). These monomers then can

either interact non-specifically to form transient dimers (U_2), or completely unfold into

polypeptide chains (U_C) [4]. The TH_4 disassembly scheme in this present study expands the

apoHb model (Figure 1B, right side, gray section) described above to include the binding of

hemin (H) to the various D, I, and U states and the formation of hemin containing tetramers

(TH_4 and TH_2 states in Figure 1). We did not include the formation of tetrameric

apohemoglobins in our models because past studies have shown that native human apoHbs

can only exist as dimers, and not as tetramers [4, 35-37]. The accompanying loss of α-helical

content in apoHb due to hemin dissociation is predicted to be occurring around the heme

cavity region which also forms the tetramer ($\alpha_1\beta_2$) interface and thereby causing the tetramer

disassembly into dimers [4, 34].

Previous studies have shown that native metHbA undergoes non-cooperative hemin

disassociation with β subunits having a faster hemin loss rate and presumably lower hemin

affinity [22, 38-40]. Therefore, we also incorporated non-cooperative hemin binding into our

model (Figure 1) by taking into account sequential hemin disassociation from the DH_2 and IH_2

states. However, we made no assumptions about the relative hemin affinities of the two

different types of subunits and instead assigned equilibrium constants for each sequential

dissociation step. In our model fitting, differences between the subunits would be manifested in

the extent of the differences between the first and second hemin dissociation or association

equilibrium constants.

Thus in our model hemin (H) disassociates first from one subunit in DH_2, resulting in the

dimeric semi-holoHb (DH) formation, and then hemin disassociation from DH leads to the

apoHb D state. Similarly for IH_2, hemin disassociation occur first from one subunit leading to

initial formation of semi-hemichrome (IH) and then the next dissociation leads to the formation

of apo molten globule (I). Past studies have suggested the semi-holohemoglobin formation

occurs by either hemin disassociation from the β subunits in holo tetramers/holo dimers or

alternatively by hemin binding to the α subunits first in the apoHbs [39, 41].

Given the weak affinity of hemin for unfolded proteins [9, 11, 42, 43], the lack of

structural evidence about unfolded states, and the limitations of spectroscopic measurements,

we assumed no differences in hemin affinity between α versus β variants of U monomers

($K_{U,UH}$) (Figure 1). Non-specific hemin binding to the U_2 and U_C could not be accurately defined

because only free hemin visible absorbance spectra could be observed after the loss of all

helical content in the unfolded monomers.

All of these reversible disassembly stages described in this section were incorporated

into our overall scheme and are represented by Model 1 (blue arrows) in Figure 1. We also

considered an additional step involving semi-holo tetramer formation, which was incorporated

into Model 2 (both blue and black arrows). In this second model, TH_4 undergoes disassociation

of two hemin molecules initially from the subunit type with lower hemin affinity to form the

tetrameric semi- holoHb (TH_2).

Spectroscopic measurements of holoHb disassembly

GdnHCl-induced disassembly of holo- aquametHbA (Figures 2 and 3) and HbF (Figure

4) was examined as function of total protein concentration by measuring simultaneously both

heme visible absorbance and protein CD spectral changes. With increasing denaturant

concentration, holoHb unfolds and undergoes both hemin disassociation (Figures 2, 3C, and

4C) and loss of α-helical content (Figures 3D and 4D). CD signal changes at 222 nm were

used as a measure of the fraction of α-helical content present in the protein relative to that of

folded holoHb (Figures 3D and 4D [4]). The changes in hemin absorbance spectra were used

to compute populations of the various holo species during disassembly and free hemin (H).

As holoHb disassembles with increasing GdnHCl concentration, we observed dramatic

shifts in the heme visible absorbance spectra as shown in Figure 2A. Initially, at low [GdnHCl],

the observed visible absorbance spectra were characterized by features that are unique to

high-spin aquametHb (Figure 2B), including a sharp, dominant Soret absorbance peak at 405

nm and additional smaller peaks at 500 nm and 630 nm. At high [GdnHCl] corresponding to

when only residual (≤ 10%) or no alpha-helical content remains in the protein (Figures 2A and

3D), the absorbance spectra were characterized by features that were unique for high-spin

free hemin in free solution, including a broad Soret absorbance peak around 370 nm and

additional smaller peaks around 510 nm and 630 nm (Figure 2B) [11, 21, 44-46]. During

intermediate stages of Hb disassembly, another distinct absorbance spectrum appears for both

HbA and HbF and is indicative of a low-spin hemichrome (Figures 2, 3C, and 4C).

Hemichrome absorbance spectra are characterized by a distinct and narrow red-shifted Soret

absorbance peak around 413 nm and Q bands or visible region peaks at 535 nm and 565 nm

(Figure 2B)[11, 26, 47, 48].

As described in Methods, we deconvoluted the heterogeneous visible absorbance

spectral curves obtained for GdnHCl induced Hb disassembly using the three component

spectra for native aquamet, hemichrome intermediates, and free hemin shown in Figure 2B.

The differing population fractions of these species were computed as the GdnHCl titrations

proceeded, and the results for aquametHbA are shown in Panels 3A, 3B, and 3C and for

aquametHbF in Panels 4A, 4B, and 4C. The hemichrome population reaches a maxima at the

beginning of the 2nd phase of unfolding, corresponding to when the protein has lost ~60 % of

its alpha-helical content and forms a molten globule (Figures 2A, 3B, 3D, 4B, and 4D).

Hemichrome mediated resistance to hemin disassociation and subunit interface

disassembly

As shown for HbA in Figure 3D, the CD curves shifted towards higher [GdnHCl] in the

presence of bound hemin in both HbA and HbF (see also [4]). This shift is indicative of the

higher stability of holoHb relative to apoHb [4]. Previously, we had determined that during

apoHb unfolding, the 1st phase likely involves the unfolding of the heme pocket while the 2nd

phase corresponds to the disassembly of the heterodimeric molten globule [4]. For apoHbs,

the protein concentration dependency of the 2nd phase of unfolding demonstrated the

disassociation of the $\alpha_1\beta_1$ and $\alpha_1\gamma_1$ dimeric molten globules into unfolded monomers [4]. The

same interpretation applies for the unfolding curves of holo- HbA and HbF in this study. Again,

increasing protein concentration shifts the second phase for loss of α-helical content to higher

[GdnHCl] (Figures 3D and 4D). Similarly, the loss of hemichrome populations (Figures 3B and

4B) and the appearance of free hemin (Figures 3C and 4C) are shifted towards higher

[GdnHCl], demonstrating that there is greater resistance to hemin disassociation with

increasing protein concentration. These CD and visible absorbance analyses together

demonstrate that the bound hemin in the hemichrome state strengthens the α1β1 and the α1γ1

interfaces in the molten globules.

In our previous studies on apoHb unfolding, we determined that the $\alpha_1\gamma_1$ dimer interface

in HbF is more resistant to disassembly relative to HbA, with a monomer to dimer association

constant (K_{UI}) of 8.4×10^9 M^{-1} whereas the $\alpha_1\beta_1$ interface in HbA has about a 10-fold lower K_{UI} at

6.6×10^8 M^{-1} [4] (Table 1). Correspondingly, the HbF hemichrome population showed a higher

resistance to hemin disassociation during the 2nd phase of unfolding (Figures 5B and

Supplement 1B). Similarly, rHb0.1, a recombinant variant of HbA with a genetic cross-linker

between the α subunits, is also observed to have a larger hemichrome population at high

[GdnHCl] than both HbA and HbF (Figures 5B and Supplement 1B). The covalently

crosslinked aporHb0.1 tetramer has been previously shown to be more resistant to

disassembly during the second phase than either HbA or HbF apo dimers because two $\alpha1\beta1$

interfaces have to be disrupted in rHb0.1 to generate 2 unfolded β monomers and a di-α

subunit [4].

Table 1. Fitted equilibrium disassembly parameters for HbA and HbF

	HbA	HbF
$K_{2,4}$ (M^{-1})	1.0×10^5	1.0×10^5
$K_{DH,DH2}$ (M^{-1})	1.3×10^{10}	1.3×10^{11}
$K_{D,DH}$ (M^{-1})	1.3×10^{11}	2.5×10^{11}
$K_{IH,IH2}$ (M^{-1})	3.3×10^8	1.4×10^9
$K_{I,IH}$ (M^{-1})	1.1×10^{11}	7.8×10^{11}
$K_{U,UH}$ (M^{-1})	2.9×10^7	8.4×10^7
$m_{2,4}$ (kJ mol^{-1} M^{-1})	20.0	20.0
$m_{DH,DH2}$ (kJ mol^{-1} M^{-1})	10.3	9.9
$m_{D,DH}$ (kJ mol^{-1} M^{-1})	10.3	9.9
$m_{IH,IH2}$ (kJ mol^{-1} M^{-1})	10.3	10.7
$m_{I,IH}$ (kJ mol^{-1} M^{-1})	10.3	10.7
$m_{U,UH}$ (kJ mol^{-1} M^{-1})	5.1	5.2
S_{TH4}	1.00	1.00
S_{DH2}	1.00	1.00
S_{DH}	0.87	0.87
S_{IH2}	0.43	0.59
S_{IH}	0.42	0.34
S_{UH}	0.20	0.30
$K_{I,D}$ [a]	164	128
$K_{U,I}$ [a] (M^{-1})	6.6×10^8	8.4×10^9
$K_{U2,U}$ [a] (M)	5.6×10^{-7}	1.7×10^{-8}
$K_{UC,U}$ [a]	1190	921
$m_{I,D}$ [a] (kJ mol^{-1} M^{-1})	16.2	17.5
$m_{U,I}$ [a] (kJ mol^{-1} M^{-1})	11.6	12.7
$m_{U2,U}$ [a] (kJ mol^{-1} M^{-1})	-3.9	-6.8
$m_{UC,U}$ [a] (kJ mol^{-1} M^{-1})	5.4	5.4
S_D [a]	0.70	0.73
S_I [a]	0.30	0.33
S_U [a]	0.062	0.08
S_{U2} [a]	0.015	0.02
S_{UC} [a]	0.0034	0.006

265 The disassembly parameters as a factor of hemin were fitted according to model 1.
266 [a]Parameters for the apoHb part of the disassembly had been determined in our previously
267 published work[4]
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For both HbA and HbF, protein concentration dependency was not observed during the

1st phase of unfolding either in the CD curves or the population fractions of the heme spectral

species (Figures 3C, 3D, 4C, and 4D). Hemichromes start to appear during the 1st phase and,

in fact, seem to be playing a key role in reducing hemin loss during the early disassembly

stages for both HbA and HbF (i.e. little free hemin is present until [GdnHCl] ≥ 1.5 to 2 M). If

significant hemin disassociation had been occurring from folded metHb or the newly formed

hemichromes, we should have observed population shifts in the curves for native aquametHb

and changes in the first phase for the appearance of free hemin as protein concentration

increased (Figures 3A,C and 4A,C). Instead, conformational changes around the heme pocket

region during the 1st phase promoted a high-spin metHb to low-spin hemichrome transition with

heme still bound in the protein. This conformation change is likely induced by the loss of α-

helicity around the heme pocket region similar to what occurs during the 1st phase of apoHb

unfolding [4].

We initially expected to see a dependence on protein concentration for the initial phase

of the CD change due to holo tetramer to holo dimer dissociation. However, none was

observed due to the large tetramer to dimer dissociation constant of aquametHb, which is ~10

µM [49]. The threshold of the CD photomultiplier limited the range of holoprotein

concentrations to be examined. Within the smaller range that we could examine (1 to ~60 µM

total heme), our modeling (Figure 1) predicted that the tetramer to dimer disassembly was

occurring at very low denaturant concentrations before any significant heme pocket unfolding

or hemin dissociation occurred. However, if we could have carried out experiments in the

millimolar hemoglobin concentration region, our model does predict that we would observe

protein concentration effects during the first phases of disassembly (see the results section,

"*Modeling HbA disassembly in erythrocytes and in blood plasma*").

The E7 histidine is the nearest amino acid that can coordinate with the hemin iron at the

distal side of the porphyrin ring in Hb. His-E7 often stabilizes exogenous ligands bound to

heme iron through hydrogen bonding with the $N\varepsilon2$ atom. However, the $N\varepsilon2$ atom is still at a

non-bonding distance of ~4.1 Å from the hemin iron in native metHbA, and a water molecule is

coordinated instead (see PDB 3P5Q [50] and [26]). Although we assume the major

hemichrome species during native HbA disassembly is most likely due to hemin iron

hexacoordination to both the F8 and E7 histidines, to the best of our knowledge, no direct

structural evidence for this assumption exists.

Different pathways for hemichrome formation are almost certainly occurring because

hemichrome spectra are still observed for the recombinant metHb α(H58L/V62F)β(H63L/V67F)

quadruple mutant, in which both distal histidines were replaced with leucines (Figure 5B).

However, the population fraction of the hemichrome intermediate formed during disassembly

of the mutant is roughly half that observed for native HbA, implying less stable

hexacoordination (Figure 5B). In our recent apoHb unfolding studies, the mutant apoHb

showed increased resistance to unfolding during the 1st phase due to the introduction of the

large and apolar residues in the heme cavity[4]. However, the mutant also showed increased

hemin disassociation during the 1st phase of disassembly (Figures 5A and Supplement 1B).

Past studies of His-E7 to Leu mutations in both Hbs and Mbs attributed the marked increases

observed in rates of hemin dissociation to loss of hydrogen bonding between the E7 side chain

and the coordinated H_2O molecule [51, 52]. Our disassembly results now suggest that the role

of hemichrome formation should also be considered when determining the rates of net hemin

dissociation since these species are clearly intermediates in holoHb unfolding.

Deriving disassembly parameters

The assembly constants defined in Figure 1B and listed in Table 1 were obtained by

extrapolation to 0 M GdnHCl concentration or x, and are represented by the K^0 values in eqs 1

and 2. This extrapolation was done using the dependence of the K constants on [GdnHCl] or x

as shown in eqs 1 and 2. The apoHb folding constants in Table 1 and eq 2 were originally

derived in our earlier work [4]. The m values describe the dependence of the assembly free

energies on [GdnHCl] and absolute temperature (T). The K parameters containing H

subscripts in eq 1 define hemin or H binding to the various apoglobin intermediates in Figure

1B, and $K_{2,4}$ defines the holo metHb dimer to tetramer association equilibrium constant.

$$K_{2,4} = \frac{[TH_4]}{[DH_2]^2} = K_{2,4}^0 \exp\left(\frac{-m_{2,4}x}{RT}\right)$$

$$K_{DH,DH2} = \frac{[DH_2]}{[DH][H]} = K_{DH,DH2}^0 \exp\left(\frac{-m_{DH,DH2}x}{RT}\right)$$

$$K_{D,DH} = \frac{[DH]}{[D][H]} = K_{D,DH}^0 \exp\left(\frac{-m_{D,DH}x}{RT}\right)$$

326    $$K_{IH,IH2} = \frac{[IH_2]}{[IH][H]} = K_{IH,IH2}^0 \exp\left(\frac{-m_{IH,IH2}x}{RT}\right) \qquad (1)$$

$$K_{I,IH} = \frac{[IH]}{[I][H]} = K_{I,IH}^0 \exp\left(\frac{-m_{I,IH}x}{RT}\right)$$

$$K_{U,UH} = \frac{[UH]}{[U][H]} = K_{I,IH}^0 \exp\left(\frac{-m_{U,UH}x}{RT}\right)$$

$$K_{TH4,TH2} = \frac{[TH_2][H]^2}{TH_4} = (K_{TH4,TH2}^0)\exp\left(\frac{m_{TH4,TH2}x}{RT}\right)$$

For our overall Hb disassembly/assembly scheme described in model 1 (Figure 1), the

theoretical values of total protein concentration [P] on per subunit basis and total free hemin

concentration [Free hemin] in solution were computed from the concentrations of the various

species in eqs 3 and 4, respectively. This theoretical [P] value should equate to y, the

experimentally determined per subunit Hb concentration. Because there is no excess hemin in

the original solutions, [free hemin] in eq 4 will equate to [H] in eq 1. [Free hemin] in eq 4 is

computed from the sum of the concentrations of the apo subunits in all the apo- and semi-holo

species. For the alternative model 2 that included semi-holo tetramers, [P] and [free hemin]

were re-derived with the addition of the terms $4[TH_2]$ and $2[TH_2]$ to the right sides of eqs 3 and

4 respectively.

$$K_{I,D} = \frac{[D]}{[I]} = K^0_{I,D} \exp\left(\frac{-m_{I,D}x}{RT}\right)$$

$$K_{U,I} = \frac{[I]}{[U]^2} = K^0_{U,I} \exp\left(\frac{-m_{U,I}x}{RT}\right)$$

339   (2)

$$K_{UC,U} = \frac{[U]}{[U_C]} = K^0_{UC,U} \exp\left(\frac{-m_{UC,U}x}{RT}\right)$$

$$K_{U2,U} = \frac{[U]^2}{[U2]} = K^0_{U2,U} \exp\left(\frac{-m_{U2,U}x}{RT}\right)$$

$[P] = 4[TH_4] + 2[DH_2] + 2[DH] + 2[IH_2] + 2[IH] + [UH] + 2[D] + 2[I] + [U] + [U_C] + 2[U_2]$ (3)

$[\text{Free hemin}] = [DH] + [IH] + 2[D] + 2[I] + [U] + [U_C] + 2[U_2]$ (4)

The general population fraction Y_i on per subunit basis for a specific folding species (i)

is described in eq 5. In eq 5, specific examples are also given for the free hemin fraction (Y_H)

and holoHb tetramer population fraction (Y_{TH4}) from model 1. The total population fraction of

folded holoHbs, encompassing both the holoHb dimers and tetramers, is defined as Y_{Folded}

$_{holoHb}$ $=Y_{TH4} + Y_{DH2} + 0.5Y_{DH}$ for model 1, and $Y_{Folded\ holoHb}= Y_{TH4} + Y_{DH2} + 0.5(Y_{TH2} + Y_{DH2})$ for

model 2. In both models, the total population fraction of hemichromes was evaluated as

$Y_{Hemichromes} = Y_{IH2} + 0.5Y_{IH} + Y_{UH.}$

The theoretical $Y_{Folded\ holoHb}$ value was assumed to correspond to the population fraction

of native metHb spectral species (red spectrum in Figure 2B) derived experimentally from the

deconvolution of the measured visible spectra as a function of [GdnHCl]. $Y_{Hemichrome}$ and Y_H

were assumed to correspond respectively to the fractions of the low spin hemichrome

spectrum (blue spectrum in Figure 2B) and high spin free hemin spectrum (green spectrum in

Figure 2B) in our measured visible absorbance data. A weight of 0.5 was assigned to the semi-

holoHb Y values in the expressions of $Y_{Hemichrome}$ and $Y_{Folded\ holoHb}$ to account for the subunits

present that already had undergone hemin disassociation. The theoretical normalized total CD

signal, S_{Total}, is a measure of fractional alpha-helicity of protein at 222 nm relative to folded

holoHb. S_{Total} was defined as the sum of the product of Y_i and intrinsic normalized CD signal S_i

for each folding species (not including H) as shown in eq 6. S_i is independent of x and y and is

defined as the fraction of α-helical content of species i relative to folded holoHb.

$$Y_i = \frac{<\text{Number of subunits per i}> [i]}{P}$$

e.g. for Model 1

364    $$Y_H = \frac{[H]}{4[TH_4] + 2[DH_2] + 2[DH] + 2[IH_2] + 2[IH] + [UH] + 2[D] + 2[I] + [U] + [UC] + 2[U_2]}$$ (5)

$$Y_{TH4} = \frac{4[TH_4]}{4[TH_4] + 2[DH_2] + 2[DH] + 2[IH_2] + 2[IH] + [UH] + 2[D] + 2[I] + [U] + [UC] + 2[U_2]}$$

$$S_{Total} = \sum S_i Y_i$$

e.g. for Model 1,

$$S_{Total} = S_{TH4}Y_{TH4} + S_{DH2}Y_{DH2} + S_{DH}Y_{DH} + S_{IH2}Y_{IH2} + S_{IH}Y_{IH} + S_{UH}Y_{UH} \ldots$$

$$+ S_D Y_D + S_I Y_I + S_U Y_U + S_{UC}Y_{UC} + S_{U2}Y_{U2}$$

(6)

The K, m, and S_i values characterizing holoHb unfolding were estimated by global fitting

of (a) specific Y_i or sum of Y_i values to corresponding native metHb, hemichrome, and free

hemin populations obtained from deconvolution of the visible spectral data; and (b) S_{Total} to the

experimental CD values that were normalized against the α-helical content of folded holoHb

(Figures 3, 4, and Table 1). The fits to these experimental measurements were done as a

function of x (GdnHCl concentration) and y (total protein concentration) variables for both

holoHbA and holoHbF. The K, m, and Si values for the apoHb unfolding steps were fixed to the

values determined from our previous, independent apoHb unfolding studies [4, 11]. The m

values for the association constants for hemin binding to the folded heme pockets were set

equal to each other (e.g. $m_{DH,DH2} = m_{D,DH}$) and the m values for the parameters describing

hemin binding to the molten dimers were also set equal to each other (e.g. $m_{IH,IH2} = m_{I,IH}$)

(Table 1, Supplement Table 1).

The $K_{2,4}$ value for metHb dimer-dimer association was initially set to 1×10^5 M^{-1} for

metHbA based on previous studies, and since there is no protein sequence difference at the

tetramer interface between HbA and HbF, the same value was used for HbF [49]. This number

did not change much during the model fittings. The lack of change was likely due to: (a) the

absolute value of $K_{2,4}$ being poorly defined but in the 1×10^5 M^{-1} range; and (b) the tetramer

dissociation being complete at very low denaturant concentrations ([GdnHCl] $\leq$ 0.5 M).

The Y_i values had to be computed first by defining them in terms of the K constants at

different x and y values (left sides of eqs 1 and 2) for comparison with the experimental data.

For this purpose, [U] and [H] values at a given x and y were obtained by numerically applying

Newton's method [53] for systems of non-linear equations (eqs 7-8) expanded from the

definitions of [P] (e.g. eq 3) and [Free hemin] (e.g. eq 4). As shown for Model 1 in eqs 7-8, all

species were expanded only in terms of [U], [H], and K values. These initial values of [U] and

[H] were used to compute new values of Y_i.

Starting from these initial estimates of Y_i, lsqcurvefit function (a non-linear least-squares

solver) in MATLAB R2017a (The MathWorks, Inc., Natick, MA) was applied to globally fit the

theoretical models to the experimental measurements as shown in Figures 3, 4, and S2. An

algorithm was written in MATLAB R2017a (The MathWorks, Inc., Natick, MA) to apply the

numerical analysis method and fitting function in an iterative loop where the K, m, and S_i

parameters updated from each optimization loop was fed back to the numerical analysis part of

the algorithm to obtain a new sets of [U] and [H] values. The implementation of the loop is

stopped when the best fit between the experiment and model is reached as defined by a

minimum steady state value of $\chi 2$, the squared residual difference between the models and

experimental measurements.

$$[P] - y = 0$$

For Model 1,

$$4[TH_4] + 2[DH_2] + 2[IH_2] + 2[DH] + 2[IH] + [UH] + 2[D] + 2[I] + [U] + [U_C] + 2[U_2] - y = 0$$

$$4K_{TH4}(K_{DH,DH2}K_{D,DH}K_{I,D}K_{U,I})^2[U]^4[H]^4 + 2K_{DH,DH2}K_{D,DH}K_{I,D}K_{U,I}[U]^2[H]^2 + ...$$

$$2K_{IH,IH2}K_{I,IH}K_{U,I}[U]^2[H]^2 + 2K_{D,DH}K_{I,D}K_{U,I}[U]^2[H] + 2K_{I,IH}K_{U,I}[U]^2[H] + ...$$

$$K_{U,UH}[U][H] + 2K_{I,D}K_{U,I}[U]^2 + 2K_{U,I}[U]^2 + [U] + \frac{[U]}{K_{UC,U}} + 2\frac{[U]^2}{K_{U2,U}} - y = 0$$

(7)

407 $[\text{Free hemin}] - [H] = 0$
For Model 1,
$[DH]+[IH]+2[D]+2[I]+[U]+[U_C]+2[U_2]-[H]=0$ (8)
$K_{D,DH}K_{I,D}K_{U,I}[U]^2[H]+K_{I,IH}K_{U,I}[U]^2[H]+...$

$2K_{I,D}K_{U,I}[U]^2+2K_{U,I}[U]^2+[U]+\dfrac{[U]}{K_{UC,U}}+2\dfrac{[U]^2}{K_{U2,U}}-[H]=0$

Our overall disassembly Model 1 (Figure 1, blue arrows) was fitted to HbA unfolding data

(Figure 3) using the optimized parameters in Table 1. A χ2 value of 0.76 was obtained.

The alternative Model 2 was proposed in order to try to determine the hemin binding

constant for the low affinity subunit (probably β) in tetrameric HbA (Figure 1). When HbA

unfolding data were fitted to the model 2 (Figure Supplement 2) with 2 additional degrees of

freedom (Table Supplement 2), there were only small changes in the goodness of the fit

(χ2=0.65). More importantly, the population fraction of semi-holo tetramers, with disassociated

hemin from β subunits, was always close to 0 (Figure Supplement 2A inset). Even this more

complex scheme indicated that small amounts of GdnHCl were sufficient to disrupt the weak

holo tetramer interface before any significant unfolding or hemin dissociation occurs at the

protein concentrations that were accessible to our spectral measurements (dashed lines in

Figures 3A and Supplement 2A).

When when Model 2 was simulated later at high protein concentrations (~mM range)

(see next section and Figure Supplement 3A), the population of semi-holotetramers was still

almost 0. This result also suggests that semi-holo tetramers are not a predominant species

during HbA assembly, and heme insertion is occurring much earlier during dimer assembly

(see Discussion). Furthermore, other kinetic studies have documented that hemin loss rates

increase dramatically when Hb tetramers dissociate into dimers [22, 24, 54]. Thus, we chose to

consider only the simpler Model 1, in which complete dissociation of holo tetramers into dimers

occurs during HbA disassembly before either hemin loss or unfolding.

This model (Figure 1) predicts sequential binding of hemin to the apo dimer with

association equilibrium constants equal to $K_{D,DH}$ = 1.3 X 10^{11} M^{-1} for the first step and $K_{DH,DH2}$=

1.3 x 10^{10} M^{-1} for the second step. By considering the binding sites to be non-interacting with

each other, we can further assess these constants in terms of a two-step binding, Adair

binding equation[55] to evaluate the intrinsic hemin affinity per subunit. In this case, the

intrinsic hemin affinity per subunit for the first step is $1/2K_{D,DH}$ or 6.5 x 10^{10} M^{-1}, and the intrinsic

affinity per subunit of the second step is $2K_{DH,DH2}$ or 2.6 x 10^{10} M^{-1}. This ~3 –fold decrease in

hemin affinity could be due to either negative cooperativity or differences in hemin binding to

the α versus β subunits. The non-cooperative, subunit differences interpretation is the most

reasonable because we and others [22, 38-40] have already shown that the rate of hemin

dissociation from β subunits is higher than α subunits in metHbA. Therefore, in our model,

these intrinsic hemin affinities can be interpreted to indicate that the initial loss of hemin from

the native folded dimer comes from the β subunit generating the semi-holo DH state with

hemin still in the α subunit.

Hemin binding to the molten globule dimer for HbA also occurs sequentially, and in this

case, the differences in hemin affinity between the first and second steps is even larger, with

$K_{I,IH}$ = 1.1 X 10^{11} M^{-1} and $K_{IH,IH2}$= 3.3 x 10^{8} M^{-1} (Table 1). In this case, the intrinsic subunit

affinity for the first step is $1/2K_{I,IH}$ =5.5 x 10^{10} M^{-1} and almost equal to that for the first step in

hemin binding to the native D state. In contrast, the intrinsic affinity for the second step is only

$2K_{IH,IH2}$ =6.6 x 10^{8} M^{-1}. Again, this significant decrease in affinity suggests the subunit

differences in the folded state increase ~30-fold in the molten intermediate. Based on the

known higher hemin loss rates from the β subunit in the folded states [22, 38-40], it appears

likely that its affinity for hemin is even lower in the molten or hemichrome I state.

The unfolded apo- and holo monomers at high [GdnHCl] only have residual α-helical

content, ranging between ~10 % and 20 % relative to folded HbA subunits (Table 1) [4]. Thus,

as expected, the fitted hemin affinity for these unstructured monomers is small $K_{U,UH}$= 2.9 x 10^7

M^{-1} (Table 1) and likely dominated by non-specific interactions [9, 11, 42, 43] (Table 1).

As shown in Figure 4, we were also able to analyze successfully holoHbF unfolding

data with the same Model 1. In this case, a somewhat higher $\chi2$ value of 1.76 was obtained. In

folded dimers, the hemin affinity for the first and second steps in hemin binding were more

similar with $K_{D,DH}$= 2.5 x 10^{11} M^{-1} (intrinsic subunit affinity, 1.3 x $10^{11}M^{-1}$) and $K_{DH,DH2}$= 1.3 x

10^{11} M^{-1} (intrinsic subunit affinity, 2.6 x $10^{11}M^{-1}$) indicating non-cooperative binding and little or

no subunit differences. However, the affinities of the first and second steps for binding to the

HbF molten intermediate were quite different, with $K_{I,IH}$ = 7.8 X 10^{11} M^{-1} (intrinsic subunit

affinity, 3.9 x 10^{11}) and $K_{IH,IH2}$= 1.4 x 10^9 M^{-1}(intrinsic subunit affinity 2.8 x 10^9 M^{-1}). Again, it

seems that the γ subunits have much lower hemin affinity in the molten globule intermediate.

The U monomers for HbF also had ~10-fold higher hemin affinity than in HbA, and the holo

monomers, UH had slightly higher α-helical content at 30% relative to folded HbF (Table 1).

Regardless of the exact interpretation, it is clear that one of the hemichrome containing

subunits in the molten dimer states of both HbF and HbA has an affinity for hemin that is

similar to that in the folded D state.

Modeling HbA disassembly in erythrocytes and in blood plasma

During maturation of red blood cells, all of the membrane organelles in the cytoplasm,

including the nucleus and mitochondria, are eliminated to allow for dense packing of

hemoglobins to a concentration of ~20 mM per subunit [56-58]. At these high concentrations,

native oxyhemoglobin exists almost completely as holo tetramers because the tetramer-dimer

disassociation constant ($K_{4,2}$ or $1/K_{2,4}$) is on the order of 10^{-6} M, and the corresponding value

for deoxyhemoglobin is even lower on the order of 10^{-12} M [49, 59, 60].

If oxidative stress occurs in erythrocytes due to either genetic disorders or

environmental chemicals or drugs, denaturation of the resultant metHb can occur in the cell

cytoplasm. At these high hemoglobin concentrations, the major disassembly pathway will be

dominated by holoprotein disassembly intermediates, represented by the left side of Figure 1.

To illustrate this effect quantitatively, we computed the population fractions of all the

intermediates for a GdnHCl-unfolding titration at 20 mM total protein using the parameters in

Table 1 (Figures 6A and B). The initial step involves dissociation of holo tetramers into holo

dimers, which is followed by unfolding into molten holo dimers with hemichrome spectra. This

IH_2 species is the dominant intermediate with small amounts of the semi–holo molten globule,

IH. The IH_2 and IH species dissociate directly into unfolded holo monomers before any

significant hemin loss occurs. The high Hb subunit concentration inhibits dissociation of both

the dimeric intermediates and bound hemin. Even the loss of hemin bound weakly to unfolded

monomers is inhibited until very high [GdnHCl]. When model 2 and the parameters in

Supplement Table 1 are used to simulate the population fractions at 20 mM Hb, the same

conclusion on the dominant role of holoprotein intermediates is reached (Figure Supplement

3A and 3B).

Hemolysis of red cells leads to the release of hemoglobin into blood plasma and can be

triggered by hemoglobinopathies, red cell aging, oxidative stress, blood infections by heme

iron extracting pathogens, and complications during cardiovascular medical procedures and

blood transfusions. Dilution into the large volume of plasma leads to very low, micromolar Hb

concentrations, which causes almost complete disassociation of tetrameric Hb into dimers and

accelerates auto-oxidation [23, 61-66]. Thus, when hemoglobin is diluted into plasma after red

cell lysis, metHb formation occurs and dissociation of both the inter-subunit interfaces and

hemin are facilitated. This situation was simulated in Figures 6C and D using the parameters in

Table 1 and 0.5 µM total protein. The disassembly pathway is quite different from that in red

cells. The initial holoprotein is primarily a dimer, DH_2 (Figures 6C) and the dominant

intermediates are the semi-holo molten dimer, IH, and the apo unfolded U monomer until the

completely unfolded chains, U_C are formed. Perhaps the most striking difference between the

red cell versus plasma pathways is the early appearance of free hemin before unfolding is

finished (Figure 6D). At low Hb concentration, dissociation of the $\alpha_1\beta_1$ dimer interface is

facilitated, resulting in low populations of the molten dimer hemichromes. Hemin dissociation is

also facilitated, leading to predominately apo intermediates and unfolded chains even before

the end of the simulated titration.

Small angle X-ray scattering (SAXS) analysis of apoHb properties in solution relative to

holoHb

SAXS intensity measurements (Figure 7) were done on both apo- and holo forms of the

recombinant crosslinked HbA or rHb0.1, which has a single-residue glycine linker covalently

connecting the N-terminal of the first α_1 subunit to the C-terminal of the second α_2 subunit. The

linker causes the aporHb0.1 to remain as a tetramer under conditions where native apoHb is a

dimer [4]. We chose the aporHb0.1 variant for SAXS analysis in order to have a covalently

crosslinked tetramer as standard for evaluating the radius of gyration (R_g) for holoHb

tetramers. The SAXS measurements were also done on the holo forms of the corresponding

non-crosslinked recombinant HbA, known as rHb0.0 [67-69]. Both rHb0.0 and rHb0.1 have a

V1M mutation in each subunit to initiate protein expression [4]. The SAXS measurements were

fitted to the Guinier plot equations (Figure 7 inset) to obtain R_g for each molecule. Minor effects

of molecular crowding were only observed for apoHb, occurring at protein concentrations >1.5

mg/ml (Figure supplement 4). Therefore for aporHb0.1, R_g and the pair distance distribution

function P(r) between atoms was determined with SAXS scattering measurements

extrapolated to infinite dilution in order to remove the molecular crowding effects [70] (Figure

7,Supplement Figure 4, and Supplement Figure 5). AporHb0.1 at 0.0 mg/ml exhibited a R_g of

~29.4 Å, whereas for the holo forms of rHb 0.0 and rHb0.1, the experimental R_g values

calculated were within the R_g range of 24 Å, as expected from the known crystal structure of

holoHbs (e.g. PDB ID 1O1L) (Figures 7 and supplement 4). Thus, aporHb0.1 is expanded in

solution. These kinds of expansions are often observed for partially unfolded proteins [70], and

support our view that hemin dissociation causes loss of the ability of hemoglobin to form

compact and cooperative tetramers. The losses of helical content are likely occurring around

the heme cavity regions, which also forms the tetramer interface, allowing expansion of the

crosslinked protein.

Interestingly, our SAXS measurements also suggest that this loss of helical content due

to hemin dissociation increases the susceptibility of Hb to molecular crowding. Similar effects

of molecular crowding probably might occur for the partially disordered assembly intermediates

and during early folding events for Hb in vivo. With more sticky regions of unfolded chain

segments exposed and expanded in solution, there is also a higher probability they can come

in closer contact and even interact weakly with other protein molecules causing the crowding.

Discussion

Global disassembly pathway for human hemoglobin

Hetero-tetrameric hemoglobins evolved from monomeric vertebrate myoglobins within

the globin metalloprotein family about 600 millions ago [1, 71, 72]. In this work, we have

shown for the first time that human hemoglobin shares a common disassembly pathway with

mammalian myoglobin. For both systems, hologlobin disassembly is mediated via a molten

hemichrome intermediate. However, relative to the simpler myoglobin monomer system

described in [11], the human hemoglobin assembly/disassembly mechanism evolved to be

much more complex. This complexity is demonstrated by the multiple steps for heme binding

to dimeric intermediates and by the sequential formation of the dimer ($\alpha 1\beta 1$) and then tetramer

($\alpha 1\beta 2$) interfaces (Figure 1).

To best of our knowledge, no one has previously tried to analyze reversible holoHb

disassembly quantitatively with a comprehensive model because of its complexity. In addition,

most studies involved thermal denaturation [73-75], which leads to precipitation of both the

globin chains and free hemin in conventional buffers. Our study has also been the first attempt

to measure equilibrium hemin binding parameters for hemoglobin, and provides reasonable

estimates for comparison with previous kinetic studies [22, 74, 76]. We were also able to

estimate heme binding parameters at different stages of Hb assembly/disassembly,

independently, and characterize how they influence inter-subunit assembly and globin folding.

Determination of hemin affinity constants is not possible by direct titration of apoglobins with

hemin because free hemin stacks to form dimers and higher molecular weight aggregates

even in the micromolar region [46]. In our experiments, the hemin equilibrium binding

constants were obtained by simultaneously analyzing GdnHCl-induced unfolding curves for

both apo- and holoHb in order to avoid problems associated with apoglobin and hemin

precipitation.

Hargrove, Barrick, and Olson [74, 76] had previously concluded that the rate constant

for hemin binding to apoglobin seems to be relatively invariant and $\approx 1 \times 10^8$ $M^{-1} s^{-1}$ per subunit

regardless of the globin structure for either apoHb or apomyoglobin. Past kinetic studies were

focused mostly on measuring heme disassociation rates from the end-state products of Hb

assembly e.g. TH_4, TH_2, and DH_2 (Figure 1) [22, 77]. However, such kinetic studies did not

attempt to consider how these rates and hemin affinities change as unfolding proceeds via

molten globules with significant reconfigurations of the hemin pockets.

The approaches that led us to successfully resolve "new" holo disassembly

intermediates in the model in Figure 1 and then determine their roles were (1) deconvolution of

visible absorbance signals into various intermediate hemin containing populations; (2) model

fitting to both apo- and holo protein unfolding spectral measurements; and (3) examination of

the protein concentration dependency of bimolecular events involving both inter-subunit

interactions or heme binding. We also developed a model evaluation approach that

incorporates numerical analysis methods to solve for the total disassembly population by

addressing the complexity of the Hb protein with four subunits and four bound hemin

molecules. For monomeric heme proteins, more straightforward quadratic equations can be

used to solve for fractions of various holo- and apo intermediates [11, 44].

Physiological relevance

The results in Figures 6A and B can also provide insights for the assembly pathway at

various protein and heme concentrations during hemoglobin biosynthesis. First, hemin binding

is a bimolecular process and required for tetramer assembly. Therefore, at high heme and

protein concentrations, the fraction of the holo molten globule and native dimers will

outcompete their apo counterparts to promote tetramer formation. In addition, at these high

concentrations, weak hemin binding to U monomers might be interpreted as an induced fit

mechanism [5], promoting holoprotein assembly from newly expressed α and β chains in

immature red blood cells or reticulocytes. This prediction seems to fit with previous

observations by Spirin's group [78] which suggest co-translational heme binding to nascent

globins chains coming off the ribosome in cell-free lysates of rabbit reticulocytes where high

concentrations of hemin were added [78]. Alternatively, at much lower concentrations of Hb

subunits and free hemin, hemin binding during assembly will proceed through conformational

selection[5] of specific partially or fully-folded conformations (Figures 6C, 6D). Even though the

total amount of holoHb being generated during erythropoiesis is large, the absolute

concentrations of the newly synthesized apo chains and free hemin in the steady state are

likely to be much lower due to rapid Hb assembly and hemin sequestering in membranes and

carrier proteins[79]. In addition, if unfolded globin chains and free hemin were present at high

concentrations, they would self-aggregate rapidly to form protein precipitates and stacked

hemin complexes[46], which would compete with assembly, causing unwanted pathological

situations.

We do acknowledge that our disassembly model derived from *in vitro* studies cannot

completely capture *in vivo* disassembly/assembly mechanisms. Within the packed red-cells,

non-specific or quinary interactions[80, 81] will occur between different combinations of α and

β chains for formation of transient U_2 species (Figures 6B) [4]. Our previous *in vitro*

experiments have shown that transient aggregation to generate apo U_2 intermediates actually

dampens the dimer formation [4]. Although these interactions could facilitate the correct

orientations for hetero-dimer assembly, such non-specific interactions could also potentially

lead to inclusion bodies in red cells. The later is especially likely if the subunit concentrations

are unequal and result in homo-oligomers (e.g. α_n or β_n occurring during thalassemia diseases)

due to disrupted expression or genetic instability of either α or β subunits [20]. *In vivo*

experiments are further needed to investigate how these non-specific interactions and

molecular crowding influence the $\alpha_1\beta_1$ dimer formation during Hb biosynthesis.

Finally, hemichromes are often observed within inclusion or Heinz bodies in red cells

[12, 17]. These particulates are found in red cells with hemoglobinopathies associated with

globin instability or oxidative stress. The results in Figures 6A and 6B provide an explanation

for the presence of these species. At the high internal hemoglobin concentration in red cells,

the unfolded holoprotein intermediates are dominant, leading to a much higher probability of

misassembled Hb being trapped in the large hemichrome populations for the UH or IH_2 states.

Our model was developed to examine unfolding as function of GdnHCl denaturant

concentrations but could be modified to include other environmental or physiological Hb

disassembly triggers such as lower pH or higher temperatures. Additionally, we can use the

population fraction values extrapolated to 0 M GdnHCl to estimate the relative absolute free

energies of different folding species *in vivo* as $-RT\ln Y_i$ for species i, where a large positive

value indicates a very unstable state (Figure 8). These values suggest the amount of free

energy that has to be added to disassemble or unfold the globin to a particular state. Figure 8

shows that the free energy differences between apo- and holo intermediates are predicted to

be higher in the red cells compared to dilute plasma environments where hemin and subunit

dissociation is promoted. However, in both environments the holo- and semi-holo disassembly

intermediates are generally favored over the apo variants.

When not contained within the reductive environment of red cells, acellular hemoglobin

in plasma undergoes rapid autooxidation forming metHb, which in turn enables rapid heme

disassociation. NO, which is generated in endothelial cells to facilitate vasodilation, can also

react with acellular oxyHb to produce metHb and nitrate. A surge of acellular Hb in the

bloodstream can lead to vasoconstriction due to NO scavenging by this dioxygenation reaction

and subsequent release of hemin from acellular metHbA [63, 82-86].

The early onset of a significant population of disassociated free hemin in the low protein

concentration pathway (Figures 6C and 6D) provides an explanation for why chronic or severe

hemolysis triggers free hemin mediated inflammatory diseases and endothelial dysfunctions

that lead to serious cardiovascular complications and kidney problems [61, 87, 88]. The

disassociated free hemin intercalates into cell membranes and leads to the generation of

reactive oxygen species (ROS) that can damage lipid, protein, and DNA [82, 89]. Free hemin

can also bind specifically to the toll-like receptor-4 (TLR4) on endothelial cells, triggering

adverse pro-inflammatory cell cascades [90, 91]. Hemin-mediated ROS generation and

oxidative stress on the blood vessel walls also promotes adhesion molecule expression on the

walls, leading to platelet aggregation and thrombosis [61, 82].

Conclusion

We have been able to derive for the first time a quantitative Hb disassembly/assembly

model, which can be used as a framework to identify the factors that resist Hb denaturation

and those that enhance expression. The key feature of our model involves formation of molten

globule intermediates with hemichrome spectral properties, and these intermediates are part of

the pathway for the disassembly and unfolding of native hemoglobin. Complementary

computational studies are also being done in our group to expand this framework to include

interpretations at an atomic level. Hopefully, our efforts to obtain a comprehensive mechanism

of Hb disassembly will lead to strategies for addressing erythropoiesis disorders and

hemoglobinopathies in a clinical setting and for engineering more stable acellular hemoglobin-

based oxygen carriers for transfusion therapy.

Materials and Methods

Cloning, expression and purification of Hb variants

The molecular biology and protein purification workflow for preparation of holoHb

samples were described in our previous paper on apoHb unfolding (see [4] and references

therein). Native HbA was extracted from expired blood units (Gulf Coast Regional Blood Bank,

Houston, TX). Recombinant native HbA and HbF were co-expressed with *E. coli* Met

aminopeptidase from pHE2 and pHE9 E. coli expression plasmids, respectively, in JM109

E.coli cells (Promega) as described by [92, 93]. The plasmids were a gifts from Dr. Chien Ho's

research group (Carnegie Mellon University). The aminopeptidase cleaves post-translationally

the initiator Met residue preceding the V1 residue in each Hb subunit. Recombinant HbA

variants with αH58L/V62F and βH63L/V67F mutations were then constructed through site

directed mutagenesis using Agilent QuckChange® Lightning Kits. rHb0.0 and rhHb0.1 were

expressed from pDL111-13e and pSGE1.1-E4 expression plasmids respectively in SGE1661

E.coli cells [67-69]. These cells and both plasmids were gifts from Somatogen Inc. (later Baxter

Hemoglobin Therapeutics).

Protein expression in cells was induced by 0.2 mM isopropyl β-D-1-

thiogalactopyranoside with the cells either grown in 500 ml terrific broth in flasks for medium-

scale recombinant Hb expression following protocols developed by Ho's group [92, 93] or in a

Biostat C 20 bioreactor for large scale expression followed protocols developed initially by

Somatogen, Inc. [68] and then modified by our group [94]. Isolation and purification of native

HbA and the recombinant hemoglobins were done as described by [68, 92-94]. Cell lysates

containing Hb were initially purified using a Zn^{2+} charged chelating Sepharose Fast Flow resin

column (GE healthcare) because Hb surfaces are rich with His residues that can bind Zn^{2+}

(also see [95]). Further Hb purification was done using a Q Sepharose anion exchange column

(GE healthcare), and a final purification step was done for the Hbs expressed from the pHE2

and pHE9 plasmids using a S Sepharose cation exchange column (GE healthcare) to

eliminate Hbs with uncleaved initiator Met residues. Quality and identify of the purified Hbs

were confirmed by: visible absorbance spectra for oxygenated, CO bound, and metHb forms;

electrospray ionization time-of-flight mass spectrometry; and reverse phase high performance

liquid chromatography (Agilent 1220 Infinity LC system) analyses [69, 96]. All hemoglobin

samples were purified and stored in their reduced CO forms to prevent auto-oxidation by

flushing cell lysates, purified proteins, and all the buffers with pure CO for ~30 minutes [97, 98].

Preparation of metHb, apoHb, and hemichrome standards

MetHb was prepared by oxidizing HbO_2 sample with a slight excess of potassium

ferricyanide. HbO_2 was initially prepared by flushing of HbCO samples with pure O_2 for ~30

minutes in chilled rotating flask that was exposed to an intense industrial light to promote

photodissociation of the bound CO [21, 97, 99].

ApoHb was prepared from the metHb according to the method developed by Antonini's

group [100]. First, the pH of a concentrated metHb sample was quickly reduced to ~2.2

followed by extraction of disassociated hemin into cold 2-butanone. Apoglobin recovered from

the extraction was then buffer exchanged into 10 mM potassium phosphate, 1mM DTT at pH

7. All ApoHb and metHb samples were prepared and examined at ~ 4 °C [4].

Hemichrome standards were prepared in order to obtain their reference spectra (Figure

2B). Excess imidazole was added to 12 μM HbA to allow for the binding of imidazole to the

hemin iron atom, thereby mimicking the hemin electronic structure of a *bis*-histidine

hemichrome. The visible absorbance spectra obtained from this species overlapped with the

hemichrome spectra generated by the addition of 600 μM sodium dodecyl sulfate, a

denaturant, to the same amount of HbA[11, 26, 48, 101].

Measurements of holoHb unfolding

MetHb samples were prepared in varying concentrations of GdnHCl in 200 mM

potassium phosphate, pH 7. The samples were incubated for 1 hour in a water bath at 10 °C.

Spectroscopic measurements of unfolding were then recorded at 10 °C in order to maintain the

solubility of unstable folding intermediates. These experimental protocols closely follow our

previous apoHb equilibrium unfolding measurements, except that for the addition of 1mM DTT.

DTT was added to the apoglobin samples to prevent formation of non-native, disulfide bonds

which occur during the preparation and storage of apoHb [4]. In our work with metHb samples,

DTT was observed to cause partial reduction, large visible absorbance changes, and a

complicated mixture of oxidation states, including deoxyHb and HbO_2. Thus, DTT was not

added to samples for holo- metHb unfolding experiments.

The CD and hemin absorbance changes were reversible when holoHb samples

incubated in concentrated GdnHCl solutions were diluted to reduce the extent of disassembly

and unfolding. CD spectral changes were measured using a Jasco J-810 CD

spectropolarimeter, and visible absorbance measurements were measured using a Cary 100

Bio UV-Visible spectrophotometer [4, 97]. During the CD spectra data collection, scanning

speed was set to 100 nm/min, data pitch was set to 0.1 nm, data integration time was set to 1

sec, and spectral bandwidth was set to 1 nm. For the visible absorbance data collection, data

interval was 1 nm. Blank buffer solutions were measured initially to enable baseline corrections

to be done on the spectral measurements with the spectrophotometer and spectropolarimeter

softwares. For a small portion of the visible spectral measurements recorded, baseline drifts

were observed. For each of the spectra with baseline drift, a new baseline was linearly

interpolated, using MATLAB R2017a (The MathWorks, Inc., Natick, MA), from the first 21

points of each spectrum at the longest wavelengths. The new baseline was then subtracted

from the corresponding spectrum.

Visible absorbance spectra at each [GdnHCl] were analyzed in terms of population

fractions of folded metHb, hemichrome species, and free hemin, using standard basis spectra

(B_i) for each species I (Figure 2B). The standard basis spectra for folded recombinant metHb

α(H58L/V62F)β(H63L/V67F) is different from that of aquametHbA because of the absence of

the H_2O ligand at the hemin iron [102]. For the mutant relative to aquametHbA, the Soret peak

is blue-shifted to 400 nm and the smaller peak at 630 nm is shifted to 600 nm (Figure

Supplement 6).

For all the visible absorbance deconvolutions, the fmincon minimizer in MATLAB

R2017a (The MathWorks, Inc., Natick, MA) was used to minimize squared residual difference

between the absorbance model defined in eq 9 and the measured experimental visible

absorbance spectrum in order to determine the population fraction (D_i) of each species.

$\text{Absorbance model} = \left(B_{MetHb}\right)xD_{Folded\ metHb} + \left(B_{Hemichrome}\right)xD_{Hemichrome} + \left(B_{Free\ hemin}\right)xD_{free\ hemin}$

753   eq 9

SAXS sample preparations, measurements, and data analysis

Analytical gel filtration was used to verify whether samples were mono-disperse. Holo-

and apoHb samples were prepared by thorough buffer exchange into degassed 10 mM

potassium phosphate, pH 7, 5 mM DTT. The buffer was degassed and contained DTT to

prevent during data collection error causing bubble formation and radiation damage (RAD) of

protein. HoloHb samples were also kept in the CO bound form to further inhibit heme

autoxidation during data collection[97, 103, 104].

SAXS measurements of the Hb samples were done using the Rigaku Bio-SAXS 1000

instrument at the Sealy Center for Structural Biology and Molecular Biophysics at the

University of Texas Medical Branch at Galveston, TX. Data were collected at 10 °C, with 12 to

16 hours of exposure to X-ray radiation, monitored hourly for RAD. SAXS measurements of

buffer alone were also done to enable blank subtraction. Data collection was done at varying

protein concentrations, ranging from 1mg/ml to 6 mg/ml. SAXSLab software (Rigaku) was

used for data collection, the SAXNS webserver (http://xray.utmb.edu/saxns) was used for

buffer subtraction, and the ATSAS software suite was used for data analysis. From within the

ATSAS software, PRIMUS, almerge, and CRYSOL packages were used for preparation and

analysis of the observed X-ray scattering curves and GNOM was used to generate P(R) plots

[97, 105].

Additional Supplementary Files

The following additional figures and table are compiled together in the Supplement.pdf file

included.

Figure supplement 1. Comparison of GdnHCl-induced equilibrium unfolding measurements of

holo- HbA, HbF, rHb0.1, and Hb α(H58L/V62F)β(H63L/V67F)

Figure supplement 2. Fittings of holoHbA GdnHCl-induced disassembly measurements to

HbA equilibrium disassembly model 2

Figure supplement 3. Simulated HbA disassembly population fractions in Hb packed red

blood cells and dilute blood plasma

Figure supplement 4. The Guinier Plots of each Hb dilution series

Figure supplement 5. The GNOM P(r) plots for aporHb0.1, holorHb0.1, and holorHb0.0

Figure supplement 6. Visible absorbance spectra for metHb α(H58L/V62F)β(H63L/V67F)

Table supplement 1. Fitted equilibrium disassembly parameters to model 2 for HbA

Abbreviations

GdnHCl, guanidine hydrochloride; Hb, hemoglobin; HbA, adult human hemoglobin A; HbF,

fetal human hemoglobin; metHb, oxidized hemoglobin with heme iron in the Fe^{3+} form; apoHb,

hemoglobin with the heme removed; rHb, recombinant hemoglobin; Mb, myoglobin; $K_{2,1}$,

heterodimer equilibrium dissociation constant; $K_{4,2}$, tetramer equilibrium dissociation constant;

CD, circular dichroism; DTT, dithiothreitol; rHb0.0, recombinant hemoglobin with V1M

mutations in both α and β globin chains; rHb0.1, recombinant hemoglobin with V1M

mutations in both di-α and -β globin chains and a single glycine linker joining the C-terminus

of the first α chain and the N-terminus of the second α chain; MALDI-TOF, matrix-assisted

laser desorption ionization time-of-flight; HPLC, high-performance liquid chromatography;

SDS, sodium dodecyl sulfate.

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805   control analyses of many of the recombinant hemoglobins.

Authors' contributions

PPS helped design the study, performed all the experiments, analyzed all of the data, and

wrote all the drafts of the paper. The experiments were part of her Ph.D. thesis at Rice

University under the guidance of JSO and GNP. She finalized the holoHb disassembly model

and analyses at Rutgers University with the support of DAC. MAW performed and analyzed

the SAXS experiments with samples provided by PPS. WCO expressed and purified the HbF

samples and performed several HbF unfolding experiments as part of his undergraduate thesis

at Rice University. GNP helped guide the work and edit various drafts of the paper. DAC also

edited the paper and focused the modeling work on medical and physiological relevance. JSO

helped to design the in vitro folding experiments, provided initial ideas about the model and

analyses, and was the major editor of the manuscript.

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824   Center for Structural Biology and Molecular Biophysics at UTMB.

Availability of data and materials

829 The data sets used in the unfolding analyses are available from the corresponding author on

830 reasonable request. Some of the protein samples and plasmids for the various Hbs are also

831 available but the amounts are limited because of the retirement of JSO.

Ethics approval and consent to participate

834 Not applicable

Consent for publication

837 Not applicable

Competing interests

840 The authors declare that they have no competing research or financial interests in this work.

Figure titles and legends

Figure 1. **Equilibrium disassembly models of HbA** A. **in cell and** B. **in vitro**. Model 1 is the

blue section and model 2 is the black and blue sections combined. In A, free hemin is the

green symbol, and bound hemin is the red symbol. Initially, the holoHb tetramer (TH_4)

disassociates into holoHb dimers (DH_2). Alternatively, TH_4 can undergo hemin (H)

disassociation from the subunits with lower hemin affinity to form the semi holotetramers (TH_2).

DH_2 can also undergo H disassociation sequentially first to form semi holo-dimers (DH),

followed by further H disassociation leading to apoHb (D) formation. DH_2 unfolds into holo-

molten globules (IH$_2$) with hemichromes characteristics. IH$_2$ also can lose H sequentially first

to form semi holo- molten hemichromes followed by formation of the apo- molten intermediates

(I). IH$_2$ next disassociates and simultaneously unfolds into holomonomers (UH). Loss of H from

UH leads into apo-monomers (U) with residual helical content. U can either associate non-

specifically with each other to form transient unfolded dimers (U$_2$) or completely unfold into

polypeptide chains (U$_C$). The apoHb unfolding mechanism had been derived independently in

our previous study[4].

Figure 2. **Visible absorbance spectra measurements for HbA disassembly**. A.

aquametHbA disassembly induced by GdnHCl; and B. aquametHbA, hemichrome, and free

hemin at 12 µM per subunit protein concentration. Unfolding measurements in 2A panel were

done at 12 µM per subunit total protein concentration. All holoHb unfolding measurements

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prepared in aqua-met oxidation states.

Figure 3. **Fittings of holoHbA GdnHCl-induced disassembly measurements to HbA**

equilibrium disassembly model 1. The solid circles are the measured data at different per

subunit total protein concentration, the solid lines are the fits, and the dashed lines are the

theoretical population fraction of folded holoHbA tetramers (TH$_4$) derived from the fittings. A.

Population fractions of aquamet- holoHb dimers and tetramers. B. Population fractions of

hemichromes. C. Population fractions of free hemin. D. Fractional CD change at 222 nm, a

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apoHbA unfolding measurements (curves at the far left) were obtained from previously

published data[4]. The experimental measurements in A-C panels were obtained from

deconvolution of visible absorbance spectra recorded between 350 nm and 660 nm. All holoHb

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triplicate titrations of protein with denaturant.

Figure 4. **Fittings of holoHbF GdnHCl-induced disassembly measurements to HbA**

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Figure 6. **Simulated HbA disassembly population fractions** in A. **Hb packed red blood**

cells and B. **dilute blood plasma** demonstrate Hb concentration directed disassembly

pathways. Simulations were done using Model 1 (Figure 1) and the parameters in Table1 as a

factor of GdnHCl.

Figure 7. **SAXS Log intensity profiles and Guiner plot (inset) for recombinant**

hemoglobins. For the Log-Log plots, the samples shown are the zero-concentration

extrapolated aporHb0.0 (green), 2.0 mg/ml Hb0.1 (blue) and 2.0 mg/ml Hb0.0 (red) Inset: The

Guinier plots with their residuals (darker lines) and $(1.3/R_g)$ Guinier q-upper limits (dashed lines

for R_g=29.4, 24.3, 23.6 Å). The expansion of the apoprotein is obvious from the less distinct

minima in the higher angle data.

Figure 8. **Predicted relative free energies of various Hb disassembly intermediates** in A.

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Figures

A. Denaturation of Hb in red cells

B. Scheme for holoHb disassembly

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