Structural basis for the allosteric regulation of the SbtA bicarbonate transporter by the P_{II}-like protein, SbtB, from *Cyanobium* sp. PCC7001

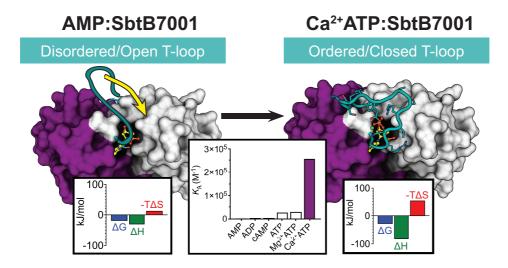
Joe A. Kaczmarski^a, Nan-Sook Hong^a, Bratati Mukherjee^b, Laura T. Wey^b, Loraine Rourke^b, Britta Förster^b, Thomas S. Peat^c, G. Dean Price^{*b} and Colin J. Jackson^{*a}.

^aResearch School of Chemistry, 137 Sullivans Creek Road, The Australian National University, Canberra, ACT 0200, Australia

^bARC Centre of Excellence for Translational Photosynthesis, Division of Plant Sciences, Research School of Biology, 134 Linnaeus Way, The Australian National University, Canberra, ACT 0200, Australia

^cCSIRO Biomedical Program, 343 Royal Parade, Parkville, VIC 3052, Australia

GRAPHICAL ABSTRACT



ABSTRACT

Cyanobacteria have evolved a suite of enzymes and inorganic carbon (C_i) transporters that improve photosynthetic performance by increasing the localized concentration of CO₂ around the primary CO₂-fixating enzyme, Rubisco. This CO₂-concentrating mechanism (CCM) is highly regulated, responds to illumination/darkness cycles and allows cyanobacteria to thrive under limiting C_i conditions. While the transcriptional control of CCM activity is well understood, less is known about how regulatory proteins might allosterically regulate C_i transporters in response to changing conditions. Cyanobacterial sodium-dependent bicarbonate transporters (SbtAs) are inhibited by P_{II}-like regulatory proteins (SbtBs), with the inhibitory effect being modulated by adenylnucleotides. Here, we used isothermal titration calorimetry to show that SbtB from Cyanobium sp. PCC7001 (SbtB7001) binds AMP, ADP, cAMP and ATP with micromolar-range affinities. X-ray crystal structures of apo- and nucleotide-bound SbtB7001 revealed that while AMP, ADP and cAMP have little effect on the SbtB7001 structure, binding of ATP stabilizes the otherwise flexible T-loop and that the flexible C-terminal C-loop adopts several distinct conformations. We also show that ATP binding affinity is increased ten-fold in the presence of Ca²⁺ and we present an X-ray crystal structure of Ca²⁺ATP:SbtB7001 that shows how this metal ion facilitates additional stabilizing interactions with the apex of the T-loop. We propose that the Ca²⁺ATP-induced conformational change observed in SbtB7001 is important for allosteric regulation of SbtA activity by SbtB and is consistent with changing adenylnucleotide levels in illumination/darkness cycles.

Keywords: SbtB, SbtA-mediated bicarbonate transport, cyanobacteria, CO_2 concentrating mechanism, regulatory protein, P_{II} -like protein

1. Introduction

Cyanobacteria possess a highly regulated network of enzymes, inorganic carbon (C_i) transporters and specialized compartments that enhance photosynthetic performance by increasing the localized concentration of CO₂ around the primary CO₂-fixating enzyme, Rubisco [1,2]. This CO₂-concentrating mechanism (CCM) responds to illumination/darkness cycles and allows cyanobacteria to thrive under limiting C_i conditions [3]. Bicarbonate (HCO₃⁻) transporters, which maintain high concentrations of HCO₃⁻ in the cell (up to 40 mM), are of vital importance to the CCM [3]. While the transcriptional regulation of these HCO₃⁻ transporters has been characterized [4], the molecular signals and post-translational mechanisms that transiently control the activity of cyanobacterial HCO₃⁻ transporters remain unclear.

The high-affinity, low-flux, sodium-dependent bicarbonate transporter A (SbtA), which is found in the plasma membrane of many cyanobacterial species, is a key component of the cyanobacterial CCM [3]. The dicistronic operon that typically encodes both SbtA and a cytoplasmic P_{II}-like regulatory protein (SbtB) is upregulated under C_i-limiting conditions in *Synechocystis* and *Synechococcus* species [5-8]. We have previously shown that SbtB from *Synechococcus elongatus* PCC7942 (SbtB7942) forms a stable complex with SbtA7942, and that several SbtB homologs from *Synechococcus, Synechocystis* and *Cyanobium* sp. inhibit their cognate SbtA when coexpressed in *E. coli* [9]. Since SbtA-mediated HCO₃⁻ transport must be coupled with the concurrent activity of ATP-dependent sodium exporters, we proposed that SbtB limits futile cycling by directly regulating SbtA-mediated HCO₃⁻ transport in cyanobacteria through the formation of an inhibitory SbtA-SbtB complex [9]. This general model was recently confirmed by Selim *et al.* [10], who showed that SbtB from *Synechocystis* sp. PCC6803

(SbtB6803) forms a stable complex with membrane-bound SbtA6803 under limiting C_i conditions *in vivo* and responds to changes in the relative levels of adenylnucleotides in the cell.

SbtB is a non-canonical member of the P_{II} signal-transduction superfamily [9-11]. P_{II} proteins are homotrimeric proteins that typically sense and tune the metabolic state of cells by binding adenylnucleotides and small molecules and forming regulatory complexes with diverse protein targets including transcription factors, enzymes and membrane transporters [12]. The functions of P_{II} proteins are primarily driven by large, ligand-induced conformational changes in the three flexible, solvent-exposed T-loops that mediate interactions with target proteins. Competitive binding of adenylnucleotides (typically ADP/Mg²⁺ATP) and/or other effector molecules at the clefts between the subunits can alter the conformation of the T-loops and modulate the affinity of the P_{II} protein for its target [13].

Selim *et al.* [10] recently reported that SbtB6803 binds AMP, ADP, cyclic-AMP (cAMP) and ATP with micromolar-range affinities, although the reported range of K_D values was relatively narrow (11–46 μ M from microscale thermophoresis experiments, 75–250 μ M for K_{D3} from ITC experiments). AMP and ADP were observed to stabilize the SbtA–SbtB6803 complex *in vivo*, and the authors suggested that the cAMP:AMP ratio may act as a signal to control SbtA6803-mediated HCO₃⁻ transport *in vivo*. However, the structures of SbtB6803 in complex with AMP and cAMP did not show any ligand-induced T-loop rearrangements that are typically required for the differential binding of P_{II} proteins to their target proteins [10]. Therefore, the structural basis for SbtA-inhibition by the formation of the SbtA–SbtB complex remains unclear. Furthermore, the functional consequence of sequence differences among SbtB homologs have not been explored. For example, many SbtB homologs, such as SbtB from *Cyanobium* sp. PCC7001 (SbtB7001), lack the C-terminal extension that forms a putative redox-sensing domain

in SbtB6803 [10]. Additional structural and functional data on phylogenetically distinct SbtB homologs are required to achieve a deeper understanding of the molecular basis for the regulation of CCM activity, *via* regulation of SbtA-mediated HCO₃⁻ transport, by SbtB.

Here we show that although SbtB7001 binds AMP, ADP, ATP and cAMP with micromolarrange affinities (like SbtB6803 [10]), the presence of Ca²⁺ dramatically increases the affinity of SbtB7001 for ATP affinity such that is ~50-fold and 100-fold greater than the affinity for ADP/cAMP and AMP (in the absence of Ca²⁺), respectively. High resolution crystal structures reveal that AMP, ADP and cAMP have little effect on the structure of SbtB7001, while ATP/Ca²⁺ATP binding causes dramatic rearrangements in the structure of SbtB7001, providing a molecular mechanism for the allosteric modulation of SbtB7001 function by ligand binding.

2. Materials and Methods

2.1 Cloning, expression and purification of SbtB7001

The gene encoding SbtB7001 (CPCC7001_1671) was codon-optimized for expression in *E. coli*, synthesized and cloned into pUC57 by GenScript, and then sub-cloned into the pHUE vector [14] between BamHI and HindIII sites to yield pHUE-SbtB7001. *E. coli* strain DH5 α cells were used for cloning. pHUE adds polyhistidine and ubiquitin tags (His₆–Ub) to the N-terminus of SbtB7001. Successful cloning was confirmed by Sanger sequencing at Garvan Molecular Genetics.

Protein expression was carried out in BL21(DE3) cells. Briefly, cells were grown in autoinduction media (20 g/l tryptone, 5 g/l yeast extract, 5 g/l NaCl, 6 g/l Na₂HPO₄, 3 g/l KH₂PO₄, 6 ml/l glycerol, 2 g/l lactose, 0.5 g/l glucose, 100 mg/l ampicillin), and protein

expression was carried out overnight at 25 °C with shaking. His₆–Ub-tagged SbtB7001 was purified from cell lysate using a 5 ml HisTrap FF column (GE Healthcare): the pre-equilibrated and lysate-loaded column was washed with binding buffer (20 mM NaH₂PO₄, 500 mM NaCl, 20 mM imidazole, pH 8), non-specific binders were removed using a step of 7% elution buffer, and tagged-SbtB7001 was eluted using 100% elution buffer (20 mM NaH₂PO₄, 500 mM NaCl, 500 mM imidazole, pH 8.0). Fractions containing tagged-SbtB7001 were pooled and transferred into binding buffer using a HiPrep 26/10 Desalting column (GE Healthcare). The His₆–Ub tags were removed using a ubiquitin carboxyl-terminal hydrolase 2 catalytic core domain (USP2cc):SbtB7001 molar ratio of 1:25 at 4 °C overnight as previously described [14]. Untagged SbtB7001 was recovered using a 5 ml HisTrap FF column (GE Healthcare) using binding buffer and concentrated using a 3 kDa high molecular weight cut-off Amicon Ultra Centrifugal Filter (Merck). SDS-PAGE was used to assess the purity of untagged SbtB7001. For preparative purification, untagged SbtB7001 was loaded onto a HiLoad Superdex 75 16/600 size exclusion chromatography (SEC) column (GE Healthcare), and eluted in SEC buffer (20 mM HEPES, 150 mM NaCl, pH 8.0). For analytical size-exclusion, untagged SbtB7001 was loaded on a Superdex 75 10/300 GL column (GE Healthcare) and eluted in SEC buffer. The sizeexclusion column was calibrated using a set of standard proteins (Gel Filtration HMW Calibration Kit, GE Healthcare) in SEC buffer. Protein concentration was measured spectrophotometrically using the molar absorption coefficient for USP2-cleaved SbtB7001 (12950 M⁻¹ cm⁻¹) as calculated using ProtParam (<u>http://expasy.org/tools/protparam.html</u>). SEC-purified, untagged SbtB7001 was used for isothermal titration calorimetry and crystallization.

2.2 Isothermal Titration Calorimetry

All isothermal titration calorimetry (ITC) experiments were performed on a Nano-ITC lowvolume calorimeter (TA Instruments). ITC experiments were performed at 25 °C with stirring at 350 rpm. Protein and ligand solutions were prepared in matched SEC buffer and degassed before use. Solutions of HCO₃⁻, AMP, ADP, ATP, and cAMP (all sodium salts, Sigma) were prepared volumetrically. Titrations were performed using 100-400 µM (monomer concentration) of USP2-cleaved SbtB7001, and typically involved $1 \times 1 \mu$, followed by $22 \times 2 \mu$ injections of 2– 25 mM ligand with 250–300 s intervals between injections. Data from each titration were analyzed using NITPIC [15] and SEDPHAT [16]; the baseline-subtracted power was integrated, and the integrated heats were fit to the single binding site model (A + B <-> AB, heteroassociation) model to obtain the macroscopic association constant (K_a) and enthalpy of binding (Δ H). Values for the incompetent fraction of protein were constrained between -0.2 and 0.2 and were fitted locally for each experiment, as suggested in the documentation for SEDPHAT [16]. Global fitting was achieved by iteratively cycling between Marquardt-Levenberg and Simplex algorithms in SEDPHAT until modelling parameters converged. Model parameters were obtained through the global fitting of single titrations or replicates. 68.3% confidence intervals were calculated using the automatic confidence interval search with projection method using Fstatistics in SEDPHAT. Other binding parameters and associated confidence intervals were obtained by propagation.

2.3 Protein crystallization

High throughput, sitting-drop vapor diffusion crystallization screens and subsequent hangingdrop optimization screens were set up at 18 °C or 4 °C in-house or at the Collaborative Crystallisation Centre (C3, CSIRO) and crystals formed in a number of conditions. For apo-SbtB7001 crystals, SEC-purified, untagged SbtB7001 was transferred to MES crystallization buffer (50 mM MES, pH 6.0, 100 mM NaCl). Crystals in sitting drops at 18 °C containing 0.15 µl of 10 mg/ml protein solution and 0.15 µl of reservoir solution (0.025 M disodium hydrogen-potassium dihydrogen phosphate, 20 % PEG 3350) were cryo-protected with 30 % (v/v) 1:1 mix of glycerol:ethylene glycol. Crystals of AMP:SbtB7001 formed at 18 °C in hanging drops containing 1 µl of protein solution (22 mg/ml of untagged SbtB7001 in 20 mM HEPES, 100 mM NaCl, pH 8.0, 10 mM MgCl₂, 12 mM AMP, 50 mM NaHCO₃) and 1 µl of reservoir solution (0.2 M sodium acetate, 18% PEG 3350). These crystals were cryo-protected using 30% (v/v) 2-methyl-2,4-pentanediol (MPD) and 100 mM NaHCO₃. Crystals of ADP:SbtB7001 formed at 18 °C in sitting drops containing 0.15 µl of protein solution (22 mg/ml of untagged SbtB7001 in 20 mM HEPES, 100 mM NaCl, pH 8.0, 10 mM MgCl₂, 12 mM ADP, 50 mM NaHCO₃) and 0.15 µl of reservoir solution (2.5 M ammonium sulfate). Crystals were cryoprotected with 30% (v/v) MPD and 100 mM NaHCO3. Crystals of cAMP:SbtB7001 formed at 18 °C in hanging drops containing 1 µl of protein solution (22 mg/ml of untagged SbtB7001 in 20 mM HEPES, 100 mM NaCl, pH 8.0, 10 mM MgCl₂, 12 mM cAMP, 50 mM NaHCO₃) and 1 µl of reservoir solution (0.2 M potassium acetate, 16% PEG 3350). Crystals were cryoprotected with 30% (v/v) MPD and 100 mM NaHCO₃. Crystals of ATP:SbtB7001 formed at 18 °C in sitting drops containing 0.15 µl of protein solution (22 mg/ml of untagged SbtB7001 in 20 mM Tris, 100 mM NaCl, pH 8.0, 10 mM MgCl₂, 12 mM ATP, 20 mM sodium bicarbonate) and 0.15 µl of reservoir solution (0.2 M lithium chloride, 20 % PEG 6000, 100 mM sodium acetate-acetic acid, pH 5). Crystals were cryo-protected with 30 % (v/v) glycerol. Crystals of Ca²⁺ATP:SbtB7001 formed at 18 °C in sitting drops containing 0.15 µl protein solution (20

mg/ml of untagged SbtB7001 in 20 mM Tris, 100 mM NaCl, pH 8.0, 10 mM MgCl₂, 12 mM ATP, 20 mM NaHCO₃) and 0.15 μl of reservoir solution (30 % MPD, 0.02 M CaCl₂, 0.1 M sodium acetate-acetic acid pH 4.6). Crystals were cryoprotected with 30 % (v/v) MPD.

All crystals were flash-cooled in liquid nitrogen. X-ray diffraction data were collected at the MX2 beamline at The Australian Synchrotron. Data was processed using XDS [17] and Aimless [18], and molecular replacement was performed using either MolRep [19] or Phaser [20]. Chain A of apo-SbtB6803 (PDB: 5O3P, 1.7 Å) was used for molecular replacement of Ca²⁺ATP:SbtB7001. All other structures were solved by molecular replacement using the resulting Ca²⁺ATP:SbtB7001 structure. Iterative cycles of manual model building and refinement were performed using Coot 0.8.2 [21], refmac [22], and/or *phenix.refine* [23]. TLS parameter refinement was also used, using TLS groups automatically selected by *phenix.refine*. Anisotropic B-factor refinement (all atoms except water) was used for refinement of Ca²⁺ATP:SbtB7001 (1.42 Å). Models were optimized using the PDB_REDO server [24]. Data collection and refinement statistics are provided in **Table 2**.

3. RESULTS

3.1 Nucleotide binding to SbtB7001

We used isothermal titration calorimetry (ITC) to investigate the affinity of SbtB7001 for the proposed P_{II} protein ligands AMP, ADP, cAMP and ATP (**Table 1**, **Figure 1**). Titrations with AMP (**Figure 1A**) gave a macroscopic dissociation constant (K_D , averaged across the three binding sites in the homotrimer) of approximately 450 µM, with an enthalpic contribution to binding of $\Delta H = -32$ kJ/mol and an entropic cost of T $\Delta S = -12.7$ kJ/mol. Titrations with ADP (**Figure 1B**) and cAMP (**Figure 1C**) gave intermediate K_D values of 224 µM and 238 µM,

respectively, with similar enthalpic (Δ H of -44 kJ/mol for ADP, -61 kJ/mol for cAMP) and entropic contributions (T Δ S of -23.3 kJ/mol for ADP, of -40.1 kJ/mol for cAMP) to binding. In contrast, ATP showed the highest binding affinity with a K_D of approximately 37 μ M (**Figure 1D**). The binding of ATP was associated with a much larger change in enthalpy (Δ H of -81.1 kJ/mol), indicative of the formation of numerous additional electrostatic interactions between ATP and SbtB7001. In addition, ATP binding resulted in a substantial loss in entropy (T Δ S of -55.9 kJ/mol), which is consistent with the stabilization of a mobile region of the protein upon binding of ATP.

Since Mg²⁺ is often required for the binding of ATP to P_{II} proteins [12], we also tested the affinity of SbtB7001 for ATP in the presence of 10 mM Mg²⁺ (**Figure 1E**) and in the presence of the Mg²⁺-chelator EDTA (**Figure 1F**). The resulting values for the K_D , ΔH , and T ΔS of binding were not significantly different to when ATP was tested alone. We then tested the effect of Ca²⁺, which can also coordinate nucleotides [25,26]. ITC measurements revealed that the affinity for ATP was increased 10-fold, to ~4 μ M (approximately 100-fold higher affinity than AMP in the absence of Ca²⁺) when 7 mM Ca²⁺ was present in the reaction conditions (**Figure 1G**). In the presence of a lower concentration of Ca²⁺ (500 μ M), ATP binding was still enhanced, with a K_D of ~7 μ M (**Table 1, Supp. Figure 11**). In the absence of nucleotides, binding of Ca²⁺ could not be detected by ITC (**Figure 1H**), suggesting that Ca²⁺ binding to SbtB7001 is dependent on prebound ATP or that Ca²⁺ and ATP bind cooperatively. Indeed, when Ca²⁺ was titrated into a solution of SbtB7001 in the presence of saturating concentrations of ATP (400 μ M, > 10 × K_D), isotherms indicated that Ca²⁺ bound with a K_D of approximately 60 μ M (**Table 1, Supp. Figure 13**). The effect of Ca²⁺ on the binding of nucleotides was most pronounced for ATP; K_D values

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for cAMP and AMP were reduced by less than a factor of 2 and 4, respectively, in the presence of 500 μ M Ca²⁺, and the K_D of ADP was unaffected by Ca²⁺ (**Table 1**, **Supp. Figure 4**).

Despite the role of SbtB in regulating SbtA-mediated HCO₃⁻ transport, specific binding of HCO_3^- to SbtB7001 could not be detected using ITC under the conditions tested (i.e. K_D for $HCO_3^- > 10$ mM, **Figure 1I**). At the concentration of HCO_3^- used in these titrations, the minor changes in heat observed in the control-subtracted thermograms are most likely due to non-specific binding events or artefacts. Indeed, when the same concentration of HCO_3^- was titrated into the lac repressor (LacI) from *E. coli* similar heats were observed (**Supp. Figure 15**).

| | $K_{\rm D}(\mu{ m M})$ | $\Delta H (kJ/mol)$ | $T\Delta S (kJ/mol)$ |
|-------------------------------------|-----------------------------|---------------------------------|---------------------------------|
| | 447 | -31.8 | -12.7 |
| AMP (n=3) | (398, 502) | (-36.2, -29.5) | (-17.4, -10.14) |
| | 127 | -30.6 | -8.40 |
| + 500 µM CaCl ₂ (n=2) | (107, 150) | (-41.2, -24.8) | (-19.4, -2.24) |
| ADP (n=4) | 244 | -43.9 | -23.3 |
| | (213, 264) | (-46.1, -35.1) | (-25.7, -14.2) |
| + 500 µM CaCl ₂ (n=2) | 270 (221, 301) | -49.3 (-54.1, -37.7) | -29.0 (-34.0, -16.8) |
| cAMP (n=4) | 238 | -60.8 | -40.1 |
| | (213, 266) | (-65.6, -57.6) | (-45.2, -36.6) |
| + 500 µM CaCl ₂ (n=2) | 144 (140, 147) | 67.6 (68.1,65.8) | -45.7 (-46.3, -43.8) |
| ATP (n=5) | 37.7 | -81.1 | -55.9 |
| (11.5) | (33.2, 43.5) | (-86.3, -76.3) | (-61.4, -50.8) |
| + 10 mM MgCl ₂ (n=2) | 34.3 (19.2, 56.5) | 79.0 (-89.6, -70.2) | -53.5 (-65.4, -43.3) |
| | 52.0 | (-89.0, -70.2) - 90.3 | (-05.4, -45.5) - 65.9 |
| + EDTA (n=1) | (44.2, 61.0) | (-94.3, -86.6) | (-70.3, -61.7) |
| + 7 mM CaCl ₂ (n=3) | 3.92 (3.16, 4.84) | -83.2 (-85.2, -81.2) | -52.3 (-54.9, -49.8) |
| + 500 µM CaCl ₂ (n=2) | 7.07 | -83.2 | -53.4 |
| + 500 μινι CaCl ₂ (II-2) | (6.46, 7.70) | (-84.3, -82.1) | (-55.2, -52.4) |
| CaCl ₂ (n=2) | ND | ND | ND |
| + 400 µM ATP (n=4) | 58.9 | -11.3 | 12.8 |
| + 400 μWI ATT (II-4) | (49.4, 66.5) | (-11.9, -10.1) | (11.9, 14.5) |
| NaHCO ₃ (n=2) | ND | ND | ND |

Table 1. Parameters for the binding of effector molecules to SbtB7001, measured by isothermal titration calorimetry (ITC).*

* Binding parameters were determined by ITC at 25 °C in 20 mM HEPES, 150 mM NaCl, pH 8.0 using a single-site binding model. Global fitting was performed using data from replicate titrations (number of replicates indicated in table) to obtain best-fit values (bold). 68.3% confidence intervals for the fit of the model to the data are given in parentheses. ND, specific binding not detected.

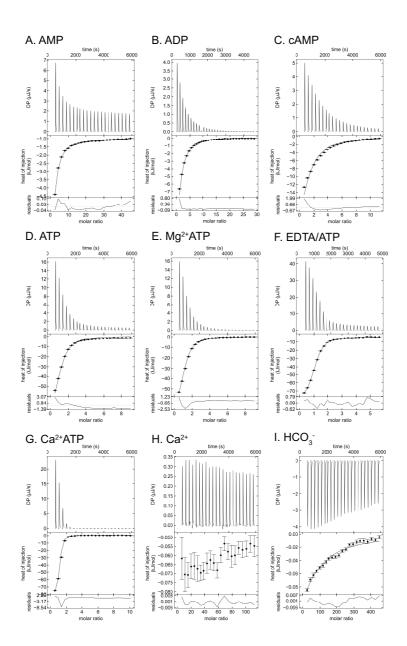


Figure 1. Isothermal titration calorimetry thermographs for the binding of nucleotides and potential effector molecules to SbtB7001. Representative ITC thermographs showing the titration of (A) 2 mM AMP into 110 μ M SbtB7001, (B) 10 mM ADP into 110 μ M SbtB7001, (C) 5 mM cAMP into 143 μ M SbtB7001, (D) 4 mM ATP into 143 μ M StbB7001, (E) 4 mM ATP into 143 μ M SbtB7001 in the presence of 10 mM MgCl₂, (F) 7 mM ATP into 400 μ M SbtB7001 in the presence of 0.5 mM EDTA, (G) 4 mM ATP into 143 μ M SbtB7001 in the presence of 7 mM CaCl₂, (H) 50 mM CaCl₂ into 143 μ M SbtB7001 and (I) 200 mM NaHCO₃ into 143 μ M SbtB7001. The upper panels represent baseline-corrected power traces. The middle panels represent the integrated heat data and best fit (globally fitted data from replicate experiments where possible) of the independent binding site model in SEDPHAT [16]. Figures were produced using GUSSI [27]. The bottom panels show the residuals of the fit. Error bars represent the standard error in the integration of the peaks as calculated by NITPIC [15]. Additional ITC data are provided in **Supp. Figures 1-15**.

3.2 Crystal structures of SbtB7001

To understand the structural basis of nucleotide binding to SbtB7001, we crystallized and collected diffraction data on StbB7001 alone (P1, 2.3 Å) and co-crystallized SbtB7001 with AMP (P1, 1.86 Å), cAMP (P1, 2.02 Å), ADP (P6322, 1.42 Å), ATP (P62, 1.9 Å) and Ca²⁺ATP (R3, 1.04 Å). SbtB7001 is trimeric in solution (**Figure 2A**), and all crystal structures showed a trimeric P_{II}-like core structure (**Figure 2B**) that was essentially identical to SbtB6803, with an RMSD of 0.45 Å between monomer α -carbons of apo-SbtB6803 (PDB: 5O3P) and apo-SbtB7001 (**Figure 2C**). The canonical nucleotide binding sites, which sit between the subunits of the trimer, are easily accessible from the solvent in apo-SbtB7001 (**Figure 2D**). StbB7001 shares 56% sequence identity with SbtB6803 (**Figure 2E**), which is consistent with the low RMSD value. However, SbtB6803 contains a C-terminal extension that is only present in a small subset of SbtB homologs, while SbtB7001 does not contain this extension.

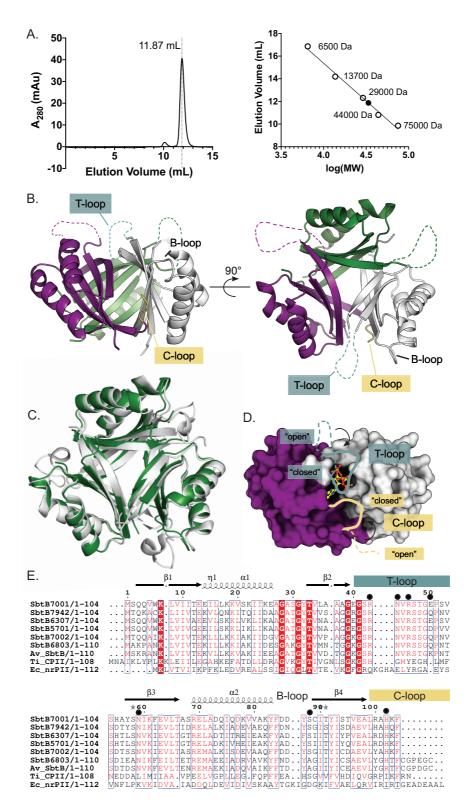


Figure 2. Overall fold of SbtB7001. (A) Size-exclusion chromatogram of SbtB7001 (left) and calibration curve for analytical size-exclusion chromatography (right). Open circles represent molecular weight standards and the closed circle represents untagged SbtB7001. The calculated molecular weight of SbtB7001 is consistent with a trimeric structure (calc. 33.7 kDa,

theor. 34.5 kDa for the trimer). (B) Crystal structure of apo-SbtB7001 (Chains A, B, C; colored by chain) showing the conserved trimeric P_{II}-like fold. The approximate positions of the flexible T-loop regions are indicated by the dashed lines, and the locations of the B-loop and C-loop are labelled. (C) Structural alignment of apo-SbtB7001 (green) and apo-SbtB6803 (PDB: 5O3P, white) showing conserved P_{II}-like fold. (D) Nucleotides (e.g. ATP, shown as ball-and-stick) bind at the interface between subunits. The surface representation of SbtB7001 with T-loop and Cloop residues removed highlights the accessibility of the nucleotide binding site. Lines indicate the approximate position of "open" (dashed line) and "closed" (solid line) states of the T-loop and C-loop. (E) Sequence alignment of SbtB7001, SbtB homologs from other cyanobacterial strains and related P_{II} and P_{II}-like proteins. Residues important for interactions with the phosphates of ATP are indicated by black circles. The primary and secondary structure of SbtB7001 is shown above the alignment. Sequences were aligned using Clustal Omega [28] and the figure was produced using EPScript 3.0 [29]. SbtB7001, Cyanobium sp. PCC7001 (WP 006910696.1); SbtB6803, Synechocystis sp. PCC6803 (WP 010872646.1); SbtB7942, Synechococcus elongatus PCC7942 (WP_011244772.1); SbtB6307, Cyanobium PCC6307 (WP 015110057.1; SbtB5701, Synechococcus WH5701 (WP 037981031.1); SbtB7002, Synechococcus sp. PCC7002 (WP 012306104.1); AvSbtB, Anabaena variabilis (yp 323533.1), TiCPII, carboxysome-associated P_{II} protein (CPII) from *Thiomonas Intermedia* K12 (WP 013104276.1); EcnrP_{II}a, P_{II} protein from *Escherichia coli* (CCK45606.1).

| | apo-SbtB7001 | AMP:SbtB7001 | ADP:SbtB7001 | cAMP:SbtB7001 | ATP:SbtB7001 | Ca ²⁺ ATP:SbtB7001 |
|--|--|--|---|--|---|--|
| PDB | 6N4A | 6MMO | 6MMC | 6MMQ | 6NTB | 6MM2 |
| Data collection | | | | | | |
| Space group Cell dimensions | P 1 | P 1 | P 63 2 2 | P 1 | P 62 | R 3 |
| a, b, c (Å) | 46.6 51.0 66.6 | 46.7 51.3 66.8 | 68.6 68.6 89.1 | 46.6 50.2 66.6 | 132.9 132.9 33.8 | 52.0 52.1 95.2 |
| α, β, γ (°) | 94.3 106.1 103.6 | 97.9 102.5 103.1 | 90 90 120 | 97.6 102.9 102.5 | 90 90 120 | 90 90 120 |
| Resolution (Å) R _{merge} | 49.03–2.30 (2.38–2.30) 0.040/0.015/1.05 | 43.22–1.86 (1.93–1.86) 0.046/0.033/1.922 | 49.44–1.42 (1.47–1.42) 0.105/0.037/6.445 | 35.32–2.02 (2.09–2.02) 0.05/0.029/1.270 | 43.52–1.90 (1.97–1.90) 0.107/0.054/2.680 | 15.03–1.04 (1.08–1.04) 0.039/0.035/1.337 |
| (overall/inner/outer) CC _{1/2} Completeness (%) Multiplicity | 0.999/0.999/0.574 98.0/96.6/95.2 3.8/3.8/3.8 | 0.998/0.996/0.335 97.1/86.5/96.3 3.6/3.2/3.8 | 1.000/1.000/0.380 100.0/99.7/100.0 37.8/26.9/36.3 | 0.998/0.998/0.307 96.4/93.5/96.6 2.4/2.4/2.5 | 0.999/0.999/0.560 100/99.3/100.0 20/17.7/18.1 | 0.999/0.997/0.405 95.2/53.4/50.9 9.0/1.34/0.55 |
| Refinement | | | | | | |
| Resolution (Å) | 49.03–2.30 | 41.58–1.86 | 49.44–1.42 | 35.32-2.02 | 43.52–1.90 | 15.03-1.04 |
| No. of reflections | (2.38–2.30) 24661 (2413) | (1.93–1.86) 47132 (4612) | (1.47–1.42) 24048 (2343) | (2.09–2.02) 35629 (3583) | (1.97–1.90) 27442 (2714) | (1.08–1.04) 44104 (2995) |
| Rwork / Rfree | 0.2170/0.2767 | 0.2234/0.2579 | 0.2029/0.2202 | 0.2208/0.2794 | 0.210/0.2475 | 0.1452/1632 |
| No. of atoms | 3989 | 4189 | 788 | 4142 | 2541 | 997 |
| Protein | 3961 | 4002 | 722 | 3985 | 2350 | 845 |
| Ligands/Ions | 10 | 138 | 1 | 132 | 95 | 33 |
| Solvent | 18 | 49 | 65 | 34 | 96 | 119 |
| Average B-factor | 75.0 | 61.25 | 27.31 | 66.26 | 45.8 | 24.1 |
| Protein | 75.1 | 61.3 | 26.44 | 66.28 | 45.9 | 20.9 |
| Ligands/Ions | 84.12 | 59.9 | 28.82 | 60.20 | 38.8 | 15.5 |
| Solvent | 56.5 | 57.0 | 36.94 | 64.02 | 49.9 | 48.7 |
| R.m.s. deviations | | | | | | |
| Bond length (Å) | 0.011 | 0.007 | 0.005 | 0.008 | 0.007 | 0.013 |
| Bond angles (°) | 1.46 | 0.91 | 0.84 | 1 | 1.17 | 1.6 |

Table 2. Data collection and refinement statistics for structures described in this work.

*Statistics for the highest-resolution shell are shown in parentheses.

3.3 Binding of AMP, ADP and cAMP at the inter-subunit clefts of SbtB7001.

When SbtB7001 was co-crystallized with AMP (**Figure 3A**), ADP (**Figure 3B**), and cAMP (**Figure 3C**), electron density was observed for the respective ligands within each of the clefts between the monomers. The ADP:SbtB7001 co-crystal displayed ambiguous electron density around the β -phosphate of ADP, consistent with either a mobile β -phosphate, or hydrolysis of ADP and occupancy by AMP (as has been observed in other studies of related P_{II} proteins [10,11,30]).

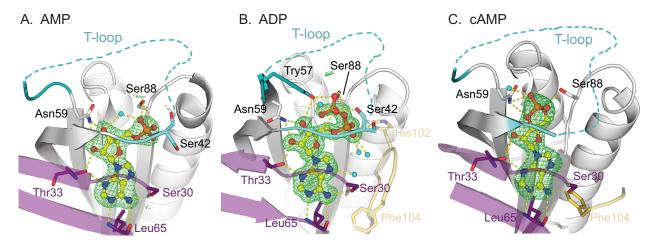


Figure 3. Ligand binding sites of SbtB7001. Representative binding sites from structures of SbtB7001 co-crystallized with (A) AMP, (B) ADP, and (C) cAMP. Ligands (yellow ball-and-stick) bind in the clefts between subunits (neighboring subunits are colored white and purple). Key residues for binding are shown as sticks. F_o - F_c polder omit maps are shown as a green mesh contoured at 3 σ and carved at 3 Å around the ligand. T-loop residues are colored cyan or teal, and the C-loop is colored yellow. A dashed line indicates the approximate location of the unmodelled, flexible T-loop. Coordinated water molecules are shown as cyan spheres.

Each nucleotide binds with the shared adenine motif sitting within a mostly hydrophobic cleft (lined by the side chains of Ile11, Leu65, Ile90), while the backbone carbonyl and amides of Gly31 and Leu65 and the hydroxyl groups of Ser30 and Thr33 form specific polar interactions with the base. The hydroxyl groups of the ribose sugar form hydrogen bonds with the side chains of Thr33 and Asn59 in each case, and the phosphate groups are positioned near the base of the flexible T-loop. In AMP:SbtB7001 (Figure 3A) and ADP:SbtB7001 (Figure 3B), the α phosphate interacts with the side chain of Ser88, as well as with the backbone of Cys89 and Ser42. The interactions between the α -phosphate and the backbones of Ser42 and occasionally Gly41 seem to be important for stabilizing residues 40–42; in apo-SbtB7001 and cAMP:SbtB7001 these interactions are not observed, and this region becomes unstructured. Since AMP and ADP share common adenosine and ribose moieties that bind to SbtB7001 in an identical manner, the higher affinity of ADP (225 μ M versus 387 μ M; Table 1) must be attributed to differences in the way that the phosphate groups interact with SbtB7001. Indeed, in the ADP:Sbtb7001 complex we observe additional interactions between the β -phosphate and Ser88 and Tyr57, which is consistent with the ITC experiments that reveal a larger change in enthalpy upon binding of ADP compared with AMP (Table 1). Interactions between ADP and T-loop residues 40–56 are likely to only be transient, at most, since additional interactions with T-loop residues were not observed in the crystal structure of ADP:SbtB7001 and the electron density around the β -phosphate of ADP is consistent with a mobile β -phosphate that is not restrained by such interactions.

Due to the cyclic nature of cAMP, its binding mode is somewhat different to AMP/ADP; the interaction with Ser42 is not observed in cAMP:SbtB7001 and the phosphate forms an additional interaction with side chain of Asn59 (**Figure 3C**). Both SbtB7001 and SbtB6803 bind cAMP with higher affinity than AMP (e.g. 238 μ M versus 447 μ M for SbtB7001; **Table 1**) [10]. This is consistent with the crystal structures, which show the different position of the phosphate moiety of cAMP (versus AMP) facilitates an additional bond with Asn59, a residue that is conserved amongst SbtB homologs (**Figure 1E**). This interaction is not observed in AMP:SbtB7001 and could explain the increased affinity for cAMP by SbtB7001 and SbtB6803 [10].

3.4 Ca²⁺ and ATP stabilize the T-loop.

In contrast to AMP and ADP, the γ -phosphate of ATP supports additional interactions with the highly conserved residues at the base of the T-loop (residues 40–46) (Figure 4A, 4B), explaining why SbtB7001 binds ATP with a much higher affinity and has a more favorable enthalpy of binding compared with AMP, ADP or cAMP. The salt bridge formed between the γ -phosphate of ATP and the side chain of Arg46 appears to be particularly important, given the high conservation of this residue among SbtB homologs (Figure 1E). This ATP-specific interaction appears to stabilize an interaction between Arg46 and Asn59, which pulls the base of the T-loop towards the core of the protein, where it tightly encircles the phosphate groups of the ligand. In canonical P_{II} -proteins, the residue at the position analogous to Asn59 in SbtB7001 is involved in forming a key salt bridge with T-loop residues, and the formation and breaking of this salt bridge has been shown to play an important role in conferring functionally-relevant T-loop conformations [31,32]. In SbtB7001, when the T-loop is held in position by the Arg46–Asn59 interaction, additional interactions are also formed between the phosphates of ATP, Ser88, Arg43 and multiple points on the backbone of the T-loop, which further stabilize this region. While the interaction between the α -phosphate and Arg43 could be possible when AMP or ADP is bound, it seems that the additional ATP-specific interaction with Arg46 is required to first stabilize this basal T-loop region and promote this interaction between the α -phosphate and Arg43. Thus, in comparison to the largely disordered T-loop residues 43–55 in apo-SbtB7001, AMP:SbtB7001, ADP:SbtB7001 and cAMP:SbtB7001 (as was also observed in structures of SbtB6803 bound to AMP and cAMP [10]), the binding of ATP to SbtB7001 resulted in stabilization of the T-loop. These structural observations are consistent with our ITC experiments

for ATP, which revealed a substantially greater entropic cost associated with ATP binding, compared with the binding of AMP, ADP and cAMP (**Table 1**).

Although both ATP:SbtB7001 and Ca²⁺ATP:StbB7001 co-crystals were grown in the presence of 10 mM Mg²⁺, electron density consistent with an octahedrally coordinated metal ion near the γ -phosphate of ATP was only observed in the crystal grown in the presence of 20 mM Ca²⁺ (Figure 4B). Metal-oxygen distances (~2.4 Å) and environmental B-factor analysis (B-factor 20.38 for Ca²⁺, averaged B-factor of 21.94 for coordinated atoms) are consistent with a Ca²⁺ binding site. This was confirmed using the 'place elemental ions' tool in phenix.refine (valence sum for Ca^{2+} is 1.989, expected 2; valence sum for Mg^{2+} is 0.948, expected 2) [33]. The Ca^{2+} ion is coordinated by the side chains of Glu50 and Asn44, the backbone carbonyl of Gly41, the βand γ -phosphates of ATP, and a single water molecule that is held in place by Ser54 (Figure 4C). The interactions with Glu50 and Asn44 appear to be important for stabilizing the apex of the T-loop; when Ca²⁺ was absent from the crystallization conditions, the complete T-loop could only be completely modelled in one of the three chains (ATP:SbtB7001, chain C), consistent with less order between residues 46 and 57. Even when this region could be completely modelled, the T-loop of ATP:SbtB7001 adopts a conformation distinct from that seen in Ca²⁺ATP:SbtB7001 (Figure 4D): Asn44 and Tyr57 flip away from the Ca²⁺ binding site, Glu50 moves away from the ATP binding site, and Pro51 adopts a cis- conformation. Thus, it seems that these Ca²⁺-specific interactions further stabilize the 'closed' conformation of the T-loop, which is consistent with our observation from ITC that affinity for ATP was increased in the presence of Ca2+, but not Mg2+. This 'closed' Ca2+ATP:SbtB7001 T-loop conformation almost completely seals the nucleotide binding site from one side, leaving only a narrow channel near the β - and γ -phosphates that could allow for the passage of an ion to and from the metal binding

site, as well a thin access tunnel near the C-terminus of the protein (**Figure 4E**). In contrast, the flexibility observed in the ATP:SbtB7001 T-loop in the absence of Ca^{2+} means that the phosphate tail of ATP is not completely occluded by the T-loop.

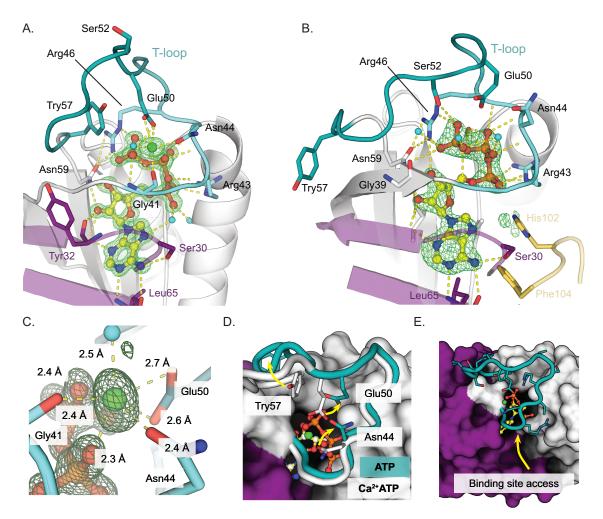


Figure 4. Ca²⁺ATP and ATP stabilize the T-loop of SbtB7001. (A) Ligand binding site of Ca²⁺ATP:SbtB7001. Ligands are shown as yellow ball-and-sticks and key residues for binding are shown as sticks. F_{o} - F_{c} polder omit maps are shown as a green mesh contoured at 3 σ and carved at 3 Å around the ligand. T-loop residues are colored cyan (residues 40–46) or teal (residues 47–57). The C-loop is colored yellow. Coordinated water molecules are shown as cyan spheres. (B) Ligand binding site of ATP:SbtB7001. (C) Close up of Ca²⁺ binding site showing distances to coordinating atoms. (D) When completely modelled, the T-loop of ATP:SbtB7001 (teal) forms a conformation distinct from that observed in Ca²⁺ATP (white). Movement of key residues is indicated by yellow arrows. (E) The T-loop is stabilized against the core of the protein by Ca²⁺ATP: and occludes the binding site from above. The binding site can still be accessed from the direction of the yellow arrow.

3.5 The dynamic C-loop forms part of the nucleotide binding site.

The C-terminal C-loop forms part of the nucleotide binding site in canonical P_{II} proteins [34]. In SbtB7001, the C-terminal residues Leu99–Phe104 could only be completely modelled in AMP:SbtB7001 (Chain A only), cAMP:SbtB7001 (Chain B only), ADP:SbtB7001, and ATP:SbtB7001 (**Figure 5**). All other chains showed poor electron density in this region, consistent with conformational heterogeneity. In crystals where the C-terminal residues could be modelled, this region occupies part of the inter-subunit cleft, where it forms part of the ligandbinding cavity, occludes access to the binding site, and can come within bonding distance of both the ligand and the T-loop.

In apo-SbtB7001, residues Leu99–His102 adopt an extended conformation: these residues run antiparallel to β 4 of the neighboring subunit with backbone hydrogen bonds holding it in place (**Figure 5A**). In this pose, His102 is positioned between Phe84 and the nucleotide binding site. His102 is located in a similar position in ADP:SbtB7001, where it is within hydrogen bonding distance (~3 Å) of the α -phosphate of the ligand (**Figure 5B**). While this interaction could aid in binding the α -phosphate of ADP/AMP, it is sterically incompatible with the 'closed' position of the T-loop observed in Ca²⁺ATP:SbtB7001 and ATP:SbtB7001, clashing with the position of Arg43 (**Figure 5C**). The rigid C-terminal helix observed in SbtB6803 holds His102 in a similar location and would cause a similar steric clash with the position of Arg43 that appears to be important for ATP-binding (**Figure 5D**).

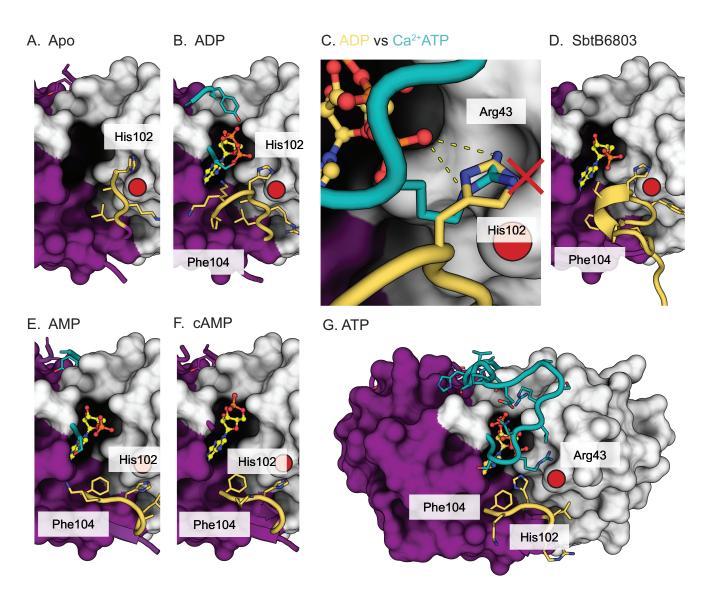


Figure 5. Structural changes at C-loop of SbtB7001. The C-loop (residues 99–104, yellow) adopted distinct conformations in the crystal structures of SbtB7001 and SbtB6803. T-loop residues are colored teal. The position of Phe88 is indicated by the red circle. In apo-SbtB7001 (A) and ADP:SbtB7001 (B) the C-loop extends up into the intersubunit cleft. These conformations would not be compatible with the closed T-loop observed when ATP is bound; for example, in ADP:SbtB7001 the C-loop residue His102 occupies the space that T-loop residue Arg43 occupies in Ca²⁺ATP:SbtB7001 (C). The rigid C-terminal region of SbtB6803 (e.g. PDB: 5O3R) holds His102 in a similar position that would likely clash with the closed T-loop. (E-G) The C-loop adopts a more compact pose in chains of AMP:SbtB7001 (E), cAMP:SbtB7001 (F) and ATP:SbtB7001 (G). As exemplified in ATP:SbtB7001, this conformation is compatible with the closed T-loop conformation. When the C-loop and T-loop are both closed, the nucleotide binding site is completely occluded.

In AMP:SbtB7001 (chain A, **Figure 5E**) and cAMP:SbtB7001 (chain B, **Figure 5F**), $\beta4$ continues to run antiparallel with $\beta1$ until Ala101 before wrapping back around towards the ligand-binding site. In this more compact conformation, His102 is no longer within bonding distance of the ligand. The C-loop of ATP:SbtB7001 traces a similar route, although His102 adopts an alternate rotamer that positions it within bonding distance of the α -phosphate of ligand (**Figure 5G**). Unlike the extended conformation observed in apo-SbtB7001 and ADP:SbtB7001, the more compact conformation observed in AMP:SbtB7001, cAMP:SbtB7001 and ADP:SbtB7001 is compatible with the closed conformation of the T-loop (e.g. as observed in ATP:SbtB7001). When both the T-loop and C-loop adopt a closed conformation (as in ATP:SbtB7001), the nucleotide binding site access is almost completely occluded.

In the related, carboxysome-associated P_{II} protein from *Thiomonas intermedia* (TiCPII), HCO₃⁻ binds near the C-loop and alters the affinity of the protein towards nucleotides [11]. While most ligand-bound crystals of SbtB7001 were grown in the presence of 50–100 mM NaHCO₃, electron density that unambiguously corresponded to HCO₃⁻ could not be identified in any of the structures. This is consistent with our ITC results that showed only low-affinity, binding of HCO₃⁻ to SbtB7001, consistent with non-specific protein stabilization.

4. DISCUSSION

In this work we have shown that, like SbtB6803 [10], SbtB7001 binds AMP, ADP, cAMP and ATP with micromolar-range affinities. However, in contrast to SbtB6803, we found that SbtB7001 binds ATP with an affinity 5- to 10-fold greater than the other nucleotides tested, and that the affinity for ATP increases a further 10-fold (from a K_D of ~40 μ M to ~4 μ M) in the presence of Ca²⁺. In contrast, the presence of Ca²⁺ had a less substantial effect on the binding of AMP, ADP or cAMP. ITC experiments revealed that binding of ATP and Ca²⁺ATP to SbtB7001 is associated with a large change in enthalpy and loss of entropy, consistent with the formation of a number of charged interactions between the ligand and SbtB7001, and a significant stabilization of a mobile region of the protein, respectively. Structures of SbtB7001 cocrystallized with ligands support the ITC results and (i) provide a structural explanation for the preferential binding of ATP, (ii) account for differences in the affinities of SbtB7001 and those reported for SbtB6803 [10], and (iii) reveal how binding of ATP stabilizes the otherwise flexible T-loop. Together, these results provide a structural explanation for how SbtB could regulate SbtA, via changes in the adenylate charge ratio ([ATP]/[ATP+ADP+AMP]) observed in a transition from light to darkness in cyanobacteria [35].

The mode of ATP binding to SbtB7001 is distinct from ATP binding in the canonical P_{II} proteins, in which a conserved RXR motif in the C-terminal C-loop is responsible for forming interactions with the terminal phosphates of the ligand [34]. In SbtB7001 these interactions are instead provided by arginine residues from the T-loop. Indeed, in SbtB7001 (and the majority of the cyanobacterial SbtBs; **Figure 1E**) the C-terminal extension is largely truncated, suggesting that it is less essential for ligand binding in SbtBs from most cyanobacteria. The reduced C-loop in SbtB7001 displayed conformational heterogeneity, adopting extended and compact states,

with the extended conformation being incompatible with Ca²⁺ATP binding. It is unlikely that the C-loop substantially contributes to the differentiation between AMP, ADP, or ATP, since C-loop residues (e.g. His102) are unlikely to be able to form interactions with the β - or γ -phosphates of ADP or ATP.

The T-loop residues involved in binding ATP are highly conserved amongst most SbtB homologs (Figure 1E), suggesting that all SbtB homologs should bind ATP more tightly than AMP, ADP or cAMP. However, this was not observed for Sbtb6803 [10]. The C-terminal extension found in SbtB6803, which is only present in a small subset of SbtB homologs, may contribute to the lower affinity between SbtB6803 and ATP. Indeed, based on the structures presented here, the rigid, disulfide-linked C-terminal region of SbtB6803 would clash with the position that Arg43 is required to adopt for the T-loop to accommodate ATP. Interestingly, the C-terminal region is unresolved in the structure of a related SbtB from Anabaena Variabilis ATCC 29413 (AvSbtB, PDB: 3DFE), which also has the putative redox-sensing C-terminal extension, suggesting that this region is able to switch between open and closed states. It is therefore possible that that the C-terminal extension found in SbtB6803 and a small subset of SbtBs could alter the affinity of SbtB for ATP based on the cellular redox conditions of the cell by preventing conformations of the T-loop that are required for ATP binding [10]. The ATPbinding activity of SbtB6803 was not tested in different reducing/oxidizing conditions [10], and it is possible that SbtB6803 has a higher affinity for ATP under certain redox conditions.

In the cyanobacterial cell, the ligand-bound state of SbtB will depend on both the affinity of the ligands, and the local concentrations of each nucleotide and/or effector molecules. Our results suggest that the apo-form of SbtB7001 is unlikely to be physiologically relevant because there will always be AMP/ADP/cAMP/ATP present at concentrations in excess of the K_D [36].

The ratio of [ATP]/[ATP+ADP] has been reported to be around 0.8 in *Synechococcus* under illumination [37], but varies significantly in cyanobacteria in response to changing light and C_i availability. For example, the [ATP]/[ADP+ATP] ratio drops from 0.9 to around 0.4 within 5-6 hours of *Synechococcus elongatus* PCC7942 being moved to darkness, and returns to about 0.85 within an hour of being returned to the light [35]. Thus, when ATP levels drop (such as during darkness), the binding sites of SbtB7001 are likely to become occupied by ADP (or possibly by AMP or cAMP), rather than ATP. Since the SbtA-SbtB6803 complex is stabilized by AMP and ADP [10], we expect that binding of these ligands would also stabilize the inactive SbtA-SbtB7001 complex (with open T-loops) to prevent futile metabolic cycling in the dark. We envision that when ATP levels rise again (upon exposure to light), ATP could displace AMP/ADP and cause the T-loops to adopt the closed conformation, which we envisage would be the allosteric signal to cause the complex to dissociate. Further *in vivo* studies will be required to test this model.

The finding that Ca^{2+} enhances ATP binding to SbtB7001 raises the possibility that Ca^{2+} might act a second messenger to influence how SbtB regulates SbtA-mediated transport. Interestingly, transient increases in Ca^{2+} have been shown to upregulate expression of CCM-related genes, including the *sbt* operon, in *Anabaena* sp. PCC7120 [38]. Ca^{2+} is known to play an important role in the regulation of other cyanobacterial systems as well. For example, Ca^{2+} acts as a signal for heterocyst formation [39] and other regulatory proteins involved in sensing C_i , such as soluble adenylyl cyclases, can simultaneously sense ATP, Ca^{2+} , and $HCO_3^-/CO_2/pH$ [40]. Further evidence that Ca^{2+} plays an important role in the regulation of cyanobacterial HCO_3^- transport includes a structure of the periplasmic domain of another cyanobacterial HCO_3^- transporter, CmpA, which revealed that HCO_3^- bound strongly only if Ca^{2+} was present [41]. Future experiments could focus on testing whether physiological levels of Ca^{2+} could act as a secondary effector for SbtB *in vivo* since free calcium levels in cyanobacteria can vary between 100–200 nM in normal conditions [42], to low micromolar-range concentrations following light to dark transitions [43] or heat-, cold-, or ammonia-shock [44]. If ATP binding does cause the complex to dissociate, then variations in the level of Ca^{2+} in the cell could further control this equilibrium by altering the affinity of SbtB for ATP. For example, transient increases in the concentration of free Ca^{2+} following light-to-dark transitions could conceivably help delay the formation of the SbtA-SbtB complex following the onset of darkness.

In summary, this study has provided a molecular basis for the design of future *in vivo* studies to test a number of hypotheses relating to the regulation of the SbtA-SbtB complex that will be required to propose a comprehensive model describing the regulation of SbtA-mediated HCO_3^- transport by SbtB. First, rising ATP concentrations (due to exposure to light) are likely to be an allosteric signal to release the inhibitory SbtB protein from the StbA HCO_3^- transporter. Second, fluctuations in Ca²⁺ levels are likely to modulate the effect of changing nucleotide concentrations. Third, the C-terminal extension, which is present in a subset of SbtBs, may play an additional role in modulating allosteric signaling by controlling ATP binding in a redox-sensitive fashion. These findings will guide future experiments that will ultimately contribute to attempts to use transgenic SbtA to improve the photosynthetic performance and yield of C3 crop plants by increasing the steady state concentration of C_i in the chloroplast stroma [2].

Conflict of Interest

The authors declare that they have no conflicts of interest with the contents of this article.

Corresponding Author

*To whom correspondence may be addressed. Email: colin.jackson@anu.edu.au,

dean.price@anu.edu.au

ABBREVIATIONS (non-standard)

C_i, inorganic carbon CCM, CO₂-concentrating mechanism SbtA, sodium-dependent bicarbonate transporter A SbtB, sodium-dependent bicarbonate transport protein B

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ACCESSION CODES

The atomic coordinates and structure factors for apo-SbtB7001 (PDB: 6N4A), AMP:SbtB7001 (PDB: 6MMO), ADP:SbtB7001 (PDB: 6MMC), cAMP:SbtB7001 (PDB: 6MMQ),

ATP:SbtB7001 (PDB: 6NTB) and Ca2+ATP (PDB: 6MM2) have been deposited in the Protein

Data Bank (<u>http://wwpdb.org/</u>).

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