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## 1 **RESEARCH ARTICLE:**

- 2 Quantification of reactive oxygen species production by the red fluorescent proteins
- 3 KillerRed, SuperNova and mCherry.
- 4

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#### 22 ABSTRACT

Fluorescent proteins can generate reactive oxygen species (ROS) upon absorption of photons 23 via type I and II photosensitization mechanisms. The red fluorescent proteins KillerRed and 24 SuperNova are phototoxic proteins engineered to generate ROS and are used in a variety of 25 26 biological applications. However, their relative quantum yields and rates of ROS production are unclear, which has limited the interpretation of their effects when used in biological 27 systems. We cloned and purified KillerRed, SuperNova, and mCherry - a related red 28 29 fluorescent protein not typically considered a photosensitizer - and measured the superoxide 30  $(O_2^{-})$  and singlet oxygen (<sup>1</sup>O<sub>2</sub>) quantum yields with irradiation at 561 nm. The formation of the O2<sup>--</sup>-specific product 2-hydroxyethidium (2-OHE<sup>+</sup>) was quantified via HPLC separation with 31 fluorescence detection. Relative to a reference photosensitizer, Rose Bengal, the O2-32 quantum yield ( $\Phi O_2^{-}$ ) of SuperNova was determined to be 0.00150, KillerRed was 0.00097. 33 and mCherry 0.00120. At an excitation fluence of 916.5 J/cm<sup>2</sup> and matched absorption at 561 34 nm, SuperNova, KillerRed and mCherry made 3.81, 2.38 and 1.65 µM O<sub>2</sub> /min, respectively. 35 Using the probe Singlet Oxygen Sensor Green (SOSG), we ascertained the <sup>1</sup>O<sub>2</sub> quantum yield 36  $(\Phi^1O_2)$  for SuperNova to be 0.0220, KillerRed 0.0076, and mCherry 0.0057. These 37 photosensitization characteristics of SuperNova, KillerRed and mCherry improve our 38 understanding of fluorescent proteins and are pertinent for refining their use as tools to 39 40 advance our knowledge of redox biology.

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44 KEYWORDS: redox, optogenetics, superoxide, singlet oxygen, photosensitizer, quantum
45 yield.

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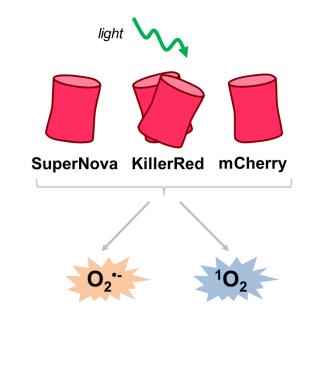
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# 48 ABBREVIATIONS:

49	2-OHE+:	2-hydroxyethidium
50	CALI:	Chromophore assisted light inactivation
51	DHE <sup>+</sup> :	Dihydroethidium
52	E+:	Ethidium
53	O <sub>2</sub> :	Superoxide
54	<sup>1</sup> O <sub>2</sub> :	Singlet oxygen
55	Ф:	Quantum yield
56	ΦO2 <sup></sup> :	Superoxide quantum yield
57	Φ <sup>1</sup> O <sub>2</sub> :	Singlet oxygen quantum yield
58	ROS:	Reactive oxygen species
59	SOSG:	Singlet oxygen sensor green
60	SOD:	Superoxide dismutase
61	X:	Xanthine
62	ХО	Xanthine oxidase
63		

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# **GRAPHICAL ABSTRACT**



#### 69 **INTRODUCTION**

70 Fluorescent proteins generate reactive oxygen species (ROS) upon irradiation by type I or type II photosensitization mechanisms [1-4]. The type I mechanism involves electron 71 transfer reactions that ultimately reduce molecular oxygen to form superoxide ( $O_2^{-}$ ) [3, 5]. 72 73 Type II photosensitization involves the direct energy transfer from excited triplet state of the 74 photosensitizer to oxygen to generate singlet oxygen  $({}^{1}O_{2})$  [4-7]. Both O<sub>2</sub> and  ${}^{1}O_{2}$  can be formed by fluorescent proteins [4, 5] but the relative yields or fluxes depend on various factors, 75 76 including the protein structure surrounding the chromophore, the oxygen concentration, 77 temperature, and pH of the environment [3, 5].

78 A range of phototoxic fluorescent proteins have been developed such as KillerRed, 79 KillerOrange, SuperNova, miniSOG and their derivatives; however their phototoxic properties 80 are not fully characterized [1-3, 8-11]. KillerRed, a dimeric red fluorescent protein, was derived from a random and site-directed mutations of a jellyfish protein, anm2CP [1, 3, 10, 12]. 81 82 KillerRed has a unique structure with a water channel to the chromophore that is responsible for its phototoxicity [1, 3, 10, 12]. The original KillerRed protein is prone to variable levels of 83 dimerization, which can lead to artifacts and mislocalization of fusion proteins within a 84 85 biological system [8]. These confounding factors can be mitigated by using the pseudo-86 monomeric version tandem KillerRed (tdKillerRed), which consists of two repeats of the KillerRed coding sequence, meaning that all copies are expressed as a dimer. SuperNova 87 was derived from KillerRed and retains similar phototoxic properties but exists as a monomer. 88 thereby limiting potential mislocalization events [8]. Both KillerRed and SuperNova are used 89 90 in a variety of applications ranging from localized ROS production to cell ablation, however the quantities or the species of ROS responsible for the effect are often unclear. KillerRed has 91 been used for chromophore- assisted light inactivation (CALI) in cells and organelles [1, 13-92 16]. These phototoxic effects have been shown to be sensitive to superoxide dismutase 93 94 (SOD), catalase, and sodium azide [1, 8], suggesting that KillerRed possesses the capacity to generate both O2<sup>-</sup> (and subsequently hydrogen peroxide) and <sup>1</sup>O2 oxidants [1, 2, 8]. Likewise, 95

SuperNova has been shown to oxidize DHE and ADPA probes, implying that it too generates both  $O_2$  and  ${}^1O_2$  oxidants [8, 17].

Although the phototoxic effects of these fluorescent proteins to cellular functioning 98 have been widely demonstrated, their precise ROS quantum yields, i.e. the ratio of ROS 99 100 molecules generated per photon absorbed by the fluorophore, and intrinsic rates of ROS production have not previously been reported. Therefore, the aim of this study was to 101 determine the quantum yields and rates of ROS production by phototoxic fluorescent proteins. 102 Using Rose Bengal, a well-characterized chemical photosensitizer molecule with a defined 103 104  $O_2^{-}$  quantum yield ( $\Phi O_2^{-}$ ) of 0.2 and  $^1O_2$  quantum yield ( $\Phi^1O_2$ ) of 0.75 as a standard [18], we determined the relative  $O_2$  and  $^1O_2$  quantum yields of KillerRed and SuperNova. As a 105 negative control for photosensitization we used mCherry, a red fluorescent protein commonly 106 used as an 'inert' fluorophore in many cellular imaging applications [3, 8]. Overall, we report 107 108 the  $O_2^{-}$  and  $O_2^{-}$  quantum yield of the fluorescent proteins tdKillerRed and SuperNova, as well as mCherry. 109

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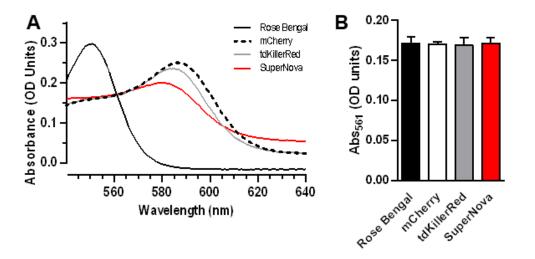
#### 111 MATERIALS AND METHODS

#### 112 **Protein cloning and purification**

SuperNova, tdKillerRed, and mCherry were transformed and grown in a culture as 113 114 previously described [8, 17]. SuperNova/pRSETB was a gift from Dr. Takeharu Nagai (Addgene plasmid # 53234) [8]. mCherry (pmCherry-C1) and tdKillerRed (#FP963, Evrogen) 115 were amplified and ligated into pRSETB using BamHI and EcoRI. Plasmids were then 116 117 transfected into JM109 (DE3) XJ autolysis cells, and protein expression was induced with 118 isopropyl β-D-1-thiogalactopyranoside (IPTG). Cultures were centrifuged at 3200 g for 10 min, washed with PBS and flash frozen. Cell lysate was run through nickel beads, then protein was 119 eluted with 100 µM imidazole in the presence of protease inhibitors (Roche) and desalted 120 using a PD-10 column. Protein concentration was determined by Lowry assay, and 121

absorbance scans were performed on a spectrophotometer (Shimadzu) to identify a region of spectral overlap in absorbance maxima between the proteins and Rose Bengal dye (# 330000, Sigma). The most robust overlap occurred between 550-580 nm (Fig. 1). Based on this, a 561 nm laser was chosen for subsequent experimentation. Proteins and Rose Bengal were diluted to achieve equal molar absorptivity at 561 nm using the Beer-Lambert equation (A= $\epsilon$ \*b\*c).

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Fig. 1. Equal photosensitizer absorbance at 561 nm. (a) Absorbance spectrum of photosensitizers. (b) Absorbance at 561 nm after adjustment of concentration of Rose Bengal dye (0.0026 mg/mL), mCherry (0.22 mg/mL), tdKillerRed (0.25 mg/mL), and SuperNova (0.76 mg/mL). Values are mean  $\pm$  SD for n = 3 independent experiments; p > 0.05 by one-way ANOVA.

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#### 134 Irradiation parameters

Irradiation of fluorescent proteins and the photosensitizing dye, Rose Bengal, were performed using a 561 nm class IIIb 50 mW diode laser (#1230935, Coherent® OBIS<sup>™</sup>, Edmund Optics, NJ, USA). The 0.7 mm diameter beam was focused through a 20x, 0.4 NA microscope objective lens (Swift) into a 200 µm core diameter, 0.22 NA SMA-terminated fiber optic cable (Part # M25L05, ThorLabs, Inc., Newton, NJ) for delivery to the sample. The fiber and objective lens were positioned using a Multimode Fiber Coupler Assembly (Part # F-91C1-T, Newport Corporation, Irvine, CA). Fiber output was collimated with an aspheric lens
(Part # A397TM-B, Thorlabs) to create a 2.5 mm-diameter collimated beam to irradiate each
200 µL sample volume contained within a 1.5 mL, 1 cm polystyrene cuvette (#97000-586,
VWR). The irradiance was measured as 25 mW at the front surface of the sample cuvette
using thermopile detector (818P-010-12, Newport Corporation, Irvine, CA) for all irradiation.
Fluence/light dose (J/cm<sup>2</sup>) was modulated by adjusting irradiation time while maintaining a
consistent fluence rate (mW/cm<sup>2</sup>).

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#### 149 **Determination of photobleaching rates**

150 Photobleaching rates of photosensitizers (Rose Bengal, 0.0026 mg/ml; mCherry, 0.22 mg/ml; KillerRed, 0.25 mg/ml; SuperNova, 0.76 mg/ml) and the probe DHE alone and in 151 combinations were determined in buffer (D-MRB; 220 mM Manitol, 70 mM Sucrose, 5 mM 152 MOPS, 2 mM EGTA, 0.4% FFBSA, 0.1 mM DTPA, pH 7.3) at 20 °C. The fluorescence signal 153 154 (Ex 525 nm; Em 550 nm) was acquired using a fluorescence spectrophotometer (Cary Eclipse, Agilent Technologies) during a cumulative time exposure (0-30 min) at 561 nm irradiation for 155 determination of the reduction in fluorescence. To determine the bleaching rates with SOSG, 156 DHE was replaced with SOSG in the buffer, and the change in absorbance was measured 157 158 between 400 – 800 nm using a spectrophotometer.

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#### 160 Xanthine oxidase superoxide production

161 Xanthine oxidase (XO) production of  $O_2^{-}$  was determined as the rate of SOD-sensitive 162 cytochrome *c* reduction, as previously described [7, 19]. Briefly, XO (0.25, 0.50, 1.0 and 4.0 163 mU/mL) was added to a 1 cm cuvette containing cytochrome *c* (40 µM) in PBS containing 164 DTPA (D-PBS: 7.78 mM Na<sub>2</sub>HPO<sub>4</sub>, 2.20 mM KH<sub>2</sub>PO<sub>4</sub>, 0.1 mM DTPA, pH 7.3). All reactions 165 were carried out at ambient O<sub>2</sub> and where indicated catalase (4200 U/mL) or SOD (800 U/mL) 166 was present. Baseline was collected for 2 min before 1 mM of xanthine (X) was added to initiate the reaction. Cytochrome *c* reduction was monitored at 550 nm for 10 min, and the rate was calculated using an extinction coefficient of  $18.7 \text{ mM}^{-1} \text{ cm}^{-1}$  [20].

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# 170 Superoxide quantification

The oxidation of dihydroethidium (DHE) yields the  $O_2$  specific fluorescence product 2-171 172 hydroxyethidium (2-OHE<sup>+</sup>) along with non-specific fluorescent products including ethidium (E<sup>+</sup>), which were separated using HPLC as previously described [7, 17, 21, 22]. Briefly, XO (4 173 mU/mL) and X (1 mM) were incubated in D-PBS at 20 °C for the indicated time (0 – 60 min). 174 Rose Bengal (0.0026 mg/mL), mCherry (0.22 mg/ml), tdKillerRed (0.25 mg/ml), or SuperNova 175 (0.76 mg/ml) were irradiated at 561 nm for the indicated time (0 – 30 min) in D-MRB in the 176 presence of DHE (100 µM). For experiments containing photosensitizers, the absorbance was 177 measured (400-800 nm) both pre- and post-irradiation at 561 nm. To these samples, an equal 178 volume of 200 mM HClO<sub>4</sub>/MeOH was added, centrifuged at 17,000 x g, and the supernatant 179 180 transferred to an equal volume 1 M K<sup>+</sup>PO<sub>4</sub><sup>-</sup> at pH 2.6.

Samples were separated using a Polar-RP column (Phenomenex, 150 x 2 mm; 4µm) on a Shimadzu HPLC with fluorescence detection (RF-20A). The flow rate was constant (0.1 mL/min) using a gradient of two mobile phases (A: 10% ACN, 0.1 %TFA; B: 60% ACN, 0.1 %TFA). The gradient was the following: 0 min, 40% B; 5 min, 40% B; 25 min, 100% B; 30 min, 100% B; 35 min, 40% B; 40min, 40% B. Standard curves were generated against known concentrations of E<sup>+</sup> and 2-OHE<sup>+</sup>, and peaks were quantified using Lab Solutions (Shimadzu) [7, 17].

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#### 189 Singlet oxygen quantification

190 The  ${}^{1}O_{2}$  production of photosensitizers (Rose Bengal, 0.0026 mg/mL; mCherry, 0.22 191 mg/ml; KillerRed, 0.25 mg/ml; SuperNova, 0.76 mg/ml) was measured using SOSG (1  $\mu$ M, #S36002, Invitrogen) in D-MRB at 20°C [7]. The SOSG signal (Ex 525 nm; Em 550 nm) was
acquired using a Cary Eclipse fluorescence spectrophotometer (Agilent Technologies) preand post- 561 nm irradiation for determination of the change in SOSG fluorescence intensity
[7].

196

#### 197 Calculations and statistical analysis

Fluorescent protein O<sub>2</sub> and <sup>1</sup>O<sub>2</sub> quantum yields were determined after correcting for 198 the bleaching rates of the photosensitizers, as we have previously demonstrated the 199 importance of photobleaching in explaining time-dependent ROS production by 200 photosensitizers [23]. Measurements of fluorescence/absorbance vs. illumination duration 201 202 were first normalized to the value prior to illumination, and then fit with an equation of the form  $B(t) = ae^{-bt}$ , where a and b are fit coefficients and t is the illumination duration in seconds. 203 The total number of absorbed photons for a sample can then be expressed as A =204  $A_0 \int_0^{t_d} B(t) dt$ , where  $A_0$  is the absorption prior to illumination,  $t_d$  is the illumination duration, 205 206 and B(t) is the bleaching curve described above. Relative to a reference quantum yield ( $\Phi_R$ ), the quantum yield of a sample ( $\Phi_s$ ) can be determined by  $\Phi_s = \frac{out_{S/A_s}}{out_{R/A_p}} \cdot \Phi_R$ , where *out* is the 207 output of interest and A is the total number of absorbed photons, as described above. 208 209 Incorporating correction for bleaching of the sample and reference, with knowledge that pre-210 illumination ( $A_0$ ) is equal for all samples, the quantum yield can be expressed as:

211 
$$\Phi_S = \frac{out_S}{out_R} \cdot \frac{\int_0^{t_R} B_R(t)dt}{\int_0^{t_S} B_S(t)dt} \cdot \Phi_R,$$

where  $out_S$  and  $out_R$  are measured outputs for illumination durations of  $t_S$  and  $t_R$  for the sample and reference, respectively, and  $B_s$  and  $B_R$  are the corresponding bleaching curves. All fitting and calculation was performed in MATLAB (The MathWorks, Inc., Natick, MA).

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216 O2<sup>-</sup> production rates of the fluorescent proteins were calculated based upon the standard curve generated using X/XO. Since the apparent number of O<sub>2</sub><sup>--</sup> molecules required 217 to generate one 2-OHE<sup>+</sup> molecule is dependent on the rate of  $O_2^{-}$ , photosensitizer  $O_2^{-}$ 218 production was matched with 4 mU/mL XO superoxide generation. Under these conditions 219 220 X/XO produced 2.24 µM O<sub>2</sub><sup>-</sup>/min. X/XO was incubated (0-60 min) of DHE and 2-OHE<sup>+</sup> was measured and plotted against the expected cumulative O2<sup>-</sup> concentration generated during 221 that time, as previously described [7]. At these lower rates the ratio of  $O_2^{-1}$  to 2-OHE<sup>+</sup> was 222 linear (y = 55.62(x) + 326.2;  $R^2 = 0.98$ ). 223

224 Statistical analysis: Data were first tested for normality of variance, and were then 225 analyzed by one- or two-way ANOVA with Tukey's post hoc using GraphPad Prism (v7).

226

#### 227 **RESULTS**

#### 228 **Purification and characterization of fluorescent proteins.**

Fluorescent proteins subjected to SDS-PAGE migrated at their expected molecular 229 weight (Supplemental Fig. 1). In order to measure protein photosensitization characteristics 230 relative to a reference dye (Rose Bengal), we first sought to determine i) a wavelength that 231 232 was near the absorption maxima for each chromophore, ii) a concentration of each chromophore in solution that would allow all of the photosensitizers absorb an equal number 233 234 of photons and iii) is not confounded by absorption of photons by other reagents used for detection of ROS. We determined from absorbance spectra that excitation at 561 nm met each 235 of these criteria (Fig. 1A), and photosensitizer concentrations were then optically matched for 236 237 equal absorbance at 561 nm (Fig. 1B).

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## 242 Superoxide quantum yield and superoxide generation rate of fluorescent proteins

We measured light-dependent photosensitizer O<sub>2</sub><sup>-</sup> generation using HPLC to quantify 243 2-OHE<sup>+</sup>, a O<sub>2</sub><sup>-</sup> specific reaction product of DHE [7, 24-26]. Since the known yield of Rose 244 Bengal served as our reference, we confirmed that Rose Bengal produced 2-OHE<sup>+</sup> in a light 245 246 dose-dependent manner (Fig. 2A). Similarly, the fluorescent proteins tdKillerRed, SuperNova and mCherry also produced 2-OHE<sup>+</sup> in a light dose-dependent manner (Fig. 2B), yet the 247 magnitude of 2-OHE<sup>+</sup> for the protein photosensitizers was considerably lower than that of 248 Rose Bengal. For example, after 60 seconds of illumination Rose Bengal generated ~17,000 249 250 pmol/mL 2-OHE<sup>+</sup>, while after 300 seconds the fluorescent proteins produced ~500 pmol/mL (Fig. 2B). 251

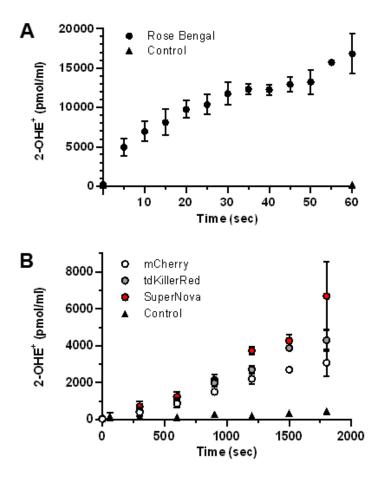


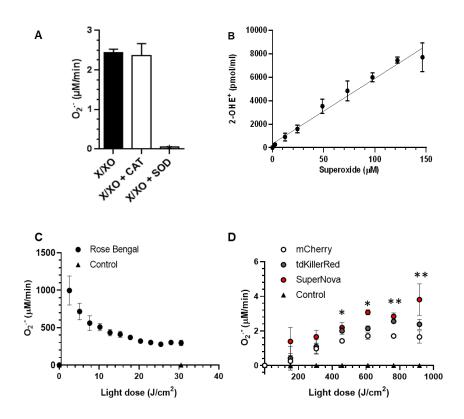
Fig. 2. Light-dependent superoxide generation by photosensitizers. (a) Rose Bengal, (b) tdKillerRed, Supernova, mCherry and control (no photosensitizer) were irradiated with equal molar absorptivity at 561 nm in the presence of DHE (100  $\mu$ M) for quantification of 2-OHE<sup>+</sup>. Values are mean ± SD for n = 3 independent experiments.

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Next, we sought to determine the O2<sup>-</sup> quantum yield of tdKillerRed, SuperNova and 258 mCherry relative to Rose Bengal, with a known  $\Phi O_2^{-}$  of 0.2 [18]. The determination of quantum 259 yields relies on the equal absorbance of photons, yet photobleaching results in a decrease in 260 photon absorbance over time that occurs at different rates between photosensitizers. We 261 262 therefore measured the rate of photosensitizer bleaching by assessing the change in fluorescence in response to a cumulative light-dose. We then corrected for the bleaching rates 263 264 of the individual fluorophores and the probe, DHE, (Supplemental Fig. S2) in order to calculate the  $\Phi O_2^{\bullet}$  relative to Rose Bengal. We thus determined that SuperNova had a  $\Phi O_2^{\bullet}$  of 0.0015, 265 266 and tdKillerRed's  $\Phi O_2^{-}$  was 0.00097; mCherry had a comparable  $\Phi O_2^{-}$  (Table 1).

267 We next sought to calculate the  $O_2$ <sup>-</sup> production rate of fluorophores. However, the 268 apparent ratio of O<sub>2</sub><sup>-</sup> molecules necessary to form one molecule of 2-OHE<sup>+</sup> is highly dependent of the rate of O<sub>2</sub><sup>-</sup> generation, possibly due to competition with spontaneous 269 dismutation [7, 26]. Therefore, we generated a standard curve using a concentration of 270 271 xanthine oxidase that produces  $O_2$  at a similar rate to that of the photosensitizers. Based on the results of the dose response (Supplementary Fig. S3), we selected 4 mU/mL of xanthine 272 oxidase (Fig. 3A) to match the 2-OHE<sup>+</sup> production rates from our photosensitizers at this 273 concentration and light dose. We determined that 4 mU/mL of xanthine oxidase produces 2.44 274  $\mu$ M/min of O<sub>2</sub>, which was SOD-sensitive and catalase-insensitive (Fig. 3A). We incubated the 275 same amount of xanthine oxidase in the presence of DHE and measured the formation 2-276 OHE<sup>+</sup> over time and expressed it as a function of expected cumulative O<sub>2</sub><sup>-</sup> production. Our 277 results show a linear increase of 2-OHE<sup>+</sup> with increasing amounts of O<sub>2</sub><sup>-</sup> across the tested 278 range ( $R^2 = 0.98$ ; Fig. 3B). Given that the photosensitizers absorbed an equal amount of light 279

280 and hence have the same ability to make ROS (Fig. 1), we then used this equation to derive the rate of O<sub>2</sub><sup>-</sup> production by photosensitizers per unit light dose (Fig. 3C & D) from the data 281 in Fig. 2. Rose Bengal had the highest rates of O2<sup>-</sup> production across light doses (~300 µM 282  $O_2^{-}$  /min at 30.55 J/cm<sup>2</sup>) with mCherry producing the least amount  $O_2^{-}$  per light dose (~1.65 283 284 µM O<sub>2</sub><sup>--</sup> /min at 916.5 J/cm<sup>2</sup>) (Fig. 3C & D). The rate of O<sub>2</sub><sup>--</sup> production by Rose Bengal decreased with increasing light dose, which is consistent with the bleaching rate of Rose 285 Bengal (Supplemental Fig. 2). The progressive loss of absorption resulted in a fluence-286 dependent decrease in the O2<sup>-</sup> production rate. At the light doses tested, each of the 287 288 fluorescent proteins showed an increasing O<sub>2</sub><sup>-</sup> production rate that reached a plateau around 600 J/cm<sup>2</sup> (Fig. 3D). The gradual increase in the measured O<sub>2</sub> - production rate could be due 289 290 to saturation of local reaction sites, such as amino acids, that can quench ROS, or a 291 conformational change in the protein resulting in a maximal observed production rate, as has 292 been reported for other fluorescent proteins [27]. As the fluorescent proteins bleach at a slower rate compared to Rose Bengal (Supplemental Fig. 2), we did not observe the same decrease 293 294 in  $O_2^{-}$  production rate that was observed for Rose Bengal.

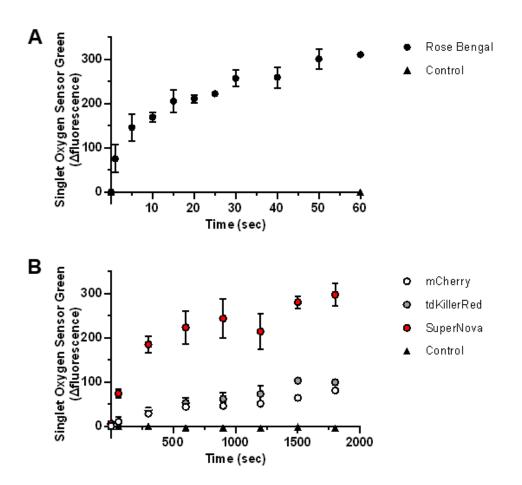


296 Fig. 3. Determination of superoxide production per light dose. (a) xanthine/xanthine oxidase (X/XO) O2. 297 production was assessed using cytochrome c reduction assay. Xanthine oxidase (XO, 4 mU/mL) and xanthine (X, 298 1 mM) were incubated with catalase (CAT) superoxide dismutase (SOD) where indicated. \*p < 0.05 vs X/XO and 299 X/XO+CAT, one-way ANOVA, Tukey post hoc. (b) Time course (0-60 min) of X/XO O2<sup>+</sup> generation was measured 300 using HPLC separation of 2-OHE<sup>+</sup> and then plotted against the expected O<sub>2</sub><sup>--</sup> production. (c) Rose Bengal O<sub>2</sub><sup>--</sup> 301 production rate per light dose. (d) Fluorescent protein (mCherry, tdKillerRed and SuperNova) O2<sup>+</sup> production per 302 light dose. Data from (c) and (d) are derived from data presented in Fig. 2. \*p < 0.05 SuperNova vs mCherry, \*\* p 303 < 0.05 SuperNova and tdKillerRed vs mCherry, two-way ANOVA, Bonferroni post hoc. Values are mean ± SD for 304 n = 3 independent experiments.

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#### Singlet oxygen quantum yield of fluorescent proteins 306

Singlet oxygen sensor green (SOSG) specifically detects <sup>1</sup>O<sub>2</sub> [4, 7, 27, 28] and does 307 308 not react with other ROS, such as  $O_2^{-1}$  or the hydroxyl radical, making it a suitable  ${}^1O_2$  detector under conditions were multiple ROS are being generated [29]. We assessed the <sup>1</sup>O<sub>2</sub> 309 production of the photosensitizers by measuring the relative change of SOSG fluorescence 310 and correcting for the bleaching rate of the individual fluorophores (Supplemental Fig. 2) [18]. 311 Rose Bengal had the greatest SOSG fluorescence change with irradiation time (Fig. 4A) 312 relative to those of the fluorescent proteins (Fig. 4B). The  $\Phi^1O_2$  of the fluorescent proteins 313 were then calculated relative to the Rose Bengal reference  $\Phi^1O_2$  of 0.75 [18]. We determined 314 that SuperNova had the highest  $\Phi^1O_2$  at ~0.022, while mCherry had the lowest  $\Phi^1O_2$  of 315 316 ~0.0057 (Table 1). This demonstrates that mCherry, KillerRed and SuperNova are each capable of generating  ${}^{1}O_{2}$  in an irradiation dose dependent manner. 317



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Fig. 4. Singlet oxygen generation by photosensitizers in response to 561 nm irraditation. (a) Rose Bengal, (b) tdKillerRed, Supernova, mCherry and control (no photosensitizer) were irradiated with equal molar absorptivity at 561 nm in the presence of 0.1  $\mu$ M SOSG. The initial fluorescence reading (Ex 525 nm; Em 550 nm) was subtracted from the post-illumination reading and presented as the relative fluorescence change. Values are mean  $\pm$  SD for n = 3 independent experiments.

325

#### 326 **DISCUSSION**

The main findings from this study are that the red fluorescent proteins tdKillerRed, SuperNova, and mCherry each generate  $O_2^{-}$  and  ${}^1O_2$  via type I and II mechanisms, respectively. We also report for the first time quantitative ROS quantum yields for tdKillerRed, SuperNova and mCherry fluorescent proteins. 331 Genetically-encoded photosensitizers are used in a variety of biological applications to generate ROS in a light-dependent manner. They have the advantage of being targeted to 332 precise regions in the cell to provide spatial control over ROS production [2]. However, their 333 precise ROS producing characteristics are often overlooked provided that a biological 334 335 phenotype has been observed. In contrast to  $\Phi^1O_2$ , very little is known about fluorescent protein  $\Phi O_2^{-}$ . This may be a result of the limited methods to selectively detect  $O_2^{-}$ , although 336 one study has reported the  $\Phi O_2^{-}$  and  $\Phi^1 O_2$  of red fluorescent protein, TagRFP [4]. Using a 337 similar SOSG detection approach, the  $\Phi^1O_2$  was estimated at 0.004, while the  $\Phi O_2^{-}$  was 338 estimated at <0.0002 using DHE bleaching as a measure of  $O_2$  [4]. 339

The first developed photosensitizer protein, KillerRed, was initially reported to make 340 O<sub>2</sub>- and <sup>1</sup>O<sub>2</sub> [1, 10]. Subsequently, literature has suggested that the KillerRed 341 photosensitization mechanism selectively produces  $O_2^{-}$  and relies on the water channel to the 342 343 chromophore for its phototoxicity [10, 30]. Depending on the application, one type of ROS may predominate in contributing to the light-induced effect. For example, <sup>1</sup>O<sub>2</sub> played a role in 344 KillerRed CALI experiments [1], while O2<sup>-</sup> mediated phototoxicity [31]. While our results 345 demonstrate that both <sup>1</sup>O<sub>2</sub> and O<sub>2</sub> are capable of being produced, researchers should 346 347 consider which species is relevant to their particular biological application.

348 SuperNova was derived from KillerRed, and it would be reasonable to assume that the photosensitization mechanisms would be similar. Accordingly, SuperNova has been thought 349 to produce O<sub>2</sub><sup>-</sup> and <sup>1</sup>O<sub>2</sub>, as measured by 2-OHE<sup>+</sup> formation [17] and ADPA photobleaching 350 [8], respectively. In the present study, SuperNova's comparatively larger ROS quantum yield 351 352 than KillerRed is consistent with previous reports of greater phototoxicity [8]. Specifically, at 916.5 J/cm<sup>2</sup> of fluence, we show that the SuperNova  $O_2$  production rate is ~1.6 fold higher 353 than KillerRed. However, the O2<sup>-</sup> production rate was not consistent across fluences tested, 354 and plateaued at the highest light dose tested. While the quantum yields provide a direct 355 356 comparison of the phototoxic mechanisms of the red fluorescent proteins tested, caution is 357 warranted when extrapolating these findings in vivo. The  $O_2$  and pH gradients or endogenous

358 chromophores present in the cellular milieu can affect the ROS generation by 359 photosensitizers. For example, a high  $O_2$  tension could favor  ${}^1O_2$  production, while hypoxic 360 conditions could favor  $O_2^{-}$  production [32].

Unlike KillerRed and SuperNova that were derived from the jellyfish protein anm2CP, 361 mCherry was derived from the sea anemone protein DsRed. Owing to the structural 362 differences that exist due to their independent lineage, mCherry lacks a water channel, 363 suggesting that it would not be as phototoxic as KillerRed. Indeed, it is widely used in biological 364 applications under the assumption that it is photochemically inert. However, some previous 365 366 reports have also shown that mCherry can be phototoxic [33] and that it produces  $O_2$  [8, 34] and  ${}^{1}O_{2}$  [8] upon irradiation. Our present findings are in agreement with this and indicate that 367 mCherry actually displays similar  $\Phi O_2^{-}$  and  $\Phi^1 O_2$  as 'professional' photosensitizer proteins. 368

The genetically-encoded photosensitizers display  $\Phi O_2^{-1}$  and  $\Phi^1 O_2$  that are orders of 369 magnitude lower than the chemical photosensitizer Rose Bengal. Yet, despite their lower 370 371 quantum yields, their ability to generate a biologically relevant effect is well established [1, 2, 372 8, 17]. Once formed by the excited chromophore, the superoxide anion must escape the protein barrel structure in order to be released to the surrounding environment and react with 373 374 the ROS probe [10]. The protein barrel likely shields the release of ROS, potentially explaining 375 the lower observed  $\Phi O_2^{\bullet}$  of the protein photosensitizers compared to Rose Bengal which can directly release oxidants to the surrounding aqueous environment. Nevertheless, our current 376  $\Phi^1O_2$  findings are generally in agreement with other fluorescent proteins that have been 377 reported to range from  $\Phi^1O_2$  0.004 to 0.030 [3]. Recently, optimized variants have reportedly 378 379 reached  $\Phi^1 O_2 \sim 0.6$  [35]. New approaches are aimed at combining the large quantum yields of chemical photosensitizers with the advantages of genetically-encoded photosensitizers [36]. 380

381

#### 382 Conclusion

Overall, we demonstrate that the red fluorescent proteins tdKillerRed, SuperNova and mCherry are able to photosensitize  $O_2^{-}$  and  ${}^1O_2$ . Our studies provide  $\Phi O_2^{-}$ ,  $\Phi^1O_2$  and rates of  $O_2^{-}$  production across light doses. Our findings will help elucidate mechanisms mediated by phototoxic proteins and aid in the development of efficient or selective ROS production by genetically-encoded photosensitizers [35, 37].

388

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397

398 **Conflicts of interest**: There are no conflicts of interest to declare.

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# 500 **TABLE 1:**

# 501 Superoxide and singlet oxygen quantum yield of mCherry, tdKillerRed, and SuperNova.

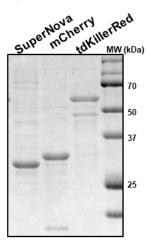
Fluorescent proteins	O₂ <sup></sup> Quantum yield (φO₂ <sup></sup> )	$^{1}O_{2}$ Quantum yield ( $\phi^{1}O_{2}$ )
mCherry	0.00120±0.000044	0.0057±0.00027
tdKillerRed	0.00097±0.000042	0.0076±0.00026
SuperNova	0.00150±0.000016	0.0220±0.00180

502

503 Data are mean  $\pm$  SD for n=3 independent experiments

# 505 Supplemental Fig. 1.

Fig S1



Coomassie stain

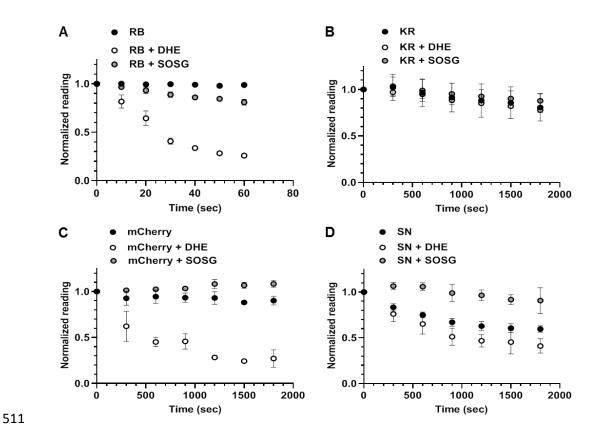
506

507 Supplementary Fig. 1. Purified fluorescent proteins. Purified SuperNova, mCherry, and tandem KillerRed protein

508 were denatured in buffer containing 100 mM Tris HCl, 4% w/v SDS, 10% v/v glycerol, 0.2% w/v bromophenol blue,

509 2% v/v  $\beta$ -mercaptoethanol then heated for 5 min at 95°C before separation by SDS-PAGE (10  $\mu$ g per lane) and

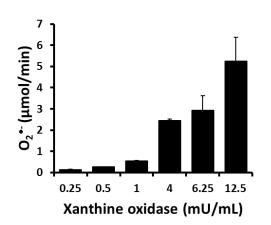
510 then stained with coomassie blue.



Supplementary Fig. 2. Photobleaching rates of photosensitizers with DHE or SOSG. (A) Rose Bengal, (B) KillerRed, (C) mCherry, and (D) SuperNova alone or in the presence of dihydroethidium (DHE) or Singlet Oxygen Sensor Green (SOSG). The cumulative fluorescence (photosensitizer alone or with DHE) or cumulative absorbance (photosensitizer with SOSG) was measured (0-60 sec for Rose Bengal; 0-30 min for KillerRed, mCherry, and SuperNova) following irradiation at 561 nm. Data were then normalized to baseline after bleaching fit correction using MATLAB. Data are N = 3, mean  $\pm$  SD.

518

520



521 522

- 523 Supplementary Fig. 3. Dose dependent superoxide production of xanthine oxidase. Xanthine oxidase (0.25,
- 524 0.5, 1, 4, 6.25 and 12.5 mU/mL) superoxide production was assessed with in the presence xanthine (1mM) using
- 525 cytochrome c reduction. Data are N = 4 independent values, mean ±SD.