Research Article

Title:

Laboratory testing of very low-copper-treated water to prolong pupation and

emerging time of mosquito larvae: an alternative method to delay mosquito

breeding capability

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Abstract

Introduction: Larvicide application in ovitrap is one of the currently available

methods used in mosquito eradication campaign because they eliminate the larval

stage. We previously reported that copper in liquid form is a promising candidate

due to its potent larvicide properties in a laboratory setting and even in the field. In

the field study, several larvae survive in the ovitraps due to the dilution of copper

concentration by the rain. The surviving larvae were smaller and less motile. This

led our interest to study the effect of sub-lethal dose of copper in ovitrap on larvae

development, pupation time and lifespan in adult stage.

Methods: First instar larvae of three species of mosquito (*Aedes albopicus*, *Culex*

pipiens and *Anopheles stephensi*) in copper-treated water started from 0.60 ppm,

0.30 ppm, and 0.15 ppm and compared with the control. The surviving larvae were

recorded every day in terms of pupa emerging time and adult emerging time. The

number of adult mortality was recorded and compare with the control.

Results: Copper showed potent larvicide effect in the 0.60 ppm concentration and

demonstrate prolong in pupation and showed significantly lower number of

emerging number of mosquitoes up to the lowest 0.15 ppm. Adult lifespan was not

different with the control.

Conclusion: This study demonstrated the capability of copper below 1 ppm to

prolong pupation time and emerging time of mosquito larvae which provide the

possibility in application of copper to cut mosquito breeding capacity in order to

reduce their ability to transmit disease.

Keywords: *Aedes albopictus*, *Anopheles stephensi*, *Culex pipiens*, ovitrap, copper,

larval control, pupation time

INTRODUCTION

Eradication campaigns for mosquito-borne diseases are of the utmost

importance for public health worldwide. Vector-based interventions are

considered a more rational and feasible choice than the challenges and expenses

associated with discovering cures and vaccines for these diseases. This approach is

particularly suitable for under-developed and developing countries with limited

financial, technical, and human resources. Over the past century, the dominant

methods recommended for vector control have shifted [1] from larval source

management (LSM), house screening, and bednets [2] to indoor residual spraying

with effective contact pesticides [3, 4].

In urgent efforts to develop new vector control methods, novel approaches

have emerged, for example toxic nectar baits [5], repellents [6], oviposition traps

[7], and genetic modifications to mosquitoes [8]. These approaches need to be

combined in order to optimally function for mosquito eradication, and increasing

the potential of combinations of interventions is also required, including LSM [1]

Our recent findings on ability and properties of copper to kill mosquito

larvae [9] and laboratory testing of the effect of liquid copper at a concentration of

10ppm on 3 species of mosquito larvae [10] prompted us to perform a field test in

a dengue endemic area of West Sumatra, Indonesia. The result is satisfactory (11)

however rain dilution became a factor to be considered, since we observed some

survived larvae in the outdoor ovitraps. However, during our observation, the

survived larvae appear weak and smaller in size in comparison with the control.

Another study is required to determine if the survived larvae capable to continue

life cycle as mosquito until adults. This urge us to do a laboratory test of sub-lethal

concentration of copper to mosquito larvae to see the effect on mosquito

development and breeding capacity.

MATERIALS AND METHODS

Ethical Approval

This research was approved by the ethics committee of Faculty of

Medicine Andalas University (Approval No.036/KEP/FK/2019).

Mosquito Eggs and Mice

An. stephensi (strain SDA 500), were reared in our laboratory under room

temperature (26°C), 50-70 % relative humidity, and a 13:11 h light : dark cycle.

From the day of emergence, the mosquitos were provided with a 5% fructose

solution soaked in a filter paper. Females five days after emergence were allowed

to feed on anesthetized mice. Three days later, an ovipositing dish was placed in a

cage containing gravid females. The eggs were laid on a filter papers soaking in

the dish. The filter paper with eggs was placed on a 12 x 20 cm hatching tray

containing 500 ml of water. After hatching, 3-5 mg of carp food per tray was

sprinkled on the surface of the water twice daily. Twelve days later, pupae were

collected daily and transferred to cages for adult emergence.

Ae. *albopictus* and Cx. *Pipiens'* eggs were purchased and first instar newly

hatched larvae were utilized in the experiment. Female BALB/c mice were

purchased from SLC (Shizuoka, Japan). The mice were fed *ad libitum* and

exposed to a 13:11 h of light : dark cycle. This mosquito-mouse cycle was used to

maintain the mosquitoes in the laboratory.

Preparation of Copper Sulfate Solutions

Four $CuSO_4$ solutions were used for the experiments. The copper

concentrations were 0.6, 0.3, 0.15, and 0 ppm. Each solution was allocated to a

pre washed square plastic plate and cover with a plastic transparent plate. First, we

prepared a standard solution of 100 mM of $CuSO_4$ by mixing copper sulfate

powder with filtered water (Fine Ceramic Filter NGK insulators, Nagoya, Japan).

Then we prepared the four concentration for the experiments by diluting them

from the standard solution (100mM $CuSO_4$) and confirming the ppm level using a

Copper Measuring Device (Hanna Instruments, Tokyo, Japan) and Z-5010

Polarized Zeeman Flame Atomic Absorption Spectrophotometer (Hitachi Ltd,

Tokyo, Japan).

Preparation of Mosquito Larvae

First instar newly hatched larvae were used in this experiment. Fifty larvae

were separated and put inside each container. Three containers for each

concentration of copper were prepared. Larvae were allowed to hatch and

observed every 24 hours under the microscope for mortality. A fine ceramic filter

(NGK Insulators, Nagoya, Japan) was used to remove chlorine ions from tap water

for the experiments.

RESULT

After 7 days, copper treated larvae of An. *stephensi* and Cx. *pipiens*

showed one hundred percent mortality under 0.60 ppm of copper, with some small

number of larvae survived in Ae. albopictus group. This followed by about fifty

percent mortality under 0.30 ppm copper concentration. The concentration of 0.15

ppm is not showing significant effect on larvae mortality in all three mosquito

species (**Figure 1**).

Surviving larvae exposed to 0.15 and 0.30 showed prolonged pupation

time in all species observed when compared to the control. We also observed that

the surviving larvae of Ae. *albopictus* on 0.60 ppm did not became pupae (**Figure**

2).

Surviving pupae of all observed species exposed to 0.15 ppm of $CuSO_4$

become adult mosquito, albeit at a statistically significantly lower number when

compare to the control. Adult mosquito of all observed species emerged at later

days compared to control and survived up to 21 days of observation. Almost none

of the pupae of all observed species of 0.30 ppm of copper solution emerged as

adult mosquito (F**igure 3**).

However, all the adult emerged in copper treated groups and control appear

to be viable and showed no difference in term of the lifespan. It seemed that

copper effect to the survived larvae is not carried to adult stage and limited to

larvae stage and only prolonged emerging in the pupae stage.

DISCUSSION

The result of the present study demonstrated that copper adequately led to

a high rate of larval mortality in copper-treated group even below 1 ppm (0.60

ppm). The concentration below that (0.30) still killed half of the larvae and

together with the lowest concentration of 0.15 ppm prolonged pupation time and

emerging time when compared to the control. The number of mosquitos emerged

in afterward is also significantly lower than those in the control group.

However, the copper effect is limited in the larvae and pupae stage. Copper

showed no effect to adult stage. The effect on the larvae probably caused by

copper effect to commensal bacteria in the larvae midgut, which disrupt their

intestinal function. This prevent them to eat properly and collect enough energy

for pupation and emerge as adult mosquito. Once the larvae become adult, this

copper effect is no longer play a role in their life cycle.

This is an interesting finding, since in the very low concentration of copper

prolonged breeding time of mosquito. This might lead to the possibility to utilize

copper in the tap water to jeopardize mosquito-breeding capability. Mosquito will

need a longer time to breed, which then reduce their capability to transmit

mosquito borne disease such as dengue hemorrhagic fever.

It is also known that *Aedes* sp. had been resistant to insecticide in indoor or

outdoor spraying. A study conducted in Sri Lanka showed *Aedes aegypti* and

Aedes albopictus highly resistant to DDT and they now have the ability to oviposit

indoors and outdoors [12]. Another study in Bangladesh reported abundant

potential larval habitats for *Aedes* sp. in containers or jars spread around the city

[13]. These situations need another approach of vector control, such as laying

ovitraps or another effort to cut or delayed mosquito breeding capability, or even

provide insecticide in the tap water if possible.

The US EPA is suggesting 1 ppm as safe limit of copper concentration in

the drinking water (14), thus make the application of very low 0.15 ppm of copper

might be logically acceptable in the future.

We believed that, together with the application of copper as ovitrap, the

concentration of 0.15 ppm of copper in the tap water in houses and large scale

water reservoir might be a potent strategy in cutting mosquito breeding capability

and their ability to transmit mosquito borne disease, especially dengue

hemorrhagic fever, which mainly breed in the city water reservoir and container.

This result might be a chance to reevaluate the allowance of copper in drinking

water as one of possible strategy in eradicating mosquito borne disease in the

future.

Authors` contributions: MR conceived and designed the study protocol. MR and

CI formulated the study proposal and obtained funding. MR conducted research in

the laboratory. MR carried out data analysis and interpretation and wrote the

manuscript while CI conducted the statistical analysis. All authors read and

approved the final manuscript.

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25

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Figure 1. The observed mortality of *A. albopictus*, *A. stephensi*, and *C. pipiens* larvae on exposure to 0.15, 0.30, and 0.60 ppm of $CuSO_4$. *significantly different from control, $p < 0.05$ (t-test or Mann-Whitney U test);

Figure 2. The number of emerging pupae of *A. albopictus*, *A. stephensi*, and *C. pipiens* on exposure to 0.15, 0.30, and 0.60 ppm of CuSO₄. *significantly different from control, p<0.05 (Mann-Whitney U test)

Figure 3. The number of emerging *A. albopictus*, *A. stephensi*, and *C. pipiens* mosquito on exposure to 0.15, 0.30, and 0.60 ppm of CuSO₄. *significantly different from control, p<0.05 (Mann-Whitney U test)