Enhanced biodegradation of naphthalene by *Pseudomonas* sp. consortium immobilized in calcium alginate beads

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36 Abstract

37 Polycyclic aromatic hydrocarbons (PAHs) belong to a large group of organic
 38 pollutant which considers as a potential health hazard to living beings. Herein,
 39 naphthalene biodegradation potential by free and immobilized *Pseudomonas putida*
 40 strain KD10 and *Pseudomonas* sp. consortium were studied. Additionally, naphthalene
 41 1, 2-dioxygenase (*nahAc*) was sequenced and analyzed, which reveals two altered
 42 amino acid residues. However, the altered amino acid residues are not present in the
 43 vicinity of the active site. The gas-phase binding free energy (ΔG_{London}) of the mutant
 44 variant of naphthalene 1, 2-dioxygenase was $-7.10 \text{ kcal mol}^{-1}$ which closely resembles
 45 the wild type variant. Naphthalene biodegradation rate by *Pseudomonas putida* strain
 46 KD10 was $79.12 \text{ mg L}^{-1} \text{ day}^{-1}$ and it was significantly elevated up to $123 \text{ mg L}^{-1} \text{ day}^{-1}$
 47 by the immobilized *Pseudomonas* sp. consortium. The half-life ($t_{1/2}$) for naphthalene
 48 biodegradation was 3.1 days with the inhibition constant (k_i), substrate saturation
 49 constant (k_s) and maximum specific degradation rate constant (q_{max}) of 1268 mg L^{-1} ,
 50 395.5 mg L^{-1} and 0.65 h^{-1} , respectively, for the *Pseudomonas putida* strain KD10.
 51 However, the $t_{1/2}$ value was significantly reduced to 2 days along with k_i , k_s and q_{max}
 52 values of 1475 mg L^{-1} , 298.8 mg L^{-1} and 0.71 h^{-1} , respectively, by the immobilized
 53 *Pseudomonas* sp. consortium. The GC-MS data suggest that KD10 might follow D-
 54 gluconic acid mediated meta-cleavage pathway of catechol biodegradation. It is
 55 concluded that naphthalene biodegradation performance by immobilized *Pseudomonas*
 56 sp. consortium was superior to free or immobilized *Pseudomonas putida* KD10.
 57 Microbial consortium immobilization could be a useful tool for water quality
 58 management and environmental remediation.

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60 *Keywords:*

61 *Pseudomonas* sp., petroleum wastes, biodegradation, cell immobilization, mutant

62 naphthalene 1, 2-dioxygenase, rigid-flexible molecular docking.

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82 *Highlights*

- 83 • Superior naphthalene biodegradation by *Pseudomonas* sp. consortium
- 84 immobilized in calcium alginate beads.
- 85 • A common mutation prone amino acid stretch inside chain A of naphthalene 1,
- 86 2-dioxygenase has been identified.
- 87 • A new naphthalene biodegradation pathway by *Pseudomonas putida* strain
- 88 KD10 has been proposed.

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104 1. Introduction

105 Polycyclic aromatic hydrocarbons (PAHs) are considered as the potential health
 106 hazard for living beings (Kumari et al., 2018). Environmental agencies including US-
 107 EPA, European Union, Environment-Canada, registered PAH as the priority pollutants
 108 that require immediate human intervention (Wang et al., 2018). The physicochemical
 109 properties of PAH make them a major contributor for soil and groundwater
 110 contamination through bio-magnification (Norris, 2017). Moreover, according to the
 111 Environmental Health Hazard Assessment, U.S.A., naphthalene (NAP) concentration
 112 beyond 170 ppb is not safe drinking (Bruce et al., 1998). Different orthodox and
 113 expensive techniques for environmental remediation, *viz.*, incineration, gasification,
 114 plasma-gasification have been replaced by green-technologies such as bioremediation,
 115 phytoremediation, nanoremediation, *etc.* (Thomé et al., 2018). Bioremediation is
 116 considered as the most cost-effective and eco-friendly oil spill management technique
 117 (Wilson and Jones, 1993). However, while considering the vast volume of a mobile
 118 open water system, bioremediation stumble upon several limiting factors such as the
 119 low local concentration of the effective microorganisms, loss of active microorganisms,
 120 *etc.* (Chen et al., 2017). Conversely, cell immobilization provides several advantages
 121 such as, it helps to retain high local concentration of the effective microorganisms, keep
 122 intact bacterial cell membrane stability and can be stored for future reuse (Bhardwaj et
 123 al., 2000). Overall, cell immobilization can markedly improve the stability and
 124 efficiency of the bioremediation process (Mrozik and Piotrowska-Seget, 2010; Tyagi et
 125 al., 2011). Cell immobilization using calcium alginate beads (CABs) is a convenient
 126 option where maximum cells remain viable and can tolerate high concentration of the
 127 toxicant (Lee and Heo, 2000). Moreover, calcium alginate is nontoxic to the bacterial

cell and it has low production cost which facilitates easy reuse (Bhardwaj et al., 2000). Biodegradation of a toxicant (complex nutrient for bacteria) by the microbial consortium can efficiently enhance biodegradation rate (Kumari et al., 2018). This rate enhancement could be achieved by several ways such as, different biodegradation pathways of each individual bacterium (Dutta et al., 2018), a metabolic intermediates of one bacteria may act as the starting material of other bacteria (Surkatti and El-Naas, 2018), different genetic makeup (Woyke et al., 2006), synergetic effects of different microbial species, (Ghazali et al., 2004) or by synthesising different variant of catalytic enzymes. Bacterial cells in the microbial consortium can also combine their metabolic capabilities to utilize the common complex nutrient (Gilbert et al., 2003). Biodegradation of complex hydrocarbon mixture by the microbial consortium offers a combination of diverse enzymes which promotes the biodegradation processes (Wongwilaiwalin et al., 2010). However, microbial consortium immobilization for biodegradation of PAHs was not studied previously. The aim of the present study is to evaluate the naphthalene biodegradation performance by *Pseudomonas putida* strain KD10 and *Pseudomonas* sp. consortium as a free and immobilized format. Previous study showed that *Pseudomonas putida* strain KD9 cells decrease in size and shape from rod to sphere during naphthalene biodegradation and their specific growth rate was also little slower (Dutta et al., 2018). Keeping this fact in the mind, both morphological types of bacteria were applied in this present set of study. Additionally, the naphthalene 1, 2-dioxygenase (*nahAc*) was sequenced and analyzed.

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151 2. Materials and methods

152 2.1. Chemicals

153 Naphthalene was purchased from Sigma-Aldrich chemicals Pvt. Ltd. (USA) and
154 all other chemicals used for media preparation were procured from HiMedia Laboratory
155 (Mumbai, India). GC-MS and HPLC grade solvents were procured from Fisher
156 Scientific (Mumbai, India). Sodium alginate (CAS No. 9005-38-3) of medium viscosity
157 was purchased from Merck Pvt. Ltd. (USA).

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159 2.2. Microorganisms, growth media, growth condition and consortium preparation

160 Soil samples were collected from petroleum refinery waste sites near Indian Oil
161 (Haldia, West Bengal, India). The enrichment isolation and strain identification were
162 carried out according to the standard protocol described previously (Dutta et al., 2017).
163 Carbon deficient minimum medium (CSM), with a pH of 7.1, was used to cultivate the
164 bacteria. Naphthalene was used as a sole source of carbon and energy in CSM with
165 following compositions: 0.2 g L⁻¹ MgSO₄, 7H₂O; 0.08 g L⁻¹ Ca (NO₃)₂, 4H₂O; 0.005 g
166 L⁻¹ FeSO₄, 7H₂O; 4.8 g L⁻¹, K₂HPO₄; 1.2 g L⁻¹ KH₂PO₄. *Pseudomonas putida* strain
167 KD6 (KX786159.1) and *Pseudomonas putida* strain KD9 (KX786158.1) and the newly
168 isolated *Pseudomonas putida* strain KD10 (KX786157.1) were used to prepare the
169 blend of *Pseudomonas* sp. consortium. The consortium was maintained in Luria-
170 Bertani broth at 31°C with 150 rpm in order to grow the bacterial cell in its normal size
171 and shape. Alternatively, bacterial cells were grown in CSM with naphthalene as sole
172 source of carbon and energy to obtain altered morphological variant.

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174 2.3. Gene sequence analysis

175 2.3.1. Detection of naphthalene 1, 2-dioxygenase and catechol 2, 3-dioxygenase

176 The conventional polymerase chain reaction (PCR) for naphthalene 1, 2-
177 dioxygenase (*nahAc*) and catechol 2, 3-dioxygenase (*nahH*) using specific primers
178 (Table S1) were performed. Additionally, PCR product of *nahAc* was sequenced and
179 analyzed according to the methods described previously (Dutta et al., 2017).

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181 2.3.2. Clustering and phylogenetic analysis

182 The evolutionary distance of naphthalene 1, 2-dioxygenase among different
183 bacterial species was analyzed using BLOSUM weighted matrix followed by pairwise
184 distance computation using MEGA (v7.0) (Kumar et al., 2016). The distance matrix
185 was then clustered using R (Team, 2013) to create the cladogram and heatmap. The
186 phylogenetic position of the isolated *Pseudomonas putida* strain KD10 was analyzed
187 using the previous method (Dutta et al., 2017).

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189 2.3.3. Molecular docking

190 The rigid-body molecular docking was conducted using Auto Dock (v4.2.1)
191 (Morris et al., 1998). Briefly, the center grid dimensions were set to
192 20.271×61.989×87.168 with a grid spacing of 0.375 Å. The virtual screening was

repeated for 10 times with the unaltered docking parameters having 2.0 Å cluster tolerance. Additionally, the rigid-flexible molecular docking was performed using Molecular Operation Environment (Chemical Computing Group, Montreal Inc., Canada). The latter scoring function was employed to identify the most favourable docked poses and to estimate the binding affinity of the protein-ligand complexes. The non-covalent interactions were analyzed using the previous method (Salentin, S., et al., 2015).

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2.4. Enzyme kinetic assay

The enzyme kinetic parameters of the naphthalene 1, 2-dioxygenase_{I250, V256} was performed using the cell-free extract of the *Pseudomonas putida* strain KD10, grown in 250 mL of CSM with naphthalene (500 mg L⁻¹) as a sole source of carbon and energy as described by previously (Dutta et al., 2017).

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2.4. Detection of solvent efflux pumps system

The detection of solvent efflux pump system (*srpABC*) in *Pseudomonas putida* strain KD10 was performed using conventional polymerase chain reaction in a thermal cycler (Mastercycler® nexus gradient, Eppendorf, (Germany). Standard reaction mixtures were prepared using forward, reverse primers (Table S1) as described previously (Dutta et al., 2018).

213

214 2.5. Cell immobilization

215 2.5.1. Cell immobilization in calcium alginate beads

216 *Pseudomonas putida* strain KD10 and *Pseudomonas* sp. consortium were
 217 immobilized according to the standard protocol described previously (Daâssi et al.,
 218 2014). Briefly as sodium alginate was dissolved in 0.9 wt. % NaCl (1 gm in 40 mL 0.9
 219 wt. % NaCl) for 24 h and sterilized by autoclaving (121°C for 15 min). Two grams of
 220 bacterial cell mass was added to 8 ml of NaCl solution and again added to the sterile
 221 alginate solution. The mixture was then gently vortex for complete homogenization and
 222 extruded dropwise through a hypodermic syringe into chilled sterile CaCl₂ solution
 223 (Figure 2). The beads were hardened in the same solution at room temperature with
 224 gentle stirring for 1 h. Finally, the beads were washed several times with 0.9 wt. %
 225 NaCl to remove excess calcium ions and free cells. The beads have an average diameter
 226 of 0.5 mm and stored at 4°C. Sterile beads (without microorganisms) were used to
 227 monitor the abiotic loss of naphthalene. Sodium alginate of medium viscosity
 228 ($\geq 2,000$ cP) was used to prepare the calcium alginate beads (CABs).

229

230 2.5.2. Cell viability count

231 The viable cell enumeration in the CABs was performed by using a protocol
 232 described previously (Usha et al., 2010). In brief, CABs was washed in saline and keep
 233 submerged for 10 min (for saline soaking). Following soaking, the CABs were shaken
 234 with glass beads for 15 min and 1 g CABs were homogenized in 9 ml saline. The saline
 235 was then used for viable cell enumeration onto a nutrient agar plate. Plating was done

with this treated saline by series dilution method up to 10^{-5} dilutions on nutrient agar plates. Plating was also done for the initial CFU count. For every dilution, 10 μ l of the solution was plated. Plating was done by the pour plate method. Plates were incubated at 37°C for 24 h. (Jain and Pradeep, 2005).

2.6. Scanning electron microscopic (SEM) study

Samples for SEM analysis were prepared by a protocol described elsewhere (Dutta et al., 2018). The dried samples were coated in a sputter coater (Quorum-SC7620) under vacuum with a thin gold layer right before SEM analysis using a scanning electron microscope (Zeiss, EVO 18, Germany) with an accelerating voltage of 5 kV.

2.7. Biodegradation kinetics

2.7.1. Biodegradation of naphthalene in the liquid medium

Naphthalene biodegradation study was conducted by the protocol described previously (Dutta et al., 2017). The conical flasks were incubated at 31°C with 150 rpm and uninoculated conical flask were used as control. Culture medium was collected at regular interval of 72 h for degradation and growth kinetic study. Additionally, the effect of initial concentration of naphthalene (150-2500 mg L⁻¹) was studied with different immobilized systems.

257 2.7.2. Chemical analysis

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259 Naphthalene biodegradation was analyzed by using the 1260 infinity series HPLC
 260 system (Agilent, Santa Clara, CA, USA) equipped with Zorbax SB-C18 reversed-phase
 261 column (4.6 × 12.5 mm, 5 µm). The analysis of naphthalene concentrations was
 262 conducted using isocratic elution conditions with the mobile phase 80:20 (v/v)
 263 methanol: water at a flow rate of 1 ml min⁻¹. The detection was performed at 254 nm
 264 according to the protocol described in the literature (Dutta et al., 2017). Additionally,
 265 the metabolic intermediates of naphthalene biodegradation were analyzed using the GC-
 266 MS system (GC Trace GC Ultra, MS-Polarisq, Thermo Scientific India Pvt. Ltd)
 267 equipped with a capillary column (TR-WaxMS, 30 m × 0.25 mm [ID] × 0.25 µm film
 268 thickness) by the protocol described elsewhere (Dutta et al., 2017). The entire analysis
 269 was performed in electron ionization, at full scan mode. The metabolite identification
 270 was based on the mass spectra comparison using the NIST Mass Spectral library (v2.0,
 271 2008).

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273 2.8. Data analysis

274 The first-order degradation kinetics model was used to estimate the residual
 275 naphthalene in CSM using equations 1, the algorithms as expressed determine the half-
 276 life ($t_{1/2}$) values of naphthalene in CSM. The substrate inhibition kinetic parameters
 277 were calculated using equation 2.

$$C_t = C_0 \times e^{-kt} \quad [1]$$

278

$$q = q_{max} \frac{S}{K_s + S + S^2/K_i} \quad [2]$$

279

280 The effect of SCS on growth pattern was measured by calculating the difference
281 between optical densities at 600_{nm} and expressed by slight extension (Equation 4) of the
282 Gompertz's sigmoid growth fit equation (Equation 3).

283

$$y = ae^{-\exp(-k(x-x_c))} \quad [3]$$

284

$$\Delta X_c = X_{c2} - X_{c1} \quad [4]$$

285

286 The growth pattern change by SCS was considered as CFU shift (Equation 5). Where,
287 X_{c2} = Optical density at X_{c2} and X_{c1} = Optical density at X_{c1} .

$$CFU_{shift} = \Delta X_c \quad [5]$$

288

289 The bacterial growth kinetics were analyzed by applying the Gompertz's model using
290 Levenberg-Marquardt algorithm in Origin[®] 2016 (California, USA) and the degradation
291 kinetics were analyzed by using Graphpad Prism[®] 6.01 (San Diego, CA, USA).

292

293 3. Results and discussion

294 3.1. Isolation, identification and growth patterns of NAP degrading bacterial strain

295 *Pseudomonas putida* strain KD10 was isolated from petroleum refinery waste
296 with its distinguished colony morphology on CSM agar plate (Figure S1a, b). The PCR
297 product of 16S rRNA gene of the strain KD10 was sequenced. Moreover, the multiple
298 sequence alignment followed by phylogenetic assessment suggests that the strain
299 belongs to the *Pseudomonas putida* group and it showed 99% sequence similarity with
300 the previously deposited sequences of *Pseudomonas putida* (Figure S2). The 16S rRNA
301 gene sequence was deposited at NCBI (<https://www.ncbi.nlm.nih.gov/>) under the
302 GenBank accession number KX786157.1.

303 Naphthalene biodegradation by the *Pseudomonas putida* KD10 was
304 preliminarily confirmed by the catechol test (Figure S1C) followed by growth pattern
305 analysis in CSM with naphthalene as a sole source of carbon and energy. Additionally,
306 growth pattern with 0.5 gm % sucrose as a secondary carbon supplement (SCS) was
307 compared with growth pattern of non-supplemented CSM. Gompertz's model fit
308 (Equation 3) was used to analyze the growth curve (Table S4).

309

310 3.2. *Pseudomonas* sp. consortium, growth pattern and NAP biodegradation

311 The growth pattern of the *Pseudomonas* sp. consortium, as shown in Figure 1
312 was optimized at 31°C and pH 7.1 which reveal successful cooperation among
313 *Pseudomonas putida* strain KD6, KD9 and KD10. Further, the growth curve model
314 indicates a significant increase in total biomass of the *Pseudomonas* sp. consortium and
315 they also showed a high concentration of naphthalene tolerance potential.

316

317 3.3. Gene sequence and phylogenetic analysis

318 Naphthalene 1, 2-dioxygenase (*nahAc*), encoded by *Pseudomonas putida* strain
 319 KD10 have only two point mutations at I250, and V256 which were replaced by
 320 methionine and glycine respectively (Figure 3A, S6). However, additional mutations
 321 were found in the same chain A at K200, A210, E264, M284 and N334 by replacing
 322 glutamic acid, glycine, aspartic acid, isoleucine, respectively (Dutta et al. 2017, 2018).
 323 The three dimensional structural analysis suggests that the mutations at 200, 210, 284
 324 and 334 causes a little structural change in comparison to the wild type variant of
 325 naphthalene 1, 2-dioxygenase (Figure 3B). Conversely, the amino acids stretch at the
 326 close proximity of the active site residues exactly from 248 to 266 showed a structural
 327 mismatch among all three mutant and wild type variants of naphthalene 1, 2-
 328 dioxygenase (Figure 3B). The active site of an enzyme tends to evolve fast to attain its
 329 maximum performance and functionality in a particular environment (James and
 330 Tawfik, 2003).

331 The alteration of amino acids bases from 248 to 266 of chain A, was a common
 332 feature in all three mutant variants of *nahAc* and this particular stretch of amino acid is
 333 very close to the active site residues (Figure 3C). This implies that the “248-266” amino
 334 acid stretch is highly prone to mutation and it may influence on the enzymatic efficiency
 335 and environmental adaptability. Thus, our study provides a new insight, which could be
 336 beneficial for rational approaches of enzyme redesigning.

337 The biodegradation performance of a bacterial strain is linked with the three
 338 dimensional structure of the catalytic enzyme and ligand binding posture (Singh et al.,
 339 2019). Moreover, alteration of a single amino acid in the catalytic domain of the enzyme
 340 caused different ligand binding postures which eventually lead to a unique

biodegradation pathways (Ferraro et al., 2006). In addition, site directed mutagenesis in the catalytic domain offers superior enzyme activity (Parales et al., 1999; Parales, 2003). The comparative analysis of binding free energy helps to comprehend the superior naphthalene biodegradation performance by the *Pseudomonas* sp. consortium, which is, in fact, the summative activity of all three mutant variants of naphthalene 1, 2-dioxygenase.

The evolutionary trace on the naphthalene 1, 2-dioxygenase among different species were studied through pairwise distance matrix analysis (Figure 4), which suggests a significant intra and inter species difference among *Sphingopyxis* sp., *Croceicoccus naphthovorans*, *Burkholderia multivorans*, *Burkholderia* sp. *Massilis* sp. and *Cycloclasticus* sp. Conversely, other *Pseudomonas* sp. particularly, *Pseudomonas benzenivorans*, *Pseudomonas balearica*, *Pseudomonas stutzeri*, *Pseudomonas kunmingensis*, *Pseudomonas frederikbergensis* have similarities in their version of naphthalene 1, 2-dioxygenase. A few other species such as *Xenophilus* sp., *Ralstonia annitolilytica*, and *Paraburkholderia aromaticivorans* does not show significant evolutionary distance in their version of naphthalene 1, 2-dioxygenase (Figure 4). However, a few strains of *Burkholderia multivorans* showed similarities and some other does not (Figure 4). The source of collection of these two *Burkholderia multivorans* species may be the reason for such variation (Li et al., 2007). In a study by (Su et al., 2016), the epigenetic impacts on the metabolic enzymes have been suggested. Moreover, the divergence of evolutionary distances of the same enzyme between different and same bacterial species indicates thrive in a particular microenvironment. However, further studies are needed to identify the underlying cause and mechanisms of such variation intra-species variation.

365

366 3.4. Molecular docking

367 The result depicted from the rigid body molecular docking, suggests *nahAc*-
 368 KD10_{I250, V256} mutant has a little higher binding free energy than that of mutant *nahAc*
 369 encoded by KD9 but little lower than the *nahAc* mutant variant encoded by the KD6
 370 (Table 1). However, the rigid-flexible molecular docking using MOE algorithms
 371 showed *nahAc* I₂₅₀, G₂₅₆ mutant variant has ΔG_{London} of -7.1 kcal mol⁻¹ and $\Delta G_{\text{GBVI/WSA}}$
 372 of -1.68 kcal mol⁻¹, which is little higher than other two mutant variants of *nahAc*
 373 encoded by KD6 and KD9 (Table 1). The interacting amino acid residues in the active
 374 site were confined to be quite same in three mutant variants, except one variation, *i.e.*,
 375 His₂₀₈ which is located about 5.49 Å away from the bicyclic ring of naphthalene (Figure
 376 3D). The altered binding free energy of the *nahAc*-KD10_{I250, V256} may be due to the fact
 377 that the two altered amino acid residues reside in “248-266” and participation of His₂₀₈
 378 as a unique interacting residue (Figure 3C). Furthermore, the non-covalent interactions
 379 of the *nahAc*-KD10_{I250, V256} with naphthalene, phenanthrene and anthracene suggest that
 380 Phn 352 was common residue and His 208 is involve in π -staking with a bond angle of
 381 80.58° and 77.19° respectively for naphthalene and anthracene (Table S5).

382

383 3.5. Enzyme kinetic assay

384 Enzyme activity by cell free extract of *Pseudomonas putida* KD10 was studied
 385 using Naphthalene as substrate. The cell free extract of strain KD10 degraded
 386 Naphthalene, with R^2 of 0.935 indicating the experimental data are well correlated with
 387 the model (Table 2). The performance constant ($K_{\text{cat}}/K_{\text{m}}$) was $0.142 \times 10^3 \text{ ml}^{-1} \text{ mol}^{-1} \text{ s}^{-1}$
 388 which further validate the chemical data obtained from chemical analysis (Table 1). In

389 previous study, the performance constant of *Pseudomonas putida* KD6 was little higher
 390 than *Pseudomonas putida* strain KD10, suggesting the six altered amino acid residues of
 391 the mutant naphthalene 1,2-dioxygenase of the KD6 may be main cause of this
 392 difference (Dutta et al. 2017). The performance constant (K_{cat}/K_m) is the indicator of
 393 the performance of an enzyme and usually the higher K_{cat}/K_m meaning better
 394 enzymatic performance (Koshland Jr DE, 2002).

395

396 3.6. Detection of solvent efflux system

397 The cellular microenvironment and internal homeostasis are crucial for
 398 maintaining normal cellular functions and cell tends to adopt several strategies in order
 399 to achieve it (Blanco et al., 2016). One such adaptation strategy is the solvent efflux
 400 pump system, that control intracellular toxicity to some extent (Kusumawardhani et al.,
 401 2018). The role of *srpABC* solvent efflux pump system in the biodegradation of PAHs
 402 are common in literature (Bugg et al., 2000). Besides the other cellular activities,
 403 *srpABC* also assist in gaining antimicrobial resistance (Schweizer, 2003).
 404 Biodegradation studies on chlorpyrifos indicate that the intermediates formed during
 405 biodegradation, act as an antimicrobial agent to other species, meaning that the
 406 metabolic intermediates may have a role on inter-species competition in a particular
 407 micro-environment to thrive in nutrient-limiting condition (Anwar et al., 2009), (Raes
 408 and Bork, 2008). Besides, the role of *srpABC* on biodegradation enhancement process
 409 is still poorly understood. The presence of *srpABC* in *Pseudomonas putida* strain KD10
 410 (Figure S5) and the efficient naphthalene biodegradation property can be interlinked
 411 (Dutta et al., 2018).

412

413 3.7. Cell immobilization and viability count

414 The viscosity of CABs determines its efficiency of the immobilization and its
415 performance of detoxification of environmental pollutant (Young et al., 2006).
416 Therefore, sodium alginates of medium viscosity were chosen for this study. The
417 efficiency of cell immobilization in CABs was evaluated by enumeration of viable cells
418 (Table S3). Bacteria immobilized in CABs do not significantly lose its viable biomass
419 after 21 days of incubation at 4°C.

420

421 3.8. Naphthalene biodegradation studies

422 3.8.1 By free *Pseudomonas putida* strain KD10

423 Biodegradation kinetics of naphthalene (500 mg L⁻¹) by the *Pseudomonas putida*
424 strain KD10 was first tested to determine its naphthalene biodegradation potential
425 (Figure 4). It was found that after 12 days of incubation at 31°C the amount of residual
426 naphthalene reduced significantly ($p < 0.05$) (Figure 5). Results are also summarised at
427 Table 3 along with names of the immobilized systems. The first order degradation
428 kinetic data suggest that at the end of the incubation *Pseudomonas putida* strain KD10
429 efficiently decompose 95.22 % naphthalene in CSM as a sole source of carbon and
430 energy. The values of degradation rate constant (k) and half-life ($t_{1/2}$) were 0.2193 and
431 3.1 days with R^2 of 0.981. In the previous study, 99.1 % of initial naphthalene was
432 removed within 96 h by strain *Bacillus fusiformis* (BFN). However, the initial
433 concentration was very low (50 mg L⁻¹ of initial naphthalene) (Lin et al., 2010).

434 After confirmation of naphthalene biodegradation potential of *Pseudomonas*
435 *putida* strain KD10, the stain was allowed to grow in association with other two strains
436 of *Pseudomonas putida*, namely *Pseudomonas putida* strain KD6 and *Pseudomonas*

437 *putida* strain KD9 (Dutta et al., 2017; Dutta et al., 2018). These two strains were
 438 selected to develop the blend of *Pseudomonas* sp. consortium, because they were
 439 collected from different isolation points, they encode different variant of *nahAc* and
 440 they have common optimized growth parameters (temperature 31°C and pH 7.1) (Dutta
 441 et al., 2017 & Dutta et al., 2018). Further, *Pseudomonas putida* strain KD6 encodes a
 442 six point mutant *nahAc* with the capability to co-degrade high concentration of
 443 naphthalene, phenanthrene (PHN) and pyrene (PYR), (500 mg L⁻¹ each). Moreover, the
 444 stain KD9 encodes a four point mutant variant of *nahAc* with rhamnolipid production
 445 capabilities (Dutta et al. 2018).

446 In the previous study, $t_{1/2}$ of naphthalene co-biodegradation with phenanthrene
 447 and pyrene by *Pseudomonas putida* strain KD6 was 4.1 days, which was significantly
 448 reduced to 2.2 days when bacterial cells were allowed to grow as *Pseudomonas* sp.
 449 consortium with only naphthalene (Table 3). Further, this value was found as 2.7 days
 450 for *Pseudomonas putida* KD9, suggesting *Pseudomonas putida* strain KD6 might face
 451 some sort of substrate inhibition by the co-presence PHN and PYR in the system (Jiang
 452 et al., 2018) and the cooperative nature of the *Pseudomonas* sp. consortium helps to
 453 enhance the naphthalene bio-utilization. The microbial consortium works several ways,
 454 viz., by the division of labor, cross-feeding, etc. (Smid and Lacroix, 2013). Moreover, a
 455 successful consortium could also overcome shortcomings of single bacteria (Bhatia et
 456 al., 2018). However, it is a fact that, bacteria select PAH among mixed PAHs based on
 457 their structural simplicity first (Dutta et al., 2017) and the velocity could be optimized
 458 by reducing the initial concentration of the PAHs gradually according to the structural
 459 simplicity (Jiang et al., 2018).

460

461 3.8.2. By immobilized *Pseudomonas* sp. consortium

462 The biodegradation kinetic of naphthalene by individual *Pseudomonas putida*
463 strain KD6, KD9, KD10 and *Pseudomonas* sp. consortium immobilized in CABs,
464 depicted further enhancement of overall naphthalene bio-utilization (Table 3). Cell
465 immobilization using hydrogel, such as CAB found to be advantageous rather than free
466 cells (Hameed and Ismail, 2018).

467 This phenomenon may be attributable due to the increased level of tightness of
468 the cross-linked polymers of the calcium alginate beads that render bacteria adequate
469 amount of protection from harsh environment (Chen et al., 2013). However, the free
470 bacterial cell lacks the capabilities to degrade a high initial concentration of the toxicant
471 because they followed the conventional growth phases (Marrot et al., 2006). In addition,
472 exposure of free bacterial cells to the high initial concentration of toxicants may
473 challenge them to experience shock-concentration (Zhao et al., 2006). Conversely, cells
474 immobilized in calcium alginate beads can tolerate high concentration of the toxicant
475 and decrease the lag phase duration (Kao et al., 2014). Moreover, the diffusion
476 limitation natures of the CABs matrix provide a high local concentration of the cell
477 population (Bezbaruah et al., 2009). Furthermore, CABs provides remarkable stability
478 and reusable features that effectively reduces the production cost (Daâssi et al., 2014).

479 Biodegradation of naphthalene by immobilized *Pseudomonas* sp. consortium
480 significantly elevated the overall naphthalene bio-utilization efficiency with $t_{1/2}$ and R^2
481 values of 2 days and 0.998 respectively, suggesting experimental data are well
482 correlated to the model (Table 3). The individual cell population also displayed
483 improved biodegradation efficiency. However, the biodegradation efficiency was
484 maximum by *Pseudomonas putida* strain KD6, which showed a marked reduction of $t_{1/2}$

485 from 4.1 to 3.0 days (Table 3). It promptly suggests that CABs facilitate KD6 optimized
486 naphthalene bio-utilization, that might be masked-up by the co-presence of NAP, PHN,
487 PYR (Jiang et al., 2018). However, further studies are required to investigate
488 behavioural patterns of a consortium with mixed PAH.

489

490 Bacterial cell grown on CSM with naphthalene as a sole source of carbon and
491 energy showed a delayed growth rate and their total biomass was also low (Figure 1).
492 Nevertheless, growing cells on CSM prior to immobilization in CABs provide them
493 essential adaptation to the toxicant and mature them for such stress condition (Table 3).
494 Immobilized cell system was useful for bioremediation of a toxicant after prior
495 adaptation to the surrounding environment (Partovinia and Rasekh, 2018). However,
496 growing cells in Luria-Bertani broth does not adapt cell adequately, and their
497 naphthalene removal performance was very poor (Figure S3). The inoculum was
498 normalized ($OD_{600nm} = 0.002$) for each *Pseudomonas putida* strain in order to prepare
499 the *Pseudomonas* sp. consortium. In aqueous medium, the immobilized bacterial cell
500 mainly on the surface was exposed to naphthalene and primarily involved in the
501 naphthalene bio-utilization process. However, due to the micro-porous feature of the
502 calcium alginate beads, microbial cells immobilized other than surface are also
503 participates in the bio-utilization and bio-sorption process of naphthalene. Further, with
504 time due to the mechanical force generated by shaking, a few microbial cells may
505 release from the calcium alginate beads. However, to fuel up the cell, it is necessary to
506 uptake naphthalene for bio-utilization through step by step intracellular enzymatic
507 reactions (Lin et al., 2014).

508

509

510 3.8.3. Effect of initial naphthalene concentration on biodegradation kinetics

511 The effect of different initial concentration of naphthalene (150 – 2500 mg L⁻¹)
 512 on degradation kinetics was evaluated with immobilized KD6, KD9, and KD10 as
 513 individual and as a consortium (Table S2). The results suggest cell immobilization in
 514 CABs facilitates bacteria to cope with a high initial concentration of NAP (Table S2).
 515 The substrate inhibition kinetic parameters, viz., the maximum specific degradation rate
 516 ($q_{\max}$), and inhibition constant (k_i) were highest for immobilized *Pseudomonas* sp.
 517 consortium and these values were 0.707 h⁻¹ and 1475 mg L⁻¹ respectively (Table S2).
 518 However, we did not find any significant change on half saturation constant (k_s),
 519 suggesting the reaction does not depend on its initial concentration and the
 520 biodegradation process follows pseudo-first order reaction kinetics. In previous study,
 521 free *Pseudomonas putida* KD9 in CSM was capable to tolerate relatively low initial
 522 concentration of naphthalene with k_i value of 1107 mg L⁻¹ and addition of sucrose as
 523 SCS provides quite similar potential of naphthalene tolerance (k_i of 1429 mg L⁻¹) with
 524 that of immobilized *Pseudomonas* sp. consortium (Dutta et al., 2018). However, sucrose
 525 (0.5 gm. %) in the mobile open water system would not be beneficial for bacteria to
 526 overcome the high shock concentration of the toxicant, again suggesting cell
 527 immobilization and development of effective microbial consortium as systematic
 528 optimization of biodegradation process.

529 3.9. Detection of metabolic end products

530 The metabolic pathway of naphthalene biodegradation that might be followed by
 531 *Pseudomonas putida* strain KD10 was elucidated through GCMS analysis of the

metabolites (Figure S4). In the previous study, the major metabolites were restricted to be salicylaldehyde, catechol, D-gluconic acid and pyruvic acid (Dutta et al., 2018). However, in the present study, we have detected salicylic acid as an additional metabolite (Table 4). Catechol step into the TCA cycle by two possible pathways one is ortho-cleavage and another is meta-cleavage pathway. The meta-cleavage pathway is led by catechol-2, 3-dioxygenase (*nahH*), and presence of *nahH* (Figure S5) suggesting in *Pseudomonas putida* strain KD10 opt the meta-cleavage pathway. Furthermore, we assume that D-gluconic acid enters in the bio-conversion of naphthalene, possibly from glucose as a precursor (Figure 6) and it is finally converted to pyruvate via aldolase. D-gluconic acid implies impotence between bacteria-plant mutualistic association e.g., induction of phosphate solubilization processes (Rodriguez et al., 2004) which can inhibit fungal growth (Kaur et al., 2006). The metabolic intermediates viz., salicylic acid, catechol has plant-growth promoting activity (Lee et al., 2010) e.g., the antioxidant property of the catechol help and promotes seed germination (Schweigert et al., 2001). The presence of d-gluconic acid as a major metabolite and its subsequent entry in TCA cycle from glucose precursor, suggest that KD10 may be able to sequester catechol upon needs to promote the plants' growth (Figure 6).

549

550 4. Conclusion

551

Naphthalene biodegradation by free and immobilized *Pseudomonas putida* strain KD10 and *Pseudomonas* sp. consortium were studied. HPLC analysis showed 80.1% of initial naphthalene (500 mg L⁻¹) was utilized by immobilized *Pseudomonas*

555 sp. consortium after 72 h of incubation. Further, initial naphthalene tolerance by
 556 immobilized *Pseudomonas* sp. consortium was highest (1475 mg L⁻¹) with maximum
 557 specific degradation of 0.707 h⁻¹. The gene sequence analysis of the naphthalene 1, 2-
 558 dioxygenase suggests a significant evolutionary distance among different microbial
 559 species and in few cases, intra-species variation was observed. A common mutation
 560 prone amino acid stretch inside Chain A of all three natural mutant variants of
 561 naphthalene 1, 2-dioxygenase were found at close proximity of the active site. Further,
 562 the rigid-flexible molecular docking showed better binding free energy of the mutant
 563 variant encoded by *Pseudomonas putida* strain KD10 than that of wild-type variant of
 564 naphthalene 1,2-dioxygenase. This common mutation prone amino acid stretch could
 565 aid the rational approaches of enzyme redesigning. Overall, this study summarises the
 566 application of bacterial cell immobilization in calcium alginate beads and development
 567 of the microbial consortium together for enhanced naphthalene biodegradation.
 568 However, further studies are required for the systematic optimization of naphthalene
 569 biodegradation in a real environment.

570

571 *Authors Contributions*

572 KD and CG conceive the main hypothesis. KD design, performed all
 573 experiments, and wrote the manuscript. SS, IK, SB, DJ assisted KD in some
 574 experiments. MK, TM, PR, KCG performed the statistical analysis and wrote the
 575 manuscript. CG critically proofread and wrote the manuscript. All authors read the
 576 manuscript.

577

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588

589 *Conflict of Interest*

590 The authors declare that they have no conflict of interests.

591

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774 Figure legends

775 **Figure 1.** Growth kinetic model of *Pseudomonas putida* and *Pseudomonas* sp.
776 consortium during naphthalene biodegradation. **A.** Gompertz's growth kinetic
777 model fit of the biodegradation of naphthalene by *Pseudomonas putida* strain
778 KD10 and by the *Pseudomonas* sp. consortium. **B.** Effect of 0.5 gm. % sucrose
779 supplementation on growth pattern of *Pseudomonas putida* strain KD10 and
780 *Pseudomonas* sp. consortium.

781 **Figure 2.** Cell immobilization in calcium alginate beads (CABs) with different
782 cell morphology of the *Pseudomonas putida* strain KD10.
783

784 **Figure 3.** Sequence and structure analysis of the naphthalene 1, 2-dioxygenase
785 (*nahAc*) encoded by *Pseudomonas putida*. **A.** The amino acid sequence of wild
786 type and three mutant variants of the naphthalene 1, 2-dioxygenase (*nahAc*).
787 Mutated residues are highlighted in blue and the common mutation prone amino
788 acid stretch is highlighted with red box. **B.** Structural mismatches among mutant
789 variants and wild type naphthalene 1, 2-dioxygenase. Each mutated residues and
790 its subsequent neighbour residues of all the mutant variant of naphthalene 1, 2-
791 dioxygenase is overlaid with cartoon representation. The highlighted red circle
792 indicates local mismatch among the mutant variant of naphthalene 1, 2-
793 dioxygenase **C.** Different ligand binding posture of the naphthalene 1, 2-
794 dioxygenase encoded by *Pseudomonas putida* strain KD10. Naphthalene (red),
795 phenanthrene (cyan) and anthracene (green) binding postures with major
796 interacting amino acid residues (left). Two-dimensional presentation of the
797 interacting residues of *nahAc*_{1250, V256} (right). **D.** Docking pose of naphthalene 1,
798 2-dioxygenase (Chain A) encoded by *Pseudomonas putida* strain KD10. The
799 four major naphthalene interacting residues and their molecular distance are
800 labelled in black.

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802

Figure 4. Distance matrix and cladogram of the naphthalene 1, 2-dioxygenase. The evolutionary distance of the naphthalene 1, 2-dioxygenase I₂₅₀, V₂₅₅ variant encoded by *Pseudomonas putida* strain KD10 other variants by different microbial species is scaled by colour index (upper-left side).

Figure 5. Naphthalene degradation kinetics of *Pseudomonas putida* strain KD10 and *Pseudomonas* sp. consortium free and cell immobilized in calcium alginate beads.

Figure 6. Proposed pathway of naphthalene biodegradation by *Pseudomonas putida* strain KD10. Naphthalene (I), salicylic acid (II), salicylaldehyde (III), catechol (IV), pyruvate (V), glucose (VI), d-gluconic acid (VII). The grey arrow indicates alternative pathway via d-gluconic acid.

Table captions

Table 1. Summary of rigid-flexible molecular docking for the analysed mutant naphthalene 1, 2-dioxygenase encoded by different *Pseudomonas putida* strains.

Table 2. Apparent enzyme kinetic parameters of naphthalene 1, 2-dioxygenase I₂₅₀, V₂₅₆ encoded by *Pseudomonas putida* strain KD10.

Table 3. Parameters of first-order biodegradation kinetics of naphthalene by different free and immobilized *Pseudomonas putida* strains.

Table 4. Major metabolite detected by GCMS analysis during naphthalene biodegradation by *Pseudomonas putida* KD10.

835 *Supplementary section*

836

837 **Table S1.** Nucleotide sequences used as primer in polymerase chain reactions

838

839 **Table S2.** Parameters of substrate inhibition kinetic model of naphthalene
840 biodegradation by different *Pseudomonas putida* strains immobilized in calcium
841 alginate beads.

842 **Table S3.** Enumeration of viable cell in calcium alginate beads.

843 **Table S4.** Gompertz's growth curve model fit of *Pseudomonas putida* strain
844 KD10 and *Pseudomonas* sp. consortium.

845

846 **Table S5.** Major amino acid residues of mutant variant of naphthalene 1, 2-
847 dioxygenase ^{I250, V256} involve in hydrophobic interaction with different ligands.

848

849 **Figure S1.** Colony morphology and catechol confirmatory test. **A.** Colony
850 morphology of *Pseudomonas putida* strain KD10 after 24 h of incubation. **B.**
851 Colony morphology of *Pseudomonas putida* strain KD10 after one week of
852 incubation.

853 **Figure S2.** Molecular phylogenetic analysis of *Pseudomonas putida* strain
854 KD10.

855 **Figure S3.** Effect of cell adaptation on naphthalene biodegradation.

856 **Figure S4.** Mass spectra of the major metabolites detected during
857 biodegradation of naphthalene in CSM as sole source of carbon and energy by
858 *Pseudomonas putida* KD10.

859

860 **Figure S5.** Agarose gel electrophoresis of PCR products of catechol 2, 3-
861 dioxygenase (*nahH*) gene and solvent efflux pump (*srpABC*) system.

862 **A.** Catechol 2, 3-dioxygenase (*nahH*). Lane 1 and 2 *nahH* pcr product of
863 *Pseudomonas putida* strain KD9 and *Pseudomonas putida* strain KD10. **B.**
864 Solvent efflux pump (*srpABC*) system. Lane 1: *Pseudomonas aeruginosa*
865 ATCC 9027, Lane 2: *Pseudomonas putida* strain KD10. Lane M: 100bp DNA
866 ladder (HiMedia, India), Lane 3: Negative control.

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868 **Figure S6.** Nucleotide sequence chromatogram of the mutant naphthalene 1, 2-
869 dioxygenase (*nahAc*), (chain-A) encoded by *Pseudomonas putida* strain KD10.

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Table 1. Summary of rigid-flexible molecular docking of naphthalene and different mutant version of naphthalene 1, 2-dioxygenase encoded by different *Pseudomonas putida* strains

ΔG^*			LB ⁴	IC ⁵	IME ⁶	VDW ⁷	RMSD ^c	RMSD ^r	References
BE ¹	London ²	GBVI/WSA ³							
-5.92 ^a	-7.18	-4.61	-0.59	46.14	-5.92	-5.91	0.00	77.29	Dutta et al., 2017
-5.41 ^b	-7.13	-4.02	-0.54	109.07	-5.41	-5.4	0.00	96.32	Dutta et al., 2019
-5.83 ^c	-7.10	-1.68	-0.58	53.13	-5.83	-5.84	0.00	96.08	This study

*PubChem ligand CID 931 = Naphthalene, BE¹ = Binding free energy, ²London free energy, ³Generalized born volume integral/weighted surface areas, LB⁴ = ligand binding, IC⁵ = inhibition constant, IME⁶ = Intermolecular energy, VDW⁷ = Vdw hb desolv energy, RMSD^c = Cluster RMSD, RMSD^r = Reference RMSD. Energy unit = kcal mol⁻¹. Inhibition constant unit = μ M.at 298.15 K. Naphthalene 1, 2-dioxygenase encoded by (a) *Pseudomonas putida* strain KD6, (b) *Pseudomonas putida* strain KD9 and (c) *Pseudomonas putida* strain KD10.

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Table 2. Apparent enzyme kinetic parameters of naphthalene 1, 2-dioxygenase I₂₅₀, V₂₅₆ encoded by *Pseudomonas putida* strain KD10

Substrate	Kinetic parameters*				
	K _m (μmol mL ⁻¹)	V _{max} (μmol min ⁻¹)	K _{cat} (s ⁻¹)	K _{cat} /K _m (mL ⁻¹ mol ⁻¹ s ⁻¹)	R ^{2**}
Naphthalene	0.867	2.474	123.72	0.142×10 ³	0.938

*The kinetic constants were determined at 30°C and pH 7.5 using 0.5-10μM ml⁻¹ substrate concentration (each) by Lineweaver–Burk plot.

** Non-linear regression between initial substrate concentration 1/[S] and degradation rate constant 1/V yielded regression equation and regression coefficient (R²)

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Table 3. Parameters of first-order biodegradation kinetics of naphthalene by different free and immobilized *Pseudomonas putida* strains

CSM	Cells immobilized in CABs			free cells			References
	k (Days ⁻¹)	$t_{1/2}$ (Days ⁻¹)	R^2	k (Days ⁻¹)	$t_{1/2}$ (Days ⁻¹)	R^2	
A	0.228±0.005	3.0±0.077	0.99	0.167±0.012	4.1±0.307	0.98	Dutta et al., 2017
B	0.312±0.014	2.2±0.102	0.93	0.252±0.019	2.7±0.193	0.99	Dutta et al., 2018
C	0.284±0.017	2.4±0.143	0.97	0.219±0.009	3.1±0.137	0.92	This study
D	0.341±0.294	2.0±0.186	0.99	0.309±0.020	2.2±0.141	0.98	This study

[#]Initial naphthalene concentration = 500 mg L⁻¹

A. *Pseudomonas putida* strain KD6, B. *Pseudomonas putida* strain KD9, C. *Pseudomonas putida* strain KD10. D. *Pseudomonas* sp. consortium.

Each figure of the table represents the mean of three replicates.

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Table 4. Major metabolites detected by GCMS analysis during naphthalene biodegradation by *Pseudomonas putida* KD10

Metabolite	R _t (min)	Major ion peaks (m/z)	Suggested structure
I	9.41	64.19, 92.17, 120.21, 138.22, 169.14, 191.18, 211.10	Salicylic acid
II	15.27	65.17, 74.41, 104.73, 121.10, 122.57, 142.21, 174.80, 201.22, 249.31	Salicylaldehyde
III	18.71	64.29, 81.35, 91.61, 110.47, 131.47, 149.71	Catechol
IV	31.89	61.21, 73.44, 76.24, 104.18, 117.41, 133.27, 177.23	D-gluconic acid
V	10.47	43.21, 61.27, 71.37	Pyruvic acid

R_t = Retention time

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Figure 1.

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978 **Figure 2.**

979 **Figure 3.**

Figure 4.

Figure 5.