Recycling of energy dissipated as heat accounts for high activity of Photosystem II

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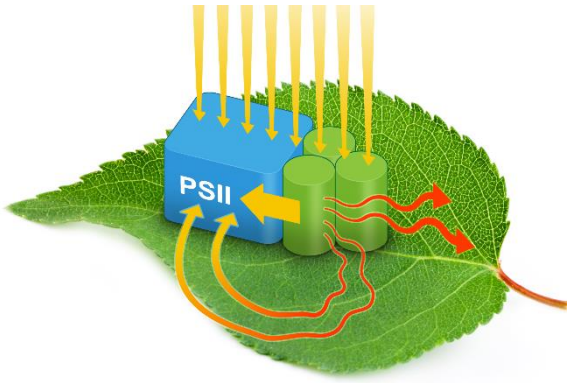
ABSTRACT. Photosystem II (PSII) converts light into chemical energy powering almost entire life on Earth. The primary photovoltaic reaction in the PSII reaction centre requires energy corresponding to 680 nm that is higher than in the case of low-energy states in complexes involved in the harvesting of excitations driving PSII. This promotes the process of thermal dissipation, reducing photosynthesis efficiency. Here we show that part of energy dissipated as heat is used to drive PSII thanks to the thermally-driven up-conversion. We demonstrate the operation of this mechanism both in leaves and in isolated pigment-protein complex LHCII. A mechanism is proposed, according to which the effective utilization of thermal energy in the photosynthetic apparatus is possible owing to the formation of LHCII supramolecular structures, leading to the coupled energy levels, corresponding to approx. 680 nm and 700 nm, capable of exchanging excitation energy through the spontaneous relaxation and the thermal up-conversion.

TOC GRAPHICS



MAIN TEXT. Life on Earth is powered by sunlight and photosynthesis is practically a sole process able to convert the energy of electromagnetic radiation to the forms which can be directly utilized to drive biochemical reactions in living organisms [1]. Importantly for life in our biosphere, oxygenic photosynthesis supplies molecular oxygen to the atmosphere, which most of the organisms use for respiration [2]. Photosynthetic oxygen evolution is directly associated with activity of Photosystem II (PSII), in which electrons can be detached from water molecules [3] due to the relatively high redox potential of the $P680^{\bullet+}$ radical cation created by a photoreaction in the reaction center [4], estimated to be 1.26 V [5]. PSII has originally evolved 2.5 billion years ago and is located in the photosynthetic membranes of cyanobacteria, algae and plants [6]. Oxidation of water would not be possible in Photosystem I (PSI) –type complexes present in all photosynthesizing organisms due to the insufficient redox potential of the $P700^{\bullet+}$ radical cation created by a photoreaction in its reaction center. Most of the photosynthetic pigments in plants, energetically coupled to PSII and to PSI contribute to the separate chlorophyll *a* (Chl *a*) fluorescence emission bands centering respectively at ca. 680 nm and 735 nm [7-9] (see Fig. 1a). Intriguingly, at the low temperature, the fluorescence emission from Chl *a* molecules associated with PSI is substantially more intensive than from the pool associated with PSII. The opposite proportion can be observed at room temperature, in accordance with the numerous previous reports [8]. Determined at 77 K, the fluorescence yield of PSII core complex is about 2 times higher and the fluorescence yield of PSI about 20 times higher, as compared to room temperature [7, 8]. The integration of the Chl *a* fluorescence emission spectra in leaves in the spectral region representing mostly PSII (wavelengths below 703 nm) and in the spectral region representing largely PSI (wavelengths higher than 703 nm) gives the ratio of photons emitted by PSII and PSI as high as 1.16 ± 0.22 at 298 K but only 0.20 ± 0.02 at 77 K (a mean value from three different leaves ± S.D., Fig. 1a). One of the possible interpretation of such an observation is that the process of thermally-driven up-conversion of the low-energy

states corresponding to wavelengths higher than 700 nm, is necessary for effective excitation supply to the P680 reaction center of PSII. Such a process cannot take place at 77 K due to the insufficient thermal energy. In principle, a process of the thermally-driven up-conversion can operate effectively in the photosynthetic apparatus of plants owing to the fact that the overall energy conversion efficiency in photosynthesis does not exceed 6 % and most of energy of absorbed light quanta is dissipated as heat [10, 11]. A direct demonstration of operation of the uphill energy transfer in the photosynthetic apparatus in leaves is shown in Fig. 2c.

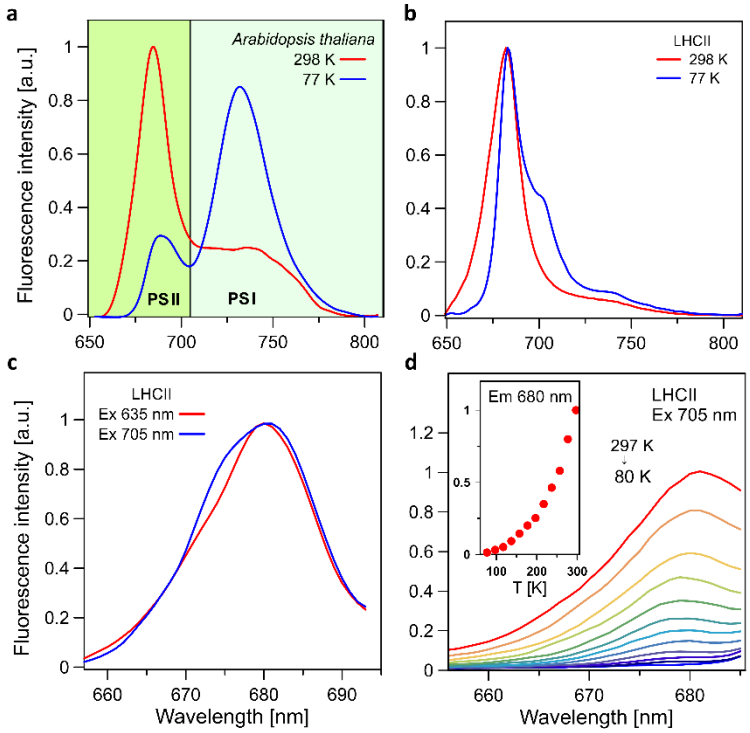


Figure 1. Chlorophyll *a* fluorescence emission spectra recorded from leaves and pigment-protein complex LHCII. (a) Fluorescence emission spectra recorded from *A. thaliana* leaves. Excitation at 635 nm. The spectra were recorded at 77 K and at 298 K, indicated. The spectra were area-normalized and presented on the arbitrary scale. The short-wavelength and the long-wavelength emission bands are assigned to PSII and PSI respectively. (b) Fluorescence emission spectra recorded from the sample containing supramolecular structures of LHCII formed spontaneously in the environment of the lipid membrane. Excitation at 635 nm. The spectra normalized at the maximum. (c) Fluorescence emission spectra recorded from the LHCII sample as in panel b, at 298 K with the laser excitations set at 635 nm or at 705 nm, indicated. (d) Fluorescence emission spectra recorded from the LHCII

sample as in panel b, recorded at different temperatures in the range from 297 K to 80 K. Excitation was set in the lower energy spectral region, at 705 nm. The inset shows the temperature dependence of fluorescence intensity at 680 nm, based on the spectra presented. The set of the spectra is presented on the arbitrary scale.

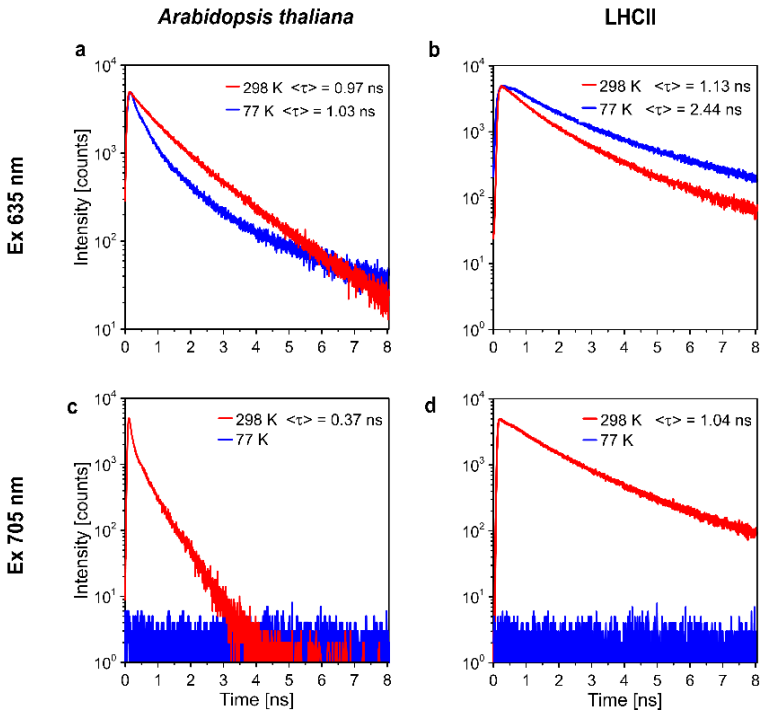


Figure 2. Chlorophyll *a* fluorescence decay kinetics recorded from intact leaves and from LHCII samples. The kinetics were recorded from intact leaves of *A. thaliana* (a and c) and from the samples containing supramolecular structures of LHCII formed spontaneously in the environment of lipid membrane (b and d), at 77 K and at 298 K, indicated. Two combinations of the excitation and fluorescence observation wavelengths were applied: Ex 635 nm/Em 680 nm (panels a and b) and Ex 705/Em 680 nm (panels c and d). Displayed are amplitude weighted average fluorescence lifetimes (<τ>) calculated for each system.

Fluorescence of Chl *a* excited in the lower energy reagion (at 705 nm) and detected at higher energies (at 680 nm) can be detected in intact leaves at the room temperature (298 K) but not at 77 K (Fig. 2c). Efficient and fluent operation of photosynthesis is assured by the activity of pigment-protein complexes, called antenna, collecting photons and transferring excitation energy towards the reaction centres responsible for the primary electric charge separation [12, 13]. The largest light-harvesting pigment-protein complex of plants, referred to as LHCII, is particularly well suited to play a photosynthetic antenna function, owing to the relatively high

concentration of chlorophylls, the presence of xanthophylls (effectively protecting the complex against photo-damage) and internal pathways of extremely efficient excitation energy transfer within the network of the protein-embedded chromophores [12, 13]. A consequence of exceptionally high protein crowding in the thylakoid membranes is the clustering of antenna complexes, potentially resulting in excitation quenching and thermal energy dissipation [14-19]. Molecular organization of LHCII can be modelled in the experimental system composed of isolated LHCII embedded into the lipid membranes formed with chloroplast lipids [17, 20-22]. A spontaneous self-association of the antenna complexes in such a system gives rise to the low-energy band in a low-temperature Chl *a* fluorescence emission spectrum, centered in the region of 700 nm, accompanying the principal band centered in the region of 680 nm (Fig. 1b), in accordance with the previous reports [18, 23-25]. The long-wavelength band in this particular spectral region (~700 nm) has been assigned to LHCII clusters in the natural thylakoid membranes [26]. There are several lines of evidence for the formation of such supramolecular structures of LHCII in the thylakoid membranes of plant chloroplasts, including the one based on the circular dichroism analyses [27], low-temperature fluorescence spectroscopy [26] and direct imaging based on electron microscopy [15]. The long-wavelength band centred close to 700 nm can be also resolved at 77 K in isolated thylakoid membranes [28] and leaves [8]. The "700 nm band" is a hallmark of aggregated LHCII [23-25] although it can be also detected from the trimeric LHCII particles in single-molecule experiments [29]. Importantly, the "700 nm band" (referred to as E700) can be resolved exclusively in the fluorescence emission spectra of aggregated LHCII recorded at low temperatures but not at physiological temperatures [18, 23-26]. This observation suggests possible depopulation of the E700 state by an uphill energy transfer, e.g. to the E680 energy level. Direct evidence for the operation of such a process is presented in Fig. 1c showing a comparison of the fluorescence emission spectra recorded from the same LHCII sample excited in the higher and in the lower energy regions (at 635 nm and at 705 nm) with respect to

the emission spectral window. Almost identical shapes of both the emission spectra recorded are consistent with the interpretation according to which both the fluorescence emissions originate from the same energy level (Fig. 1c). Comparison of the quantum yields of fluorescence excited at 635 nm and at 705 nm shows that the quantum yield of the emission excited at 705 nm is lower by a factor of 6.3 than the fluorescence excited in the higher energy region (at 635 nm). In principle, the ratio of the uphill and downhill rate constants multiplied by the state populations should follow the Boltzmann distribution [30]. The difference between the E680 and E700 states (420 cm^{-1}) expressed in the kT units corresponds to the temperature of 627 K that is higher than the room temperature (298 K). On the other hand, the energy gap and kT ratio ($\Delta E/kT$) that is only as high as 2.1 at 298 K, makes the uphill energy transfer relatively efficient ($\exp(-\Delta E/kT)=0.12$), as manifested by the fluorescence quantum yield of the anti-Stokes excitation. It is very likely that the effective energy gap between the E680 and E700 states is even lower than 420 cm^{-1} owing to the degeneracy of states at room temperature. The fact that fluorescence spectrum can be recorded under the anti-Stokes conditions and that the quantum yield is relatively high is a direct manifestation of the activity of the up-conversion process. Fluorescence intensity in such a system, recorded at 680 nm and excited at lower energies, drops down with the temperature decrease (see Fig. 1d) and with increasing a distance between the excitation and observation wavelengths (see Supporting Information Fig. S1). Importantly, application of a two-dimensional electronic spectroscopy, which correlates the fluorescence excitation and observation wavelengths, enabled to detect a temperature-dependent uphill excitation energy transfer even in the trimeric LHCII [31]. Another kind of evidence for the operation of the thermally-driven up-conversion in the supramolecular structures of the complex is shown in Fig. 2d presenting fluorescence emission kinetics, excited at 705 nm and detected in the higher energy region, at 680 nm. The process of a thermally-driven up-conversion can be observed at room temperature but it is not possible to operate at

77 K, for energy reasons (Fig. 2d). This means that the observed thermally-driven up-conversion has to be combined with a thermal deactivation of chromophores electronically-excited and localized in the close neighbourhood. It should be emphasized that the illuminated LHCII proved to be a very efficient emitter of heat that can be transmitted over long distances in the supramolecular structures of the protein [32]. The fact that fluorescence emission from the E700 can be detected at low temperatures, enables to determine fluorescence lifetime of this state (Fig. 3, Fig. S2). It appears that the average lifetime of the E700 is substantially longer than that of the E680 state: 4.18 ns versus 2.44 ns, very close to the previous determinations of similar LHCII systems [25]. This suggests that the presence of the E700 energy level, below the E680 state, creates conditions for effective excitation quenching from this latter state, by a downhill energy transfer. Most probably, due to such a quenching, the fluorescence lifetime of E680 is even shorter at the higher temperatures (1.13 ns, see Fig. 2b). Intriguingly, fluorescence emission from the E700 state of aggregated LHCII is not effectively observed at room temperatures (see Fig. 1b and the literature references [18, 23-26]). This observation implies that under such conditions the E700 is almost entirely depopulated via non-radiative processes including the uphill energy transfer to E680 and thermal dissipation. It is also possible that the E700 state is depopulated via energy transfer to the low-energy states of LHCII (730-780 nm, see Fig. 1b, Fig. S3). Such a long-wavelength fluorescence emission band is particularly observed at low temperatures, under the absence of the thermally-driven up-conversion. Interestingly, the long-wavelength spectral forms can be detected in a single molecule emission spectra of LHCII [29]. Coexistence of the two-direction energy transfer pathways between the E680 and E700 states that are present in supramolecular structures of LHCII, namely the spontaneous down-conversion from E680 to E700 and the thermally-driven up-conversion from E700 to E680, can be discussed in the context of overall excitation energy flows in the photosynthetic apparatus of plants (see the model presented in Fig. 4).

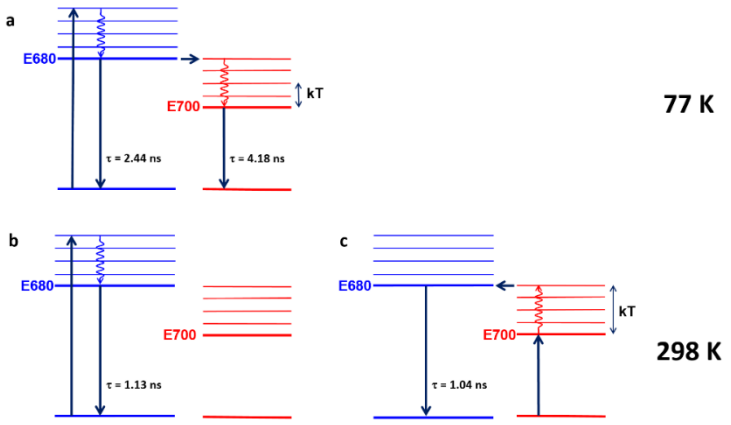


Figure 3. Energy level diagrams representing selected electronic states and transitions in supramolecular structures of LHCII. The average fluorescence lifetime values are also reported, determined on the basis of the decay kinetics shown in Fig. 2 and Fig. S2. Vertical arrows up represent light absorption, vertical arrows down represent fluorescence, wavy and horizontal arrows represent nonradiative processes.

The central element of this model, which may be referred to as the "energy recycling unit", is based upon spontaneously formed supramolecular structures of LHCII. We have modelled formation of such structures in LHCII-containing lipid membranes. Dependently of an actual protein-lipid ratio, the proportion between the E680 and E700 bands can be different in the fluorescence emission spectra, reflecting ability of LHCII to self-aggregate in a system [18]. One of the weaknesses of the model applied is a possibility of incorporation of the protein in two opposite orientations, i.e. the N- and C-termini facing to both sides of the lipid bilayers, in contrast to the native thylakoid membranes. On the other hand, very close agreement of the spectral shapes representing the aggregated structures of LHCII formed in the natural thylakoid membranes and under laboratory conditions justifies the application of such a system for model study [26]. Nevertheless, for the sake of caution, the conclusions drawn on the basis of the experiments carried out on model membranes were confirmed in the present study by the results of experiments carried out on intact leaves.

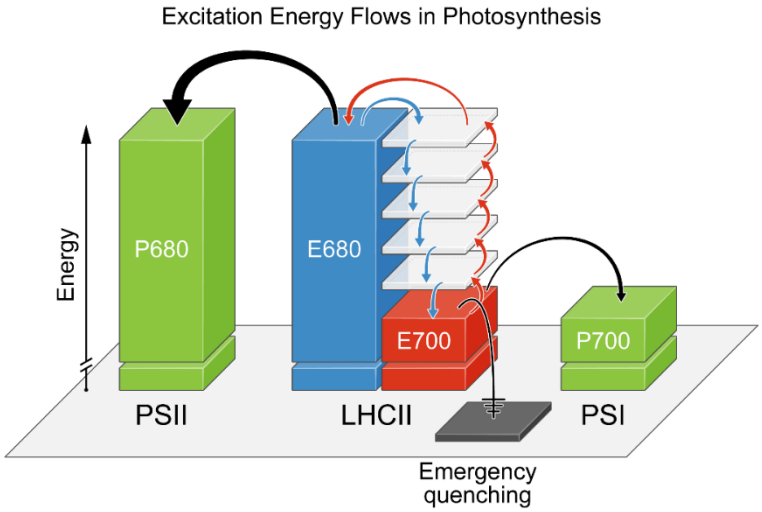


Figure 4. A simplified "energy recycling unit" model representing the excitation energy flows in the photosynthetic apparatus of plants. The scheme presents the excitation energy flows between the light-harvesting antenna complexes (LHCII) and the reaction centres of Photosystem II (P680) and Photosystem I (P700). Two experimentally identified, energetically coupled excitation energy states of supramolecular structures formed by LHCII are marked as E680 and E700. Energy circulation between these states is represented by arrows: the quenching of E680 by E700 and the thermally-driven up-conversion from E700 to E680. An emergency excitation quenching channel leading to thermal dissipation of excess excitation energy is also shown.

Numerous consequences of the operation of the molecular system based on supramolecular structures of LHCII can be envisaged, important for the activity of PSII and for the efficiency of photosynthetic energy conversion in plants. The main consequence seems to be the ability to use low-energy excitations (corresponding to wavelengths longer than 680 nm) to drive the photo-physical processes in the PSII reaction centre, thanks to the utilization of a fraction of the excitation energy dissipated as heat in the pigment-protein antenna systems. Concerning the utilization of near infrared radiation, it is worth mentioning that the core complex of PSII also contains some red-shifted states, in the core antenna CP43/CP47 and possibly some state emitting light up to more than 700 nm, which might be part of the reaction center excimer and less fluorescent. The 695 nm band was identified in the low-temperature fluorescence emission spectra recorded from isolated CP47 complexes of PSII [33]. This shows that the presence and

activity of LHCII significantly enhance photophysical processes that are already potentially present in PSII. On the one hand, recovering a certain fraction of the excitation energy already dissipated as heat shall increase the overall energy efficiency of PSII and therefore of photosynthesis. On the other hand, operation of this mechanism extends the action spectrum of PSII, towards the far-red spectral region, beyond the limit defined by the energy of the PSII reaction centre. A steep decrease in light absorption at wavelengths greater than 680 nm is associated with a very sharp decrease in the photosynthetic activity, referred to as the "red drop" [11]. Importantly, a significant photosynthetic activity of light from the far-red spectral region has been demonstrated by direct measurements of photosynthetic oxygen evolution [34, 35] and by means of EPR spectroscopy [36]. This effect has been observed with a single wavelength excitation [35, 36] and with isolated PSII particles [36] and therefore does not correspond directly to the classical Emerson effect [37]. The photo-physical mechanism, reported in the present work, of thermally-driven up-conversion of the low-energy excitations to the energy level sufficiently high to drive the photochemical activity of PSII provides a direct explanation for the photosynthetic activity of light beyond the "red drop", long-wavelength threshold. In our opinion, the mechanism of the thermally-driven up-conversion proved here to operate both in model systems and *in vivo*, is more realistic than involvement of unidentified "long-wavelength chlorophylls" [35] or hypothetical excited X^* states [36]. Importantly, the uphill activation energies for the far-red light-driven oxygen evolution in sunflower, determined based on the temperature dependencies, have been found to correspond directly to the energy gaps between the level attributed to PSII (680 nm) and the energies of light representing excitations in the long-wavelength spectral region [35]. Interestingly, the quantum yield levels of PSII activity corresponding to 680 nm and to the far-red region differ only by the factor of ~6, despite pronounced differences in light absorption in the corresponding spectral regions [35]. A very close factor (6.3, see above) has been determined in the present work for the chlorophyll fluorescence

quantum yield in LHCII, in the direct, downhill fluorescence excitation and in the process of the uphill fluorescence excitation, mediated by the processes of thermally-driven up-conversion. The fact that two energetically coupled energy states reported in the present work, E680 and E700, are apparently precisely tuned to the energies of the reaction centres of both photosystems, namely P680 and P700, provides favourable conditions for LHCII to act as a universal antenna complex. From the standpoint of energy supply to PSI, it can be assumed that there is no particular need to transfer low-energy excitations to this photosystem from LHCII. PSI of plants contains light absorption forms practically extending to 715 nm and emitting fluorescence in the 720-730 nm spectral region [9]. Moreover, LHCII energetically coupled to PSI remains most probably in the trimeric form [9]. Interestingly, the thermally-driven up-conversion has been recently shown to operate efficiently in isolated PSI [38]. This can lead to a more general conclusion that uphill energy transfer is a common mechanism in the photosynthetic apparatus and potentially important for the process of photosynthesis.

Natural photosynthesis constantly inspires technological activity focused on a construction of biomimetic solar cells based on isolated elements of the photosynthetic apparatus [39]. The results of the present study show that engineering of solar cells containing not only photosynthetic reaction centers but additionally pigment-protein antenna complexes, in the form of supramolecular structures, can improve an overall energy yield of light energy conversion owing to the process of recycling of a fraction of excitation energy dissipated as heat.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://

Experimental details, Excitation wavelength dependency of up-conversion-induced chlorophyll fluorescence in LHCII, Chlorophyll *a* fluorescence decay kinetics in LHCII, Fluorescence analysis of long-wavelength LHCII spectral forms (PDF).

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Notes

The authors declare no competing financial interest.

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