Validation of carbon isotopologue distribution measurements by GC-MS and application to ¹³C-metabolic flux analysis of plant tricarboxylic acid cycle

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Abstract

The estimation of metabolic fluxes in photosynthetic organisms represents an important challenge that has gained interest over the last decade with the development of ¹³C-Metabolic Flux Analysis at isotopically non-stationary steady-state. This approach requires a high level of accuracy for the measurement of Carbon Isotopologue Distribution in plant metabolites, which has still not been accurately evaluated at the isotopologue level for GC-MS. Thus, uncertainties remain about the accuracy of the CID measured and subsequently, the metabolic fluxes associated. Moreover, a significant part of our knowledge about the functioning and the regulation of plant TCA cycle is solely based on ¹³C-labeling experiments analyzed with Nuclear Magnetic Resonance. But this method is not very sensitive at low ¹³C abundance and can hardly fully resolve isotopologue dynamics inherent to the TCA cycle compared to MS. Here, we developed a workflow to validate the measurements of CIDs from plant metabolites with GC-MS by producing tailor-made *E. coli* standard extracts harboring a predictable binomial CID for all metabolites. Most of our TMS-derivatives mass fragments were validated with these standards and at natural isotope abundance in plant matrices. Then, we applied this validated MS method to investigate the light/dark regulation of plant TCA cycle metabolic fluxes by incorporating U-¹³C-pyruvate to *Brassica napus* leaf discs. We took advantage of pathway-specific isotopologues/isotopomers to show that the TCA cycle has a nearly four-fold higher contribution to Citrate and Malate production than phosphoenolpyruvate carboxylase (PEPc). Overall, the dynamic of ¹³C-enrichment in Citrate, Glutamate, Succinate and Malate ruled out the hypothesis of the stored citrate contribution to TCA cycle and highlighted the major contribution of the forward plant TCA cycle to malate biosynthesis in both dark and light conditions. Interestingly, the light-dependent ¹³C-

incorporation into Glycine and Serine showed that nearly 5% of TCA cycle-derived CO₂ was actively reassimilated by the photosynthesis. We also observed an increase of PEPc contribution to malate production in the light while there was a higher involvement of PEPc-derived carbons to citrate production in dark condition. The results suggested an important role for Malate import into mitochondria in sustaining TCA cycle in dark conditions.

1 INTRODUCTION

Metabolic fluxes within central metabolism are among the most important determinants of plant cell physiology, supporting central energetical functions and allowing them to quickly adapt to environmental changes by controlling metabolite turnover (Clark et al., 2020). Given that most of fluxes supported by a metabolic network cannot be measured directly, Metabolic Flux Analysis (MFA) technics were developed to infer information about metabolic fluxes from isotopic labeling experiments (Ratcliffe and Shachar-Hill, 2006). ¹³C-MFA relies on the tracing of ¹³C throughout a metabolic network in a relative metabolic steady-state to model and/or compare metabolic fluxes, *i.e.* that the metabolic fluxes are stable across the experiment because the associated metabolite pool sizes are relatively constant over the time of labeling except for the source (consumption) and sink (production) metabolites (Allen and Young, 2019; Antoniewicz, 2021). ¹³C-MFA is usually performed at isotopic steady-state for heterotrophic tissues for practical reasons, and also because they can reach this isotopic state (Hay et al., 2014). Conversely, ¹³C-MFA in phototrophic tissues can only be performed at isotopically non-stationary steady-state (INST-MFA), thus requiring a higher accuracy in the measurement of ¹³C enrichment within metabolites (Cheah and Young, 2018). Over the last decades, ¹³C-MFA has been applied successfully to microorganisms to highlight new metabolic routes and identify metabolic bottlenecks for the purpose of metabolic engineering (Cheah et al., 2020; Antoniewicz, 2021; Treves et al., 2022). In plants, ¹³C-INST-MFA has been scarcely applied to date, due to the complexity in modelling accurately metabolic fluxes in a multicompartmental system (Heise et al., 2015; Kruger and Ratcliffe, 2021; Wieloch, 2021). However, accurate measurement of ¹³C-enrichment is a prerequisite for precise evaluation of isotopic labeling dynamics and accurate modelling of metabolic fluxes. Indeed, small errors in mass isotopologue distribution can propagate to large errors in estimated fluxes (Antoniewicz et al., 2006).

Two major analytical methods are currently employed to analysis the Carbon Isotope Distribution (CID) within plant metabolites during ¹³C-isotope tracing experiments: Nuclear Magnetic Resonance (NMR) and Mass Spectrometry (MS). NMR analysis of the ¹³C enrichment is generally performed with ¹³C-decoupling ¹H sequences and provides information respectively at isotopomer level (by integrating ¹³C-¹³C interactions) (Kruger et al., 2007; Masakapalli et al., 2010; Abadie et al., 2017). Although this method is quantitative for the ¹³C at the isotopomer level and allows to significantly separate carbon atoms of different chemical structures, it has a low sensitivity at natural ¹³C abundance (large amounts of fresh material required), thereby affecting the accuracy for ¹³C enrichment measurements regarding metabolites found in small quantities (Dellero et al., 2016; Abadie and Tcherkez, 2021). In addition, this method is time-consuming due to the longer relaxation time for ¹³C nucleus and the need to accumulate many spectrums to get a good signal (Abadie and Tcherkez, 2021). As a consequence, most of the ¹³C-tracing isotope experiments recently performed in plants and analyzed by ¹³C-NMR have only one time-point (Abadie et al., 2016; Abadie et al., 2018; Abadie and Tcherkez, 2019a;b), which questions the estimation of metabolic fluxes therein (metabolic/isotopic steady-state and treatment of inactive metabolite pools). To date, the accurate estimation of metabolic fluxes in phototrophic tissues requires to determine the isotopic and metabolic state of the system by coupling transient kinetics of ¹³C-labeling with absolute quantification of metabolite pools (Kruger and Ratcliffe, 2015; Cheah and Young, 2018; Clark et al., 2020; Antoniewicz, 2021). Some methodological

improvements were recently made with highly sensitive ^1H -derived sequences to improve the resolution of the ^{13}C atoms at the isotopomer level (Sinnaeve et al., 2018), and its future use for ^{13}C -INST-MFA in plants will provide interesting perspectives. Conversely, mass spectrometry has been preferentially used to analysis the Carbon Isotope Distribution (CID) within plant metabolites during time-course ^{13}C -isotope tracing experiments (Szecowka et al., 2013; Ma et al., 2014; Medeiros et al., 2018; Treves et al., 2022). This method is more sensitive at natural ^{13}C abundance and can be implemented at high throughput. However, MS only gives information at the isotopologue level and an absolute quantification of the metabolite content is generally required to subsequently infer metabolic fluxes. In addition, the analysis of CID for amino acids by LC-MS revealed some bias and errors in the measurement of CID (Heuillet et al., 2018), thus highlighting the need to evaluate measurement accuracy with GC-MS fragments for subsequent ^{13}C -MFA in plants.

Among plant metabolites, organic and amino acids are major players for plant primary metabolism. They are actively involved in crucial physiological processes such as photorespiration, *de novo* nitrogen assimilation and mitochondrial respiration (Xu et al., 2012; Zhang and Fernie, 2018; Dellero, 2020; Timm, 2020). Most of the carbon backbones for amino acid biosynthesis arise from organic acids of the tricarboxylic acid (TCA) cycle (Galili et al., 2016). Interestingly, an important part of our knowledge about the functioning and regulation of plant TCA cycle is almost exclusively based on ^{13}C -labeling studies that used NMR. These previous works suggested that the plant TCA cycle is not cyclic in the light, with two branches acting separately (Tcherkez et al., 2009; Sweetlove et al., 2010): i) a first branch stimulated by the anaplerotic activity of phosphoenolpyruvate carboxylase (PEPc) leading to the production of Oxaloacetate-malate-fumarate ; ii) a second branch stimulated by the pyruvate dehydrogenase activity leading to the production of citrate-isocitrate-2-oxoglutarate-glutamate. Based on the NMR analysis of combined $^{13}\text{CO}_2$ and $^{15}\text{NH}_4^{15}\text{NO}_3$ labeling experiments in oilseed rape leaves, it was proposed that a high proportion of non-labeled citrate (vacuolar pool probably) was involved in the synthesis of 2-oxoglutarate under light conditions to sustain *de novo* nitrogen assimilation through glutamate (Gauthier et al., 2010). Under this scenario, the citrate pool would be replenished during dark conditions from glycolysis-derived ^{13}C -pyruvate, leading to a higher ^{13}C enrichment of the citrate when compared to light conditions. Thus, the light and dark environmental conditions are expected to introduce important variations of metabolic fluxes within the plant TCA cycle and its associated PEPc anaplerotic pathway. However, such conclusions and hypotheses remain to be carefully investigated with mass spectrometry and a ^{13}C -MFA framework, *i.e.* with time-course ^{13}C -isotope tracing experiments and pool size quantification.

Currently, the routine analysis of CID on plants is often assessed for organic and amino acids by GC-MS using TMS-derivatives (Szecowka et al., 2013; Araujo et al., 2014; Lima et al., 2018; Treves et al., 2022). This TMS-based method notably allowed to follow the fate of sucrose breakdown within guard cells in connection with TCA cycle (Medeiros et al., 2018). More recently, a ^{13}C -positionnal isotopomer approach was also proposed for glutamate and malate to investigate the fate of PEPc-derived carbon from ^{13}C -labeling experiments (Lima et al., 2021). The comparison of two derivatization methods suggested that TBDMS-derivatives were potentially less prone to errors for amino acid analysis than TMS-derivatives (Antoniewicz et al., 2007; Young et al., 2014). However, the suitability of a TMS-derivative molecular ion in each of these studies was only performed using U- ^{13}C -metabolites, which do not evaluate the entire CID but only the carbon distribution between the unlabeled isotopologue and the fully labeled isotopologue (M0 and M5 for a 5-carbon metabolite). Unfortunately, isotopic effects can occur at all stages of the analytical procedures and depend on the tested metabolite and its isotopic content. For instance, the present methodologies used have not assessed the impact of each isotopologue to the final isotopologue distribution, which can lead to important bias during interpretation of the results. Thus, the reliability of CID measurements by GC-

MS has to be evaluated for each individual isotopologue of all the compounds of interest to subsequently ensure a reliable use for monitoring isotopic labeling dynamics. This is even more important for plants, given that plant metabolites show strong differences of ^{13}C -enrichment dynamics between amino and organic acids (Szecowka et al., 2013). For this purpose, the recent development of an *E. coli* ^{13}C biological standard with controlled patterns of CID for organic and amino acids may help to address this issue (Millard et al., 2014). Indeed, the recent use of these reference samples allowed to identify important analytical biases in the CID measurement of Leucine and Asparagine in a concentration-dependent manner by LC-MS (Heuillet et al., 2018). Interestingly, this validated LC-MS method successfully measured nitrogen isotope distribution within plant amino acids and led to the accurate estimation of metabolic fluxes from ^{15}N -isotope transient labeling experiments in oilseed rape leaves at a metabolic steady-state using the ScalaFlux approach (Dellero et al., 2020b; Heuillet et al., 2020; Millard et al., 2020).

Here, we proposed a workflow to assess the measurement accuracy of CID from organic and amino acids by GC-MS for applications in plants. To do so, we built up an analytical method maximizing information on carbon backbones and analyzed two specific *E. coli* ^{13}C biological standards with controlled patterns of CID for organic and amino acids (^{13}C -PT). We also analyzed chemical standards and plant samples at natural ^{13}C abundance to identify potential bias arising from matrix and analytical systems. Next, we applied our MS method to investigate the light/dark regulation of metabolic fluxes within plant TCA cycle by performing U- ^{13}C -pyruvate labeling kinetic experiments on oilseed rape leaves under light or dark conditions. Using ^{13}C -MFA principles and some specific isotopologues, we calculated the fractional contribution of phosphoenolpyruvate carboxylase (PEPc) and TCA cycle to malate and citrate production and proposed a simplified carbon flux map with relative metabolic fluxes. Our results notably revealed that under non-limiting pyruvate conditions, PEPc activity remained nearly four-fold lower than the TCA cycle activity independently of the light/dark stimulation. In addition, we showed that pyruvate dehydrogenase and TCA cycle-derived CO_2 contributed to around 5% of net CO_2 assimilation under light conditions.

2 MATERIALS AND METHODS

Organisms and culture

The strain *Escherichia coli* K-12 MG1655 was grown on minimal medium containing 5 mM KH_2PO_4 , 10 mM Na_2HPO_4 , 9 mM NaCl , 40 mM NH_4Cl , 0.8 mM MgSO_4 , 0.1 mM CaCl_2 , 0.3 mM thiamine, and 45 mM of the ^{13}C -labeled Acetate mixture as the sole carbon source. Acetate and thiamine were sterilized by filtration, whereas the other compounds were autoclaved separately. Batch cultures were carried out in 1-L baffled flasks with 500 mL of medium sparged with synthetic gas (80% N_2 /20% O_2 containing <1 ppm CO_2 , Air Liquide). Cultivation was performed in a Multifors Bioreactor (Infors HT, Bottmingen-Basel, Switzerland) at 37 °C and 200 rpm of orbital shaking. The pH was maintained at 7.0 by the addition of 2 M HCl. Cell growth was monitored by measuring optical density at 600 nm with a Genesys 6 spectrophotometer (Thermo, Carlsbad, CA, USA). Amount of biomass in each sample was calculated using a coefficient of 0.37 g of cell dry weight per OD unit. *Arabidopsis thaliana* seeds were sterilized for 10 min with absolute ethanol containing 7.5% NaClO and rinsed two times with absolute ethanol. Seeds were sowed on ½ MS media containing 1% Agar and incubated to the dark at 4°C for 48-72 hours to break seed dormancy. Seedlings were grown at ambient air under a day/night cycle with 16 h light/8h dark at a light intensity of 120 $\mu\text{mol photons.m}^{-2}.\text{s}^{-1}$ and 20°C day/18°C night. Seedlings were harvested after 2-3 weeks by instant sampling with liquid nitrogen and lyophilized for 96 h before to be stored at -80°C until analysis. 2-month old *Brassica napus* plants were

obtained from seeds exactly as described before (Dellero et al., 2020a). The leaf L9 was used for all experiments (natural abundance and U-¹³C-pyruvate incorporation).

U-¹³C-pyruvate labeling experiments

Four hours after the beginning of the light period, each leaf was cut at the basis of the petiole and then 66 leaf discs (0.8 cm²) were randomly punched with a cork-borer in both laminas of each leaf. Leaf discs were first floated in a 10 mM MES-KOH (pH 6.5) buffer for 15 min in order to process the L9 leaf for four plants. In this experiment, one leaf of a plant corresponded to a biological replicate. Then for each time point, six leaf discs from the same leaf were transferred in a well of a 6-well microplate filled with 4 mL/well of a buffer containing 10 mM MES-KOH (pH 6.5) and 10 mM U-¹³C-Pyruvate (¹³C isotopic purity of 99%) (Eurisotop, France). All incubations were performed at 20 °C with an orbital shaking of 70 rpm under either a continuous light of 120 μmol photons m⁻² s⁻¹ or under dark conditions using multiple opaque plastic bags, similarly to the previous experimental set-up used for dark-induced senescence (Dellero et al., 2020a). The timepoint 0 was the same for both conditions as it corresponded the beginning of the experiment. For each time point (0, 0.5, 1, 2, 4, 6 hours of incubation), the leaf discs were harvested and rinsed three times (10 sec) with a neutral buffer (i.e., without U-¹³C-Pyruvate) and dried for 10 s on a clean tissue before being frozen in liquid nitrogen and stored at -80 °C. The timepoint 0 of the experiment was incubated 2 sec with the labeled buffer before being rinsed in order to evaluate the fraction of ¹³C-pyruvate detected in our analysis that was not incorporated inside leaf discs (leading to a mean ¹³C enrichment close to 35% for pyruvate at T0 in our study).

Analysis of ¹³C-labeled Acetate by NMR

An equimolar mixture of all ¹³C-isotopomers of acetate was produced from ¹²C-Acetate, 1-¹³C-Acetate, 2-¹³C-Acetate and U-¹³C-Acetate (13C-purity 99%, Eurisotop, France; **Figure 1**). Quantitative 1H 1D-NMR was used to control the isotopic composition of the ¹³C-labeled substrate before use. NMR spectra were recorded with an Avance 500 MHz spectrometer (Bruker, Rheinstetten, Germany) at 298 K, using a 30° pulse and a relaxation delay of 20 s. The proportion of each isotopic form was quantified by fitting using a mixed Gaussian–Lorentzian model (Millard et al., 2017).

Metabolite extraction and quantification

¹³C-PT full metabolites and ¹³C-PT protein hydrolysates extracts were obtained following procedures described previously (Heuillet et al., 2018). Briefly, for the ¹³C-PT full metabolites extract, 1 mL of *E. coli* broth at OD600 nm = 1.7 was micro-filtered and the pellet was rapidly quenched at -20 °C into a mixture of acetonitrile/methanol/H₂O (2:2:1) with 0.1% formic acid. After a centrifugation step (7000g, -20 °C, 15 min), the supernatant was evaporated and resuspended in 250 μL of Milli-Q water, to be stored at -80 °C until analysis. For the ¹³C-PT protein hydrolysates extract, the pellet obtained with the cellular extract was subjected to an acid hydrolysis with 6N HCl at 110 °C overnight. Samples were evaporated and rinsed 3 times with Milli-Q water. After a centrifugation step (10 min, 12 000g), the supernatant was stored at -80 °C. Polar metabolites of freeze-dried Arabidopsis seedlings and *B. napus* leaves were extracted with a MeOH/Chloroform/H₂O mixture following procedures described previously (Dellero et al., 2020b). The quantification of amino acid content was performed by HPLC-UV exactly as described previously (Dellero et al., 2020b). The quantification of organic acids was performed by GC-FID exactly as described previously (Bianchetti et al., 2021).

Determination of Carbon Isotopologue Distribution by GC-MS

Dried aliquots of samples were derivatized on a MultiPurpose Sampler Gerstel and analyzed by GC-MS (GC-2010 Plus Shimadzu coupled to a triple quadrupole mass spectrometer GCMS-TQ8040 Shimadzu). Polar metabolites were first derivatized for 1 h at 40°C with 50 µL of pyridine containing 20 g.l⁻¹ methoxyamine chloride and then were incubated with 50 µL of N-Methyl-N-(trimethylsilyl)-trifluoroacetamide (MSTFA) for 0.5 h at 40°C. 1 µL was injected with a split mode 1/20 at 260°C on a TG-5MS column (26098-1420, Thermo Scientific). Separation was performed in a Helium gas-stream at 1 mL.min⁻¹ using the following temperature ramp : 4 min at 70°C, +10°C.min⁻¹ until 198°C, 2 min at 198°C, +1°C.min⁻¹ until 202°C, +15°C.min⁻¹ until 268°C, 3 min at 268°C, +1°C.min⁻¹ until 272°C, +10°C.min⁻¹ until 310°C and 7 min à 310°C. The transfer line was maintained at 300°C and the ion source at 200°C. Ionization was performed by electron impact at 70 eV and the MS acquisition was carried out after a solvent delay of 5 min in full scan mode at 7 spectra.s⁻¹ over the m/z range 45-700. MS Spectra were processed with the software “GCMS Solutions” supplied by Shimadzu to adjust retention times between each run using a mixture of alkanes (C7 to C40, Sigma-Aldrich). Peak identity was established according to their retention time and mass fragmentation spectra with respect to authentic standards derivatized (**Table S1**), and by reference to mass spectra of derivatives in the US National Institute of Standards and Technology database and in the Golm metabolome database. All mass isotopologue raw areas were subjected to correction for the contribution of naturally occurring isotopes of all elements except the carbon arising from the metabolite backbone using the IsoCor v2.1.4 package of Python (<https://github.com/MetaSys-LISBP/IsoCor>) in low resolution mode (Millard et al., 2019). In addition, unlabeled chemical standards and Arabidopsis and *B. napus* leaves were additionally corrected for the contribution of naturally occurring isotopes of the carbon arising from the metabolite backbone (giving a residual molecular ¹³C enrichment of 0 for molecules labeled at natural ¹³C abundance). ¹³C-PT samples were not concerned by this additional correction since they held fully controlled ¹²C/¹³C patterns at all positions. A correction method dedicated to IsoCor was specifically created for the TMS-derivative fragment described in this study (**Table 1**), *i.e.* by considering the contribution of all atoms added by the derivatization on the isotopic measurement of the monitored fragments (chemical separation of all atoms arising from the metabolite and the derivative). All files required to run this method on IsoCor are available in the supplementary material of this article (**Supplementary files**) and all information will be updated indefinitely in the following INRAE dataverse (Dellero and Filangi, 2021). The method is publicly available and can also be run directly from GC-MS source files using the following Galaxy workflow (<https://galaxy.genouest.org/>, section “Metabolomics”, subsections “Conversion GCMS PostRun Analysis to IsoCor” and “Isotope Correction for mass spectrometry labeling experiments”). For this purpose, each raw GC-MS dataset should contain a column “Name” for the fragments considered and a column “Area”. The name of each fragment must be written exactly as specified in the “Metabolite.dat” (an example file is provided in the **supplementary material**). Fractional Carbon Isotopologue Distribution and mean ¹³C enrichment were directly obtained from IsoCor outputs (isotopologue_fraction and mean_enrichment) while CIDs were converted to binomial distributions by multiplying with 2ⁿ, where n is the number of carbon atoms (Millard et al., 2014). The accuracy of measurements was assessed from the closeness of experimental and predicted values for CIDs and mean ¹³C enrichments at either ¹³C-PT abundance or at natural ¹³C abundance. CID accuracy was formally defined as the mean of absolute differences between measured and predicted values for each isotopologue (Heuillet et al., 2018).

Pathway fractional contribution

Pathway fractional contribution (FC) can be applied to a metabolic network in a metabolic and isotopic steady-state to compare the relative contribution of the considered pathway to the production of a metabolite, as long as these pathways produced specific isotopologues that can be monitored (Buescher et al., 2015). Here, FC was calculated from PEPc and TCA cycle-specific mass

isotopomers/isotopologues for Malate and Citrate from the timepoint 4 and 6 hours (metabolic steady-state and assumed similar pseudo-isotopic steady-state). In the following equations, mi refers to the isotopologue enrichment calculated with respect to the CID (sum of all mi for a given metabolite equal to 1). As all these calculations are relative, it is also possible to use corrected areas of mass isotopologues (outputs from IsoCor). For Malate_2TMS_C1C4, M1 isotopologue measurement reflected both PEPc and TCA cycle isotopomers. However, the ratio between M1 and M3 isotopologues of Malate_2TMS_C1C4 produced by the TCA cycle (denoted X in the following equations) can also be observed in Succinate_2TMS_C1C4 and Glutamate_2TMS_C2C5, thereby reflecting the M0 and M2 from mitochondrial active pool of Acetyl-CoA, *i.e.*:

$$SuccinateC1C4\left(\frac{m3}{m1+m3}\right) = GlutamateC2C5\left(\frac{m3}{m1+m3}\right) = Acetate\left(\frac{m2}{m0+m2}\right) = X$$

From this parameter, we calculated the FC of PEPc and TCA cycle to Malate and Citrate production as follows:

$$FC\ Malate_{PEPc} = MalateC1C4\left(\frac{m1 - \left(\left(\frac{1-X}{X}\right) \times m3\right)}{m1+m2+m3}\right)$$

$$FC\ Malate_{TCA\ cycle} = MalateC1C4\left(\frac{m2 + \frac{m3}{X}}{m1+m2+m3}\right)$$

$$FC\ Citrate_{PEPc} = CitrateC1C6\left(\frac{m1+m3}{m1+m2+2 \times m3+m4}\right)$$

$$FC\ Citrate_{TCA\ cycle} = CitrateC1C6\left(\frac{m2+m3+m4}{m1+m2+2 \times m3+m4}\right)$$

It should be noted that $m3$ was used twice in the FC Citrate calculations because this isotopologue is produced by both pathways. Finally, the FC of pyruvate dehydrogenase (PDH) and TCA cycle-derived CO₂ to photosynthesis was calculated using X as a proxy for the ¹³C/¹²C ratio of this CO₂ fraction in light conditions. Considering the isotopic and metabolic steady-state for these metabolites, the mean ¹³C enrichment of X should be observed in Serine if this fraction of CO₂ accounted for 100% of CO₂ fixed by photosynthesis, *i.e.*:

$$FC\ photosynthesis_{PDH\ and\ TCA\ cycle\ CO_2} = \frac{mean\ ^{13}C\ enrichment\ (SerineC1C3)}{X}$$

Mean ¹³C enrichment of Serine C1C3 was recalculated from M1 isotopologue of Serine_2TMS_C1C2 and Serine_2TMS_C2C3 assuming that Serine was only labeled in the C2 of these fragments based on the validated hypothesis (**Figure 7B**). Statistical comparisons were performed using a Student-test (2 groups) or an ANOVA followed by a post-hoc Tukey-HSD test (3 or more groups) with a level of p-value <0.05 and Rstudio (RStudio Team, 2016).

3 RESULTS

The proposed workflow (**Figure 1**) consisted in three steps: 1) the build-up of a method to identify and select abundant molecular ions deriving from TMS-derivatives with a large number of carbons; 2) the production of two reference ^{13}C -labeled samples with a controlled pattern for CID using *E. coli* as a metabolic factory; 3) the validation of TMS-derivatives mass fragments for CID measurements with our GC-MS method for applications in plants using unlabeled and ^{13}C -labeled plant samples.

Selection of GC-MS mass fragments maximizing information on carbon back-bones of organic and amino acids

We combined a literature survey analyzing common pathway of TMS-derivative mass fragmentations with the injection of chemical standards in our analytical system to select some abundant fragments maximizing the information on carbon backbones. Their chemical identity was further confirmed by the observed CID pattern with our two reference ^{13}C -labeled samples with a controlled pattern for CID. Derivatives with only one TMS group were not considered in this study due to the low yield of these fragments after fragmentation and the occurrence of important contaminations for low m/z . Overall, TMS-derivatives of most α -amino acids were characterized by an α -cleavage of the carboxylic function in the form of a TMS- CO_2 (m/z 117) (Harvey and Vouros, 2020), thus leading to the abundant formation of fragments containing all others atoms of the α -amino acids except the C1 (**Table 1**). Only Ornithine(4TMS) could be detected without any mass loss (**Table 1**, (Harvey and Vouros, 2020)). Aromatic amino acids showed an additional α -cleavage of the amine function leading to the additional production of an abundant m/z 218 fragment containing the C1-C2 for Phenylalanine, Tryptophan and Tyrosine (**Table 1**, (Lai and Fiehn, 2018; Harvey and Vouros, 2020)). Only tryptophan did not proceed through an α -cleavage of the carboxylic function compared to Phenylalanine and Tyrosine, but the other fragment containing the C3 to C11 atoms resulting from the α -cleavage of the amine function could be abundantly detected. Serine(3TMS) and Valine(2TMS) showed also some specific abundant molecular ions in addition to the α -cleavage of the carboxylic function (**Table 1**, (Harvey and Vouros, 2020)). Arginine, Histidine and Cysteine could not be detected by GC-MS using TMS derivatization and were not further considered in our study. Beside this, β -Alanine, an important β -amino acid from plants, showed a classical abundant fragment resulting from the loss of methyl from a TMS group, which is known to be very low in α -amino acids (**Table 1**, (Lai and Fiehn, 2018)). Interestingly, fragmentation of the TMS-derivatives of organic acids related to the TriCarboxylic Acid cycle proceeded through an α -cleavage within a TMS group leading to the loss of a methyl radical and producing a [M-15] fragment containing all carbon atoms. However, this fragment was very weak for Citrate and Malate. The highest fragments maximizing information on carbon backbones were m/z 273 (C1 to C5) for Citrate(4TMS) and m/z 233 (C2 to C4) for Malate(3TMS). The identity of the carbons hold by these fragments has been recently assigned using position-specific ^{13}C -labeled standards prepared by *in vitro* enzymatic reactions (Okahashi et al., 2019).

Evaluation of CID measurements by GC-MS using reference ^{13}C -PT samples and plant samples at natural isotope abundance

Next, we produced reference ^{13}C -labeled samples using *E. coli*, denoted here ^{13}C -PT. This strategy relies on the use of an accurate equimolar mixture of all ^{13}C isotopomers of acetate to biologically produce metabolites with fully controlled and predictable ^{13}C -labeling patterns (Millard et al., 2014). Thus, molecular ^{13}C enrichment of each carbon-containing metabolite will be theoretically equal to 50% and all isotopomers will be present in equimolar concentrations, leading to a binomial distribution of isotopologues following Pascal's triangle. For this purpose, we grew *E. coli* in a minimal media with this acetate mixture as the sole carbon source of the media (**Figure 1**) and checked its isotopic content by NMR (molecular ^{13}C enrichment expected at $49.9\% \pm 0.8$). From cell harvest, two specific

reference ^{13}C -PT samples were produced during the extraction: i) a ^{13}C -PT full metabolite extract containing all polar metabolites for the evaluation of CID for organic acids; ii) a ^{13}C -PT protein hydrolysates extract containing proteogenic amino acids in high concentrations for the evaluation of CID for amino acids. This strategy allowed to maximize the amount of amino acids that can be weakly abundant as soluble pools when *E. coli* is grown in a minimal media. It should be mentioned that the hydrolysis process deamidated Asparagine into Aspartate and Glutamine into Glutamate, thus these amino acids could not be evaluated (Kambhampati et al., 2019). In addition, non-proteogenic amino acids from the **Table 1** could not be evaluated with this extract (Ornithine, β -Alanine).

To facilitate comparison of CIDs between the different amino acids, we converted the measured ^{13}C -enrichments among isotopologues of each fragment into binomial distribution, thus reflecting the number of isotopomers considered for each isotopologue (**Figure 2** and Material and Methods). Overall, all the considered amino acids were detected in the ^{13}C -PT protein hydrolysates except Tryptophan (**Figures 2A, S1A**). We found very a high level of correlation between the measured CID (blue bars) and the expected CID (black bars) for numerous amino acid fragments considered in our method: α -Alanine_2TMS_C2C3, Aspartate_C2C4 either in 2TMS or 3TMS, Glutamate_3TMS_C2C5, Glycine_3TMS_C2, Isoleucine_2TMS_C2C6, Leucine_2TMS_C2C6, Phenylalanine_2TMS_C1C2 and C2C9, Serine_2TMS_C2C3, Serine_3TMS_C1C2 and C2C3 and Tyrosine_2TMS_C1C2. CID measurements for Methionine_2TMS_C2C5 showed a high variability coupled to a lower estimation of M3 and M4 isotopologues at the expense of M1 and M2 (**Figure 2A**). However, this amino acid was in a very low abundance in the ^{13}C -PT protein hydrolysates compared to the others amino acids (**Figure S1A**). Lysine and Tyrosine_2TMS_C2C9 had also important variability for their CID measurements, leading to misestimations of their M3 isotopologues. The most important and significant errors for CID measurements of amino acids concerned Proline_2TMS (overestimation of M4), Threonine_3TMS (overestimation of M3 at the expense of M0 and M1), Valine_2TMS_C1C2 (overestimation of M2) and Valine_2TMS_C2C5 (overestimation of M3 at the expense of M1 and M2). Considering the recent establishment of a ^{13}C -positional isotopomer approach for Glutamate to analysis its C1 (Lima et al., 2021), we also analyzed the CID for the two low abundant fragments Glutamate_3TMS_C1 (CO₂-TMS, M0 at m/z 117) and Glutamate_3TMS_C1C5 fragment (CH₃ loss, M0 at m/z 348). However, they both showed large and significant misestimations of their CIDs, thus questioning the accuracy of this method (**Figure S1B**). The analysis of organic acids from ^{13}C -PT full metabolites extract showed very good CID for Citrate_2TMS_C1C5, Malate_3TMS_C2C4 and Fumarate_2TMS_C1C4 (**Figure 2B**). Citrate_4TMS_C1C6 and Malate_3TMS_C1C4 showed also relatively good CID but they had a higher variability compared to their respective C1C5 and C2C4 fragments. There was a very high overestimation of M0 in Succinate_2TMS_C1C4 at the expense of M2, M3 and M4 isotopologues, However, these misestimations have been previously attributed to contaminations inherent in the ^{13}C -PT full metabolites extract (Millard et al., 2014), thus preventing the analytical validation for this fragment. For Pyruvate_MEOXTMS_C1C3, the variability was too high to assess the CID measurements, perhaps reflecting its weak derivatization yield and low abundance in ^{13}C -PT full metabolites extract (**Figure S1A**). To further assess the distribution of measurements errors among isotopologues, we analyzed the CID accuracy, using threshold values of ± 0.05 (**Figure 3A**). Overall, the bias of measurement for each isotopologue of each metabolite was relatively low except for Methionine, Valine, Threonine, Succinate and Pyruvate. Interestingly, these very high levels of accuracy for CID measurements were translated into very high levels of accuracy for mean ^{13}C enrichment measurements, leading to very small shifts between expected (0.499 ± 0.08) and measured values (below 1% of error), except for Methionine, Valine, Threonine, Succinate and Pyruvate (**Figure 3B**). However, Malate_3TMS_C1C4, Lysine_2TMS_C2C6 and Serine_2TMS_C2C3 showed important variability for mean ^{13}C enrichment measurements although no strong biases in their CID measurements were previously observed. Conversely, mean ^{13}C

enrichment of Phenylalanine_2TMS_C1C2 and proline_2TMS_C2C5 were slightly overestimated and reflected the overestimation of their respective isotopologues M2 and M4. To ensure that the tested fragments are suitable for isotopic analyses of metabolites from plant samples, we analyzed their accuracy at natural ^{13}C abundance. For this purpose, we produced two types of plant samples (seedlings from *Arabidopsis* and leaves from *Brassica napus*) and we measured the CID for each fragment of our method to calculate the residual molecular ^{13}C enrichment (**Figure 3C**). This variable corresponded to the ^{13}C enrichment remaining after removing the contribution of natural abundance for all isotopes (a value of 0 is expected if no contamination occurs). Among the metabolites that were detected in the plant samples, most of the fragments for organic and amino acids showed relatively high accuracy at natural ^{13}C abundance, with most of the mean values of residual molecular ^{13}C enrichment ranging between 0.05 and 3% (**Figure 3C**). Interestingly, some discrepancies were observed between *Arabidopsis* seedlings and *B. napus* leaves, notably for the full carbon backbone fragments Citrate_4TMS_C1C6 and Malate_3TMS_C1C4, which were more accurately measured in *B. napus* at low enrichment levels (critical values close to 5% for *Arabidopsis*). Conversely, *Arabidopsis* seedlings showed slightly higher accuracies than *B. napus* leaves for AlphaAlanine_2TMS_C2C3, Aspartate_2TMS_C2C4, Fumarate_2TMS_C1C4 and Serine_3TMS fragments. Phenylalanine_2TMS fragments showed residual molecular ^{13}C enrichments close to 5% but only in plants samples. Surprisingly, Proline_2TMS_C2C5, Threonine_3TMS_C2C4 and Valine_2TMS_C2C5 showed relatively important residual molecular ^{13}C enrichments (nearly 10%) in both chemical standards and unlabeled plant extracts, suggesting the occurrence of analytical bias. A detailed survey showed that the isotopic cluster of Proline_2TMS_C2C5 was contaminated in the M4 isotopologue (m/z 146) by a mass fragment deriving from near elution of Isoleucine_2TMS, and correlated with previous measurements of CIDs (**Figures 2, 3D**). Similarly, the isotopic cluster of Valine_2TMS_C2C5 was contaminated in the M3 isotopologue (m/z 147) while Threonine_3TMS_C2C4 showed a surprising very high abundance of the M1 isotopologue (40% of the CID) in both chemical standards and unlabeled plant extracts.

Assessment of the light/dark regulation of plant TCA cycle using a ^{13}C -MFA approach

To further validate our MS method, we applied it to ^{13}C -labeled plant samples. For this purpose, we performed transient kinetics of U- ^{13}C -pyruvate incorporation to *B. napus* leaf discs subjected to either light or dark conditions for up to 6 hours. The aim of this experiment was to obtain an overview of the light/dark regulation of metabolic fluxes associated to TCA cycle and PEPc using MS methodology and isotopologue-based approaches. We used U- ^{13}C -pyruvate as a probe since the mitochondrial Pyruvate DeHydrogenase (PDH) will be able to produce simultaneously $^{13}\text{C}_2$ -Acetyl-CoA molecules suitable for the investigation of the TCA cycle and $^{13}\text{CO}_2$ molecules suitable for the investigation of PEPc activity through the combined action of carbonic anhydrase ($\text{H}^{13}\text{CO}_3^-$).

Following our experiments, the analysis of mean ^{13}C enrichment for all conditions revealed a significant increase for α -Alanine, Aspartate, Citrate, Glutamate, Glycine, Malate, Pyruvate, Serine and Succinate (**Figure 4A**). There was a very high biological variability and small ^{13}C enrichment for Fumarate_2TMS_C1C4. Surprisingly, these enrichment dynamics remained relatively equivalent between light and dark conditions for all the fragments considered, except for Glycine and Serine, which were almost exclusively enriched in light conditions. Considering the importance of the light-dependent photorespiratory pathway for glycine and serine production in leaves, this result clearly showed that a significant part of $^{13}\text{CO}_2$ arising from PDH and TCA cycle decarboxylations (isocitrate dehydrogenase and 2-oxoglutarate dehydrogenase) was reassimilated by photosynthesis. Overall, Glycine_3TMS_C2, Serine_3TMS_C1C2 and Serine_3TMS_C2C3 reached a pseudo-isotopic steady-state after 2 hours of illumination. Other fragments showed a relatively linear evolution of their mean

¹³C enrichment, except for Pyruvate_MEOXTMS_C1C3 which rapidly reached an isotopic steady-state after 0.5 hours only. Interestingly, labeling data from Pyruvate_MEOXTMS_C1C3 and Succinate_2TMS_C1C4 fragments had a very good resolution (weak analytical and biological variability) compared to previous evaluation with ¹³C-PT full metabolites extract. Next, we performed quantification of metabolite pool sizes along the experiment to determine whether the considered metabolic network was in a metabolic steady-state (**Figure 4B**). For this purpose, we used GC-FID for organic acids and HPLC-UV for amino acids using AccQTag methodology as they have a very high level of stability for absolute quantification. Indeed, TMS-derivatives of amino acids are highly unstable along a GC-MS run, leading to drastic differences of response when quantifying them with external standards curves (Data not shown). In addition, the reliable absolute quantification of metabolites using their CID masses can only be achieved by using metabolite-specific isotopic standards (not our case in the present study) (Heuillet et al., 2020; Evers et al., 2021). While metabolite pool sizes were relatively different between them, they all reached a remarkable pseudo steady-state starting at 0-0.5 hours, except for Succinate and Serine where this state was reached at 2 hours. Nevertheless, the range of pool size variation for these two metabolites was below $\pm 2 \mu\text{mol.g}^{-1}\text{DW}$. Glycine was only quantified in light conditions (not detected in “dark condition” samples with HPLC-UV). To summarize, we have found some similar ¹³C enrichment dynamics for TCA cycle derived metabolites in both light and dark conditions combined to similar metabolite pool sizes at a metabolic steady-state.

Next, we focused our analysis on the isotopologue dynamics of these metabolites in order to depict the considered metabolic network and compare metabolic fluxes associated to TCA cycle and PEPc (**Figure 5**). Mean ¹³C enrichment of Citrate, Glutamate and Succinate was essentially driven by an increase of the M2 isotopologue for light and dark conditions, thus reflecting the incorporation of M2 Acetyl-CoA arising from U-¹³C-pyruvate. However, there was also an increase of M1 and M3 isotopologues for Glutamate and Succinate after 2 hours while M1, M3 and M4 isotopologues of Citrate_4TMS_C1C6 and Citrate_2TMS_C1C5 were increased starting from 4 hours. M1 and M4 of Citrate_4TMS_C1C6 were essentially increased in dark conditions compared to light conditions after 4 hours. There was a very high variability for Fumarate_2TMS_C2C4 CID in both conditions, which prevented its use for a reliable isotopic investigation. Interestingly, Malate_3TMS_C1C4 fragment showed a higher enrichment of M1 compared to M2 in the light condition while M2 remains higher than M1 in the dark condition. This shift may reflect an increase of the PEPc activity under light condition (M1 arising from the reassimilation of the mitochondrial ¹³CO₂ decarboxylations). Since there was no difference of mean ¹³C enrichment between light and dark conditions for this fragment, this result reinforced the importance of analyzing specifically CIDs for the interpretation of labeling data. In addition, the CID of Malate_3TMS_C2C4 showed a similar dynamic for both light and dark conditions, hence ruling out the use of this fragment to disentangle PEPc and TCA cycle contribution to Malate production. Aspartate_3TMS_C2C4 showed a very high variability for the light condition, but the CID dynamic in the dark condition was similar to the CID dynamic of Malate_3TMS_C2C4. Thus, both fragments seemed to mimic Oxaloacetate labeling dynamics. Pyruvate_MEOXTMS_C1C3 was only enriched with the M3 isotopologue (nearly 80% M3/20% M0), thereby suggesting that the activity of the malic enzyme, converting malate to pyruvate, was negligible in our metabolic network.

From these observations, we build a simplified metabolic network for TCA cycle composed of two “turns” to explain the production of M1 and M3 in both Succinate_2TMS_C1C4 and Glutamate_2TMS_C2C5 (**Figure 6A**). In this network, the second “turn” reflected the commitment of a molecule that has already experienced all reactions of the TCA cycle to another round of reactions (Orange and blue labels to compare acetyl-CoA-derived labeling from 1st and 2nd “turn”). Based on previous works (Abadie and Tcherkez, 2019a), the reverse reaction of the fumarase was also considered

in the network, although it has no effect on the M1 isotopologue monitored by mass spectrometry (C4 vs C1 and C4 isotopomers of M1 isotopologue). The involvement of a light-dependent stored ^{12}C -Citrate pool was not considered here since it only provided M0 isotopologues. From this metabolic network, we proposed to evaluate PEPc-dependent and TCA cycle-dependent contributions to Citrate production through M1, M2, M3 and M4 isotopologues and to Malate production through M1, M2 and M3 isotopologues. These contributions were only calculated using timepoints 4 and 6 hours, because all the fragments to consider were mostly visible at these moments. The main idea of this pathway fractional contribution (FC) to a given metabolite production rely on the use of pathway-specific mass isotopomer ratios for the metabolite, as long as the metabolic network is in a metabolic and isotopic pseudo steady-state. Here, we observed a rapid metabolic steady-state and a relatively negligible evolution of the relative proportions between the considered isotopologues (comparison of M1, M2 and M3 only for Malate for example) between 4 and 6 hours, *i.e.* the relative contribution of PEPc and TCA cycle to either Malate and Citrate production remains in the same order of magnitude between T4 and T6 hours. An important issue to address for calculating FC concerned the M1 from Malate_2TMS_C1C4, which monitored at the same time PEPc and TCA cycle dependent isotopomers. To deal with, we took advantage of Succinate_2TMS_C1C4 and Glutamate_2TMS_C1C4 fragments which both harbored M1 and M3 isotopologues that are the precursor of TCA cycle-dependent M1 and M3 of Malate_2TMS_C1C4 (**Figure 5**). Analysis of the M1/(M1+M3) ratio for these two fragments showed a remarkable stability between dark and light conditions, thus reflecting a stable M2 enrichment of 40% for the active pool of acetyl-CoA (**Figure 6 B** and Equations). Using this ratio, we calculated the FC of PEPc and TCA cycle for malate and citrate respectively (See Equations). We found that TCA cycle had a significantly four-fold higher contribution to either Malate or Citrate production compared to PEPc in all conditions (around 80% versus 20%). Interestingly, there was a 1.6-fold increase of the FC of PEPc to Malate production from dark to light conditions (14% versus 23%), although it remained not statistically significant (high variability, t-test p-value =0.08). Conversely, PEPc contribution to Citrate production was significantly 1.5-fold higher in dark condition than in light condition (26% versus 17%). Thus, the results suggested a higher commitment of PEPc-derived carbons to the forward part of the TCA cycle in the dark, while commitment of PEPc-derived carbons to the reverse part of the TCA cycle (Malate and perhaps Fumarate) was reduced at the same time.

Finally, we explored the isotopologue distribution of Glycine and Serine, some amino acids which were previously identified as a proxy for photosynthetic reassimilation of PDH and TCA cycle-derived $^{13}\text{CO}_2$ (**Figure 7A**). Our results showed that the increase of mean ^{13}C enrichment for the three considered fragments in light condition arisen from an increase of their M1 isotopologue. Given that the light-dependent photorespiratory enzymes glycine decarboxylase and serine:hydroxymethyl aminotransferase convert 2 glycine into 1 serine, we wanted to know whether the fragments Serine_2TMS_C1C2 and Serine_2TMS_C2C3 were labeled in the same or different carbons, thus revealing if the C1 of Glycine was also labeled (**Figure 7B**). By reviewing all the possible scenario with different combinations of labeled glycine, we showed that the CID associated to our fragments (isotopologue occurrence) could only be produced with one molecule of glycine labeled at the C2 and an unlabeled molecule of glycine, leading to the production of a molecule of Serine labeled only at the C2 and an unlabeled CO_2 molecule. Since we have shown that Glycine and Serine were both at the isotopic and metabolic steady-state between 2 and 6 hours, we used Serine fragments to evaluate the contribution of PDH and TCA cycle-derived CO_2 to the photosynthetic activity. For this purpose, the steady-state enrichment of Acetyl-CoA calculated before was used as a proxy of the $^{12}\text{C}/^{13}\text{C}$ ratio of the CO_2 produced by pyruvate dehydrogenase since a significant proportion of PDH-derived ^{12}C could still contribute to photosynthesis (See equations), and given its major contributor for light respiration (CO_2 emissions) (Tcherkez et al., 2005). After recalculating the mean ^{13}C -enrichment of SerineC1C3

from both Serine fragments, we showed that the light-dependent assimilation of PDH and TCA cycle decarboxylations represented around 5% of photosynthesis (**Figure 7C**).

4 DISCUSSION

The estimation of metabolic fluxes in photosynthetic organisms represents an important challenge that has gained interest over the last decade with the development of ^{13}C -INST-MFA. This approach requires a high level of accuracy for CID measurements to avoid propagation of small measurement errors to large errors in estimated fluxes (Antoniewicz et al., 2006). Considering the need to perform high throughput measurements at isotopically non-stationary steady-state, MS has been preferentially selected by the field in the recent years to measure CID in organic and amino acids. However, the suitability of TMS-derivative molecular ions for these metabolites has never been evaluated at the isotopologue level, *i.e.* by monitoring the impact of each isotopologue to the final CID measurement. Conversely, a significant part of our knowledge about the functioning and the regulation of plant TCA cycle is based on ^{13}C -labeling experiments with NMR detection of the ^{13}C signal. Thus, the use of a validated MS methodology to investigate ^{13}C -labeling dynamics within plant TCA cycle will definitely help to counterbalance our vision of the plant TCA cycle.

A robust methodology to evaluate CID measurements and identify analytical and matrix biases

In this study, we proposed a workflow to assess the measurement accuracy of CID from TMS-derivatives of organic and amino acids by GC-MS for applications in plants, based on tailor-made *E. coli* ^{13}C biological standards with controlled patterns of CID (^{13}C -PT) (**Figure 1**). From the TMS-derivatives (selected to maximize information on carbon backbones), we showed that most of them had a remarkable accuracy, leading to very small bias in the measurement of CIDs (below 5%) and mean ^{13}C enrichments (below 1%). We further validated their suitability for application in plants by monitoring very low residual molecular ^{13}C -enrichments in Arabidopsis seedlings and *Brassica napus* leaves, although these accuracies could be variable between the two plant matrices (Citrate_4TMS_C1C6, Malate_3TMS_C1C4, α -Alanine_2TMS_C2C3, Aspartate_2TMS_C2C4 and Serine_3TMS). However, among amino acid TMS-derivatives, we identified three important biases at the isotopologue level. The first bias concerned the contamination of the isotopic cluster of Proline_2TMS_C2C5 at the m/z 146 (**Figure 3D**). A previous study reported this contaminant using chemical standards at natural ^{13}C abundance, but no explanation was provided for this bias (Antoniewicz et al., 2006). Here, we showed that the m/z 146 arisen from the co-elution with Isoleucine, *i.e.* a separation by 0.02 seconds in our chromatograms, equivalent to 0.33 °C with the ramp time for temperature during the elution (+10°C.min⁻¹). Nevertheless, this contamination only led to a slight overestimation of the M4 isotopologue (^{13}C -PT samples), which did not strongly impact the accuracy measurement for mean ^{13}C enrichment (**Figure 3B**). Thus, this fragment may still be suitable for ^{13}C -MFA in plants, as long as the ^{13}C -labeling strategy for the metabolic network is adapted to prevents the production of M4 isotopologues. The second bias identified concerned the contamination of the isotopic cluster of Valine_2TMS_C2C5 at the m/z 147, which was observed in both chemical standards and plant matrices. Interestingly, when a siliconium ion (loss of a methyl in a TMS group) operates a cyclization by attacking another TMS group of the same molecule, it produces an ion $[(\text{CH}_3)_3\text{Si-O-Si}(\text{CH}_3)_2]^+$ of the same mass 147 (Harvey and Vouros, 2020). We also detected a significant abundance of an m/z 246 in the fragmentation spectra of Valine_2TMS, which seems to support this hypothesis for the contamination (**Figure S2**). The third bias concerned the very high level of M1 isotopologue for Threonine_3TMS_C2C4 detected in both chemical standards and plant matrices. In its fragmentation spectra, we found an additional M1 isotopologue contamination for the peak m/z 291, an important fragment for the identification of Threonine_3TMS by GC-MS (**Figure**

S2, NIST database). Given that we are working at a nominal mass resolution, we cannot determine whether this contamination is due to: i) a co-elution of a contaminant chemically similar to Threonine; ii) a strong isotopic effect during the derivatization. The use of GC coupled to high-resolution MS may help to resolve this analytical issue (instrument not available at present in our platform). Besides this, we observed a high level of M0 contamination for CID measurement of Succinate using the ^{13}C -PT full metabolites extract, that has been reported previously (Millard et al., 2014), but prevented its accurate evaluation with our workflow. Nevertheless, there was a small biological and analytical variability for Succinate_2TMS_C1C4 in our ^{13}C -labeling experiments and we measured similar ratios for M1 and M3 with Glutamate_2TMS_C2C5 that was already validated by our method (**Figure 6B**). Since M1 and M3 were both impacted by the M0 contamination in ^{13}C -PT samples, our results suggest that this fragment can be reliably used for ^{13}C -MFA.

Using our validated method, we provided important information about the role of to the activity of the TCA cycle that contrasted with some proposed flux modes for TCA cycle in illuminated leaves (Tcherkez et al., 2009; Gauthier et al., 2010; Sweetlove et al., 2010; Tcherkez et al., 2017; da Fonseca-Pereira et al., 2021).

Stored Citrate is unlikely to contribute to TCA cycle activity

The hypothesis about the role of night-stored citrate molecules relies on two pioneer works. In the first work, after 50 min of $^{13}\text{CO}_2$ incorporation in *Xanthium stomarium* leaves, ^{13}C signal was analyzed by NMR and led the authors to propose “transmission coefficients” for metabolic reactions based on ^{13}C redistribution towards metabolites normalized to the $^{13}\text{CO}_2$ assimilated (Tcherkez et al., 2009). To ensure the fitting with their experimental data, the authors had to introduce the contribution of an important pool of unlabeled Citrate. The second work used the following experimental set-up ($^{13}\text{CO}_2$ 6h light/ $^{12}\text{CO}_2$ 6h dark/ $^{12}\text{CO}_2$ 6h light, with $^{15}\text{NH}_4^{15}\text{NO}_3$ available all the time) with *B. napus* leaves and a petiole-feeding assay with (Gauthier et al., 2010). Analysis of the results by NMR showed a weak ^{13}C -enrichment for Citrate, Glutamate, Glutamine and Malate at the end of the first light period and 30 min after the beginning of the dark period compared to the end of the experiment. Combined to a higher occurrence of ^{12}C - ^{15}N interactions compared to ^{13}C - ^{15}N interactions in Glutamate after 6h of light with $^{13}\text{CO}_2$, the results were interpreted as evidences for the contribution of night-stored vacuolar Citrate and Malate molecules to TCA cycle activity in the subsequent light period. Here, using U- ^{13}C -pyruvate labeling experiment on *B. napus* with a leaf disc-feeding assay, we provided evidences for the absence of Stored Citrate contribution to the TCA cycle in a light/dark dependent manner in our conditions (**Figure 8**). Notably, we showed that the M2/(M0+M2) ratio of the Acetyl-CoA active pool was close to 0.4 after 4 to 6 hours in our experiments, thereby showing that most of M0 isotopologues from metabolite active pools arisen from M0 Acetyl-CoA and M0 OAA and not Citrate (**Figure 6B**). Given the important production of PEPc-derived molecules with non-labeled PEP in our experiments, ^{12}C -Pyruvate must be produced by the glycolysis and dilute the mitochondrial U- ^{13}C -Pyruvate pool (nearly 20% of M0 detected in Pyruvate). It is also possible that thioredoxin-mediated activation of the ATP citrate lyase activity in light conditions leads to the conversion of stored ^{12}C -Citrate pool into ^{12}C -OAA and ^{12}C -Acetyl-CoA (Sweetlove et al., 2010; da Fonseca-Pereira et al., 2021). However, we showed that the dynamic of ^{13}C -labeling for Citrate, Glutamate and Succinate was not diluted by light conditions compared to dark conditions while the system was in a metabolic steady-state (**Figure 4**). The remobilization of ^{12}C -stored Citrate in the light was expected to introduce a ^{12}C isotopic dilution to Glutamate and Succinate compared to the dark, if this contribution was relatively important (Gauthier et al., 2010), but this was not observed here. Interestingly, the ^{13}C -enrichment of Acetyl-CoA was calculated from ^{13}C -labeled fragments of either Succinate or Glutamate (independent of ^{12}C -molecules), thus this measurement cannot be biased with the contribution of stored ^{12}C -Citrate. Besides

this, we did not observe any significant modification of the Citrate pool during the 6 hours of our experiment while a strong decrease in light conditions and a strong increase in dark conditions were expected with the hypothesis of the stored citrate remobilization (**Figure 4B**). Since the previous pioneer experiments performed with $^{13}\text{CO}_2$ and not U- ^{13}C -pyruvate and used different methods for the detection of ^{13}C , one might think that it is difficult to reconcile them. However, a ^{13}C -MFA performed in Arabidopsis rosette leaves with $^{13}\text{CO}_2$ showed that Pyruvate was rapidly labeled and reached a pseudo-isotopic state at nearly 40% of mean ^{13}C enrichment after 1 hour, similarly to the ^{13}C -enrichment of Acetyl-CoA we calculated between 4 and 6 hours (Szecowka et al., 2013). In addition, another ^{13}C -MFA performed in tobacco mature leaves showed that after 5 hours of $^{13}\text{CO}_2$ incorporation, only 20% of the starch produced was ^{13}C -labeled while ADP-Glucose pool was ^{13}C -labeled by up to 88% (Chu et al., 2022). Thus, a high proportion of non-labeled carbon is still expected to enter the glycolysis and the TCA cycle during the first hours of $^{13}\text{CO}_2$ incorporation, without necessarily implying the contribution of stored Citrate. Given the similar dynamic of ^{13}C -enrichment for Glutamate in both light and dark conditions in our experiments, the higher commitment of old ^{12}C than photosynthetic ^{13}C towards Glutamate biosynthesis previously observed in the light seems to be essentially attributed to both: i) an isotopic dilution of newly assimilated ^{13}C with pre-existing ^{12}C metabolic pools; ii) a relatively slow turn-over of TCA cycle metabolic intermediates independently of the light/dark condition. In addition, Glutamate is produced from 2-Oxoglutarate and Glutamine, thus stored ^{12}C carbons are also incorporated at this step from Glutamine. Overall, our present findings confirmed that stored Citrate is unlikely to contribute to plant TCA cycle in either light or dark conditions (**Figure 8**). Since most of the Citrate and Malate pools are found in the vacuole independently of the leaf developmental stage (Szecowka et al., 2013; Destailleur et al., 2021), their remobilization may be expected only under stress conditions.

The forward cyclic flux mode of the plant TCA cycle majorly contributes to Malate biosynthesis

To date, non-cyclic nature of the TCA cycle in illuminated leaves has been proposed based on few studies that used NMR to interpret ^{13}C signals in terms of ^{13}C - ^{13}C interactions (Sweetlove et al., 2010; Abadie et al., 2017). One branch is supposed to be fed essentially by PEPc (OAA, Malate and Fumarate) while the other branch is only fed by (Citrate, 2-Oxoglutarate and Succinate) and the conversion of Succinate to Fumarate is almost non-existent. In addition, it has also been proposed that 80% of Glutamate molecules in the light arisen from PEPc-derived carbons (Abadie et al., 2017). However, one evidence is based on the absence of ^{13}C enrichment detected in the Succinate compared to Fumarate with a 2-hours $^{13}\text{C}_1$ -pyruvate incorporation while both metabolites showed similar ^{13}C -enrichments with a 2-hours U- ^{13}C -Pyruvate (Tcherkez et al., 2009). Unfortunately, the reassimilation of $^{13}\text{CO}_2$ from $^{13}\text{C}_1$ -Pyruvate decarboxylation by PEPc cannot be metabolically transmitted from Malate to Succinate since the C1 and the C4 of the OAA backbone are lost during Citrate to Succinate conversion (**Figure 6A**) (Alves et al., 2015). Another evidence is based on “transmission coefficient” reflecting the ^{13}C -redistribution towards TCA cycle metabolites after 50 min of $^{13}\text{CO}_2$ labeling (Tcherkez et al., 2009). The authors suggested that PEPc provided carbons essentially to OAA, Malate and Fumarate, while stored Citrate supplied most of the carbon precursors for Glutamate (lower PDH activity compared to PEPc). To assess the flux modes of the TCA Cycle, we took advantage of pathway-specific mass isotopologues/isotopomers and a ^{13}C -MFA approach to investigate the TCA cycle and PEPc metabolic fluxes after a U- ^{13}C -Pyruvate incorporation in *B. napus* leaf discs. First, we showed that Glutamate and Succinate rapidly harbored M2 isotopologues, corresponding to *de novo* incorporation of M2 Acetyl-CoA from the TCA Cycle, followed by an increase of M1 and M3 isotopologues over the time, corresponding to M2 isotopologues that had experienced a complete turn of the forward cyclic mode of the TCA cycle (**Figure 5**). A brief comparison shows that the (M1+M3)/M2 isotopologues ratio is close to 0.5 after 6 hours of labeling for both Glutamate and

Succinate (**Figure 5**). Since M2 isotopologues can be produced by both M0 OAA fixing M2 Acetyl-CoA and M2 OAA fixing M0 acetyl-CoA, the commitment of M2 isotopologue to the forward TCA cycle is necessarily higher than this ratio expected. Using a ^{13}C -MFA approach combined to pathway specific MS fragments and the indirect measurement of M2/(M0+M2) ratio for the Acetyl-CoA active pool, we showed that the forward part of the TCA cycle contributed to nearly 80% of both Malate and Citrate production compared to PEPc (4:1 activity ratio), while malic enzyme activity was not significantly detected (**Figure 6**). Interestingly, this modelling approach integrate the M0 isotopologues into the metabolic pools without the need to measure them, *i.e.* a potential contribution of any additional ^{12}C pool can still be compatible. However, this pathway fractional contribution assumes that most of the CO_2 is produced by the PDH, and its conversion to HCO_3^- are not rate-limiting for PEPc activity. While the second hypothesis may not be relevant given that CO_2 was actively reassimilated by photosynthesis in our light conditions (**Figure 7**), the second hypothesis can be questioned given the metabolic network (**Figure 6**). Indeed, isocitrate and 2-oxoglutarate dehydrogenases necessarily decarboxylate more $^{12}\text{CO}_2$ than $^{13}\text{CO}_2$ compared to the PDH, which may imbalance the reassimilation of $^{13}\text{CO}_2$ by PEPc if the first part of the TCA cycle operates at a high rate. Assuming an equivalent metabolic flux from PDH to 2oxoglutarate dehydrogenase, the overall mitochondrial $^{12}\text{CO}_2/^{13}\text{CO}_2$ ratio should not exceed a 2:1 dilution compared to the $^{12}\text{CO}_2/^{13}\text{CO}_2$ produced by the PDH, based on the isotopologue distribution of Citrate, Glutamate and Succinate and their associated CO_2 losses. Thus, the forward and cyclic mode of the TCA cycle will still majorly contribute to Malate and Citrate compared to PEPc (2:1 ratio with this scenario) (**Figure 8**). Interestingly, a ^{13}C -MFA performed at isotopic and metabolic steady-state in heterotrophic suspension cells culture under non-limiting glucose concentrations showed nearly similar ratio of 2:1 for forward TCA cycle/PEPc metabolic fluxes, with a negligible malic enzyme activity (10-fold less than PEPc) (Masakapalli et al., 2010). Considering Arabidopsis leaves, a ^{13}C -MFA was performed recently using a 1-hour kinetic of $^{13}\text{CO}_2$ incorporation and suggested that PEPc activity was 1.5 fold higher than the TCA cycle flux, with fumarase and malate dehydrogenase being inactive (value below 0.1 compared to PEPc at $1.2 \mu\text{mol.g}^{-1} \text{FW.h}^{-1}$) (Xu et al., 2021). However, such conclusions were drawn with questionable modelling, *i.e.* M1, M2 and M3 isotopologues of Citrate were already present at high levels when starting the experiment and metabolic flux variations were only recorded as the production of M4 isotopologue (unrealistic metabolic network compared to other work (Szecowka et al., 2013)). Based on ^{13}C -positionnaly accumulated in Malate after a $^{13}\text{CO}_2$ incorporation, another study proposed an important flux for PEPc at high photorespiration rates in sunflower (Abadie and Tcherkez, 2019a), a condition which may happen in our experiments using leaf disc floating on a solution. However, the conclusion for PEPc flux were also drawn with questionable modeling approach: i) calculations were made without the dynamic of ^{13}C -enrichment for both the substrate and the product (mandatory (Buescher et al., 2015; Millard et al., 2020) ; ii) the estimated metabolic fluxes associated to the use of HCO_3^- or PEP were largely different while both metabolites are the substrates of the same reaction (same metabolic flux); iii) the metabolic steady-state of the system was not considered (accumulation of Malate is unlikely to occur). To date, the accurate estimation of absolute PEPc metabolic flux remains challenging in plants, as it is difficult to routinely get access to ^{13}C -enrichments for HCO_3^- and PEP by either NMR or GC-MS. Here, we showed that the anaplerotic contribution of PEPc to the mitochondrial metabolism remains lower than the contribution of the TCA cycle, perhaps reflecting the constraints imposed by PEPc-derived Malate transport from cytosol to mitochondria.

Besides this, the forward cyclic mode of the TCA cycle has to be reconnected to the relative ^{13}C content found in Glutamate and Aspartate, which can represent important escapes for ^{13}C that will drive the relative differences of metabolic flux for the biosynthesis of Glutamate and Succinate, and Malate and OAA respectively (branching points) (**Figure 5**). However, a significant proportion of Glutamate and Aspartate may come back to the TCA cycle. In astrocyte cells, U- ^{13}C -Glutamate labeling

incorporations showed that Glutamate M5 was rapidly metabolized into Glutamate M3 after one turn of the TCA cycle (Nissen et al., 2015) thereby proposing a similar role for plant mitochondrial Glutamate dehydrogenase (Terce-Laforgue et al., 2013). The potential role of GABA (Gamma-aminobutyric acid) shunt has been proposed to reroute Glutamate towards Succinate (Tcherkez et al., 2009). Finally, mitochondrial transaminases such as Alanine:2-oxoglutarate and Aspartate:2-oxoglutarate could also participate in this process. Since Glutamate is necessarily produced from ^{13}C -2-Oxoglutarate and ^{12}C -Glutamine molecules in our metabolic network, Glutamate rerouting pathways are supposed to introduce a 50% isotopic dilution between ^{12}C and ^{13}C molecules for mitochondrial active pools of either Succinate or 2-Oxoglutarate. But it is not possible for instance to have access to the size of the mitochondrial active pool for both metabolites and by extension their exact ^{13}C -enrichments without a complete isotopic steady-state of the plant TCA cycle. Given the fractional contributions of PEPc and TCA cycle calculated in this work for Malate and Citrate, a higher metabolic flux for PDH (^{13}C -Acetyl-CoA) than 2-Oxoglutarate dehydrogenase (^{13}C -Succinate) is expected and necessarily constraints the PEPc flux to Malate (**Figure 8**). To which extent Glutamate and Aspartate molecules escape the TCA cycle remains to be determined. Overall, this forward cyclic flux mode of the TCA cycle in both light and dark conditions strongly supports its key roles in *de novo* nitrogen assimilation and mitochondrial ATP production (through Succinate dehydrogenase activity) (Sweetlove et al., 2010).

Light/dark regulation of PEPc and TCA cycle metabolic fluxes: the potential roles of glycolysis and Malate import into mitochondria

We showed that Malate_2TMS_C1C4 fragment experienced a strong shift in its M1/M2 isotopologues in light conditions compared to dark conditions, which was associated to a relatively higher contribution of PEPc to Malate production in light condition than in dark condition. Such shift could be attributed to both increased PEPc activity and TCA cycle inhibition in the light compared to dark conditions. Interestingly, PEPc activity is positively regulated by its light-dependent thioredoxin-mediated phosphorylation in C_3 and C_4 plants (Caburatan and Park, 2021). In the light, the lower respiration rate measured compared to dark conditions has been partly attributed to an inhibition of PDH and TCA cycle-derived decarboxylations (Tcherkez et al., 2005; Xu et al., 2021). In our experiments, it was not possible to compare absolute fluxes of TCA cycle in light and dark conditions, due to the relative nature of the fractional contribution measurements (total of 100% in each condition). However, we found very similar dynamics of ^{13}C -enrichment for TCA cycle metabolites combined to similar metabolic pool sizes in a metabolic steady-state (**Figure 5**). Thus, variations of TCA cycle metabolic fluxes between light and dark conditions are expected to be very low in our experimental set-up, and subsequently the variation of light/dark respiration. It is tempting to think that our experimental set-up badly reflects plant physiology and its light/dark associated regulation (leaf discs floating on a solution). However, by integrating a ^{13}C -positionnal isotopomer approach to the photorespiratory metabolism, we showed that PDH and TCA-cycle derived CO_2 were actively reassimilated by the photosynthesis, leading to a contribution of around 5% to net CO_2 assimilation (**Figure 7**). Under the light irradiance of around $120 \mu\text{mol photons.m}^{-2}.\text{s}^{-1}$, net CO_2 uptake of *B. napus* leaves at a similar developmental stage (L11-L7) is close to $5 \mu\text{mol CO}_2. \text{m}^{-2}.\text{s}^{-1}$ (Dellero et al., 2021), leading to a reassimilation rate of $0.25 \mu\text{mol CO}_2. \text{m}^{-2}.\text{s}^{-1}$. Since the dark respiration rate for these leaves is close to $1.25 \mu\text{mol CO}_2. \text{m}^{-2}.\text{s}^{-1}$ (Dellero et al., 2021), and given the proposed inhibition of day respiration by nearly 50% in plants (Tcherkez et al., 2005), this means that the current measurement/estimation of day respiration rate through CO_2 mass balance at the leaf level (Tcherkez et al., 2017; Berghuijs et al., 2019) is still underestimated by nearly 40% (photosynthetic reassimilation not taken into account because not evaluated). Therefore, the inhibition of day respiration at the metabolic flux level is likely to be lower than measured by gas-exchanges (**Figure 8**). For instance, the

very similar dynamics of ^{13}C -enrichment observed in our experiments for light and dark conditions can be explained by two major hypotheses: i) a relatively higher control of TCA cycle activity by glycolysis rather than by pyruvate dehydrogenase, given that starch degradation and glycolysis are essentially inhibited by light conditions (Plaxton and Podestá, 2006;Thalmann and Santelia, 2017); ii) a specific activation of pyruvate dehydrogenase in both conditions, given that we incorporated a strong amount of U- ^{13}C -pyruvate into leaf discs (Tovar-Mendez et al., 2003). Nevertheless, the shift in the fractional contribution of PEPc to malate production between dark and light conditions still confirms the robustness of our method to infer reliable information about the light/dark regulation of metabolic fluxes. Besides this, we observed another unexpected but interesting result: the higher contribution of PEPc derived-carbons to Citrate production in the dark condition compared to the light condition. Since we have shown that PEPc contribution to Malate production was reduced in dark compared to light conditions, this result can only be explained by a higher transport of OAA/Malate from the cytosol to the mitochondria in dark conditions (**Figure 8**). This idea is supported by the fact that PEPc is majorly localized to the cytosol (Caburatan and Park, 2021) and the recent identification of a mitochondrial malate importer in Arabidopsis with a decisive impact on mitochondrial respiration (Lee et al., 2021). Whether a higher malate import into mitochondria allows to increase the absolute metabolic flux of the TCA cycle in dark conditions in our experimental set-up remains to be elucidated.

Conclusions

Overall, we proposed a robust workflow to evaluate the accuracy of CID measurements by GC-MS for organic and amino acids and successfully applied our validated method to the investigation of plant TCA cycle and PEPc metabolic fluxes. Notably, our results do not support the hypothesis of the stored citrate contribution to TCA cycle and demonstrated that the forward and cyclic flux mode of the TCA cycle largely contributed to Malate and Citrate production compared to PEPc (4:1 ratio). Thus, the activity of the succinate dehydrogenase in both dark and light conditions has been largely underestimated until now. Reassimilation of PDH and TCA-cycle derived CO_2 by photosynthesis represented nearly 5% of the net CO_2 assimilation rate in our conditions, and this should be considered when day respiration rate is measured by CO_2 mass balance at the leaf level. Although the regulation of the PEPc activity by light was still operating, the higher involvement of PEPc-derived carbons to citrate production in dark conditions proposed an important role for Malate import into mitochondria in sustaining TCA cycle in dark conditions. The next stage of this work should be focused on the accurate modeling of this metabolic network to infer absolute metabolic fluxes. This perspective sounds stimulating given the development of multiple modeling approaches over the last decade but still remains highly challenging (Kruger and Ratcliffe, 2021).

5 Conflict of Interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

6 Author Contributions

Conceptualization, Y.D., A.B.; methodology, Y.D.; software, Y.D., O.F.; validation, Y.D., S.B. and C.B.; formal analysis, Y.D.; investigation, Y.D., S.B. and C.B.; data curation, Y.D.; writing-original draft preparation, Y.D.; writing-review and editing, Y.D., A.B.; visualization, Y.D.; supervision, Y.D.; project administration, Y.D.; funding acquisition, Y.D., A.B. All authors have read and agreed to the published version of the manuscript.

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9 Supplementary Material

Supplementary Material is attached to the preprint.

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Table 1. Molecular ions selected for Carbon Isotopologue Distribution analysis of organic and amino acid TMS-derivatives by GC-MS.

Compounds	TMS derivative		Loss		Molecular ion		
	Formula	mass	Formula	mass	Formula	m/z	Carbon skeleton
<i>Organic acids</i>							
Citrate(4TMS)	C18H40O7Si4	480	CH3	15	C17H37O7Si4	465	1-2-3-4-5-6
Citrate(4TMS)	C18H40O7Si4	480	TMS-CO2, TMS-OH	207	C11H21O4Si2	273	1-2-3-4-5
Fumarate(2TMS)	C10H20O4Si2	260	CH3	15	C9H17O4Si2	245	1-2-3-4
Malate(3TMS)	C13H30O5Si3	350	CH3	15	C12H27O5Si3	335	1-2-3-4
Malate(3TMS)	C13H30O5Si3	350	TMS-CO2	117	C9H21O3Si2	233	2-3-4
Pyruvate(MEOX)(TMS)	C7H15NO3Si	189	CH3	15	C6H12NO3Si	174	1-2-3
Succinate(2TMS)	C10H22O4Si2	262	CH3	15	C9H19O4Si2	247	1-2-3-4
<i>Amino acids</i>							
Asparagine(2TMS)	C10H24N2O3Si2	276	TMS-CO2	117	C6H15N2OSi	159	2-3-4
Asparagine(3TMS)	C13H32N2O3Si3	348	TMS-CO2	117	C9H23N2OSi2	231	2-3-4
Aspartate(2TMS)	C10H23NO4Si2	277	TMS-CO2	117	C6H14NO2Si	160	2-3-4
Aspartate(3TMS)	C13H31NO4Si3	349	TMS-CO2	117	C9H22NO2Si2	232	2-3-4
Glutamate(3TMS)	C14H33NO4Si3	363	TMS-CO2	117	C10H24NO2Si2	246	2-3-4-5
Glutamine(3TMS)	C14H34N2O3Si3	362	TMS-CO2	117	C10H25N2OSi2	245	2-3-4-5
Glycine(3TMS)	C11H29NO2Si3	291	TMS-CO2	117	C7H20NSi2	174	2
Isoleucine(2TMS)	C12H29NO2Si2	275	TMS-CO2	117	C8H20NSi	158	2-3-4-5-6
Leucine(2TMS)	C12H29NO2Si2	275	TMS-CO2	117	C8H20NSi	158	2-3-4-5-6
Lysine(4TMS)	C18H46N2O2Si4	434	TMS-CO2	117	C14H37N2Si3	317	2-3-4-5-6
Methionine(2TMS)	C11H27NO2SSi2	293	TMS-CO2	117	C7H18NSSi	176	2-3-4-5
Ornithine(4TMS)	C17H44N2O2Si4	420	none		none		1-2-3-4-5
Phenylalanine(2TMS)	C15H27NO2Si2	309	TMS-CO2	117	C11H18NSi	192	2-3-4-5-6-7-8-9
Phenylalanine(2TMS)	C15H27NO2Si2	309	C7H7	91	C8H20NO2Si2	218	1-2
Proline(2TMS)	C11H25NO2Si2	259	TMS-CO2	117	C7H16NSi	142	2-3-4-5
Serine(2TMS)	C9H23NO3Si2	249	TMS-CO2	117	C5H14NOSi	132	2-3
Serine(3TMS)	C12H31NO3Si3	321	TMS-CO2	117	C8H22NOSi2	204	2-3
Serine(3TMS)	C12H31NO3Si3	321	TMS-CH2O	103	C8H20NO2Si2	218	1-2
Threonine(3TMS)	C13H33NO3Si3	335	TMS-CO2	117	C9H24NOSi2	218	2-3-4
Tryptophan(2TMS)	C17H28N2O2Si2	348	2TMS-C2H2NO2	218	C9H8N	130	3-4-5-6-7-8-9-10-11
Tryptophan(2TMS)	C17H28N2O2Si2	348	C9H8N	130	C8H20NO2Si2	218	1-2
Tryptophan(3TMS)	C20H36N2O2Si3	420	2TMS-C2H2NO2	218	C12H16NSi	202	3-4-5-6-7-8-9-10-11
Tryptophan(3TMS)	C20H36N2O2Si3	420	C12H16NSi	202	C8H20NO2Si2	218	1-2
Tyrosine(3TMS)	C18H35NO3Si3	397	C10H15OSi	179	C8H20NO2Si2	218	1-2
Tyrosine(3TMS)	C18H35NO3Si3	397	TMS-CO2	117	C14H26NOSi2	280	2-3-4-5-6-7-8-9
Valine(2TMS)	C11H27NO2Si2	261	TMS-CO2	117	C7H18NSi	144	2-3-4-5
Valine(2TMS)	C11H27NO2Si2	261	C3H7	43	C9H20NO2Si2	218	1-2
α -Alanine(2TMS)	C9H23NO2Si2	233	TMS-CO2	117	C5H14NSi	116	2-3
β -Alanine(3TMS)	C12H31NO2Si3	305	CH3	15	C11H28NO2Si3	290	1-2-3

Figure legends

Figure 1. Workflow used in this study.

Figure 2. Evaluation of Carbon Isotopologue Distribution (CID) measurements for organic and amino acids by GC-MS using ^{13}C -PT reference samples. A, Evaluation of amino acids using ^{13}C -PT protein hydrolysates. B, Evaluation of organic acids using ^{13}C -PT full metabolites extract. Black bars correspond to expected CIDs and blue bars to the measured CIDs. The results are presented as the mean $\pm$ SD of four independent biological replicates.

Figure 3. Evaluation of the measurement accuracy for CID and mean ^{13}C enrichment at ^{13}C -PT and ^{13}C natural abundance in plants. A, CID accuracy and B, Mean ^{13}C enrichment for the selected TMS-derivatives of organic and amino acids using ^{13}C -PT samples. C, Residual mean ^{13}C enrichment measured in chemical standards, Arabidopsis seedlings and *Brassica napus* leaves after correction for isotope natural abundance. D, Isotopologue-specific biases identified at ^{13}C natural abundance for Proline, Valine and Threonine fragments. The results are presented as the mean $\pm$ SD of four independent biological replicates or independent preparations for commercial chemical standards.

Figure 4. Dynamics of U- ^{13}C -pyruvate incorporation in *B. napus* leaves exposed to light or dark conditions. A, Mean fractional ^{13}C enrichment and B, Metabolite pool sizes. Mean fractional ^{13}C enrichment were monitored with GC-MS, while absolute quantification of organic and amino acids was performed respectively with GC-FID and HPLC-UV. The results are presented as the mean $\pm$ SD of four independent biological replicates.

Figure 5. Fractional ^{13}C enrichment at the isotopologue level for TCA-cycle derived metabolites. A, Light conditions and B, Dark conditions. The M0 isotopologue was not represented to facilitate the visualization of low isotopologue enrichments. The results are presented as the mean $\pm$ SD of four independent biological replicates.

Figure 6. Fractional contribution of PEPc and TCA cycle to Malate and Citrate production in light and dark conditions. A, Network reconstruction at the isotopomer level based on the isotopologues detected. B, Fractional contribution of PEPc and TCA cycle to Malate and Citrate production based on pathway-specific mass isotopologues/isotopomers. In the proposed network, the second "turn" reflected the commitment of a molecule that has already experienced all reactions of the TCA cycle to another round of reactions (Orange and blue labels to compare acetyl-CoA-derived labeling from 1st and 2nd "turn"). Note that M0 isotopologues of all metabolites will still continue to produce M0 isotopologues at a significant proportion after a real turn of the TCA cycle, given the ^{13}C -enrichment for Acetyl-CoA active pool. For these calculations, only the timepoint T4 and T6 hours were used and pooled. The results are presented as the mean $\pm$ SD of four independent biological replicates. Statistical comparison of Acetyl-CoA enrichment for light and dark conditions was performed with a Student-test, while fractional contributions were compared with an ANOVA followed by a post-hoc Tukey-HSD test (letters indicate groups that were separated with a p-value <0.05).

Figure 7. Fractional contribution of PDH and TCA-cycle derived decarboxylations to photosynthesis. A, Fractional ^{13}C enrichment at the isotopologue level for Glycine and Serine. B, Working hypotheses to explain the photorespiratory-dependent labeling pattern of Glycine and Serine. C, Estimation of the fractional contribution of PDH and TCA-cycle CO_2 to photosynthesis with Serine (expressed in %). For these calculations, only the timepoint T4 and T6 hours were used and pooled (metabolic and isotopic steady-state requirements, assumption with $^{13}\text{C}/^{12}\text{C}$ of CO_2 released). The

1073 results are presented as the mean $\pm$ SD of four independent biological replicates. Statistical comparison
1074 of the two Serine fragments was performed with a Student-test.

1075 **Figure 8. Proposed metabolic fluxes for plant TCA cycle and PEPc in light and dark conditions**
1076 **based on this study.** The contribution of stored citrate to is unlikely to occur based on our results,
1077 while the forward TCA cycle has a higher contribution to both malate and citrate production than PEPc.
1078 Questions remains about the exact proportion of Glutamate and Aspartate carbon backbones escaping
1079 the TCA cycle. Despite a reduction of PEPc activity in dark conditions compared to light conditions,
1080 PEPc-derived carbons show a higher commitment to Citrate production in dark conditions due to a
1081 higher rate of Malate import into mitochondria. Reassimilation of PDH and TCA-cycle derived CO₂
1082 by photosynthesis may represent up to 40% of the TCA cycle metabolic flux. See Discussion for
1083 detailed explanations. TCA, Tricarboxylic acids; PEPc, Phosphoenolpyruvate carboxylase; OAA,
1084 Oxaloacetate; GDH, Glutamate dehydrogenase; GABA, Gamma-aminobutyric acid; ALAAT,
1085 Alanine:2-oxoglutarate aminotransferase; ASAT, Aspartate:2-oxoglutarate aminotransferase.

Identification of TMS-derivative mass fragments maximizing carbon backbones

Production of reference labelled samples

Validation of TMS-derivative mass fragments for CID measurements

A

B

A **B** **C** **D**

A

Mean fractional ^{13}C enrichment

B

Metabolite pool sizes

A **B**

anova p-value = e-16

A **B**

			H 1	H 2	H 3	H 4	H 5	H 6
Glycine	C2	M1						
Serine	C1C2	M1						
		M2						
	C2C3	M1						
		M2						

C

Contribution of PDH and TCA cycle derived CO_2 to photosynthesis

t-test *p*-value = 0.07